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## **TiO<sub>2</sub> –SiO<sub>2</sub> nanocomposite thin films deposited by direct liquid injection of colloidal solution in an O<sub>2</sub> /HMDSO low-pressure plasma**

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1 TiO<sub>2</sub>-SiO<sub>2</sub> Nanocomposite thin films deposited by  
2 Direct Liquid Injection of colloidal solution in an  
3 O<sub>2</sub>/HMDSO low-pressure plasma

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11 **Keywords:** low-pressure plasma processing, direct-liquid injection, colloidal solution, TiO<sub>2</sub>  
12 nanoparticles, TiO<sub>2</sub>-SiO<sub>2</sub> nanocomposite thin films

13 **Abstract:** TiO<sub>2</sub> nanoparticles (NPs), 3 nm in size, were injected inside a very-low-pressure O<sub>2</sub>  
14 plasma reactor using a liquid injector and following an iterative injection sequence.  
15 Simultaneously, hexamethyldisiloxane (HMDSO) vapor precursor was added to create a SiO<sub>2</sub>  
16 matrix and a TiO<sub>2</sub>-SiO<sub>2</sub> nanocomposite (NC) thin film. Both the liquid injection and vapor  
17 precursor parameters were established to address the main challenges observed when creating  
18 NCs. In contrast to most aerosol-assisted plasma deposition processes, Scanning/Transmission  
19 Electron Microscopy (S/TEM) indicated isolated (i.e. non-agglomerated) NPs distributed in a  
20 rather uniform way in the matrix. The fraction of the TiO<sub>2</sub> NPs inside the SiO<sub>2</sub> matrix was  
21 estimated by SEM, Spectroscopic Ellipsometry (SE), and X-ray Photoelectron Spectroscopy.  
22 All techniques provided coherent values, with percentages between 12 and 19%. Despite the  
23 presence of TiO<sub>2</sub> NPs, SE measurements confirmed that the plasma-deposited SiO<sub>2</sub> matrix was  
24 dense with an optical quality similar to the one of thermal silica. Finally, the percentage of TiO<sub>2</sub>  
25 NPs inside the SiO<sub>2</sub> matrix and the effective refractive index of the NCs can be tuned through  
26 judicious control of the injection sequence.

27

## 1. Introduction

Nowadays, there is an ongoing demand for smaller, faster, and low-cost integrated systems. Achieving this often requires new materials and structures that combine the more conventional ones. In this context, nanocomposite (NC) thin films formed of nanoobjects embedded in a matrix have gained great attention, as the properties of the resulting composite may be more likely to resemble those of the interface zones rather than those of the original constituents[1]. The high surface-area-to-volume ratio in NCs can also lead to enhanced macroscopic properties compared to the bulk materials.

One of the most significant challenge in the development of NC thin films is the control of the growth mechanisms, the final morphology and the incorporation of the nanoobjects using reproducible, versatile and low-cost processes. Several chemical and physical methods have been used in recent years. For example, sol-gel is a well-established process based on hydrolysis and condensation reactions.[2,3] The main drawback of this approach is the need of several steps to produce NC thin films. In addition, the toxicity of the nanoparticles (NPs) during the drying of the film[4] limits the industrialization. Vacuum processes such as PECVD or CVD have also demonstrated their ability to synthesize NC thin films by controlling the plasma- or thermally-activated reaction pathways.[5,6] In parallel, sputtering approaches have also been used to create metal, metal-oxide or metal-nitrides NC thin films. [4,7]

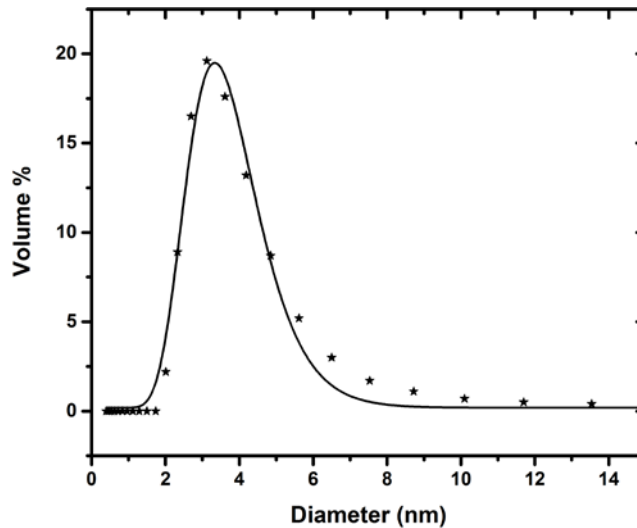
Recently, there have been some attempts using hybrid (aerosol-assisted, direct liquid injection etc.) approaches for the creation of NC thin films. Profili et al. [8] implemented a one-step approach for the deposition of ZnO-SiO<sub>2</sub> NC coatings on wood by the atomization of a stable suspension of ZnO dispersed in hexamethyldisiloxane (HMDSO) in an atmospheric pressure dielectric barrier discharge (DBD). Ross et al.[9] elaborated composite films using an ultrasonic atomizer and a PECVD system using Tetraethoxysilane (TEOS) as a matrix precursor and polystyrene nanoparticles with a mean size of 96 nm. The pressure in the specific system was varied between 100 and 500 mTorr. More recently, Kahn et al.[10] proposed a new method that allowed to synthesize NPs from organometallic precursor in a reactor-injector prior to their injection in a RF low-pressure plasma reactor at 750 mTorr. The resulted NC thin films consisted of small ( $6 \pm 3$  nm) and isolated ZnO NPs homogeneously dispersed in an amorphous hydrogenated carbon matrix. This increased interest in hybrid approaches led to the development of new direct liquid injection systems[11] and studies regarding the interactions of the liquid droplets with the low pressure plasma systems [12] (misty plasmas). As a general trend, most of these studies focus on the deposition of inorganic NPs-organic matrix NC film. The majority of the reposted works involve high-pressure and atmospheric-pressure plasma system. These systems have been thoroughly studied mainly because of the technically-challenging conditions existing at very low pressures (*e.g.* risk of contamination of the vacuum chamber, malfunction of the turbomolecular pump, fast liquid vaporization without reaching the substrate).[13,14] In most cases, however, the NPs in NC thin films are agglomerated, and their size is increased compared to the one in the colloidal solution before their injection in the plasma. [13,15–22]

72 The main bottlenecks for producing NC thin films are incorporating a high-volume fraction,  
73 preventing the agglomeration of NPs and dispersing them homogeneously in the matrix. Using  
74 a very-low-pressure system, the objective is to deposit thin films incorporating 3-nm TiO<sub>2</sub> NPs  
75 in a high-quality inorganic SiO<sub>2</sub> matrix. This is accomplished by the direct liquid injection of  
76 the 3-nm TiO<sub>2</sub> NPs colloidal solution without a carrier gas and the addition of HMDSO vapor  
77 precursor in the very-low-pressure O<sub>2</sub> plasma. Furthermore, as several works involving  
78 atmospheric-pressure plasma systems reported challenges to deposit in one-step and acquire  
79 good quality NC films, the differences and prospects in using very-low-pressure hybrid systems  
80 is investigated. The evaluation of the NC's quality, homogeneity and distribution of the TiO<sub>2</sub>  
81 NPs are achieved by various characterization techniques such as high-resolution Scanning  
82 Transmission Electron Microscopy (HR-STEM), Energy Dispersive X-ray spectrometry  
83 (EDX), Atomic Force microscopy (AFM) and X-ray photoelectron spectroscopy (XPS).  
84 Finally, with this work, besides establishing this novel and versatile approach, the objective is  
85 to identify experimental conditions for acquiring thin films oriented towards optical  
86 applications. Therefore, the optical characteristics of the film, such as its effective refractive  
87 index, are being extracted with Spectroscopic Ellipsometry (SE).

## 88 89 **2. Experimental Details**

### 90 91 **2.1 TiO<sub>2</sub> NPs colloidal solution**

92  
93 The colloidal solution used for the injection of the TiO<sub>2</sub> NPs in the very-low-pressure plasma  
94 has been synthesized following a specific protocol developed at the Institut des Matériaux Jean  
95 Rouxel.[23,24] The NPs enclosed in the colloidal solution were well-defined and  
96 monodispersed with an anatase crystalline structure. Their mean hydrodynamic diameter  
97 (including species of the solvent surrounding the NPs to counterbalance surface charge)  
98 obtained by Dynamic Light Scattering (DLS) with Zetasizer NanoZS (Malvern) was centered  
99 on 3.7 nm (Figure 1). The double layer surrounding the NPs has been estimated at 1 nm.[24]  
100 Therefore, the actual TiO<sub>2</sub> nanocrystallite diameter lay at 2.7 nm. As synthesized, the solvent  
101 of this solution was a mixture of 4-methyl-1,3-dioxolan-2-one also known as propylene  
102 carbonate (PC) and 1,2-propanediol or propylene glycol (PG), with a molar fraction of  
103 27%:73% respectively. From previous studies of our group [23,25], we identified that some  
104 solvent molecules remain attached to the surface of the NPs, creating an 4.7 Å hydroxylated  
105 and organic shell around TiO<sub>2</sub> NPs.



107  
 108 **Figure 1:** Hydrodynamic size distribution of the TiO<sub>2</sub> NPs inside the as-prepared (PC-PG)  
 109 colloidal solution deduced from DLS. Line corresponds to a Log-Normal distribution fit, based  
 110 on which the mean diameter size was extracted.

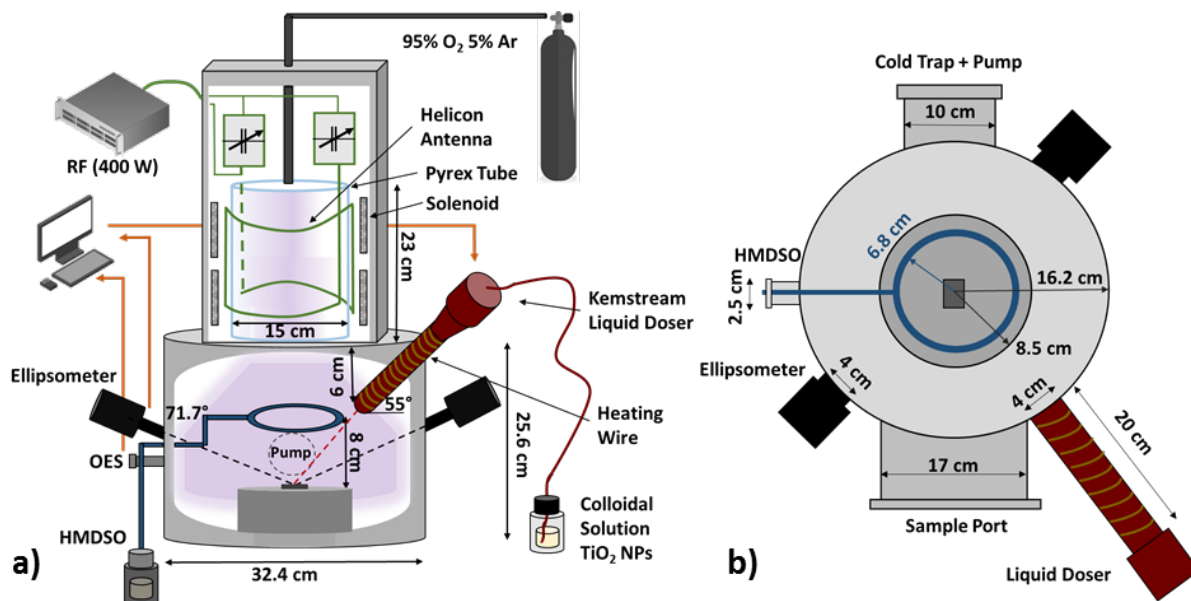
111 For the effective injection and evaporation of the colloidal solution in the very-low-pressure  
 112 PECVD system, the dilution of the colloidal solution with a more volatile one was carried out.  
 113 Therefore, the final colloidal solution used was a mixture of PC:PG:Methanol with a volume  
 114 fraction of 20.25:54.75:25 . The size distribution of TiO<sub>2</sub> NPs in the diluted solution was  
 115 verified with DLS and was not altered by either dilution or long-time storage..

116  
 117 **2.2 Deposition of TiO<sub>2</sub>-SiO<sub>2</sub> NC thin films using the hybrid approach**

118  
 119 For the elaboration of the TiO<sub>2</sub>-SiO<sub>2</sub> NC thin film, the colloidal solution was injected in a  
 120 very-low-pressure PECVD reactor. The schematic representation of the hybrid experimental  
 121 setup is given in Figure 2. The reactor comprises a very-low-pressure high-density Inductively  
 122 Coupled Plasma (ICP) source created at 13.56 MHz (400 W) in a fused silica discharge tube  
 123 using a helicon antenna which is also connected to a matching network. A static magnetic field  
 124 (50 G) is applied to the source in order to confine the electrons and densify the plasma. The  
 125 plasma then expands into a stainless-steel processing chamber where the substrate holder is  
 126 located. The substrate holder is 17 cm in diameter located in the middle of the processing  
 127 chamber with a distance of 16.8 cm from the bottom of the ICP antenna. The Si substrates used,  
 128 which were positioned in the middle of the substrate holder, were 1 cm x 2 cm in dimension  
 129 and their temperature was measured at 400 K through temperature-sensitive labels located at  
 130 the substrate's backside. In this system, plasma gases (95% O<sub>2</sub>, 5% Ar here) are introduced at  
 131 the top of the plasma source and the high-vacuum pumping system is located 11 cm from the  
 132 bottom of the processing chamber. The outlet of the vacuum pumping system that includes also  
 133 a cold trap, is 10 cm in diameter. In Figure 2b, its location related to the substrate holder and  
 134 the liquid doser can be seen. More information about the reactor can be found in [26,27].

135  
 136 For the deposition of the SiO<sub>2</sub> matrix, HMDSO vapor was introduced from a heated-  
 137 container (at 35.8 °C) without any carrier gas through a distribution ring (indicated with blue  
 138 in Figure 2), located 8 cm above the substrate. The injection of the TiO<sub>2</sub> colloidal solution was

139 achieved through a separate industrial direct-liquid doser (indicated with red in Figure 2). It is  
 140 an injection system from Kemstream that features a normally closed fast solenoid valve (or  
 141 injector), with a small conductance (flow coefficient), commonly called Cv. This valve is  
 142 controlled in pulsed regime. The orifice of the injection system includes 8 holes with a diameter  
 143 of 70  $\mu\text{m}$  each. This injection system was mounted on the PECVD reactor with an angle of 55°  
 144 with respect to the substrate holder. The distance of the injector to the reactor was fixed at 20  
 145 cm and therefore the distance of the injector to the substrate can be calculated at 48.24 cm. The  
 146 pipe connecting the injector to the reactor was heated (yellow lines indicating the heating wire  
 147 in Figure 2) at 423 K to enhance the evaporation process of the produced droplets.  
 148



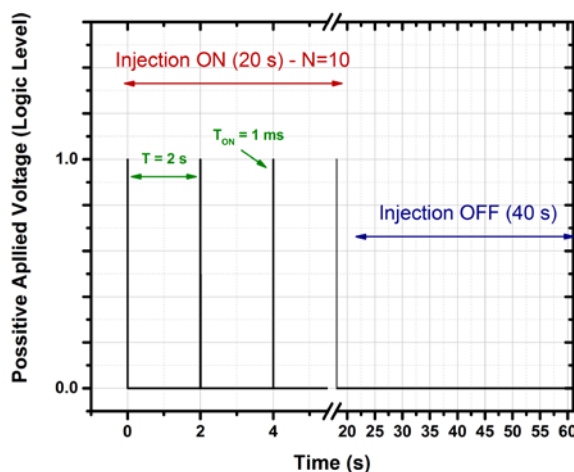
149  
 150 **Figure 2:** Schematic of the hybrid experimental setup including the low pressure ICP O<sub>2</sub>  
 151 reactor, the liquid doser apparatus (with red) and the HMDSO vapor distribution system (with  
 152 bleu) at a) side view and b) top view. The *in situ* characterization techniques are also indicated  
 153 on this schematic.  
 154

155 For maintaining very-low pressure during the deposition (3 mTorr) and therefore obtaining  
 156 a high-quality inorganic matrix, no additional carrier gas has been used to assist the injection  
 157 of TiO<sub>2</sub> NPs. In such conditions, the droplet creation and the formation of the aerosol were  
 158 mainly driven by the difference in pressure between the reservoir of the colloidal solution (760  
 159 Torr) and the plasma reactor (3 mTorr).[28] For this specific pressure difference, the mean  
 160 droplet size obtained by Phase Doppler Anemometry (PDA) was 65.0  $\mu\text{m}$  with a standard  
 161 deviation of 29  $\mu\text{m}$ . The opening of the injection valve was triggered by applying a positive pulse  
 162 controlled by the VapSoft software of the direct-liquid injector. It is worth highlighting that  
 163 during each pulsed injection of the colloidal solution in the plasma reactor the pressure  
 164 instantaneously increased from 3 to 15 mTorr due to solvent evaporation[12] and then returned  
 165 back to the steady-state pressure (3 mTorr) within less than 2 s. The pressure variation was  
 166 detected from the capacitance diaphragm gauge (MKS Instruments France SA) and was read  
 167 on the go. Additionally, the 15 mTorr was the maximum value that was provided by the gauge  
 168 reading. Upon injection and pressure increase, the intensity of the ICP O<sub>2</sub> plasma is decreased  
 169 retrieving within less than 2 seconds its usual brightness. Despite this fact, the plasma is never

170 extinguished remaining ignited during the total duration of the process. As a matter of fact, the  
171 intensities of O and Ar lines (at 394.5 nm and 419.8 nm, respectively) measured by time  
172 resolved OES are supplied as Electronic Supporting Information (ESI). These results, indicate  
173 that during each injection (within the DLI-ON), the emission intensities of the Ar and O lines  
174 are decreased trying to retrieve their steady state value within the pulses (Figure S1). This  
175 variation, especially for the Ar emission intensity is related to the pressure variation. Both  
176 intensities never reach a zero value verifying once more that the plasma remains ignited during  
177 the total duration of the hybrid deposition process. Regarding the RF power supply, during the  
178 droplet injection, a reflected power with a mean value of 2 W is provided by the CESAR RF  
179 Power Generator.

180  
181 For deposition experiments, the total sequence length was set at one minute (Figure 2). More  
182 specifically, during the first 20 s,  $N=10$  pulses of 1 ms ( $T_{ON} = 1$  ms) were triggered each two  
183 seconds ( $T=2$  s). Thus, after this first-time lapse of  $N*2$  s = 20 s, the valve remained closed for  
184  $60-N*2$  s = 40 s ( $T_{OFF}$ ). Such sequence was repeated for 50 minutes. The solvent mass rate was  
185 estimated at 2.33 mg per pulsed injection and therefore 23.3 mg per minute for  $N=10$  cycles.  
186 During this intermittent mode of injection (Figure 3), the oxygen mass flow rate of the working  
187 gas was set at 24 SCCM (standard cubic centimeters per minute) while the continuous mass  
188 flow rate of HMDSO vapor was maintained at 0.11 SCCM. The fraction between HMDSO and  
189  $O_2$  was chosen in a way to produce a silica-like matrix with a very low carbon content[29].  $O_2$   
190 was also used to ensure the removal of solvent-based organic moieties surrounding  $TiO_2$   
191 NPs[30].

192



193  
194 **Figure 3:** Description of the plasma deposition sequence including the first part of pulsed  
195 injection of the colloidal solution (10 cycles of  $T_{ON}$  with an off time of 2 s between each pulse  
196 for a total of 20 s) and the second part without liquid injection (for a total off time  $T_{OFF} = 40$ s).

### 197 2.3 Characterization of the NC thin films

198  
199 The cross section and distribution of the  $TiO_2$  NPs inside the  $SiO_2$  matrix were investigated  
200 (Figure 4 a, b ,c) using transmission electron microscopy (S/TEM Themis Z G3, 300kV,  
201 Thermo Fisher Scientific). A typical image of high-angle annular dark-field scanning  
202 transmission electron microscopy (HAADF-STEM, 28.5 mrad, probe corrected) is reported in

203 Figure 4 d. The elemental mapping (Figure 4 e f g) was performed using EDX (energy  
204 dispersive X-ray spectrometer) analysis. For the effective characterization of the samples  
205 through TEM, the deposition was carried on Tin doped Indium oxide deposited on polyethylene  
206 terephthalate (ITO (50 nm) / PET (125  $\mu\text{m}$ ), Sheldahl). Cross-sections of the sample were  
207 prepared by using an ultramicrotome (Leica UC7/FC7) equipped with diamond knife  
208 (Diatome). The morphological analysis of the  $\text{TiO}_2\text{-SiO}_2$  NC film on Si was carried out through  
209 Scanning Electron Microscopy (SEM) with a JEOL JSM 7600F, in a secondary electron mode,  
210 operating at 5 kV. The chemical analysis by EDX was conducted with a SAMx SDD system  
211 on a SEM JEOL JSM 5800LV. To increase the EDX signals, a dedicated thicker film was  
212 prepared (224 nm). To ensure that no additional Si coming from the substrate would interfere  
213 with the quantification of the silica matrix the dedicated thicker NC film was deposited on  
214 ITO/PET substrate. To probe different thicknesses, EDX spectra were recorded at three  
215 accelerating voltages: 15 kV, 10 kV and 7 kV and a magnification of X1000 ( $2552 \mu\text{m}^2$ ) was  
216 chosen to smooth any local anomalies of chemical concentrations. The software Stratagem was  
217 used for EDX measurements because first the volume irradiated by the electron beam and  
218 second the X-ray collected are related to a larger depth than the thickness of the top layer. [31]  
219 The atomic percentages were determined by considering a theoretical model for the probed  
220 zone [32]. For this, three layers were considered: first, 224 nm of NC (Ti, Si, O), second, 50  
221 nm of ITO (In, Sn, O), and third, PET (C, O). The different atomic percentages were fitted to  
222 reproduce the intensity of EDX signals as a function of the accelerating voltage, accounting for  
223 the thicknesses of the NC and ITO and attenuation of signals throughout the different layers.  
224 The determination of carbon and oxygen was not reliable because the corresponding signals  
225 may come from ITO and PET underneath (one must also consider that the effective thickness  
226 of PET analyzed by EDX depends on the accelerating voltage). However, the precautions taken  
227 before acquisition and the good fit between theoretical curves generated by Stratagem and  
228 experimental values make it possible to rely on the results concerning silicon and titanium  
229 atomic percentages.

230  
231 The optical characteristics and composition of the  $\text{TiO}_2\text{-SiO}_2$  NC thin films deposited on  
232 silicon substrates were also examined through a rotating compensator spectroscopic  
233 ellipsometer (J.A. Woollam M-2000). The measurements were conducted in situ (i.e. during  
234 plasma deposition) in the wavelength range of 245 to 1000 nm at  $71.7^\circ$  incident angle.

235  
236 Atomic Force Microscopy (AFM) was used in tapping mode to investigate the nanostructured  
237 topography and roughness of the NC thin films. AFM Nanowizard II (JPK Instruments), Al-  
238 coated cantilever (PPP-NCHR-50 from NANOSENSORS) with a free resonance frequency of  
239 330 kHz and a typical spring constant of about 42 N/m were used. The size distribution of the  
240 NPs was deduced from the AFM scans by using the ImageJ software after adjustment of the  
241 threshold and transformation in binary format.

242  
243 Raman spectroscopy measurements were performed using a microconfocal Raman  
244 microscope (Renishaw inVia). The instrument was equipped with a double edge filter to  
245 eliminate the Rayleigh scattering and a charged couple device (CCD) camera working at a  
246 temperature of 220 K with a 1024x256 pixel array. The setup was composed of a confocal



247 microscope that was equipped with an automated XYZ table and an X50 magnification  
248 objective. The laser excitation source used was set at 514 nm and the focused power was  
249 adjusted to avoid any degradation of the samples. Therefore, the power was fixed at 1.5mW  
250 and a recording time of 1 s with an accumulation of 10 measurements was used.

251  
252 The chemical characterization of the topmost surface was investigated with X-ray  
253 photoelectron spectroscopy (XPS) on a Kratos Nova spectrometer with Al K $\alpha$  radiation at  
254 1486.6 eV. Neutralization of the surface was applied to avoid differential charging effect.  
255 Survey spectra were measured from 1200 to -2 eV (Binding Energy) using a pass energy of 160  
256 eV whereas high-resolution spectra over specific regions of interest were recorded with a pass  
257 energy of 40 eV. Short time measurements were recorded (dwell of 100 ms per 0.1 eV step-  
258 scan for O1s, Ti 2p and 200 ms for C1s). After checking that no X-ray damage was induced,  
259 the scans were summed up to increase the signal-to-noise ratio. The decomposition and fitting  
260 of the measured XPS peaks was carried out using CasaXPS software [33]. After accounting for  
261 the background using a U2-Tougaard background, Gaussian-Lorentzian (30-70%) functions  
262 were selected to model the different components. The signals were calibrated in energy  
263 according to the position of the C 1s band at the binding energy of 285 eV.

264  
265 To be able to compare the results obtained from different characterization techniques, detailed  
266 calculations have been achieved (which are supplied as Electronic Supporting Information) to  
267 transform at% of elements to vol% of oxides.

268

### 269 **3. Results**

270

#### 271 **3.1 Structural characterization of the TiO<sub>2</sub>-SiO<sub>2</sub> NC thin films**

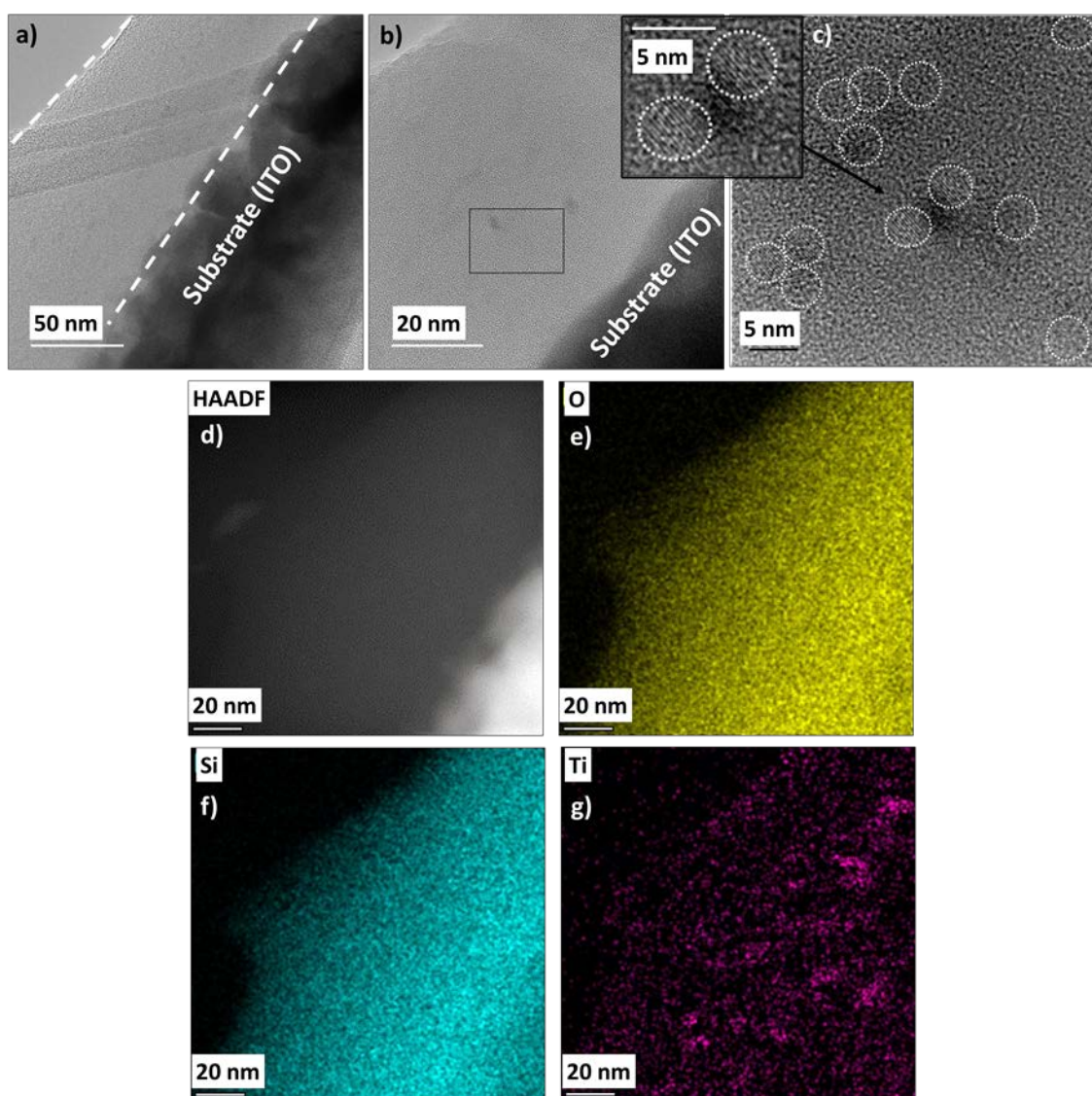
272

273 Cross section structural analysis of the plasma-deposited NC thin film was performed using  
274 TEM and STEM (Figure 4). Through multiple measurements on low magnification images (Fig.  
275 4a), the thickness of the film was estimated at 85 +/- 6 nm (a growth rate of 1.7 nm/min for a  
276 total plasma deposition time of 50 min). This value is comparable to the one achieved in  
277 O<sub>2</sub>/HMDSO plasmas without injection of the TiO<sub>2</sub> colloidal solution[29]. Using high-  
278 resolution images on the rectangular area (black contour) shown in Fig. 4b and based on the  
279 contrast between the amorphous matrix and the basal planes of the nanocrystals, some TiO<sub>2</sub>  
280 NPs can be identified (enlarged NPs as an inset in Figure 4) and stressed with white-rounded  
281 contour shapes (Figure 4c). Interestingly, no agglomeration of these NPs is observed but rather  
282 an assembly of small and isolated TiO<sub>2</sub> NPs in the SiO<sub>2</sub> matrix. Using ImageJ software, the  
283 size of each of these NPs (14 nanoparticles identified in the specific zone probed by electron  
284 microscopy) was extracted and their mean value was found to be at  $2.7 \pm 0.4$  nm. By comparing  
285 this value with the hydrodynamic diameter of TiO<sub>2</sub> NPs in the colloidal solution obtained by  
286 DLS (2.7 nm + 1 nm, Figure 1), it can be confirmed that the NPs' size remains intact following  
287 the aerosol formation and plasma deposition. This is a remarkable result with respect to other  
288 experiments on hybrid plasma deposition of NC thin films for which agglomeration phenomena  
289 were deemed unavoidable[13,15–22]. Moreover, TiO<sub>2</sub> NPs displayed in Figure 4c seem  
290 dispersed in a rather random way in the SiO<sub>2</sub> matrix, with no evidence of preferential

291 incorporation sites. No cracking of the SiO<sub>2</sub> layers is visible, showing that the insertion of the  
292 NPs does not strongly affect the growth of the silica layer.

293  
294 The typical HAADF-STEM image given in the same figure confirms that no bright contrast  
295 associated to possible agglomeration of the electronically denser TiO<sub>2</sub> NPs is detected,  
296 indicating the homogeneity of the NPs dispersion, at this scale of observation. Moreover, EDX  
297 elemental mapping also confirms that both oxygen and silicon (Fig.4 e,f) are homogeneously  
298 distributed along the cross section of the NC thin film. On the other hand, Ti (Fig. 4g) signal,  
299 which is only linked to the TiO<sub>2</sub> NPs, is identified with some small dots verifying the TEM  
300 images (Fig. 4c) and their distribution inside the film. Since the data are a projection of 80 nm  
301 thin sample, the reinforcement in contrast, (in some small parts of Fig.4g) may come from  
302 particles distributed along this thickness.

303

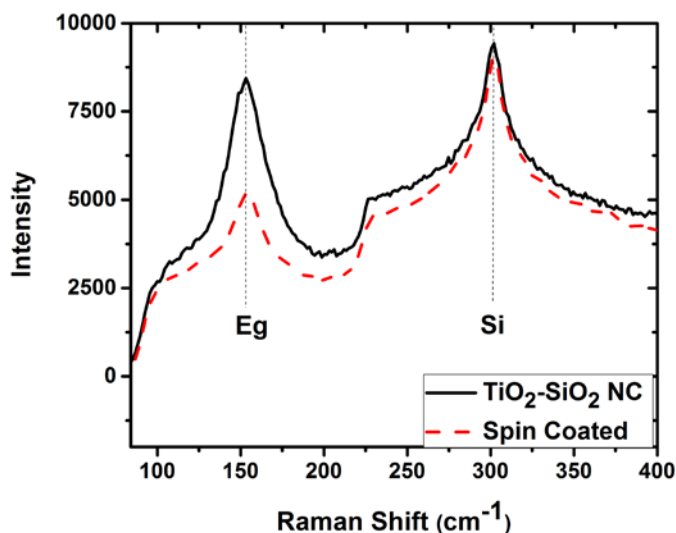


304  
305 **Figure 4:** TEM cross section of the TiO<sub>2</sub>-SiO<sub>2</sub> NC thin film a) in low magnification indicated  
306 with white dashed lines, b) high magnification and c) higher magnification image focusing on  
307 the distributed TiO<sub>2</sub> NPs inside the SiO<sub>2</sub> matrix. An inset is given with higher magnification

308 focusing on the basal planes of the anatase nanoparticles. d) Typical HAADF-STEM image of  
309 the SiO<sub>2</sub>-TiO<sub>2</sub> NCs in a selective area. Elemental mapping for e) O, f) Si and g) Ti.

310 The crystalline phase of the TiO<sub>2</sub> NPs was identified using Raman spectroscopy. Figure 5  
311 presents the Raman spectra of the plasma-deposited NC thin film (black line) and the TiO<sub>2</sub> NPs  
312 spin coated thin films (red line) using the as synthesized colloidal solution, with a focus on the  
313 Eg anatase peak. The analysis was limited to using the peak positions since extreme caution is  
314 required when using the Raman intensities of very thin films. Raman modes can be assigned to  
315 the Raman spectra of the anatase crystal: 144 (Eg), 200 (Eg), 400 (B1g), 513 (A1g), 518 (B1g)  
316 and 642 cm<sup>-1</sup> (Eg) [34,35]. In this figure, it can be observed that the Eg peak lies at 154 cm<sup>-1</sup>, a  
317 value shifted from the expected anatase peak at 144 cm<sup>-1</sup> [36,37]. It is well-established  
318 experimentally that Raman spectra of low-dimensional semiconductor crystals are modified  
319 when compared to the one of bulk crystals: (1) the Raman peaks are asymmetrically broadened  
320 and (2) their positions are shifted [38,39]. The former is not observed mainly due to the size  
321 distribution of the specific nanocrystals. This effect, results in masking the confinement effect  
322 and averaging the Raman response [40]. However, the results presented in Figure 5 (black line)  
323 confirm the strong shift of the Eg band that the plasma-deposited TiO<sub>2</sub> NPs embedded in the  
324 SiO<sub>2</sub> matrix are anatase crystallites and in nanometric size. In addition to this, the Raman  
325 spectrum of the TiO<sub>2</sub> NPs thin film through the spin-coated technique is given with red color.  
326 The fact that for both cases the anatase peak is found in the same position (154 cm<sup>-1</sup>) confirms  
327 that upon the colloidal solution dilution, the TiO<sub>2</sub> NPs injection and the SiO<sub>2</sub> matrix deposition,  
328 the nano crystallite is not affected.

329



330  
331 **Figure 5:** Raman spectra indicating the Anatase Eg band at 154 cm<sup>-1</sup> and the second order Si  
332 band peak at 303 cm<sup>-1</sup> for the TiO<sub>2</sub>-SiO<sub>2</sub> NC thin film through the hybrid approach (black color)  
333 and the spin coated TiO<sub>2</sub> NPs thin film using the as synthesized colloidal solution.

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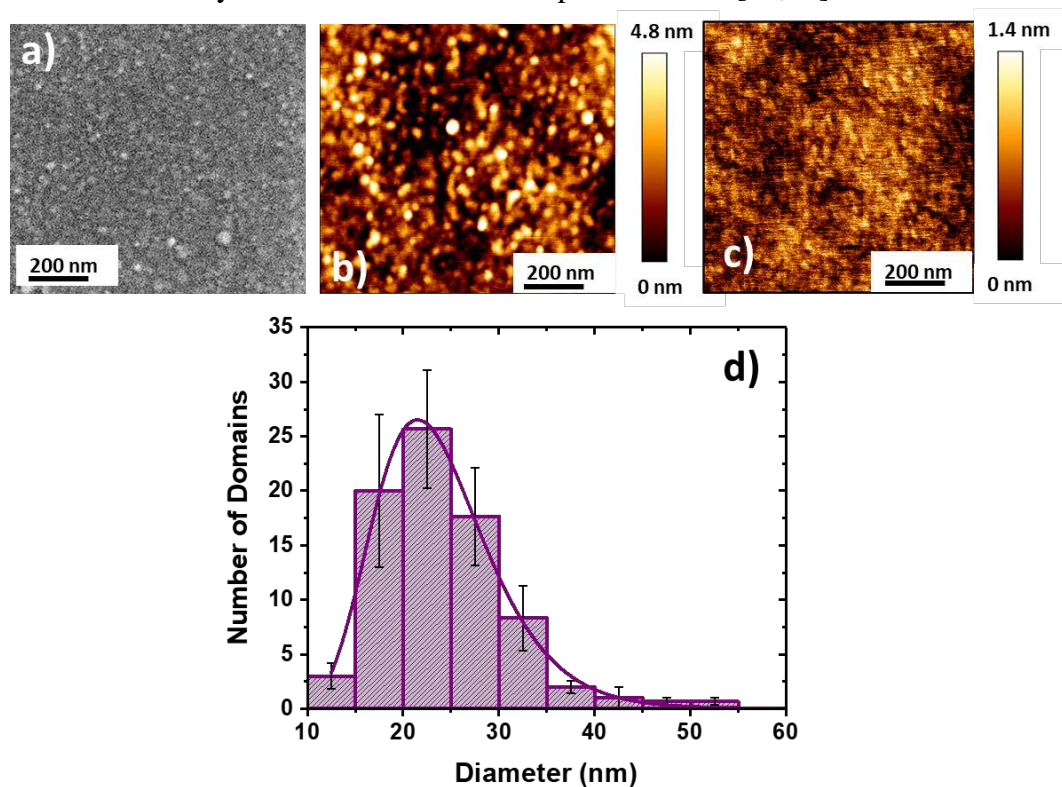
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### 337 3.2 Morphological analysis of the TiO<sub>2</sub>-SiO<sub>2</sub> NC thin films

338

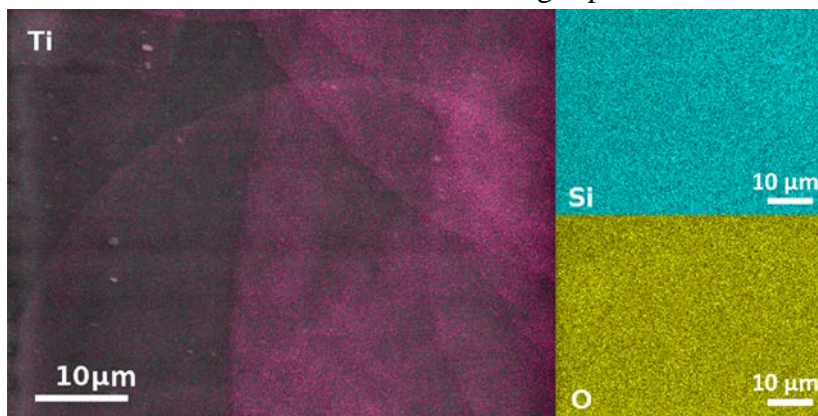
339 The surface morphological analysis of the TiO<sub>2</sub>-SiO<sub>2</sub> NC thin film was investigated by SEM  
 340 and AFM as shown in Figure 6. From the SEM top view image (Fig. 6a) in high magnification,  
 341 the surface of the film is homogeneous and nanostructured, revealing some rounded domains.  
 342 Since the resolution of the SEM is at 10 nm, one can understand that the exact identification  
 343 and quantification of these domains is impossible. These same rounded domains are highlighted  
 344 in the AFM measurements (bright yellow-white color in Fig 6b). They are located higher on  
 345 the z axis and their size is higher than the one of the initial NPs in solution measured by DLS  
 346 in Figure 1. From detailed analysis of multiple AFM scans at low and high magnification, the  
 347 mean size of the domains was extracted and plotted as a histogram in Fig 6c. The fitting of a  
 348 LogNormal distribution provided a mean size of 23 nm of these rounded domains and a standard  
 349 deviation of 0.3 nm. The Rq (RMS) roughness of the NC film was also deduced. A value of 2  
 350 nm was obtained, which was far different from the measured roughness of the silica thin film  
 351 (Fig. 6c, Rq (RMS) = 0.3 nm) obtained by PECVD with the same reactor in comparable  
 352 experimental conditions. The AFM scan for this case is given in Figure 6c, revealing the much  
 353 smoother topography having a Rq (RMS) roughness lower than 5 nm. Since the deposition on  
 354 flat silicon substrate should not lead to additional roughness or nanostructuration, especially for  
 355 the deposition of amorphous SiO<sub>2</sub> (matrix), the nanometric pattern observed by AFM must be  
 356 linked to the presence of TiO<sub>2</sub> NPs inside the SiO<sub>2</sub> matrix. It is worth highlighting, however,  
 357 that the mean domain size displayed in Figure 6 (23 nm) is much larger than the mean size of  
 358 TiO<sub>2</sub> NPs obtained from both DLS (Figure 1) and TEM (Figure 4a). This is because TiO<sub>2</sub> NPs  
 359 are located at different locations across the thickness of the SiO<sub>2</sub> matrix (85 +/- 6 nm) and the  
 360 PECVD process used in this study provides highly conformal plasma deposition.[41–45]  
 361 Therefore, the deeper the particle is located inside the matrix, the larger is the size of the  
 362 structure observed by SEM and AFM on the topmost surface[15,46].



363  
 364

365 **Figure 6:** a) SEM secondary electron image of the surface, b) AFM scan (1  $\mu\text{m}$  x 1  $\mu\text{m}$ ) of the  
 366  $\text{TiO}_2\text{-SiO}_2$  NC thin films, c) AFM scan (1  $\mu\text{m}$  x 1  $\mu\text{m}$ ) of  $\text{SiO}_2$  only when deposited on Si  
 367 substrate and d) size distribution of the rounded domains deduced from the AFM scans.

368 From the high-magnification and highly-localized analysis obtained both from TEM and  
 369 SEM/AFM, the plasma-deposited  $\text{TiO}_2\text{-SiO}_2$  NC thin films indicated a rather homogenous  
 370 spatial distribution. To have an overall view of this film, low-magnification EDX mappings of  
 371 the film's surface were recorded. To extract adequate resolution for each element, a thicker film  
 372 was especially deposited on the ITO/PET substrate (224 nm). The results are presented in Figure  
 373 7. While homogeneous distributions are observed for silicon and oxygen elements over the  
 374  $2552 \mu\text{m}^2$  surface, abnormal rounded-ring shapes with diameters of several tens of micrometers  
 375 can be seen for the titanium element. Such feature suggests the well-known coffee-ring effect,  
 376 happening when the solute migrate in the edge of the evaporating droplet due to the formation  
 377 of preferential flow and faster evaporation at this area. [47–49] Based on such large-scale SEM  
 378 and EDX observations, it seems that  $\text{TiO}_2$  species are provided to the substrate during growth  
 379 as droplet and that evaporation and removal of organic species from the solvent[30] occur, at  
 380 least partially, on the substrate surface rather than in the gas phase.



381 **Figure 7:** SEM backscattered electrons image and EDX map of Ti superimposed (left) and  
 382 corresponding EDX maps of Si and O (right).  
 383

384 A quantitative EDX analysis was conducted on the large region ( $2552 \mu\text{m}^2$ ) avoiding an area  
 385 exhibiting droplet traces such as the ones shown in Figure 7. In order to avoid all inhomogeneity  
 386 aspects (in particular those related to the Ti element), only averaged values are shown in Table  
 387 1. The results are presented only for Ti and Si due the weak energy of the carbon and oxygen  
 388 EDX peaks and the strong interactions throughout the layers for these elements (see details in  
 389 Sec. 2.2). From such analysis, 15% Ti and 85% Si is obtained (atomic %). Based on the  
 390 calculations provided in the ESI, the volume percentage corresponding to 15 at% is estimated  
 391 to reach 16 %.  
 392

394 **Table 1:** Quantitative results by EDX on large surface ( $2552 \mu\text{m}^2$  - magnification X1000).  
 395

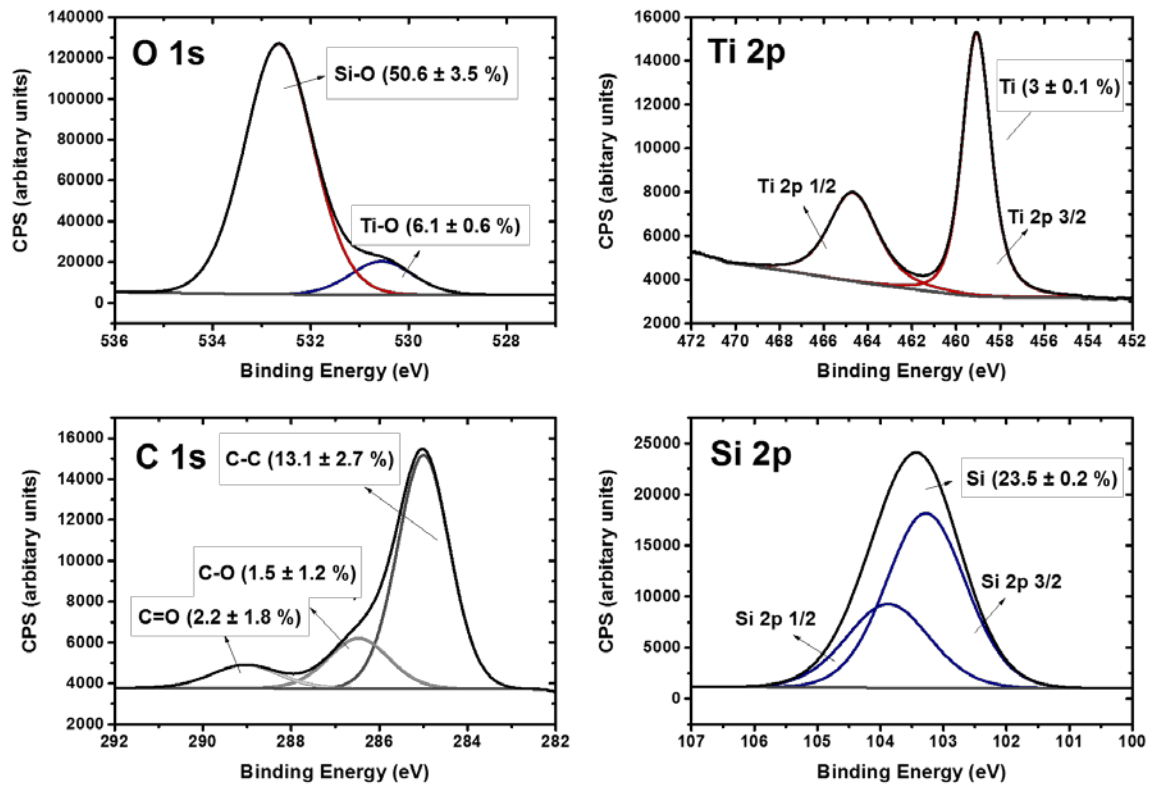
	Weight %	Atomic %
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Silicon	76.9	85
Titanium	23.1	15

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### 3.3 Surface chemical composition investigation of the TiO<sub>2</sub>-SiO<sub>2</sub> NC thin films

The chemical analysis of the topmost surface of the TiO<sub>2</sub>-SiO<sub>2</sub> NC thin films was carried out with XPS. Four different positions were measured on the same sample. Figure 8 represents the intensity of the O 1s, Ti 2p, C 1s and Si peaks. From these elementary spectra, the mean atomic concentration and the standard deviation in percentage is given by considering the components included in each atomic peak. The contribution of oxygen bonded to silicon atoms, noted as O-Si appears at the binding energy of 532.6 eV whereas the one corresponding to the bond with titanium, noted as O-Ti, at 530.5 eV. [6] A noteworthy observation in this figure is that no peak corresponding to Ti-O-Si (at 531.5 eV [50]) can be found, as expected for a NC, confirming that only two different phases are detected in the film.[50] As reported by many authors [24,51], the signal of titanium element in XPS results from two separated peaks that can be attributed to Ti<sub>2p3/2</sub> and Ti<sub>2p1/2</sub>. The 2p<sub>3/2</sub> peak lies at 458.5 eV as expected for Ti<sup>4+</sup> cations in TiO<sub>2</sub>. Since the injected colloidal solution of TiO<sub>2</sub> NPs was synthesized in an organic solvent, the contribution of carbon element was investigated. Therefore, three components were identified in the C 1s peak. The first and most prominent one, at 285eV, is usually assigned to C-C bonds of both adventitious elemental carbon and the solvent, while the second, at 286.0eV, and the third, 288.5eV, can be related to -C-O bonds and C=O bonds, respectively. Finally, the Si 2p<sub>3/2</sub> peak has a binding energy of 103 eV, which is related to the fully oxidized SiO<sub>2</sub> matrix [51]. Consequently, from the Ti2p and Si 2p signals only, the estimated atomic percentage in Ti is at 12 ± 0.004 % and for Si at 88 ± 0.004 %; these values are consistent with those obtained by EDX at the SEM (Table 1). The atomic percentage from XPS can be treated in the same manner as in Section 3.2 and the calculations provided in the ESI, leading to a volume percentage of 13 %.



422  
 423 **Figure 8:** O 1s, Ti 2p, C 1s and Si 2p peaks of the TiO<sub>2</sub>-SiO<sub>2</sub> NC thin film measured by XPS  
 424 (mean atomic percentages and standard deviation of the components from 4 different positions  
 425 in between brackets).

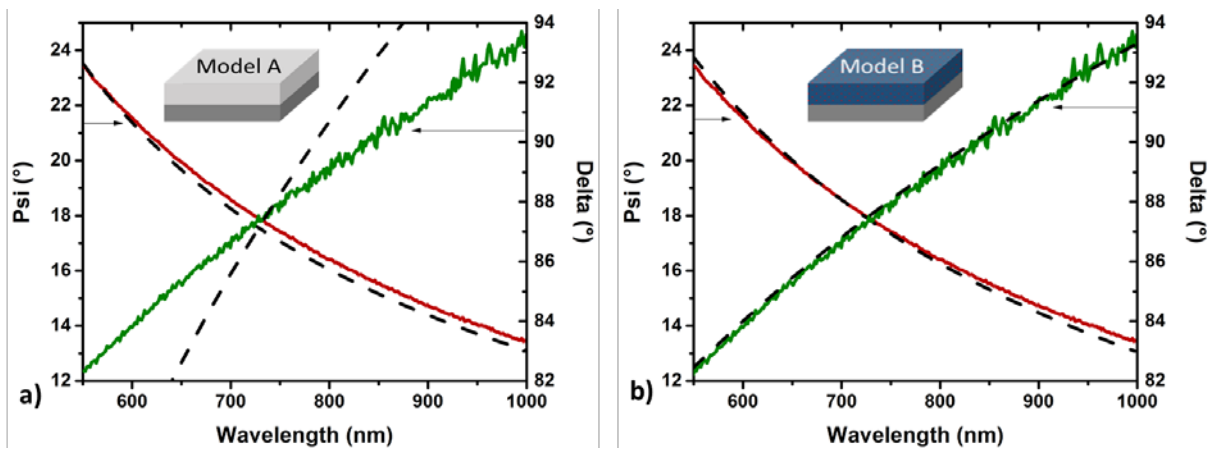
### 426 3.4 Spectroscopic ellipsometry of the TiO<sub>2</sub>-SiO<sub>2</sub> NC thin films

427  
 428 The investigation of the optical and physical characteristics was carried out with  
 429 spectroscopic ellipsometry. The ellipsometric ratio  $\rho = \tan(\Psi)\exp(j\Delta)$  is analyzed from  
 430 the measured  $\Psi$  and  $\Delta$ . To extract the structural and optical information from these data,  
 431 the sample was modeled as a homogenous layer on silicon substrate, without any top-layer  
 432 roughness. Due to droplet traces observed at low magnification in the Section 3.1, a more  
 433 pronounced diffusion of the incident light was observed (in the measurement) at shorter  
 434 wavelengths. Hence, the fitting was carried out in the limited range of 550 - 1000 nm. The  
 435 validity of the model and therefore the evaluation of the accuracy of the fit was estimated by  
 436 the Mean Square Error (MSE) factor. [52]

437  
 438 Assuming the ellipsometry was not sensitive enough to detect the NC nature of the sample,  
 439 the first ellipsometric model used was a one phase material with Tauc-Lorentz dispersion law  
 440 (Figure 9a, Model A).[52] However, the experimental and simulated data show discrepancies,  
 441 providing a high value of the error value (MSE=22). Therefore, a two-phase material (SiO<sub>2</sub> and  
 442 TiO<sub>2</sub>) was set up to account for the ellipsometric data of the TiO<sub>2</sub>-SiO<sub>2</sub> NC thin films. First,  
 443 the dispersion laws for the SiO<sub>2</sub> matrix and the TiO<sub>2</sub> NPs have to be assessed. Conducting an  
 444 experiment of depositing the SiO<sub>2</sub> matrix while injecting the solvent without the NPs, using the  
 445 same injection conditions as the plasma-deposited TiO<sub>2</sub>-SiO<sub>2</sub> NC thin films, allowed us to  
 446 acquire a reference SiO<sub>2</sub> matrix film and assess its quality, in the range between 550 and 1000

447 nm. The model used to fit the ellipsometry data of the matrix was the one of the classical thermal  
 448 SiO<sub>2</sub> material, as expressed by Herzinger et al. [53] using the Sellmeier/Cauchy dispersion law.  
 449 The measured and simulated Psi and Delta signals are given in Figure 10a. In this figure, the  
 450 quality of the fit appears to be high and the MSE for all the wavelength range was given at  
 451 MSE=10. In Figure 10b, the refractive index (n) and the extinction coefficient (k) of the SiO<sub>2</sub>  
 452 matrix are given. For the TiO<sub>2</sub> NPs, the Tauc-Lorentz dispersion law was used based on a spin-  
 453 coated 50 nm TiO<sub>2</sub> NPs film. The effective refractive index for this spin coated film was found  
 454 at 1.68 at 633 nm wavelength (as voids were evidenced in the film) and the optical bandgap  
 455 (E<sub>g</sub>) at 3.4 eV. The detailed results can be found in a very recent publication of the group.[25]  
 456 The fit of the NC film using the second model (Model B), resulted in an improved error  
 457 (MSE=4).

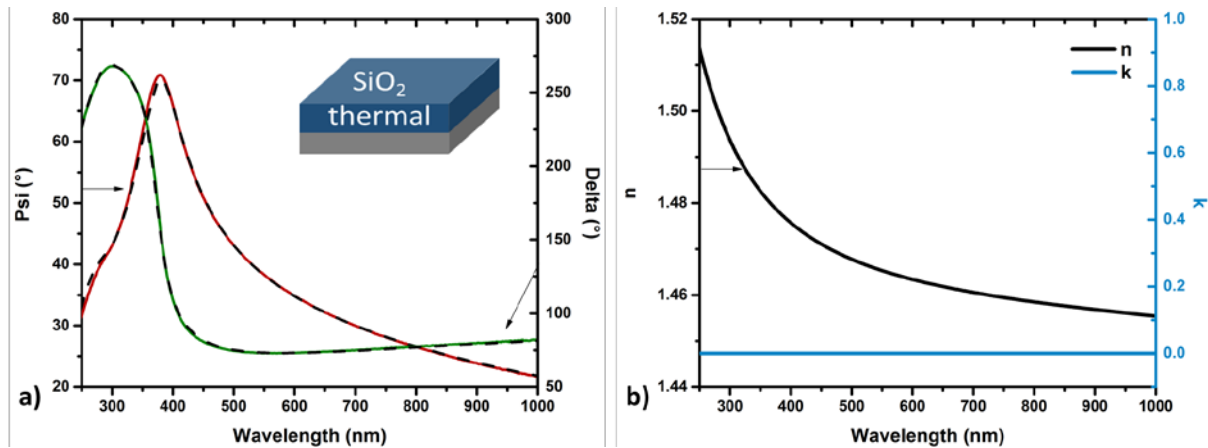
458 Then, based on the cross-section results obtained with TEM (Fig. 4, Section 3.1) for the  
 459 nanocomposite film, the Bruggeman Effective Medium Approximation (BEMA) was used  
 460 (Figure 9b, Model B) to model the TiO<sub>2</sub> NPs dispersed in the SiO<sub>2</sub> matrix. A much better  
 461 agreement was observed between the experimental and the simulated data.  
 462



463 **Figure 9:** Psi (red) and Delta (green) results of the measured (continuous) and modeled  
 464 (dashed) data with ellipsometry, based on Model A (a) and Model B (b).  
 465

466 Additionally, from the BEMA, the volume fraction of TiO<sub>2</sub> NPs in the film could be deduced  
 467 reaching 19 vol.% and 81 vol.% of SiO<sub>2</sub>. This observation confirms the EDX and XPS results  
 468 (Section 3.2), where no mixed (Ti-O-Si) bonding was observed. The optical TiO<sub>2</sub> energy gap  
 469 is determined at 3.4 eV due to the nanometric size of the TiO<sub>2</sub> [54,55]. In addition, the effective  
 470 refractive index at 633 nm wavelength was estimated at 1.59, an increased value compared to  
 471 the SiO<sub>2</sub> matrix itself (1.46).

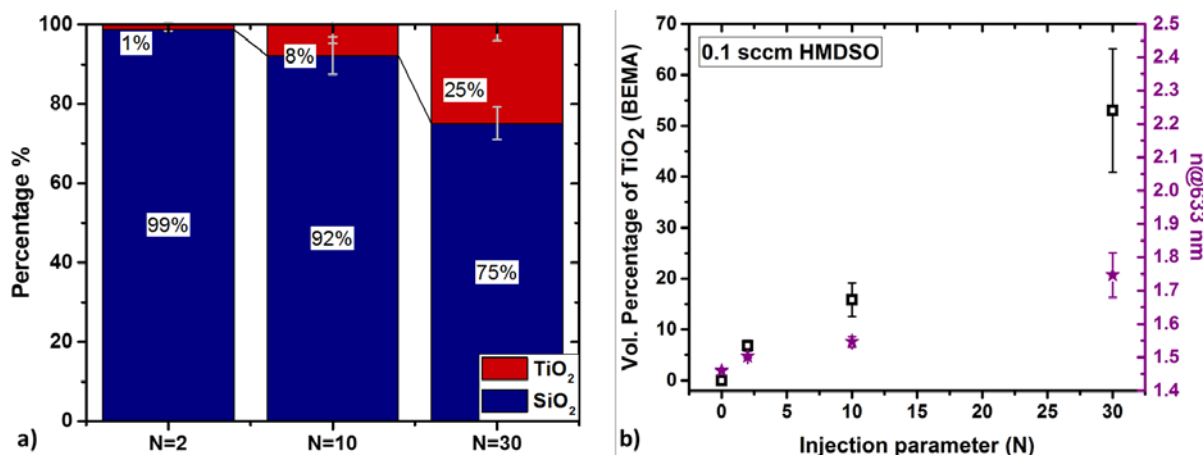




472  
 473 **Figure 10:** a) Psi (red) and Delta (green) results of the measured (continuous) and modeled  
 474 (dashed) data with ellipsometry of the matrix only fitted with SiO<sub>2</sub> thermal model.

475  
 476 **3.5 Comparison with lower and higher TiO<sub>2</sub> NPs injection rate**  
 477

478 To assess the agility of this hybrid process into acquiring NC thin films with different  
 479 composition and optical characteristics, experiments with lower and higher injection rate were  
 480 attempted. The N of the injection sequence (for deposition of TiO<sub>2</sub> NPs) was set either at N=2  
 481 or at N=30 (instead of 10) and the flow rate of the HMDSO precursor (for deposition of the  
 482 SiO<sub>2</sub> matrix) was kept at 0.11 sccm. The total deposition time was maintained at 50 min. Figure  
 483 11a provides the stacked column (%) resulting from the XPS core level spectra fitting as  
 484 explained in Section 3.3. In order to assess the reproducibility of the hybrid approach the XPS  
 485 stacked column provide the mean value and the standard deviation (as error bars) for two to  
 486 four different experiments in each condition (including the N=10). As expected, for lower N  
 487 and therefore lower amount of TiO<sub>2</sub> NPs injected in a minute sequence, the atomic percentage  
 488 of TiO<sub>2</sub> on the surface of the TiO<sub>2</sub>-SiO<sub>2</sub> NC thin film was much lower than N=10 (1% for N=2  
 489 and 8% for N=10). Additionally, for the highest TiO<sub>2</sub> NPs injection rate (N=30), we can observe  
 490 a three times higher TiO<sub>2</sub> percentage (25% for N=30) than the N=10. Furthermore, using the  
 491 data obtained by spectroscopic ellipsometry, and the Model B proposed in Section 3.4, the  
 492 percentage of the TiO<sub>2</sub> NPs and the effective refractive index of the NC at 633 nm wavelength  
 493 was obtained (shown in Figure 11b). Once more, to account for the reproducibility of this hybrid  
 494 process, the results shown in Figure 11b are extracted from four different experiments for each  
 495 condition. From these results, a clear dependency of the NC's characteristics on the injection  
 496 parameters is observed. More specifically, for N=0 when no NPs are injected (SiO<sub>2</sub> matrix  
 497 only) the refractive index is found at 1.47. As we start injecting a low amount of NPs, hence  
 498 N=2, the NC film contains 7 vol.% of TiO<sub>2</sub> and reveals a refractive index of 1.50. In accordance,  
 499 for higher injection (N=10), the NC film contains 16 vol.% and reveals an index of 1.54. For  
 500 N=30, during the one minute sequence, pulses are applied on the injector constantly and  
 501 consequently no DLI-OFF exists. Hence, this induces a strong pressure variation (from 16 to  
 502 20 mTorr) and may produce less stable plasma conductions both the NPs deposition and the  
 503 silica matrix. Therefore, the significantly higher error bars than the rest conditions could be  
 504 explained. Nevertheless, for the highest amount of the injected NPs (N=30), the NC film can  
 505 reach up to 53 vol.% in TiO<sub>2</sub> NPs leading to a refractive index of 1.74.



507 **Figure 11:** a) Stacked columns (%) of the NC surface composition obtained from XPS for N=2,  
 508 N=10 and N=30 with measurements obtained from two to four samples. b) Volume percentage  
 509 of TiO<sub>2</sub> extracted from the BEMA and refractive index at 633 nm wavelength for N=0,  
 510 N=2, N=10 and N=30 extracted from four samples.  
 511

#### 512 4. Discussion

513  
 514 As demonstrated above, over the range of experimental conditions investigated, the deposition  
 515 of TiO<sub>2</sub>-SiO<sub>2</sub> NC thin films using a hybrid approach leads to small and isolated (non-  
 516 aggregated) TiO<sub>2</sub> NPs well-distributed into a dense and inorganic SiO<sub>2</sub> matrix. Compared to  
 517 other processes reported in scientific literature (mostly based on atmospheric-pressure  
 518 systems)[15,22,46,56,57], the proposed method significantly limits agglomeration of the NPs  
 519 and improves their dispersion in the matrix. An additional important factor leading to the  
 520 limited agglomeration of the NPs could be the initially very small size of the NPs in the stable  
 521 (over time and composition) colloidal solution. Changes in the agglomeration dynamics could  
 522 partially be attributed to the differences affecting the aerosol formation dynamics between  
 523 operating at low pressure compared to atmospheric conditions. In particular, at low pressure,  
 524 the nature of the flow is mainly molecular flow whereas it is laminar or even turbulent at  
 525 atmospheric pressure. [12,58,59] It could also be related to the low injection rates of the colloidal  
 526 solution. Using a direct-liquid injection in a CVD system at low pressure, Varvaele et al. [11]  
 527 have identified that low colloidal solution injection rates (as in this study) promotes the  
 528 dispersion of the NPs whereas at higher rates, they tend to coalesce such that the mean size  
 529 distribution shifts towards higher values.

530  
 531 Because of the overall good homogeneity of the film, the optical and dielectric properties of  
 532 the films could also be deduced by proper modelling of the ellipsometric signals (especially  
 533 between 550 and 1000 nm). This feature qualifies the process as suitable for the preparation of  
 534 TiO<sub>2</sub>-SiO<sub>2</sub> NC thin film for optical applications in this wavelength range. Despite the solvent  
 535 injection, the plasma-deposited matrix proved to be of high quality as the fit with thermal silica  
 536 was significantly accurate. This is a proof that carbon is not incorporated inside the matrix  
 537 confirming its inorganic nature. By tuning the injection sequence, it was further possible to tune  
 538 the effective refractive index, having the possibility to reach up to 53 vol% in TiO<sub>2</sub>. To the best

539 of our knowledge, these results are unique compared to the reported approaches.  
540 [12,15,22,46,56,60] Due to the set of experimental conditions used, the combination of adjusted  
541 volatility of solvents, remarkable stability of the colloidal solution and condition of spraying in  
542 a low-pressure plasma reactor resulted in an efficient transport of the droplets containing the  
543 non-agglomerated NPs towards the substrate. The optimization of the injection sequence  
544 permitted to keep the NPs protected inside the liquid droplets down to the substrate, thus  
545 preventing them to aggregate following their interaction with the plasma species. A downside  
546 of these liquid droplets as observed by the low magnification SEM-EDX images is the coffee  
547 ring patterns of the TiO<sub>2</sub> NPs. Several works report the depiction of droplets due to the limited  
548 volatility of the injected solution or the increased mean size (due to system limitations) of the  
549 produced droplets [9,61,62]. Given that some of the produced droplets in the specific system (by  
550 the direct-liquid injector) do not have enough time to evaporate (low volatility of the solvent),  
551 the evaporation process happens at the surface of the film. Further works are under progress to  
552 get better insight in the mechanisms involved during the process in terms of droplet formation,  
553 droplet travel speed and evaporation of droplet solvents due to the interaction with the low-  
554 pressure plasma. This will open a promising chapter for understanding the impact of low-  
555 pressure plasma on the physical and chemical state of the liquid droplet during plasma  
556 deposition of NC thin films.

557

## 558 **5. Conclusion**

559

560 In this work, the creation of TiO<sub>2</sub>-SiO<sub>2</sub> NC thin films by injecting a colloidal solution in a  
561 very-low-pressure O<sub>2</sub>/HMDSO plasma was reported for the first time. The iterative injection  
562 sequence was selected in a way to control and maintain in low levels the deposition rate of the  
563 NC films. The pressure was maintained at 3 mTorr (with slight variations) leading to the  
564 deposition of good-quality inorganic SiO<sub>2</sub> matrix. The cross section of the film revealed a fairly  
565 homogeneous distribution of the NPs inside the matrix, at the local scale. The amount of these  
566 3-nm NPs was evaluated by multiple characterization techniques and was estimated at 16 vol.  
567 % at the SEM scale and 19 vol. % as deduced from ellipsometry *i.e.* on a larger area. The surface  
568 composition revealed a slightly lower value (13 vol. %), which could be explained due to the  
569 coating effect of the SiO<sub>2</sub> matrix (during the last second of the injection sequence). Notably,  
570 not significant agglomeration of NPs was observed. Adjusting the parameters of the injection  
571 sequence (from N=2 to N=30), NC films with lower and higher TiO<sub>2</sub> NPs content and thus  
572 different optical properties could be achieved. Due to the originality of this work and its novelty,  
573 large-surface area deposition has still to be improved. This requires conducting theoretical and  
574 further experimental studies to understand the behavior and evaporation of the liquid droplets  
575 inside low-pressure plasmas. Through these first results it is evidenced that potentially, an agile  
576 approach like this could give the freedom to tune and adjust the NC's properties by adjusting  
577 both the nature of the NPs and the matrix but also of the injection parameters reaching a large  
578 range of NPs content inside the matrix.

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