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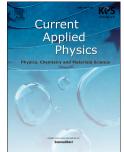
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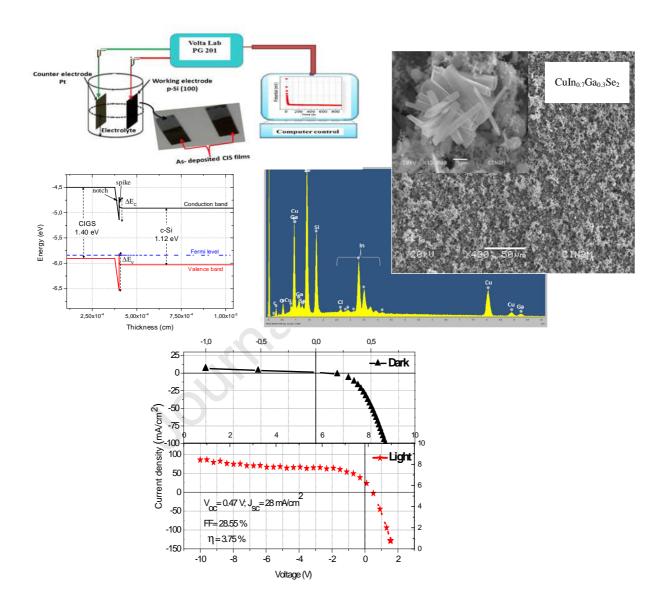
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Graphical abstract



Physical properties of electrodeposited CIGS films on crystalline silicon: Application for photovoltaic heterojunction

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Abstract

P-type CIGS (CuIn_{1-x}Ga_xSe₂) thin films are electro-deposited on a p-type c-Si substrate with a galvanostatic mode to form CIGS(p)/c-Si(p) hetero-junction. The Ga content is varied up to x = 30%. The physical properties of formed CIGS films are characterized by XRD, SEM, EDS and UV-Visible spectroscopy. With x = 30%, we obtain a single chalcopyrite phase of CIGS with a tetragonal crystal structure, a high crystallinity, an orientation toward the (112) direction and a band gap energy of 1.40 eV.

AM1.5 J-V performed on the $CuI_{0.7}G_{0.3}Se_2/c$ -Si hetero-junction reveals interesting photovoltaic parameters with an efficiency of 3.75%. In addition, using the energy diagram of the hetero-junction calculated with the Anderson model, we show that it could play a dual role when combined to a c-Si cell in a Ag-Al/c-Si(n⁺)/c-Si(p)/CIGS(p)/Al new architecture. Therefore, in addition to its interesting photovoltaic parameters, this hetero-junction can substitute the BSF.

Keywords: CIGS; Silicon, Electro-deposition, Hetero-junction.

1. Introduction

Nowadays, photovoltaic (PV) energy plays a key role in renewable energy (RE) scenarios. The conversion efficiency of silicon solar cells is being 25.8 % at a laboratory scale and a stability of 20 years [1]. Improving solar cell efficiencies while simultaneously lowering its cost is an essential challenging aim in R&D (Research and Development) in photovoltaic. In the last few years, thin film solar cells based on polycrystalline chalcopyrite materials (I-III-VI₂), have received great deal of attention due to their versatile optical and electrical properties [2]. Copper Indium Gallium Diselenide Cu(In,Ga)Se₂ (CIGS) with a chalcopyritetype structure has proven to be one of the prominent candidates to compete with the efficiency and stability of conventional crystalline silicon cells, because of its high light absorption coefficient, lower materials cost and great stability [3-4]. Additionally, CIGS has a direct band gap which can be varied from 1.02 to 1.60 eV by increasing the Ga proportions ratio from 0 to 1 [5]. Further, solar cells based on CIGS thin films have the highest efficiencies among second generation photovoltaic devices with a record laboratory scale efficiency reaching about 22.6%) [1]. Many efforts have been invested to obtain a low cost CIGS absorber layer by different approaches including chemical and physical methods [6-8]. Electrochemical deposition techniques of CIGS thin films have been extensively investigated [9]. Especially, the one-step electro-deposition method (ECD) is competitive, economic and convenient. Moreover, this approach has many advantages such as; high deposition speed, low processing cost, no vacuum system required and the use of a large area of substrate at room temperature [10]. The cell efficiencies based on CIGS thin films fabricated by the ECD process are close to 7% for 30x30 cm² surface area and about 10 % for laboratory-scale device modules [11].

Bifacial solar cells can use direct and diffuse sunlight to create electric energy from both front and back surfaces of the cell. This kind of device provides high-efficiency cell design and it is competent of generating more electrical power per cell by absorbing photons from both sides of the device [12]. In a previous work [13] we reported on the physical properties of CIS/c-Si hetero-junction with a conversion efficiency of ~ 0.41%. Nevertheless, the efficiency of such cell can be improved with good reproducibility by widening the band gap to an optimal value of 1.5eV that could be adjusted through Ga, Al and/or S incorporation in the CIS matrix.

The partial substitution of In^{3+} by Ga^{3+} ions is important in increasing the band gap to values between 1.04 eV of CuInSe₂ and 1.68 eV of CuGaSe₂ and to reach enhanced conversion

efficiency [14]. In other hand, it is necessary to optimize the atomic ratio $x = \frac{Ga}{In+Ga}$ in order

to obtain high quality of fabricated layers. The effect of Ga content on the properties of chalcopyrite $CuIn_{1-x}Ga_xSe_2$ thin films have been investigated by many authors [15]. The best CIGS solar cell efficiency was obtained with a 30% Ga-content.

The main emphasis and purpose of this work is the investigation of Ga content on the physical properties of the CIGS film fabricated on crystalline silicon substrate. Furthermore, these physical properties were used to analyze and discuss the electrical performances of fabricated CIGS(p)/c-Si(p) hetero-junction solar cell deduced from AM1.5 current density – voltage characterization.

2. Experimental details

CuIn_{1-x}Ga_xSe₂ thin films with x ratio ranging from 0.0 to 0.3 were electro-deposited on monocrystalline silicon substrates using a Volta Lab potentiostat/galvanostat (PGP 201) connected to a two-electrode cell. We used boron doped p-type monocrystalline silicon wafers with a (100) crystal orientation, a thickness of $525 \pm 25\mu$ m and a resistivity in the 1-2 Ω .cm range. The set-up contains the silicon substrate as the working electrode and a platinum

counter electrode. The electro-deposition mixture is formed by 3 Mm of CuCl₂, 3 Mm of InCl₂, 0.3 - 0.9 Mm of Ga₂O₃, 8 Mm of SeO₂ and 0.1 M of sodium citrate (complexing agent) in de-ionized (DI) water. The pH of the bath has been adjusted to 1.65 via the addition of hydrochloric acid (HCl). During the electro-deposition, the solution was stirred and a constant current density of 10 mA.cm⁻² was applied during a deposition time of 10 minutes at room temperature.

Before experiment, the silicon wafers were ultrasonically cleaned with acetone, ethanol and DI water. Then, they are dipped into dilute hydrofluoric acid (HF, 5%) solution for 1 minute in order to remove the native SiO₂ insulating film to permit the electro-deposition of CIGS on the silicon substrates. After electro-deposition, all samples were annealed in a nitrogen atmosphere at optimized temperature of 350 °C for 5 min by an infrared Rapid Thermal Processing (RTP). The RTP annealing permits a formation of an excellent crystallized CIGS film [13-16]. Hall effect characterization at ambient temperature (300 K) confirms the p-type nature of CIGS films. Thus, for the CuIn_{0.7}Ga_{0.3}Se₂ film, we obtained a concentration of majority carriers of 3.55 10^{16} cm⁻³, a conductivity of 3.73 Ω^{-1} .cm⁻¹ and a mobility of 656 cm².V⁻¹.s⁻¹.

The structural properties of elaborated CIGS thin films were studied by means of X-ray diffraction (XRD) analysis using an automated Bruker D8 advanced diffractometer with a CuK_{α} ($\lambda = 1.540$ Å) radiation. Surface morphology of the samples was characterized by a Scanning Electron Microscope (SEM) JEOL JSM-5910, with an accelerating voltage of 20 kV. Energy Dispersive Spectroscopy (EDS), as well as the UV–Visible-NIR LAMBDA 950 spectrophotometer were used to determine chemical compositions and optical properties, respectively. Thermal evaporation instrument was further used to deposit a Back and grid front aluminum (Al) contacts. Then, samples were heated in argon atmosphere using RTP furnace at specific annealing temperatures during 15 min (i.e.; 630 °C for the back and 250 C°

for the grid front Al contacts). AM1.5 short current-voltage characterization of the fabricated photovoltaic cell was measured by a digital source meter (Keithley Instruments Inc., Model 2400) in the dark and under AM1.5 (100 mW.cm⁻²) illumination.

3. Results and discussion

We plot in Fig. 1 XRD patterns of $CuIn_{1-x}Ga_xSe_2$ films with different gallium percentages. For undoped sample (x = 0), in addition to the Si (100) diffraction peaks, we notice three main diffraction peaks at approximately 26.56°, 44.26° and 52.40° assigned respectively to (112), (204) (220) and (116) (312) plans of the well-identified CIS chalcopyrite phase (Ref. JCPDS 40-1487). After Ga doping, we observe a high 20 shift of CIS peaks. This shift indicates the incorporation of Ga in CIS matrix and hence a formation of the quaternary CuIn_{1-x}Ga_xSe₂ which is consistent with the standard reported values for CIGS (JCPDS NO. 03-065-7027). This result was demonstrated by several researchers [17]. Furthermore, the crystal phase of CuIn_{1-x}Ga_xSe₂ films is affected by varying the gallium content. In fact, the (112) peak intensity decreases clearly with increasing gallium concentration especially for the specimens obtained at $0.1 \le x < 0.3$, so a noticeable increase in the full width at half maximum FWHM (table.1) can be noticed, indicating a low crystallization of films. Whereas, increasing Ga content to 30% XRD pattern shows the formation of a single chalcopyrite tetragonal structure with a strong and preferential orientation along the (112) direction. This provides that Ga in higher concentration improves crystallization. In order to confirm this result, we investigate the effect of gallium composition on the preferential orientation in fabricated films. As reported in Fig. 2, the (112) peak relative intensity has a maximum value when the substitution rate is 30%.

The low intensity of (112) peaks for samples synthesized with low percentage of gallium suggests their worse crystallinity, which may be attributed to the large compositional

deviation from the stoichiometry and/or to structural disorder introduced in the chalcopyrite crystal lattice by inhomogeneous gallium incorporation.

Improved crystallinity was checked by the change in crystallite sizes which were calculated from the main peak intensity by the Scherrer formula expressed by Eq. (1) [18]:

$$D = \frac{k\lambda}{\beta\cos\theta} \tag{1}$$

Where λ is the X-ray wavelength of CuK α line ($\lambda = 1.540$ Å), θ is the Bragg angle and β is the values of the FWHM of the (112) peak.

According to calculated values of grain size given in table 1, we guess that the average crystallite size is changed by varying the relative amounts of gallium. In addition, the largest crystallite with mean grain diameter of about 82.50 nm is obtained for the CIGS thin film with high x ratio. This result confirms the XRD analysis which shows an improvement in crystallinity at this corresponding ratio.

As mentioned above, Ga incorporation causes a shift of the diffraction peaks to higher Bragg angles, in particular the highest peak (112), which causes a change of the lattice parameters a and c. Indeed, the variation of the two lattice parameters a and c with gallium composition given in figure 3 is found to be approximately linear as the latter is increased from zero (CuInSe₂) to one (CuGaSe₂), which is in agreement with Vegard law [19]. Therefore, CIGS films prepared using one step electro-deposition process is very close to the perfect tetragonal structure for which the lattice distortion is $\mu = c/a \simeq 2$) [20]. The variation of the lattice constants as a function of x in CIGS thin films is given by the following two equations:

$$a_{Culn_{(1-x)}Ga_xSe_2} = 5.78\text{\AA} - 0.17\text{\AA}x$$
(2)

$$c_{CuIn_{(1-x)}Ga_xSe_2} = 11.61\text{\AA} - 0.57\text{\AA}x$$
(3)

These expressions are in good agreement with those reported in the literature, e.g. a = 5.783Å-0.171 Å and c = 11.618Å- 0.586 Å x [21].

As shown in Fig. 3 and table.1, the lattice parameters a and c of all CIGS films were found to decrease when the percentage of gallium rises. This reduction could be explained by the small ionic radius of Ga (r(Ga) = 1.81Å) compared to In (r(In) = 2Å), an effect which was already known and widely studied in the literature [22].

The micro-strain (ϵ) developed in CIGS thin films is calculated using the following relation: [23]

$$\varepsilon = \frac{\beta}{4\tan\theta} \tag{4}$$

As reported in table 1, values of ε decreases from 0.275 % for sample synthesized with low percentage of Ga (x= 0.1) to 0.177 % for deposited film with high gallium rate (x =0.3). This is attributed to a reduction in defect centers in the structure and the improvement in sample crystallinity. Therefore, a Ga composition of 0.3 appears beneficial to obtain interesting structural quality of such films.

The surface morphology of deposited CIGS layers is depicted in Fig. 4. Different morphologies can be observed prior and after Ga doping. In fact, micrograph (Fig. 4a) reveals a homogeneous surface with a uniform and dense distribution of aggregates composed of small rounded grains. For compounds prepared with low gallium content (Figs. 4b and 4c), the surface morphology is deteriorated displaying grains/particles with uneven sizes which leads to non-homogeneous surface. However, a more consistent variation in morphology is observed in insert of Fig. 4d when the Ga content is 30%. The surface of the film appears to be more homogenous and formed by aggregates consisting of larger size crystals having rod-

like shapes. The length of the latter varies from 1 to 3 μ m. Therefore, the crystallinity of the CIGS films is noticeably improved with increasing Ga content, especially at 30%. In addition, the grain size became larger, which is consistent with XRD analysis.

Fig. 5 shows a typical EDS spectrum for $CuIn_{1-x}Ga_xSe_2$ films with x = 0% and x = 30%. The chemical composition of CIS (x = 0%) layer characterized by EDS (Fig. 5a) reveals the presence of copper, indium and selenium elements. The compositional analysis of this CIS film is summarized in table 2. Otherwise, as shown in Fig. 5b, in addition to copper, indium and selenium elements, we note the appearance of new peaks assigned to the gallium element. This confirms the Ga incorporation in CIS film and the formation of a single quaternary $CuIn_{0.7}Ga_{0.3}Se_2$ phase that has been already mentioned by XRD studies. As well as that, the presence of other elements such as residual oxygen and carbon may be due to interface and surface contamination however silicon is attributed to the substrate.

The compositional analysis of $CuIn_{1-x}Ga_xSe_2$ films with various Ga content is summarized in table 2. We notice that specimens synthesized with 10% and 20% of Ga contents are not stoichiometric and the x ratio is less than one. While, increasing gallium content above 30%, the elementary composition tends to improve in quaternary CIGS thin film and the best stoichiometry was achieved.

The optical properties (absorption coefficient, band gap energy...) of quaternary CIGS thin films depend on the relative of component element and especially on the x ratio. In addition, the partial substitution of In^{3+} by Ga^{3+} ions leads to increase the band gap energy (for heterostructure with monocrystalline silicon) and hence to improve the opto-electrical properties of materials [24].

From the diffuse reflection spectra R (λ), we have assessed the absorption coefficient of Cu (In_{1-x}Ga_x) Se₂ layers using the Kubelka–Munk function F(R), determinate with respect to reflectivity of dense films (R = R_{∞}) [15].

$$F(R) = \frac{(1-R)^2}{2R}$$

(5)

A plot of $[F(R) hv]^2$ vs. hv as illustrated in Fig. 6 is extrapolated to be a straight line portion of the absorption edge whose intercepts the energy axis giving the optical band gap of the CIGS films. As a result, we obtain a shift of the absorption edges towards higher energies when the Ga composition increases. Hence, we found that value of the band gap energy rises from 1.01 eV to 1.40 eV when gallium proportion rises from 0 to 30 % (see inset graph). Similarly, Cao Jie et al. [25] obtained a value of 1.38 eV at the same Ga content. However, we notice that in other works [26], reported values are below 1.2 eV. In fact, the increase in direct band gap with change of gallium rate is studied by many investigators [27-28]. According to these researches, authors reported two main causes. The first reason might be linked to differences between the electro-negativity of the In⁺³ and Ga³⁺ ions which leads to broadening of valence/conduction bands inducing the increase of the gap. The second one is related to an important distortion of the tetragonal lattice in films generated by Ga- addition. In the present study, we attribute the increase of the optical band gap to two major effects; the improvement in crystallinity and the reduction of defects like strain and dislocations density, which provide less contribution beneath the absorption edge in this corresponding substitution rate. This result is confirmed by XRD analysis. Furthermore, the increase of optical band gap to 1.4 eV improves the open circuit voltage Voc of the CIGS/c-Si heterojunction by increasing the difference between the quasi Fermi levels during illumination.

As previously highlighted in micro-structural and optical analysis, the sample with 30 % Ga contents exhibits a better crystallization and has a more propitious band gap compared to our other fabricated films. To achieve the fabrication of the based hetero-junction solar cell, we thermally evaporate aluminum to form back and front (grid) contacts to perform dark and

AM1.5 illumination characterizations. To a better understanding of such characterizations, we simulate the energy diagram of the CIGS(p)/c-Si(p) hetero-junction using the Anderson model. Result is plot in figure 7(a). We notice that in these calculations, the Anderson model uses the electron affinity and band gap values of both semiconductors to calculate offsets of conduction and valence bands (ΔE_C and ΔE_V). As depicted in figure 7(a), we observe a "spike" and "notch" at the interface. These simulation were performed with x = 30%. We notice that since N_A(Si) is in the round of 10⁺¹⁵ cm⁻³, and obtained N_A(CIGS) from Hall effect measurements is 3.55 10⁺¹⁶ cm⁻³, this hetero-junction can play a double role when it is used in c-Si bifacial solar cells (Ag-Al/c-Si(n⁺)/c-Si(p)/CIGS(p)/Al). Hence, in addition to the enhancement in the photovoltaic characteristics of the cell, this heterojunction can substitute the costly annealing step of the Al contact to form the BSF (Back Surface Field).

Typical J–V characteristics of this cell under dark and AM1.5 illumination are plot in Fig. 7(b). We obtain an open circuit voltage V_{OC} of 470 mV and a short circuit-current density J_{SC} of 28 mA.cm⁻². The series R_S and shunt R_{Sh} resistances are 8.33 Ω and 1162 Ω , respectively. The FF is 28.5% and the conversion photovoltaic efficiency is 3.75%. These photovoltaic performances are close to those obtained for solar cells prepared by others cost-effective techniques [29-30], leading to conclude that synthesized CIGS films could be an interesting absorber material for bifacial solar cells.

4. Conclusion

Highly polycrystalline quaternary CuIn_{0.7}Ga_{0.3}Se₂ thin film is obtained using a relatively simple and convenient one-step electro-deposition upon galvanostatic mode on monocrystalline silicon substrate. We obtain interesting micro- structural and electrical properties of CIGS absorber layers following Ga incorporation into CIS.

The effect of Ga content on the physical properties of samples was investigated. At 30% of Ga content in the CIGS material, XRD shows the formation of the single chalcopyrite

structure phase with a preferential orientation along the (112) direction. The average crystallite size and the lattice parameters are greatly influenced by varying the gallium compositions from 0 to 30%. SEM and EDS analysis prove once again that at the 30% of Ga proportion, the CIGS layer has a sufficient surface homogeneity covered with agglomerated crystallites (rod-like grains) and nearly stoichiometric composition. From optical measurements, the band gap of $CuIn_{0.7}Ga_{0.3}Se_2$ is found to be 1.40 eV which is appropriate for photovoltaic application.

Using Hall effect measurements, we confirm that the CIGS film is of p-type. Furthermore, to a better understanding of the measured AM1.5 J-V characterization of the CIGS(p)/c-Si(p) hetero-junction, we calculate its energy diagram using the Anderson model. Results permit us to conclude that this hetero-junction can play a double role when it is used in a new architecture of c-Si bifacial solar cells (Ag-Al/c-Si(n⁺)/c-Si(p)/CIGS(p)/Al). Therefore, in addition to its interesting photovoltaic parameters (V_{oc}, J_{SC}, R_S, R_{Sh}, FF and η) as compared to other solar cells prepared by others cost-effective techniques, this hetero-junction can substitute the costly annealing step of the back Al contact when the BSF is formed during the fabrication of c-Si solar cell.

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Tables

Table 1: Structural characteristics of the quaternary alloy Cu (In_{1-x}, Ga_x) Se₂ thin films for different values of x, derived from XRD analysis.

Table 2: Elemental composition of EDS patterns of CIGS thin films at different Ga contents.

Table 1: Structural characteristics of the quaternary alloy Cu (In_{1-x}, Ga_x) Se₂ thin films for different values of x, derived from XRD analysis.

X	Bragg angle 2θ (°)	FWHM (°)	D (nm)	Lattice parameters (Å)		Distorsion	Strain (ε x10 ⁻²)
			2)	a	c	$\mu = c/a$	
0.0	26.50	0.121	67.30	5.78	11.61	2.007	0.224
0.1	26.75	0.150	53.82	5.76	11.55	2.005	0.275
0.2	26.95	0.152	53.15	5.74	11.50	2.003	0.277
0.3	27.15	0.098	82.50	5.73	11.47	2.001	0.177

Table 2: Elemental composition of EDS patterns of CIGS thin films at different Ga contents.

x (%)	Precursor composition (Molar ratio Cu: In: Ga: Se)	EDS estimated composition (% Atomic ratio Cu:In:Ga:Se)	Cu+In+Ga+Se =1 (Theoritecally 100%)				
0	1/1/0/2	26.14/22.83/0.00/51.03	1.04				
10	1/0.9/0.1/2	25.12/22.71/3.72/48.45	0.93				
20	1/0.8/0.2/2	24.30/20.33/5.80/49.57	0.98				
30	1/0.7/0.3/2	25.57/15.41/8.54/50.48	1.02				

Figures caption

Fig. 1: XRD patterns of Cu (In_{1-x}, Ga_x) Se₂ layers prepared by one-step electrodeposition onto silicon (Si (100)) substrate at different Ga contents.

Fig. 2: Variation of relative intensity of the main peak (112) as a function of $x = \frac{Ga}{In+Ga}$

ratio.

Fig. 3: The dependence of the lattice parameters a and c on the Ga contents.

Fig. 4: SEM images of top surface of electrodeposited Cu(In_{1-x}Ga_x) Se₂ layers on Si before and after Ga doping: (a) Undoped CIS film (x=0); (b), (c) and (d) doped samples at x = 0.1, x = 0.2 and x = 0.3, respectively.

Fig. 5: Typical EDS spectrum of Cu (In _{1-x}Ga_x) Se₂ thin films: (a) ternary CIS and (b)

quaternary Cu (In _{0.7}Ga_{0.3)} Se₂ films.

Fig. 6: Plot of $(F(R) hv)^2$ vs. (hv) using the Kubelka–Munk approximation for CIGS thin films deposited on silicon substrates with different Ga composition.

Fig. 7: (a): Simulated energy diagram of the CIGS(p)/c-Si(p) heterojunction using the Anderson model. (b): Dark and AM1.5 illumination J–V curves of the fabricated Al/ Cu (In_{0.7} Ga_{0.3)} Se₂/c-Si: p/Al hetero-junction solar cells.



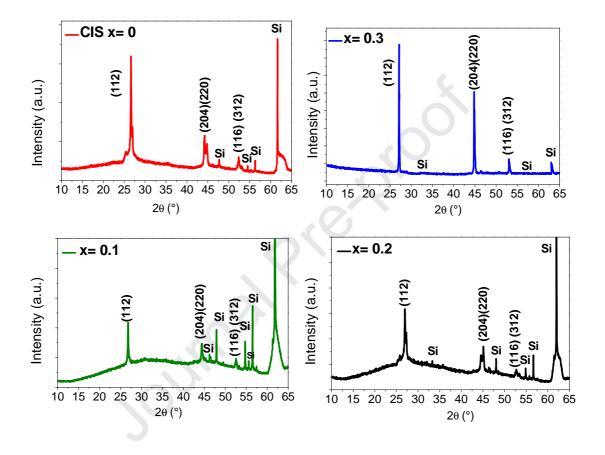


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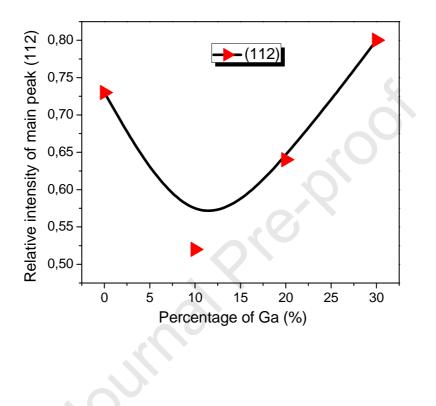


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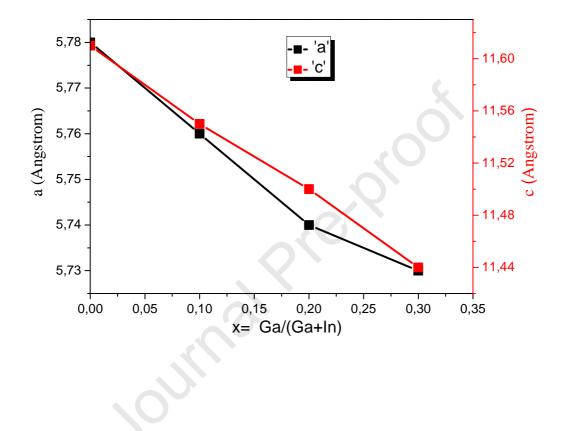


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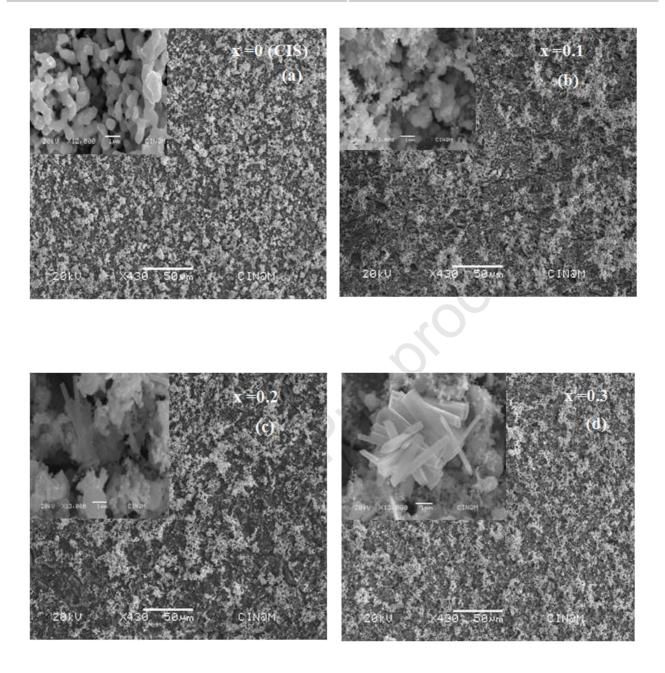
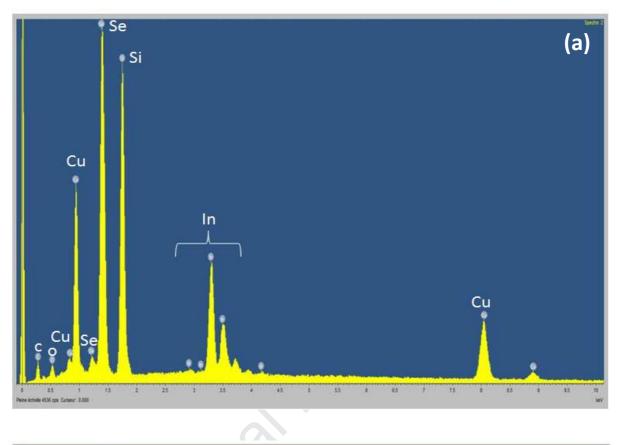


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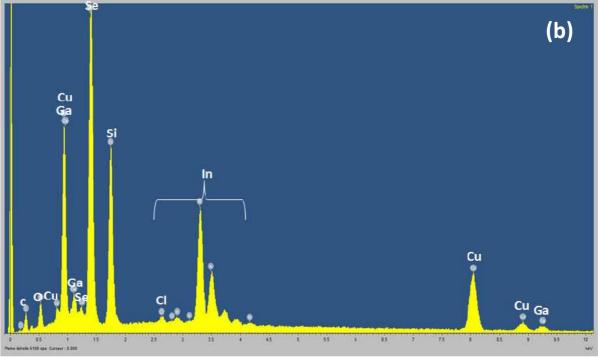


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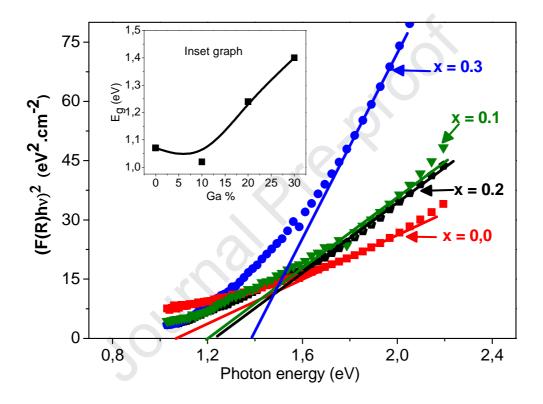


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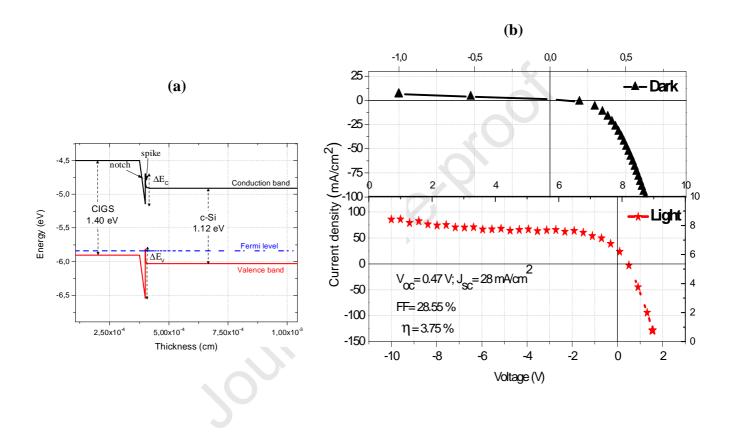


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Highligths

- Highly polycrystalline CIGS film is electro-deposited on c-Si to form CIGS/c-Si • hetero-junction.
- AM1.5 J-V performed on the CuI_{0.7}G_{0.3}Se₂/c-Si hetero-junction reveals interesting • photovoltaic parameters with an efficiency of 3.75%.
- Using the Anderson model, we show that the CIGS/c-Si hetero-junction could play a ٠ dual in a new architecture solar cell.