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CRediT statement

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35 **Abstract**

36 Industrial metalworking facilities emit a variety of air toxics including volatile organic 37 compounds, polycyclic aromatic hydrocarbons (PAHs) and heavy metals. In order to investigate 38 these emissions, a 1-month multi-instrument field campaign was undertaken at an industrial site 39 in Grande-Synthe, Dunkirk (France), in May and June 2012. One of the main objectives of the 40 study was to provide new information on the chemical composition of particulate matter with 41 aerodynamic diameters smaller than 2.5 μ m (PM_{2.5}) in the vicinity of metalworking facilities. An 42 aerosol time-of-flight mass spectrometer (ATOFMS) was deployed to provide size-resolved 43 chemical mixing state measurements of ambient single particles at high temporal resolution. This 44 mixing state information was then used to apportion $PM_{2.5}$ to local metalworking facilities 45 influencing the receptor site. Periods when the site was influenced by metalworking sources were 46 characterised by a pronounced increase in particles containing toxic metals (manganese, iron, 47 lead) and polycyclic aromatic hydrocarbons (PAHs) with a variety of chemical mixing states. 48 The association of specific particle classes with a nearby ferromanganese alloy manufacturing 49 plant was confirmed through comparison with previous analysis of raw materials (ores) and 50 chimney filter particle samples collected at the facility. Particles associated with emissions from 51 a nearby steelworks were also identified. The contribution of local metalworking activities to 52 PM_{2.5} at the receptor site for the period when the ATOFMS was deployed ranged from 1-65% 53 with an average contribution of 17%, while the remaining mass was attributed to other local and 54 regional sources. These findings demonstrate the impact of metalworking facilities on air quality 55 downwind and provide useful single particle signatures for future source apportionment studies 56 in communities impacted by metalworking emissions. light mass spectrometer (ATOFMS) was deployed to prostate measurements of ambient single particles at high tempormation was then used to apportion PM_{2.5} to local met ceptor site. Periods when the site was influenced by m

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58 Keywords:

59 Particulate matter; source apportionment; mass spectrometry; industrial emissions; 60 metalworking; air quality

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64 **1 Introduction**

65 Negative health outcomes, lower crop yields, building damage and other impacts associated with 66 industrial air pollution and greenhouse gas emissions were estimated to cost Europe over ϵ 300 67 billion between 2008 and 2012 (European Environment Agency, 2014). Almost a third of 68 emissions of particulate matter with aerodynamic diameters smaller than 10 μ m (PM₁₀) in France 69 (CITEPA, 2014) are attributed to industrial sources. The city of Dunkirk lies in the Hauts de 70 France region of Northern France and is home to over 200 industrial facilities from many sectors 71 including refining and cement production. Dunkirk also contains one of Europe's largest 72 steelworks, which emits approximately 3,000 t of PM per year (Flament et al., 2008). Numerous 73 local industrial facilities generate plumes that impact air quality in the surrounding areas 74 depending on local meteorological conditions, and local PM_{10} concentrations regularly exceed 75 the European threshold value (24-hour average of 50 μ g m⁻³) (Setyan et al., 2019).

76 The physico-chemical nature of industrial PM varies considerably, particularly in terms of 77 composition, depending on the type of industrial processes involved (Sanderson et al., 2014; 78 Taiwo et al., 2014; Riffault et al., 2015). Ambient concentrations of particulate phase metals tend 79 to be greater at locations directly influenced by industrial activities than at rural or urban 80 locations (Osornio-Vargas Alvaro et al., 2003; Schaumann et al., 2004; Lim et al., 2010; 81 Mohiuddin et al., 2014). Heavy metals and polycyclic aromatic hydrocarbons (PAHs) are 82 important contributors to particle toxicity (Harrison and Yin, 2000). The generation of reactive 83 oxygen species (ROS) and the development of asthma and cardiopulmonary disease have been 84 linked to exposure to metals (Schaumann et al., 2004; Majestic et al., 2007). PM $_{2.5}$ collected 85 from the same industrial area of Dunkirk investigated in this work has been recently 86 demonstrated to damage DNA through an *in vitro* assessment of immortalized lung cells (Platel 87 et al., 2020). Thus, the impact of industrial emissions on downwind air quality needs to be 88 assessed in order to better understand human exposure to the associated air toxics. g and cement production. Dunkirk also contains one or emits approximately 3,000 t of PM per year (Flament et a facilities generate plumes that impact air quality in the cal meteorological conditions, and local PM₁₀ conc

89 Particulate PAH and metal content studies have been performed in the Dunkirk area previosuly 90 due to the importance of metalworking sources and their proximity to large residential areas 91 (Ledoux et al., 2006; Ledoux et al., 2009; Flament et al., 2008; Alleman et al., 2010; Marris et 92 al., 2012; Marris et al., 2013; Hleis et al., 2013; Mbengue et al., 2014; Cazier et al., 2016; Crenn 93 et al., 2018). In terms of source apportionment, the tracers used have included elemental ratios,

94 for examples Mn/Fe as a signature for a ferromanganese alloy metalworking facility, V/Ni for 95 petrochemical facilities and isotopes of Pb and Fe as a signature for a local steelworks (Veron et 96 al., 1999; Ledoux et al., 2006; Flament et al., 2008; Ledoux et al., 2009; Alleman et al., 2010; 97 Marris et al., 2012). Emissions from the steelworks are also the dominant source of air emissions 98 of Fe in Dunkirk, and the facility is estimated to emit 620 t of Fe annually (Flament et al., 2008). 99 Anthropogenic particulate Fe emissions are of additional concern because of their higher 100 solubility relative to Fe present in naturally occurring dust. Thus anthropogenic particulate Fe is 101 more bioavailable than natural Fe dust, and the former can modulate primary productivity in 102 high-nutrient, low chlorophyll (HNLC) ocean regions (Ito and Shi, 2016). Anthropogenic 103 sources are responsible for approximately 7% of the 8.4 Mt of Fe deposited to oceans annually, 104 and therefore have significant impacts upon carbon dioxide fixation globally (Wang et al., 2015).

105 The density and diversity of sources and emission profiles in Dunkirk complicates source 106 apportionment efforts based on off-line analysis significantly (Alleman et al., 2010). One 107 technique that is well suited to providing information on the size-resolved chemical composition 108 and mixing state of particles emitted by industrial activities is single particle mass spectrometry. 109 A review of industrial studies utilising this technique can be found in Riffault et al. (2015). The 110 capability to provide mixing state information for individual particles is key to the role single 111 particle mass spectrometry plays in source apportionment. Ambient particles containing specific 112 internal mixtures of metals and other primary organic and inorganic species have been 113 successfully assigned to their respective sources using this technique. For example, internally 114 mixed Pb/Zn/Cl particles have been assigned to industrial waste incineration (Moffet et al., 115 2008b), Ga- and Se-containing particles particles have been associated with coal-fired power 116 generation (Pekney et al., 2006; Snyder et al., 2009) and internally mixed V/Ni/sulfate particles 117 have been associated with ship exhaust (Ault et al., 2009; Healy et al., 2009). Fe, Zn and Pb-118 containing particles have also been previously assigned to steelwork facility operations 119 (Dall'Osto et al., 2008). e than natural Fe dust, and the former can modulate prin
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120 In this context, the NANO-INDUS field campaign was performed in the multi-industry 121 influenced zone of Grande-Synthe in Dunkirk, France, close to a ferromanganese alloy facility 122 and a steelworks (Setyan et al., 2019). An aerosol time-of-flight mass spectrometer (ATOFMS) 123 was deployed to provide size-resolved chemical composition and mixing state data for industrial 124 PM_{2.5} with high temporal resolution. The ATOFMS data were used in combination with support 125 instrumentation measurements to apportion $PM_{2.5}$ to local and regional sources and to assess the 126 impact of local metalworking facilities on air quality at the receptor site. Comparison of the 127 ambient ATOFMS data with previous analysis of raw materials (ores) and particles collected on 128 industrial filters from the ferromanganese alloy facility's chimneys (Arndt et al., 2016), enabled 129 more effective source assignments for the particle classes observed during the field campaign.

130 **2 Experimental**

131 **2.1 Sampling site and Instrumentation**

132 All measurements were made in the industrial area of Grande-Synthe, Dunkirk (51.028809 N, 133 2.269448 E, 10 m above sea level), approximately 800 m southwest of a ferromanganese alloy 134 manufacturing facility and 1 km southwest of a major steelworks (Figure S1). The adjacency of 135 these two sources precludes separation of their air pollutant contributions based on wind 136 direction alone (Setyan et al., 2019). Other nearby sources include a petrochemical storage 137 facility <500 m to the west of the site and urban centres located to the east of the site (Fort-138 Mardyck and Dunkirk town centre, \sim 2 km and \sim 7 km from the site, respectively). The field 139 campaign was undertaken from 15 May 2012 to 12 June 2012. An overview of the campaign, 140 local meteorology, the instrumentation deployed and broad PM composition is described in 141 detail elsewhere (Setyan et al., 2019). Date and time are reported in Coordinated Universal Time 142 (UTC). PM_{2.5} mass concentrations were determined every 2 hours with a MP101M Beta Gauge 143 monitor. 24 to 48-hour filter samples were also collected on 25mm polycarbonate membrane 144 filters (WhatmanTM) using 13- and 3-stage cascade impactors, and weighed with a micro-145 balance (UMT2, Mettler Toledo, Switzerland) (Mbengue et al., 2014, 2017). s were made in the industrial area of Grande-Synthe, Durn above sea level), approximately 800 m southwest of a fecility and 1 km southwest of a major steelworks (Figure Si ess precludes separation of their air pollutant c

146 The ATOFMS (TSI model 3800) was operated continuously from 18th May – 8th June 2012. A 147 detailed description of the ATOFMS can be found elsewhere (Gard et al., 1997). Briefly, it 148 consists of (i) an aerodynamic focusing lens (TSI AFL100) (Su et al., 2004) used to optimize 149 particle transmission in the aerodynamic diameter (d_a) range 100-3000 nm, (ii) a particle sizing 150 region, and (iii) a bipolar reflectron time-of-flight mass spectrometer. Single particles are 151 desorbed/ionised using a pulsed Nd:YAG laser ($\lambda = 266$ nm, ~1 mJ pulse⁻¹) and positive and 152 negative ion mass spectra of individual aerosol particles are simultaneously collected for 153 successfully ionised particles. ATOFMS particle counts are scaled to account for sizing and other 154 instrument biases as discussed below.

156 **2.2 ATOFMS Data Analysis**

157 Mass spectral data for over 800,000 single particles was collected using the ATOFMS during the 158 deployment period and was clustered using the *K*-means algorithm (*K*=80), as described 159 elsewhere (Gross et al., 2010; Healy et al., 2010; Healy et al., 2013). Clusters with similar mass 160 sepctral profiles, size distributions and temporality were combined. The final merged clusters 161 were then identified as particle "classes"; 33 in total. Approximately 15% of the particles ionised 162 did not produce positive ion mass spectra. The absence of positive ions is likely a result of 163 desorption/ionization matrix effects and precludes the assigment of these particles to a specific 164 category due to the absence of useful marker ions.

165 The particle class nomenclature used here is regularly used in the literature (Dall'Osto and 166 Harrison, 2006; Spencer et al., 2006; Ault et al., 2010; Pratt and Prather, 2012) and typically 167 denotes the dominant species in the positive ion mass spectra. For example, a particle class with 168 high intensities for sodium and elemental carbon is labelled Na-EC. In some cases this is 169 followed by secondary species detected in the negative ion mass spectra (e.g. $EC\text{-}NH₄\text{-}NO₃$), 170 referring to elemental carbon internally mixed with ammonium and nitrate. From French Control (19 and Control Cyprocential Cyprocential Cyprocential positive ions matrix effects and precludes the assignent of these pie absence of useful marker ions.

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171 The ATOFMS does not measure particle number or mass concentrations quantitatively, but 172 instead provides particle counts classified by aerodynamic diameter. The size-dependent 173 transmission efficiency of the focusing lens, data acquisition "busy time", and limited detection 174 of particles <150 nm complicate accurate counting of particles over the ATOFMS size range 175 (100-3000 nm *d*a). The desorption/ionization laser used by the ATOFMS also complicates 176 quantitative speciation efforts. Differences in laser output power and variations in power density 177 across the laser beam between pulses (Steele et al., 2005; Wenzel and Prather, 2004) alter the 178 mass of each particle that is effectively desorbed impacting mass spectral signal intensities 179 (Gross et al., 2000a; Reinard and Johnston, 2008).

180 Thus the measured ATOFMS counts were scaled using two quantitative particle counting 181 instruments operated concurrently; an optical particle counter (OPC, TSI model 3300) and a 182 scanning mobility particle sizer (SMPS, TSI DMA model 3080 and CPC model 3010). Scaling 183 the ATOFMS number concentration data requires conversion of d_a (aerodynamic diameter) into 184 the corresponding volume equivalent diameter (d_{ve}) , using the following relationship:

$$
d_a = \frac{\rho_p}{\rho_0} \frac{d_{ve}}{\chi} \tag{1}
$$

186 where ρ_p is the particle density, assumed for the purposes of number-size distribution scaling to 187 be 1.7 g cm⁻³ as per previous applications of the method (Morawska et al., 1999; Reinard et al., 188 2007), *dv*^e is the volume equivalent diameter (operationally equivalent to electrical mobility 189 diameter d_m measured by the SMPS and optical diameter d_0 measured by the OPC), ρ_0 is 190 standard density (1 g cm⁻³) and γ represents the dynamic shape factor (e.g. 1, representing 191 spherical shape). This is a simplified version of particle property relationships that are covered in 192 detail elsewhere (DeCarlo et al., 2004). Mass concentrations can then be calculated from the 193 scaled number concentrations as follows (Reinard et al., 2007):

$$
m = \frac{\pi}{6} \rho_p d_{\nu e}^3 \tag{2}
$$

195 The accuracy of the conversion of number concentrations to mass concentrations is dependent on 196 the representativeness of χ and ρ _p for each type of particle. In the absence of morphology data χ 197 is assumed to be 1. The use of a single density value, ρ_p , for conversion of ATOFMS scaled 198 particle number concentrations to mass concentrations has previously resulted in good agreement 199 with other quantitative measurements (Qin et al., 2006; Healy et al., 2013; Zhou et al., 2016). 200 However, this simplification is not entirely accurate because different particle types exhibit 201 different densities depending upon their chemical composition (Spencer and Prather, 2006; 202 Reinard et al., 2007). A range of more representative densities, estimated from the bulk densities 203 of the chemical components observed in the mass spectra (Bein et al., 2006; Reinard et al., 204 2007), was therefore used to convert the scaled number-size distributions to mass concentrations 205 for each particle class, as listed in Table S1. ATOFMS-derived mass concentrations for $PM_{2.5}$ 206 were averaged to hourly and 24 to 48-hour time resolutions for comparison with $PM_{2.5}$ measured 207 concurrently on-line using the beta gauge instrument and off-line through gravimetric filter 208 analysis, respectively. Similar ATOFMS-based source apportionment techniques have been 209 performed previously to separate local and regional contributions to ambient PM in Paris, France 210 and the apportionment results were found to agree well with those modelled using a regional 211 chemical transport model (Healy et al., 2013; Skyllakou et al., 2014). This is a simplement version of particle property relationship
(DeCarlo et al., 2004). Mass concentrations can then be
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212 **3 Results and Discussion**

213 **3.1 ATOFMS Results Overview**

214 ATOFMS-derived mass concentrations of $PM_{2.5}$ were in reasonable temporal agreement with 215 concurrently measured on-line beta gauge $PM_{2.5}$ mass concentrations at 2-hour resolution ($R^2 =$ 216 0.62) and in good temporal agreement with gravimetric $PM_{2.5}$ mass concentrations at 24 to 48-217 hour resolution ($R^2 = 0.91$) as shown in Figure 1. This temporal agreement is in line with previous studies involving similar scaling procedures for single particle mass spectrometers (R^2) 218 219 = 0.64-0.91) (Qin et al., 2006; Jeong et al., 2011; Healy et al., 2013; Zhou et al., 2016) and 220 indicates that changes in ATOFMS-derived mass concentrations are representative of variations 221 in ambient PM_{2.5} mass concentrations at the site during the field study.

222 The 33 distinct ATOFMS particle classes identified through clustering were grouped into ten 223 general "categories" for simplicity based on the dominant elements in their respective mass 224 spectra. Briefly, the categories identified are: K-rich, EC-rich (elemental carbon), Na-rich, Fe-225 rich, Ca-rich, V-rich, Mn-rich, OC-rich (organic carbon), Pb-rich/Li-rich and Others. The 226 relative contributions of each category in terms of mass concentration is shown in Table S1. The 227 dominant categories are K-rich (24%), EC-rich (23%) Na-rich (10%) and Fe-rich (10%), 228 followed by OC-rich (4%), Ca-rich (4%), V-rich (4%), Mn-rich (3%) and Pb-rich/Li-rich (3%). 229 The "Others" category contains particle classes with minimal mass contributions (Sulfur and 230 Amines, <1%). Temporal trends for each category are shown in Figure 2. It is clear from these 231 temporal trends that the Na-rich, K-rich and EC categories exhibit prolonged, persistent mass 232 contributions, consistent with regional background sources. Briefly, the K-rich category is 233 dominated by a single particle class (K-NO₃) often associated with regional biomass combustion 234 (Silva et al., 1999), routinely observed in regional background air in ATOFMS datasets collected 235 across Europe (Dall'Osto et al., 2009; Healy et al., 2015; Arndt et al., 2017). The ATOFMS is 236 highly sensitive to potassium (Gross et al., 2000a), and while the potassium ion is a dominant 237 feature of biomass burning particle mass spectra, it is typically present as a trace component, 238 with EC, OC and secondary inorganics dominating the composition in terms of mass (Healy et 239 al., 2013). Particles in the EC-rich category are associated with various combustion sources, 240 including vehicle exhaust and biomass burning (Spencer et al., 2006; Moffet et al., 2008a), and 241 are often observed to be internally mixed with OC and secondary inorganics (Cahill et al., 2012; 242 Healy et al., 2012). The Na-rich category is predominantly comprised of supermicron sea salt 243 internally mixed with nitrate, which is expected for a coastal urban site (Gard et al., 1998). These 244 three major particle categories are routinely observed in ATOFMS field studies. However, mass concentrations at the site during the field study.
ATOFMS particle classes identified through clustering we
es" for simplicity based on the dominant elements in th
the categories identified are: K-rich, EC-rich (eleme

245 several of the less abundant particle classes were found to be associated with local metalworking 246 activities.

247 As outlined by Setyan et al. (2019), two distinct meteorological sea-breeze periods when winds 248 were predominantly from the northeast were identified during the NANO-INDUS field campaign 249 when the sampling site was downwind of the nearby ferromanganese and steelworks facilities 250 (May 25-28 and June 02-03). During these events, particle number concentrations increased 251 considerably, with concurrent increases in non-refractory chloride, organic aerosol and metals 252 including manganese and iron. As shown in Figure 2, concentrations of several metal-rich 253 particle categories detected by the ATOFMS were enhanced during these periods (shaded in 254 grey). Particle classes associated with the ferromanganese alloy facility were identified through 255 comparison with previous scanning electron microscopy/electron dispersive X-ray spectrometry 256 (SEM-EDX) and ATOFMS analysis of raw materials (ores) and cooling, firing and smelting 257 chimney filter samples previously collected at the facility (Arndt et al., 2016). Particle classes 258 were assigned to the steelworks based on previous off-line elemental analysis of particles 259 collected downwind of this facility (Mbengue et al., 2017) and previous ATOFMS measurements 260 made in the vicinity of a large steelworks in the UK (Dall'Osto et al., 2008; Dall'Osto et al., 261 2012). nese and iron. As shown in Figure 2, concentrations of
s detected by the ATOFMS were enhanced during these
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262 **3.2 Metalworking Particle Source Identification**

263 Temporal trends for individual particle classes exhibiting elevated concentrations during the 264 prolonged periods when the site was downwind of the ferromanganese facility and the 265 steelworks (Setyan et al., 2019) are shown in Figure 2. These include metal-rich particles with a 266 variety of chemical mixing states and particles containing OC and polycyclic aromatic 267 hydrocarbons (PAHs).

268 **Mn-rich particles:**

269 Three distinct Mn-containing particle classes were identified. Mn-K particles were characterized 270 by the presence of Mn⁺ (m/z 55) and K⁺ (m/z 39) while Mn-Al-K particles exhibited a strong 271 additional signal for aluminium $(AI^+, m/z 27)$ and Mn-K-Fe particles exhibited a strong 272 additional signal for iron (Fe⁺, m/z 56) Average mass spectra for the three Mn-containing classes 273 are shown in Figure 3, and the positive ion mass spectra are consistent with ATOFMS positive

274 ion mass spectra for particles collected at the ferromanganese facility (Arndt et al., 2016) (Figure 275 S2). Mn-K particles are consistent with samples of the ores used at the facility as well as filter 276 samples from the cooling chimney. Mn-Al-K particles also resemble both ore samples and the 277 filter samples from the chimneys. Finally, Mn-K-Fe particle mass spectra are consistent with 278 mass spectra for ore samples exclusively. Ambient Mn-K, Mn-Al-K and Mn-K-Fe are all 279 enriched in nitrate (NO_2/NO_3) , m/z -46/-62) and chloride (Cl, m/z -35) relative to the facility 280 samples (Arndt et al., 2016), likely due to secondary accumulation of inorganic ions either 281 through mixing with either inorganic gases from the metalworking facilities themselves or with 282 ambient air after emission. All three Mn-rich particle classes are therefore assigned to the 283 ferromanganese facility, which has been previously identified as the major local source of Mn in 284 a previous apportionment study based on off-line filter analysis (Alleman et al., 2010).

285 **Fe-rich particles:**

286 Mass spectra for particles containing internally mixed iron, potassium and calcium (Fe-K-Ca) 287 feature a dominant signal for Fe⁺ (m/z 56) followed by K⁺ (m/z 39), Ca⁺ (m/z 40), Na⁺ (m/z 23) 288 and $Al⁺$ (m/z 27) (Figure 4). Negative ion mass spectra contain peaks corresponding to chloride 289 (CI, m/z -35) and nitrate (NO₂ /NO₃, m/z -46/-62). Fe-K-Na particles are quite similar to Fe-K-290 Ca particles, differing only in the intensities of sodium and potassium relative to iron. The mass 291 spectral features of both particle classes are similar to internally mixed Fe/K/Ca particles 292 collected from the chimneys of the ferromanganese facility (Arndt et al., 2016) (Figure S3), 293 although the ambient particles shown in Figure 4 exhibit higher signals for nitrate $(NO₂/NO₃)$, 294 *m/z* -46/-62), again indicating uptake of nitrate after emission. While similarities exist between 295 ambient and ferromanganese facility sample mass spectra for Fe-containing particles, the nearby 296 steelworks remains the largest source of particulate iron in Dunkirk. A previous analysis of the 297 steelworks particulate emissions demonstrated that most particles emitted were comprised of 298 either Fe₂O₃, CaCO₃, KCl or were graphite flakes (Flament et al., 2008). Iron-rich particles 299 similar to the ATOFMS Fe-K-Ca and Fe-K-Na classes were also detected by ATOFMS near a 300 steelworks in Port Talbot (UK) (Dall'Osto et al., 2008), where they accounted for 3% of the 301 particles detected and were attributed to the hot and cold mills. In the absence of any unique ions 302 to differentiate between iron-containing particles from the steelworks and those emitted by the 303 ferromanganese facility, and given that Fe/K/Ca particles only contributed a small fraction of the 304 particles identified in the ferromanganese facility samples (Arndt et al., 2016), it is likely that the with either inorganic gases from the metalworking facilities

Exerce emission. All three Mn-rich particle classes are theref-

acility, which has been previously identified as the major k

ionment study based on off-line

305 majority of iron-containing particles detected during the field study originated from the 306 steelworks. Setyan et al (2019) found that Mn/Fe elemental ratios in filter samples collected 307 during the periods when the receptor site was downwind of the metalworking facilities were 308 between the ratios expected for emissions from the ferromanganese facility and the steelworks, 309 suggesting that emissions from both facilities were impacting the site. Contributions of Fe from 310 both facilities were also observed in the previous source apportionment study (Alleman et al., 311 2010).

312 **K-rich particles:**

313 As shown in Figure 5, K-Cl particles are characterized by peaks for K^+ (m/z 39), Na⁺ (m/z 23), 314 Cl⁻ (m/z -35), and nitrate (NO₂⁻/NO₃⁻, m/z -46/-62). The local steelworks emits potassium 315 chloride during the sintering process (Setyan et al., 2019), and potassium chloride emissions are 316 estimated to represent ~40% of total sintering emissions from the facility (Flament et al., 2008; 317 Hleis et al., 2013). K-Cl particles were not observed in the chimney filter or ore samples from the 318 ferromanganese facility (Arndt et al., 2016). K-Cl particles have however, been detected 319 downwind of the Port Talbot steelworks (Dall'Osto et al., 2008) and are thus assigned to the steelworks here. K-Al-Si particles are characterized by signals for $K^+(m/z)$, Na⁺ (m/z 23), Fe⁺ 320 321 (m/z 56), AlO⁻ (m/z -43), AlO(OH)⁻/SiO₂⁻/CO₃⁻ (m/z -60), SiO₃⁻ (m/z -76), AlSiO₃⁻ (m/z -103), 322 nitrate $(NO_2/NO_3, m/z - 46/–62)$, phosphate $(PO_3, m/z -79)$, and chloride $(Cl, m/z -35/–37)$ 323 (Figure 5). No aluminium signal was observed in K-Al-Si positive mass spectra, likely due to 324 matrix-related ion suppression (Reinard and Johnston, 2008). K-Cl and K-Al-Si particles were 325 not observed in the ferromanganese facility samples and are therefore assigned to the steelworks. 326 However, K-Al particles were positively identified in the ferromanganese facility firing chimney 327 samples (Figure S4). Ambient K-Al particles produced stronger signals for nitrate $(NO₂/NO₃)$, 328 m/z -46/-62) and sulfate (HSO₄, m/z -97) but both the ambient and the firing chimney K-Al 329 particles exhibited similar signals for $F(m/z -19)$. Therefore fluoride represents a useful 330 discriminatory marker for identifying potassium-containing particles from the ferromanganese 331 facility. are 5, K-Cl particles are characterized by peaks for K⁺ (*m*/
nd nitrate (NO₂/NO₃, *m/z* -46/-62). The local steelwor
ne sintering process (Setyan et al., 2019), and potassium ch
seent ~40% of total sintering emissi

332 **Ca-rich particles:**

Ca particles are characterized by a strong signal for $Ca^+(m/z, 40)$ and smaller signals for Na⁺ (m/z 23) and CaO⁺(m/z 56), while Ca-K particles contain stronger signals for Na⁺ (m/z 23) and K⁺ 334 335 (*m/z* 39) as shown in Figure 6. Based on comparison with the ferromanganese facility samples, 336 both Ca and Ca-K particles appear to originate from the firing chimney, although ambient Ca-K 337 particles exhibit higher signals for Na than the chimney samples (Figure S3). Negative ion mass spectra for both ambient Ca and Ca-K particles also exhibited higher signals for nitrate $(NO₂)$ 338 339 /NO₃, m/z -46/-62) and sulfate (HSO₄, m/z -97) than their respective chimney sample particles 340 again likely due to secondary accumulation after emission. A signal for fluoride is present in the 341 ambient Ca negative ion mass spectra $(F, m/z -19)$, consistent with the firing chimney samples, 342 however this ion is absent in the ambient Ca-K particle spectra, indicating that ambient Ca-K 343 particles may be associated with the steelworks instead. Flament et al. (2008) noted that calcium 344 salts, including calcite and lime, are used in the sintering process at the steelworks, however the 345 Ca-containing particles produced in that case are aggregates of $Fe₂O₃$, CaCO₃ and KCl. Ca-346 containing particles constituted only 8% of the particles analyzed from the steelworks sintering 347 process by Flament et al. (2008), while Ca-rich particles accounted for approximately 80% of 348 the particles in the firing chimney filter sample (Arndt et al., 2016). Ca-containing particles have 349 also been assigned to the blast furnace unit of the Port Talbot steelworks (Taiwo et al., 2014), but 350 those particles were also internally mixed with OC, and no Ca particles were reported in the 351 previous Port Talbot steelworks studies by Dall'Osto et al. (2008,2012). Thus, while there 352 remains a possibility that some of the Ca-rich particles could have originated from the 353 steelworks, the ferromanganese facility is likely the dominant source. o secondary accumulation after emission. A signal for fluorive ion mass spectra (F, m/z -19), consistent with the firin is absent in the ambient Ca-K particle spectra, indicating associated with the steelworks instead. Fl

354 **OC-rich particles:**

355 Three OC-containing particle classes were identified, of which the OC class was the dominant 356 contributor in terms of mass. Average mass spectra are shown in Figure 7. All three classes are 357 characterized by the presence of hydrocarbon fragments in the positive mass spectra $(C_2H^{3+}/C_3H^+/C_2H_3O^+/C_4H_3^+/C_5H_3+$ / $C_6H_2^+/C_7H_3^+$, m/z 27/37/43/51/63/74/87) together with K⁺ 358 $(m/z 39)$ and Na⁺ ($m/z 23$). Carbon fragment ions ($C/C_2/C_3...C_n$, m/z -12/-24/-36....-12_n), CN⁻ 359 360 (m/z -26) and sulfate (HSO₄, m/z -97) are evident in the negative mass spectra, while nitrate 361 (NO₂⁻/NO₃⁻, m/z -46/-62) is dominant in the OC and PAH classes. The presence of internally 362 mixed sulfate is consistent with emissions from the coke plant at the steelworks (Zhang et al., 363 2021). Mass spectra similar to the three OC-containing classes reported here were also observed

364 downwind of the Port Talbot steelworks (Dall'Osto et al., 2012), with *m/z* 202 assigned to 365 pyrene/fluoranthene as per previous ATOFMS field studies (Gross et al., 2000b). In the case of 366 Dall'Osto et al. (2012) OC-PAH particles were attributed to the hot and cold mills, while OC and 367 PAH particles were related to the steel/coke-making sector. Thus, all three organic carbon-368 containing classes are attributed to the steelworks here.

369 **Pb-rich and Li-rich particles:**

370 Average ATOFMS mass spectra for Pb-rich particles are shown in Figure 8, characterized by 371 internally mixed K⁺ (m/z 39), Pb⁺ (m/z 208), Cl⁻ (m/z -35) and nitrate (NO₂/NO₃, m/z -46/-62). 372 No equivalent particles were observed in the ferromanganese facility samples. Pb-containing 373 particles similar to the Pb-rich class were also observed in the Port Talbot study, and represented 374 <1% of the particles ionized. In that case Pb-containing particles were assigned to the hot and 375 cold mills (Dall'Osto et al., 2008). The Port Talbot Pb-containing particles were internally mixed 376 with KCl, which is also the case for Dunkirk as shown in Figure 8. Thus Pb-rich particles are 377 attributed to the steelworks here. IS mass spectra for Pb-rich particles are shown in Figure K^+ (m/z 39), Pb⁺ (m/z 208), Cl⁻ (m/z -35) and nitrate (NO₂ articles were observed in the ferromanganese facility sand o the Pb-rich class were also obs

Li-Zn particles contain Li⁺ (*m/z* 7), Na⁺ (*m/z* 23), K⁺ (*m/z* 39), Zn⁺ (*m/z* 65/66/68), nitrate (NO₂⁻ 378 379 /NO₃, m/z -46/-62) and phosphate (PO₃, m/z -79), as shown in Figure 8, and were not observed 380 in the ferromanganese facility samples. Particles detected by ATOFMS containing Li have been 381 assigned to coal combustion sources in previous studies (Guazzotti et al., 2003; Furutani et al., 382 2011). Zn-containing particles were also associated with the Port Talbot steelworks, but those 383 particles did not contain internally mixed Li (Dall'Osto et al., 2008). Li-Zn particles have a 384 minimal mass contribution overall and it cannot be easily determined whether they originate 385 from the ferromanganese facility or the steelworks.

386 **3.3 Source Apportionment**

387 Summing the mass concentrations of the different particle classes associated with the 388 ferromanganese facility and the steelworks enabled an estimation of the $PM_{2.5}$ associated with 389 metalworking activities at the receptor site at hourly resolution. Elevated metalworking PM_{2.5} 390 concentrations are strongly dependent on northeasterly wind directions, consistent with the 391 locations of the two metalworking facilities (including ore and coal stockpiles), as shown in

392 Figure 9. Elevated non-metalworking (background) PM2.5 mass concentrations are observed 393 under northerly, easterly and southerly wind directions, consistent with more diffuse regional 394 background sources. The contribution of metalworking activities to $PM_{2.5}$ at the receptor site 395 ranged from 1-65% over the entire measurement period, with an average contribution of 17%. 396 The average contribution for those periods when wind was predominantly from the northeast 397 (shaded grey periods in Figure 2) is higher, at 28%. The overall average contribution of 17% is 398 somewhat lower than the 28% contribution these two facilities were estimated to contribute to 399 PM₁₀ at a site approximately 3 km to the southeast through PMF analysis of filters collected in 400 2003-2005 (Alleman et al., 2010). Although the local metalworking facilities are not the 401 dominant contributor to $PM_{2.5}$ mass at the receptor site, their dominant contributions to toxic 402 metal and PAH-containing particle concentrations is of concern from an exposure perspective 403 (Schroeder et al., 1987; Durant et al., 1996; Majestic et al., 2007; Zhang et al., 2008; Hu et al., 404 2012). The metalworking facilities were also the dominant contributor to ultrafine particles at the 405 site (Setyan et al., 2019), a size fraction receiving increased attention with respect to its potential 406 toxicological impacts (Ohlwein et al., 2019; Saleh et al., 2019; Sotty et al., 2019). proximately 3 km to the southeast through PMF analysis c
man et al., 2010). Although the local metalworking fa
utor to $PM_{2.5}$ mass at the receptor site, their dominant co
containing particle concentrations is of concern

407 **4 Conclusions**

408 The unique particle mixing states associated with local metalworking facilities identified at the 409 Dunkirk receptor site will be useful for future researchers aiming to apportion particulate matter 410 to metalworking sources in other locations globally. The facilities investigated here use industrial 411 processes that are common in the pyrometallurgic industry (sintering, smelting/reduction in an 412 electric/blast furnace, refining, rolling and crushing of finished product) and therefore can be 413 considered representative of that sector. The number of distinct chemical mixing states identified 414 highlights the complexity and diversity in primary particle composition associated with 415 emissions from metalworking. Particles containing Mn and Ca were associated predominantly 416 with the ferromanganese facility, while particles containing Fe, K, Pb and PAHs were associated 417 predominantly with the steelworks. Internally mixed fluoride was also found to be a useful 418 discriminating marker for identifying ferromanganese firing chimney emissions specifically. The 419 two metalworking facilities combined are estimated to contribute 17% on average to PM_{2.5} mass 420 concentrations at the receptor site during the measurement period, but are the dominant 421 contributors of particles containing transition metals and PAHs; species associated with negative

422 health outcomes. The single particle source apportionment approach used here is also expected to 423 be useful in other regions where the relative contributions of industrial emissions, including 424 metalworking emissions, to ambient $PM_{2.5}$ concentrations need to be assessed. Knowledge of 425 these contributions is particularly valuable for designing effective local air quality strategies.

426

427 **CRediT statement**

428 **Jovanna Arndt**: Investigation, Methodology, Formal Analysis, Visualization, Data curation, 429 Writing- original draft preparation. **Robert M. Healy**: Investigation, Methodology, Formal 430 Analysis, Visualization, Writing- review and editing. **Ari Setyan**: Investigation, Methodology, 431 Writing- review and editing. **Pascal Flament**: Conceptualization, Writing- review and editing, 432 Supervision, Project Administration, Funding Acquisition. **Karine Deboudt**: Investigation, 433 Conceptualization, Writing- review and editing. **Véronique Riffault**: Investigation, Formal 434 Analysis, Conceptualization, Writing- review and editing, Supervision, Project Administration, 435 Funding Acquisition. **Laurent Y. Alleman**: Investigation, Writing- review and editing. **Saliou** 436 **Mbengue**: Investigation, Writing- review and editing. **John C. Wenger**: Conceptualization, 437 Writing- review and editing, Supervision, Project Administration, Funding Acquisition Investigation, Methodology, Formal Analysis, Visualiza

I draft preparation. **Robert M. Healy**: Investigation, M

zation, Writing- review and editing. **Ari Setyan:** Investigation, M

zation, Writing- review and editing Acq

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- 451
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718 **Figures**

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720 Figure 1: Top: Comparison of 2-hour mass concentrations of $PM_{2.5}$ measured by the beta gauge 721 instrument and derived using the ATOFMS. Bottom: Comparison of 24 to 48-hour mass 722 concentrations of $PM_{2.5}$ determined by gravimetric filter analysis and derived using the 723 ATOFMS. Red lines represent least squares regression fits. Days when wind was predominantly 724 from the northeast (the direction of metalworking facilities) are shaded grey.

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731 Figure 2: Top: Temporal trends of hourly resolution mass concentrations of the ten ATOFMS 732 particle categories. Bottom: Temporal trends of hourly resolution mass concentrations of 733 metalworking ATOFMS particle classes. Days when wind was predominantly from the northeast 734 (the direction of metalworking facilities) are shaded grey.

736 Figure 3: Average mass spectra for Mn-rich metalworking particle classes.

739 Figure 4: Average mass spectra for iron-rich metalworking particle classes.

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744 Figure 5: Average mass spectra for K-rich metalworking particle classes.

747 Figure 6: Average mass spectra for Ca-rich metalworking particle classes.

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753 Figure 7: Average mass spectra for OC-rich metalworking particle classes.

756 Figure 8: Average mass spectra for Pb-rich and Li-rich metalworking particle classes.

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762 Figure 9: Dependence of concentrations of metalworking and background ATOFMS-derived $PM_{2.5}$ mass concentrations upon local wind speed (m s⁻¹) and wind direction. Periods with wind 764 speed $< 1 \text{ m s}^{-1}$ are not included in the analysis. Plots generated using the Zefir Toolkit (Petit et 765 al., 2017) running in Igor Pro v6.37 (Wavemetrics Inc.). dence of concentrations of metalworking and background
entrations upon local wind speed (m s⁻¹) and wind direction
re not included in the analysis. Plots generated using the Z
in Igor Pro v6.37 (Wavemetrics Inc.).

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Highlights

Particles assigned to a ferromanganese facility and a steelworks based on composition

Metalworking particle composition and mixing state is complex and diverse

Metalworking emissions contributed 17% to $PM_{2.5}$ on average at the receptor site

Metalworking is the dominant source of metals and polycyclic aromatic hydrocarbons

Just not leader the polycyclic aromat

Declaration of interests

 \boxtimes The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

☐The authors declare the following financial interests/personal relationships which may be considered as potential competing interests:

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