

New determination of the 229Th half-life

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40 The ²²⁹Th massic activity was confirmed by isotope dilution alpha spectrometry measurements.

The ²²⁹Th half-life calculated from the activity measurements is (7825 \pm 87) years ($k = 2$).

42 **Key Words**

²²⁹ Th, half-life, liquid scintillation counting, NaI counting, isotope dilution alpha spectrometry

44 **Introduction**

45 The ²²⁹Th isotope is a relatively long-lived nuclide in the ²³⁷Np decay series and primarily occurs as a decay product of ²³³U [1]. The isotope decays by α emission to ²²⁵Ra which, in turn, decays 47 to ²²⁵Ac followed by a series of short-lived nuclides to ²⁰⁹Bi (Fig. 1). The ²²⁹Th isotope does not 48 occur in nature making it useful to the environmental and geological sciences where it is used as 49 a yield tracer for thorium α spectrometry [2] and as an isotope dilution mass spectrometry 50 (IDMS) spike for thorium amount measurements [3]. In addition, 229 Th is a source of high-purity 2^{225} Ac for research in nuclear medicine [4] and is being evaluated for use as a high accuracy 52 nuclear clock [5]. The nuclide is particularly important for nuclear forensics where it is used as 53 an IDMS spike for 230 Th "dating" of uranium materials associated with the nuclear fuel cycle, 54 e.g. [6,7]. Due to ongoing interest in the radionuclide, the National Institute of Standards and 55 Technology (NIST) has produced four issues of 229 Th Standard Reference Material (SRM) 56 certified for massic activity ($Bq g^{-1}$). These standards are SRM 4328 issued in 1984, SRM 57 4328A issued in 1995, SRM 4328B issued in 1996, and SRM 4328C issued in 2008 [8-11].

Fig 1. Simplified decay scheme for 229 Th. The uncertainties for the half-lives are standard 60 uncertainties. Half-life and decay mode data for 229 Th are from [12]. Half-life and decay mode 61 data for all other isotopes are from [13].

62

 \overline{a}

63 To date, four half-life measurements have been published for 229 Th, but these values have been 64 somewhat inconsistent. Hagemann *et al.* [1] published a ²²⁹Th half-life of $(7340 \pm 160)^1$ years as 65 part of a study of the ²³³U decay chain. This value is significantly shorter than subsequent

¹ Uncertainties for values in text are combined standard uncertainties (u_c) unless otherwise indicated with a coverage factor value (*k*) of 2 or greater for the expanded uncertainties ($U = k u_c$).

66 measurements (\approx 7%) with the discrepancy attributed to a combination of factors including a 67 biased ²³³U half-life value used to calculate the quantity of ²²⁹Th in the analysis samples, a 68 possible underestimate of ²²⁸Th contributions to ²²⁹Th activity measurements, or ²³³U 69 interference in ²²⁹Th activity measurements [14-15]. Goldstein *et al.* [14] performed a study to 70 refine the ²²⁹Th half-life value by measuring the molality of ²²⁹Th in a solution with a well-71 characterized massic activity, SRM 4328. The resulting measurement indicated a half-life of 72 (7880 \pm 120) years ($k = 2$). More recently, Kikunaga *et al.* [15] calculated a half-life of (7932 \pm 55) years ($k = 2$) based on measured $A(^{233}U)/A(^{232}U)$ and $A(^{229}Th)/A(^{228}Th)$ activity ratios. 74 Finally, Varga *et al.* [16] measured the molality of the SRM 4328C massic activity standard by 75 IDMS using two different Th spikes. One spike was a commercially prepared natural Th solution 76 [Method A: (7889 ± 32) years] and the second spike was a high purity U material with a z 7 calculated quantity of ingrown ²³⁰ Th [Method B: (7952 \pm 36) years]. The half-lives calculated for 78 each set of IDMS measurements were combined for a reported half-life value of (7917 ± 48) 79 years $(k = 2)$.

80 Fitzgerald *et al.* [17] describe a massic activity calibration for the ²²⁹Th Master (M) solution used 81 to prepare SRM 4328C. To produce the stock solution for the SRM, an aliquot of the M solution 82 was quantitively diluted by a factor (D_s) of 308.85 ± 0.09. The ²²⁹Th massic activity of the M 83 solution was then measured by live-timed $4\pi\alpha\beta$ -γ anticoincidence counting. The massic activity 84 of the SRM solution was calculated using the measured massic activity of the M solution and the 85 gravimetrically determined dilution factor. Relative liquid scintillation (LS) counting of the M 86 and SRM 4328C solutions was used to verify the gravimetric dilution factor for the SRM, with 87 the relative difference between the activity ratio and the dilution factor of 0.04 % and a 88 combined standard uncertainty of about 0.4 % in the activity ratio. The calculated activity value

103 reference material (referred to here as 229 Th RM) by direct comparison with the SRM 4328C 104 massic activity standard and the confirmation of the measured value by isotope dilution alpha spectrometry (IDAS). A new value for the ²²⁹Th half-life is calculated from the measured ²²⁹Th 106 massic activity and the previously measured 229 Th molality.

l

 2^2 The acronym CIEMAT/NIST refers to the two laboratories that collaborated in developing the protocol for this LS tracing methodology; viz., the Centro de Investigaciones Energéticas, Medioambientales y Tecnológicas (CIEMAT) and the National Institute of Standards and Technology (NIST).

³ The dilution factor for the ²²⁹Th RM (D_X) is only used for comparison purposes. During the dilution of the M solution to create the 229 Th RM stock solution, the masses measured on a largecapacity Jupiter 3000 balance appear to be inconsistent with normal operation of the balance. Accordingly, the calculated dilution factor value is an approximation.

Direct Activity Comparison

109 A massic activity value for 229 Th in the 229 Th RM was determined by measuring the ratio of 110 relative activities (i.e. activity per gram solution) for the ²²⁹Th RM solution and the SRM 4328C activity standard. Three counting methods were used to make the relative activity measurements. The first method compared the relative LS counting rates for carefully prepared sources of the two solutions. Then, the relative integral photon emission rates of the LS counting sources were directly compared using a NaI(Tl) well counter. Finally, a standard addition LS counting method was used to determine the activity ratio from a series of mixtures of the two reference materials.

Relative LS Counting

118 The ²²⁹Th decay series consists of a seven-nuclide-long decay chain (Fig. 1) with an LS 119 efficiency of > 6.9 s⁻¹ Bq⁻¹. To achieve accurate results for the comparative LS counting 120 measurements, it is imperative that the two 229 Th solutions are in identical states of radioactive equilibrium and that the solutions were counted with identical detection efficiency. Nearly identical counting efficiency was achieved by gravimetrically determining all of the components in the counting source cocktails to assure similar quenching. Identical radioactive equilibrium is also a reasonable assumption when the history and handling of the solutions are considered. 125 Both Th solutions were prepared from quantitative dilutions of aliquots from the same M 126 solution; both solutions were diluted with 1 mol L^{-1} HNO₃; and no additional separations or processing were performed on either solution prior to sealing in glass ampoules. Ampoules of both reference materials were opened specifically for the activity measurements performed in 129 April 2017, however, the SRM 4328C solution was prepared, dispensed into ampoules, and 130 sealed in October 2007 while the ²²⁹Th RM was prepared and sealed in March 2012. Although it 131 is not possible to categorically rule out chemical fractionation of the sealed M solution aliquots 132 over the 4.5 year period between preparation of the ²²⁹Th solutions (e.g., preferential adsorption 133 or precipitation of an element in the ²²⁹Th decay chain), there is no *a priori* reason to suspect a 134 difference in the radioactive equilibrium of the two solutions.

135 Counting sources for high precision LS were prepared from the ²²⁹Th RM and SRM 4328C 136 solutions (Table 1). Two different scintillation fluids were used to create liquid scintillation 137 sources for both 229 Th solutions, Ultima Gold AB⁴ (PerkinElmer, Waltham, MA) and Ecoscint 138 (National Diagnostics, Tampa, FL). Three sources were gravimetrically prepared in 20 mL glass 139 LS vials for each scintillant-²²⁹Th solution combination, resulting in a total of 12 sources. To 140 prepare the sources, aliquots from ampoules of the SRM and 229 Th RM were dispensed using 141 polyethylene aspirating pycnometers with masses measured on an electronic microbalance 142 (AT20, Mettler Toledo, USA). Other cocktail components were weighed with a mechanical 143 analytical balance (Mettler B5, Mettler Toledo, USA). Close matching of the aqueous mass 144 fraction in the sets of cocktails was achieved by the addition of 1 mol L^{-1} HNO₃ to some of the 145 sources. In addition, each cocktail contained approximately 60 mg of di-(2-ethylhexyl) 146 phosphoric acid (DEHPA or HDEHP), $(C_8H_{17}O)_2PO_2H$, which is a complexing agent. Matched 147 blank sources of similar composition over the same quench range were also prepared for 148 background subtraction.

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⁴ Certain commercial equipment, instruments, software, or materials are identified in this paper to foster understanding. Such identification does not imply recommendation or endorsement by the National Institute of Standards and Technology, nor does it imply that the materials or equipment identified are necessarily the best available for the purpose.

 Each source was counted in triplicate on a LS 6500 counter (Beckman Coulter, Fullerton, CA, USA) and on an Accuflex LSC-8000 (Hitachi Aloka Medical Ltd., Tokyo, Japan). Two LS counters were used for this set of experiments to demonstrate that results are independent of the operating characteristics of a particular instrument (detection threshold, photomultiplier efficiency, deadtime, amplification, signal conversion, etc.). Individual LS measurements consisted of integrating the entire LS spectrum, and were of sufficient duration to accumulate 155 over 10^6 counts, such that the relative Poisson counting statistics on each was < 0.1 %. In addition to closely matching quenching conditions and cocktail compositions in counting sources, both instruments use quench indicating parameters (QIP) to monitor quenching. The Accuflex LSC-8000 (Counter H) uses the External Standard Channel Ratio (*ESCR)* quench parameter, which relates quenching to the ratio of two counting windows in the Compton 160 spectrum of ¹³³ Ba. The Beckman LS 6500 (Counter B) uses a Horrocks Number ($H#$) to monitor the quench level, which is based on the downward spectrum shift of the Compton edge of an 162 external 137 Cs source with increasing quenching in the cocktail.

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LS Source Series	Mass of Scintillant (g)	Mass of 229 Th Solution Aliquot (g)	Mass of Blank HNO ₃ (g)	Total Mass Cocktail (g)	$f_{\rm{aq}}$ $(\%)$	H#
229 Th RM UGAB	9.850(3)	0.2039(9)	0.798(23)	10.912(19)	9.18(19)	89.0(3)
4328C UGAB	9.852(1)	0.9909(10)		10.906(11)	9.09(8)	91.2(5)
229 Th RM ECOS	8.916(17)	0.1993(40)	0.790(26)	9.969(42)	9.92(2)	118.4(3)
4328C ECOS	8.910(7)	1.0103(26)	$---$	9.987(8)	10.12(2)	121.1(3)

164 Table 1²²⁹Th RM and SRM 4328C LS source preparation details and characteristics. Each LS 165 Source Series consisted of 3 separately prepared counting sources. UGAB and ECOS refer to the 166 Ultima Gold AB and Ecoscint scintillation solutions, respectively. The values for the masses (g) 167 and aqueous mass fraction f_{aq} (in %) are averages for the three sources in the series with the

 values in parentheses representing the standard deviation of the means. *H#* is the average quench indicating parameter (as measured in Counter B) which is an additional indicator of the close composition matching.

Relative Photon Emission

173 As an independent measurement of the ²²⁹Th RM and SRM 4328C activity ratio, relative photon emission rates were measured on the LS sources described above. Six replicate measurements of each source were made on a NaI (Tl) well-type counter (2480 WIZARD2, PerkinElmer, Waltham, MA). The detector crystal is 75 mm in diameter with a height of 80 mm and is shielded with a minimum of 50 mm of lead. A total of 6 trials for each source were measured 178 sequentially, alternating between SRM and 229 Th RM vials and the spectrum for each measurement was accumulated for an interval of 10800 s (live time). Stability between trials was 180 monitored with a 1.97 kBq ¹²⁹I check source (Half-life = 1.6 x 10⁷ years). The region from, nominally, 20 keV to 2000 keV was integrated to obtain an integral counting rate that was corrected for background. The Poisson counting statistics on any single measurement was approximately 0.13 %.

Results for LS and Photon Emission Counting Rates

 The measurement results are provided in terms of grand-means and great-grand-means of the massic count ratios (Table 2). The measured count rates from the LS and NaI(Tl) experiments are proportional to the massic activity of the respective solutions. Therefore, if 229 Th decay chain equilibrium is the same for both solutions, then the ratio of the massic count rates is equivalent to 189 the ²²⁹ Th relative activity ratio (A_X / A_S) which, in turn is equivalent of the inverse ratio of M

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Table 2 Count Ratios of ²²⁹Th RM - SRM 4328C as measured by LS and NaI(Tl) counters.

Counters H and B are as described in the text. UGAB and ECOS are as previously defined. $S_{\bar{x}}$ is 202 a relative precision estimator equivalent to the standard deviation of the mean.

203 **Activity Ratio by LS Standard Addition**

204 A LS standard addition experiment was performed using a series of counting sources prepared as 205 scintillation cocktails that contain mixtures of SRM 4328C and the 229 Th RM in varying 206 proportions (Table 3). A total of five standard addition LS sources were prepared using the 207 Ecosint scintillation solution. Aliquots of both SRM 4328C and the ²²⁹Th RM were added to 208 each source but source preparation was, otherwise, as previously described. The LS counting 209 rate (R_c) for the mixtures of the SRM 4328C, with massic activity (A_S) , and the ²²⁹Th RM, with 210 massic activity (A_X) , can be given by

$$
R_c = \varepsilon (m_S A_S + m_X A_X)
$$
 (1)

212 where ms and mx are the respective aliquot masses of the two solutions in the cocktail and where 213 ε is the overall ²²⁹ Th decay chain detection efficiency for the mixture in matched cocktails. 214 Simple rearrangement, by normalization of each term in Eq. (1) by m_S , yields a linear function

$$
\frac{R_c}{m_S} = \varepsilon A_S + \varepsilon A_X(m_X/m_S) \tag{2}
$$

216 with intercept εA_S and slope εA_X . From a series of matched counting sources with rates R_c for 217 varying m_S and m_X , a linear fit of R_c / m_S as a function of m_X / m_S will determine slope and 218 intercept parameters whose ratio gives A_x/A_y . Additionally, the linear equation, in two 219 independent variables of Eq. (1), can be fit by multiple regression to yield εA_S and εA_X directly. 220 The two calculational approaches give identical results, though extraction of the covariance 221 between ϵA_S and ϵA_X from the covariance matrix is easier with multiple regression.

 Ten (10) replicate measurements of the five sources were made in both Counter B and Counter H. The counting rates for the sources integrated over the entire LS spectrum ranged from about $\,$ 520 s⁻¹ to 950 s⁻¹. So, a duration of 3600 s was used for the measurements in order to accumulate 225 approximately 1.9 10^6 to 3.4 10^6 integral counts in each spectrum (corresponding to relative Poisson counting statistics of about 0.07 % to 0.05 % for each of the 10 measurements).

227

 Table 3 Standard Addition Source Preparation Details and Characteristics. Mass values have a standard uncertainty of < 0.01 %. *H#* is a quench indicating parameter (as measured in Counter B 230 for $n = 10$ replications) which is an additional indicator of the close composition matching. The values in parentheses represent the numerical values of the standard deviation of the mean for the quoted H#.

233

234 **Results for LS Standard Addition Measurement**

235 The standard deviation of the mean for the average of the 10 replicate measurements of a source 236 on Counter B ranged from 0.25 % to 0.37 %, and from 0.060 % to 0.083 % on Counter H. 237 Multiple regressions of the Counter B and Counter H data sets for the R_c (m_X , m_S) two-238 independent-variable function of Eq. (3) were performed with the LAB Fit code [19]. Table 4 239 summarizes the results of these regressions. The ratio A_X/A_S was obtained from the fitted 240 parameters ϵA_X and ϵA_S and its relative standard deviation was computed from

241
$$
S = \left[\left(\frac{S_X}{A_X} \right)^2 + \left(\frac{S_S}{A_S} \right)^2 - 2 \frac{S_{XS}}{A_X A_S} \right]^{1/2}
$$
 (3)

242 where S_X and S_S are the standard deviations on A_X and A_S , respectively, and where S_{XS} is the 243 covariance between A_X and A_S (refer to Table 4). The reduced χ^2 values for the two regressions 244 for $v = 3$ degrees of freedom were 3.1 and 0.61, with corresponding percentile points of $p =$

- 245 0.024 and $p = 0.61$ for Counters B and H data, respectively.
- 246

Table 4 Standard Addition Activity Ratio Measurement Results. The fitted parameters εA_X and ϵA_S are for the ²²⁹Th RM and SRM 4328C solutions, respectively, with corresponding variances S_X^2 and S_S^2 and covariance S_{XS} . The estimator *S* is the propagated standard deviation of the A_X / *A^S* massic activity ratio.

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253 **Isotope Dilution Alpha Spectrometry**

254 The massic activity of the 229 Th RM was measured on five subsamples that were prepared for 255 IDAS from a single unit of the reference material. This measurement method is independent of 256 the three activity ratio methods and is not reliant upon comparison with SRM 4328C. The IDAS 257 analysis is a nuclide-specific measurement of 229 Th activity so any perturbation of the radioactive 258 equilibrium of the decay chain for the isotope will not bias the measurement results.

 Accordingly, the analyses represent a robust verification of massic activity calculated from the previously described relative activity measurements.

- 282 with the known aliquot weights were then used to calculate the 229 Th massic activity using a
- 283 standard isotope dilution algorithm.

Source ID	229 Th Aliquot (g)	4342A aliquot (g)	229 Th corrected counts	230 Th corrected counts	Massic activity $(Bq g^{-1})$
1A	0.04681	0.23224	201134	208015	195.7
2A	0.03359	0.23442	146521	214837	194.2
3A	0.04935	0.22853	193851	187416	195.5
4A	0.03323	0.21738	121015	160688	201.0
5A	0.03362	0.23496	138180	203920	193.2

Table 5²²⁹Th IDAS sample data. Mass values have a relative standard uncertainty of 0.3 %.

286 **Results from IDAS**

287 The combined IDAS measurement data indicate a ²²⁹Th massic activity of (195.9 \pm 3.2) Bq g⁻¹ (*k* $=$ 2) for a measurement date of March 5, 2012. The uncertainty of the measured value is dominated by the variability of the replicate massic activity measurements (Table 6) with the uncertainty for solution masses, the SRM 4342A tracer, and peak tail corrections representing minor contributions.

292

Discussion

300 The physical and analytical linkages between the M solution, SRM 4328C, and the 229 Th RM 301 allow for the measurement of the ²²⁹Th RM massic activity through a direct comparison to SRM 4328C (Fig. 2). The three counting methods used to measure the activity ratios of the ²²⁹Th RM and SRM 4328C should yield the same values, within uncertainties. Results from the LS and NaI(Tl) relative counting methods are consistent among all three counters and both scintillation cocktails used to prepare the sources (Table 2). The LS standard additions results are also consistent for both LS counters (Table 4) but the relative difference between the standard addition and the relative counting methods (δ) is 0.7 %, while the relative standard deviation for the results of the two standard addition measurements are both approximately 0.16 %. No clear 309 explanation for the apparent discrepancy between the A_X / A_S ratio for the methods has been identified. Firstly, the composition matching of the five standard addition sources dictated nearly identical detection efficiencies across the series. Secondly, the quality of the fits, as reflected in 312 the adjusted R^2 , reduced χ^2 , and *F*-values (for ANOVA ratio of variances), suggests that there

313 should have been better agreement. The 95.4 % coverage factor on the standard deviation S_x and *S_S* for A_X and A_S is $k = 3.31$, such that the expanded relative uncertainties kS on A_X / A_S of 0.18 % 315 and 0.40 % are roughly a factor of 2 to 3 smaller than the δ from the A_X / A_S ratio. In the absence of identifiable analytical problems for the three activity ratio measurement methods, the overall 317 mean values for each method were averaged to yield a combined activity ratio $(A_X/A_S)_{c}$ of 5.510 with a standard uncertainty of 0.026. This standard uncertainty is calculated by adding, in quadrature, two variance components. These components are the within-method variability, represented by the typical standard deviation of the mean for the great-grand-means multiplied by the Student's *T* distribution for 2 degrees of freedom (*n* = 3 measurement methods), and the between-method variability represented by the standard deviation of the 3 activity ratio values.

 Fig. 2 Schematic of material and analytical linkages between the M solution, SRM 4328C, and the ²²⁹ Th RM. Bold arrows indicate the activity measurement linkages. The lighter arrow 327 indicates direct production of the 229 Th RM from the M solution. Activity values shown in the figure are for the December 31, 2007 reference date of the massic activity value of the M solution. All uncertainties are combined standard uncertainties.

330 As previously described, the inverse ratio of dilution factors (D_S/D_X) for the two reference 331 materials should be identical to the 229 Th activity ratio, representing an independent confirmation 332 of the measured ratio. Although the dilution factor for the ²²⁹Th RM is an approximation, a 333 calculated D_s/D_x of 5.511 is essentially indistinguishable from the measured activity ratio value 334 indicating the measured ratio is consistent with the values expected based on preparation of the 335 reference material from the M solution.

336 The massic activity of the ²²⁹Th RM solution (A_X) can be calculated (Eq. 4) from the combined 337 activity ratio $(A_X / A_S)_{c}$, as described above, and the measured activity of the M solution (A_m) and 338 the dilution factor for production of SRM 4328C stock solution (D_S) , as reported in [17].

339
$$
A_X = \frac{A_m (A_X / A_S)_c}{D_S}
$$
 (4)

340 The resulting massic activity value for the ²²⁹Th RM is (194.4 \pm 1.1) Bq g⁻¹. For comparison, the ²²⁹Th massic activity measured by IDAS can be corrected to the reference date for SRM 4328C 342 (December 31, 2007). The resulting massic activity value is 196.0 Bq g^{-1} with an expanded 343 uncertainty of 3.2 Bq g^{-1} ($k = 2$). The relative difference between the activity measurements is 344 0.8 % which is well within the 1.6 % relative uncertainty of the IDAS measurement.

345 The massic activity from Eq. 4 (A_X) and the ²²⁹Th molality (b_X) of (0.11498 \pm 0.00008) nmol g⁻¹, as measured for the ²²⁹Th RM in [18], can then be used to calculate a decay constant (λ) for ²²⁹Th 347 but a decay correction for the (4.7 ± 0.25) year interval between the reference dates for the 348 massic activity and the molality measurements (*t*) must also be incorporated into the calculation 349 (Eq. 5).

$$
\lambda e^{\lambda t} = \frac{A_X}{b_X L} \tag{5}
$$

351 Where *L* is the Avogadro's Constant. This function is solved for the decay constant by numerical 352 iteration, which converges rapidly because of the small magnitude of the exponential term. A 353 half-life (*T*) is then calculated from the decay constant (Eq. 6).

 $T = \frac{\ln(2)}{1}$ 354 $T = \frac{m(z)}{\lambda}$ (6)

355 The best-fit decay constant for the ²²⁹Th data is 8.858 $10^{-5} a^{-1}$ with a combined standard 356 uncertainty of 0.049 $10^{-5} a^{-1}$ which corresponds to a half-life and expanded uncertainty of (7825) ± 87) years ($k = 2$). The uncertainty for the half-life determination is dominated by components 358 from the M solution activity measurement and the measured activity ratio for the 229 Th RM (see Table 7). The uncertainties for the measured molality of the ²²⁹Th RM solution, the SRM 4328C 360 dilution factor, and the decay interval between the SRM reference date and the 229 Th molality 361 measurements represent minor contributions to the overall uncertainty for the half-life.

362

382

Fig. 3 Reported half-lives for ²²⁹Th. Error bars are expanded uncertainties ($k = 2$).

384 Kikunaga *et al.* measured the half-life by an indirect method based on known ingrowth periods 385 for Th in a high purity ²³³U material that also contains a small proportion of ²³²U. In their study, 386 the measured α count ratios of $A(^{233}U)/A(^{232}U)$ and $A(^{229}Th)/A(^{228}Th)$ and the half-lives of ²³²U 387 and ²²⁸Th were used to calculate a ²²⁹Th half-life without the necessity of measuring the absolute 388 amount or activity for the nuclide. Accordingly, their calculations are highly sensitive to the 228Th half-life and the measured ²²⁹Th to ²²⁸Th count ratio, for which they indicate an average 390 uncertainty of 0.2 %. The ²²⁸Th half-life appears to be well constrained [21], and Kikunaga, *et* 391 *al*. addressed potential sources for significant bias in their data such as corrections for background and interferences in the α spectra. However, the combined alpha spectra for ²²⁹Th 393 and ²²⁸Th are complex and there is potential that their choice of region-of-interest for the

 nuclides, the magnitude of applied corrections, and/or uncertainties in decay data for the nuclides could result in a bias that is not covered by the 0.15 % relative "systematic" uncertainty cited for their measurements.

 Varga *et al.* [13] provided detailed uncertainty budgets for each of the measurement methods used for their half-life determination (Methods A and B). The uncertainties for each method is dominated by the Type B uncertainties for SRM 4328C massic activity. Three replicates of the $\frac{229}{10}$ Th molality measurement were performed for each method but no variability component associated with the replicate IDMS measurements was specified. Varga et al. then combined the two measured half-lives to obtain a weighted mean value of 7917 years with an expanded 403 uncertainty of only 48 years $(k = 2)$, despite a 63 year difference in the half-lives for the two methods. It appears that the combined measurement uncertainty did not incorporate a component for the variability evident between measurement methods and it is unclear whether the type B uncertainty components shared by the methods were appropriately propagated.

Conclusion

408 A new measured half-life of (7825 \pm 87) years ($k = 2$) is reported for ²²⁹Th. This value is based 409 on the previously determined 229 Th molality of a new thorium IDMS reference material and the detailed characterization of the massic activity value for the reference material, presented here. The measured massic activity is tied directly to the SRM 4328C activity standard through multiple replicate activity ratio measurements using three independent methods and was also 413 independently confirmed by IDAS measurements. This new ²²⁹Th half-life measurement is between 0.7 % and 1.4 % shorter than other modern half-life determinations but is otherwise consistent with these values (overlapping uncertainties at the 95 % level of confidence).

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References

- 1. Hagemann F, Katzin L I, Studier M H, Seaborg G T, Ghiorso A (1950) The 4n + 1
- 421 radioactive series: The decay products of U^{233*} . Phys Rev. 79(3):435-443
- 2. de Lavison P, Husband L J, Jerome S M, Keightley J D, Woodman A P, Woods D H,
- 423 Woods S A (2000) The standardization of the ²²⁹Th for an environmental yield tracer. Appl
- Rad Iso. 53:243-249
- 3. Aggarwal S K (2016) A review on the mass spectrometric analysis of thorium. Radiochim Acta, 104 (7):445-456.
- 4. Apostolidis C, Molinet R, Rasmussen G, Morgenstern A (2005) Production of Ac-225 from
- Th-229 for targeted alpha therapy. Anal Chem. 77:6288-6291
- 5. Campbell C J, Radnaev A G, Kuzmich A, Dzuba V A, Flambaum V V, Derevianko A
- (2012) A Single-Ion Nuclear Clock for Metrology at the 19th Decimal Place. Phys Rev Lett. 108 (120802):1-5
- 432 6. Lamont S P, Hall G J (2005) Uranium age determination by measuring the ²³⁰Th/²³⁴U ratio. Radioanal Nuc Chem. 264:423-427
- 434 7. Williams R W, Gaffney A M (2011)²³⁰Th-²³⁴U model ages of some uranium standard
- reference materials. Proc Radiochim Acta, 1:31-35

- 22. Metrodata GmbH, (2009) GUM Workbench. Weil am Rhein, Germany.
- 23. Pomme S (2015) The uncertainty of the half-life. Metrologia, 52:S51-S65.