Nanomaterials for Photovoltaic Conversion

Abdelilah Slaoui¹, Daniel Lincot²,

Jean François Guillemoles^{2,4}, Ludovic Escoubas³

¹Laboratoire des Sciences de l'Ingénieur, de l'Informatique et de l'Imagerie (Icube) -UMR CNRS-UdS, 23 rue du Loess, F-67037 Strasbourg , France

²Institut de Recherche et Développement sur l'Energie Photovoltaîque (IRDEP) -UMR CNRS-EDF-Chimie paristech, EDF R&D, 6 Quai Watier, F-78401 Chatou, france

³*Institut Matériaux Microélectronique Nanosciences de Provence (IM2NP)* – UMR CNRS- Aix-Marseille University, Campus de Saint-Jérôme, Avenue Escadrille Normandie Niemen , F-13397 Marseille, France

4, NextPV, LIA CNRS-U. Bordeaux-U. Tokyo, RCAST 4-6-1 Komaba, Meguro Ku, Tokyo, 153-8904 Japan

Abstract :

This chapter provides an overview on the concepts, materials and devices that involve nanoscience for photovoltaics. First, the general basics of photovoltaics in terms of principle and technologies will be presented. Then, the features specifically associated with the use of nanomaterials in current and future photovoltaic devices will be developed. The optical effects (quantum confinement, photonics, plasmonics), electrical effects (work function) and geometrical effects (interpenetrating structures) - which are intrinsic to nanomaterials - will be described at theoretical and experimental levels. Properties of solar cells devices involving nanomaterials such as dye and organic cells, ultrathin cells implementing metal nanostructures, new concept based cells (up and down conversion, intermediate band cells, multiexciton generation, hot carriers ...) will be presented.

1-Introduction

Since its emerging in 1954, the field of photovoltaics, direct sunlight-electricity conversion phenomenon, has been closely associated with a class of semiconductors, basically silicon, III-V semiconductor compounds (GaAs-based), II -VI such as CdTe, or more recently I-III-VI₂ such as Cu(In,Ga,S)Se₂ noted CIGS. The associated photovoltaic cells are based on concepts derived from p-n junctions. The absorbing materials consist in either thick to thin (50 – 300 μ m) silicon wafers or single or stack of micrometers thick $(0,5 - 2 \mu m)$ layers for other inorganic materials¹. More recently, extensive studies have shown that nanochemistry and nanomaterials can provide numerous opportunities for a new generation of photovoltaics with high solar energy conversion efficiencies at low fabrication cost. For instance, threedimensional (3-D) arrays of nanostructures such as nanowires (NW), nanopilars (NP), nanocone, etc. have demonstrated significantly improved photon absorption and photo-carrier collection efficiencies^{2,3,4}. Other unique physical features of the nanomaterials, e.g large surface-to- volume ratio⁵,⁶ and quantum confinement⁷, have also shown a large potential to enable photovoltaics with new mechanisms, like in the case of dye-sensitized solar cells and quantum dot solar cells^{8,9}. On the other hand, photon management realized in nanostructures are offering great opportunities for harvesting underutilized ultraviolet and currently unutilized infrared photons.

In this chapter, we will summarize recent progress involving nanomaterials and nanostructures in the photovoltaic field. We will beging with a brief review of state-ofart research on PV technologies. Thereafter, we will summarize the recent development of studies on optical properties of nanomaterials/nanostructures and examples of solar cell devices based on these materials and structures. We will highlight in particular how quantum-confinement effect in some nanostructures can be used to harvest sun light over a broad range of the spectrum ; also how some geometric confinment effects, using plasmons for instance, can offer effective ways to reduce the thickness of light-absorbing layers. Finally, we will present concepts and nanostructures that are of significant interest for harvesting underutilized ultraviolet and currently unutilized infrared photons. It concerns photon downconversion and dow-shifting, photon up-conversion and multiple exciton generation.

¹ L.L. Kazmerski, "Solar photovoltaics R&D at the tipping point: a 2005 technology overview," J. Electron. Spectroscopy 150, 105 (2006).

Z.Y. Fan, H. Razavi, J.W. Do, A. Moriwaki, O. Ergen, Y.L. Chueh, P.W. Leu, J.C. Ho, T. Takahashi, L.A. Reichertz, S. Neale, K. Yu, M. Wu, J.W. Ager, A. Javey, Nature Materials 8 (2009) 648.

³ E.C. Garnett, P.D. Yang, Nano Letters 10 (2010) 1082.

⁴ K. Cho, D.J. Ruebusch, M.H. Lee, J.H. Moon, A.C. Ford, R. Kapadia, K. Takei, O. Ergen, A. Javey, Applied Physics Letters 98 (2011) 203101.

⁵ A.J. Nozik, M.C. Beard, J.M. Luther, M. Law, R.J. Ellingson, J.C. Johnson, Chemical Review 110 (2010) 6873.

⁶ Octavi E. Semonin, Joseph M. Luther, and Matthew C. Beard, Materials Today, V.15 (11), Issue 11 pp. 508–515 (2012) ⁷ .M. Kayes, H.A. Atwater, N.S. Lewis, Journal of Applied Physics 97 (2005) 114302.

⁸ Qing Shen Junya Kobayashi, Lina J. Diguna and Taro Toyoda ; J. Appl. Phys. 103, 084304 (2008)

⁹ Francesco Priolo, Tom Gregorkiewicz, Matteo Galli & Thomas F. Krauss, Nature Nanotechnology 9, 19-32 (2014)

2- Photovoltaic Materials and Technologies: State of Art

2.1- Principle and limiting factors

Photovoltaic consist in a direct conversion of sun light into electricity. The principle is very simple: absorption of photons by an appropriate semiconductor, generation of electron-hole pairs, dissociation of the pairs through a pn junction or interfaces, and finally collection of the carriers by metal contacts.

Figure 1 shows the results of theoretical calculation of loss mechanisms in single pn junction solar cells versus the band gap of the semiconductor. We distinguish between the losses associated with lower energy photons, the thermalization losses associated with higher energy photons, in addition to other loss mechanisms. In the absence of such losses, the theoretical efficiency of photovoltaic cells may reach 80 to 90%. The figure shows that for a pn junction involving one single semiconductor, the theoretical conversion efficiency varies with the bandgap energy and reaches an optimum at about 30% for bandgap values in the range 1.2-1.5 eV. This is because the maximum photovoltage is proportional to the bandgap of the semiconductor, $V_{co}^{max} = Eg / q$ (in fact, only 2/3 of this value can be reached at the best), while its generated photocurrent decreases because all the photons of energy below the bandgap are not collected. Such opposite behaviors result in the presence of an optimum of efficiency.



Figure 1. Theoretical analysis of the loss mechanisms in a photovoltaic cell based on the bandgap. The blue area corresponds to the theoretical efficiency of a single junction cell under illumination AM1.5 (Hirst et al, 2011).

2.2- Present Photovoltaic technologies

Figure 2 depicts the cell structures of the most used devices. Figure 2a is the structure using crystalline or multicrystalline silicon wafers, which represents about 90% of the market today. It is a typical N-P homojunction where conventionally, the N region is a phosphorus-doped silicon and the P region is a boron doped silicon substrate. Figure 2b concerns thin-film amorphous silicon (about 5% of the market). This is also a kind of homojunction where the intrinsic absorbing amorphous layer is sandwished in between the n and p type regions. Figure 2c displays the cell structure based on cadmium telluride (CdTe) whose market share increased from 1% in 2005 to nearly 8% in 2014. This is a heterojunction cell as the n- and p-type semiconductor materials are different, namely here CdS and CdTe resoectiveley. Finally, figure 2d gives the scheme of a CIGS (copper- indium- gallium diselenide) cell which cover approximately 1% of the market, still under strong development at labs and industry levels.



Figure 2. Schematic representation of the different current photovoltaic technologies: A) Conventional cell based on mono or multicrystalline silicon wafers. The pn junction is completed by a front electrical contact for collecting photogenerated electrons (in the case of a silver grid), and a back contact usually consisting of aluminium and deposited on the entire surface (screen printing process). The latter can collect the photogenerated holes. A texturing of the surface and the setting of an antireflection coating makes it possible to limit the optical losses by reflection. B) thin film cell based on amorphous silicon ; the front contact is transparent conductive tin oxide layer, while the back contact is a silver layer. C) thin film cell based on cadmium telluride (CdTe). The pn junction is made with cadmium sulfide (CdS). Contacts are tin oxide on one side and carbon on the other one. D) Cell based on thin-film copper indium and gallium diselenide (CIGS). The pn junction is made with CdS. The front contact is made of zinc oxide while the back contact is molybdenum.

Figure 3 shows the evolution of record efficiencies for solar cells employing the different technologies¹⁰. The data concerns mature technologies such as those involving crystalline silicon, thin films technologies based CIS, CdTe, or amorphous Si, single-junction GaAs, III-V multijunction compounds, as well as emerging technologies like in the case of organic or hybrid structures. One can notice that the record conversion efficiency reaches more than 46% for the multijunction based on GaAs, and it is lying between 22,1 and 25,6% for single junction cells (crystalline Si, CIS, CdTe), while it is between 13 to 21% for recently developed solar cells. Figure 3 plots also efficiencies data for cells made using nanostructures such as the dye sensitized cells (DSSC) and interpenetrating network organic cells (ORSC) or using nanomaterials such as the PbS quantum dots cells (QDSC). Much more recently, hybrid organic-inorganic cells like those based on perovskite materials have emerged and exhibit a very high potential with a record efficiency approaching 21% on small area.

One important message from Figure 3 is that a strong research and a smart engineering is necessary over years to reach a maturity for any technology.

The second message is that, besides the multijunction cells, none of the technologies based on a single junction have passed the Schockley-Queisser limit as the limiting losses are still present.



Figure 3. Evolution of record efficiencies obtained for the different photovoltaic technologies (Source NREL, http://www.nrel.gov/ncpv/images/efficiency_chart.jpg).

3- Nanomaterials for photovoltaics

To overcome the limitations mentioned in figure 1, innovation on material and device structure has become the cutting-edge of the current research. The focus is mainly on the improvement of the photon absorption and photon-carrier collection

¹⁰ http://www.nrel.gov/ncpv/images/efficiency_chart.jpg

efficiencies. Nanochemistry and nanomatériaux, provide numerous opportunities for a new generation of photovoltaics with high solar energy conversion efficiencies at low fabrication cost, provided it is possible to engineer the material properties through its chemical composition but also by tailoring the size and the shape. Here, we will Quantum-confined nanomaterials polymer-inorganic examples of and nanocomposites that can be tailored to harvest sun light over a broad range of the Spectrum. We will also review how plasmonic structures offer effective ways to reduce the thickness of light-absorbing layers. Finally, we will mention some concepts such as multiple exciton generation, photon down-conversion, and photon up-conversion realized in nanostructures that have significant interest for harvesting underutilized ultraviolet and currently unutilized infrared photons.

3.1- Quantum-confinement effects for photovoltaic

3.1.1- Principle

The nanostructures are characterized by novel and interesting properties that may be seriously different from those of the bulk material. At the nanoscale, the energies which are characteristics of the electronic levels, their interactions with photons, as well as the vibration modes of the atoms are strongly modified compared to their values in the bulk structure. This structuring at the nanoscale will thus open new opportunities in the photovoltaic conversion process. Lower is the size of these nano-objects compared to the bulk material, greater will be the gap between their physical properties and those at conventional shape. In particular, reducing the size of crystalline semiconductors to the nanoscale leads to remarguable effects such as the quantum confinement. At nanometric scales, the electrons behave more like waves than particles, and thus as phenomena that cannot be localized without affecting strongly the reachable energy levels (Heisenberg uncertainty principle). Based on the confinement direction, a quantum-confined structure will be classified into three categories as quantum well, quantum wire and quantum dots or nanocrystals. As more number of the dimension is confined, more discrete energy levels can be found, in other words, carrier movement is strongly confined in a given dimension. The quantum states densities and the energy dispersions versus the wave vector (k = $2\pi/\lambda$) for the above-mentioned cases are shown in Figure 4. In the quantum dots for example, the energy does not vary linearly anymore. It is this feature that makes quantum dots of great interest to a number of advanced concepts and to the fabrication of a new branch of photovoltaics devices. These quantum dots can also be, in many ways, considered as "artificial atoms". The quantum dots as well as nanolayers or nanowires are often embedded in a material in which the electron wave functions do not overlap significantly. In the case of semiconductor nanocrystals such as silicon, germanium or lead sulphide, the host material is generally an insulating matrix allowing the transfer of electrons from one nanostructure to another by the tunnel effect.



Figure 4. Density of electron states of a semiconductor as a function of its dimension. The optical absorption spectrum is roughly proportional to the density of states.

Figures 4 b to d show that the reduction of the semiconductor crystal size to the nanoscale induces the discretization of the energy. The singular feature is that the width of the band gap E_g can be tuned versus the size of the nanowire or nanodot for instance. This property can be exploited to absorb and convert more photons and / or generate more carriers with respect to the bulk material. The materials can be structured in the form of nanotubes, nanowires, quantum wells or quantum dots. In the case of spherical nanoparticles (nanodots) with a radius *R*, the energy bandgap is given by the relationship:

$$E_{min} = \frac{\hbar^2 \pi^2}{2R^2} \left[\frac{1}{m_{\theta}^*} + \frac{1}{m_{h}^*} \right] - \frac{1.8q^2}{\epsilon_{\theta}R} - 0.25 E_{Ryd}^*$$

 E_{Ryd} is the binding energy of excitons while m_e, m_h are the effective masses of electrons and holes, and h is Planck's constant. The second term takes into account the electrostatic interactions between the electron and the hole.

Figure 5 shows the variation of the optical gap as a function of the size of the nanoparticle for various semiconductors.



Figure 5. Theoretical evolution of the optical bandgap of various types of spherical nanocrystals, according to their size and the nature of the material.

It can be seen that the quantum effect can increase significantly the value of the gap by several eV. Thus using the same material but varying its size, it becomes possible to tune the energy bandgap at a desired value. Figure 5b shows the results obtained in the case of silicon nanoparticles, which is the subject of numerous investigations as we will see below. The energy bandgap can be changed from 3 eV to 1.3 eV for Si-QDs of 1.2nm to 5nm respectively. Thus, one can say that the use of quantum size effects generates the possibility of engineering the bandgap of conventional semiconductors to a large extend, which is possible only by changing the chemical composition by introducing massive amount of impurities in a bulk material.

Figure 6 illustrates the dramatic effect of changing the bandgap in the case of CdTe nanoparticles, on the optical abosrption. The effect of particle size is clearly illustrated in the optical absorbance spectra (left) that show a blue shift (increase of the optical energy bandgap) when decreasing the particles size. The photograph at the right shows vials containing colloidal suspensions of CdTe QDs of different sizes under UV excitation.





Figure 6. (Left) Absorption spectra of colloidal suspensions of CdTe nanocrystal (NC) quantum dots of different sizes. (Right) Photograph of vials containing colloidal suspensions of CdTe QDs of different sizes under UV excitation. Reprinted with permission from Ref. [Groeneveld, E.: Synthesis and optical spectroscopy of (hetero)-nanocrystals. Ph.D. Thesis, Utrecht University, Utrecht (2012)

Another interesting consequence of the quantum confinement effect is the modification of the electron affinity of the semiconductor¹¹, which may result in a change of the energy diagrams. Figure 7 shows the example of the effect of quantum confinement in the case of CdSe nanoparticles on the photogenerated electron injection into the titanium oxide conduction band (Kongkanand et al., 2008).



Figure 7. Schematic representation of changes in the energy diagram of CdSe nanoparticles in contact with the titanium oxide (ref. 11). In Y axis is represented the energy with the conduction band and valence positions for the titanium oxide and for the CdSe particles according to their size. The abscissa axis represents the distance. The effect of the size of the CdSe particle is represented by different graphs on the x-axis for convenience.

In the following are reported few examples of solar cells illustrating whow the quantum-confinement effect is exploited.

3.1.2- Solar cells based on PbS or PbSe nanoparticles

The high absorption cross-section of colloidal nanocrystals or CQDs at tunable wavelengths makes them attractive for thin film solar cells, since this characteristic allows for optimal harvesting of the solar spectrum, thereby opening a promising route to high efficiency¹²,¹³. Moreover, just like for QD-LEDs, solution processability is a great asset, offering the prospect of low fabrication costs and easy upscaling. The exciton dissociation in QDs is also relatively easy and can be boosted by using heteronanocrystals architectures. These properties have led to the utilization of CQDs in a variety of different solar cell concepts¹⁴,¹⁵.

¹¹ Kongkanand A., Tvrdy K., Takechi K., Kuno M., Kamat P.V. (2008). J. Am. Chem. Soc. <u>130</u>, 4007.

¹² Konstantatos, G., Sargent, E.H.: Colloidal quantum dot optoelectronics and photovoltaics. Cambridge University Press, Cambridge (2013),

¹³ Nozik, A.J., Conibeer, G., Beard, M.C.: Advanced concepts in photovoltaics. Royal Society of Chemistry, Oxford (2014).

¹⁴ Lunt, R.R., Osedach, T.P., Brown, P.R., Rowehl, J.A., Bulovic, V.: Practical roadmap and limits to nanostructured photovoltaics. Adv. Mater. 23, 5712–5727 (2011)

As an example, bulk PbSe semiconductor has a low band gap of 0.27 eV which is obviuously not suitable for efficient photovoltaic conversion of solar energy. In contrast, the quantum size effect is potentially interesting to be able to increase the bandgap of a layer composed of PbSe nanoparticles (PbSe-NP). Similarly, bulk PbS with a gap of 0.37 eV is a good candidate for these studies¹⁶.

Synthesis of a PbS film composed of monodisperse nanoparticles (3 to 5 nm in diameter) can be performed by self-assembly of nanoparticle solution. The overlapping of the wave functions of the particles decreases the quantum confinement effect and therefore they stay spatially separated. In contrast, the transfer of electrons (and holes) from one particle to the other must remain possible and fast enough to avoid introducing series resistances which are detrimental to a high conversion efficiency. This question is central for the use of nanoparticles based films as there is indeed a conflict between quantum confinement effect and electron mobility. This highlights the interest for structures in which the confinement is possible in one or two directions only, preserving a direction for the efficient transfer of electrons. The transfer can be done by tunneling if the particles medium is insulating or by conduction if the particles medium is conducting. The results obtained with cells using PbS films have reached conversion efficiencies as high as 4% which is particularly encouraging (Figure 8). The performances are mostly due to low charge carrier mobility through the CQD-film and contact losses. Also, there is a demand for proper band alignment at electrodes for efficient charge extraction at minimal cost to voltage.



Figure 8. Photovoltaic properties of PbSe-NP based cells with bandgap (particle size) of 1,1 eV (4,3 nm) et 1,3 eV (3,5 nm). The conversion efficiency is of about 4%. *(Ref. 16).*

Recently, self-assembled monolayers (SAMs) were be used to modify interface energy levels locally. This was done through a process that secured anchoring of aromatic SAMs, aided by deposition of the SAMs in a water-free deposition environment¹⁷. The energy alignment at the rectifying interface was tailored by tuning the robust SAM (R-SAM) for optimal alignment relative to the CQDs quantum-

¹⁵ Kramer, I.J., Sargent, E.H.: Colloidal quantum dot photovoltaics: a path forward. ACS Nano 5, 8506–8514 (2011)

¹⁶ Szandrei et al., 2010

¹⁷ Gi-Hwan Kim, F. Pelayo García de Arquer Yung Jin Yoon, Xinzheng Lan, Mengxia Liu, Oleksandr Voznyy, Zhenyu Yang, Fengjia Fan, Alexander H. Ip, Pongsakorn Kanjanaboos, Sjoerd Hoogland, Jin Young Kim, and Edward H. Sargent, *Nano Lett.*, 2015, *15* (11), pp 7691–7696

confined electron energy levels. Figure 9a gives the schema of the PbS CQDs based cell using the SAMs structure. Figure 9b displays the illuminated J-V characteristics of the CQDs/SAM cell exhibiting a record power conversion efficiency (PCE) of 10.7% (17).



Figure 9: (a) schema of the PbS CQDs based cell using the SAMs structure; (b) J-V characteristics under illumination of the CQDs/SAM cell (ref. 17).

3.1.3- Solar cells based on silicon nanoparticles

One of the most interesting examples of quantum effects using silicon nanostructures (wells, wires, dots) is to produce multijunctions solar cells based solely on silicon (figure 10a). In the case of silicon nanocrystals, they are often embedded in a dielectric matrix obtained by several deposition techniques such as Plasma enhanced CVD, evaporation or sputtering. Figure 10b shows the example of a superlattice composed of a stack of sub-stochiometric SiO_x and stochiometric SiO₂ nanolayers. The control of the bandgap energy can be obtained by controlling the size of the silicon nanocrystals (figure 9c), which is governed by the tickness of the nanolayers. The nanoparticle size is the key parameter to adjust the absorption of the total thin layer. The conduction properties depend on the matrix wherein the Si-ncs are integrated (the barrier height between the particules controls the probability of charges transfer between them). Reducing the distance between the silicon nanocrystals leads to the formation of mini-bands which results in an effective energy bandgap larger than that of bulk silicon. This perspective generate the concept of « all silicon » tandem cell suggesting the possibility of a significant increase in the open circuit voltage (larger bandgap) in comparison to a single junction. Theoretically, a configuration like that of figure 9a is expected to deliver a quantum efficiency of about 29%¹⁸.

¹⁸ Green M. A. (2003). Third Generation Photovoltaics: Advanced Solar Electricity Generation, Springer, Berlin.



Figure 10. (a) scheme of a "QDs based silicon" tandem cell; (b) TEM of alternated layers of pure followed by silicon-rich oxide, nitride or carbide. Upon thermal annealing, silicon precipitates out of the silicon-rich phase as spherical quantum-dots of diameter nominally equal to the layer thicknesss (figure at right).(Ref. 18)

Fabrication of solar cells involving silicon nanoparticles is quite complicated and it is still under development. While the optical effects were already observed, the transport properties are very limiting. However, first demonstrators employing a single junction¹⁹ have been produced and an example is shown in Figure 11. An open circuit voltage up to 350mV was obtained but the current is low. In such structures, the question of n-doped and p-doped nanoparticles was also introduced so as to allow the realization of pn or pin type cells.



Figure 11. First results from cells using quantum effects in silicon. The band gap of the top layer containg silicon nanoparticles is about 1.8 eV (Conibeer, Green et al, 2010).

¹⁹ Conibeer G., Green M., König D., Perez-Wurfl I., Huang S., Hao X., Di D., Shi L., Shrestha S., Puthen-Veetil B., SoY., Zhang B., Wan Z. (2010). Proceedings 25th European Photovoltaic Solar Energy Conference, Valencia, (2010).

3.2 Geometric-confinement effects for photovoltaics

3.2.1 - Principe

In the case of a conventional photovoltaic cell, the optical absorption depth usually has a thickness less than the active thickness for the separation and efficient collection of the electron-hole pairs. One alternative way to collect efficientyly the electron-hole pairs is to reduce the absorption volume into small domains of thicknesses lower that the carriers' diffusion length. These domains should have thicknesses from a few nanometers to a few tens of nanometers. The concept was applied to nanostucture the space between n-type (collection of electrons) and p-type (collection of holes) regions. This design, far from that of a conventional photovoltaic cell, is called "interpenetrated junctions" or "bulk heterojunction" (**Figure 12**). It has in fact been introduced in early 90ths for the photoelectrochemical dye cells²⁰, and later applied to organic cells. It can be also extended to inorganic materials based on nanowires or nanorods.



Figure 12. Schematic of an ideal interpenetrating junctions structures composed of donor and acceptor nanomaterials

3.2.2- Organic and hybrid nanostructured cells

Figures 13a gives a schema of a bulk heterojunction (BHJ) structure consisting of a donor material (polymer) and an acceptor material (fullerene, polymer, inorganic) finely mixed to obtain an active layer morphology. It presents randomized domains whose characteristic dimensions are close to the exciton diffusion length (10 to 20 nm). This morphology, by providing sufficient donor-acceptor interfaces, enhances the probability of exciton dissociation. Figure 13b shows a microscope image of a bulk heterojunction composed of P3HT and PCBM, the most studied organic mix. Current high-efficiency polymer solar cells (PSCs)²¹ with efficiencies above 9.0% are restricted to materials combinations that are based on limited donor polymers and only one specific fullerene acceptor, PC₇₁BM. Furthermore, best-efficiency PSCs are

²⁰ O'Regan B., Grätzel M. (1991). Nature **353** 737.

²¹ Liu, Y., Zhao, J., Li, Z., Mu, C., Ma, W., Hu, H., Jiang, K., Lin, H., Ade, H. and Yan, H., "Aggregation and morphology control enables multiple cases of high-efficiency polymer solar cells." Nat. Commun., 5, 5293, (2014).

mostly based on relatively thin (100 nm) active layers. Thick-film PSCs generally exhibit lower fill factors and efficiencies compared to the best thin-film PSCs. Recently were repoted multiple cases of high-performance thick-film (300 nm) PSCs based on conventional PCBM and many non-PCBM fullerenes with efficiencies up to 10.8%²². The simple aggregation control and materials design rules allowed to develop three new donor polymer, six fullerenes (including C60-based fullerenes), and over ten polymer:fullerene combinations, all of which yielded higher efficiency than previous state of art devices. A record certified efficiency of 11.7% was demonstrated (ref.22).



Figure 13. Structure of the active layer of an organic solar cell. (29a): Junction consists of an interpenetrating mixture of donor and acceptor materials (P3HT and PCBM in the figure). (11b) a schema of figure 13a for illustration purpose.

One of the major limitation of the interpenetrating network structure is it does not fulley support the transport of charges to the electrodes. The carriers must indeed make their way through the donor and acceptor materials to the electrodes. To improve the carriers transport, while maintaining optimal exciton separation, controlled nanostructure junction structure is necessary. Among the options, the donor material can be a vertical nanowires embedded in an acceptor material matrix. The nanowires should have a diameter in the order of the exciton diffusion length, typically 10 to 20 nm, and a length compatible with the carriers diffusion length to the electrodes (of the order of 100 nm). The nanowires are used to generate percolation paths to the electrodes. However, the fabrication of such structures is difficult and it is still a subject of intense research. It can be noted that the transport properties in these disordered systems can be improved by controlling the morphology of the mixture (domain size and degree of crystallinity). For instance, some groups developed methods of structuring the P3HT as fibrils that are obtained in solution prior to the spinning step²³. In these fibrils, the polymer chains are highly ordered and conversion efficiencies around 4-5% were reported while overcoming the conventional post-deposition annealing. Further work has concerned the replacement of the fullerene molecules by carbon nanotubes in order to facilitate the organization of the layers and improve the electron transport properties.

Another nanostructured type of cell is the hybrid dye solar cells composed of TiO_2 nanoparticles, decorated with dye molecules and embedded in a iodine based electrolyte. The transport properties in the porous oxide matrix are also an issue that

²² Jingbo Zhao, Yunke Li, Guofang Yang, Kui Jiang, Haoran Lin, Harald Ade, Wei Ma & He Yan, *Nature Energy* 1, Article number: 15027 (2016)

²³ Monestier F., Simon J.-J., Torchio Ph., Escoubas L., Flory F., Bailly S., De Bettignies R., Guillerez S., Defranoux C. (2007). Solar Energy Materials & Solar Cells <u>91</u>, 405 – 410.

should be improved. There may be mentioned in particular the embodiment of monocrystalline ZnO porous matrices obtained by the electrochemical deposition in the presence of structuring agents. In this case, there is no barrier at the grain boundaries as in the case of conventional structures obtained by sintering. Figure 14a shows a typical example of an ETA cell²⁴. It uses a network of nanocolumns as electron carriers ("electron highways'). Figure 14b shows a network of ZnO nanocolumns as produced by electrolysis. As an example of application, Optimized ZnO/CdSe/CuSCN solar cells exhibiting 3.2 % solar energy conversion efficiency were obtained by using 230 nm diameter ZnO NWs²⁵.



Figure 14. left: typical scheme of an ETA cell; Right: array of nanowires of ZnO single crystal electrochemically deposited on a layer of TCO (SnO2) (ref. 25).

3.2.2- Nanostructured cells based on silicon or III-V compounds

The use of nanostructures tends to extend also to conventional materials commonly employed in planar configurations. A trivial example concerns the cells based on nanostructured silicon. The first approach is to use arrays of SiNWs to improve light absorption in Si in both different ways²⁶,²⁷: while random multiple scattering in SiNW arrays enhances broadband light absorption, resonant optical phenomena such as Mie scattering at individual SiNWs constitute the optical response of the entire absorber, i.e. cause an overall wavelength-, angle- or polarization- selective photonic absorption enhancement²⁸. The second approach is targeting to fabricate silicon nanowires based solar cells. In this case, the orthogonal

²⁴ I. Kaiser, K. Ernst, Ch.-H. Fischer, R. KoK nenkamp, C. Rost, I. Sieber, M.Ch. Lux-Steiner; Solar Energy Materials & Solar Cells 67 (2001) pp.89-96

²⁵ C. Levy-Clement, J. Elias, Journal of Chemistry Physics and Physical Chemistry, V.14(10), 2013 pp. 2321–2330

²⁶ Tsakalakos, L.; Balch, J.; Fronheiser, J.; Korevaar, B. A.; Sulima, O.; Rand, J. Silicon Nanowire Solar Cells. *Appl. Phys. Lett.* **2007**, *91*, 233117

²⁷ Garnett, E. C.; Yang, P. Light Trapping in Silicon Nanowire Solar Cells. *Nano Lett.* **2010**, *10*, 1082–1087.

²⁸ Brönstrup, G.; Garwe, F.; Csáki, A.; Fritzsche, W.; Steinbrück, A.; Christiansen, S. Statistical Model on the Optical Properties of Silicon Nanowire Mats. *Phys. Rev. B* **2011**, *84*, 125432.

direction between the charge-carrier collection path and incident light enables the use of low-quality silicon in the production of solar cells. The material cost and amenability to eventual large scale fabrication determine the viability of novel concepts in solar cell design in addition to efficiency. The SiNWs can be grown on solids using the thoroughly studied vapor-liquid-solid (VLS) process or modifications of which, e.g. the vapor-solid-solid (VSS) growth. These processes are relying on catalytic growth of Si from metal-Si eutectics under continuous supply of the silicon growth species from the gas phase in chemical-vapor-deposition (CVD) reactors²⁹. Here, the size and arrangement of grown SiNW is dependent on the previous deposition of a metal catalyst mask on top of the solid substrate. However, even though these so called *bottom-up* processes produce SiNW of uniform shape and good crystalline structure, their electrical quality was found to be strongly influenced by the incorporation of metal catalyst atoms that cause inter band trap states in silicon³⁰.

Another method of fabricating hexagonally aligned arrays of silicon nanostructures is the cryogenic RIE in combination with polystyrene nanosphere (PSNS) lithography³¹. The process flow is shown in Figure 15. Nanowires and nanorods can be fabricated using this method.

It is also possible to fabricate symmetric or antisymmetric NWs using top-down lithography with a dry etching process and silicon oxyde as a mask³². Figure 16a illustrates the asymmetric silicon nanowire (SiNW) solar cell structure. A maximum short circuit current density of 27.5 mA/cm² and an efficiency of 7.53% were realized without anti-reflection coating. Changing the silicon nanowire (SiNW) structure from conventional symmetric to asymmetric nature improves the efficiency due to increased short circuit current density. This is well demonstrated in figure 14b. The proposed asymmetric structure has great potential to effectively improve the efficiency of the SiNW solar cells.

²⁹ Wagner, R. S.; Ellis, W. C. Vapor-Liquid-Solid Mechanism of Single Crystal Growth. *Appl. Phys. Lett.* **1964**, *4*, 89.

³⁰ Hannon, J. B.; Kodambaka, S.; Ross, F. M.; Tromp, R. M. The Influence of the Surface Migration of Gold on the Growth of Silicon Nanowires. *Nature* **2006**, *440*, 69–71.

³¹ Cheung, C. L.; Nikolic, R. J.; Reinhardt, C. E.; Wang, T. F. Fabrication of Nanopillars by Nanosphere Lithography. *Nanotechnology* **2006**, *17*, 1339–1343.

³² Ko, M.-D. *et al.* High efficiency silicon solar cell based on asymmetric nanowire. *Sci. Rep.* 5, 11646; doi: 10.1038/srep11646 (2015).



Figure 15 : Process steps for the fabrication of SiNWs by PSNS lithography and RIE : On a clean silicon surface a PSNS colloidal crystal was deposited via a Langmuir Blodgett (LB) through (a, b). Subsequently, the sample was transferred to the RIE reactor and the PSNS were reduced to the desired size via oxidation in an O₂ plasma (c). With the PSNS of remaining size masking the surface, the poly-Si sample were exposed to a cryogenic RIE plasma to selectively etch the non-covered surface area (d. Cleaning was performed with a dip in an ultra-sonic bath to remove the spheres and rinsing in ethyl acetate/isopropanol to dissolve the thermal coupling oil (Fomblin) for the cryogenic RIE process. The processing was finalized by a thermal annealing in O₂ ($500^{\circ}C / 30$ min) and a dip in hydrofluoric acid to remove remaining surface damage and roughness from the as-etched structures (e). SEM image of SiNWs produced by the PSNS lithography method and RIE on silicon on glass. With the courtesy of Y. Schmitt.

Beyond photovoltaics, the individual and interconnected silicon nanowire elements can serve as robust power sources to drive functional nanoelectronic sensors and logic gates. These coaxial silicon nanowires provide a new nanoscale test bed for studies of photoinduced energy/charge transport and artificial photosynthesis, and might find general usage as elements for powering ultralow-power electronics and diverse nanosystems.



Figure 16: (a) Schematic illustration of the asymmetric silicon nanowire (SiNW) solar cell consisting of an array of radial p-n junction asymmetric SiNWs, back surface field (BSF) layer, Al back reflector and Ag top electrode. ; (b) The distribution of the efficiency of four types of SiNW solar cells. D_B is the core diameter at the bottom of the nanowire. Each distribution comes from the J-V characteristics of the solar cells fabricated with identical design and process parameters.

Recently, highly efficient hybrid solar cells based on silicon nanowires (SiNWs) and poly(3,4-ethylenedioxythiophene):poly(styrenesulfonate) (PEDOT:PSS) using transfer-imprinted metal mesh frnt electrodes were reported³³. Compared with hybrid cells formed using indium tin oxide (ITO) electrodes, they find an increase in power conversion efficiency from 5.95% to 13.2%, which is attributed to improvements in both the electrical and optical properties of the Au mesh electrode. The proposed fabrication strategy for metal mesh electrode is suitable for the large-scale fabrication of flexible transparent electrodes, paving the way towards low-cost, high-efficiency, flexible solar cells.

This organic-inorganic approach was also applied to the III-V GaAs-based semiconductor. For instance, hybrid solar cells based on poly(3-hexylthiophene)-coated GaAs nanopillars grown on a patterned GaAs substrate using selective-area metal organic chemical vapor déposition were reported.³⁴

3.3- Optical Management based on nanostructures

3.3.1 Plasmonics for photovoltaics

The study of plasmonics in solar cells has developed over the past ten years. The issue was to increase light absorption in materials used for solar cells. Coupling of light energy in the form of plasmons is thus of high interest in thin-film solar cells, when the price of the material can be a hindrance to the development of the technology.

In physics, a plasmon is a quantum of plasma oscillation arisen from the free electron "gas" in conducting material³⁵. There are three types of plasmons, bulk plasmon (located in a bulk conducting material), surface plasmon (located at planar metal-dielectric interfaces) and localized surface plasmon (highly confined on a small surface, e.g., on the surface of a nanoparticle). In an infinite solid (e.g., metal), the **bulk plasmons** are longitudinal waves³⁶. According to the Drude model³⁷, if the frequency of an incident electromagnetic wave is higher than bulk plasmon frequency, the metal becomes transparent for this wave. The bulk plasmon frequencies of most metals are in the ultraviolet region, associating energies within 5-15 eV. But for some metals, e.g., noble metals (copper, silver and gold), they have filled d-shells. These d-electron bands have lower energies than the Fermi level of the conduction band and thus allowing new narrow transitions close to or in visible region³⁸. Due to the fact that light waves are transverse, the wave vector mismatch makes the bulk plasmons incapable to be excited from direct irradiation, thus strongly limiting their utilization in the photovoltaic domain. Surface plasmons are plasmons at the metal-dielectric interfaces and tightly bound to the interfaces. Since their wavevector is larger than light waves, the surface plasmon cannot be directly excited by illumination. Instead, the surface plasmon wave can be generated under special

³³ Park, K.-T. *et al.* 13.2% efficiency Si nanowire/PEDOT:PSS hybrid solar cell using a transfer-imprinted Au mesh electrode. *Sci. Rep.* 5, 12093; doi: 10.1038/srep12093 (2015).

³⁴ Giacomo Mariani et al., Hybrid conjugated polymer solar cells using patterned GaAs nanopillars, Appl. Phys. Lett. **97**, 013107 (2010)

³⁵ D. Pines, "Collective energy losses in solids," Rev. Mod. Phys. 28, 184–199 (1956)

³⁶ R. A. Ferrell, "Predicted Radiation of Plasma Oscillations in Metal Films," Phys. Rev. **111**, 1214 (1958).

³⁷ L. Tonks and I. Langmuir, "Oscillations in ionized gases," Phys. Rev. **33**, 195 (1929)

³⁸ F. Wooten, « Optical properties of solids » (Academic Press, 1972).

geometries (Kretschmann coupler³⁹, gratings⁴⁰). The surface plasmon propagates along the dielectric/metal interface and it has a highly concentrated electric field near the interface. This electric field exponentially decays away from the surface. The decay rate is much higher in metal than in dielectric. **Localized surface plasmons** excited in metallic nanoparticles are non-propagating plasmon excitations. Since the size of a metallic nanoparticle is on the same scale of the penetration depth of electromagnetic waves in metals (e.g., $20 \sim 30$ nm for Ag and Au), the external field can penetrate the whole particles and shift the conduction electrons with respect to the rigid ion lattice. Thus, the charges are separated and this charge separation results in a restoring force and then an oscillation. The oscillation frequency is mainly related to effective electron mass, charge density and geometry of the particle, as well as the properties of the surrounding medium. The amplitude of the induced electromagnetic field is much stronger than exciting fields (over 10 times). The properties of localized surface plasmons in noble metallic nanoparticles can be tuned by the composition, size, shape and their environment⁴¹ ⁴² ⁴³.

A comprehensive review on the applications of plasmon effect to solar cells has been published by Atwater and Polman^{44 45}. More specific work concerns the wafer or thin film silicon solar cell types with a strong increase of the optical absorption for wavelengths close to the gap⁴⁶. In organic solar cell, an improvement of the conversion efficiency was obtained with gold nanoparticles in the active layer and silver electrodes including a diffraction grating^{47 48}.

For solar cells, particularly in thin film form, the coupling of **surface plasmons** can therefore allow a trapping, at the interface between the semiconductor material and the back electrode, of a part of the radiation that is not or little absorbed within the semiconductor, for example, the radiation part which is close to the gap of the semiconductor. To do so the back electrode of the cell is periodically structured in the form of mono- or two-dimensional grating. This grating allows both to excite surface plasmon in a specific spectral range but it can also play the role of a rear mirror and have an influence on the coupling of Fabry-Perot type modes in a wider spectral range, in the overall structure of the solar cell^{49 50} (fig. 17).

³⁹ E. Kretschmann, "The Determination of the Optical Constants of Metals by Excitation of Surface Plasmons," Zeitschrift für Phys. **241**, 313–324 (1971)

⁴⁰ E. Devaux, T. W. Ebbesen, J.-C. Weber, and "A. D., "Launching and decoupling surface plasmons via microgratings," Appl. Phys. Lett. **83**, 4936 (2003).

⁴¹ E. Stefan Kooij and B. Poelsema, "Shape and size effects in the optical properties of metallic nanorods.," Phys. Chem. Chem. Phys. **8**, 3349–57 (2006).

⁴² C. Noguez, "Surface Plasmons on Metal Nanoparticles: The Influence of Shape and Physical Environment," J. Phys. Chem. C **111**, 3806–3819 (2007).

⁴³ K.-S. Lee and M. A. El-Sayed, "Dependence of the enhanced optical scattering efficiency relative to that of absorption for gold metal nanorods on aspect ratio, size, end-cap shape, and medium refractive index.," J. Phys. Chem. B **109**, 20331–8 (2005).

⁴⁴ H. Atwater, A. Polman, « Plasmonic for improved photovoltaic devices », Nature Materials **9** (3), 205-213 (2010)

⁴⁵ K. Catchpole, A. Polman, « Plasmonic solar cells », Optics Express **16** (26), 21793 – 21800 (2008)

⁴⁶ S ; Pillai, K. Catchpole, T. Trupke et al., « Surface plasmon enhanced solar cells », Journal of Applied physics **101** (9), 093105 (2007)

⁴⁷ X. Li, W. Choy, L. Huo et al., « Dual plasmonic nanostructures for high performance inverted organic solar cells », Adv. Mat. **24** (22), 3046-3052 (2012)

⁴⁸ D. Duche et al., "Improving light absorption in organic solar cells by plasmonic contribution", Solar Energy Materials & Solar Cells **93**, 1377–1382 (2009)

⁴⁹ B. Behaghel et al., « Absorption enhancement through Fabry-Perot resonant modes in a 430nm thick InGaAs/GaAsP multiple quantum wells solar cell », APL **106**, 081107 (2015)

⁵⁰ I. Massiot, N. Vandamme et al., « Metal Nanogrid for Broadband Multiresonant Light-Harvesting in Ultrathin GaAs Layers », ACS Photonics **1** (9), 878–884 (2014)



Figure 17: Reproduced from ref 34 (I. Massiot et al.) Measured (a) and calculated (b) total absorption spectra at normal incidence of the SiNx (Au)/GaAs/Au structure. Inset of (a) SEM image of the 2D gold nanogrid. Inset of (b) : schematic of the structure modeled with a rectangular surface profile.

The excitation of these resonance modes in surface or in volume allows exalting the light absorption within the semiconductor material, and thus increases the photocurrent.

If we now look at the **localized plasmons**, the principle of increasing photon absorption is totally different. Indeed, the incident light radiation is partially scattered by the metal nanoparticles and partly absorbed in the metal. In the latter case, light energy is lost through thermalization and does not contribute to increase absorption in the semiconductor material. The light radiation which is scattered by metal nanoparticles is usually more efficiently absorbed within the semiconductor material just because the optical path in the material is increased⁵¹. Depending on the density of metal nanoparticles and thus the distance between the nanoparticles, an electromagnetic coupling may occur between the latter. Under certain conditions of size and density of the metal nanoparticles, this coupling can lead to an exalted light scattering, enabling to further increase the absorption of radiation in the semiconductor material. The shape of the metal nanoparticles, which may be spherical, prismatic, cylindrical or in tetrapod forms, has an influence on the excitation wavelength of the plasmon^{52 53 54} (fig. 18). Compared to a sphere, in nanoparticles with sharper curvatures area along the polarization direction of light there is weaker restoring strength leading to a lower plasmon resonance frequency (redshift) 55.

⁵¹ D. Derkacs, S. H. Lim, P. Matheu et al., « Improved performance of amorphous silicon solar cells via scattering from surface plasmon polaritons in nearby mettalic nanoparticles », APL **89** (9), 093103 (2006)

⁵² K. L. Kelly, E. Coronado, L. L. Zhao, and G. C. Schatz, "The Optical Properties of Metal Nanoparticles: The Influence of Size, Shape, and Dielectric Environment," J. Phys. Chem. B **107**, 668–677 (2003).

⁵³ X. Zhang, E. M. Hicks, J. Zhao, G. C. Schatz, and R. P. Van Duyne, "Electrochemical Tuning of Silver Nanoparticles Fabricated by Nanosphere Lithography," Nano Lett. **5**, 1503–1507 (2005).

⁵⁴ Z. Cao, Z. Chen, L. Escoubas, « Optical, structural and electrical properties of PEDOT :PSS thin films doped with silver nanoprisms », Opt. Mat. Express **4** (12), 2525-2534 (2014)

⁵⁵ P. K. Jain, "Plasmons in assembled metal nanostructures: radiative and nonradiative properties, near-field coupling and its universal scaling behavior," Dissertation presented at The Academic Faculty of Georgia Institute of Technology (2008)



Figure 18 : Reproduced from ref 38 (Z. Cao et al.) Normalized solution UV-Vis spectra of Ag nanoprisms synthesized from different volumes of a seed solution: black - seeds only; red - 2 mL; blue - 1mL; dark cyan - 650 μ L; pink - 500 μ L. The use of different amount of seed solution leads to different batches of Ag nanoprisms of different edge-length/thickness aspect ratios. Inset: TEM image of flat-lying Ag nanoprisms which show an in-plane localized surface plasmon resonance band at l=560 nm.

3.3.2- Up / Down photon Conversion for photovoltaics

The main limitations of photovoltaic efficiency of a single junction solar cell are the thermalization of the high energy photons and the non absorption of photons having energies lower than the bandgap energy of the absorbing semiconductor. Thus, potentially, changes in the incident solar spectrum may allow a standard cell to achieve a very high efficiency, provided to convert the wavelength of the incident photons to that corresponding to the bandgap of the semiconductor. The envisaged concepts to overcome partially these limitations are illustrated in Figure 19.

Basically, the challenge is to fabricate a component that absorbs a photon of energy at least twice the energy gap of the used semiconductor and to generate two incident photons in the cell (downconverter or DC (Figure 19.a), sometimes referred to as quantum-cutting). Down-shifting or DS (Figure 19.b) is also an optical process similar to DC with the exception that one high-energy photon (blue) is absorbed and converted into a single lower-energy photon (red). In this case, the only effect is to 'shift' high-energy photons into a more efficient region of the solar cell's spectral response which is typically at a lower energy. The other challenge for up-conversion or UC (Figure 19.c) is to find a component that absorbs at least two photons having less energy than the bandgap and that emit a photon of energy above the bandgap. For all processes, the resulting photons have energy just above the bandgap of the active semiconductor, which corresponds to the most efficient energy absorption.

From a purely theoretical side, a maximum efficiency of 38.6% is achievable for a down-converter with bandgap of 2.2 eV located on the front surface of a solar cell with bandgap of 1.1 eV (corresponds to that of silicon). On the other hand, detailed balance calculations show that the efficiency of silicon solar cells can increase to 40.2% when coupled to an UC layer. From the processing point of view, no modification of the active part of the cell is necessary in contrast to the case for multijunction cells mentioned above. However, few experimental developments exist involving nanomaterials.



Figure 19. Principle of spectrum modification by (a) down-convesion, (b) down-shifting and (c) upconversion to increase the performance of a solar cell.

Different materials have been investigated in the literature for down-shifting layers such as organic dyes, quantum dots and rare earth ions. Organic dyes exhibit the highest optical conversion efficiency but lack stability and have an overlap between the absorption and emission peaks. Rare earth doped glass fluorescent materials have lower efficiencies but higher stability and larger stokes shift. Quantum dots have the distinct advantage of a tunable absorption profile by changing their size. Thus, the use of a layer containing CdS QDs (emitting at 610 nm) covering the surface of a polycrystalline silicon cell⁵⁶ showed an ~ 28% increase in the circuit current and a relative enhancement of the yield ~ 6% in the case of a diffuse AM1.5 radiation. Other works have investigated the effects of the down shifting layers and to demonstrate a gain in conversion efficency. Svrcek et al.⁵⁷ have demonstrated a gain in the UV region of the spectral response of a conventional P-N junction silicon solar cell using Si nanoparticles dispersed in a sol-gel solution. Pi⁵⁸ et al. have also been able to improve the absorption in the blue using a film rich in silicon nanoparticles deposited by spin coating on a crystalline silicon solar cell. The overall cell efficiency increased from 16.9 to 17.5%. More recently, silicon oxide layers containing silicon nanoparticles deposited by PECVD have been applied on back contact silicon solar cells⁵⁹. An enhancement in the blue region of the internal quantum efficiency was measured which has been translated in an improvement in efficiency for the best cell from 19.4 to 20.3%. Other approaches use oxides matrices (ZnO, SnOx, CeO2...) doped with rare earths such as Nd, Yb or Pr⁶⁰. Thanks to their thickness of less than

⁵⁶ Van Stark W.G.J.H. (2005). Appl. Phys. Lett. **<u>87</u>**, 151117

⁵⁷ Švrček V., Slaoui A., Muller J.-C. (2004). J. Appl. Phys. <u>95</u>, 3158

⁵⁸ Pi, X.*et al.* Air-stable full-visible-spectrum emission from silicon nanocrystals synthesized by an all-gas-phase plasma approach. *Nanotechnology* 19, 245603 (2008).

⁵⁹ Delachat F., Slaoui A. (2009). Nanotechnology **20**, 415608; F. Sgrignuoli, P. Ingenhoven, G. Pucker, V.D. Mihailetchi, E. Froner, Y. Jestin, E. Moser, G. Sànchez, L. Pavesi ; Solar Energy Materials & Solar Cells 132 (2015) 267–274

⁶⁰ P M. Balestrieri, M. Gallart, M. Ziegler, P. Bazylewski, G. Ferblantier, G. Schmerber, G.S. Chang, P. Gilliot, D. Muller, A. Slaoui, S. Colis, A. Dinia, J. Phys. Chem. C 118 (25), 13775-13780 (2014)

100nm and the appropriate refractive index, such layers are also playing the role of an antireflection coating for the solar cell. Though still very few breaktroughs are reported, the results for their use as DS converters are encouraging.

Concerning the up-conversion process, since the UC layers absorb photons that could have not be absorbed by the solar cell⁶¹, those layers are coupled to the back surface of bifacial solar cells, and their presence cannot decrease the efficiency of the cell as in the case of poor choice of the down-shifting layers. Trivalent erbium, Er^{3+} , is ideally suited for UC of near-infrared light due to its ladder of nearly equally spaced energy levels that are multiples of the $4I_{15/2}$ to $4I_{13/2}$ transition. The Er^{3+} ion is well suited for silicon solar cells as the absorption spectrum is centered at 1520 nm while the emission peaks occur at 980 nm, 880 nm and 650 nm. Another common choice for UC are coupled Er^{3+}/Yb^{3+} ions, which are suitable for larger bandgap devices such as CdTe or amorphous silicon solar cells since the absorption profile of Yb³⁺ is centered at 980 nm. Such elements are usually inserted in less than 100nm thick oxide (ZnO, TiO₂, ...) or fluroide (NaYF4) matrices⁶². One can mention the results obtained with erbium and ytterbium doped NaYF4, which demonstrate a conversion efficiency of 17% of the infrared radiation at 1.57 micron to 1 a micron radiation⁶³.

Note that the process of up-convertion is implementing several basic steps of absorption. Its effectiveness increases with the light concentration. One must therefore consider working in high concentration. Another possibility is the coupling of UC process with nanophotonics, especially plasmonic resonances modes since the increasing of the local field will enhance the probability of the UC process. Thus, application of plasmonic structure for up-conversion enhancement has been realized using Er^{3+} doped YF3 as up- converter layer⁶⁴. In spite of the low thickness of films (<100nm) and the low concentration of doping ions, up-conversion luminescence visible with naked eye has been obtained under 1540 nm wavelength excitation. Luminescence cartography showed that up-conversion enhancement up to a factor 5 has been obtained. Comparison between the structured layers and layers of the same thickness and the same Er^{3+} concentration shows that the enhancement factor of up-conversion luminescence can reach up to 35 by using plasmonic structures.

3.3.3- Multiple exciton generation Solar cells

The principle of multiple exciton generation (MEG) is to exploit the surplus energy of hot carriers to create one or more additional electron-hole pairs. A photon with very high energy will create a first high energy exciton, which can subsequently decay

⁶¹ Strümpel C., McCann M., Beaucarne G., Arkhipov V., Slaoui A., Švrček V., Del Cañizo C. and Tobias I. (2007). Solar Energy Materials and Solar Cells <u>91</u> (4) 238-249

⁶² Petersen J., Brimont C., Gallart M., Crégut O., Schmerber G., Gilliot P., Hönerlage B., Ulhaq-Bouillet C., Rehspringer J. L., Leuvrey C., Colis S., Slaoui A. and Dinia A. (2008). J. Appl. Phys. <u>104</u> 113539

⁶³ Ivanova S., Pellé F., Esteban R., Laroche M., Greffet J.J., Colin S., Pelouard J.L., Guillemoles J.F. (2008). Proc. 23rd European Photovoltaic Solar Energy Conference, Valencia, (2008).

⁶⁴ C. Andriamiadamanana A. Ferrier, L. Lombez, A-L. Joudrier, N. Naghavi, P. Ghenuche, N. Bardou, J-L. Pelouard, S. Collin, F. Pellé and J-F. Guillemoles, Physics, Simulation, and Photonic Engineering of Photovoltaic Devices, Edited by Alexandre Freundlich, Jean-Francois F. Guillemoles, Proc. of SPIE Vol. 8256, 825608 (2012.

into as many excitons as allowed by energy conservation, by an impact ionization phenomenon. This is schematically illustrated in Figure 20 taken from the review done by A. Nozik⁶⁵. This process could help to increase the theoretical maximum efficiency for a single junction from 30% to about 40% or more depending on the threshold energy and on the processes considered^{66,67,68,69}. Such phenomena have been observed in the context of solar energy conversion in bot bulk and QD materials, and it is still debated whether the processes are more efficient in QD than in bulk materials^{70,71}.

This multiple exciton generation phenomenon was observed experimentally after excitation by a single photon of InAs, CdS, PbS or PbSe nanoparticles (NPs) with the generation of up to 8 excitons (ref. A. Nozik). More recently, investigations of MEG in close-packed Si QDs embedded in a silica matrix have yielded some interesting results. The first demonstration of MEG in Si utilized colloidal QDs with a diameter of 9.5 nm (Figure 21) and Eg= 1.2 eV, and reported hv (threshold)= 2.4 Eg and a quantum yield of 2.6 for a pump photon energy of 3.4 Eg (Figure 21)⁷².



Figure 20. Left : relaxation process of hot carriers in a conventional semiconductor. Excess photon energy is transferred as heat. Right: generation process of two excitons by absorption of a photon in a quantum dot (e.g. PbSe or PbS) thanks to the phenomenon of ionization by impact. The generation of a second exciton is faster than the conventional thermal transfer by interaction with phonons due to the quantization of energy levels. (ref.65).

By using photoluminescence quantum yield measurements and also subsequently ultrafast transient absorption spectroscopy, the step-like enhancement

⁶⁵ Nozik A. J. (2008). Chemical Physics Letters <u>457</u> 3.

⁶⁶ R. Brendel, J. Werner, H. Queisser, Solar Energy Materials and Solar Cells 41/42 (1996) 419.

⁶⁷ P. Landsberg, H. Nussbaumer, G. Willeke, J. Appl. Phys. 74 (1993) 1451;

⁶⁸ W. Spirkl, H. Ries, Phys. Rev. B 52 (1995) 11319;

⁶⁹ De Vos, Alexis, et Bart Desoete. « On the ideal performance of solar cells with larger-than-unity quantum efficiency ». Solar energy materials and solar cells 51, nº 3 (1998): 413-24]

⁷⁰ Geiregat, Pieter, Christophe Delerue, Yolanda Justo, Michiel Aerts, Frank Spoor, Dries Van Thourhout, Laurens D. A. Siebbeles, Guy Allan, Arjan J. Houtepen, et Zeger Hens. « A Phonon Scattering Bottleneck for Carrier Cooling in Lead Chalcogenide Nanocrystals ». ACS Nano 9, nº 1 (27 janvier 2015): 778-88. doi:10.1021/nn5062723

⁷¹ Charles Smith and David Binks, Multiple Exciton Generation in Colloidal Nanocrystals, Nanomaterials 2014, 4, 19-45

⁷² M. C. Beard , K. P. Knutsen , P. Yu , J. M. Luther , Q. Song , W. K. Metzger, R. J. Ellingson , and A. J. Nozik, "MEG in colloidal silicon nanocrystals" , *Nano Letters* 7: 2506. 2007].

in the quantum yield of photo-generated charges that has been predicted for very high efficiency MEG has been reported⁷³.



Figure 20: (left) a TEM image of a silicon nanocrystal; (right) Quantum efficiency of silicon nanocrystals of different sizes and compared to that of a bulk silicon (ref. 72).

The most convincing result, in terms of application, presented in this line of research was published⁷⁴ in 2011, where a solar cell (with efficiency 5%) made from a stack of ZnO/PbSe QD/ Au presented quantum efficiencies in excess of 100% figure 22). These results are very spectacular, but surprisingly did not lead to further development since.



Figure 21: Red curves are the measured EQE for QD solar cells made from PbSe QDs with associated bandgaps of 0.72 eV. The estimate of the uncertainty of these EQE curves was ~3% The IQE is determined either by EQE/[1 - R] (purple curves) or by EQE/A (blue curves), where A is the modeled absorptance of the PbSe plus ZnO layers. The modeled reflectance is shown as the dashed black line. (ref. 74)

⁷³ Trinh, M.T.; Limpens, R.; de Boer, W.D.A.M.; Schins, J.M.; Siebbeles, L.D.A.; Gregorkiewicz, T. Direct generation of multiple excitons in adjacent silicon nanocrystals revealed by induced absorption. Nat. Photonics 2012, 6, 316–321

⁷⁴ Semonin, Octavi E., Joseph M. Luther, Sukgeun Choi, Hsiang-Yu Chen, Jianbo Gao, Arthur J. Nozik, et Matthew C. Beard. « Peak external photocurrent quantum efficiency exceeding 100% via MEG in a quantum dot solar cell ». Science 334, nº 6062 (2011): 1530-33

3.3.4- Hot carriers solar cells

When the dominant process for excess energy relaxation is not through impact ionization, there is still a possibility to use the excess heat generated provided a hot electron-hole gas can be formed, decoupled from the lattice temperature at least for times comparable to the carrier extraction times. This is the hot carrier solar cell concept, reported in a recent review⁷⁵.

The system operates as an ideal thermoelectric element, in series with a solar cell operating at high temperature. It could reach the higher efficiencies compatible with thermodynamics (86% under full concentration)⁷⁶, if the an absorber can be found with very slow electron to lattice coupling and selective energy contacts can be made as illustrated in figure 22. The former has been demonstrated under pulsed⁷⁷ or CW⁷⁸ laser excitation in MQW heterostructures from III-W materials, where carrier confinement was shown to help reduce the thermalization, and that it could even be made to saturate. Using experimentally measured thermalization rates, efficiencies as high as 50% are in principle within reach under very high concentration^{79,80}. In such systems, the excess carrier energy yields a higher voltage, collected by the selective energy contacts, than what is available in a solar cell of the same gap with thermalized carriers. This voltage could exceed the band gap, clearly passing the Schockley-Queisser limit. The voltage gain is roughly proportional to the excess carrier temperature, and therefore reminiscent of a Seebeck effect^{81,82}.

A first confirmation that the conversion of hot carriers energy is possible yielding Quasi Fermi level above the band gap (measured by calibrated photoluminescence) was presented recently⁸³, as illustrated figure 23.

The main challenge remaining today is that if energy selective contacts to really harvest the energy. Again nanotechnology, and in particular silicon nanocrystals, can contribute to this concept via the discretization of the energy levels due to quantum confinement. Thus, Si-QDs can be envisaged as "selective contacts"⁸⁴.

⁷⁵ Conibeer, Gavin, Jean-François Guillemoles, Feng Yu, and Hugo Levard. "CHAPTER 12 Hot Carrier Solar Cells." In Advanced Concepts in Photovoltaics, RSC pub. 2014.

 ⁷⁶ Ross R.T., Nozik A.J. (1982). J. Appl. Phys. <u>53</u> 3813; Würfel P. (1997). Sol. Energy Mat. and Sol. Cells. <u>46</u> 43
⁷⁷ Rosenwarks 1993

⁷⁸ A. Le Bris, L. Lombez, S. Laribi, G. Boissier, P. Christol and J-F. Guillemoles,"Thermalization rate study of GaSb-based heterostructures by continuous wave photoluminescence and their potential as hot carrier solar cell absorbers", Energy Environ. Sci., 2012, DOI: 10.1039/C2EE02843C

⁷⁹ Conibeer, G., D. Konig, M.A. Green, and J.F. Guillemoles "Slowing of carrier cooling in hot carrier solar cells". *Thin Sol. Films*, **516** (2008), 6948-6953;

⁸⁰ Arthur Le Bris, Jean Rodière, Clément Colin, Stéphane Collin, Jean-Luc Pelouard, Ruben Esteban, Marine Laroche, Jean-Jacques Greffet, and Jean-François Guillemoles, "Hot carrier solar cells: controlling thermalization in ultra thin devices" JOURNAL OF PHOTOVOLTAICS, 2 (2012), 506 - 511, doi: 10.1109/JPHOTOV.2012.2207376

⁸¹ S. Kettemann and J.F. Guillemoles, "*Thermoelectric enhancement of efficiency of solar cells*", J. Phys. E, *14*, **2002**, p. 101-106;

⁸² François Gibelli, Laurent Lombez, Jean Rodière and Jean-François Guillemoles, Optical imaging of lightinduced thermopower in semiconductors, Accepted by Phys Rev Applied, 2016

⁸³ Rodière, Jean, Laurent Lombez, Alain Le Corre, Olivier Durand, et Jean-François Guillemoles. « Experimental evidence of hot carriers solar cell operation in multi-quantum wells heterostructures ». Applied Physics Letters 106, no 18 (4 mai 2015): 183901. doi:10.1063/1.4919901

⁸⁴ Conibeer G., Ekins-Daukes N., Guillemoles J.F., Kőnig D., Cho E. C., Jiang C. W., Shrestha S., Green M. (2009). Progress on hot carrier cells. Solar Energy Materials and Solar Cells <u>93</u> 713



Figure 22. Energy band diagram of a hot carrier cell. Solar photons are absorbed in the central portion (absorber) where is formed a hot electron gas (at a different temperature TH of the network). Their additional kinetic energy allows them to be trapped at a higher energy than the conduction band edge (for electrons) and valence band edge (for holes). The energy selectivity of the contact is the condition for hot carrier distribution to not heat the contacts, which must remain at the lowest possible temperature.



Figure 23: Quasi Fermi level splitting in the quantum wells energy region (green squares) and in the barrier energy region (blue triangles). Dotted lines are guides for the eyes. Inset: Equivalent circuit of the hot carrier device with hot quantum wells (green diode) in parallel with cold barrier (blue diode) and equivalent Seebeck device (rectangle with color gradient). (ref. 83)

4. Conclusion and Outlook

This chapter has given a overview, far to be exhaustive, on past and present nanomaterials and nanostructures for photovoltaic applications. We have first shown that the research on materials and devices for photovoltaics has moved from the first generation based on thick silicon wafers to the second generation employing micrometer thin semiconductors. It is nowadays more and more open to concepts involving the use of structures at the nanoscale. This is the case in particular for organic cells and hybrid celles (inclusing dyes and perovskites) where few hundred nanometers of absorbing material are enough for an efficient photon conversion. We have shown through examples that some nanostructured materials can bring the singularities of their optical properties, particularly the effects of quantum confinement that will allow tuning the bandgap heights and generate new materials without changing their chemical composition. Such properties are of interest to develop efficient multijunction using basically the quantum size effects. This can be the case for silicon quantum dots for instance.

On the other hand, we have shown that nanostructuring of materials can also be used for optical management of photovoltaic devices to allow the containment of electromagnetic radiation in ultrathin areas. This is the case when using advanced concepts such as plasmonics and photonics.

Finally, nanostructures can play a major role in the development of new photovoltaic devices with very high efficiency based on some advanced concepts. Such approaches involve the conversion of photons, generating multiple excitons, and direct collection of electron-hole pairs before thermalization in "hot carrier" cells. Yet, despite of the significant potential advantages of these nanostructures for both increasing the conversion efficiency and reducing manufacturing costs, the technology is not yet mature, and the efficiency of proof-of-principle devices lies well below those of established technologies. Furthermore, it is not always possible to independently verify the reported efficiency of such devices.

In conclusion, next-generation solar cells must meet stringent requirements in terms of conversion efficiency, cost and long-term stability (10–15 years). The nanostructured solar cells have the potential to achieve these objectives.