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Structural and Optical Properties of Metal-Nitrosyl Complexes

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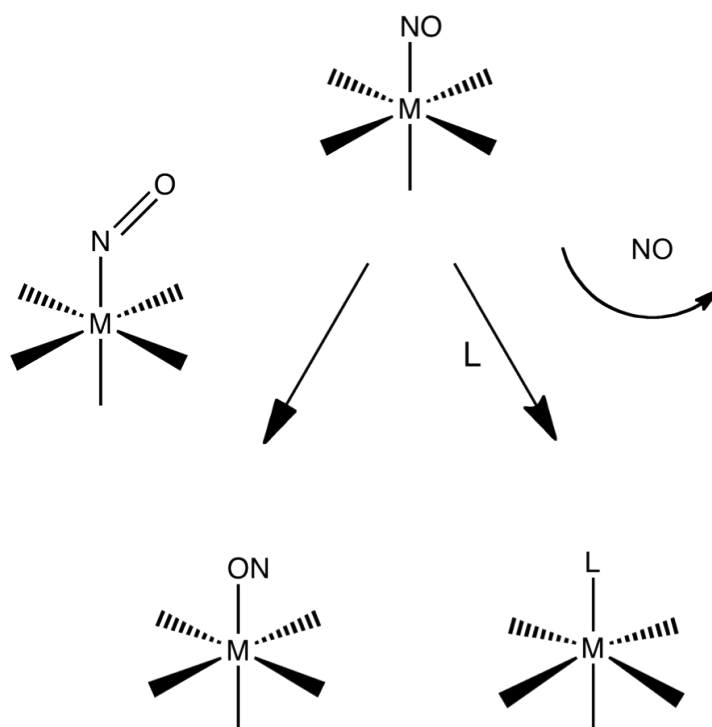
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Abstract

The electronic, structural and optical properties (including Spin-Orbit Coupling) of metal nitrosyl complexes $[M(CN)_5(NO)]^{2-}$ (M=Fe, Ru or Os) are investigated by means of Density Functional Theory, TD-DFT and MS-CASPT2 based on RASSCF wavefunction. The energy profiles connecting the N-bound (η^1 -N), O-bound (η^1 -O) and side-on (η^2 -NO) conformations have been computed at DFT level for the closed shell singlet electronic state. For each structure the lowest singlet and triplet states have been optimized in order to get insight of the energy profiles describing the conformational isomerism in excited states. The energetics of the three complexes is similar with the N-bound structure being the most stable, with one exception, namely the triplet ground state of the O-bound isomer for the iron complex. The conformational isomerism is highly unfavorable in the S_0 electronic state with the occurrence of two energy barriers higher than 2 eV. The lowest bands of the spectra are assigned to $MLCT_{NO}/LLCT_{NO}$ transitions with an increasing MLCT character going from iron to osmium. Two low-lying triplet states, T1 ($MLCT_{NO}/LLCT_{NO}$) and T2 ($MLCT_{NO}/IL_{NO}$), seem to control the lowest energy profile of the excited state conformational isomerism.

1. Introduction

Metal-Nitrosyl coordination compounds are of great interest because they exhibit two essential photo-induced primary reactions namely NO/ON linkage isomerism and NO release (Scheme I) of crucial importance in cardiovascular treatments and cancer therapies [1].



Scheme I. Concurrent deactivation pathways of Metal-Nitrosyl complexes (Adapted from Ref. 2 with permission of Elsevier).

Metal-nitrosyl coordination compounds characterized by high quantum yields of NO release are particularly adapted to site-specific delivery in tumor cells within photodynamic therapy (PDT). Photo-activated nitrogen monoxide releasing moieties

(PhotoNORMs) have been developed on the basis of Fe and Ru complexes already in the late 90's [3] and recent reviews are dedicated to these complexes [4-7].

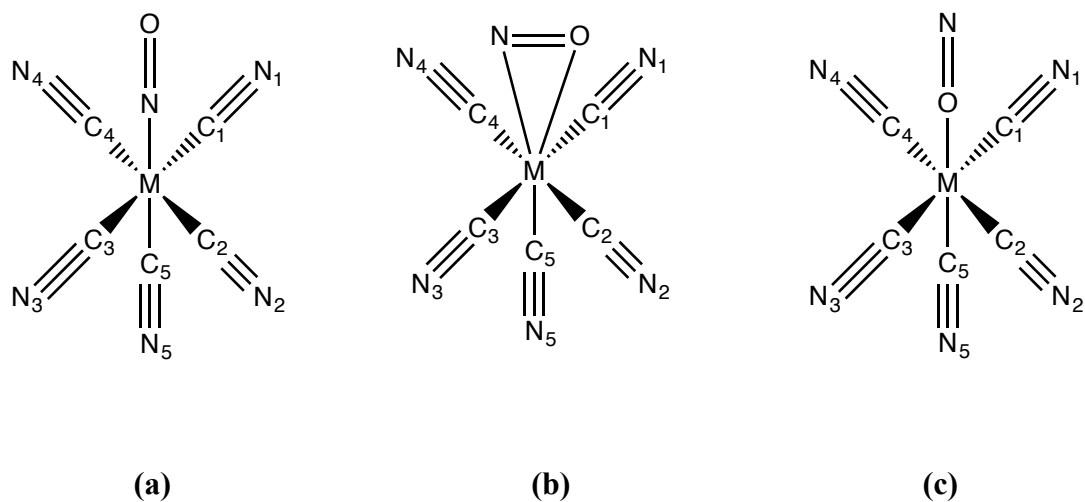


Figure 1. Schematic structures of the investigated $[M(CN)_5(NO)]^{2-}$ ($M = Fe, Ru, Os$) complexes in their *standard* N-bound ($\eta^1\text{-N}$) **(a)** and metastable (MS) structures: *flat* side-on ($\eta^2\text{-NO}$) **(b)** and *reverse* O-bound ($\eta^1\text{-O}$) **(c)**.

The discovery in the late 1970s of metastable isomers of sodium nitroprusside $Na_2[Fe(CN)_5(NO)]$ [8] and the synthesis of related complexes, opened the way to a wealth of experimental and theoretical studies based on the development of both spectroscopic techniques and quantum chemical methods [9-13]. The co-existence of several structural isomers related to different M-NO binding modes (Figure 1) were put in evidence upon irradiation of M-NO complexes [14].

Alongside their interest in the understanding of important biological processes implicating nitric oxide [3] these molecules possess the potential for efficient optical data storage based on long-lived metastable states (MS) [15, 16].

Under visible irradiation a number of $\{MNO\}^6$ ($M=Fe, Ru, Os$), $\{MNO\}^8$ ($M=Pt$) and

$\{\text{MNO}\}^{10}$ ($\text{M}=\text{Ni}$) nitrosyl complexes reversibly switch between the so-called N-bound ($\eta^1\text{-N}$) ground states (*standard*), O-bound ($\eta^1\text{-O}$) (*reverse*) and side-on ($\eta^2\text{-NO}$) (*flat*) conformations (Figure 1) [17]. For optimal efficiency, MS states should be long-lived enough and the process has to be thermally or photo-chemically reversible, not only at the molecular level but also in solid-state.

Most of the theoretical studies dedicated to Fe and Ru nitrosyl photo-isomerizable complexes recently investigated experimentally have focus on the structural and electronic investigation of the ground state and MS isomers coupled to an analysis of the potentially photo-active states in term of frontiers orbitals [12, 18-25]. However, because of the non-innocent nature of the nitrosyl ligand the character of the photo-active excited states accessible in the visible energy domain between 330 nm and 550 nm is strongly dependent of the surrounding ligands. This character may vary from pure MLCT to pure LLCT with intermediate situations such as mixed LLCT/MC/MLCT or MC/LLCT/LC.

Light-induced linkage NO isomerism has to be initiated in the absorbing state and fast enough for competing with direct NO release or other ligands dissociation (Scheme I). The electronic density alteration is governed by the bending of the Metal-N-O bond angle to form the intermediate MS_{inter} . This can be only accomplished if the absorbing singlet state possesses a significant metal contribution (MC) that will weaken the strong M-N bond. This fragile bond will allow the isomerization and facilitate the stabilizing metal-oxygen bonding interaction in the excited state [2]. Indeed three competitive deactivation channels of $[\text{Fe}(\text{CN})_5(\text{NO})]^{2-}$ dissolved in methanol, namely Fe-NO linkage isomerism, NO release and CN dissociation, have been put in evidence within the first 500 fs after 400 nm pump irradiation using ps transient IR spectroscopy [26]. These experiments indicate that population of the absorbing states detected at 393 nm and 520 nm at room temperature in methanol, most probably corresponding to $\text{LLCT}_{\text{NO}}/\text{MLCT}_{\text{NO}}/\text{MC}$ excited states, opens the

route to both ligand dissociation and Fe-NO linkage isomerism. Ultra-fast ISC to the associated low-lying triplet states cannot be excluded at this stage. However the usual wavelength dependence of the measured quantum yields is in favor of singlet photo-reactivity. Consequently by adjusting the initial wavelength of irradiation it should be possible to target specific photo-active states and to control the branching ratio between the different primary reactions.

The present work aims at comparing the electronic, structural and optical properties of the $[\text{M}(\text{CN})_5(\text{NO})]^{2-}$ ($\text{M} = \text{Fe}, \text{Ru}, \text{Os}$) series of complexes (Figure 1). Energy profiles and critical geometries characterizing the conformational isomerism in the three molecules are investigated for the S_0 electronic ground states as well as for the low-lying excited states in order to get some insight on the mechanisms of inter-conversion.

2. Computational Details

Three series of calculations were undertaken. A first set was computed using ADF2013 software [27]. The structures of various isomers of the three complexes $[\text{M}(\text{CN})_5(\text{NO})]^{2-}$ ($\text{M} = \text{Fe}, \text{Ru}, \text{Os}$) (Figure 1) have been fully optimized for the electronic ground state $^1\text{A}'$ under the Cs symmetry constraint at the density functional theory (DFT) [28] level with the B3LYP functional [29] and triple- ζ basis sets [30] on all atoms. Scalar relativistic corrections were included through the zero-order relativistic approximation (ZORA) [31] and solvent effects were taken into account by means of the CONductor-like Screening Model COSMO [32] of methanol ($\epsilon = 32.7$). Electronic ground state minima were characterized by a complete set of real frequencies. The low-lying electronic excited states were optimized at the same level of theory. The theoretical absorption spectra were computed by means of time-dependent DFT (TD-DFT) [33] performed on the ground state

optimized structures. The spin-orbit coupling (SOC) has been included at a perturbative level [34]. The Tamm-Dancoff approximation (TDA) [35] was used to avoid over stabilization of the lowest triplet states.

A second set of calculations was undertaken in order to compute the Gibbs Free Energy surface leading from the standard structure to the reverse one and to locate the transition states using GAUSSIAN 09 quantum chemistry package [36]. These calculations were carried out at the DFT level of theory with the B3LYP functional using the 6-31+G** basis set for C, N and O [37] and the Stuttgart-Dresden SDD basis sets and associated small core pseudopotentials for Fe, Ru and Os atoms [38]. Solvation (methanol) was included through a PCM model [39]. Calculations were performed without symmetry on the S_0 electronic ground state singlet state.

For the iron complex Restricted-Active Space Self-Consistent field (RASSCF) [40], including ten electrons correlated in ten active orbitals, supplemented by multi-state complete active space perturbation theory 2nd order (MS-CASPT2) [41] calculations have been performed (on the ADF optimized structures) including solvent corrections by means of the polarized continuum model (PCM) for MeOH and using quadruple- ζ quality atomic natural relativistic correlated corrected basis sets (ANO-RCC) [42] and the Molcas 8.2 quantum chemistry package [43].

3. Structural and Electronic Properties

3.1. Structures

Some important optimized bond distances and angles of $[M(CN)_5(NO)]^{2-}$ (M = Fe, Ru, Os) computed with ADF in their so-called *standard structure* (Figure 1a) are reported

in Table 1 and compared to the X ray data for the iron [44] and ruthenium [45] complexes. Whereas the agreement between the calculated values and the experimental data is rather good for the 1st-row transition metal complex, some discrepancies are observed for 2nd-row transition metal compound, especially for the Ru-N distance which is overestimated by 0.051 Å at this level of calculation. The structures of the 2nd and 3rd-row complexes are very similar.

Table 1. Important bond distances (in Å) and bond angles (in °) of $[\text{M}(\text{CN})_5(\text{NO})]^{2-}$ (M = Fe, Ru, Os) in their so-called *standard structure* (Figure 1a) computed with ADF. (Atoms numbers according to Figure 1). a) Ref 44 b) Ref 45

	Fe	Fe (RX) ^a	Ru	Ru (RX) ^b	Os
M-N	1.66	1.65	1.81	1.75	1.79
M-C ₁	1.96	1.93	2.09	2.07	2.08
M-C ₂	1.96	1.93	2.09	2.07	2.08
M-C ₃	1.96	1.93	2.09	2.07	2.08
M-C ₄	1.96	1.94	2.09	2.07	2.08
M-C ₅	1.95	1.93	2.08	2.07	2.10
NO	1.13	1.13	1.14	1.15	1.14
M-N-O	179.8	178.2	179.9	177.0	179.9
N-M-C ₁	94.1	95.9	94.1	94.3	94.2
N-M-C ₂	94.0	95.5	94.1	93.3	94.1
N-M-C ₃	93.9	94.1	94.1	95.9	94.0
N-M-C ₄	94.0	96.1	94.1	95.3	94.1

Selected optimized bond distances and angles of $[M(CN)_5(NO)]^{2-}$ (M = Fe, Ru, Os) in their so-called *flat* (Figure 1b) and *reverse* (Figure 1c) structures computed with ADF are reported in Table 2 together with their energy gap ΔE (in eV) with the *standard* structures reported in Table 1. For the iron complexes both the lowest singlet and triplet states are reported. All the Cs optimized structures are characterized by real frequencies in the singlet ground state. In the triplet state, symmetry is broken and the reverse structure was optimized without symmetry.

Table 2. Important bond distances (in Å) and bond angles (in °) of $[M(CN)_5(NO)]^{2-}$ (M = Fe, Ru, Os) in their so-called *flat structure* (Figure 1b) and *reverse structure* (Figure 1c) and associated energy gap ΔE (in eV) with the *standard* structures. (Atoms numbers according to Figure 1). ^a The structure was computed without symmetry.

	Fe			Ru		Os	
	flat	reverse	reverse ^a	flat	reverse	Flat	Reverse
	S ₀	S ₀	T ₁	S ₀	S ₀	S ₀	S ₀
M-N	1.92			2.04		2.01	
M-O	2.04	1.77	2.28	2.21	1.95	2.20	1.92
M-C ₁	1.96	1.96	1.96	2.09	2.08	2.09	2.08
M-C ₂	1.97	1.96	1.96	2.09	2.08	2.08	2.08
M-C ₃	1.95	1.96	1.95	119.3	2.08	2.08	2.08
M-C ₄	1.97	1.96	1.96	2.09	2.08	2.08	2.08
M-C ₅	1.91	1.91	1.89	2.01	2.02	2.02	2.02
NO	1.15	1.11	1.15	1.16	1.12	1.18	1.13
M-N-O	79.3			82.8		82.9	

M-O-N	67.0	179.8	128.2	65.9	179.8	64.9	179.9
N-M-C ₁	79.7	92.6	87.3	78.0	92.6	77.8	92.0
N-M-C ₂	91.0	92.6	87.2	91.5	92.4	90.9	92.0
N-M-C ₃	120.4	92.7	90.5	119.3	92.5	119.4	92.0
N-M-C ₄	91.0	92.6	88.4	91.5	92.6	90.9	92.1
ΔE_{DFT}	1.46	1.65	1.49	1.33	1.67	1.59	2.05

While the *standard* structures remain the most stable structures at this level of calculation, the *reverse* ones in which the nitrosyl ligand is bounded to the metal centre by the O atom are less stable ($\Delta E = 1.5\text{-}2.0$ eV or $145\text{-}193$ kJ Mol⁻¹) for all compounds. The *flat* structures, where both N and O are bounded to the metal atom, are at about 1.5 eV above the *standard* ones as well as the *reverse* structure in the lowest triplet state of $[\text{Fe}(\text{CN})_5(\text{NO})]^{2-}$. This latter structure, at 1.49 eV above the *standard* isomer, is characterized by a bent geometry with a Fe-N-O bond angle of 128° and is nearly degenerate with the *flat* structure calculated at 1.46 eV.

According to the energetics reported in Table 2, the adiabatic conformational isomerism on the electronic ground state S₀ potential energy surface (PES) seems to be easier for the ruthenium complex than for the osmium complex for which the *flat* and *reverse* structures are highly destabilized. A S₀/T₁ crossing characterizes the conformational isomerism pathway of the iron complex.

3.2. Electronic structures and potential energy profiles

One important point concerns the validity of DFT at describing correctly the electronic structure of 1st-row transition metal complexes. In particular, metal-nitrosyls may be the

seat of near-degeneracies and unusual spin densities. A DFT based theoretical study coupled to Mossbauer spectroscopy of a series of iron compounds, including nitrosyl substituted complexes, by Ghosh A. et al [46] has concluded that experimental isomer shifts are well reproduced. To further support our approach, MS-CASPT2 calculations were undertaken. At this level of theory, the S_0 electronic state of the *standard* isomer is again the global energy minima of the three isomers of $[\text{Fe}(\text{CN})_5(\text{NO})]^{2-}$. The stability of the *flat* and *reverse* isomers is similar to that found at DFT level, the *flat* being 1.62 eV and the *reverse* being 1.50 eV above the *standard* structure, the *reverse* structure is now slightly more stable than the *flat* one at MS-CASPT2 level. This difference between the ADF and MS-CASPT2 result may come from a greater multi reference character of *reverse* structure. Indeed, the *standard* and *flat* isomers are mainly described by one electronic configuration weighting respectively 81% and 85% of the total wavefunction. This weight significantly drops to 70% in the *reverse* case showing a greater multireference character of the electronic ground state. This may be the source of the different stability of the flat and reverse isomers, the mono reference DFT approach being unable to catch this contribution. Though the reasonable agreement between the MS-CASPT2 and the DFT approach validates our approach.

We then studied the mechanism of the NO isomerization in the ground state using GAUSSIAN. The potential energy profile describing the conformational isomerism for the iron complex is depicted in Figure 2. The data for all complexes are gathered in Table 3.

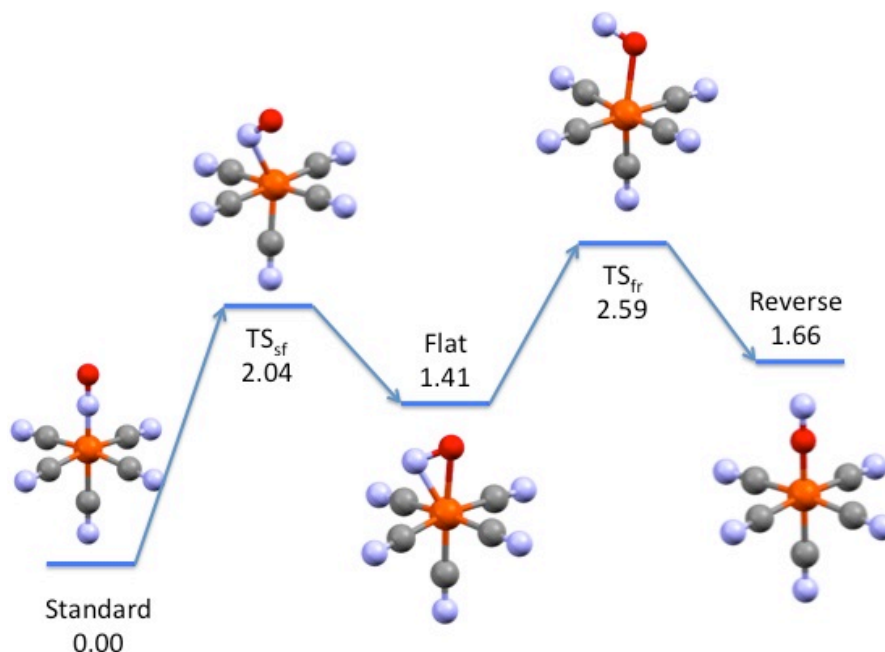


Figure 2. Calculated low-lying S_0 potential energy profile of the conformational isomerism in $[\text{Fe}(\text{CN})_5(\text{NO})]^{2-}$ in methanol computed with GAUSSIAN. The potential and Gibbs free energies are reported in Table S1 of the SI section. Energies are in eV.

Table 3. Calculated low-lying S_0 potential energy profiles of the conformational isomerism for the three complexes in methanol and in the triplet state for iron. For the latter, the reference energy is the S_0 standard structure. Gibbs Free energies are in eV.

	<i>Standard</i>	TS_{SF}	<i>Flat</i>	TS_{FR}	<i>Reverse</i>
Fe	0.0	2.03	1.41	2.59	1.66
Fe (triplet)	0.42				0.87
Ru	0.0	2.09	1.35		
Os	0.0	2.40	1.56	3.12	2.05

The results are similar for the three molecules, although no transition state has been located for the flat to reverse conversion for the ruthenium complex. The transition states are very high in energy for the *standard* to *flat* (TS_{SF}) conversion, more than 2 eV. Such a high

energy can easily be explained by the structure of the TS itself. For the three complexes, in the TS_{SF} structure, the Cs symmetry is broken and to ensure the conversion, the M-N bond is significantly lengthened (from 1.66 to 2.03 Å in TS_{SF} for Fe). The overall shape of the TS_{SF} is a lacunar octahedron in which the NO ligand is close to de-coordinate. The same features can characterize the structures of the two calculated transition states from flat to reverse structure (TS_{FR}) we were able to locate, with a lengthening of the M-O bond of roughly 0.5 Å moving from the reverse structure to that of TS_{FR}. In the iron case, the M-O distance is of 2.271 Å and the M-N distance of 2.943 Å. The need for a significant metal cation - NO bond weakening for ensuring the conversions is at the origin of the high barrier. This can also explain the difficulty to locate the TS_{FR} in the ruthenium case. The barrier approaches the dissociation channel and all attempt to locate this TS_{FR} led either to dissociation of NO or to the rotation of the ligand around the M-O axis.

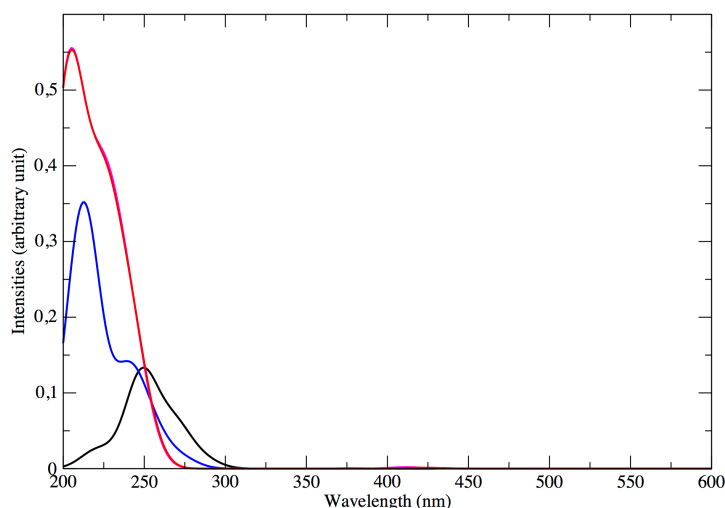
We tentatively compute the NO isomerization pathway in the lowest triplet state of the Fe complex. As mentioned above, in this state, the Cs symmetry is broken and the Fe-N-O angle is no longer linear with a value of 138.8° for the Fe-N-O angle. The minimum *standard* structure in the triplet state is very similar to that of the TS_{SF} in the singlet state (0.42 eV above the singlet state). A minimum was also found for the binding through the O atom (at 0.87 eV, Table 3) like in the *reverse* isomer but again in a bent structure with a Fe-O-N angle of 130.0°. Our attempts at finding an equivalent of the singlet state *flat* isomer in the triplet state converge either to the equivalent of the *standard* or of the *reverse* structure with no evidence of a transition state connecting these two structures. A scan performed on the triplet potential energy surface (PES) following the bending of the Fe-N-O angle starting from triplet *standard* structure did not lead to triplet *reverse* isomer. At some point appears a discontinuity on the PES and the structure evolved to the de-coordination of the NO. Starting from this structure, geometry optimization leads a

structure degenerate with the triplet O-bound structure at 0.87 eV (this structure still present imaginary frequencies despite fulfilling the convergence criteria which we were unable to get rid of). This explains why we failed in locating the isomerization TS as the triplet PES probably present several state crossing. This shows that searching for the TS would require multiconfigurational approaches as several PES seems to be close for those structures. This is beyond the scope of the present study.

4. Optical Properties

4.1. Absorption spectra of the standard structures

The TD-DFT absorption spectra of $[\text{Fe}(\text{CN})_5(\text{NO})]^{2-}$, $[\text{Ru}(\text{CN})_5(\text{NO})]^{2-}$ and $[\text{Os}(\text{CN})_5(\text{NO})]^{2-}$ are represented in Figure 3. The three *standard* complexes of C_{4v} symmetry absorb between 200 and 500 nm with a very weak absorption starting at 500 nm. Whereas the Spin-Orbit Coupling (SOC) effects are negligible for the Fe and Ru complexes a SO splitting of the lowest band in three peaks (490, 420 and 375 nm) is observed for the Os compound. The lowest bands of the Fe and Ru complexes are calculated at 450 nm and 425 nm, respectively.



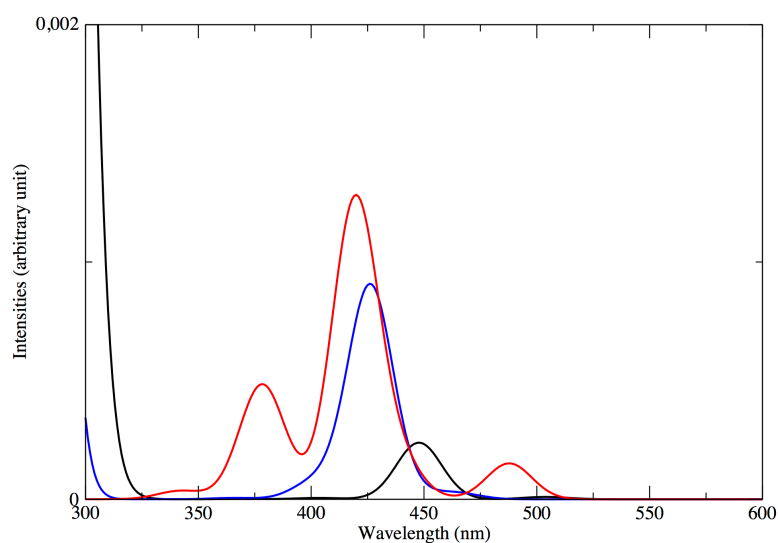
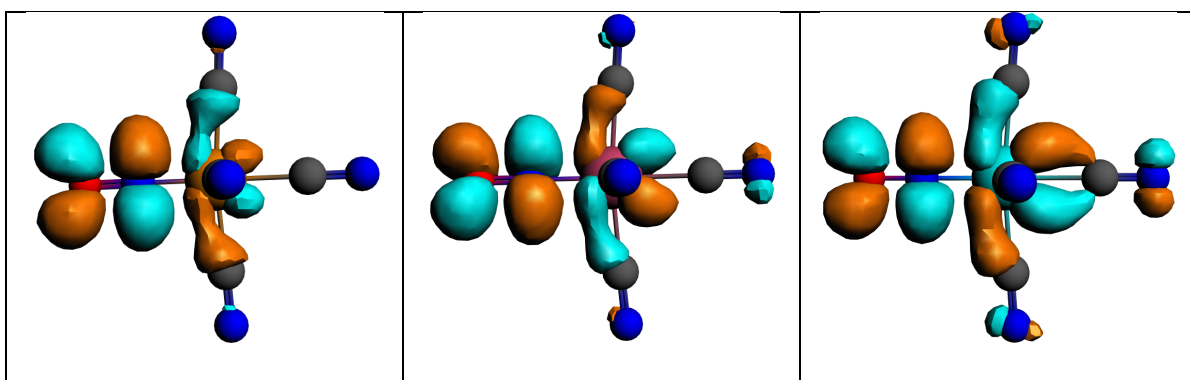


Figure 3. TD-DFT absorption spectra including SOC of $[\text{Fe}(\text{CN})_5(\text{NO})]^{2-}$ (black), $[\text{Ru}(\text{CN})_5(\text{NO})]^{2-}$ (blue) and $[\text{Os}(\text{CN})_5(\text{NO})]^{2-}$ (red). Bottom: a zoom in the visible domain.

The lowest absorbing bands in the visible domain are assigned to $\text{MLCT}_{\text{NO}}/\text{LLCT}_{\text{NO}}$ transitions described by HOMO-LUMO excitations (Figure 4), namely $d\pi_{\text{CN}} \rightarrow \pi^*_{\text{NO}}$. The MLCT character of these mixed MLCT/LLCT transition increases from Fe to Os correlating with expanding SOC effects in the 3rd-row transition metal complex in addition to the usual heavy atom influence.



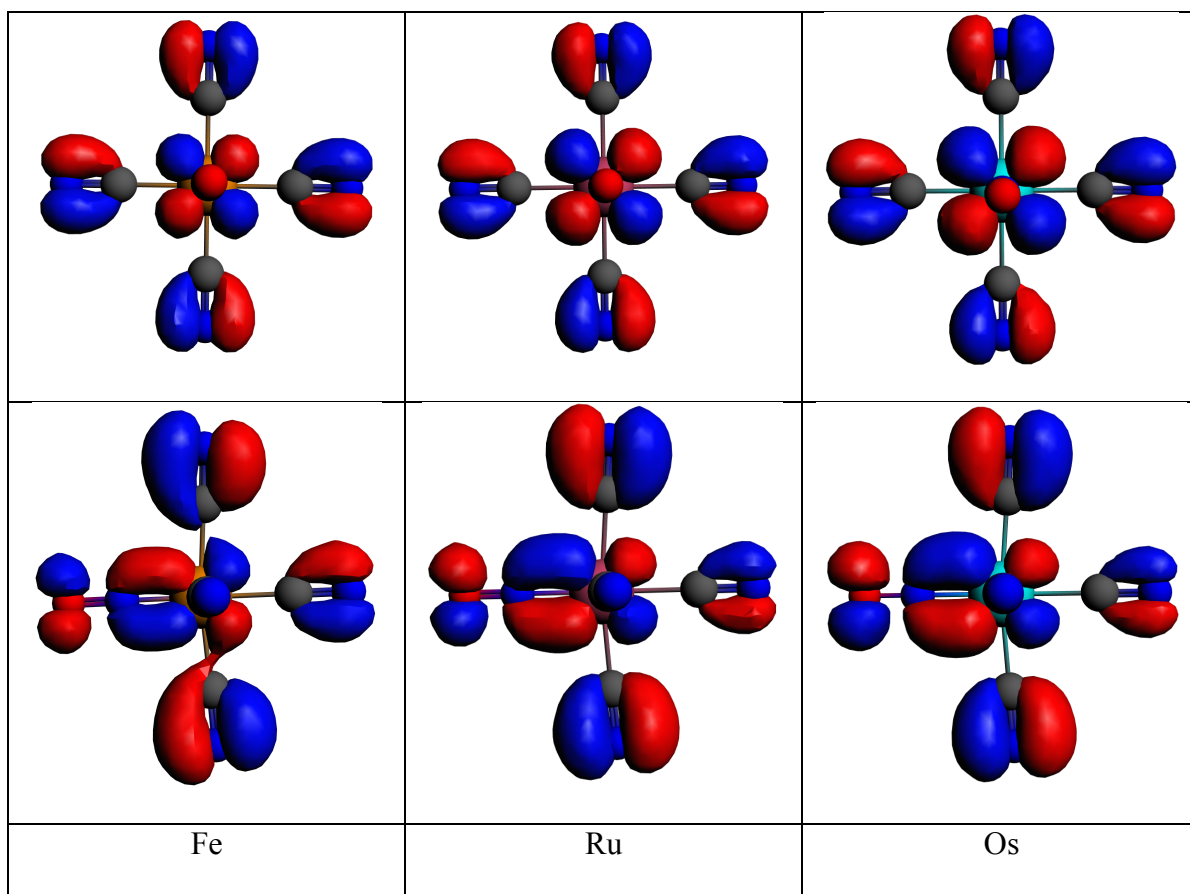


Figure 4. Frontier Kohn-Sham orbitals of the $[M(\text{CN})_5(\text{NO})]^{2-}$ complexes; HOMO-1 (bottom, side view), HOMO (middle, front view) and LUMO (top, side view).

The visible band (350-500 nm) of $[\text{Os}(\text{CN})_5(\text{NO})]^{2-}$ is shifted to the red with respect to the one of the 1st-row and 2nd-row transition metal complexes and broadened due SOC effects. The three complexes present upper absorption peaks between 300 to 200 nm of increasing intensity and slightly blue-shifted from Fe to Os. These intense peaks are attributed to charge transfer from the cyanide π orbitals towards the LUMO. Below 200 nm, the most intense computed peak is a mixture of $\pi_{\text{CN}} \rightarrow \pi^*_{\text{MNO}}$ and of $\pi_{\text{MNO}} \rightarrow \pi^*_{\text{MNO}}$.

4.2. Low-lying singlet and triplet states

Under C_s symmetry constraint, the lowest singlet S_1 is doubly degenerate with one component in A' and the other in A'' symmetry point group, corresponding to pure transitions from the HOMO to the doubly degenerate LUMO (Figure 4). The situation is trickier for the triplet states. In competition with the doubly degenerate triplet counterparts T_1 (MLCT_{NO}/LLCT_{NO}) of the previous singlet states, another triplet T_2 (MLCT_{NO}/IL_{NO}) of A' symmetry is present. This T_2 is even the most stable in the iron complexes and lays close upper the T_1 in the ruthenium and osmium complexes. This T_2 correspond to a transition from the HOMO-1 to the LUMO (Figure 4). The calculated vertical transition energies to these low-lying excited states at Franck-Condon of the *standard* structure (Figure 1a) are reported in Table 4.

Table 4. Calculated TD-DFT vertical transition energies to the low-lying excited states of $[M(CN)_5(NO)]^{2-}$ ($M = Fe, Ru, Os$) at the standard structure and MS-CASPT2 results for iron (a).

	Fe	Fe ^a	Ru	Os
T_1A'	2.45	2.41	2.70	2.82
T_1A''	2.45	2.33	2.70	2.82
T_2A'	2.31		2.76	2.99
S_1A'	2.79		2.92	3.03
S_1A''	2.79		2.92	3.03

Whereas the T_1 state calculated at 2.45 eV (Fe), 2.70 eV (Ru) and 2.82 eV (Os) is the lowest state of the Ru and Os complexes, it is above the T_2 excited state calculated at 2.31

eV in the Fe complex. The MS-CASPT2 results are again in good agreement with the TD-DFT ones validating our approach.

In order to investigate more closely the potential energy profiles of the conformational isomerism in the excited states, the structures of the five excited states reported in Table 3 have been fully optimized under Cs symmetry constraint and starting from the *standard*, flat and *reverse* geometries. In addition to the structures depicted in Figure 1, three structures represented in Figure 5 are put in evidence, namely the NO-bent, MNO-bent, and ON-bent conformations.

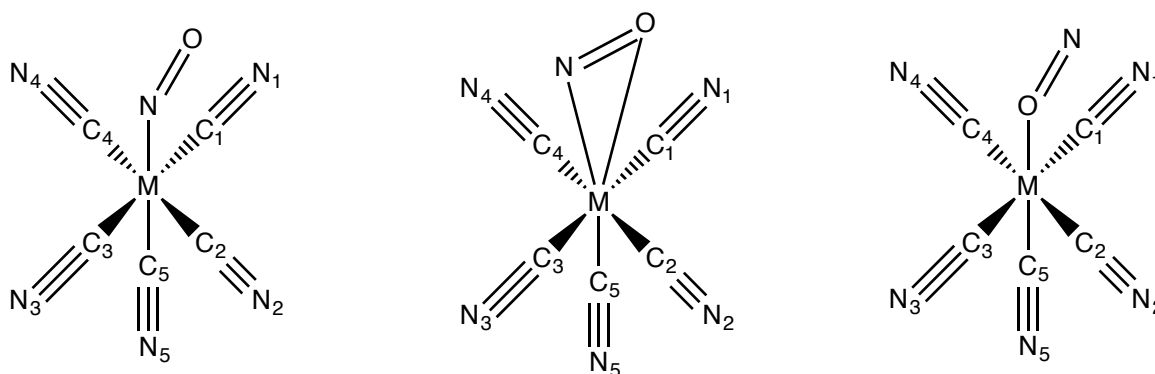


Figure 5. Structures of the $[M(CN)_5(NO)]^{2-}$ ($M = Fe, Ru, Os$) complexes in their low-lying electronic excited states. The NO-bent structure is on the left, The MNO-bent one is in the middle, and the ON-bent one on the right.

The energetics (in eV) obtained for the different optimized structures is reported in Tables 5, 6 and 7 for the Fe, Ru and Os complex, respectively. The reference energy (0.0) is given by the *standard* structure in its S_0 electronic ground state.

Table 5. Potential energy (in eV) of $[Fe(CN)_5(NO)]^{2-}$ calculated for the different optimized structures (Figures 1 and 5) with respect to the S_0 electronic ground state energy of the standard geometry (Figure 1a).

Structure	<i>Standard</i>	<i>Flat</i>	<i>Reverse</i>	NO-bent	MNO- bent	ON-bent
S ₀	0.0	1.46	1.65		2.09^a	2.20^b
S ₁ (A')	2.42	3.41	2.99			
T ₁ (A')	2.12	-NO	2.62			
T ₂ (A')	1.77				1.86^a	2.10^b
S ₁ (A'')	-		2.85	2.22	2.65	
T ₁ (A'')	-			1.75	2.25	2.45

^a On the S₁(A'') optimized structure

^b On the T₁(A'') optimized structure

Whereas the S₁(A') and T₁(A') excited states converge to the *standard* structure, the S₁(A'') and T₁(A'') states are more stable in the NO-bent structures and should play an important role in the conformational isomerism leading to the *reverse* structure. At some critical geometries the T₂(A') excited state is more stable than the S₀ electronic ground state leading to several potential energy surface crossings potentially active in the isomerism mechanism. The T₁(A') excited state in the flat structure leads to de-coordination of the NO ligand.

Table 6. Potential Energy (in eV) of [Ru(CN)₅(NO)]²⁻ calculated for the different optimized structures (Figures 1 and 5) with respect to the S₀ electronic ground state energy of the standard geometry (Figure 1a).

Structure	<i>Standard</i>	<i>Flat</i>	<i>Reverse</i>	NO-bent	MNO- bent
S ₀	0.0	1.33	1.67		
S ₁ (A')	2.54	3.22	3.10		

T ₁ (A')	-	-		
T ₂ (A')	2.19	2.58		
S ₁ (A'')	-	-NO	2.37	
T ₁ (A'')	-	2.83	2.13	2.66

The excited states energetics of $[\text{Ru}(\text{CN})_5(\text{NO})]^{2-}$ differs significantly of the one of the Fe complex. The de-coordination of the NO should occur via the S₁ (A'') excited state in its *flat* structure. This pathway is not energetically favorable. All the excited states remain above the S₀ electronic ground state by more than 1.5 eV whatever the structure is. There is no evidence at this level of calculation of energetically favorable ON-bent structures or S₀/T₂(A') potential crossings. Consequently the mechanism of conformational isomerism seems to be much more simple for the 2nd-row complex than for the iron compound for which low-lying triplet excited states play an important role both for NO dissociation (T₁(A')) and in a non-adiabatic mechanism (T₂(A')).

Table 7. Potential energy (in eV) of $[\text{Os}(\text{CN})_5(\text{NO})]^{2-}$ calculated for the different optimized structures (Figures 1 and 5) with respect to the S₀ electronic ground state energy of the standard geometry (Figure 1a).

Structure	<i>Standard</i>	<i>Flat</i>	<i>Reverse</i>	NO-bent	MNO-bent	ON-bent
S ₀	0.0	1.59	2.05			
S ₁ (A')	2.69	3.41	3.56			
T ₁ (A')	2.50	3.08	-			
T ₂ (A')	-	-	3.03			
S ₁ (A'')	-		-NO	2.54	3.31	

T ₁ (A'')	-	-	2.33	3.09	3.31
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Similarly to the ruthenium complex the osmium complex does not exhibit low-lying excited state critical geometries, the S₀ electronic ground state remaining the lowest state whatever the structure is. Likewise the S₁(A'') excited state in the reverse conformation is dissociative for the NO departure. When optimizing the T₂(A') state in the standard structure it does converge to the T₁(A') state, indicating a potential degeneracy of these two excited states in the osmium compound. Again the mechanism of conformational isomerism in [Os(CN)₅(NO)]²⁻, analogously to the one proposed for [Ru(CN)₅(NO)]²⁻, should follow an adiabatic process along the S₀ electronic ground state potential energy surface.

Conclusion

On the basis of density functional theory (TD)-DFT and MS-CASPT2 based on RASSCF wavefunction, we have investigated the electronic, structural and optical properties of the group 8 series metal nitrosyl complexes. The energy profiles connecting the N-bound (η^1 -N), O-bound (η^1 -O) and side-on (η^2 -NO) conformations show that conformational isomerism is unlikely in the S₀ electronic ground state. The presence of too significant barriers (> 2 eV) is due to the almost decoordination of the NO ligand to ensure the isomerization. We have shown that the energetics of the three complexes is similar with the N-bound structure being the most stable, the main difference being the triplet ground state of the O-bound isomer for the iron complex. The optical spectra and the low lying triplet excited states exhibit an important MLCT, LLCT and IL mixed character. The photo-induced isomerism is certainly controlled by the presence of two competing triplet

T₁ and T₂. Further studies should include computation of the low lying triplet potential energy profile at the multi reference level, beyond the capabilities of DFT.

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