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12 5 **THE SULFUR BUDGET AND SULFUR ISOTOPIC COMPOSITION OF MARTIAN**
13 6 **REGOLITH BRECCIA NWA 7533**
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3 34 Abstract. The sulfur isotope budget of Martian regolith breccia (NWA 7533) has been addressed
4 35 from conventional fluorination bulk-rock analyses and ion microprobe in-situ analyses. The bulk
5 36 rock analyses yield 865 ± 50 ppm S in agreement with LA-ICP-MS analyses. These new data
6 37 support previous estimates of 80% S loss resulting from terrestrial weathering of NWA 7533
7 38 pyrite. Pyrite is by far the major S host. Apatite and Fe oxyhydroxides are negligible S carriers,
8 39 as are the few tiny igneous pyrrhotite-pentlandite sulfide grains included in lithic clasts so far
9 40 identified. A small non-zero $\Delta^{33}\text{S}$ (-0.029 ± 0.010 ‰) signal is clearly resolved at the 2σ level in
10 41 the bulk-rock analyses, coupled with negative CDT-normalized $\delta^{34}\text{S}$ (-2.54 ± 0.10 ‰), and near
11 42 zero $\Delta^{36}\text{S}$ (0.002 ± 0.09 ‰). In-situ analyses also yield negative $\Delta^{33}\text{S}$ values (-0.05 to -0.30 ‰)
12 43 with only a few positive $\Delta^{33}\text{S}$ up to $+0.38$ ‰. The slight discrepancy compared to the bulk-rock
13 44 results is attributed to a possible sampling bias. The occurrence of mass-independent
14 45 fractionation (MIF) supports a model of NWA 7533 pyrite formation from surface sulfur that
15 46 experienced photochemical reaction(s). The driving force that recycled crustal S in NWA 7533
16 47 lithologies—magmatic intrusions or impact-induced heat—is presently unclear. However, in-situ
17 48 analyses also gave negative $\delta^{34}\text{S}$ values ($+1$ to -5.8 ‰). Such negative values in the hydrothermal
18 49 setting of NWA 7533 are reflective of hydrothermal sulfides precipitated from $\text{H}_2\text{S}/\text{HS}^-$ aqueous
19 50 fluid produced via open-system thermochemical reduction (TSR) of sulfates at high temperatures
20 51 ($>300^\circ\text{C}$).
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36 53 INTRODUCTION

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38 55 In the absence of ozone, terrestrial sulfur of Archean age exhibits prominent mass
39 56 independent isotope fractionation (MIF) effects due to UV irradiation of atmospheric SO_2 to
40 57 form a variety of reduced and oxidized sulfur species (Farquhar et al., 1998; Pavlov and Kasting,
41 58 2002; Farquhar et al., 2002). The martian atmosphere is likely to have had similar UV irradiation
42 59 effects on volcanically produced SO_2 over most of martian history, thus producing mass-
43 60 independent fractionation in S isotopic signatures (Farquhar et al., 2000; Halevy et al., 2007;
44 61 Righter et al., 2009). In contrast to Earth, the martian sulfur cycle may have been dominated
45 62 throughout its geochemical history by processes operating at the martian surface due to the lack
46 63 of an efficient sulfur crustal recycling mechanism (Foley et al., 2003; King and McLennan,
47 64 1990; Franz et al., 2019 and references therein). Landers, rovers, and remote sensing
48 65 observations by orbiting spacecraft using different spectrometers identified a great diversity of
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3 66 Ca-, Mg-, Fe-sulfate minerals as major components of the martian surface (see a review in Franz
4 et al., 2019a). The abundance and diversity of sulfate minerals testify to a wide range of
5 67 formation processes such as sulfur output through volcanic outgassing of SO₂ (Gaillard et al.
6 68 2009; Tian et al. 2015; Kerber et al. 2015; Franz et al, 2017), oxidation or acid weathering of
7 69 pre-existing igneous pyrrhotite/pyrite to Fe-sulfates (e.g. Zolotov and Shock, 2005; Dehouck et
8 70 al., 2012), hydrothermal precipitation (e.g. McCubbin et al., 2009), and brines from low pH
9 71 waters that contained sulfuric acid (e.g. McLennan et al., 2005), to near neutral, low-salinity
10 72 water that precipitated Ca sulfates by evaporation (e.g. Grotzinger et al., 2013; Nachon et al.,
11 73 2014). The Curiosity rover has identified Ca-sulfate-bearing fractures of diagenetic origin
12 74 deposited from near neutral water in the fluvio-lacustrine sedimentary sequence filling the
13 75 Noachian Gale Crater (Grotzinger et al., 2013; Nachon et al., 2014). Thick veins of Ca-sulfate
14 76 (gypsum) were also identified at Endeavour crater by the Opportunity rover (Squyres et al.,
15 77 2012).
16 78

17 79 Sulfates of martian origin were also identified in the interior of martian meteorites. The
18 80 Chassigny meteorite shows Ca-sulfate associated with carbonate veins (Wentworth and Gooding
19 81 1994). Nakhlite meteorites show evaporitic Ca-sulfate (gypsum, anhydrite) associated with
20 82 halite, Fe-rich carbonate, saponite, Fe³⁺ oxides and silica gels, all well preserved in the Nakhla
21 83 meteorite (Gooding et al. 1991; Bridges and Grady, 2000). Jarosite was found in a melt inclusion
22 84 inside clinopyroxene in MIL 03346 nakhlite (McCubbin et al. 2009). These different sulfates
23 85 point to episodic periods of water flow that exchanged sulfur processed with the atmosphere and
24 86 then deposited oxidized sulfur-bearing minerals through fractures, millions of years after
25 87 complete solidification of the igneous mass (Changela and Bridges, 2011; Bridges et al., 2019,
26 88 and references therein). Sulfates were also assimilated into the nakhlite parent magma at the
27 89 igneous stage, as shown by isotopically anomalous igneous pyrrhotite (Franz et al., 2014; Dottin
28 90 et al., 2018; Mari et al., 2019) that reequilibrated its metal-to-sulfur ratios toward metal-deficient
29 91 compositions (Chevrier et al., 2011; Mari et al., 2019). The nakhlite meteorites which contain
30 92 both sulfate and magmatic sulfide minerals display the largest range of MIF anomalies (Farquhar
31 93 et al., 2007b; Franz et al., 2014 ; Dottin et al., 2018). The Lafayette meteorite, another nakhlite
32 94 contains sulfates and/or sulfides with positive $\Delta^{33}\text{S}$ anomalies (like the Chassigny meteorite)
33 95 while Fe sulfides from the 4.1 Ga-old orthopyroxenite ALH 84001 display large variations from
34 96 negative to positive $\Delta^{33}\text{S}$ values ($-1.10 \pm 0.14\text{‰}$ to $0.66 \pm 0.22\text{‰}$) (Farquhar et al., 2000b; Franz
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97 et al., 2014).

98 Sulfur isotope exchange between oxidized and reduced forms of sulfur during
99 hydrothermal circulation can also generate large mass-dependent fractionations of S isotopes.
100 Impact-driven hydrothermal activity is thought to explain $\delta^{34}\text{S}$ of -9 to -10 ‰ in carbonate-
101 associated pyrites of the 4.1 Ga-old orthopyroxenite ALH 84001 (Shearer et al., 1996). Franz et
102 al (2017) documented Fe-sulfides strongly depleted in ^{34}S ($\delta^{34}\text{S} = -47 \pm 14\%$) in the Cumberland
103 (CB) and Oudam (OU) fluvio-lacustrine sediments of Gale Crater. These values were interpreted
104 to be produced by equilibrium isotopic fractionation between sulfides and sulfates in
105 groundwater warmed by impact or igneous intrusions (Franz et al., 2017). Regolith breccia
106 NWA 7533 (paired with the so-called “Black Beauty” meteorite NWA 7034 (Agee et al., 2013)
107 and paired stones) that has sampled 4.4 Ga old impact lithologies (Humayun et al., 2013) is of
108 special interest for testing this suggestion of Franz et al. (2017). Unlike SNC meteorites, regolith
109 breccias are almost devoid of magmatic sulfides and contain sulfides of mostly hydrothermal
110 origin (pyrite and scarce pyrrhotite; Lorand et al., 2015). This Ni-rich (up to 4.5 wt.% Ni) pyrite
111 is primary unlike other hydrothermal pyrite occurrences so far reported in SNC meteorites and
112 ALH 84001 that are instead the oxidation product of more reduced sulfides (Shearer et al, 1996;
113 Greenwood et al., 2000; Lorand et al., 2018a). This pyrite likely reflects S-rich hydrothermal
114 fluids that percolated through the breccia billions of years after breccia formation (Lorand et al.,
115 2015; Goderis et al., 2016).

116 Our aim is to address the origin of S in the late circulating fluids using the S budget from
117 both bulk-rock S contents and S multi-isotope systematics. We focus on NWA 7533 because it is
118 the only one among the impact breccia samples that has been studied in detail for pyrite
119 distribution and composition including chalcophile trace elements (Lorand et al., 2015; 2018b).
120 The sulfur isotope analyses were done both with digestion of powdered rocks and by ion
121 microprobe *in-situ* measurements on pyrite and pyrrhotite. These data are used to build a
122 comprehensive picture of hydrothermal S cycling in this polymict regolith breccia.

123 124 **MAIN PETROGRAPHIC FEATURES OF NWA 7533** 125

126 The petrogenesis of NWA 7533 impact breccia has been reported in a series of recent
127 papers (Wittmann et al., 2015; McCubbin et al., 2016; Hewins et al., 2017). Its impact origin was

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3 128 identified from the rock textures (various lithic clasts and impact melt rocks) and high (up to
4 129 1,000 ppm Ni, $1 < \text{Ir} < 100$ ppb) concentrations of siderophile elements (Ni, Ir and other highly
5 130 siderophile elements - HSE) (Humayun et al., 2013; Wittmann et al., 2015; Goderis et al., 2016).
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7 131 NWA 7533 meteorite was classified as a polymict breccia because it is composed of separate
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9 132 clasts with chemical compositions that require multiple igneous, sedimentary or impact-melt
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11 133 sources (see Hewins et al., 2017, and references therein).

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14 134 Briefly, its fine-grained Inter-Clast matrix (ICM) contains lithic clasts (norite,
15 135 monzonite) and single mineral clasts derived from these rocks and probably orthopyroxenite, as
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17 136 well as clast-laden melt rocks (CLMR), microbasalt, and melt spherules. Noritic clasts are
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19 137 composed of ferroan pyroxene ($\text{En}_{<72}$), orthopyroxene or inverted pigeonite, plagioclase (An_{50-30})
20 138 and Cr-rich magnetite. Monzonitic clasts show alkali feldspar, often perthitic, associated with
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22 139 plagioclase ($<\text{An}_{30}$), ferroan pyroxene, chlorapatite and Fe-Ti oxides (magnetite, ilmenite and
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24 140 accessory rutile). Noritic and monzonitic clasts contain zircon and baddeleyite that may also
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26 141 occur as individual mineral clasts. Zircons in NWA 7533 and paired meteorites yielded pre-
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28 142 Noachian, U-Pb crystallization ages of 4.47-4.35 Ga with a discordance implying a disturbance
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30 143 at around 1.4 Ga (Humayun et al., 2013; Bellucci et al., 2015; McCubbin et al., 2016; Bouvier et
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32 144 al., 2018). Crystal clasts are fragments of disaggregated lithic clasts. Noritic and monzonitic
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34 145 clasts show widespread evidence of impactor-derived contamination: their concentration levels
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36 146 of siderophile elements (Ni, Ir...) indicate that 3-5 % impactor materials accumulated over
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38 147 repeated impact events (Humayun et al., 2013). This pervasive siderophile enrichment in the
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40 148 melt rocks suggests a thick sequence of impact-generated rocks including mature regolith. The
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42 149 monzonite clasts were formed by either differentiation of these large impact melt sheets or re-
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44 150 melting of primary martian crust at depth in the presence of volatiles. A few pyroxene clasts
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46 151 contain exsolution implying prolonged cooling at depth. Another zoned pyroxene population
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48 152 suggests near surface crystallization (Leroux et al., 2016; Hewins et al., 2017 and references
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50 153 therein).

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52 154 Microbasalts and CLMR clasts show closely similar bulk composition and similar
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54 155 siderophile element enrichment (Humayun et al., 2013). CLMRs are composed of plagioclase
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56 156 laths and subophitic pyroxene indicating rapid quenching due to incorporation of abundant
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58 157 clasts. Microbasalts are composed of orthopyroxene (En_{73-63}) or pigeonite (En_{63-49}), augite (En_{46-}
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60 158 29), plagioclase (An_{66-30}) and Fe-rich spinel. Because of their subophitic to granoblastic textures,

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3 159 microbasalts are interpreted as impact melt that crystallized less rapidly than CLMR (Hewins et
4 160 al., 2017). Regarding lithophile trace element geochemistry, the spherules, the groundmass of the
5 161 melt rock and the microbasalts resemble melted wind-blown dust and regolith debris (Humayun
6 162 et al., 2013). The dense ICM consists of anhedral micrometer-sized plagioclase embedded in
7 163 sub-micrometer sized pyroxene, plus fine-grained Fe-(Ti) oxides – magnetite and maghemite
8 164 (Muttik et al., 2014; Leroux et al., 2016). Disturbances in several isotopic systems (U-Pb in
9 165 zircon; K-Ar/Ar-Ar; Pb-Pb in phosphates; bulk-rock Re/Os systematics) indicate a probable
10 166 major reheating event that annealed the breccia at 1.35–1.4 Ga (Humayun et al., 2013;
11 167 McCubbin et al., 2016; Bellucci et al., 2015; Goderis et al., 2016; Hewins et al., 2017; Cassata et
12 168 al., 2018; see also Leroux et al., 2016).

170 **SULFIDE MINERALOGY OF NWA 7533**

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172 The sulfide mineralogy in NWA 7533 impact breccia was previously studied from the
173 seven polished sections (7533-1 to -7) and one doubly polished transparent thick section (7533-
174 LM) available at the Museum National d'Histoire Naturelle (MNHN) (Lorand et al., 2015).
175 NWA 7533 is unique among martian meteorites in being almost devoid of the magmatic sulfide
176 assemblage of igneous origin (i.e. pyrrhotite-pentlandite-chalcopyrite) repeatedly reported in
177 SNC meteorites (e.g. Chevrier et al., 2011; Lorand et al., 2005; 2018a; Baumgartner et al., 2017;
178 Franz et al., 2014; 2019a,b; Dottin et al., 2018; Mari et al., 2019). Ni-bearing pyrrhotite (3 wt%
179 Ni) + pentlandite; was identified only as three inclusions in plagioclase clasts over the eight
180 polished thick sections investigated. By contrast, accessory pyrite (0.6-1% by weight according
181 to an X-Ray distribution map of S and point counting estimates) is widespread and evenly
182 distributed in all of the impact lithologies (monzonitic and noritic clasts, clast-laden melt rocks
183 (CLMR), melt spherules and microbasalts), in orthopyroxenite clasts from the pristine martian
184 crust, as well as in ICM and the late veins cutting across the meteorite. Hence, pyrite is a late-
185 stage mineral that crystallized after the final assembly of the breccia, during the annealing and
186 lithification event presumed to have occurred 1.4 Ga ago (Lorand et al., 2015).

187 Pyrite crystals show combinations of cubes, truncated cubes and octahedra (average grain
188 size 30-40 μm). These euhedral crystals were observed mostly inside open cracks, microbasalts
189 and ICM (Fig. 1A). In addition to occurring as euhedral crystals, many pyrites are anhedral

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3 190 grains, sometimes of very small size (down to 1 μm across), especially when disseminated inside
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5 191 ICM (Fig. 1B). Some anhedral pyrite grains replace low-Ca pyroxene clasts. Whether euhedral
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7 192 or anhedral, many pyrite grains are poikiloblastic and enclose relict micron-sized Fe-(Ti) oxides
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9 193 identified as being the same composition and habit as the magnetite-maghemite that is so
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11 194 abundant in NWA 7533 (Fig. 1B). Pyrite-producing reactions thus involved sulfidation, either
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13 195 from magnetite/maghemite, from the low-Ca pyroxene, or from dissolved divalent iron in a fluid.
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15 196 Regardless of their mode of formation, NWA 7533 pyrite grains show nickel-rich areas (up to
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17 197 4.5 wt.%; EMPA and SEM data) representing former Fe-Ni metal particles, as well as highly
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19 198 siderophile element-rich micro-inclusions (Os, Ir, Ru, Pt, Pd, Re, Mo) interpreted as altered
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21 199 impactor debris of the pre-Noachian meteoritic bombardment (Lorand et al., 2015, 2018b). It
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23 200 was inferred from this maximum Ni content (positively correlated with Se) that pyrite started
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25 201 crystallizing at 400-500°C. Compared to pyrite, hydrothermal pyrrhotite is very scarce, always
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27 202 anhedral, and highly spongy (see Fig 1C). The assemblage pyrite-pyrrhotite-magnetite and
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29 203 maghemite defines Eh-pH conditions of hydrothermal fluids as near neutral H_2S -HS-rich
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31 204 ($6 < \text{pH} < 10$) oxidizing fluids (minimum $\log f_{\text{O}_2}$ of $> \text{FMQ} + 2$ log units; Lorand et al., 2015) in
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33 205 agreement with the redox conditions deduced from pyroxene clast decomposition into Mg-
34
35 206 rich pyroxene, silica, and Fe-oxide (Leroux et al., 2016). Hydrothermal alteration under
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37 207 oxidizing conditions appears to have given rise to fine-grained Fe-rich phases
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39 208 influencing the magnetic properties of the breccia (Gattaceca et al., 2014).

37 209 Pyrite, a very fragile mineral, shows planar fractures (Fig. 1) resembling planar
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39 210 deformation features (PDF). The shock creating such planar fractures was a very late event,
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41 211 because pyrite was one of the last phases to crystallize in the breccia on Mars (Lorand et al.,
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43 212 2015). The fractures were possibly related to the excavation event that liberated the meteorite
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45 213 from the martian subsurface crust, yet its shock intensity could not be determined due to the lack
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47 214 of olivine and quartz (Hewins et al., 2017). These fractures acted as preferential pathways for
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49 215 partial replacement of pyrite by iron oxyhydroxides of terrestrial origin according to their D/H
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51 216 values and their distribution defining an alteration gradient throughout the meteorite (Lorand et
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53 217 al., 2015). Electron Microprobe analyses (EMP) of altered and hydrated regions in pyrite show
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55 218 compositions similar to that of goethite, but containing $\sim 4\%$ SiO_2 while electron diffraction data

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3 219 indicate the presence of hematite, as well. These oxide mixtures will be treated as iron
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5 220 oxyhydroxides in the present paper (see Table S20 in Hewins et al., 2017).

6 221 7 8 **ANALYTICAL METHODS** 9 223 10 224

11 225 The bulk analyses of the meteorite were performed on two powder splits (BB UMD 1 and
12 226 BB UMD 2) of 0.6 grams each. Both replicates yielded very similar results in terms of S contents
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14 227 and S isotopic compositions (see below), so we assume that these powder splits were
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16 228 representative of the entire breccia. Bulk sulfur abundances and sulfur isotopic compositions of
17
18 229 the bulk sulfur fractions were obtained at the University of Maryland, according to a technique
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20 230 described in detail in Labidi et al. (2012). Briefly, powdered samples were digested in Teflon
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22 231 vessels while continuously flushed with pure N₂. Twenty milliliters (ml) of 2.1 M CrCl₂ solution
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24 232 and 5 ml of 29 N HF (48%, ultrapure) were used to digest the samples. The H₂S released in this
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26 233 process was flushed to a sulfide trap filled with AgNO₃ solution (0.3 M) where it reacted to
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28 234 precipitate Ag₂S. After each extraction and for all samples, weighed Ag₂S precipitate is
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30 235 compared to the S content obtained with EMP and used to determine extraction yields. Weighed
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32 236 silver sulfide samples were subsequently wrapped in Al-foil and placed in Ni reaction vessels for
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34 237 fluorination with at least 10 times excess of pure F₂ at 250 °C overnight. The SF₆ produced was
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36 238 isolated from impurities using cryogenic and chromatographic techniques. The volume of
37
38 239 purified SF₆ was then measured with a manometer and transferred to the dual inlet of a
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40 240 ThermoFinnigan MAT 253 isotope ratio gas-source mass spectrometer that was used to measure
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42 241 ion beams at m/z = 127+, 128+, 129+ and 131+. Once δⁿS values are determined (n = 33, 34,
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44 242 36), Δ³³S and Δ³⁶S are calculated ($\Delta^{33}\text{S} = \delta^{33}\text{S} - 1000((\delta^{34}\text{S}/1000 + 1)^{0.515} - 1)$ and $\Delta^{36}\text{S} = \delta^{36}\text{S} -$
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46 243 $1000((\delta^{34}\text{S}/1000 + 1)^{1.90} - 1)$). IAEA S1 values are used to evaluate our long-term accuracy and
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48 244 precision on standards, and to anchor our δ³⁴S to the V-CDT scale. Analyses of the S1 standard
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50 245 performed along with the Mars rock analyses yield average δ³⁴S, Δ³³S and Δ³⁶S values of -
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52 246 5.28±0.09‰, +0.083±0.008‰ and -0.78±0.26‰ (n=13, all 2s.d.) relative to our reference gas
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54 247 (see details in Labidi et al. 2017). Relative to the same reference gas, our CDT yield δ³⁴S, Δ³³S
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56 248 and Δ³⁶S values at -4.88±0.15‰, -0.035±0.008‰ and -0.08±0.08‰ (n=6, all 2 s.d.). The S
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58 249 isotope shift for δ³⁴S, Δ³³S and Δ³⁶S between CDT and S1 is comparable to what was obtained
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60 250 elsewhere (Ono et al., 2006; Labidi et al., 2012). We use the CDT scale to anchor our δ³⁴S, Δ³³S

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3 251 and $\Delta^{36}\text{S}$ values. We additionally report our $\delta^{34}\text{S}$ values on the V-CDT scale to allow
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5 252 comparisons with other datasets that also reported $\delta^{34}\text{S}$ values relative to V-CDT (e.g., terrestrial
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7 253 mantle-derived rocks, Labidi et al. 2012). We obtained 2σ uncertainties of 0.10, 0.012 and
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9 254 0.25‰ for $\delta^{34}\text{S}$, $\Delta^{33}\text{S}$ and $\Delta^{36}\text{S}$ values, respectively, for all measurements with a SF_6 amount > 1
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11 255 μmol .

12 256 The *in-situ* analyses of S multi-isotopes were done with ion probes on two polished thin
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14 257 sections (7533-4 and 7533-5) over the 8 available. These two polished thin sections show
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16 258 unaltered pyrite crystal cores large enough to be analysed. The dimensions and localisation of the
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18 259 final ion microprobe spots were checked with a scanning electron microscope (Tescan VEGA II
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20 260 LSU Scanning Electron Microscope (SEM) operating in conventional (high-vacuum) mode, and
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22 261 equipped with an SD3 (Bruker) EDS detector (Muséum National d'Histoire Naturelle Paris,
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24 262 France = MNHN) operating in the backscattered mode (BSE) (see Lorand et al., 2015 for more
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26 263 details). Major element compositions of pyrite (Fe, Ni, S) were determined with the same
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28 264 standard-less Energy Dispersive procedure as in Lorand et al. (2015) inside the ion beam craters
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30 265 of SIMS analysis; these latter analyses agree quite well with pyrite compositions reported in
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32 266 Lorand et al. (2015; 2018b)

31 267 Two Cameca IMS 1280 facilities (CRPG, Nancy, France; NordSIMS facility, Stockholm,
32
33 268 Sweden) (Table 1) were used. The analytical method for the CRPG Cameca IMS 1280 HR2 is
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35 269 described in detail in Thomassot et al. (2009) and in Muller et al. (2016). Sulfur isotope
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37 270 compositions of sulfides from thick section 7533-5 were measured by simultaneous
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39 271 measurements of $^{32}\text{S}^-$, $^{33}\text{S}^-$, $^{34}\text{S}^-$, and $^{36}\text{S}^-$ in multicollection mode with three off-axis Faraday
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41 272 cups (L'2, C and H1) that were intercalibrated before each session of analyses, and one electron
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43 273 multiplier (H2) for the most rare isotope (^{36}S). The analytical settings were: 5 nA, Cs^+ - primary
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45 274 beam focused to a spot of about 15 μm , with secondary intensities from 8.2×10^8 to 10.2×10^8
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47 275 cps for ^{32}S , depending on the mineralogy. Gains and backgrounds of the Faraday cups were
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49 276 regularly measured during the entire session. Measurements were performed with an automatic
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51 277 centering process and consist of 30 cycles of 4 seconds each after a pre-sputtering time of 300
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53 278 seconds. The background of the detectors was measured during the pre-sputtering and was then
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55 279 used to correct each analysis. The internal precision achieved under these conditions was better
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57 280 than 0.05‰ for $\delta^{34}\text{S}$ and 0.03‰ for $\delta^{33}\text{S}$ values (2σ). The external precision, determined from
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59 281 repeated measurements on various reference materials, was 0.40‰ (2σ) for $\delta^{34}\text{S}$, 0.09‰ (2σ) for

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3 282 $\Delta^{33}\text{S}$ values and 0.24‰ for $\Delta^{36}\text{S}$. Several pyrite in-house reference materials (Maine, Philippot et
4 al., 2012, Balmat pyrite, courtesy of M. Whitehouse, and Spain-CR, Muller et al., 2016) were
5 283 used to determine (i) the instrumental mass fractionation, and (ii) the reference mass
6 284 discrimination line, from which $\Delta^{33}\text{S}$ and $\Delta^{36}\text{S}$ values were calculated. The values of the in-house
7 285 reference materials can be found in Muller et al. (2017). The bulk isotopic compositions of these
8 286 materials were measured by conventional gas source mass spectrometer at IPGP following
9 287 Labidi et al. (2012).
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11 289 Quadruple S-isotopes were measured in-situ on section 7533-4 at a ca. 10 μm scale using
12 290 a CAMECA ims1280 SIMS instrument (NordSIMS facility, Stockholm), closely following
13 291 analytical protocols outlined by Whitehouse (2013). The sulfur isotopes ^{32}S , ^{33}S and ^{34}S were
14 292 measured on Faraday detectors, while ^{36}S was measured on an ion counting electron multiplier,
15 293 corrected for drift using interspersed reference materials (Balmat and Ruttan pyrites), which
16 294 were also used to correct for instrumental mass bias. Isua pyrite was used to monitor the mass
17 295 independent values.
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19 297 Sulfur contents of Fe oxyhydroxides and apatite that may be potential S reservoirs (e.g.
20 298 Parat et al., 2011) were analysed *in-situ* using a Laser Ablation Inductively Coupled Plasma
21 299 Mass Spectrometer (LA-ICP-MS) at the «Laboratoire de Planétologie et Géodynamique» in
22 300 Nantes (France) following the procedure described in detail by Lorand et al (2018b). The
23 301 samples were ablated and analysed for ^{34}S using a Bruker 880 quadrupole ICP-MS coupled with
24 302 a Photon Machine Analyte™ G2 equipped with an excimer laser (193 nm laser wavelength) and
25 303 a dual volume sample cell that keeps the sample volume small and constant. The laser was
26 304 operated with a repetition rate of 10 Hz and a laser output energy of 90 mJ with a 50% attenuator
27 305 and 20x demagnification, resulting in low fluences on the sample ($<4 \text{ J/m}^2$). Spot sizes for
28 306 standards and samples were set to 25-50 μm depending on the size of the mineral grains. Six Fe
29 307 oxyhydroxide grains replacing pyrite were analysed in section 7533-5 with a spot size of 25 μm ,
30 308 NIST 610 glass as calibration standard and SiO_2 content as internal standard, based on the values
31 309 published by Lorand et al. (2015) and Hewins et al. (2017). Eight apatite crystals from NWA
32 310 7533 - 5 were analysed with a laser spot size of 50 μm , NIST 612 glass as calibration standard
33 311 and Ca as internal standard based on the apatite analyses reported by Hewins et al. (2017).
34 312 Concentrations in both cases were determined using Glitter™ software (Table 2).
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RESULTS

The two bulk-rock analyses yield quite reproducible bulk S concentrations (805-920 ppm). These bulk chemistry values are very close to the mean value calculated from LA-ICP-MS raster analyses of Humayun et al. (2013) although the latter method detected huge variations (100 to 1500 ppm) between each of the breccia components (CLIMR, ICM, microbasalts, melt spherules; see section 2), reflecting small scale variations in modal abundances and weathering degrees of pyrite (Lorand et al., 2015; 2018b). The samples yield an average $\Delta^{33}\text{S}$ of -0.029 ± 0.010 ‰ reflecting a small but resolvable non-zero signal in $\Delta^{33}\text{S}$ at the 2σ level. This non-zero $\Delta^{33}\text{S}$ is coupled with negative $\delta^{34}\text{S}$ (-2.54 ± 0.10 ‰). The NWA 7533 meteorite S isotopic composition is distinct from that of the juvenile martian S isotopic composition calculated by Franz et al. (2019) from 47 shergottite analyses ($\delta^{34}\text{S} = -0.24 \pm 0.05$ ‰, $\Delta^{33}\text{S} = 0.0015 \pm 0.0016$ ‰, $\Delta^{36}\text{S} = 0.039 \pm 0.054$ ‰).

In addition to bulk isotope measurements, 29 spot analyses were performed on 18 pyrite and 3 pyrrhotite grains of sufficient size (>20 μm) that were not significantly altered to Fe-oxyhydroxides. Ion microprobe measurements yielded more negative $\Delta^{33}\text{S}$ values ($-0.05 < \Delta^{33}\text{S} < -0.38$ ‰,) compared to the bulk-rock analyses, and a wider range of negative $\delta^{34}\text{S}$ signatures ($-1.5 < \delta^{34}\text{S} < -3.1$ ‰; with outliers up to -5.5 and $+0.6$ ‰) (Fig. 2). The Nancy SIMS analyses show a larger scatter in the $\delta^{34}\text{S}$ and $\Delta^{33}\text{S}$ values than the bulk analyses and include two positive values for $\Delta^{33}\text{S}$ (0.24 - 0.35 ‰ corresponding to Site 16-1 and Site F-4 pyrites; Table 1). The ranges of $\Delta^{36}\text{S}$ are similar: -0.55 to $+0.49$ ‰ (NORDSIMS) vs. -0.9 to $+0.49$ ‰ (Nancy SIMS), excluding one outlier at -2.2 ‰ (**Site B-1** Pyrite). Correlation between $\Delta^{33}\text{S}$ and $\delta^{34}\text{S}$ is lacking (Fig. 2). There are no correlations between sulfur isotopic compositions and pyrite Ni contents (<0.1 - 3 wt.%; Table 1)(Fig. 3), suggesting that the correction procedure of SIMS analyses was not affected by the small deviation of NWA 7533 pyrite compared to pure FeS_2 . The spread in the $\delta^{34}\text{S}$ values seems to be due to beam overlap over polymineralic aggregates (pyrite-Fe oxides or Fe oxyhydroxides), matrix silicates or vein-filling terrestrial products (calcite). The analyses having incorporated some calcite seem to shift $\delta^{34}\text{S}$ toward more positive values (7533-5-19; 7533-5-17; 7533-5-F-2; 7533-5-1a) (Fig. 1). No systematic trend can be seen for those analyses overlapping Fe oxyhydroxides or matrix silicates ($\delta^{34}\text{S}$ down to -5.58 ‰ for NWA7533-5-G).

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3 343 The six pyrite crystals analysed with the NORDSIMS ion microprobe show a very
4 similar range of negative $\Delta^{33}\text{S}$ values ($-0.05\text{‰} < \Delta^{33}\text{S} < -0.36\text{‰}$) to the 12 Nancy SIMS pyrite
5 344 analyses, however with no positive values. Both data sets also yield consistent results with
6 345 regard to the negative $\delta^{34}\text{S}$ value of NWA 7533 sulfides ($-3.3\text{‰} < \delta^{34}\text{S} < -1.5\text{‰}$). Duplicate
7 346 analyses generated variable reproducibility that reflected the different degrees of sulfide
8 347 alteration from one section to another (e.g. site 16a,b; Table 1), and the grain size of the
9 348 preserved pyrite. Pyrite and the three highly spongy pyrrhotite grains of hydrothermal origin
10 349 analysed here carry similar isotopic compositions at the 2-sigma level (Table 1; Fig. 2). Due to
11 350 their lower count rate the NordSIMS analyses of pyrrhotite are less precise compared to the
12 351 analyses of pyrite, as shown by the two sigma values of the three S isotope ratios. This lower
13 352 precision may result from the highly spongy nature of that pyrrhotite, showing pore sizes much
14 353 smaller ($< 1\text{ }\mu\text{m}$) than the ion beam diameter ($> 10\text{ }\mu\text{m}$).
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26 27 28 356 DISCUSSION

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30 357 At first sight, the occurrence of Mass Independent Fractionation of sulfur (S-MIFs)
31 supports a martian origin for sulfur in NWA 7533 as deduced from microtextural and
32 358 compositional characteristics of NWA 7533 pyrite (Lorand et al., 2015). Indeed, S-MIFs are
33 359 generated by UV-induced photochemical processes in an atmosphere poor in O_2 like the martian
34 360 atmosphere since the earliest times (e.g. Farquhar et al., 2000, 2001; Pavlov and Kasting, 2002).
35 361 In oxygen isotopes, MIF effects also affect the bulk-rock composition of NWA 7034 (Agee et
36 362 al., 2013), phosphates (Belluci et al., 2020) and are largest in NWA 7533 zircons (Nemchin et
37 363 al., 2014). Moreover, the variation in $\Delta^{33}\text{S}$ with an absence of significant anomalies in $\Delta^{36}\text{S}$ (Fig.
38 364 4) is considered to be diagnostic of S-MIF production on Mars, by different pathways from those
39 365 that operated on the early Earth and in Earth's stratosphere today (see e.g. Masterson et al.,
40 366 2011). The calculated mean $\Delta^{36}\text{S}$ value from both the bulk-rock and SIMS analyses is close to
41 367 0 ‰, a characteristic feature of S multi-isotope systematics of martian meteorites compared to
42 368 Archean terrestrial samples showing mass independent signatures (Franz et al., 2014,
43 369 2019a,b)(Fig. 4). The plot in Fig. 4 also rules out alternative fractionation processes, such as, for
44 370 example, mass-dependent fractionation during Rayleigh distillation with alternative mass laws
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3 372 (e.g. Young et al., 2002). This latter process should produce a regression line characterized by a
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5 373 slope of -7 ± 1 whereas our data set does not align on any correlation at all. However, it is worth
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7 374 noting that all but three analyses of NWA 7533 sulfides plot within the scattergram of nakhlites
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9 375 in the $\Delta^{36}\text{S}$ vs. $\Delta^{33}\text{S}$ plot of Fig. 4. This may be more than a coincidence because NWA 7533
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11 376 sulfides were dated at 1.4 Ga, like nakhlites that recorded sulfate contamination at each stage of
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13 377 their petrogenetic history (e.g. Franz et al., 2014; Dottin et al., 2018; Lindsay et al., submitted).
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15 378 This could reflect the S isotopic composition of surficial sulfur at that time on the martian crust,
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17 379 or a specific S reservoir sampled by both martian breccias and nakhlites.

17 380 The presence of sulfur isotopic variations that are both mass-independent and
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19 381 mass-dependent compared to the inferred composition of martian mantle-derived
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21 382 sulfur, argues in favor of a crustal origin of NWA 7533 pyrite consistent with its
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23 383 hydrothermal origin (Lorand et al., 2015). Even taking into account only the bulk-rock
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25 384 data, the small but significant negative $\Delta^{33}\text{S}$ values overlap with the isotopic S compositions
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27 385 reported for some shergottites that are now considered to have assimilated crustal sulfur (NWA
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29 386 11300; NWA 7635; Franz et al., 2019). Before further discussion, it is necessary to address any
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31 387 bias in the database. Hence, the following three points will be addressed 1) how can the slight
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33 388 discrepancy between in-situ analyses and bulk-rock determination of $\Delta^{33}\text{S}$ be explained, 2) how
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35 389 was anomalous S transferred to impact breccias, and 3) what was the ultimate process that
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37 390 triggered hydrothermal cycling of S in these martian regolith breccias.

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39 392 **The balance of the bulk-rock S multi-isotope budget of NWA 7533**

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41 393 Anomalous $\Delta^{33}\text{S}$ for pyrite was observed with two different ion microprobes in two
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43 394 different laboratories. This argues for a significant negative $\Delta^{33}\text{S}$ anomaly as measured by SIMS,
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45 395 which seems to conflict with the near-zero $\Delta^{33}\text{S}$ value of $-0.029\pm 0.010\text{‰}$ measured with precise
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47 396 bulk S isotope measurements. It is not the first time that in-situ analyses and bulk-rock
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49 397 determination of S isotopes do not match. Franz et al. (2014) documented quite a large
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51 398 discrepancy between SIMS analyses and chemical extractions of MIL 03346 pyrrhotite ($\Delta^{33}\text{S} = -$
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53 399 0.73 ± 0.13 vs. -0.43‰) but did not propose a reason for the discrepancy. Dottin et al. (2018)
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55 400 also reported SIMS analyses of nakhlites paired with MIL 00336 yielding lower $\Delta^{33}\text{S}$ compared
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57 401 to bulk rocks, and argued that the SIMS measurements captured a different population of sulfides

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3 402 than the bulk analyses. Another possible source of analytical bias is that S-bearing phases other
4 403 than the two Fe-sulfides, pyrrhotite and pyrite, were overlooked by petrographic studies.
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7 404 Sulfur in NWA 7533 was identified as residing in both magmatic sulfide assemblages,
8 405 apatite and Fe-oxyhydroxides (Lorand et al., 2015; this study). As said before, the impact breccia
9 406 is almost devoid of magmatic sulfides, which is interpreted to result from repeated impact and
10 407 shock melting. It is well known that shock melts are strongly degassed (Pierazzo and Artemieva,
11 408 2012; Righter et al., 2015). Nickel, a strongly chalcophile element was demonstrated to behave
12 409 as a lithophile element in impact-melt derived noritic and monzonitic clasts that crystallized
13 410 without evolving sulfide melts (Hewins et al., 2017 and references therein). The paucity of
14 411 magmatic sulfides in NWA 7533 clearly reflects the repeated heavy bombardment of the martian
15 412 regolith. Another potential S carrier in igneous rocks is apatite (Parrat et al., 2011) which is a late
16 413 magmatic mineral in lithic clasts and microbasalts of the breccia (Hewins et al., 2017). NWA
17 414 7533 apatite S content determined by LA-ICP-MS is quite homogeneous (200-300 ppm; Table
18 415 2). However, apatite is a minor mineral (<6 vol% in NWA 7034, paired with NWA 7533; Agee
19 416 et al., 2013) and its contribution to the whole rock S budget is expected to be negligible (< 15
20 417 ppm vs. 865 ppm). Its effect on the S isotopic composition is difficult to predict without a
21 418 reliable apatite standard for SIMS isotope analyses. However, we can speculate that magmatic
22 419 sulfides or apatite containing juvenile martian sulfur would drive the S multi-isotope
23 420 composition of NWA 7533 toward 0 ‰ in the $\delta^{34}\text{S}$ vs. $\Delta^{33}\text{S}$ diagram of Fig. 2. Alternatively,
24 421 Humayun et al. (2014) suggested that NWA 7533 lithic clasts crystallized from impact
25 422 melts that incorporated already weathered clay-rich martian soils. In that case, evidence
26 423 from SNC meteorites and analyses of Gale crater sulfates (Greenwood et al., 2000;
27 424 Farquhar et al., 2007; Franz et al., 2014, 2017, 2019, Dottin et al., 2018; Mari et al., 2019)
28 425 predict deviation of $\delta^{34}\text{S}$ toward strongly positive values in Fig. 2. Clearly, neither
29 426 magmatic sulfides nor apatite could account for the dilution of the pre-existing non-zero $\Delta^{33}\text{S}$ at
30 427 near constant $\delta^{34}\text{S}$ between in-situ SIMS analyses and the two bulk-rock data.
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52 428 Fe oxyhydroxides contain ten times more S than does apatite (3000±1000 ppm; now
53 429 measured *in-situ* with both EMP and LA-ICP-MS, Table 2). However, simple mass balance
54 430 calculations suggest that Fe oxyhydroxides are a negligible S reservoir, too. They would
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3 431 represent c.a. 40 ppm S (5% of the bulk-rock S concentration) at best taking into account the
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5 432 highest in-situ measured S content ($\approx 4,000$ ppm) and modal abundances of ca. 1 wt.% (implying
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7 433 a total oxidation of pyrite, which is an over-estimate). Some of the sulfur of Fe oxyhydroxides
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9 434 may come from both oxidized pyrites and terrestrial sources with near-zero $\Delta^{33}\text{S}$ values, which
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11 435 would explain the shift in the bulk-rock $\Delta^{33}\text{S}$ values compared to ion microprobe data. From the
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13 436 mass balance calculations, Fe oxyhydroxides would have to have unrealistically positive $\Delta^{33}\text{S}$ of
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15 437 about +6 to +9 ‰ to explain the shift of $\Delta^{33}\text{S}$ from -0.2 ± 0.1 ‰ (SIMS data) relative to the
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17 438 bulk value of -0.03 ± 0.01 ‰.

17 439 Sulfates, also major S-bearing phases on Mars (Franz et al., 2017, 2019 and reference
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19 440 therein), have not been detected in martian breccia NWA 7533 despite careful SEM scanning of
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21 441 the 8 polished pieces studied so far (Lorand et al., 2015; Hewins et al., 2017). There are no
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23 442 coupled Ba+S spikes that could correspond to barite micro-inclusions in the time-resolved
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25 443 diagram of the LA-ICP-MS analyses of Fe oxyhydroxides (Fig. 5). Barium was systematically
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27 444 detected along with S (in agreement with EMP and TEM data for Ca, S and Si). The Sr and Ba
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29 445 ion signals are highly correlated and the ^{34}S signal follows the Ba and Sr signals very closely in
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31 446 the first portion of the scan in Fig. 5, while the Ca signal is independent of the Ba-Sr-S signals.
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33 447 The high Sr might indicate that a Sr-Ba sulfate is the only sulfate present and that it is
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35 448 sufficiently widely distributed at a nanometric scale in the Fe-oxyhydroxides not to form discrete
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37 449 spikes. However, the trend of strongly decreasing S/Se ratio in NWA 7533 Fe oxyhydroxides
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39 450 ($137 < \text{S/Se} < 1,193$) compared to their pyrite precursors ($3,500 < \text{S/Se} < 54,000$) supports the
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41 451 hypothesis of extensive leaching of sulfates from the breccia during terrestrial weathering (Fig.
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43 452 6). Such a drop in the S/Se ratios is consistent with the well documented differences in the
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45 453 behaviour of S and Se in terrestrial weathering (e.g. Lorand et al., 2003 and ref. therein). Under
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47 454 acidic conditions that characterize terrestrial weathering, both Se^{4+} and Se^{6+} tend to adsorb onto
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49 455 naturally occurring Fe oxides and oxyhydroxides while S is leached as SO_4^{2-} (e.g. Ziemkiewicz et
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51 456 al., 2011). From pyrite modal abundance (up to 1 wt.%; Lorand et al., 2015), NWA 7533 should
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53 457 show a bulk-rock S content of 5,400 ppm S instead of 865 ± 80 ppm measured in the present
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55 458 study. This suggests a loss of 80% S as dissolved sulfates during terrestrial weathering, which is
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57 459 consistent with previous estimates, either based on pyrite/Fe oxyhydroxide modal ratios (Lorand
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59 460 et al., 2015) or *in-situ* analyses of pyrite S and Se contents (Lorand et al., 2018b).

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3 461 The discussion above casts serious doubt on whether another S reservoir with a
4 462 significant effect on bulk S isotopic composition is present in these breccias. Another possible
5 463 source of discrepancy between the bulk-rock analyses and in-situ measurements is analytical bias
6 464 generated by one or both of the two techniques. The bulk-rock analysis involved an extraction
7 465 procedure of S using a Cr(II) solution (c.f. Labidi et al., 2012) which has been shown to be very
8 466 inefficient for sulfates. (Gröger et al., 2009). However, this analytical bias was likely negligible
9 467 because as shown above, NWA 7533 meteorite lacks detectable amounts of sulfates. A
10 468 systematic analytical bias from both the Nancy and Stockholm SIMS can be ruled out because
11 469 multiple sulfur analyses were processed in both places the same way using the same reference
12 470 materials and both were mass resolved from $^{32}\text{S}^1\text{H}$. One may speculate that hydride generation
13 471 could be a bit higher on the sulfide standard mount block (in epoxy) (or if the standard sulfides
14 472 themselves are hydrated) than on the meteorite section, thus generating $\Delta^{33}\text{S}$ values biased
15 473 towards higher values that would in return make the unknowns lower in $\Delta^{33}\text{S}$. Another effect that
16 474 could alter the $\Delta^{33}\text{S}$ values measured by SIMS is the Faraday amplifier background drift. Many
17 475 of the sulfides analysed in our paper are yielding somewhat lower 70-80% count rates than the
18 476 standards so a small drift would have a disproportionate effect in any drift correction, probably at
19 477 around the 0.1 permil level (around 1000 cps in 10^7 cps), and it would lower $\Delta^{33}\text{S}$ too. However,
20 478 there is no direct evidence that this putative drift might have operated during the analytical
21 479 session of NWA 7533 sulfides. Moreover the background was systematically measured during
22 480 the pre-sputtering time.

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38 481 The question that arises now is whether the SIMS analyses sampled one or several S
39 482 populations from the same sulfides (pyrite/and or pyrrhotite). This issue can be addressed by
40 483 looking at the uncertainty data on the average triple S compositions reported Table 1 as the
41 484 simple 2x standard deviation on the values, and the weighted average uncertainty on the data,
42 485 along with its MSWD value. Both the NORDSIMS and Nancy data set yield similar results :

43 486 □ large spread of data for $\delta^{34}\text{S}$ (MSWD of 42 for the NORDSIMS analyses and 132 for the
44 487 Nancy analyses), beyond what can be explained by analytical uncertainties. This scatter could
45 488 represent a real Mass Dependent Fractionation (MDF) trend or a subtle analytical issue. Hence,
46 489 the weighted average uncertainty is obviously not a useful measure of
47 490 the scatter for $\delta^{34}\text{S}$ that integrates a component of geological scatter (MDF).

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3 492 -□ $\Delta^{33}\text{S}$ values define statistically single low MSWD (1.1 vs. 1.5 for the Nancy and NORDSIMS
4 data set, respectively) populations that are very slightly negative. In fact, combining both data
5 493 sets gives a weighted average of -0.165 ± 0.032 permil, not overlapping the bulk value. Of
6 494 course, the fact that $\delta^{34}\text{S}$ shows variation means that our analyses may have sampled two (or
7 495 more) reservoirs that are very slightly different in $\Delta^{33}\text{S}$. Testing that interpretation from a $\Delta^{33}\text{S}$
8 496 vs. $\delta^{34}\text{S}$ plot would demand many more analyses with much better individual data uncertainties
9 497 than we have. In that case, the calculation with standard deviations may apply, which would give
10 498 a mean $\Delta^{33}\text{S}$ value of -0.2 ± 0.18 (2SD) for the Stockholm data, -0.12 ± 0.22 (2SD) for the
11 499 Nancy data and a -0.16 ± 0.11 (2SD) for the combined
12 500 dataset. Taken individually, both mean $\Delta^{33}\text{S}$ values overlap the bulk rock data, but when
13 501 combined, they do not.
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15 503 - Our $\Delta^{36}\text{S}$ averages are similar to each other and within error of the
16 504 bulk. However, the Nancy data display a greater spread (MSWD of 13) perhaps inconsistent with
17 505 sampling a single population.
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19 507 To summarize, the fact that SIMS and bulk-rock data do not overlap at 95% confidence
20 508 suggests that the two techniques may have sampled different sulfur isotopic compositions from
21 509 the same Fe-sulfides, assuming that each data set (the UMD bulk-rock data, the NordSIMS data
22 510 and the Nancy data) are single populations. Thus, the case for multiple S populations could be
23 511 stronger than the case for analytical bias. Deciphering which parameter really played the major
24 512 role in our data set is made difficult by the fact that we are comparing data at the 0.1 permil level
25 513 which is very close to the limitations of the SIMS method. We also cannot exclude that the tiny
26 514 pyrite crystals that are well below the size of the SIMS beam (1 vs. 10 μm , respectively) in
27 515 impact breccia NWA 7533 show less negative $\Delta^{33}\text{S}$ compared to the larger crystals measured.
28 516 Clearly, resolving both issues will demand better spatial resolution and analytical precision than
29 517 those of present-day SIMS techniques .
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31 518 **Hydrothermal cycling of MIF-bearing S in NWA 7533**

32 519 As recalled in the introduction, SNC meteorites that have sampled sub-effusive igneous
33 520 rocks of the martian crust (i.e. nakhlites) have provided multiple evidence of crustal S
34 521 contamination that took place throughout the cooling history of these rocks, from the magma
35 522 crystallization stage to late exposure to hydrothermal fluids and evaporitic brines (see also Dottin
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3 523 et al., 2018; Franz et al., 2019; Bridges et al., 2019 and reference therein). These rocks produce
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5 524 trends of positive $\delta^{34}\text{S}$ values coupled with negative $\Delta^{33}\text{S}$ values with water soluble (Ca sulfates)
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7 525 and acid soluble sulfates (jarosite) in Antarctic nakhlites (McCubbin et al., 2009) defining the
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9 526 highest $\delta^{34}\text{S}$ endmembers (Fig. 7). Likewise ALH 84001 (a proxy of NWA 7533 orthopyroxene
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11 527 crystal clasts with similar ages of formation; Hewins et al., 2017) incorporated crustal sulfate
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13 528 enriched in ^{34}S along with anomalous ^{33}S ($\delta^{34}\text{S} = 7.20 \pm 0.15\%$, $\Delta^{33}\text{S} = -0.072 \pm 0.008\%$, $\Delta^{36}\text{S} =$
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15 529 $0.02 \pm 0.15\%$; Franz et al., 2014). We note here that the MIF observed in NWA 7533 (this
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17 530 study) and in the nakhlites (Franz et al., 2014; Dottin et al., 2018) are an order of magnitude
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19 531 smaller than the effects observed in terrestrial Archean pyrites (Farquhar et al., 2000). During the
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21 532 photolysis of SO_2 , the magnitude of the S-MIF effect is dependent on the formation of at least
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23 533 two distinct species, an oxidized sulfate (or gaseous SO_3) and a reduced sulfur polymer (S_2 or S_8)
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25 534 between which the sulfur isotopes are distributed in a mass-independent manner (Pavlov and
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27 535 Kasting, 2002). The presence of oxidation on Mars' surface evidenced by oxychlorine
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29 536 compounds would oxidize the reduced species, so that practically all sulfur ends up as sulfate at
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31 537 the surface. This would prevent the formation of large MIF effects in surficial sulfate, contrary to
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33 538 initial expectations.

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35 539 As suggested by NWA 7533 pyrite, negative $\delta^{34}\text{S}$ values seem to be a characteristic
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37 540 feature of pyrite occurrences in SNC meteorites. Shearer et al. (1996) and Greenwood et al.
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39 541 (2000) documented a wide range of $\delta^{34}\text{S}$ ($-9.69 \pm 0.22\%$ to $7.8 \pm 0.7\%$), as well as anomalous
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41 542 $\Delta^{33}\text{S}$ signals ($-0.74 \pm 0.39\%$ and $-0.51 \pm 0.38\%$) in two pyrite grains located in "crushed zones"
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43 543 of ALH 84001 and associated with carbonates interpreted to have hydrothermal origin. Negative
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45 544 $\delta^{34}\text{S}$ values were also published for Lafayette pyrite resulting from partial replacement of
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47 545 pyrrhotite by sulfur-bearing fluid influx into the nakhlite unit (Greenwood et al., 2000; Fig. 7).
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49 546 The same is true for Chassigny pyrite that replaced pyrrhotite ($-4.6\% < \delta^{34}\text{S} < -1.5\%$; Greenwood
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51 547 et al., 2000). According to Shearer et al. (1996) and Greenwood et al. (2000), igneous pyrrhotite
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53 548 in ALH 84001 was largely converted to pyrite by a secondary alteration process, such as an
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55 549 impact-driven hydrothermal system. A low-temperature (200°C) hydrothermal influx of volatile
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57 550 components (S, C, H) was postulated for this reaction in Chassigny, too (Floran et al., 1978;
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59 551 Peslier et al., 2019; Lorand et al., 2018a and references therein). Using results from the Sample
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552 Analysis at Mars (SAM) facility aboard the Curiosity rover, Franz et al (2017) found large

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4 553 variations in $\delta^{34}\text{S}$ in ten sites of fluvio-lacustrine sediments filling the Gale Crater that exceed
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6 554 those measured for martian meteorites. They identified quite low $\delta^{34}\text{S}$ (-47 ± 14 ‰) at both
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8 555 Cumberland (CB) and Oudam (OU) sites where sulfides (pyrrhotite, pyrite) occur and
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10 556 enrichment in ^{34}S ($0 < \delta^{34}\text{S} < 28 \pm 7$ ‰ (CDT normalized)) in mudstone-hosted sulfates. They
11
12 557 concluded that sulfur isotopic equilibrium between sulfides and sulfates could have
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14 558 been reached through warmed groundwater within several thousand years at 100–
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16 559 150°C to yield $\delta^{34}\text{S}$ depleted sulfides.

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18 560 A similar process known as Thermochemical Sulfate Reduction (TSR) is assumed to
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20 561 have transferred S-MIF-bearing S into martian impact breccias. Thermochemical Sulfate
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22 562 Reduction is associated with kinetic fractionation that scales with temperature (Kiyosu and
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24 563 Krouse, 1990). Anhydrite-pyrite fractionation coefficients indicate a difference of 15 per mil at
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26 564 300°C (Shanks, 2014 and references therein) with more positive (heavier) compositions in
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28 565 sulfates compared to pyrite. Typical TSR S isotope fractionations are usually less than 20 ‰
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30 566 (Machel et al., 1995; Seal, 2006). A crude estimate for initial martian sulfate source enriched in
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32 567 ^{34}S is suggested by $\delta^{34}\text{S}$ as high as 4.7 ± 0.2 ‰ in sulfate extracted from Nakhla (Farquhar et al.,
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34 568 2007). Shergottite sulfates show $\delta^{34}\text{S}$ of 6.68 ± 0.15 ‰ (Farquhar et al., 2007). Higher values
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36 569 recorded by sulfate of Antarctic nakhrites MIL 03346 and its pairs (up to 12.39 ‰) may reflect
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38 570 input of terrestrial sulfates (Franz et al., 2014), but are not unrealistic for martian sulfates as
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40 571 suggested by analyses of sulfate in fluvio-lacustrine sediments of Gale crater (Franz et al., 2017).
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42 572 Hence, taking into account the extreme isotopic composition of martian sulfates, the isotopic
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44 573 fractionation relationship between NWA 7533 pyrite and putative martian sulfates ($\delta^{34}\text{S}_{\text{sulfate}} -$
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46 574 $\delta^{34}\text{S}_{\text{pyrite}}$) is about 10 ‰. Such $\delta^{34}\text{S}$ values require TSR temperatures in excess of 250°C using the
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48 575 experimentally derived equation of Kiyosu and Krouse (1990) (see also Machel et al. (1995)). To
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50 576 preserve negative $\delta^{34}\text{S}$ values, this system must have been open with respect to the supply of
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52 577 sulfate (sulfate supply > sulfate consumption), otherwise pyrite would develop positive $\delta^{34}\text{S}$
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54 578 values upon completion of closed system fractionation effects (e.g. Magnall et al., 2016). Such
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56 579 an open system is consistent with the widespread distribution of pyrite inside porosity of each
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58 580 lithological component of martian breccias, including late veinlets. The TSR process may have
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60 581 contributed along with other petrogenetic processes (magma degassing, weathering.....) to the

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3 582 highly oxidized state of NWA 7533 of \sim FMQ + 2 log units (Leroux et al., 2016; Hewins et al.,
4 583 2017), well above typical martian magmatic conditions. Reduction of one mole of S^{6+} to S^{2-}
5 584 consumes eight moles of electrons and has the potential to oxidize eight moles of Fe^{2+} to Fe^{3+}
6 585 (see detailed calculation in Dottin et al., 2018). Our scenario of precursor sulfates is also
7 586 supported by the S/Se ratio of NWA 7533 pyrite (3,500-54,000; Lorand et al. 2018b; Fig. 5).
8 587 Selenium chemistry is strongly tied to S chemistry: both elements can substitute as anions into
9 588 sulfides while forming oxycompounds hosted in sulfates (Howard, 1977; Huston et al., 1995;
10 589 Lorand et al., 2003; Ziemkiewicz et al., 2011 and references therein). Although Se concentration
11 590 data (and hence S/Se ratio) are not available for martian sulfates, one nakhlite (MIL 03346) that
12 591 bears multiple evidence of martian sulfate contamination shows a bulk-rock S/Se ratio of up to
13 592 13,385 whereas it is only 3566 ± 798 for the shergottites that have incorporated juvenile (mantle-
14 593 derived) martian sulfur (Wang and Becker, 2017). Owing to its S/Se ratio up to 54,000, there can
15 594 be little doubt that NWA 7533 pyrite formed from precursor sulfates (Fig. 6).

16 595 The Eh-pH redox conditions deduced for NWA 7533 from the Fe-S-O system, and the
17 596 lack of any firmly identified sulfate, imply that sulfate reduction products were recycled by
18 597 hydrothermal fluids in the martian breccia as reduced to near-neutral H_2S/HS^- -rich aqueous
19 598 fluids (Lorand et al., 2015). Reduction of sulfate during the hydrothermal process would
20 599 generate H_2S and potentially H_2Se that are likely to precipitate as pyrite in the presence of Fe^{2+}
21 600 (Auclair et al., 1987). Experimental studies on terrestrial hydrothermal vent systems demonstrate
22 601 that at 300 and 350 °C (a temperature range corresponding to that deduced from the Ni content
23 602 of NWA 7533 pyrite; Lorand et al., 2015, 2018b), pyrite is expected to be depleted in the heavy
24 603 isotopes of S relative to H_2S at equilibrium (Seal, 2006; Wu et al., 2010; Gregory et al., 2015;
25 604 Magnall et al., 2016). Lorand et al. (2015, 2018b) suggested that NWA 7533 pyrite precipitated
26 605 by direct nucleation or sulfidation reactions between H_2S/HS^- -rich aqueous fluids and Fe from
27 606 different sources (dissolved ions in the fluid, pyroxene clast, matrix oxides (magnetite,
28 607 maghemite)). Like pyrite, the negative $\Delta^{33}S$ coupled with negative $\delta^{34}S$ values makes the few
29 608 fine-grained spongy anhedral pyrrhotite grains identified in the interclast matrix (ICM) of NWA
30 609 7533 crystallization products from fluids that reduced crustal sulfates (Fig. 2). However, the
31 610 highly resorbed and denticulated shape of these pyrrhotite blebs and their spongy textures
32 611 compared to euhedral pyrite crystals are noteworthy. Such sulfide textures were reported in SNC
33 612 meteorites that experienced shock-induced sulfur degassing (Lorand et al., 2012, Gattaceca et al.,

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3 613 2013). Lorand et al. (2012) concluded that pyrite in the strongly shocked chassignite NWA 2737
4 614 was desulfurized into FeS due the high shock pressure (55 GPa) coupled with strong heating
5 615 conditions (perhaps up to 800 °C) published for this meteorite. In NWA 7533, the shock event
6 616 postdating pyrite crystallization generated only planar deformation features in pyrite. A
7 617 desulfurization of pyrite into spongy pyrrhotite is thus considered to be very unlikely. For
8 618 comparison, pyrite was preserved in Chassigny that experienced a shock pressure of much
9 619 greater intensity (about 35 GPa with a post-shock temperature of about 300 °C; Lorand et al.,
10 620 2018a). The NWA 7533 spongy pyrrhotite must be interpreted as a metastable hydrothermal
11 621 product of sulfate reduction reactions reflecting local, transient more reducing conditions during
12 622 percolation by hydrothermal fluids (see Lorand et al., 2015).
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23 624 **Meteorite impact vs. igneous intrusion as source of NWA 7533 hydrothermal pyrite?**

24 625 According to a detailed analysis of pyrite textures and its relationships with silicate
25 626 lithologies of NWA 7533, hydrothermal alteration took place at a very late stage of the history of
26 627 the meteorite during and after a lithification and annealing event that made the regolith breccia
27 628 very cohesive. This annealing event responsible for the lithification of the meteorite likely
28 629 occurred at ~1.4 Ga from the apatite U-Pb ages (Bellucci et al., 2015), the zircon lower intercept
29 630 ages and the K-Ar ages (Humayun et al., 2013; Lindsay et al., 2014; McCubbin et al., 2016;
30 631 Cassata et al., 2018) and the whole-rock $^{187}\text{Re}/^{187}\text{Os}$ model ages (Goderis et al., 2016). Taking
31 632 into account the Amazonian age of NWA 7533 pyrite, the water responsible for the hydrothermal
32 633 event on Mars was likely produced by melting of ice from a subsurface permafrost reservoir.
33 634 Chlorine was likely present too, as indicated by core-rim increase in Cl/F ratio of NWA 7533
34 635 apatite (Bellucci et al., 2016); however, Lorand et al. (2018b) suggested that the chlorine content
35 636 was significantly low when NWA 7533 pyrite precipitated because this sulfide shows very low
36 637 contents of the base metals (Cu, Zn...) that are preferentially transported by Cl-rich acidic fluids.
37 638 This hydrothermal cycling process bears similarities to that documented for nakhlites from the
38 639 Fe-carbonate + Ca-sulfates + halite veins: Changela and Bridges (2011) argued for an impact-
39 640 induced heating event which triggered upward percolation of H₂O-CO₂ fluids from buried ice
40 641 into the nakhlite igneous mass, although both upward and downward percolations were also
41 642 suggested (see Bridges et al., 2019 and reference therein). A downward percolation of sulfate-
42 643 bearing aqueous fluid is considered to be more likely for martian impact breccias; however the
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3 644 source of heat that triggered this percolation, either an impact or a magmatic event is still
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5 645 debated (Lindsay et al., submitted).

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7 646 Meteorite impacts on rocky planetary bodies have the potential to initiate transient
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9 647 hydrothermal systems if: (i) the target contains sufficient volatiles (e.g. liquid water or ice), and
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11 648 (ii) a substantial heat source is generated (e.g. melt sheet, nearby pluton, raised geothermal
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13 649 gradient) (Newsom, 1980; Abramov and Kring, 2004; Naumov, 2005; Osinski et al., 2013;
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15 650 Koeberl, 2014). Sulfides are common hydrothermal products in impact craters on Earth and
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17 651 mineralisation dominated by pyrite is thought to occur during the latest stages of cooling (Parnell
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19 652 et al., 2010; Simpson et al., 2005; 2017). Geochemical modeling of impact-generated
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21 653 hydrothermal systems for Mars also frequently return pyrite among hydrothermal products (e.g.
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23 654 Schwenzer and Kring, 2013). As shown by these authors, pyrite becomes a major mineral along
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25 655 with hematite for water/rock ratios exceeding 1,000. Such conditions are met inside veins, as
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27 656 observed in NWA 7533 where some cracks are filled with pyrite + maghemite (Lorand et al.,
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29 657 2015). Cassata et al. (2018) proposed for NWA 7034 (paired with NWA 7533) a rather different
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31 658 interpretation linking the lithification and annealing event with a thermal dome above deeper
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33 659 intrusives because thermal metamorphism recorded by the K-Ar and U-Pb systems yield plateau
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35 660 age differences that were greater than durations (10's of Ma or less) required to cool shock-
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37 661 heated basement terrains following large (100 to 200 km) impact events that might have
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39 662 occurred at ~1,300 Ma. Magmatism is known to have occurred on Mars at 1.4-1.35 Ga from the
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41 663 nakhlite and chassignite meteorites (Nyquist et al., 2001; Cohen et al., 2017 and references
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43 664 therein), while the impact cratering rate of the martian surface was already considerably reduced
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45 665 at that time (e.g., Moser et al., 2019). However, MacArthur et al. (2019) related the maximum
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47 666 ^{40}Ar - ^{39}Ar age of 1130 – 1250 Ma in the NWA 8114 meteorite clast (paired with NWA 7533)
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49 667 analysed by them to an impact shock event and the subsequent high-temperature oxidative
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51 668 breakdown of many of the pyroxenes (also documented in NWA 7533 by Leroux et al., 2016). A
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53 669 simple Fourier cooling model suggested that a burial depth of 5 m was enough to maintain
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55 670 sufficiently high temperatures (>600°C) for ~30 days (MacArthur et al., 2019). With this
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57 671 constraint in hand, one may assume that sulfates in martian breccias may have been reduced to
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59 672 sulfides almost in-situ, implying very little transport of aqueous fluids.
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673 674 CONCLUSIONS

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5 676 Pyrite is by far the major host of sulfur in the polymict NWA 7533 breccia and paired
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7 677 meteorites. Apatite and Fe oxyhydroxides are negligible S carriers, as are the tiny magmatic
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9 678 sulfide grains so far identified. Sulfates (barite or gypsum) have not been firmly identified in the
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11 679 breccia. The initial sulfur content of the breccia inferred from the modal abundance of pyrite and
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13 680 its pseudomorphous Fe-oxyhydroxides is estimated to be ~5,400 ppm. The two bulk-rock S
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15 681 analyses obtained here confirm previous estimates of 80% S loss resulting from terrestrial
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17 682 weathering of NWA 7533 pyrite.

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19 684 The occurrence of S-MIF in the S multi-isotope composition supports a model of NWA
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21 685 7533 pyrite formation from martian surficial sulfur that experienced photochemical reaction(s).
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23 686 The slight discrepancy between *in-situ* SIMS analyses ($\Delta^{33}\text{S} = -0.2 \pm 0.1 \text{ ‰}$; mean of 29
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25 687 analyses) and the bulk IRMS analyses ($\Delta^{33}\text{S} = -0.029 \pm 0.010 \text{ ‰}$) is tentatively interpreted as a
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27 688 likely sampling bias since only grains $> 20 \text{ }\mu\text{m}$ in size were analyzed.

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29 689 Negative $\delta^{34}\text{S}$ values were previously reported for several occurrences of secondary
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31 690 pyrite in SNC meteorites. Such negative values in the hydrothermal setting of NWA 7533 are
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33 691 reflective of authigenic sulfide precipitated from H_2S produced via open-system reduction of
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35 692 sulfates. A transient more reducing stage likely produced the spongy pyrrhotite that carries
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37 693 similar S multi-isotope compositions as that of pyrite.

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39 694 It is concluded that sulfur was recycled by downward percolation of reduced S-bearing
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41 695 aqueous fluid produced by melting of permafrost. However, the driving force that recycled
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43 696 crustal S in NWA 7533 lithologies, underlying magmatic intrusions or impact-induced heating,
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45 697 is presently unclear.

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3 **1013 Figure caption**

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5 1015 Figure 1 : Backscattered electron (BSE) images of NWA 7533 Fe sulfides. (A) euhedral
6 1016 pyrite (1) affected by late fracturing that also guided terrestrial weathering (2) dark gray
7 1017 Fe oxyhydroxides); (B) poikilitic pyrite surrounded by a fine-grained dust of ICM
8 1018 minerals; solid inclusions are skeletal Fe oxides (maghemite-magnetite; 3) and perfectly
9 1019 euhedral apatite (4); (C) interstitial porous pyrrhotite (5) within interclast matrix (ICM).
10 1020 Note silicate clast inclusions (I) in the pyrrhotite.
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18 1021 Figure 2 : $\delta^{34}\text{S}$ vs $\Delta^{33}\text{S}$ diagram for NWA 7533 sulfides and bulk-rock. $\Delta^{33}\text{S} = \delta^{33}\text{S} - 1000 \times$
19 1022 $[(\delta^{34}\text{S}/1000 + 1)0.515 - 1]$. Compositional field of shergottites and weighted mean used as
20 1023 revised estimate for the juvenile Martian sulfur composition after Franz et al. (2019),
21 1024 excluding the samples that were described as contaminated by crustal materials. Note the overall
22 1025 negative $\delta^{34}\text{S}$ values of NWA 7533 sulfides for slightly negative $\Delta^{33}\text{S}$ values compared to
23 1026 shergottites. Typical error bars on shergottite data display 2σ uncertainties.
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32 1027
33 1028 Figure 3 : $\delta^{34}\text{S}$ vs. Ni and $\Delta^{33}\text{S}$ vs. Ni diagrams for NWA 7533 pyrite (NWA7533-5) (SEM
34 1029 analyses for Ni).
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37 1030
38 1031 Figure 4 : $\Delta^{33}\text{S}$ vs. $\Delta^{36}\text{S}$ diagrams for NWA 7533 sulfides. $\Delta^{36}\text{S} = \delta^{36}\text{S} - 1000 \times$
39 1032 $[(\delta^{34}\text{S}/1000 + 1)1.9 - 1]$. Data for SNC meteorites from Franz et al., (2014, 2019).
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42 1033
43 1034 Figure 5. Time-resolved LA-ICPMS spectra collected along the laser traverse through a Fe
44 1035 oxyhydroxide pseudomorph replacing pyrite.
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47 1036
48 1037 Figure 6 : Se vs. S plots for NWA 7533. Color symbols correspond to the rastered LA-ICPMS
49 1038 analyses of the different lithological components of NWA 7533 (Humayun et al., 2013). NWA
50 1039 pyrite and Fe oxyhydroxide compositions from Lorand et al (2018b). The blue arrow depicts the
51 1040 preferential leaching of S with respect to Se resulting from terrestrial weathering. Putative
52 1041 composition of martian sulfates after Wang and Becker (2017).
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1043 Figure 7. Comparison of NWA 7533 pyrite compositions with SNC meteorites in a $\delta^{34}\text{S}$ vs $\Delta^{33}\text{S}$
1044 diagram. A = ALH 84001; L = Lafayette; N = Nakhla; Y = Y000593 (nakhlite); M = Miller
1045 Range nakhlites; NWA 998 and NWA 6148 : nakhlites; WSS : water-soluble sulfates; AVS :
1046 acid-volatile sulfides; CRS : Chromium-reducible sulfides; Py : pyrite. Other data : bulk-rock
1047 analyses. Note the overall negative $\delta^{34}\text{S}$ values of NWA 7533 sulfides for slightly negative $\Delta^{33}\text{S}$
1048 values compared to nakhlite sulfates. Arrow materializes possible thermochemical reduction of
1049 sulfates into NWA 7533 hydrothermal sulfides.

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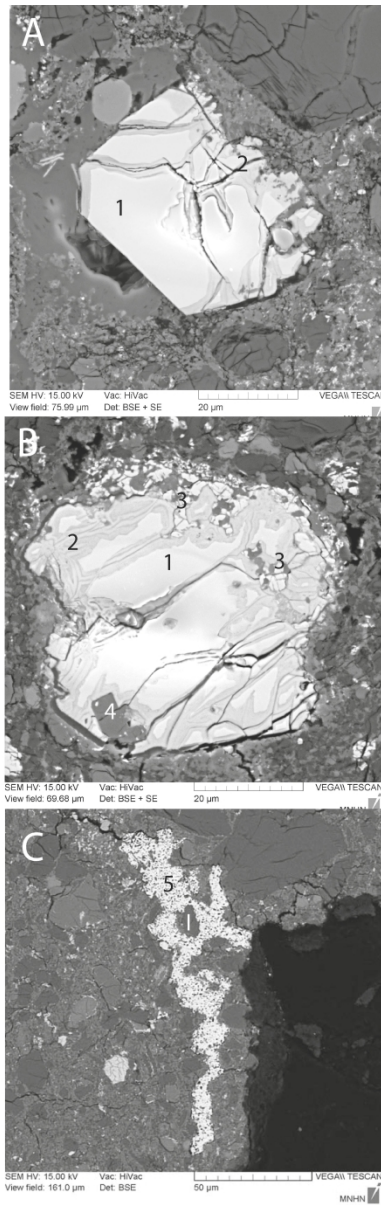


Fig. 1

210x297mm (600 x 600 DPI)

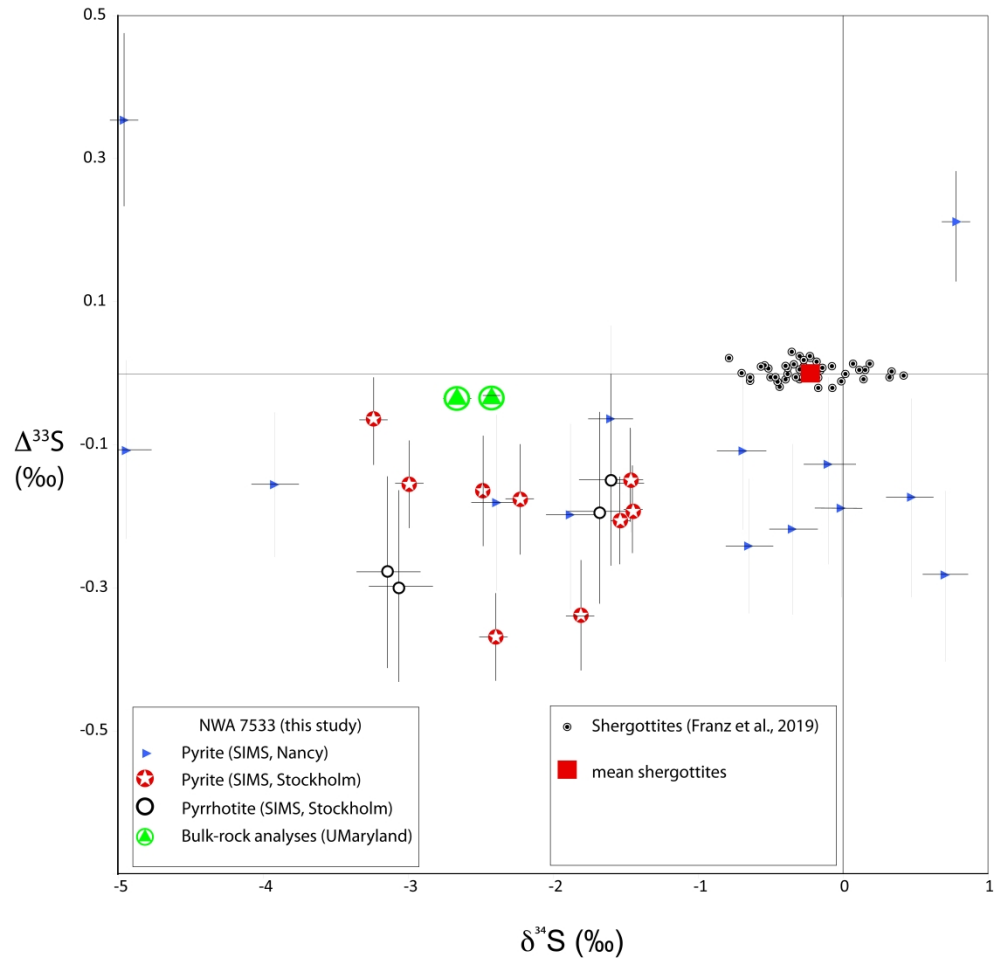


Fig. 2

698x855mm (600 x 600 DPI)

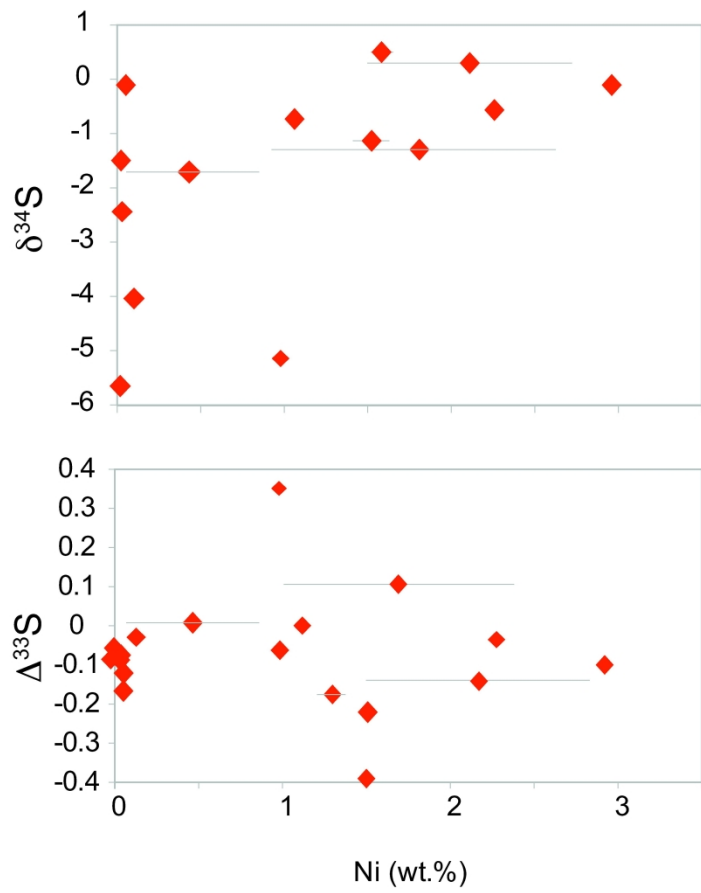


Fig. 3

228x231mm (600 x 600 DPI)

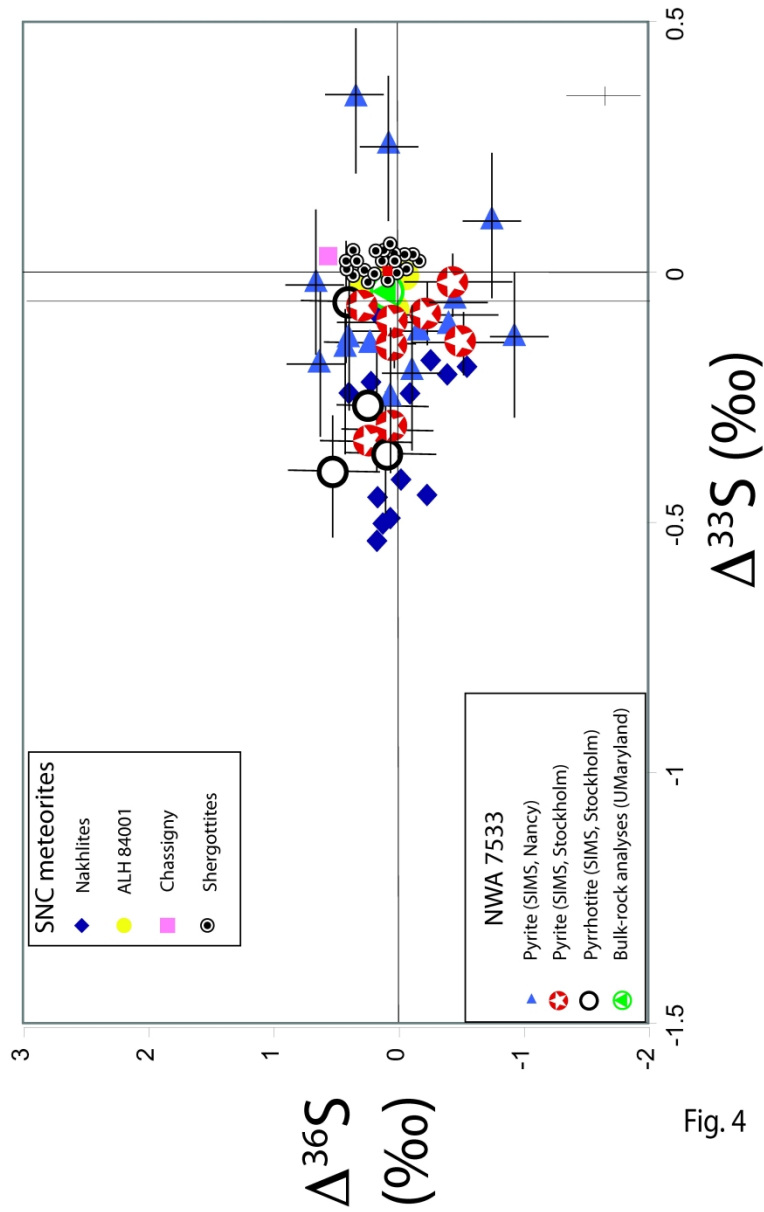


Fig. 4

188x295mm (600 x 600 DPI)

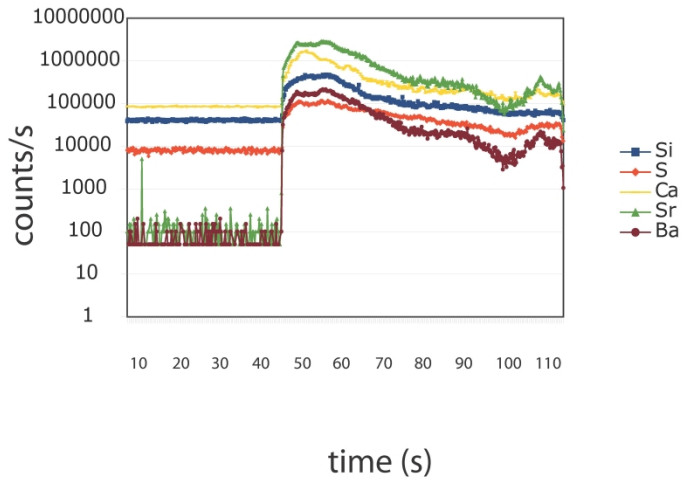


Fig. 6

210x297mm (600 x 600 DPI)

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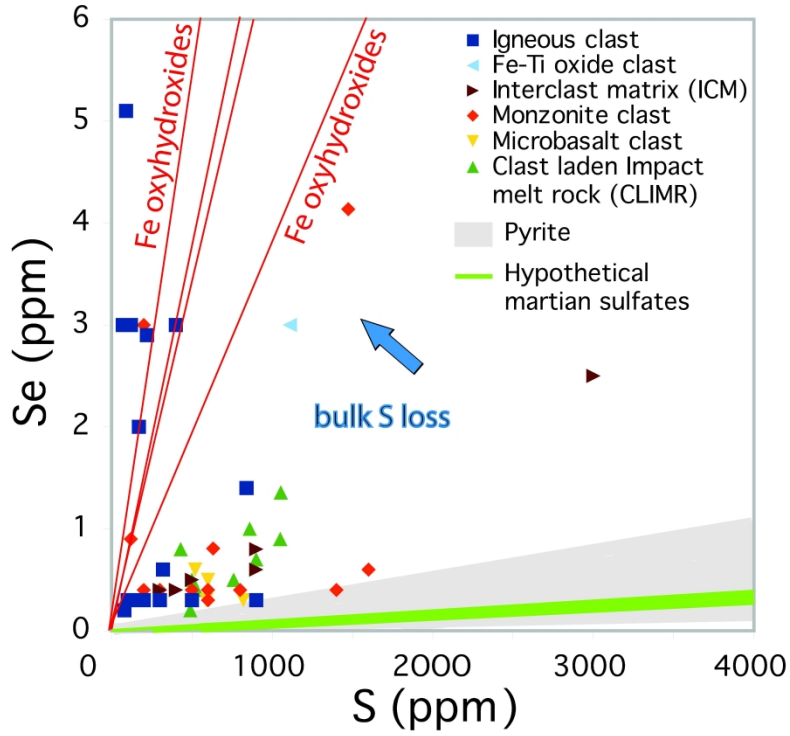


Fig. 5

233x252mm (600 x 600 DPI)

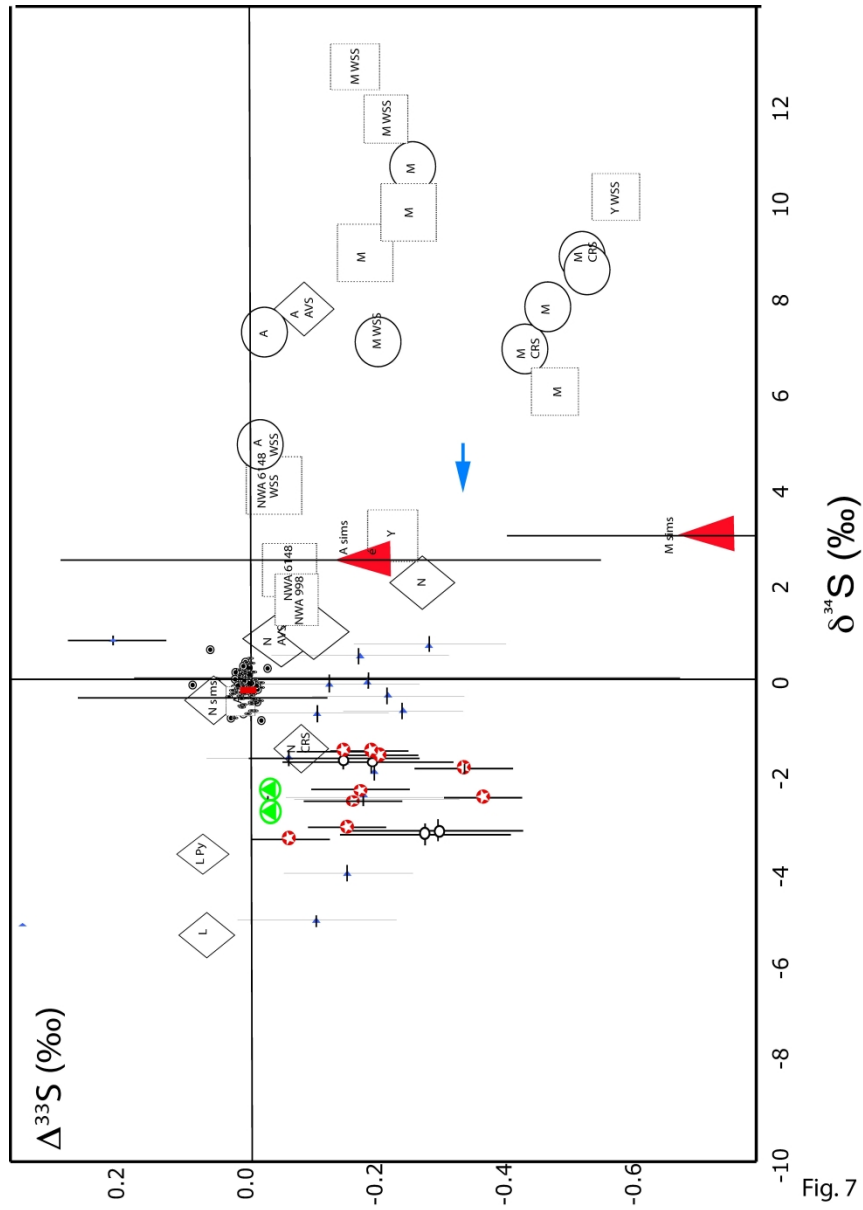


Fig. 7

211x296mm (600 x 600 DPI)

Table 1. S isotopic compositions of sulfides in NWA 7533 (Canon Diablo-normalized)

	S (ppm)	$\delta^{34}\text{S}$ (‰)	2σ	$\Delta^{33}\text{S}$ (‰)	2σ	$\Delta^{36}\text{S}$ (‰)	2σ
Bulk-rock (U Maryland)							
BB-UMD-1	920	-2.39	0.013	-0.026	0.012	0.002	0.105
BB-UMD-2	805	-2.673	0.027	-0.031	0.01	0.001	0.076
average	863	-2.532	0.015	-0.029	0.010	0.002	0.09
NWA 7533-4 (NORDSIMS)							
Site 8a Pyrite		-1.82	0.23	-0.34	0.17	0.08	0.53
Site 8b Pyrite		-2.5	0.23	-0.36	0.18	0.12	0.66
Site 5b Pyrite		-1.54	0.18	-0.21	0.12	0.03	0.44
Site 15 Pyrite		-2.5	0.17	-0.12	0.14	0.49	0.51
Site 16a Pyrite		-1.63	0.17	-0.21	0.14	-0.55	0.53
Site 16b Pyrite		-3.25	0.17	-0.05	0.14	-0.43	0.49
Site 17 Pyrite		-2.99	0.18	-0.16	0.13	-0.004	0.45
Site b1a Pyrite		-2.22	0.19	-0.16	0.16	-0.083	0.54
Site b1b Pyrite		-1.5	0.17	-0.13	0.14	-0.39	0.53
Site 9a Pyrrhotite		-1.65	0.38	-0.19	0.28	0.12	0.81
Site 9b Pyrrhotite		-1.61	0.41	-0.14	0.28	0.39	0.71
Site 11 Pyrrhotite		-3.16	0.4	-0.26	0.3	0.05	0.6
Site 12 Pyrrhotite		-3.11	0.32	-0.28	0.23	0.49	0.67
average		-2.22		-0.186		0.01	
2SD		1.38		0.18		0.66	
wtd. Av. (95% conf.)		0.41		0.044		0.20	
MSWD		42.		1.11		1.4	
NWA 7533-5 (Nancy).							
Site A Pyrite		-1.71	0.32	-0.01	0.18	0.59	0.48
Site 8 Pyrite		0.6	0.32	-0.22	0.18	0.03	0.48
Site -1a Pyrite		-0.44	0.32	-0.17	0.18	0.57	0.48
Site -1b Pyrite		-2.49	0.32	-0.12	0.18	0.3	0.48
Site -17 Pyrite		-0.11	0.32	-0.12	0.18	0.15	0.48
Site -19 Pyrite		0.39	0.32	-0.13	0.18	0.37	0.48
Site -12 Pyrite		-0.79	0.32	-0.04	0.18	-0.49	0.48
Site -C Pyrite		-4.02	0.32	-0.1	0.18	-0.21	0.48
Site -6-1 Pyrite		-1.33	0.32	0.11	0.18	-0.74	0.48
Site 6-2 Pyrite		-0.75	0.32	-0.18	0.18	-0.15	0.48
Site -B1 Pyrite		-1.21	0.32	-0.38	0.18	-2.27	0.48
Site G Pyrite		-5.08	0.32	-0.08	0.18	-2.22	0.48
Site -F-1 Pyrite		-1.57	0.32	-0.09	0.18	-0.4	0.48

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3	Site -F-2 Pyrite	-0.19	0.32	-0.11	0.18	-0.92 0.48
4	Site F-4 Pyrite	-5.58	0.32	0.35	0.18	0.26 0.48
5	Site 16-1 Pyrite	-2.01	0.32	0.24	0.18	0.02 0.48
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7	average*	-1.65		-0.12		0.32
8	2SD	3.69		1.5		1.74
9	wtd. av. (95% conf.)	0.98		0.064		0.46
10	MSWD	132		1.5		13.0
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13 * : exclude the positive values of $\Delta 33S$ in Site F-4 Pyrite and Site 16-1 Pyrite

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15 2SD is the simple 2x standard deviation on the values (measures variation in a data set regard
16 wtd.av. is the weighted average uncertainty on the data along its MSWD value (measures the
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Table 2. Laser Ablation inductively-coupled mass spectrometry (LA-ICPMS) analyses of pyrite alteration products.

Fe oxyhydroxides (pyrite alteration products)

	1	2	3	4	5	6	EMPA (n=15)*
	LA- ICPMS	LA- ICPMS	LA- ICPMS	LA- ICPMS	LA- ICPMS	LA- ICPMS	
S (ppm)	2392	2059	3061	4128	3728	2299	2400 +/-130
Ca	4455.5	13158.6	10681.2	5349.9	11609.4	8009.6	9500 +/-201
Sr	79.5	75.6	372.3	386.2	411.3	283.4	
Ba	227.1	91.1	198.2	283.6	271.6	233.6	
detection limits							
S (ppm)	2.83	3.09	4.48	4.14	7.76	3.77	
Ca	3.68	4.17	6.20	5.82	11.20	5.81	
Sr	0.001	0.001	0.002	0.002	0.004	0.002	
Ba	0.009	0.009	0.014	0.013	0.027	0.014	
	0.003	0.003	0.005	0.005	0.010	0.004	
Apatite							
	1	4	6	7	8	9	10
	LA- ICPMS	LA- ICPMS	LA- ICPMS	LA- ICPMS	LA- ICPMS	LA- ICPMS	LA- ICPMS
S (ppm)	229.9	230.2	209.3	217.2	229.6	231.1	267.6
Ca	IS	IS	IS	IS	IS	IS	IS
Sr	137.3	157.6	145.6	112.7	179.9	197.7	143.9
Ba	6.73	8.98	8.05	5.09	7.28	12.4	5.13
detection limits							
S (ppm)	2.01	2.81	2.40	3.37	3.10	3.42	3.87
Sr	0.001	0.001	0.001	0.001	0.009	0.001	0.001
Ba	0.006	0.009	0.007	0.009	0.008	0.009	0.011

*Electron Microprobe analyses-EMPA (Lorand et al., 2015) ; IS = Internal Standard.