

# Influence of preferred orientation of clay particles on the diffusion of water in kaolinite porous media at constant porosity

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#### 16 Abstract

17 Solute transport in natural or artificial compacted clay porous media is receiving particular attention in the contexts of waste storage and the design of materials with tuneable physical 18 properties. In these contexts, the porosity is commonly considered as a primary parameter 19 controlling the diffusional properties of water and solutes in these systems. However, little 20 attention has been given to the role played by anisotropy in the particle orientation. In this study, 21 the influence of the preferred orientation of clay particles on the water diffusion anisotropy in 22 two kaolinite porous media obtained by compaction and centrifugation methods (for a constant 23 porosity value of ~0.5) was investigated by coupling experiments and simulations. An increase 24 25 in the preferred orientation of kaolinite particles, as quantified by X-ray scattering analysis, was found to be logically associated with an enhanced anisotropy in water diffusion obtained from 26 pulsed gradient spin echo attenuation measurements by nuclear magnetic resonance of protons. 27 28 Brownian dynamics simulations performed on three-dimensional virtual porous media, mimicking the shape and orientation of the particles in the samples, led to calculated water 29 diffusion coefficients in agreement with experimental data. Once validated, this computational 30 work was extended to a wide range of degrees in the preferred orientation of particles. The 31 results showed that this parameter leads to an increase and a decrease in pore water diffusion 32 33 coefficients along and across the mean orientation plane, respectively, up to a factor  $\sim 2$ . The directional diffusion anisotropy was also found to range between 1 and ~5 for the most isotropic 34 and anisotropic organisations, respectively. This study hence provides quantitative insights into 35 the impact of the preferred orientation for the prediction of water diffusion in compacted clay 36 media. 37

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Keywords: clay minerals; porous media; anisotropy; particle orientation; water diffusion

#### 40 **1. Introduction**

The understanding of solute transport in compacted clay-based porous media has considerable importance in the fields of nuclear waste storage in deep geological formations (Altmann et al., 2012; Charlet et al., 2017) and the design of innovative materials with fluid barrier efficiency (Aulin et al., 2012). For these systems, macroscopic transport models based on the Fickian expression of the diffusion process of a non-sorbing tracer rely on its effective diffusion coefficient  $D_e$ , defined by (Bourg et al., 2006; Tournassat and Steefel, 2015; Tinnacher et al., 2016):

$$D_e = \frac{\varepsilon}{c} D_0 \tag{1}$$

49 where  $\varepsilon$  is the porosity of the porous medium,  $D_0$  the diffusion coefficient of the tracer in 50 bulk liquid water, and *G* the tortuosity factor related to the geometry of the pore network. The 51 diffusion coefficient of the tracer *D* in the pores, related to this  $D_e$  entity by  $D_e = \varepsilon$ . *D*, is thus 52 expressed as follows:

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$$D = \frac{1}{G}D_0 \tag{2}$$

In Eq. (2), all information regarding the geometry of the pore network, including the distributions in the size, shape, orientation, and connectivity of the pores, is contained in the entity *G*. Although different experimental methods such as microscopy techniques can be used to access specific properties of the porous network (Keller et al., 2014; Gaboreau et al., 2016; Leu et al., 2016; Backeberg et al., 2017; Takahashi and Tachi, 2019), the quantitative prediction of the geometric factor *G* value for a given porous medium remains a challenging task in the field of macroscopic transport modelling.

Because of their lamellar shape, clay particles most often display a preferred orientation, leading to anisotropy in the morphology of the pore network and having a considerable impact on the transfer properties of water and solutes. For instance, experimental studies have evidenced a significant variation in the measured water effective diffusion coefficients in the

direction parallel or perpendicular to the bedding of different natural argillaceous rocks (Van 65 Loon et al., 2004; García-Gutiérrez et al., 2006; Xiang et al., 2013; Gimmi et al., 2014; Jacops 66 et al., 2017). Furthermore it was also demonstrated that the increase in the diffusion anisotropy 67 was qualitatively linked to a more pronounced preferred orientation of clay particles in the 68 different samples (Wenk et al., 2008). Although the quantitative correlation between the water 69 diffusion and porosity parameter has been subjected to a large number of studies (Van Loon 70 and Mibus, 2015; Charlet et al., 2017 and references therein), little is known regarding the 71 quantitative impact of the degree of preferred orientation on the diffusional properties of water 72 and solutes. To obtain a complete understanding of this specific contribution, additional work 73 74 on samples of model clay porous media with controlled porosity and anisotropic properties in the particle orientation is required. 75

In that context, the present study focuses on the water diffusion properties in two model 76 77 porous media prepared from KGa-2 kaolinite particles with similar porosity but different particle orientations. The quantitative measurement of preferred orientation of the particles in 78 79 these samples will be performed using X-ray scattering (XRS) analysis. An experimental analysis of the anisotropy in the water diffusion will be performed using pulsed gradient spin 80 echo attenuation measurements by nuclear magnetic resonance (PGSE-NMR) of protons. 81 Information regarding the particle size, shape and orientation will then be used to build 3D 82 virtual porous media, for which Brownian dynamics simulations will be performed to analyse 83 the water diffusional properties in the pore network. This computational methodology was 84 recently shown to be efficient in connecting the different time scales of the diffusional process 85 of water and solutes in compacted clay porous media (Tyagi et al., 2013; Bacle et al., 2016). 86 Once validated against experimental data, the computational methodology will be extended to 87 a wide range of degrees in preferential orientation of particles. The obtained results will 88

eventually help us to obtain quantitative insights into the role played by the preferentialorientation in the variation of the diffusional properties in these systems.

#### 91 2. Materials and methods

#### 92 2.1. Starting material

The kaolinite KGa-2 from the Source Clay Repository of the Clay Mineral Society was used 93 to prepare the different porous media. The average size of kaolinite particles is approximately 94 0.5 µm al., 2005) and their structural formula 95 (Hassan et is  $[(Al_{3,80}Ti_{0,13}Fe^{3+}_{0.07}) (Si_{3,84}Al_{0.16}) O_5(OH_4)]$  (Mermut and Cano, 2001). The original kaolinite 96 97 powder was Na-saturated using three saturation cycles in a 1 mol/L NaCl solution to obtain homoionic samples. Dialysis in deionised water was then used to remove the excess salt until a 98 99 silver nitrate test for Cl<sup>-</sup> was negative. Finally, the kaolinite dispersion was dried in an oven at 100  $60^{\circ}$ C and then sieved through a 50  $\mu$ m mesh to remove coarse aggregates.

## 101 2.2. Preparation of porous media with contrasted anisotropy

102 Two porous media made of kaolinite powder were prepared in order to obtain a similar 103 porosity value  $\varepsilon$  but contrasted anisotropy degrees in particle orientation. The first sample was obtained by uniaxial compaction of the initial powder in a poly(tetrafluoroethylene) (PTFE) 104 cylinder with a diameter of 0.64 cm and a height of 7.5 cm. The kaolinite powder was 105 introduced in the tube, the tube placed in a metallic cell with a bottom cap to prevent widening, 106 107 and the powder then compressed to end up with a compacted kaolinite sample of 1 cm in height. 108 The second sample was obtained by a centrifugation process directly in the PTFE tube placed in a poly(methyl methacrylate) (PMMA) cell to avoid deformation. To do so, the initial 109 kaolinite powder was first dispersed in water (50 g/L). A 1 mL aliquot of the obtained dispersion 110 was then introduced in the PTFE tube and centrifuged horizontally at approximately 18 000 g 111 (Centrifuge Avanti J 301 and rotor JS-24.38 from Beckman Coulter) for 10 min. After 112 centrifugation, the excess of water was removed before the addition of a new dispersion aliquot. 113

Ten steps were sufficient to obtain a sufficient amount of kaolinite material. Finally, the sample
was dried at 60°C and slightly compressed to a 1 cm height using the same protocol as for the
first sample.

The final bulk porosity  $\varepsilon$  of 0.45±0.02 for both samples was determined based on the weight 117 of the samples (488±1 mg in both cases), their height (measurement after compression: 1.0 to 118 1.1 cm), and the kaolinite grain density. This latter value was estimated at 2.62 g.cm<sup>-3</sup> based on 119 the structural formula mentioned above and the crystal structure parameters defined by 120 Sakharov et al. (2016) for KGa-2 kaolinite. Both samples were realised twice following exactly 121 the same protocol. The first set of samples was used for quantifying the particle orientation 122 using XRS measurements after sample induration (see Section 2.3). The second batch of sample 123 124 was used for the PGSE-NMR attenuation measurements of protons to extract the self-diffusion tensor of the water probes (see Section 2.4). 125

## 126 2.3. Sample inducation and X-ray scattering measurements

Both dried porous media were indurated and sliced to perform XRS measurements in 127 transmission mode. For the induration, methyl methacrylate (MMA; C<sub>5</sub>H<sub>8</sub>O<sub>2</sub>) resin was chosen 128 because of its fluidity higher than that of water allowing it to quickly fill the porosity, its small 129 molecular volume compared to other organic molecules (19 nm<sup>3</sup>), and its dipole moment being 130 131 similar to that of water. The protocol used was based on the work of Sammaljärvi et al. (2012). The first step consists of setting the sample under a primary vacuum in a hermetic cell for few 132 minutes. This step is necessary to fully dehydrate the sample, as water molecules can disturb 133 the polymerisation reaction of MMA without impacting their orientation (Hubert et al., 2013). 134 In the second step, MMA mixed with benzoyl peroxide (BPO; added in a BPO/MMA ratio of 135 0.5 wt.% as a thermal initiator of the polymerisation) is introduced in PTFE tubes under vacuum 136 to facilitate the vaporisation of MMA and its condensation into the smallest pores to fill all the 137 porosity. For both preparation conditions (compacted and centrifuged samples), the saturation 138

of porosity by the resin was achieved after 3 days. The third step, consisting of the polymerisation of the MMA into PMMA, was achieved by transferring the saturated sample contained in the waterproof PTFE tube into a water bath at 55°C for 24 h.





**Fig. 1.** Experimental measurement of particle orientation within kaolinite porous media using X-ray scattering technique with a 2D detector. (**a**) Sketches of the 500 µm-thick longitudinal and transverse lamellas relative to the tube axis and extracted from indurated samples to characterise the particle orientations. (**b**) Schematic representation of 2D-XRS measurements of lamellas in transmission mode. The modulation of the *hkl* Bragg reflection intensity rings along the detector angle  $\tau$  is used to probe the particle orientation. (**c**) Spherical coordinates ( $\theta$ ,  $\phi$ ) defining the orientation of the normal (*Oz'*) of an individual particle with respect to the laboratory frame (*x*, *y*, *z*).

After induration, the cylindrical porous media were extracted from the PTFE tubes and 150 sawed in both the longitudinal and transverse directions of the tube axis (Fig. 1a). These two 151 directions were chosen to investigate the particle orientation in the two main directions relative 152 153 to the compaction/centrifugation forces. The obtained slices were then polished into 500 µm thickness lamellas using grinding paper (with a grain diameter of 5 µm). The XRS 154 measurements were performed at the Laboratoire de Physique des Solides (Orsay, France). A 155 156 copper rotating anode generator (RU H3R, Rigaku Corporation, Japan) equipped with a multilayer W/Si mirror (Osmic) allows obtaining a monochromatic beam ( $\lambda_{CuK\alpha} = 1.5418$  Å) of 157 600 x 600 µm<sup>2</sup> at the sample position. Two-dimensional (2D) X-ray scattering patterns were 158 159 collected on a MAR345 2D-detector (marXperts GmbH, Germany, 150 µm pixel size). The

sample-to-detector distance D was set to 250 mm with a sample-to-beam stop distance of 30 160 mm. This configuration allows us to obtain scattering vector moduli down to  $Q_{\min} = 0.2 \text{ Å}^{-1} (Q)$ 161 =  $4\pi/\lambda \sin(\theta_B)$ , where  $\lambda$  is the incident wavelength and  $2\theta_B$  is the scattering angle), i.e., d-162 spacings up to 31.5 Å ( $d = 2\pi/O$ ). Sample lamellas were mounted on a goniometer head and 163 aligned perpendicularly to the incident X-ray beam with a typical exposure time of 300 to 900 164 s per sample position (Fig. 1b). For the XRS analyses, the limited thickness of the lamellas (i.e., 165  $500 \,\mu\text{m}$ ) is a compromise to optimise both the percentage of the transmitted beam and the 166 quantity of particles analysed over a 600 x 600  $\mu$ m<sup>2</sup> surface. 167

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## 169 2.4. <sup>1</sup>H NMR Pulsed Gradient Spin Echo experiments

170 The PGSE-NMR analyses were performed at the ICMN laboratory (Orléans, France). The 171 dried kaolinite samples in their PTFE tubes (see Section 2.2) were first water-saturated by 172 adding successive small drops of deionised water followed by equilibration over 2-3 days. As 173 the samples were hydrated in unconstrained conditions, a limited decompaction occurred for 174 both samples, leading to  $\varepsilon$  values increasing from 0.45±0.02 to 0.50±0.02, a final value 175 confirmed by helium pycnometry (Porion et al., 2018).

The NMR measurements were performed using a Bruker DSX100 spectrometer with a static field of 2.35 T, equipped with a saddle detection coil and a micro-imaging probe (Micro5 Bruker) with gradient coils able to generate magnetic field gradients in three perpendicular directions. <sup>1</sup>H PGSE-NMR attenuation measurements (Stejskal and Tanner, 1965; Cotts et al., 1989; Callaghan, 1991) were used to determine the macroscopic water mobility along any preselected direction within the sample.





Fig. 2. Schematic view of the pulse sequence used to perform Pulse Gradient Spin Echo (PGSE-NMR)
attenuation measurements. Details on the time delays, pulse durations and strengths of the magnetic
field gradients are given in the text.

Fig. 2 illustrates the pulse sequence used to perform the PGSE-NMR attenuation 186 measurements with which a wave vector q may be associated (Callaghan, 1991) according to 187 the relationship  $q = \gamma g \delta / \pi$ , where g is the intensity of the applied field gradient (between 188 0 and 1.6 T/m), 2 $\delta$  its effective duration (1 ms) and  $\gamma$  the proton gyromagnetic ratio 189  $(2.6752 \times 10^8 \text{ rad/s for }^1\text{H})$ . With the fixed duration  $\delta$  of the pulsed field gradient set equal to 190 500  $\mu$ s, the maximum value of the probed wave vector reaches 6.8 10<sup>4</sup> m<sup>-1</sup>. As a consequence, 191 192 our diffusion measurements are macroscopic and result from averages over sample sizes larger than 15  $\mu$ m with respect to the particle size (~0.5  $\mu$ m). Under such macroscopic conditions 193 (Porion et al., 2003), the attenuation of the intensity of the NMR echo I(q)/I(0) evolves 194 according to a Gaussian relationship from which the components of the self-diffusion tensor 195 are easily extracted (Cotts et al., 1989; Callaghan, 1991): 196

197 
$$\frac{I(q)}{I(0)} = exp\left[-4 \pi^2 q^2 \overrightarrow{e_g}^T \overrightarrow{D} \overrightarrow{e_g} (\Delta + 3\tau/2 - \delta/6)\right]$$
(3)

where  $\vec{e_g}$  is the direction of the applied field gradient and  $\vec{D}$  the self-diffusion tensor. This equation is usually simplified by replacing the tensorial contraction  $\vec{e_g}^T \vec{D} \ \vec{e_g}$  by the value of the self-diffusion coefficient  $D_g$  along the selected  $\vec{e_g}$  director. The diffusion time  $\Delta$  (20 ms) and the delay  $\tau$  (760 µs) are also displayed in Fig. 2. The PGSE-NMR attenuation measurements were performed at 292 K and the associated experimental bulk water selfdiffusion coefficient  $D_0$  was measured at 2.0±0.2.10<sup>-9</sup> m<sup>2</sup>/s in these conditions.

204

## 205 2.5. Generation of 3D virtual kaolinite porous media and Brownian dynamics

206 2.5.1. Simulated porous media of kaolinite with contrasted anisotropy and controlled 207 porosity

208 The 3D virtual porous media (VPM) mimicking the organisation of kaolinite porous media for different degrees of anisotropy in particle orientation were obtained using a one-by-one 209 210 deposition algorithm of elliptic disc-shaped particles (Ferrage et al., 2015, 2018). This type of simulation was recently shown to be able to reproduce the organisation and porosity of disc-211 shaped particle packings (Dabat et al., 2018). According to this algorithm, each particle with a 212 set of Euler angles  $(\varphi, \theta, \psi)$  following the ZXZ convention (Ferrage et al., 2015), is introduced 213 at the top of a square simulation box with periodic conditions along the x and y axes (z axis 214 215 pointing upward; Fig. 3a) and settles to provide the steepest descent of the barycentre altitude. 216 The settling process of the particle, either at the bottom of the box or onto an existing bed of particles with fixed positions, is thus obtained through the repetition of individual movements 217 218 around one or several contact points with one (or several) particles. The movements allow the particle to slide, swivel or rotate with a random amplitude ranging from zero to a maximum 219 value to provide the steepest descent of the particle altitude. The variable parameters for the 220 each simulation thus include a total number of particles (10 000), a width for the square 221 simulation box set at  $15\tilde{d}$ , with  $\tilde{d}$  the mean particle diameter, a total number of movement 222 attempts (2400), an initial angle  $\theta_{ini}$  of the particle introduced in the box, and a set of maximum 223 amplitudes to swivel  $(A_{swiv}^{max})$ , to rotate  $(A_{rot}^{max}$  with here  $A_{rot}^{max} = A_{swiv}^{max})$ , and to slide  $(A_{slide}^{max})$ 224 (Ferrage et al., 2015). For these simulations, the lognormal distribution in the geometric 225 dimensions (i.e., basal surface, particle diameter, ratio between thickness and diameter, and 226

ellipticity degree) of the individual elliptic particles was obtained from the work (Ferrage et al.,
2015) and based on the experimental morphological study of Reinholdt et al. (2013) for the 0.10.2 µm size fraction of vermiculite from Santa Olalla, Spain. This vermiculite sample was used
here as a proxy, because it displays particles having very similar average aspect ratios (i.e., ratio
between thickness and diameter of particles) to that of KGa-2 kaolinite (0.08 and 0.07 for
vermiculite and kaolinite, respectively; Hassan et al., 2005, Reinholdt et al., 2013).

The degree of anisotropy of particle orientation in the obtained packings was extracted by calculating the average of the second-order Legendre polynomial on the angular distribution of the particle orientations as follows:

236

$$\langle P_2 \rangle = \langle P_2(\cos\theta) \rangle = \langle 3\cos^2\theta - 1 \rangle / 2 \tag{4}$$

with  $\theta$  the angle between the normal unit vector of the particle and the z axis of the simulation 237 box (Figs. 1c and 3a). This  $\langle P_2 \rangle$  order parameter takes the value of 0 for an isotropic organisation 238 and 1 when all particles are perfectly oriented in the bedding (all normal to particles aligned 239 with the z axis of the simulation box). The  $\langle P_2 \rangle$  order parameter was previously used for the 240 description of compacted clay porous media (Perdigon-Aller et al., 2005; Ferrage et al., 2015; 241 Dabat et al., 2018) but is also referred to as H, the Hermans parameter in polymer sciences 242 (Hermans and Platzek, 1939), or S, the nematic order parameter in colloid science (Chaikin and 243 Lubensky, 1995; Davidson et al., 1995). To cover a large range of anisotropy degrees, 12 244 particle packings with  $\langle P_2 \rangle$  values varying from 0.03 to 0.92 were generated according to 245 algorithm parameters reported in Table S1. 246

To allow investigating the dynamic behaviour of water tracers in these different VPM, additional numerical treatments are necessary. The generation procedure of these VPM indeed leads to different porosity values as a function of the  $\langle P_2 \rangle$  values (Ferrage et al., 2015; Dabat et al., 2018) and does not allow obtaining periodic conditions in the *z* direction. These effects are negative drawbacks for both Brownian dynamics calculations and comparisons with

experimental data on compacted/centrifuged kaolinite samples. To overcome these limitations, 252 a specific methodology was employed for the most isotropic VPM constituted by approximately 253 10 000 particles, as illustrated in Fig. 3. Each individual particle, originally defined by a set of 254 three parameters (thickness and major and minor axes), is first polygonised considering 12 in-255 plane vectors plus 2 vectors passing along the normal of the particle (Fig. 3b). A cubic sub-256 volume containing ~2000 particles is then extracted (Fig. 3c). In order to allow the numerical 257 investigation of water tracer mobility over a large spatial domain, periodic conditions under the 258 form of minimum-image convention along the directors  $\overrightarrow{e_x}$ ,  $\overrightarrow{e_y}$ , and  $\overrightarrow{e_z}$  are required for the newly 259 260 obtained simulation box (Fig. 3d). For Brownian dynamics simulations this convention indeed insures that a water tracer passing through one side of the box will re-appears on the opposite 261 side of the simulation box. However, it requires avoiding the case where a particle protruding 262 out of one side could overlap on the other side of the box. This is achieved by first detecting 263 particle overlapping after application of minimum-image convention along the directors  $\vec{e_x}$ ,  $\vec{e_y}$ , 264 and  $\overrightarrow{e_z}$  and its removal by reducing the length of the different vectors defining the geometry of 265 the two overlapping particles. The two following numerical treatments consist of reducing the 266 267 porosity of the VPM in order to reach an  $\varepsilon$  value close to that of the water-saturated kaolinite samples analysed by PGSE-NMR, i.e.,  $\varepsilon = 0.50$  (see Section 2.4). First, each particle is allowed 268 269 to grow along one of its 14 vectors, chosen randomly. The amplitude of the increase in the vector length is also chosen randomly between zero and a maximum value set at 10% of the 270 initial vector length. In the case where the modification of the particle geometry induces an 271 overlapping with another particle of the packing, the attempt is rejected. This first procedure 272 allows increasing the volume of solid in the packing and thus decreasing the porosity until all 273 274 particles become geometrically constrained by the other particles from the stack (Fig. 3e). Second, if the porosity obtained still remains higher than the target value, new solid volume is 275 added by injecting a new particle into the stacking on the surface of an existing one and letting 276

the particle grow along its 14 vectors (Fig. 3f). The newly injected particles satisfy the 277 distribution of the geometric dimensions but only the addition of the smallest ones is likely to 278 be accepted during such a procedure. Depending on the initial porosity of the original VPM, 279 280 the procedure is stopped during either the first or the second treatment, i.e., when the VPM reaches the porosity of  $\varepsilon$ =0.50. Note that if the porosity is the same for all VPM, these two 281 processes lead to a modification in the particle sizes and thus a change in the overall specific 282 surface areas (SSAs). For instance, an increase in the volume of the individual particles leads 283 to a decrease in the SSA values, whereas the injection of new particles, most often small in size. 284 in order to fill the remaining porosity leads to an increase in the SSA values. Accordingly, the 285 final step of the VPM generation procedure thus consists of applying a dilatation or contraction 286 287 of the overall packing (and thus particle) dimensions (Fig. 3e). This leads to VPM with cubic lengths of ~4  $\mu$ m (i.e., between 4.3 and 3.9 between ( $P_2$ )=0.03 and 0.92, respectively) and SSA 288 values of ~20 m<sup>2</sup>/g (similar to the experimental SSA value for KGa-2 kaolinite; Hassan et al., 289 2005). 290

## 291 2.5.2. Brownian dynamics of water in virtual porous media

Brownian dynamics (BD) simulations represent a convenient way to investigate water diffusion at the representative time scale of the micrometre-sized porous media (typically several ms). In the framework of the general Langevin equation, the displacement of the molecular probes can be defined by (Gunsteren et al., 1981; Tertre et al., 2015; Porion et al., 2018):

297 
$$m_i \frac{d\vec{v}_i(t)}{dt} = -m_i \gamma_i \vec{v}_i(t) + \vec{F}_i(x(t), t) + \vec{R}_i(t)$$
(5)

where  $\vec{v}_i$ ,  $m_i$  and  $\gamma_i$  are the velocity, molecular mass and frictional coefficient of probe *i*, respectively,  $\vec{F}_i(x(t), t)$  is a systematic force acting on the probes and depending on their positions and  $\vec{R}_i(t)$  is a random force resulting from thermal collisions within the liquid. Different algorithms can be used to solve the set of Eq. (5), mainly depending on the ratio between the friction coefficient and the time step ( $\delta t$ ) used in the simulation. In the case where the time step is much larger than the velocity correlation time (i.e.,  $\gamma_i \delta t \gg 1$ ) and considering the probe displacements to be completely independent, the solution of Eq. (5) becomes (Gunsteren et al., 1981):

$$\vec{x}_i(t+\delta t) = \vec{x}_i(t) + \vec{R}_i \tag{6}$$

307 with the random force  $\vec{R}_i$  now satisfying a Gaussian distribution function with a zero mean 308 and standard deviation given by the following:

$$\langle R_i^2 \rangle = 2D_i \delta t \tag{7}$$

where  $D_i$  is the molecular self-diffusion of the probe *i*. For diffusion within isotropic media, the modulus of the random 3D displacements is quantified by the self-diffusion propagator, i.e., the density of the probability of the displacement distribution law (Callaghan, 1991; Bacle et al., 2016):

314 
$$P(\vec{r}, \Delta \mid \vec{r}_0, 0) = \{4\pi D\Delta\}^{-3/2} exp\left[-\frac{(\vec{r} - \vec{r}_0)^2}{4D\Delta}\right]$$
(8)

where  $\Delta$  is the diffusion time (see Eq. (3)). The self-diffusion propagator quantifying the mobility along any single director is then defined by the following:

317 
$$P_{\alpha}(\overrightarrow{r_{\alpha}},\Delta \mid \overrightarrow{r_{\alpha}}_{0},0) = \left\{4\pi D_{\alpha}\Delta\right\}^{-1/2} exp\left[-\frac{\left(\overrightarrow{r_{\alpha}}-\overrightarrow{r_{\alpha}}_{0}\right)^{2}}{4D_{\alpha}\Delta}\right]$$
(9)

318 where  $D_{\alpha}$  is the component of the self-diffusion tensor describing the mobility of the probe 319 along the selected  $\overrightarrow{e_{\alpha}}$  director (Callaghan, 1991). Its spatial Fourier transform is also Gaussian 320 and describes the probe displacement in the reciprocal *q*-space:

321 
$$E_{\alpha}(q,\Delta) = \exp\left[-2\pi^2 q^2 \left\langle \left(\overrightarrow{r_{\alpha}} - \overrightarrow{r_{\alpha}}\right)^2 \right\rangle\right] = \exp\left[-4\pi^2 q^2 D_{\alpha} \Delta\right]$$
(10)

where  $\Delta$  is the diffusion time and  $\langle (\vec{r_{\alpha}} - \vec{r_{\alpha_0}})^2 \rangle$  the mean-squared displacement along the director (Callaghan, 1991; Porion et al., 2001, 2003). In the scope of comparing simulations and experiments, as done in the present study, it should be pointed of that this Fourier transform of the self-diffusion propagator derived from BD simulations corresponds to the entity directly probed by PGSE-NMR measurements (see Section 2.4). Moreover it also corresponds exactly to the intermediate scattering function ( $F_q(t)$ ) commonly probed, on a much shorter time-scale, by quasi-elastic neutron-scattering measurements (Hansen and McDonald, 1990).

For the BD simulations performed here at 300 K, N = 2500 water probes with the molecular self-diffusion  $D_0$  set at 2.3.10<sup>-9</sup> m<sup>2</sup>/s were first introduced randomly into the porosity of the VPM. A time step  $\delta t = 5$  ns was considered, and additional tests showed that smaller values did not provide any difference in the simulation results. An overall simulation time of 4 ms was used to extract the pore water diffusion coefficients  $D_{\alpha}$  along a selected  $\vec{e}_{\alpha}$  director based on the asymptotic slope of the radial mean squared displacement as follows:

335 
$$D_{\alpha} = \lim_{\Delta \to \infty} \frac{\sum_{i=1}^{N} (x_{i,\alpha}(0) - x_{i,\alpha}(\Delta))^2}{2N\Delta}$$
(11)

where  $\alpha$  refers to the single direction  $\overrightarrow{e_{\alpha}}$  on which the mobility is extracted (the uncertainty on the calculated  $D_{\alpha}$  values is estimated at 10%). During the BD simulations, the collision with the surface of the kaolinite was treated as a Maxwell's *A* wall with the angular probability  $f_A(\theta, \phi)$  of water probes rebounding from the surface given by:

340  $f_A(\theta, \phi) d\Omega = \cos \theta \, d\Omega / \pi \tag{12}$ 

where  $\theta$  and  $\phi$  are the usual polar and azimuthal angles (with respect to the normal of the surface of the particle) and  $d\Omega$  is the element of the solid angle. This condition mimics the fate of a molecule, which is supposed to reside at the interface of the wall and eventually leave the wall to return to the fluid (Valleau et al., 1991).



345

Fig. 3. Generation of virtual porous media with the same porosity but different anisotropy degrees in particle orientation. (a) Cubic 3D virtual porous medium composed of ~10 000 particles with an almost isotropic organisation ( $\langle P_2 \rangle = 0.04$ ) and generated using a sequential deposition algorithm (Ferrage et al., 2015, 2018). (b) Meshing and polygonalisation of individual particles. (c) Extraction of a cubic subvolume with ~2000 particles. (d) Application of periodic boundary conditions along  $\vec{e_x}$ ,  $\vec{e_y}$ , and  $\vec{e_z}$ directors of the simulation box and elimination of particle overlapping. (e) Decrease of porosity by

particle growth along the 14 individual axes (12 and 2 axes on the basal surface and along the normal of the particle, respectively). (**f**) Decrease of porosity by injection of new particles and subsequent growth along their 14 individual axes. (**g**) Contraction or dilation of the overall 3D cubic porous media in order to keep the total specific surface area similar for all samples (VPM obtained for  $\langle P_2 \rangle = 0.58$  and 0.92 are also shown here for illustration).

#### 357 3. Results and Discussion

### 358 *3.1 Preferred orientation analysis of kaolinite particles*

Preferred orientation of the particles in both the compacted and centrifuged samples is based 359 360 on the orientation distribution function (ODF) determined from 2D-XRS measurements (see Section 2.3). The orientation of a cylindrical particle can be described in an orthogonal 361 laboratory framework by the direction of its normal Oz' through the spherical coordinates  $\theta$  and 362 363  $\varphi$  (Fig. 1c). In addition, clay porous media are transverse isotropic systems with their z-axis as the unique symmetry axis (Cebula et al., 1979; Perdigon-Aller et al., 2005; Wenk et al., 2010; 364 Hubert et al., 2013), leading to an equal probability of the particle orientation over the  $\varphi$  angles. 365 Accordingly, the ODF only depends on the angle  $\theta$ , with the following general constraints: 366

- 367  $f(\theta) \ge 0$
- 368

$$f(\theta) = f(\pi - \theta) \tag{14}$$

(13)

369 
$$\int_0^{\pi} f(\theta) \sin(\theta) \, d\theta = 1 \tag{15}$$

Eqs. (13) and (14) indicate that the ODF  $f(\theta)$  is positive and has an equal probability of the particles pointing upward and downward, respectively. Eq. (15) allows normalising the total number of particles in the system through the integration over all  $\varphi$  angles (Labarthet et al., 2000). The experimental ODF can be extracted from the angular modulation of the scattered intensity of the 001 reflection of kaolinite along the detector angle  $\tau$  of the 2D-XRS patterns (Fig. 1b). Given the low values for the Bragg angle for the 001 reflection of kaolinite, the  $\tau = \theta$  approximation can indeed be considered, and the  $f(\theta)$  ODF is thus directly extracted from the scattered intensity  $I(\theta)$  evolution:

(16)

378 
$$f(\theta) = \frac{I(\theta)}{\int_{0}^{\pi} I(\theta) \sin(\theta) d\theta}$$



**Fig. 4.** Extraction of orientation distribution functions of kaolinite particles from 2D X-ray scattering measurements. (a) Experimental 2D-XRS patterns recorded from the compacted and the centrifuged samples in both the longitudinal and transverse directions of the tube axis. The colour scale ranges from dark blue for the smallest intensity values to red for the largest ones. (b) Corresponding orientation distribution functions from the angular scan of the 001 reflection of kaolinite over 180° (normalised intensity according to Eq. (15), resulting in the same intensity for isotropic ODF).

The experimental 2D-XRS patterns for both the compacted and centrifuged samples on lamellas extracted in the longitudinal and transverse directions with respect to the tube axis are reported in Fig. 4 with their corresponding ODF. As seen for both samples, the  $f(\theta)$  functions recorded for the lamellas extracted in the longitudinal direction display symmetric profiles,

indicating that the lamellas were sliced perpendicularly to the preferred orientation plane and 390 well aligned on the XRS setup. For lamellas corresponding to the transverse direction of the 391 tube axis, the obtained ODF displays a constant intensity whatever the azimuthal angle, in 392 agreement with the isotropic transverse nature of the sample and the correct methodology of 393 the sample preparation and analysis. Based on the shape of the  $f(\theta)$  functions for lamellas 394 sliced in the longitudinal direction of the tube axis, the degree of preferred orientation can be 395 obtained for both samples by calculating the average value of the second-order Legendre 396 polynomial on the angular distribution as: 397

398 
$$\langle P_2 \rangle = \int_0^{\pi} P_2(\cos\theta) f(\theta) \sin(\theta) d\theta$$
 (17)

399 As seen from the different shapes of the  $f(\theta)$  functions (Fig. 4b), the two samples display very different order parameters, with  $\langle P_2 \rangle$  values of 0.21±0.02 and 0.59±0.05 for the 400 compacted and centrifuged samples, respectively (Table 1). It is worth noting that the obtained 401  $\langle P_2 \rangle$  values are averages over several measurements for each sample. Although it can be 402 expected that the successive centrifugation steps lead to a more heterogeneous sample 403 organization, such a difference was not evident given the about 10% uncertainty of the structure 404 parameter in both cases (Table 1). The different organisations for the two samples can likely be 405 attributed to the preparation method for two reasons. First, the uniaxial compaction in air-dried 406 407 conditions has been shown to provide rather poor preferred orientations for montmorillonite particles (Lutterotti et al., 2010; Suuronen et al., 2014). Second, it is well accepted that the 408 dispersion of particle aggregates as individual clay platelets in water followed by sedimentation 409 promotes the preferred orientation, and a wide range of  $\langle P_2 \rangle$  values could be achieved, 410 depending on the particle size, solid-solution ratio, and hydrostatic or shear forces applied on 411 the sample  $(0.10 \le \langle P_2 \rangle \le 0.85$ ; Perdigon-Aller et al., 2005; Méheust et al., 2006; Perdigón et 412 al., 2007; Hubert et al., 2013; Ferrage et al., 2015; Carrier et al., 2016). 413

414

415 **Table 1**. Experimental measurements of the anisotropy degree of particle orientation (order parameter 416  $\langle P_2 \rangle$ ) determined from XRS analyses and associated pore water diffusion coefficients  $D_{\alpha}$  obtained from 417 PGSE-NMR experiments on compacted and centrifuged samples (porosity  $\varepsilon$ =0.50±0.02). The director

Sample	$\langle P_2 \rangle$ order parameter	$D_z/D_0$	$D_x/D_0$	$D_y/D_0$	$D_{\overline{x,y}}/D_z$
Compacted	0.21±0.02	0.43±0.04	0.51±0.05	0.50±0.05	1.2±0.2
Centrifuged	0.59±0.05	0.31±0.03	0.59±0.06	$0.57 \pm 0.06$	1.9±0.4

418  $\overrightarrow{e_z}$  is longitudinal whereas the directors  $\overrightarrow{e_x}$  and  $\overrightarrow{e_y}$  are transverse to the tube axis.

419

## 420 3.2 Experimental analysis of water diffusion in kaolinite porous media

The contrasted organisations in the kaolinite particles (i.e.,  $\langle P_2 \rangle = 0.21 \pm 0.02$  and  $0.59 \pm 0.05$ 421 422 for the compacted and centrifuged samples, respectively) for a similar porosity ( $\varepsilon = 0.50 \pm 0.02$ in wet conditions; see Section 2.4) indicate that the two samples are well adapted to investigate 423 in detail the impact of the preferred orientation on the anisotropy of the water diffusion. As 424 illustrated in Fig. 5 for the centrifuged sample, the pulsed gradients are applied along six non-425 collinear directions (Basser et al., 1994; Skare et al., 2000) noted  $\overrightarrow{e_1} = \overrightarrow{e_x} = (1,0,0), \overrightarrow{e_2} = \overrightarrow{e_y} = \overrightarrow{e_y}$ 426  $(0,1,0), \ \overrightarrow{e_3} = \overrightarrow{e_z} = (0,0,1), \ \overrightarrow{e_4} = (1,1,0), \ \overrightarrow{e_5} = (0,1,1) \ \text{and} \ \overrightarrow{e_6} = (1,0,1), \ \text{respectively},$ 427 with the direction  $\overrightarrow{e_3}$  being parallel to the compaction/centrifugation axis of the sample. From 428 the six corresponding components of the water self-diffusion tensor, one can easily extract its 429 three principal axes and the corresponding eigenvalues describing the water mobility in the 430 longitudinal or transverse direction of the tube axis (Fig. 1a). 431



432

433 Fig. 5. Illustration of water self-diffusion propagators obtained for the centrifuged sample from PGSE-434 NMR experiments. More details on the selected diffusion directors  $\vec{e}_{\alpha}$  are given in the text, leading to 435 the components  $D_{\alpha}$  of the self-diffusion tensor:  $D_1=D_x=1.142\times10^{-9}$  m<sup>2</sup>/s,  $D_2=D_y=1.165\times10^{-9}$  m<sup>2</sup>/s,

436 
$$D_3 = D_z = 0.614 \times 10^{-9} \text{ m}^2/\text{s}, D_4 = 1.173 \times 10^{-9} \text{ m}^2/\text{s}, D_5 = 0.869 \times 10^{-9} \text{ m}^2/\text{s} \text{ and } D_6 = 0.879 \times 10^{-9} \text{ m}^2/\text{s}.$$

As detailed in Table 1, the compacted sample exhibits a small anisotropy of the water 437 self-diffusion tensor, which is consistent with the measured low values for the order parameter 438  $\langle P_2 \rangle$ . Moreover, the pore water diffusion coefficient obtained for this sample along the z 439 direction is consistent with a previous analysis from magnetic resonance imaging of the time 440 441 evolution of the water concentration profiles during D<sub>2</sub>O/H<sub>2</sub>O exchange (Porion et al., 2018). In contrast, the sample obtained by the centrifugation method shows a larger anisotropy ratio 442  $D_{\overline{x},\overline{y}}/D_z$  of the water diffusion tensor, in agreement with the increase of the  $\langle P_2 \rangle$  order parameter 443 value for this sample. The water self-diffusion tensor can be better illustrated for these two 444 samples by plotting the variation of the water mobility within two orthogonal planes (i.e., either 445 parallel or perpendicular to the preferred orientation plane). The gradual variation of the pore 446 water diffusion coefficients between the two principal axes of the diffusion tensor (i.e.,  $\overrightarrow{e_{\perp}}$  and 447  $\vec{e_{\parallel}}$ , directions perpendicular or parallel to the preferred orientation plane, respectively) can 448 indeed be described by (Porion et al., 2003): 449

450 
$$D_{meas}(\gamma) = \overrightarrow{e_g}^T \overleftarrow{D} \overrightarrow{e_g} = D_{\perp} \cos^2(\gamma) + D_{\parallel} \sin^2(\gamma)$$
(18)

451 where  $\gamma$  is the angle between  $\overrightarrow{e_{\perp}}$  and  $\overrightarrow{e_{g}}$ .  $D_{\alpha}$  and  $D_{\parallel}$  are the two eigenvalues of the 452 diffusion tensor along the principal directions  $\overrightarrow{e_{\perp}}$  and  $\overrightarrow{e_{\parallel}}$ . The diffusion tensor is illustrated in 453 Fig. 6 for both the compacted and centrifuged samples. Calculations are performed either 454 parallel or perpendicular to the preferred orientation plane based on the (x,z) or (x,y) planes, 455 respectively.

The enhanced experimental anisotropy in the water mobility on the (x,z) plane for the 456 centrifuged sample compared to the compacted sample well illustrates the role played by the 457 preferred orientation of the kaolinite particles on the water mobility for a given  $\varepsilon$  value. In that 458 regard, and compared to other diffusion experiment setups, the great advantage of the PGSE-459 460 NMR attenuation measurements is to provide a direct measurement of the water diffusion tensor in the sample. Such experimental data thus represent key constraints for assessing the 461 representativeness of virtual porous media through a detailed comparison between the 462 463 experimental and simulated water mobility dependences on the particle orientation.



#### 464

**Fig. 6.** Water diffusion tensors determined from PGSE-NMR experiments (left) and Brownian dynamics simulations (right). The variations of the water mobility  $D_{\alpha}/D_0$  in the (*x*, *y*) plane and (*x*, *z*) plane are shown in blue and red, respectively. For the experimental data, the diffusion tensor is calculated from its eigenvalues  $D_x$ ,  $D_y$ , and  $D_z$  according to Eq. (18). For the simulated data, the solid lines correspond to the limiting time evolution of the mean square displacements (see Eq. (11)), whereas the dotted lines represent the calculation of the diffusion tensor from its eigenvalues using Eq. (18).

## 471 3.3 Brownian dynamics of water diffusion in 3D virtual porous media

The pore diffusion coefficients in the 12 virtual porous media (with  $0.04 \le \langle P_2 \rangle \le 0.92$ ) are calculated along the 3 directors of the 3D virtual porous media reference framework (i.e., *x*, *y*, and *z*; Fig. 7) based on the mean square displacement (MSD) of the water probes according to Eq. (11). Additional calculations were performed every  $10^{\circ}$  in the (*x*,*y*) and (*x*,*z*) planes of the diffusion tensor.



477

**Fig. 7.** Illustration of the time evolution of the mean square displacements (MSD) of the water probe obtained by Brownian dynamics simulations for the virtual porous media with  $\langle P_2 \rangle = 0.92$ . Displacements are extracted along the 3 directors of the 3D virtual porous media reference framework (i.e., *x*, *y*, and *z*) as well as every 10° in the (a) (*x*,*y*) plane and (b) (*x*,*z*) plane.

As illustrated in Fig. 7 for the VPM obtained for  $\langle P_2 \rangle = 0.92$ , the asymptotic time evolution behaviour of the MSD is rapidly obtained. The different MSD curves extracted in the (x,y) plane of the diffusion tensor are very similar (Fig. 7a). This feature confirms the transverse isotropy of the VPM, in agreement with the experimental data (see Section 3.1). Moreover, this lack of angular dependence of the calculated  $D_{\alpha}$  values likely evidence that the VPM contain sufficient numbers of particles and thus can be considered as representative elementary volumes for the analysis of the dynamical behaviour of diffusive species. The situation differs when analysing the MSD curves in the (x,z) plane of the diffusion tensor (Fig. 7b). The  $D_{\alpha}$  values are indeed significantly lower in the *z* direction compared with the ones in the *x* direction, with intermediate values of  $D_{\alpha}$  for in-between directions. This observation is consistent with the plane (x,y) as the preferred orientation plane.

The pore water diffusion coefficients  $D_{\alpha}$  derived from these different MSD curves are 493 used to draw the water diffusion tensor in Fig. 6 for the two VPM having the closest  $\langle P_2 \rangle$  values 494 to the experimental ones, i.e.,  $\langle P_2 \rangle = 0.24$  and 0.58. The overall  $D_{\alpha}$  values are found to be in 495 fair agreement when comparing the experimental and calculated water diffusion tensors. For 496 497 the compacted sample, the simulated tensor displays a slightly more anisotropic pattern, likely due to the increased  $\langle P_2 \rangle$  value ( $\langle P_2 \rangle = 0.24$  and 0.21 for simulated and experimental media, 498 499 respectively). In Fig. 6, the obtained simulated water diffusion tensor (solid line) is also compared to the tensor (black dotted line) only based on the extracted  $D_x$ ,  $D_y$ , and  $D_z$  values 500 from BD simulations and considering the angular dependence given by Eq. (18). The good 501 502 consistency between the simulated and recalculated water diffusion tensors provides additional 503 evidence of the good representativeness of the VPM as well as the correct consideration of the simulated  $\overrightarrow{e_x}$ ,  $\overrightarrow{e_y}$ , and  $\overrightarrow{e_z}$  directors of the simulation box as eigenvalues of the diffusion tensor. 504 The obtained VPM coupled to BD simulations can be used to further assess the influence of the 505 preferred orientation of particles on the diffusional properties of water. In Fig. 8a, the obtained 506 simulated  $D_{\alpha}/D_0$  values extracted from the MSD curves are plotted in the 0.04-0.92  $\langle P_2 \rangle$  range, 507 whereas the anisotropy ratio  $D_{\overline{x},\overline{y}}/D_z$  of the water diffusion tensor is reported in Fig. 8b. Again, 508 good agreement between the experimental and calculated diffusion coefficients is obtained for 509 both samples. The  $D_{\overline{x},\overline{y}}/D_0$  and  $D_z/D_0$  values show steady increasing and decreasing 510 behaviours, respectively, when increasing the degree of preferred orientation (Fig. 8a). For both 511

directions, a factor ~2 is obtained for *G* values between the more isotropic to more anisotropic systems. This amplitude observed for  $\varepsilon$ =0.5±0.02 is in agreement with the computational results obtained by (Tyagi et al., 2013) using random walk simulations on 2D microstructure maps. Such variation well illustrates the great importance of the preferred orientation on the diffusional properties of water. In the case of transport modelling, according to Eq. (1), the modification of the  $D_{\alpha}/D_0$  values through the investigated  $\langle P_2 \rangle$  range has indeed the same impact on the  $D_e$  value as a change in the porosity value by a factor of 2.



519

**Fig. 8.** Influence of particle preferred orientation on the water diffusion properties. (a) Evolution as a function of the order parameter  $\langle P_2 \rangle$  of the mean water diffusion coefficients in the (x,y) plane  $D_{\overline{x},\overline{y}}$ (circles) or along the *z* direction  $D_z$  (squares), calculated from Brownian dynamics simulations (blue) or obtained experimentally by PGSE-NMR attenuation measurements (red). (b) Evolution of experimental (red) and calculated (blue)  $D_{\overline{x},\overline{y}}/D_z$  ratios with preferred orientation of particles.

The anisotropy ratio  $D_{\overline{x},\overline{y}}/D_z$  of the water diffusion tensor displays an exponential-525 shaped evolution with the  $\langle P_2 \rangle$  value (Fig. 8b). The observed variation of the  $D_{\overline{x},\overline{y}}/D_z$  values 526 between 1 and 5 is consistent with the reported experimental diffusivity anisotropies for water 527 tracers in compacted clay-rich porous media, for pure clay-based materials (e.g., bentonite; Sato 528 and Suzuki, 2003; Suzuki et al., 2004), or for more complex polymineralic argillaceous rocks 529 (Van Loon et al., 2004; García-Gutiérrez et al., 2006; Xiang et al., 2013; Gimmi et al., 2014; 530 531 Jacops et al., 2017). For these latter, the presence of elongated non-clay grains can provide additional contribution to the overall anisotropy in the water diffusion tensor (Robinet et al., 532 2012). 533

## 534 4. Summary and conclusions

The results described in the present study further confirm the significant role played by the 535 preferred orientation of the particles on the diffusional properties of water in compacted porous 536 media made of clay minerals. The originality of this work is provided by a close connection 537 538 between experiments and simulations on representative virtual porous media, allowing quantitative agreement between the measured and calculated  $D_{\alpha}/D_0$  values. It is worth pointing 539 out that such an agreement conceals significant differences in the probed spatial and dynamical 540 domains between the experiments (~5 mm; 20 ms) and simulations (~4 µm; 4 ms). 541 Accordingly, the spatial heterogeneity in the porosity and anisotropy within the samples at 542 higher length scales are not accounted for in the computed VPM used here, or are just averaged 543 544 in such a limited volume. Moreover, the flexibility of the clay particles (Honorio et al., 2018) and the interfacial molecular forces on the dynamics of the water molecules are not considered 545 in the present study in these mesoscopic simulations. Despite these limitations and in line with 546 previous computational studies (Tyagi et al., 2013; Bacle et al., 2016), the simulated VPM can 547 be considered good toy models for estimating the underlying parameters of the pore network 548 controlling the geometric parameter G (i.e., distribution in size, shape, orientation, and 549

connectivity of pores). This estimation is achieved in the present study based on the 550 551 computation of the shape and the mean orientation of the particles, two parameters that are easily obtained experimentally. The validation of the computational methodology provides 552 quantitative information on the role played by the preferred orientation of the particles on the 553 G values, as a variation by a factor up to ~2 of  $D_{\alpha}/D_0$  for  $\varepsilon = 0.5 \pm 0.02$  is observed when 554 increasing the anisotropy in the particle orientation. Because the order parameter  $\langle P_2 \rangle$  values 555 can significantly vary in compacted materials made from clay minerals (Perdigon-Aller et al., 556 2005; Méheust et al., 2006; Perdigón et al., 2007; Lutterotti et al., 2010; Carrier et al., 2016), 557 558 this parameter undoubtedly deserves, as a porosity parameter, particular attention in the scope of predicting the macroscopic transport properties in compacted clay media. A logical 559 continuation of this work could be, then, to expand the analysis of the influence of the preferred 560 orientation on the water diffusion for different  $\varepsilon$  values. Using the same strategy combining 561 experiments and simulations, another perspective could be to extend the analysis to other types 562 of clay minerals, in particular to swelling clay minerals, leading to the presence of different 563 type of porosities, and evaluating the overall diffusional properties of water. 564

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762