

Chemically vapor deposited Eu $3+:Y \ge 0$ 3 thin films as a material platform for quantum technologies

Nao Harada, Alban Ferrier, Diana Serrano, Mauro Persechino, Emrick Briand, Romain Bachelet, Ian Vickridge, Jean-Jacques Ganem, Philippe Goldner, Alexandre Tallaire

▶ To cite this version:

Nao Harada, Alban Ferrier, Diana Serrano, Mauro Persechino, Emrick Briand, et al.. Chemically vapor deposited Eu $3+:Y\ 2$ O 3 thin films as a material platform for quantum technologies. Journal of Applied Physics, 2020, 128 (5), pp.055304. 10.1063/5.0010833 . hal-02913151

HAL Id: hal-02913151 https://hal.science/hal-02913151v1

Submitted on 23 Nov 2020

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers. L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

Chemically Vapour Deposited Eu³⁺:Y₂O₃ thin films as a material platform for quantum technologies

Nao Harada¹, Alban Ferrier^{1, 2*}, Diana Serrano¹, Mauro Persechino¹, Emrick Briand³, Romain Bachelet⁴, Ian Vickridge³, Jean Jacques Ganem³, Philippe Goldner¹ and Alexandre Tallaire¹

¹Chimie ParisTech, PSL University, CNRS, Institut de Recherche de Chimie Paris, 75005 Paris, France

² Sorbonne Universités, Faculté des Sciences et Ingénierie, UFR 933, 75005 Paris, France

³ INSP, Sorbonne Universités, UPMC Université Paris 6, CNRS, UMR 7588, Paris, France

⁴ INL, Université de Lyon, Ecole Centrale de Lyon, CNRS UMR 5270, 69134 Ecully, France

Rare earth ions hosted in solids are good candidates for quantum technologies due to their chemical stability and optical and spin transitions exhibiting long coherence lifetimes. While bulk oxide crystals are usually the preferred host material, the development of a scalable silicon-compatible thin film platform would be desirable. In this paper, we report on the growth of $Y_{2(1-x)}Eu_{2x}O_3$ thin films on silicon in the full range of Eu^{3+} concentration by direct liquid injection Chemical Vapour Deposition. Our sub-micron polycrystalline films with a strong-(111) texture were grown for all compositions into the bixbyite cubic phase. The variation of growth rates with temperature and flow indicated that deposition occurred through a mass-transport controlled regime. Optical assessment of the Eu-doped thin films showed inhomogeneous linewidths as narrow as 50 GHz and fluorescence lifetimes of 1ms for the lowest concentrations. Finally, a spectral hole was successfully burned in a 200 nm-thin film with a 2 % Eu doping leading to a homogeneous linewidth of 11 MHz. These values are still below those reported for bulk single crystals indicating that additional decoherence mechanisms exist in such nanometric films, which might be alleviated by further improvement of the crystalline quality. Nevertheless, these results pave the way to the use of CVD-grown $Eu:Y_2O_3$ thin films as a platform for integrated quantum devices.

1-Introduction

Quantum technologies (QT) are witnessing an increasing interest with the promise to offer new systems and devices with unrivalled performance: ultra-sensitive sensors, quantum computers, information processing, highly-secured communication networks etc¹. Often, these achievements are conditioned by the availability of custom-made materials with exceptional control over the purity as well as defect positioning at the atomic scale. Adopting a stable solid-state, scalable and efficient platform for hosting the required qubits would be advantageous for these devices. Among the various candidates, rare-earth ions (REI) are outstanding as they exhibit extremely narrow optical homogeneous linewidths at low temperature, in the range of a few kHz to less than 100 Hz in bulk single crystals². This behaviour stems from their particular electronic configuration in which 4f electrons are shielded from the environment by outer electronic shells³. Optical coherence lifetimes of up to T₂ = 2.6 ms (homogeneous linewidth $\Gamma_h = 1/\pi T_2 = 122$ Hz) have been reported for Eu^{3+} :Y₂SiO₅ at cryogenic temperatures² while ¹⁵¹Eu³⁺ spin transitions that can be optically addressed, have shown coherence lifetimes up to 10's of ms⁴ or even hours⁵. REI can be chosen with optical transitions in a wide range of wavelengths, including some that might be relevant for optical fibre telecommunications. In addition, they can be hosted in a variety of matrices and are easily interchangeable due to their chemical similarity. All these properties have for example allowed demonstrating quantum storage⁶ and quantum state teleportation⁷ with such systems.

Single crystals produced by crystal pulling techniques such as Czochralski-grown Y₂SiO₅, are the preferred host for manipulation of large ion ensembles since they provide a well-controlled environment^{8,9}. These bulk materials are however costly to produce especially when isotopic enrichment is needed. Due to their size, they also do not provide a platform as appropriated as nano-materials when it comes to integrating REI emitters into other nanoscale systems like high finesse and small volume optical cavities¹⁰. Such systems can enhance RE interactions with light, a pre-requisite to detect the weak emission from single ions^{11,12}. Manipulating single emitters in nanomaterials could indeed potentially unleash quantum processing and quantum computing applications¹³. Recently, we have shown that optical homogeneous linewidths as narrow as 30 kHz¹⁴ can be obtained in specially prepared yttrium oxide (Y₂O₃) nanocrystals doped with Eu³⁺ ions, the lowest ever measured for nanoscale solids^{15,16}. While this value remains above that of bulk crystals, integration of these nanoparticles into an optical fibre cavity could be successfully obtained¹⁰.

Single crystalline thin films constitute an alternative approach. They could possibly allow one to benefit from the flexibility of multi-layers of variable thicknesses with spatially localized emitters and compatibility with scalable platforms like silicon^{17–20}. Integration into a device or coupling to other quantum materials in a hybrid system^{21,22} would thus be greatly facilitated. In the context of QT and high resolution optical spectroscopy, single crystalline films with high purity and low defect content are preferred to achieve an optimized environment for RE ions in which strain, structural defects and electrical or magnetic noise are as limited as possible, possibly leading to extended coherence properties. Indeed, in polycrystalline or even amorphous films, the remaining disorder is expected to induce additional inhomogeneous broadening of the optical linewidths. Epitaxial thin films made by Molecular Beam Epitaxy (MBE) or lifted-off from bulk crystals have been recently investigated and showed promising results^{18,23,24}. Chemical methods have also been explored. Atomic

Layer Deposition (ALD) has been used to deposit thin optically active Eu or Er-doped Y_2O_3 films with precise thickness control in the 10-200 nm range and high conformity²⁵. Regarding optical properties of the emitters, encouraging results have been recently obtained with inhomogeneous linewidth of 200 GHz and 1.5 ms lifetime for the infrared transition of Erdoped ultra-thin films (10 nm)²⁶. However, ALD is limited to a restricted low-temperature window that does not easily allow for the synthesis of epitaxial oxide films. High temperature post-treatments (950-1050 °C) can partially improve the structural order, but they lead to the formation of parasitic interfacial phases²⁷ as well as unwanted dopant diffusion²⁸ which may be a limiting factor especially on silicon substrates. Chemical Vapour Deposition (CVD) offers an interesting approach for the deposition of high quality epitaxial RE oxide films due to the possibility to grow under a more flexible range of conditions and higher temperatures²⁹. Eu³⁺:Y₂O₃ films have already been deposited using standard metal-organic CVD on sapphire, and inhomogeneous and homogeneous optical linewidths have been reported for films that are several microns thick and high temperature annealed³⁰.

In this work, we used a variant of the CVD technique involving Direct Liquid Injection (DLI) with metal-organic precursors. This technique allow to partially overcomes the issue of low volatility of heavy element precursors as REI ones^{31–33} and helps having a better control over their constant feeding inside the reaction chamber. Europium ions doped in Y₂O₃ were chosen among other REI in this study because they exhibit ultra-narrow optical homogeneous linewidths and long coherent times while the matrix is a simple cubic binary oxide. Moreover, using this growth technique, almost any other REI or even oxide matrix could be prepared as long as suitable precursors in terms of temperature stability and volatility are outsourced. Deposition was performed on silicon for compatibility with standard wafer processing and silicon-on-insulator waveguide technologies³⁴. A range of deposition conditions were explored, and highly textured polycrystalline thin films were obtained with a good control over a wide composition range. We provide an assessment of their optical properties and evaluate their relevance to the development of scalable platforms for quantum technologies.

2- Experimental details

To deposit the films, we used a home-made DLI-CVD set-up composed of a double injection and evaporation head (KEMSTREAM Vapbox1500) and a horizontal tube furnace (*Carbolite*) (Fig. 1). High purity (3N-99.9 %) solid precursors from the β -diketonate family (Fig. 1a, Y(TMHD)₃ and Eu(TMHD)₃) purchased from *Strem Chemicals* were dissolved in mesitylene (purity of 99.9%) to obtain concentrations in the range of 0.1-0.01 mmol/L and stored in pressurized tanks. The liquid solution was mixed to a carrier gas (300 sccm of Ar) and brought to the evaporation head maintained at 220 °C where it was sprayed into tiny droplets and flash-evaporated (Fig. 1b). Composition of the films was adjusted by monitoring the opening time and frequency of the two injectors (one for Y and one for Eu). The injectors can be controlled independently allowing a precise control of the incoming molar flow. The gas mixture was then carried through a heated line towards a tubular quartz tube furnace at 750 °C in which Si wafer pieces $(10 \times 10 \text{ mm}^2)$ were placed. An additional oxygen flow (150 sccm) was introduced inside the tube. The gases were pumped with a rotary vane pump allowing the pressure to settle at around 10 mbar. Unreacted precursors and chemicals were collected in a nitrogen-cooled trap placed downstream the chamber. Prior to deposition, asreceived silicon (100) substrates were cleaned using the standard Radio Corporation of America (RCA) protocol. This includes: first a (1:1:5) solution of (NH₄OH:H₂O₂:H₂O) then a 5 % HF solution and finally a (1:1:5) solution of (HCI: H_2O_2 : H_2O). Each step is followed by a deionised water rinsing. This procedure allows creating a 4 nm-thin, reproducible and stable chemical oxide at the surface of the silicon wafer.



Fig. 1. (a) Schematics of the DLI-CVD set-up used in this study and that uses specially developed liquid injectors. The reaction chamber consists of a horizontal quartz tube furnace. (b) Chemical representation of $Y(TMHD)_3 \beta$ diketonate precursor used.

Spectroscopic ellipsometry (*Woollam iSE*) in the range 400-1000 nm was used to evaluate film's thickness. The ellipsometric angle Ψ and phase difference Δ (see supplementary information file, Fig. S1) were recorded at an incidence angle of 65°. The optical refractive index of yttria was obtained using a Cauchy model.

The crystallinity of the films was evaluated using X-ray Diffraction (XRD) in a *PanalyticalXPert Pro* diffractometer with a Ge monochromator. Additional high-resolution (HR) XRD characterisations were performed using a high-brilliance *Rigaku SmartLab* diffractometer with a 9 kW copper rotating anode and a 2 bounces Ge(220) monochromator.

Chemical composition of the films was obtained by Rutherford Backscattering Spectroscopy (RBS). RBS was performed with a Van de Graaff accelerator, ⁴He⁺ particles at 1.8 MeV, 100 nA and a scattering detection angle of 165 °. The areal density of atoms was determined using a reference Si sample implanted with a controlled dose of ²⁰⁹Bi ions (see supplementary information file, Fig. S3). The spectra were acquired for a 10 μ C dose of incident ions for each sample. For Eu concentration below 1 % the doping level could not be precisely assessed by RBS and the nominal doping value was used.

Field Effect Gun Scanning Electron Microscope (FEG-SEM) images of the surface of the films were recorded using a ZEISS Leo 15030 microscope. Micro photoluminescence (μ PL) measurements were performed using a Renishaw InVia apparatus with a 50 × objective and a 532 nm laser as the excitation source. Decay measurements at room temperature were also recorded using a tunable optical parametric oscillator pumped by a Nd:YAG Q-switched laser

(*Ekspla NT342B-SH*) with 6 ns pulse length, an *ACTON SP2300i* monochromator and a photomultiplier tube. The detection time constant was between 50 µs and 200 µs and data was recorded on a digital oscilloscope (*LeCroy 332A*).

Inhomogeneous linewidths of the ${}^{7}F_{0} \rightarrow {}^{5}D_{0}$ transition were measured in reflection mode using a closed-cycle cryostat (*Janis CTI-Cryogenics Model CCS-150*) with the temperature monitored by a *LakeShore model 330* controller. The ${}^{7}F_{0} \rightarrow {}^{5}D_{0}$ transition inhomogeneous linewidths were probed by monitoring the fluorescence intensity of the ${}^{5}D_{0} \rightarrow {}^{7}F_{2}$ transition at 612 nm as a function of the excitation wavelength around 580.883 nm. Excitation was provided by a single mode dye laser (*Coherent 899-21*, 1 MHz linewidth) pumped by a semi-conductor laser (*Coherent Verdi G10*). Red fluorescence was collected by a 75 mm diameter lens and focused on a photomultiplier tube after suitable filtering to eliminate scattered laser light.

For spectral hole burning (SHB) experiments, samples were cooled down to liquid helium temperatures in a bath cryostat (*Janis SVT 200*). A continuous wave narrow linewidth (250 kHz) dye laser (*Sirah Matisse DS*) provided the excitation around 580 nm ($^7F_0 \rightarrow {}^5D_0$ transition), which was modulated in time and frequency using an acousto-optic modulator driven by an arbitrary waveform generator (*Agilent N8242A*). Holes were probed by monitoring the red fluorescence (${}^5D_0 \rightarrow {}^7F_2$) at 612 nm, which was collected using a high numerical aperture lens mounted on a piezo-driven mount and located inside the cryostat. Fluorescence was filtered using low-pass and interferential filters and sent to a high sensitivity photomultiplier tube (*Hamamatsu R10699*). Further details on the SHB setup used and the procedure are given in the supplementary information file (Fig. S5).

3- Results

3.1- Structural and morphological characterization of the Y2O3 thin films

Undoped Y₂O₃ thin films were first grown on (100) silicon wafers in order to investigate the deposition process by DLI-CVD. By adjusting the deposition time while keeping all other deposition parameters constant, films with thicknesses from 90 to 650 nm as evaluated by ellipsometry (see supplementary information file, Fig. S1) were obtained and observed by SEM. The resulting morphologies are presented in Fig. 2a-c. The films show a polycrystalline structure with an increasing but uniformly distributed grain size from about 20 to 80 nm when increasing thicknesses from 50 to 650 nm and with an obvious saturation of grain coarsening above 400 nm (Fig. 2d). This behaviour is typical from a columnar growth mechanism³⁵ where nucleated grains competitively grow depending on their faster growth direction until they reach a steady-state.

In Fig. 3a, growth rates are plotted as a function of temperature. Below 500 °C no growth is observed due to the well-known limited reactivity of the β -diketonate precursors with molecular oxygen³⁶. The use of more powerful oxidizing agents such as ozone would be a possible way to increase reactivity . At higher temperatures, three regions can be distinguished indicating the different CVD growth modes³⁷. From the slope of the growth rate dependence with temperature, one can extract the activation energy of the reaction from an Arrhenius law:

$$\ln(GR) = A - \frac{E_a}{RT}$$

where GR is the growth rate, A is a constant, E_a is the activation energy and R is the universal gas constant. In region I (500-650 °C), growth rates strongly depend on the temperature with an estimated activation energy of 1.2 kJ/mol. This is characteristic of a kinetic regime where

surface reactions are the limiting process. In region II (650-1000 °C), growth rates are less dependent on temperature and activation energy reduces to 0.14 kJ/mol. This indicates that mass transport, i.e. the ability to carry the reactants towards the substrate is the main limiting step. Finally, in region III (T > 1000 °C) the trend is inverted and growth rates continuously decrease with temperature. This can possibly indicate pre-reactions of the precursors that occur before reaching the substrate and that limit the deposition process. This sets the upper temperature bound for this particular CVD system and chemistry.

To confirm that in region II growth rates can be controlled by the supply of the metallic cation precursor, a series of growth runs, in which the $Y(TMHD)_3$ flow was varied from 3 to 14 µmol/min, was carried out at a constant temperature of 750 °C. A linear dependence was found from 0.2 to 0.9 µm/h without any apparent saturation (Fig. 3b). This confirms that mass-transport is the controlling step for growth at this temperature. Although this trend may seem trivial, it indicates that the design of the reactor and the growth conditions (total flow, pressure, concentration etc.) are suitable, avoiding pre-reactions between precursors or uncontrolled turbulent flow inside the chamber.



Fig. 2. (a) –(c) Film morphologies observed by SEM for different thicknesses (t) from 90 to 650 nm showing polycrystalline grains. (d) Grain size directly estimated from SEM images as a function of film thickness.



Fig. 3. (a) Growth rate dependence on temperature showing three distinctive regions. Activation energies are extracted from the slopes of the lines and are indicated. (b) Y₂O₃ film growth rate dependence as a function of the incoming precursor flow at 750°C. This confirms the mass transport regime for region II. The dotted line is a linear fit of the data.

In order to get more insight into the structural properties of the films, HR-XRD analysis was carried out on a 100 nm-thick pure Y_2O_3 film grown on Si (100). The θ -2 θ scan presented in Fig. 4a shows that the cubic phase of Y_2O_3 was obtained. The position of the peaks is consistent with a fully-relaxed film. The absence of the (400) diffraction peak and the presence of only (hhh) reflections indicate that the film is (111)-textured. The ω -scan of Fig. 4b acquired for the (222) lattice plane indicates a relatively high out-of-plane mosaicity of about 8° which shows that an important crystalline disorder exists. Preferential orientation along (111) has already been reported for Y₂O₃ polycrystalline films obtained by various techniques^{38,39}.Such texturing is a consequence of the minimization of free surface energy of the grains as well as a competitive growth between grains of different orientations as expected from a columnar growth model. We note that, although all films produced in this study displayed preferential texture, full (111)-texture was only obtained when a thin chemical silicon oxide layer (around 4 nm) was intentionally created by using an RCA cleaning procedure ending with an oxidation step. With as-received silicon substrates, a significant fraction of (100)-oriented grains was found (see supplementary information file, Fig. S2). This suggests that surface preparation of the substrate plays an important role in the texture formation. To assess the in-plane orientation of the films, a pole figure was also acquired on an asymmetric (400) reflection (Fig. 5). The presence of a concentric circle (Fig. 5a) rather than clear diffraction spots in the map underlines that the film was not in-plane oriented. Although no epitaxial growth was obtained here, the full (111)-texture and thus the preferential out-of-plane orientation as well as the relatively narrow diffraction peaks provide positive indicators for the optical properties, even though they do not correspond to specific incorporation sites for the REI. By using appropriate substrates, surface treatments and growth conditions the possibility to epitaxially grow the films on Si will be attempted as already reported for films grown by Molecular Beam Epitaxy (MBE)^{40,41}, Pulsed Laser Deposition (PLD)⁴² and e-beam evaporation⁴³ for example.



Fig. 4. HR-XRD analysis. (a) θ -2 θ scans showing that only the (111)-related diffraction peaks of cubic Y₂O₃ are observed. (b) ω scan around the (222) diffraction peak showing an 8° out-of-plane mosaicity. The solid red line is a Lorentzian fit from which FWHM is extracted.



Fig. 5. (a) Pole figure on the asymmetric (400) reflection from which no preferential in-plane orientation is evidenced. (b) and (c) Line scans at particular values of ϕ and χ .

3.2 Eu doping of the Y₂O₃ thin films

We then evaluated the ability to produce $Y_{2(1-x)}Eu_{2x}O_3$ films with a controlled doping and in a wide range of compositions (x = 0 to 1). To this end, a variable amount of Eu(TMHD)₃ was added together with Y(TMHD)₃ using the double injection head of the DLI-CVD set-up. The films were grown at 750°C in the mass-transport regime with a 6 µmol/min flow which is a good trade-off between the deposition's duration and thickness homogeneity on the substrate. To perform RBS measurements, thinner films of 60 nm were grown. Indeed, for too high thicknesses Eu and Y peaks widen and may overlap, further complicating their quantification. The composition of the films was determined using non-channelling conditions from the Eu/(Eu+Y) ratio of each related peak in the RBS spectrum (see supplementary information file, Fig. S3). It indicates that incorporation efficiency of Eu and Y in the films was similar. This could be expected since the use of identical ligand molecules in the precursor (TMHD) and the interchangeability of rare-earth elements that possess similar radii and properties are likely to lead to similar chemistry. We note that a systematic cleaning procedure of the injectors between each deposition run using mesitylene solvent was crucial to limit cross-contaminations. This approach allowed us to address a wide Eu range from 0 to 99% as deduced from RBS analysis. For the highest doping (99%), background yttrium contamination could not be entirely suppressed despite careful cleaning of the lines preventing reaching a completely pure Eu_2O_3 film. The crystal structure was then further analysed by XRD (Fig. 6a). The (222) peak position varied in between the reported values for Y_2O_3 (a = 10.60 Å) to Eu₂O₃ (a = 10.87 Å) cubic crystalline structures in a linear fashion (Fig. 6b). Full solid solution was therefore achieved in agreement with what is expected for cation substitution by the Vegard's law⁴⁴. Lanthanide sesquioxides present three major structural polymorphs: cubic bixbyite, trigonal, and monoclinic. Under ambient conditions, the thermodynamically stable phase of Eu_2O_3 is monoclinic unlike Y_2O_3 which is cubic. Interestingly we note that even for the highest Eu contents, the crystal structure remained in the cubic phase. This might be related to a stabilizing effect of yttrium on the cubic structure or to an influence of the cubic substrate underneath.



Fig. 6. Y_{2(1-x)}Eu_{2x}O₃ thin films. (a) Position of the (222) X-ray diffraction peak for different Eu contents measured by RBS. Vertical dotted lines are the expected values of the two rare-earth oxides in the literature^{45,46}. (b) Deduced lattice parameter from the peak position using Bragg's formula. The expected trend for a full solid solution between Y₂O₃ and Eu₂O₃ cubic phases are indicated by the red line.

4- Optical properties of the Eu:Y₂O₃ thin films

The optical properties of the films grown with different Eu concentrations (thickness around 200 nm) were assessed and benchmarked against Eu:Y₂O₃ transparent ceramics and nanoparticles already investigated in our lab. The μ PL spectra of films with a similar thickness and Eu content from 1 to 99 % as measured by RBS were recorded using a non-resonant 532 nm excitation laser (Fig. 7a). All spectra were normalized to the samples' thicknesses. For all compositions, a narrow emission band around 612 nm, corresponding to the ⁵D₀ \rightarrow ⁷F₂ Eu³⁺ transition in Y₂O₃ and characteristic of the cubic phase, is observed (Fig. 7b). Furthermore, although weaker, the fluorescence of europium ions in Eu₂O₃ cubic host is detected. The integrated intensity of this main emission was found to increase up to concentrations of about 4 % of Eu^{3+} (Fig. 7c) and then decreases due to concentration quenching, i.e. a reduction of the emission related to interaction between Eu and defects or impurities (Fig. 7c)⁴⁷. In addition, migration of the excitation energy over the Eu^{3+} sub-lattice increases the probability of quenching by impurities at higher concentrations. For bulk crystals, the critical concentration was found to be around 10 %⁴⁸.

The lifetime of the ${}^{5}D_{0}$ level was then assessed by probing the red fluorescent transition (${}^{5}D_{0} \rightarrow {}^{7}F_{2}$) following a resonant excitation of the ${}^{7}F_{0} \rightarrow {}^{5}D_{2}$ level at 465 nm (Fig. 7a). The curves in Fig. 6d show a single exponential decay from which the lifetime T₁ can be extracted. T₁ exhibits a maximum around 1 ms for the lowest concentrations and then reduces drastically down to about 36 µs for the highest Eu concentration (Fig. 7e). This is in good agreement with the concentration quenching previously observed in the fluorescence spectra. We note that at a nominal concentration of 1 % the signal was weak and the error in measuring the lifetime was much larger. Nevertheless this maximum lifetime value is comparable to that measured on transparent ceramics with similar Eu³⁺ doping⁴⁹ which is an encouraging indicator of performance of this thin film material.



Fig. 7. Fluorescence and lifetime measurements at room temperature as a function of Eu content estimated by RBS. (a) Energy level of Eu^{3+} in Y_2O_3 with the 612 nm transition that is monitored. (b) μ PL spectra acquired with a non-resonant 532 nm excitation. (c) Area of the 612 nm PL emission. (d) Fluorescence dynamics of the 5D_0 level obtained by resonantly exciting 5D_2 at 465 nm. (e) Lifetime values determined by fitting the experimental fluorescence decay curve with a single exponential.

Inhomogeneous linewidths (Γ_{inh}) were also evaluated as a function of Eu³⁺nominal concentration at a temperature of 10 K. To this end, emission of the ${}^{5}D_{0} \rightarrow {}^{7}F_{2}$ transition was monitored near 612 nm while the laser excitation frequency was scanned around a central frequency to resonantly probe the ${}^{7}F_{0} \rightarrow {}^{5}D_{0}$ transition (Fig. 6a). The results presented in

Fig. 8a indicate that, as the Eu content increased, Γ_{inh} broadened from a minimum of 50 GHz (at 0.25 % Eu) to a maximum of 325 GHz (at 50 % Eu). This behaviour is consistent with the fact that the 50 % doping level corresponds to the maximum statistical disorder in the crystal structure where one half of the cations are Y³⁺ and the other half are Eu³⁺. A broadening rate of about 6 GHz per percent of Eu is thus deduced. A blueshift of the line as a function of increasing Eu concentration is also observed, in agreement with the larger Eu ionic radius compared to Y one^{50,51}. The Γ_{inh} measured here is comparable although slightly narrower than that reported for ALD-grown and annealed thin films (around 80 GHz at 2 % Eu)²⁶ as well as CVD-grown and high-temperature annealed films on sapphire (around 90 GHz at 2 % Eu)³⁰. This relatively low value of inhomogeneous linewidth suggests that RE ions experience a fairly uniform crystalline environment. Nevertheless, at low Eu concentrations, Γ_{inh} of our films are significantly broader than that measured in high-quality nanoparticles annealed at 1200 °C (10 GHz at 0.3 %)¹⁵ or transparent ceramics⁴⁹ of the same composition. This larger contribution can be attributed to the presence of intrinsic defects such as dislocations or point defects (impurities, vacancies) in the films, indicating that the insertion of Eu³⁺ ions is not the main source of broadening at low doping levels. The Lorentzian rather than Gaussian shape of the $\Gamma_{\rm inh}$ line rather points towards impurities-induced broadening⁵². We indeed know that carbon from the precursors or silicon from the substrate might be present in the films. The use of annealing post-treatment at high temperature (> 1100 °C) could possibly help in relieving this issue as demonstrated in Eu:Y₂O₃ nanoparticles⁵³.



Fig. 8. Inhomogeneous linewidth measurement (a) Emission of the 612 nm line as a function of excitation laser frequency to probe the ${}^{7}F_{0} \rightarrow {}^{5}D_{0}$ transition for various Eu concentrations at 10 K. The solid line is a Lorentzian fit of the experimental data points. (b) Variation of the inhomogeneous linewidth with Eu content.

In order to get more insight into the coherent properties of RE ions in our CVD-grown Eu:Y₂O₃ thin films, we performed a SHB experiment at 2 K on a 200 nm-thick film with 2 % Eu doping. Spectral Hole Burning (SHB) is an optical method that can probe homogeneous linewidths and therefore the effect of dynamical fluctuations in the material on RE ion transitions³. In our experiments, the spectral hole is induced by a modification of the population in the long-lived hyperfine level of the ground state of europium. While at thermal equilibrium all the hyperfine levels are equally populated, a narrow laser excitation can change this population file, Fig. S6). This change of population is directly observed by a

decrease of the red fluorescence emission. Spectral hole width is as broad as twice the homogeneous linewidth provided that it is not increased by other effects like too high excitation power and laser jitter or drift.

The spectral hole burning sequence used in this work was composed of 3 different parts (Fig. 9a). A burn pulse of 20 ms at a fixed frequency (v), a shorter probe pulse at a variable frequency ($v + \Delta v$) and a series of strong pulses scanned over 200 MHz to erase the spectral hole by distributing the population equally between the three hyperfine levels of Eu. The measurement was performed by monitoring the fluorescence intensity after the weak probe pulse. Contribution from laser frequency instability and other hole broadening effects was determined using a Eu:Y₂SiO₅ single crystal in which the homogeneous linewidth is below 1 kHz. Here a minimal hole linewidth of 2 MHz was recorded, corresponding to the spectral resolution of our setup.

Using the sequence outlined above, we observed a clear spectral hole burning phenomenon with a FWHM of about 22 MHz (Fig. 9b), which gives a homogeneous linewidth of about $\Gamma_{hom} = 11$ MHz. This is similar to the value obtained on Eu³⁺:Y₂O₃ high temperature annealed CVD films of higher thickness³⁰ and deposited on sapphire. Nevertheless, homogeneous linewidths lower by almost two order of magnitude were measured on nanoscale systems (Eu:Y₂O₃ nanoparticles^{15,16}). The origin of this broadening might be connected with local disorder that leads to additional dephasing mechanisms known as Two Level System (TLS)³⁰. As discussed above, the large inhomogeneous linewidths observed in the films is a clear indication of such a disorder. Another homogeneous broadening mechanism suggested in nanoparticles is fluctuating electric fields due, for example, to charge noise on surfaces or grain boundaries^{15,16}. Decreasing these processes could be achieved by reducing defects and increasing grain or by combining high-temperature deposition and annealing, as well as other post-treatments¹⁴. This has proved efficient in obtaining linewidths below 100 kHz in Eu³⁺:Y₂O₃ nanoparticles^{15,53}. Epitaxially depositing the film could also potentially help reducing dephasing phenomena.



Fig. 9. (a) SHB sequence used. (b) SHB spectrum obtained when varying the detuning frequency between the burn and probe pulses. A distinct fluorescence loss is observed indicating an efficient hole burning. The solid line is a Lorentzian fit of the data.

5- Conclusion

Using DLI-CVD we successfully grew $Y_{2(1-x)}Eu_{2x}O_3$ oxide thin films on (100)-oriented silicon wafers in the full range of Eu content (0 < x < 1). The films obtained through a columnar growth mode were polycrystalline with full (111)-texture. High structural order was achieved for deposition temperatures in the range 650-1050 °C which corresponds to a mass-transport dominated regime. A good control of the solid solution composition was obtained with lattice parameters matching the expected values from a relaxed structure.

The optical properties of such thin films on Si were then assessed and benchmarked against other systems (nanoparticles and ceramics) within the specific context of quantum technologies. Optical spectroscopy showed that fluorescence from the ${}^{5}D_{0} \rightarrow {}^{7}F_{2}$ transition of Eu^{3+} ions had strongest emission intensity and longest decay time (1 ms) for low concentrations (below 4 %). At high concentrations, quenching of the luminescence occurred. The lowest inhomogeneous linewidth was found to be around 50 GHz which is a relatively low value for such a nanoscale system. Notably, a 22 MHz-wide hole was spectrally burned in a 200 nm-thin film on silicon, corresponding to a homogeneous linewidth of 11 MHz. Future efforts to reduce Γ_{hom} should be directed to the improvement of the structural quality of our Y_2O_3 thin films and in particular reduce the presence of grain boundaries as well as impurities and defects that can possibly act as a source of decoherence. Nevertheless, the first structural

and optical assessment indicates that CVD-grown $Eu: Y_2O_3$ thin films on silicon offer interesting perspectives towards their integration into quantum devices.

Supplementary material

See the supplementary material for a more complete description about thickness measurement by Spectroscopic Ellipsometry (SE), effect of silicon surface preparation, composition analysis by RBS, hole burning experimental setup and energy levels of europium.

Acknowledgements

This project has received funding from the European Union's Horizon 2020 research and innovation programme under grant agreement No 820391 (Square). This research work has received funding from the European Union's Horizon 2020 research and innovation programme under grant agreement No 712721 (NanOQTech). We acknowledge the financial support of the Paris network on quantum technologies (SIRTEQ), and the French EMIR-A network for provision of irradiation beam time using the SAFIR facility. The authors would also like to thank Hervé Guillon from *Kemstream* as well as Dominique De Barros for advice and technical support throughout this project.

Data availability

The data that supports the findings of this study are available within the article and its supplementary material.

Bibliography

¹ A. Acín, I. Bloch, H. Buhrman, T. Calarco, C. Eichler, J. Eisert, D. Esteve, N. Gisin, S.J. Glaser, F. Jelezko, S. Kuhr, M. Lewenstein, M.F. Riedel, P.O. Schmidt, R. Thew, A. Wallraff, I. Walmsley, and F.K. Wilhelm, New J. Phys. **20**, 080201 (2018).

² R.W. Equall, Y. Sun, R.L. Cone, and R.M. Macfarlane, Phys. Rev. Lett. **72**, 2179 (1994).

³ P. Goldner, A. Ferrier, and O. Guillot-Noel, in *Handb. Phys. Chem. Rare-Earth Mater.*, edited by Bunzli and Pecharsky (2015), pp. 1–78.

⁴ A. Arcangeli, M. Lovrić, B. Tumino, A. Ferrier, and P. Goldner, Phys. Rev. B **89**, 184305 (2014).

⁵ M. Zhong, M.P. Hedges, R.L. Ahlefeldt, J.G. Bartholomew, S.E. Beavan, S.M. Wittig, J.J. Longdell, and M.J. Sellars, Nature **517**, 177 (2015).

⁶ P. Jobez, C. Laplane, N. Timoney, N. Gisin, A. Ferrier, P. Goldner, and M. Afzelius, Phys. Rev. Lett. **114**, 230502 (2015).

⁷ F. Bussières, C. Clausen, A. Tiranov, B. Korzh, V.B. Verma, S.W. Nam, F. Marsili, A. Ferrier, P. Goldner, H. Herrmann, C. Silberhorn, W. Sohler, M. Afzelius, and N. Gisin, Nat. Photonics **8**, 775 (2014).

⁸ N. Timoney, B. Lauritzen, I. Usmani, M. Afzelius, and N. Gisin, J. Phys. B At. Mol. Opt. Phys. **45**, 124001 (2012).

⁹ E.Z. Cruzeiro, A. Tiranov, I. Usmani, C. Laplane, J. Lavoie, A. Ferrier, P. Goldner, N. Gisin, and M. Afzelius, Phys. Rev. B **95**, 205119 (2017).

¹⁰ B. Casabone, J. Benedikter, T. Hümmer, F. Oehl, K. de O. Lima, T.W. Hänsch, A. Ferrier, P. Goldner, H. de Riedmatten, and D. Hunger, New J. Phys. **20**, 095006 (2018).

¹¹ R. Kolesov, K. Xia, R. Reuter, R. Stöhr, A. Zappe, J. Meijer, P.R. Hemmer, and J. Wrachtrup, Nat. Commun. **3**, 1 (2012).

¹² T. Zhong, J.M. Kindem, J.G. Bartholomew, J. Rochman, I. Craiciu, V. Verma, S.W. Nam, F. Marsili, M.D. Shaw, A.D. Beyer, and A. Faraon, Phys. Rev. Lett. **121**, 183603 (2018).
 ¹³ T. Zhong and P. Goldner, Nanophotonics **8**, 2003 (2019).

¹⁴ S. Liu, D. Serrano, A. Fossati, A. Tallaire, A. Ferrier, and P. Goldner, RSC Adv. **8**, 37098 (2018).

¹⁵ J.G. Bartholomew, K. de Oliveira Lima, A. Ferrier, and P. Goldner, Nano Lett. **17**, 778 (2017).

¹⁶ D. Serrano, J. Karlsson, A. Fossati, A. Ferrier, and P. Goldner, Nat. Commun. 9, 2127 (2018).
 ¹⁷ H.J. Osten, A. Laha, M. Czernohorsky, E. Bugiel, R. Dargis, and A. Fissel, Phys. Status Solidi A 205, 695 (2008).

¹⁸ M.K. Singh, A. Prakash, G. Wolfowicz, J. Wen, Y. Huang, T. Rajh, D.D. Awschalom, T. Zhong, and S. Guha, APL Mater. **8**, 031111 (2020).

¹⁹ J. Paulraj, R. Wang, M. Sellars, and B. Luther-Davies, Appl. Phys. A **122**, 422 (2016).

²⁰ T. Tawara, H. Omi, T. Hozumi, R. Kaji, S. Adachi, H. Gotoh, and T. Sogawa, Appl. Phys. Lett. **102**, 241918 (2013).

²¹ K.J. Tielrooij, L. Orona, A. Ferrier, M. Badioli, G. Navickaite, S. Coop, S. Nanot, B. Kalinic, T. Cesca, L. Gaudreau, Q. Ma, A. Centeno, A. Pesquera, A. Zurutuza, H. de Riedmatten, P.

Goldner, F.J. Garcia de Abajo, P. Jarillo-Herrero, and F.H.L. Koppens, Nat Phys **11**, 281 (2015). ²² G. Kurizki, P. Bertet, Y. Kubo, K. Mølmer, D. Petrosyan, P. Rabl, and J. Schmiedmayer, Proc. Natl. Acad. Sci. **112**, 3866 (2015).

²³ S. Dutta, E.A. Goldschmidt, S. Barik, U. Saha, and E. Waks, Nano Lett. 20, 741 (2020).
²⁴ S. Wang, L. Yang, R. Cheng, Y. Xu, M. Shen, R.L. Cone, C.W. Thiel, and H.X. Tang, ArXiv191207584 Phys. (2019).

²⁵ T.T. Van and J.P. Chang, Surf. Sci. **596**, 1 (2005).

²⁶ M. Scarafagio, A. Tallaire, K.-J. Tielrooij, D. Cano, A. Grishin, M.-H. Chavanne, F.H.L. Koppens, A. Ringuedé, M. Cassir, D. Serrano, P. Goldner, and A. Ferrier, J. Phys. Chem. C **123**, 13354 (2019).

²⁷ M. Scarafagio, A. Tallaire, M.-H. Chavanne, M. Cassir, A. Ringuedé, D. Serrano, P. Goldner, and A. Ferrier, Phys. Status Solidi A **n/a**, 1900909 (2020).

²⁸ B. Lacroix, F. Paumier, and R.J. Gaboriaud, Phys. Rev. B **84**, 014104 (2011).

²⁹ R. Lo Nigro, R.G. Toro, G. Malandrino, G.G. Condorelli, V. Raineri, and I.L. Fragalà, Adv. Funct. Mater. **15**, 838 (2005).

³⁰ G.P. Flinn, K.W. Jang, J. Ganem, M.L. Jones, R.S. Meltzer, and R.M. Macfarlane, J. Lumin. **58**, 374 (1994).

³¹ V. Astié, C. Millon, J. M. Decams, and A. Bartasyte, in *Chem. Vap. Depos. Nanotechnol.*, edited by Pietro Mandracci (IntechOpen, 2019), pp. 2–23.

³² N. Jehanathan, O. Lebedev, I. Galard, C. Dubourdieu, and G. V. Tendeloo, Nanotechnology **21**, 075705 (2010).

³³ C. Durand, C. Dubourdieu, C. Vallée, V. Loup, M. Bonvalot, O. Joubert, H. Roussel, and O. Renault, J. Appl. Phys. **96**, 1719 (2004).

³⁴ H. Zhang, J. Yang, J.A. Brown, S. Gray, T.D. Ketcham, B.G. Aitken, and D.A. Nolan, Opt. Mater. Express **10**, 99 (2020).

³⁵ F. Silva, A. Gicquel, A. Tardieu, P. Cledat, and T. Chauveau, Diam. Relat. Mater. **5**, 338 (1996).

³⁶ H.C. Aspinall, in *Rare Earth Oxide Thin Films*, edited by M. Fanciulli and G. Scarel (Springer Berlin Heidelberg, Berlin, Heidelberg, 2007), pp. 53–72.

³⁷ H. O. Pierson, in *Handb. Chem. Vap. DEPOSITIONCVD*, NOYES PUBLICATIONS Park Ridge, New Jersey, U.S.A. WILLIAM ANDREW PUBLISHING, LLC Norwich, New York, U.S.A. (n.d.).
 ³⁸ Y.-M. Wu and J.-T. Lo, Jpn. J. Appl. Phys. 4943 (1998).

³⁹ J. Xia, W. Liang, Q. Miao, and D. Depla, Appl. Surf. Sci. **439**, 545 (2018).

⁴⁰ L. Tarnawska, A. Giussani, P. Zaumseil, M.A. Schubert, R. Paszkiewicz, O. Brandt, P. Storck, and T. Schroeder, J. Appl. Phys. **108**, 063502 (2010).

⁴¹ R. Bachelet, P. de Coux, B. Warot-Fonrose, V. Skumryev, G. Niu, B. Vilquin, G. Saint-Girons, and F. Sanchez, CrystEngComm **16**, 10741 (2014).

⁴² M. E. Hunter, M. J. Reed, N. A. El-Masry, J. C. Roberts, and S. M. Bedair, Appl. Phys. Lett. **76**, 1935 (2000).

⁴³ C. W. Nieh, Y. J. Lee, W. C. Lee, Z. K. Yang, A. R. Kortan, M. Hong, J. Kwo, and C. -H. Hsu, Appl. Phys. Lett. **92**, 061914 (2008).

⁴⁴ X. Wu, J.-G. Li, Q. Zhu, J. Li, R. Ma, T. Sasaki, X. Li, X. Sun, and Y. Sakka, Dalton Trans. **41**, 1854 (2012).

⁴⁵ A. Fert, Bull. Société Fr. Minéralogie Cristallogr. **85**, 267 (1962).

⁴⁶ G. Adachi and N. Imanaka, Chem. Rev. **98**, 1479 (1998).

⁴⁷ M. Inokuti and F. Hirayama, J. Chem. Phys. **43**, 1978 (1965).

⁴⁸ M. Buijs, A. Meyerink, and G. Blasse, J. Lumin. **37**, 9 (1987).

⁴⁹ A. Ferrier, C.W. Thiel, B. Tumino, M.O. Ramirez, L.E. BausÃi, R.L. Cone, A. Ikesue, and P. Goldner, Phys. Rev. B **87**, 041102 (2015).

⁵⁰ R.D. Shannon and C.T. Prewitt, Acta Crystallogr. B **25**, 925 (1969).

⁵¹ S. Welinski, C.W. Thiel, J. Dajczgewand, A. Ferrier, R.L. Cone, R.M. Macfarlane, T.

Chanelière, A. Louchet-Chauvet, and P. Goldner, Opt. Mater. 63, 69 (2017).

⁵² A.M. Stoneham, Proc Phys Soc **89**, 909 (1966).

⁵³ K. de Oliveira Lima, R. Rocha Gonçalves, D. Giaume, A. Ferrier, and P. Goldner, J. Lumin. **168**, 276 (2015).