

Engineering Optically Switchable Transistors with Improved Performance by Controlling Interactions of Diarylethenes in Polymer Matrices

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ABSTRACT: The integration of photochromic molecules into semiconducting polymer matrices *via* blending has recently attracted a great deal of attention, as it provides the means to reversibly modulate the output signal of electronic devices by using light as a remote control. However, the structural and electronic interactions between photochromic molecules and semiconducting polymers is far from being fully understood. Here we perform a comparative investigation by combining two photochromic diarylethene moieties possessing similar energy levels yet different propensity to aggregate with five prototypical polymer semiconductors exhibiting different energy levels and structural order, ranging from amorphous to semicrystalline. Our in-depth photochemical, structural, morphological and electrical characterization reveals that the photoresponsive behavior of thin-film transistors (TFTs) including polymer/diarylethenes blends as the active layer is governed by a complex interplay between the relative position of the energy levels and the polymer matrix microstructure. By matching the energy levels and optimizing the molecular packing, high-performance optically switchable organic thin-film transistors were fabricated. These findings represent a major step forward in the fabrication of light-responsive organic devices.

INTRODUCTION

Photochromic molecules have garnered increasing attention as building blocks for applications in artificial molecular machines, data storage, imaging, and sensing.¹ Among photochromic molecules, diarylethenes (DAEs) are extremely interesting since they combine high fatigue resistance and thermal stability of the ring-open and ringclosed configurations.² Moreover, the two isomers possess significantly different energy levels of their frontier molecular orbitals, i.e. the highest occupied molecular orbital (HOMO) and the lowest unoccupied molecular orbital (LUMO). The associated ability to control this property change has been key to develop multifunctional organic devices, which just as conventional organic devices, are based on materials that can be solution-processed and, as such, are suited for roll-to-roll printing on flexible substrates at large scale.³ DAEs represent ideal systems to be combined with organic/polymer semiconductors in order to reversibly tune the output current, conductivity, or electroluminescence upon irradiation at distinct wavelengths.⁴ The use of external light stimuli provides precise temporal and spatial control on the DAEs embedded in electronic material systems and devices thereof.

A simple-yet-effective approach to introduce optical switching functions into organic devices consists in the di-

rect blending of DAEs with organic semiconductors as active materials in the devices. Optically switchable organic thin-film transistors (OTFTs) have been fabricated by blending DAEs with different semiconductor matrices including poly(3-hexyl-thiophene) (P3HT), 2,7-dialkyl-benzothieno(3,2-b)benzothiophene (BTBT) or fullerenes.⁵ The output current of these OTFTs could be controlled by choosing specific irradiation wavelengths. Moreover, polymer/DAE blends have been proven to be the key to fabricate flexible non-volatile optical memories exceeding 8-bit storage capacity.⁶

In order to design and optimize the materials for highperformance light-responsive multifunctional optoelectronic devices, it is highly desirable to gain an in-depth understanding over the relationship between polymer structure, photoswitching performance of DAEs, and the resulting electrical properties of devices using such bi-component films as the active layer. More specifically, fundamental questions need to be addressed, such as: (i) how the photoisomerization process of a given DAE molecule is affected by the surrounding semiconducting polymer matrix; (ii) how the polymer packing itself is affected by the presence of DAEs, and (iii) what are the principles governing device mobility and optical switching in the devices based on the composition and processing of polymer/DAE binary blends.

To address these open questions, we synthesized two DAEs, i.e. DAE_Me and DAE_^tBu (Figure 1B), possessing very similar HOMO levels (-5.6 eV and -4.8 eV of their open and

closed isomers, respectively, Figure 1D) yet different side chains, the latter ruling the supramolecular interaction with the semiconducting polymers. In particular, the steric hindrance brought into play by the four bulky tert-butyl substituents of DAE_'Bu leads to a lower tendency for both aggregation and association with the polymer matrix when compared to DAE_Me, which only has two pending methyl substituents. Moreover, five archetypal high-mobility ptype polymers (Figure 1A) were chosen that feature different HOMO levels (Figure 1D) and propensity to undergo self-assembly forming well-distinct morphologies. In parpoly(9,9-dioctylfluorene-alt-benzothiadiazole ticular. and poly(9,9-dioctylfluorene-*alt*-bithiophene) (F8BT)7 (F8T2)⁸ form completely amorphous structures, whereas thermally untreated P3HT organizes into poorly ordered films,⁹ and diketopyrrolopyrrole thieno[3,2-b]thiophene copolymer (DPPT-TT)¹⁰ and poly(4-decyltetradecylisoindigo-*alt*-dithiophene) (IIDDT-C3)¹¹ are known to generate highly ordered semi-crystalline structures when thermally annealed. The photochemical properties of the DAEs embedded in the polymer matrices were studied by UV/visible absorption spectroscopy. The morphology of the polymer/DAE films was investigated by Atomic Force Microscopy (AFM), whereas the structure was elucidated via grazing incidence wide-angle X-ray scattering (GIWAXS) measurements. OTFTs devices (Figure 1C) were fabricated and tested in order to characterize charge transport and light response of these ten polymer/DAE blends. By correlating the intrinsic electronic structure (i.e., energy levels) and matrix microstructure resulting from supramolecular interactions with the photochemical and electrical performance of polymer/DAE blends, we have gained insight into how to maintain the strong electrical performance of the semiconducting polymers while aptly exploiting DAE photoswitching to fabricate efficient light-tunable organic devices.

RESULTS

Photochemical properties. The photoswitching behavior of neat DAEs and their blends with the chosen semiconducting polymers was studied by UV/vis absorption spectroscopy. The open isomers of DAEs undergo ring closure (photocyclization) upon brief UV irradiation (312 nm, 35 s, 2.5 mW/cm²) in 1,2-dichlorobenzene (DCB), leading to formation of a new broad absorbance band between 400 nm and 650 nm for both DAE_Me and DAE_^tBu (Figure S1). The conversion to the closed isomer in the photostationary state (PSS) upon UV irradiation reached 92% for DAE_Me and 90% for DAE_tBu.5b Subsequent irradiation of this PSS with visible light converts the closed isomer back to the open one (photocycloreversion), resulting in recovery of the initial spectra. The photocyclization and photocycloreversion of DAE_Me and DAE_tBu show relatively good fatigue resistance, with < 20% loss of photochromic material over 8 illumination cycles in solution.5b

To gain further insight into the photochemical processes of DAEs when embedded in the polymer matrices, samples were prepared by spin-coating bi-component polymer/DAE solutions at concentrations of 2 mg/mL for the polymer and 0.5 mg/mL for the DAE, providing films with thicknesses between 20 nm and 30 nm. No further thermal annealing treatment was applied, except in the cases of DPPT-TT and its DAE blends, in which the spin-coated films were annealed at 90 °C to improve microstructural ordering.^{10a, 10c} The weight ratio of DAEs in the blends was always 20 wt% in this study, in agreement with previous works which suggested that this percentage results in a good balance between current modulation and device mobility in OTFTs.^{5a}



Figure 1. Chemical structures of semiconducting polymers and DAEs, resulting OTFT devices, and energetics. (A) Chemical structures of semiconducting polymers F8BT, F8T2, P3HT, DPPT-TT, and IIDDT-C3. (B) Molecular structures of photoswitchable DAE_Me and DAE_tBu in their open and closed isomers, which can be interconverted reversibly by irradiating with UV or visible light. (C) Structure of the OTFT device applied. (D) The HOMO energy levels of organic semiconducting polymers and DAEs used. ^{5a, 5b, 10a, 10b, 11b, 12}.

The comparative study on the photocyclization and photocycloreversion of DAEs when embedded in the five chosen polymer matrices was carried out by exploiting the same irradiation power and distance as DAEs in solution, by simply varying irradiation time to confirm that the PSS is

reached in all the polymer/DAE blends. UV irradiation induced an increase in the absorbance between 400 nm and 650 nm in all five polymer matrices (Figure 2, and difference spectra in Figure S2), indicating the formation of the closed isomer. After reaching the PSS, irradiation of the sample with visible light led to recovery of the initial spectra as a result of the DAE ring opening. The reversible photoisomerization was observed in both DAE_Me- and DAE_tBu-containing polymer matrices. The recorded spectra did not show any significant shift of the main absorption band of all five polymers upon blending with the DAE molecules (see Figure S3). The photochemical properties of the DAE derivatives in polymer matrices are summarized in Table 1, including the spectroscopic characteristics of neat DAEs in solution and thin film as well for comparison purposes. The open isomer shows a very small bathochromic shift of λ_{max} (< 7 nm) from solution to polymer matrices, providing evidence for the absence of strong interactions between each polymer and the open DAEs in the ground state. A larger bathochromic shift of $\lambda_{max} > 20$ nm was observed for the closed isomers when embedded in amorphous (F8T2, F8BT) and poorly ordered (P3HT) polymers, thereby suggesting that the electronic structure of the more planar closed isomers is perturbed by the polymer matrices.¹³ Conversely, polymer matrices with increased degree of ordering (IIDDT-C3 and DPPT-TT) induced a somewhat smaller bathochromic shift of_{max} < 17 nm.

The PSS of DAEs in thin films is considerably lower than in solution. The ring-open isomer of DAE adopts an approximate 1:1 mixture of parallel and antiparallel conformations, which can easily interconvert in solution, while only the antiparallel conformer undergoes photocyclization.² However, when DAEs are deposited into films, the parallel conformer of the ring-open DAE is somewhat trapped and not enabled to convert to the photo-active antiparallel isomer, therefore the PSS is reduced as we observed. The PSS of DAEs also shows dependence on the microstructure of the polymer matrices. In the case of DAE Me, the PSS of 67% in IIDDT-C3 and 65% in DPPT-TT decreased to 35% in F8BT, 43% in F8T2 and 40% in P3HT. For DAE_tBu, the PSS of 51% in IIDDT-C3 and 56% in DPPT-TT decreased to 23% in F8BT, 26% in F8T2 and 50% in P3HT. The higher PSS in highly ordered matrices might be attributed to the fact that a more ordered matrix can lead to a higher population of the antiparallel conformation of the ring-open isomer, thus resulting in a higher quantum yield for ring-closure. Furthermore, the PSS of DAE_tBu is lower than the PSS of DAE_Me in all polymer matrices (except in the case of P3HT) despite the similar value in solution phase, possibly due to the more sterically demanding DAE_^tBu, the isomerization requires more space to take place, which is hindered by the interaction with the surrounding matrix. The glass transition temperature of P3HT ($T_g = 12$ °C) is lower than the one of F8T2 and F8BT (70 - 130 °C),^{15c} indicating a more flexible inner space of the amorphous region of P3HT at room temperature, which may determine the moderate effect on the photoisomerization processes of DAE_tBu.



Figure 2. UV/visible absorption spectra of polymer/DAE films and the photostationary state (PSS) upon UV light irradiation (312 nm) and visible irradiation. (A) F8BT/DAE_Me and F8BT/DAE_tBu films. (B) F8T2/DAE_Me and F8T2/DAE_tBu films. (C) P3HT/DAE_Me and P3HT/DAE_tBu films. (D) DPPT-TT/DAE_Me and DPPT-TT/DAE_tBu films. (E) IIDDT-C3/DAE_Me and IIDDT-C3/DAE_tBu films.

In order to quantify the efficiency of the DAEs photoswitching in the polymer matrices, the isomerization quantum yields under UV and visible light irradiation (Φ_{UV} and Φ_{vis}) were determined, and the results are also summarized in Table 1. Φ_{UV} is 50% for DAE_Me and 31% for DAE_'Bu in solution, respectively,^{5b} and the neat DAE_Me and DAE_'Bu films show very similar values. The efficient Φ_{UV} can be attributed to the practically barrier-less reaction pathway after excitation.¹⁴ When blended with polymers, Φ_{UV} of DAEs is strongly affected by surrounding polymer matrices. Highly ordered polymers have Φ_{UV} that about double of those in amorphous polymers (e.g., $\Phi_{UV} = 61\%$ for DPPT-TT/DAE_Me, while $\Phi_{UV} = 29\%$ for F8T2/DAE_Me). A

more ordered matrix leads to a higher population of the antiparallel conformation of the ring-open isomer, only the antiparallel conformation can undergo ring closing while the parallel conformation cannot, and thus a higher quantum yield is observed in DPPT-TT and IIDDT-C3.

Table 1. Photochemical properties of DAEs in solutionand in polymer matrices.

	$\lambda_{\max}(nm)$		PSS ^b	Φ _{UV} ^c (%)	Φ _{vis} d (%)
~	open isomer	closed isomer			
DAE_Me solution ^a	279	520	92	43	0.8
DAE_Me film	286	543	56	42	0.11
F8BT/DAE_ Me film	282	544	35	23	0.34
F8T2/DAE_ Me film	285	552	43	29	0.63
P3HT/DAE_ Me film	283	552	40	36	0.34
DPPT- TT/DAE_Me film	285	537	65	61	0.08
IIDDT- C3/DAE_Me film	285	537	67	65	0.07
DAE_ ^t Bu solution ^a	282	523	90	31	0.9
DAE_ ^t Bu film	288	537	65	41	0.18
F8BT/DAE_ ^t Bu film	284	543	23	27	0.56
F8T2/DAE_ ^t Bu film	285	548	26	23	0.78
P3HT/DAE_ ^t Bu film	286	549	50	37	0.40
DPPT- TT/DAE_ ^t Bu film	285	535	56	42	0.10
IIDDT- C3/DAE_ ^t Bu film	286	538	51	49	0.09

 $^{\rm a}$ In acetonitrile. $^{\rm b}$ Amount of the closed DAE isomer in PSS upon UV irradiation (312 nm), obtained by UV/vis absorption spectra with the experimental error of 10%. $^{\rm c}$ Quantum yields upon 310 nm irradiation with the experimental error of 10%. $^{\rm d}$ Quantum yields upon 530 nm irradiation with the experimental error of 10%.

As commonly observed in DAE photoswitches, visible light-induced ring-opening of the closed isomer is typically associated with a lower Φ_{vis} as a result of an energy barrier along the excited state pathway.¹⁴ A reduction in Φ_{vis} was observed in both the neat DAEs films and polymer blended films. F8T2/DAE films have the highest Φ_{vis} , with 0.63% for DAE_Me and 0.78% for DAE_tBu. Φ_{vis} is slightly smaller in P3HT and F8BT matrices (ranging between 0.34% and 0.56%). It should be noted that Φ_{vis} of DAEs in F8T2, P3HT

or F8BT are more than three times higher than those of neat DAEs in the solid state, reflecting the fact that the conformational freedom in the matrices of these three polymers allows relatively unrestricted movement during the ringopening process. For DPPT-TT and IIDDT-C3, Φ_{vis} are much lower, below 0.1%, which can be attributed to steric hindrance brought into play by the rigid polymer matrix impeding the recovery of the open form. In general, DAE_'Bu shows slightly larger Φ_{vis} than DAE_Me in polymer matrices, possibly due to the four bulky *tert*-butyl groups of DAE-'Bu preventing deep interactions with the polymer microstructure therefore less affecting the photocycloreversion.

Morphology. As in a bottom-gate bottom-contact OTFT the charges mostly flow close to the interface between the semiconducting material and dielectric substrate, the surface morphologies in thin-films of polymer/DAE blends strongly affect the charge transport in the OTFTs applications.^{9b, 15} Atomic force microscopy (AFM) was used to study the morphology of neat polymers and polymers/DAE films. The images displayed in Figure 3 are discussed based on the appeared morphological features and on the root-meansquared (R_{RMS}) roughness, which was estimated on region of $10 \times 10 \,\mu\text{m}^2$. On the tens of micrometers scale explored by AFM, the binary films did not exhibit significant phase separation between the polymer and DAE component forming the blends, highlighting a good miscibility for both DAE_Me and DAE_tBu when blended with all the five chosen polymers. From the DAEs' perspective, the good miscibility with polymers would be advantageous for efficient physical and energetical interaction thus leading to a high current modulation upon photoisomerization.

As expected, F8BT, F8T2 and P3HT showed a very smooth morphology with little apparent structure as-cast by spin-coating, due to the amorphous nature of F8T2 and F8BT films and poorly ordered nature of P3HT when not exposed to thermal annealing.^{15b, 16} Films produced by blending F8BT with both DAE Me and DAE ^tBu exhibited very similar morphology and R_{RMS} compared to the pristine polymer films. F8T2/DAE_tBu film has a similar roughness as neat F8T2 (R_{RMS} of 0.25-0.30 nm), while blending DAE_Me molecules results in a slightly rougher morphology with $R_{RMS} = 0.42$ nm. In the case of P3HT, R_{RMS} increased after blended with DAE_tBu ($R_{RMS} = 0.87$ nm), compared to pristine P3HT films (R_{RMS} = 0.62 nm) and P3HT/DAE_Me films (R_{RMS} = 0.68 nm). Upon annealing at 90 °C, DPPT-TT films exhibited tightly packed grains that are the results of a greater propensity to undergo aggregation forming crystalline structures.¹⁰ The blending with DAE molecules resulted in a similar morphology, but accompanied with an increased surface roughness (R_{RMS} = 2.30 nm for DPPT-TT/DAE Me and $R_{RMS} = 2.56$ nm for DPPT-TT/DAE ^tBu, respectively). A larger grain size was also observed when DAEs were present, and the average width of grains increased from 150 nm for pristine DPPT-TT to about 200 nm for DPPT-TT/DAE_Me blends and 170 nm for DPPT-TT/DAE ^tBu blends. The AFM images of the as-spun IIDDT-C3 films revealed ordered crystalline fibrillary networks with individual fibers having a width of \sim 50 nm, being a typical feature for isoindigo polymeric derivatives.^{11b} Upon blending IIDDT-C3 with DAE_tBu, the surface roughness increased as evidenced by a R_{RMS} increase from 1.34 nm for IIDDT-C3 to 1.65 nm for IIDDT-C3/DAE_'Bu films. The presence of DAE_Me led in a larger fibrillary structure, with an average width of 80 nm, and greater roughness ($R_{RMS} = 2.04$ nm). The differences between DAE_'Bu and DAE_Me in IIDDT-C3 might result from a more spark propensity of DAE_Me to aggregate in the polymer matrix.



Figure 3. AFM topography images and root-mean-square roughness (R_{RMS} , and the black squares are error bars) of neat polymers and polymer/DAE blends.

Further investigation of the molecular packing and crystalline microstructure of polymer/DAE blends was carried out by GIWAXS, enabling to bestow information on a much smaller length scale. Our previous analysis on neat DAE_Me and DAE_^tBu indicated an amorphous nature of the spincoated films.^{5a,5b} When blended with P3HT, the DAE molecules preferred to reside in the amorphous regions of P3HT with no significant disruption of the molecular packing in the crystalline domains.^{5a, 5b} The GIWAXS patterns of neat F8T2 and F8BT consist of broad, featureless amorphous Xray scattering halos that likely stem from a considerable degree of disorder and lack of long-range order,^{15a,17} compared to ordered polymers such as DPPT-TT or IIDDT-C3. Blending with DAE molecules in F8T2 produced no apparent changes in its X-ray scattering pattern (Figure S4), and similar results are expected in F8T2 due to its amorphous microstructure.



Figure 4. 2D GIWAXS patterns. (A) DPPT-TT, DPPT-TT/DAE_Me and DPPT-TT/DAE_tBu films. (B) IIDDT-C3/DAE_Me and IIDDT-C3/DAE_tBu films.

Table 2. Predominant lamellar (100) and π - π stacking (010) distances extracted from fitting the diffraction peak positions of neat DPPT-TT and IDDT-C3 and polymer/DAE blends. Propagated fit error is ±0.01 Å.

System	lamellar spacing (Å)	π-π stacking spacing (Å)	
DPPT-TT	19.82	3.85	
DPPT-TT/DAE_Me	19.97	3.85	
DPPT-TT/DAE_ ^t Bu	20.34	3.81	
IIDTT-C3	24.84	3.64	
IIDTT-C3/DAE_Me	25.29	3.61	
IIDTT-C3/DAE_ ^t Bu	25.57	3.62	

Figure 4 shows the 2D GIWAXS patterns of neat DPPT-TT, IIDDT-C3 and their DAEs blends, and predominant lattice spacings extracted from the in-plane and out-of-plane scattering lineouts (Figure S5) are summarized in Table 2. Thin films of DPPT-TT and IDDT-C3 exhibit a predominant edge on texture, with lamellar and backbone ordering perpendicular to the substrate, and strong out-of-plane diffraction as evidenced by the presence of high order peaks corresponding to the (h00) set of crystal planes. In the in-plane direction, π -stacking (010) diffractions were observed. π - π stacking among conjugated backbones endows efficient charge transfer integrals thereby guaranteeing high device in OTFTs.¹⁸ As shown in Table 2, upon blending with DAEs, the crystalline structure of DPPT-TT and IIDDT-C3 polymer matrices is mildly disrupted by small molecule permeation.

An increase in the lamellar spacing suggests that some small molecules intercalate in the crystalline domains, whereas differences in π - π spacing are small and close to the experimental error, and thus, no comparisons are performed in this crystallite direction. It is interesting to note that blending with DAE_'Bu produces a larger expansion in the lamellar spacing of the polymers (2.6 and 2.9% in DPPT-TT and IIDDT-C3, respectively), whereas blending with DAE_Me produces smaller distortions in the lamellar direction (0.8 and 1.8% in DPPT-TT and IIDTT-C3, respectively), which can be ascribed to the greater steric hindrance brought into play by the DAE_'Bu molecules.

Electrical properties in OTFTs. To explore the electrical properties of the bi-component films, the polymer/DAE blends have been integrated in bottom-contact bottom-gate OTFTs, and the device mobility and photo-programmed I_{on}/I_{off} ratio (the ratio of current in the initial state and after UV irradiation) have been quantified, with the ultimate goal of casting light onto the charge transport properties of the chosen materials. Towards this end, OTFT assembled on SiO₂/Si substrates with pre-patterned gold interdigitated electrodes as source and drain contacts were used. The SiO₂ substrates were treated with octadecyltrichlorosilane (OTS). The bi-component blends were then spin-coated (see details in SI). It is already known that the output current in P3HT/DAEs-based OTFTs can be tuned by illumination at defined wavelengths. The driving mechanism is that the HOMO level of P3HT (-4.8 eV) is higher than that of DAEs in their open form (-5.6 eV), and approximately equal to the closed form (around -4.8 eV), thus the holes can be only trapped by the closed DAEs upon UV light irradiation, but not by the open-ring DAEs.^{5a} To further investigate the optically switching function of other polymer/DAE blendbased OTFTs, we firstly summarized the HOMO levels of the polymers and DAEs in Figure 1D. Amorphous/disordered F8T2 and F8BT have HOMO levels of -5.6 eV and -5.9 eV, respectively. ¹² It should be noted that the HOMO level of F8BT is lower than those of the DAEs both in their open and closed form, therefore the hole transport is energetically unfavorable in the presence of DAEs. The HOMO levels of the DPPT-TT and IIDDT-C3 (-5.2 eV and -5.5 eV, respectively) are lying in between those of the closed and open isomers of DAEs. 13, 14b



Figure 5. The device mobility of the neat polymers and polymer/DAE blends-based OTFTs (error bar in black, before irradiation), and their UV light response (error bar in red). The mobility is extracted from the transfer characteristics in the saturation regime (Figure S6), and calculated using eq. (1) provided in the Supplementary Information.

The device mobilities (extracted from the transfer characteristics in the saturation regime, see Supplementary Information for materials and characterization) of the neat polymers and polymer/DAE blends-based OTFTs are reported in Figure 5, and the transfer curves are plotted in Figure S6. Neat F8BT transistors showed a mobility of 1×10⁻ ⁵ cm²V⁻¹s⁻¹. As expected, drain current was detected within the noise of the measurement after blending with DAE molecules, since both the open and closed DAEs act as trap for the holes. Compared to the mobility of neat F8T2 film (8×10-³ cm²V⁻¹s⁻¹), F8T2/DAE_^tBu OTFT showed a very small decrease in the mobility (6×10⁻³ cm²V⁻¹s⁻¹), whereas more than one order of magnitude decrease was observed in F8T2/DAE_Me OTFTs (1.5×10⁻⁴ cm²V⁻¹s⁻¹). The reduced mobility in the latter case is possibly due to DAE Me molecules having a higher tendency to aggregate into F8T2 matrix thus impacting the polymer packing. The increased surface roughness and the similar HOMO levels of the open DAE_Me and F8T2 could easily cause the trapping and/or scattering of the hole transport. For the addition of DAEs into the matrix of P3HT a mobility in the same order of magnitudes as neat P3HT was observed, consistent with our previous results.^{5a, 5b} This can be explained by the facts of: 1) the DAE molecules reside in the amorphous region of P3HT, not strongly affecting the packing of P3HT; 2) the HOMO levels of the open DAEs is 800 meV below that of P3HT. which is not accessible for mobile holes thus minimizing the influence on charge transport in P3HT. DPPT-TT/DAE blends, with the HOMO difference of 400 meV from that of the open DAEs, also indicated the mobilities in the same order of magnitude as neat DPPT-TT. A small decrease in the mobility was observed when comparing the neat DPPT-TT (0.05 cm²V⁻¹s⁻¹) to the DPPT-TT/DAE_Me blend (0.02 cm²V⁻¹ ¹s⁻¹) and the DPPT-TT/DAE_^tBu blend (0.01 cm²V⁻¹s⁻¹). The slight change in mobility could be explained by: firstly, the smaller difference between HOMO levels compared to the case of P3HT; secondly, the fact that the crystalline structure of DPPT-TT polymer matrices is mildly disrupted by DAE molecules permeation. We note that DPPT-TT/DAE blends show the highest charge mobility in all of our investigated blends. In the case of IIDDT-C3, the mobility was dramatically affected when blending with DAE molecules, from 0.03 cm²V⁻¹s⁻¹ for neat IIDDT-C3 to 2.5×10⁻⁵ cm²V⁻¹s⁻¹ for IIDDT-C3/DAE_Me blend and 4.5×10⁻⁴ cm²V⁻¹s⁻¹ for IIDDT-C3/DAE_^tBu blends. The small injection barrier, consisting of only 100 meV from the HOMO level of the open DAEs (-5.6 eV) to that of IIDDT-C3 (-5.5 eV), provides high probability for holes scattering and/or trapping by DAE molecules and thus reducing the electrical performance. It is interesting to note that the impact of blending DAEs on the device mobility of F8T2, DPPT-TT or IIDDT-C3 is correlated with the roughness change evidenced by AFM imaging, i.e., the rougher the surface compared to that of neat polymer, the greater is the reduction in mobility. This modulation is magnified when the energy gap between the open DAE and polymers is becoming smaller. The films of DAE_Me blended with F8T2 and IIDDT-C3 exhibit both a higher roughness than those of DAE_^tBu blends, and lower hole mobilities in OTFTs. Such reduced electronic characteristics in rougher films can be due to the more disordered packing of the guest molecules within the polymer matrix and the increased content of grain boundaries hindering efficient charge transport.¹⁹



Figure 6. The reversible modulation of output current of polymer/DAEs-based OTFTs over irradiation cycles with UV irradiation (312 nm, violet shaded areas) and visible irradiation (540 nm, green shaded areas). (A) F8T2, F8T2/DAE_Me, and F8T2/DAE_'Bu-based OTFTs at Vd=-20 V and Vd=-60 V. (B) P3HT, P3HT/DAE_Me, and P3HT/DAE_'Bu-based OTFTs at Vd=-20 V and Vd=-60 V. (C) DPPT-TT, DPPT-TT/DAE_Me, and DPPT-TT/DAE_'Bu based-OTFTs at Vd=-50 V and Vd=-60 V. (D) IIDDT-C3, IIDDT-C3/DAE_Me, and IIDDT-C3/DAE_'Bu based-OTFTs at Vd=-20 V and Vd=-60 V. The irradiation steps were carried out until no variation in drain current was observed upon further illumination at the same wavelength. All values are normalized to initial value before any irradiation and the connecting lines serve as 'guides to the eye'.

A decrease in output current and device mobility in polymer/DAE blend-based OTFTs was observed upon UV illumination (Figure 5 and S7), as the consequence of holetrapping by closed DAEs. We note that the larger the HOMO gap between the closed DAEs and polymer, the higher is the photo-programmed I_{on}/I_{off} ratio. In the cases of F8T2 and IIDDT-C3, with HOMO levels ~700 meV below those of the closed-form DAEs, the I_{on}/I_{off} ratios are above 10³. The I_{on}/I_{off} ratio of DPPT-TT/DAEs-based OTFTs was found to be ~10² upon UV irradiation. With the very similar energy HOMO levels of the closed DAE and P3HT, the I_{on}/I_{off} ratio was measured at only ~10.

The reversibility of the drain current tuning of polymer/DAE blends-based OTFTs was investigated by performing two cycles of alternating irradiations at different wavelength (312 nm or 540 nm) (Figure 6). The drain current modulations of pristine polymers by light irradiation were below 10%. The variation is due to trapping of charges under bias stress and generation of mobile carriers under light irradiation. Noteworthy, no feature of reversible change was observed in neat polymers when the illuminating wavelength varied. In contrast, DAE_Me and DAE_fBu showed efficient current modulation and good reversibility in P3HT and F8T2 polymer matrices. In the latter case, the drain current is ~ 10^3 times less than the initial value when

the equilibrium upon UV irradiation is reached. Blending DAE_Me with highly ordered polymers (DPPT-TT and IIDDT-C3) led to an irreversible change of the drain current. A similar behavior was observed in DAE_Me and BTBT blends.5b It can be explained by the lower photocycloreversion quantum yield and higher PSS of DAE_Me in DPPT-TT and IIDDT-C3 matrices (as discussion in photochemical properties). Upon visible irradiation, a very small fraction of DAE_Me in their closed form still resided in the polymer matrices, sufficient to trap the transporting holes and thus keeping the drain current at a very low level. In contrast, the steric tert-butyl substituents in DAE_tBu resulted in a weaker association in the matrix of crystalline domain, associated with slightly higher photocycloreversion quantum vield and lower PSS, compared to DAE_Me. This might explain the almost full recovery of the drain current in the case of DPPT-TT and only 10% recovery in the case of IIDDT-C3 upon visible irradiation. Besides the limited propensity for photocycloreversion of the DAE, the reduction of the device performance after several switching cycles is governed by the photostability of the polymer matrix as well as intrinsic fatigue of DAE molecules, while DAE_Me and DAE_^tBu show reasonable fatigue resistance in solution and in films.^{5b, 21} The fatigue resistance of polymer/DAE-based OTFTs over alternative light irradiation can be further improved by choosing photostable polymers and DAEs featuring high fatigue resistance.

DISCUSSIONs

The study of photophysical and photochemical properties of the DAE based blends revealed a strong dependence of the PSS and photochemical quantum yield of DAE on structure of the polymer matrix. When embedded in highly ordered polymer matrices, such as those based on IIDDT-C3 and DPPT-TT, the antiparallel conformation of DAEs is likely to be constrained, yielding relatively high efficiency in the process of the ring-closure and higher PSS under UV light irradiation, but lower efficiency in the process of the ring-opening of DAEs under visible light irradiation. The closed DAE features a cofacial orientation of the coplanar planes.²⁰ which allows the molecules to easily pack in the rigid inner space of highly ordered polymers. Once the closed DAEs have been restricted in the rigid environment, they are hindered to undergo back isomerization to the more flexible and space occupying open isomer. On the other hand, the flexible inner spaces in amorphous and poorly ordered polymer matrices allow the closed DAEs to undergo relatively free movement back to the open isomer, as evidenced by an over five times higher Φ_{vis} . The reduced PSS and Φ_{uv} in F8BT, F8T2 and P3HT, compared to those of DAEs in highly ordered polymer matrices, may result from less antiparallel conformation constrained. When compared to DAE_Me, the DAE_'Bu inserted in highly ordered semicrystalline polymer matrix displays a slightly lower PSS and Φ_{UV} , but higher Φ_{vis} , indicating that when embedded in a polymer matrix, the constrains of the DAE_^tBu is more moderate than those of DAE_Me, because the four tert-butyl side group determine lower tendency to self-aggregation and aggregation in polymer matrices. The ringclosure is less efficient in the tert-butyl case because of steric strain, and the closed isomer is less stabilized favoring ring-opening, due to a smaller thermal barrier in the excited state of photocycloreversion.

The photophysical and photochemical behavior of polymer/DAE pairs is found to correlate well with the reversible modulation of output current in optically switchable OTFTs. The free movement of DAEs in amorphous and poorly ordered polymers enables effective photoisomerization of both ring opening and closing, and optical switching behavior of F8T2/DAE-based and P3HT/DAE-based OTFTs is stable over several cycles (Figure 6A and 6B). As the closed DAEs act as charge traps within the matrices, the higher PSS and Φ_{uv} of DAEs in the highly ordered DPPT-TT and IIDDT-C3 results in a significant decrease in output current upon UV irradiation (the first UV light irradiation step in Figure 6C and 6D). However, the closed DAEs is strongly hindered to undergo the photocycloreversion ($\Phi_{vis} < 0.1\%$) in the matrices of DPPT-TT and IIDDT-C3 upon further visible light irradiation, thus the transported charge is still trapped by residual closed DAEs causing the non-reversible modulation of output current of DPPT-TT/DAE_Me-based and IIDDT-C3/DAE_Me-based OTFTs. DAE_tBu has the advantage to more readily recover to the open isomer when compared to DAE_Me under visible light irradiation. We observed that output current recovery to about 10% in IIDDT-C3/DAE_tBu with a small bandgap between IIDDT-C3 and the open DAE_^tBu (100 meV), and considerable recovery in DPPT-TT/DAE ^tBu with a larger bandgap (400 meV).

The AFM images of bicomponent films provided evidence for the good miscibility of DAEs with the chosen polymers. Increase of the surface roughness was observed when DAEs were embedded in polymer matrices. Compared to the case of DAE_tBu, blending DAE_Me with F8T2 and IIDDT-C3 yielded rougher surfaces. The change of surface roughness resulted in variation of field-effect mobility of the polymers after introducing DAE molecules. We found that the rougher surface upon blending DAE_Me had lower mobility in F8T2 and IIDDT-C3, compared to DAE_^tBu in the same polymers. The increase of roughness, which is an indication of a greater amount of grain boundaries as a result of the blending with DAE molecules, determines a reduction in the device mobility. GIWAXS revealed that the crystalline structure of DPPT-TT and IIDDT-C3 polymer matrices is mildly disrupted by DAE molecules permeation, which explains the strong effect on the photoisomerization of DAEs and the decrease in device mobility in highly ordered polymersbased films. In the case of P3HT, DAE molecules reside in the amorphous regions and do not disrupt in the molecular packing of P3HT crystalline grains; as a result, the mobility of P3HT/DAE blends is comparable to that of neat P3HT films, and the photochemical processes of DAEs are similar to the case of amorphous F8T2 and F8BT.

The energy levels of polymers and DAEs is another key factor that governs the performance of optically switchable OTFTs. In order to tune the output current through the optical control, it is required that the energy levels of applied polymers are located in between phototunable energy levels of DAEs. As a counter example of F8BT, whose HOMO energy level is lying even lower than that of chosen DAEs in their open forms, blending these two components induces direct trapping of the holes, thus no hole conduction could be observed. The position of HOMO levels of the polymers, closer to that of the open DAEs or the closed DAEs, strongly influences both mobility and photo-programmed I_{on}/I_{off} ratio. When the HOMO of the polymers is far above that of the

open DAEs, as is the case for P3HT and DPPT-TT, the mobility of the blends remains as high as the pristine polymers. The large energy gap, 800 meV for P3HT and 400 meV for DPPT-TT, leads to the open DAE molecules not accessible for mobile holes thus minimize the influence of charge transport. The smaller bandgap between embedded polymers and the ring-open DAEs, together with the increased roughness, lead to stronger charge scattering and the possibility of trapping by the open DAEs in the polymer matrix, thus to some extent reduce the mobility of polymers, as was observed in the cases of F8T2/DAE Me and IIDDT-C3/DAE_Me blends. However, if the HOMO of the polymers is much lower than that of the closed DAE, the UV light irradiation can result in an increased photo-programmable Ion/Ioff ratio, surpassing 1000 in the case of F8T2/DAE pairs, due to the deep trapping. IIDDT-C3/DAE blends also exhibited high photo-programmed I_{on}/I_{off} with a similar bandgap as F8T2, however, they showed non-reversible light modulation due to the inefficient photocycloreversion. P3HT with the HOMO level adjacent to the closed DAE has a photo-programmable Ion/Ioff ratio about 10 because of the more moderate trapping.

CONCLUSION

The incorporation of photochromic diarylethenes in matrices of polymer semiconductors represents a powerful methodology to extend the functions of organic electronic devices with the response to external light. Here, we have reported an in-depth investigation on the performances of diarylethenes blended with organic semiconducting polymers featuring various microstructures and energy levels. A correlation between photoswitching, morphology, energy levels and OTFT performance of polymer/DAE blends have been thoroughly investigated. Based on our results, we have defined some important guidelines for the fabrication of high-performance optically switching organic devices: 1) photochromic molecules expected to weakly interact with the polymer matrices should be chosen to attain efficient photoisomerization quantum yields in both directions; 2) because of the amorphous nature of DAEs, polymers with lower degrees of crystallinity are better candidates, to guarantee better mixing with DAE molecules and less influence on the packing of polymer chains; 3) if a higher device mobility is required, in principle, the HOMO level of the polymer should be close to that of the closed DAE, while if the higher photo-programmable I_{on}/I_{off} ratio is required, the HOMO level of the polymer should be close to that of the open DAE. In compliance with the above guidelines, we have observed optical switchable OTFTs based on F8T2/DAE_^tBu outperforms all other combinations regarding rather high hole mobility (6×10⁻³ cm²V⁻¹s⁻¹) and high photo-programmed I_{on}/I_{off} ratio (>10³), and DPPT-TT/DAE_tBu based OTFT features highest hole mobility (0.01 cm²V⁻¹s⁻¹)) yet accompanied by a moderate photoprogrammed I_{on}/I_{off} ratio (10²). In principle, this approach can be applied to *n*-type polymer/DAE-based photoswitchable field-effect transistors, in which the electron transport can be modulated thanks to the photoactivated switching of DAEs with one isomer acting as an electron accepting level within the *n*-type semiconducting polymer matrix. We are convinced that the general component-structure-performance correlation deduced from this study will guide future rational design of molecular photoswitches/semiconducting polymers blends based high-performance light-responsive multifunctional optoelectronic devices for information storage and telecommunications.

ASSOCIATED CONTENT

Supporting Information. Details on materials; light irradiation; devices fabrication; devices characterization; UV/visible absorption; GIWAXS; transfer curves were described in supporting information. This material is available free of charge via the Internet at http://pubs.acs.org.

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