

# Toward a comprehensive study for multielemental quantitative LA-ICP MS bioimaging in soft tissues

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- 1 Toward a comprehensive study for multielemental quantitative LA-ICP MS
- 2 bioimaging in soft tissues
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#### Abstract

Quantitative localization of metals in biological tissue sections is critical to obtain insight into metal toxicity mechanisms or their beneficial characteristics. This study presents the development of a quantitative LA-ICP MS bioimaging methodology based on the polymer film strategy and internal standardization. To maximize the number of elements mapped, an aqueous soluble polymer (dextran) was selected. Among the elements studied, the great majority (eight out eleven), i.e., Co, Ni, Cu, Zn, Se, Mo, Cd and Pt, exhibited linear regression after LA-ICP MS analysis of metal-spiked polymer standards. Methodology performances were carefully assessed as a function of the three internal standards (In, Rh and Ir) considered, the analytical operational conditions (ICP power, addition of O<sub>2</sub> to ICP, and laser fluency) and the thickness of the biological tissue section. The results indicated that three groups (Co, Mo; Ni, Cu, Pt; and Zn, Se, Cd) of elements could be distinguished from their analytical response as a function of analytical conditions and the internal standard. These different element behaviors

appeared to be mainly First Ionization Potential dependent (FIP). For elements with lower FIP (Co, Ni, Cu, Mo and Pt), differential responses due to carbon load in the ICP MS plasma could be efficiently corrected as a function of analytical conditions. Matrix effects were more pronounced for higher FIP elements (i.e., Zn, Cd and Se), and analysis of <10- $\mu$ m thin sections without the addition of  $O_2$  to ICP MS plasma is recommended. LODs are in the range of 0.1-0.5  $\mu$ g g<sup>-1</sup> for Co, Mo, Cu, Ni, Pt and Cd as well as 0.9 and 1  $\mu$ g g<sup>-1</sup> for Zn and Se, respectively. The methodology was validated by means of a homemade metal-spiked kidney homogenate analyzed by LA-ICP MS imaging, and Co, Ni, Cu, Mo, and Pt provided the closest concentrations (5-29% bias) to the target values determined by ICP MS after mineralization. The methodology was applied to two types of clinical human samples undergoing different sample preparation protocols that did not affect internal standard homogeneity in the polymer film. This methodology is the first reported for the quantitative bioimaging of eight elements simultaneously.

Keywords: LA-ICP MS, bioimaging, quantification, multielemental, polymer films

# Introduction

Metals affect many aspects of life. Some metals are essential or beneficial, whereas others are deleterious or even toxic (i.e., iron is the most widely distributed metal and is essential to oxygen transport; mercury is neurotoxic, but its presence with selenium inhibits its toxicity) [1-2]. The bioinorganic chemistry of the organism/cell governs the role of each metal, stressing the requirement for a fine description of metal localization. Thus, visualizing metals distribution at the organism and cellular level is critical for understanding how metals function and how their functions might be altered [3]. Moreover, the measurement of the concentration of metal in a specific area of a biological tissue is of paramount importance. Indeed, metal deficiency or overload can be a proof of disease (i.e., iron misbalance in neurological diseases [4-5], high copper level s in sera of patients with breast cancer [6]), and studies of metal-based drug penetration into tumor facilitates treatment protocol development [7-9].

LA-ICP MS is the most advanced technique for metal species quantification given its sensitivity (low  $\mu g$  g<sup>-1</sup> level) [10-13] and sufficient spatial resolution ( $\mu m$  and sub- $\mu m$ ).

Nevertheless, quantification of metals in tissue thin sections using LA-ICP MS is problematic. Indeed, the heterogeneity of biological samples and the lack of suitable standard material give rise to difficulties (ablation, transport, ionization into ICP source, signal drift over analysis time and matrix effects) that are well documented [14]. Various approaches have been proposed to improve metal quantification by LA-ICP MS in biological tissues; however, there are still discrepancies regarding a truly universal calibration and quantification protocol [14-15].

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The quantification methodologies by LA-ICP MS can be classified into three strategies. One of the pioneer works corrected instrumental or elemental sensitivity through the addition of an internal standard into either the ablation cell or the ICP source [16]. However, correction strategies for ablated material variation and transport are lacking. External calibration strategies have emerged using matrix matched standards, i.e., gelatin-dried droplets [13] and standards [17], embedding resins [18], or a piece of the sample itself [19-20] spiked with different amounts of metal. These strategies that lack internal standard require the complete ablation of the standard as well as the sample itself, and the lack of a blank may cause issues with the measurement of endogenous elements, such as Cu, Zn, and Fe, among others. Internal standardization strategies consist using either a native element homogeneously distributed in the sample [16], but this is often difficult to find, or by introducing an internal standard by full immersion [21] of the slice sample into the metallic solution, but this method potentially bears the risk of metal redistribution. Other internal standardization strategies rely upon adding an internal standard spiked layer either at the top or below the sample. Among these strategies, printed metal spike ink [22] onto the sample surface, spraying an internal standard coating on top of the sample [23] and depositing the internal standard between the sample and the substrate [24] represent the main strategies being developed. Here, the complete ablation of the sample and internal standard ensures their simultaneous transport. Internal standard-spiked ink methods may be limited when endogenous elements, such as copper, are under study given their natural presence in ink. Coating the sample with an internal standard spray is typically performed by sputter using Au, Cr, Mo, Ni, Ti, W, Ag, Pt, Cu, Pd, Co, or Fe as commercially available metal targets. Therefore, given that most of these metals may be encountered in biological samples, applications are often limited to the use of Au as internal standard, which is only appropriate for a set of elements. Interfacing a metal spiked polymer film between the support (glass slide) and the sample requires some sample preparation similar to that noted with others methodologies and the use of a spin-coater, but

leaves free ourselves of the choice of the polymer and the internal standard according to the bioimaging application. Autin et al [24] has proposed an organic soluble (xylene and chlorobenzene) polymer (polymethylmethacrylate) requiring organometallic standards (Cu and Zn for calibration and Y and Ru as internal standards) for spiking experiments. Normalization with Y and Ru improved the accuracy of Cu and Zn quantification on a 20-µm tissue slice despite increased standard deviation compared to <sup>13</sup>C normalization. This methodology is very promising given that it can be potentially used in a wide panel of various applications (different elements to map). Nevertheless, the use of such a type of polymer is a limiting parameter in the choice of internal standards and the handling of organometallic standards (non-aqueous soluble); thus, further developments should be considered.

The choice of the internal standard for an effective normalization of the analyte is of course of paramount importance. The use of carbon (13C) has been long time discussed to correct for variation in ablated mass, but its low sensitivity (11.6 eV first ionization potential (FIP)) and its heterogeneous distribution in biological samples prevent its use for quantitative bioimaging by LA-ICPMS. The choice of an isotopic variant of the analyte itself would be ideal to correct for instrumental signal drift, matrix effects or analyte losses [25-26]. However, this is not always possible (case of monoisotopic element) and given that this method would ideally imply the mixing of the enriched isotope with the sample at the earliest point in the preparation procedure, this would limit the field of applications (impossible in case of clinical samples for example). The selection of the analyte/internal standard pair is typically made based on the criteria of a close mass and first ionization potential between the two elements. Autin et al [27] recommended the incorporation of an internal standard with an atomic mass close to the analyte for imaging experiments to correct for variation in ablated mass and sampling cell position. In addition, <sup>197</sup>Au, which has a high mass and mid-range ionization potential (9.22 eV), compensates for instrumental drifts, matrix-related ablation differences and day-to-day signal changes for <sup>63</sup>Cu (FIP 7.73 eV) and <sup>64</sup>Zn (9.39 eV) analysis despite the high mass difference [23]. Therefore, the choice of an internal standard relies on the application.

The aim of our study was to develop an internal calibration methodology based on polymer film standards to extend this methodology to the quantitative bioimaging of a wider range of elements by LA-ICP MS. Thus, the nature of the polymer was first considered by targeting aqueous soluble polymers for easier spiking with metals. The methodology aimed at the

application to a set of elements encountered in biological samples given their natural occurrence, their therapeutic use or their presence as pollutants. Several internal standards were investigated for normalization of the various elements for increased stability and sensitivity. The accuracy of our methodology was assessed by means of a homemade matrix-matched standard at three concentrations levels and different thin-section thicknesses. The influence of experimental conditions of LA-ICP MS analysis, such as ICP power and composition of ICP plasma (O<sub>2</sub> addition), were also evaluated to minimize differential analyte responses over the imaging run time.

## **Materials and methods**

# Reagents and samples

Elemental solutions at 50 000 mg L<sup>-1</sup> Fe, Cu and Zn and 10 000 mg L<sup>-1</sup> of Pt, P, Co, Ni, Se, Mo, Hg, and Cd were purchased from SCP Science (PlasmaCAL, Courtaboeuf, France). Briefly, 1 000 mg L<sup>-1</sup> solutions of In, Ir and Rh used as internal standards were purchased from SCP Science. Intermediate elemental solutions at 100 and 2500 mg L<sup>-1</sup> were prepared in 2% ultrapure HNO<sub>3</sub> (70% INSTRA-ANALYZED, Baker, Illkirch-Graffenstaden, France). Dextran polymer (Mw of ca. 500 000 g mol<sup>-1</sup>) was purchased from VWR (Fontenay-sous-Bois, France). All solutions were prepared with milliQ water (resistivity  $\geq$  18.2  $\Omega$  cm) obtained from a Milli-Q Elix water purification system (Millipore, Molsheim, France).

# Polymeric film preparation and thickness characterization

Here, 60 mg mL<sup>-1</sup> dextran solution was prepared in water by simple vortexing. A portion of this solution was spiked with the internal standards solution (In, Rh, Ir) to obtain a theoretical concentration of 10  $\mu$ g g<sup>-1</sup> before preconcentration by spin-coating and dedicated as sample deposit substrates after spin-coating. The other portion of the solution was spiked with internal standard solution and different amounts of metal solutions (Fe, Cu, Zn, Pt, P, Co, Ni, Se, Mo, Hg, Cd) to produce calibrants for quantification by LA-ICP MS after spin-coating. In each standard, two ranges of theoretical elemental concentrations were aimed in polymeric film, i.e., from 0 to 80  $\mu$ g g<sup>-1</sup> for minor elements (Co, Ni, Se, Mo, Cd and Hg) and from 0 to 400  $\mu$ g g<sup>-1</sup> for major elements (P, Fe, Cu, Zn and Pt) while maintaining the same internal standard concentration (i.e., 10  $\mu$ g g<sup>-1</sup>) as for samples deposit substrates. For sample deposit substrates

and each calibration solution, the proportion of polymer remained constant, whereas final volume was adjusted with milli-Q water. Each solution was individually spin-coated as follows. Approximately 600  $\mu$ L of each 45- $\mu$ m filtrated (45  $\mu$ m cellulose filter, VWR, France) solution was loaded onto the surface of a cleaned 2.6 x 2.6 cm glass slide (VWR international, Fontenay-sous-bois, France) and spin-coated (Polos SPIN150i spin-coater model, SPS, Vourles, France) following optimized conditions at 4500 rpm for 60 s under a light N<sub>2</sub> gas stream. After spin-coating, film substrates were ready for analysis or sample deposit. The average mass (n=5) of polymeric film was determined for each calibration point using an ultrahigh precision balance (Automated-S microbalance XP6, Mettler Toledo, Viroflay, France) by weighting glass slide before and after film deposition (providing a clean glass slide surface). Film thickness was determined by optical profilometry using a WYKO NT1100 Optical profiling System (Veeco, Dourdan, France) in Vertical Shift Interference (VSI) mode after using a scalpel to scratch the polymeric film on the glass substrate and coating with a thin Au layer (1-5 nm) for 30 s to make the surface reflective to light.

## Preparation of matrix-matched tissue standards

A piece of pig kidney was purchased at a local supermarket and was homogenized using an Ultraturax (IKA T10 basic, Imlab, Lille, France). Aliquots of homogenized kidney were used as control standard, 2.5  $\mu$ g g<sup>-1</sup> spiked standard, 50  $\mu$ g g<sup>-1</sup> spiked standard and 250  $\mu$ g g<sup>-1</sup> spiked standard (concentrations expressed as major elements). To prepare these standards, approximately 5 g of homogenized kidney tissue was placed into a 10-mL tube, nonspiked (for control standard), or spiked with an appropriate volume of 100  $\mu$ g g<sup>-1</sup> (for 2.5  $\mu$ g g<sup>-1</sup> standard) and 2500  $\mu$ g g<sup>-1</sup> (for 50 and 250  $\mu$ g g<sup>-1</sup> standard) of intermediate elemental solutions before homogenization with Ultraturax. For each condition, 2 aliquots were transferred (as compact as possible to form a cylinder) into 2-mL Eppendorf tubes, frozen under liquid N<sub>2</sub> and stored at -18°C until cryomicrotomy.

## Cryomicrotomy

Matrix-matched tissue standard- Homogenate kidney standards were removed from Eppendorf tube and embedded with OCT (Q Path® VWR). Thin sections of 8, 10, 14, 30 μm thickness were produced from the kidney standards using a cryomicrotome (CM 1950 model, Leica, France) where -18°C and -15°C temperatures were set for the specimen and the chamber, respectively, and deposited on the polymeric substrate.

**Patient tissues**- For sanitary, safety and regulatory issues, human tissue samples (ovary and tumoral nodules) were collected, frozen at -80°C and processed at the Institut du Cancer de Montpellier in accordance with standard practices. Here, 8-μm thin sections of ovary were obtained from frozen tissues using a Microm HM 550 (Thermo Scientific) cryotome set at -18°C and -20°C for specimen and chamber, respectively. Thin frozen sections were deposited on top of the polymeric film substrate and allowed to dry under vacuum. Tumor tissues were fixed with formaldehyde and paraffin-embedded prior to microtomy (MICROM HM 340 E, Thermo Scientific) sectioning at 8 μm. Paraffin-embedded sections were transferred into a heated (40°C) water bath to avoid shrinkage, and the polymeric film substrate was immersed (5 s) in the bath to recover the slice. Samples were then analyzed by LA-ICP MS without any further processing. The storage in a box of such a sample did not exceed 10 days. All patients provided written informed consent before the procedure, and the study was approved in the institutional review board in accordance with the ethical standards of the Helsinki Declaration of 1975.

# Determination of elemental content by ICP MS

*Kidney standards* - The frozen kidney standard in Eppendorf tube was cut in six 2-3 mm sections (approximately 150 mg each) to determine elemental concentrations in fresh and dried kidney. Half of the sections were quickly frozen under liquid  $N_2$  and placed into a freezedrying system until complete dryness (water content of ca. 75%). Then, the dried half and the other half of the fresh kidney section were mineralized using 400 μl 69% HNO<sub>3</sub> and heated at 80°C in the oven for 265 min. Then, digests were accordingly diluted with milli-Q water to reduce the HNO<sub>3</sub> content to 2% before analysis by ICP MS.

*Polymeric films* – For each concentration of metal-spiked polymeric films (calibrants for LA-ICP MS), over 10 substrates, 5 substrates (glass slide + polymer film) (n=4/condition and 5 additional blanks (glass slides)) were individually placed into a plastic flask containing 5 mL of 2% HNO<sub>3</sub> solution overnight to allow for quantitative polymer film solubilization and metals released into the acid solution before ICP MS analysis. Each left substrate per condition was used as a control for LA-ICP MS analysis. Glass slides were washed into two successive 2% HNO<sub>3</sub> baths to ensure complete metals release into solution. Consecutively, glass slides were dried under a laminar hood, and the absence of metal spiked at the glass slide surface was verified by running several transects (n= 3) by LA-ICP MS analysis over the glass slide surface.

The polymeric films of the 10 substrates were previously weighted using the high-precision

balance as described above.

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221 ICP MS analysis was performed with an Agilent 7700x model (Agilent Technologies, Tokyo,

Japan). The instrument was tuned daily using a 1 μg L-1 solution containing Y, Li, Tl, Ce, and Ba

in 2% HNO<sub>3</sub>. ICP MS was operated using a micromist nebulizer, Ni cones and under octopole

reaction cell with He gas to lower interferences on <sup>31</sup>P, <sup>63</sup>Cu, <sup>65</sup>Cu, <sup>64</sup>Zn, <sup>64</sup>Zn, <sup>57</sup>Fe, <sup>78</sup>Se and

<sup>80</sup>Se. Element quantification in solution was performed with internal calibration in the range

0 to 50 μg L<sup>-1</sup> for major elements (0-10 μg L<sup>-1</sup> for minor), and standard addition was performed

for element quantification in kidney samples and polymeric films.

## **LA-ICP MS imaging experiment**

A 7700s ICP MS detector (Agilent Technologies, Tokyo, Japan) mounted with platinum cones and S-lenses was used. Preliminary optimization of detection parameters was conducted under the liquid sample introduction mode by means of the 1 μg L<sup>-1</sup> Y, Li, Tl, Ce, 2% HNO<sub>3</sub> tuning solution to ensure maximum sensitivity and optimal mass calibration. A laser ablation system (NWR213, ESI, Freemont, CA, USA) was then coupled to the ICP MS. The ablated particles of sample were transported from the TV2 ablation cell with 800 mL min<sup>-1</sup> of He mixed by means of a T-piece with Ar gas (ca 1.1 mL min<sup>-1</sup>) prior to introduction to the dry ICP source. Under coupling mode, carrier gas and ionic lenses voltages were finely tuned using a NIST 612

glass standard (NIST, Washington, USA) for a U/Th moving closer to the unit and to maximize

sensitivity. ICP MS was used in collision cell mode with He (11 mL min<sup>-1</sup>) as the

collision/reaction gas to allow for monitoring of <sup>13</sup>C, <sup>31</sup>P, <sup>57</sup>Fe, <sup>59</sup>Co, <sup>60</sup>Ni, <sup>63</sup>Cu, <sup>64</sup>Zn, <sup>65</sup>Cu, <sup>66</sup>Zn,

 $^{78}$ Se,  $^{80}$ Se,  $^{98}$ Mo,  $^{103}$ Rh,  $^{111}$ Cd,  $^{115}$ In,  $^{193}$ Ir,  $^{194}$ Pt,  $^{195}$ Pt,  $^{200}$ Hg and  $^{202}$ Hg with 0.08 s integration

time except for  $^{13}$ C (0.01 s), yielding a sampling time of 1.567 s.

Imaging experiments were performed by sampling of subsequent line scans defined over the sample surface (metal-spiked film standards and biological tissues (kidney, ovary and tumoral tissue) and operating at 20 Hz laser shot repetition rate. For kidney standards and ovary samples, a laser beam size of 75 x 75  $\mu$ m, a fluency of 3.5 J cm<sup>-2</sup>, a scan speed of 45  $\mu$ m s<sup>-1</sup> per duty cycle of ICP MS acquisition and a 125  $\mu$ m distance between each consecutive line were used to obtain the desired spatial resolution and signal intensity. For smaller sized samples, such as tumoral tissue (ca 0.5-0.7 cm<sup>2</sup>), these parameters were 50 x 50  $\mu$ m, 4.2 J cm<sup>-2</sup>, 35  $\mu$ m s<sup>-1</sup> and 100  $\mu$ m, respectively. For each sample line scanned, an ICP MS acquisition file was

 $s^{\text{-}1}$  and 100  $\mu\text{m}$ , respectively. For each sample line scanned, an ICP MS acquisition file was

250 generated under a csv format.

#### LA-ICP MS calibration and quantitative element imaging

Metal-spiked polymeric standards with theoretical concentrations of 0, 2.5, 10, 100 and 400  $\mu g \, g^{-1}$  (as major element) were scanned according to the conditions used for sample imaging. For each standard, a set of 3 consecutive lines (ca 1.5-2 cm long) were scanned. Each ion intensity was normalized by the respective internal standard intensity, and an average value was calculated from the 3 scans for each standard. For each element, a linear regression was obtained from these average values as a function of the respective standard concentrations. These linear regressions were used to convert raw data (element intensity to internal standard intensity ratio) from sample analysis to elemental concentrations ( $\mu g \, g^{-1}$ ) by taking into account the difference in ablated volume (thickness of polymeric film and tissue thin-section) as well as the densities of the polymeric film and tissue.

## Data processing

An imaging experiment for a single sample consisted of n csv files, which were then concatenated. Then, a homemade program under Python was used to generate an elemental spectrum and images of element concentrations per pixel using a color code (png image format).

## **Results and discussion**

The method relies on the use of polymeric films doped with metallic (including internal standards) solutions for calibrants or internal standards for samples. Therefore, elements present in biological tissues deposited on top of the polymeric film are quantified by internal calibration after complete ablation. For reliable quantification, calibrants must be physically and chemically characterized.

#### Film characterization

The main idea was to develop a method using a hydrosoluble polymer to extend the methodology to a larger panel of metals likely to be encountered in living organisms but also to avoid the use of organic solvents and thus reduce chemical waste treatment for a more environmentally friendly and durable methodology. Furthermore, successful quantification using polymer film calibrant strictly relies on two keys parameters: the chemical purity in terms of metals of the polymer molecule and the homogeneity of the film produced.

Chitosan is a natural polysaccharide produced from chitin of skeletons of crustaceans and cephalopods and the cell wall of mushrooms. It is soluble in slightly acidified water and therefore was considered as a potential candidate in a first attempt to produce the polymeric film. However, chemical analysis of a homemade chitosan compound (Mw 20 000 g mol<sup>-1</sup>) using ICP MS revealed high concentrations of Fe (ca 300 000 μg kg<sup>-1</sup>), Ni (ca 3 000 μg kg<sup>-1</sup>), Zn (750  $\mu g \ kg^{-1}$ ), Cu (400  $\mu g \ kg^{-1}$ ), Mo (100  $\mu g \ kg^{-1}$ ) and Pt (1  $\mu g \ kg^{-1}$ ), preventing its use for the purpose of this study as it would increase the chemical background during analysis. A set of 6 commercially available dextran polysaccharides that were 100% water soluble and with Mw ranging from 50 000 to 500 000 g mol<sup>-1</sup> were then investigated. The determination of their chemical composition by ICP MS allowed the selection of one dextran molecule (Mw 500 000 g mol<sup>-1</sup>) with the lowest concentrations of elements of interest (P  $\sim$  73 000  $\mu$ g kg<sup>-1</sup>, Fe  $\sim$  2800  $μg kg^{-1}$ ; Cu  $^{\sim}$  180  $μg kg^{-1}$ ; Ni  $^{\sim}$  148  $μg kg^{-1}$ ; Zn  $^{\sim}$  77  $μg kg^{-1}$ ; Mo  $^{\sim}$  51  $μg kg^{-1}$ ; Se  $^{\sim}$  21  $μg kg^{-1}$ ; Co  $\sim$  8 µg kg<sup>-1</sup>; Cd  $\sim$  0.2 µg kg<sup>-1</sup>; Hg  $\sim$  0.2 µg kg<sup>-1</sup> and Pt<LOD of ca. 0.02 µg kg<sup>-1</sup>). Polymer concentration and spin-coating parameters (speed and acceleration speed, rpm; duration, s; N<sub>2</sub> gas stream), which impact film thickness and roughness, were optimized to first obtain a visible homogenized film. Although these dextran films are transparent, defaults were apparent when the wrong spin-coating conditions were used. Of note is the importance of the cleanness and purity of the glass slide (dust deposits must be meticulously removed, and chemical content must be assessed). Once default-free polymer films were obtained (i.e., at 60 mg mL<sup>-1</sup>; 4000 or 4500 rpm for 1 s), the homogeneity of the polymer film was assessed using LA-ICP MS by scanning in imaging mode a large surface of several polymer substrates (n=2 @ 4000 rpm and n=2 @ 4500 rpm) used for sample deposits and monitoring internal standard (In, Rh, and Ir) signals. Over the 4 film substrates analyzed, internal standards signal stability was in the range of 8 to 16% and was slightly lower for film substrates produced at 4000 rpm (RSD ~ 12%) compared with film obtained at 4500 rpm (RSD ~ 9%). For the latter condition, RSD between the 2 film substrates was better for In (10%) compared with Ir (12%) and Rh (14%). Therefore, spin-coating at 4500 rpm provided homogenized film substrates. Elemental concentrations in film substrate were determined after the spin-coating step, which involves evaporation of the solution, leading to a metal preconcentration. For film calibrants at 0, 2.5, 10, 50, 200 and 400 µg g<sup>-1</sup> (expressed for major elements), concentrations determined by total analysis of the solubilized film are presented in Table 1 for all elements present in polymer films taking into account the measured mass of each film. Compared to

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the theoretical concentration expected in the film, preconcentration factors (PF) for each element were determined and varied between 1 and 53 (average value by element) depending on the element. The low Hg concentration suggested a loss of Hg during the evaporation process such that no loss during filtration was observed. In addition to Hg, P and Fe exhibited the highest variabilities (respectively 74 and 112% RSD) in terms of preconcentration factors most likely due to measurement bias in the blank and at low concentrations, reflecting contamination issues (i.e., PF= 63 and 158 for P and Fe, respectively, at 2.5 µg g<sup>-1</sup>). For the others elements, i.e., Ni, Co, Cu, Zn, Se, Mo, Cd, Pt and the three internal standards (In, Rh, and Ir), preconcentration factors were in the range of 12 to 22 with RSD between 4% (Pt) and 34% (Ni). These preconcentration factors are in the same order of magnitude as that reported by Austin et al [25], who determined a 10-fold increase for Cu and Zn. The mass of the film varied between 99 and 126  $\mu g$  from the blank to the 400- $\mu g$  g<sup>-1</sup> standard. This finding was consistent with the thickness of 40 nm (variance 1%) and 80 nm measured for the blank and highest concentrated calibrant, respectively. This thickness difference is potentially due to a slight acid content difference (from the solution of elemental standards) between the blank and 400 µg g<sup>-1</sup> standard. This thickness determined approaches values (9 ±1 nm) obtained by means of the sputtering method [23, 29]. Thinner film is more preferable compared with that previously reported [24] as surface variation would be insignificant given the complete ablation of the film and μm spatial resolution of the laser. Thinner films would also minimize the contribution of the chemical composition of the polymer to the blank background. Therefore, quantification was performed based on preconcentration factors determined for each element and the 40-nm thickness (density was therefore determined to 3.6982) of the film, which is the substrate used for sample deposition.

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#### Calibration and normalization

The purpose of this study was to identify the most appropriate internal standard (supposedly absent from biological samples) for each element taking into account its mass and FIP (eV). Considering these two latter parameters, <sup>115</sup>In (5.79 eV), <sup>103</sup>Rh (7.46 eV), and <sup>192</sup>Ir (8.97 eV) were investigated as internal standards as they cover the large range of mass and FIP of the elements of interest. The responses of all elements under study were plotted as a function of their concentration and the internal standard considered (**Figure 1**). Good correlation

coefficients (0.995 - 1) were obtained except for P (<0.5), Fe (<0.2) and Hg (<0.09). For Hg, as expected from preconcentration factor determination, this lack of linearity originates from Hg loss during spin-coating. Several attempts to stabilize Hg in solution using an excess of chloride ions or cysteine ligand have been made but were not successful and require further research. The poor correlation coefficients of P and Fe calibration result from the presence of these elements in the polymer as it is observed in the blank and the lowest concentrations (2.5 and 10 μg g<sup>-1</sup>). Furthermore, the presence of P and Fe in the glass slide (which is also ablated at the surface (<10 µm depth)) might also contribute to increase the background, thereby degrading the linear correlation at low concentrations. For these reasons, these three elements have been excluded from the following discussion. Regardless of the element, calibration curves show that In is the internal standard with the highest sensitivity when normalization was applied. Compared to In calibration, reductions of approximately 1.5 and 3.4 were observed when element responses were normalized with Rh and Ir, respectively, and regardless of the element. This finding prompted us to first conclude that In is the best internal standard for highest sensitivity, for all elements and independently of the mass and FIP of the elements to be quantified. This feature is most likely due to its lower

FIP leading to a better response than Rh and Ir in ICP MS. At constant sensitivity parameters

(mainly ICP MS cones status), the graphs revealed good repeatability (SD < 1.1) for all element

concentrations and regardless of the internal standard for the same calibration analyzed thrice

weekly. Our data also demonstrated the stability and linearity of a calibration over one year,

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# Influence of ICP conditions

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ICP conditions were optimized for robust plasma and in an attempt to reduce potential differential elemental responses due to ablation of different materials (film, tissue (heterogeneous itself)), which might lead to incorrect quantification due to the introduction of organic material into the plasma (carbon effect). First, adjustment of ICP power revealed that the analysis could be run at 1400 until 1500 W, which favor sensitivity and signal stability (supporting text, Figures 2-SI and 3-SI).

The addition of  $O_2$  into the ICP was investigated to reduce carbon deposits and possible matrix effects. First, the impact of  $O_2$  addition was measured on calibration standards for all elements (**Figure 2a and Figure 4-SI**). When  $O_2$  is added, three groups of elements could be

differentiated according to their responses and the internal standard used for normalization, i.e., (1) a group of elements benefiting from sensitivity improvement regardless of the internal standard used; (2) a group of mixed elements with increased/stable or stable/decreased sensitivity as a function of the internal standard used; and (3) a group of elements suffering from decreased sensitivity with all internal standards. In the first group, O<sub>2</sub> addition enhanced the sensitivity of Co and Mo normalized with In, Rh or Ir. From the second group, the sensitivity of Ni and Cu increased when normalized with In and Ir, whereas sensitivity tends to decrease with Rh normalization. From the same group, the sensitivity of Pt is maintained with Ir as internal standard but tends to decrease when Pt is normalized with In and Rh. In the last group, the sensitivities of Zn, Se and Cd decrease systematically with all internal standards. The tendency of these 3 groups is likely to follow the FIP of these elements with the lowest FIP for Co (7.88 eV) and Mo (7.09 eV) followed by Ni (7.64 eV)-Cu (7.73 eV) and Pt (8.96 eV) at the limit of the 3<sup>rd</sup> group and finally the triplet of Zn (9.39 eV)-Se (9.75 eV)-Cd (8.99 eV). Therefore, if  $O_2$  is added, Ir offering close mass and FIP similar to those of Pt and should be considered as the most appropriate to maintain analytical performances and reduce possible matrix effect. Definitively, the use of O<sub>2</sub> in the ICP source must be avoided for the LA-ICP MS analysis of Zn, Se and Cd unless more convenient internal standards (with closer mass and FIP) can be identified that are not naturally occurring in biological samples. The effect of O<sub>2</sub> addition for potential matrix effect reduction was studied during ablation of biological tissue (homogenized metal-spiked kidney), which is a source of carbon introduced in the ICP. Figure 2b and 2c shows elemental images of kidney thin-sections without the addition of O2 into the plasma (upper half part of the image) and with the addition of O2 (lower half part). A representative image illustrates the tendency for each group of elements as defined above without internal standard normalization (2b and with normalization (2c). For the first group of elements, O2 addition enhanced Co signal intensity even in the presence of organic material. However, when Rh normalization was applied, no difference in relative responses were observed between the two conditions of plasma, indicating that these conditions were free of matrix effects. From the second group of elements, Pt exhibited increased signal intensity with O2 addition. In contrast, In normalization tended to decrease the discrepancy of the Pt relative response over the thin-section regardless of plasma conditions, although Pt/In responses were still slightly more pronounced at the upper part of the section. With Ir normalization of Pt signal, no differential response could be observed

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between the two statuses of  $O_2$  plasma. Zn images that represent the thin section of the last group demonstrated strong loss of sensitivity under  $O_2$  plasma that internal standard normalization did not recover, but Ir reduced the most, the differential response between the two plasma conditions. Together with results of the  $O_2$  effects on calibration curves, these data demonstrate that the response of low FIP elements can be more easily corrected even in the presence of organic material and  $O_2$  in the plasma, whereas high FIP elements are dramatically affected by the presence of  $O_2$  and carbon load in the plasma that cannot be compensated for by normalization with any internal standard considered in this study. The intermediate group of elements (Cu, Ni and Pt) requires careful attention in the selection of the most suitable internal standard based on the plasma condition.

## Influence of laser energy and tissue thickness

Thin sections of homogenized metal-spiked kidney (the matrix matched standard) were used to assess the effect of laser energy delivered and tissue thickness (10 and 30 µm) on elements response with and without O<sub>2</sub> in the ICP plasma. In most bioimaging studies, laser energy ranges in the low J cm<sup>-2</sup>. The two levels of energy (3.3 and 5.5 J cm<sup>-2</sup>) used in this study were chosen to enable complete ablation of the tissue and limit glass slide surface ablation as much as possible (Figure 3). On a 10-μm section analysis, these two levels of energy did not reveal notable differences for Co, Pt and Zn relative responses, which are representative elements of the three previously defined groups. On the other hand, when O<sub>2</sub> was added to the plasma, all relative responses of Pt and Zn were decreased by at least 2-fold with a 1.6-fold energy increase, whereas relative responses of Co were much less affected by O2 while increasing energy level. Therefore, given that increasing the laser energy did not favor sensitivity enhancement and that a white deposit appeared on the skimmer cone after several days of imaging runs, the lowest laser energy was chosen to reduce (in depth) glass ablation and thus minimize white deposits originating from glass silicon oxide. The relative response of the elements was compared for the two thicknesses of the matrixmatched standard using 3.3 J cm<sup>-2</sup> fluency. In the first group, Co relative responses increased proportionally (2.6 to 3-fold) with increases in section thickness for the majority of conditions (In, Rh, Ir, +/- O<sub>2</sub>). Mo behaved similarly to Co with or without addition O<sub>2</sub> to plasma. Considering Pt from the 2<sup>nd</sup> group, the lowest correlation (3.7-fold proportionality) was

obtained when the Pt signal was normalized by Rh in the absence of the addition of  $O_2$  to plasma (2.9-fold for In and Ir normalizations). Under  $O_2$ , the proportionality between tissue thickness increases and Pt normalized responses was slightly greater than with 3.6- to 4-fold relative response increases (as noted for Cu and Ni). Concerning the  $3^{rd}$  group, the normalized responses of Zn were on average poorly proportional (min-max: 1.7- to 2.9-fold) to the increase in tissue thickness. The closest values (2.87 and 2.95-fold) were obtained under Rh normalization without and with  $O_2$  as observed for Se and Cd. Therefore, for this group of elements, thinner sections (i.e.,  $10~\mu m$ ) are preferred for LA-ICP MS analysis without the addition of  $O_2$  to plasma.

In summary, for the majority of these elements, with the exception of those in the third group (Zn, Se, and Cd), the carbon load from biological tissue does not hamper LA-ICP MS analysis on 30- $\mu m$  sections given correct selection of internal standard for signal normalization.

Overall, this finding underlines the consistent behavior of elements within each defined group,

revealing once more that FIP is a key parameter to cope with the differences in carbon load in

the plasma and regardless of its composition  $(-/+ O_2)$ .

# LA-ICP MS methodology validation

The reliability of the developed methodology was assessed using the matrix-matched standard, and homogenized kidney spiked with different amounts of metals (i.e.,  $2.5 \ \mu g \ g^{-1}$ ,  $50 \ \mu g \ g^{-1}$  and  $250 \ \mu g \ g^{-1}$ ) was cryosectioned at 8, 10, 14 and 30  $\mu$ m (n=3/thickness conditions deposited in 3 different films). For each condition, 5 scans were acquired per kidney thin sections. In parallel, metal concentrations in each kidney standard (n= 3/conditions) were determined by conventional ICP MS analysis both in fresh and dried standards. Limits of detection of the LA-ICP MS methodology (under laser sampling conditions used) were determined according to the linear regression obtained by means of the polymer film standards (i.e., LODs = 3SD of the blanks/slope) and by taking into account the densities of the polymer film (d film = 3.6982 and d tissues=1), the thickness of the film (40 nm) and tissue section (8-30  $\mu$ m). Using In as an internal standard for an 8- $\mu$ m thin section samples (the thinnest section experimented), the limits of detections were in the  $\mu$ g  $\mu$ g range and lower with Co (0.1  $\mu$ g  $\mu$ g range and lower with Co (0.1  $\mu$ g  $\mu$ g range and Se (1  $\mu$ g  $\mu$ g range are in the expected range as LODs are usually reported at the 10-4 to 10-1  $\mu$ g  $\mu$ g range and lower will be a same in the expected range as LODs are usually reported at the 10-4 to 10-1  $\mu$ g  $\mu$ g range and lower will be a same in the expected range as LODs are usually reported at the 10-4 to 10-1  $\mu$ g  $\mu$ g range and lower will be a same range as LODs are usually reported at the 10-4 to 10-1  $\mu$ g  $\mu$ g range and lower will be a same range as LODs are usually reported at the 10-4 to 10-1  $\mu$ g  $\mu$ g range and lower will be a same range as LODs are usually reported at the 10-4 to 10-1  $\mu$ g  $\mu$ g range and lower will be a same range as LODs are usually reported at the 10-4 to 10-1  $\mu$ g  $\mu$ g range and lo

level [16]. LODs of some hundreds of ng g<sup>-1</sup> and 0.08 μg g<sup>-1</sup> were reported for Cu [23] and Pt 475 [19], respectively, and are close to our values. However, LODs are not reported for Co, Mo, 476 Cd, Ni, Zn and Se for LA-ICP MS bioimaging analysis. 477 Figure 4 presents LA-ICP MS quantification data obtained by means of metallic film calibration 478 using In as internal standard, 3.3 J cm<sup>-2</sup> fluency and no O<sub>2</sub> in the ICP plasma. The LA-ICP MS 479 data are bracketed by concentration values in fresh and dry weight samples for each 480 481 condition. First, for all elements and thickness sections, concentrations determined by LA-ICP MS are located between the values expressed as fresh weight (FW) and as dry weight (DW). 482 483 Of note, higher concentrations were obtained for 8-µm sections compared with 10-µm and 30-µm sections. This general trend is observed for all elements and is particularly pronounced 484 485 for the highest concentration of kidney standard where an average 2.8-fold decrease in 486 concentration is observed between the 8- and 30-µm sections. For the two other kidney 487 standards, this effect is reduced for all elements with an average 1.7- and 1.2-fold decrease at 2.5 and 50 µg g<sup>-1</sup>, respectively. For all kidney standard concentrations, Pt is the element least 488 489 affected by this decrease with 22% RSD (as Cd), 2% RSD (as Mo), and 64% RSD (as Cd, Mo and Se) at 2.5  $\mu$ g g<sup>-1</sup>, 50  $\mu$ g g<sup>-1</sup> and 250  $\mu$ g g<sup>-1</sup>, respectively. In contrast, Ni is the most affected with 490 the highest concentration discrepancies between the 8-μm and 30-μm sections (69% RSD at 491 2.5 μg g<sup>-1</sup>; 71% RSD at 250 μg g<sup>-1</sup>). However, these decreases does not seem to be mass 492 493 element dependent. Additionally, at 250 μg g<sup>-1</sup>, quantification values obtained for an 8-μm sections are closer to 494 495 the DW concentration values than the FW values, especially for Co, Ni, Se and Mo with an 496 approximately 37% bias. While Cu and Zn concentrations are approximately one-half of the 497 value of DW, Zn and Cd values are one-half of an order of magnitude below the DW values. Consequently, for this concentration standard, the bias of 30-µm section values with the FW 498 499 values are in the range of 5% (Pt) to 29% (Co) for most elements with an average bias of 11% 500 without taking into account Se (31%), Cd (49%) and Zn (56%). For the 50 μg g<sup>-1</sup> kidney standard, the average bias for the Co, Ni, Cu, Mo and Pt concentration values determined is 7%, and bias 501 502 for Zn, Se and Cd is increased (42-92%). At 2.5 μg g<sup>-1</sup>, the bias of the 30-μm section 503 concentrations to the FW concentration values increases (>30%), which could be explained by 504 the fact that the concentrations determined are moving closer to the LOQs. Nevertheless, the Pt concentration is still the closest to the FW value (29% bias). Using Ru and Y as internal 505 506 standards for normalization in 20-µm sections and comparing concentrations determined by LA-ICP MS and ICP MS, a slightly higher bias (35%) was obtained for Cu (approximately the same level as 2.5  $\mu g$  g<sup>-1</sup>), but the bias tended to be much lower for Zn (11% RSD at a concentration of 8  $\mu g$  g<sup>-1</sup>) [24], in contrast to our study. To summarize, Pt is the least problematic element in contrast to Zn (followed by Cd), which exhibits completely different behavior and appears to be difficult to properly quantify. Given that the kidney sample is directly analyzed without a drying step, one could expect a value closer to that expressed as fresh weight. This notion is likely to reflect the drying of the biological sample due to high He flow in the ablation cell during the imaging run or during the storage of the section before analysis. Thus, thinner sections would dry faster than the thicker

during the ablation process. The use of a cryocell would be beneficial to better control this parameter in fresh biological samples analysis. Another alternative would be to dry (under light vacuum for example) the biological section immediately after its deposition on the

polymer film without damaging the film substrate. In summary, the values determined by LA-

ICP MS fall within intervals of fresh-dry weight concentrations, ensuring the validity of our

methodology. In addition, the bias with the FW value can be assessed according to the

thickness of the section under such analytical conditions.

# Application to human clinical samples (ovaries and tumoral tissues)

The methodology was applied to map elements in two types of samples from different patients. The standard protocol for woman treated by Hyperthermic Intraoperative Peritoneal Chemotherapy (HIPEC) for a peritoneal carcinomatosis is the removal of both ovaries. The HIPEC protocol involves the intraperitoneal application of a circulating bath of a chemotherapeutic solution heated at 42°C for 30 minutes. Platinum salts (cisplatin or oxaliplatin) are the onco-drugs most commonly used in this protocol. We have access to peritoneal tumoral nodules and ovaries resected before and after HIPEC. **Figure 5** shows the quantitative distribution of Pt derived from the drug, endogenous elements (Cu, Zn and Mo) and Cd. Of note, the qualitative and quantitative distributions were identical for all these elements, confirming that polymeric film and internal standards were homogeneous even after the drying step under a light vacuum. The highest concentrations (15-20  $\mu$ g g<sup>-1</sup>) of Pt were mostly determined at the periphery of the ovary given the type of treatment, which facilitated direct contact of the drug with the tissues and organs. In addition, approximately 5-10  $\mu$ g g<sup>-1</sup> of Pt spots could be quantified in some central parts of the ovary, which potentially

correspond to residual Pt from prior systemic chemotherapy. Zn is homogeneously distributed in the ovary at approximately 10  $\mu g$  g<sup>-1</sup>. However, considering the quite important bias determined from the kidney standard, this value is certainly underestimated. Cu is also homogeneously distributed at lower concentrations, i.e., 2.5-5  $\mu g$  g<sup>-1</sup>. Mo, an element involved in some biological processes, was also detected at quantifiable (0.1  $\mu g$  g<sup>-1</sup>) levels but close to background level. The presence of 1-2  $\mu g$  g<sup>-1</sup> Cd, which is quite unexpected, leaves us with unanswered questions about its origin and its potential toxic effects on ovary functionality [30].

This methodology was also applied to paraffin-embedded biological tissues, which is the most common protocol used for the preparation of tissue sections for the creation of clinical tissue banks. As shown in **Figure 6**, this preparation involving paraffin and polymer film immersion in a warm bath did not hamper quantification given that internal standards remained homogeneous after this entire process. Quantitative maps could be obtained for Pt, Cu, Zn and Ni among all the set of elements monitored. The others were under LODs. Although Pt was primarily found in the tissues surrounding the ovary, Pt concentrations were increased and in the range of 30-60 µg g<sup>-1</sup>. In contrast to the ovary sample, the highest concentrations (30-90 μg g<sup>-1</sup>) of Cu and Zn mere mostly noted at the periphery of the tumor tissue where some (10 g<sup>-1</sup>) Ni was also quantified. μg This bioimaging analysis allowed to assess the penetration of Pt drug into tissues of patients, thus enabling comparisons of clinical protocols and their efficacy in a long term study. Finally, the quantification of endogenous elements (Cu, Zn, Ni, Mo) is also of paramount importance

as it could enable us to follow the potential impact of the drug on these essential elements.

Conclusion

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Given that most bioimaging studies report quantitative mapping for one or two elements, the developed methodology based on the use of a water-soluble polymer to produce metal-spiked polymer film standards allows the simultaneous quantitative mapping of a set of eight elements in biological samples for the first time. The 8 elements exhibited different behaviors as a function of the internal standards used for normalization and experimental conditions for analysis. Elements with low and medium first ionization potential (<9 eV) can be easily quantified in 8- to 30-μm tissue sections. For higher FIP elements, which are more subjected to matrix effects, quantification is limited to <10-μm sections. The methodology was validated

- on kidney matrix standards, and LODs in the range of  $0.1 1 \mu g g^{-1}$  were determined for 8- $\mu$ m thin-sections. This quantitative bioimaging methodology was successfully applied to two
- 573 different samples following different sample preparations (fresh-frozen and fixed-embedded
- 574 sample) without impacting polymer film homogeneity, thus ensuring the use of this
- 575 methodology for clinical tissue banking.

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- 688 Table and figures caption
- **Table 1.** Chemical composition, weight and thickness of polymer film standards prepared by
- 690 spin-coating. Concentration ranges (0-400 μg g<sup>-1</sup> and 0-80 μg g<sup>-1</sup>) are reported for major
- 691 elements (P, Fe, Cu, Zn, and Pt) and minor (Ni, Co, Se, Mo, Cd, and Hg) elements, respectively.
- 692 Rh, In and Ir are internal standards.
- 693 Figure 1. LA-ICP MS element calibration curves obtained as a function of internal standard
- 694 (Normalized Intensity on the Y-axis) after spin-coating and LA-ICP MS analysis.

696 Figure 2. Influence of O<sub>2</sub> addition to ICP plasma on A) Co, Pt and Zn sensitivity according to internal standard (Y-axis: Normalized Intensity) and element responses normalized C) or not 697 B) during analysis of the 10-µm sections of matrix-matched kidney standard. 698 Figure 3. Influence of laser energy and section thickness on element response according to the 699 three internal standard normalizations. Unless specified on the X-axis, experiments are run at 700 3.3 j cm<sup>-2</sup> fluency. 701 702 Figure 4. Methodology validation with determination of element concentrations by LA-ICP MS imaging in matrix-matched kidney standards at 2.5, 50 and 250 µg g<sup>-1</sup> (expressed as major 703 704 element concentrations) in 8-, 10-, 14- and 30-μm sections with In normalization. 705 Concentrations in fresh weight and dry weigh kidney standards were also determined by ICPMS following acid mineralization of the standards. 706 707 Figure 5. Example of quantitative mapping of elements in an 8-μm section of a fresh frozen ovary from patients under oxaliplatin treatment with a photography before LA-ICP MS 708 709 imaging.. 710 Figure 6. Example of quantitative mapping of elements in an 8-μm section of a fixed and resin-711 embedded tumor tissue from patients undergoing oxaliplatin treatment with a photography 712 before LA-ICP MS imaging.

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