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# The third life of CSP : Solar high temperature processes

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Developed in the 80's to produce electricity, Concentrated Solar Power (CSP) is today more expensive than photovoltaic panels (PV) for direct production during sunny hours. CSP keeps a competitive advantage if coupled with thermal storage that permits to meet daily electricity production and user needs.

The third step of CSP is chemical storage from high temperature processes. Produced solar fuels present a great energetic density and allow to tackle solar spatial

heterogeneity and temporal variation.

#### Candidate reactions for high temperature solar processes

High temperatures reachable by concentrated solar processes, above 800°C, permit to carry out endothermic reactions which are presented in Figure 1

The exergy is stored in the products of the reaction : materials, (Ca0, Fe, ...), materials for energy (Zn/02 batteries...), fuel (syngas, hydrogen...). Some of these reactions produces both fuels and materials (i.e. metal oxide methano-reduction)



processes. Red blocks shows the total cost of hydrogen production. It is based on the energy needs of reaction, assuming that thermodynamic equilibrium is achieved, and considering thermal losses and ideal product separation.



Figure 1 : recapitulative scheme of candidate for high temperature processes (Bizeau, O.)

Figure 2 shows an estimation of the cost of hydrogen production for different reaction. Thermal decomposition of H<sub>2</sub>S is economically the best candidate, but is technically challenging. Gasification of biomass is the best compromise. Figure 3 shows the influence of hybridization on the cost of hydrogen production. Hybrid process seems to be more cost effective.

Because of the variability of solar energy, it's possible to hybridize the process: during the night or a cloudy day, the energy is provided by combustion of fossil fuel carbonaceous material.



Figure 3 : Influence of hybridization on the cost of hydrogen production by gasification of biomass. Hybridization is managed by combustion of  $CH_4$ 





- · There can be a fluid passing through to fluidize
- the bed

A fluidized bed presents a better mass/energy transfer but





heated tubes.

The tube can be transparent, so the reactant are heated directly

Good mass distribution and energy transfer provided that turbulent flow can be

achieved The tubes can be obstructed by condensed byproducts.



 Good mass and heat transfer, complex flow.



Gravity distribution: · Particles go down through the

- reaction zone by gravity.
- Simple construction. Short residence time
- (8)

#### How to maximize the solar energy amount in the products: reactor design key points

- Minimize thermal loss, reducing the reaction zone's and absorber's temperature, Allow a good product separation and avoid product losses.
- maximize heat transfer to the reactants.
- Ensure good temperature distribution in reaction zone.
- Lead a free-flowing transport of reactant in the reaction zone (avoid fouling).
- Reduce construction cost with a optimized design matching the mechanical needs. •
- Ensure reliable process and materials to reduce operating cost.
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