



HAL
open science

The Strong Confinement Regime in HgTe Two-Dimensional Nanoplatelets

Nicolas Moghaddam, Charlie Greboval, Junling Qu, Audrey Chu, Prachi Rastogi, Clément J Livache, Adrien Khalili, Xiang Zhen Xu, Benoit Baptiste, Stefan Klotz, et al.

► **To cite this version:**

Nicolas Moghaddam, Charlie Greboval, Junling Qu, Audrey Chu, Prachi Rastogi, et al.. The Strong Confinement Regime in HgTe Two-Dimensional Nanoplatelets. *Journal of Physical Chemistry C*, 2020, 10.1021/acs.jpcc.0c07533 . hal-02948654

HAL Id: hal-02948654

<https://hal.science/hal-02948654>

Submitted on 24 Sep 2020

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

The Strong Confinement Regime in HgTe Two-Dimensional Nanoplatelets

Nicolas Moghaddam¹, Charlie Gréboval², Junling Qu², Audrey Chu², Prachi Rastogi², Clément Livache², Adrien Khalili¹, Xiang Zhen Xu¹, Benoit Baptiste³, Stefan Klotz³, Guy Fishman⁴, Francesco Capitani⁵, Sandrine Ithurria¹, Sébastien Sauvage^{6*}, Emmanuel Lhuillier^{2*}

¹Laboratoire de Physique et d'Etude des Matériaux, ESPCI-ParisTech, PSL Research University, Sorbonne Université UPMC Univ Paris 06, CNRS, 10 rue Vauquelin 75005 Paris, France.

²Sorbonne Université, CNRS, Institut des NanoSciences de Paris, INSP, F-75005 Paris, France.

³Sorbonne Université, CNRS, Institut de Minéralogie, de Physique des Matériaux et de Cosmochimie, IMPMC, F-75005 Paris, France.

⁴Kadope, Centre de Nanosciences et de Nanotechnologies, 10 boulevard Thomas Gobert, 91120, Palaiseau, France.

⁵Synchrotron-SOLEIL, Saint-Aubin, BP48, F91192 Gif sur Yvette Cedex, France.

⁶Université Paris-Saclay, CNRS, Centre de Nanosciences et de Nanotechnologies, 91120, Palaiseau, France.

Abstract: The confinement in colloidal HgTe nanocrystals enables this material to be promising for colloidal optoelectronics over a wide range of energies, from the THz spectral range up to the visible region. Herein, by using a combination of high energy absorption HgTe nanoplatelets and low energy absorption HgTe nanocrystals, we probe optical transmission of HgTe nanoparticles over the 0.26–1.8 eV range, from 0 K to 300 K temperatures and under simultaneous pressure, up to 4 GPa. While the pressure dependence of nanoplatelets follows the one observed for bulk and nanocrystals, the temperature dependence dramatically differs for nanoplatelets. The modeling of the electronic energy dispersion using up to 14-band **k.p** formalism suggests that the second conduction band and higher bands of HgTe play a vital role to describe and explain the HgTe nanoparticle spectroscopies.

*To whom correspondence should be sent: el@insp.upmc.fr and sebastien.sauvage@c2n.upsaclay.fr

reduce the device dark current. Alternatively, the development of HgTe based colloidal heterostructures appears promising to expand the reachable range of wavelength of the 2D NPLs,^{30,31} but also raises questions relative to the understanding of the optical spectrum to disentangle charge delocalization effect from pressure effect resulting from non-lattice matched shells.³²

In this paper, we explore in a systematic way the effect of confinement, temperature and pressure on the energy dispersion of HgTe. In practice, we grow a series of HgTe NCs with various confinement energies from 0.26 to 1.8 eV. The temperature is tuned from 0 K to 300 K which corresponds to cryogenic operating condition for IR sensors and finally the pressure is tuned up to 4 GPa, corresponding to the full range of existence of the zinc blende low pressure phase of HgTe nanoparticles. While the pressure dependence of the spectroscopic properties appears to be weakly affected by the confinement, we show that the temperature dependence of the 2D NPLs is opposite to the one observed for bulk and weakly confined NCs. Our multiband **k.p** modelling points to the central role of the second conduction band in the origin of this observation. Finally, thanks to a 14 bands **k.p** model we are able to propose a set of parameters which correctly account for temperature and pressure dependence and describe the dispersion relation of HgTe over an extended range of the Brillouin zone.

METHODS

Chemicals: Mercury chloride (HgCl₂, Strem Chemicals, 99%), Hg(Acetate)₂ (Hg(Ac)₂ Aldrich, > 99.0 %), **Mercury compounds are highly toxic. Handle them with special care.** Tellurium powder (Te, Sigma-Aldrich, 99.99%), cadmium oxide (CdO) (Aldrich 99.99 %), trioctylphosphine (TOP) (Alfa, 90 %), oleic acid (OA, Sigma, 90%), propionic acid (Aldrich, 99 %), oleylamine (OLA, Acros, 80-90%), dodecanethiol (DDT, Sigma-Aldrich, 98%), Trioctylamine (TOA, Aldrich), Octadecene (ODE) (Aldrich, 90 %), chloroform (Carlo Erba), methanol (VWR, 98.5%) acetone (VWR rechapur), ethanol (absolute VWR), toluene (VWR, 99.8 %). All chemicals are used without further purification.

Cadmium propanoate (Cd(Prop)₂): 1.036 g of CdO is mixed in 10 mL of propionic acid under Ar for 1 hour. Then the flask is open to air and the temperature risen to 140 °C up to the point the volume gets divided by a factor of two. The whitish solution is precipitated by addition of acetone. After centrifugation, the solid is dried under vacuum for 24 hours.

1 M TOP:Te precursor: 2.54 g of Te powder is mixed in 20 mL of TOP in a three neck flask. The flask is kept under vacuum at room temperature for 5 min and then the temperature is raised to 100 °C. Furthermore, degassing of flask is conducted for the next 20 min. The atmosphere is switched to Ar and the temperature is raised to 275 °C. The solution is stirred until a clear orange coloration is obtained. The flask is cooled down to room temperature and the color switches to yellow. Finally, this solution is transferred to an Ar filled glove box for storage.

Synthesis of 3 MLs CdTe NPLs: The procedure follows the procedure proposed by Pedetti *et al.*³³ In a three-neck flask, Cd(prop)₂ (390 mg, 1.5 mmol), oleic acid (240 μL, 0,75 mmol), and 30 mL of octadecene are degassed under vacuum at 90 °C for 1 hour. Under argon flow, TOP:Te at 1 M (300 μL) diluted in 1.5 mL of octadecene is swiftly injected at 210 °C. After 30 minutes of reaction, the mixture is cooled down and 1.7 mL of oleic acid is introduced. The resulting nanoplatelets are precipitated twice with an excess of ethanol and then are redispersed in 30 mL of hexane.

Cation exchange for synthesis of 3 MLs HgTe NPLs with TOA: In a glass tube, 2 mL of hexane and 200 μL of HgTe NPLs are introduced. 60 μL of a fresh 0.1 M HgAc₂ in trioctylamine solution is added at room temperature (The Cd:Hg ratio is 1:2). The cation exchange is followed by absorption spectroscopy. Once the first excitonic peak has reached 824 nm, 80 μL of oleic acid are added to stabilize the HgTe NPLs. The excess of precursor is removed by a two-step centrifugation (6000 rpm, 30 s and we keep the supernatant then 6000 rpm, 5 min). The resulting NPLs are resuspended in 1 mL of toluene. Procedures for other surface chemistry on the HgTe NPL surface are given in the SI

HgTe NCs synthesis with band edge at 4000 cm⁻¹: 513 mg of HgCl₂ was added to 60 mL of oleylamine in a 100 mL round flask. The solution was placed under vacuum and heated to 110 °C for 1 h. Then, the temperature is decreased to 80 °C and solution is placed to Ar atmosphere. 1.9 mL of TOP:Te (1 M) with 10 mL of oleylamine is added to the mercury solution. The solution color gradually turns to dark brown and the reaction is stopped at 3 min. A solution made of 1 mL of dodecanethiol and 9mL of toluene is quickly added to quench the reaction. The nanocrystals are then precipitated with ethanol. After centrifugation, the nanocrystals are redispersed in chloroform. The washing step is repeated one more time. The solution is filtered with a 0.2 µm filter and redispersed in 6mL of chloroform. Procedures for other NC sizes are given in the SI.

Electron microscopy: For transmission electron microscopy (TEM) pictures, a drop of the NC solution is drop-casted onto a copper grid covered with an amorphous carbon film. The grid is degassed overnight to reduce future contamination. A JEOL 2010F is used for acquisition of pictures and operated at 200 kV.

Infrared spectroscopy as a function of temperature: The solution of nanocrystal is drop-casted onto a double side polished Si wafer for the sample in the mid-infrared and onto a CaF₂ substrate for the sample in the near-infrared. The sample is mounted on the cold finger of a closed cycle cryostat. The latter is used in a transmission configuration with two ZnSe windows. The cryostat is then introduced into the Fourier Transform Infrared spectrometer (FTIR, Thermo Fischer iS 50). Two sources have been used: a white light in the near infrared and a globar in the mid infrared. Two beam splitters have been used: a CaF₂ beam splitter in the near infrared and an extended KBr in the mid infrared. Two detectors have been used: an InGaAs sensor in the near infrared and a DTGS in the mid infrared. In all case, the background is made at room temperature using the same optical configuration but without sample onto the substrate. Spectra are acquired with a 4 cm⁻¹ resolution and typically averaged 100 times.

X-ray diffraction under pressure: Experiments were carried out using a membrane diamond anvil cell equipped with Böhler-type anvils with 500 micrometer culets. A stainless-steel gasket of 200 micrometer thickness was pre-indented to 70 µm, provided with a 200 µm hole, and loaded with the sample, a 4:1 methanol-ethanol pressure transmitting medium and a 10 µm diameter ruby sphere which served as pressure marker. The "Mao hydrostatic" pressure scale was used to determine pressures. X-ray powder diffraction measurements were performed at the XRD platform of the IMPMC on a Rigaku MM007HF diffractometer equipped with a Mo rotating anode ($\lambda_{K\alpha 1} = 0.709319 \text{ \AA}$, $\lambda_{K\alpha 2} = 0.713609 \text{ \AA}$), Varimax focusing optics, and a RAXIS4++ image plate detector. X-ray data were collected at 20 °C. A LaB₆ standard sample was measured in the same experimental conditions to calibrate the Fit2D program, the image processing software used to integrate the intensities around the Debye-Scherrer rings and to get the 1D patterns.

Infrared spectroscopy under pressure: Membrane diamond anvil cells (DACs), equipped with IIa-type diamonds having a 400-500 µm culets, are used. Stainless steel gaskets with a thickness of 200 µm are indented down to 50 µm. A 150 µm hole is then drilled by electro-erosion. The gasket is placed on the top of one of the DAC diamonds, and the gasket hole filled with NaCl. After introducing a ruby crystal, the DAC is closed under gentle pressure until the salt forms a clear window. A drop of a diluted suspension of NCs in toluene is then added and dried on top of the NaCl window to form a dry and uniform film of NCs. The DAC is then closed and introduced in the Cassegrain microscope. the pressure is generated by the means of a metallic membrane inflated with He gas. The *in-situ* pressure is monitored by the ruby luminescence technique. A 532 nm laser is shone through one of the Cassegrain objectives and focused on the ruby crystal embedded in the cell. Ruby photoluminescence light is collected and sent to an OceanOptics spectrometer, showing a characteristic doublet around 694 nm at ambient pressure. This photoluminescence is fitted and the pressure is computed using the main peak position. Once the pressure is stabilized, an infrared absorbance spectrum is acquired using a Thermo Fisher Nicolet iS50 FTIR with a globar source, a quartz beam splitter, and a silicon detector in transmission configuration. This operation is repeated for various pressure from 0 GPa to 4 GPa. When the maximum pressure is reached, the membrane is deflated and several spectra are acquired to check the reversibility of the pressure effect.

k.p modeling and simulations: 3x3 and 14x14 **k.p** $H(k, T, P)$ hamiltonian matrices are processed symbolically and diagonalized numerically for each couple of temperature and pressure (T, P) using the parameters in Table 1. Bulk material energy dispersion diagrams $E(k)$ are calculated using the wave vector k as a free-varying independence variable. On the contrary, spectroscopic energies of the form $E(T, P)$ are extracted by modifying k , with the following procedure. Since thermal dilation or pressure contraction occurs, the nanoparticles are implicitly labelled by their discretized wave vector $k_0 = 2\pi/L_0$ in (0 K, 0 GPa) conditions. Based on the starting energy E_{start} in the starting conditions (T_{start}, P_{start}) , the dispersion relation $E(k)$ is first inverted giving out a reference wave vector k_{start} . This k_{start} wavevector is then dilated or contracted back to (0 K, 0 GPa) conditions following equation S6b, thus giving the k_0 wavevector labelling the nanoparticle. The temperature T and pressure P are then freely scanned. For each (T, P) , the norm of the discretized wave vector k_0 is changed into $k(k_0, T, P)$ according to the dilation or contraction of the lattice parameter. The **k.p** matrix is finally numerically evaluated at this altered wave vector k and the eigen energies are reported against T and/or P .

RESULTS AND DISCUSSION

From weak to strong confinement in HgTe Nanocrystals

We grow a series of HgTe NCs with various sizes to widely tune the confinement energy. As a strongly confined form of HgTe, 2D NPLs are grown.¹⁵ The HgTe NPLs are obtained using a two-step approach where CdTe NPLs are first synthesized,³⁴ see Figure S1. The NPLs are 3.5 monolayers thick (3 planes of Te surrounded by 4 planes of Cd) corresponding to a thickness of 1.1 nm.³⁴ Those NPLs are then exposed to a bulky Hg precursor (Hg oleate), which transforms them into pure HgTe NPLs due to cation exchange.¹⁵ Given that CdTe (0.648 nm) and HgTe (0.646 nm³⁵) have similar lattice parameter (0.3 % difference) the final thickness of the HgTe NPL is 1.1 nm¹⁵ which is confirmed by the fact that the two types of material present similar X-ray diffractogram, see Figure S2.

The lateral extension of the particles is large and can reach several hundred of nm according to electronic microscopy, see **Figure 2a** and S3-S5. Due to the thickness, the material is strongly confined along $\langle 001 \rangle$ crystal direction and the band edge absorption appears around 900 nm ($\approx 11000 \text{ cm}^{-1}$ or $\approx 1.4 \text{ eV}$ at room temperature), see **Figure 2c**. It has been demonstrated that the exact value of the band-edge energy can be further tuned with the capping ligands which also take part in the effective delocalization length.¹⁵ This enables a fine tuning of the confinement energy while keeping the inorganic core unchanged.

In addition, we grow a series of HgTe NCs with room temperature band edge between 9000 cm^{-1} (1.11 eV) and 2000 cm^{-1} (248 meV), see **Figure 2b-c** and S6. The syntheses follow the procedure of Keuleyan *et al.*³⁶ (for large NCs) and Geiregat *et al.*³⁷ (for the smallest NCs). The combination of the HgTe NPLs and NCs allows us to tune the confinement energies over more than a factor of 5, from strongly to weakly confined materials as we switch from NPLs to NCs.

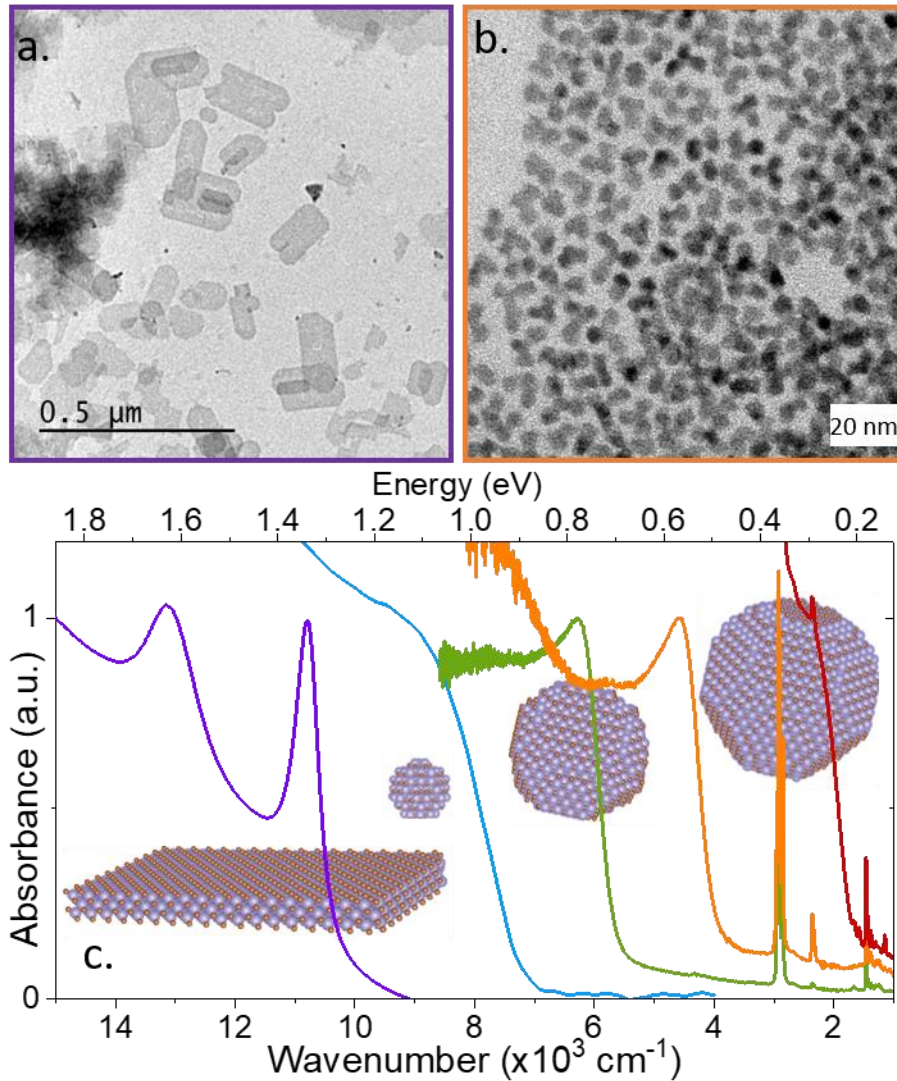


Figure 2 Confined NCs of HgTe. TEM images of a. HgTe NPLs and b. NCs. The associated absorption spectrum appears in purple and orange, respectively. c. Infrared absorption spectra for HgTe NPLs and NCs with various sizes.

Effect of temperature of the infrared absorption

We then probe the effect of temperature on the IR spectrum of the material, see the spectra in **Figure 3a-c** and S8-S9. The temperature dependence on the material is dramatically influenced by the material confinement, as reported in **Figure 3a-c**. For large (*ie* weakly confined) HgTe NCs (**Figure 3c**), the E_G confinement energy redshifts as the temperature is lowered, corresponding to a positive value for dE_G/dT . This behavior is similar to the one observed in bulk HgTe,^{38,39} but is opposite to the trend observed for wider band gap II-VI semiconductors such as CdTe⁴⁰ or CdSe^{41,42} where whatever the particle size only negative value have been observed for dE_G/dT . In NCs, such positive value for dE_G/dT has been observed for lead halide perovskites⁴³ or large PbS NCs.⁴⁴ Nevertheless, as the confinement is increased, we observe a reduction of the value of dE_G/dT (**Figure 3b**) and surprisingly even an inversion of its sign in the case of NPLs (**Figure 3a**).

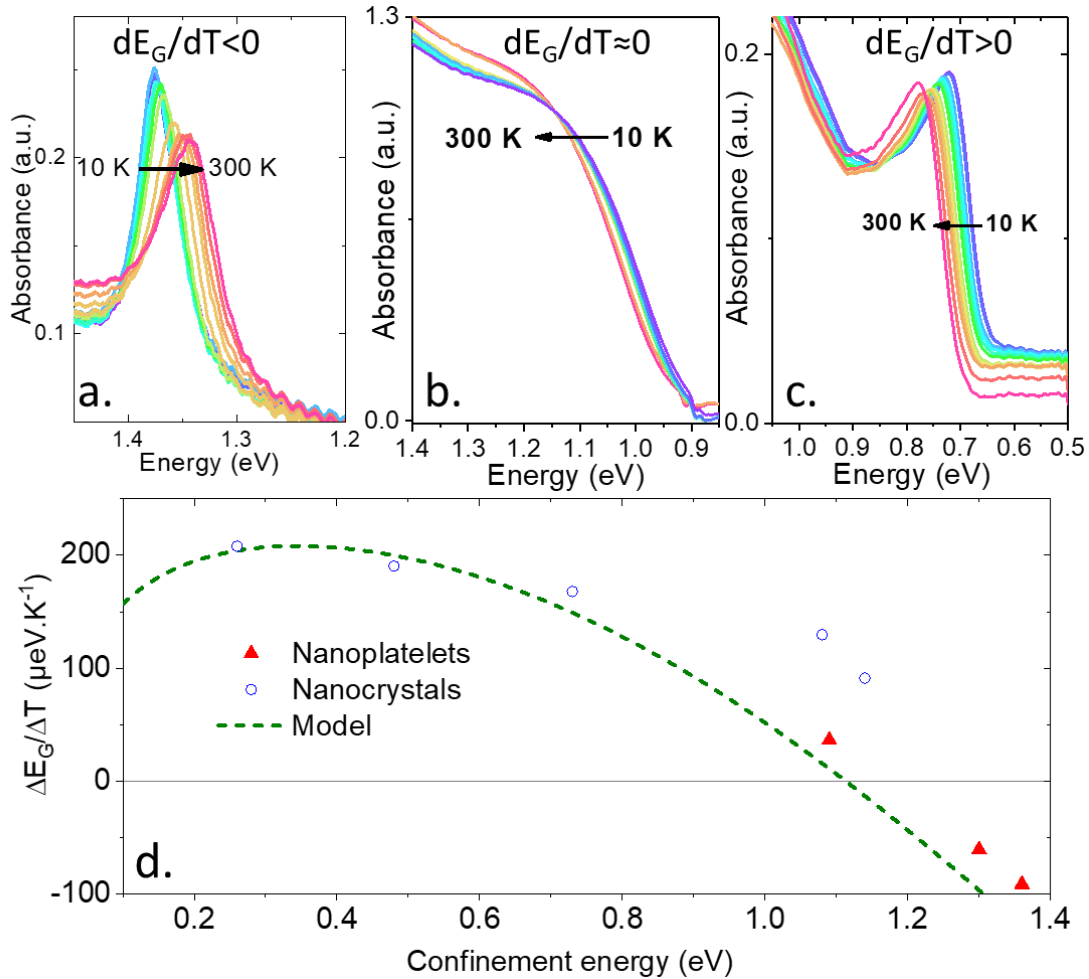


Figure 3 Effect of the temperature on the infrared spectra of HgTe NCs. *a.* Infrared absorption spectra for HgTe NPLs at various temperature between 10 K and 300 K. The absorption blueshifts as temperature is lowered. *b.* Infrared absorption spectra for small HgTe NC at various temperature between 10 K and 300 K. The absorption is barely affected by temperature. *c.* Infrared absorption spectra for large HgTe NC at various temperature between 10 K and 300 K. The absorption redshifts as temperature is lowered. *d.* Experimental temperature dependence of the band gap variation as a function of HgTe NC and NPL E_G confinement energy (at $T=10$ K). The green dotted line is from the (3+1) $k.p$ model.

This change of sign for dE_G/dT excludes that the observed shifts may result solely from a thermal dilation or contraction of the nanoparticles, the size of which will itself tune the confinement. In addition, this contribution of temperature has been found to be quantitatively weak (few $\mu\text{eV} \cdot \text{K}^{-1}$)²³ due to the very small thermal dilation, see Figure S11b. Note that the dimensionality does not seem to be a driving parameter for the magnitude of dE_G/dT since NCs and NPLs with the same E_G confinement energy present similar values for dE_G/dT , see circles and triangles in **Figure 3d**. A similar trend for dE_G/dT (reduction of the amplitude and change of sign) as a function of the inverse of the particle size was observed in the case of PbS NCs. By excluding other possibility, the band gap shift with temperature was finally attributed to a size dependent electron-phonon coupling.^{23,44}

Effect of pressure of the infrared absorption

Similarly, we measured the pressure dependence of the band gap for the HgTe NPLs at various fixed temperatures. Note that for HgTe NCs the pressure dependence of the band gap was already reported in ref⁴⁵ at room temperature. This is why the effect of pressure is discussed in the case of NPL here. In practice, the sample is loaded in a diamond anvil cell together with NaCl which is used as an infrared transparent pressure transmission medium. A ruby is also loaded in the cell and used to determine the applied pressure,

see **Figure 4a** for a scheme of the experiment and the Supporting Information (SI) for details. All the reported observations appear to be fully reversible in the targeted range of pressure (*i.e.* zero pressure spectrum is recovered after pressure releasing). On the other hand, the process is strongly hysteretic: the full release of pressure is required to recover the initial spectrum.

To understand pressure induced change in the infrared spectrum, it is first necessary to determine if the changes might be the result of structural phase transition. Bulk HgTe crystallizes in the cubic zinc blende structure at ambient pressure and undergoes several phase changes as pressure is increased. For bulk material, the cinnabar phase appears around 1.4 GPa.^{46,47} Under NC form, the zinc-blende phase remains stable up to higher pressures.⁴⁵ In the range between 3 and 5 GPa, we can notice the presence of both zinc blende and cinnabar phase (Figure S7). It is however unclear if the two phases exist within the same NPL or if the signal is associated with different particles. In the zinc-blende phase, we identify a decrease in the lattice parameter with initial pressure ($da/dP \approx 7$ pm/GPa), which corresponds to a bulk modulus of 30 GPa for the HgTe NPLs. This value is similar to the one measured for the HgTe NCs⁴⁷ (33 GPa) which was relatively close to the one of the bulk (34 GPa).⁴⁸

The E_G confinement energy appears to be dependent on the applied pressure, see **Figure 4b-c**, 6c and S10. The confinement energy gradually increases with pressure corresponding to a positive value for dE_G/dP . This is similar to what have been measured for HgTe NCs,⁴⁵ and wider band gap II-VI semiconductor quantum dots such as CdSe⁴⁹ and core shell made of CdSe/ZnS.^{50,51} Applying pressure in the range of existence of the zinc blende phase can leads to a shift of the confinement energy up to 300 meV, which is far more than the shift induced by cooling the sample from room temperature to cryogenic condition (≈ 35 meV). This order of magnitude is partially expected from the lattice parameter change under pressure around 6% in Figure S7c and estimated around only 0.08% over 300K temperature range in Figure S11b. Above 4 GPa, no excitonic feature is identified. This observation is correlated to the disappearance of the zinc-blende phase under similar pressure, see Figure S7.

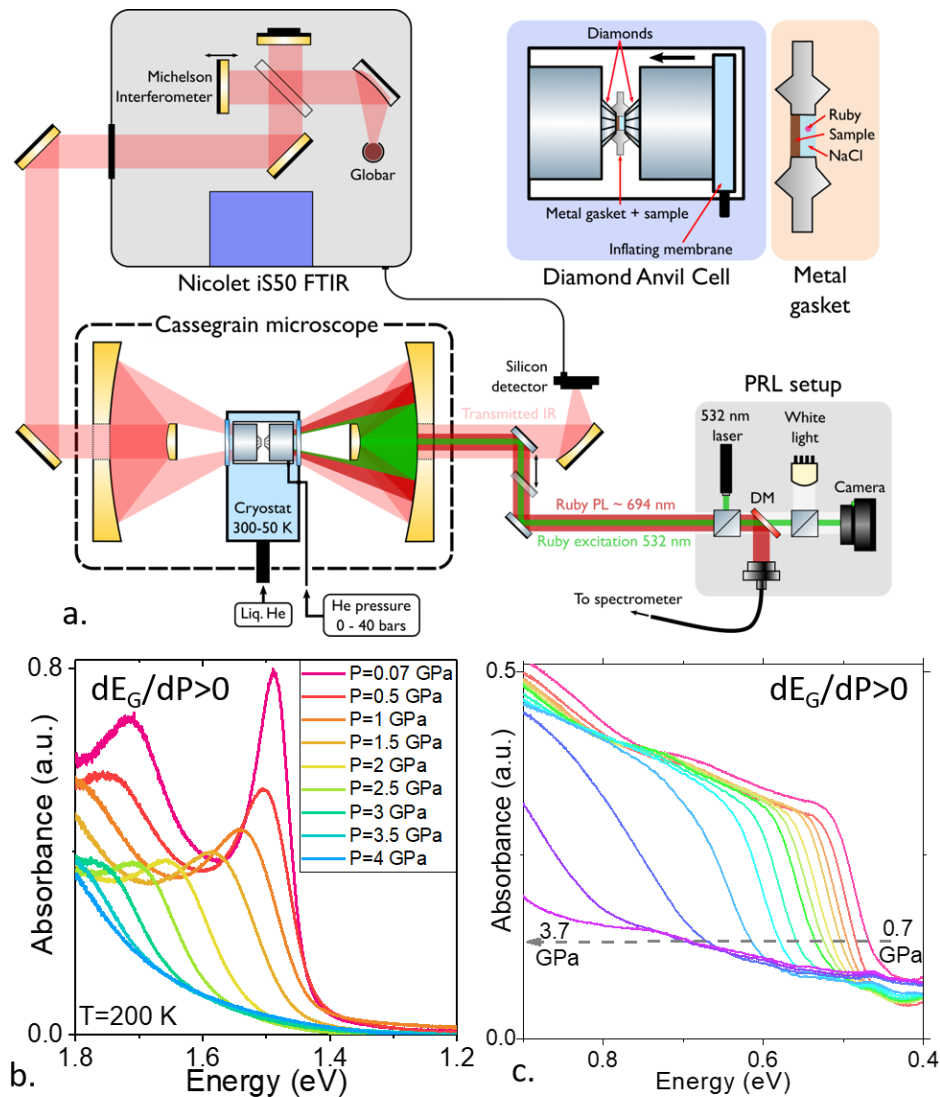


Figure 4 Effect of the pressure on the infrared spectra of HgTe NCs. a. Scheme of the experimental setup of the SMIS beamline of synchrotron SOLEIL to probe the infrared spectrum of a HgTe NPLs under pressure. b. Infrared spectrum of HgTe NPLs at 200 K and under various pressure in the range of 0 to 4 GPa. c. Infrared spectrum of HgTe NCs at 300 K and under various pressure in the range of 0 to 4 GPa

Multiband $\mathbf{k}\cdot\mathbf{p}$ modelling

In order to address the striking temperature and pressure dependence of the spectra observed in **Figure 3** and **Figure 4**, we model the nanoparticles energies using multiband $\mathbf{k}\cdot\mathbf{p}$ simulations. HgTe dispersion appears to be non-parabolic^{26,52} and the effective mass approximation (1 band model) is only valid over 1% of the Brillouin zone. For wavevector far away for the Γ point, the dispersion becomes first linear and then further slightly bends down for wavevectors corresponding to the confinement of NPL (confinement energy above 1 eV). More bands need to be included in the $\mathbf{k}\cdot\mathbf{p}$ model to obtain an accurate description of the dispersion relation.

When a detailed description of the valence band is required, 8 band $\mathbf{k}\cdot\mathbf{p}$ model describing the Γ_6 , Γ_7 and Γ_8 bands and their degeneracy is used.^{53,54} It has been estimated that the 8 band $\mathbf{k}\cdot\mathbf{p}$ model correctly described the Brillouin zone over 8% of it, corresponding to energies significantly smaller than the ones of the NPLs which are around 1.1-1.8 eV.⁵³ While, the effective mass approximation only describes the very bottom parabolic part of the light hole like (Γ_8) band, the inclusion of additional conduction band is critical to capture the linear dispersion occurring at larger wavevector and higher energies.

In HgTe contrary to PbS, the hole and electron effective masses strongly differ, the valence band mass being more than 10 times the conduction band mass. So the confinement is mostly the result of the conduction band and the latter is the one requiring the most accuracy. In the SI (Figure S14-S17), we provide a 14-band **k.p** model, which includes the minimum amount of bands to properly take into account additional conduction bands.⁵⁵

This 14-band model aims at supporting a simpler (3+1) effective bands model later described in the main text. This simple model provides the driving concepts leading to the realistic calculated 14-band structure. The 14-band model includes notably the spin-orbit coupling, disregarded in the simplified model, and all relevant interactions. Note that coulombic renormalization of the band gap has been neglected since these effects have found to be small in the case of HgTe NCs.²⁶ Even though the conclusions are qualitatively the same using each version of the model, the complete version may somehow blurry the message with a large amount of involved parameters. This is why, in the following, we describe a simplified approach. We use a (3+1) bands one-dimensional **k.p**-like model, referred to as the simple model and described in the SI. Its illustrative role is to be tractable and to account for all the experimental features with a minimal set of necessary ingredients, defined in **Figure 1**. The associated parameters are given in Table 1.

The (3+1) band model accounts for the 2 bands with Γ_8 symmetry at $\vec{k} = 0$ (called HH and LH in **Figure 1**) which are the bands around the Fermi level and involved in the absorption. We also include the Γ_6 valence band and a second conduction band. The introduction of this effective high energy band in the conduction band is central to accurately describe the bending of the LH band occurring at large wave vector.

All the experimental data are then accounted from the variation with temperature and pressure of only 2 bulk band gap energies: $E_g(T,P)$ and $E_{gC}(T,P)$. As these gaps will be affected by pressure and temperature, the bands curvature will be modified (through the E_p and E_{pC} parameters coupling the bands), unveiling effect of pressure and temperature in presence of confinement (*i.e.* away from the Γ point). The strong confinement in the HgTe nanoparticles is modeled by the discretization of the \vec{k} wavevector in the reciprocal space, typically at $k = \pm \frac{\pi}{L}$ along [001] where L is size of the nanoparticle, accounting for the contraction and thermal dilation of the nanoparticle with pressure and temperature respectively.

Table 1 Values of the simple (3+1) and the 14-band model parameters used to describe the confinement, temperature and pressure dependences. n.a. means non applicable.

Parameter	Simple band value	(3+1)-model	14-band model value	k.p	Reference
E_g^0	0.31 eV		0.375 eV		0.303 eV in Ref. 56
Simulated $E_g(k=0)$			0.305 eV		
E_{gC}^0	4 eV		2.5 eV		4.11 eV (QSGW) and 3.07 eV (LDA) in Ref. 52
Simulated $E_{gC}(k=0)$			2.78 eV		
E_p	25 eV		30 eV		Fitted, 24.8 eV in Ref. 56
E_{pC}	15 eV		20 eV		Fitted
E_{pV}	2.5 eV		10 eV		Fitted
$\partial E_g / \partial T$	from $E_g(T,P)$, typically $\approx -0.61 \text{ meV.K}^{-1}$		from $E_g(T,P)$ plus a constant 0.1 meV.K^{-1} , typically $\approx -0.51 \text{ meV.K}^{-1}$		See $E_g(T,P)$ formula in SI
$\partial E_{gC} / \partial T$	-2.4 meV.K^{-1}		-3 meV.K^{-1}		Fitted
$\partial E_g / \partial P$	-85 meV/GPa		-95 meV/GPa		Fitted
$\partial E_{gC} / \partial P$	$+220 \text{ meV/GPa}$		$+600 \text{ meV/GPa}$		Fitted

a bulk HgTe lattice parameter	0.6465 nm		Ref 35 at 300K
a_0^{NP} HgTe nanoparticle lattice parameter	0.6500 nm		From measurement Fig. S6c on NPLs at 300K
B_0 HgTe nanoparticle elasticity modulus	30 GPa		From measurement Fig. S6c on NPLs at 300K
$\partial B_0/\partial P$	-1.5		From measurement Fig. S6c on NPLs at 300K
m_{LH}^*	$0.013 m_0$	$0.013 m_0$	Output
m_{HH}^*	$0.33 m_0$	$0.69 m_0$	Output
Spin-orbit coupling Δ	n.a.	0.5 eV	0.775 eV (hQSGW) in Ref. 52 1.08 eV in Ref. 52
Simulated spin-orbit coupling at $k=0$	n.a.	0.732 eV	
Δ_c	n.a.	1 eV	1.06 eV (QSGW) 1.17 eV (LDA) in Ref. 52
Simulated spin-orbit coupling at $k=0$	n.a.	0.768 eV	
Δ'	n.a.	-1.5 eV	Fitted

Figure 5a and **Figure 6a** show respectively the effects of temperature and pressure on the relation dispersion. When temperature or pressure increases, the band gaps E_g and E_{gC} are altered as depicted in Figure S11c. The effect on the LH band effective mass m_{LH}^* , defined at small k , is given by:

$$E_{LH}(k, T, P) \approx \left(1 + \frac{E_p}{E_g(T, P)} - \frac{E_{pC}}{E_{gC}(T, P)} \right) \frac{\hbar^2 k^2}{2m_0} = \frac{\hbar^2 k^2}{2m_{LH}^*} \quad (1)$$

As seen in the equation (1), these variations lead to a competition between the two gaps $E_g(T, P)$ and $E_{gC}(T, P)$ in the alteration of $E_{LH}(k)$ relation dispersion.

At low energy (*ie* small k , see the left arrow in the dispersion relation) the dispersion is ruled by the m_{LH}^* effective mass reduction. The energy blue-shifts as shown in **Figure 5a** for temperature, and **Figure 6a** for pressure as the result of the drastic reduction of E_g from 0.31 eV at 0 K down to 0.12 eV at room temperature. E_g being one order of magnitude smaller than E_{gC} , and $\partial E_g/\partial T/E_g$ and $\partial E_{gC}/\partial T/E_{gC}$ partial derivatives being comparable, the effective mass is essentially governed by E_g (*ie* the second conduction band plays no role in the vicinity of the Γ point). The same rule of thumb holds for pressure.

However, at high energies (*ie* high k , see the right arrow in the dispersion relation), the parabolic equation (1) ceases to be valid. When temperature increases, the second conduction band counteracts m_{LH}^* reduction. As the result of the shrinking of E_{gC} , the second conduction band becomes closer to the LH band and repeals it downwards as seen in **Figure 5a**. In other words, the second conduction band is responsible for the sublinear relation dispersion observed at large k and leads to an inversion of the confinement energy variation with a simulated red-shift of the E_G confinement energy.

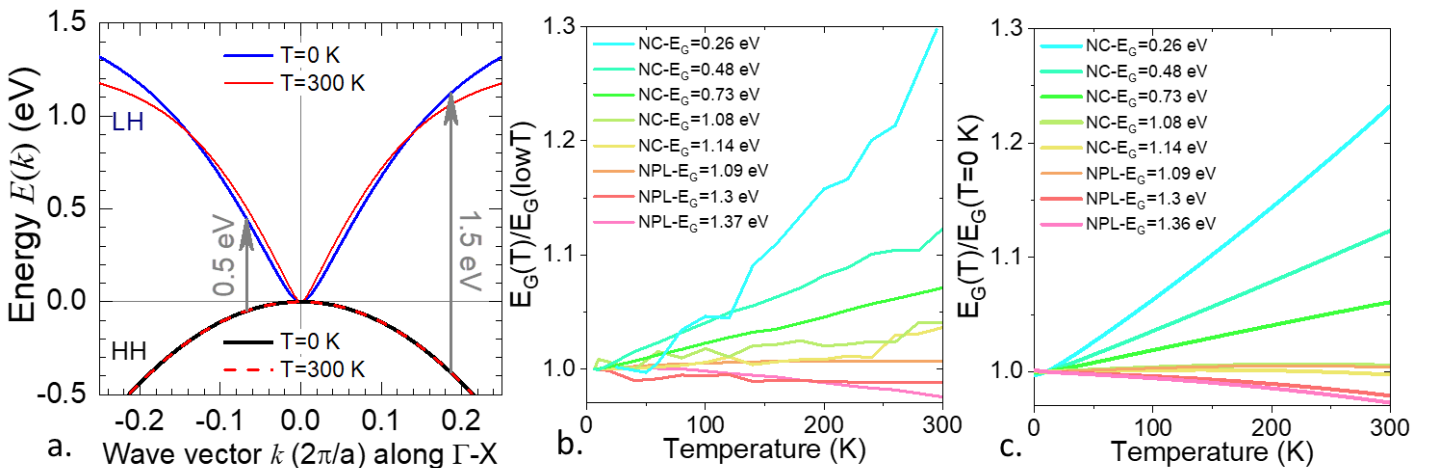


Figure 5 Modelling the temperature effect of the dispersion relation. a. Effect of temperature on $E(k)$ when the two bands gaps E_g and E_{gC} varies. The gray arrows corresponds to typical confinement energies for the NPL (1.5 eV) and for NCs (0.5 eV). b. Experimental normalized shift of the band gap as a function of temperature for different 10 K sizes of HgTe NCs and 3 ML HgTe NPLs with different surface chemistry. c. Simulated normalized shift of the confinement energy E_G as a function of temperature for different sizes of HgTe NCs and 3 ML HgTe NPLs with different surface chemistry.

Using the (3+1) band model and -2.4 meV.K^{-1} value for $\frac{\partial E_{gC}}{\partial T}$, the experimental temperature dependence of the band gap (**Figure 5b**) is reproduced, as shown in **Figure 5c**. With low $T=10\text{K}$, the blue shift ($E_G(T)/E_G(\text{low}T)>1$) at small energies is due to the shrinking of the E_g band gap. However, at large energies, E_{gC} also shrinks and the closer second conduction band pushes the first one downward, leading to a red-shift ($E_G(T)/E_G(\text{low}T)<1$). This interpretation is further analyzed quantitatively by plotting the simulated slopes $\frac{\Delta E_G}{\Delta T}$ as a function of the confinement energy of the particles, see the dashed green line in **Figure 3d**. We see that the model appears in fair agreement with the observed slopes over the whole range of experimental energies supporting the interpretation of the shrinking of the two gaps energies.

Regarding the pressure dependence, the observed energy shifts with hydrostatic pressure present an average slope $\frac{\Delta E_G}{\Delta P} \simeq 100 \text{ meV/GPa}$ in the case of NPL (**Figure 4b**, **Figure 6c**, and S10) which has to be compared with the value of $\approx 60 \text{ meV/GPa}$ for the nanocrystals⁴⁵(**Figure 4c** and S13). Contrary to the case of temperature, the confinement only quantitatively affects the pressure dependence of the confinement energy. This result contrasts to CdSe where the size dependence of the $\frac{\Delta E_G}{\Delta P}$ parameter has been found to be larger with an increase of $\frac{\Delta E_G}{\Delta P}$ from 40 to 80 meV/GPa as the size changes by only a factor two.⁴⁹

Under increasing pressure, the second conduction band goes away from the first one. Pressure simply pushes E_G confinement energy even higher, see **Figure 6a**. In this case, due to pressure, the second conduction band adds an additional contribution to the one of E_g to the observed blue-shift of the confinement energy E_G .

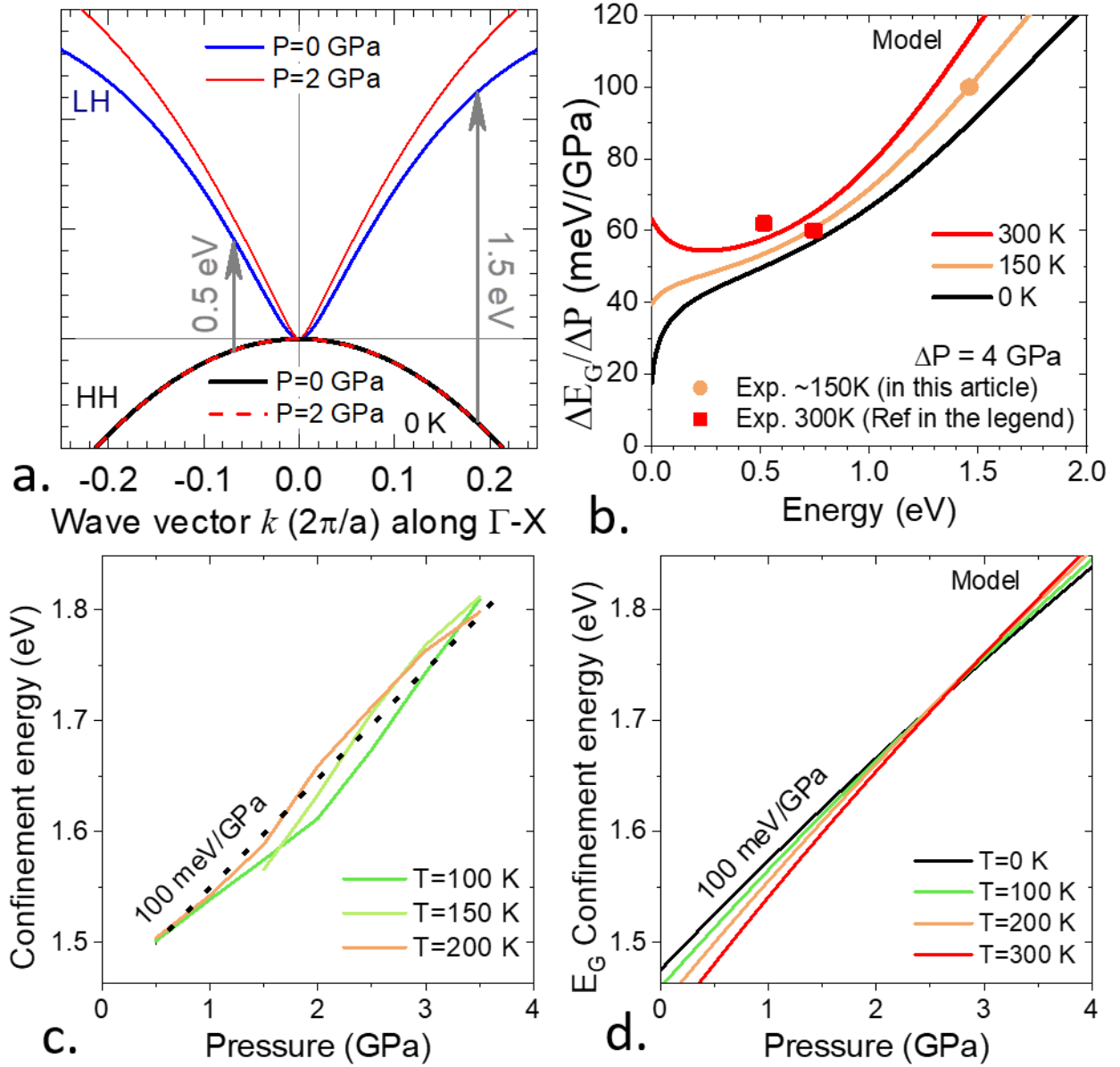


Figure 6 Modelling the pressure effect of the dispersion relation a. Effect of pressure on the conduction and valence bands. The gray arrows corresponds to typical confinement energies for the NPL (1.5 eV) and for NCs (0.5 eV). b. Simulated confinement energy variation over $\Delta P=4$ GPa pressure difference for 3 temperatures. The light orange dot is the experimental measurement around 150 K. The two red squares are the measurements from Ref⁴⁵ at 300 K. c. Experimental confinement energy E_G shift as a function of an applied hydrostatic pressure for HgTe NPLs for three different temperatures. d. Simulated confinement energy E_G shifts as a function of an applied pressure for HgTe NPLs from 0K to 300K. The indicated slope corresponds to 0 K

Using -85 meV/GPa and $+220$ meV/GPa respectively for $\partial E_g/\partial P$ and $\partial E_{gC}/\partial P$, the (3+1) bands model can reproduce the pressure dependence of the confinement energy of the NPL, see **Figure 6c** and d. Also note that using the temperature dependence determined previously, it is possible to reproduce the weak temperature dependence of the pressure potential dE_G/dP of the confinement energy in the case of NPL, see the four simulated temperatures in **Figure 6d**, while capturing the moderate confinement dependence as depicted in **Figure 6b** and S13a.

In all, though both temperature and pressure alter the bulk band gaps E_g and E_{gC} , only blue shifts are observed with increasing pressure, while increasing temperature also shows red-shift. We thus conclude that the unusual change of the optical band gap with temperature and its change of sign reflects the bulk material electronic structure and that phonon does not have to be invoked.⁴⁴

CONCLUSION

In conclusion, we have shown that the strong confinement in NPL causes a non-linear dispersion of high energy state energies in the electronic structure of HgTe nanoparticles. Measurements of the temperature dependence of the optical band gap appears as a direct probe of the dispersion relation. We found that a single parameter set in the (3+1) effective band model can account for all the observed features in energy, temperature, and pressure, providing that the relative proximity of the second conduction band of HgTe be included. We show that the second conduction band appears as a paramount ingredient for absorption energies above ≈ 0.5 –1 eV to describe realistically the energies and electronic states of strongly confined HgTe nanoparticles.

ASSOCIATED CONTENT

Supporting information

The Supporting Information is available free of charge at

Supporting Information include details about material synthesis, material characterization, X-ray diffraction measurement under pressure, raw spectrum about material temperature and pressure dependence and additional information regarding the (3+1)-band **k.p**-like and 14 band **k.p** models.

ACKNOWLEDGMENTS

We thank Mayank Goyal for participation to synchrotron experiments. This work has been supported by the Region Ile-de-France in the framework of DIM Nano-K (grant dopQD). This work was supported by French state funds managed by the ANR within the Investissements d'Avenir programme by Labex Matisse (ANR-11-IDEX-0004-02), and Labex NanoSaclay (ANR-10-LABX-0035), and also by the grant FRONTAL (ANR-19-CE09-0017), IPER-Nano2 (ANR-18CE30-0023-01), Copin (ANR-19-CE24-0022), Graskop (ANR-19-CE09-0026). JQ thanks Chinese Scholar council for PhD grant while AC thanks Agence Innovation Defense. EL thanks the support ERC starting grant blackQD (grant n° 756225). SI thanks the support ERC starting grant Ne2Dem (grant n° 853049).

REFERENCES

- (1) Bourzac, K. Quantum Dots Go on Display. *Nature* **2013**, *493*, 283.
- (2) Green, M.; Mirzai, H. Synthetic Routes to Mercury Chalcogenide Quantum Dots. *J. Mater. Chem. C* **2018**, *6*, 5097–5112.
- (3) Nozik, A. J. Quantum Dot Solar Cells. *Phys. E Low-Dimens. Syst. Nanostructures* **2002**, *14*, 115–120.
- (4) McDonald, S. A.; Konstantatos, G.; Zhang, S.; Cyr, P. W.; Klem, E. J. D.; Levina, L.; Sargent, E. H. Solution-Processed PbS Quantum Dot Infrared Photodetectors and Photovoltaics. *Nat. Mater.* **2005**, *4*, 138–142.
- (5) Hafiz, S. B.; Scimeca, M.; Sahu, A.; Ko, D.-K. Colloidal Quantum Dots for Thermal Infrared Sensing and Imaging. *Nano Converg.* **2019**, *6*, 7.
- (6) Livache, C.; Martinez, B.; Goubet, N.; Ramade, J.; Lhuillier, E. Road Map for Nanocrystal Based Infrared Photodetectors. *Front. Chem.* **2018**, *6*, 575.
- (7) Lhuillier, E.; Guyot-Sionnest, P. Recent Progresses in Mid Infrared Nanocrystal Optoelectronics. *IEEE J. Sel. Top. Quantum Electron.* **2017**, *23*, 6000208.
- (8) Böberl, M.; Kovalenko, M. V.; Gamerith, S.; List, E. J. W.; Heiss, W. Inkjet-Printed Nanocrystal Photodetectors Operating up to 3 μm Wavelengths. *Adv. Mater.* **2007**, *19*, 3574–3578.
- (9) Cryer, M. E.; Halpert, J. E. 300 nm Spectral Resolution in the Mid-Infrared with Robust, High Responsivity Flexible Colloidal Quantum Dot Devices at Room Temperature. *ACS Photonics* **2018**, *5*, 3009–3015.

- (10) Chu, A.; Martinez, B.; Ferré, S.; Noguier, V.; Gréboval, C.; Livache, C.; Qu, J.; Prado, Y.; Casaretto, N.; Goubet, N. *et al* HgTe Nanocrystals for SWIR Detection and Their Integration up to the Focal Plane Array. *ACS Appl. Mater. Interfaces* **2019**, *11*, 33116–33123.
- (11) Chatterjee, A.; Pendyala, N. B.; Jagtap, A.; Rao, K. S. R. K. Uncooled Mid-Wave Infrared Focal Plane Array Using Band Gap Engineered Mercury Cadmium Telluride Quantum Dot Coated Silicon ROIC. *E-J. Surf. Sci. Nanotechnol.* **2019**, *17*, 95–100.
- (12) Buurma, C.; Pimpinella, R. E.; Ciani, A. J.; Feldman, J. S.; Grein, C. H.; Guyot-Sionnest, P. MWIR Imaging with Low Cost Colloidal Quantum Dot Films. In *Optical Sensing, Imaging, and Photon Counting: Nanostructured Devices and Applications 2016*; International Society for Optics and Photonics, 2016; Vol. 9933, p 993303.
- (13) Lhuillier, E.; Scarafagio, M.; Hease, P.; Nadal, B.; Aubin, H.; Xu, X. Z.; Lequeux, N.; Patriarche, G.; Ithurria, S.; Dubertret, B. Infrared Photodetection Based on Colloidal Quantum-Dot Films with High Mobility and Optical Absorption up to THz. *Nano Lett.* **2016**, *16*, 1282–1286.
- (14) Goubet, N.; Jagtap, A.; Livache, C.; Martinez, B.; Portalès, H.; Xu, X. Z.; Lobo, R. P. S. M.; Dubertret, B.; Lhuillier, E. Terahertz HgTe Nanocrystals: Beyond Confinement. *J. Am. Chem. Soc.* **2018**, *140*, 5033–5036.
- (15) Izquierdo, E.; Robin, A.; Keuleyan, S.; Lequeux, N.; Lhuillier, E.; Ithurria, S. Strongly Confined HgTe 2D Nanoplatelets as Narrow Near-Infrared Emitters. *J. Am. Chem. Soc.* **2016**, *138*, 10496–10501.
- (16) Tenney, S. M.; Vilchez, V.; Sonnleitner, M. L.; Huang, C.; Friedman, H. C.; Shin, A. J.; Atallah, T. L.; Deshmukh, A. P.; Ithurria, S.; Caram, J. R. Mercury Chalcogenide Nanoplatelet–Quantum Dot Heterostructures as a New Class of Continuously Tunable Bright Shortwave Infrared Emitters. *J. Phys. Chem. Lett.* **2020**, *11*, 3473–3480.
- (17) Chang, Y.-C.; Schulman, J. N.; Bastard, G.; Guldner, Y.; Voos, M. Effects of Quasi-Interface States in HgTe-CdTe Superlattices. *Phys. Rev. B* **1985**, *31*, 2557–2560.
- (18) Rogalski, A. HgCdTe Infrared Detector Material: History, Status and Outlook. *Rep. Prog. Phys.* **2005**, *68*, 2267–2336.
- (19) Orłowski, N.; Augustin, J.; Gołacki, Z.; Janowitz, C.; Manzke, R. Direct Evidence for the Inverted Band Structure of HgTe. *Phys. Rev. B* **2000**, *61*, R5058–R5061.
- (20) Bernevig, B. A.; Hughes, T. L.; Zhang, S.-C. Quantum Spin Hall Effect and Topological Phase Transition in HgTe Quantum Wells. *Science* **2006**, *314*, 1757–1761.
- (21) König, M.; Wiedmann, S.; Brüne, C.; Roth, A.; Buhmann, H.; Molenkamp, L. W.; Qi, X.-L.; Zhang, S.-C. Quantum Spin Hall Insulator State in HgTe Quantum Wells. *Science* **2007**, *318*, 766–770.
- (22) Leubner, P.; Lunczer, L.; Brüne, C.; Buhmann, H.; Molenkamp, L. W. Strain Engineering of the Band Gap of HgTe Quantum Wells Using Superlattice Virtual Substrates. *Phys. Rev. Lett.* **2016**, *117*, 086403.
- (23) Lhuillier, E.; Keuleyan, S.; Guyot-Sionnest, P. Optical Properties of HgTe Colloidal Quantum Dots. *Nanotechnology* **2012**, *23*, 175705.
- (24) Chen, M.; Guyot-Sionnest, P. Reversible Electrochemistry of Mercury Chalcogenide Colloidal Quantum Dot Films. *ACS Nano* **2017**, *11*, 4165–4173.
- (25) Hudson, M. H.; Chen, M.; Kamysbayev, V.; Janke, E. M.; Lan, X.; Allan, G.; Delerue, C.; Lee, B.; Guyot-Sionnest, P.; Talapin, D. V. Conduction Band Fine Structure in Colloidal HgTe Quantum Dots. *ACS Nano* **2018**, *12*, 9397–9404.
- (26) Allan, G.; Delerue, C. Tight-Binding Calculations of the Optical Properties of HgTe Nanocrystals. *Phys. Rev. B* **2012**, *86*, 165437.
- (27) Keuleyan, S. E.; Guyot-Sionnest, P.; Delerue, C.; Allan, G. Mercury Telluride Colloidal Quantum Dots: Electronic Structure, Size-Dependent Spectra, and Photocurrent Detection up to 12 μm . *ACS Nano* **2014**, *8*, 8676–8682.
- (28) Chen, M.; Guyot-Sionnest, P. Reversible Electrochemistry of Mercury Chalcogenide Colloidal Quantum Dot Films. *ACS Nano* **2017**, *11*, 4165–4173.
- (29) Jagtap, A.; Martinez, B.; Goubet, N.; Chu, A.; Livache, C.; Gréboval, C.; Ramade, J.; Amelot, D.; Troussset, P.; Triboulin, A. *et al* Design of a Unipolar Barrier for a Nanocrystal-Based Short-Wave Infrared Photodiode. *ACS Photonics* **2018**, *5*, 4569–4576.
- (30) Izquierdo, E.; Dufour, M.; Chu, A.; Livache, C.; Martinez, B.; Amelot, D.; Patriarche, G.; Lequeux, N.; Lhuillier, E.; Ithurria, S. Coupled HgSe Colloidal Quantum Wells through a Tunable Barrier: A Strategy To Uncouple Optical and Transport Band Gap. *Chem. Mater.* **2018**, *30*, 4065–4072.
- (31) Gréboval, C.; Izquierdo, E.; Livache, C.; Martinez, B.; Dufour, M.; Goubet, N.; Moghaddam, N.; Qu, J.; Chu, A.; Ramade, J. *et al* Impact of Dimensionality and Confinement on the Electronic Properties of Mercury Chalcogenide Nanocrystals. *Nanoscale* **2019**, *11*, 3905–3915.

- (32) Goubet, N.; Livache, C.; Martinez, B.; Xu, X. Z.; Ithurria, S.; Royer, S.; Cruguel, H.; Patriarche, G.; Ouerghi, A.; Sully, M. *et al* Wave-Function Engineering in HgSe/HgTe Colloidal Heterostructures To Enhance Mid-Infrared Photoconductive Properties. *Nano Lett.* **2018**, *18*, 4590–4597.
- (33) Pedetti, S.; Nadal, B.; Lhuillier, E.; Mahler, B.; Bouet, C.; Abécassis, B.; Xu, X.; Dubertret, B. Optimized Synthesis of CdTe Nanoplatelets and Photoresponse of CdTe Nanoplatelets Films. *Chem. Mater.* **2013**, *25*, 2455–2462.
- (34) Pedetti, S.; Ithurria, S.; Heuclin, H.; Patriarche, G.; Dubertret, B. Type-II CdSe/CdTe Core/Crown Semiconductor Nanoplatelets. *J. Am. Chem. Soc.* **2014**, *136*, 16430–16438.
- (35) Śniadower, L.; Psoda, M.; Gałazka, R. R. X-Ray Measurements of Lattice Dilatation in HgTe. *Phys. Status Solidi B* **1968**, *28*, K121–K123.
- (36) Keuleyan, S.; Lhuillier, E.; Guyot-Sionnest, P. Synthesis of Colloidal HgTe Quantum Dots for Narrow Mid-IR Emission and Detection. *J. Am. Chem. Soc.* **2011**, *133*, 16422–16424.
- (37) Geiregat, P.; Houtepen, A. J.; Sagar, L. K.; Infante, I.; Zapata, F.; Grigel, V.; Allan, G.; Delerue, C.; Van Thourhout, D.; Hens, Z. Continuous-Wave Infrared Optical Gain and Amplified Spontaneous Emission at Ultralow Threshold by Colloidal HgTe Quantum Dots. *Nat. Mater.* **2018**, *17*, 35–42.
- (38) Guenzer, C. S.; Bienenstock, A. Temperature Dependence of the HgTe Band Gap. *Phys. Rev. B* **1973**, *8*, 4655–4667.
- (39) Dobrowolska, M.; Mycielski, A.; Dobrowolski, W. Determination of Temperature Dependence of Energy Gap in HgTe by Oscillatory Magnetotransmission Measurements. *Solid State Commun.* **1978**, *27*, 1233–1235.
- (40) Fonthal, G.; Tirado-Mejía, L.; Marín-Hurtado, J. I.; Ariza-Calderón, H.; Mendoza-Alvarez, J. G. Temperature Dependence of the Band Gap Energy of Crystalline CdTe. *J. Phys. Chem. Solids* **2000**, *61*, 579–583.
- (41) Ithurria, S.; Tessier, M. D.; Mahler, B.; Lobo, R. P. S. M.; Dubertret, B.; Efros, A. L. Colloidal Nanoplatelets with Two-Dimensional Electronic Structure. *Nat. Mater.* **2011**, *10*, 936–941.
- (42) Kushavah, D.; Mohapatra, P. K.; Ghosh, P. Reduced Carrier Trapping in CdSe/ZnS/CdSe Heterostructure Quantum Dots Inferred from Temperature Dependent Spectral Studies. *Phys. E Low-Dimens. Syst. Nanostructures* **2018**, *102*, 58–65.
- (43) Ramade, J.; Andriambarijaona, L. M.; Steinmetz, V.; Goubet, N.; Legrand, L.; Barisien, T.; Bernardot, F.; Testelin, C.; Lhuillier, E.; Bramati, A. *et al* Exciton-Phonon Coupling in a CsPbBr₃ Single Nanocrystal. *Appl. Phys. Lett.* **2018**, *112*, 072104.
- (44) Olkhovets, A.; Hsu, R.-C.; Lipovskii, A.; Wise, F. W. Size-Dependent Temperature Variation of the Energy Gap in Lead-Salt Quantum Dots. *Phys. Rev. Lett.* **1998**, *81*, 3539–3542.
- (45) Livache, C.; Goubet, N.; Gréboval, C.; Martinez, B.; Ramade, J.; Qu, J.; Triboulin, A.; Cruguel, H.; Baptiste, B.; Klotz, S. *et al* Effect of Pressure on Interband and Intraband Transition of Mercury Chalcogenide Quantum Dots. *J. Phys. Chem. C* **2019**, *123*, 13122–13130.
- (46) Wright, N. G.; McMahan, M. I.; Nelmes, R. J.; San-Miguel, A. Crystal Structure of the Cinnabar Phase of HgTe. *Phys. Rev. B* **1993**, *48*, 13111–13114.
- (47) San-Miguel, A.; Wright, N. G.; McMahan, M. I.; Nelmes, R. J. Pressure Evolution of the Cinnabar Phase of HgTe. *Phys. Rev. B* **1995**, *51*, 8731–8736.
- (48) Düz, I.; Erdem, I.; Ozdemir Kart, S.; Kuzucu, V. First Principles Investigations of HgX (X=S, Se and Te). *Arch Mater Sci Eng* **2016**, *79*, 5–11.
- (49) Meulenbergh, R. W.; Strouse, G. F. Pressure-Induced Electronic Coupling in CdSe Semiconductor Quantum Dots. *Phys. Rev. B* **2002**, *66*, 035317.
- (50) Li, B.; Liu, W.; Zhu, X.; Lin, S.; Yang, Y.; Yang, Q.; Jin, P. Pressure-Dependent Photoluminescence of CdSe/ZnS Quantum Dots: Critical Point of Different Pressure Regimes. *Phys. Lett. A* **2019**, *383*, 1483–1486.
- (51) Fan, H. M.; Ni, Z. H.; Feng, Y. P.; Fan, X. F.; Kuo, J. L.; Shen, Z. X.; Zou, B. S. High Pressure Photoluminescence and Raman Investigations of CdSe/ZnS Core/Shell Quantum Dots. *Appl. Phys. Lett.* **2007**, *90*, 021921.
- (52) Svane, A.; Christensen, N. E.; Cardona, M.; Chantis, A. N.; van Schilfgaarde, M.; Kotani, T. Quasiparticle Band Structures of β -HgS, HgSe, and HgTe. *Phys. Rev. B* **2011**, *84*, 205205.
- (53) Man, P.; Pan, D. S. Infrared Absorption in HgTe. *Phys. Rev. B* **1991**, *44*, 8745–8758.
- (54) Pokatilov, E. P.; Fonoberov, V. A.; Fomin, V. M.; Devreese, J. T. Electron and Hole States in Quantum Dot Quantum Wells within a Spherical Eight-Band Model. *Phys. Rev. B* **2001**, *64*, 245329.
- (55) El kurdi, M.; Fishman, G.; Sauvage, S.; Boucaud, P. Comparison between 6-Band and 14-Band **k.p** Formalisms in SiGe/Si Heterostructures. *Phys. Rev. B* **2003**, *68*, 165333.
- (56) Laurenti, J. P.; Camassel, J.; Bouhemadou, A.; Toulouse, B.; Legros, R.; Lusson, A. Temperature Dependence of the Fundamental Absorption Edge of Mercury Cadmium Telluride. *J. Appl. Phys.* **1990**, *67*, 6454–6460.

TOC graphic

