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Room Temperature Synthesis of Tetrasubstituted 1,3-Dithioles by Dimerizing Sulfuration of Chalcones with Elemental Sulfur

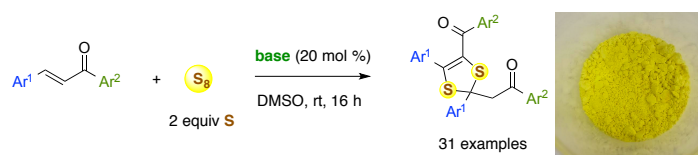
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ABSTRACT: Chalcones were found to undergo a sulfurative dimerization with elemental sulfur to tetrasubstituted 1,3-dithioles. The reaction was found to proceed at room temperature in the presence of a nitrogen base catalyst in DMSO.

Introduction

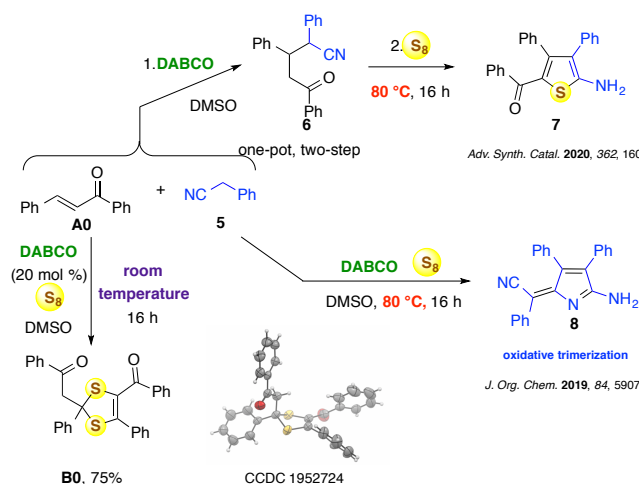
Organosulfur compounds, including sulfa heterocycles, are ubiquitous in both natural products and artificial molecules with interesting properties. Consequently, the finding of efficient and direct approaches to such molecules starting from simple starting materials and readily available in terms of both quantity and variety plays undoubtedly a central role in accelerating their research and development.

To this end, a wide range of synthetic methods has been developed to incorporate one or more sulfur atoms to the organic scaffolds using simple inorganic sulfur sources. Among these sulfur precursors, elemental sulfur is unarguably the best sulfur source. Using elemental sulfur in organic synthesis, we can benefit from its unique redox chemistry as an element, its user-friendliness along with its easy

An interesting aspect of elemental sulfur is its capability to promote oligomerization of single starting materials to provide oligomers that are much more difficult or impractical to obtain otherwise. Via this strategy, simple and readily available starting molecules could undergo dimerization,⁹ trimerization,¹⁰ tetramerization¹¹ and hexamerization¹² with various degree of sulfuration.¹³ Herein, we report a direct dimerization of chalcones with sulfur to provide original tetrasubstituted 1,3-dithiole scaffold. More strikingly, the reactions could be performed under simple conditions even at room temperature.¹⁴

Results and Discussion

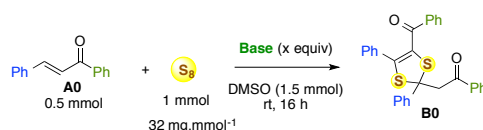
During the course of our study on new reactivities of the combination S₈/DMSO,¹⁵ which could act as both oxidizing and sulfurating agent, we developed a three-component access to 2-aminothiophene **7** from phenylacetonitrile **5**, chalcone **A0** and elemental sulfur (Scheme 1).¹⁶ Reasonable yield was obtained if the Michael adduct **6** between phenylacetonitrile **5** and chalcone **A0** were formed prior to the DABCO-catalyzed thiophenation with elemental sulfur. On the contrary, if three components of the reaction were mixed together in the presence of DABCO catalyst, 2-aminothiophenes **7** were formed in low yield along with pyrrole **8** from trimerization of **5**^{10a} and a byproduct **B0** which could be generated in another control experiment performed without **5** (Table 1, entry 1). Interestingly, high yield of **B0** could be achieved even at rt. The exact structure of **B0** was unambiguously determined to be a tetrasubstituted dithiole by single crystal X-ray crystallographic analysis.



Scheme 1. Initial observation for the formation of dichalcone disulfide B0

Other basic additives such as liquid organic bases (triethylamine, tri-*n*-propylamine, *t*-butylamine, DIPEA, *N*-methylpiperidine, 2,2,6,6-tetramethylpiperidine, *N*-methylpyrrolidine in stoichiometric amount) (Table 1, entries 2-10) or solid inorganic base (CsF, Na₂S•5H₂O, K₂CO₃ in 20 mol % amount) were also found to promote this sulfurative dimerization in good yields in almost all cases (Table 1, entries 11-13).

Table 1. Screening of the reaction conditions

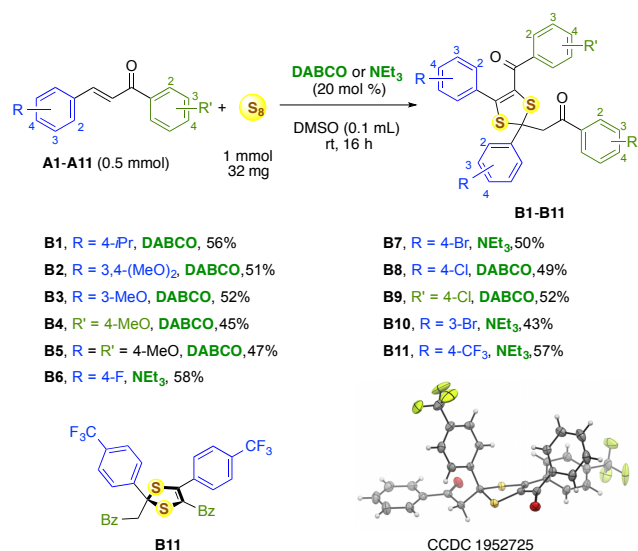


Entry ^a	Base (x equiv)	Yield (%) ^b
1	DABCO (0.2)	75
2	NEt ₃ (1)	75
3	N(<i>n</i> -Pr) ₃ (1)	70
4	<i>t</i> -BuNH ₂ (1)	60
5	DIPEA (1)	71
7	<i>N</i> -methylpiperidine (1)	67
8	2,2,6,6-tetramethylpiperidine (1)	52
9	<i>N</i> -methylpyrrolidine (1)	55
10	DBU (1)	73
11	CsF (0.2)	70
12	Na ₂ S•5H ₂ O (0.2)	62
13	K ₂ CO ₃ (0.2)	55
14	NEt₃ (0.2)	75
15	NEt ₃ (0.1)	47

16 ^c	NEt ₃ (0.2-10)	0
17 ^d	NEt ₃ (0.2)	50
18 ^e	NEt ₃ (0.2)	30

^aReaction conditions: chalcone **A0** (0.5 mmol, 104 mg), S (1 mmol 32 mg), base (x equiv), DMSO (1.5 mmol, 0.1 mL) at rt. ^bIsolated yield. ^cReaction performed in the absence of DMSO. ^dReaction performed in DMF (0.1 mL) instead of DMSO. ^eReaction performed with S (1 equiv, 0.5 mmol, 16 mg). Comparably good yield 75% was obtained when triethylamine was used in catalytic amount (20 mol %) (entry 14). Lowering NEt₃ amount to 10 mol % resulted in lower yield (54%, entry 15). When reaction was performed in the absence of DMSO, chalcone **A0** remained unchanged regardless of NEt₃ amount (entry 16). We noted that in this case, since both chalcone **A0** and sulfur are solid at rt, their mixtures with NEt₃ remained slurry states with increasing amounts of NEt₃ (0.2-10 equiv). When DMF was used instead of DMSO (entry 17), the reaction was not complete, the expected dimer **B0** was formed in low yield along with many side-products.

When only 1 equiv sulfur was used in the reaction, i.e. to provide exactly two sulfur atoms in dimer product **B0**, a sharp drop lower yield occurred along with the formation of unidentified products (entry 18). Since the expected sulfurative dimerization reaction required the removal of one hydrogen atom per each consumed chalcone, it is clear that sulfur acted also as a dehydrogenative agent.

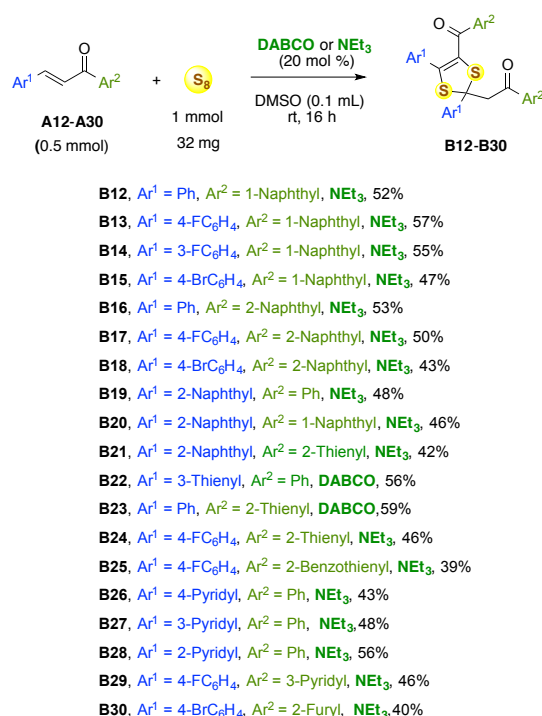


Scheme 2.

Gratifyingly, the dimer **B0** could be readily precipitated as a bright yellow solid by simply diluting the reaction mixtures with methanol and thus readily isolated by filtration. To further demonstrate the efficiency of this procedure, we explored the reaction with chalcones **A1-A11** (Scheme 2).

Two sets of optimized conditions involving either DABCO (entry 1) or NEt₃ (entry 14) in 20 mol % amount were employed. A wide range of functional substituents (alkyl **A1**, alkoxy **A2-A5**, halogen **A6-A10**, trifluoromethyl **A11**) in either or both aromatic rings were shown to be compatible with the reaction conditions. The structure of disulfide dimers **B11** was determined by X-ray diffraction, confirming again the common 1,3-dithiole scaffold of all products obtained by our reaction.

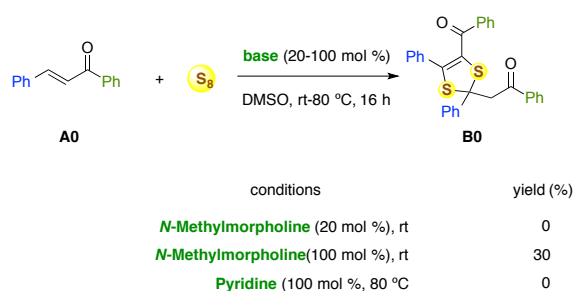
Next, a series of naphthalene and heterocyclic chalcones **A12-A30** was tested and these substrates gave moderate yields ranging from 40% to 59% (Scheme 3). Chalcones bearing naphthyl substituents **A12-A21** as well as heterocyclic ring such as thienyls (**A21-A24**), benzothienyl (**A25**), pyridyls (**A26-A29**) and furyl (**A30**) displayed good reactivity, providing the corresponding disulfurative dimeric products **B12-B30**.



Scheme 3.

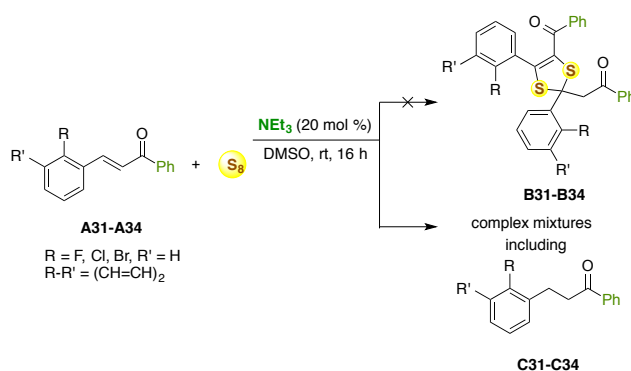
We emphasized that the best yield was obtained with the most simple chalcone **A0**, which could be readily crystalized out from its reactions mixture by simply diluting with methanol, whereas other dimers **B1-B30** were shown to be much more soluble than in this solvent and thus could not be efficiently isolated by this manner. It is possible that the higher crystallinity of **B0** allowed this dimer to be separated from the reaction mixture when it was formed and thus avoided other side reactions that could lower the global yields.

To gain further insight into the reaction mechanism, several control experiments were performed. First, NEt₃ (pK_b 3.25) was replaced by a weaker base in stoichiometric amount (1 equiv) such as *N*-methylmorpholine (pK_b 6.62) and pyridine (pK_b 8.75) (Scheme 4). Chalcone **A0** was converted only partially to dimer **B0** (30% conversion) with *N*-methylmorpholine (1 equiv) at rt and remained unchanged with the lower loading of this base (20 mol %) or with pyridine (1 equiv) even upon heating up to 80 °C, indicating that the formation of sulfur-base adduct via ring opening of S₈ by the base was a key step in initiating the reaction. This step depended on both basicity and concentration of the base.



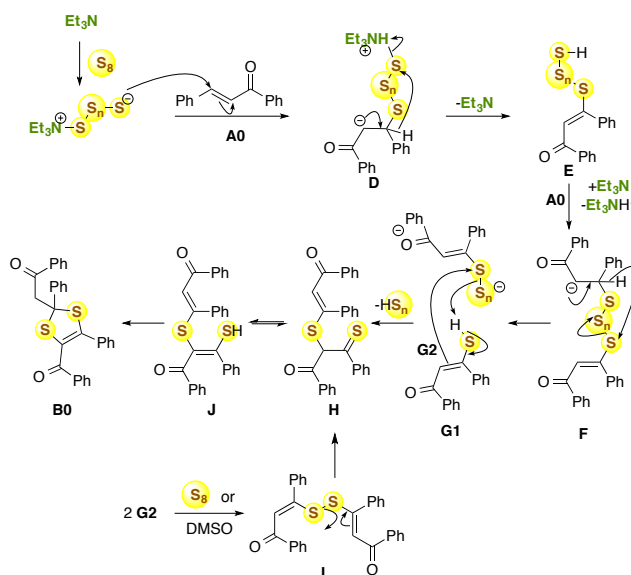
Scheme 4. Control experiment with weaker bases additives

Second, we found that chalcone substrates bearing a 2-substituent such as 2-fluoro-, 2-chloro-, 2-bromo-chalcones **A31-A33** or 1-naphthyl analog **A34** did not undergo the expected sulfurative dimerization (Scheme 5). Their reactions under the same conditions led to complex mixtures in which we could detect easily the presence of dihydrochalcone issued from reduction of the olefin based on their ¹H NMR signals of aliphatic protons. This side reaction was previously observed with *o*-halochalcones.^{17a} From our previous studies on sulfur-chalcone reactions,^{15f,17} we speculated that the reaction pathway should be initiated by a Michael nucleophilic addition reaction to the chalcone and an *ortho* substituent on the ring B of chalcone substrate would slower or even inhibit the reaction.



Scheme 5. Failed reactions with *o*-substituted chalcones

Based on the obtained results, the above-mentioned control experiments, and the precedent literature,¹⁸ a plausible reaction mechanism is outlined in Scheme 6.



Scheme 6. Plausible reaction mechanism

The reaction was initiated by a Michael addition of NEt_3 -sulfur adduct to chalcone **A0**. The resulting adduct **D** underwent a disproportionation in which the sulfurated chalcone moiety was oxidized and the terminal N-S bond was reduced, leading to polysulfane **E**. Subsequent addition of **E** to another molecule of chalcone **A1**, followed by fragmentation of the resulting adduct **F** would give two fragments **G1** and **G2**, different only in the number of sulfur atom and interconvertible. The transformation of **G2** into thioketone **H** could proceed by combining with **G1** with the simultaneous extrusion of hydropolysulfide

or by S₈- or DMSO- promoted oxidative dimerization into **I** then rearrangement.¹⁸ Cyclization of enethiol **J**, which is a tautomer of **H**, would finally provide the dithiole **B0**.

In conclusion, we explored the reactivity of elemental sulfur on chalcones and developed an efficient strategy for the construction of tetrasubstituted 1,3-dithiole compounds using NEt₃ in DMSO as a promoter. The reaction went through a sulfurative dimerization of chalcone under mild conditions with broad substrate scope. Densely substituted 1,3-dithioles were obtained in reasonable yields for the majority of tested substrates. The result presented here confirms once again the very rich organic chemistry of elemental sulfur even under very simple conditions. Since chalcones are readily available in a wide range of structures from both commercial suppliers and syntheses, our method opens a new and convenient access to a large collection of original structures for bioactive screening. Further experiments on the mechanism as well as synthetic application are currently ongoing in our laboratory.

Experimental Section

General Information

Reagents were obtained from commercial supplier and used without further purification. Analytical thin layer chromatography (TLC) was purchased from Merck KGaA (silica gel 60 F254). Visualization of the chromatogram was performed by UV light (254 nm) or KMnO₄ or vanilline stains. Flash column chromatography was carried out using kieselgel 35-70 µm particle sized silica gel (230-400 mesh). NMR Chemical shifts are reported in (δ) ppm relative to tetramethylsilane (TMS) with the residual solvent as internal reference (CDCl₃, δ 7.26 ppm for ¹H and δ 77.0 ppm for ¹³C). Data are reported as follows: chemical shift, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet), coupling constants (Hz) and integration. Mass spectra were obtained using electron spray ionization (ESI). The HRMS data were measured on MALDI-TOF instrument.

General procedure

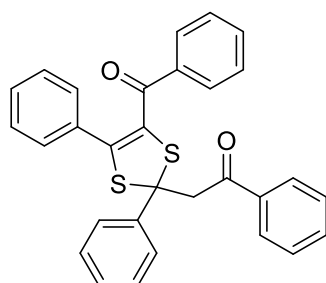
A mixture of chalcone **A** (0.5 mmol, 1 equiv), **S** (32 mg, 1 mmol, 2 equiv), triethylamine or DABCO (0.1 mmol) in 0.1 mL of DMSO in a 7-mL test-tube closed with a septum was stirred at rt for 16 h. In case of **A0**, upon dilution with MeOH (2-4 mL), dimer **B0** was precipitated as a bright yellow solid, which was filtered, washed with MeOH (2 mL x 3) and dried in vacuo to give pure **B0**. In cases of chalcones **A1-A30**, due to higher solubilities of the corresponding dimers **B1-B30**, the crude mixture

was diluted with MeOH:CH₂Cl₂ (4 mL, v:v 1:1) and filtered over Celite. After removal of volatiles in vacuo, the residue was purified by column chromatography (hexane/ethyl acetate: 19/1 to 9/1) to give the dimer product a bright yellow solid.

Large-scale synthesis of **B0**: A mixture of chalcone **A0** (5.20 g, 25 mmol), S (1.60 g, 50 mmol), and DABCO (560 mg, 5 mmol) in 5.0 mL of DMSO in a 25-mL round-bottom flask closed with a septum was stirred at rt for 16 h. Dilution of the reaction mixture with MeOH (20 mL) precipitated the yellow product, which was filtered and washed with MeOH (2 mL x 5). The yellow powder was dried in vacuo (4.18 g, 70%).

Characterization of the products

2-(4-Benzoyl-2,5-diphenyl-1,3-dithiol-2-yl)-1-phenylethan-1-one (**B0**)



Upon dilution with MeOH (2-4 mL), the product was precipitated as a bright yellow solid, which was filtered, washed with MeOH (2 mL x 3) and dried in vacuo to give pure **B0**.

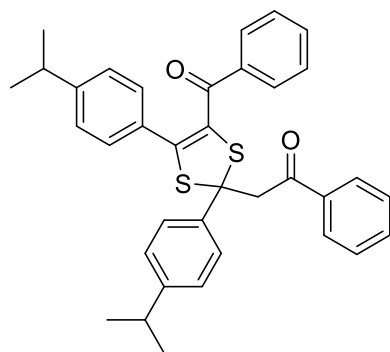
Bright yellow solid (90 mg, 75%).

¹H NMR (300 MHz, CDCl₃) δ 8.0 (d, *J* = 7.7 Hz, 2H), 7.9 (d, *J* = 7.9 Hz, 2H), 7.6 (t, *J* = 7.4 Hz, 1H), 7.6 – 7.4 (m, 3H), 7.4 (t, *J* = 7.6 Hz, 2H), 7.4 – 7.2 (m, 4H), 7.2 – 7.0 (m, 5H), 4.5 (s, 2H).

¹³C{¹H} NMR (75 MHz, CDCl₃) δ 194.7, 190.4, 144.5, 142.2, 136.7, 136.2, 133.7, 132.5, 132.1, 129.9, 129.5, 129.3, 128.8, 128.4, 128.2, 128.1, 128.0, 127.9, 126.5, 66.4 (CS₂), 51.2.

HRMS *m/z* calculated for [M+Na]⁺ C₃₀H₂₂NaO₂S₂ 501.0959. Found 601.0963.

2-(4-Benzoyl-2,5-bis(4-isopropylphenyl)-1,3-dithiol-2-yl)-1-phenylethan-1-one (**B1**)



Eluent: hexane/ethyl acetate: 19/1 to 9/1.

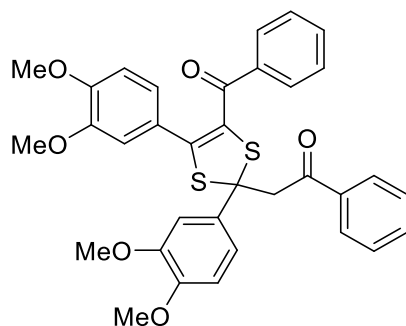
Bright yellow solid (79 mg, 56%).

^1H NMR (300 MHz, CDCl_3) δ 8.0 – 7.9 (m, 2H), 7.8 – 7.7 (m, 2H), 7.6 – 7.5 (m, 1H), 7.5 – 7.4 (m, 4H), 7.2 – 7.2 (m, 2H), 7.1 (td, $J = 6.2, 1.7$ Hz, 2H), 7.1 – 7.0 (m, 1H), 6.9 (d, $J = 8.2$ Hz, 2H), 4.5 (s, 2H), 2.9 (p, $J = 6.9$ Hz, 1H), 2.7 (p, $J = 6.9$ Hz, 1H), 1.2 (d, $J = 6.9$ Hz, 6H), 1.1 (d, $J = 6.9$ Hz, 6H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (75 MHz, CDCl_3) δ 194.9, 190.5, 150.5, 148.4, 145.6, 139.3, 136.9, 133.6, 132.1, 130.0, 129.2, 128.7, 128.1, 127.8, 127.7, 127.7, 126.4, 66.0, 51.2, 33.8, 33.6, 23.8, 23.6.

HRMS m/z calculated for $[\text{M}+\text{Na}]^+$ $\text{C}_{36}\text{H}_{34}\text{NaO}_2\text{S}_2$ 585.1898. Found 585.1996.

2-(4-Benzoyl-2,5-bis(3,4-dimethoxyphenyl)-1,3-dithiol-2-yl)-1-phenylethan-1-one (B2)



Eluent: hexane/ethyl acetate: 19/1 to 9/1.

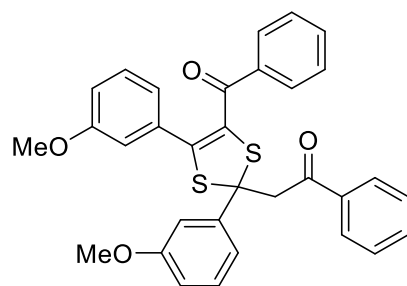
Bright yellow solid (76 mg, 51%).

^1H NMR (300 MHz, CDCl_3) δ 8.0 (d, $J = 7.3$ Hz, 3H), 7.6 – 7.4 (m, 9H), 7.3 – 7.2 (m, 3H), 7.2 – 7.1 (m, 3H), 6.9 (dd, $J = 8.3, 2.1$ Hz, 1H), 6.8 (d, $J = 8.5$ Hz, 1H), 6.6 (t, $J = 2.5$ Hz, 1H), 6.6 (d, $J = 8.4$ Hz, 1H), 4.5 (s, 2H), 3.9 (s, 3H), 3.9 (s, 3H), 3.8 (s, 3H), 3.7 (s, 3H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (75 MHz, CDCl_3) δ 195.0, 190.6, 150.1, 148.6, 148.5, 148.2, 145.0, 136.8, 136.3, 134.4, 133.7, 132.6, 129.2, 128.8, 128.6, 128.1, 128.0, 125.2, 124.7, 123.2, 118.6, 113.0, 110.7, 110.6, 110.5, 65.6, 56.0, 55.9, 55.7, 50.9.

HRMS m/z calculated for $[\text{M}+\text{Na}]^+$ $\text{C}_{34}\text{H}_{30}\text{NaO}_6\text{S}_2$ 621.1381. Found 621.1387.

2-(4-Benzoyl-2,5-bis(3-methoxyphenyl)-1,3-dithiol-2-yl)-1-phenylethan-1-one (B3)



Eluent: hexane/ethyl acetate: 19/1 to 9/1.

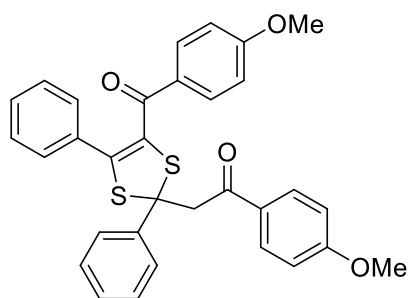
Bright yellow solid (70 mg, 52%).

^1H NMR (500 MHz, CDCl_3) δ 7.9 (dd, $J = 8.4, 1.3$ Hz, 2H), 7.6 – 7.5 (m, 3H), 7.5 – 7.4 (m, 3H), 7.4 (ddd, $J = 7.9, 2.0, 0.9$ Hz, 1H), 7.3 – 7.3 (m, 2H), 7.2 – 7.1 (m, 2H), 7.0 (t, $J = 7.9$ Hz, 1H), 6.9 – 6.8 (m, 1H), 6.8 (ddd, $J = 8.2, 2.5, 0.9$ Hz, 1H), 6.7 (dd, $J = 2.6, 1.6$ Hz, 1H), 6.7 (ddd, $J = 8.3, 2.6, 1.0$ Hz, 1H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 194.8, 190.4, 159.6, 159.1, 144.1, 143.9, 136.8, 136.3, 133.8, 133.3, 132.6, 129.5, 129.3, 129.2, 128.8, 128.2, 128.0, 126.6, 122.5, 118.7, 115.8, 115.0, 113.2, 113.0, 66.2, 55.4, 55.3, 51.3.

HRMS m/z calculated for $[\text{M}+\text{Na}]^+$ $\text{C}_{32}\text{H}_{26}\text{NaO}_4\text{S}_2$ 561.1170. Found 561.1175.

2-(4-(4-Methoxybenzoyl)-2,5-diphenyl-1,3-dithiol-2-yl)-1-(4-methoxyphenyl)ethan-1-one (B4)



Eluent: hexane/ethyl acetate: 19/1 to 9/1.

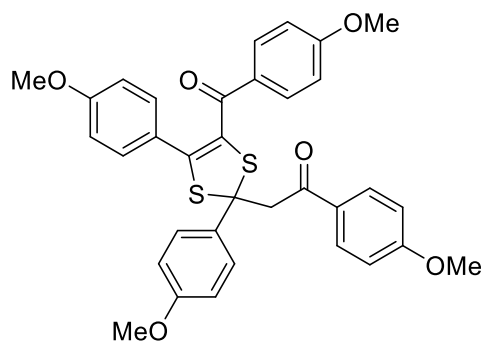
Bright yellow solid (61 mg, 45%).

^1H NMR (300 MHz, CDCl_3) δ 8.0 – 7.9 (m, 4H), 7.6 (d, J = 9.1 Hz, 2H), 7.4 (dd, J = 8.4, 6.8 Hz, 2H), 7.3 (ddd, J = 7.5, 4.8, 1.7 Hz, 2H), 7.1 (td, J = 5.9, 2.8 Hz, 2H), 6.9 (d, J = 8.4 Hz, 2H), 6.7 (d, J = 10.2 Hz, 2H), 4.5 (s, 2H), 3.9 (s, 3H), 3.7 (s, 3H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (75 MHz, CDCl_3) δ 192.2, 188.2, 162.9, 162.2, 141.5, 140.7, 131.3, 130.7, 129.4, 128.7, 128.2, 128.2, 127.3, 127.1, 126.7, 126.6, 125.4, 125.0, 112.8, 112.3, 112.2, 65.9, 54.5, 54.3, 49.6.

HRMS m/z calculated for $[\text{M}+\text{Na}]^+$ $\text{C}_{32}\text{H}_{26}\text{NaO}_4\text{S}_2$ 561.1170. Found 561.1175.

2-(4-(4-Methoxybenzoyl)-2,5-bis(4-methoxyphenyl)-1,3-dithiol-2-yl)-1-(4-methoxyphenyl)ethan-1-one (B5)



Eluent: hexane/ethyl acetate: 19/1 to 9/1.

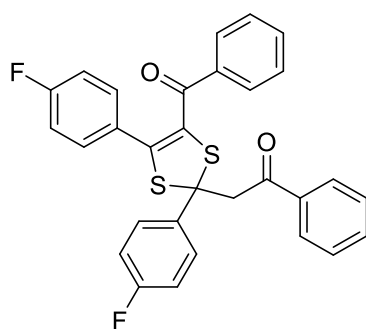
Bright yellow solid (70 mg, 47%).

^1H NMR (300 MHz, CDCl_3) δ 7.9 (d, J = 8.9 Hz, 2H), 7.8 (d, J = 8.9 Hz, 2H), 7.6 (d, J = 8.8 Hz, 2H), 7.2 (d, J = 8.8 Hz, 2H), 6.9 (dd, J = 11.9, 8.9 Hz, 4H), 6.6 (t, J = 8.9 Hz, 3H), 4.4 (s, 2H), 3.9 (s, 3H), 3.8 (s, 3H), 3.8 (s, 3H), 3.7 (s, 3H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (75 MHz, CDCl_3) δ 193.4, 189.5, 163.9, 163.2, 160.3, 158.9, 142.4, 134.4, 131.7, 131.6, 131.3, 131.1, 131.0, 130.5, 130.4, 130.3, 130.1, 129.4, 129.3, 129.2, 129.1, 129.0, 127.9, 124.9, 124.6, 113.9, 113.6, 113.4, 113.3, 66.2, 55.5, 55.4, 55.3, 55.2, 50.4.

HRMS m/z calculated for $[\text{M}+\text{Na}]^+$ $\text{C}_{34}\text{H}_{30}\text{NaO}_6\text{S}_2$ 621.1381. Found 621.1392.

2-(4-Benzoyl-2,5-bis(4-fluorophenyl)-1,3-dithiol-2-yl)-1-phenylethan-1-one (B6)



Eluent: hexane/ethyl acetate: 19/1 to 9/1.

Bright yellow solid (75 mg, 58%).

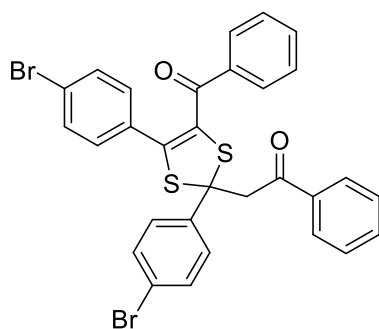
^1H NMR (500 MHz, CDCl_3) δ 7.9 (d, J = 8.2 Hz, 2H), 7.9 – 7.8 (m, 2H), 7.6 (t, J = 7.4 Hz, 1H), 7.5 (t, J = 7.9 Hz, 3H), 7.3 (t, J = 7.4 Hz, 1H), 7.2 – 7.2 (m, 1H), 7.1 (t, J = 7.9 Hz, 2H), 7.1 (t, J = 8.5 Hz, 2H), 6.8 (t, J = 8.6 Hz, 2H), 4.5 (s, 2H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 194.7, 190.2, 163.6 (d, J = 132.1 Hz), 161.6 (d, J = 129.2 Hz), 143.1, 138.2, 136.5, 136.0, 133.9, 132.8, 131.8, 131.7, 129.2, 128.8, 128.5 (d, J = 8.1 Hz), 128.1 (d, J = 4.3 Hz), 115.3 (d, J = 4.3 Hz), 115.2 (d, J = 4.3 Hz), 65.9, 51.1.

^{19}F NMR (471 MHz, CDCl_3) δ -110.2, -114.2.

HRMS m/z calculated for $[\text{M}+\text{Na}]^+$ $\text{C}_{30}\text{H}_{20}\text{F}_2\text{NaO}_2\text{S}_2$ 537.0770. Found 537.0775.

2-(4-Benzoyl-2,5-bis(4-bromophenyl)-1,3-dithiol-2-yl)-1-phenylethan-1-one (B7)



Eluent: hexane/ethyl acetate: 19/1 to 9/1.

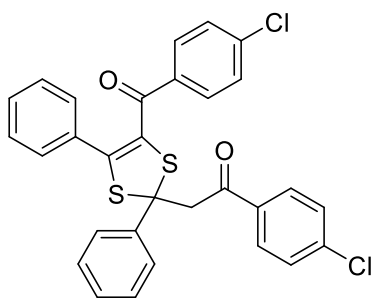
Bright yellow solid (80 mg, 50%).

^1H NMR (500 MHz, CDCl_3) δ 7.92 (dd, $J = 8.1, 1.4$ Hz, 2H), 7.74 (d, $J = 8.6$ Hz, 1H), 7.60 (t, $J = 7.4$ Hz, 1H), 7.54 – 7.43 (m, 5H), 7.38 – 7.31 (m, 1H), 7.25 – 7.10 (m, 4H), 7.08 (d, $J = 8.5$ Hz, 1H), 4.45 (s, 2H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 194.5, 190.0, 142.3, 141.6, 136.4, 135.9, 134.0, 132.9, 131.5, 131.4, 131.2, 130.8, 129.2, 128.8, 128.3, 128.2, 128.1, 127.0, 123.9, 122.1, 66.2, 51.0.

HRMS m/z calculated for $[\text{M}+\text{Na}]^+$ $\text{C}_{30}\text{H}_{20}\text{Br}_2\text{NaO}_2\text{S}_2$ 656.9169. Found 656.9176.

2-(4-(4-Chlorobenzoyl)-2,5-diphenyl-1,3-dithiol-2-yl)-1-(4-chlorophenyl)ethan-1-one (B8)



Eluent: hexane/ethyl acetate: 19/1 to 9/1.

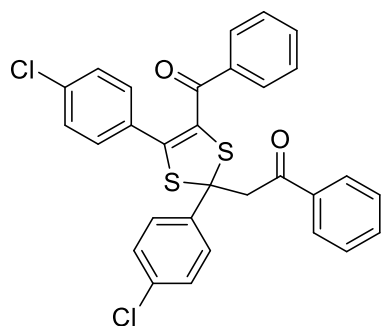
Bright yellow solid (67 mg, 49%).

^1H NMR (500 MHz, CDCl_3) δ 7.9 – 7.8 (m, 5H), 7.4 – 7.4 (m, 5H), 7.4 (d, $J = 8.0$ Hz, 2H), 7.3 – 7.3 (m, 1H), 7.2 – 7.2 (m, 2H), 7.2 – 7.1 (m, 1H), 7.1 (d, $J = 8.4$ Hz, 2H), 7.1 (d, $J = 9.4$ Hz, 2H), 4.4 (s, 2H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 193.6, 189.1, 145.0, 142.1, 140.4, 138.9, 135.1, 134.5, 131.9, 130.6, 129.9, 129.8, 129.6, 129.1, 128.5, 128.3, 128.1, 126.5, 126.2, 66.5, 51.2

HRMS m/z calculated for $[\text{M}+\text{Na}]^+ \text{C}_{30}\text{H}_{20}\text{Cl}_2\text{NaO}_2\text{S}_2$ 569.0179. Found 569.0184.

2-(4-Benzoyl-2,5-bis(4-chlorophenyl)-1,3-dithiol-2-yl)-1-phenylethan-1-one (B9)



Eluent: hexane/ethyl acetate: 19/1 to 9/1.

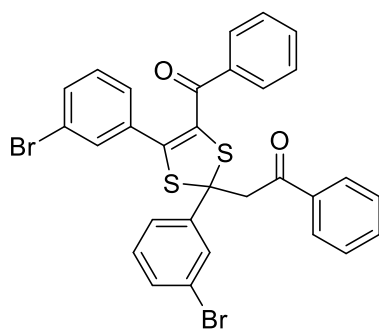
Bright yellow solid (71 mg, 52%).

^1H NMR (500 MHz, CDCl_3) δ 7.9 (d, $J = 6.3$ Hz, 2H), 7.8 (d, $J = 8.7$ Hz, 2H), 7.6 (t, $J = 7.7$ Hz, 1H), 7.5 (d, $J = 7.1$ Hz, 2H), 7.5 (t, $J = 7.8$ Hz, 2H), 7.4 – 7.3 (m, 3H), 7.2 – 7.1 (m, 4H), 7.1 (d, $J = 8.5$ Hz, 2H), 4.5 (s, 2H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 194.6, 190.2, 142.5, 141.1, 136.5, 135.9, 135.6, 134.0, 133.9, 133.0, 131.1, 130.4, 129.8, 129.6, 129.3, 129.1, 128.9, 128.6, 128.6, 128.5, 128.4, 128.3, 128.2, 128.1, 128.0, 127.0, 66.1, 51.1.

HRMS m/z calculated for $[\text{M}+\text{Na}]^+ \text{C}_{30}\text{H}_{20}\text{Cl}_2\text{NaO}_2\text{S}_2$ 569.0179. Found 569.0188.

2-(4-Benzoyl-2,5-bis(3-bromophenyl)-1,3-dithiol-2-yl)-1-phenylethan-1-one (B10)



Eluent: hexane/ethyl acetate: 19/1 to 9/1.

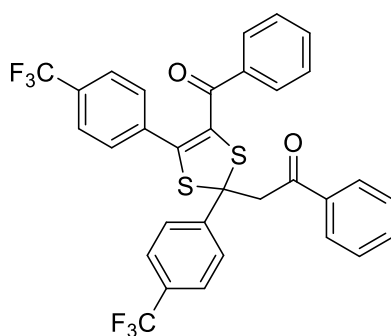
Bright yellow solid (68 mg, 43%).

^1H NMR (500 MHz, CDCl_3) δ 7.97 (t, $J = 2.0$ Hz, 1H), 7.89 – 7.83 (m, 2H), 7.69 (dd, $J = 8.0, 2.0$ Hz, 1H), 7.52 (t, $J = 7.5$ Hz, 1H), 7.41 (dd, $J = 8.2, 6.7$ Hz, 5H), 7.35 (dd, $J = 8.0, 1.8$ Hz, 1H), 7.27 (d, $J = 1.9$ Hz, 1H), 7.26 – 7.21 (m, 1H), 7.20 – 7.15 (m, 3H), 7.12 – 7.05 (m, 3H), 6.86 (t, $J = 7.9$ Hz, 1H), 4.40 (d, $J = 1.5$ Hz, 2H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 194.4, 190.0, 144.8, 141.9, 136.5, 135.8, 134.0, 133.6, 132.9, 132.7, 132.5, 131.1, 130.0, 129.8, 129.6, 129.1, 128.9, 128.3, 128.2, 127.7, 125.0, 122.7, 122.1, 65.8, 51.1.

HRMS m/z calculated for $[\text{M}+\text{Na}]^+$ $\text{C}_{30}\text{H}_{20}\text{Br}_2\text{NaO}_2\text{S}_2$ 656.9169. Found 656.9174.

2-(4-Benzoyl-2,5-bis(4-(trifluoromethyl)phenyl)-1,3-dithiol-2-yl)-1-phenylethan-1-one (B11)



Eluent: hexane/ethyl acetate: 19/1 to 9/1.

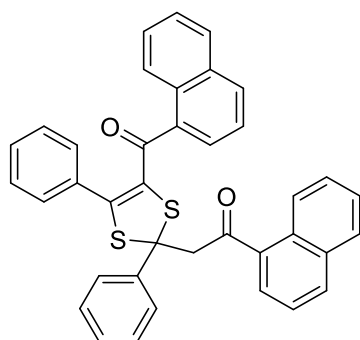
Bright yellow solid (87 mg, 57%).

^1H NMR (300 MHz, CDCl_3) δ 8.0 (d, $J = 8.2$ Hz, 2H), 8.0 (d, $J = 7.6$ Hz, 2H), 7.7 (d, $J = 8.3$ Hz, 2H), 7.6 (d, $J = 7.9$ Hz, 1H), 7.6 – 7.5 (m, 4H), 7.4 – 7.3 (m, 6H), 7.2 (t, $J = 7.0$ Hz, 2H), 4.6 (s, 2H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (75 MHz, CDCl_3) δ 194.4, 189.7, 146.5, 141.3, 136.2, 135.6, 135.3, 134.1, 133.2, 130.0, 129.1, 128.9, 128.3, 128.1, 127.0, 125.5 (d, $J = 3.8$ Hz), 125.1 (d, $J = 4.1$ Hz), 66.5, 51.2.

HRMS m/z calculated for $[\text{M}+\text{Na}]^+$ $\text{C}_{32}\text{H}_{20}\text{F}_6\text{NaO}_2\text{S}_2$ 637.0707. Found 637.0712.

2-(4-(1-Naphthoyl)-2,5-diphenyl-1,3-dithiol-2-yl)-1-(naphthalen-1-yl)ethan-1-one (B12)



Eluent: hexane/ethyl acetate: 19/1 to 9/1.

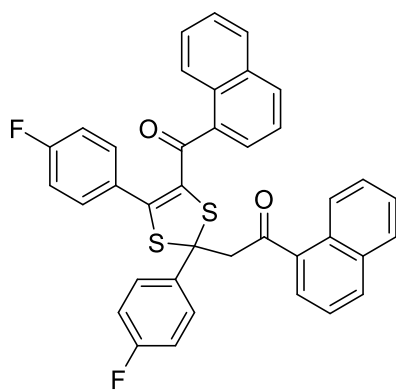
Bright yellow solid (75 mg, 52%).

^1H NMR (500 MHz, CDCl_3) δ 8.38 – 8.32 (m, 1H), 8.20 – 8.10 (m, 1H), 8.00 (d, $J = 8.2$ Hz, 1H), 7.96 (d, $J = 7.5$ Hz, 2H), 7.88 – 7.81 (m, 2H), 7.64 (dd, $J = 6.2, 3.3$ Hz, 1H), 7.58 (d, $J = 8.2$ Hz, 1H), 7.54 – 7.47 (m, 3H), 7.45 – 7.38 (m, 4H), 7.33 (dd, $J = 8.8, 7.1$ Hz, 2H), 7.03 (t, $J = 7.7$ Hz, 1H), 6.99 (d, $J = 7.1$ Hz, 2H), 6.86 (t, $J = 7.5$ Hz, 1H), 4.61 (d, $J = 17.4$ Hz, 1H), 4.56 (d, $J = 17.3$ Hz, 1H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 198.8, 190.7, 148.8, 142.8, 135.4, 134.9, 133.9, 133.4, 133.2, 131.6, 131.6, 130.6, 130.0, 129.7, 129.3, 129.1, 128.5, 128.4, 128.3, 128.2, 128.0, 128.0, 127.8, 127.3, 127.2, 126.7, 126.0, 125.6, 125.3, 124.3, 124.0, 77.3, 77.0, 76.8, 65.5, 54.3.

HRMS m/z calculated for $[\text{M}+\text{Na}]^+$ $\text{C}_{38}\text{H}_{26}\text{NaO}_2\text{S}_2$ 601.1272. Found 601.1278.

2-(4-(1-Naphthoyl)-2,5-bis(4-fluorophenyl)-1,3-dithiol-2-yl)-1-(naphthalen-1-yl)ethan-1-one (B13)



Eluent: hexane/ethyl acetate: 19/1 to 9/1.

Bright yellow solid (87 mg, 57%).

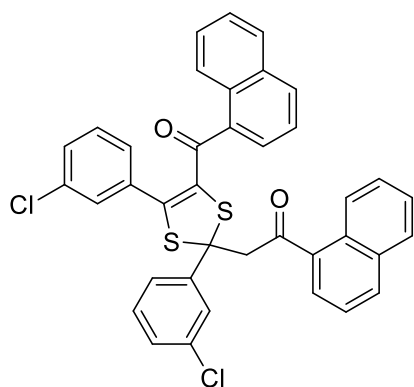
^1H NMR (500 MHz, CDCl_3) δ 8.45 – 8.36 (m, 1H), 8.07 (d, J = 8.1 Hz, 1H), 8.01 (d, J = 8.2 Hz, 1H), 7.97 (dd, J = 8.8, 5.2 Hz, 2H), 7.85 (dd, J = 8.5, 6.0 Hz, 2H), 7.68 (d, J = 6.8 Hz, 1H), 7.64 (d, J = 8.2 Hz, 1H), 7.53 (td, J = 5.5, 4.8, 3.2 Hz, 2H), 7.51 – 7.46 (m, 1H), 7.45 – 7.35 (m, 2H), 7.31 (dd, J = 7.1, 1.1 Hz, 1H), 7.10 (dt, J = 15.1, 8.1 Hz, 3H), 6.93 (dd, J = 8.6, 5.4 Hz, 2H), 6.38 (t, J = 8.6 Hz, 2H), 4.56 (d, J = 17.4 Hz, 1H), 4.52 (d, J = 17.4 Hz, 1H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 198.6, 190.5, 163.5 (d, J = 76.9 Hz), 161.5 (d, J = 74.5 Hz), 147.3, 138.9 (d, J = 3.2 Hz), 135.3, 134.5, 133.7, 133.2, 131.7, 131.1 (d, J = 8.5 Hz), 130.5, 130.1, 130.0, 128.7 (d, J = 8.2 Hz), 128.5, 128.3, 128.2 (d, J = 5.1 Hz), 128.0, 127.5, 127.5, 126.8, 126.2, 125.5, 125.0, 124.3, 124.1, 115.3 (d, J = 21.9 Hz), 114.4 (d, J = 22.0 Hz), 65.0, 54.2.

^{19}F NMR (471 MHz, CDCl_3) δ -110.52, -114.05.

HRMS m/z calculated for $[\text{M}+\text{Na}]^+$ $\text{C}_{38}\text{H}_{24}\text{F}_2\text{NaO}_2\text{S}_2$ 637.1083. Found 637.1085.

2-(4-(1-Naphthoyl)-2,5-bis(3-chlorophenyl)-1,3-dithiol-2-yl)-1-(naphthalen-1-yl)ethan-1-one
(B14)



Eluent: hexane/ethyl acetate: 19/1 to 9/1.

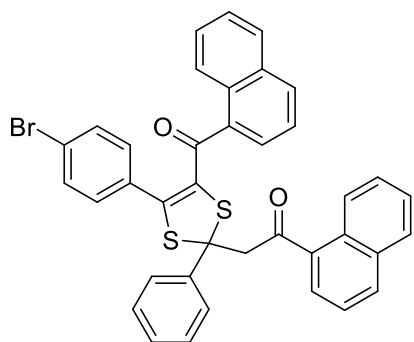
Bright yellow solid (84 mg, 55%).

^1H NMR (500 MHz, CDCl_3) δ 8.4 (d, $J = 8.4$ Hz, 1H), 8.1 (d, $J = 9.5$ Hz, 1H), 8.0 – 8.0 (m, 2H), 7.9 – 7.8 (m, 2H), 7.8 (d, $J = 7.8$ Hz, 1H), 7.7 – 7.7 (m, 1H), 7.6 (d, $J = 8.2$ Hz, 1H), 7.6 – 7.5 (m, 4H), 7.4 (qd, $J = 7.0, 3.6$ Hz, 2H), 7.4 – 7.3 (m, 4H), 7.1 (t, $J = 7.6$ Hz, 1H), 6.9 (s, 1H), 6.8 (d, $J = 8.1$ Hz, 2H), 6.6 (t, $J = 7.9$ Hz, 1H), 4.6 (d, $J = 17.6$ Hz, 1H), 4.5 (d, $J = 17.6$ Hz, 1H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 198.3, 190.3, 146.3, 145.3, 135.2, 134.6, 134.2, 134.0, 133.8, 133.5, 133.2, 132.8, 131.9, 130.9, 130.4, 130.0, 129.8, 129.2, 129.2, 128.5, 128.4, 128.3, 128.2, 128.1, 128.0, 127.5, 127.3, 127.2, 126.8, 126.3, 125.5, 125.0, 124.7, 124.3, 124.1, 64.9, 54.0.

HRMS m/z calculated for $[\text{M}+\text{Na}]^+$ $\text{C}_{38}\text{H}_{24}\text{Cl}_2\text{NaO}_2\text{S}_2$ 669.0492. Found 669.0798.

2-(4-(1-Naphthoyl)-5-(4-bromophenyl)-2-phenyl-1,3-dithiol-2-yl)-1-(naphthalen-1-yl)ethan-1-one (B15)



Eluent: hexane/ethyl acetate: 19/1 to 9/1.

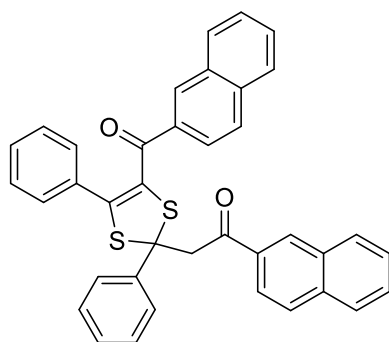
Bright yellow solid (86 mg, 47%).

^1H NMR (500 MHz, CDCl_3) δ 8.4 (d, $J = 9.5$ Hz, 1H), 8.0 (d, $J = 8.3$ Hz, 3H), 7.9 (dd, $J = 7.5, 4.1$ Hz, 6H), 7.7 – 7.6 (m, 3H), 7.6 (d, $J = 8.7$ Hz, 3H), 7.5 – 7.5 (m, 4H), 7.5 – 7.4 (m, 2H), 7.4 (ddd, $J = 8.1, 6.7, 1.4$ Hz, 2H), 7.3 – 7.3 (m, 1H), 7.1 (t, $J = 7.7$ Hz, 2H), 6.8 (d, $J = 8.5$ Hz, 3H), 6.8 (d, $J = 8.5$ Hz, 3H), 4.6 (d, $J = 17.7$ Hz, 1H), 4.5 (d, $J = 17.4$ Hz, 1H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 194.5, 190.1, 141.7, 141.4, 135.9, 135.5, 133.6, 133.2, 132.4, 132.2, 132.0, 131.7, 131.5, 131.5, 131.0, 130.1, 129.9, 129.7, 129.4, 129.0, 128.8, 128.7, 128.5, 128.4, 127.9, 127.7, 127.1, 126.8, 126.7, 124.3, 123.8, 123.5, 122.2, 67.2, 50.9.

HRMS m/z calculated for $[\text{M}+\text{Na}]^+ \text{C}_{38}\text{H}_{25}\text{BrNaO}_2\text{S}_2$ 679.0377. Found 679.0385.

2-(4-(2-Naphthoyl)-2,5-diphenyl-1,3-dithiol-2-yl)-1-(naphthalen-2-yl)ethan-1-one (B16)



Eluent: hexane/ethyl acetate: 19/1 to 9/1.

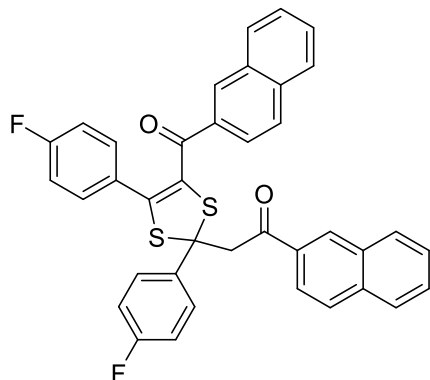
Bright yellow solid (77 mg, 53%).

^1H NMR (500 MHz, CDCl_3) δ 8.49 (d, $J = 1.6$ Hz, 1H), 8.02 (d, $J = 1.5$ Hz, 1H), 7.99 – 7.92 (m, 4H), 7.88 (dd, $J = 8.5, 4.6$ Hz, 2H), 7.71 (d, $J = 8.2$ Hz, 1H), 7.68 – 7.54 (m, 6H), 7.53 – 7.47 (m, 1H), 7.44 – 7.38 (m, 4H), 7.34 – 7.29 (m, 3H), 6.99 (dd, $J = 5.2, 1.9$ Hz, 3H), 4.67 (s, 2H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 194.7, 190.5, 143.9, 142.3, 135.8, 135.2, 133.9, 133.6, 132.4, 132.2, 132.2, 132.0, 131.6, 130.1, 129.7, 129.3, 129.3, 128.9, 128.7, 128.4, 128.3, 128.1, 128.0, 127.8, 127.6, 127.0, 126.6, 126.4, 124.6, 123.6, 67.3, 51.1.

HRMS m/z calculated for $[M+Na]^+$ $C_{38}H_{26}NaO_2S_2$ 601.1272. Found 601.1278.

2-(4-(2-Naphthoyl)-2,5-bis(4-fluorophenyl)-1,3-dithiol-2-yl)-1-(naphthalen-2-yl)ethan-1-one
(B17)



Eluent: hexane/ethyl acetate: 19/1 to 9/1.

Bright yellow solid (77 mg, 50%).

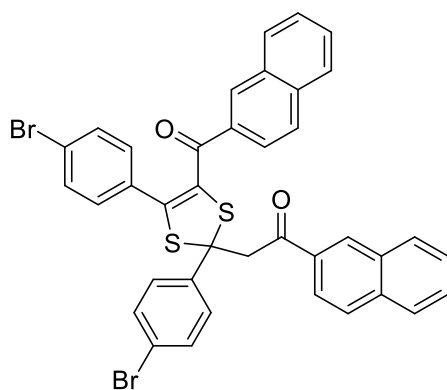
1H NMR (500 MHz, $CDCl_3$) δ 8.48 (d, $J = 1.7$ Hz, 1H), 8.01 (s, 1H), 7.96 (ddd, $J = 13.8, 8.3, 3.9$ Hz, 4H), 7.88 (dd, $J = 8.3, 6.1$ Hz, 2H), 7.74 (d, $J = 8.4$ Hz, 1H), 7.69 – 7.59 (m, 4H), 7.57 (s, 0H), 7.28 (dd, $J = 8.6, 5.4$ Hz, 2H), 7.09 (t, $J = 8.6$ Hz, 2H), 6.70 (t, $J = 8.6$ Hz, 2H), 4.63 (d, $J = 2.7$ Hz, 2H).

$^{13}C\{^1H\}$ NMR (126 MHz, $CDCl_3$) δ 194.6, 190.3, 163.6 (d, $J = 107.6$ Hz), 161.6 (d, $J = 104.4$ Hz), 142.4, 138.2, 135.9, 135.3, 133.7, 133.4, 132.4, 132.0, 131.6, 131.5, 131.5, 130.1, 129.7, 129.6, 129.0, 128.8, 128.7, 128.6, 128.6, 128.2, 127.9, 127.7, 127.1, 126.7, 124.5, 123.5, 115.4 (d, $J = 8.5$ Hz), 115.2 (d, $J = 8.2$ Hz), 66.8, 51.0.

^{19}F NMR (471 MHz, $CDCl_3$) δ -110.35, -114.11.

HRMS m/z calculated for $[M+Na]^+$ $C_{38}H_{24}F_2NaO_2S_2$ 637.1083. Found 637.1095.

2-(4-(2-Naphthoyl)-2,5-bis(4-bromophenyl)-1,3-dithiol-2-yl)-1-(naphthalen-2-yl)ethan-1-one
(B18)



Eluent: hexane/ethyl acetate: 19/1 to 9/1.

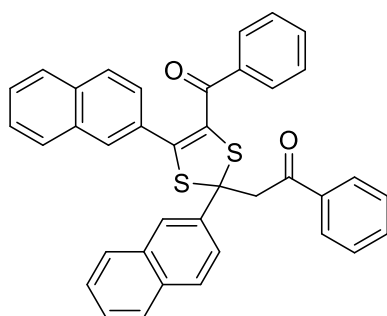
Bright yellow solid (79 mg, 43%).

^1H NMR (500 MHz, CDCl_3) δ 8.47 (d, $J = 1.6$ Hz, 1H), 8.00 – 7.94 (m, 4H), 7.91 – 7.86 (m, 2H), 7.84 (d, $J = 8.6$ Hz, 2H), 7.77 (d, $J = 8.2$ Hz, 1H), 7.71 – 7.61 (m, 4H), 7.60 – 7.51 (m, 6H), 7.48 (ddd, $J = 8.1, 6.9, 1.2$ Hz, 1H), 7.16 (s, 4H), 4.62 (d, $J = 2.2$ Hz, 2H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 194.5, 190.1, 141.7, 141.4, 135.9, 135.5, 133.6, 133.2, 132.4, 132.2, 132.0, 131.7, 131.7, 131.5, 131.5, 131.0, 130.1, 129.9, 129.7, 129.4, 129.0, 128.8, 128.7, 128.5, 128.4, 127.9, 127.7, 127.1, 126.8, 126.7, 124.3, 123.8, 123.5, 122.2, 67.2, 50.9.

HRMS m/z calculated for $[\text{M}+\text{Na}]^+$ $\text{C}_{38}\text{H}_{24}\text{Br}_2\text{NaO}_2\text{S}_2$ 756.9482. Found 756.9488.

2-(4-Benzoyl-2,5-di(naphthalen-2-yl)-1,3-dithiol-2-yl)-1-phenylethan-1-one (B19)



Eluent: hexane/ethyl acetate: 19/1 to 9/1.

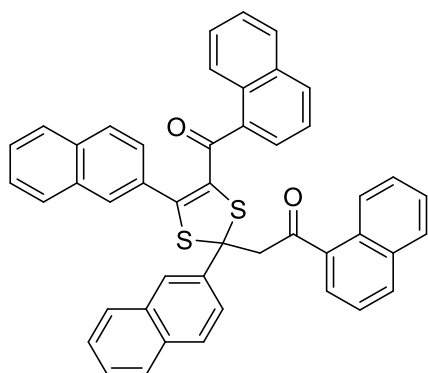
Bright yellow solid (69 mg, 48%).

^1H NMR (500 MHz, CDCl_3) δ 8.42 (d, $J = 2.0$ Hz, 1H), 7.96 (dd, $J = 8.6, 1.9$ Hz, 3H), 7.92 – 7.85 (m, 2H), 7.85 – 7.79 (m, 2H), 7.71 – 7.62 (m, 2H), 7.61 – 7.52 (m, 3H), 7.53 – 7.38 (m, 7H), 7.29 (dd, $J = 8.5, 1.8$ Hz, 1H), 7.11 (t, $J = 7.4$ Hz, 1H), 6.98 (t, $J = 7.7$ Hz, 2H), 4.65 (s, 2H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 194.8, 190.6, 144.2, 139.3, 136.8, 136.2, 133.7, 133.3, 132.8, 132.7, 132.5, 132.5, 130.0, 129.5, 129.1, 128.8, 128.5, 128.5, 128.2, 128.2, 127.9, 127.7, 127.5, 127.5, 127.0, 126.8, 126.7, 126.6, 126.5, 126.5, 126.0, 124.2, 66.6, 51.0.

HRMS m/z calculated for $[\text{M}+\text{Na}]^+ \text{C}_{38}\text{H}_{26}\text{NaO}_2\text{S}_2$ 601.1272. Found 601.1278.

2-(4-(1-Naphthoyl)-2,5-di(naphthalen-2-yl)-1,3-dithiol-2-yl)-1-(naphthalen-1-yl)ethan-1-one
(B20)



Eluent: hexane/ethyl acetate: 19/1 to 9/1.

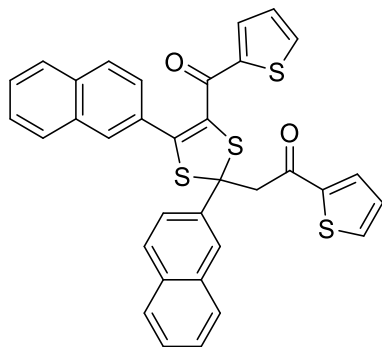
Bright yellow solid (78 mg, 46%).

^1H NMR (500 MHz, $\text{DMSO}-d_6$) δ 8.6 (d, $J = 2.0$ Hz, 1H), 8.4 (d, $J = 7.2$ Hz, 1H), 8.2 – 8.2 (m, 3H), 8.1 – 8.0 (m, 3H), 8.0 (d, $J = 8.1$ Hz, 1H), 8.0 – 7.9 (m, 1H), 7.7 – 7.7 (m, 2H), 7.6 (ddd, $J = 20.0, 14.6, 7.8$ Hz, 5H), 7.5 (dd, $J = 13.0, 7.9$ Hz, 2H), 7.4 – 7.3 (m, 5H), 7.3 (dd, $J = 11.9, 8.0$ Hz, 2H), 7.1 (dd, $J = 8.6, 1.7$ Hz, 1H), 6.9 (t, $J = 7.6$ Hz, 1H), 5.2 (d, $J = 18.3$ Hz, 1H), 5.0 (d, $J = 18.2$ Hz, 1H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, DMSO) δ 200.2, 190.7, 148.5, 140.5, 135.6, 134.6, 133.9, 133.8, 133.1, 132.8, 132.7, 132.0, 131.6, 130.2, 129.9, 129.8, 129.7, 129.5, 129.0, 129.0, 128.9, 128.8, 128.5, 128.5, 128.4, 128.1, 127.9, 127.6, 127.6, 127.5, 127.5, 127.2, 127.2, 127.0, 126.7, 126.6, 126.4, 126.0, 125.4, 125.3, 125.2, 125.0, 124.4, 66.0, 53.4.

HRMS m/z calculated for $[M+Na]^+$ $C_{46}H_{30}NaO_2S_2$ 701.1585. Found 701.1592.

2-(2,4-Di(naphthalen-2-yl)-5-(thiophene-2-carbonyl)-1,3-dithiol-2-yl)-1-(thiophen-2-yl)ethan-1-one (B21)



Eluent: hexane/ethyl acetate: 19/1 to 9/1.

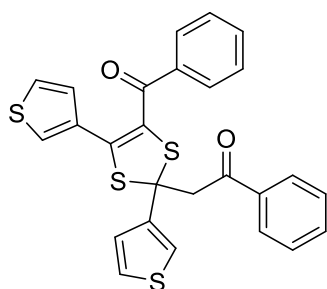
Bright yellow solid (62 mg, 42%).

1H NMR (500 MHz, $CDCl_3$) δ 8.4 (d, J = 2.1 Hz, 1H), 7.9 – 7.9 (m, 2H), 7.9 – 7.8 (m, 2H), 7.8 – 7.7 (m, 1H), 7.6 (d, J = 4.9 Hz, 1H), 7.6 (d, J = 8.5 Hz, 1H), 7.5 – 7.4 (m, 5H), 7.4 (dd, J = 8.6, 1.9 Hz, 1H), 7.4 (d, J = 4.9 Hz, 1H), 7.3 (d, J = 3.8 Hz, 1H), 7.1 (t, J = 4.4 Hz, 1H), 6.6 (t, J = 4.4 Hz, 1H), 4.5 (d, J = 17.0 Hz, 1H), 4.5 (d, J = 17.1 Hz, 1H).

$^{13}C\{^1H\}$ NMR (126 MHz, $CDCl_3$) δ 187.5, 182.4, 143.5, 142.7, 142.2, 138.6, 134.6, 134.5, 133.4, 132.8, 132.5, 129.7, 129.6, 128.5, 128.5, 128.4, 128.3, 128.0, 127.7, 127.6, 127.5, 127.0, 126.7, 126.6, 126.6, 126.5, 126.1, 124.2, 67.4, 51.3.

HRMS m/z calculated for $[M+Na]^+$ $C_{34}H_{22}NaO_2S_4$ 613.0400. Found 613.0409.

2-(4-Benzoyl-2,5-di(thiophen-3-yl)-1,3-dithiol-2-yl)-1-phenylethan-1-one (B22)



Eluent: hexane/ethyl acetate: 19/1 to 9/1.

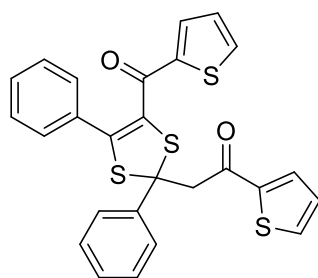
Bright yellow solid (69 mg, 56%).

^1H NMR (300 MHz, CDCl_3) δ 8.0 – 7.9 (m, 2H), 7.7 – 7.6 (m, 5H), 7.5 – 7.4 (m, 2H), 7.4 – 7.4 (m, 3H), 7.3 – 7.3 (m, 2H), 7.2 (ddt, J = 8.1, 6.6, 1.2 Hz, 3H), 7.0 (dd, J = 5.0, 3.0 Hz, 1H), 6.9 (dd, J = 5.0, 1.3 Hz, 1H), 4.4 (d, J = 17.5 Hz, 1H), 4.4 (d, J = 17.5 Hz, 1H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (75 MHz, CDCl_3) δ 194.8, 190.5, 143.7, 136.7, 136.1, 133.8, 132.8, 132.6, 129.3, 128.8, 128.3, 128.2, 128.0, 127.4, 126.5, 126.3, 125.7, 123.2, 50.3.

HRMS m/z calculated for $[\text{M}+\text{Na}]^+$ $\text{C}_{26}\text{H}_{18}\text{NaO}_2\text{S}_4$ 513.0087. Found 513.0093.

2-(2,4-Diphenyl-5-(thiophene-2-carbonyl)-1,3-dithiol-2-yl)-1-(thiophen-2-yl)ethan-1-one (B23)



Eluent: hexane/ethyl acetate: 19/1 to 9/1.

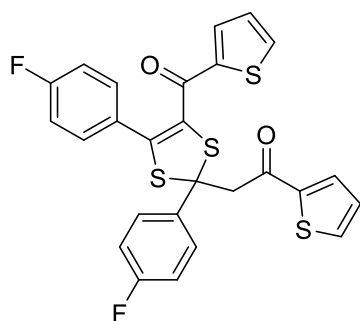
Bright yellow solid (72 mg, 59%).

^1H NMR (300 MHz, CDCl_3) δ 7.9 – 7.8 (m, 2H), 7.7 (dd, J = 3.8, 1.2 Hz, 1H), 7.6 (dd, J = 5.0, 1.1 Hz, 1H), 7.5 (dd, J = 4.9, 1.2 Hz, 1H), 7.4 (ddd, J = 6.2, 4.7, 2.7 Hz, 4H), 7.3 – 7.3 (m, 3H), 7.2 (tq, J = 4.3, 1.6 Hz, 4H), 7.1 (dd, J = 5.0, 3.8 Hz, 1H), 6.8 (dd, J = 4.9, 3.8 Hz, 1H), 4.5 (d, J = 17.1 Hz, 1H), 4.4 (d, J = 17.1 Hz, 1H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (75 MHz, CDCl_3) δ 187.6, 182.3, 143.5, 143.4, 142.7, 142.4, 141.6, 134.6, 134.5, 132.6, 129.7, 129.6, 128.4, 128.4, 128.3, 128.1, 127.7, 126.6, 125.3, 67.4, 51.5.

HRMS m/z calculated for $[\text{M}+\text{Na}]^+$ $\text{C}_{26}\text{H}_{18}\text{NaO}_2\text{S}_4$ 513.0087. Found 513.0098.

2-(2,4-Bis(4-fluorophenyl)-5-(thiophene-2-carbonyl)-1,3-dithiol-2-yl)-1-(thiophen-2-yl)ethan-1-one (B24)



Eluent: hexane/ethyl acetate: 19/1 to 9/1.

Bright yellow solid (60 mg, 46%).

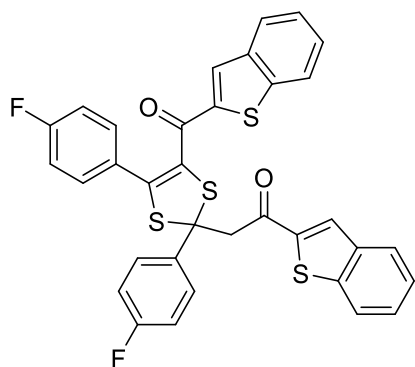
^1H NMR (500 MHz, CDCl_3) δ 7.8 (dd, J = 8.8, 5.2 Hz, 2H), 7.7 (d, J = 3.8 Hz, 1H), 7.7 (d, J = 5.0 Hz, 1H), 7.5 (dd, J = 5.0, 1.1 Hz, 1H), 7.3 (dd, J = 8.6, 5.4 Hz, 2H), 7.1 (t, J = 4.4 Hz, 1H), 7.0 (t, J = 8.6 Hz, 2H), 6.9 (t, J = 8.6 Hz, 2H), 6.8 (t, J = 4.4 Hz, 1H), 4.4 (d, J = 17.0 Hz, 1H), 4.3 (d, J = 17.0 Hz, 1H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 187.4, 182.0, 163.7 (d, J = 127.8 Hz), 161.7 (d, J = 125.1 Hz), 143.3, 142.5, 141.0, 137.5 (d, J = 3.3 Hz), 134.8 (d, J = 9.8 Hz), 134.4, 132.6, 131.5 (d, J = 8.5 Hz), 128.5 (d, J = 8.1 Hz), 128.3, 127.8, 125.4, 115.6 (d, J = 21.9 Hz), 115.3 (d, J = 21.7 Hz), 66.9, 51.3.

^{19}F NMR (471 MHz, CDCl_3) δ -110.2, -113.9.

HRMS m/z calculated for $[\text{M}+\text{Na}]^+$ $\text{C}_{26}\text{H}_{16}\text{F}_2\text{NaO}_2\text{S}_4$ 548.9899. Found 548.9890.

1-(Benzo[b]thiophen-2-yl)-2-(4-(benzo[b]thiophene-2-carbonyl)-2,5-bis(4-fluorophenyl)-1,3-dithiol-2-yl)ethan-1-one (B25)



Eluent: hexane/ethyl acetate: 19/1 to 9/1.

Bright yellow solid (61 mg, 39%).

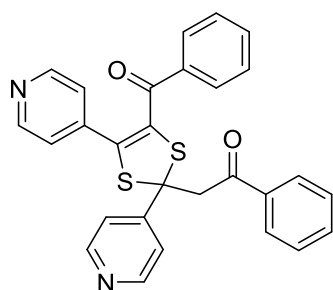
^1H NMR (500 MHz, CDCl_3) δ 8.0 (s, 1H), 7.9 – 7.8 (m, 5H), 7.8 (d, J = 8.2 Hz, 1H), 7.6 (d, J = 8.0 Hz, 1H), 7.5 (s, 1H), 7.5 (d, J = 7.2 Hz, 1H), 7.4 – 7.4 (m, 4H), 7.3 (d, J = 7.8 Hz, 1H), 7.1 (t, J = 8.6 Hz, 2H), 6.9 (t, J = 8.6 Hz, 2H), 4.5 (d, J = 17.0 Hz, 1H), 4.4 (d, J = 16.9 Hz, 1H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 189.0, 183.4, 164.3, 163.3, 162.3, 161.3, 142.8, 142.8, 142.7, 142.0, 141.6, 138.9, 138.5, 137.2, 137.2, 131.8, 131.5, 131.4, 130.0, 128.7, 128.6, 128.0, 127.7, 126.2, 126.0, 125.3, 125.1, 123.0, 122.8, 115.8, 115.6, 115.5, 115.3, 67.5, 51.2.

^{19}F NMR (471 MHz, CDCl_3) δ -109.92, -113.60.

HRMS m/z calculated for $[\text{M}+\text{Na}]^+$ $\text{C}_{34}\text{H}_{20}\text{F}_2\text{NaO}_2\text{S}_4$ 649.0212. Found 649.0218.

2-(4-Benzoyl-2,5-di(pyridin-4-yl)-1,3-dithiol-2-yl)-1-phenylethan-1-one (B26)



Eluent: hexane/ethyl acetate: 19/1 to 9/1.

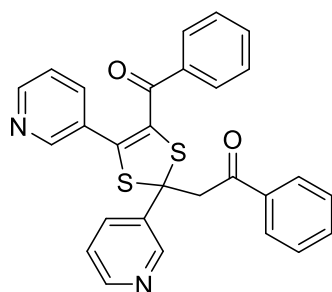
Bright yellow solid (52 mg, 43%).

^1H NMR (500 MHz, CDCl_3) δ 8.65 (d, $J = 5.4$ Hz, 1H), 8.35 (d, $J = 5.2$ Hz, 2H), 7.90 (d, $J = 7.7$ Hz, 2H), 7.76 (d, $J = 6.8$ Hz, 2H), 7.60 (t, $J = 7.4$ Hz, 1H), 7.55 (d, $J = 7.6$ Hz, 2H), 7.47 (t, $J = 7.7$ Hz, 2H), 7.38 (t, $J = 7.3$ Hz, 2H), 7.20 (t, $J = 7.7$ Hz, 2H), 7.09 (d, $J = 6.8$ Hz, 2H), 4.49 (s, 2H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 194.1, 189.5, 151.7, 150.2, 149.9, 139.2, 138.6, 135.9, 135.5, 134.2, 133.7, 129.3, 129.2, 128.9, 128.5, 128.1, 123.5, 121.1, 66.1, 50.7.

HRMS m/z calculated for $[\text{M}+\text{Na}]^+$ $\text{C}_{28}\text{H}_{20}\text{N}_2\text{NaO}_2\text{S}_2$ 503.0864. Found 503.0870.

2-(4-Benzoyl-2,5-di(pyridin-3-yl)-1,3-dithiol-2-yl)-1-phenylethan-1-one (B27)



Eluent: hexane/ethyl acetate: 19/1 to 9/1.

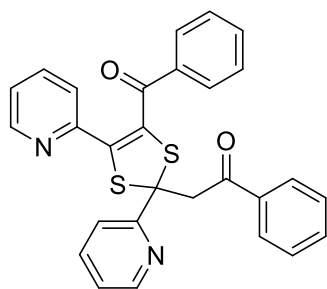
Bright yellow solid (58 mg, 48%).

^1H NMR (500 MHz, CDCl_3) δ 9.10 (d, $J = 2.7$ Hz, 1H), 8.60 – 8.50 (m, 1H), 8.45 (d, $J = 2.4$ Hz, 1H), 8.34 (dd, $J = 4.9, 1.8$ Hz, 1H), 8.26 (dt, $J = 8.3, 2.1$ Hz, 1H), 7.92 (d, $J = 7.8$ Hz, 2H), 7.59 (d, $J = 7.3$ Hz, 1H), 7.54 – 7.45 (m, 5H), 7.37 – 7.30 (m, 2H), 7.16 (t, $J = 7.7$ Hz, 2H), 7.01 (dd, $J = 8.0, 4.7$ Hz, 1H), 4.53 (s, 2H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 194.3, 189.5, 150.3, 149.9, 149.1, 147.8, 140.0, 138.5, 136.8, 136.2, 135.6, 134.5, 134.1, 133.1, 129.7, 129.2, 128.9, 128.4, 128.3, 128.1, 123.0, 122.8, 64.9, 50.9.

HRMS m/z calculated for $[\text{M}+\text{Na}]^+$ $\text{C}_{28}\text{H}_{20}\text{N}_2\text{NaO}_2\text{S}_2$ 503.0864. Found 503.0870.

2-(4-Benzoyl-2,5-di(pyridin-2-yl)-1,3-dithiol-2-yl)-1-phenylethan-1-one (B28)



Eluent: hexane/ethyl acetate: 19/1 to 9/1.

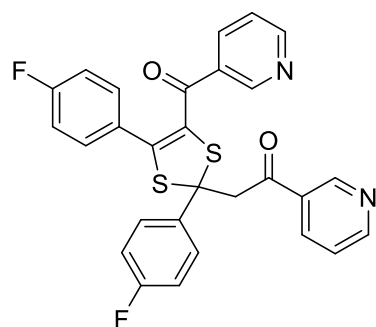
Bright yellow solid (67 mg, 56%).

^1H NMR (500 MHz, CDCl_3) δ 8.41 (d, $J = 3.0$ Hz, 0H), 8.36 (d, $J = 8.0$ Hz, 1H), 8.29 (d, $J = 3.8$ Hz, 1H), 8.04 – 7.90 (m, 2H), 7.90 – 7.71 (m, 3H), 7.57 (t, $J = 7.6$ Hz, 1H), 7.50 – 7.35 (m, 3H), 7.21 (d, $J = 7.9$ Hz, 1H), 7.15 (dd, $J = 7.6, 4.9$ Hz, 1H), 6.97 (dd, $J = 7.6, 4.7$ Hz, 1H), 4.78 (dd, $J = 17.2, 3.5$ Hz, 2H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 195.8, 190.2, 161.1, 150.4, 149.0, 148.6, 137.2, 136.9, 136.8, 136.2, 136.1, 133.5, 132.9, 129.6, 129.2, 128.6, 128.3, 128.2, 123.3, 122.7, 122.3, 67.5, 51.3.

HRMS m/z calculated for $[\text{M}+\text{Na}]^+$ $\text{C}_{28}\text{H}_{20}\text{N}_2\text{NaO}_2\text{S}_2$ 503.0864. Found 503.0870.

2-(2,4-Bis(4-fluorophenyl)-5-nicotinoyl-1,3-dithiol-2-yl)-1-(pyridin-3-yl)ethan-1-one (B29)



Eluent: hexane/ethyl acetate: 19/1 to 9/1.

Bright yellow solid (62 mg, 46%).

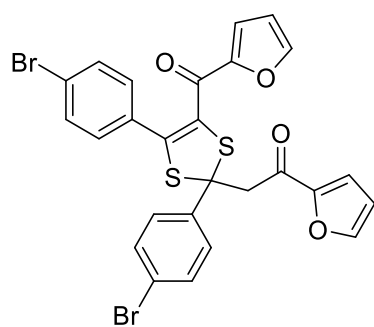
^1H NMR (500 MHz, CDCl_3) δ 9.1 (s, 1H), 8.8 (d, $J = 5.3$ Hz, 1H), 8.6 (s, 1H), 8.5 (d, $J = 4.9$ Hz, 1H), 8.2 (d, $J = 8.1$ Hz, 1H), 7.8 (dd, $J = 9.0, 5.0$ Hz, 2H), 7.7 (d, $J = 7.6$ Hz, 1H), 7.4 (dd, $J = 8.0, 4.5$ Hz, 1H), 7.2 (dd, $J = 8.7, 5.2$ Hz, 2H), 7.1 (m, 3H), 6.8 (t, $J = 8.5$ Hz, 2H), 4.5 (s, 2H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 193.6, 188.2, 163.8 (d, $J = 149.7$ Hz), 161.8 (d, $J = 146.4$ Hz), 145.6, 137.7 (d, $J = 2.9$ Hz), 136.0, 135.5, 132.3, 132.1 (d, $J = 8.6$ Hz), 131.3, 128.4 (d, $J = 8.6$ Hz), 127.4 (d, $J = 3.3$ Hz), 126.7, 123.8, 122.9, 115.7 (d, $J = 21.9$ Hz), 65.4, 51.5.

^{19}F NMR (471 MHz, CDCl_3) δ -108.8, -113.4.

HRMS m/z calculated for $[\text{M}+\text{Na}]^+$ $\text{C}_{28}\text{H}_{18}\text{F}_2\text{N}_2\text{NaO}_2\text{S}_2$ 539.0675. Found 539.0682.

2-(2,4-Bis(4-bromophenyl)-5-(furan-2-carbonyl)-1,3-dithiol-2-yl)-1-(furan-2-yl)ethan-1-one (B30)



Eluent: hexane/ethyl acetate: 19/1 to 9/1.

Bright yellow solid (62 mg, 40%).

^1H NMR (500 MHz, CDCl_3) δ 7.7 (d, $J = 8.9$ Hz, 1H), 7.6 (d, $J = 2.0$ Hz, 1H), 7.5 (d, $J = 8.9$ Hz, 1H), 7.4 (s, 0H), 7.4 – 7.3 (m, 2H), 7.2 – 7.2 (m, 2H), 6.9 (d, $J = 3.7$ Hz, 1H), 6.5 (dd, $J = 3.6, 1.8$ Hz, 1H), 6.3 (dd, $J = 3.6, 1.8$ Hz, 1H), 4.3 (d, $J = 1.7$ Hz, 1H).

$^{13}\text{C}\{^1\text{H}\}$ NMR (126 MHz, CDCl_3) δ 183.5, 176.3, 152.1, 150.9, 146.9, 146.8, 142.7, 140.8, 131.9, 131.6, 131.6, 131.6, 131.5, 131.5, 131.5, 131.4, 131.3, 130.9, 130.8, 130.8, 130.1, 129.9, 128.3, 128.3, 128.3, 125.5, 123.9, 122.2, 120.0, 120.0, 120.0, 118.0, 112.7, 112.4, 66.6, 50.3.

HRMS m/z calculated for $[\text{M}+\text{Na}]^+$ $\text{C}_{26}\text{H}_{16}\text{Br}_2\text{NaO}_4\text{S}_2$ 636.8754. Found 636.8760.

ASSOCIATED CONTENT

Supporting information

The Supporting Information is available free of charge on the

ACS Publications website at DOI: 10.1021/acs.joc.xxxxxxx.

Copies of ^1H and ^{13}C NMR spectra data for all new compounds. X-ray crystallographic data for **B0** and **B11** (CCDC 1952724 and 1952725, respectively).

This material is available free of charge via the Internet at <http://pubs.acs.org>.

REFERENCES

- (1) For reviews on the use of elemental sulfur in organic synthesis, see: (a) Nguyen, T. B. Recent Advances in Organic Reactions Involving Elemental Sulfur. *Adv. Synth. Catal.* **2017**, *359*, 1106-1130. (b) Nguyen, T. B. Elemental sulfur and molecular iodine as efficient tools for carbon-nitrogen bond formation via redox reactions. *Asian J. Org. Chem.* **2017**, *6*, 477.
- (2) (a) Iyoda, M.; Hasegawa, M.; Miyake, Y. Bi-TTF, Bis-TTF, and Related TTF Oligomers. *Chem. Rev.* **2004**, *104*, 5085-5114. (b) Canavet, D.; Sallé, M.; Zhang, G.; Zhang, D.; Zhu, D. Tetrathiafulvalene (TTF) derivatives: key building-blocks for switchable processes. *Chem. Commun.* **2009**, 2245-2269.
- (3) (a) Fleck, N.; Hett, T.; Brode, J.; Meyer, A.; Richert, S.; Schiemann, O. C–C Cross-Coupling Reactions of Trityl Radicals: Spin Density Delocalization, Exchange Coupling, and a Spin Label. *J. Org. Chem.* **2019**, *84*, 3293. (b) Liu, Y.; Villamena, F. A.; Rockenbauer, A.; Song, Y. Zweier, J. L. Structural Factors Controlling the Spin–Spin Exchange Coupling: EPR Spectroscopic Studies of Highly Asymmetric Trityl–Nitroxide Biradicals. *J. Am. Chem. Soc.* **2013**, *135*, 2350-2356. (c) Hintz, H.; Vanas, A.; Klose, D.; Jeschke, G.; Godt, A. Trityl Radicals with a Combination of the Orthogonal Functional Groups Ethyne and Carboxyl: Synthesis without a Statistical Step and EPR Characterization. *J. Org. Chem.* **2019**, *84*, 3304-3320.
- (4) Uchimasu, H.; Matsumura, K.; Tsuda, M.; Kumagai, K.; Akakabe, M.; Fujita, M. J.; Sakai, R. Mellpaladines and dopargimine, novel neuroactive guanidine alkaloids from a Palauan Didemnidae tunicate. *Tetrahedron* **2016**, *72*, 7185-7193.

(5) Rivero, R. A.; Greenlee, W. J.; Chang, R. S. L. Substituted 1,3-benzodioxole & 1,3-benzodithiole-2- carboxylates and their tetrazole analogs with potent binding affinity to the angiotensin II AT1 receptor. *Bioorg. Med. Chem. Lett.* **1994**, *4*, 747-750.

(6) (a) McKinnon, D. M.; Buchshriber, J. M. The Reaction of 5-Unsubstituted-1,2-Dithiole-3-thiones with some Activated Acetylenes. *Can. J. Chem.* **1971**, *49*, 3299-3304. (b) Meijer, J.; Vermeer, P.; Bos, H. J. T.; Brandsma, L. Thermal rearrangement of 2-alkynyl alkanedithioates. *Recl. Trav. Chim. Pays-Bas* **1973**, *92*, 1067-1072. (b) Konstantinova, L. S.; Lysov, K. A.; Amelichev, S. A.; Obruchnikova, N. V.; Rakitin, O. A. A one-pot synthesis and 1,3-dipolar cycloaddition of [1,2]dithiolo[4,3-*b*]indole-3(4*H*)-thiones. *Tetrahedron* **2009**, *65*, 2178-2183. (c) Benzyne with dithioesters: Kentaro, O; Akiko, N.; Yoshinobu, Y. Reaction of Dithiolactones with Benzyne: A Novel Synthesis of Benzo-1,3-dithioles. *Heterocycles* **2008**, *75*, 1417-1424.

(7) (a) Kim, K.-T.; Okazaki, R.; Inamoto, N. Synthesis and Properties of *o*-Thioquinone Methides Having Ketene Aminal, Ketene Acetal, Ketene Monothioacetal, or Ketene Dithioacetal Group. *Bull. Chem. Soc. Jpn.* **1979**, *52*, 3640-3646. (b) Burton D. J.; Inouye Y. Synthesis of Perfluorinated 1,3-Dithioles from Fluoroolefins. *Chem. Lett.* **1982**, *11*, 201-204. (c) Ngounda, M.; Bozec, H. L.; Dixneuf, P. New Synthesis of 1,3-Dithiole and 1,3-Thiazole-2-thiones Promoted by Iron Complexes. *J. Org. Chem.* **1982**, *47*, 4000-4002. (d) Kreitsberga, Y. N.; Vilyuma, é. V.; Khodorkovskii, V. Y.; Neiland, O. Y. Synthesis and transformations of 1,3-dithiol-2-ylideneisopropylidene malonates. *Chem. Heterocycl. Compd.* **1988**, *24*, 1339-1342. (e) Hartke, K.; Rettberg, N.; Dutta, D.; Gerber, H. Dithio- und Thionester, 59. Zur Reaktion von aliphatischen Dithiosäure-dianionen mit Schwefelkohlenstoff. *Liebigs Ann. Chem.* **1993**, 1081-1089. (f) Frère, P.; Gorgues, A.; Jubault, M.; Riou, A.; Gouriou, Y.; Roncali, J. Electrochemically induced intramolecular cyclization of 1,2-bis(1,4-dithiafulven-6-yl)benzenes. *Tetrahedron Lett.* **1994**, *35*, 1991-1994. (g) Reimann-Andersen, S.; Pritzkow, H. Sundermeyer. H. Halogen-Kohlenstoff-Schwefel-Verbindungen: Zur Chemie von 1,1,1,4,4,4-Hexafluor-2-buten-2,3-bissulfenylchlorid. *Chem. Ber.* **1994**, *127*, 533-539. (h) Soliman, A. M.; Sultan, A. A.; El-Shafei, A. K.

Synthesis of Some New Dispiro[dipyrano-(2,4'.6,4'')bidithiolo(4,f-b.4,5-e)- 4,8-benzoquinones. *Monat. Chem.* **1995**, *126*, 615-619.

(8) (a) Mayer, R.; Gebhardt, B. Schwefel-Heterocyclen und Vorstufen, XXXIII. Präparative Synthese und Folgereaktionen des Isotrithions (1.3-Dithiol-thions-(2)) und Isodithions (1.3-Dithiol-ons-(2)). *Chem. Ber.* **1964**, *97*, 1298 – 1307. (b) Bock, H.; Rittmeyer, P. Zur Darstellung von Benzodithiet-Derivaten durch Thermolyse 1,2-Dithio-Substituierter Benzole [*Phosphorus Sulfur Silicon Relat. Elem.* **1988**, *35*, 291-308. (c) Ando, W.; Tokitoh, N.; Kabe, Y. Synthesis and Reaction of Novel Sulfur Containing Heterocyclic Systems. *Phosphorus Sulfur Silicon Relat. Elem.* **1991**, *58*, 179-205. (d) Gusarova, N. K.; Chernysheva, N. A.; Yas'ko, S. V.; Korchevin, N. A.; Dolgushin, G. V.; Trofimov, B. A. Effect of microwave irradiation on the reaction of elemental sulfur with Phenylacetylene. *Doklady Chem.* **2004**, *399*, 240-241. (e) Petrov, V. A.; Marshall, W. Remarkable effect of metal fluoride catalyst on reaction of hexafluoropropene, sulfur and vinyl ethers. Convenient synthesis of 2,2-bis(trifluoromethyl)-4-R-thietanes, 3,3-bis(trifluoromethyl)-5-R-1,2-dithiolanes and 2,2-bis(trifluoromethyl)-4-R-1,3-dithiolanes. *J. Fluorine Chem.* **2010**, *131*, 1144-1155. (f) Arisawa, M.; Ichikawa, Takuya; Tani, Saori; Yamaguchi, M. Synthesis of Symmetrical and Unsymmetrical 1,4-Dithiins by Rhodium-Catalyzed Sulfur Addition Reaction to Alkynes. *Synthesis* **2016**, *48*, 3107-3119. (g) Petrov, Viacheslav; Dooley, Rebecca J.; Marchione, Alexander A.; Diaz, Elizabeth L.; Clem, Brittany S. Simple protocol for preparation of Diels-Alder adducts of perfluorinated thioketone. *J. Fluorine Chem.* **2019**, *225*, 1-10.

(9) Thioamides from amines: (a) Nguyen, T. B.; Ermolenko, L.; Almourabit, A. Efficient and Selective Multicomponent Oxidative Coupling of Two Different Aliphatic Primary Amines into Thioamides by Elemental Sulfur. *Org. Lett.* **2012**, *14*, 4274-4277. Thiophenes: (b) from acetophenones or cyclopentanone: Nguyen, T. B.; Retailleau, P. Sulfurative Self-Condensation of Ketones and Elemental Sulfur: a Three-Component Access to Thiophenes Catalyzed by Aniline Acid-Base Conjugate Pairs. *Green Chem.* **2018**, *20*, 387-390. From acetylenes: (c) Nakayama, J.; Yomoda, R.; Hoshino, M. Reactions of Elemental Sulfur and Selenium with Some Asetylenic Compounds. Formation of

Thiophenes and Selemophenes. *Heterocycles* **1987**, 26, 2215-2222. (d) Blum, J.; Badrieh, Y.; Shaaya, O.; Meltser, L.; Schumann, H. On the Various Modes of Interaction of Sulfur with Phenylated Diynes. *Phosphorus Sulfur Silicon Relat. Elem.* **1993**, 79, 87. (e) Abdel-Wahab, A. A.; Abdel-Rahman, A. E. Molecular Rearrangements: Part I-Pyrolysis of Dibenzyl Sulphide. *Indian J. Chem. B* **1981**, 20, 636-639. (f) Liu, W.; Chen, C.; Liu, H. Synthesis of Polysubstituted Thiophenes via Base-Induced [2+2+1] Cycloaddition Reaction of Alkynes and Elemental Sulfur. *Adv. Synth. Catal.* **2015**, 357, 4050-4054. (g) Alizadeh, A.; Hosseinpour, R. An efficient synthesis of tetraalkyl 2,3,4,5-thiophenetetracarboxylate derivatives. *Synthesis*, **2009**, 1960-1962. (h) From chalcone: Joshi, S.; Dhar, D. N. *Indian J. Chem. B*, **1987**, 26, 179. *o,o'*-Diaminodiphenyltrisulfide from *o*-aminothiophenol: (i) Nguyen, T. B.; Ermolenko, L.; Dean, W. A.; Almourabit, A. Benzazoles from Aliphatic Amines and *o*-Amino/Mercaptan/Hydroxyanilines: Elemental Sulfur as a Highly Efficient and Traceless Oxidizing Agent. *Org. Lett.* **2012**, 14, 5948-5951.

(10) (a) Oxidative trimerization of arylacetonitriles: Nguyen, T. B.; Retailleau, P. Sulfur-Promoted DABCO-Catalyzed Oxidative Trimerization of Phenylacetonitriles. *J. Org. Chem.* **2019**, 84, 5907-5912. (b) Desulfurative trimerization of dibenzyl disulfide: see reference 12.

(11) Tetraphenylthiophenes from: benzyl halides: (a) Przewoski, K.; Voronkov, M. G.; Przewoska, L.; Warnke, Z.; Szafrank, J. Reaction of sulfur with benzyl halides. *Chem Sci, Bull. Acad. Sci. USSR div. Chem. Sci.* **1989**, 38, 579. (b) Lai, Chung-Tin; Hong, J. Aggregation-induced emission in tetraphenylthiophene-derived organic molecules and vinyl polymer. *J. Phys. Chem. B* **2010**, 114, 10302-10310. (c) Lai, C.; Chien, R.; Liu, C.; Hong, J. Thermal and spectral stability of fluorescent copolymers containing tetraphenylthiophene-quinoline unit. *J. Polym. Sci. A* **2011**, 49, 2059-2069.

(12) Nguyen, T. B.; Retailleau, P. Sulfur-Promoted Decarboxylative Sulfurative Hexamerization of Phenylacetic Acids: Direct Approach to Hexabenzylidyne Tetrasulfides. *Org. Lett.* **2019**, 21, 279-282.

(13) When no sulfur was incorporated to the final molecules, the reactions could be classified as oxidative self-condensation.

- (14) For selected reactions involving elemental sulfur that could be performed at rt, see: (a) Nguyen, T. B.; Retailleau, P. Cooperative Activating Effect of Tertiary Amine-DMSO on Elemental Sulfur: Direct Access to Thioaurones from 2'-Nitrochalcones under Mild Conditions. *Org. Lett.* **2018**, *20*, 186-189. (b) Nguyen, T. B.; Ermolenko, L.; Almourabit, A. Three-Component Reaction between Isocyanides, Aliphatic Amines and Elemental Sulfur: Preparation of Thioureas under Mild Conditions with Complete- Atom Economy. *Synthesis* **2014**, *46*, 3172-3179.
- (15) (a) Nguyen, T. B.; Nguyen, L. A.; Retailleau P. Strategy for Contiguous Tetramination of Cyclohexanones with *o*-Phenylenediamines with Elemental Sulfur and DMSO. *Org. Lett.* **2019**, *21*, 6570-6574. (b) Nguyen, T. B.; Retailleau P. Sulfur-Promoted Synthesis of 2-Aroylquinazolin-4(3*H*)-ones by Oxidative Condensation of Anthranilamide and Acetophenones. *Adv. Synth. Catal.* **2019**, *361*, 3337-3341. (c) Nguyen, T. B.; Retailleau P. Sulfur-Catalyzed Stereo and Regioselective Synthesis of Heteropropellanes via Oxidative Condensation of Cyclohexanones with 2-Aminophenols. *Adv. Synth. Catal.* **2019**, *361*, 3588-3592. (d) Nguyen, L. A.; Retailleau P.; Nguyen, T. B. Elemental Sulfur/DMSO-Promoted Multicomponent One-pot Synthesis of Malonic Acid Derivatives from Maleic Anhydride and Amines. *Adv. Synth. Catal.* **2019**, *361*, 2864-2869. (e) Nguyen, T. B.; Retailleau, P. Sulfur-Promoted Aminative Aromatization of 1,2,3,4-Tetrahydrophenazines with Amines: Flexible Access to 1-Aminophenazines. *Adv. Synth. Catal.* **2018**, *360*, 2389-2393.
- (16) Nguyen, T. T. T.; Le, V. A.; Retailleau P.; Nguyen, T. B. Access to 2-Amino-3-Arylthiophenes by Base-Catalyzed Redox Condensation Reaction Between Arylacetonitriles, Chalcones, and Elemental Sulfur. *Adv. Synth. Catal.* **2020**, *362*, 160-165.
- (17) (a) Nguyen, T. B.; Retailleau, P. DIPEA-Promoted Reaction of 2-Nitrochalcones with Elemental Sulfur: An Unusual Approach to 2-Benzoylbenzothiophenes. *Org. Lett.* **2017**, *19*, 4858-4860. (b) Nguyen, T. B.; Retailleau, P. Redox-Neutral Access to Sultams from 2-Nitrochalcones and Sulfur with Complete Atom Economy. *Org. Lett.* **2017**, *19*, 3879-3882.
- (18) (a) Timokhina, L. V.; Sokol'nikova, O. V.; Kanitskaya, L. V.; Voronkov, M. G. 5',5',6,6-Tetramethyl-6,7-dihydro-3'H-spiro[1,3-benzodithiole-2,1'-cyclohexane]-3',4(5H)-dione. *Russian J.*

Org. Chem. **2009**, *45*, 466-467. (b) Dalgaard, L.; Lawesson, S. O. Enethiols. VIII. 3-Mercapto-5,5-dimethyl-2-cyclohexen-1-one ("Thiodimedone") and Derivatives. Thermal and Photochemical Rearrangements.. *Acta Chem. Scand. B* **1974**, *28*, 1077-1090.