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Hot target magnetron sputtering process: effect of infrared radiation on the deposition of titanium and titanium oxide thin films

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Abstract: Titanium-based thin films are synthesized by the magnetron sputtering of a titanium target in pure argon or in a mixture of argon and oxygen (reactive regime) in three experimental configurations: a hot target configuration for which the target is thermally disconnected from the cooled magnetron body, a cold/conventional target configuration, and a cold target configuration with an additional infrared source introduced in the chamber to understand the role of the radiation on thin film properties. In hot configuration, the target temperature and its emissivity are measured and compared to a 0D steady state model taking into account the main different cooling mechanisms. IR radiations constitute the major part of the total energy deposited on the substrate during hot target magnetron sputtering, whereas an increase of target temperature allows a rise of sputtered Ti and Ti⁺ species fluxes and slightly affect their kinetic energy. In reactive regime, the obtained films are constituted of polycrystalline anatase TiO₂, and an increase of target/IR source temperature leads to an improvement of crystallinity. In pure argon, hot configuration seems to enhance its porosity, induces columns exhibiting sharp-features, slightly modifies its crystalline properties and helps to decrease internal compressive stress.

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Keywords: hot target; magnetron sputtering; titanium; infrared radiation.

1. Introduction

Magnetron sputtering and evaporation (vacuum, arc or electron beam based) are nowadays the most widely used physical vapour deposition (PVD) processes to eject atoms from a solid target allowing the formation of nanoparticles or thin films. In vacuum evaporation process [1], especially adapted for elements with low melting point, material is heated until atoms acquire enough energy to leave surface, and cross vacuum chamber with a very low energy (between 0.1 and 0.3 eV depending on the material) toward substrate. Deposition rate is very sensitive to the nature of materials, making thus difficult to evaporate high melting point elements and to obtain thin films of complex stoichiometry (polycationic oxides or alloys). In sputtering techniques, target is submitted to the bombardment of ions generated in a plasma or issued from an ion beam. This results in the ejection of energetic atoms (few 10's eV depending on sputtering conditions) that are able to form a dense thin film (compared to the thermal/vacuum evaporation technique), which can enhance the film hardness [2,3] but induces tensile or compressive intrinsic stress depending and the deposition conditions [4-6]. During sputter deposition, substrate is submitted to action of other energetic particles (ions, electrons, high-energy neutrals) that can transfer energy and thus affect film formation (sputtering, desorption, implantation, diffusion...) and morphology [7-10]. In turn, film properties such as adhesion, stress, density, grain size, orientation and defects could be modified [11-14]. Even if magnetron sputtering techniques are widely used in industry, the synthesis of defect-free compounds containing labile (as fluor, lithium, sodium ...) and low melting point (as gallium, indium ...) is a difficult task. Moreover, the synthesis of oxides and nitrides with appropriate stoichiometry, good crystalline quality at high deposition rate is really challenging [15-17]. This leads manufacturers to mix various deposition techniques and use post annealing treatments, which complicates the process and lowers its efficiency (in terms of energy saving and environmental fingerprint). A way to overcome these issues would be to perform sputtering of a hot target heated by ion bombardment, so emitting infrared (IR) radiations and acting as a substrate front heater. The effect of target radiation has been already observed almost 20 years ago [18]. But it was rather regarded as an undesired side effect. Whereas a cooling system is usually used to prevent the overheating of magnets and to limit the temperature of target in conventional sputtering configuration, it is possible to promote target surface heating by thermally disconnecting the rear side of target from cooled magnetron. This can even lead to a liquid target surface. Performing sputtering with a high target temperature leads to an increase of deposition rate [15-26], improves crystalline quality (especially oxides) and influences texture [19-26] without any-annealing

1 treatments, modifies stoichiometry, especially for thin film containing volatile/labile elements [27], enhances
2 reactive sputtering stability by reducing hysteresis loop [28], enhances thin film hardness, decreases average
3 surface roughness [24,25], allows heteroepitaxy framework film growth that usually requires high substrate
4 temperature to overcome large lattice parameter mismatch between film and substrate [29]. This could also
5 help igniting/sustaining magnetron discharge during the sputtering of a ferromagnetic target by reaching its
6 target paramagnetic state above the Curie temperature [30-32]. This emerging technology could be very
7 promising, but so far the temperature dependence of mechanisms related to material ejection (sputtering,
8 evaporation), plasma properties (composition, energy distribution, density, ...) and the mechanisms of film
9 growth (under the effect of kinetic and radiative energy transfers) are not yet fully understood, and actually
10 difficult to access and interpret. Moreover, the heating of target induced by ion bombardment cannot be
11 adjusted and controlled properly regardless the plasma properties. The precise knowledge (and the control) of
12 the target temperature is also experimentally challenging either by using thermocouples, thermopiles (heat flux
13 sensor), pyrometers or by analysing thermal radiation spectrum using visible or near-IR spectrometer [33,34].
14 Modelling all the physical mechanisms occurring inside target/magnetron system (IR radiation loss, sputtering
15 at high temperature, evaporation, enthalpy modification, heat propagation, residual cooling by conduction) and
16 at surface (ion reflection, electron secondary emission, atom ejection ...) to finally estimate target temperature
17 is an interesting approach developed by different groups [34-38].

18 To explain the modification of thin film properties, two main energetic contributions can be pointed out. IR
19 radiations emitted by hot target have been suspected by some authors [39] to play a key role (in the heating of
20 growing film for example), but have never been quantified and clearly related to thin film quality. Second,
21 high-energy species which are generally produced at target in conventional (cold) magnetron sputtering
22 transfer their energy to growing film and modify its physical properties. The influence of target temperature
23 on flux and on the energy of these species has never been measured even if research groups investigated the
24 influence of hot target during sputtering [40,41] and pointed out the fact that this hot process should induce
25 lower energy ejected species (due to evaporation mechanism).

26 The present paper aims at investigating both target temperature in hot configuration using a straightforward
27 steady state model validated by experimental measurements and the effect of high target temperature on the
28 sputtering process on the properties of titanium-based thin films (morphology, composition and structure) in
29 pure Ar and in an Ar/O₂ gas mixture. As model material in research (largely documented in the literature) and

1 for various applications, titanium is a good candidate for this study. Especially the relative influence of IR
2 radiations (emitted by the target) and the energy of condensing species (only Ti atoms and Ti⁺ ions, here) on
3 film properties are studied.

4

5 **2. Experimental**

6 *2.1. Sputtering apparatus*

7 Experiments were performed in a 6-way cross-shaped HV deposition chamber pumped down to $8 \cdot 10^{-5}$ Pa using
8 a 500 l/s turbomolecular pump. The pressure is regulated to 1 Pa during deposition with 20 sccm Ar flow rate,
9 whereas the O₂ one is fixed to 0 for non-reactive sputtering (in metallic regime) and 2 sccm (50W) or 3 sccm
10 (100 and 150 W) for reactive sputtering (in compound/oxide regime). The Ti target (thickness of 3 mm and a
11 radius R of 25.4 mm) without (cold configuration) or with (hot configuration) a 1.8 mm thick MACOR® 2
12 inches ceramic disk, is clamped on magnetron (2 inches TORUS® from K.J. Lesker) backing plate. This
13 ceramic disk thermally disconnects the target and promotes its temperature rise. However, clamping
14 surrounding target to ensure electrical connection between magnetron and target induces a residual cooling.
15 The magnetron is powered by a DC power supply (Pinnacle plus 5 kW) between 50 and 200 W in this study.
16 A bichromatic pyrometer (IGAR 6 from Lumasense) placed outside vacuum chamber in front of a ZnSe
17 viewport is used to measure target surface temperature (using the Wien law) and emissivity, denoted ϵ_{pyro} . The
18 distance between ZnSe viewport and target is sufficient (about 300 mm) to neglect deposition on viewport and
19 the modification of its transmission. The pyrometer is placed at the focusable position of 350 mm from the
20 target in normal incidence and the spot size on the target is close to 3 mm. This emissivity naturally depends
21 on the chemical properties of the target surface (metallic, partially or fully oxidized, roughness etc.) but also
22 on the emitted angle Θ , the temperature T and the wavelength λ . In this study, ϵ_{pyro} is defined as the emissivity
23 in the direction of the pyrometer (normal to the target surface, $\Theta = 0^\circ$) in the 2-2.5 μm wavelength range at a
24 temperature T and is a quasi-monochromatic directional emissivity. The precision of the measured temperature
25 is estimated to 2%. Figure 1a displays the experimental set up during the deposition of thin films or during the
26 measurement of the target temperature. During the cooling step of the target, the temperature measured with
27 the pyrometer displays an inflection point corresponding to Ti solid state transition (β -Ti transforms into α -
28 Ti below 882°C) as previously reported by Tesař et al [42]. This transition temperature is used to calibrate

1 directional emissivity ratio between both spectral ranges (1.5-1.6 and 2-2.5 μm) in pure Ar gas, which is found
2 to be between 1.28 and 1.35. These values indicate that the emissivity decreases when wavelength increases,
3 which is commonly observed for metals [43] and Ti in particular [44]. A mean value of 1.3 will be considered
4 in this study. In hot configuration, target temperature reaches 660, 870 and 1010°C ($\pm 50^\circ\text{C}$, high uncertainty
5 between runs of experiments due to the progressive erosion of target, see after) for the 3 respective power
6 values (50, 100 and 150 W) in pure Ar gas. These temperature values are measured in racetrack area (defined
7 by $r=13$ mm) and thus are higher than temperature averaged on the whole surface area. The upper temperature,
8 1010 °C, is largely inferior to titanium melting temperature at 1 atm (1668°C). A thermocouple placed on rear
9 side of Si sample indicates a rise of its temperature up to values ranging from 120 to 220°C in hot configuration
10 depending on DC power, whereas temperature inferior to 50°C are obtained in cold configuration. A higher
11 temperature on the front side of substrate is probable considering that silicon is a good thermal insulator.
12 The substrate holder is placed in front of target with a substrate/target distance of 45 mm and an angle of 45°
13 as shown on Figure 1a. A heating source (covered by a titanium thick film) is installed in front of substrate
14 holder in order to mimic (at the best) target IR radiation emission with the same distance and angle (see Figure
15 1a). Its temperature can be regulated between 20 and 800°C, but only two temperatures have been tested in
16 this study (600 and 800°C). The addition of this heating source allows to work in three different deposition
17 configurations: hot target, cold target and cold target assisted by IR radiation.

18 2.2. *In-situ / in-operando analysis*

19 As shown on Figure 1b, substrate holder can be removed and replaced by a 6 mm in diameter thermopile sensor
20 (also called heat flux sensor, HFM) [45]. This latter is covered by a zynolite back paint and a copper disk (6
21 mm in diameter and 100 μm in thickness) is fixed on it using a thermal conductive paste. Due to the presence
22 of cold junctions (whose temperature is fixed to 5°C) just below the active junctions of sensor, the thermopile
23 acts as a bolometer and directly gives the value of heat flux arriving to sensor (based on Seebeck effect)
24 expressed in mW/cm^2 . All the energy of species impacting to the surface of sensor is converted into heat flux
25 and so such system allows to measure the energy flux density (Φ) coming from gas conduction/convection
26 (Φ_{gas} , neglected in this study), the plasma species (Φ_{plasma}) and IR radiations emitted by the hot parts of
27 chamber, especially the target ($\Phi_{\text{IR,target}}$). A more detailed list of energy contributions will be given in the results
28 section. In this Figure 1b, surface sensor is perpendicular to the glowing target surface (denoted \perp geometry),
29 but the sensor has also been placed parallel to the target ($//$ geometry) as indicated by dashed blue arrow. Sensor

1 has been previously calibrated under vacuum according to the NIST protocol and using a homemade blackbody
 2 (ensuring absorption and emission coefficients close to 1) [45]. In our study, the black paint is covered by a
 3 copper disk and a Ti based thin film (10 min deposition time in pure Ar), so the sensor system cannot be
 4 considered as a blackbody anymore. A part of energy is reflected at thin film surface and must be taken into
 5 account. The reflected and absorbed parts of energy are difficult to precisely estimate. Considering emitting
 6 source as diffuse grey-body, we can estimate its steady-state temperature using the Stefan-Boltzmann law from
 7 the IR radiation (emitted by the hot target $\Phi_{IR,target}$ or the IR source $\Phi_{IR,source}$ and expressed in mW/cm²)
 8 collected on sensor. Moreover, we consider that sensor temperature is much smaller than the emitting source
 9 one.

$$\Phi_{IR(x)}(T) = A\sigma T^4 \text{ with } A = \alpha \varepsilon F_x \frac{S}{S_{HFM}} \quad (1)$$

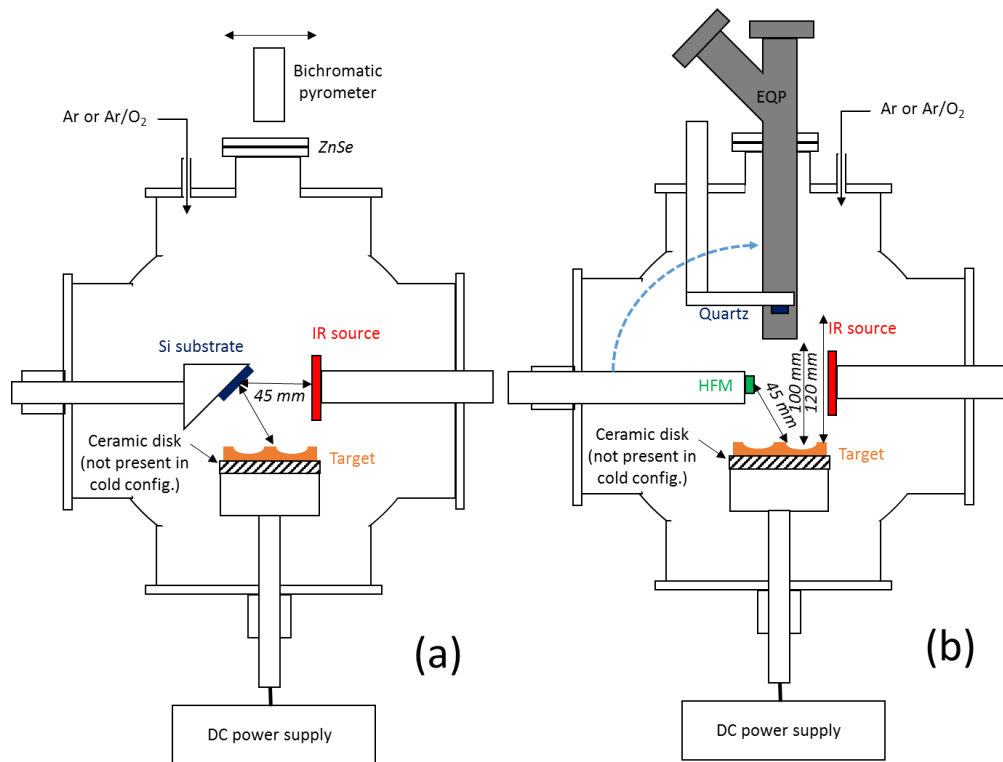
10 where σ is the Stefan-Boltzmann constant ($5.67 \cdot 10^{-8}$ W/m².K), A is a constant (without dimension), ε is the
 11 total emissivity which corresponds to ratio between the excittance of material and that of a blackbody, both
 12 integrated on Θ , T and λ (Θ defined the angle from the normal to the target surface). ε is considered as the total
 13 emissivity even if sensor only measured radiation emitted in its direction Θ . α is the total absorptivity of the
 14 sensor surface and F_x is radiative view factor (geometrical factor taking into account that only a portion of
 15 target IR emission can be detected by sensor) with x indicates the corresponding geometry (x= //, \perp , 45°).
 16 Using a ray-trace software and assuming that the spectral radiance of target is Lambertian, $F_{//}$ and F_{\perp} have been
 17 found to be equal to $3.7 \cdot 10^{-3}$ and $1.3 \cdot 10^{-3}$, respectively. $F_{//}$ can also be deduced from the analytical expression
 18 given in [46]. For substrate (Figure 1a) positioned with an inclination of 45°, software gives a view factor F_{45°
 19 of $3.3 \cdot 10^{-3}$. We found that the ratio $F_{45^\circ} / F_{//}$ is equal to 0.89 (superior to the expected $\cos(45^\circ)$ value), whereas
 20 2.5 is obtained for F_{45° / F_{\perp} ratio. This 0.89 factor (close to 1) indicates that measurement performed in //
 21 geometry gives a good estimation of the flux energy density incoming onto substrate.

22 According to the large thickness of disk (100 μ m) covering the black paint, we consider that all un-reflected
 23 energy is absorbed, converted into heat and transferred to thermopile. In both regimes (oxide and metallic),
 24 copper disk is covered (before recording the measurement) by a Ti metallic thin film (containing residual
 25 oxygen) prior to the recording of experiment. The thickness of this thin film reaches few hundreds of
 26 nanometers in // or in 45° geometry. This thickness is high enough to consider Ti thin film as an IR transducer
 27 absorbing all IR radiation up to few μ m [47, 48]. So the absorptivity of sensor surface is not equal to the copper

1 one but to the Ti thin film in 45° or // geometry. Considering the evolution of refractive index and of the
2 extinction coefficient of titanium bulk in near IR between 1.5 and 2 μm wavelength, an absorptivity of 0.3-
3 0.35 has been found for such Ti thin film (obtained from Openfilter software and from the Fresnel laws).
4 However, the Ti thin film is not purely metallic and contains oxygen species which increased its global
5 absorptivity. Previously experiments have shown that an absorptivity of 0.5 for such Ti thin film was a good
6 approximation [49,50]. This is the value that we took in this study. During the recording of heat flux in oxide
7 regime, an oxide thin film with a thickness inferior to 100 nm covers the previously deposited Ti metallic thin
8 film but the absorptivity of sensor surface is not modified considering that such additional TiO_x thin layer onto
9 Ti, is transparent to IR radiation. In this study, the absorptivity α of the Ti thin film (acting as IR transducer)
10 is close to its emissivity. Indeed, considering local thermodynamic equilibrium, most metal can be considered
11 as a diffuse grey body (contrary to most non-conductive materials).

12 An EQP energy resolved mass spectrometer (Hiden Analytical, EQP1000) is used to estimate the amount and
13 the energy of species interacting with the substrate. Only the energy of Ti sputtered atoms and of Ti^+ ions have
14 been considered in this study. The 200 μm orifice is used and the EQP faces target center at a distance of 100
15 mm. The distribution of Ti atoms has been obtained after their ionisation (electron bombardment) in EQP
16 internal ioniser (RGA configuration). During ionisation, the initial energy of Ti atoms are of course modified,
17 but we assume that this modification is similar for all energy distributions taken into account that these
18 distributions are close to each other. In the mass spectrometer, ioniser is surrounded by a cage biased to 3.0 V
19 (in our experiments) giving energy to ions and ensuring their transport up to the detector, so 3.0 eV has been
20 subtracted to the distributions. An auto-tune procedure has been performed in order to adjust the different
21 internal parameters of EQP and to optimize signal intensity.

22 A quartz microbalance (8 mm in diameter, 6 MHz quartz covered by gold) is placed in front of the center of
23 Ti target (at a 120 mm distance) using water cooled crystal holder in order to measure deposition rate through
24 a STM-2 transducer (INFICON).



1

2 Figure 1. Experimental set up during deposition (a) or during in-situ diagnostic measurements (b) performed
 3 with heat flux microsensor (HFM) perpendicular to target surface (\perp geometry), mass spectrometer (EQP),
 4 and quartz microbalance. The blue arrow indicates HFM is also placed parallel to magnetron (\parallel geometry) at
 5 a 45 mm distance from target surface.

6 2.3. Ex-situ analysis

7 Eight metallic and eight oxide thin films were deposited on (100) Si substrates at three DC power values (50,
 8 100 and 150 W) in cold configuration, with or without the presence of IR source, and in hot configuration as
 9 reported in Table 1. The samples are denoted $Ti_X_Y_Z$ or $TiO_x_X_Y_Z$ (for metallic and reactive sputtering,
 10 respectively) where X stands for sputtering configuration (*cold* or *hot*), Y magnetron power and Z indicates if
 11 IR source was (or not) ignited and its corresponding temperature. Cathode voltage is reported for most
 12 conditions.

13 Rutherford backscattering spectroscopy (RBS) was used to compare the evolution of Ti and O amount in thin
 14 films deposited on Si substrates in the three configurations. The energy of probing ^4He particles and scattering
 15 angle were fixed to 2 MeV and 165° , respectively. Film cross-section (for morphology, and thickness analysis)
 16 was observed by scanning electron microscopy (SEM, Zeiss Supra 40). Structural characterizations were
 17 performed by X-ray diffraction in Bragg-Brentano geometry on a diffractometer (Bruker D8 Discover) using
 18 Cu $K\alpha$ radiation (0.15406 nm). XRD patterns were compared to JCPDS files corresponding to anatase TiO_2

1 (00-021-1272 JCPDS files), rutile TiO₂, hexagonal α -Ti (00-044-1294) and cubic β -Ti (01-074-7075). Results
 2 obtained on the 16 thin films will be presented and discussed.

3 Table 1. Deposition conditions for the 16 thin films obtained in the 3 configurations in pure Ar and in Ar/O₂.

Name	Power (W)	Cathode voltage (V)*	Target temp. (°C)**	IR source temp.(°C)	Dep. time (min)
In pure Ar (metallic regime)					
<i>Ti_Cold_50</i>	50	270	<250	-	10
<i>Ti_Cold_100</i>	100	285	<250	-	10
<i>Ti_Cold_150</i>	150	-	<250	-	10
<i>Ti_Cold_100_IR600°C</i>	100	295	<250	600	10
<i>Ti_Cold_100_IR800°C</i>	100	295	<250	800	10
<i>Ti_Hot_50</i>	50	300	600	-	10
<i>Ti_Hot_100</i>	100	330	870	-	10
<i>Ti_Hot_150</i>	150	391	1010	-	10
In Ar/O₂ (oxide/reactive regime)					
<i>TiOx_Cold_50</i>	50	334	<250	-	180
<i>TiOx_Cold_100</i>	100	373	<250	-	80
<i>TiOx_Cold_150</i>	150	-	<250	-	80
<i>TiOx_Cold_100_IR600°C</i>	200	369	<250	600	80
<i>TiOx_Cold_100_IR800°C</i>	200	367	<250	800	80
<i>TiOx_Hot_50</i>	50	384	600	-	180
<i>TiOx_Hot_100</i>	100	422	800	-	80
<i>TiOx_Hot_150</i>	150	476	1000	-	80

4 * Voltage and power values correspond to the ones given by power supply.

5 ** Temperature measured with bichromatic pyrometer focused on racetrack area (r=13 mm from target center).

6

7

8 **3. Results and discussion**

9 *3.1. Target temperature evolution in hot configuration in metallic regime*

10 This section focuses on the estimation of target temperature in hot configuration. First, a simple model based
 11 on energy balance at target is presented. Spatial distributions of the target surface temperature and directional
 12 emissivity are measured using bichromatic pyrometer. Last, mean target surface temperatures measured at
 13 steady state are compared to values calculated from model.

14 The control of process requires to precisely know target temperature as function of the power injected into the
 15 discharge. This is the reason why the measured temperature of target (T) is compared to values calculated from
 16 a 0D stationary model based on Bleykher et al [36-38]. This model takes into account all contributions: power

1 delivered on target by ion bombardment (W_{ion}), and power evacuated by heat radiation (W_{rad}), evaporation
 2 (W_{evap}) and sputtering processes (W_{sputt}), power due to the flux of reflected ions and to the gas convection being
 3 negligible. Target temperature is derived from the energy balance equation in stationary operation mode where
 4 W_{ion} , W_{rad} , W_{evap} , W_{sputt} represent the respective powers (expressed in W):

$$W_{ion} = W_{rad} + W_{evap} + W_{sputt} \quad (2)$$

5 W_{ion} and W_{sputt} can be estimated as function of discharge parameters whereas W_{rad} , W_{evap} mainly depends on
 6 temperature, material emissivity, saturated vapour pressure, the specific heat of sublimation and the specific
 7 heat of condensed matter.

$$W_{ion} = S E_{ion} F_{ion} = P(1 - \gamma)(1 - \beta) \quad (3)$$

8 Where S corresponds to target surface (πR^2 in cm^2), F_{ion} is the ion flux density impinging onto target
 9 (ion/ cm^2/s), E_{ion} is mean energy of the ions (J/ion), γ is secondary electron emission (estimated to 5% here for
 10 titanium metal and for argon ion energies in the range of 200-500 eV [51,52]) and considered unchanged with
 11 temperature, β represents the energy loss of argon ions in cathodic sheath (estimated to 40% for an argon
 12 pressure of 1 Pa [37] and consistent with collisional plasma sheath model [53]). In our study, we consider a
 13 homogeneous ion flux over the whole target surface. But this is not the case in reality and the distribution of
 14 the current density should be taken into account in order to have a more precise calculation of the target
 15 temperature. P is the DC power delivered by power supply.

$$W_{rad} = S \varepsilon \sigma T^4 \quad (4)$$

16 Where ε is the total emissivity of target surface (integrated on all values of λ , Θ and T), σ is Stefan – Boltzmann
 17 constant. Losses at the rear side of target are neglected, we only consider the front side of the target (S).

$$W_{evap} = S F_{evap} \Delta H \quad (5)$$

18 Where $F_{evap} = \sqrt{\frac{1}{2\pi mkT}} (P_{sat} - P')$, $\Delta H = L_0 + \frac{3}{2} kT - c_p T$ and $P_{sat} = \exp(2.303(C - \frac{D}{T}))$

19 Where F_{evap} is the flux density of metal atoms leaving target (at/ $m^2.s$) according to Hertz-Knudsen law, ΔH
 20 is the difference of enthalpy between the two phases (vapour and condensed state) (J/at), $\frac{3}{2} kT$ is the mean
 21 energy of atoms leaving target (J/at) for monatomic gas, k is the Boltzmann constant, m is the mass of
 22 evaporated atom (kg), L_0 is standard heat of sublimation per atom ($7.23 \cdot 10^{-19}$ J/at) at 0 K, c_p is the specific heat
 23 of condensed matter (in $4.3 \cdot 10^{-29}$ J/at.K) close to $3k$ according to Dulong et Petit law, P_{sat} is saturated vapour

1 pressure at surface temperature T (Pa), P' is vapour pressure at target surface temperature T including the
 2 pressure of ejected metal and working gas (both neglected here compared to P_{sat}), the 2.303 multiplier is used
 3 to convert the base-10 logarithm to the natural one, C and D are metal coefficients determining the saturated
 4 vapor pressure of target material in the Hertz-Knudsen equation. Different values of these coefficients can be
 5 found in literature [54, 55]. In this study, mean values of 10 and 23000 have been taken for C and D,
 6 respectively.

$$W_{sputt} = S F_{sputt} E_{sputt} = S Y F_{ion} E_{sputt} \quad (6)$$

7 Where $Y = \frac{3}{4\pi^2 U_s} (1 - \alpha) U_s (0.15 + 0.13 \frac{M_{Ti}}{M_{Ar}}) (\frac{4M_{Ti}M_{Ar}}{(M_{Ti}+M_{Ar})^2})$, $F_{ion} = \frac{I_d}{qS} (1 - \gamma)$ and $E_{sputt} = U_s/2$

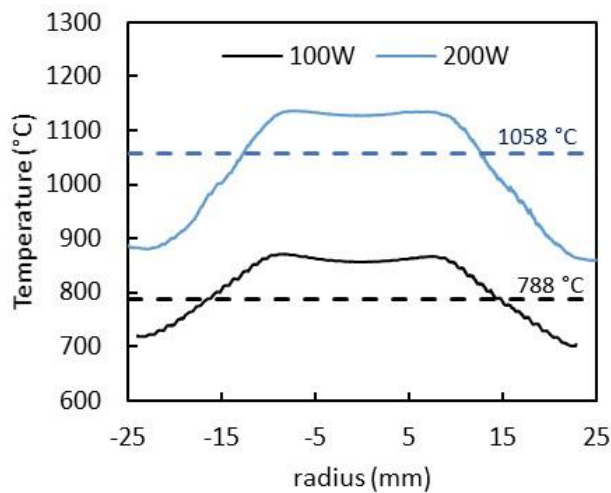
8 Where F_{sputt} is the atomic flux density of sputtered atoms, E_{sputt} is the mean energy of sputtered atoms, Y is the
 9 sputtering yield, M_{Ti} is the molar weight of Ti atom, M_{Ar} is the molar weight of Ar atoms, I_d is the discharge
 10 current, U is the cathode voltage, I_d is the cathode current, U_s is the Ti binding energy (4.85 eV).

11

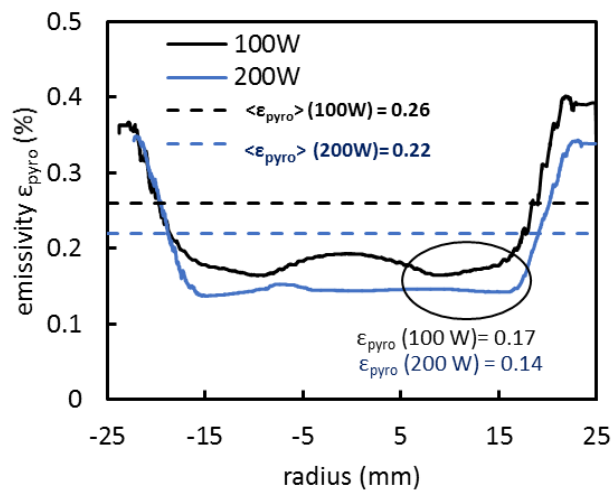
12 From equation (2), the temperature of Ti target can be estimated assuming that there are no conduction losses
 13 between target and magnetron (rear side and target area in contact to the clamping ring). Such conduction
 14 losses could induce a decrease of steady state temperature and eventually modify its uniformity (thermal
 15 gradient along the radius). In hot configuration, since thermal conduction is intentionally impeded at rear side,
 16 the first assumption sounds reasonable. On the contrary, it seems likely that thermal losses occur at clamping
 17 ring. Figure 2a displays the measured temperature of Ti target surface for 100 and 200 W (in pure Ar) along
 18 diameter (origin at target center). The temperature measured at target center is 860 °C for 100 W, and a
 19 temperature decrease of 150°C along a 8 mm distance is observed to target edge (corresponding to thermal
 20 gradient of 2 10⁴ K/m). The mean temperature calculated on the whole circular surface is 788°C as reported
 21 on graph (dashed line). The temperature decrease is equal to 300°C for 200 W (thermal gradient of 4x10⁴ K/m)
 22 with a temperature of 1130°C at center and a mean one of 1058°C. In the case of highly eroded targets, we
 23 have observed that temperature gradient between center and outer limits becomes higher due to the decrease
 24 of thermal conduction in the racetrack area. Thus, the temperature of target centre is higher for an used target
 25 than for a fresh one.

26 The directional emissivity in the 2-2.5 μm spectral range ϵ_{pyro} is reported on Figure 2b along target diameter in
 27 pure Ar. ϵ_{pyro} is minimum in the racetrack area between |6| and |16| mm and reaches minimum values of 0.14

1 at 200 W and 0.17 at 100 W, but largely increases in the outer limit of target to reach values above 0.3. A small
 2 dump (at 0.2 value) in the target is observed at center for 100W. These spatial evolutions of ϵ_{pyro} are due to
 3 differences in the physical state of target: very smooth surface in sputtered area, but rough and partly oxidized
 4 surface at center and around edge. The emissivity averaged over the whole target surface $\langle \epsilon_{pyro} \rangle$ (dashed lines
 5 on graph 2b) is found to decrease from 0.26 to 0.22 between 100 and 200 W, as reported on the graph. This
 6 could appear surprising at first glance (ϵ generally increases with temperature for metals [43]), but is related
 7 to the physical state of target that may not be the same, depending on sputtering power.



8

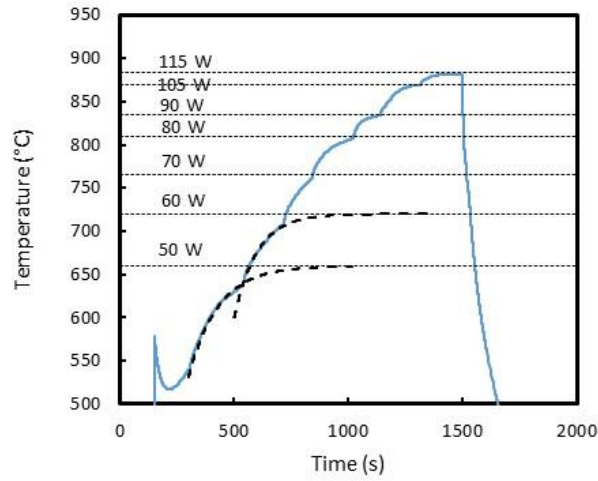


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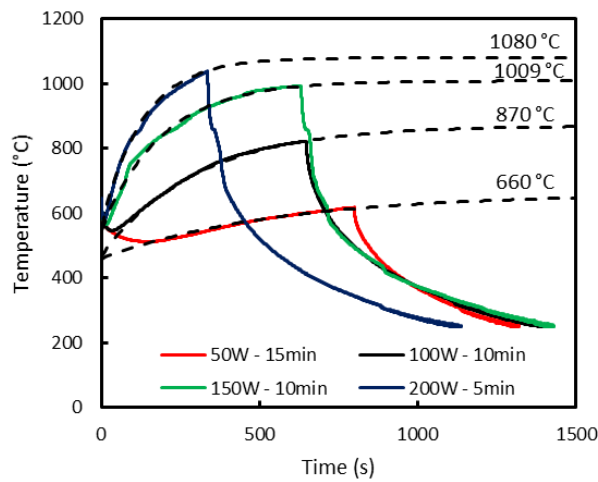
10 Figure 2. (a) Evolution of the target temperature and (b) emissivity ϵ_{pyro} measured using a bichromatic
 11 pyrometer along the target radius at two power values in metallic regime (pure argon). The dashed lines
 12 represent the mean value of temperature and of emissivity calculated on the whole circular target surface.

13

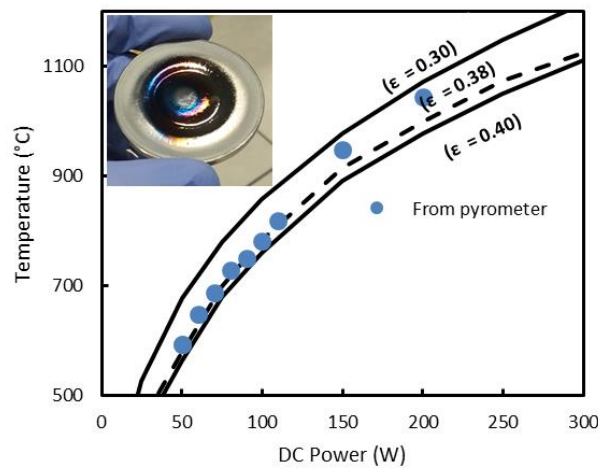
1 Figures 3a and 3b display the time evolution of the target temperature (in the racetrack area, $r \sim 13$ mm).
2 Figure 3a is obtained by progressively increasing (by steps) power between 50 and 115 W, whereas on Figure
3 3b power remains unchanged over time. It is clearly seen that when plasma is lighted on, or when power is
4 increased, the target temperature rises until it reaches a steady state. For both figures, the steady state
5 temperature evolution is extrapolated by using the empirical function $T(t)=T_{\infty}(1 - e^{-t/\tau})$ where the two fitted
6 parameters are τ , a time constant depending on the emissivity and the specific heat of the target (in particular),
7 and T_{∞} , the achieved steady state temperature, as displayed in dashed line on Figure 3a and 3b. The temperature
8 evolution obtained with the model is reported on Figure 3c (for a total emissivity ε equal to 0.30, 0.38 and
9 0.40) and compared to the steady state mean temperature T measured with pyrometer. This mean temperature
10 over the whole target has been evaluated considering a temperature decrease of $7\pm 2\%$ from the temperature
11 measured at $r \sim 13$ mm. This is the percentage that we observed when comparing Figure 2a and 3b. Figure 3c
12 shows that model with $\varepsilon = 0.38$ and experimental temperature measured by pyrometer are in good accordance
13 for power up to 100 W. This total emissivity value is higher than $\langle \varepsilon_{\text{pyro}} \rangle$ value measured by the pyrometer
14 (0.22 at 200 W and 0.26 at 100 W) which can be due to the angular and wavelength dependence of emissivity.
15 This shows that target is not a real diffuse grey-body as we previously assume. But the difference can also
16 come from the emission of IR radiation at the rear side of target which would modify equation (4) by increasing
17 the real emitting surface and so decreasing ε value. For 150 and 200 W, the experimental temperatures are
18 above the modelling curve obtained for $\varepsilon = 0.38$ which confirms that emissivity decreases when its temperature
19 increases or when its surface roughness decreases (both parameters being connected to the power). In addition
20 to the temperature, this model allows to quantify the contribution of each energy loss mechanism. For a power
21 ranging from 50 to 200 W, the mean target temperature is inferior to 1100°C and the IR radiation power density
22 W_{rad} represents 99% of the total incoming power density W_{in} .



1 (a)



2 (b)



3 (c)

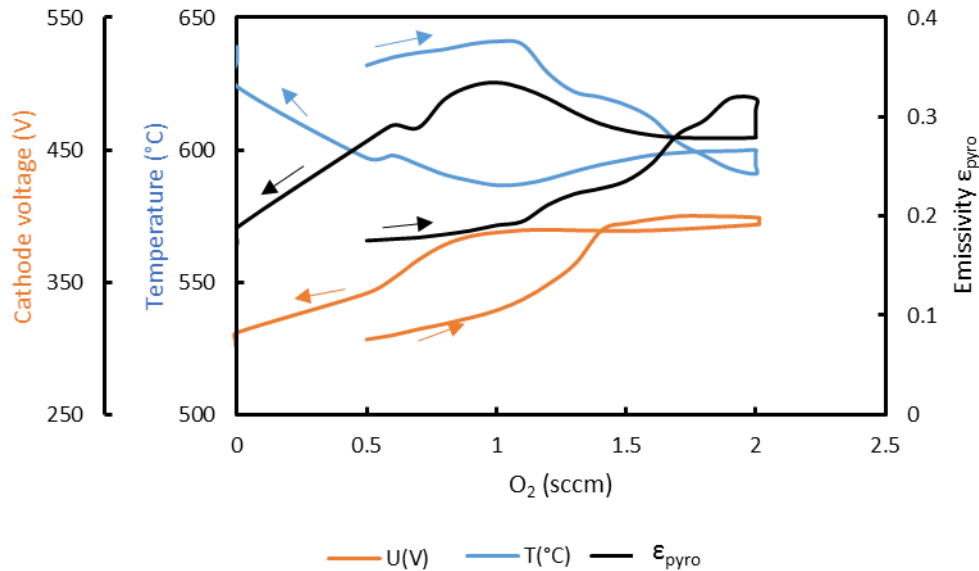
4 Figure 3. Time evolution of the Ti target temperature (in the racetrack area, $r \sim 13$ mm) for powers ranging
 5 from 50 to 200 W applied to cathode (a,b) and temperature evolution (in stationary mode and averaged on the
 6 whole target surface) as function of the power compared to modelling curves (for 3 ϵ values : 0.30, 0.38 and
 7 0.40). A picture of the Ti target after sputtering under Ar/O₂ in hot configuration is added as inset in figure (c).

1 *3.2. Effect of oxygen incorporation on the target temperature and emissivity*

2 In reactive mode, O₂ introduction modifies the properties of target surface, its emissivity and so its cooling by
3 radiation can drastically change. Figure 4 displays cathode voltage evolution, its temperature (in the racetrack)
4 and its emissivity ϵ_{pyro} (in the 2-2.5 μm range) measured by the pyrometer facing the target at a DC power of
5 50 W for increasing and decreasing oxygen flow rates (0.1 sccm min⁻¹). Note that all temperature values and
6 emissivities reported on this Figure are instantaneous measurements and do not correspond to steady state
7 values for a given oxygen flow rate in steady state. The absolute value of voltage increases from 275 to 370 V
8 when cathode goes through the metal/compound transition. Meantime, ϵ_{pyro} emissivity increases from 0.19 in
9 pure argon at 50 W (which is slightly higher than the directional one obtained on Figure 2b at 100 W) to 0.30
10 corresponding to a 1.6 increased factor. Since oxides exhibit higher emissivity than metals, this trend was
11 expected. At 100 W, pyrometer gives an emissivity of 0.17 in racetrack position whereas the total emissivity
12 deduced from model is about 0.38 corresponding to a 2.2 increased factor. Consequently, the total emissivity
13 in oxide regime is probably higher than 0.30. The spread between the way in and the way back result from the
14 different dynamics of oxidation and thermal balance of the target, both processes evolving with its proper time
15 constant. The minimum observed on the temperature curve around 1 sccm on way back, which corresponds
16 strictly to the bump in emissivity at same flow, is interpreted as resulting from the very slow thermal response
17 of target. Moreover, it is well-known that de-oxidation process does not occur in the same way than oxidation
18 route, especially when competing with sputtering. Consequently, target surface state (oxidation degree,
19 roughness etc.) at a given oxygen flow rate may not be the same on way in and back.

20 In addition, the target temperature and emissivity exhibit opposite trend. This is easily explained by the fact
21 that reached temperature depends on the ability of target to be cooled by IR radiations, which is directly
22 connected to the emissivity value. The target temperature decreases from 640°C to 580°C (by a 1.1 factor)
23 when turning from metallic to oxide mode, whereas the emissivity increases from 0.19 to 0.30 (by a 1.6 factor).
24 This indicates that the power released by radiation is rather constant ($1.1^4 \sim 1.5$) at 50 W.

25



1

2 Figure 4. Target temperature, cathode voltage and emissivity ϵ_{pyro} as a function of the O₂ flow rate for 50 W.

3

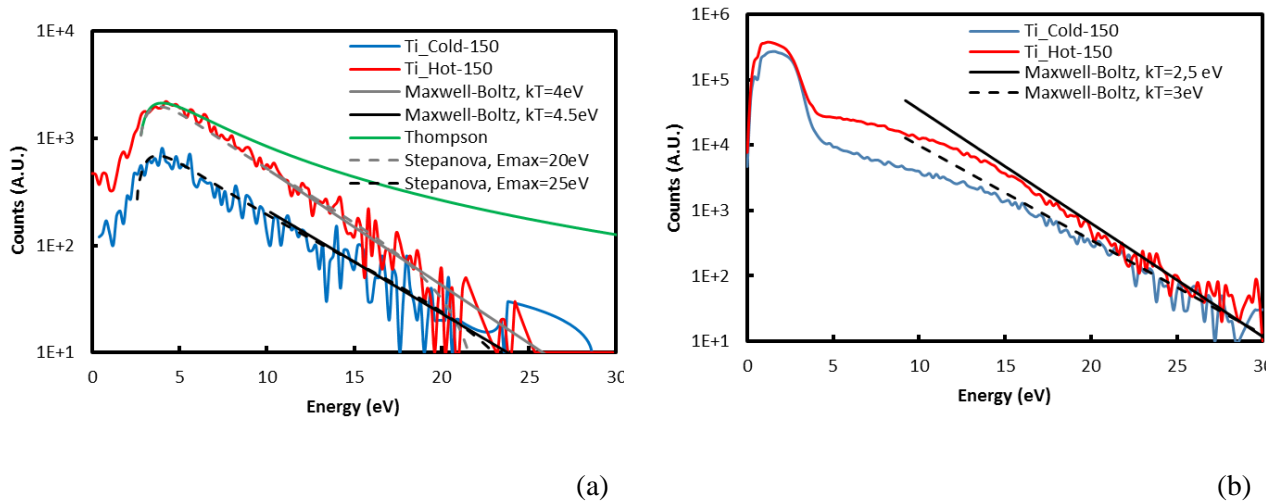
4 The above results show how it is challenging to control target surface temperature in hot configuration.
 5 Actually, the obtained value at steady state is mainly determined by the absorbed power by target and its
 6 emissivity, which is spatially inhomogeneous and target-history dependent. Nevertheless, the experimental
 7 set-up developed in this work and the simple model used allows to get a good estimation of the target
 8 temperature.

9

10 3.3. Energy distributions of Ti atoms and Ti⁺ ions by mass spectrometry

11 The energy distributions of Ti atoms and Ti⁺ ions are measured and compared in hot/cold configuration on
 12 Figure 5 using mass spectrometry in pure argon. Only the distributions obtained at 150 W are presented. The
 13 maximum of the atomic distributions (a) is obtained for about 3.5 eV ($\sim 0.72 U_s$) and seems to be unchanged
 14 between both configurations. Absolute intensity is increased in hot configuration, confirming the increase of
 15 deposition rate when target temperature rises. We do not observe any energy shift (to lower and higher value)
 16 or the appearance of Ti population group at low energy, which could have corresponded to evaporated atoms
 17 ($3/2 kT = 0.15 \text{ eV}$ for $T=1150^\circ\text{C}$). Nevertheless, this does not rule out the existence of evaporation process
 18 since low intensity / low energy contributions may not be detected. Other techniques, as laser induced
 19 fluorescence, could be more relevant for this observation. As shown on Figure 5a, Thompson distribution
 20 largely overestimates the measured distribution for high energies due to collisions in gas phase at 1 Pa.

1 Stepanova et al [56] corrected this distribution by introducing a multiplicative factor $(1-E/E_{\max})$. Both
 2 distributions on Figure 6a are well fitted by the Stepanova function with a cutting energy E_{\max} of 25 eV for
 3 cold configuration and 20 eV for the hot one. The tail of both distributions at high energy can be also adjusted
 4 by a Maxwell-Boltzmann distribution with a kT constant of 4.5 and 4 eV for cold and hot configurations,
 5 respectively. These results show that a high target temperature slightly decreases the kT constant (or the cutting
 6 energy E_{\max}) which leads to decrease the proportion of high energy sputtered atoms and so to a lower mean
 7 energy of the sputtered Ti species. This mean energy is equal to 6.9 eV for the distribution obtained in cold
 8 configuration and 6.35 eV for the one obtained in hot configuration. This decrease may be related to the
 9 decrease of the binding energy in materials when their temperature increases or other physical mechanisms as
 10 the ejection of clusters and the modification of cascade collisions process inside the materials. On Figure 5b,
 11 the Ti^+ ions energy distributions is composed of a low energy population between 0 and 4 eV and a high energy
 12 population up to 30 eV. As for the Ti distribution, absolute intensity is increased in hot regime, especially
 13 between 5 and 20 eV; but the amount of high energy ions seems to decrease in hot regime. The high energy
 14 tails are found to correspond to kT constant of 3 and 2.5 eV for respective cold and hot regime, which confirms
 15 the trend evidenced on Ti atom distributions.
 16



17 Figure 5. Ti (a) and Ti^+ (b) energy distributions obtained using EQP mass spectrometer in cold (blue) and hot
 18 configuration (red) for metallic regime.

19
 20
 21

1 3.4. Energy flux density

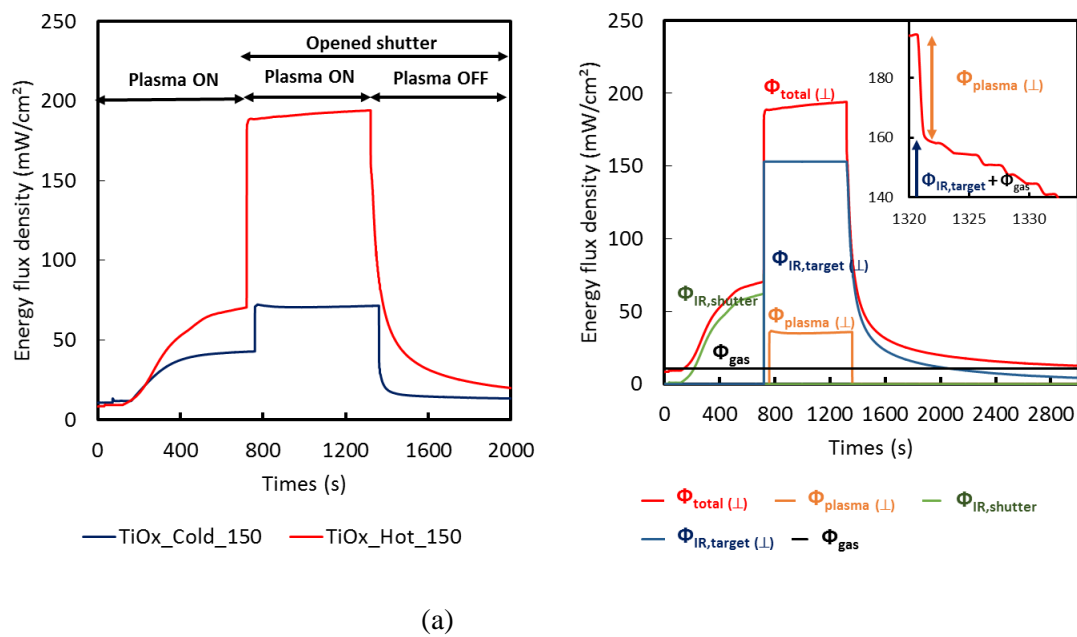
2 The energy flux density measured by a thermopile sensor is given in Figure 6a for the *TiOx_Cold_150* (blue)
3 and *TiOx_Hot_150* (red) deposition conditions with thermopile surface perpendicular to target surface (\perp
4 geometry). Between 0 and 750 s, plasma is running, but the shutter placed in front of target is closed. Thus,
5 the increase observed is only due to IR radiation from the heated shutter (up to 35-70 mW/cm² at 750 s for the
6 cold-hot configuration). When shutter is opened (750 s), a large increase of energy flux density is observed in
7 both cases. These values reach about 70 and 190 mW/cm² for cold and hot target configuration (blue and red,
8 respectively). These overall energies correspond to the sum of several energy contributions coming from:

- 9 (1) the kinetic energy of species sputtered and condensed on sensor and energy eventually
10 released/absorbed during crystallization of these species or during surface chemical reactions.
- 11 (2) the kinetic energy of eventual species evaporated and condensed on sensor and energy eventually
12 released/absorbed during the crystallization of these species or during surface chemical reactions.
- 13 (3) the conduction/convection energy coming from gas atoms (acting as cooling),
- 14 (4) the kinetic energy of high energy plasma species (neutralized and reflected species, negative and
15 positive ions, secondary target electrons etc.)
- 16 (5) the eventual IR radiation emitted by the discharge
- 17 (6) the IR radiation emitted by all vacuum parts placed in front of the sensor and especially the target.

18 As already explained in previous studies [30, 39], the energy contribution corresponding to IR radiation can
19 be measured by analysing the time recording of thermopile. When power is switched off (1321 s), all energy
20 contributions coming from (1), (4) and (5) disappear in few ms and energy flux density quickly drops as shown
21 on inset of Figure 6b. The (3) energy contribution measured just before starting the experiment (plasma off
22 and experiment at room temperature) is equal to 8.2 mW//cm² in this work performed at 1 Pa [57] and we
23 assume that this value is unchanged during the experiment. After subtracting this low value, the energy flux
24 density measured just after plasma extinction (1321 s) results from the addition of the (2) contribution
25 corresponding to the eventual evaporated species and the (6) contribution corresponding to the IR radiation
26 emitted by the heated target (all other vacuum parts are much colder). But the (2) energy contribution can be
27 neglected in the study since EQP measurements have shown that no significant low energy population has
28 been detected at 150 W.

29

1 For indication, Figure 6b displays the “plasma contribution” in \perp geometry denoted $\Phi_{\text{plasma}(\perp)}$ (orange color)
 2 mainly corresponding to (4) energy contribution with the summation of the low (1) and (5) contributions. IR
 3 radiation arriving on surface sensor from target, denoted $\Phi_{\text{IR,target}(\perp)}$, and deduced by this procedure are shown
 4 on the graph (blue color). Finally, $\Phi_{\text{plasma}(\perp)}$ and $\Phi_{\text{IR,target}(\perp)}$ are respectively equal to 33 and 153 mW/cm² in hot
 5 target regime and 36 and 22 mW/cm² in cold target regime for these conditions (oxide regime, 150 W and \perp
 6 geometry). In hot target regime, this measurement reveals that the energy incoming onto the sensor is mainly
 7 due to the IR radiation emitted by target at 150 W in oxide regime, whereas $\Phi_{\text{plasma}(\perp)}$ represents a small part
 8 of total energy.



9 Figure 6. (a) Energy flux density versus time in hot and cold configurations (with O₂, 150 W) for
 10 *TiOx_Hot/Cold_150*. (b) Overview of the different contributions of the energy flux density for the
 11 *TiOx_Hot_150*: IR radiations coming either from the shutter ($\Phi_{\text{IR,shutter}(\perp)}$) or from the target ($\Phi_{\text{IR,target}(\perp)}$),
 12 density coming from the gas (Φ_{gas}) and density coming from the energetic plasma species ($\Phi_{\text{plasma}(\perp)}$).
 13

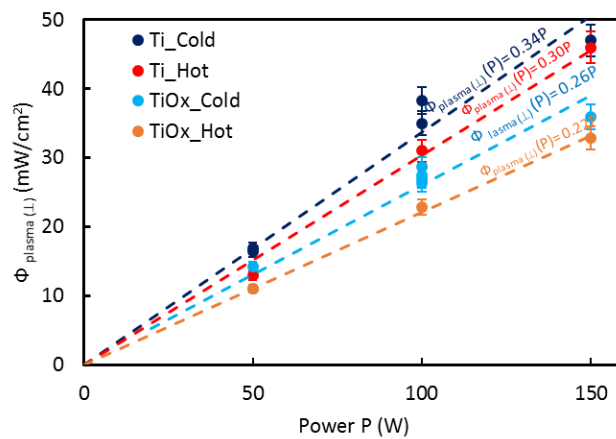
14 The measurement of energy flux density, $\Phi_{\text{IR,target}}$ and Φ_{plasma} , has been performed in all the conditions described
 15 in Table 1 with the HFM surface perpendicular to target surface (\perp geometry) and for few conditions with the
 16 HFM surface parallel to target surface (\parallel geometry). Figure 7a displays the density evolution of energy flux
 17 due to plasma species in \perp geometry, denoted $\Phi_{\text{plasma}(\perp)}$, as a function of the power applied to the cathode in
 18 pure Argon or in an Ar/O₂ mixture. $\Phi_{\text{plasma},\perp}$ is proportional to the power delivered by the power supply from

1 0 to 150 W and the function expression is reported on the graph. The flux is 25% lower in oxide regime (light
2 blue and orange symbols) than in metallic one (blue and red). In addition, hot configuration induces a small
3 flux reduction of about 10% as compared to cold configuration in both cases (metallic and oxide regimes).
4 This may be related to the decreased amount of high energy species (see EQP results) and to the production of
5 low energy species.

6
7 Figure 7(b) and 7(c) shows the density evolution of the energy flux coming from the hot target $\Phi_{IR,target}$ or from
8 the IR source $\Phi_{IR,source}$ as a function of target (or IR source) temperature in an Ar/O₂ gas mixture (b) or in pure
9 Ar (c). To perform such measurements, HFM surface has been placed perpendicular and parallel to target or
10 IR source surface (giving access to $\Phi_{IR,target}(\perp)$, $\Phi_{IR,target}(\parallel)$, $\Phi_{IR,source}(\perp)$ and $\Phi_{IR,source}(\parallel)$). Experimental data are
11 found proportional to the fourth power of temperature as expected from Stefan-Boltzmann law. Fitting curves
12 described by the function $\Phi_{IR}(T) = \Phi_{IR,source}(T) = \Phi_{IR,target}(T) = A\sigma T^4$ are used on both figures for both
13 geometries. The good agreement, which can be noticed between the absorbed IR radiations emanated from
14 the hot target (red symbols) or the IR source (blue symbols), proves that this setup efficiently mimics the
15 radiation coming from the hot target at a given temperature and will permit to isolate the effect of IR radiation
16 on the substrate.

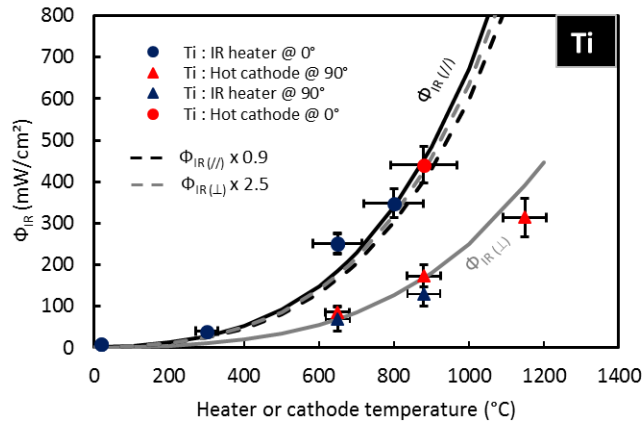
17
18 In // geometry, fitting curves are obtained with A equal to 0.045 and 0.075 for metal and oxide regime,
19 respectively (where A is given by equation (1)). Considering an α absorptivity of 0.5 for sensor surface and a
20 view factor of $3.4 \cdot 10^{-3}$ in // geometry, a ϵ emissivity of 0.36 in metallic and 0.6 in oxide regime is deduced.
21 Both emissivities are largely superior to the directional one measured by the bichromatic pyrometer but are
22 consistent with the total one of target surface deduced from the model for metallic regime (0.38) and to the
23 expected range in oxide regime. In \perp geometries, A values obtained from the fitting curves (Figure 7b and 7c)
24 is equal to $17 \cdot 10^{-3}$ and $6.5 \cdot 10^{-3}$ for metallic and oxide regime, respectively. If a value of 0.5 is used for
25 absorptivity, a ϵ total emissivity of 0.36 in metallic is deduced, which is in accordance with the measurement
26 performed in // geometry. In oxide regime, the thickness of the Ti film, deposited prior the TiO_x one, is too
27 thin to guarantee that it acts as a good IR absorber and the copper absorptivity is much lower than the titanium

1 one. Indeed α is inferior to 0.1 for Cu in the 1-5 μm range using the Openfilter software based on refractive
 2 index and the extinction coefficient values from [58].
 3
 4 These measurements performed in both geometries give the upper and lower limits of energy flux density on
 5 the substrate in the deposition configuration (45° from the target surface). For metallic and oxide mode and
 6 assuming again that emission is Lambertian, the power density received on the substrate $\Phi_{\text{ir,target}}(45^\circ)$ is obtained
 7 by multiplying the power density obtained in // geometry $\Phi_{\text{ir,target}}(//)$ by the $F_{45^\circ}/F_{//}$ ratio (0.9). $\Phi_{\text{ir,target}}(45^\circ)$ can
 8 also be obtained by multiplying the density obtained in \perp geometry $\Phi_{\text{ir,target}}(\perp)$ by the F_{45°/F_{\perp} (2.5) in metallic
 9 regime. Results is given in dashed line on Figure 7b and 7c. The comparison of graphs (a) and (b) confirmed
 10 that the energy flux density coming from the IR radiations emitted by the hot target is much higher than the
 11 one brought by plasma species in metallic conditions. For instance, at the substrate position (45°) and 100W
 12 corresponding to 870°C in metallic regime, the plasma contribution is around $75 \text{ mW}/\text{cm}^2$ and that of the IR
 13 radiation close to $400 \text{ mW}/\text{cm}^2$. At this power, 85% of the energy flux density arriving on the growing Ti thin
 14 film is due to IR radiation in metallic mode and this percentage increases with power/temperature. In oxide
 15 regime and for the same power, the plasma contribution is equal to $55 \text{ mW}/\text{cm}^2$ (22×2.5) and the IR radiation
 16 are also close to $400 \text{ mW}/\text{cm}^2$, which means that 88% of the energy flux density arriving on the growing TiO_x
 17 thin film is due to IR radiation in oxide mode. We must keep in mind that all these energy flux densities
 18 correspond to the absorbed part by sensor but not to the energy flux densities emanating from target.



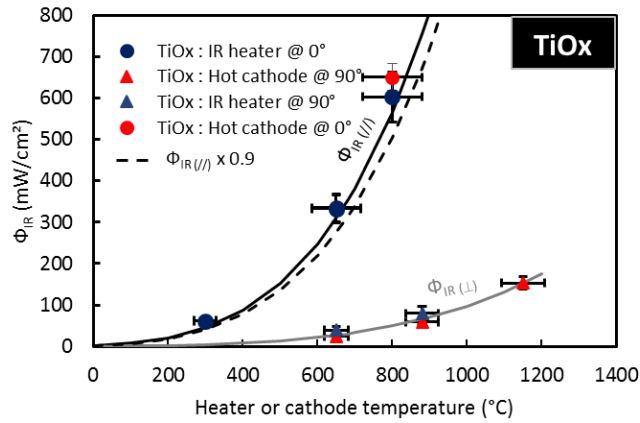
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(a)



1

(b)



2

(c)

3 Figure 7. (a) Energy density flux $\Phi_{\text{plasma}(\perp)}$ as function of the power applied to the cathode and (b,c) energy
 4 flux density flux coming from the hot target or the IR source, in parallel (//) or in perpendicular geometry (\perp),
 5 and absorbed by the sensor surface as function of the target or heater temperature for (b) metallic and oxide
 6 (c) regime. Solid lines stand for Φ_{IR} estimated from the Stefan-Boltzmann law and dashed lines give the
 7 estimated values for the substrate position (angle of 45°).

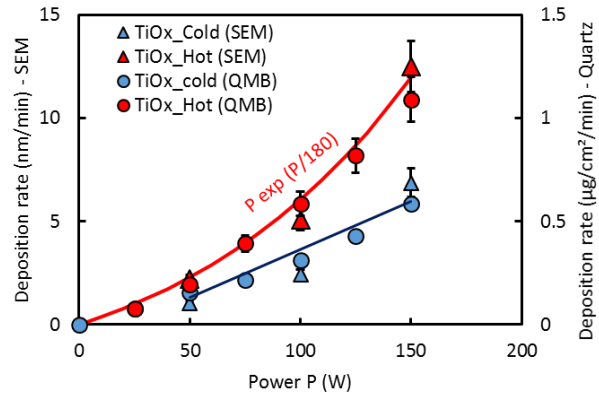
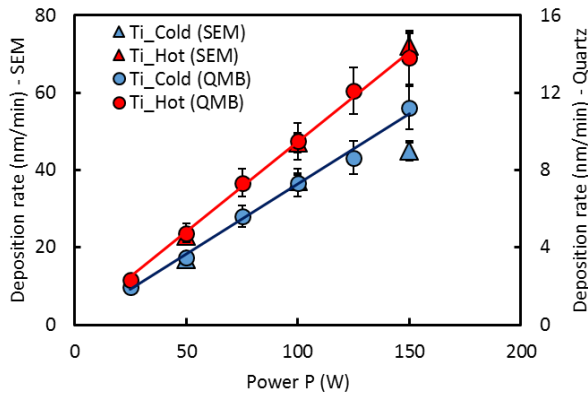
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9 *3.5. Sputtering process and deposition rate*

10 The thickness of each thin film has been measured on SEM images and the deposition rate, expressed in
 11 nm/min, has been calculated and reported in Figure 8 in pure Ar (a) and Ar/O₂ mixture (b) . These values (Δ
 12 symbols) are compared to the deposition rate obtained with quartz microbalance (o symbols). The deposition
 13 rate obtained using the microbalance, expressed in $\mu\text{g}/\text{min}$, are reported in nm/min on Figure 8a assuming a Ti
 14 density of $4.5 \text{ g}/\text{cm}^3$ and in $\mu\text{g}/\text{cm}^2$ on Figure 8b as TiO_x density is not known. In our setup, the quartz
 15 microbalance cannot be placed at the same location as the substrate during deposition (120 mm from target

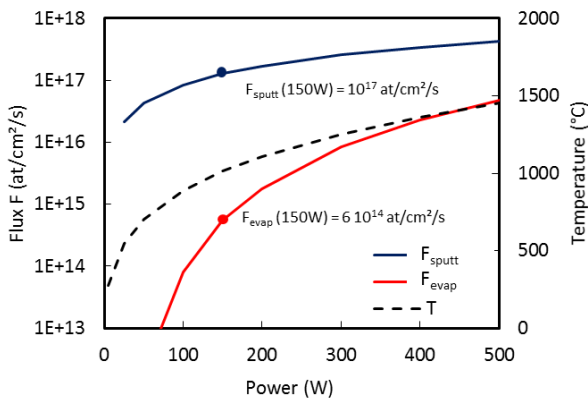
1 instead of 45 mm in deposition configuration) and so the values of the deposition rates measured by
2 microbalance and by SEM cannot be directly compared. However, the evolutions with respect of the DC power
3 applied to the cathode and differences between metallic / oxide modes and cold / hot configurations can be
4 discussed. Clearly, all the trends displayed in Figure 8 are in good agreement for both measurement techniques,
5 which indicates that the density of films does not significantly evolve in the investigated ranges of parameters.
6 In metallic mode and on the explored power range, the deposition rate linearly increases with rising power for
7 cold and hot configuration in metallic mode and only for cold configuration in oxide mode. In this last case,
8 in hot configuration, d exponentially rises (b, red marks). Such trends and behaviour in oxide have been already
9 observed in literature and Musil et al [19] use two straight lines (representing the sputtering and the evaporation
10 mechanisms) as fitting curves in oxide regime. Moreover, the values of deposition rate are found higher in the
11 hot configuration than in the cold one. At 150 W, the deposition rates calculated from SEM cross sections
12 reach 70 nm/min and 14 nm/min for respective metallic and oxide modes in hot configuration, whereas 50 and
13 7 nm/min are obtained in cold configuration. This corresponds to an increased factor of 1.4 and 2. This first
14 value for metallic regime is relatively low compared to previous results obtained by Bleykher et al for Titanium
15 [38], but the specific power was much higher in their study (up to 40 W/cm²) than one used here (up to 7.5
16 W/cm²). Both behaviours (enhanced deposition rate in hot configuration and higher increase in oxide regime)
17 are not fully explained at that time. At such low specific power, evaporation mechanism is negligible according
18 [40] and the EQP measurement supposing that low energy species are detectable. Our model indicates that the
19 flux of evaporated atoms is $6 \cdot 10^{14}$ at/cm².s, whereas the one of sputtered atoms is 10^{17} at/cm².s at 150 W (and
20 1010°C), as shown on Figure 8c. As sputtering is apparently the only ejection mechanism, this deposition rate
21 increase could be related to a decrease of the binding energy or apparition/modification of other physical
22 mechanisms. The addition of ceramic disk in hot configuration leads to an increase of the distance between the
23 magnets and the target surface which modifies the discharge parameters (an increase of the cathode voltage in
24 hot configuration is observed in Table 1 or for Ni in [30]) and may change the distribution of magnetic field
25 lines. This can also be an explanation of the change of the deposition rate. Finally, the increase of deposition
26 rate could also be related to the modification of sputtering yield between hexagonal α and cubic β Ti phases.
27
28 Finally, on Figure 8c is reported the deposition rate measured on SEM cross sections depending on the
29 temperature of the additional IR source for metallic and oxide modes. The graph shows that the temperature

1 of IR source does not influence deposition rate without (blue) or with O₂ (orange). This result proves that IR
 2 radiations emitted by the cathode would certainly have no influence on condensation mechanism and that the
 3 enhancement of deposition rate in hot configuration (see Figure 8a and b) is rather due to a modification of
 4 atom ejection efficiency from hot target. A mean value of 45 nm/min is found for Ti thin film and 2.4 nm/min
 5 for TiO_x at 100 W. The lower value observed for TiO_x films (more than a ten factor) was expected and is
 6 typical of deposition carried out in oxide mode.

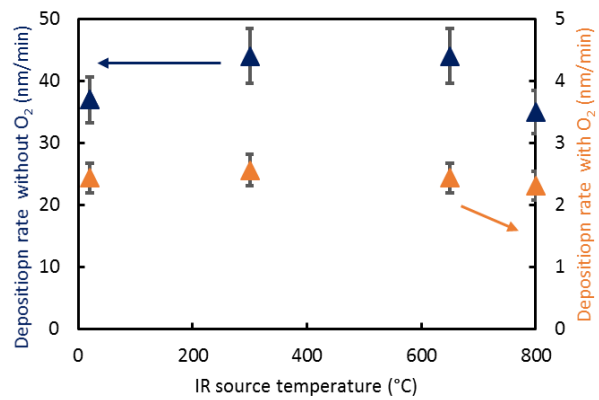


(a)

(b)



(c)



(d)

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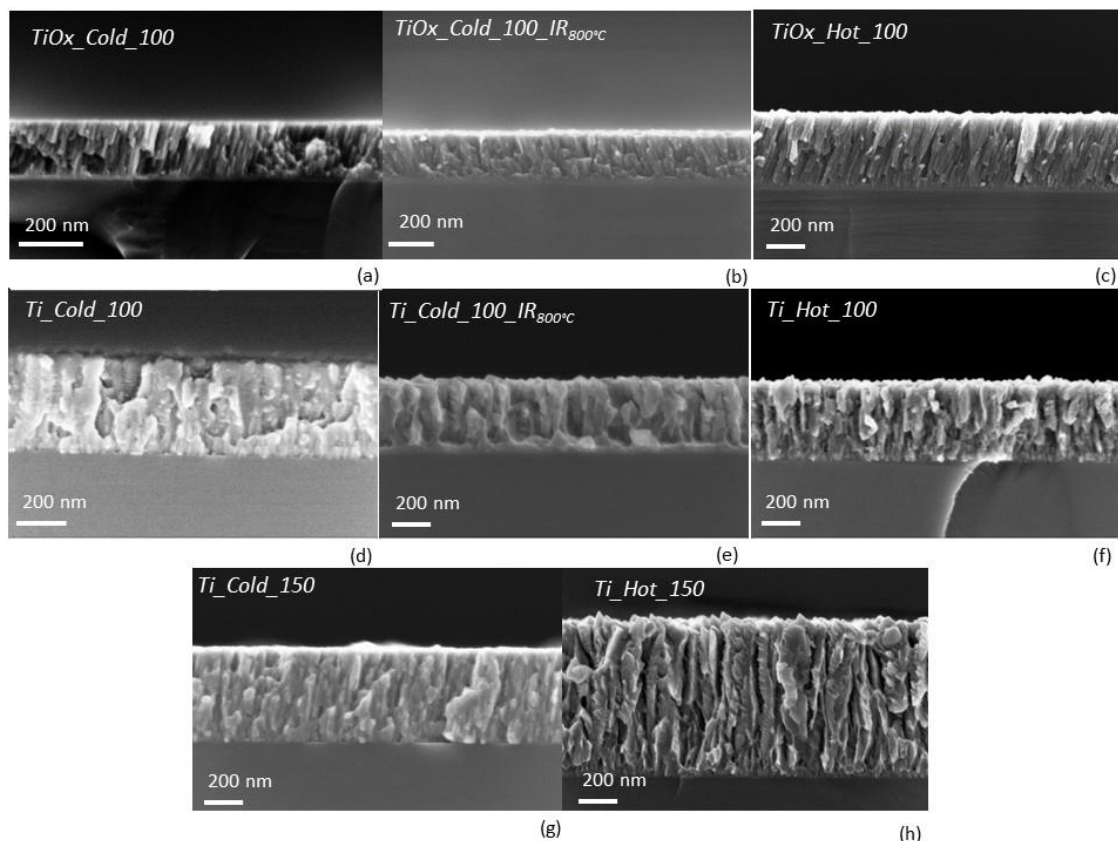
11 Figure 8. Deposition rate determined from SEM cross-section (Δ) and quartz microbalance (QMB)
 12 measurement (o) as function of the power applied to magnetron (a) without and (b) with O₂ in cold (blue) and
 13 hot (red) configuration. (c) Modelled sputtered F_{sputt} evaporated F_{evap} and target temperature as function of
 14 power. (d) SEM deposition rate in cold configuration as function of IR source temperature without oxygen
 15 (blue) and with oxygen (orange) at 100 W.

16

17 *3.6. Morphology of thin films*

1 Figure 9 compares the SEM cross section images corresponding to oxide thin films obtained in an Ar/O₂ gas
2 mixture at 100 W in 3 configurations: cold (a), cold with IR (b) and hot (c). The images on Figures 9 c,d,e
3 corresponds to the metallic thin films obtained in pure Ar in the same 3 configurations. These 6 thin films,
4 whatever the configuration and the processing gas are, present a columnar morphology which is typical for
5 thin film obtained by DC magnetron sputtering at this pressure (1 Pa) with tilted columns coming from the 45°
6 deposition angle. No clear differences between the 6 first images are observed even if the structure appears
7 more disordered (higher porosity, non-straight columns) in metallic thin films than in TiO_x ones. The hot
8 configuration or the cold one with IR radiation does not modify thin film morphology for these deposition
9 parameters and more specifically for 100 W. However, for higher power, here 150W, the morphology of the
10 metallic thin film depends on the configuration as shown on Figures 9g) and 9h). *Ti_Hot_150* thin film (h)
11 seems to display a more ordered structure (than that of all other deposits and notably the *Ti_Cold_150*) and its
12 columns appear faceted and sharp-ended, whereas “smooth” ones are observed in the other deposition
13 conditions. This change in column features from domed tops to sharp ends is typical of growth mode evolution
14 from zone 1 to zone T [59,60] or zone 1a to zone 1c [61] according to the well-known structure zone models.
15 This is usually related to an increase of the deposition temperature and is a proof that hot target configuration
16 influences the mobility of adatoms on the surface.

17 However, this modification in the structure zone models generally goes with a thin film densification as it has
18 been reported by Sidelev et al for chromium thin films [25,26,30] and a V-shaped column appearance. This is
19 not the case here where thin film porosity seems to be enhanced in hot configuration. This higher porosity,
20 typical of thin film obtained by vacuum evaporation, could be related to the decrease of the energy of
21 condensing species evidenced by mass spectrometry even if evaporation mechanism can be neglected in our
22 study taken into account the relative low specific power and low temperature.



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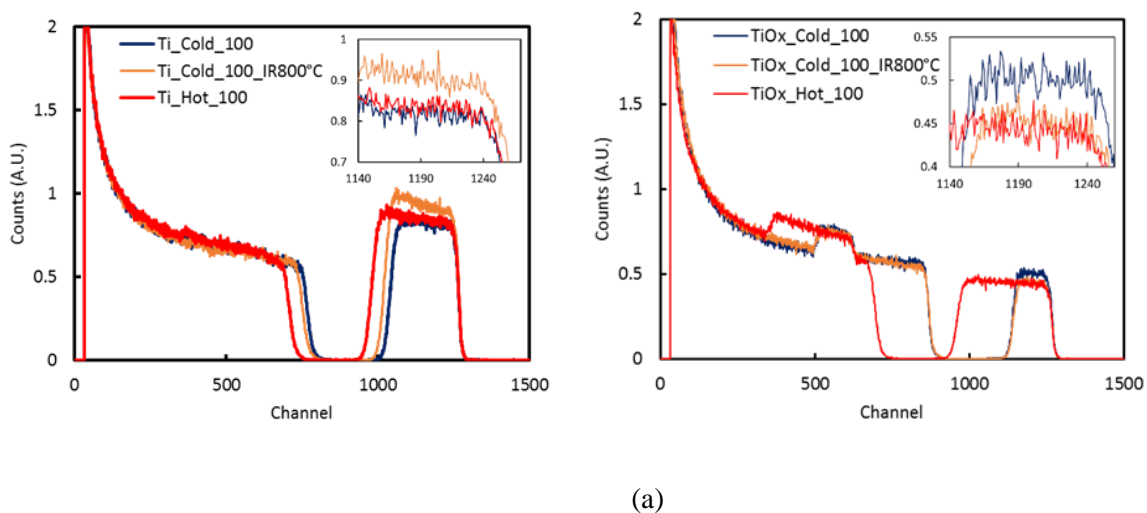
Figure 9. SEM cross section images corresponding to the 3 oxide thin films (*TiOx_Cold_100* (a), *TiOx_Cold_100_IR800°C* (b) and *TiOx_Hot_100* (c)), the 3 metallic thin films obtained for 100 W (*Ti_Cold_100* (d), *Ti_Cold_100_IR800°C* (e) and *Ti_Hot_100* (f)) and the 2 metallic thin films obtained for 150 W (*Ti_Cold_150* (g) and *Ti_Hot_150* (h)).

3.7. Oxygen content in films

Figure 10 displays the RBS spectra for titanium (a) and titanium oxide (b) thin films obtained at 100 W in the three configurations: hot (red), cold (blue) and cold with IR source (orange). On these graphs, the Ti peaks (between the 900th and 1250th channels) become larger in hot configuration (red) especially in Ar/O₂ mixture, whereas deposition time was unchanged. This confirms that larger thicknesses, and consequently deposition rates, are obtained in hot configuration, and that IR radiations do not impact condensation process. The small widening of the Ti peak observed in the case of cold configuration and IR source in metal mode (Figure 10b, orange curve) is not significant. On Figure 10a (TiO_x thin films), it can be seen that the maximum intensity of Ti peak decreases from 0.5 to 0.45 (as shown in the insert) when comparing the cold configuration to the two other configurations (hot and cold with presence of the IR source). This indicates that the deposition assisted by IR radiations (coming from the heated target or the IR source) enhances oxygen incorporation into thin

1 film. The O peak intensity (between the 300 and 700 channels) is not clearly modified but this is not surprising
 2 due to the low sensibility of RBS signal at low channels. On Figure 10b, the highest Ti peak corresponds to
 3 the thin film realized with IR source. If we assume, as suggested by the results obtained in Ar/O₂ mixture, that
 4 the incorporation of O is furthered in hot titanium, this trend can be explained. In pure Ar sputtering, residual
 5 oxygen is mainly consumed by the hottest surface in the vacuum experiment, i.e. the surface of the IR source
 6 that has been coated with Ti during previous runs of experiments. Additional IR source helps to consume
 7 residual oxygen into the chamber and so acts as an efficient O trapping device, in this case.

8 In order to obtain the absolute values of oxygen amount in these thin films, the spectra have been fitted using
 9 the Simnra® software, with Ziegler/Biersack stopping power data and Rutherford cross sections. In oxide
 10 mode, stoichiometric TiO₂ thin films are obtained in hot configuration or with the presence of the IR source,
 11 whereas in cold configuration under-stoichiometric thin films are synthesized (typically TiO_{1.9}). In pure Ar,
 12 when no oxygen is introduced in vacuum chamber, the oxygen percentage in the Ti- based thin film reaches
 13 large values: 25% in cold and hot configuration for 100 W. This high level of oxygen contamination inside Ti
 14 films is correlated to the very high affinity of this metal with oxygen. Of course, this percentage highly depends
 15 on the power applied on magnetron: 50% for 50 W, 25% for 100 W and 20% for 150W, since increasing the
 16 sputtered atom flux leads to higher Ti/O ratios inside films. As mentioned above, the addition of IR source
 17 allows to decrease percentage down to 15% for 100 W. All thin films obtained in pure argon are denoted
 18 “metallic Ti” film for convenience in the following section even if oxygen is present in these films.



19 Figure 10. RBS spectra for Ti (a) and TiO_x (b) thin films obtained in hot configuration (red), cold
 20 configuration (blue) and with the presence of the IR source (orange).

1 3.8. Structural properties of films.

2 3.8.1. Films obtained in oxide mode

3 By XRD, we have evidenced that the obtained TiO_x films are either amorphous or constituted by anatase TiO_2 ,
4 no rutile or brookite phases being observed whatever deposition conditions. In literature, the stabilization of
5 anatase or rutile phase during magnetron sputter deposition at low substrate temperature has been reported to
6 widely depends on deposition conditions: total pressure, oxygen percentage in gas phase, target to substrate
7 distance *etc.* [16,17,62]. Anatase phase is usually observed when energy transferred to the substrate during
8 thin film growth is low [45]. In the present work, we focused our study on the comparison between cold and
9 hot target configurations.

10 XRD analysis shows that all TiO_x thin films obtained at 50 W with a thickness of 400 nm are amorphous
11 whatever the configuration. The diffractogram of *TiOx_Hot_50* film corresponding to a target temperature of
12 660°C is shown on Figure 11a. JCPDS data of tetragonal anatase TiO_2 (# 00-021-1272) are reported on Figure
13 11a (black dashed bars). When increasing power to 100W, the structure still remains amorphous as observed
14 on the *TiOx_Cold_100* (Fig 11a in blue). At this power, IR radiation emitted by target or by IR source seems
15 to promote crystallinity, since the (101) diffraction line of TiO_2 anatase phase is detected at 25.28° . By
16 increasing power, the formation of polycrystalline tetragonal anatase phase is promoted as several peaks
17 corresponding to the (004), (112), (200) (105) and (211) lattice planes are clearly visible on Figure 11a for the
18 film obtained at 150W. However, this conclusion has to be handled with care, because the thickness of
19 *TiOx_hot_150* is more than twice that of other samples. It is well established in magnetron sputtering technique
20 that the crystallinity of a film is directly related to its thickness, reason why the study of effect of deposition
21 conditions can only safely be performed on films of similar depth [61]. To overcome this issue, a second set
22 of 3 deposits *TiOx_Cold_100*, *TiOx_Cold_100_IR* and *TiOx_hot_100* with the same thickness (1 μm) were
23 performed by adjusting the deposition time (respectively fixed to 360, 360 and 180 min). Figure 11b gives the
24 evolution of the main (101), (004), (112), (105) and (211) diffraction lines. The 3 films present a powder-like
25 diffractogram of the TiO_2 anatase phase for which crystallites are randomly distributed on substrate; one can
26 simply notice that the relative intensities of these 3 peaks are modified in hot target and cold + IR source
27 configurations : the intensity of the (112) and (211) peaks is enhanced with the presence of IR radiation.

1 3.8.2. Films obtained in metallic mode

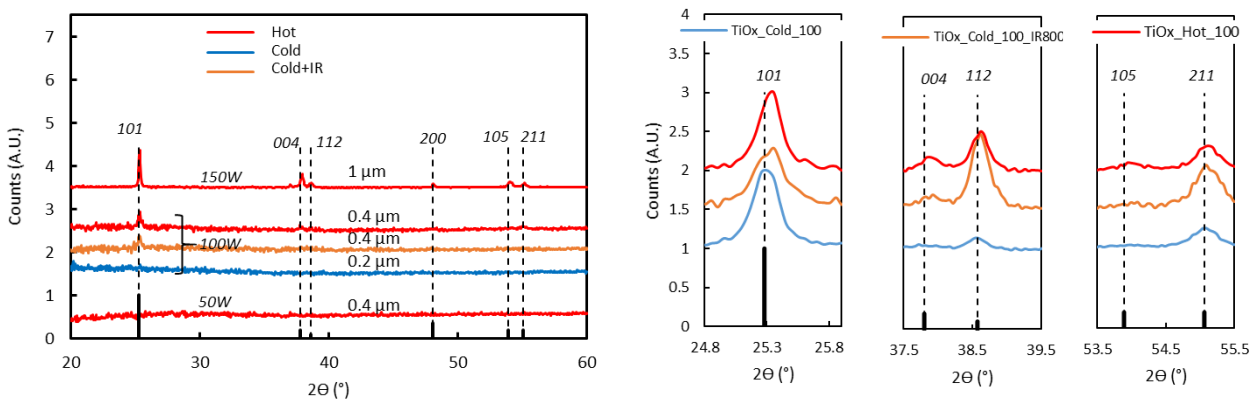
2 The XRD patterns of these films present the diffraction lines of hexagonal α -Ti phase (Figure 11c). The film
3 obtained at the lowest power in cold configuration (*Ti_Cold_50*) is amorphous, while all other films exhibit
4 diffraction lines corresponding to α -Ti bulk reference material (JCPDS #00-044-1294). Globally, same
5 conclusions than before can be suggested: increasing power or adding IR radiations promotes crystallinity and
6 slightly modifies the relative intensities of peaks. The crystallographic orientation of deposited Ti films
7 changes from the (100) to (002) plane exaltation for 100 W. Inverse trend (from (002) to (100)) is generally
8 observed when changing the DC process to a high power impulse magnetron sputtering (HiPIMS) process
9 [63,64]. Inverse changes found between these papers and our work show that increasing energy impinging
10 onto the substrate by an increase of IR radiation or by an increase of ion energy does not play the same role
11 and does not alter film structure in the same way. At the same time, XRD line broadening remains nearly
12 unaffected, implying negligible changes in average grain size.

13 On Figure 11c, it is important to note that no oxide phase is detected on these films, whereas a large amount
14 of oxygen is revealed by RBS analyses. It comes that an amorphous TiO_x phase is certainly present in addition
15 to metallic Ti. This is confirmed by the increase of the background signal intensity observed on XRD patterns
16 at low angle diffraction, which can be related to diffusion process by amorphous matter.

17 Depending on experimental growth conditions, a shift of the angular positions of the (100) and (002) Ti planes,
18 can be noticed. As reference, the (100) and (002) peak positions (dash bars) of bulk α -Ti are given in Figure
19 11c ; they are located at 35.09° and 38.42° , respectively, and the corresponding “a” and “c” cell parameters
20 are 0.2950 nm and 0.4683 nm (JCPDS # 00-044-1294). The “a” and “c” lattice parameters determined from
21 the (100) and (002) peak positions on the XRD patterns (Figure 11c) are in the ranges 0.2911 – 0.2928 nm and
22 0.4682 – 0.4704 nm, respectively. Compared to reference values, the “a” parameter is lower while the “c”
23 parameter is higher, meaning that compressive stresses exists in crystallites. In addition, the “c/a” ratios, which
24 range from 1.59 to 1.62 in films, are higher than that of bulk α -Ti (1.587). When increasing power up to 100
25 W, the positions of the (100) and the (002) are shifted to the corresponding positions of the bulk α -Ti,
26 especially for hot and cold+IR configuration. This means that the 2 cell parameters and the “c/a” ratio become

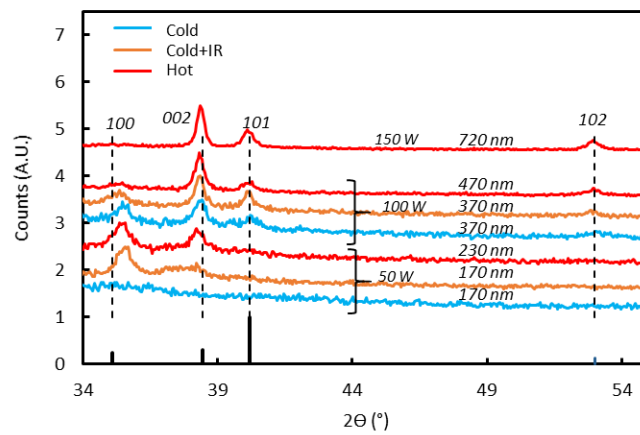
1 closer to the ones corresponding bulk α -Ti. Indeed, the “c/a” ratio is slightly inferior to 1.60 for
 2 *Ti_Cold_100_IR* and *Ti_Hot_100* thin films.

3 Thus, it may be concluded from this part that the IR radiations coming from hot target or the IR source have
 4 an effect on the structural properties of Ti films. While the *Ti_Cold_50* film is amorphous, *Ti_Hot_50* and
 5 *Ti_Cold_50_IR* ones are crystalline and *Ti_Cold_100_IR* and *Ti_Hot_100* exhibit a “c/a” ratio inferior to 1.60
 6 indicating that the presence of IR radiation tends to decrease the compressive stress inside film. Moreover, the
 7 crystallographic orientation of deposited Ti films changes from the (100) to (002) plane dominance for 100 W
 8 by using hot target configuration inducing IR radiation.



(a)

(b)



(c)

11
 12 Figure 11. X-ray diffractograms of (a) TiO_x thin films with various thicknesses, (b) TiO_x films with same
 13 thickness (1 μm) and, (c) Ti thin films in the three configurations: hot (red), cold (blue) and cold with IR
 14 radiation (orange). The XRD patterns are compared to JCPDS files corresponding to anatase TiO_2 (# 00-021-
 15 1272), and hexagonal α -Ti (# 00-044-1294). No peak related to cubic β -Ti has been found.

1 **4. Conclusion**

2 We have investigated the sputtering of titanium in hot target configuration in pure Ar and in an Ar/O₂ mixture
3 by comparing three configurations: cold target, hot target (insertion of a ceramic disk between the target and
4 the cooled magnetron) and cold target assisted by IR radiations. The effect of target surface temperature (up
5 to about 1000°C) has been evaluated on sputtering process and on properties of Ti, and TiO_x thin films for
6 relatively low power (inferior to 150 W). This study confirms the trends reported in literature: the sputtering
7 of a hot target induces an increase of deposition rate. In the present work, we found a 1.4 factor in metallic
8 mode and a 2 factor in oxide mode for 150 W for which target temperature (measurement and model) reaches
9 about 1000°C. The introduction of O₂ in sputtering apparatus induces an increase of target emissivity and so a
10 slight decrease of temperature.

11 No evidence of evaporation process, even at the highest temperature, has been found by mass spectrometry. A
12 slight decrease of high energy Ti and Ti⁺ species proportion has been detected using spectrometry leading to a
13 decrease of their mean energy when incoming onto the substrate. The influence of the target temperature –
14 especially in reactive (oxygen) sputtering – on the formation of oxygen radicals and negative oxygen ions near
15 the target surface has not been studied but could be of interest. This suggests a modification of the sputtering
16 process efficiency of heated material, leading to a rise of the sputtering yield. This can be related to the binding
17 energy that is known to be lowered at high temperature and may also be the reason for the decrease of the high
18 energy populations of the ejected species. However, this parameter cannot justify alone the deposition rate
19 increase and other ones are probably affected by high temperature. Energy flux measurements have shown that
20 the energy brought by plasma species slightly decreases with the rise of temperature. The important result is
21 that the IR radiation emitted from hot target is the main part (about 80%) of the global energy transferred to
22 the substrate in hot configuration. The influence of this contribution that has been suspected by several authors
23 has clearly been found in the present work to determine final film properties.

24 All films are found columnar whatever the deposition configuration. Concerning the thin films obtained in
25 Ar/O₂ gas, no clear influence of IR radiation has been obtained on TiO_x thin film microstructure, oxides being
26 slightly denser than metals, as usually reported. The enhancement of oxygen incorporation in thin film is
27 observed in oxide regime from RBS measurement. Working in hot target configuration seems to help
28 synthesizing a nearly-stoichiometric titanium dioxide. This is consistent with an enhancement of the crystalline
29 state of TiO₂ anatase films characterized by crystallites randomly distributed on substrate (even if an

1 enhancement of the (112) and (211) peaks is observed). Concerning the Ti-based film obtained in pure argon,
2 no oxide phase is detected on these films, whereas a large amount of oxygen is revealed by RBS analyses. It
3 comes that an amorphous TiO_x phase is certainly present in addition to metallic Ti. Raman spectroscopy
4 analysis could have been of interest to precise such microstructure. Hot configuration seems to enhance
5 porosity of Ti-based film only at 150 W and induces columns exhibiting sharp-features. Moreover, crystallinity
6 and crystallographic orientation are slightly improved and modified with a decrease of the compressive stress
7 inside the film.

8 Finally, the IR radiations emitted in hot target magnetron sputtering are thought to be responsible for the
9 modifications of thin film properties as compared to cold configuration.

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