### Your abstract submission has been received

Print this page

You have submitted the following abstract to GSA 2020 Connects Online. Receipt of this notice does not guarantee that your submission was complete or free of errors.

### PHONOLITES, S-SOLUBILITY, AND SODALITE-HAÜYNE STABILITY

**CARROLL, Michael**<sup>1</sup>, SCAILLET, Bruno<sup>2</sup> and ANDUJAR, Juan<sup>2</sup>, (1)Geology Division, Camerino University, via Gentile III da Varano, Camerino, 62032, Italy, (2)CNRS/INSU-Université d'Orléans, ISTO, UMR 6113, 1a rue de la Férollerie, Orléans, 45071, France

Sulfur solubility in magmas is known to strongly depend on such factors as melt composition, magma oxidation state ( $fO_2$ ), P, and T. Unlike the more common magmatic volatiles ( $H_2O$ ,  $CO_2$ ), sulfur can occur in a variety of forms, mainly controlled by magma oxidation state; the most important S-species in natural systems appear to be  $H_2S$ ,  $S_2$ , and  $SO_2$  as gas/fluid phase components, and  $S^{2-}$ ,  $HS^-$ , or  $SO_4^{2-}$ , as species dissolved in silicate melts. As a result of this complexity (and possible S-rich crystals), it is not always easy to interpret the significance of observed variations in S-emissions in active volcanic systems in terms of what is happening at depth. Furthermore, estimation of S-yields to the atmosphere in older volcanic eruptions requires accurate knowledge of not only how much S was in the pre-eruptive melt (e.g., via melt inclusions), but also an estimate of the mass and fugacities of sulfur species in any pre-eruptive magmatic fluid phase.

We have conducted experiments on S-solubility in a sodic phonolitic melt composition (with, in wt% ~60 SiO<sub>2</sub>, 18.8 Al<sub>2</sub>O<sub>3</sub>, 10 Na<sub>2</sub>O, 5.5 K<sub>2</sub>O), at T= 930°C, P = 50-390 MPa, using mainly IHPV with controlled H2 fugacity and S fugacity roughly controlled by using samples with different amounts of S added as elemental S. Oxidation states range from approximately -3 to +3 log fO2 units relative to the NNO buffer. Depending on the amount of S added, the experiments range from nearly water saturated (low S added) to significantly H<sub>2</sub>O undersaturated (P(H<sub>2</sub>O) down to ~0.5\*P<sub>total</sub>), with significant fugacities of H<sub>2</sub>S or SO<sub>2</sub> in the fluid phase, depending on experimental oxidation state. Sulfur solubilities in these alkali-rich phonolites are generally higher than those for dacitic to rhyolitic melts reported in other studies at similar fO<sub>2</sub> and fS<sub>2</sub> conditions. While previous studies have reported anhydrite stable in oxidized, S-rich andesitic to rhyolitic melts, in the studied phonolite we observe the stabilization of S-rich Sodalite (Hauyne) with up to  $\sim$ 12 wt% SO<sub>3</sub> at pressures up to 150 MPa, whereas at higher pressures the phonolitic melts coexist with an oxygen-rich Fe-S-O immiscible liquid, but no anhydrite was observed. Additional work will help understand if Sodalite-Hauvne S (and CI) contents in natural samples may be useful for estimating pre-eruptive S and CI activities in magmas.

# **Abstract ID#:** 358075

**Password:** 434175

Meeting: GSA 2020 Connects Online

Session Type:

### **Topical Sessions**

### **Primary Selection:**

T36. Experimental and Petrologic Investigation of Halogens, Sulfur, and Other Volatile Species in Igneous Systems: In Honor of Jim Webster

### **Final Session Number:**

### Abstract Title: PHONOLITES, S-SOLUBILITY, AND SODALITE-HAÜYNE STABILITY

**Preferred Presentation Format:** 

Either Oral or Poster

## Discipline Categories:

Petrology, Igneous

### Abstract Submission Fee:

Paid (gsa-2020AM-2852-6159-7588-4218)

### **Presenting Author**

Michael Carroll **Email:** michael.carroll@unicam.it

Camerino University Geology Division via Gentile III da Varano Camerino 62032 Italy

### Student? N

Bruno Scaillet Email: bscaille@cnrs-orleans.fr

ISTO, UMR 6113 CNRS/INSU-Université d'Orléans 1a rue de la Férollerie Orléans 45071 France

Juan Andujar **Email:** juan.andujar@cnrs-orleans.fr

ISTO, UMR 6113 CNRS/INSU-Université d'Orléans 1a rue de la Férollerie Orléans 45071 France

### Student? N

## If necessary, you can make changes to your abstract submission until Wednesday, 5 August 2020.

- To access your submission in the future, use the direct link to your abstract submission from one of the automatic confirmation emails that were sent to you during the submission.
- *Or* point your browser to /gsa/reminder.cgi to have that URL mailed to you again. Your username/password are 358075/434175.

Any changes that you make will be reflected instantly in what is seen by the reviewers. You DO NOT need to go through all of the submission steps in order to change one thing. If you want to change the title, for example, just click "Title" in the abstract control panel and submit the new title.

When you have completed your submission, you may close this browser window.

Tell us what you think of the abstract submission process

Home Page