

Antimicrobial and anthelmintic activities of the ethanolic extract, fractions and isolated compounds from Manilkara zapota L. P. Royen (Sapotaceae)

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Antimicrobial and anthelmintic activities of the ethanolic extract, fractions, and isolated compounds from *Manilkara zapota* L. P. Royen (Sapotaceae)

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20 ABSTRACT

21

Objectives: To evaluate the antibacterial, antifungal, and anthelmintic activities of the ethanolic extract (EEMz), fractions (LPFMz and HPFMz), and compounds isolated from the leaves of *Manilkara zapota* L. P. Royen.

Methods: Extract and fractions were produced by turbolization. LPFMz fraction was analyzed by GC-MS. The isolated compounds from HPFMz were purified by flash and preparative chromatographic methods, and chemically characterized by UPLC-ESITOFMS, optical rotation, and one- and two-dimensional ¹H and ¹³C NMR techniques. Anthelmintic against *Strongyloides venezuelensis* and antimicrobial activities against *Candida albicans*, *Trichophyton rubrum* and *Staphylococcus aureus* were evaluated.

32 Key findings: EEMz showed mainly phenolic compounds and pentacyclic triterpenes 33 from $\Delta 12$ -oleane/ $\Delta 12$ -ursane series. Chlorogenic acid, myricetin-3-O-β-D-34 glucopyranoside, mearnsitrin, germanicol, and germanicol acetate were reported to M. 35 zapota leaves for the first time in this work. EEMz, HPFMz, LPFMz showed significative activity against C. albicans (16 µg/mL), while isolated flavonoids were 36 37 active against S. aureus (<32 µg/mL). EEMz, phenolic-rich compounds (F₂), and chlorogenic acid were potentially active against S. venezuelensis at 24 h. 38

39 Conclusions: *M. zapota* and its bioactive compound can be eligible such as a potential
40 phytomedicine for the treatment of microbial and strongyloidiasis drug-resistant
41 infections.

42 KEY-WORDS: Traditional medicine Meso- and Southern America; phytochemistry;
43 anthelmintic; antimicrobial; chromatography.

45	Highl	ights

- Phenolic and pentacyclic triterpenes from $\Delta 12$ -oleane/ $\Delta 12$ -ursane were detected
- 47 Chlorogenic acid, myricetin-3-O-β-D-glucopyranoside, mearnsitrin were
 48 identified
- 49 The pentacyclic triterpenes presented promising antimicrobial activity
- 50 Ethanolic extract and phenolic fractions showed high antifungal activity
- Ethanolic and phenolic fractions, and chlorogenic acid exhibited anthelmintic
 action
- 53 Abbreviation List
- 54 CAD: Charged Aerosol Detector
- 55 Compound 1: chlorogenic acid
- 56 Compound 2: myricetin-3-O-β-D-glucopyranoside
- 57 Compound 3: myricitrin
- 58 Compound 4: mearnsitrin
- 59 Compound 5: germanicol
- 60 Compound 6: β amyrin
- 61 Compound 7: α-amyrin
- 62 Compound 8: germanicol acetate
- 63 COSY: Correlation Spectroscopy
- 64 EEMz: dry ethanolic extract
- 65 ESI: Electrospray interface
- 66 F: fraction obtained from High polarity insoluble fraction
- 67 HMBC: Heteronuclear Multiple Bond Correlation
- 68 HPFMz: High polarity insoluble fraction
- 69 HRMS: High Resolution Mass Spectroscopy
- 70 HSQC: Heteronuclear Single Quantum Coherence
- 71 LPFMz: lower polarity fraction
- 72 Mz: Manilkara zapota
- 73 PVP: polyvinylpyrrolydone
- 74 TOF: Time-of-flight

76 1. INTRODUCTION

Infectious diseases caused by bacteria, viruses, fungi, and parasites are the greatest cause of death, disability, and social and economic difficulties for millions of people worldwide ^[1]. This problem is especially high in developing countries, due to poor availability of drugs and the emergence of widespread drug resistance ^[2].

Infections caused by microorganisms such as Staphylococcus aureus can promote 81 disorders in the skin and mucosa, toxic shock syndrome, endocarditis, osteomyelitis, 82 and infections of the urinary and intestinal tracts.^[3]. Additionally, dermatophyte fungi 83 are also able to invade the keratinized tissues of animals and humans ^[4]. Among them, 84 Trichophyton rubrum is most widespread, representing 80% of clinical isolates ^[5]. 85 Candida albicans is another important opportunistic pathogenic fungus that causes 86 candidiasis responsible for a variety of conditions in humans, ranging from superficial 87 cutaneous lesions to generalized infection^[6]. 88

89 In the same way, helminthiases caused by intestinal nematodes and transmitted via the 90 soil are the most common and persistent neglected parasitic infections, especially in 91 developing countries. Strongyloides stercoralis is the main neglected geohelminths responsible to infect 30-100 million people in the world ^[7, 8]. The autoinfection ability 92 93 of S. stercoralis leads to syndromes of persistent infection that can result in chronic 94 illness with various clinical manifestations. In immunocompromised patients, the 95 uncontrolled multiplication of the parasite (hyperinfection) and the potentially fatal 96 distribution of larvae in various organs represents a major concern in terms of public health ^[8; 9]. 97

98 Therefore, the discovery of new compounds with chemical structures and mechanisms 99 of action different from those current antimicrobial and anthelmintic drugs are urgently 100 need. This fact is due to the increased incidence of new and reemerging infectious 101 diseases, the ineffectiveness of treatments using broad-spectrum anthelmintics, and 102 especially the increasing inefficacy of the drugs employed in the current therapy against 103 pathogenic microorganisms and intestinal parasites ^[10; 11].

Manilkara zapota (L.) (Synonyms: *M. zapotilla*, *M. achras*, *Mimusops manilkara*, *Achras zapota*), commonly known as sapoti, is a member of the Sapotaceae family and it represents a potential natural resource to treat infectious process ^[12]. In folk medicine, the leaves are used to treat coughs, colds, and diarrhea, the crushed seeds are used as a

diuretic and febrifuge, and the husks have astringent and febrifuge activities ^[13]. In the 108 literature, *M. zapota* has been reported to exhibit antioxidant properties ^[14; 15], as well as 109 analgesic ^[16], antimicrobial ^[17; 18], antihyperglycemic, and hypocholesterolemic 110 activities ^[19; 20], due the presence of phenolic acids, flavonoids, tannins (mainly in the 111 immature fruit), triterpenes, and saponins (mainly in seeds) compounds ^[19; 21-23]. Despite 112 these important properties, the bioguided phytochemical study of the compounds from 113 114 the *M. zapota* leaves has not yet been developed. In this sense, the goal of this work was to characterize the phytochemistry of the ethanolic extract, fractions, and isolated 115 116 compounds from the Manilkara zapota L. P. Royen leaves and to evaluate their 117 antimicrobial action and, for the first time, their anthelmintic activity.

118

119 2. MATERIALS AND METHODS

120 2.1 Materials

Leaves of *Manilkara zapota* (Mz) were collected from the orchard of the Multidisciplinary Center for Chemical, Biological, and Agricultural Research (CPQBA) in Campinas, São Paulo State, Brazil (22° 47' 52" S, 47° 6' 49" W). A sample was deposited in the Herbarium of the Institute of Biology of the State University of Campinas, under registration no. UEC 186541.

Ivermectin and chlorogenic acid (95%) were puchased from Sigma-Aldrich (St. Louis, MO, USA) and used as a control drug and an analytical standard, respectively, in the anthelmintic activity assays. Polyvinylpyrrolidone (PVP-K30) was obtained from Synth (São Paulo, Brazil). Oxacillin, itraconazole, and fluconazole from Molekula (Dorset, UK) were used as control drugs in the antimicrobial activity assays. All solvents employed in the study were HPLC or UPLC grade.

132

133 **2.2. Bioactive compounds of the** *M. zapota*

134 **2.2.1. Ethanolic extract of the** *M. zapota*

The ethanolic extract was produced by turbolization. 900 g of the triturated and dried
leaves of *M. zapota* were dispersed in 3 L of ethanol under constant stirring at 16,000
rpm for 10 min using a mechanical disperser (Ultra Turrax T50, IKA Works Inc.,
Wilmington, NC, USA) at 25 °C. The suspension was filtrated in nylon filter to remove

non-soluble particles. Afterthat, the residue was re-extracted with 3 L of ethanol and
filtrated. The extracts were combined and the solvent was evaporated under vacuum
until complete dryness, resulting in the dry ethanolic extract (EEMz, 96.47 g).

142

143 **2.2.2. Fractions of the** *M. zapota*

30 g of the EEMz was suspended in 300 mL of hexane/ethyl acetate mixture (1:1) and extracted in the Ultra Turrax system for 5 min at 16,000 rpm. Lower polarity fraction (LPFMz) and the high polarity insoluble fraction (HPFMz) were obtained. These fractions were gently separated by centrifugation at 5,000 g for 15 min and posteriorly dried under vacuum in order to produce 16.25 g and 11.44 g of LPFMz and HPFMz, respectively.

150

151 **2.2.3. Isolation of the** *M. zapota* **compounds**

1.5 g of the HPFMz was purified by flash chromatography using a Combi Flash[®] Rf^+ 152 Teledyne ISCO system (Lincoln, NE, EUA) equipped with a UV detector and C18 153 154 column (86 g). A flow rate of 60 mL/min and H₂O/CH₃CN gradient of: 80/20 for 6 min 155 (F₁), 80/20 to 70/30 in 1 s, 70/30 for 6 min (F₂), 70/30 to 60/40 in 1 s, 60/40 for 6 min (F₃), 60/40 to 50/50 in 1 s, 50/50 for 6 min (F₄), 50/50 to 0/100 in 1 s, and 0/100 for 6 156 157 min (F₅), was used to obtain five diferents fractions. Subsequently, the fractions F₂ and F₃ were combined and purified by preparative HPLC system (Gilson, Middleton, WI, 158 159 USA) equipped with a model 322 pump, a GX-271 fraction collector, a DAD 171 diode 160 array detector, and a prepELSII electrospray accessory. A Phenomenex Luna 161 preparative C18 column (21.2 \times 250 mm, 5 μ m) (Torrance, CA, USA) was employed at a flow rate of 17 mL/min. The gradient apllied was H₂O/CH₃CN (98/2 to 77/23 in 5 162 163 min, 77/23 to 55/45 in 15 min, 55/45 to 0/100 in 5 min). Four compounds 1 (3.5 mg, Rt 164 2.14 min); 2 (0.7 mg, Rt 3.51 min); 3 (13.8 mg, Rt 4.14 min); 4 (4.6 mg, Rt 5.26 min) 165 were isolated and chracterized.

166

167 **2.3 Identification of the components by chromatography analyses**

168 **2.3.1 Thin layer chromatography**

169 Thin layer chromatography analyses were performed using aluminum plates coated with

170 silica gel 60 F₂₅₄ (Merck, Darmstadt, Germany). The lower polarity compounds were

elueted with hexane: ethyl acetate (70:30) and ethanolic anisaldehyde was used as stain.
For the high polarity compounds, the mobile phase was ethyl acetate: formic acid:
glacial acetic acid: water (52:11:11:26) and a solution of diphenylboryloxyethylamine
and polyethyleneglycol (NP/PEG) was used as the developer. UV detection was
performed at 254 and 366 nm.

176

177 2.3.2 Gas chromatography-mass spectrometry (GC-MS)

178 A gas chromatograph (Agilent, model 6890, Santa Clara, CA, USA) coupled to an 179 Agilent 5975 mass spectrometer was used to identify compounds present in LPFMz. 180 The analyses were performed using an HP-5MS capillary column (30 m \times 0.25 mm, 181 0.25 µm) (J&W Scientific, Santa Clara, CA, USA). The column temperature was 182 increased from 110 °C to 280 °C during 26 min, using a heating rate of 5 °C/min. The 183 injector and detector were maintained at 250 °C and 300 °C, respectively. The carrier 184 gas was helium at a flow rate of 1 mL/min. The mass spectra were obtained using an ionization energy of 70 eV and m/z from 40 to 600 a.m.u.. Injection of 1 µL of sample 185 was made in split mode (1:30). The compounds were identified by comparison of the 186 187 fragmentation patterns with data in the NIST 11 spectral library and data reported in the 188 literature.

189

190 2.3.3 HPLC analyses

The analytical profiles were obtained with a Dionex UltiMate 3000 system equipped with dual PDA and Corona CAD detectors (Thermo Fisher Scientific, Waltham, MA, USA). A Cortecs C18 column was used $(4.6 \times 150 \text{ mm}, 2.7 \mu\text{m})$ (Waters, Milford, MA, USA). Elution was performed using a linear gradient of H₂O mixed with an increasing proportion of CH₃CN. The flow rate was of 1.6 mL/min. The data were obtained with ChromeleonTM Chromatography Data System (CDS) Software.

197

198 **2.4 Structural elucidation of the HPFMz**

199 **2.4.1 Optical rotation**

200 The optical rotation was measured in methanol (MeOH) using an Anton Paar MCP 300

- 201 polarimeter (Graz, Austria) equipped with a 350 µL cell with a length of 100 mm.
- 202

203 2.4.2. ¹H and ¹³C NMR and 2D-NMR spectra

NMR analyses were performed in a Bruker (Billerica, MA, USA) 500 MHz spectrometer equipped with a 1 mm inverse detection probe. The samples were dissolved in deuterated methanol (CD₃OD) in 5 mm tubes. The chemical shifts (δ , in ppm) were reported based on the signal for tetramethylsilane (TMS).

208

209 2.4.3 High resolution mass analyses

High resolution mass measurements (ESITOFMS) were performed in Acquity UPLC system (Waters, Milford, MA, USA) coupled to a Waters Micromass LCT Premier mass spectrometer equipped with an electrospray interface (ESI) and a time-of-flight (TOF) detector.

214 **Compound 1:** Yellow powder; $R_f \sim 0.45$ with blue fluorescence at 366 nm. λ_{max} : 327 215 nm; ¹H NMR (CD₃OD) 7.56 (1H, d, J = 15.8 Hz, H-7), 7.05 (1H, d, J = 1.8 Hz, H-2'), 216 6.96 (1H, dd, J = 7.9, 1.8 Hz, H-6'), 6.77 (1H, d, J = 7.9 Hz, H-5'), 6.37 (1H, d, J =

217 15.8 Hz, H-8'), 4.20 (1H, m), 3.85 (1H, m), 2.05 (1H, br.s), 1.93 (br.s); HRESITOFMS 218 $[M - H]^-$ calculated for C₁₆H₁₇O₉, *m/z* 353.0872.

Compound 2: Yellow powder; $R_f \sim 0.65$ with orange fluorescence at 366 nm. $[\alpha]_{D}^{25}$: -219 12° (c 0.1 g/100 mL, CH₂OH); λ_{max} : 261, 356 nm; ¹H NMR (CD₃OD): δ 3.23 (1H, ddd, 220 221 J = 9.5, 5.2, 2.4 H-5"), 3.38 (1H, dd, J = 9.5, 8.8 Hz, H-3"), 3.42 (1H, t, J = 8.8 Hz, H-4"), 3.51 (1H, dd, *J* = 8.8, 7.9 Hz, H-2"), 3.62 (1H, d, *J* = 11.9, 5.2 Hz, H-6β"), 3.73 222 $(1H, dd, J = 11.9, 2.4 Hz, H-6\alpha")$, 5.26 (1H, d, J = 7.9 Hz, H-1"), 6.20 (1H, d, J = 2.1)223 Hz, H-6), 6.38 (1H, d, J = 2.1 Hz), 7.30 (2H, s); HRESITOFMS $[M - H]^{-}$, m/z224 479.0820 (calculated for $C_{21}H_{19}O_{13}$, m/z 479.0826) and $[M + H]^+$, m/z 481.0986 225 (calculated for $C_{21}H_{21}O_{13}$, *m/z* 481.0982). 226

Compound 3: Yellow powder; $R_f \sim 0.8$ with orange fluorescence at 366 nm. $[\alpha]_{D}^{25}$: -227 61° (c 0.1 g / 100 mL, CH₂OH), λ_{max}: 261, 350 nm; ¹H NMR (CD₃OD): δ 0.96 (3H, d, J 228 = 6.1, H-6"), 3.52 (1H, dd, J = 9.2, 6.1, H-5"), 3.34 (1H, t, J = 9.2 Hz, H-4"), 3.79 (1H, t, J = 9.2 Hz, H-4")), 3.79 (1H, t, J = 9.2 Hz, H-4"))) 229 dd, J = 9.2, 3.0 Hz, H-3"), 4.22 (1H, dd, J = 3.0, 1.2 Hz, H-2"), 5.32 (1H, d, J = 1.2 Hz, 230 231 H-1"), 6.20 (1H, d, J = 1.8 Hz, H-6), 6.36 (1H, d, J = 1.8 Hz), 6.95 (2H, s, H-2', H-6'); 232 ¹³C NMR (CD₃OD): 179.58 (C-4), 166.55 (C-7), 163.15 (C-5), 159.34 (C-2), 158.55 (C-9), 146.84 (C-3', C-5'), 137.91 (C-4'), 136.25 (C-3), 121.91 (C-1'), 109.55 (C-233 2',6'), 105.67 (C-10), 103.60 (C-1"), 100.00 (C-6), 94.83 (C-8), 73.34 (C-4"), 72.11 (C-234

- 235 3"), 72.00 (C-5"), 71.87 (C-2"), 17.64 (C-6"). HRESITOFMS $[M H]^-$, *m/z* 463.0877 236 (calculated for C₂₁H₁₉O₁₂, *m/z* 463.0877) and $[M + H]^+$, *m/z* 465.1032 (calculated for 237 C₂₁H₂₁O₁₂, *m/z* 465.1033).
- 238 **Compound 4:** Yellow powder; $R_f \sim 0.85$ with yellow fluorescence at 366 nm $[\alpha]_{D}^{25}$ = -
- 239 58° (c 0.1 g/100 mL, CH₂OH), λ_{max} : 263, 337 nm. ¹H NMR (CD₃OD): δ 0.95 (3H, d, J
- 240 = 5.7, H-6"), 3.32 (m, H5"), 3.33 (m, H4"), 3.75 (1H, dd, J = 9.2, 3.5 H-3"), 3.88 (3H, s,
- 241 4'-OCH₃), 4.22 (1H, dd, J = 3.5, 1.9 Hz, H-2"), 5.30 (1H, d, J = 1.9 Hz, H-1"), 6.13
- 242 (1H, d, J = 2.2 Hz, H-6), 6.27 (1H, d, J = 2.2 Hz), 6.87 (2H, s, H-2', H-6'); ¹³C NMR
- 243 (CD₃OD): 168.27 (C-7), 158.77 (C-2), 152.28 (C-3', C-5'), 139.28 (C-4'), 127.42 (C-
- 244 1'), 109.91 (C-2', C-6'), 103.63 (C-1"), 100.65 (C-6), 95.85 (C-8), 73.40 (C-5"), 72.27
- 245 (C-4"), 72.15 (C-3"), 72.05 (C-2"), 61.04 (4'-OCH₃), 17.86 (C-6"). HRESITOFMS [M
- 246 H]⁻, m/z 477.1039 (calculated for C₂₂H₂₁O₁₂, m/z 477.1033) and [M + H]⁺, m/z
- 247 479.1180 (calculated for $C_{22}H_{23}O_{12}$, *m*/z 479.1190).
- 248

249 2.5 Antimicrobial activity

The antimicrobial activities of the EEMz, HPFMz, LPFMz, and isolated flavonoids 2, 3, 250 251 and 4 were analyzed against S. aureus (ATCC 29213), C. albicans (ATCC 10231), and T. rubrum (SNB-TR1). The broth microdilution technique was used to determine the 252 minimal inhibitory concentration (MIC) according to the guidelines of the Clinical and 253 Laboratory Standards Institute (CLSI)^[24; 25]. Previously, the samples were solubilized 254 with dimethyl-sulfoxide (DMSO). Then, the solutions were placed in 96-well 255 microplates, diluted in a growth medium. Lastly, the pathogen suspension was added, 256 257 resulting in concentrations ranging from 512 to 1 µg/mL. The microplates were for 258 incubated at 37 °C for 24 h (bacteria), 48 h (yeast), and 5 days (dermatophyte). 259 Oxacillin (16-0.03 µg/mL) was used as a reference antibiotic, while itraconazole (16-260 0.03 μ g/mL) and fluconazole (64-0.125 μ g/mL) were employed as positive controls for 261 antifungal activity. The concentrations of DMSO were less than 0.5 %, thus not toxic for the micro-organisms. MIC was determined as the lowest concentration that 262 263 prevented microbial growth.

264

265 **2.6 Anthelmintic activity**

266 All the experiments employing animals were approved by the Ethics Committee for the 267 Use of Animals (CEUA/UNICAMP, protocol no. 4089-1/2015) and complied with the ethical principles for animal experimentation adopted by CEUA, State University of 268 269 Campinas. Adults (triploid parthenogenetic females) of Strongyloides venezuelensis 270 were isolated from the rodent Bolomys lasiuris and maintained in Rattus norvegicus. 271 Infective larvae (L3) of S. venezuelensis were obtained by coproculture. The cultures 272 were collected, concentrated and mantained at 28 °C for 48 h. For the recovery of adult 273 female parasites, male SPF Wistar rats were deliberately infected with S. venezuelensis 274 (2,000 larvae) during 15 days. Subsequently, rats were euthanasied (by concussion) and 275 15 cm of the intestines were removed, longitudinally sectioned, and placed in Petri 276 dishes containing RPMI 1640 medium. The parasites leave the intestinal mucosa after 2 277 h of incubation at 37 °C.

278 EEMz, HPFMz, LPFMz, F₂, F₃, and compounds **1** (chlorogenic acid) and **3** (myricitrin) 279 were analysed. In order to increase the solubility in the culture medium, samples were 280 dispersed in polyvinylpyrrolidone (PVP-K30) at 1:4 ratio. The assays were performed 281 in 24-well plates containing RPMI 1640 medium (2 mL/well) supplemented with 282 gentamicin (50 mg/L) and amphotericin B (2.5 mg/L). Two parasites were transferred to 283 each well and the plates were incubated under 5% CO₂ at 37 °C for 72 h. 50, 100, 200 284 and 400 µg/mL were analysed. Positive control was treated under the same 285 concentrations of ivermectin, while negative control only fresh medium containing PVP 286 was used. At 6, 24, 48, and 72 h, the motility and survival of the worms were evalueted 287 and morphological changes was observed by DM500 microscope (Leica, Wetzlar, 288 Germany). All the tests were performed in triplicate and survival curves were plotted 289 using Prism 5.0 software (GraphPad, San Diego, CA, USA).

290

291 2.7 Statistical analyses

All analyses were performed in triplicate. Statistical significance was determined by the Mantel-Cox test, using GraphPad Prism 5.0 software. The differences between the survival curves were considered statistically significant when the p-value was smaller than 0.001.

296 **3. RESULTS AND DISCUSSION**

Bioguided studies of the antibacterial, antifungal, and anthelmintic (against *S. venezuelensis*) activities of the extract, fractions, and isolated compounds obtained from the leaves of *M. zapota* was firstly developed in this work. EEMz was obtained with a yield of 10.71% while LPFMz and HPFMz showed yields of 54.16% and 38.13%, respectively.

303

304 **3.1 Identification of compounds in the LPFMz fraction**

305 GC-MS analyses revealed a predominance of pentacyclic triterpene compounds 306 showing typical fragmentation patterns of ursane and oleane skeletons, with retention 307 times between 40 and 47 minutes corresponding at 75% of the all LPFMz fraction (Fig. 308 1). Table 1 shows the compounds identified from their fragmentation patterns, 309 compared to data obtained from the NIST 11 library. The presence of long chain 310 hydrocarbons and sterols in the aerial parts of *M. zapota* has been reported previously 311 ^[21; 26] and was confirmed here by the presence of compounds **A**, **B**, and **C**.

312 Please, insert here the Figure 1 and Table 1

313 Compound 5 presented a base peak at m/z 204, accompanied by the ions with m/z 189 and 177, which were characteristic of the compound germanicol (olean-18-en-3 β -ol)^[27] 314 (Fig. 2). Peaks at m/z 218, 203, and 189 were associated with retro-Diels-Alder type 315 316 fragmentation of ring C of olean-12-enes and ursan-12-enes. These characteristic peaks 317 were observed for compounds 6 and 7 identified as β - and α -amyrin, respectively. 318 Triterpene analogues of β -amyrin show greater abundance of the peak at m/z 203, compared to the peak at m/z 189, while the α -amyrin analogues show similar 319 abundances of these peaks ^[27; 28]. The mass spectra of triterpene esters reflect the 320 321 fragmentation pattern of the main skeleton, with smaller ions derived from the acid portion ^[29]. Hence, the fragmentation pattern of compound 8 corresponded to that 322 323 expected for germanicol acetate (m/z, 468), with the fragments characteristic of germanicol (at m/z 204 and 169), together with fragments at m/z 453 and 408 reflecting 324 the loss of methyl and acetyl groups, respectively, and a fragment at m/z 43 325 corresponding to COCH₃^[27]. 326

327 Please, insert here the Figure 2

328 Compounds **9** and **10** corresponded to α - and β -amyrin acetates, respectively, with 329 fragments peaks at *m/z* 218, 203, and 189 characteristics of amyrins and fragments with loss of the acetyl group ^[27]. Compound **11** was identified as lupeol acetate, with peaks at m/z 453, 408, 365, 218, 189, and 43 ^[30]. Amyrins, amyrin acetates, and lupeol acetate have been reported in leaves of *M. zapota* ^[14; 16]. However, germanicol and its acetate were firstly reported in the leaves of *M. zapota* in the current study. However, the current study is the first time that germanicol and its acetate have been reported in *M. zapota* leaves.

336

337 3.2 Identification of compounds isolated from HPFMz

Four compounds from HPFMz can be isolated and purified by preparative HPLC. Compounds **1** (3.5 mg), **2** (0.7 mg), **3** (13.8 mg), and **4** (4.6 mg) showed retention times of 2.01, 3.51, 4.14, and 5.26 min, respectively, detected in the HPLC analysis (Fig. 3). The structures of the compounds were determined based on the one- and twodimensional NMR (¹H, ¹³C, COSY, HSQC, and HMBC), high-resolution mass spectrometry, and optical rotation analyses.

344 **Please, insert here the Figure 3**

345 Compound 1 presented the molecular formula $C_{16}H_{18}O_9$, with a mass of 354. Its 346 structure was identified as that of chlorogenic acid (Fig. 4). The high-resolution negative mode mass spectrometry analysis resulted in extraction of the [M–H]⁻ ion with 347 m/z 353.0872. The structural elucidation data corroborates Fayek et al. (2013) ^[19], who 348 isolated this compound from the fruit of *M. zapota*. Other earlier studies have also been 349 identified chlorogenic acid in the fruit and seeds of *M. zapota* ^[31; 32]. It could be inferred 350 from the ¹H NMR spectra of substances 2, 3, and 4 that they possessed similar base 351 352 nuclei and belonged to the flavonoid class of compounds (Fig. 4). The pair of doublets at δ 6.18 (± 0.8) (J = 2.0 δ ± 0.5 1H) and δ 6.35 (± 0.8) (J = 2.1 ± 0.4 1H) could be 353 354 attributed to the 5,7,9,10-tetrasubstituted ring A, while the presence of two singlets at δ 6.95 (\pm 0.8) was indicative of a flavonol with a 1',3',4',5'-tetrasubstituted ring B, 355

356 corresponding to myricetin ^[33].

357 Please, insert here the Figure 4

358 Considering the sugars linked to C-3 of the aglycones, compound **2** showed the 359 presence of β -D-glucopyranoside unit, due to the high coupling constants (J_{HH}) 360 involving each consecutive pair of H-1, H-2, H-3, H-4, and H-5 protons of the sugar, 361 varying between 8.8 and 11.9 Hz, which is characteristic of protons in axial positions.

- For compounds **3** and **4**, the same saccharide unit was attached to C-3. The signal at $\delta_{\rm C}$ 17.6 ± 0.2, assigned to the C-6 methyl carbon of the sugar, together with the set of signals at $\delta_{\rm C}$ 71.8 (± 0.2), 72.1 (± 0.4), 73.3 (± 0.1), and 72.0 (± 0.2), enabled its classification as a rhamnose.
- 366 HRESITOFMS spectrum of compound 2 presented pseudomolecular ion peaks [M +
- 367 H]⁺ with *m/z* 481.0986 (calculated for C₂₁H₂₁O₁₃, with *m/z* 481.0982) and [M H]⁻ with 368 *m/z* 479.0820 (calculated for C₂₁H₁₉O₁₃, with *m/z* 479.0826), giving the molecular 369 formula C₂₁H₁₉O₁₃ with mass of 480. The structural elucidation of this compound was 370 reported in other plants and identified as myricetin-3-O-β-D-glucopyranoside ^[34; 35]. The
- absence of a smaller value of $J_{\rm HH}$ between the H-3" and H-4" protons confirmed that it was not a 3-O-galactoside ^[34]. The literature confirms the presence of myricetin-3-O- β -
- 373 D-galactopyranoside in species of the Sapotaceae family ^[36].
- Compound **3** presented the molecular formula $C_{21}H_{19}O_{13}$, with mass of 464, derived of the base formula $C_{21}H_{19}O_{12}$ resulting from the negative mode adduct $[M - H]^-$ with m/z
- 463.0877 (calculated for C₂₁H₁₉O₁₂, with *m/z* 463.0877) and the positive mode ion [M + H]⁺ with *m/z* 465.1032 (calculated for C₂₁H₂₁O₁₂, with *m/z* 465.1033). This compound was identified as myricetin-3-O-α-L-rhamnopyranoside, also known as myricitrin. This compound has also been identified in previous studies using the leaves of *M. zapota* ^{[21;} $^{33; 37]}$.
- 381 ¹H NMR data for compound **4** showed a signal characteristic of a methoxyl (δ 3.88). 382 The location of this group was determined from the HMBC data, which showed its 383 correlation with C-4' (δ 139.2) and with the two protons H-2' and H-6' responsible for the singlet at 6.87 ppm of the aglycone. Using the ions $[M - H]^{-}$ with m/z 477.1039 384 (calculated for $C_{22}H_{21}O_{12}$, with m/z 477.1033) and $[M + H]^+$ with m/z 479.1180 385 386 (calculated for $C_{22}H_{23}O_{12}$, with m/z 479.1190), it was possible to obtain the molecular formula of compound 4 as C₂₂H₂₂O₁₂, with mass of 478, identified as myricetin-3-O-4'-387 388 methyl-ether-3-O- α -L-rhamnopyranoside, also known as mearnsetin-3-O-a-L-389 rhamnopyranoside (mearnsitrin). Take all results together, it is important to highlight 390 that chlorogenic acid, myricetin-3-O-β-D-glucopyranoside, and mearnsitrin compounds 391 in leaves of *M. zapota* were firstly detected in this work.
- 392
- 393 **3.3** *In vitro* antimicrobial activity

394 MIC of the extracts, fractions, and isolated flavonoids of *M. zapota* are shown in Table 395 2. These results showed that EEMz, HPFMz, LPFMz were potentially active against C. 396 albicans (MIC = 16 μ g/mL). In the case of T. rubrum, EEMz and HPFMz presented 397 moderate activity, while LPFMz was ineffective. HPFMz showed high activity against 398 S. aureus and moderate activity against the other microorganisms. About the flavonoid, 399 compound 2 showed significant activity against S. aureus (MIC = 8 μ g/mL) and 400 promising activity against T. rubrum (MIC = 16 μ g/mL). Myricitrin (compound 3) 401 presented promising activity against the gram-positive bacterium and the dermatophyte 402 (MIC values of 32 and 64 µg/mL, respectively). Mearnsitrin (compound 4) presented 403 activity against S. aureus (MIC = 32 μ g/mL) and was ineffective against the other 404 microorganisms.

405 Please, insert here the Table 2

406 It has been found that the presence of terpenoids and glycosidic flavonoids in crude 407 extracts plays an important role in determining their antimicrobial activity ^[38]. The 408 presence of such compounds in the extract and the higher polarity fraction could 409 provide an explanation for the observed antimicrobial action against the pathogens tested. The results obtained to C. albicans corroborate previous studies using aqueous 410 extract of leaves collected in India^[39] and various *Candida* species^[40]. The 411 412 antimicrobial properties of M. zapota extracts against S. aureus have been observed previously ^[39; 41]. These activities may be related to the isomeric mixture of α - and β -413 amyrin detected in the LPFMz that possesses known antimicrobial activity ^[42]. The 414 presence of one or more sugar units at different positions of the aglycone can increase 415 solubility and enhance the inherent medicinal properties of flavonoid compounds ^[43]. 416 417 Taken together, all the samples originated from the leaves of M. zapota presented 418 antimicrobial activities (of different magnitudes). These findings demonstrate the high 419 antimicrobial potential of the leaves of *M. zapota* in combating three important human 420 pathogenic microorganisms.

421

422 **3.4** *In vitro* anthelmintic activity assays

423 The anthelmintic effect of EEMz, HPFMz, LPFMz, F_2 , F_3 , and isolated flavonoids 424 (compounds 1 and 3) were evaluated by the survival percentage and cuticular integrity 425 of female (L4) *S. venezuelensis* parasites. The positive control (with addition of PVP) 426 presented 100% survival of the parasites. All samples tested at 400 µg/mL were lethal 427 after 24 h to the parasites (Fig. 5). For EEMz, no survival of parasites at 72 h was observed to 100 and 200 µg/mL (Fig. 5A). HPFMz was lethal at 200 µg/mL after 48 h, 428 429 although survival at 72h have been occurred for 100 and 50 µg/mL corresponding at 430 5.5% and 72%, respectively (Fig. 5B). F₂ was lethal at all concentrations tested after 72 431 h of incubation (Fig. 5C), while F_3 was only lethal at the two highest concentrations 432 (Fig. 5D). The parasites exposed to chlorogenic acid at 200 µg/mL died within the first 433 24 h, while after 72 h there was no survival at any concentration (Fig. 5E). At 72 h, 434 there was no survival of the parasites exposed to myricitrin (compound 3) at 400 and 435 200 µg/mL. However, the survive of the 11.1 and 61.1% were determined to 100 and 50 436 µg/mL, respectively (Fig 5F). It was possible to observe that all treatment had 437 statistically significant difference of the control (p < 0.0001).

438 **Please, insert here the Figure 5**

439 Ivermectin (pharmacological control) reduced the motility of the organisms, but there 440 was survival of all the individuals after 72 h. Similar behavior to infective larvae (L3) of 441 *S. stercoralis* using 250 μ g/mL of ivermectin has been observed with reduction of the 442 parasite motility and survive during 6 days ^[44]. The lower polarity fractions were also 443 tested (data not shown) but showed unsatisfactory and imprecise results at all 444 concentrations evaluated due the formation of insoluble clusters in the culture medium.

445 At 100 μ g/mL, chlorogenic acid was lethal in the first 24 h, while EEMz and F₂ caused total mortality at the end of the incubation time (Fig. 6). At 200 µg/mL, only 446 chlorogenic acid was lethal after 24 h. However, there was no survival for any of the 447 448 samples tested after 72 h. Images of the parasites after 72 h of incubation are shown in Fig. 7. Parasites exposed to myricitrin (Fig. 7B) exhibited distortions and cuticular 449 450 wrinkling. Those treated with F2 (Fig. 7C) showed distortions, depressions, and 451 alterations in cuticular integrity, suggestive of direct cytotoxic effects on the 452 morphology of the worms. The parasites exposed to ivermectin (Fig. 7D) exhibited 453 reduced motility, but the integrity of the cuticular surface remained similar to that of the 454 positive control (Fig. 7A).

455 Please, insert here the Figure 6 and 7

Thus, the first study describing the direct action of extracts, fractions, and isolated compounds obtained from *M. zapota* leaves on adult females of *S. venezuelensis* have been presented in this work, highlighting the potential anthelmintic activities, especially of the ethanolic extract, F_2 , and chlorogenic acid. The chemical analyses showed that both the ethanolic extract and the fractions contained glycosidic flavonoids as well as phenolic acids and their derivatives, suggesting that the effective anthelmintic action against the nematode was due to the presence of these compounds in the leaves of *M*. *zapota*.

464 Various studies of plant species have been undertaken in the search for new treatments 465 for veterinary and human parasitosis. Free-living females of Strongyloides ssp. exposed 466 to a methanolic extract of Lawsonia inermis have been analyzed taking into account the presence of distortions ,and longitudinal and transverse cracks ^[45]. Sarojini et al. 467 (2012)^[46] reported strong anthelmintic action of *L. inermis* extract due to the presence 468 of glycosides, terpenoids, and flavonoids. Carvalho et al. $(2012)^{[47]}$ described the action 469 470 of extracts of Piper tuberculatum, Lippia sidoides, and Mentha piperita against 471 Haemonchus contortus

- 472 Ndhlala et al. (2016) ^[48] described the chemical composition and anthelmintic activity
 473 of extracts of *Achyranthes aspera* against the free-living nematode *Caenorhabditis*474 *elegans*. The aqueous and acetone extracts of *A. aspera* exhibited strong activity, with
 475 UHPLC-MS/MS analyses revealing the presence of rutin, chlorogenic acid, and
 476 genistein, which were associated with the activity.
- 477 Comparison of the analogous flavonoids revealed a close relationship between their 478 structures and antiparasitic activity. The presence of substituents (hydroxyl, methyl, 479 methoxy, and glycosides) is fundamental to determine its effectiveness. In the case of 480 hydroxyls, the number and pattern of hydroxylations in ring B has been found to have a major influence on the antiparasitic activity of flavonoids ^[49]. However, the mechanisms 481 482 of action of flavonoids in helminths remain to be fully elucidated, even if several 483 possibilities have been proposed: an increase in the supply of nitric oxide acting as a 484 neurotransmitter in neuromuscular junctions and causing myoinhibition; nonspecific 485 inhibition of esterases and cholinesterases leading to neuromuscular disorganization of 486 the parasites; increased efflux of calcium causing depolymerization of the 487 microtrabecular network, with consequent disorganization and disaggregation of the myofilaments; and decreased levels of free amino acids and increased levels of glutamic 488

acid, glutamine, citrulline, and gamma-aminobutyric acid (GABA), leading to the
 release of nitric acid and resulting in neurological alterations ^[49].

491

492 **CONCLUSIONS**

493 Antimicrobial, antifungal, and anthelmintic activities of the extract, fractions, and isolated compounds obtained from the leaves of Manilkara zapota were firstly 494 495 evaluated. Another innovative find of this work was the identification of the compounds 496 1 (chlorogenic acid), 2 (myricetin-3-O-β-D-glucopyranoside), 4 (mearnsitrin), 5 497 (germanicol), and 8 (germanicol acetate) in *M. zapota* leaves. Ethanolic extract and 498 fractions showed high antifungal activity against C. albicans. Compounds 3 (myricetin-499 3-O- α -L-rhamnopyranoside) and 4 presented promising activity against S. aureus while 500 the high polarity fraction and compound 2 exhibited high activity against S. aureus and 501 T. rubrum. The ethanolic extract and F₂ containing phenolic compounds, and 502 chlorogenic acid showed effective antiparasitic action against female (L4) S. 503 venezuelensis parasites. This study contributes to the scientific basis for the possibility 504 of using plants and compounds of natural origin to combat strongyloidiasis and 505 microbial infections. The development of new therapeutic agents from *M. zapota* leaves 506 will require further in-depth studies, but nonetheless this plant material appears to offer 507 an effective alternative for the treatment of these diseases that are widespread and 508 difficult to control.

509

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518

519 CONFLICTS OF INTEREST

520 The authors declare that they have no conflicts of interest.

521

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647 **FIGURE LEGENDS**

- 648 **Fig. 1-** Expanded chromatogram profile of the LPFMz
- 649 Fig. 2- Compounds detected by GC-MS in the LPFMz
- **Fig. 3-** HPLC elution of the compounds 1, 2, 3, and 4 in the HPFMz.
- 651 **Fig. 4-** Compounds isolated from HPFMz.
- 652 Fig. 5- Anthelmintic activity of M. zapota against S. venezuelensis after 72 h of
- 653 incubation. Percentage survival of parasites exposed to: EEMz (A), HPFMz (B), F₂ (C),
- 654 F₃ (D), chlorogenic acid (compound 1) (E), and myricitrin (compound 3) (F). Control
- 655 (●), 50 µg/mL (■), 100 µg/mL (▲), 200 µg/mL (▼), and 400 µg/mL (♦).
- 656 Fig. 6- Comparison of the anthelmintic activities of the extract, fractions, and
- 657 compounds obtained from the leaves of *M. zapota* at 100 μ g/mL (A) and 200 μ g/mL
- 658 (B). Control (•), EEMz (\blacksquare), HPFMz (\blacktriangle), F₂ (\triangledown), F₃ (•), chlorogenic acid (compound
- 659 1) (\circ), and myricitrin (compound 3) (\Box).
- 660 **Fig. 7-** Parthenogenetic females of *S. venezuelensis* after exposure to 400 μg/mL for 72
- 661 h of incubation. Parasite of the control group with preserved cuticle (A); Parasite
- 662 exposed to myricitrin, with cuticular wrinkling (arrow) (B); Parasite exposed to F_2 , with
- 663 cuticular alterations (arrows) (C); Parasite treated with ivermectin, with preserved
- 664 cuticular integrity (D).

Туре	RT ^a (mints)	RT ^a (mints) Compound ^b		Molecular formula	Peak area (%)	
Α	32.76	Squalene	410	$C_{30}H_{50}$	5.94	
В	36.61	n-Hentriacontane	436	C ₃₁ H ₆₄	2.83	
С	37.35	α-Tocopherol	430	$C_{29}H_{50}O_2$	0.86	
5	41.69	Germanicol	426	C ₃₀ H ₅₀ O	14.79	
6	42.13	β-Amyrin	426	$C_{30}H_{50}O$	3.89	
7	43.46	α-Amyrin	426	C ₃₀ H ₅₀ O	6.54	
8	44.54	Germanicol acetate	468	$C_{32}H_{52}O$	4.84	
9	45.15	β-Amyrin acetate	468	$C_{32}H_{52}O_2$	13.90	
10	45.59	α-Amyrin acetate	468	$C_{32}H_{52}O_2$	31.04	
11	46.60	Lupeol acetate	468	$C_{32}H_{52}O_2$	51.04	

Table 1: GC-MS analysis of the chemical composition of the LPFMz

^a Retention time (mints).

^b Compounds listed in order of elution from an HP-5MS capillary colum	ın.
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Table 2: Minimum inhibitory concentrations (MICs) of the EEMz, HPFMz, LPFMz,

6/0 and isolated flavonoids (compounds 2, 5, and 4) obtained from the leaves of A	670	and isolated flavonoids	(compounds 2, 3, and 4) obtained from the	leaves of M. zap	ota.
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MIC (µg/mL)							
Microrganism	EEMz	HPFMz	LPFMz	2	3	4	Control [*]
C. albicans (ATCC 10231)	16	16	16	>128	>128	>128	8^{a}
T. rubrum (SNB-TR1)	256	128	>512	16	64	128	3.125x10 ^{-3 a} 2 ^b
S. aureus (ATCC 29213)	128	32	128	8	32	32	2^{c}

⁶⁷¹ ^{*}Positive control; ^afluconazole; ^bitraconazole; ^coxacillin.

Figure 1 Expanded chromatogram profile of the LPFMz.



















Figure 5 Anthelmintic activity of M. zapota against *S. venezuelensis* after 72 h of incubation. Percentage survival of parasites exposed to: EEMz (A), HPFMz (B), F_2 (C), F_3 (D), chlorogenic acid (compound 1) (E) and myricitrin (compound 3) (F). Control (•), 50 µg/mL (•), 100 µg/mL (▲), 200 µg/mL (▼) and 400 µg/mL (♦).





Figure 6 Comparison of the anthelmintic activities of the extract, fractions, and compounds obtained from the leaves of *M. zapota* at 100 μ g/mL (A) and 200 μ g/mL (B). Control (•), EEMz (•), HPFMz (▲), F2 (♥), F3 (•), chlorogenic acid (compound 1) (\circ) and myricitrin (compound 3) (\Box).



Figure 7 Parthenogenetic females of *S. venezuelensis* after exposure to 400 µg/mL for 702 72 h of incubation. Parasite of the control group with preserved cuticle (A); Parasite 703 exposed to myricitrin, with cuticular wrinkling (arrow) (B); Parasite exposed to F2, with 704 cuticular alterations (arrows) (C); Parasite treated with ivermectin, with preserved 705 cuticular integrity (D).



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