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1 Low-phase spectral reflectance and equivalent “geometric 2 albedo” of meteorites powders

3

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5

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10

11 **Abstract:** Generally, the reflectance of a particulate surface depends on the phase angle
12 at which it is observed. This is true for laboratory measurements on powders of natural
13 materials as well as remote observations of Solar System surfaces. Here, we measured
14 the dependences of reflectance spectra with phase angles, of a suite of 72 meteorites in
15 the 400-2600 nm range. The 10-30° phase angle range is investigated in order to study
16 the contribution of Shadow Hiding Opposition Effect (SHOE) to the phase behavior. The
17 behavior is then extrapolated to phase angle of 0° using a polynomial fit, in order to
18 provide grounds for comparison across meteorite groups (enabling to remove the
19 contribution of shadows to reflectance) as well as to provide “equivalent albedo” values
20 that should be comparable to geometric albedo values derived for small bodies. We find
21 a general behavior of increasing strength of the SHOE with lower reflectance values
22 (whether between samples or for a given samples with absorption features). This trend
23 provides a first order way to correct any reflectance spectra of meteorite powders
24 measured under standard conditions ($g=30^\circ$) from the contribution of shadows. The
25 $g=0^\circ$ calculated reflectance and equivalent albedos are then compared to typical values
26 of albedos for main-belt asteroids. This reveals that among carbonaceous chondrites
27 only Tagish Lake group, CI, and CM chondrites have equivalent albedo compatible with
28 C- and D-type asteroids. On the other hand equivalent albedo derived with CO, CR and
29 CK chondrites are compatible with L- and K-type asteroids. The equivalent albedo
30 derived for ordinary chondrites is related to petrographic types, with low-grade
31 petrographic type (type 3.6 and less) being generally darker than higher petrographic

32 types. This work provides a framework for further understanding of the asteroids
33 meteorite linkage in particular when combining with colors and spectroscopy.

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38 **1. Introduction**

39

40 Solar System small bodies are a make-up of heterogeneous objects in term of size,
41 orbit and colors. Observations available are much more numerous and detailed for the
42 largest members, while for the smallest - and faintest - objects, orbit is generally the only
43 available information. Still, there are 10 000s of objects for which, size, colors and
44 albedo have been determined in addition to orbit (Izevic et al., 2001, Mainzer et al.,
45 2011, Usui et al., 2019). Albedo is then one of the properties that is known for the largest
46 number of small bodies and, combined with colors, could in principle be used to help
47 mapping the composition distribution of small bodies across the solar system. However,
48 the interpretation of albedo in term of composition is degenerate; the brightness of a
49 material can be related to chemistry (the presence of Fe-related absorption features, the
50 presence of opaque phases, metals, oxides, organics,...) as well as to physical properties
51 such as grain size. Still, albedo has been used as a zero order connection between
52 asteroid classes to meteorites groups by qualitatively comparing albedos to reflectance
53 levels measured on meteorites: the dark asteroid families are often named C-complex,
54 where C refers to “carbonaceous”, by analogy to the dark carbonaceous chondrites. This
55 has so far remained a qualitative comparison since reflectance levels cannot be
56 compared to albedos: they are not the same physical quantity by definition, which
57 forfeits a direct comparison.

58 There are multiple types and definitions of albedo. In the case of large asteroid
59 surveys, the geometric albedo in the visible, p_v , is determined on the basis of mid-
60 infrared observations coupled with a thermal model and visible observations (Lebofsky
61 et al., 1986). The geometric albedo is defined as the brightness ratio between the
62 reflected sunlight (averaged spectrally over the visible range) to that of a perfectly
63 lambertian disk with the same cross-section, everything being observed at a phase angle
64 of $g=0^\circ$. This quantity is therefore different from laboratory measurements obtained on

65 meteorite samples, which typically provide biconical or bidirectional reflectance spectra
66 in a phase angle range around 30° . While the definition of p_v is at zero phase angle, note
67 that asteroid observations are essentially conducted at low phase angles but only rarely
68 obtained at exactly $g=0^\circ$. They are extrapolated to $g=0^\circ$ using measurements (or
69 guesses) of the phase coefficient (the dependence of brightness on phase angle). This
70 approach is complicated by the fact that the opposition effect, the strong increase of
71 reflectance towards 0° phase, shows variability in both its magnitude and shape
72 depending on the composition and physical properties of the surface material.

73 Here, we present an attempt to provide laboratory measurements that can in
74 principle be used to compare with the geometric albedo of a smooth-surfaced asteroid,
75 covered by particulate meteorite-like material. Previous investigations have looked at
76 some aspects of this topic on extra-terrestrial samples ([Gradie et al., 1980](#); [Gradie and](#)
77 [Veveřka, 1986](#); [Capaccioni et al., 1990](#)), but the originality of our approach is to focus on
78 the low phase angle range and to sample a large diversity of samples. For that we
79 present multispectral measurements on a suite of 72 meteorite samples belonging to 10
80 different classes, with a special emphasis on carbonaceous chondrites. For each sample,
81 multispectral measurements were obtained in the 400-2500 nm spectral range, at phase
82 angles of 10° , 20° and 30° . This enables to extrapolate the reflectance spectra at 0° phase
83 angle for each sample, and to calculate the reflectance at 0° phase angle averaged over
84 the Solar spectrum, two quantities that can be seen as equivalent to p_v and p . We then
85 compare these values to albedo values derived for various asteroids classes, and discuss
86 the connection between meteorite classes and asteroid families.

87

88

89 **2. Samples & Method**

90

91 **2.1 Samples**

92 A suite of 72 meteorite samples was analyzed for this study. We particularly
93 focused on carbonaceous chondrites, which have been related in the past to C-type
94 asteroids. Samples from the CI, CM, CR, CO and CV groups were analyzed together with
95 ungrouped C2 chondrites. Many of these samples were provided through the Antarctic
96 meteorite research program. Meteorites were grinded in an agate mortar, in order to
97 obtain a particulate sample. Previous samples prepared with the same protocol resulted

98 in typical grain size in the range 30-200 μm (Beck et al., 2012). A list of samples used in
99 this study and their reflectance values is provided in [supplementary materials](#).

100

101 2.2 Bidirectional reflectance measurements

102

103 The reflectance measurements were obtained with the SHADOWS spectro-gonio
104 radiometer (Potin et al., 2018). For each sample, we measured the reflectance between
105 400 nm and 2500 nm, every 100 nm at a spectral resolution of around 4 nm in the 360-
106 670 nm range, 8 nm in the 680-1420 nm range and 16 nm in the 1430-2640 nm range
107 (Figure 1). This wavelength range was measured in order to cover a significant fraction
108 (88%) of the solar energy spectral distribution.

109 The SHADOWS instrument enables to measure bi-directional reflectance over a
110 range of incidence and emergence angles, down to phase angles of 5° . For this work
111 spectra of each sample were measured at 10, 20 and 30° phase angle for nadir incidence
112 (0°). The instrument was used in standard mode with a spot size of 5.2 mm (much larger
113 than the grain size) and the typical sample mass used was around 50 mg. Spectra were
114 obtained under ambient conditions and calibrated using SpectralonTM and InfragoldTM.
115 The system used a 20mm diameter diaphragm placed in front of the optics of both
116 detectors to increase the bidirectionality (i.e. the angular resolution) of the observation
117 to $\pm 0.8^\circ$ (therefore the angular resolution of the system is of 3° , constrained by
118 incidence).

119 For each sample, the reflectance values obtained are extrapolated to zero phase
120 angle using a second order polynomial fit of the measurements at 10, 20 and 30° . This is
121 possible in the 400-1000 nm range measured by a silicon detector with a high signal to
122 noise ratio (SNR). This is not possible in the near-IR range where the SNR of the InSb
123 detector was not sufficient to enable a reliable extrapolation to zero-phase. The
124 significance of the 3-points polynomial approach is investigated in the case of four
125 samples (1 CO, 1 CV, 1 CR, and one ordinary chondrite) by measuring the phase-curve
126 every two degrees from $g=8$ to $g=40^\circ$ (Fig. 2) and comparing the interpolated values to
127 $g=0^\circ$ using either the 3-point approach or the full dataset (second-order polynomial fit
128 of the 17 data points). This test reveals that using 3 points or 17 points does not change
129 the zero-phase interpolated value by more than a few percent (<0.005 in absolute
130 reflectance) (Fig. 2).

131 Astronomical observations of small bodies, and derivation of zero-phase
132 magnitude from phase curves are generally performed by using an exponential fit to the
133 magnitude vs. phase angle relation. We chose not to use the same analytical formalism
134 because these observations are not directly analogous to our laboratory measurements.
135 Indeed the phase curves of small bodies are a combination of photometric behavior of
136 the surface, and the fraction of the object that is illuminated when seen from the
137 observer. As a consequence, the polynomial approach was used since it provided a
138 simple and satisfactory fits to the 17 points curves.

139

140

141 **3. Calculation of « albedo » and zero phase reflectance.**

142 3.1 Method

143 At present, the SHADOWS instrument does not allow to perform measurement at
144 zero phase angle. Therefore, for each wavelength the reflectance at zero phase angle is
145 obtained by extrapolating the reflectance at 10°, 20°, 30° to a phase of 0° using a second
146 order polynomial. At low phase angle ($g < 30^\circ$), two contributions result in an increase of
147 reflectance with decreasing phase angle ([Hapke, 2012](#)). First, the shadow-hiding
148 opposition effect (SHOE) that typically occurs below 30° is related to the fact that at
149 lower phase, there are less shadows at grain scale, and then the surface is brighter. It is a
150 geometrical effect although its magnitude may depend on the scattering properties of
151 the grains. A second contribution is the coherent back-scattering opposition effect
152 (CBOE), which is related to additive or destructive interactions of photons and occurs
153 for phase angles typically below 2° for many particulate samples and for the lunar
154 surface ([Hapke, 2012](#)).

155 In the case of small bodies observations, the geometric albedo that was computed
156 is based on thermal modeling of mid-IR thermal emission observations as well as on the
157 values of the visible absolute magnitude (H) and phase slope parameter (G). The latter is
158 measured from telescopic observations that are only rarely obtained at very low phase
159 angle ([Harris, 1989](#)) and therefore may not take into account the CBOE. This is why we
160 prefer in this work not to use zero-phase angle measurements directly but rather to use
161 an interpolation approach.

162 Two quantities are calculated that could in principle be used for comparison with
163 small bodies observations. The first one is the 550 nm reflectance at zero-phase angle,

164 which is calculated by averaging the 500 and 600 nm zero-phase reflectances, that are
165 estimated using the polynomial interpolation. This quantity can in principle be
166 compared to the visible geometric albedo p_v and may be called *calculated zero-phase*
167 *reflectance* or *equivalent visible geometric albedo*.

168

169 A second quantity that is calculated is the solar-spectrum averaged zero-phase
170 reflectance. For each sample and each phase angle, the spectral integral of a black body
171 spectrum at 5500°C multiplied by the reflectance spectrum of the sample is divided by
172 the spectral integral of the same blackbody multiplied by the spectrum of a lambertian
173 surface (reflectance of 1).

174

$$175 \quad A_{Lab}^g = \frac{\int_{\lambda=400}^{2500} R_{\lambda} \times BB_{\lambda} d\lambda}{\int_{\lambda=400}^{2500} BB_{\lambda} d\lambda} \quad (\text{eq. 1})$$

176

177 This calculated quantity represents the ratio between the solar radiation
178 reflected by the sample surface at a given phase angle to that reflected by a lambertian
179 surface. This quantity was calculated for $g=10, 20$ and 30° and then interpolated to $g=0^{\circ}$
180 using a second order polynomial fit, in order to estimate the fraction of solar radiation
181 reflected by the surface at zero-phase angle. This quantity is somehow analogous to the
182 geometric albedo p and may be called *equivalent geometric albedo*.

183

184 3.2 Cautions and underlying hypothesis

185

186 When comparing calculated zero-phase reflectance and equivalent geometric
187 albedo from our work to asteroid observations, there are some strong underlying
188 hypotheses that the reader needs to keep in mind:

189

190 - Photometric properties are hypothesized to be dependent on phase angle only,
191 while in reality incidence and emergence angles also have an independent role (see for
192 instance Beck et al., 2012 or Potin et al., 2019), an observation also noted at larger
193 phase angles and having led to the weighting of measurements made at high incidence
194 angles in the photometric inversion procedures (e.g., Souchon et al., 2011).

195

196 - The contribution from the CBOE is not taken into account. The zero-phase
197 reflectance derived here cannot be used to compare with observations obtained at very-
198 low phase angles ($<2^\circ$). For low-albedo samples (i.e. carbonaceous chondrites) the CBOE
199 might be moderate (Skhuratov and Helfenstein, 2001) but detailed investigations of
200 CBOE on carbonaceous chondrites are not available at present.

201

202 - The impact of large-scale shadows is not taken into account when comparing
203 disk-integrated observation to laboratory measurements. The effects of large-scale
204 shadows are expected to be modest at low phase making this hypothesis reasonable
205 (Hapke et al., 2012).

206

207 - A particulate material with a wide range of particle sizes is covering asteroids
208 surfaces. The measurements in this work were obtained on fine powders and the
209 behavior of a surface covered by large particles ($>1\text{ cm}$) could differ from that of a
210 surface covered by smaller ones (especially for weakly absorbing materials). Grain
211 shape as well as the style of topography at all scales may also play a role (e.g., Shepard
212 and Campbell, 1998; Helfenstein and Shepard, 1999; Cord et al., 2003; Skhuratov et al.,
213 2005; Souchon et al., 2011).

214

215 - We did not carry out any measurement on pure Fe-metal, so this work does not
216 apply to metal dominated surfaces.

217

218

219 **4. The scattering behavior of meteorites at low phase angle**

220

221 4.1 A back-scattering behavior at low phase angle

222

223 All 72 meteorite samples analyzed in this study reveal a similar behavior of
224 increasing reflectance at low phase angle. We quantify the relative magnitude of the
225 opposition effect as the ratio between the extrapolated reflectance at zero phase angle
226 and the reflectance in standard geometry ($i=0^\circ$, $e=30^\circ$). Note that this definition is close
227 but slightly different from that in Beck et al. (2012) ($i=3^\circ$, $e=30^\circ$). In figure 3, the
228 relative intensity of the opposition effect is shown as a function of the reflectance at

229 standard geometry for the six wavelengths measured below 1 μm (zero-phase angle
 230 reflectance could not be extrapolated for the near-IR range due to the lower SNR). This
 231 graph reveals a general behavior among all samples and wavelengths, namely that the
 232 relative magnitude of the opposition effect is greater for darker surfaces. The values
 233 found in this work are in fair agreement with previous results we obtained on 6 samples
 234 (Beck et al., 2012, Fig. 3,5).

235 For asteroid observations this translates into the fact that darker asteroids
 236 should present higher phase coefficient values, which is indeed the case (Belskaya and
 237 Shevchenko, 2000; Longobardo et al., 2014, 2016). This increased back-scattering
 238 behavior for low-reflectance surfaces can be explained by the shadow-hiding opposition
 239 effect (SHOE). Indeed darker materials have a steeper phase slope due to a combination
 240 of surface texture and radiative transfer effects. When grain sizes are of the order of the
 241 surface roughness length scale, multiple scattering dilutes the effect of shadowing for
 242 transparent materials (i.e. bright surfaces). In the case of more opaque materials, the
 243 effect of shadowing is more pronounced compared to the reduced multiple scattering
 244 contribution (Skhuratov et al., 2005).

245
 246 Several analytical functions were tested in order to model the distribution
 247 observed in figure 3. Among those investigated, the function that resulted in the best
 248 quality of fit is a log-normal distribution, that is expressed as:

249

$$250 \quad \frac{R_{\lambda,g=0^\circ}}{R_{\lambda,g=30^\circ}} = C_1 + C_2 \exp\left(-\left[\frac{\ln\left(\frac{R_{\lambda,g=30^\circ}}{C_3}\right)}{C_4}\right]^2\right) \quad (\text{eq. 2})$$

251
 252 with $C_1=5.48 \pm 1.14$, $C_2=-4.40 \pm 1.14$, $C_3=0.388 \pm 0.019$, $C_4=6.08 \pm 0.95$

253
 254 This law should in principle enable to calculate a low phase reflectance from a
 255 reflectance value measured under standard conditions. Note that this law does not take
 256 into account the CBOE, and should not be used to compare laboratory data to
 257 observations at very low-phase angle ($g < 2^\circ$) where the CBOE starts to play a significant
 258 role. Note also that this law was obtained for powders and should not be applied to bulk
 259 samples (i.e. not powdered) until it is proven valid for such samples as well (which is
 260 most likely not the case). law also only applies to reflectance lower than 0.5 which is the

261 range investigated here (this is fortuitous, only related to the arbitrary choice of a
262 mathematical description of this dataset, which is not a bijection in the range of value
263 investigated here). Conversely, this law may enable to convert asteroid albedo values
264 into reflectance under standard laboratory conditions, for comparison with laboratory
265 data measured under “standard” conditions.

266

267 4.2 The low-phase relative blueing

268

269 The behavior observed in figure 3 could also be modeled by a law $f(x)=x/(a+x)$,
270 but with a slightly lower quality of the fit. This means somehow that the shadow-hiding
271 opposition effect studied here is close to being an additive contribution:

272

$$273 R_{\lambda,g=0^\circ} \approx C + R_{\lambda,g=30^\circ} \text{ with } C \text{ a constant value}$$

274

275 This means that in relative intensity the effect will appear stronger for darker
276 samples. Another consequence is that when working in relative reflectance, which is a
277 common way to analyze ground-based observations, phase angle can have an effect on
278 the calculated spectral slope. To illustrate this point, the spectra obtained for 4
279 meteorite samples at different phase angles are normalized at 500 nm and displayed in
280 figure 1. These graphs reveal that a red sample will appear bluer at low phase angle if
281 the slope is calculated after spectral normalization, and particularly in the case of dark
282 meteorites (see the CV chondrite LAP 02206 in figure 1). Such a behavior is also
283 revealed when the visible slope is calculated for each meteorite in absolute and relative
284 reflectance, at $g=0$ and $g=30^\circ$ (Figure 4). This is due to the fact that the SHOE behaves
285 almost like an additive contribution, because it is only due to external reflections on the
286 grains, therefore without absorption. This also means that at low-phase a spectrally flat
287 sample will not become blue, but that a sample with a spectral slope will have a
288 decreasing spectral slope at decreasing phase angle, if the slope is calculated in
289 normalized reflectance. The redder the spectra, the more pronounced the low-phase
290 blueing (Figure 4). Note that, at least in the case of S-type asteroids, these effects appear
291 minor when compared to the magnitude of space weathering effects (Vernazza et al.,
292 2009).

293

294 4.3 The opposition effect as a function of meteorites groups in carbonaceous chondrites

295

296 In figure 5, the same data as figure 3 are shown but each meteorite group is color
297 coded, in order to investigate differences in magnitudes of SHOE among the various
298 groups investigated. Within that plot, meteorites groups with particular high opposition
299 effect will appear above the average trend (the global fit presented in Figure 3) while
300 meteorites groups with relatively low opposition effect will appear below. This graph
301 reveals that among carbonaceous chondrites, some CV chondrites appear to have an
302 unusually pronounced SHOE. On the other hand heavily aqueously altered meteorites
303 (C2, CI and CM) appear to have a relatively less pronounced SHOE, and meteorites from
304 the CO and CR groups appear to be in between.

305 The existence of these group behaviors may be related to different mineralogy
306 (optical properties of grains), grain sizes and grain shapes. It is certainly difficult to
307 disentangle which property is responsible for these contrasted behaviors. The first
308 observation that should be made is that meteorites groups that experienced the
309 strongest aqueous alteration tend to lie below the global fit. Aqueous alteration has
310 mineralogical consequences, since it induces the hydrolysis of primary phases and the
311 production of phyllosilicates and secondary opaques. Aqueous alteration has also
312 physical consequences, for instance to increase porosity (Britt and Consolmagno, 2000)
313 or decrease strength of the samples (personal experience of the authors with grinding
314 such samples).

315 The presence of a significant amount of metal may produce a peculiar behavior,
316 but metal abundance varies in the order CM-C2-CI < CV < CO < CR (Krot et al., 2006).
317 Also note that the only metal rich meteorites studied here (mesosiderite) tend to show a
318 rather low SHOE. The presence of an elevated amount of CAI in CV could provide a lead
319 to explain the unusually strong opposition effect observed, but CAI are abundant in CO
320 as well.

321

322

323 **5. Low-phase reflectance and equivalent geometric albedos among meteorite**

324 **families**

325

326

327 5.1 Comparison to RELAB data

328

329 Several parameters may in principle impact the reflectance value measured for a
330 given sample. These include grain size, sample heterogeneity (meteorites are often
331 breccia), as well as measurement uncertainty. Also, because the final goal of this work is
332 to provide a way to compare laboratory data to observations, it is important to assess
333 how laboratory measurements may differ between different laboratories. In order to
334 investigate interlaboratory variability, we chose to compare to data from the RELAB
335 facility since it provides at the moment the most extensive suite of measurements on
336 extra-terrestrial materials. In figure 6 we present the range of reflectance values at 550
337 nm for different classes of meteorites using the IPAG setup (SHADOWS, this study and
338 Eschrig et al., submitted) and data obtained on meteorite powders with the RELAB setup
339 (Brown University). This graph shows a good agreement for all classes of meteorite
340 studied here. The only difference found is for type 3 ordinary chondrites (OC), which
341 appear to show a globally lower reflectance in the dataset measured with SHADOWS,
342 but this difference may be related to the different definition of the petrographic types
343 used at IPAG (based on Raman studies, Bonal et al. 2018).

344

345

346 5.2 Equivalent albedo values

347

348 5.2.1 Carbonaceous chondrites

349

350 The equivalent albedo values derived for our suite of meteorites vary
351 significantly from family to family (Fig. 7). In the case of carbonaceous chondrites, the
352 average equivalent albedo decreases in the order CK > CR > CV > CO > CM > CI > TL. A
353 first explanation for this variation could be the different amount of carbonaceous
354 compounds in those meteorite samples. In figure 7, the average equivalent albedo per
355 meteorite class is plotted against the average carbon content. The carbon content was
356 calculated using data in Alexander et al. (2012, 2013) as well as Jarosewich (1990) and
357 Pearson et al. (2006). As can be seen in Fig. 7 a rough correlation exists between carbon
358 content and equivalent albedo but correlation does not imply causality. Indeed, the
359 carbon content in carbonaceous chondrites is a proxy for the amount of matrix and this

360 correlation might also be explained by the fact that meteorites rich in carbon also tend
361 to have an elevated amount of dark matrix, which is enriched in iron-rich opaque
362 minerals.

363

364 5.2.2 The impact of thermal processing on carbonaceous chondrites

365

366 A significant fraction of CM chondrites has experienced a post-aqueous alteration
367 heating process (Nakamura et al., 2005). The measurements obtained here on heated
368 CM chondrites enable to assess the impact of thermal processing on equivalent albedo of
369 primitive carbonaceous chondrites. Results show that on average heated CM chondrites
370 are not darker than more primitive CM chondrites, but they show more variability in
371 equivalent albedo values (see supplementary table). This variability was explained in
372 Beck et al. (2018) by the fact that upon heating CM chondrites first become darker due
373 to modifications of organic compounds, and then become brighter upon more intense
374 thermal processing, as a consequence of the recrystallization of the matrix.

375

376

377 5.2.3 Ordinary chondrites

378

379 Our dataset contains mostly ordinary chondrites of low petrographic types since
380 we focus here on “primitive meteorite samples”. It is therefore important to remark that
381 this dataset is not representative of the average of the whole range of metamorphic
382 grades of ordinary chondrites.

383 We find that primitive ordinary chondrites of petrologic type 3, with an average p
384 of 0.154 ± 0.024 (1 sigma, $n=6$), appear darker than petrologic types > 4 . This
385 difference can be explained by the fact that small opaque phases that are present in low
386 petrographic grade meteorites tend to coalesce during the metamorphic process, and
387 becomes less efficient in darkening the sample. Low-metamorphic grade ordinary
388 chondrites have a higher equivalent albedo than CI and CM chondrites, but their
389 equivalent albedo values are in the range of CO, CV and CR chondrites.

390

391 5.2.4 HED meteorites

392

393 Values derived for HED meteorites vary typically in the 0.2-0.5 range (see
394 supplementary [table](#)). Within this suite the howardite sample has the lowest albedo
395 when compared to eucrite and diogenite but the dataset is too small to address whether
396 this is a general behavior.

397

398 5.3 p vs pv

399

400 When asteroid albedo values are determined, an assumption often made is that
401 $p/p_v=1$ ([Lebofsky et al., 1986](#)). In [Figure 8](#), we present the values of the ratio of
402 equivalent albedos to equivalent visible albedos, which can be considered as analogous
403 to p/p_v . The values found range from 0.85 to 1.25 with an average of 1.04. The values
404 are mostly above 1 except for HED meteorites and type 4-5 ordinary chondrite. The
405 specificity of these meteorites is the presence of a strong absorption band around 850
406 nm that absorbs a significant fraction of the solar energy. On the other hand, most
407 samples studied have a red slope, which may explain the overall values above 1.

408

409 **6. Comparison to small bodies**

410

411 In [figure 9](#) we present a comparison of equivalent albedo values derived for our
412 series of meteorites, to the asteroid family modal distribution albedos as obtained by
413 [DeMeo and Carry \(2013\)](#). The [DeMeo and Carry \(2013\)](#) data were obtained by mapping
414 [DeMeo et al. \(2009\)](#) spectral endmembers on multi-filter sky surveys, enabling to study
415 objects down to diameters as low as 5 km and providing the most detailed taxonomy of
416 main-belt objects. The values used for comparison are the modes of the albedo value
417 distribution for each taxonomic group (Table 4 in [DeMeo and Carry, 2013](#)).

418

419 6.1 “Carbonaceous” asteroids vs carbonaceous chondrites

420

421 While the denomination “C”-type asteroid initially stands for carbonaceous, a first
422 important observation is that only a fraction of carbonaceous chondrites appears
423 compatible with C-type asteroids. The CO, CV, CR and CK chondrites have albedos
424 significantly higher than CM chondrites and are too bright to be related to C-type. These

425 meteorites classes represent about 50% of the carbonaceous chondrite falls and their
426 possible parent bodies will be discussed in sections 6.3-6.4 and 6.5.

427 Among C-type asteroids, a significant fraction ($\frac{1}{3}$ to $\frac{1}{2}$) is hydrated and shares
428 very strong spectral similarities with CM chondrites (Vernazza et al., 2015, 2016;
429 Vernazza and Beck, 2017). The average albedo value derived for CM chondrite
430 (including heated samples) is 0.066 ± 0.019 which is in good agreement with C-complex
431 asteroid (0.054 ± 0.023 , DeMeo and Carry, 2013; Fig. 9). This good agreement reinforces
432 the connection between hydrated C-complex asteroids (Ch, Cgh) and CM chondrites.

433 The rest of the C-complex that do not show evidences of hydration remains
434 puzzling and there are at present two competing hypothesis. The first one is that they
435 represent thermally metamorphosed CM chondrites (Hiroi et al., 1993). From an albedo
436 perspective this hypothesis remains valid since heated CM chondrites have albedo
437 values consistent with C-type asteroids (Table 1) except for those that have been the
438 most extensively thermally processed (Beck et al., 2018). However, this hypothesis is
439 challenged by the fact that heated CM chondrites do show evidence of hydration at 3- μ m
440 (Garenne et al., 2016; Beck et al., 2018) and they often show olivine feature in their mid-
441 infrared spectra (Beck et al., 2014) unlike “non-hydrated” C-type (Vernazza et al., 2015).
442 An alternative interpretation is that these objects may represent loosely consolidated
443 material related to Interplanetary Dust Particles (Vernazza et al., 2015).

444

445 6.2 S-type: the scarcity of low-metamorphic grade objects

446

447

448 The albedo values calculated for type 3 ordinary chondrites are much lower than
449 those obtained for the LL4-5 chondrites, as well as values found for type 4 and above
450 ordinary chondrites measured under low phase angles (Capaccioni et al., 1990). Type 3
451 ordinary chondrites escaped the thermal metamorphism event experienced by most
452 ordinary chondrites (Bonal et al., 2016). The thermal metamorphism process is
453 explained by the build-up of heat within the parent body due to decay of short and long-
454 lived radionuclides, and type 3 ordinary chondrites are thus expected to have been at
455 some point the surficial lithology of their parent bodies.

456

457 The albedo values of S-type (0.247 ± 0.084 , [DeMeo and Carry, 2013](#)) is
458 significantly higher than values obtained for type 3 ordinary chondrites ([Fig. 9](#)). This
459 difference cannot be attributed to space weathering that tends to darken asteroid
460 surfaces with time ([Pieters and Noble, 2016](#)). This observation suggests that the surface
461 of S-type asteroids is not covered by material analogue to type 3 chondrites, but rather
462 by more thermally processed material analogous to type 4 or above. Such an
463 observations is in agreement with a fast accretion of S-type parent bodies, followed by
464 fragmentation and surface brecciation ([Vernazza et al., 2014](#)).

465

466 6.3 K -and L-types

467

468 L-type asteroids are a peculiar class of objects with an absorption band above 2-
469 μm , which has been explained by the presence of FeO bearing spinel ([Burbine et al.,](#)
470 [1991](#)) in excess of what is observed in carbonaceous chondrites ([Sunshine et al., 2008](#)).
471 This led to the idea that they could represent a reservoir of early solar system material
472 not sampled by meteorites, enriched in refractory inclusions ([Sunshine et al., 2008](#)).
473 Another peculiarity of L-type asteroids is the presence of unusual polarimetric curves
474 ([Devogèle et al., 2018](#); [Cellino et al., 2005](#)) with an inversion angle much larger than
475 other asteroid taxonomic types (i.e. $> 25^\circ$).

476 K-type asteroids are characterized by the presence of a red-slope and a moderate
477 1- μm silicate band (olivine). K-type asteroids are distributed among the entire main-
478 belt according to multi-color surveys but a large number of K-type objects are found
479 among the EOS family ([Mothes-Diniz et al., 2008](#); [DeMeo and Carry, 2013](#)).

480 K-type objects can be distinguished from the L-type by the presence of a weaker
481 2 μm band and a stronger 1- μm band. The average albedos derived for K- and L-type by
482 [DeMeo and Carry \(2013\)](#) are similar (so it is difficult to distinguish between the two
483 types in multicolor surveys). Among the meteorites studied, only CO, CK and CV groups
484 appear consistent with a connection to L and K-types based on their low-phase
485 reflectance ([Fig. 9](#)); L- and K-type are thus very likely related to carbonaceous
486 chondrites but not all types of carbonaceous chondrites.

487 Since albedo does not discriminate between CO, CK and CV, to further investigate
488 the nature of L- and K-types, it is necessary to turn to spectroscopy. Spectroscopic
489 surveys of the EOS family have revealed the presence of a strong 1- μm band but only

490 weak 2- μm features (Mothes-Diniz et al., 2008). These spectral characteristics are
491 typically found in laboratory spectra of CK meteorites, while spectra obtained for CO and
492 CV tend to show a lot a variability from featureless spectra, to spectra containing both a
493 1- and a 2- μm band. Therefore, some CO and CV chondrites have affinities with L-type
494 asteroids but not all meteorites of these groups.

495

496 6.4 D-type vs Tagish Lake group

497

498 The Tagish Lake meteorite was the first identified as having a reflectance
499 spectrum similar to D-type asteroids (Hiroi et al., 2011). Since then, a few other
500 meteorites have been recognized as having spectra similar to Tagish Lake and D-types.
501 From an albedo perspective, the connection between D-type and Tagish Lake appears
502 reinforced (Fig. 9) given the similarities found in “albedo” values. The spectra similarity
503 and the albedo similarity are therefore making a strong case for a meteorite-parent body
504 relation, but there are some major difficulties in this connection. A first one is the
505 difference in mid-IR emissivity spectra that was noticed by Vernazza et al. (2011), which
506 suggests a very fine-grained surface (not lithified) similar to comets. The second one is
507 the lack of a strong 3- μm band on the only D-type object visited by a spacecraft, comet
508 67P/CG (Cappaccioni et al., 2015; Quirico et al., 2016), while all Tagish Lake samples
509 show the presence of abundant hydrated minerals, leading to a well-defined 3- μm band
510 (Gilmour et al., 2019). Therefore Tagish Lake has strong affinities to D-type asteroids
511 when looking at the VNIR, but its mineralogy appears to be different from D-type surface
512 when looking at the 3- an 10- μm regions.

513 An explanation to reconcile these observations is that Tagish Lake arises from the
514 interior of a D-type object where sufficient heat enabled to transform primary phases
515 into phyllosilicates but without strongly altering its VNIR properties (and then its
516 spectrum and low-phase reflectance). *If the D-type population accreted late, ^{26}Al decay
517 was insufficient to induce a total differentiation of 100 km sized objects (Neveu and
518 Vernazza, 2019), and while a muddy interior may have existed, their surface may have
519 remained pristine. The remaining difficulty is to explain how aqueous alteration may
520 have produced abundant phyllosilicates, without changing the VNIR spectra. One
521 possibility is that the peculiar spectrum of Tagish Lake is related to the abundant
522 macromolecular organics, (Herd et al., 2011) and that they were not modified during*

523 aqueous alteration. Experiments on soluble organics mixed with phyllosilicates under
524 hydrothermal conditions reveal that they experience important transformations
525 (Vinogradoff et al., 2018; Vinogradoff et al., 2020). In the case of more refractory
526 organics (IOM, insoluble organic matter), the impact of hydrothermal alteration is less
527 understood, in particular due to difficulties in generating a valuable laboratory
528 analogue. An approach to investigate the impact of aqueous alteration on the D-type
529 signature that can be pursued, is to look at the VNIR signature of different lithologies of
530 Tagish Lake, that experienced different level of aqueous alteration. Such an approach
531 was started in Gilmour et al. (2019), and showed that all lithologies investigated have a
532 D-type signature, while they experienced different levels of aqueous alteration.

533

534

535 7. Summary and conclusions

536

537 The results and conclusions can be summarized as follows:

538

539 An approach is proposed using low-phase angle measurements of meteorites to
540 provide observable quantities analogous to geometric albedo for meteorites powders.
541 This is done under a number of hypotheses listed in 3.2.

542

543 There appears to be a general trend in the low-g behavior of meteorite powders.
544 The increase of reflectance due to the shadow-hiding opposition effect (SHOE) is an
545 almost additive contribution. A relation is derived to obtain the zero-phase angle
546 reflectance from standard geometry observations (Fig. 3). Note that this relation (eq. 2)
547 does not take into account the coherent back-scattering opposition effect.

548

549 Albedo values derived for Tagish-Lake lithologies and “Tagish-Lake group”
550 meteorites are in agreement with D-type asteroids. We propose that they represent
551 samples from the interior of a D-type or a piece of D-type asteroid that experienced
552 heating and hydrothermal alteration.

553

554 Albedos values derived for CM chondrites are in good agreement with values
555 found for C-type asteroids. Heated CM chondrites tend to show more variability in

556 albedo with both brighter and darker samples when compared to “standard” CM
557 chondrites. This reinforces the relation between CM chondrites and hydrated C-complex
558 asteroids.

559

560 Carbonaceous chondrites from the CO, CV, CR and CK groups are brighter than
561 those from the CI, CM and Tagish Lake groups. The derived albedo values are higher
562 than for C-type asteroids, while in the range of K and L-type. This confirms that a
563 significant fraction of carbonaceous chondrites does not originate from C-type.
564 Combined with spectroscopic observations, this also reinforces the relation between K-
565 type (mostly EOS family) and CK meteorites. This dataset further delves into the
566 mystery of the parent bodies of CO-CV-CR meteorites for which there are no well-
567 identified parent bodies that match the spectra and albedo of these meteorites.

568

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758

Figure captions

Figure 1: Multispectral observations of powdered meteorites (LL4, CO, CR, CV) at phase angles ranging from 8 to 40° (incidence=0°). Spectra are shown in reflectance factor (left graphs) or in relative reflectance normalized at 500 nm (right graphs). The black line is the extrapolated reflectance at $g=0^\circ$ using a polynomial fit of the data (see Fig. 2).

Figure 2: Examples of fits of the phase curve for the extrapolation of the reflectance at $g=0^\circ$ using a 2nd order polynomial. The graphs show the reflectance at 500 nm as a function of phase angle for the same samples as in figure 1 and compare the extrapolated value at $g=0^\circ$ using a 3-point interpolation (all 72 samples from this study) against a 17-points interpolation (tested only for these 4 samples). The difference between the two extrapolated values at $g=0^\circ$ is less than 0.005.

Figure 3: « Relative magnitude » of the opposition effect as a function of the reflectance at standard geometry. This magnitude is defined as the ratio between the extrapolated reflectance at $g=0^\circ$ and the reflectance in standard geometry.

Figure 4: The effect of phase angle ($g=0$ or $g=30^\circ$) on the 400-900 nm visible spectral slope (in μm^{-1}) whether calculated in reflectance or in relative reflectance. In the case of reflectance, a slight redening is present for low-phase angle. When slope are calculated for relative reflectance spectra, a significant blueing is observed at low phase angle for the spectra with initially red slopes.

Figure 5: « Relative magnitude » of the opposition effect as a function of the reflectance measured under standard geometry. This graph shows the same data as figure 3, but color- and symbol-coded according to meteorite groups. The data from [Beck et al. \(2012\)](#) is also shown (circles) with the same color code.

Figure 6: Comparison of the reflectance at 550 nm for meteorite powders of carbonaceous chondrites measured at IPAG and at RELAB for different meteorite groups. **This graph shows that while some variability is present for each meteorite group, the range of value is in agreement for measurements obtained in two different laboratories.**

Figure 7 : Relation between meteorite equivalent albedo and average carbon content for different meteorite groups. Carbon contents are from [Alexander et al. \(2012\)](#), [Jarosewich \(1990\)](#) and [Pearson et al. \(2006\)](#)

Figure 8: Ratio between equivalent geometric albedos and equivalent visible geometric albedos for different meteorites belonging to different classes.

Figure 9: Comparison of the reflectance at 550 nm from this work for different meteorites groups, the calculated equivalent geometric albedo, and geometric albedos of asteroids for different taxonomical families ([DeMeo and Carry, 2013](#)). Top: linear scale. Bottom: logarithmic scale.

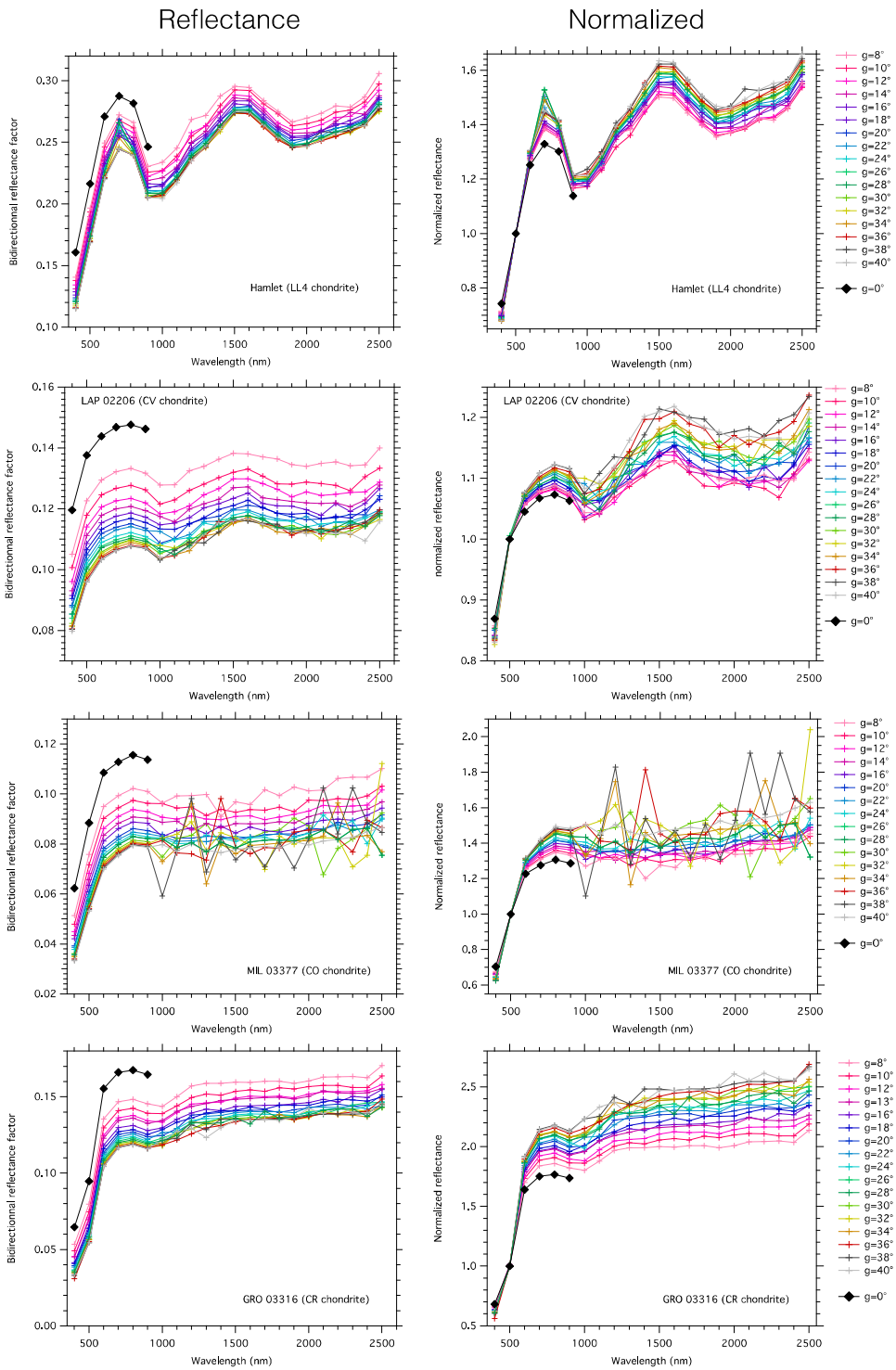


FIGURE 1

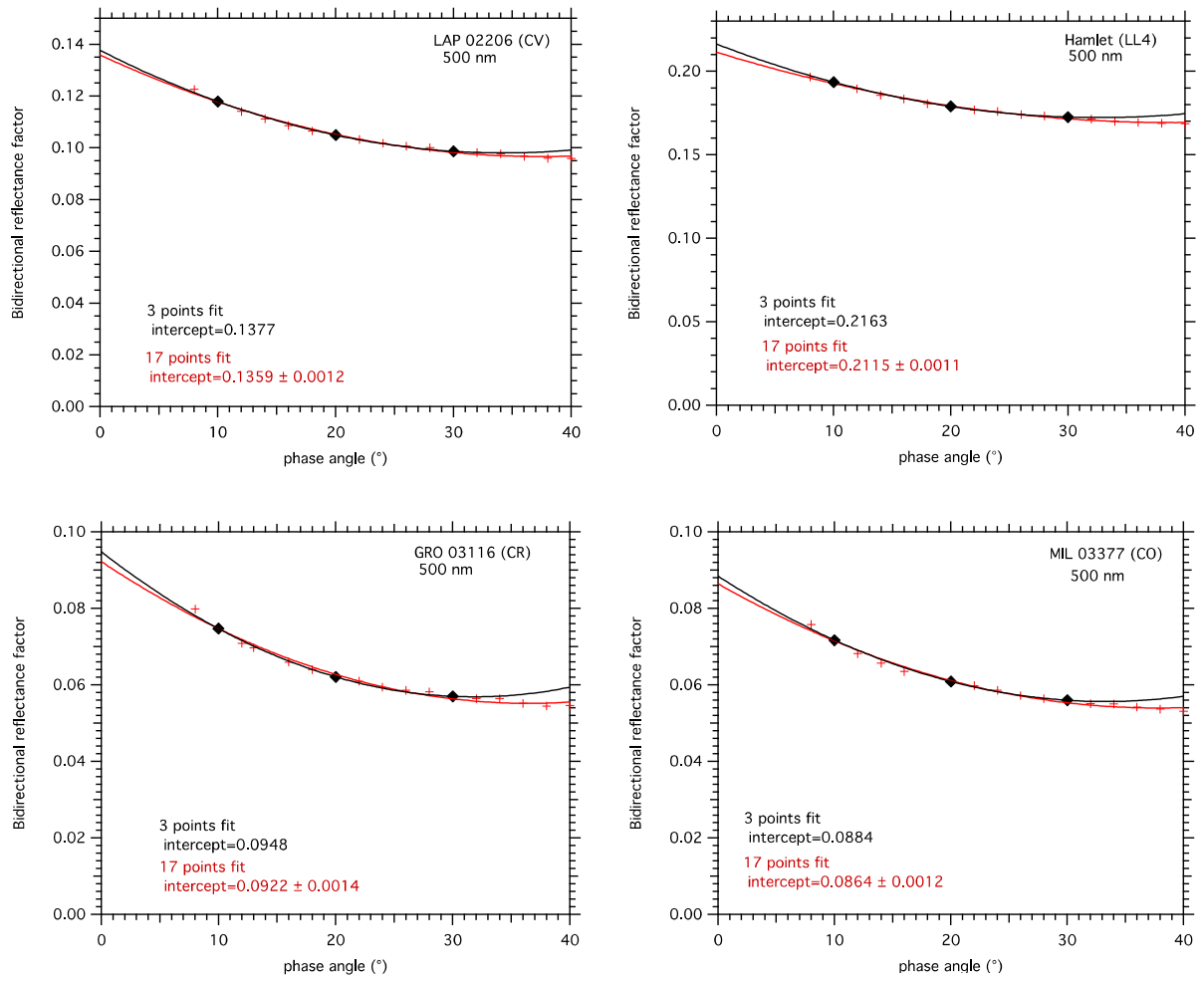


FIGURE 2

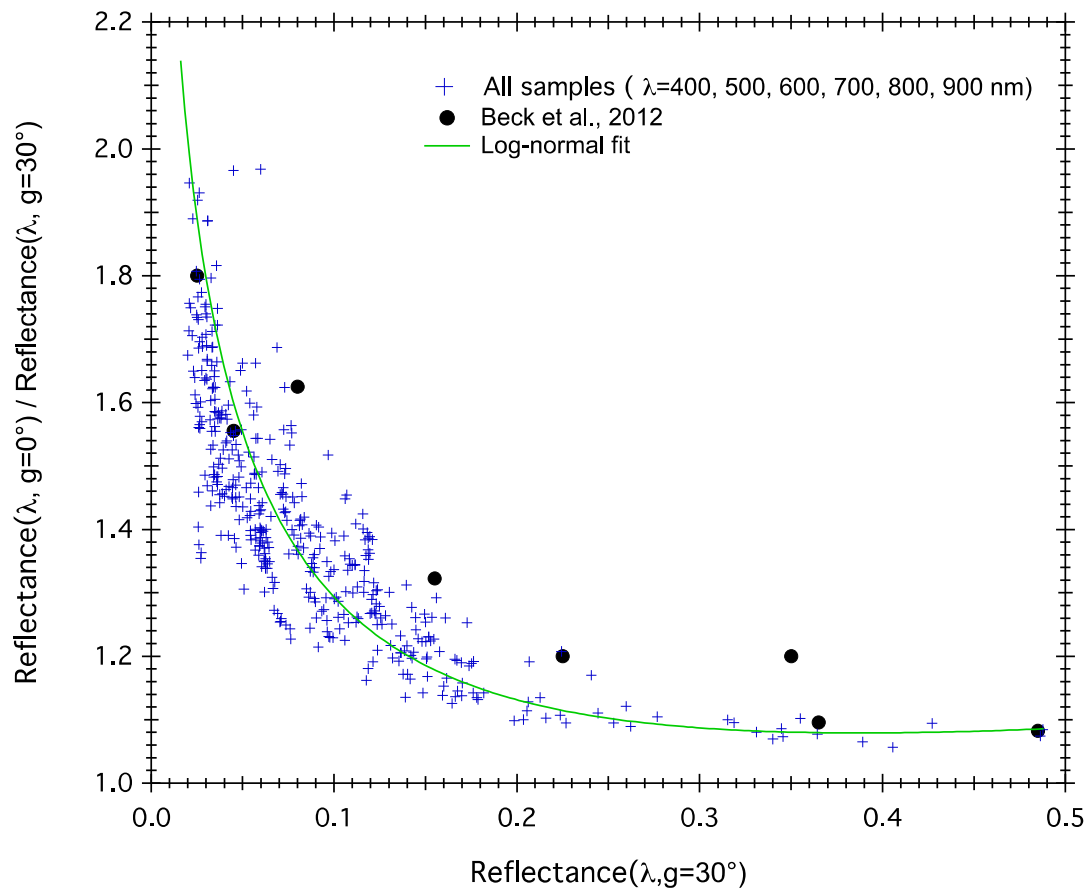
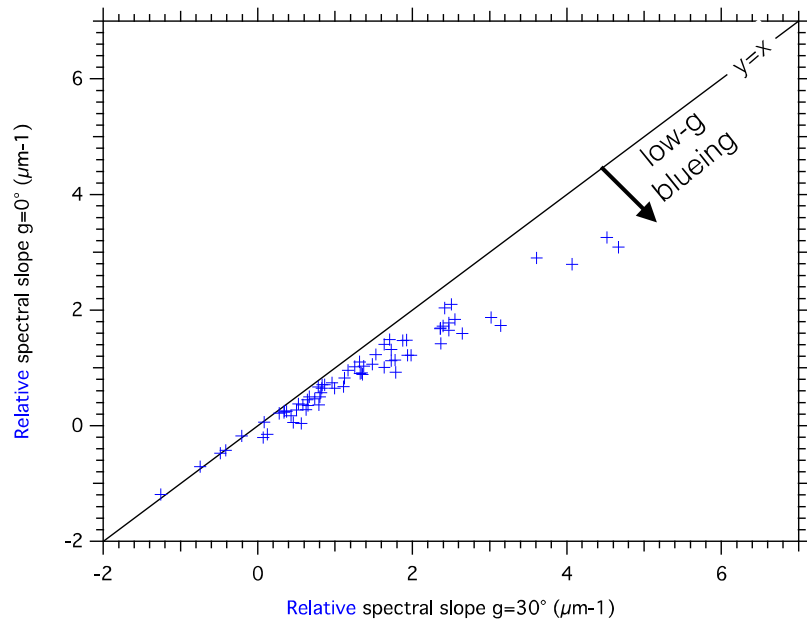
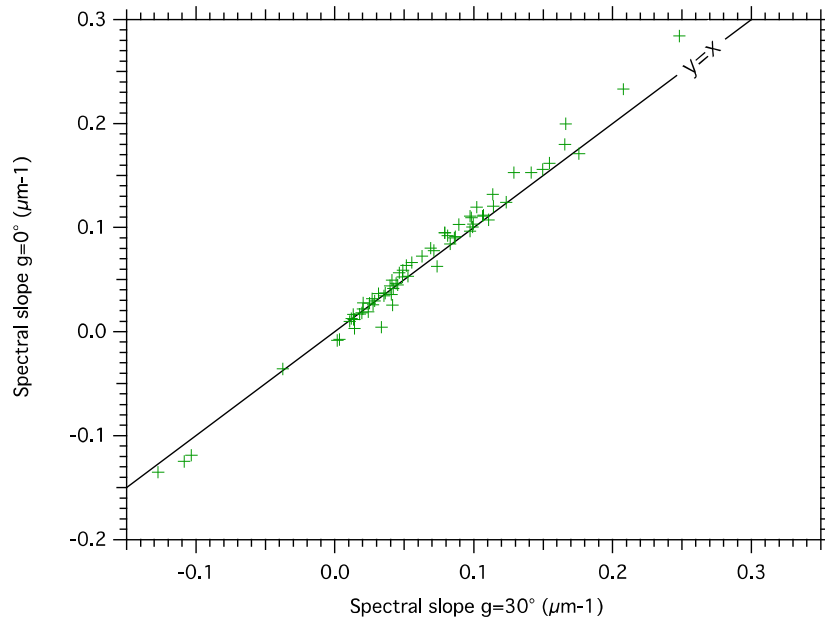


FIGURE 3



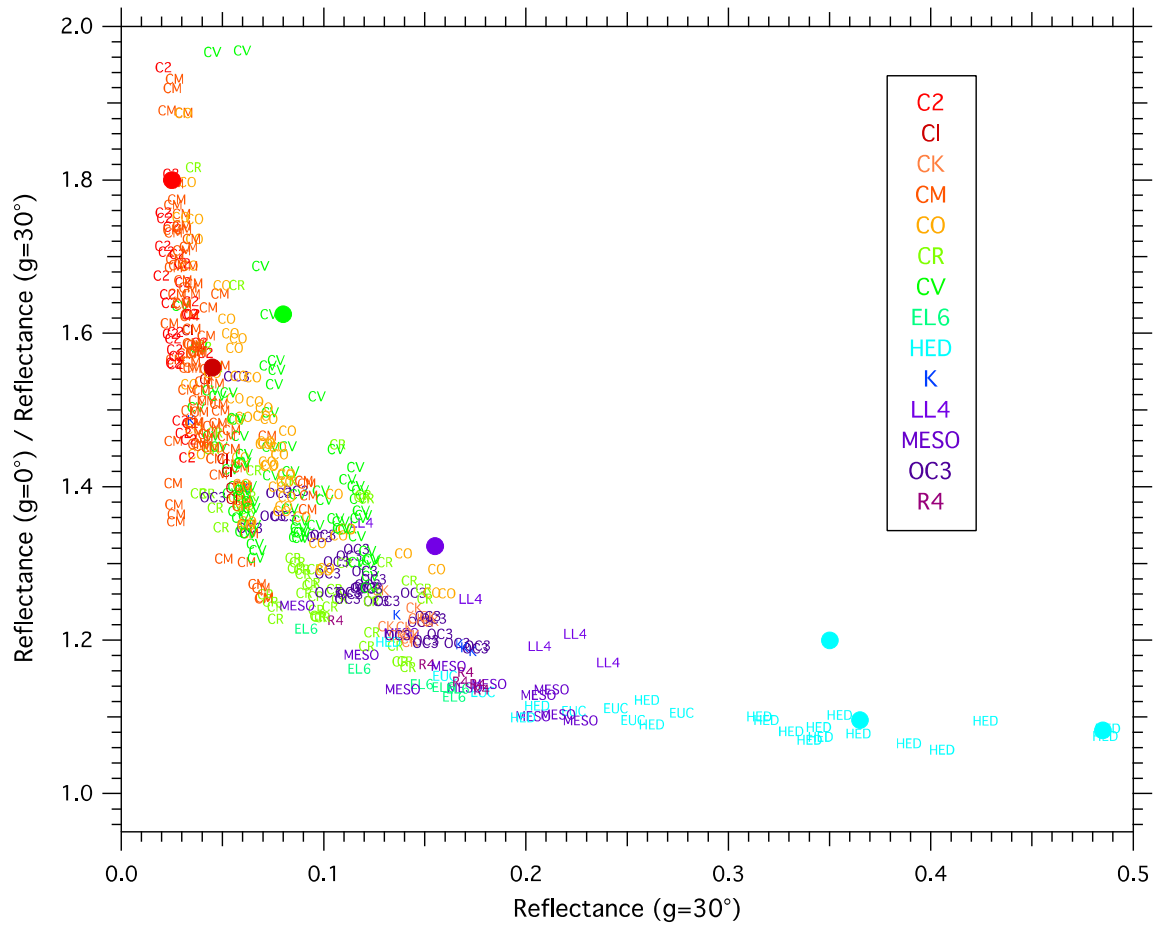


FIGURE 5

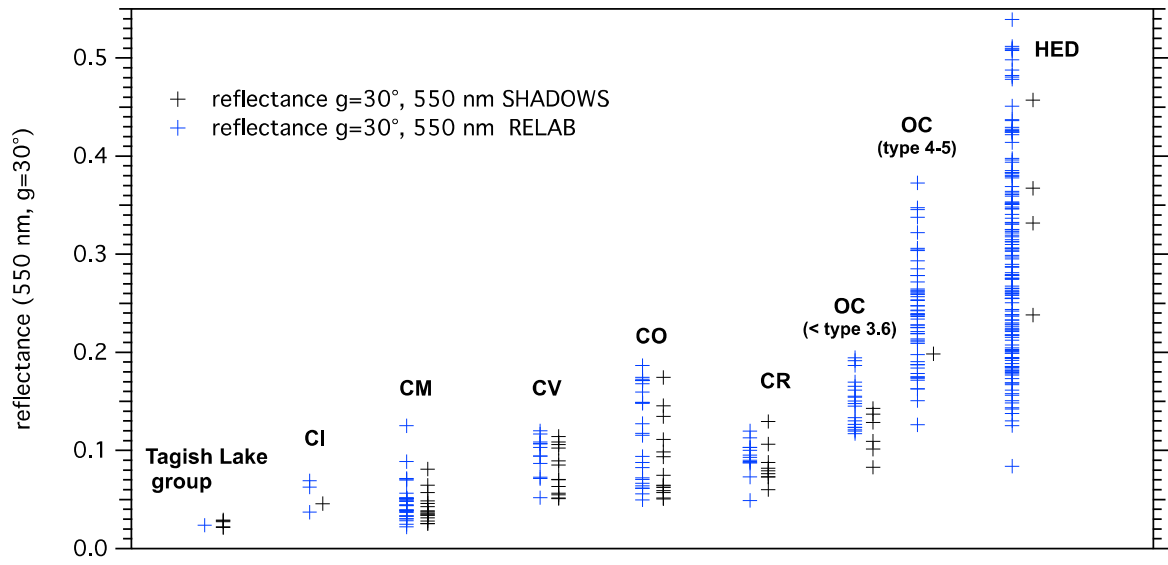


FIGURE 6

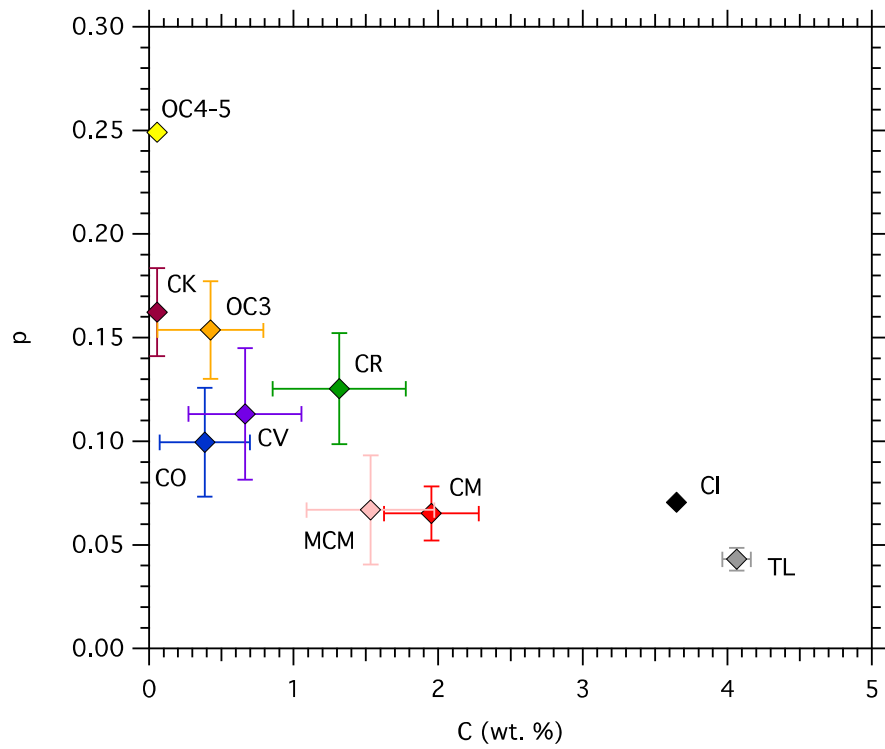


FIGURE 7

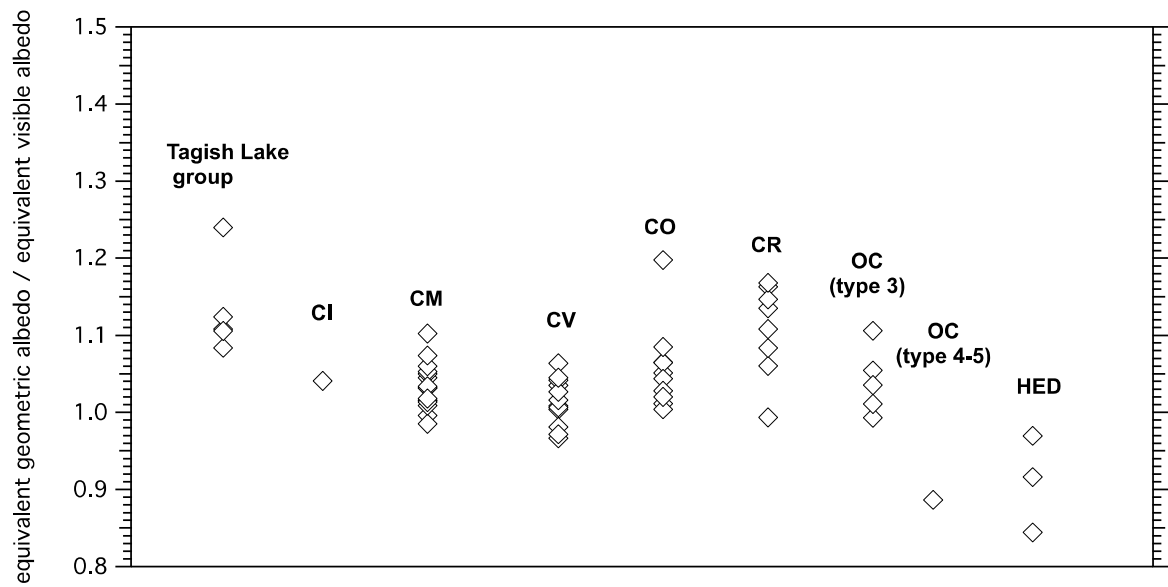


FIGURE 8

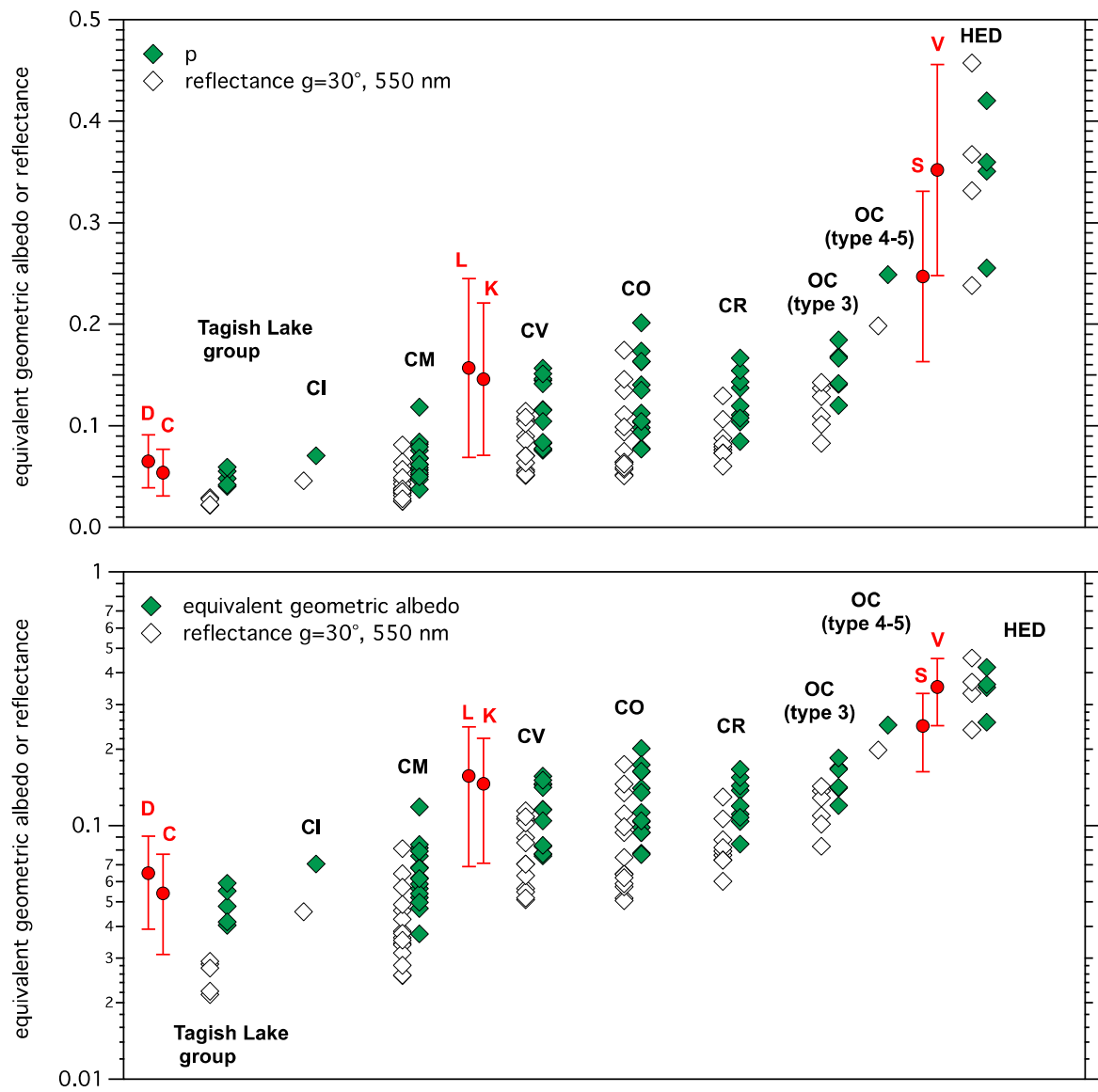


FIGURE 9

Class	Name	reflectance	reflectance	reflectance	reflectance	reflectance	reflectance	"albedo"	reflectance	reflectance	reflectance	reflectance	reflectance	reflectance	reflectance
		400 nm g=30°	500 nm g=30°	600 nm g=30°	700 nm g=30°	800 nm g=30°	900 nm g=30°	550 nm g=0°	550 nm g=0°	400 nm g=0°	500 nm g=0°	600 nm g=0°	700 nm g=0°	800 nm g=0°	900 nm g=0°
C2	MET 00432	0.0268	0.0276	0.0292	0.0306	0.0325	0.0335	0.0481	0.0434	0.043	0.0433	0.0434	0.045	0.0467	0.0489
C2	Tagish Lake 5b	0.0247	0.0278	0.0305	0.0341	0.0339	0.0349	0.0552	0.0491	0.0429	0.0473	0.0509	0.0506	0.0551	0.0568
C2	Tagish Lake 11i	0.0208	0.0246	0.0302	0.0344	0.0386	0.0414	0.0593	0.0478	0.0405	0.0444	0.0511	0.0565	0.0612	0.0652
C2	Tagish Lake 4	0.0199	0.0208	0.0223	0.0238	0.0256	0.0265	0.0405	0.0373	0.0334	0.0366	0.038	0.039	0.0408	0.0418
C2	WIS 91600	0.0206	0.0215	0.023	0.0244	0.026	0.0266	0.0417	0.0378	0.0352	0.0376	0.038	0.039	0.0405	0.0415
CI	Orgueil	0.0325	0.0414	0.0501	0.0523	0.0543	0.0545	0.0706	0.0678	0.0522	0.0637	0.0719	0.0742	0.0759	0.0754
CK	ALH 85002	0.1309	0.1399	0.1419	0.1434	0.1426	0.1362	0.1653	0.1704	0.1593	0.1702	0.1705	0.173	0.1707	0.1643
CK	EET 92002	0.1282	0.1448	0.15	0.1528	0.1516	0.1459	0.1778	0.1817	0.162	0.1798	0.1835	0.1871	0.1866	0.1792
CM	ALH 83100	0.0239	0.033	0.0358	0.034	0.0362	0.0373	0.052	0.0522	0.0385	0.0513	0.0531	0.051	0.0532	0.0543
CM	ALH 84033	0.0321	0.0433	0.0544	0.059	0.0604	0.0598	0.0791	0.0737	0.0505	0.0672	0.0802	0.084	0.0844	0.0837
CM	ALH 84044	0.0258	0.0346	0.0378	0.0364	0.0388	0.0401	0.0565	0.0556	0.0435	0.0541	0.0571	0.054	0.0565	0.0582
CM	Cold Bokkeveld	0.0298	0.0431	0.0492	0.0491	0.052	0.0542	0.0758	0.0735	0.0522	0.0703	0.0767	0.0736	0.0763	0.0785
CM	DOM 03183	0.0309	0.0348	0.0358	0.0348	0.0367	0.038	0.0618	0.0597	0.0582	0.0599	0.0596	0.0559	0.0581	0.0598
CM	EET 96029	0.0253	0.0333	0.042	0.0464	0.0482	0.0479	0.0679	0.0616	0.0486	0.0561	0.0671	0.0712	0.0726	0.071
CM	LEW 87022	0.0304	0.0374	0.0394	0.0387	0.0408	0.0424	0.0621	0.0595	0.0527	0.0589	0.06	0.0579	0.0597	0.0622
CM	LEW 90500	0.0274	0.0332	0.0346	0.0335	0.0353	0.0363	0.0588	0.057	0.0487	0.0569	0.0571	0.0543	0.056	0.0572
CM	MAC 88100	0.025	0.0299	0.033	0.0338	0.0347	0.035	0.0538	0.0534	0.0434	0.052	0.0547	0.0548	0.0551	0.0543
CM	MET 00761	0.0506	0.0619	0.0672	0.0691	0.0707	0.0704	0.0818	0.083	0.0661	0.0806	0.0855	0.0875	0.0888	0.0882
CM	MET 01070	0.0424	0.0544	0.0597	0.0593	0.0618	0.0634	0.0841	0.0801	0.0641	0.0777	0.0824	0.0815	0.0834	0.0848
CM	MET 01072	0.0263	0.0255	0.0257	0.0265	0.0276	0.0279	0.0472	0.0448	0.0509	0.0451	0.0445	0.0449	0.0455	0.0471
CM	MET 01075	0.0264	0.0258	0.0257	0.026	0.027	0.0273	0.0374	0.0368	0.0414	0.0376	0.0361	0.0358	0.0365	0.0372
CM	MIL 07700	0.0227	0.0264	0.0298	0.0305	0.0303	0.0298	0.0498	0.0492	0.0429	0.0474	0.051	0.0509	0.0497	0.0488
CM	Murchison	0.0325	0.0408	0.0447	0.044	0.0463	0.0481	0.0682	0.0644	0.0496	0.0627	0.0661	0.0638	0.0665	0.068
CM	PCA 02012	0.0488	0.0721	0.0902	0.0916	0.0925	0.0921	0.1184	0.1163	0.0805	0.1057	0.1269	0.1286	0.1285	0.1262
CO	ALH 77003	0.0577	0.0821	0.1052	0.1114	0.1113	0.1075	0.1351	0.1336	0.0919	0.1209	0.1463	0.1497	0.1497	0.1436
CO	DOM 08006	0.0372	0.0461	0.0572	0.0606	0.0625	0.0617	0.0779	0.0731	0.0537	0.0669	0.0792	0.0835	0.0846	0.0833
CO	MIL 03377	0.0362	0.056	0.073	0.0783	0.0816	0.0809	0.1035	0.0985	0.0623	0.0885	0.1086	0.1128	0.1156	0.1138
CO	MIL 03442	0.0357	0.0604	0.0892	0.097	0.101	0.1001	0.1125	0.1057	0.0527	0.0901	0.1213	0.1287	0.1307	0.1292
CO	MIL 05013	0.0363	0.0562	0.0721	0.077	0.0802	0.0793	0.0983	0.0942	0.0634	0.0851	0.1033	0.1077	0.1102	0.1085
CO	MIL 05024	0.0337	0.045	0.0562	0.0589	0.0603	0.0594	0.0769	0.0766	0.0516	0.0696	0.0836	0.0847	0.0843	0.0834
CO	MIL 09001	0.0307	0.0498	0.0648	0.0693	0.072	0.0713	0.0939	0.0914	0.0579	0.0827	0.1	0.1035	0.1048	0.104
CO	MIL 090038	0.0335	0.0521	0.0659	0.0705	0.0735	0.073	0.0938	0.092	0.0566	0.0844	0.0996	0.1026	0.1049	0.1041
CO	MIL 090073	0.0326	0.0539	0.0706	0.0773	0.0813	0.0818	0.1043	0.0961	0.0586	0.0862	0.106	0.1122	0.115	0.1134
CO	MIL 090785	0.0294	0.0582	0.1394	0.1558	0.1609	0.1535	0.1634	0.1364	0.0515	0.0898	0.183	0.2012	0.2028	0.1936
CR	EET 92042	0.0444	0.0627	0.0903	0.0971	0.0993	0.0976	0.1108	0.0992	0.0644	0.0846	0.1138	0.1195	0.1221	0.12
CR	EET 92159	0.0407	0.0654	0.1103	0.1212	0.1254	0.1235	0.1375	0.1182	0.0644	0.0929	0.1434	0.1444	0.1581	0.1544
CR	GRA 95229	0.0461	0.0627	0.0956	0.1053	0.1077	0.1031	0.1195	0.1053	0.0678	0.087	0.1236	0.1333	0.135	0.1281
CR	GRO 03116	0.0356	0.057	0.1069	0.1195	0.121	0.1187	0.1435	0.1251	0.0647	0.0948	0.1555	0.166	0.1674	0.1646
CR	LAP 04720	0.0464	0.0618	0.0848	0.0891	0.0898	0.086	0.1036	0.0978	0.0637	0.0847	0.1108	0.1151	0.1154	0.1112
CR	MET 00426	0.0422	0.0591	0.087	0.0943	0.096	0.0932	0.1077	0.0972	0.0587	0.0812	0.1132	0.1202	0.1206	0.1185
CR	MIL 90657	0.038	0.0493	0.0712	0.0759	0.0762	0.0736	0.0845	0.078	0.0529	0.0664	0.0896	0.0944	0.0936	0.0919
CR	PRE 95411	0.0962	0.1237	0.1354	0.1402	0.1416	0.1377	0.1545	0.1556	0.1192	0.1497	0.1614	0.1642	0.1649	0.1613
CR	QUE 99177	0.046	0.0826	0.1302	0.1423	0.1493	0.1499	0.1668	0.1428	0.0715	0.1163	0.1694	0.1817	0.1891	0.188
CV	ALH 81003	0.0966	0.1118	0.1169	0.1185	0.1196	0.1173	0.1567	0.1597	0.1466	0.1575	0.1618	0.1621	0.1629	0.1594
CV	ALH 85006	0.0444	0.0542	0.0587	0.0604	0.0626	0.0625	0.077	0.0776	0.0651	0.0754	0.0798	0.081	0.0837	0.0837
CV	Axtell	0.045	0.0729	0.106	0.1159	0.118	0.1157	0.1447	0.136	0.0884	0.1184	0.1536	0.1623	0.1647	0.1649
CV	Efremovka	0.0737	0.096	0.1162	0.1212	0.1224	0.1224	0.1462	0.1403	0.1069	0.1295	0.1512	0.1573	0.1578	0.1552
CV	Grosnaja	0.0741	0.0837	0.0872	0.0884	0.0893	0.0878	0.1159	0.12	0.1047	0.1188	0.1211	0.1198	0.1196	0.1177
CV	Kaba	0.0475	0.0531	0.0565	0.0586	0.0606	0.06	0.0828	0.0825	0.0721	0.0809	0.0841	0.0858	0.0874	0.0858
CV	LAP 02206	0.0824	0.0987	0.1059	0.1084	0.1099	0.1087	0.1416	0.1407	0.1196	0.1376	0.1438	0.1468	0.1476	0.1463
CV	LAR 06317	0.0369	0.0481	0.0539	0.0568	0.0594	0.06	0.0758	0.0732	0.0555	0.0698	0.0767	0.0796	0.0824	0.0838
CV	MET 00430	0.0569	0.0622	0.0647	0.066	0.0674	0.0665	0.0836	0.086	0.0779	0.0847	0.0873	0.0875	0.0887	0.0869
CV	MIL 07671	0.0735	0.1004	0.1165	0.1217	0.1238	0.123	0.1511	0.1472	0.11	0.1388	0.1556	0.1601	0.1617	0.1603
CV	Mokoia	0.0597	0.0688	0.0724	0.0756	0.0767	0.0765	0.1154	0.1144	0.1175	0.116	0.1128	0.1159	0.1191	0.1197
CV	RBT 04133	0.0382	0.0573	0.0833	0.0877	0.0885	0.0869	0.1043	0.0998	0.0604	0.0853	0.1142	0.118	0.118	0.1159
CV	RBT 04302	0.0292	0.0442	0.0594	0.0627	0.0643	0.0637	0.0774	0.0762	0.0477	0.0674	0.085	0.0878	0.0882	0.0879
EL6	Khaipur	0.0914	0.1176	0.1485	0.1593	0.1644	0.1662	0.1716	0.1531	0.1109	0.1367	0.1696	0.1813	0.1851	0.1888
HED	Howardite	0.1786	0.2236	0.2528	0.2768	0.2443	0.1598	0.2554	0.2622	0.2022	0.2475	0.2769	0.3058	0.2713	0.1843
HED	Millbillillie	0.2597	0.3188	0.3446	0.3643	0.331	0.2054	0.3507	0.3617	0.2913	0.3491	0.3742	0.3925	0.3577	0.2289
HED	NWA 5356	0.2621	0.3455	0.3891	0.4055	0.34	0.1984	0.3598	0.3927	0.2856	0.3709	0.4144	0.4285	0.3638	0.218
HED	Tatahouine	0.355	0.427	0.4875	0.4863	0.3152	0.1319	0.4204	0.4979	0.3911	0.4673	0.5286	0.5225	0.3467	0.1579
K	LEW 87232	0.0346	0.0612	0.1362	0.1671	0.1737	0.1696	0.1623	0.1267	0.0514	0.0855	0.1679	0.1995	0.2059	0.2019
OC	Hamlet	0.1188	0.1726	0.2243	0.2583	0.2407	0.2067	0.249	0.2436	0.1607	0.2163	0.2709	0.2551	0.2817	0.2463
MESO	CMS 04021	0.1385	0.1817	0.2061	0.2126	0.1616	0.0868	0.1951	0.22	0.1674	0.2076	0.2325	0.2413	0.1884	0.108
MESO	LEW 87006	0.1186	0.1698	0.2158	0.2269	0.2035	0.139	0.2075	0.2156	0.14	0.1932	0.2379	0.2484	0.2239	0.1578
OC	ALH 76004	0.0751	0.0982	0.1206	0.1247	0.1205	0.1064	0.1407	0.142	0.1023	0.1309	0.1531	0.1595	0.1553	0.1385
OC	EET 83248	0.0454	0.0636	0.1022	0.1131	0.112	0.1022	0.1201	0.1086	0.0629	0.0856	0.1315	0.1425	0.1404	0.1289
OC	LEW 87248	0.0999	0.1261	0.148	0.1508	0.1504	0.1367	0.1682	0.1694	0.1335	0.1577	0.181	0.1808	0.1799	0.1648
OC	MET 00489	0.078	0.1127	0.1445	0.1545	0.1516	0								