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ACCEPTED MANUSCRIPT

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**On the contribution of fullerene to the current of planar heterojunction
organic solar cells.**

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Summary

Recently, significant progress in the field of organic photovoltaic cells was obtained by substituting new electron acceptor molecules to the fullerene, which was attributed to the fact that the fullerene absorption is quite small. Nevertheless, we demonstrate in the present work that, in the case of inverted cells, *i.e.* when the transparent bottom electrode is used as cathode, the contribution of fullerene to the J_{sc} short-circuit current of the cells, if not dominant, is not negligible; and that mainly in the short wavelength spectral range. The experimental results are confirmed by an optical simulation. Due to this significant contribution to J_{sc} , the light transmission of the transparent electrode towards the UV-part of the spectrum is crucial for inverse cell performances. When a transparent conductive electrode based on an alternative dielectric/metal/dielectric structure is substituted to ITO, such as ZnS/Ag/TiO₂, the study allows to obtain promising results, although there is a loss of performance due to the decrease of transmission of TiO₂ below 400 nm in wavelength.

Keywords: Organic solar cells, Planar heterojunction, External Quantum Efficiency, Absorption, Optical Modeling

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1. Introduction

Due to their lightweight, flexibility, semi-transparency [1] organic photovoltaic cells (OPVs) are complementary to silicon cells. The spectacular progress achieved in the field of OPVs is due to the use of low energy gap molecules [2], new electron acceptors [3], and optimization of their geometry [4, 5]. A basic OPV is made up of an organic active layer, consisting of an electron donor (ED) and an electron acceptor (EA), sandwiched between two electrodes, one of which being transparent. It is usual to employ this transparent layer, generally an ITO layer, as an anode. Such a configuration is called classic OPV (COPV). In contrast, when the transparent conductive electrode (TCE) is used as a cathode it is called inverted OPV (IOPV). It was shown that IOPVs are more stable than COPVs, hence the interest in studying these inverted structures, even if, at times, their performances are inferior to that of COPVs [6], but this is not systematically the case when the ED is SubPc [7].

Whatever their geometry, IOPV or COPV, two configurations are possible for the organic layer, either bulk-heterojunction (BHJ) or planar-heterojunction (PHJ). If BHJ-OPVs allow achieving highest efficiencies, PHJ-OPVs, due to their simplicity, permit obtaining more reproducible results [7]. Moreover, while when a fullerene derivative is used as electron acceptor in BHJ-OPV it is well admitted that its contribution to the OPV current is small, it appears that it is not so evident in the case of inverted PHJ-OPVs (i.e. PHJ-IOPV). As a matter of fact, in the present work, using SubPc/C₆₀ as heterojunction ED/EA in PHJ-IOPVs, we show that, if the SubPc contribution to the external quantum efficiency (EQE), is dominant, that of C₆₀ is far to be negligible. This experimental result is comforted by optical simulation of the OPVs. On the other hand, it is well admitted that flexible optoelectronic is promised rapid development in the coming years [8]. Therefore, ITO being poorly flexible, it is necessary to look for new flexible transparent electrodes. Among the possible substituents, Dielectric/Metal/Dielectric (DMD) structures are one of the most promising [9]. We have probed two of these DMD structures as cathodes in the inverted OPVs studied in the present work. We show that the transmission spectrum of the TCEs is primordial for OPVs performances.

2. Description of approach and techniques

As described in supporting S1, OPVs were realized by deposition under vacuum on TCE coated glass substrates [10]. The optically and electrically active core of the cells consisted of the planar heterojunction SubPc/C₆₀. The transparent electrode was either ITO or a DMD

structure. Two kinds of DMD structures were used, ZnS/Ag/TiO₂, with the thicknesses 31 nm/10nm/31nm for the different layers [11], and MoO₃/Ag/MoO₃, with 35 nm/11 nm/15 nm respectively [12]. The second electrode was an aluminum layer thick of 100 nm. It is known that to optimize the charge collection it is necessary to introduce buffer layers at the electrode/organic material interfaces [13]. At the interface anode/SubPc we have introduced a thin MoO₃ layer as hole transporting layer (HTL), MoO₃ being well known as very efficient hole collector, even if the work function of the anode itself is not very high [14]. As electron transporting layer (ETL), Alq₃ (Tris (8-hydroxyquinoline) aluminum), was introduced between the cathode and the fullerene. When used in COPVs, the Alq₃ ETL is called exciton blocking layer (EBL) and was proved to be very efficient [15]. All chemical products were provided by Aldrich and CODEX. The different layers evocated above were sequentially deposited such as, in the classical configuration, we had Glass/TCE/MoO₃/SubPc/C₆₀/Alq₃/Al and in the inverted geometry Glass/Alq₃/C₆₀/SubPc/MoO₃/Al (supporting information S2). The device energy diagrams for both OPV configurations are shown in Figure 1.

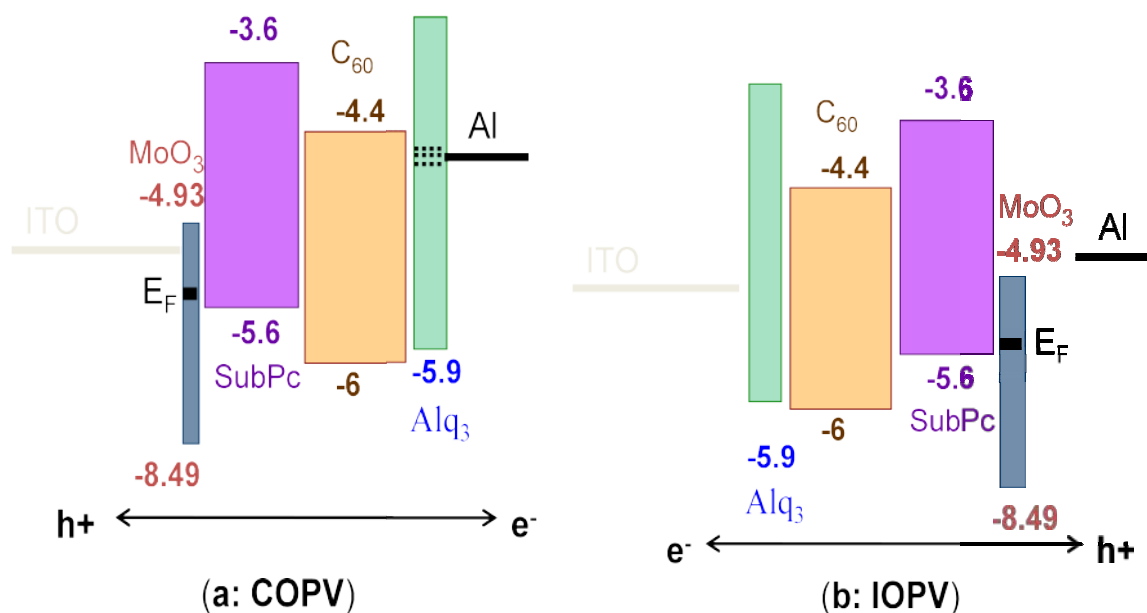


Figure 1: Device energy diagrams for both OPV configurations: (a) COPV, (b) IOPV.

After realization, the OPVs were characterized by J-V, External Quantum Efficiency (EQE) and optical absorption measurements (these experimental techniques are presented in the supporting information S3).

3. Experimental results and discussion

3.1 Optimization of the performances of the inverted OPVs, ITO being the transparent electrode.

As evocated before, the main goal of this work is to study the contribution of C₆₀ to the photocurrent of OPVs and, more precisely, to compare its contribution in the classical OPVs to that in the inverted OPVs. To carry out this comparison, all other things being equal, we used in the inverse cells, the same materials as in the classical cells. Therefore, knowing that for classical cells the configuration was Glass/ITO/MoO₃/SubPc/C₆₀/Alq₃/Al, for reverse structures we used the configuration Glass/ITO/Alq₃/C₆₀/SubPc/MoO₃/Al. We have already shown that, in the case of inverted OPVs it was difficult to achieve reproducible performances [16]. Actually, it was demonstrated that the open circuit voltage (V_{oc}), the fill factor (FF) and therefore the OPV efficiency (η) depend on the morphology of the SubPc layer, which itself depends on its deposition rate. Optimum results were obtained when the deposition rate of SubPc was 0.02±0.01nm/s, the SubPc films being homogeneous. For higher deposition rate, the V_{oc} and FF values and the reproducibility of the results decrease. It was also shown that the optimum thickness of the SubPc layer was 16 nm, for smaller or higher thickness the OPV efficiency decreases [16]. At the beginning of the present work, even after optimization of the deposition rate and film thickness of SubPc, some variations in V_{oc} and FF values persist. Actually, the SubPc purity given by our supplier was 85 at. %, which means that from one batch to another one some significant differences can be encountered. This crucial point should be overcome. Therefore we had to purify the material. It was shown earlier that it was possible to purify phthalocyanine dyes by simply carrying out successive sublimations. Indeed, by using the same charge in the sublimation crucible, there is a progressive "auto purification" of the phthalocyanine dye [17]. Taking advantage of our experience acquired with the CuPc [18] we undertaken the purification of SubPc by a series of initial deposits which aimed to optimize the efficiency of the cells. Typical results are summarized in Figure 2. It can be seen that the performances of the OPVs depend very significantly of the number of deposits made, mainly through the values of V_{oc} and FF.

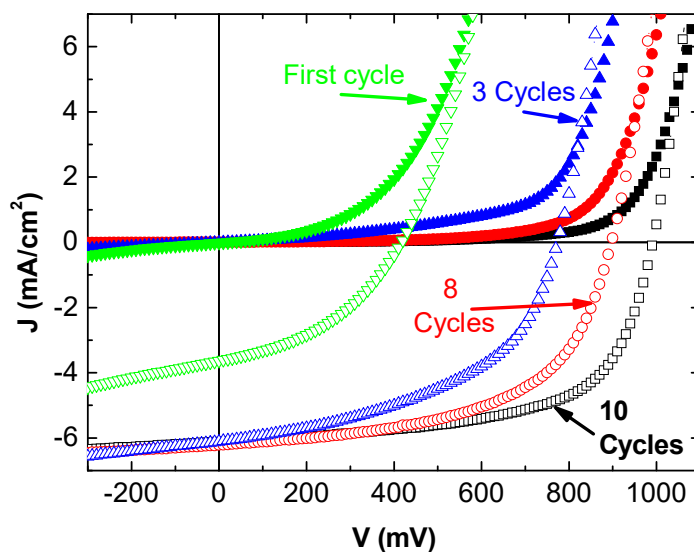


Figure 2: Optimization of the J-V characteristics of inverted OPV by purifying SubPc through deposition cycles: 1 cycle (\blacktriangledown), 3 cycles (\blacktriangle), 8 cycles (\bullet) and 10 cycles (\blacksquare) (the filled symbols correspond to dark conditions while the open symbols correspond to light AM 1.5 conditions).

It is clear that V_{oc} and FF increase with the number of cycles of deposits. For the first deposition, SubPc layer quality is very poor as shown by the J-V characteristics obtained. Then, up 10 cycles, there is a progressive improvement of these characteristics due to the increase of V_{oc} and FF while J_{sc} does not vary significantly after 3 cycles. From 10 cycles the characteristics are optimal and stable. It must be noted that our experience with CuPc discard the hypothesis of a possible artifact due to a problem on our equipment to deposit the film. Actually when the purity of CuPc is 99%, it is not necessary to carry out successive sublimations to obtain stable results, while it is necessary when the purity of CuPc is only 95% [18]. The effectiveness of our process in purifying these phthalocyanine dyes shows that some impurities present in the commercial powder have smaller sublimation temperature than the dyes, which allows the purification through deposition cycles.

After stabilization of the OPCs performances, *i.e.* purification of the SubPc present in the sublimation crucible, we proceeded to the comparison of the J-V characteristics and the EQE spectra of inverted and classical OPVs.

3.2 Classical and inverted OPVs using ITO as transparent electrode

The J-V characteristics of the inverted and classical OPVs manufactured during the present work are represented in figure 3. The champion IOPV produced a J_{sc} of 6.12 mA/cm^2 , a V_{oc} of 0.99 V , a FF of 0.61 and η of 3.76% , while $R_s = 20 \text{ } \Omega$ and $R_{sh} = 1800 \text{ } \Omega$. On the other hand, the COPV produced a J_{sc} of 7.65 mA/cm^2 , a V_{oc} of 1.00 V , a FF of 0.58 and η of 4.44% , while $R_s = 15 \text{ } \Omega$ and $R_{sh} = 1300 \text{ } \Omega$. The obtained results are rather close, the main difference being due to J_{sc} , which is smaller in the case of IOPV, as it is often the case in PHJ-OPVs.

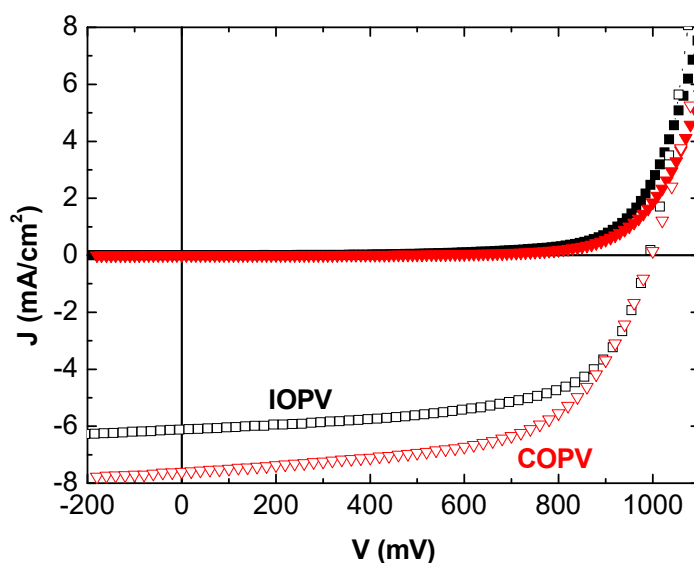


Figure 3: J-V characteristics of IOPV (■) and COPV (▼) (the filled symbols correspond to dark conditions while the open symbols correspond to light AM 1.5 conditions).

Typical EQE spectra of both types of OPVs are presented in Figure 4 and absorption spectra are shown in Figure 5. In the EQE spectra the signal situated between 500 nm and 650 nm can be attributed to SubPc. In the short wavelengths spectral range, the signal below 400 nm corresponds to the second absorption band of SubPc, but above all to C_{60} . As shown by the absorption curves, the intermediate domain between these two contributions must be mainly attributed to C_{60} . It is remarkable to note that the C_{60} contribution in EQE is far to be negligible in the case of IOPV. In order to check the importance of the contribution of C_{60} to the short circuit current we have studied series of IOPVs with different thicknesses of C_{60} : 20 nm , 30 nm and 40 nm . Typical J-V characteristics are reported in the supporting information S4. The main information, which confirms the importance of the contribution of C_{60} , is that the value of J_{sc} decreases when the thickness of C_{60} decreases: $J_{sc} = 4.10 \text{ mA/cm}^2$ when the

thickness of C_{60} is 20 nm, $J_{sc} = 4.95 \text{ mA/cm}^2$ for 30 nm and $J_{sc} = 6.12 \text{ mA/cm}^2$ for 40 nm of C_{60} . Beyond, J_{sc} starts to decrease due to the increase of the series resistance.

It can be seen in Figure 5 that the overall shape of the absorption spectra of IOPV and COPV are rather similar. Nevertheless, the ratio of the values measured at 450 nm and 590 nm ($OD_{450}/OD_{590} \text{ nm}$) is different. It is greater than 1 in the case of IOPVs and smaller than 1 for COPVs, *i.e.* the contribution of C_{60} is dominant in IOPVs. Moreover, it can be noted that the C_{60} contribution in the UV spectral domain is very important.

Since the EQE spectra are clearly dependent of the absorption spectra of the organic layers, as shown by the Figures 3 and 4, we have proceeded to a simulation of the optical absorption of the different structures in order to try to discriminate the contribution of each constituent to the OPVs current.

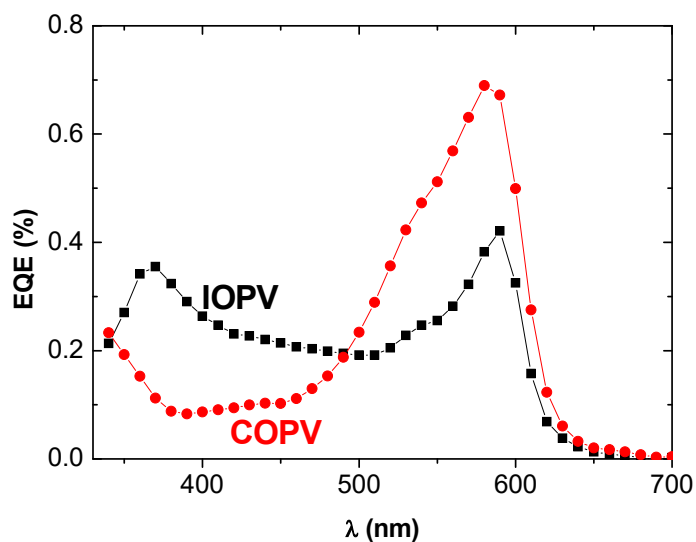


Figure 4: Typical EQE spectra of IOPV (■) and COPV (●).

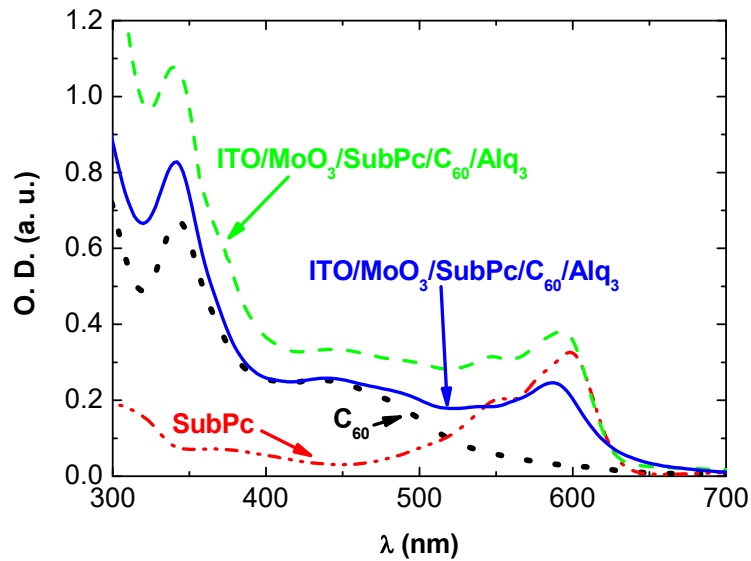


Figure 5: Optical absorption spectra of SubPc (●●—●●—●●), C_{60} (●●●●●), ITO/Alq₃/C₆₀/SubPc/MoO₃ (—●—●—●) and ITO/MoO₃/SubPc/C₆₀/Alq₃ (—●—).

3.3 Optical simulation

A numerical model based on a Transfer Matrix Method (TMM), is employed to predict the optical behavior of the considered electrodes or organic solar cells [19, 20]. It supposes that the stacks are composed of flat, massive and homogeneous layers, that the air surrounds the multilayer, and that light penetrates at normal incidence. The input data of our simulation algorithm are the thicknesses of each material and their complex optical constants, previously obtained by ellipsometric spectroscopy or by scientific literature [21, 22]. The parameters used for the simulation of transmission spectrum are given in the supporting information S5 and in the references [21, 23].

This method makes it possible to calculate the optical properties such as the reflectance R , the transmittance T and the absorbance A , and to optimize the thicknesses of our structures [22, 23]. As an example, we are able to compare the modeled transmittance to those measured (figure 6) from a typical ZnS/Ag/TiO_x electrode manufactured by a vacuum electron beam evaporator on a glass substrate, then characterized by UV-Visible spectrophotometer. Results are globally in good agreement for such Glass/ZnS (31 nm)/Ag (10 nm)/TiO_x (31 nm) electrode (the measured curve is in full line while the simulated curve is in dashed line).

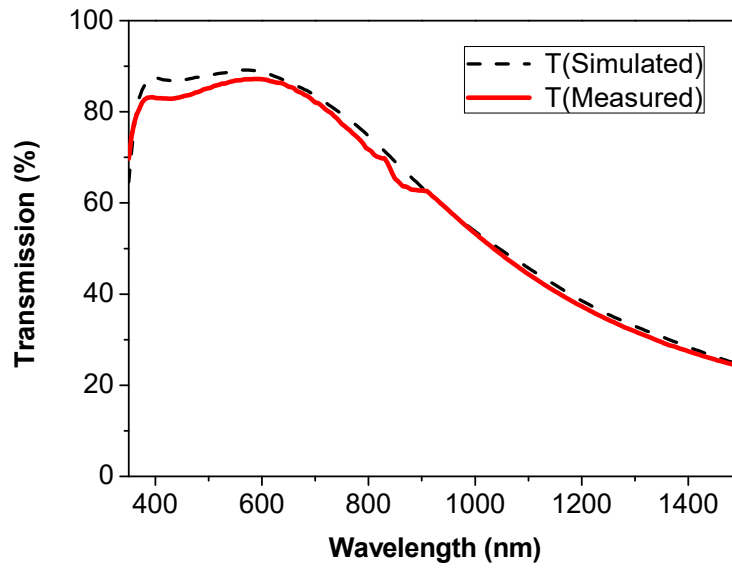


Figure 6: Simulated (dashed line) and measured (full line) Transmittance (T), for Glass/ZnS(31 nm)/Ag(10 nm)/TiOx(31 nm) electrode.

Our optical numerical method was used to calculate the total absorbance in a classical design (COPV): ITO (80) / MoO₃ (4) / SubPc (20) / C₆₀ (45) / Alq₃ (9) / Al, and in an inverted design (IOPV): ITO (80) / Alq₃ (9) / C₆₀ (45) / SubPc (16) / MoO₃ (4) / Al (the values in brackets are the thicknesses of each layer in nm). We observe in figure 7 the absorbance of such solar cells in the [400, 1200]-nm spectral range, and note a significant difference of optical behavior between COPV and IOPV in the [400, 700]-nm range.

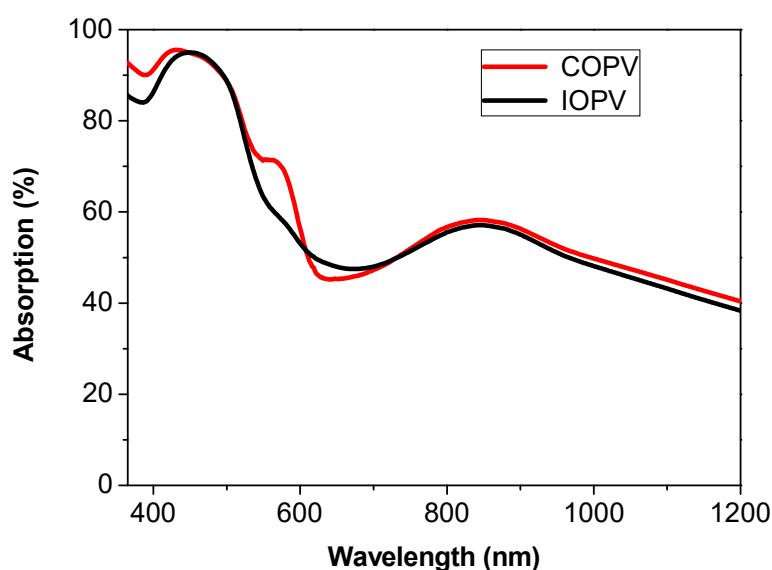


Figure 7: Total absorption in the following OPV cells:

- COPV classical design: ITO (80)/MoO₃ (4)/SubPc (20)/C₆₀ (45) /Alq₃ (9)/Al
- IOPV inverted design: ITO (80) / Alq₃ (9)/ C₆₀ (45) / SubPc (16) / MoO₃ (4) / Al

However, only the absorbance in the photo-active layers is likely to produce photocurrent in the cells. Therefore, we calculated the useful intrinsic absorption in the only SubPc and C₆₀ active layers in order to correlate our results with the EQE ones in the same spectral domain (figure 8). It appears clearly in the classical design two absorption bands: one below the wavelength of 550 nm which is mainly due to the absorption of the C₆₀ acceptor material, and one above this wavelength that is mainly attributed to the SubPc donor molecule. In the inverted cell, the optical contribution of the C₆₀ is increased between 425 and 525 nm, while those of the SubPc is decreased between 525 and 600 nm. These variations can be explained by the optical interferences taking place in the whole stack, which modify the distribution of the electromagnetic field and thus the absorbance, and by the thickness of the SubPc that is a little smaller in the inverted cell.

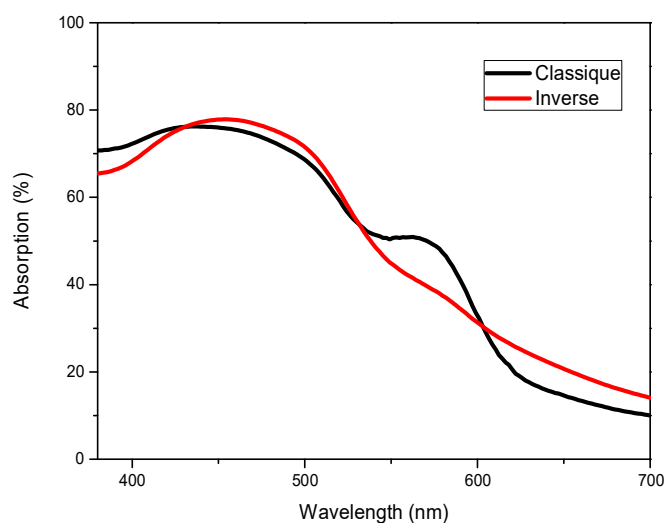


Figure 8: Absorption of the only (SubPc+C₆₀) active materials inside inverted (IOPV) and classical (COPV) design.

It is also remarkable to note some strong similarities with the EQE curves of the figure 4 as the presence of a "crossing" wavelength (around 500-525 nm) between 2 spectral domains where the COPV is better than IOPV (in the [500-520, 600]-nm range), and where the IOPV is better than COPV (in the [350-400, 500-520]-nm range). This demonstrates the significant influence of the absorption on the EQE of the OPVs, and the non-negligible contribution of the C₆₀ acceptor molecule in its absorption range.

3.4 Inverted OPVs with different transparent conductive electrodes

After the study of OPVs using ITO as TCE, we have substituted DMD to ITO, using two different types of DMD: ZnS/Ag/TiO₂ and MoO₃/Ag/MoO₃. We have used them as cathode, the main goal of the present study being not to match the performances obtained with ITO, but to study the influence of the transmission curves of the new TCEs on the electrical characteristics of IOPVs. The transmission curves of the different TCEs used are visible in Figure 9. It can be seen that there is a decrease of the light transmission of the ZnS/Ag/TiO₂ structure for wavelengths below 375 nm, while in the spectral range above 550 nm, the MAM structure is the TCE that transmits the least. In Figure 9 we have also introduced the absorption spectra of SubPc and C₆₀ in order to facilitate the discussion of the shape of the EQE curves further shown. If the main contribution of the absorption spectrum of SubPc is situated in the wavelength range that corresponds more or less to the maximum transmission range of the three TCEs, it is not the case with C₆₀. Actually, the main contribution of the

absorption spectrum of C_{60} is situated in the very short wavelengths range (below 400 nm), with the second absorption peak of SubPc.

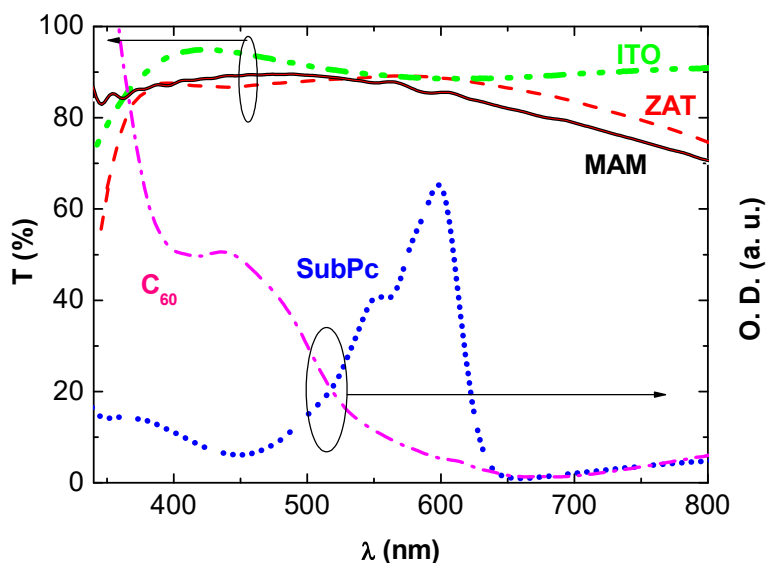


Figure 9: Typical transmission spectra of $MoO_3/Ag/MoO_3$ (—), $ZnS/Ag/TiO_2$ (---) and ITO (····) and absorption spectra of C_{60} (---) and SubPc (····).

Typical J-V characteristics obtained with the new TCEs, when they are substituted to ITO in IOPVs, are shown in Figure 10 and summarized in Table 1, while the corresponding EQE spectra are shown in Figure 11. The performances are sensibly smaller than that achieved with the ITO electrode. Actually, in the case of $ZnS/Ag/TiO_2$ -TCE, we have already shown that the performances of the OPVs, are limited by the diffusion of some Ag towards the surface of the TCE [24]. In the case of $MoO_3/Ag/MoO_3$ it can be seen that with the ETL used in the present work, Alq_3 , the result is very bad. As a matter of fact, it is known that the value of the work function (W_F) of MoO_3 is higher than 5 eV, which makes that it is a very efficient HTL [25] but reversely a very bad ETL.

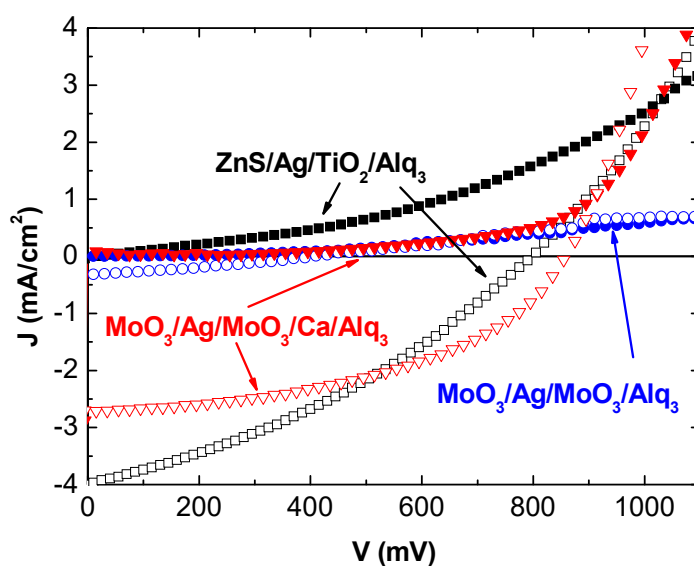


Figure 10: J-V characteristics of IOPV with ZnS/Ag/TiO₂/Alq₃ (■), MoO₃/Ag/MoO₃/Alq₃ (●) and MoO₃/Ag/MoO₃/Ca/Alq₃ (▼) as cathode/ETL (the filled symbols correspond to dark conditions while the open symbols correspond to light AM 1.5 conditions).

So, when used as cathode in IOPVs the efficiency of the cells is very small. Nevertheless, when a small W_F material, Ca, is introduced at the interface cathode/organic material, a far better result is obtained. Knowing that the goal of the present study was not to overpass some record but to study the effect of the transmission spectra of the new TCE on the IOPVs performances we focused our attention on the EQE spectra. It can be seen in Figure 11, that the contribution of the main SubPc absorption peak is clearly visible for all samples. However, in the small wavelengths range, the EQE signal decreases when ZnS/Ag/TiO₂ is the cathode, while a contribution to the current is clearly visible with other TCEs.

Cathode	ETL	Voc (V)	Jsc (mA/cm ²)	FF (%)	η (%)
ZnS/Ag/TiO ₂	Alq ₃	0.80	4.00	36	1.15
MoO ₃ /Ag/MoO ₃	Alq ₃	0.40	0.32	31	0.04
MoO ₃ /Ag/MoO ₃	Ca/Alq ₃	0.86	2.71	48	1.10

Table 1: Typical parameters of IOPVs (Cathode/ETL/C₆₀/SubPc/MoO₃/Al) with different cathodes.

More precisely, relatively to the maximum of the SubPc contribution, the MoO₃/Ag/MoO₃-TCE permits the best answer. These results can be discussed in the light of the spectra of transmission of TCEs and absorption of SubPc and C₆₀. We have seen (Figure 9) that there is a decrease of the light transmission of the ZnS/Ag/TiO₂-TCEs just in the range where the C₆₀ absorption is maximum, while that of MoO₃/Ag/MoO₃ is higher than that of ITO, which justifies the shape of the EQE spectra.

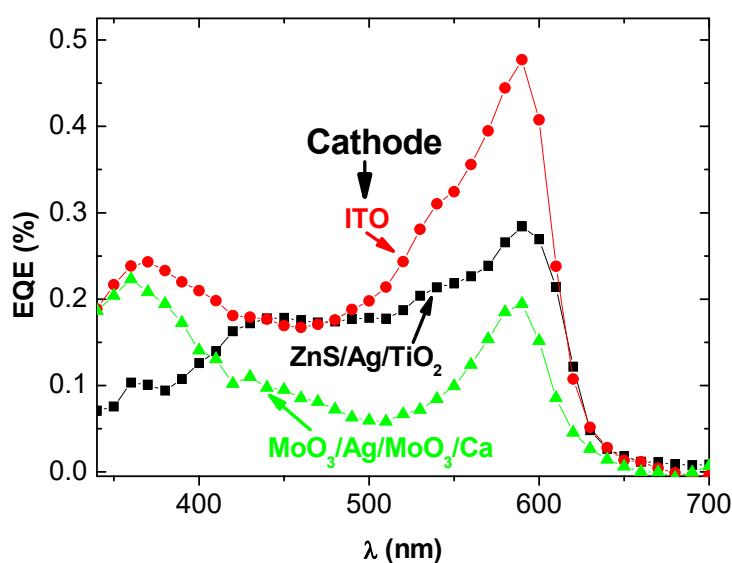


Figure 11: Typical EQE spectra of IOPV with different cathodes: (■)ZnS/Ag/TiO₂, (●) ITO and (▲)MoO₃/Ag/MoO₃.

5. Conclusion

In the present work we demonstrate that, in the case of PHJ-IOPVs, the contribution of the C₆₀ layer to the J_{sc}, if not principal, is far to be negligible. As a matter of fact, the contribution of SubPc to J_{sc} is dominant in the visible range, while that of C₆₀ prevails in the short wavelengths spectral range. These results are corroborated by the optical simulation, which demonstrates the non-negligible contribution of the C₆₀ acceptor molecule to the absorption of inverse OPVs in the UV-part of the spectral range and therefore to J_{sc}. Moreover, we show that the transmission spectrum of the TCE is determinant for the shape of the EQE spectrum and thus for J_{sc} and the OPV efficiency. The alternative ZnS/Ag/TiO₂ indium-free TCE is also proved to be a credible alternative to ITO. However, if TiO₂ is a good electron collector,

due to its weak extraction work, the decrease in its transmission below the wavelength of 400 nm penalizes the PHJ-OPVs performances.

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