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 the dereplication of highly conjugated natural products using
 molecular networks

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30 32

9 Abstract :

10 Natural products are sources of inspiration and reservoir of high valuable molecules. Recently, analytical tools based on liquid chromatography coupled to tandem mass 11 12 spectrometry to generate molecular network became widely employed for dereplication. This 13 strategy greatly accelerates the identification of known and structural hypothesis of unknown. 14 Despite the availability of different ionization sources, alternatives to classical electrospray 15 ionization (ESI), such as atmospheric pressure chemical ionization (APCI) or photoionization 16 (APPI), have been neglected. In particular, APPI has been described for its ionization 17 efficiency on non-polar molecules bearing no acid or basic groups. For that reason, we investigated APPI potential to generate molecular network and compare it to ESI on several 18 criteria that are generation of ion species, sensitivity and signal-to-noise ratio (SNR) for 19 20 different extracts rich in highly conjugated natural products. We first optimized APPI experimental conditions on crude extract from a fungus, Penicillium sclerotiorum (SNB-21 22 CN111), producing polyketones belonging to the azaphilone family. Then we compared APPI 23 and ESI on different fractions of the fungus and on two plant extracts, French Guyanese Swartzia panacoco (Aubl.) R.S. Cowan (arial parts) and Indian Cassia auriculata L. (leaves) 24 25 containing phenolic compounds, such as flavonoids. While ESI generated more ion species and displayed a better sensitivity, APPI generated only protonated adduct and better SNR. 26 27 Comparing ESI and APPI generated species on molecular network reveal that both 28 strategies overlap for the majority of protonated ions.

29 Keywords: APPI, Molecular Network, Metabolomic, Azaphilone, Flavonoid

Natural products are a source of inspiration as well as an important reservoir of molecules of 30 high values in many economic fields such as medicine [1], nutrition [2], materials science [3] 31 32 or cosmetics [4]. The analytical strategies used in metabolomics become gold standard in 33 natural product research [5,6,7]. In particular, metabolomics accelerates the dereplication of 34 complex mixtures, i.e. rapid annotation of already published compounds, for the discovery of undescribed products. One of the most widely used dereplication tool is the strategy based 35 36 on liquid chromatography coupled to tandem mass spectrometry to generate molecular 37 networks (LC-MS/MS-MN) [5]. Briefly, a data-dependent analysis (DDA) is performed 38 throughout the chromatographic run. Cyclically, a MS1 scan is performed, then a fixed 39 number of charged species, typically from 3 to 5, will be sequentially fragmented and their 40 respective MS2 spectra will be acquired. These fragmentation patterns will then be compared leading to the calculation of a cosine-score related to a structural homology 41 distance. The generated distance matrix is finally used to generate the molecular networks 42 43 (MNs). T-SNE algorithm can be finally implemented to improve the visualization of the MN. (gnps.ucsd.edu) [8]. 44

45

LC-MS/MS experiments to produce data for MN are usually performed with an electrospray ion source. More recently, methodologies to generate MNs from GC-EI-MS data have been introduced [9], (gnps.ucsd.edu). Nevertheless, alternative ion sources such as Atmospheric Pressure Chemical Ionization (APCI) or Photoionization (APPI) have been largely neglected for such dereplication approaches. In particular, APPI has been previously described to be efficient for detection of non-polar or highly conjugated molecules bearing no acid or basic groups [10].

During APPI process, direct photoionization can occur if the ionization energy of the analyte 53 54 is lower than the energy of photons emitted by the discharge lamp, typically 10.0 and 10.6 eV for krypton. This phenomenon can be efficient on aromatic compounds for example. 55 56 However, the major part of the analytes are not ionized or not sufficiently ionized by APPI 57 leading to the introduction of a dopant molecule into the eluent. This electron or proton donor is usually a small organic molecule showing an ionization energy below 10 eV such as 58 59 anisole, toluene or acetone whereas [10] the common solvents, i.e. water and acetonitrile, 60 are not photoionized due to ionization potential at 12.6 and 12.2 eV, respectively [11].

Some families of natural products have already been studied by APPI such as pentacyclic triterpene [12], mycotoxins [13], lipids [14,15,16], chlorophyll derivatives [17], flower absolutes [18] or polyphenols [19]. However, these studies were performed only from a diagnostic or quantification perspective and never in an unsupervised way. We choose to

investigate two families of molecules likely to be studied with APPI: first,. highly conjugated 65 and colored polyketides (azaphilones and derivatives), with biological properties such as 66 antiparasitic [20] and antiviral [21] and proposed as potential natural food colorant [22,23] 67 and second, flavonoids, a molecular class with well-described antioxidant properties [24]. 68 Therefore, extracts from a microscopic fungus *Penicillium sclerotiorum* SNB-CN111, which 69 70 species is mentioned to produce azaphilone and derivatives [25] and two plants, *i.e.* French Guyanese Swartzia panacoco (arial parts) and Indian Cassia auriculata, (leaves), both from 71 72 the Fabales order, were investigated. Cassia auriculata is used in traditional Indian medicine 73 and its extract possesses antioxidant properties [26]. We thus propose to compare APPI and 74 ESI for the acquisition of MS2 data to generate MN in terms of number and type of detected species, sensitivity or MN information. 75

76

78 2. Material and method

79 2.1.1 Fungus material

The fungus strain Penicillium sclerotiorum SNB-CN111 of strain library collection of ICSN 80 (Sequencing allowed for strain identification by NCBI sequence comparison (Blastn®); the 81 82 sequence has been registered in the NCBI GenBank database (http://www.ncbi.nlm.nih.gov) under the registry number KJ023726) was cultivated on a large scale using 330 Petri dishes 83 (14 cm diameter) at 28 °C for 15 days on potato dextrose agar (PDA) medium (Dominique 84 Dutscher SAS, Brumath, France). The culture medium containing the mycelium was cut into 85 small pieces and macerated three time at room temperature with ethyl acetate (EtOAc) on a 86 rotary shaker (70 rpm) for 48h, 24h and 48h, respectively. The contents were extracted each 87 88 time with 10 L of EtOAc using a separatory funnel. Insoluble residues were removed via 89 filtration and the organic phase was washed three times with an equivalent volume of water 90 (H₂O), dried with anhydrous solid Na₂SO₄ and evaporated using a rotary evaporator under reduced pressure to yield a crude extract (6.5 g). The whole crude extract was fractionated 91 92 by reversed-phase flash chromatography (Grace Reveleris, Grace, Maryland, USA) using a 93 120 gram C18 column and ultraviolet (UV) and evaporative light scattering detector (ELSD) 94 detectors. A linear gradient of H_2O /formic acid (99.9/0.1) - acetonitrile/formic acid (99.9/0.1) (v/v, 95:5 to 0:100 over 40 min, flow rate at 80 mL.min⁻¹) followed by another gradient of 95 96 acetonitrile - methylene chloride (v/v, 50:50 to 0:100 over 15 min, flow rate at 80 mL.min⁻¹) was set up to generate 11 fractions that were labeled F1 to F11. 97

98 2.1.2 Plant material

99 Swartzia panacoco stems and leaves were collected on the Couy River near Kourou (N 100 05'01.523', W 052.43.490'), French Guiana, in April 2014. This species is not a protected 101 plant and can be collected without restriction at this location. A herbarium voucher (XC27) 102 was deposited in Guyane IRD Herbarium (CAY). Plant samples were dried (room 103 temperature, 10% air relative humidity) and finely powdered. Two sample of 300 g of stems 104 and leaves were extracted at room temperature by maceration using 3 × 1 L of EtOAc or 105 MeOH providing four different extracts after evaporation of the solvents.

106 *Cassia auriculata* leaves (1 g, Aroma-Zone, Paris, France) from India were macerated during 107 24 h at room temperature on a rotary shaker (70 rpm) with either methanol, ethyl acetate or 108 methylene chloride (200 mL). Insoluble residues were removed *via* filtration and the organic 109 phase was washed three times with an equivalent volume of H_2O using a separatory funnel, 110 dried with anhydrous solid Na₂SO₄ and evaporated using a rotary evaporator under reduced pressure to yield a crude mixture. The first equivalent of water from EtOAc extract was kept to be dried leading to 6 mg for H_2O fraction, 82 mg for MeOH, 62 mg for EtOAc and 56 mg for CH_2Cl_2 .

114 2.2 Data acquisition

Fractions from Penicillium sclerotiorum SNB-CN111 or crude extract from Cassia auriculata 115 leaves and Swartia panacoco leaves and stems were prepared at 1 mg.ml⁻¹ in methanol and 116 filtered on 0.45 µm PTFE membrane. LC-MS/MS experiments were performed with a 1260 117 Prime HPLC (Agilent Technologies, Waldbronn, Germany) coupled with an Agilent 6540 Q-118 ToF (Agilent Technologies, Waldbronn, Germany) tandem mass spectrometer. LC 119 separation was achieved with an Accucore RP-MS column (100 × 2.1 mm, 2.6 µm, Thermo 120 121 Scientific, Les Ulis, France) with a mobile phase consisting of H_2O /formic acid (99.9/0.1) (A) and acetonitrile with 0.1% formic acid (B). The column oven was set at 45 °C. Compounds 122 123 were eluted at a flow rate of 0.4 mL.min⁻¹ with a gradient from 5% B to 100% B in 20 min and then 100% B for 3 min. Injection volume was fixed at 5 µL for all analyses. For electrospray 124 125 ionization source, mass spectra were recorded in positive ion mode with the following 126 parameters: gas temperature 325 °C, drying gas flow rate 10 L.min⁻¹, nebulizer pressure 30 127 psi, sheath gas temperature 350 °C, sheath gas flow rate 10 L.min⁻¹, capillary voltage 3500 V, nozzle voltage 500 V, fragmentor voltage 130 V, skimmer voltage 45 V, Octopole 1 128 RFVoltage 750 V. The collision energy was fixed at 30 eV for all the MS/MS experiments. 129 For ESI, internal calibration was achieved with two calibrants (m/z 121.0509 and m/z130 922.0098) providing a high mass accuracy better than 3 ppm. For atmospheric pressure 131 photoionization source mass spectra were recorded in positive ion mode with the following 132 starting parameters for the optimization: gas temperature set at 325 °C, vaporizer set at 300 133 °C, capillary voltage set at 2500 V, fragmentor voltage at 130 V. The dopant flow was set at 134 30 µL per minute. For the analysis once optimized, APPI parameters were set to gas 135 temperature set at 325 °C, vaporizer set at 300 °C, capillary voltage set at 2000 V, 136 fragmentor voltage at 130 V. The dopant was acetone and its flow was set at 10 µL.min⁻¹. 137 External calibration was achieved with LC/MS tuning mix, for APCI/APPI (Agilent 138 Technologie, Massy, France) and mass accuracy was better than 5 ppm at m/z 383. The 139 data-dependent MS/MS events were acquired for the five most intense ions detected by full-140 141 scan MS (maximum of five precursors per cycle of 1.2 s) above an absolute threshold of 142 1000 counts. Selected parent ions were fragmented at a fixed collision energy of 30 eV and with an isolation window of 1.3 amu. The mass range of the precursor and fragment ions was 143 144 set as *m/z* 200–1000.

145 2.3 Data processing and analysis

The data files were converted from the .d (Agilent Technologies, Massy, France) standard 146 data format to .mzXML format using the MSConvert software, part of the ProteoWizard 147 package 3.0 [28]. All .mzxml were processed using MZmine2v51 as previously described 148 [29]. The mass detection was realized with MS1 noise level at 1000 and MS2 noise level at 149 0. The ADAP chromatogram builder was employed with a minimum group size of scans of 3, 150 a group intensity threshold of 1000, a minimum highest intensity of 1000, and m/z tolerance 151 of 0.008 (or 20 ppm). Deconvolution was performed with the ADAP wavelets algorithm 152 according to the following settings: S/N threshold=10, minimum feature height=1000, 153 154 coefficient/area threshold=10, peak duration range0.01-1.5min, RT wavelet range 0.00-155 0.04min. MS2 scans were paired using an m/z tolerance range of 0.05 Da and RT tolerance 156 range of 0.5 min. Isotopologues were grouped using the isotopic peak grouper algorithm with 157 an m/z tolerance of 0.008 (or 20 ppm) and a RT tolerance of 0.2 min. Peaks were filtered using Feature list row filter, keeping only peaks with MS2 scan (GNPS). Adduct identification, 158 159 i.e. sodium- or potassium-cationized species, was performed on the peak list with a retention time tolerance of 0.1 min, an m/z tolerance of 0.008 or 20 ppm, and a maximum relative peak 160 height of 150%. Complex search, such as dimers, was performed with a retention time 161 tolerance of 0.1 min, an m/z tolerance of 0.008 or 20 ppm and a maximum relative peak 162 height of 150%. Peak alignment was performed using the join aligner with an m/z tolerance 163 of 0.008 (or 20 ppm), a weight for m/z at 20, a retention time tolerance of 0.2 min and a 164 weight for RT at 50. The MGF file and the metadata were generated using the export/submit 165 to GNPS option. 166

167 Cosine score were calculated on MetGem 1.3 software [8], MetGem parameters. MS2 168 spectra were window-filtered by choosing only the top 6 peaks in the \pm 50 Da window 169 throughout the spectrum. The data was filtered by removing all peaks in the \pm 17 Da range 170 around the precursor m/z. The m/z tolerance windows used to find the matching peaks was 171 set to 0.02 Da and cosine score were kept in consideration for spectra sharing 2 matching 172 peaks at least. The number of iterations, perplexity, learning-rate and early exaggeration 173 parameters were set to 5000, 25, 200 and 12, respectively.

Annotations of molecules in MN were made using "Find standards in library" or "Find analogs in library" functions of MetGem. Queries were made against Massbank NA (<u>https://mona.fiehnlab.ucdavis.edu</u>), GNPS Public Spectral Library (<u>https://gnps.ucsd.edu</u>), NIST14 Tandem (<u>https://www.nist.gov</u>), NIH Natural Product (<u>https://www.nih.gov</u>) and MS-Dial (<u>http://prime.psc.riken.jp/compms</u>) libraries with absolute *m/z* tolerance of 0.02, 4 minimum matched peaks and minimal cosine score of 0.5. 180 To annotate molecules with high confidence in MN using standard search in library, both 181 exact mass, comparison of MS² spectra and previous report of molecules within genera or 182 species are taken into account.

183 Ion intensity and signal-to-noise ratio (SNR) for diagnostic ions were extracted with 184 MassHunter software (AgilentTechnologie, Massy, France).

Figures were generated using R and related packages (ggplot2, VennDiagram, Rcolorbrewerand gridextra).

188 3. Results and discussion

189 3.1 Parameter optimization

In order to optimize APPI conditions, several parameters were investigated such as vaporizer 190 191 temperature, the capillary voltage, the dopant chemistry and flow rate. The influence of these parameters on ionization and signal-to-noise ratio was then evaluated. Therefore, we 192 selected several ions belonging to the azaphilone family from a crude extract from 193 Penicillium sclerotiorium SNB-CN111. Five of them were the most abundant species, i.e. 194 ochrephilone ([M+H]⁺, m/z 383.1860); sclerotioramine ([M+H]⁺, m/z 390.1472); sclerotiorin 195 ([M+H]⁺, *m/z* 391.1312); geumsanol B ([M+H]⁺, *m/z* 417.1914); isochromophilone VI ([M+H]⁺, 196 m/z 434.1735)) and three were less abundant ones (unknown([M+H]⁺, m/z 353.1967), 197 isorotiorin([M+H]⁺, m/ 385.2015) and unknown([M+H]⁺, m/z 451.1525). Formation of radical 198 199 ion [M]^{**} is expected on APPI experiment but none were observed with acetone as dopant, 200 and less than 2% with toluene (supplementary figure 1). It can be explained by several of our experimental condition: protic solvent like water are known to support apparition of [M+H]⁺ 201 202 [30], acetone as dopant is described to enhance the formation of protonated species [31]. As for toluene, working with acetonitrile inhibit toluene [M]⁺ formation, thus leading to major 203 204 [M+H]⁺ population [32]. Optimizations of drying gas temperature, capillary voltage and dopant 205 flow are illustrated below with acetone as a dopant.

206

207 3.1.1 Effect of the vaporizer temperature

Because ionization under APPI happens in a vapor like state, vaporizer temperature is a key factor in order to maximize the ion intensity or the signal-to-noise ratio (SNR). A low vaporizer temperature could not vaporize enough mobile phase leading to signal dilution whereas too high temperature could lead to molecular degradation. A temperature range between 200 to 450°C with 50°C step was thus investigated for acetone at a flow rate of 10 μ L.min⁻¹.

Two types of behavior can be described based on figure 1. For 5 over 8 diagnostic ions (m/z353.1967, 390.1472, 417.1914 434.1735 and 451.1525), both intensities and SNR are the lowest at 200°C then reach a plateau between 250 and 350°C then dramatically decrease. For the other 3 diagnostic ions (m/z 383.1860, 385.2015 and 391.1312), a slight intensity decrease or a plateau is occurring from 200 to 450 °C. Therefore, a vaporizer temperature at 300°C was chosen for the following optimization steps because this source condition led to the best intensity and SNR for 6 over 8 diagnostic ions while being acceptable for the twolast ones (figure 1a, a1b).

222

3.1.2 Effect of the capillary voltage

The capillary voltage can affect the transmission efficiency of the ions at the entrance of the mass spectrometer and is rather independent to the vaporizer temperature. A voltage range from 2000 to 3500V by 500V step was scanned. As depicted on Figure 2, this parameter has no significant effect on the final intensity or SNR of the diagnostic ions and was optimal at 2000 V. This value was chosen for the following optimization steps.

229

230 3.1.3 Effect of the dopant flow rate

When direct photoionization of compound is not efficient, the addition of a dopant is 231 necessary. The dopant will exchange its charge or radical hydrogen to the analyte [33]. In the 232 case of acetone, the main ionization process is the formation of protonated acetone followed 233 234 by a proton transfer to the analyte. One important reported parameter in the literature is the 235 flow rate of the dopant [34,35]. If the dopant quantity is too low, the ionization process will not be optimal whereas a too high quantity will lead to increasing noise signals. The dopant flow 236 237 rate has been empirically established to 5-10% of the total flow rate. Therefore, typical dopant flow rates from 10 to 50 μ L.min⁻¹ were investigated (figure 3a, 3b). 238

Without dopant, proportion of [M]⁺ significantly increase (Supplementary figure 2) suggesting 239 that highly conjugated azaphilone are subject to direct photoionization but both signal 240 241 intensity and SNR for all diagnostic ions are drastically increased when introducing the dopant. The dopant flow rates had no significant impact on the intensity of the 8 diagnostic 242 ions (figure 3a) whereas increasing flow rate of the dopant decreases the SNR (figure 3b). 243 Thus the optimal value is 10 μ L.min⁻¹ for acetone representing 2.5 % of the eluent flow rate. 244 This value is slightly under described ones [34] 5 to 10 % of eluent flow rate but it could be 245 related to the use of acetone instead of toluene. 246

247

248 3.1.4 Effect of the dopant chemistry

249 Acetone, anisole and toluene are classical dopants handled to increase APPI efficiency. We excluded anisole as a dopant from the analysis due to several intense ions at m/z 206.1666, 250 251 230.0937, 263.0917, 530.4691 and 647.4576 throughout the chromatographic time saturating the signal. The same optimization procedure as for acetone was operated for 252 toluene (supplementary figure 3). The selected parameters for toluene were as follows: 253 vaporizer temperature at 350 °C, capillary voltage at 2500 V and dopant flow at 30 µL.min⁻¹. 254 255 The optimized flow rate of toluene was higher compared to acetone as previously described 256 in the literature [34].

Acetone as a dopant outperforms toluene for the intensity of 6 diagnostic ions and displays same intensities for the two others (m/z 417 and 434) (figure 4a). Regarding SNR, acetone fully exceeds toluene by a factor of 3 (figure 4b). Some author reported this phenomenon and suggest that toluene impurities could explain this discrepancy [35]. Based on this experiment, acetone was kept as a dopant for further experiments using the previously optimized ionization conditions.

263

3.2 Comparison of ESI and APPI for the dereplication of *Penicillium sclerotiorum* SNB-CN111

In order to evaluate APPI-MS for dereplication of complex mixture of natural products by molecular networking [8,28, 36], *Penicillium sclerotiorum* SNB-CN111 crude extract has been fractionated by polarity. Nine fractions were thus analyzed by reversed-phase liquid chromatography couple to tandem mass spectrometry (LC-MS/MS) in ESI and APPI positive modes. Then, the data have been processed by MZmine2 and MetGem software to construct molecular network containing data from both ionization sources.

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273 3.2.1 Qualitative analysis of ESI and APPI analysis

Chromatographic run with ESI source produce significantly more ion signals than APPI 274 275 (Figure 5a). Indeed, over the 9 analyzed fractions, ESI generated 3152 nodes (precursor ion / retention time couples) for which a MS² was acquired. This is 4.2 time more than APPI 276 277 (747). About 15 % (462) of ESI signals were annotated by MZmine2 fragment search algorithm as [M+H]⁺, [M+Na]⁺, [M+K]⁺, [2M+H]⁺, [2M+Na]⁺, [M+M'+H]⁺ or [M+M'+Na]⁺ of the 278 same molecule M. A manual search within the most intense ions, as well as the clustering of 279 280 ESI nodes in the t-SNE representation, suggests that this number is largely underestimated. 281 This phenomenon is common within ESI experiment and manual inspection is often

282 mandatory to avoid misinterpretation or redundant information. On the contrary, APPI ion 283 signals were strictly annotated as protonated species. Common features between ESI and 284 APPI are estimated to be 497 corresponding to 66% of APPI signals. Moreover, these 285 common features are related to ion population of high intensity (Figure 5b, 5c) and can be 286 employed for reducing the redundant information from non-protonated species. Furthermore, 287 ESI led to the detection of 125 features present in each sample and in the blank (pure 288 methanol) whereas APPI only exhibited 7 interferences from the blank, *i.e.* pure methanol.

An intensity threshold value of 100 000, corresponding to two orders of magnitude between the most and less intense feature, was added to simplify the reading of the molecular networks (figure 5c, 6a) and get rid of very low intensity ions. The dataset is finally composed of 691 ions, *i.e.* 374 APPI features, 643 ESI ones which 102 of them are non-protonated species and 326 are common between the two ionization methods (figure 6b). This clearly highlights the fact that major differences between ESI and APPI features are for low abundant species and non-protonated species.

It is interesting to note that the species present in the south of the t-SNE MN is mainly constituted of bimolecular species such as [2M+H]⁺, [2M+Na]⁺, [M+M'+H]⁺ or [M+M'+Na]⁺ (figure 6a). To understand better the differences of ionization between several scaffold of azaphilone such as ochrephilone, isochromophilone VI or geumsanol B, MN annotation is provided in Figure 7 and led to the annotation of 16 molecular classes using dereplication tools such as MS² database search, the Natural Product Atlas [37] and a review of literature on Azaphilone and *Penicillium sclerotiorum* or *multicolor*. [38,39]

Several signals have been directly annotated through MS² database comparison, the 303 304 database compound with the highest homology cosine score was used for annotation, such 305 as isochromophilone VI (cluster1, cosine score = 0.85), sclerotioramine (cluster2, cosine score = 0.97), ochrephilone (cluster4, cosine score = 0.92), hypocrellone A (cluster 6, cosine 306 307 score = 0.76), geumsanol C (cluster 7, cosine score = 0.59), several diketopiperazine 308 (cluster 8, CycloPhe-Pro, cosine score = 0.97, CycloPheLeu, cosine score = 0.92, CycloPhe-HydroxyPro, cosine score = 0.80), monoelaidin (cluster 9, cosine score = 0.86), 309 isochromophilone IV (cluster 13, cosine score = 0.75), sclerotiorine (cluster 14, cosine score 310 311 =0.82), lauryl ethanolamide (cluster 15, cosine score = 0.87). Ergosterol derivatives (cluster 312 10) and sclerotioramine analogs (cluster 12) were found through analogs search into MS² 313 databases. Their scaffolds were common and redundant over several nodes present in their 314 respective cluster. Geumsanol C (cluster 7) annotation with a homology cosine score from 315 databases of 0.59 was low for a highly confident annotation, we decided to include it because we manually found other evidence that increased our confidence: its chlorinated 316

analog is present within the same cluster and also annotated through databases search with 317 the highest compound hit of 0.66. . Finally, Peniazaphilone A (cluster 3, 0 ppm from 318 calculated mass), Geumsanol B (cluster 5, +1.7 ppm from calculated mass) are dereplicated 319 using the structural cluster search of Natural Product Atlas. Most of these molecule share a 320 common scaffold and belongs to the azaphilone family. Nevertheless, there is a diversity of 321 substructure that can impact ionization: addition of hydroxyl (Geumsanol B, Geumsanol C, 322 323 Hypocrellone A), a lactone ring (Peniazaphilone A, Ochrephilone, Geumsanol B), substitution of the oxygen by a nitrogen (Isochromophilone VI, Sclerotioramine, Peniazaphilone A). This 324 is why we decided to make a dilution range of the fractions from 1000 µg.mL⁻¹ to 25 µg.mL⁻¹ 325 in order to evaluate sensitivity and SNR for APPI compare to ESI (figure 8b-8k). 326

ESI outperformed APPI in term of ionization for 4 out of 6 species (Ochrephilone (I), 327 Sclerotiorine (II), Sclerotioramine (III), Isochromophilone VI (IV)) and displayed similar 328 ionization to APPI for the two others (Geumsanol B (V) and 5-chloroisorotiorin (VI)) at 25 and 329 100 µg.ml⁻¹. At 250 µg.ml⁻¹ ESI performed better for 3 out of 6 ions (Ochrephilone (I), 330 Sclerotioramine (III) and Isochromophilone VI (IV)). This is not the case anymore for 331 Sclerotiorine (II) as APPI and ESI equally perform, as well for Geumsanol B (V) and 5-332 chloroisorotiorin (VI). Despite a saturation phenomenon at 1000 µg.ml⁻¹, it is interesting to 333 note that only the 2 nitrogen molecules (Sclerotioramine (III) and Isochromophilone VI (IV)) 334 are better ionized by ESI (figure 8a - 8g). 335

APPI leads to higher SNR for 3 molecules out of 6 (Ochrephilone (I), Isochromophilone VI (IV), Geumsanol B (V)) compared to ESI, similar values for (5-chloroisorotiorin (f)) and lower for 2 molecules (Sclerotiorine (II), Sclerotioramine (III)) at 25 μ g.ml⁻¹. The capability of APPI to achieve higher SNR compared to ESI increases for the higher concentration: 5 out of 6 molecules (Ochrephilone (I), Sclerotiorine (II), Sclerotioramine (III), Isochromophilone VI (IV) and 5-chloroisorotiorin (VI)) display better SNR value with APPI at 100 μ g.ml⁻¹, for example (figure 8a, 8h - 8k).

Overall, ESI and APPI displayed different but complementary features for the analysis of the specialized metabolism of *Penicillium sclerotiorum*. While ESI displays better sensitivity overall, APPI showed better SNR and produces only protonated species leading to an easier data interpretation. Similar results were obtained on other natural product families, indicating that overall, APPI is less susceptible to matrix effect [40, 41]. Once data from ESI and APPI are combined together, they allow a highly valuable molecular snapshot of a complex sample.

350

351 3.3 Comparison of ESI and APPI for the analysis of flavonoid containing plants.

To extend the scope of this study, we have chosen to investigate another class of natural product, *i.e.* flavonoids in crude plant extracts. *Swartia panacoco* and *Cassia auriculata* specialized metabolites were extracted with solvent of different polarity (water, methanol, ethyl acetate and dichloromethane). The extract was thus analyzed by LC-MS/MS using either ESI or APPI in positive mode.

357

358 3.3.1 Qualitative analysis of ESI and APPI analysis

359 In accordance with the results previously described with *Penicillium sclerotiorum* fractions, 360 LC-MS with ESI source produces significantly more ion signals than APPI (Figure 9a, 9c). Indeed, over the analyzed 8 crude extract (2 plants and 4 extraction solvents each), ESI 361 generated 3560 nodes (precursor ion / retention time couples) for which a MS² was acquired. 362 This is 5.2 time more nodes than APPI (675). About 6% (222) of ESI signals were annotated 363 by MZmine2 fragment search algorithm as [M+H]⁺, [M+Na]⁺, [M+K]⁺, [2M+H]⁺, [2M+Na]⁺, 364 [M+M'+H]⁺ or [M+M'+Na]⁺ of the same molecule M. On the contrary, APPI ion signals were 365 strictly annotated as protonated species. Common features between ESI and APPI are 384 366 corresponding to 56% of APPI signals. Moreover, these common features are related to ion 367 368 population of high intensity (Figure 9b) and can be employed for reducing the redundant 369 information from non-protonated species. Furthermore, ESI led to the detection of 152 370 features present in each sample and in the blank (pure methanol) whereas APPI only exhibited 9 interferences from the injection solvent. All these observations are in perfect 371 accordance with results obtained on Penicillium sclerotiorum. 372

Following the same procedure as before, an intensity threshold was added to simplify the reading of the molecular networks and get rid of very low intensity ions. This time value of 50 000 has been defined, to keep to two orders of magnitude between the most and less intense feature. The dataset is finally combining 898 nodes, *i.e.* 295 APPI features, 870 ESI ones which 49 of them are non-protonated species and 267 are common between the two ionization methods (figure 10a, 10b). Once again major differences between ESI and APPI are concentrated on low abundant or non-protonated species.

It is interesting to note that the west of the t-SNE MN is mainly constituted of high intensity nodes produced by both APPI and ESI. The nodes specific to APPI are concentrated into two clusters (in the south west and the south east) and one isolated node in the north. MN annotation is provided in Figure 11 to better understand the specificity of APPI and ESI, and to verify if our hypothesis considering APPI as a technic of interest to search for flavonoid in a complex extract. It led to the annotation of 14 clusters using MS² database search dereplication tools. Cluster1 was annotated as glycosylated flavonoids (quercetin-4O-

glucoside, m/z 465.0988, cosine score = 0.79), cluster 2 as flavonoids (3-O-methylquercetin, 387 m/z 317.0628, cosine score = 0.84, luteolin, m/z 287.0724, cosine score = 0.88 and catechin, 388 389 m/z 291.0848, cosine score = 0.87), cluster3 as Pheophorbide A (m/z 593.275, cosine score = 0.75), cluster 4 as Pheophytin A analogs (Pheophytin A, m/z 871.5748, cosine score 390 =0.69), cluster 5 as ceramide (analog search on the whole cluster), cluster 6 as 391 phytosphingosine (m/z 318.2977, cosine score = 0.73), cluster 7 as phosphatidylcholine 392 393 (Lyso-PC(16:0), m/z 496.3387, cosine score = 0.87), cluster 8 as adenosine and derivatives (adenosine, m/z 268.1211, cosine score = 0.83), cluster 9 as phtalates (dioctyl phtatalate, 394 395 m/z 391.2822, cosine score = 0.80), cluster 10 as monoglyceride (glycerol 1-stearate, m/z396 359.313, cosine score = 0.89), cluster 11 as fatty acids (octadecadienoic acid, m/z 295.225, 397 cosine score = 0.66), cluster 12 was pentacyclic triterpene (glycyrrhetinic acid, m/z 471.3433, 398 cosine score = 0.71), cluster13 as diglyceride/triglyceride (analogs search on the whole 399 cluster) and cluster 14 as fatty acid amide (linoleyl-ethanolamide, m/z 324.287, cosine score 400 = 0.79).

A single cluster at the extreme east, and annotated as polyethylene glycol (PEG) derivatives, has been cut off because it was too far away and took away legibility. From the two clusters constituted of APPI-only feature, one is constituted of ceramide analogs while the other seems to be a subclass of pentacyclic triterpene. Moreover, the APPI generated ion intensity of pentacyclic triterpenes is directly depending to their polarity (Supplementary figure 4); the less polar, the better the intensity was. As previously described in the literature, highly apolar pentacyclic triterpenes are well suited for APPI analysis in the positive mode [12].

408 Because our aim was to characterize and compare APPI performances over ESI on highly 409 conjugated molecules, we extracted both intensity and SNR and intensity for a panel of 6 410 molecules representative of flavonoid clusters. Two molecule from the purple sub-cluster (3-O-methylquercetin, cosine score = 0.84, and formononetin, cosine score = 0.84, cluster 2), 1 411 molecule from the north pink sub-cluster (luteolin, cosine score = 0.88, cluster 2), 2 molecule 412 from the brown sub-cluster (Epigallocatechin, cosine score =0.7 and Catechin, cosine score 413 414 = 0.87, cluster 2) and one molecule from the glycosylated flavonoid green sub-cluster (Quercetin-4O-glucoside, cosine score = 0.79, cluster 1) of the t-SNE representation (Figure 415 416 11).

For 3-o-methylquercetin (I), APPI outperforms ESI both in intensity and SNR. ESI and APPI displays equal sensitivity for Formononetin (II) and Luteolin (III) and similar SNR for Formononetin (II) while SNR of Luteolin (III) is better for ESI. For the last 3 diagnostic molecules (Epigallocatechin (IV), Catechin (V) and Quercetin-4-o-glucoside (VI)) ESI operates much better than APPI for both intensity and SNR. The global behavior is following the retention times and thus the polarity as described for pentacyclic triterpenes. This results
confirmed that APPI can be highly performant for the ionization of apolar molecules, as
described within several lipids classes [14], leading to highly informative molecular networks.

426 Conclusion

427 APPI-MS parameters were optimized for the analysis of highly conjugated and apolar 428 compounds from a fungus crude extract with an emphasis on sensitivity and SNR. While ESI-MS generates more species and displayed a better sensitivity overall, APPI-MS only 429 generates protonated species, displays better SNR and generated less interfering species. 430 431 These conclusions were extended by the study of arial parts extracts from French Guyanese Swartzia panacoco and Indian Cassia auriculata, especially for flavonoid and pentacyclic 432 triterpene families. Using complementarity between APPI-MS and ESI-MS, a strategy for the 433 analysis of complex natural extracts by molecular network could start with an APPI-MS 434 435 analysis to characterize and annotate ions. This will avoid redundant signals because APPI-MS generate only protonated species with high SNR. Then a second batch of analysis could 436 be performed using ESI-MS in order to detect and annotate low abundance ions thanks to 437 the better sensitivity of ESI-MS. 438

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Figure legends:

- Figure 1: Effect on (a) intensity or (b) signal-to-noise ratio (SNR) of vaporizer temperature for
 8 ions from *Penicillium sclerotiorum* SNB-CN111 crude extract. Logarithmic scale.
- Figure 2: Effect on(a) intensity or (b) signal-to-noise ratio (SNR) of capillary voltage for 8 ions
 from *Penicillium sclerotiorum* SNB-CN111 crude extract. Logarithmic scale.
- Figure 3: Effect on(a) intensity or (b) signal-to-noise ratio (SNR) of acetone dopant flow for
 8 ions from *Penicillium sclerotiorum* SNB-CN111 crude extract. Logarithmic scale.
- Figure 4: Effect on(a) intensity or (b) signal-to-noise ratio (SNR) of dopant chemistry for 8 ions from *Penicillium sclerotiorum* SNB-CN111 crude extract. Logarithmic scale.
- Figure 5: (a) Ion distribution from *Penicillium sclerotiorum* SNB-CN111 fractions between ESI
 and APPI; (b) their respective intensity their (c) t-SNE representation constructed on MS²
 homology the size of nodes is related to their intensity.
- Figure 6: *Penicillium sclerotiorum* SNB-CN111 fractions: (a) t-SNE representation constructed on MS² homology and (b) Ion repartition between ESI and APPI. An intensity threshold value of 100 000 is applied to the dataset, the size of nodes is related to their intensity.
- Figure 7: Structural annotation of the molecular network related to *Penicillium sclerotiorum*SNB-CN111 fractions. Contaminants are composed of polydimethylsiloxane (PDMS) from
 vial septum (hit on databases).
- Figure 8: (a) Diagnostic molecules (Ochrephilone (I), Sclerotiorine (II), Sclerotioramine (III),
 Isochromophilone VI (IV), Geumsanol B (V) and 5-chloroisorotiorin (VI)). Ion counts (b, c, d,
 e, f, g) and SNR (h, I, j, k) for the respective protonated species. Logarithmic scale.
- Figure 9: (a) Ion distribution of specialized metabolites from *Swartia panacoco* and *Cassia auriculata* crude extracts analyzed by E|SI and APPI, (b) their respective intensity (logarithmic scale), and (c) t-SNE representation constructed on MS² homology, the size of nodes is related to their intensity.
- Figure 10: (a) t-SNE representation –from *Swartia panacoco* and *Cassia auriculata* crude extracts– constructed on MS² homology, the size of nodes is related to their intensity. (b) Ion repartition between ESI and APPI. An intensity threshold value of 50 000 is applied to the dataset.

Figure 11: Annotation of the molecular network (t-SNE view) from 8 extracts of two plant
leaves (*Swartia panacoco* and *Cassia auriculata*). 1: Glycosylated flavonoid, 2: Flavonoid,
3: Pheophorbide A, 4: Pheophytin A analogs, 5: Ceramide, 6: Phytosphingosine,
7: Phosphatidylcholine, 8: Adenosine and derivative, 9: Phtalate, 10: Monoglyceride, 11:
Fatty acid, 12: Pentacyclic triterpene, 13: Diglyceride/Triglyceride, 14: Fatty acid amide.

- 484 Figure 12: (a) Diagnostic molecules (3-O-methylquercetin (I), Formononetin (II), Luteolin (III),
- 485 Epigallocatechin (IV), Catechin (V) and Quercetin-4-O-glycoside (VI)), Ion counts (b) and
- 486 SNR. (c) for the respective protonated species Logarithmic scale.

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