

Initial investigation of B4C–TiB2 composites as neutron absorption material for nuclear reactors

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Initial investigation of B₄C-TiB₂ composites as neutron

2 absorption material for nuclear reactors

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23 Abstract

In this study, a specifically designed B₄C-TiB₂ composite with the 24 typical microstructural feature of a TiB₂ network (cages) that 25 encapsulates a B₄C matrix was fabricated by the molten-salt and spark 26 plasma sintering (SPS) method. The finite-element (FE) calculation 27 results show that the connected TiB₂ cages constitute a thermally 28 conductive network, which effectively improves the overall thermal 29 conductivity of the composite; these results agree well with the 30 experimental results. Moreover, the Vickers indentation results reveal that 31 the TiB₂ network (cages) can effectively impinge/block the propagation 32 of cracks, which increases the composite toughness. The composite was 33 subjected to helium (He) ion irradiation to simulate the situation in which 34 the B₄C-TiB₂ composites serve as neutron absorption material, and for 35 which case a high quantity of He atoms is produced by the B^{10} (n, α) Li⁷ 36 nuclear reaction. According to the transmission electron microscopy 37 (TEM) results, the interfaces between TiB₂ and B₄C act as effective sinks 38 for He atoms, and are preferential nucleation sites for He bubbles. The 39 theoretical and experimental results show that when the B₄C-TiB₂ 40 composites serve as neutron absorption pellets in nuclear reactors, they 41 exhibit a better resistance to their disintegration than pure B₄C pellets. 42 Consequently, the performance of the control rods of nuclear reactors can 43 be improved. 44

- 45 Keywords: neutron absorber; B_4C pellet; He irradiation; crack
- 46 propagation; finite element.

1. Introduction

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B₄C is widely used as a neutron absorption material for control rods 49 (CRs) in many nuclear reactor systems, in particular in fast reactor 50 systems [1] (e.g., Phenix (France), BN (Russia), JOYO (Japan), ALMR, 51 FFTF (United States)) owing to its high melting temperature, outstanding 52 thermal stability, good mechanical properties, and the high neutron 53 absorption cross section of ¹⁰B; B₄C neutron absorbers play an important 54 role regarding the performance of CRs by, for example, adjusting the 55 reactivity compensation during the normal operation cycles of nuclear 56 reactors, initiating/terminating an operation cycle for refueling or reactor 57 component maintenance, and rapidly shutting down the reactor if 58 abnormal conditions occur. Therefore, B₄C neutron absorbers are of 59 primary importance for the safe and efficient operation of nuclear 60 reactors. 61 In many CR designs (e.g., JOYO), B₄C materials are usually 62 presented in the form of pellets. However, during the operation lifetime of 63 CRs, B₄C pellets often fracture into pieces and even powders [2]. These 64 fractured B₄C pellet pieces progressively fill the gaps between the pellets 65 cladding materials, which accelerates the absorber-cladding 66 and mechanical interaction [2]. This leads to an incompatibility between the 67 cladding materials and other components of reactors (e.g., induced 68 absorber swelling cladding cracking (IASCC)) and shortens the CR 69

lifetime. Moreover, the cracked materials are very sensitive to the subsequent radiation-assisted dissolution (i.e., radiolysis effects [3-5]). Regarding water-cooled nuclear systems, when cooling water leaks into the pellet capsule owing to IASCC-induced cracking [6], the radiolysis effect accelerates the dissolution of the absorber materials and promotes the disintegration of B_4C pellets [5]. Both effects degrade the CRs and safe operation of nuclear reactors.

According to previous research studies, the disintegration of bulk 77 B_4C ceramic pellets is closely associated with 78 the thermal-gradient-induced macrocracks and swelling-gradient-induced 79 microcracks [7]. During the service lifetime of CRs, the (n, α) nuclear 80 reaction releases an average energy of 2.78 MeV per event [1]; the most 81 energy is deposited within the B₄C matrix. For instance, at a CR burnup 82 of 10^{22} cap/cm³ per year, a volume power of 1.5×10^2 W/cm³ is deposited 83 in the B₄C pellets. Owing to the low thermal conductivity of B₄C and the 84 high heat-generation rate caused by the ¹⁰B (n, α) ⁷Li reaction, the 85 temperature gradient of B₄C pellets can reach up to 10³ K/cm. Such a high 86 temperature gradient causes a great thermal stress and results in 87 thermal-stress-induced macrocracks. Moreover, a high quantity of He 88 atoms produced by the transmutation reaction accumulate in the form of 89 flat lenticular He bubbles [8,9]. These He bubbles cause significant 90 volume swelling. When the neutron absorption pellets are placed in 91

thermal neutron reactors, owing to the self-shielding effect [7,10], the burnup of the pellet periphery is usually stronger than that of the interior. Because the He atom production is proportional to the burnup level, a gradient distribution of He bubbles leads to a steep swelling gradient of the B₄C material and an abrupt local strain [7]: the consequence are swelling-gradient-induced microcracks. In general, the released heat and He atoms produced by the 10 B (n, α) 7 Li reaction are two factors that lead to the formation of cracks. Both factors promote the formation of macro-scale cracks and facilitate the crack propagation through pellets, which result in their disintegration [2,11]. Therefore, it is of vital importance to improve the thermal

conductivity and irradiation-induced swelling resistance of B₄C absorber materials to guarantee the good long-term performance of CRs and a safe and efficient operation of nuclear reactors. B₄C-diborides composites have been explored a long time ago to improve the performance of B₄C in the irradiation field. For example, B₄C-HfB₂ composite was designed to improve fracture toughness [12, 13] and the thermal-mechanical properties of the neutron absorber [12]. In this study, a low economic cost and honeycomb-structured B₄C-TiB₂ composite was elaborated to attain the aforementioned objective. The main feature of the composite is a highly thermally conductive cage-like TiB₂ network that encapsulates conventional B₄C matrix grains. TiB₂ has a high melting point (3498 K),

- exceptional hardness (25-35 GPa), highly thermal conductivity, excellent 114 wear resistance, a low thermal expansion coefficient $(4.6 \times 10^{-6} \text{ K}^{-1})$, 115 which is almost the same as that of B_4C , 4.5×10^{-6} K⁻¹), and a high 116 oxidation resistance [14-18]. Owning to the inimitable characteristics of 117 TiB₂, it is therefore selected as the cage-like structure of the B₄C-TiB₂ 118 composites. The good features of conventional B₄C material are 119 maintained, and the thermal conductivity of the composite is significantly 120 improved with respect to that of the pure B₄C material. In addition, the 121 composite exhibits many interfaces. 122
- The B_4C -Ti B_2 composites were subjected to He ion irradiation to simulate the situation in which the B_4C -Ti B_2 composite is exposed to a high neutron flux, and a high quantity of He atoms is produced by the (n, α) reaction. Moreover, a series of post-irradiation analyses and FE calculations were conducted to evaluate the behavior of the B_4C -Ti B_2 component in a nuclear reactor environment.
- 129 2. Experimental details
- 130 2.1. Material synthesis
- A two-step synthesis method was employed to synthesize the B_4C –
 132 TiB_2 composite [19]. In the first step, Ti and B_4C powders were mixed
 133 with NaCl and KCl in an Al_2O_3 crucible. The mixtures with initial B_4C/Ti 134 molar ratios of 6:1, 4:1, and 2:1 were exposed to an Ar atmosphere at
 135 1273 K. After the (1) and (2) in-situ reactions in the melted salt bath, a

uniform TiB₂ coating occurred on the B₄C particles. The remnant NaCl, KCl, and other impurities were washed away with deionized water, and the samples were subsequently dried to obtain dry TiB₂-coated B₄C powders. Note that the melt point of NaCl and KCl mixing is low (~ 933 K) and the liquid mass transfer mode between B₄C and Ti in the molten salts guarantees the uniform nucleation and growth of TiB2 on the surface of B₄C particles. Therefore, the NaCl and KCl were employed as the medium for reactions.

$$3Ti + B_4C \rightarrow TiC + 2TiB_2 \qquad (1)$$

$$2TiC + B_4C \rightarrow 2TiB_2 + 3C \qquad (2)$$

In the second step, the mixtures containing uniformly TiB₂-coated B₄C powders with different TiB₂ contents were sintered in a spark plasma sintering (SPS) system (SPS-20T-15, Chen Hua) with the same procedure to obtain three cylindrical bulk samples with diameters of 13 mm and heights of 2 mm (denoted as S1, S2, and S3). In the SPS procedure, the heating rates are 50 K/min from RT to 1673 K, and 25 K/min from 1673 K to 1973 K, respectively. The holding time and temperature are 10 min and 1973 K, respectively. The pressure during the holding stage is 45 MPa. The cooling rate is 50 K/min from the holding temperature to RT. Each bulk sample has a relative density of 100%. The volume fractions of TiB₂ phases in B₄C-TiB₂ composites are determined by the equation (3).

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$$\frac{V_{TiB_2}}{V_t} = \frac{\frac{m_{TiB_2}}{\rho_{TiB_2}}}{\frac{m_t}{\rho_t}} = w_{TiB_2} \cdot \frac{\rho_t}{\rho_{TiB_2}}$$
 (3)

Where V_{TiB_2} and V_t are the volume of TiB_2 phases and the B_4C-TiB_2 composites, respectively. w_{TiB_2} is the weight percent of TiB₂ measured by using XRD, the calculated weight percent of TiB₂ for S1, S2 and S3 are 18 %, 26% and 44%, respectively. ρ_{TiB_2} (4.5 g/cm³) is the density of TiB_2 and ρ_t is the density of the B_4C - TiB_2 composites (2.637, 2.796 and 3.051 g/cm³ for S1, S2 and S3, respectively). The density of the B₄C-TiB₂ composites were measured by and Archimedes method. The chemical composition and the calculated volume fraction of each sample are shown in Table 1.

167 2.2. Vickers indentation tests and irradiation procedure

A pyramidal diamond tip with an apical angle of 136° was used to perform Vickers indentation tests. The tests were performed on the surface of the virgin samples to investigate the crack propagation behavior in the composite. During the indentation tests, the maximal load was set to 9.8 N, and the holding time was 15 s. Each indentation was set more than 100 μ m apart to avoid an overlap of the indentation-induced deformations or crack outspread regions.

The He ion irradiation procedure was conducted in a terminal chamber with a 320 kV multidiscipline research platform for highly

- charged ions at the Institute of Modern Physics (IMP), Lanzhou, China. 177 Samples S1, S2, and S3 were irradiated with 500 keV He ions with 178 1×10¹⁷ ions/cm² at 773 K. After the irradiation process, the samples were 179 annealed at 1273 K for 25 min in a vacuum of 10⁻⁵ Pa. In addition, 180 Stopping Range of Ions in Matter (SRIM) codes were conducted with the 181 Kinchin-Pease quick-calculation mode to predict the distribution of the 182 He atoms and displacement damage in B₄C and TiB₂, respectively. The 183 calculated distribution is a function of the penetration depth, as shown in 184 Fig. 1; the Bragg peak with a He concentration of 8.92×10^{21} cm⁻³ (7.02%) 185 is located at a depth of $\sim 1.38 \, \mu m$ in the B₄C material, and 7.18×10^{21} 186 cm⁻³ (6.14%) at a depth of $\sim 1.56 \,\mu m$ in TiB₂ material, respectively. 187
- 188 2.3. Characterization
- The morphology of the sample surface and Vickers indentation 189 impression were recorded by scanning electron microscopy (SEM, 190 Thermo Fisher Quanta FEG 250). In addition, the focused ion beam 191 technology (Thermo Fisher, Helios G4 CX) was employed to lift up thin 192 foils for the transmission electron microscopy (TEM) analysis. The thin 193 foils were analyzed with a Talos F200X TEM at 200 kV in the 194 bright-field mode, and a HV-1000 indentation test machine (Shanghai 195 Lianer Testing Equipment Corporation) was used to perform the Vickers 196 indentation tests. 197
- 198 3. Results and discussions

199	Before the SEM analysis, the composites were mechanically
200	polished with diamond papers (from 800 to 12500 grit) until mirror-like
201	surfaces were obtained. Fig. 2 shows the typical SEM images of the B ₄ C-
202	TiB_2 composites with TiB_2 volume fractions of 10.5% (Fig. 2(a)), 16.2%
203	(Fig. 2(b)), and 29.8% (Fig. 2(c)). The white contrast indicated by black
204	arrows in the SEM images are TiB2 phases, and the left, dark contrast
205	represents B ₄ C grains. The TiB ₂ phases are distributed along the B ₄ C
206	grain boundaries. With increasing TiB2 phase volume fraction, the
207	fraction of isolated TiB2 particles decreases, and more and more TiB2
208	phases along the grain boundaries become connected, which results in an
209	interconnected network from the two-dimensional (2D) point of view
210	(Fig. 2(c)). Therefore, from the three-dimensional point of view, the TiB_2
211	network forms a cage-like structure, which encapsulates the B ₄ C matrix
212	grains. The average thickness of the TiB_2 network of S3 is 0.9 μm , and
213	the statistical measurement based on the line-intercept method [20]
214	reveals that the mean grain sizes of B ₄ C in the S1, S2, and S3 samples are
215	9.5, 8.6, and 6.4 μ m, respectively. Thus, the TiB ₂ fraction increases at the
216	expense of consumed B ₄ C.
217	To evaluate the effect of the TiB ₂ network (cages) and its volume
218	fraction on the heat transfer performance in the B ₄ C-TiB ₂ composites, a
219	series of numerical studies based on the typical microstructures of pure
220	B ₄ C and B ₄ C-TiB ₂ composites were implemented in the FE program

ABAQUS. The geometric constructions of the 2D pure B₄C and B₄C-221 TiB₂ composites FE model is shown in Fig. 3. A thermal continuity 222 constraint was defined at each interface to connect the B₄C and TiB₂ 223 domains without heat dissipation. An initial temperature of 773 K was 224 assigned to the entire computational region to simulate the reactor 225 environment temperature, and 1273 K (boundary condition) was assigned 226 to the left end of each model to generate a stable heater source; the other 227 adiabatic boundary conditions. boundary conditions were 228 temperature dependence of the physical parameters of B₄C and TiB₂ used 229 in the numerical calculation are listed in Table 2 [18,21]. 230 The temperature distributions used with different time scales for the 231 pure B₄C and B₄C-TiB₂ composites models are shown in Fig. 4. The 232 temperature distribution in pure B₄C is homogeneous along the vertical 233 direction, and the isotherms are perpendicular to the horizontal axis. 234 Whereas the temperature in the B₄C-TiB₂ composites presents an 235 irregular gradient distribution. Furthermore, the temperature gradient in 236 the B₄C-TiB₂ composites with a high TiB₂ volume fraction is lower than 237 that of the low TiB₂ volume fraction. Particularly, the maximum 238 temperature gradient appears in pure B₄C. Thus, heat is transmitted faster 239 in the B₄C-TiB₂ composite than in the pure B₄C, the higher the volume 240 fraction of TiB₂, the faster the heat transfers, which is consistent with the 241 experimental results in [19]. 242

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To compare the heat transfer performance of the four structures quantitatively, the temperature variations at the middle point of the right-end boundary in each model are presented with respect to time in Fig. 5. The heating rate in the B₄C-TiB₂ composite with higher TiB₂ volume fraction is higher than that of lower ones. For example, the temperature at point P_3 in the B_4C-TiB_2 composite with the highest TiB_2 volume fraction (29.8%) reaches 1181 K within 1 µs, while the temperature at point P_0 in the pure B₄C becomes only 1076 K within the same time. The FE analysis results demonstrate that the lower temperature gradient caused by the faster heat propagation in the B₄C-TiB₂ composites can be ascribed to the existence of the TiB₂ network in the B₄C matrix, which has a higher thermal conductivity. If the composites are used as neutron-absorbing pellets in nuclear reactors, their high thermal conductance can rapidly cool the local high temperature caused by the nuclear reactions and greatly reduce the thermal stress caused by the thermal gradient, which reduces the risk of structural failure.

To further determine the equivalent average parameters and average heat conducting characteristics of the B₄C-TiB₂ composite, the equivalent homogeneous models with all the homogenized model parameters, such as conductivity, specific heat and density, determined by the volume-weighted-average were implemented for the three cases of

different TiB₂ volume fractions (10.5%, 16.2% and 29.8%). Taking the 265 temperature of the middle point of the right-end boundary as the 266 assessment during simulation from 0.2 to 1 µs, the comparison of the 267 results calculated from real composite structure and equivalent 268 homogeneous models are summarized in Table 3. It can be seen that the 269 results of the equivalent homogeneous models are slightly overestimated 270 for the three cases, but the maximum relative error is less than 5%, which 271 indicates that the simple volume-weighted-average method can be used to 272 roughly predict the equivalent parameters and average heat conducting 273 characteristics of the B₄C-TiB₂ composite for variational volume fraction. 274 It demonstrates the expected conductivity enhancement brought by the 275 interconnectivity of the TiB₂ phases as well. 276 In real nuclear reactors, the thermal gradient He and 277 swelling-induced crack propagation in B₄C materials are mainly 278 responsible for the disintegration of B₄C pellets in fast neutron reactors 279 and in thermal neutron reactors, respectively. To investigate the effect of 280 the TiB₂ network (cages) on the crack propagation behavior in the B₄C-281 TiB₂ composites, Vickers indentation was employed to simulate the 282 external stress that was applied to the B₄C-TiB₂ composites. 283 The typical Vickers indentation marks on the surfaces of S1, S2, and 284 S3 are shown in Figs. 6(a), (b), and (c), respectively. The beginnings and 285 endings of the cracks are indicated by double-headed arrows. Note that in 286

the Fig. 6(c), the gradient of sunken area is steep, outline of the Vickers indentation impression is irregular, but four cracks of the indentation are obvious. Therefore, each crack length is measured from the end of the crack, and along the inverse direction of crack propagation to the point where crack origins from the steeply sunken area (i.e. the indentation impression area). At a load of 9.8 N, the length of the cracks on S1 range from 20.7 to 27.9 µm; those on S2 and S3 range from 12.8 to 20.3 µm and 6.8 to 13.5 µm, respectively. The statistical results of the cracks show that the average values are 22.9 ± 2.8 , 16.1 ± 4.3 , and 12.2 ± 3.8 µm for S1, S2, and S3, respectively. Fig. 7 presents the statistical results of the crack lengths on S1, S2, and S3, respectively. Evidently, the crack lengths decrease with increasing TiB₂ volume fraction. Thereby the statistical results prove that the TiB₂ phase distribution along the grain boundaries can impinge/block the crack propagation.

Figs. 8(a) and (b) present the enlarged SEM images of the crack propagation paths in S1 and S3, respectively. The white contrast in Figs. 8(a) and (b) are TiB₂ phases: the TiB₂ phases in Fig. 8(a) are isolated particles, which are distributed in the B₄C matrix; the TiB₂ phases in Fig. 8(b) are interconnected, thereby forming a network (cages). The solid arrow in Fig. 8(a) indicates clearly that the propagating crack departs from its original straight trajectory and curves along the B₄C/TiB₂ interface, thereby proving that the isolated TiB₂ islands in the B₄C matrix

can redirect the crack propagation. A similar phenomenon was reported in 309 [22]. In this study, the network (cages) topography in Fig. 8(b) clearly 310 shows that the crack propagates through the TiB₂ walls (dashed arrows 2 311 and 3) and extends along the interface between the B_4C and TiB_2 phases 312 (dashed arrow 1), which facilitates the energy release of the crack tip. 313 314 This is confirmed by the shorter crack length (12.2 \pm 3.8 μ m) of S3 than those of S1 (22.9 \pm 2.8 μ m) and S2 (16.1 \pm 4.3 μ m), as shown in Fig. 6. 315 It can be concluded that there are two underlying ways in which the 316 TiB₂ phases affect the crack propagation in B₄C-TiB₂ composites: first, 317 when the volume fraction of the TiB₂ phase is low, TiB₂ distributes in the 318 B₄C matrix as isolated particles (shown in Fig. 8(a)), and when the 319 propagation of a crack tip encounters a TiB₂ particle, the crack moves 320 along the B₄C/TiB₂ interface, prolongs the crack propagation length and 321 dissipates more energy, finally results in an improvement of the fracture 322 toughness of the B₄C-TiB₂ composites. Similar dissipation mechanisms 323 at the B₄C/TiB₂ interfaces are also found in Refs. [23, 24]. Second, when 324 the volume fraction of the TiB₂ phase is high, the TiB₂ phase forms an 325 enclosed cage-like structure that encapsulates the B₄C grains (Fig. 8(b)). 326 Thus, the cracks have to penetrate the TiB₂ walls before extending along 327 the B₄C/TiB₂ interfaces or must propagate through the neighboring B₄C 328 grains. According to the model of Cook et al. on crack propagation in 329 brittle systems [25], additional energy is required to deflect the crack 330

- propagation and initiate secondary crack propagation across the interface;
- the primary crack requires much energy.
- To quantitatively measure the fracture toughness of the B₄C-TiB₂ composites, following equations are used [26]:

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$$K_{IC} = 0.016(E/H_V)^{0.5}Pc^{-1.5}, \tag{4}$$

336 where

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$$H_V = \frac{1.8544 \times P}{4a^2 \times 1000},\tag{5}$$

- in which E is the Young's modulus, H_V the Vickers hardness, P the 338 Vickers indentation load, and a and c are the half length of the residual 339 Vickers indenter impression diagonal and half length of the radial crack 340 diagonal, respectively. The fracture toughness of the composites K_{IC} 341 were calculated to be 3.06 ± 0.28 (MPa·m^{1/2}), 3.81 ± 0.33 (MPa·m^{1/2}), and 342 4.38 ± 0.67 (MPa·m^{1/2}) for S1, S2, and S3, respectively. In other words, 343 increasing the TiB₂ volume fraction and optimizing the distribution of 344 TiB₂ can impinge/block the propagation of cracks and improve the 345 fracture toughness of the B₄C-TiB₂ composites. This promotes the 346 integration of B₄C pellets and improves the CR performance in the harsh 347 environments of nuclear reactors. 348
- He behavior of the composites under irradiation conditions is important for the evaluation of the composites serving in the reactors, however, it was seldomly studied. Therefore, we conducted He ions irradiation on the composites and initially investigated the He behavior by

TEM observation.

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The TEM images in Fig. 9 show the He bubble distribution and morphology in B₄C phase of samples S1, S2, and S3. The images were recorded with the same underfocus of 2.8 µm. The He bubbles, which are small bright dots in the images, are located within a band area. The width of He bubbles deposition band is smaller than that of the implanted band as calculated by SRIM, similar phenomenon is observed by Motte et al. [8]. It is suggested that this is due to the different depths of the He ions induced damage profile and the He ions implanted profile. In the area of the damage profile, lot of vacancies were introduced by energetic ions collision, while in the area of He atoms implanted band, He atoms greatly outnumbered the vacancies. In the overlapped areas (the width of which was smaller than the implanted band as calculated by SRIM), abundant vacancies and He atoms promoted the formation of He-vacancy clusters and increased the density of He bubble germs. After annealing, He-vacancy clusters evolved into visible He bubbles under TEM observation and formed a bubbles band in the overlapped area. Moreover, Gillet et al. found that the dissociation energy of He-vacancy is $\sim 2 \text{ eV}$, the high energy barrier inhibited the He diffusion by the formation of He-vacancy clusters [27], and this explained that no He bubbles outside the band area can be found (excepted the He bubbles in the grain boundaries).

375	The He bubbles in the bubbles band are small and dense; such
376	feature has been argued to arise partially from the diffusion behavior of
377	He atoms in B ₄ C. According to Refs. [27-30], He atoms must overcome a
378	high energy barrier to migrate even through the most probable migration
379	paths owing to the intrinsic chemical bond and lattice structure of B ₄ C.
380	Schneider et al. discovered that He atoms prefer to migrate between <111
381	planes by first-principle calculations. Moreover, they discovered that the
382	activation energy of He ion diffusion is approximately 2 eV [28], which is
383	consistent with the experimental results in [30]. The 2 eV migration
384	energy means that the aggregation of He atoms in B ₄ C materials is
385	retarded and explains the observation of small-sized He bubbles in this
386	study rather than large-sized He bubbles in metals under similar
387	irradiation conditions [31].
388	The He bubbles that are distributed along the interfaces are bigger
389	than the bubbles distributed in the B ₄ C and TiB ₂ grains (however, bubbles
390	along the TiB ₂ -TiB ₂ interface are not investigated in this study), as shown
391	in Fig. 10. This is due to the special structure of the interface. Misfits in
392	the lattice structure of the interfaces contain plenty of open volume [32],
393	which provides adequate free space for the deposition of
394	irradiation-induced point defects (e.g., vacancies, interstitials, transmuted
395	elemental atoms) [33]. Owing to the high binding energy between

vacancies and He atoms (~ 2 eV [27,28]), the open volume of interfaces

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can act as efficient trapping sites for He and preferential nucleation sites for He bubbles [8,9,34]. Therefore, once the He atoms are captured by the interfaces, they cannot easily be released within the temperature range investigated in this study. As shown in Fig. 10, no He bubbles depleted zone was detected, which is consistent with the results of Motte et al.'s work [28] that it was hard for He atoms to diffuse through or in the B₄C grain boundaries below 1373 K. This might prove that long-range He diffusion is negligible at 1273 K. Moreover, TEM image in Fig. 10 shows that He bubbles deposition band in B₄C is closer to the surface than that in TiB₂, this phenomenon is also consistent with SRIM calculation as shown in Fig. 1 that the overlap of He distribution and damage profile is shallower in B₄C than that in TiB₂. In a realistic environment of nuclear reactors, increasing the density of sinks for the precipitation of He atoms is an effective way to suppress the formation of large He bubbles [35], and reduce the He-induced property degradation (e.g. He embrittlement). Therefore, the network (cages) of the TiB₂ phases provides numerous sinks for He precipitation, which might inhibit the formation of large He bubbles, and mitigate the effects of He-accumulation-induced shear stress at interfaces, and the resulting microcracks, especially in the thermal neutron reactors. However, other factors may affect performance of the B₄C-TiB₂

composites, such as shear stress induced by He accumulation at the

interfaces, different thermal dilation between B_4C and TiB_2 . To have a better evaluation of the B_4C - TiB_2 composites in a realistic environment, a detailed numerical evaluation will be carried out in the future study.

As an important component of the B_4C -diborides composites, the behavior of diborides under irradiation environment is very crucial to the performance of composites. Cheminant et al. conducted neutron irradiation on HfB_2 , and found that porosity appeared at grain boundaries, and resulted in a decrease of the thermal diffusivity [36], which lead to the degradation of B_4C - HfB_2 composites. Regarding the TiB_2 in this study, despite the inimitable merits of TiB_2 , however, its irradiation behavior was seldomly studied, which is very important for the application of TiB_2 as neutron absorbers. To the best of our knowledge, only two studies reported the microstructural evolution of TiB_2 under irradiation conditions [37, 38]. Therefore, a systematic study on the irradiation tolerance of TiB_2 is needed, which will be part of our future work.

3. Conclusion

To improve the performance of B_4C neutron absorption material in nuclear reactors, a set of property-optimized B_4C -TiB₂ composites were designed and fabricated through a combination of the molten-salt and SPS methods. The TiB₂ phase distributes along the B_4C grain boundaries and forms a connected network (cages), which encapsulates the B_4C

matrix phase. The thermal conductivity and fracture toughness of the B_4C - TiB_2 composites are greatly improved owing to the special cage-like structure. The FE analyses show that the connected TiB_2 network promotes the heat transfer and improves effectively the overall thermal conductivity of the B_4C - TiB_2 composite. The improved thermal conductivity results in a lower thermal stress, which mitigates thermal-stress-induced crack propagation.

Vickers indentation tests were performed on the composites to study the crack propagation behavior. The TiB_2 network (cages) can effectively impinge/block the crack propagation, thereby improving the composite fracture toughness. This is important for maintaining the integration of B_4C – TiB_2 composites if the composite serves as neutron absorption material in nuclear reactors.

Moreover, He ion irradiation was used to investigate the He behavior in the composites and to simulate the situation in which a high quantity of He is produced by the B^{10} (n, α) Li⁷ nuclear reaction. The TEM results show that the interfaces of B_4C and TiB_2 are effective trapping sites for He atoms and nucleation sites for He bubbles. Note that in a fast neutron reactor, the ballistic damage induced by energetic neutrons scattering and the nuclear reaction products (e.g., He, Li) scattering are much higher than the sole He ions implantation, this leads to a higher vacancy/He ratio and the possible consequences on the defects dynamics [28].

Overall, this study provides a synthesis method for fabricating neutron absorption materials for the application in nuclear reactors. The special property-optimized B_4C-TiB_2 composites with designed microstructure exhibit a better resistance to disintegration under irradiation and in the thermodynamic environment of nuclear reactors. In addition, they have a great potential for the application as neutron-absorbing pellets in nuclear reactors.

In the future study, other issues will be addressed to have a comprehensive evaluation of $B_4C\text{-}TiB_2$ composites under irradiation conditions. For example, helium behavior in TiB_2 and irradiation resistant property of TiB_2 , effects of the special cage-like structure and TiB_2 content on He diffusion and release during annealing, other synthesis routes to optimize the structure and properties of the $B_4C\text{-}TiB_2$ composites, influence of He bubbles aggregation along the $B_4C\text{-}TiB_2$ and $TiB_2\text{-}TiB_2$ interface on the thermal conductivity of the composites and the crack propagation behavior, etc.

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- 486 Data availability
- The raw data required to reproduce these findings cannot be shared at
- 488 this time as the data is part of an ongoing study. The processed data
- required to reproduce these findings cannot be shared at this time as the
- 490 data is part of an ongoing study.
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573 26-39.

- 575 Figure captions
- 576 Fig. 1. Displacement damage and He concentration as a function of
- penetration depth in B_4C and TiB_2 material, respectively.
- 578 Fig. 2. SEM images of surface morphology of (a) 10.5 Vol.% TiB₂
- composites, (b) 16.2 Vol.% TiB₂ composites, and (c) 29.8 Vol.% TiB₂
- 580 composites; arrows indicate the TiB₂ phases.
- Fig. 3. Geometric construction and boundary conditions of FE models: (a)
- pure B₄C model; (b)-(d) B₄C-TiB₂ composite model based on SEM
- 583 image of sample S1, S2 and S3, respectively; light and dark contrasts
- represent TiB₂ network and B₄C matrix, respectively; P_i (i=0, 1, 2 and 3)
- represents middle point of the right-end boundary of each calculated
- 586 domain.
- Fig. 4. Temperature distribution in pure B₄C and B₄C–TiB₂ composites
- with different TiB₂ volume fractions at different time scales calculated by
- FE method: (a) pure B₄C; (b)-(d) B₄C-TiB₂ composites of sample S1, S2
- and S3, respectively.
- Fig. 5. Temperature variation at points P_i (center of the right-end
- boundary in each model) with respect to time.
- Fig. 6. Typical Vickers indentation impressions on (a) 10.5 Vol.% TiB₂
- composites, (b) 16.2 Vol.% TiB₂ composites, and (c) 29.8 Vol.% TiB₂

- composites. The double arrows indicate the starting to the ending of the cracks.
- Fig. 7. Statistical results of Vickers indentation-induced crack length for
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- Fig. 8. SEM images of Vickers indentation-induced crack propagation; solid arrows in (a) indicate deflection of cracks by isolated TiB₂ particles, and dashed arrows in (b) indicate crack penetration through TiB₂ cages.
- Fig. 9. TEM images of He bubble morphology in He deposition band in B₄C phase of (a) 10.5 Vol.% TiB₂ composites, (b) 16.2 Vol.% TiB₂ composites, and (c) 29.8 Vol.% TiB₂ composites.
- Fig. 10. TEM image of He bubble morphology in He deposition band and at B₄C-TiB₂ interfaces. The arrows indicate the incident direction of He ions. The dash lines indicate the boundaries of He bubbles band.

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Table 1. Molar ratio of raw B₄C powder and Ti powder for molten-salt reaction, volume fraction of TiB₂, weight percent of TiB₂, B₄C and C in synthesized bulks.

Samples	Molar ratio (B ₄ C:Ti)	Volume fraction of TiB ₂	Weight percent of TiB ₂	Weight percent of B ₄ C	ent of Weight percent of C	
S1	6:1	10.5%	18.0%	80.4%	1.6%	
S2	4:1	16.2%	26.0%	72.0%	2.0%	
S3	2:1	29.8%	44.0%	52.0%	4.0%	

Table 2. Material properties of B₄C and TiB₂ used in heat transfer

analysis [16, 19]

		ctivity	Specific Hea	Density (kg/m ³)		
Temperature (K)	773	1273	773	1273	773	1273
B ₄ C	20.00	11.00	1680.80	2072.68	2550	2550
TiB_2	81.00	78.10	1073.00	1186.00	4450	4390

Table 3. Comparison of the results calculated from real composite structure models (CM), equivalent homogeneous models (HM) and the relative error (PE)

617	relative error	(RE).
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Time		S1			S2			S 3	
(µs)	CM(K)	HM(K)	RE(%)	CM(K)	HM(K)	RE(%)	CM(K)	HM(K)	RE(%)
0.2	819	836	2.1	832	856	2.9	867	901	3.9
0.4	917	948	3.4	941	981	4.3	997	1043	4.6
0.6	996	1033	3.7	1024	1069	4.4	1093	1129	4.2
0.8	1057	1093	3.7	1085	1128	4.0	1142	1180	3.3
1	1102	1137	3.2	1130	1168	3.4	1181	1212	2.6

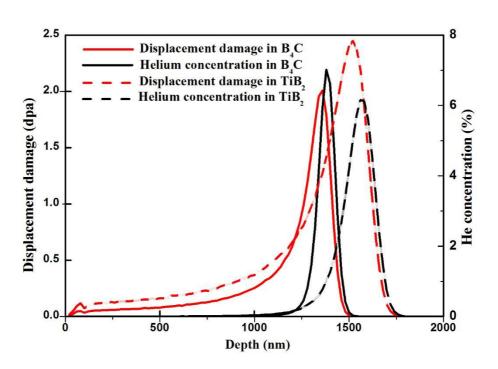


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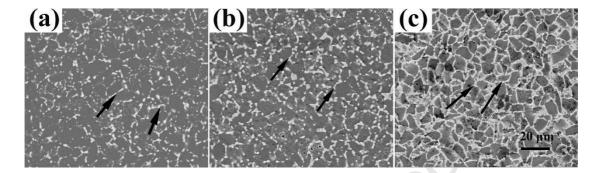


Fig. 2. SEM images of surface morphology of (a) 10.5 Vol.% TiB₂ composites, (b) 16.2 Vol.% TiB₂ composites, and (c) 29.8 Vol.% TiB₂ composites; arrows indicate the TiB₂ phases.

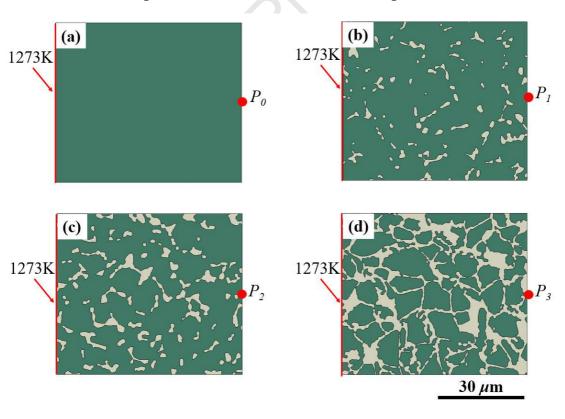


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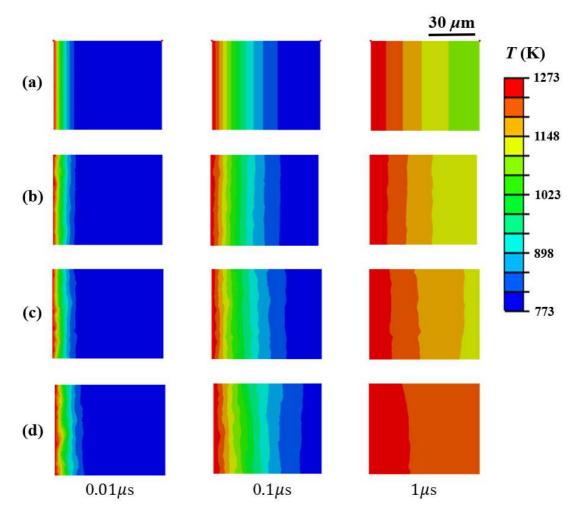


Fig. 4. Temperature distribution in pure B_4C and B_4C – TiB_2 composites with different TiB_2 volume fractions at different time scales calculated by FE method: (a) pure B_4C ; (b)-(d) B_4C – TiB_2 composites of sample S1, S2 and S3, respectively.

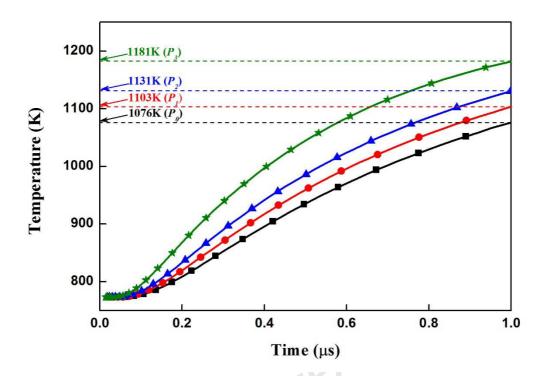


Fig.5. Temperature variation at points P_i (center of the right-end boundary in each model) with respect to time.

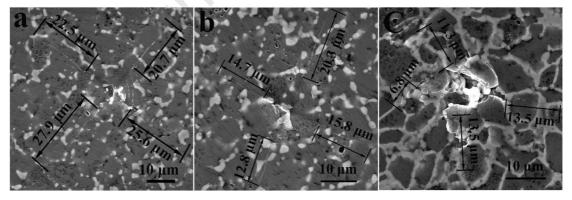


Fig. 6. Typical Vickers indentation impressions on (a) 10.5 Vol.% TiB_2 composites, (b) 16.2 Vol.% TiB_2 composites, and (c) 29.8 Vol.% TiB_2 composites. The double arrows indicate the starting to the ending of the cracks.

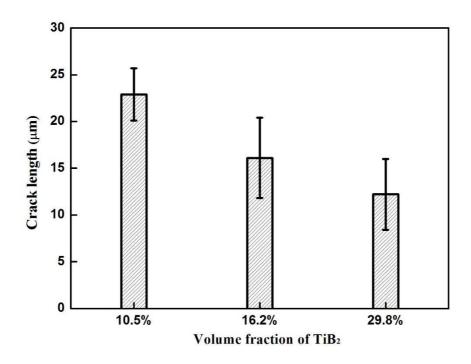


Fig. 7. Statistical results of Vickers indentation-induced crack length for different TiB₂ volume fractions of composites.

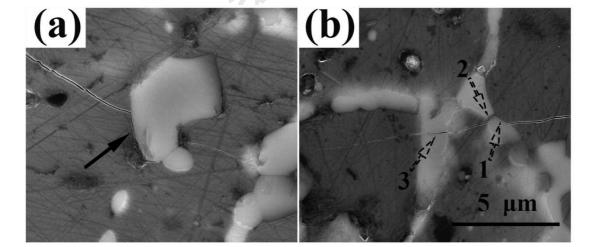


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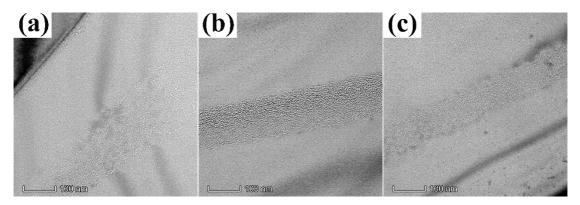


Fig. 9. TEM images of He bubble morphology in He deposition band in B_4C phase of (a) 10.5 Vol.% TiB_2 composites, (b) 16.2 Vol.% TiB_2

composites, and (c) 29.8 Vol.% TiB_2 composites.

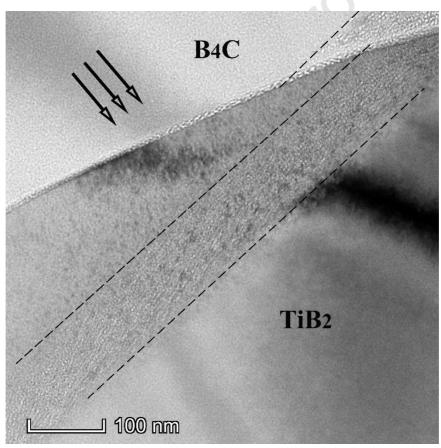


Fig. 10. TEM image of He bubble morphology in He deposition band and at B_4C – TiB_2 interfaces. The arrows indicate the incident direction of He ions. The dash lines indicate the boundaries of He bubbles band.

 B_4C -Ti B_2 composites with B_4C grains encapsulated by Ti B_2 network (cages) are fabricated.

 TiB_2 network (cages) greatly improve thermal conductivity and fracture toughness of B_4C – TiB_2 composites.

The interconnected interfaces between TiB_2 network (cages) and B_4C grains act as effective sinks for the aggregation of He atoms.

Declaration of interests
\square The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.
☐ The authors declare the following financial interests/personal relationships which may be considered as potential competing interests: