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COMMUNICATION

High temperature quantum tunnelling of magnetization and thousand kelvin anisotropy barrier in a Dy₂ single-molecule magnet

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We report here a dinuclear Dy^{III} iodine-bridged single-molecule magnet self-assembled by *cis/trans* coordination chemistry that displays a large anisotropy barrier of *ca.* 1300 K and a hysteresis opening temperature of 16 K. High temperature quantum tunnelling of magnetization is observed up to 56 K in zero-field and explained by the combination of large anisotropy barrier and the local transverse field at the *trans* site. The results provide a model for thorough understanding of the effect of electronic structure on the magnetic behavior of lanthanide complexes.

The strong magnetic anisotropy of lanthanide complexes often induces single-molecule magnets (SMM) behavior,¹ which spurs their various applications in prototype magnetic devices.^{2,3} Among the lanthanide based SMMs, Dy^{III} is acclaimed as an unrivalled candidate owing to the strongly oblate spheroidal shape of its electron density and the presence of Kramers doublets.⁴ The anisotropy of Dy^{III} is therefore enhanced in axial coordination environments.^{5,6} The design of Dy^{III} complexes with low coordination numbers presents a particularly efficient avenue to realize high degree of axiality but remains quite synthetically challenging. The pitfall in lanthanide chemistry is the significant quantum tunneling of the magnetization (QTM) between the M_j level that plays crucial roles on the magnetic properties.⁷ The existence of fast QTM often causes the reduction or even loss of memory effect of the magnetization at zero field, limiting their potential applications.^{8,9} But the QTM can be reduced by consideration of the local symmetry around lanthanide ions.¹⁰ It has been shown that high CF symmetries, such as $C_{\infty v}$, D_{4d} , S_8 , D_{5h} , D_{6d} , and $D_{\infty h}$, etc. eliminate transverse CFs that enlarge CF and reduce QTM.¹⁰ However, in a real

system, the symmetry is most likely to be lower than ideal, and the high CF symmetry is most obtained from more coordinatively saturated molecules instead of low-coordinate systems.¹¹ How to control structural factors and take advantage of the symmetry strategies to suppress QTM while maintaining the magnetic axiality in low-coordinate systems is still a current challenge that needs to be further explored.

Octahedral six-coordinate complexes, as the threshold of the regular coordination number (7-12) of lanthanide complexes towards the low-coordinate ones and with a well-defined axial and equatorial distribution, are excellent candidates to realize performant SMMs. Increasing efforts have been made to understand magnetic relaxation and optimize electrostatic distribution of Dy^{III} in pseudo-octahedral geometry.¹²⁻¹⁷ We have recently reported the *cis/trans* stereoisomers in a distorted octahedral bromine-bridged Dy^{III} dimer [Dy(Cy₃PO)₂(μ -Br)(Br)₂·2C₇H₈ (**1**·2C₇H₈)] with a large effective barrier of 684 K.¹⁵ Interestingly, it is characterized by the very weak magnetic interaction between Dy^{III} ions as a consequence of the mutual orthogonal anisotropy arrangement and the bridging bromide ions. In this system the magnetic relaxation is dominated by the strong single ion anisotropy of the *trans*-Br₄O₂ site.¹⁵ We have decided to take advantage of this situation as we go down the periodic table: to substitute bromine with iodine. Weaker coordinating anions such as iodine 1) minimize the in-plane CF at *trans* site and, in principle, should maximize the axial anisotropy 2) minimize the superexchange interaction between Dy^{III} centers. It is striking that the energy barrier (*ca.* 1300 K) of the new dinuclear Dy^{III} SMM [Dy(Cy₃PO)₂(μ -I)(I)₂·4C₇H₈ (**2**·4C₇H₈)] is the highest among the reported polynuclear complexes obtained *via* conventional coordination chemistry approaches while QTM is visible up to *ca.* 56 K. Solid-state magnetic site dilution and *ab-initio* calculations allow rationalizing these prominent features.

The addition of DyI₃ and Cy₃PO at a 1:2 molar ratio in ethyl ether produced a white suspension solution. The volatiles were removed under reduced pressure and faint yellow crystals of **2**·4C₇H₈ were obtained by recrystallization from toluene (Scheme S1). It crystallizes in the monoclinic space group of $P2_1/n$ (Table S1). The asymmetric unit contains two unique stereoisomeric [Dy(Cy₃PO)₂I₄] molecules with metal ions

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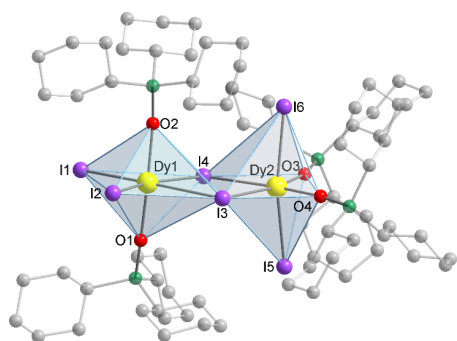


Fig. 1 Molecular structure of **2**. Hydrogen atoms are omitted for clarity. For unlabeled atoms: P green, and C grey

bridged by two μ_2 -I ions and each Dy^{III} center is hexacoordinated adopting a distorted octahedral geometry (Fig. 1). Dy1 is ligated by two strong Cy₃PO ligands in *trans* position which are in *cis* position for Dy2. The Dy–O distances (2.172–2.195 Å) are considerably shorter than Dy–I bonds (2.985–3.192 Å) (Table S2). Consequently, the coordination polyhedra are strongly distorted with CShMs values for O_h symmetry at 2.79 and 2.56 (Table S3).¹⁸ In **1**, the distances are equivalent for Dy–O (2.188–2.199 Å) but much shorter for Dy–Br (2.735–2.950 Å) and less distorted polyhedra were observed (CShMs values at 1.921 and 1.342).¹⁵ The intramolecular Dy–Dy distance (4.80 Å) is longer than in **1** (4.46 Å) while maintaining the same Dy–X–Dy angles. Within the lattice, the shortest intermolecular Dy...Dy distance is 10.661 Å (Fig. S1).

Static magnetic susceptibility measurements on **2**·4C₇H₈ are carried at 1000 Oe between 2 to 300 K. The room-temperature $\chi_M T$ value is equal to 27.71 cm³ K mol⁻¹ in agreement with the expected values for two free Dy^{III} ions (28.34 cm³ K mol⁻¹) (Fig. S4). $\chi_M T$ decreases monotonically on cooling down with a low temperature limit that is close to 24.01 cm³ K mol⁻¹. A small drop below 5 K is visible with a value of 22.95 cm³ K mol⁻¹ at 2 K. At this stage, there is no evidence of interactions (dipolar and/or superexchange) between Dy^{III} magnetic moments. The *M* vs. *H* curve at 2 K saturates at value close to 10 N β in agreement with the stabilization of the *M_J* = ±15/2 Ising component of the ⁶H_{15/2} ground state multiplet (Fig. S4). The slow relaxation of the magnetization is evidenced through ac susceptibility measurements. A maximum shows up on the out-of-phase component of the ac susceptibility below 70 K in zero external dc field (Fig. 2a and Fig. S5). The maximum shifts to low frequency on lowering the temperature but becomes temperature independent below 56 K, displaying a tunneling frequency of 412 Hz. At each temperature, the frequency dependence of the ac signals is analyzed quantitatively with an extended Debye model (Fig. S6, Table S4).¹⁹ Important information are extracted at this stage: 1) A narrow distribution of the relaxation times (α) of 0.05–0.18 is obtained, in agreement with a single relaxation process 2) The relaxation time follows a Arrhenius law combined with a thermally independent process ($\tau^{-1} = \tau_{\text{QTM}}^{-1} + \tau_0^{-1} \exp(-U_{\text{eff}}/T)$) with $U_{\text{eff}} = 1290$ K, $\tau_0 = 1.26 \times 10^{-12}$ s and $\tau_{\text{QTM}} = 3.88 \times 10^{-4}$ s (Fig. 2b) 3) The non-relaxing fraction (χ_s/χ_T) is close to 40% in all the temperature range (Table S4). This last characteristic suggests that only one

Dy^{III} site is slowly relaxing and that the second one is not involved. Clearly, the energy barrier jumped from ~700 K in **1**·2C₇H₈ up to ~1300 K in **2**·4C₇H₈.¹⁵ It proves that our approach which consists in minimizing CF in the plane rather than maximizing it along one axis is valid for Dy^{III}. However, QTM in **2**·4C₇H₈ is faster than in **1**·2C₇H₈ (Fig. S7).¹⁵ The application of an external dc field annihilates the thermally independent process and shifts the relaxation process to lower frequencies (Fig. S8). The relaxation becomes so slow on cooling that Zero-Field Cooled (ZFC) and Field-Cooled (FC) curves measured at a sweep rate of 0.5 K min⁻¹ diverge at 12 K that reveal the onset of the memory effect (Fig. 2c). On Fig. S8 the relaxation is the slowest at 1 kOe and the frequency dependence at various temperatures of the ac signals has been recorded at this optimum field (Fig. S9) and analyzed in the framework of the generalized Debye model (Table S5). The non-relaxing fraction remains close to 40% and now the Raman process ($\tau^{-1} = CT^n$) becomes the fastest at low temperature. The best fit is obtained with $C = 1.04 \times 10^{-4}$ s⁻¹ K⁻ⁿ, $n = 3.31$, $U_{\text{eff}} = 1323$ K, $\tau_0 = 7.3 \times 10^{-12}$ s (Fig. 2b). As expected, τ_0 and U_{eff} are almost unchanged between zero external dc field and 1 kOe while QTM is too slow to be numerically extracted.

The disappearance of QTM at low temperature with the magnetic field is common in Dy^{III} SMMs.²⁰ Hyperfine coupling, dipolar and/or superexchange interactions are known to be the sources of this thermally independent effect.^{21–23} Dilution in a diamagnetic matrix and isotopic enrichment are standard tools employed to eliminate QTM.^{24,25} However, in polynuclear complexes, *a fortiori* in dimers, the doping with non-magnetic metal ion (Y^{III}) may result in either Dy₂ complexes within a matrix of diamagnetic dimers or heteronuclear Dy–Y units. In the present case, 5% of Dy^{III} ion has been injected in the diamagnetic Y^{III} matrix to yield **2@Y** with a final ratio Dy/Y: 1/15.2. The magnetic investigation in zero external field on **2@Y** reveals a low frequency shift of χ_M'' peak at low temperature compared with **2**·4C₇H₈ (Fig. S10). The QTM is then efficiently released by the dilution. Remarkably, the characteristic relaxation times superimpose with those of the condensed **2**·4C₇H₈ at 1 kOe (Table S6, Fig. 2b). Here again, 40% of the magnetic moment is non relaxing. The best fit is obtained with Raman+Orbach processes ($U_{\text{eff}} = 1311$ K, $\tau_0 = 1.08 \times 10^{-12}$ s, $C = 6.69 \times 10^{-4}$ s⁻¹ K⁻ⁿ, $n = 2.82$). The dilution experiment indicates that intramolecular and/or intermolecular magnetic interactions are highly responsible for the prevalence of fast tunneling in **2**·4C₇H₈. However, there is no possible discrimination between the two types of dilution since we cannot dilute magnetic dimers into a diamagnetic matrix in frozen solution with ensuring the integrity of the complex in solution.

In-field magnetic hysteresis is clearly observed for **2**·4C₇H₈ up to 14 K, for a sweep rate of 0.12 kOe s⁻¹ (Fig. S11) which matches with the divergence of ZFC-FC curves at 12 K. The closing at the origin whatever the temperature is clearly due to the QTM fast relaxation. On contrary, the magnetic hysteresis curves for **2@Y** is open at the origin up to 16 K (Fig. S12). The normalized hysteresis loops are represented for both **2@Y** and **2**·4C₇H₈ at 2 K on Fig. 2d highlighting remanence (24% of saturation

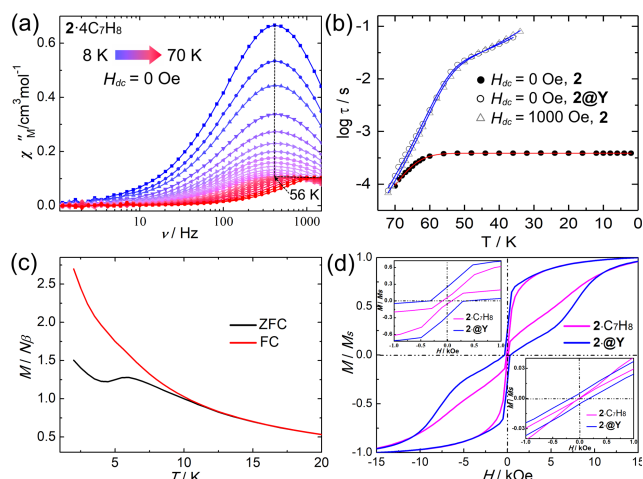


Fig. 2 (a) Frequency-dependent out-of-phase χ''_M ac susceptibility signals for **2** at zero external dc field. (b) Temperature dependences of the relaxation time for **2** at zero external dc field (\bullet) **2** at 1 kOe (Δ) and **2@Y** at zero external dc field (\circ) with the best fitted curves: in red QTM+Orbach and in blue Raman+Orbach. (c) ZFC-FC magnetization curves at 2 kOe for **2-4C₇H₈** at a sweep rate of 0.5 K min⁻¹. (d) Normalized M vs. H hysteresis profiles of **2@Y** and **2** at 2 K. In inset: top left) zoom at the origin of the 2 K hysteresis; bottom right) zoom at the origin of the 16 K hysteresis.

magnetization) and coercivity (360 Oe) achieved through the dilution.

Ab-initio Complete Active-Space Self-Consistent Field (CASSCF) calculations were performed on the X-ray structure of **2**, using the MOLCAS 8.2 and SINGLE_ANISO programs (see Supporting Information for computational details). Calculations were carried out on each asymmetric Dy^{III} center of the dimer unit (namely Dy1/Dy2) while the second metal ion was replaced by a diamagnetic Y^{III} ion. The resulting energy spectrum and g -tensors (g_{xx} , g_{yy} , g_{zz}) of the eight lowest Kramers Doublets (KD) of **2** are reported in Table S7. Dy1 and Dy2 show a strong ground state anisotropy with a main g_{zz} component at 19.9 for Dy1 and only 19.1 for Dy2 ($g_{xx}=g_{yy}=0.00$ for Dy1; $g_{xx}=0.03$ and $g_{yy}=0.07$ for Dy2). The orientations of the magnetic axes for the KD ground states are reported in Fig. 3. Due to the large intermolecular Dy-Dy distances (10.66 Å), only intramolecular dipolar magnetic interactions are calculated. The intramolecular dipolar term was computed in the effective spin $\tilde{S} = 1/2$ formalism using the POLY_ANISO routine (see experimental section for details).^{26,27} The resulting intramolecular dipolar constant is $J = -0.012$ cm⁻¹. This very weak antiferromagnetic coupling is explained by the orthogonal configuration of the anisotropy axes in the KD ground states (92°). In this frame, the computed molar magnetic susceptibility and magnetization curve at 2 K are in good agreement with experimental data (Fig. S4).

The electrostatic potential (Fig. 3), computed from the *ab-initio* LoProp charges using the home-made software CAMMEL,^{28,29} governs the Dy^{III} anisotropy axis orientation that follows the most negative potential induced by the O atoms of the C₃PO ligands (Fig. 3 and Fig. S13). The energy spectra of the ground state multiplet $^6H_{15/2}$ for each Dy^{III} ions are reported in Fig. 3. First, the total splitting is twice larger for Dy1 (1365 K) than for Dy2 (685 K). Second, the gap between the ground state KD and the first excited KD state is four times larger for Dy1 (488

K) than for Dy2 (122 K). The computed transition moments between each state (Fig. 3) for both Dy1 and Dy2 show two drastically different relaxation pathways. For Dy2, the non-zero transversal components of the ground state g -tensor and a mixed low-lying 1st excited state (Table S8) lead to sizable QTM and thermally assisted QTM suggesting a fast relaxation mechanism. On the contrary for Dy1, the strong Ising-type anisotropy of the ground and 1st excited state (Table S8) results in a relaxation path through the third excited state at 1165 K (Fig. 3), which corresponds to the effective energy barrier (Orbach) in almost perfect agreement with the experiment (~1300 K). The computed results support the rationale that at high temperature **2** relaxes *via* an over-barrier (Orbach) mechanism associated with the large single ion anisotropy of the *trans*-Dy₄O₂ site.

It is noticed that the change of the ancillary halogen donors connecting Dy^{III} is shown not only to induce differences in the octahedrally distorted DyX₆O₂ coordination sphere that adjusts the magnetic anisotropy and the superexchange interaction but also to modify the packing and intermolecular dipolar interactions. The principal axes of monoclinic **2** are canted by 45° between neighboring molecules. Whereas **1** crystallizes in the triclinic $P\bar{1}$ space group¹⁵ without a canted angle between two independent molecules (Fig. S14). Despite there is no possible to discrimination the origin of QTM (intramolecular vs. intermolecular processes), the large tilting of the principle axes between neighboring molecules should enhance the transverse components of fluctuating magnetic fields, thus improving tunneling of the magnetization in **2**.

In fact, **2-4C₇H₈** displays the best performance of six-coordinate SMMs with an effective barrier of up to 1300 K and hysteresis loops open up to 16 K (Table S9). This strong anisotropy combined with weak coercivity appears to be a common feature in the limited reported examples of D_{3h} and O_h geometries.^{12-17,30-32} Compared with two widely studied families of high-performance pentagonal bipyramidal and the metallocene dysprosium single-ion magnets,³³⁻³⁸ it has been found that higher charge symmetry and weaker thermal vibration of atoms favor stronger coercivity. For the reported trigonal bipyramidal and octahedral complexes,^{12-17,30-32} the uneven and the monodentate coordinated transverse sites

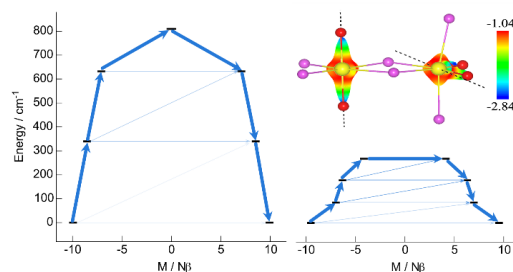


Fig. 3 Computed magnetization blocking barriers of Dy1 (left) and Dy2 (right) of **2**. The four lowest KDs (black lines) are represented according to their magnetic moment along the main magnetic axis. The arrows are drawn according to the mean values of the transversal matrix element of the transition magnetic moment. The total electrostatic potential at 1.6 Å of each Dy center is also represented (top right).

diminish the CF symmetry and increase the molecular flexibility, which contributes to the off-diagonal elements of the matrix representations and thereby enhancing QTM.

In summary, we observed a large anisotropy barrier of 1300 K and magnetic hysteresis opened up to 16 K in a dimetallic octahedral Dy^{III} system along with significant QTM up to 56 K. Strong QTM in the presence of large barrier is associated with the magnetic environment around Dy^{III}. The *cis/trans* arrangement of strong Cy₃PO moieties combined with weak iodine ligands around the Dy^{III} centers produces two drastically different crystal fields. The exceptional properties originate from the combination of large single-ion anisotropy at *trans* site and cancelation of magnetic interactions through near-orthogonality of the magnetic axes. The magnetic dilution experiment proved that the dipolar coupling accounts for the significant zero-field QTM. Comparative analysis of dysprosium complexes with different configurations suggests that the synthesis of high axial low coordination systems with uniform molecule anisotropy orientations, high charge symmetry and restricted molecular vibrations are expected to achieve a qualitatively new breakthrough for the development of SMMs.

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Conflicts of interest

There are no conflicts to declare.

Notes and references

- D. N. Woodruff, R. E. Winpenny and R. A. Layfield, *Chem. Rev.*, 2013, **113**, 5110-5148.
- R. Vincent, S. Klyatskaya, M. Ruben, W. Wernsdorfer and F. Balestro, *Nature*, 2012, **488**, 357-360.
- S. Thiele, F. Balestro, R. Ballou, S. Klyatskaya, M. Ruben and W. Wernsdorfer, *Science*, 2014, **344**, 1135-1138.
- J. D. Rinehart and J. R. Long, *Chem. Sci.*, 2011, **2**, 2078-2085.
- L. Ungur and L. F. Chibotaru, *Inorg. Chem.*, 2016, **55**, 10043-10056.
- P.-B. Jin, Y.-Q. Zhai, K.-X. Yu, R. E. P. Winpenny and Y.-Z. Zheng, *Angew. Chem. Int. Ed.*, 2020, **59**, 9350-9354.
- J. R. Friedman, M. P. Sarachik, J. Tejada and R. Ziolo, *Phys. Rev. Lett.*, 1996, **76**, 3830-3833.
- Y.-S. Ding, K.-X. Yu, D. Reta, F. Ortu, R. E. P. Winpenny, Y.-Z. Zheng and N. F. Chilton, *Nat. Commun.*, 2018, **9**, 3134.
- F. Ortu, D. Reta, Y.-S. Ding, C. A. P. Goodwin, M. P. Gregson, E. J. L. McInnes, R. E. P. Winpenny, Y.-Z. Zheng, S. T. Liddle, D. P. Mills and N. F. Chilton, *Dalton trans.*, 2019, **48**, 8541-8545.
- J. L. Liu, Y. C. Chen and M. L. Tong, *Chem. Soc. Rev.*, 2018, **47**, 2431-2453.
- M. A. Sørensen, U. B. Hansen, M. Perfetti, K. S. Pedersen, E. Bartolomé, G. G. Simeoni, H. Mutka, S. Rols, M. Jeong, I. Zivkovic, M. Retuerto, A. Arauzo, J. Bartolomé, S. Piligkos, H. Weihe, L. H. Doerrer, J. van Slageren, H. M. Rønnow, K. Lefmann and J. Bendix, *Nat. Commun.*, 2018, **9**, 1292.
- M. Gregson, N. F. Chilton, A.-M. Ariciu, F. Tuna, I. F. Crowe, W. Lewis, A. J. Blake, D. Collison, E. J. L. McInnes, R. E. P. Winpenny and S. T. Liddle, *Chem. Sci.*, 2016, **7**, 155-165.
- B.-C. Liu, N. Ge, Y.-Q. Zhai, T. Zhang, Y.-S. Ding and Y.-Z. Zheng, *Chem. Commun.*, 2019, **55**, 9355-9358.
- M. Li, H. Wu, Z. Xia, L. Ungur, D. Liu, L. F. Chibotaru, H. Ke, S. Chen and S. Gao, *Inorg. Chem.*, 2020, **59**, 7158-7166.
- M. Li, H. Wu, Z. Xia, V. Montigaud, O. Cador, B. Le Guennic, H. Ke, W. Wang, G. Xie and S. Chen, *Chem. Commun.*, 2019, **55**, 14661-14664.
- M. J. Giansiracusa, S. Al-Badran, A. K. Kostopoulos, G. F. S. Whitehead, D. Collison, F. Tuna, R. E. P. Winpenny and N. F. Chilton, *Dalton trans.*, 2019, **48**, 10795-10798.
- X.-L. Ding, Y.-Q. Zhai, T. Han, W.-P. Chen, Y.-S. Ding and Y.-Z. Zheng, *Chem. Eur. J.*, doi: 10.1002/chem.202003931.
- M. Llunell, D. Casanova, J. Cirera, J. M. Bofill, P. Alemany and S. Alvarez, SHAPE (Version 2.1), 2013.
- D. Gatteschi, R. Sessoli and J. Villain, *Molecular Nanomagnets*, Oxford University Press, Oxford, 2006.
- M. Li, H. Wu, Q. Yang, H. Ke, B. Yin, Q. Shi, W. Wang, Q. Wei, G. Xie and S. Chen, *Chem. - Eur. J.*, 2017, **23**, 17775-17787.
- N. Ishikawa, M. Sugita and W. Wernsdorfer, *Angew. Chem. Int. Ed.*, 2005, **44**, 2931-2935.
- P. H. Lin, T. J. Burchell, R. Clerac and M. Murugesu, *Angew. Chem. Int. Ed.*, 2008, **47**, 8848-8851.
- T. Han, M. J. Giansiracusa, Z.-H. Li, Y.-S. Ding, N. F. Chilton, R. E. P. Winpenny and Y.-Z. Zheng, *Chem. Eur. J.*, 2020, **26**, 6773-6777.
- F. Pointillart, K. Bernot, S. Golhen, B. Le Guennic, T. Guizouarn, L. Ouahab and O. Cador, *Angew. Chem. Int. Ed.*, 2015, **54**, 1504-1507.
- E. Moreno-Pineda, M. Damjanović, O. Fuhr, W. Wernsdorfer and M. Ruben, *Angew. Chem. Int. Ed.*, 2017, **56**, 9915-9919.
- L. F. Chibotaru, L. Ungur and A. Soncini, *Angew. Chem. Int. Ed.*, 2008, **47**, 4126-4129.
- L. Ungur, W. Van den Heuvel and L. F. Chibotaru, *New J. Chem.*, 2009, **33**, 1224-1230.
- G. Huang, G. Fernandez-Garcia, I. Badiane, M. Camarra, S. Freslon, O. Guillou, C. Daiguebonne, F. Totti, O. Cador, T. Guizouarn, B. Le Guennic and K. Bernot, *Chem. Eur. J.*, 2018, **24**, 6983-6991.
- The CAMMEL code is available under GNU General Public License v3.0 and can be downloaded at <https://github.com/rmarchal1/CAMME>.
- K. L. Harriman, J. L. Brosmer, L. Ungur, P. L. Diaconescu and M. Murugesu, *J. Am. Chem. Soc.*, 2017, **139**, 1420-1423.
- T. Han, Y.-S. Ding, Z.-H. Li, K.-X. Yu, Y.-Q. Zhai, N. F. Chilton and Y.-Z. Zheng, *Chem. Commun.*, 2019, **55**, 7930-7933.
- R. J. Blagg, L. Ungur, F. Tuna, J. Speak, P. Comar, D. Collison, W. Wernsdorfer, E. J. McInnes, L. F. Chibotaru and R. E. Winpenny, *Nat. Chem.*, 2013, **5**, 673-678.
- Y. C. Chen, J. L. Liu, L. Ungur, J. Liu, Q. W. Li, L. F. Wang, Z. P. Ni, L. F. Chibotaru, X. M. Chen and M. L. Tong, *J. Am. Chem. Soc.*, 2016, **138**, 2829-2837.
- J. Liu, Y. C. Chen, J. L. Liu, V. Vieru, L. Ungur, J. H. Jia, L. F. Chibotaru, Y. Lan, W. Wernsdorfer, S. Gao, X. M. Chen and M. L. Tong, *J. Am. Chem. Soc.*, 2016, **138**, 5441-5450.
- S. K. Gupta, T. Rajeshkumar, G. Rajaraman and R. Murugavel, *Chem. Sci.*, 2016, **7**, 5181-5191.
- F.-S. Guo, B. M. Day, Y.-C. Chen, M.-L. Tong, A. Mansikkamäki and R. A. Layfield, *Science*, 2018, **362**, 1400-1403.
- C. A. P. Goodwin, F. Ortu, D. Reta, N. F. Chilton and D. P. Mills, *Nature*, 2017, **548**, 439-442.
- K.-X. Yu, J. G. C. Kragsskov, Y.-S. Ding, Y.-Q. Zhai, D. Reta, N. F. Chilton and Y.-Z. Zheng, *Chem*, 2020, **6**, 1777-1793.