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1 **OPTICAL THEORY BASED SIMULATION OF ATTENUATED TOTAL**
2 **REFLECTION INFRARED SPECTRA OF MONTMORILLONITE FILM**

3

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8

ABSTRACT

10 Infrared analyses of clay minerals samples are usually performed by transmission techniques. While
11 transmission measurements are easy and inexpensive, the sample preparation plays a primordial role in the quality
12 of the data. Alternatively, attenuated total reflection (ATR) provides a powerful and often simpler analysis method.
13 However, the ATR spectra reveal strong differences when compared to transmission spectra leading sometimes to
14 confusion in the interpretations. Indeed, optical effects play a prominent role in the ATR spectral profile and their
15 identification is mandatory for obtaining quantitative information regarding molecular/particle orientation or film
16 thickness. Here, exact spectral simulations of montmorillonite films are performed by making use of the optical
17 theory, following the determination of the anisotropic optical constants from the experimental reflectance spectra
18 by Kramers-Kronig transformation. This methodology can advantageously be used (i) to choose the appropriate
19 optical conditions for advanced and reliable characterization of clay minerals, (ii) to extract quantitative
20 information such as the estimation of the film thickness, and (iii) to discriminate optical phenomena (optical
21 interferences) from chemical/structural features of the sample.

22

KEYWORDS

23 Attenuated Total Reflectance, Clay minerals, numerical simulation, infrared optical constants.

24

INTRODUCTION

The chemistry of mineral surfaces and the processes taking place thereon in aqueous environment play a pivotal role in a wide spectrum of chemical, physical and geological processes (Stumm 1997; Lambert 2008; Polubesova and Chefetz 2014). The large surface area in combination with the high retention capability give to clay minerals important adsorption performances for organics and metal ions (Brigatti et al. 2013). To date, probing interfacial processes remains a challenging task for experimentalists because of the complexity of the system. Surface analytical techniques usually require separating the solid from the aqueous solution prior to analysis with the undesirable consequence of completely changing the surface characteristics of the solid or the adsorbed species. Though such methods may be helpful in providing some insights into the nature of the interface under investigation, comparison with the batch experiments may be ambiguous. With the constant effort given by experimentalists to integrate molecular data into macroscopic adsorption models extracted from isotherm measurements, it is of prime importance to probe the interface in-situ for capturing an accurate picture of the interactions and surface modifications occurring in real systems.

In this context, attenuated total reflectance Fourier transform infrared spectroscopy (ATR-FTIR) is particularly valuable as a molecular spectroscopic technique sensitive to both the inorganic and organic parts, and with the possibility of adapting diffusion cells on the top of the internal reflection element (IRE) for probing in-situ dynamical processes (Hind et al. 2001; Lefèvre et al. 2012). Although the transmission method is usually preferred for characterizing materials using the Beer Law, the dispersion of the refractive index and the interference phenomena are commonly ignored. In addition, the delicate task of preparing pellets of samples homogeneously diluted in a transparent matrix suitable for transmission analysis contributes to the frequency variations commonly observed in literature for similar systems (Iglesias et al. 1990). Unlike transmission, ATR gives a quasi-absorption spectrum which is similar but not identical to transmission (Yamamoto and Ishida 1994). This technique greatly facilitates sample preparation and the recorded signal is usually enhanced with respect to transmission enabling the measurements of single inorganic or organic layers. However, interpretation of ATR spectra is more difficult because of the prominence of optical effects (Hansen 1965, 1968). A direct consequence of these effects is the apparent shift of the vibrational bands and intensity variations when compared to transmission spectra making the assignment of vibrational bands more challenging.

Given this strong connection between band profiles and instrumental setup used for analyses, infrared spectra should be regarded as a sample "fingerprint" only for a given condition. The most relevant information that can be obtained from IR measurement is the optical constants of the sample which are independent of the measurement

55 methods and allow cross-referencing with results obtained under different experimental conditions (Tolstoy et al.
56 2003; Hasegawa 2017). Optical constants describe phenomenologically the interaction of the radiation with the
57 matter, providing insight into the molecular structure of a material. Extracting the optical constants from the
58 spectroscopic observables is not straightforward. Two different approaches have been adopted for extracting the
59 optical constants from reflectance spectra of clay minerals. The first one is based on Kramers-Kronig transform to
60 analyze normal or quasi-normal reflectance data. For example, Karakassides et al. performed Kramers-Kronig
61 analysis on reflectance spectra at 10° off-normal incidence to compute the optical and dielectric properties of
62 montmorillonite clay minerals in pellet form (Karakassides et al. 1997). These analyses enable to evidence the
63 migration of Li⁺ cations into the layer structure upon thermal treatment. Alternatively, some authors model the
64 optical constants of phyllosilicate minerals by making use of dispersion theory using non-linear least square
65 optimization routines and a best-fit approach (Glotch and Rossman 2009; Roush et al. 1991; Esposito et al. 2000).
66 This approach consists in defining the vibration in a crystal lattice as a sum of the vibrations of numerous oscillators
67 leading to dispersion equations associating the complex dielectric function as a function of the frequency. The
68 optical constants, $n(\nu)$ and $k(\nu)$, are then directly determined from the dielectric function. The slight discrepancy
69 usually observed in the determined optical constants by these two approaches likely originates from the sample
70 preparation and the calculation methods. Optical constants are therefore macroscopic parameters, and can be
71 hardly transferable if acquisition or sampling conditions are not identical.

72 The determination of optical parameters is a prerequisite for advanced interpretation of structural features based
73 on optical spectroscopy. As an example, Ras and coworkers carried out numerous studies to determine the structure
74 and orientation of hybrid organo-clay preparations by various infrared spectroscopic methods (Ras et al. 2003,
75 2004b, 2004c, 2007b). These authors successfully prepared highly oriented monolayer films of clay minerals by
76 the Blodgett-Langmuir method, enabling them to investigate the orientation and conformation of organic
77 molecules adsorbed onto the clay minerals by attenuated total reflectance spectroscopy. Orientational
78 determination consists in relating changes in intensity of a vibrational band obtained with the electric field oriented
79 parallel or perpendicular to the plane of incidence to the average angle between the transition moment at the origin
80 of this vibrational band and the electric field of the infrared radiation. These calculations require the determination
81 of the mean squared electric field amplitudes which need the sample optical constants as input parameters. For the
82 reasons explained above, the correctness of the orientational order determination relies strongly on the sample
83 optical constants.

84 These examples illustrate that macroscopic properties such as the orientational order of a given sample are
85 transferable only if all the input parameters enabling the determination of that property are determined on the same
86 preparation. Motivated by this statement, this work aims at providing ATR users with the necessary equations for
87 computing the optical constants from polarized ATR measurement that can be further used for the quantitative
88 interpretation of infrared spectra. The determination of the anisotropic optical constants is based on a Kramers-
89 Kronig transform analysis applied to polarized ATR spectra of thick Na – SWy-3 montmorillonite film (Bardwell
90 and Dignam 1985; Dignam and Mamiche-Afara 1988; Bertie and Lan 1996). These optical constants are then used
91 to simulate spectra under a variety of conditions. Because the orientational order has already been treated in the
92 literature (Grégoire et al. 2020), this manuscript specifically focuses at the determination of film thickness and its
93 influence on the measured spectra, the absorption coefficient and the qualitative influence of the presence of water.

94 The benefit of this methodology is two-fold. First, and for a sample with given optical constants, the simulated
95 spectra can reveal the effects of interferences due to e.g. thickness, orientation, etc. First, the calculations being
96 based on optical theory, the simulated ATR spectra only evaluate optical interferences because the sample property
97 are described by their experimental optical constants. Therefore, deviations between the experimental and
98 calculated spectra may originate from changes in the sample property (such as porosity or orientation) giving
99 evidence for specific interactions or changes in sample chemistry which affect optical properties. Second, it can
100 help at choosing the most relevant experimental conditions for the acquisition of high-quality spectra such as the
101 appropriate IRE element, angle of incidence, polarization angle, film thickness, optical configurations (e.g. internal
102 or external reflectance, normal or off-normal transmission, etc.).

103 **THEORETICAL BACKGROUND**

104 Spectral simulation

105 The development of the equations for extracting the optical constants treat the system as a series of stratified
106 parallel optical boundaries as depicted in **Figure 1**. The propagating radiation can pass through n phases, where
107 phase j is described by the complex refractive index $\hat{n}_j(\nu)$ given by :

$$108 \quad \hat{n}_j(\nu) = \sqrt{\hat{\epsilon}_j(\nu)} = n_j(\nu) + ik_j(\nu) \quad (1)$$

109 Where $n_j(\nu)$ is the real refractive index of the medium in phase j, $k_j(\nu)$ is the absorption index and $\hat{\epsilon}_j(\nu)$ is the
110 related complex dielectric function. To simplify the notation, the frequency dependence of the optical constants
111 will be deliberately omitted. In an ATR experiment as depicted in **Figure 1**, phase 1 corresponds to the ATR IRE
112 which is optically described by a constant real refractive index assuming no absorption of the incident

113 electromagnetic radiation ($n_1 = 4$ for Germanium IRE and $n_1 = 2.4$ for Diamond and ZnSe IRE). Phase 2
 114 corresponds to the sample film, optically defined by a complex optical constant. In the case where the sample
 115 thickness is below the penetration depth of the evanescent wave, the optical constants of phase 3 needs to be
 116 explicitly described i.e. assuming air ($n_3 = 1$) or water (defined with complex \widehat{n}_3) for example. The extent to which
 117 radiation is reflected or transmitted at an interface depends on the light polarization, the angle of incidence of the
 118 incoming radiation and the optical constants of the phases at the interface. The observable in an ATR experiment
 119 is the reflectance spectrum, which is determined from the Fresnel amplitude coefficient, \widehat{r}_{jk}^i :

$$120 \quad R_i = |\widehat{r}_{jk}^i|^2 = \frac{E_r^2}{E_{inc}^2} \text{ with } i = s, p \quad (2)$$

121 The script i refers to the polarization direction which is either s when the electric field vector is perpendicular to
 122 the incident plane or p for a parallel electric field vector. The Fresnel coefficients are defined as the ratio of the
 123 complex amplitude of the electric field vectors of the reflected waves, E_r , to the one of the incident waves, E_{inc} .
 124 They can be calculated at the interface between phases j and k , if the incident angle to this interface, α_j , and the
 125 optical constants of phase j and k are known according to :

$$126 \quad \widehat{r}_{jk}^s = \frac{\xi_j - \xi_k}{\xi_j + \xi_k} \text{ and } \widehat{r}_{jk}^p = \frac{\widehat{n}_k^2 \xi_j - \widehat{n}_j^2 \xi_k}{\widehat{n}_k^2 \xi_j + \widehat{n}_j^2 \xi_k} \quad (3)$$

127 Where $\xi_{j,k}^i = \widehat{n}_{j,k} \cos \alpha_{j,k} = \sqrt{(\widehat{n}_{j,k}^2 - n_1^2 \sin^2 \alpha_1)}$ can be regarded as the refractive coefficient for phase j or k .

128 Here, α_1 , corresponds to the angle of incidence which is set at 45° in our study. Equations 2 and 3 allow to compute
 129 the reflectance spectra of a two-phase system, i.e. when the thickness of the sample filmed on the ATR IRE is
 130 higher than the evanescent wave penetration depth (few microns). In the case where a thin film is deposited on the
 131 ATR IRE, more than one interfaces have to be described and the incident radiation undergoes multiple reflections
 132 at each interface. Therefore, the amplitude of the reflected radiation, $\widehat{r}_{123}^{s,p}$, is calculated by the summation of the
 133 reflected rays of the multiple reflection at each interface.

$$134 \quad \widehat{r}_{123}^{s,p} = \frac{r_{12}^{s,p} + r_{23}^{s,p} \exp(2i\delta_{s,p})}{1 + r_{12}^{s,p} r_{23}^{s,p} \exp(2i\delta_{s,p})} \quad (4)$$

135 Where $\delta_i = \frac{2\pi * d * \xi_2^i}{\lambda}$ indicates the change of the phase of the wave between phase 2 and 3 which depends on the

136 film thickness, d . Therefore, the reflectance spectra of a three phase system, $R = \left| \frac{\widehat{r}_{123}^{s,p}}{r_{13}^{s,p}} \right|^2$ can be computed for

137 various thicknesses, d , if \widehat{n}_j of the three phases are known. For total reflectance, $\widehat{r}_{13}^{s,p} = 1$, the reflectance spectra

138 for ATR geometry is therefore easily obtained from equation 4. It is noteworthy that this method can be generalized
 139 to anisotropic layers. In this case, \hat{n}_j must be replaced by \hat{n}_y and \hat{n}_x for s and p polarization, respectively. The
 140 refractive index for the p polarization become , $\xi_j^p = \sqrt{\frac{(\hat{n}_x^2 - \hat{n}_x^2 n_1^2 \sin^2 \alpha_1)}{\hat{n}_z^2}}$. Therefore, spectral simulations can also
 141 be carried out for uniaxial ($\hat{n}_x = \hat{n}_y \neq \hat{n}_z$) and biaxial symmetry ($\hat{n}_x \neq \hat{n}_y \neq \hat{n}_z$).

142 Determination of the optical constants of uniaxial thick film.

143 The following method is applicable only if the stratified model is composed of a single interface, and therefore,
 144 the thickness of the sample is larger than the penetration depth (few microns thick). In the condition of total
 145 reflection when the angle of incidence is greater than the critical angle, $(\xi_j^i)^2 = (\hat{n}_j^2 - n_1^2 \sin^2 \alpha_1)$ become negative
 146 even for non-absorbing sample ($k_j \rightarrow 0$) and therefore the Fresnel equation (Equation 3) becomes a complex
 147 number that can be written with polar coordinate as :

$$148 \quad \hat{r}_{jk}^i = |\hat{r}_{jk}^i| \exp(i\delta_i) = R_i^{1/2} \exp(i\delta_i) \quad (5)$$

149 Where $\delta_i = \arg(\hat{r}_{jk}^i)$ correspond to the phase shift. Therefore, in the ATR mode, the incident electric fields undergo
 150 phase shifts upon reflection even for non-absorbing sample, in contrast to external reflection at normal incidence.
 151 Given that \hat{r}_{jk}^i is function of the optical constants n_j and k_j , $\ln(R^{1/2})$ and δ_i must satisfy the Kramers-Kronig
 152 relations. This method was successfully applied by Karakassides *et al.* for recovering the optical constants from
 153 the external reflectance spectra obtained at quasi-normal incidence (Karakassides et al. 1997). Unlike normal
 154 reflectance experiments, in the case of ATR and because of the phase shift induced by reflections under total
 155 reflection conditions, the δ_i determined from $\ln(R_i^{1/2})$ has to be corrected (Plaskett and Schatz 1963; Bardwell and
 156 Dignam 1985; Dignam 1988; Dignam and Mamiche-Afara 1988; Ohta and Ishida 1988; Buffeteau et al. 1999;
 157 Boulet-Audet et al. 2010). Dignam et al showed that this correction factor, $I_{s,p}$, takes a particularly simple form
 158 that can be readily estimated. The phase shift for the ATR geometry is therefore expressed as (Dignam and
 159 Mamiche-Afara 1988) :

$$160 \quad \delta_s(v_a) = I_s - \frac{2}{\pi} P \int_0^\infty \frac{v \ln(R_s)^{1/2}}{v - v_a} dv \quad \text{and} \quad \delta_p(v_a) = I_p - \frac{2}{\pi} P \int_0^\infty \frac{v \ln(R_p)^{1/2}}{v - v_a} dv \quad (6)$$

$$161 \quad \text{for the s-polarisation : } I_s = -2 \arctan \frac{\sqrt{n_1^2 \sin^2 \alpha_1 - n_\infty}}{n_1 \cos \alpha_1} \quad (6a)$$

$$162 \quad \text{for the p-polarisation : } I_p = -2 \arctan \frac{n_1 \sqrt{n_1^2 \sin^2 \alpha_1 - n_\infty}}{n_{x,y} n_z \cos \alpha_1} \quad (6b)$$

163 where n_∞ represents the refractive index of the sample at infinite wavenumber in a region of zero absorbance. This
 164 value was estimated based on a best fit approach between the simulated and experimental ATR spectra for various
 165 film thicknesses. The best agreement was obtained by setting $n_\infty = 1.4$. The calculation of the phase shift
 166 according to Equation 6 requires reflectivity data over the complete frequency range, zero to infinity. In practice,
 167 the spectra are obtained on a limited frequency range. Therefore, the truncation at high and low wavenumber
 168 imposes a limitation on the accuracy achievable using the KK transformation but should be relatively small
 169 because the experimental values of $\ln(R^{1/2})$ at the cut-off wavenumbers are very small and constant. The KK
 170 transformation was numerically implemented by the Maclaurin formula providing a better accuracy than the
 171 double Fourier transform (Ohta and Ishida 1988; Huang and Urban 1992). The Fresnel reflection coefficient was
 172 then calculated from Equation 5 and the in-plane, $\widehat{\varepsilon}_{xy}$ and out-of-plane, $\widehat{\varepsilon}_z$ complex dielectric function was
 173 obtained through the following relations:

$$174 \quad \widehat{\varepsilon}_{xy}(\nu) = n_1^2 \left[\sin^2 \alpha_1 + \left(\frac{1 - \widehat{r}_s(\nu)}{1 + \widehat{r}_s(\nu)} \right)^2 \cos^2 \alpha_1 \right] \quad (7a)$$

$$175 \quad \widehat{\varepsilon}_z(\nu) = \frac{n_1^2 \sin^2 \alpha_1}{1 + \left(\frac{\widehat{\varepsilon}_{xy}}{n_1^2} \right) \left(\frac{1 - \widehat{r}_p(\nu)}{1 + \widehat{r}_p(\nu)} \right)^2 \cos^2 \alpha_1} \quad (7b)$$

176 Then, \widehat{n}_x , \widehat{n}_y and \widehat{n}_z are directly determined using Equation 1. It is noteworthy that using solely the s-polarized
 177 spectra for the KK transform according to the corresponding equations allows the determination of the isotropic
 178 optical constants.

179 MATERIALS AND METHODS

180 *Clay minerals*

181 The SWy-3 montmorillonite was obtained from the Source Clays Repository of the Clay Minerals Society. The
 182 <1 μm size fraction was collected by centrifugation and Na-saturated using three saturation cycles in a 1 M NaCl
 183 solution. The excess of salt was removed by dialysis until Cl⁻ anion was no more detected by AgNO₃ test. The
 184 concentration of the resulting suspensions of Na-SWy-3 montmorillonite is 24.7 g/L.

185 *Spectral acquisition*

186 All spectra were recorded with a Nicolet IS50 Fourier transform infrared spectrometer (ThermoScientific)
 187 equipped with a DLaTGS detector, and a KBr beamsplitter. ATR spectra were acquired with the VeeMaxIII
 188 accessory (Pike Technologies) fitted with a 45° angle germanium IRE having a crystal diameter of 20 mm and

189 giving a penetration depth of 650 nm at 1000 cm⁻¹ (assuming $n_2 = 1.4$). The germanium IRE was
190 thermoregulated at 25°C for all experiments. The angle of incidence, α_1 , was set to 45°. The electric field of the
191 incident infrared beam was polarized either perpendicular (s) or parallel (p) to the plane of incidence using a ZnSe
192 wire grid polarizer placed in the optical beam path before the sample compartment. Given that for a 45° incidence
193 angle and isotropic samples, the intensity ratio $A_p/A_s = 2$, the efficiency of the polarization was checked on a
194 methanol solution deposited on the ATR IRE and tightly covered to prevent evaporation. In our experimental
195 setup, the A_p/A_s ratio was measured to be 1.97 suggesting an efficient polarization of the electric field. To
196 compensate for this deviation, a factor of 1.016 was applied to all p-polarized spectra.

197 Spectra were acquired with a 4 cm⁻¹ resolution and 100 scans were co-added in the range 4500 – 400 cm⁻¹ using
198 a Happ-Genzel apodization. The spectra were never smoothed or deconvolved. The only correction was a baseline
199 adjustment in some cases. Background spectra were acquired in the same condition as for sample using s and p
200 polarized radiation.

201 *Spectral simulation*

202 The computed program for the calculation of ATR spectra of anisotropic layers was realized using MatLab©
203 (Software version R2017b). Some input parameters had to be introduced such as the incidence angle set at 45°,
204 the electric field polarisation (i.e. s or p), the number of layers (i.e. 2 or 3), and the IRE refractive index (2.4 and
205 4 for diamond and germanium IRE, respectively).

206 *Film preparation*

207 The simulation of ATR spectra under various experimental conditions required the prior determination of the
208 optical constants of the SWy – 3 Montmorillonite film. A volume of 500 µL of the stock dispersion (24,7 g/L) was
209 firstly deposited on the Ge IRE and allowed to dry at 25 °C. Then, s and p polarized spectra were acquired. This
210 methodology ensures that the thickness of the sample film is much larger than the penetration depth (few microns
211 thick). The as-obtained s and p-polarized spectra are then used for the extraction of the anisotropic optical constants
212 ($\hat{n}_x = \hat{n}_y \neq \hat{n}_z$) using Equations 5,6 and 7.

213 The as-determined optical constants are further used for the simulation of ATR spectra on films of various
214 thicknesses which are compared to experimental ones. The thickness of the experimental film, d , was estimated
215 based on the Na-SWy-3 Montmorillonite formula, assuming one layer of water as expected at room temperature
216 and ambient humidity, corresponding to an estimated density of 2.3 g/cm³. The germanium IRE having a surface

217 area of 314 mm², a volume of 100 μL was pipetted from diluted suspensions prepared from the SWy-3 stock
218 dispersion and was dropped on the IRE surface for targeting film thicknesses ranging from about 7 to 230 nm.
219 Because the germanium surface is hydrophobic, the original aqueous suspension was diluted by methanol resulting
220 in better wettability and ensuring homogeneous film thickness over the IRE surface. When the methanol was fully
221 evaporated and the sample was equilibrated to ambient humidity, s and p polarized spectra were acquired. The
222 estimated thickness was input in the model for the simulation of the ATR spectra. Based on a best fit approach, a
223 variation of 10% of the targeted d value was allowed accounting for changes in surface rugosity and porosity and
224 to compensate for measurement precision and accuracy during suspension preparation.

225 For one experiment, a film with an estimated thickness of 14 nm was firstly deposited on the Ge IRE, and
226 washed with deionized water several times until no more changes in the vibrational intensity of the SWy-3 film
227 were observed. ATR spectral simulation was performed by adjusting the thickness of the sample in the simulation
228 until qualitative good match is obtained between the experimental and calculated intensity of the clay mineral film.
229 The best fit was obtained with an input thickness of 7 nm. In order to evaluate the influence of water on the top of
230 clay film, a drop of deionized water was added and polarized spectra were acquired where the corresponding
231 polarized background spectra was only bulk water.

232 RESULTS AND DISCUSSION

233 In order to confirm the optical anisotropy of clay platelet, FTIR transmission measurement was firstly
234 performed. The absorption intensity of an infrared radiation relies on a scalar product of the vibrational transition
235 moment amplitude and the amplitude of the electric field of the radiation. Accordingly, absorption bands are
236 observable only if the angle between the electric vector and the transition moment is different from 90°. In the
237 case of oriented samples measured through transmission at normal incidence with s- and p-polarized radiation,
238 only transition moments within the plane of the absorbate result in absorption (Margulies et al. 1988). The
239 transmission spectra of Na-SWy-3 sample oriented by drop-casting on a ZnSe substrate and acquired at normal
240 incidence using polarized radiation in the spectral range 700-1400 cm⁻¹ (**Fig. 2a**), show the vibrational signatures
241 of the clay layers. Roughly, the region between 700-950 corresponds to hydroxyl bending vibrations from the
242 octahedral sheet, whereas the region 1000-1400 cm⁻¹ is usually assigned to Si-O stretching vibration of the
243 tetrahedral sheet (Farmer and Russell 1964). For both s- and p-polarization, the absorbance spectra are similar,
244 suggesting that the in-plane transition moments are isotropically distributed within the plane of the sample. Taking
245 advantage of the Berreman effect, tilting the sample with respect to the radiation at an angle of 40° enables the
246 out-of-plane transition moment to be excited (Berreman 1963). The absorbance spectra (**Fig. 2b**) display noticeable

247 differences around 1085 cm^{-1} that correspond to a transition moment almost perpendicular to the film plane.
248 Samples with transition moments that are isotropically distributed within the sample plane and anisotropically
249 distributed in the perpendicular direction are called transverse isotropic systems. As a consequence, a proper
250 description of the optical properties of clay minerals requires the determination of the optical constants in an
251 uniaxial symmetry ($\widehat{n}_{xy} \neq \widehat{n}_z$).

252 *Determination of anisotropic optical constants from polarized ATR measurements*

253 The advantages of ATR geometry enable the computation of both the in-plane \widehat{n}_{xy} and out-of-plane optical
254 constants \widehat{n}_z on a thick film of Na-SWy-3 montmorillonite sample with a single preparation. As stated in the
255 introduction, the Fresnel reflection coefficients were determined from the experimental ATR reflectance spectra
256 (**Fig. 3a**) by taking advantage of the Kramers-Kronig relationship (Equation 5,6). As a result, no fitting procedure
257 is required because all calculations are based on the recorded reflectance spectra and the ATR experimental setup.
258 While the dielectric function described the response of the materials to an electromagnetic radiation, the exact
259 resonance frequency of vibrational modes cannot be accessed experimentally. Indeed, even if the position of
260 reflectance bands is close to the wavelength of absorption, a shift in position is expected to occur accounted for
261 porosity, particle shape and size that affect the dielectric function. The dielectric function determined from the
262 Fresnel coefficients using Equations 7 has to be considered as a macroscopic function that may be different from
263 the microscopic dielectric function which depends only on the resonant vibrational frequency of an IR-active
264 mode. Both macroscopic and microscopic dielectric properties can be connected using an effective medium
265 approach although retrieving the intrinsic dielectric functions from the measured ATR spectra and the Fresnel
266 reflection coefficient is a challenging task. Alternatively, a modelling strategy has been developed by Balan et al
267 for calculating the ATR spectra of powder samples by ATR spectroscopy (Balan et al. 2002, 2011; Aufort et al.
268 2016). It consists in calculating the ATR spectra from an oscillator model using vibrational mode frequencies
269 determined by first principle quantum mechanics to derive the microscopic dielectric functions. An approximate
270 macroscopic dielectric function is then calculated by considering that the substance of interest is embedded in a
271 matrix which is described by its own dielectric property. Given that the determined macroscopic dielectric
272 properties is a “fingerprint” of the sample being analyzed, the reported optical constants in this study are solely
273 those of the film which is likely different from the mineral itself. The in-plane \widehat{n}_{xy} and out-of-plane optical
274 constants \widehat{n}_z (**Fig. 3b,c**) was determined from the anisotropic dielectric function using Equation 1. The optical
275 constants, n and k being related to each other by the Kramers-Kronig relationship, the n spectrum follows a

276 derivative shape at the maximum intensity of the k spectrum. Given that the determination of \widehat{n}_{xy} and \widehat{n}_z has
277 never been performed on clay minerals in the infrared region, comparison with already published optical constants
278 can only be achieved by averaging the anisotropic optical constants to obtain the isotropic optical constants \widehat{n}_2 .
279 The \widehat{n}_2 value determined in this work are slightly different from those already published based on the Drude-
280 Lorentz dispersion theory (Roush et al. 1991; Esposito et al. 2000; Glotch and Rossman 2009). Given that the
281 optical constants are always determined from the recorded reflectance spectra, it is very likely that the
282 discrepancies originate from the sample preparation and the calculation methods.

283 Strong dispersion of n_{xy} associated with intense absorption of k_{xy} is observed in the Si-O stretching region at
284 1039 cm^{-1} . the anomalous dispersion of the n_{xy} curve reaches a maximum value of 2.73 at 1004 cm^{-1} (**Fig. 3b**).
285 Recalling that total reflection occurs only when the incident angle is greater than the critical angle, one may
286 calculate the maximum value that n_2 can reach to satisfy this condition. For Ge IRE ($n_1 = 4$) and assuming an
287 incident angle of 45° , the maximum n_2 value is 2.83 while for diamond IRE ($n_1 = 2.4$) this value was found to be
288 1.70. Because the experimental value of 2.73 is below the maximum value for Ge IRE, the condition of total
289 reflection is respected for both polarizations and within the whole wavelength range investigated. Using diamond
290 or ZnSe IRE is expected to be more problematic for reliable ATR measurements because the condition of total
291 reflection is no more fulfilled and the critical angle will be higher than the incident angle in the wavenumber range
292 $850\text{-}1030\text{ cm}^{-1}$. It is therefore predictable that frequency shift, or distorted band would appear on the ATR spectra
293 when using a low refractive index crystal like diamond or ZnSe. The easiest and more convenient way to reduce
294 these effects is to select high refractive index crystal, such as Ge IRE, to ensure that the angle of incidence will be
295 well above the critical angle (Boulet-Audet et al. 2010). It is noteworthy that verifying the optical conditions for
296 ATR measurement is a prerequisite for quantitative analysis of ATR spectra.

297 The maxima in the k_{xy} and k_z spectra are directly related to the magnitude of the transition moment (Wilson
298 Edgar 1980) and to its direction (Fraser and MacRae 1973). Therefore, all the information about particle orientation
299 is inherently included in the in-plane, k_{xy} and out-of-plane, k_z extinction coefficients. Examining **Fig. 3c**,
300 vibrations in the range $1000 - 1200\text{ cm}^{-1}$ assigned to tetrahedral sheet transition moments do not present clear
301 orientation because each vibrational mode includes in-plane and out-of-plane components. The bending of the
302 hydroxyl groups in the octahedral sheet decomposes in four different vibrations usually assigned to AlAlOH,
303 FeAlOH, MgAlOH and MgFeOH located at $919, 883, 842$ and 800 cm^{-1} respectively. These bands are highly
304 responsive to polarization with some having mainly in-plane (MgAlOH and AlAlOH) or out-of-plane (FeAlOH

305 and MgFeOH) transition moments in good agreement with the results from Ras et al (Ras et al. 2003). These
306 authors rationalized their results by suggesting that the orientation of the hydroxyl groups depend on the nature of
307 the nearest cations in the octahedral sheets. Although the nature of the octahedral cations may substantially
308 influence hydroxyls orientations, these modes are likely coupled with other vibrations in the tetrahedral units that
309 may alter their responses toward polarization (Larentzos et al. 2007).

310 *Spectral simulations from the determined optical constants*

311 Now that phase 2 in **Fig. 1** is optically defined by the determined \widehat{n}_{xy} and \widehat{n}_z values on Na-SWy-3 film,
312 exact spectral simulations can be performed to examine the spectral profiles under various conditions. Given that
313 ultra-thin film in combination to surface-sensitive techniques can provide valuable information about interface
314 phenomena, emphasis is given to the influence and benefits of different optical configurations (e.g. transmission,
315 internal or external reflections, etc.) that could give complementary information or improved sensitivity when used
316 together. Focusing more specifically to ATR methods, the spectral simulation allows to estimate the sample
317 thickness and to highlight changes on the spectral profile as due to the variation of the sample thickness. On a final
318 example, the influence of water or air as the top layer on the ATR spectra will be quantitatively investigated.

319 *Choosing the appropriate optical configuration for Na-SWy-3 thin film analysis:* Because the wavelength of
320 infrared radiation (2-20 μm) is much greater than the dimensions of the crystal unit cell, only long-wavelength
321 lattice modes interact with the incident light. These optical modes are described with respect to the phonon
322 propagation direction: the transverse optical modes, TO, in which the atomic displacements are perpendicular to
323 the direction of the periodicity of the wave-vector and the longitudinal vibration (LO) where the displacements are
324 parallel to the wave vector. The frequencies of the LO and TO phonon modes are defined with respect to the
325 intrinsic dielectric function. Because of the macroscopic nature of the dielectric function obtained through the KK
326 transform, only the TO_i and LO_i energy loss functions, defined as $\text{imag}(\widehat{\varepsilon}_i(\nu))$ and $\text{imag}\left(\frac{-1}{\widehat{\varepsilon}_i(\nu)}\right)$, respectively
327 are accessible, and described the TO and LO surface modes of a thin crystallite layer (**Fig. 4a,b**). They are
328 conveniently used for describing the spectral shape depending on the optical configuration regardless of whether
329 the TO and LO phonon modes are known. Therefore, the determination of the TO and LO energy loss functions
330 could be a precious help for adequately choosing the appropriate optical configuration thanks to the various surface
331 selection rules observed on thin film. For example, simulated spectra under various optical configurations
332 assuming a film thickness, d , of 20 nm was calculated with Equation 2-4 (**Fig. 4c-g**). Because of the transverse
333 nature of electromagnetic radiation, infrared optical measurement of uniaxial oriented thin film using either s- and

334 p-polarized light only activates in-plane TO_{xy} and LO_z surface modes from surface-parallel or surface-
335 perpendicular transition moments whichever the experimental setup used. Therefore, the TO_{xy} and LO_z energy loss
336 function are plotted along the calculated spectra to better identify the nature of the transition moment involved as
337 a function of the optical configuration. As far as normal transmission is concerned, the simulated spectrum profile
338 is close to the TO_{xy} energy loss function and therefore, this optical method activate only surface-parallel transition
339 moments, independently of the polarization (**Fig. 4c**). In a very similar way, the same information can be obtained
340 from external reflectance measurement at normal (or near-normal) incidence (**Fig. 4d**). These two optical methods
341 are sometimes used for calculating the in-plane optical constants \hat{n}_{xy} . (Berreman 1963) demonstrated that
342 acquisition of IR transmittance spectra at off normal incidence allowed the activation of LO surface modes using
343 p-polarized radiation (**Fig. 4e**). Indeed, at oblique incidence, the p-polarized radiation gives rise to two components
344 that are parallel and perpendicular to the film surface and hence, excites both TO_{xy} and LO_z surface modes
345 respectively. Both surface-parallel and surface-perpendicular components of the transition moment are then
346 activated and the spectral profile is therefore a combination of the TO_{xy} and LO_z energy loss functions. Using s-
347 polarized radiation enables to record only surface-parallel component of a transition moment very similarly to
348 what is obtained using classical transmission techniques. This is very similar to ATR spectroscopy, and further
349 demonstrates the benefit of this technique for extracting anisotropic optical constant or for carrying out
350 orientational measurements (**Fig. 4e,f**). Reflection-absorption at grazing angle on thin film deposited on metal
351 substrate may be of interest because in this condition the p-polarized radiation couple with the surface-
352 perpendicular component of the transition moment (Ras et al. 2004a). The spectra therefore is essentially driven
353 by the LO_z energy loss function (**Fig. 4g**). Only the most classical optical configurations are illustrated here.
354 However, the methodology based on the calculations of the Fresnel coefficient enable to determine the best
355 experimental conditions simply by changing some variables like film thickness, incidence angle, etc.

356 *Estimation of sample thickness and its influence on the ATR spectra:* From the determination of the in-plane and
357 out-of-plane optical constants of the studied montmorillonite, infrared spectra of multi-layer systems can be
358 computed for a variety of incidence angle, polarization, layer thickness conditions using the formalism detailed in
359 the theory section. The benefit of such simulations lies in a better understanding of the origin of the experimental
360 spectra profile, and allows to evaluate whether the observed spectral changes originate from optical effects or
361 chemical effects resulting from specific interaction or ordering. To illustrate this point, clay films of different
362 thicknesses were deposited on the IRE surface and the polarized spectra were acquired and compared to the
363 calculated ones where the estimated thickness was input in the model (**Fig. 5**). To account for potential changes in

364 particle orientation as a function of the film thickness, the s- and p-polarised ATR spectra, A_s and A_p , were
365 converted to isotropic spectra, A_0 , according to the relation:

$$366 \quad A_0 = \frac{A_p + 2A_s}{3} \quad (27)$$

367 The calculated spectra and the experimental ones are in a fairly good agreement suggesting that the calculation
368 method of \hat{n}_{xy} and \hat{n}_z is accurate (**Fig. 5**). Thus, the variation in position of the vibrational bands account mainly
369 for optical interferences within the film thickness. Deviations between the calculated and experimental spectra
370 may arise from the surface rugosity and probably from the variation of film porosity for different thicknesses which
371 are not explicitly taken into account in the simulations.

372 Using the n_2 and k_2 optical constants, s- and p- polarised ATR spectra were simulated for various film
373 thicknesses (**Fig. 6**). In this condition, the orientation of the clay particles is reflected in the determined anisotropic
374 optical constants and implicitly taken into account. The spectral profiles for s- and p- polarization display
375 appreciable variations as the thickness of the clay film increases. The origin of these changes is only attributed to
376 optical effects which depends on the surface to volume ratio of the film as well as the electric field amplitude along
377 the film thickness (Harrick 1965). Interestingly, a mode appears for thin film at 1085 cm^{-1} which shifts to lower
378 wavenumber when the thickness increases. The frequency of this mode is correlated to the LO_z surface modes of
379 Si-O describing the vibration of the silicon with apical oxygen (**Fig. 4b**). This mode being strongly z-oriented has
380 been used to determine the orientation of particle with respect to the laboratory frame (Johnston and Premachandra
381 2001; Ras et al. 2003, 2007a; Amarasinghe et al. 2008).

382 This mode shifts linearly to lower wavenumber when the film thickness increases in the range 1-150 nm (**Fig. 7a**).
383 If an electromagnetic radiation propagates along an oriented film (and the electric field is perpendicular to the film
384 surface), positive ions and negative ions will be displaced to one and the other extremity of the film respectively.
385 This separation of charges creates an additional electric field perpendicular to the film plane whose frequency
386 varies with the incident wave frequency. This charge displacements contribute to an additional restoring force with
387 respect to their equilibrium position. This force can be assimilated to the one perpendicular to the front of a
388 longitudinal wave. Therefore, the frequency of this transverse wave will be close to the LO surface modes. When
389 a thick film is deposited on an IRE, the frequency of LO surface modes decreases and shifts toward the TO surface
390 modes because the contribution of surfaces charges to the created internal field decreases. When the electric field
391 is parallel to the film plane (the wave propagates in the normal direction to the film surface), no additional field

392 arises because the crystal is considered as infinite plane, so the frequency corresponds to that of the transverse
393 vibration in the infinite crystal.

394 Note that the intensity of this LO surface mode also varies with the film thickness modifying the dichroic ratio
395 (A_s/A_p) although the orientation of the particle is unchanged. As a consequence, the sample film thickness needs
396 to be carefully controlled if the determination of particle orientation in thin clay film is targeted. In order to
397 experimentally determine the film thickness, the most intense Si-O stretching mode at 1039 cm^{-1} can be used. The
398 absorbance is linearly correlated to the film thickness in a rather narrow range from 1 to 100 nm (**Fig. 7b**). This
399 contrasts with transmission measurements where the absorbance is proportional to the concentration of particles.
400 The methodology presented here based on the prior determination of \hat{n}_{xy} and \hat{n}_z thus represents an efficient way
401 to estimate the film thickness (**Fig. 5**). Alternatively, a modified form of Beer-Lambert law accounting for
402 reflectivity is often used in literature.

$$403 \quad -\log R = \alpha d_e \quad (28)$$

404 The application of this equation requires the determination of the Napierian absorption coefficient, α , and the
405 effective thickness, d_e . The effective thickness introduced by Harrick is defined as the hypothetical thickness that
406 would result in the same absorbance value in a transmission measurement as obtained from ATR experiment
407 (Harrick and Beckmann 1974). The effective thickness is actually much greater or much less than the actual film
408 thickness because it only represents an equivalent pathlength in a transmission spectrum. Owing to the exponential
409 decay of the electric field along the film thickness, the determination of the electric field amplitude need to be
410 estimated (Harrick 1965). The approximate equations proposed by Harrick for the determination of the electric
411 field amplitude are only applicable for very weak absorbers (k tends to zero) which is not the case of the Si-O
412 stretching vibrations (**Fig. 3c**).

413 The Napierian absorption coefficient, α , is a more reliable parameter that can be directly compared to transmission
414 measurements. It is defined by :

$$415 \quad \alpha_{Si-O}^s = 4\pi k_y \nu \quad (29)$$

416 In order to enable comparison with transmission measurement, and recalling that only in-plane vibrational
417 modes are accessible at normal incidence transmission, the extinction coefficient, k_y , has to be selected. From
418 Equation 29, the experimentally determined absorption coefficient for the Si-O band is $2.94 \times 10^4\text{ cm}^{-1}$. For
419 comparison, a value of $3.60 \times 10^4\text{ cm}^{-1}$ were reported by Johnston (Johnston and Premachandra 2001). These

420 authors determined the absorption coefficient by transmission method where the Na-SWy-3 montmorillonite film
421 thickness was evaluated by weighting the dried residue on an IR transparent substrate of known dimensions. The
422 slight difference in the determination of the absorption coefficient with these two methods is likely due to particle
423 orientations. Note that the methodology presented here requires the preparation of a single sample, whose thickness
424 is controlled by the limited penetration depth of the evanescent wave. Therefore, the sample thickness can be
425 assumed as homogeneous through the sample, well above the penetration depth for the entire wavelength range.
426 For a 1 nm thick film, corresponding to the thickness of a unique smectite layer, intensity of the Si-O vibration is
427 calculated to be 1.6×10^{-3} which confirms the value of 1.5×10^{-3} determined by Johnston et al. and Ras et al. by
428 transmission measurements (Johnston and Premachandra 2001; Ras et al. 2003). As suggested by these authors,
429 single layer of smectite is measurable through modern FTIR spectrometer, where the sensitivity is in the range 1
430 $\times 10^{-5}$. The very similar results obtained by ATR and transmission methods clearly demonstrates that ATR
431 spectroscopy is a versatile method enabling quantitative measurements.

432 *Contribution of surrounding medium to ATR spectra: air and water:* The benefit of ATR spectroscopy lies in the
433 possibility of in-situ analysis of dynamical processes. Typically, a liquid cell is mounted on the ATR IRE where a
434 thin film of a material was previously deposited on. Solutions of interest are then injected through the liquid cell
435 and spectra are recorded to monitor the interactions between the film and the solute (Schampera and Dultz 2009;
436 Lefèvre et al. 2012; Schampera et al. 2016). For quantitative analysis of these interactions, the effect of the solution
437 on top of the sample on the ATR spectra of materials has to be determined. Intuitively, the presence of water
438 instead of air on top of the deposited film at the IRE surface contributes to the variation of the Fresnel reflection
439 coefficient which is correlated with changes of the optical indices at the interfaces according to Snell law. In order
440 to ascertain whether vibrational changes are correlated to interactions at the solid/liquid interface, optical effects
441 have to be firstly determined.

442 The calculation of s- and p-polarized spectra of a thin montmorillonite layer deposited on a Ge IRE was performed
443 with an estimated thickness of 7 nm (**Fig. 8a**). Given that this thickness is below the penetration depth of the
444 evanescent wave, a third semi-infinite layer of either air or water was considered on the top of the sample film.
445 The optical constants of water were determined using the same methods as in the theory section on a thick layer
446 of water. While the position of vibrational bands is strictly identical for both models, clear modification of band
447 intensity is observed for some specific bands. The out-of-plane Si-O stretching located at 1085 cm^{-1} appears with
448 a higher intensity when water is considered as the third layer. This variation is associated to drastic changes of the
449 mean squared electric field components along the z axis. Therefore, only dipole moments having non-zero

450 component in the z direction will be affected by changes of the refractive index of the third layer. Unlike the s-
451 polarized spectra where only in-plane transition moments are excited, the influence of a water layer instead of air
452 layer can only be observed on the p-polarized spectrum (**Fig. 8b**). The intensity of the strongly polarized out-of-
453 plane Si-O stretching is enhanced when the film is immersed in water. However, these results have to be considered
454 in a qualitative way. Indeed, the simulation of the thin layer assumed that the orientation of particles is similar to
455 that of the bulk material from which the anisotropic optical constants, \hat{n}_{xy} and \hat{n}_z were determined. The
456 discrepancies between the simulated and experimental spectra account for this variation of particle organisation
457 which is well reflected in the intensity of the in plane and out-of-plane Si-O band.

458 CONCLUSIONS

459 ATR-FTIR technique is a powerful method for studying interactions at the interface between clay surface and
460 solute at the molecular scale. Compared to transmission measurement, ATR spectra appear more complicated
461 because of the significant contribution of optical effects. In this paper, a general formalism to simulate ATR spectra
462 for clay minerals relying on the experimental determination of the optical constants is introduced. In combination
463 to polarized incident radiation, the anisotropic optical constant in the infrared region has been determined for the
464 first time on montmorillonite film. These optical constants allow for the quantitative determination of the sample
465 thickness, and to assess the effect of experimental conditions on the recorded spectra. The formalism being solely
466 based on the classical theory of electromagnetic wave propagation in layered media, differences between the
467 calculated and the experimental spectra allow optical and chemical effects to be discriminated.

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473 manuscript.

474 **Conflict of Interest.**

475 The authors declare that they have no conflict of interest

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581

582

583

584 Figure Captions

585 **Fig. 1** Schematics of the three-phase system. Notations are defined in the text.

586 **Fig. 2** a) Normal and b) 40° off-normal transmission spectra of Na-SWy-3 montmorillonite ed on ZnSe substrate with s- (green) and p- (blue)
587 polarization.

588 **Fig. 3** a) ATR reflectance spectra on a thick Na-SWy-3 montmorillonite film and the determined (b) real part, n_2 and (c) imaginary part k_2 of
589 the anisotropic optical constants. The dashed line in Fig. 3b represents the maximum n_2 value permitted to satisfy the total reflection conditions
590 using Ge ATR.

591 **Fig. 4** TO and LO energy loss functions for a) in-plane and b) out-of-plane components. Spectral simulations of a 20 nm thick Na-SWy-3 film
592 under different optical configurations: c) Normal transmission, d) External reflection on dielectric substrate (SiO₂), e) Off-normal transmission,
593 f) ATR, g) External reflection at grazing angle on metallic substrate (gold). The transmission Fresnel coefficients were taken from (Milosevic
594 2004)

595 **Fig. 5** Comparison of experimental spectra of variable thickness Na-SWy-3 montmorillonite films on Ge IRE and calculated ones using the
596 three layer model described in the theory section for the same estimated sample thicknesses. From bottom to top, film thickness is 14, 22, 33,
597 63, 120 and 230 nm.

598 **Fig. 6** Calculated s- and p-polarised spectra of Na-SWy-3 Montmorillonite of different thicknesses on Ge IRE. For the sake of clarity, all
599 spectra are normalized to the most intense vibrational band.

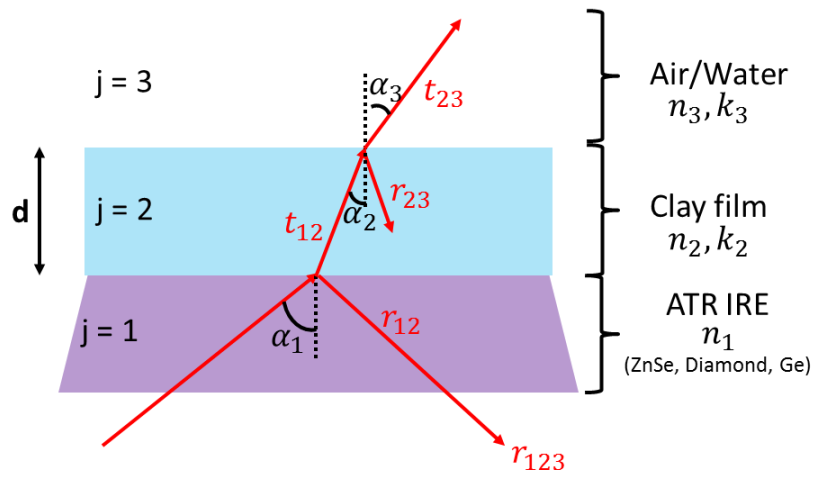
600 **Fig. 7** Thickness dependence evolution of a) LO_z surface mode frequency from 1085 cm⁻¹ and b) TO_{xy} surface mode intensity at 1039 cm⁻¹

601 **Fig. 8** a) Calculated and b) Experimental spectra of thin Na-SWy-3 montmorillonite film at the interface between air or water.

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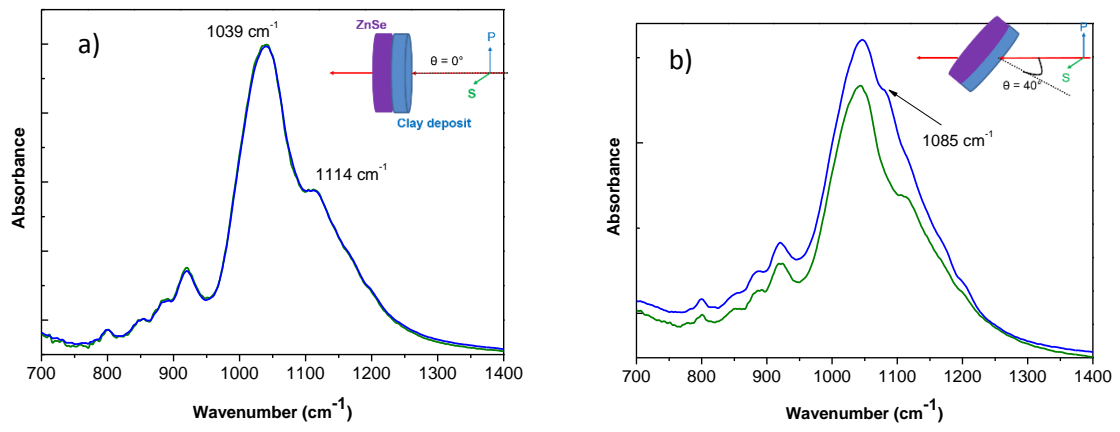
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Fig. 1

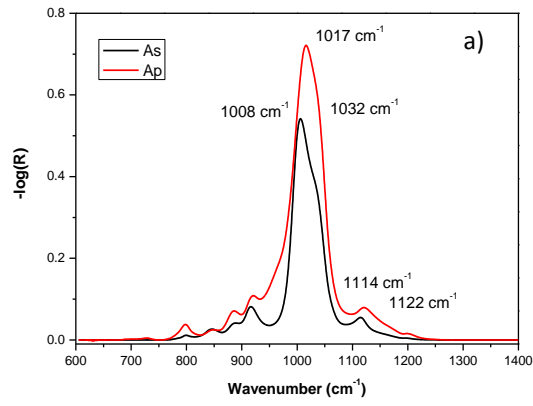


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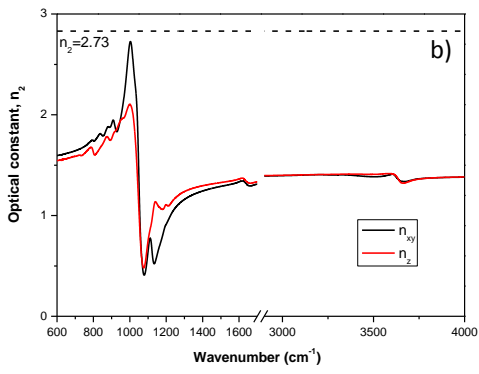
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Fig. 2



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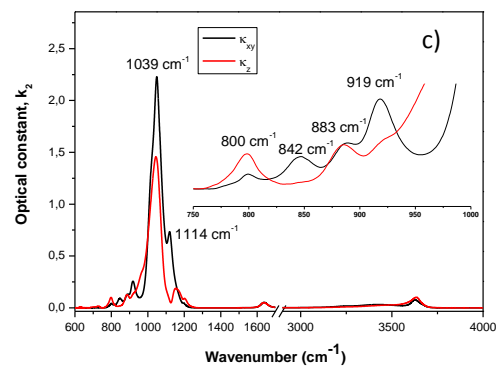
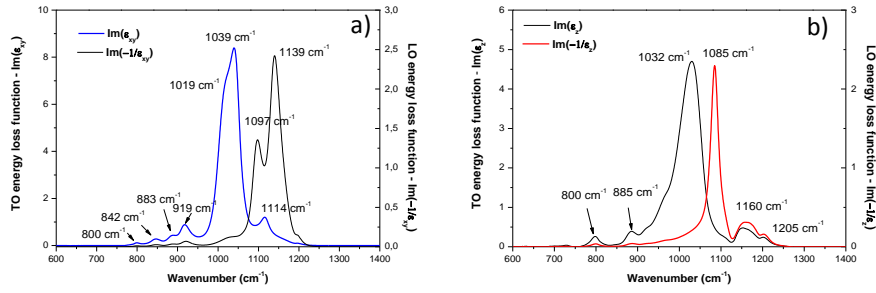
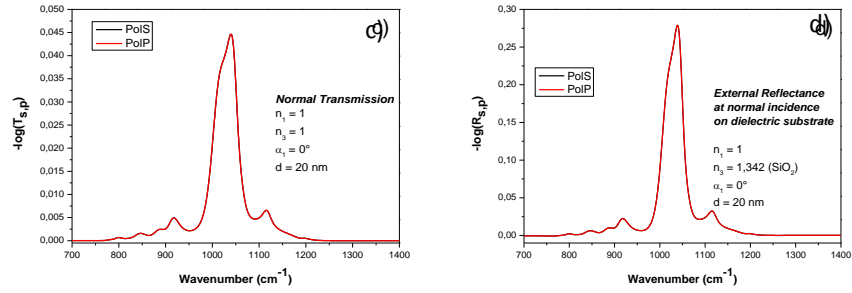


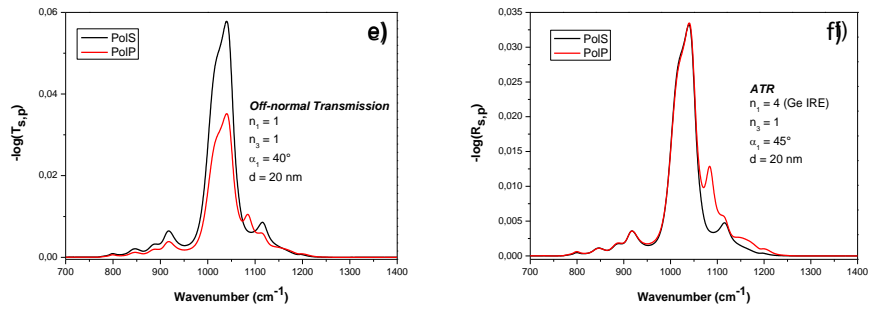
Fig. 3



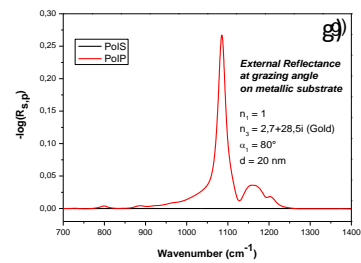
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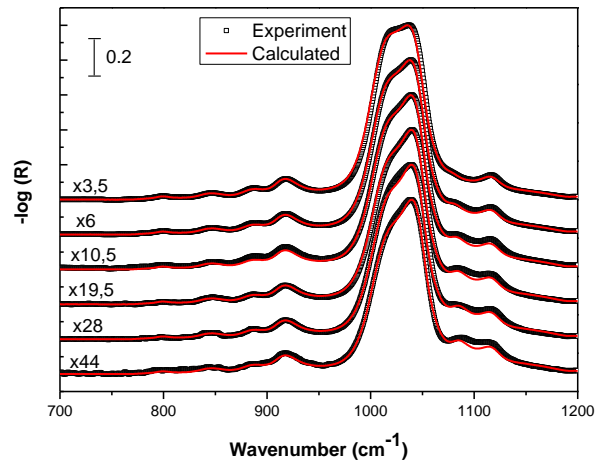
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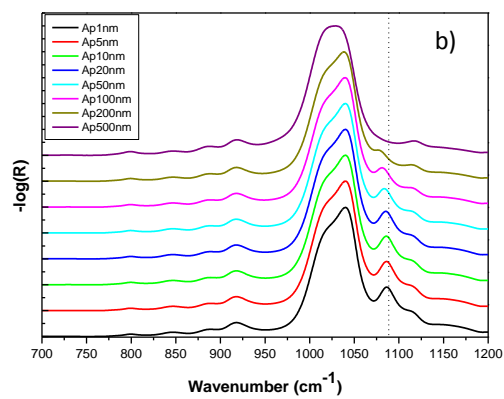
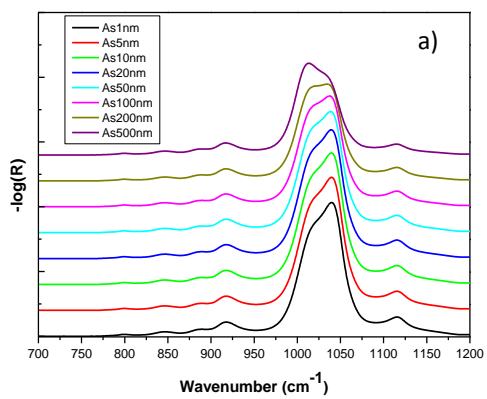
Fig. 4



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Fig. 5

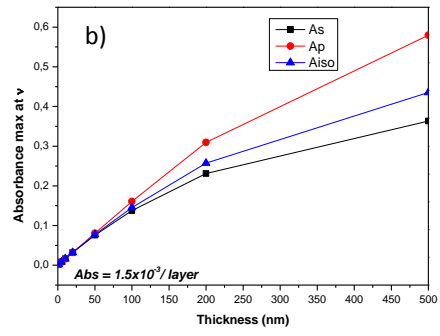
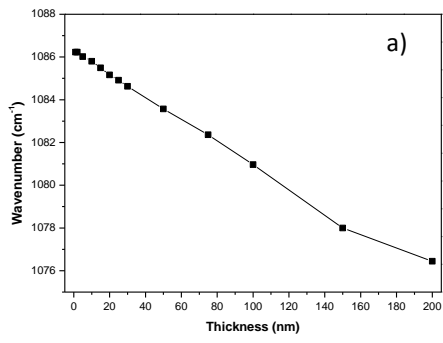


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Fig. 6

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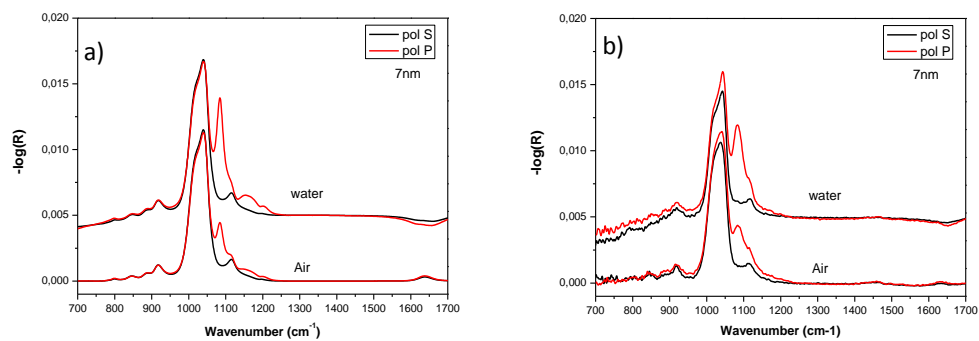
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Fig. 7

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Fig. 8

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