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# Chiral Self-Sorting in Truxene-Based Metallacages

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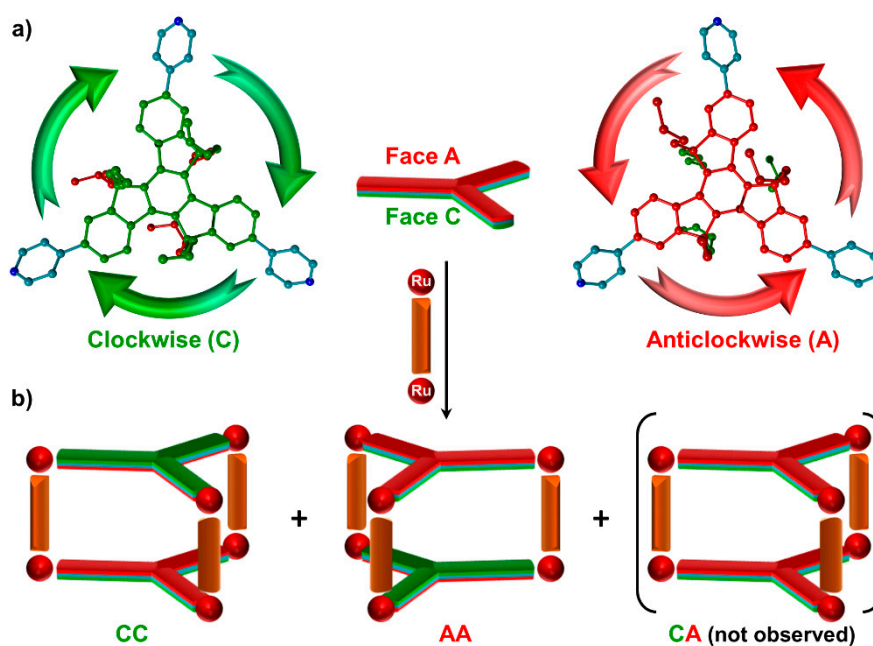
**Abstract:** Two chiral face-rotating metalla-assembled polyhedra were constructed upon self-assembling achiral components, i.e., a tritopic ligand based on a truxene core (10,15-dihydro-5H-diindeno [1,2-a;1',2'-c]fluorene) and two different hydroxyquinonato-bridged diruthenium complexes. Both polyhedra were characterized in solution as well as in the solid state by X-ray crystallography. In both cases, the self-sorting process leading to only two homo-chiral enantiomers was governed by non-covalent interactions between both truxene units that faced each other.

**Keywords:** supramolecular chemistry; self-assembly; self-sorting; metalla-cage; truxene

## 1. Introduction

Coordination-driven self-assembly methodology has been extensively used during the last few decades to construct bi- or tri-dimensional discrete architecture [1–17]. Beyond the challenge of producing more and more sophisticated compounds, special attention has been given to functional architectures that are responsive to an external stimulus such as light [18–22], redox [23–31], or the addition of a chemical [32] in order to target original properties [33–41]. Those systems are generally achiral and highly symmetric. New synthetic methodologies have therefore been developed to allow for chiral metalla-assembled architecture [5,42–46]. One of these methodologies consists of thermodynamically controlling the self-sorting of multiple building blocks [47–49]. In particular, chiral self-sorting can be spontaneously achieved from a racemic mixture upon metal coordination, leading either to homochiral [50–56] or heterochiral [50,51,57,58] assemblies.

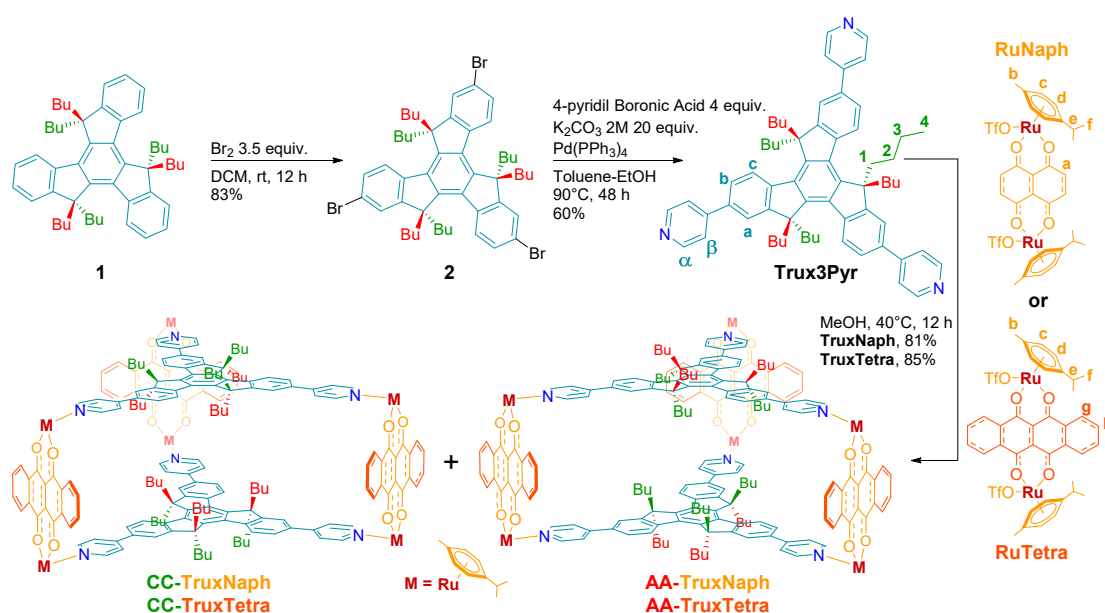
On the other hand, face-rotating polyhedra constitute a recent class of chiral architecture that can be constructed from chiral or achiral linkers and face-rotating component [59–65]. The latter can consist of achiral compounds that exhibit two faces with opposite directionalities such as truxene [62–65], triazatruxene [59,60], and tetraphenylethylene [61]. In particular, the  $C_{3h}$ -symmetric hexa-alkylated truxene derivatives exhibit both clockwise (C; green) and anti-clockwise (A; red) faces that are defined by the rotation of the three  $sp^3$  methylene bridges along the  $C_3$  axis (Figure 1a). Upon self-assembly with achiral hydroxyquinonato-bridged diruthenium complexes, the formation of sandwich-type supramolecular cages is expected (Figure 1b). In those cages, the truxene ligands lose their mirror symmetry, thus resulting in several possible stereoisomers. We herein show that through-space interactions between both face-rotating ligands guide the self-assembly process toward only the CC/AA enantiomers couple, whereas the hetero-chiral CA stereoisomer is not observed in this process.



**Figure 1.** (a) Truxene-based ligand (X-ray) presenting two rotational faces (clockwise (C), green face, and anticlockwise (A), red face) and (b) the three possible metalla-cage structures obtained upon self-assembling with bis-ruthenium complexes.

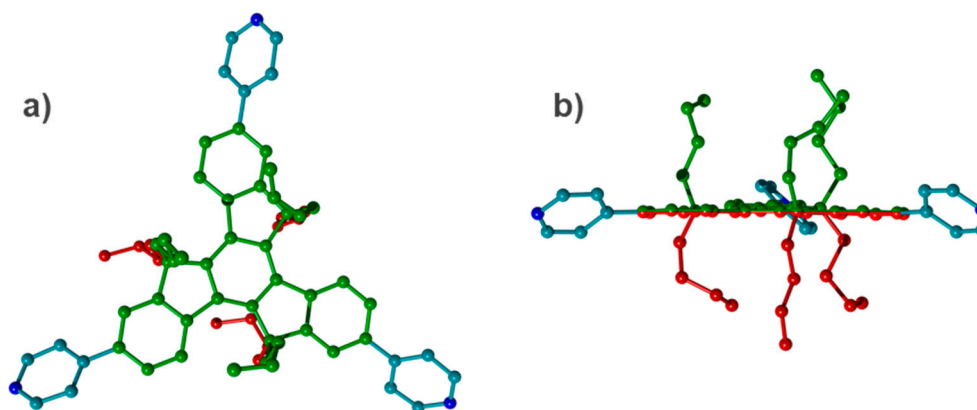
## 2. Results and Discussion

The **Trux3Pyr** ligand was synthesized in two steps from hexabutyl truxene derivative **1** in a good yield (Scheme 1) through the adaptation of described procedures [66–68]. Butyl chains were selected to ensure the good solubility of the target ligand and to prevent aggregation. The first step consisted of the regioselective tri-bromination of the truxene core in the presence of  $\text{Br}_2$ , thus affording compound **2** in an 83% yield. The target **Trux3pyr** ligand was then obtained in a 60% yield after purification through a palladium-catalyzed Suzuki–Miyaura cross-coupling reaction with 4-pyridinylboronic acid (Figures S3–S9).



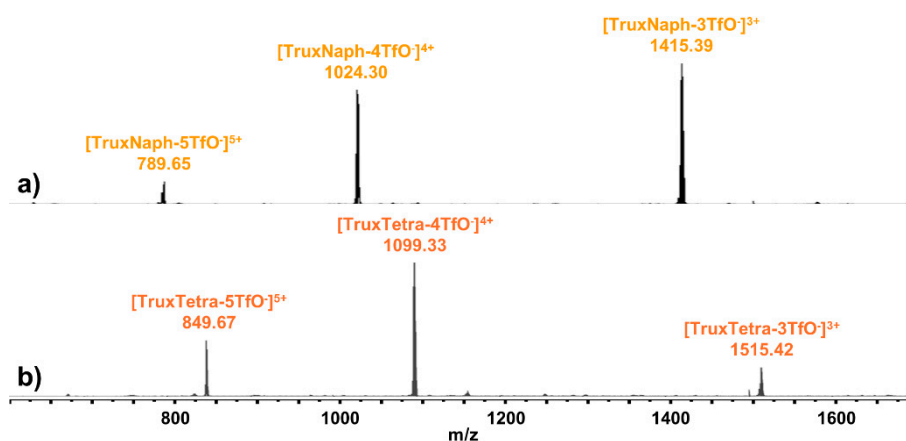
**Scheme 1.** Synthesis of the **Trux3Pyr** ligand and the CC-TruxNaph, AA-TruxNaph, CC-TruxTetra, and AA-TruxTetra metalla-cages.

Single crystals of the **Trux3Pyr** ligand that were suitable for X-ray analyses were obtained by the slow evaporation of a  $\text{CH}_2\text{Cl}_2/n$ -hexane solution, and the corresponding solid-state crystal structure is depicted in Figure 2. Thanks to the planar geometry of the central truxene core, the three peripheral nitrogen were found to be in the same plane. The pyridine moieties are twisted around the  $\text{C}_{\text{trux}}\text{-C}_{\text{pyr}}$  axis with an average angle of  $40.8(1)^\circ$ , thus minimizing H–H interactions, as reported for similar compounds [69]. An angle of  $120^\circ$  was found between each pyridine axis in accordance with the  $\text{C}_3$  symmetry of the **Trux3Pyr** ligand. It was found that the alkyl chains tend to lie away perpendicularly to the truxene plane.



**Figure 2.** X-ray crystal structure of **Trux3Pyr**: (a) top view (clockwise face in green) and (b) lateral view showing both rotating faces (clockwise in green and anticlockwise in red).

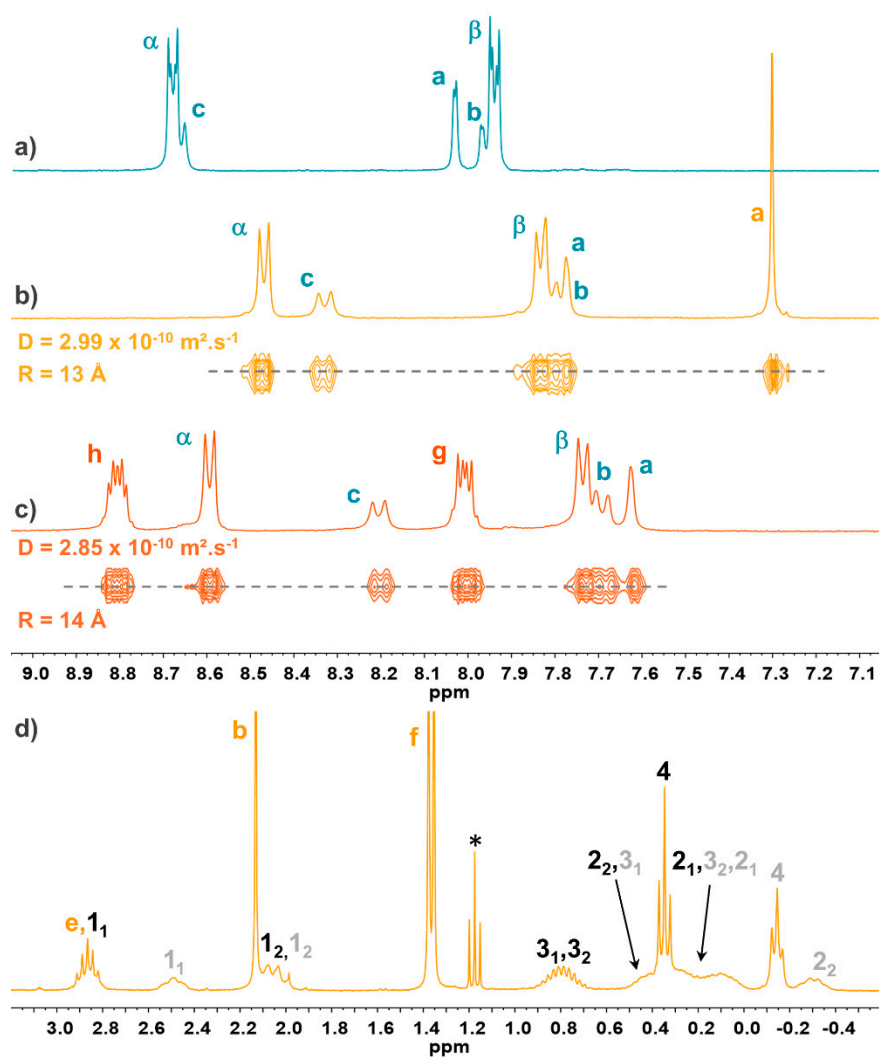
The self-assembly processes between the face-rotating **Trux3Pyr** ligand (two equivalents) and the bis-ruthenium acceptors (three equivalents), **RuNaph** or **RuTetra**, were followed by  $^1\text{H}$  NMR in MeOD at  $C = 3 \times 10^{-3}$  M. These ruthenium complexes were selected for their accessible syntheses, their rigidity, and their intermetallic distance of ca.  $8.3 \text{ \AA}$ , a value that we considered sufficient for the six alkyl chains to fit inside the cavity. Note that the reaction of **Trux3Pyr** with the smaller oxalate bis-ruthenium complex (Ru–Ru distance:  $5.5 \text{ \AA}$ ) [70] did not converge into a discrete species. After one night at  $50^\circ\text{C}$  with the use of **RuNaph** or **RuTetra**, the spectra became symmetrical and well-resolved. The resulting products were then isolated by precipitation with  $\text{Et}_2\text{O}$ . For both samples, high-resolution ESI-FTICR-MS spectrometry experiments were carried out in MeOH at  $C = 10^{-4}$  M (Figure 3). The corresponding spectra confirmed an  $\text{M}_6\text{L}_2$  stoichiometry in both cases, with characteristic multi-charged isotopic patterns localized at  $m/z = 789.6516$ ,  $1024.3024$ , and  $1415.3877$  (main contributions, Figure S16) and at  $m/z = 849.6704$ ,  $1099.3257$ , and  $1515.4192$  (main contributions, Figure S17) for the compounds isolated from the reaction of **Trux3Pyr** with **RuNaph** and with **RuTetra**, respectively.



**Figure 3.** ESI-FTICR mass spectra recorded in MeOH ( $C = 10^{-4}$  M) of: (a) **TruxNaph** and (b) **TruxTetra**.

The  $^1\text{H}$  NMR,  $^1\text{H}$  COSY NMR and  $^1\text{H}$  DOSY NMR spectra of both self-assembled cages are shown in Figure 4 and Figures S10–S15. Both the **TruxNaph** and **TruxTetra**  $^1\text{H}$  DOSY spectra exhibited only one set of signals, with  $D$  values of 2.99 and  $2.85 \times 10^{-10} \text{ m}^2 \cdot \text{s}^{-1}$ , respectively, meaning that they had similar sizes. The corresponding hydrodynamic radii estimated from the Stokes–Einstein equation [71] were found to be 13 and 14 Å ( $T = 298 \text{ K}$ ). As expected, these values were much larger than the one calculated for the **Trux3Pyr** ligand ( $D = 4.78 \times 10^{-10} \text{ m}^2 \cdot \text{s}^{-1}$ , MeOD,  $R = 8 \text{ Å}$ , Figure S9) and were in good agreement with the formation of  $\text{M}_6\text{L}_2$  self-assemblies. Compared to the **Trux3Pyr** ligand, both isolated discrete assemblies showed upfield shifted  $\text{H}_\alpha$  and  $\text{H}_\beta$  protons after coordination to the ruthenium complex (Figure 4), with chemical shifts in accordance with reported values [72]. Each  $^1\text{H}$  NMR spectrum revealed only one set of signals for all **Trux3Pyr** protons ( $\text{H}_\alpha$ ,  $\text{H}_\beta$ ,  $\text{H}_a$ ,  $\text{H}_b$  and  $\text{H}_c$ ), this indicating the absence of a mixture of diastereoisomers. Therefore, the reaction between the **Trux3Pyr** ligand and the bis-ruthenium complexes afforded either a mixture of  $D_3$  symmetric CC and AA enantiomers or the  $C_3$  symmetric CA (or AC) stereoisomer alone. This selectivity suggests that a through-space communication existed between both facing panels and orientated the self-assembly process through a chiral self-sorting. This behavior can be compared to what is reported for, e.g., chiral helicates, for which a strong mechanical coupling between the metal units orientates the self-assembly process [43,73]. Additional 2D COSY experiments (Figures S11 and S14) allowed for the assignation of all butyl chains protons. One should note that those located on carbons 1, 2 and 3 of the starting **Trux3Pyr** ligand (see Scheme 1) are diastereotopic [74], two different signals being therefore observed for each methylene group, as illustrated in Figures S1, S3 and S7. In the case of the **TruxNaph** and **TruxTetra** cages, two sets of butyl chains were observed, i.e., those pointing outside the cavity and those pointing inside, more shielded [8], therefore generating 12 signals for the protons located on carbons 1, 2 and 3 (Figure 4d).

Single crystals were obtained for both self-assemblies from the slow diffusion of methyl *tert*-butyl ether in solutions of **TruxNaph** and **TruxTetra** in MeOH. In each case, several crystals were analyzed and gave the same result (Figure 5). Both complexes crystallized in non-centrosymmetric trigonal chiral space groups,  $P3_121$  and  $P3_221$ , respectively, for **TruxNaph** and **TruxTetra**, and exhibited very large unit cells volume of ca.  $23,500 \text{ Å}^3$ . Their Flack parameters were of 0.20 and 0.42, which indicated enantio-enriched crystals. An analysis of the crystal structures gave evidence of chiral self-sorting along the self-assembly process, since only AA and CC enantiomers were present in the solid state. Both self-assemblies exhibited a similar geometric arrangement, organized from two approximatively planar truxene moieties that faced each other. Contrary to what is usually observed for similar metalla-cages in which the two facing polyaromatic ligands tend to get closer to maximize the pi–pi interactions and structure compactness thanks to a large tilt of the lateral ruthenium-based panels [75,76], the six butyl chains present inside the cavity of **TruxNaph** and **TruxTetra** maintained, in this case, both truxene moieties at distances of 7.30 and 7.58 Å, respectively (average value between mean planes). These values were nevertheless smaller than the Ru–Ru distance (8.30 and 8.33 Å) in the bis-ruthenium units of **TruxNaph** and **TruxTetra**, meaning a tilt of the lateral ruthenium side panels of ca.  $12^\circ$  out of the truxene planes. As a consequence, the trigonal prisms were slightly deformed and showed average Bailar twist angles of  $7.94$  and  $8.11^\circ$ , respectively, (Figure S18) [77]. While the external butyl chains remained nearly linear, those located inside the polyhedra were bent upon confinement within the cavity. Their role was of critical importance in the self-sorting process since they allowed for inter-ligand communication. Indeed, the metalla-prism distortion was minimized when the six butyl chains pointing inside the cavity were alternated. This was the case only for the AA and CC geometries. On the other hand, a self-assembly of two opposite faces (e.g., a CA geometry) would necessitate a much larger tilt of the lateral ruthenium panels in order to preserve the regular alternation, explaining why only the AA and CC enantiomers were observed.



**Figure 4.**  $^1\text{H}$  NMR (298 K,  $C = 10^{-3}$  M, MeOG) downfield region of (a) the **Trux3Pyr** ligand, (b) **CC-** and **AA-TruxNaph**, (c) **CC-** and **AA-TruxTetra**, and highfield region of (d) **CC-** and **AA-TruxNaph**. See Scheme 1 for  $^1\text{H}$  NMR assignments. Grey and black assignments correspond to the inner cavity and cavity protons, respectively. D corresponds to the diffusion coefficient extracted from an  $^1\text{H}$  DOSY NMR experiment, and R to the corresponding hydrodynamic radii calculated from the Stokes–Einstein equation. \* Residual  $\text{Et}_2\text{O}$ .

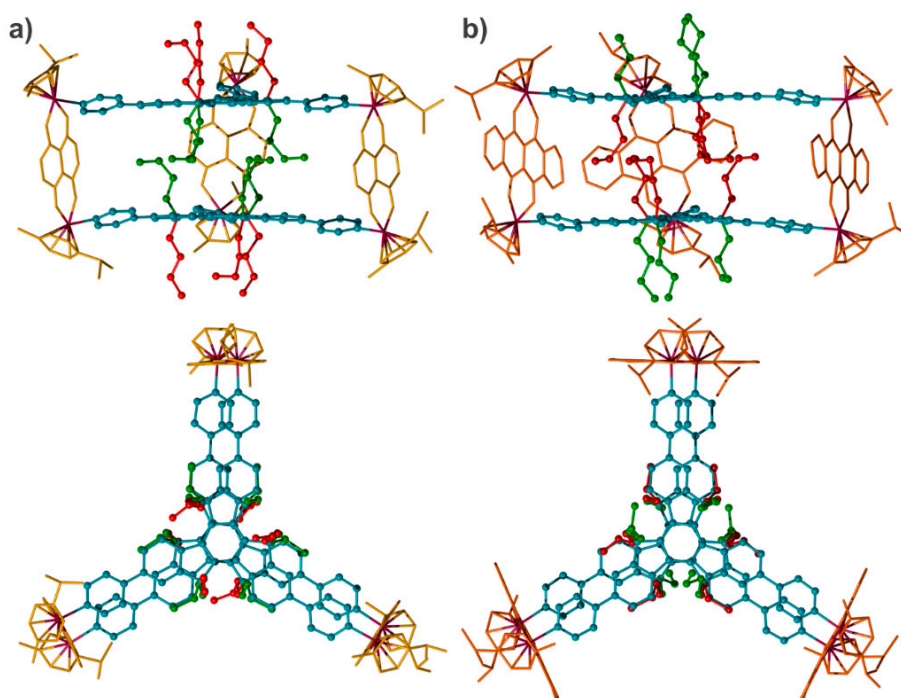


Figure 5. X-ray crystal structures of (a) AA-TruxNaph and (b) CC-TruxTetra.

### 3. Materials and Methods

#### 3.1. Chemicals

Hexa-alkylated truxene **1** [68], as well as the RuNaph [78] and RuTetra [78] complexes, were synthesized by using procedures described in the literature. All reagents were of commercial reagent grade and were used without further purification. Silica gel chromatography was performed with a SIGMA Aldrich Chemistry SiO<sub>2</sub> (pore size 60 Å, 40–63 µm technical grades) (Sigma-Aldrich, Steinheim, Germany).

#### 3.2. Instrumentation

Characterizations and NMR experiments were carried out on an NMR Bruker Avance III 300 spectrometer or an NMR Bruker Avance III HD 500 spectrometer (Bruker, Wissembourg, France) at 298 K by using perdeuterated solvents. <sup>1</sup>H DOSY NMR spectra were analyzed with the MestReNova software (12.0.1, Mestrelab Research, Santiago de Compostela, Spain). MALDI-TOF-MS spectra were recorded on a MALDI-TOF Bruker Biflex III instrument (Bruker, Wissembourg, France) by using a positive-ion mode. Very high resolution ESI-FTICR mass spectra were performed in positive detection mode on a 7T Solarix 2xR (Bruker Daltonics, Wissembourg, France).

#### 3.3. Experimental Procedure and Characterization Data

##### 2,7,12-tribromo-5,5,10,10,15,15-hexabutyl-10,15-dihydro-5H-diindeno[1,2-a:1',2'-c]fluorene (**2**)

Compound **2** was synthesized after the modification of a reported procedure [68]. Bromine (0.1 mL, 1.94 mmol, 3.5 equivalents) was added to a stirred suspension of truxene **1** (370 mg, 0.545 mmol) in dichloromethane (20 mL) over a 5 min period at room temperature with protection from light. After one night, methanol (50 mL) was added. The resulting precipitate was filtered off and washed with methanol, diethyl ether, and pentane to give compound **2** (412 mg, 0.445 mmol, 83%) as a pale yellow solid. <sup>1</sup>H NMR (500 MHz, 298 K, CDCl<sub>3</sub>): δ 8.20 (d, *J* = 8.3 Hz, 3H), 7.57 (d, *J* = 2.1 Hz, 3H), 7.52 (dd, *J* = 8.3 Hz, *J* = 2.1 Hz, 3H), 2.89–2.83 (m, 6H), 2.07–2.01 (m, 6H), 0.98–0.81 (m, 12H), 0.58–0.33 (m, 30H).

$^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  155.89, 145.04, 138.87, 137.71, 129.41, 125.99, 125.61, 121.09, 55.93, 36.50, 26.47, 22.76, and 13.80. HRMS: found: 912.2487; calculated: 912.2480.

### Trux3Pyr Ligand

An aqueous solution of  $\text{K}_2\text{CO}_3$  (0.64 g in 2.5 mL, 4.63 mmol, 20 equivalents) and 4-pyridinylboronic acid (86 mg, 0.699 mmol, 3.2 equivalents) were added to a stirred suspension of **2** (200 mg, 0.218 mmol) in toluene (7 mL) and ethanol (3.5 mL). The mixture was degassed by bubbling argon for 30 min. Then, tetrakis(triphenylphosphine)palladium (0) (70 mg, 0.060 mmol) was added, and the mixture was vigorously stirred and heated to 90 °C. After 48 h, the mixture was cooled to room temperature and extracted with dichloromethane. The organic extracts were washed with water and dried over magnesium sulfate, and then the solvent was evaporated. The residue was purified by chromatography on silica gel by using dichloromethane/ethyl acetate/methanol/triethylamine (from 99/0/0/1 *v/v/v/v* to 47/47/5/1 *v/v/v/v*) to give **Trux3Pyr** (119 mg, 60%) as a pale yellow solid.  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ ):  $\delta$  8.73 (d,  $J = 8.7$  Hz, 6H), 8.51 (d,  $J = 8.7$  Hz, 3H), 7.77 (s, 3H), 7.75 (d,  $J = 8.7$  Hz, 3H), 7.68 (d,  $J = 8.7$  Hz, 6H), 3.09–2.99 (m, 6H), 2.26–2.17 (m, 6H), 1.00–0.83 (m, 12H), 0.68–0.54 (m, 12H), 0.47 (t,  $J = 8.7$  Hz, 18H).  $^{13}\text{C}$  NMR (76 MHz,  $\text{CDCl}_3$ ):  $\delta$  154.57, 150.29, 148.28, 146.28, 141.17, 138.00, 136.17, 125.32, 125.24, 121.57, 120.54, 55.97, 36.78, 26.60, 22.86, 13.85. HRMS: found: 910.6043; calculated: 909.5961.

### TruxNaph Self-Assembly

A mixture of **Trux3Pyr** (10 mg, 11  $\mu\text{mol}$ , 2 equivalents) and the **RuNaph** complex (15.78 mg, 16.5  $\mu\text{mol}$ , 3 equivalents) in methanol (2 mL) was stirred overnight at 50 °C. Then, diethyl ether (5 mL) was added, and the resulting suspension was centrifuged and washed two times with diethyl ether to give **TruxNaph** (21 mg, 4.5  $\mu\text{mol}$ , 81%) as a dark solid.  $^1\text{H}$  NMR (300 MHz, MeOD):  $\delta$  8.47 (d,  $J = 6.3$  Hz, 12H), 8.33 (d,  $J = 8.4$  Hz, 6H), 7.83 (d,  $J = 6.3$  Hz, 12H), 7.79 (m, 12H), 7.30 (s, 12H), 5.87 (d,  $J = 6.1$  Hz, 12H), 5.64 (d,  $J = 6.1$  Hz, 12H), 2.91–2.82 (m, 12H), 2.54–2.44 (m, 12H), 2.13 (s, 18H), 1.99–2.08 (m, 12H), 1.37 (d,  $J = 6.9$  Hz, 36H), 0.86–0.74 (m, 12H), 0.48–0.03 (m, 30H), 0.35 (t,  $J = 7.3$  Hz, 18H),  $-0.14$  (t,  $J = 7.0$  Hz, 18H),  $-0.24$ –( $-0.34$ ) (m, 6H).  $^1\text{H}$  DOSY NMR (300 MHz, MeOD)  $D = 2.99 \times 10^{-10} \text{ m}^2 \cdot \text{s}^{-1}$ . FTICR-HRMS ( $m/z$ ), [**TruxNaph-3TfO**] $^{3+}$ : found: 1415.38771; calculated 1415.38762, [**TruxNaph-4TfO**] $^{4+}$ : found: 1024.30249; calculated 1024.30257, [**TruxNaph-5TfO**] $^{5+}$ : found: 789.65157; calculated 789.65155.

### TruxTetra Self-Assembly

A mixture of **Trux3Pyr** (13 mg, 14.3  $\mu\text{mol}$ , 2 equivalents) and the **RuTetra** complex (22.66 mg, 21.4  $\mu\text{mol}$ , 3 equivalents) in methanol (2.5 mL) was stirred overnight at 50 °C. Then, diethyl ether (5 mL) was added, and the resulting suspension was centrifuged and washed two times with diethyl ether to give **TruxTetra** (22 mg, 4.4  $\mu\text{mol}$ , 85%) as a dark solid.  $^1\text{H}$  NMR (300 MHz, MeOD)  $\delta$  8.83–8.79 (m, 12H), 8.59 (d,  $J = 6.2$  Hz, 12H), 8.21 (d,  $J = 8.4$  Hz, 6H), 8.04–7.98 (m, 12H), 7.74 (d,  $J = 6.2$  Hz, 12H), 7.68 (d,  $J = 8.4$  Hz, 6H), 7.63 (s, 6H), 6.04–6.01 (m, 12H), 5.80–5.78 (m, 12H), 3.01–2.92 (m, 6H), 2.80–2.70 (m, 6H), 2.38–2.28 (m, 6H), 2.23 (s, 18H), 2.01–1.82 (m, 12H), 1.37 (d,  $J = 6.9$  Hz, 36H), 0.85–0.64 (m, 12H), 0.30 (t,  $J = 7.3$  Hz, 18H), 0.28–0.11 (m, 18H), 0.02–( $-0.15$ ) (m, 12H),  $-0.36$  (t,  $J = 7.1$  Hz, 18H),  $(-0.34)$ –( $-0.40$ ) (m, 6H).  $^1\text{H}$  DOSY NMR (300 MHz, MeOD)  $D = 2.85 \times 10^{-10} \text{ m}^2 \cdot \text{s}^{-1}$ . FTICR-HRMS ( $m/z$ ), [**TruxTetra-3TfO**] $^{3+}$ : found: 1515.41921; calculated 1515.41892, [**TruxTetra-4TfO**] $^{4+}$ : found: 1099.32576; calculated 1099.32605, [**TruxTetra-5TfO**] $^{5+}$ : found: 849.67041; calculated 849.67033.

### 3.4. X-Ray Crystallographic Analysis

X-ray single-crystal diffraction data were collected at 150 K on an Agilent SuperNova diffractometer equipped with an Atlas CCD detector and micro-focus Cu  $K\alpha$  radiation ( $\lambda = 1.54184 \text{ \AA}$ ). The structures were solved by direct methods, expanded, and refined on  $F^2$  by full matrix least-squares techniques by using *SHELX* (G.M. Sheldrick, 2008–2016) package. All non-hydrogen atoms were anisotropically refined, and the H atoms were included in the calculation without refinement. Multiscan empirical absorption was corrected with the *CrysAlisPro* program (*CrysAlisPro* 1.171.38.41r, Rigaku Oxford Diffraction, 2015).



The three structure refinements showed disordered electron density that could not be reliably modeled, and the program PLATON/SQUEEZE was used to remove the corresponding scattering contribution from the intensity data. This electron density can be attributed to solvent molecules (*n*-hexane for **Trux3pyr**, methanol for **TruxNaph** and **TruxTetra**) and missing triflate molecules (CF<sub>3</sub>SO<sub>3</sub> anions for **TruxNaph** and **TruxTetra**). The assumed solvent composition and missing anion molecules were included in the calculation of the empirical formula, formula weight, density, linear absorption coefficient, and F(000).

Crystallographic data for **Trux3pyr**: C<sub>138</sub>H<sub>164</sub>N<sub>6</sub>, M = 1906.75, colorless prism, 0.203 × 0.192 × 0.047 mm<sup>3</sup>, monoclinic, space group Pn, *a* = 11.7717(4) Å, *b* = 28.966(1) Å, *c* = 16.9134(4) Å, β = 91.318(2)°, *V* = 5765.6(3) Å<sup>3</sup>, *Z* = 2, ρ<sub>calc</sub> = 1.098 g/cm<sup>3</sup>, μ = 0.470 mm<sup>-1</sup>, F(000) = 2068, θ<sub>min</sub> = 3.03°, θ<sub>max</sub> = 76.65°, 25,637 reflections collected, 15,430 unique (R<sub>int</sub> = 0.0459), parameters/restraints = 1247/4, R1 = 0.0759 and wR2 = 0.2084 using 12,872 reflections with I > 2σ(I), R1 = 0.0869 and wR2 = 0.2305 using all data, GOF = 1.058, absolute structure parameter = 0.1(8), -0.275 < Δρ < 0.454 e.Å<sup>-3</sup>. CCDC 1956952.

Crystallographic data for **TruxNaph**: C<sub>268</sub>H<sub>418</sub>F<sub>18</sub>N<sub>6</sub>O<sub>70</sub>Ru<sub>6</sub>S<sub>6</sub>, M = 5984.84, black prism, 0.262 × 0.201 × 0.157 mm<sup>3</sup>, trigonal, space group P3<sub>1</sub>21, *a* = 25.1943(5) Å, *b* = 25.1943(5) Å, *c* = 42.5843(7) Å, α = 90°, β = 90°, γ = 120°, *V* = 23409(1) Å<sup>3</sup>, *Z* = 3, ρ<sub>calc</sub> = 1.274 g/cm<sup>3</sup>, μ = 3.364 mm<sup>-1</sup>, F(000) = 9450, θ<sub>min</sub> = 2.90°, θ<sub>max</sub> = 72.812°, 124745 reflections collected, 30645 unique (R<sub>int</sub> = 0.0277), parameters/restraints = 1107/45, R1 = 0.0503 and wR2 = 0.1546 using 15,980 reflections with I > 2σ(I), R1 = 0.0771 and wR2 = 0.1904 using all data, GOF = 0.945, absolute structure parameter = 0.447(9), -0.233 < Δρ < 0.200 e.Å<sup>-3</sup>. CCDC 1956951.

Crystallographic data for **TruxTetra**: C<sub>292</sub>H<sub>418</sub>F<sub>18</sub>N<sub>6</sub>O<sub>70</sub>Ru<sub>6</sub>S<sub>6</sub>, M = 6273.10, black prism, 0.35 × 0.34 × 0.31 mm<sup>3</sup>, trigonal, space group P3<sub>2</sub>21, *a* = 25.0842(6) Å, *b* = 25.0842(6) Å, *c* = 43.3591(10) Å, α = 90°, β = 90°, γ = 120°, *V* = 23627(1) Å<sup>3</sup>, *Z* = 3, ρ<sub>calc</sub> = 1.323 g/cm<sup>3</sup>, μ = 3.360 mm<sup>-1</sup>, F(000) = 9882, θ<sub>min</sub> = 2.27°, θ<sub>max</sub> = 72.88°, 1,250,008 reflections collected, 30,698 unique (R<sub>int</sub> = 0.0493), parameters/restraints = 1203/39, R1 = 0.0605 and wR2 = 0.1777 using 19,178 reflections with I > 2σ(I), R1 = 0.0762 and wR2 = 0.1956 using all data, GOF = 0.965, absolute structure parameter = 0.204(7), -0.387 < Δρ < 0.469 e.Å<sup>-3</sup>. CCDC 1956958.

#### 4. Conclusions

In summary, two metalla-cages were synthesized from a hexa-alkylated truxene ligand and hydroxyquinonato-bridged diruthenium complexes. While all starting components were achiral, the resulting **TruxNaph** and **TruxTetra** ruthenium based self-assemblies were chiral. Remarkably, only the homo-directional CC and AA enantiomers were observed in solution and in the solid state, an observation which is explained by the combination of the space filling between both facing alkylated truxene units and the tilt value of the lateral ruthenium panels. Work is now under progress to extend this approach toward interlocked supramolecular systems.

**Supplementary Materials:** The following are available online at <http://www.mdpi.com/2304-6740/8/1/1/s1>, NMR spectra, cif files and check cif files of **Trux3pyr**, **TruxNaph** and **TruxTetra**.

**Author Contributions:** M.S. and S.G. conceived the study, designed the experiments, analyzed the data, and wrote the paper; S.S., A.L., C.D. and A.B. performed the experiments; M.A. performed X-ray analyses; V.C. and F.A. performed ESI-FTICR analyses. All authors have read and agreed to the published version of the manuscript.

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