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# From Canals to the Coast: Dissolved Organic Matter and Trace Metal Composition in Rivers Draining Degraded Tropical Peatlands in Indonesia

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**Abstract.** Worldwide, peatlands are important sources of dissolved organic matter (DOM) and trace metals (TM) to surface waters and these fluxes may increase with peatland degradation. In Southeast Asia, tropical peatlands are being rapidly deforested and drained. The black rivers draining these peatland areas have high concentrations of DOM, and the potential to be hotspots for CO<sub>2</sub> release. However, the fate of this fluvial carbon export is uncertain, and its role as a trace metal carrier has never been investigated. This work aims to address these gaps in our understanding of tropical peatland DOM and associated elements in the context of degraded tropical peatlands of Indonesian Borneo. We quantified dissolved organic carbon and trace metals concentrations in the dissolved and fine colloidal (<0.22µm) and coarse colloidal (0.22 – 2.7 µm) fractions and characterized the characteristics (δ<sup>13</sup>C, Absorbance, Fluorescence :excitation-emission matrix and PARAFAC analysis) of the peatland-derived DOM as it drains from peatland canals, flows along the blackwater Ambawang River, and eventually mixes with Kapuas Kecil River before meeting the ocean near the city of Pontianak in West Kalimantan, Indonesia. We observe downstream shifts in indicators of in-stream processing. The increase in the δ<sup>13</sup>C of DOC, along with an increase in the C1/C2 ratio of PARAFAC fluorophores, and decrease in SUVA (Specific UV Absorbance) along the continuum suggest the predominance pf photo-oxidation. However, we also observe very low dissolved oxygen concentrations, suggesting that oxygen is quickly consumed by microbial degradation of DOM in the shallow layers of water. Black rivers draining degraded peatlands show significantly higher concentrations of Al, Fe, Pb, As, Ni, and Cd. A strong association is observed between DOM, Fe, As, Cd and Zn in the dissolved and fine colloid fraction, while Al is associated to Pb and Ni and present in a higher proportion in the coarse colloidal fraction. We additionally measure the isotopic composition of lead released from

degraded tropical peatlands for the first time and show that Pb originates from anthropogenic atmospheric deposition. Degraded tropical peatlands are important sources of DOM and trace metals to rivers and a secondary source of atmospherically deposited contaminants.

**Keywords:** Tropical peatlands, Dissolved Organic Matter, Absorbance, Fluorescence, PARAFAC, Stable isotopes, Trace metal, Lead isotopes

## 1. Introduction

Most Southeast Asian tropical peatlands developed as domes beneath ombrotrophic peat swamp forests (Page et al., 2006; Cobb et al. 2017). They store at least 68.5 Pg C, or 15-19% of the global peat carbon stocks (Page et al., 2011; Dargie et al., 2017; Lahteenoja et al., 2012). They have experienced widespread degradation as a result of deforestation, conversion to agriculture and drainage, which all accelerated in the late 1970s (Miettinen et al., 2011). This abrupt change in land use, and corresponding lowering of the water table, has led to subsidence and a massive release of carbon from peatlands to the atmosphere due to enhanced aerobic decomposition of organic matter from the drained peat. Extensive work has focused on quantifying the resulting CO<sub>2</sub> fluxes (Couwenberg et al., 2010; Jauhiainen et al., 2012; Miettinen et al., 2017) and land surface subsidence (e.g. (Hooijer et al., 2012; Carlson et al., 2015).

Drainage canals are dug in forested peatlands for multiple reasons: first, as a mechanism to transport timber out of the peatland during deforestation, and later to lower the water table, making the land suitable for agriculture. These peatland drainage canals channel water from the peatlands to surrounding surface waters. The resulting fluvial export of dissolved organic matter (DOM) has been recognized as a significant component of the carbon budget of tropical peatlands, that could increase with deforestation and peatland exploitation (Moore et al. 2011; Gandois et al. 2013). Indonesia alone contributes over 10% of the global riverine dissolved organic carbon (DOC) input into the ocean (Baum et al., 2007), as a result of both high peatland coverage and high precipitation rates. This proportion is likely to increase with rapid peatland conversion to agriculture, which destabilizes long-term peat C stocks (Moore et al., 2013).

Another implication of DOM transfers from peatlands to surface water is the transport of associated elements, especially trace metals (TM). Tropical peatlands in Southeast Asia are mainly ombrotrophic systems, which receive critical nutrients through atmospheric deposition, and serve as a sink for atmospheric pollutants (Weiss et al., 2002). Northern peatlands have been shown to constitute a source of major and trace elements to surface waters (J. J. Rothwell et al. 2007; Tanja Broder and Biester 2017; Jeremiason et al. 2018). This has important implication : as a result of colloidal association between peatland-derived organic molecules and Fe, northern peatlands are responsible for a significant transfer of Fe to the Atlantic Ocean (Krachler et al. 2010, 2012). In the UK, peat degradation and erosion has led to the dispersion of lead into watersheds, which previously accumulated through atmospheric deposition over decades (Rothwell et al., 2008). Although drainage of tropical peatlands is occurring at a rapid rate across Southeast Asia, to our knowledge no data are available on trace metal release in black rivers draining tropical peatlands.

Black rivers draining peatlands (as defined in Alkhatib et al (2007)) also have the potential to be hotspots of fluvial carbon degassing (Müller et al., 2015; Wit et al., 2015). By measuring pCO<sub>2</sub> in Indonesian and Malaysian black rivers, Wit et al. (2015) estimated that 53% of DOC entering surface waters was converted to CO<sub>2</sub>, which is similar to global averages for inland waters. In contrast, blackwater river measurements and incubations by Martin et al., (2018) found a smaller proportion of DOC was processed in rivers. Rixen et al., (2008) also found a large proportion of the DOM was resistant to decomposition in a laboratory incubation study. These studies have focused on CO<sub>2</sub> measurements and incubations to assess the potential for DOM processing.

Monitoring changes in DOM composition in canals and rivers can provide complementary information on the extent of in-stream processing of fluvial carbon, and potential reemission of greenhouse gas (GHG) to the atmosphere. Qualitative evaluation of in-stream DOM transformation by UV light and microbial processes can be performed using isotopic and optical characterization of DOM. The stable isotope signature of DOM is an indicator of its origin (Hood et al., 2005; Barber et al., 2017), but also transformation processes. Lalonde et al (2014) assessed photochemical processing of DOM in major large rivers worldwide, and found that it caused an increase in the  $\delta^{13}\text{C}$  of DOM of 0.5 to 2.3‰. However, microbial processing is also expected to lead to an increase in the  $\delta^{13}\text{C}$  of DOM. Optical properties of DOM are also sensitive indicators for DOM processing (Spencer et al. 2009; Harun et al. 2015; Hansen et al. 2016), with contrasted effects on optical properties. Microbial processing is generally found to increase the aromaticity of DOM by selective processing of less aromatic molecules, while photo-oxidation tends to decrease aromaticity, because of selective photo-oxidation of aromatic moieties (Spencer et al., 2009; Hansen et al., 2016).

In summary, although there has been an increase in efforts to quantify DOC exports from tropical peatlands, our complementary understanding of the transfer of associated elements and in-stream processing of DOM remains limited. This work aims to address these gaps in our understanding of the composition and evolution of tropical peatland DOM and how it could act as a carrier of trace metals to surface waters, in the context of highly degraded tropical peatlands of Indonesia. We characterize the quality of the peatland-derived DOM and trace metals as they drain from peatland canals, flow along blackwater rivers, and eventually mix with a whitewater river before meeting the ocean. We assess spatial and seasonal changes in the organic matter quality, and document changes in DOM composition due to transport, mixing, and processing. We also assess black river trace metal release to surface waters, analyzing trace metal concentrations and the isotopic composition of lead released from degraded tropical peatlands for the first time.

## **2. Material and methods**

### **2.1 Study area**

The study area is located in West Kalimantan, Indonesia, near the city of Pontianak (0.09°N, 109.24°E) on the island of Borneo. The climate is humid equatorial with 3041±532mm of rainfall and a mean annual temperature of 27°C (1985-2017 data). The study focused on the Ambawang River, which flows into the Landak and Kapuas Kecil rivers. It is a black river draining a watershed (approximately 706 km<sup>2</sup>) entirely covered with peatlands. This river was

selected to represent water of exclusively peatland origin. All peatlands in the sampling area have been drained and converted to agriculture.

## 2.2 Sample collection and treatment

Two sampling campaigns were conducted during the dry season in June 2013 and during the wet season in January 2014. Using a boat, samples were collected in the center of the river, from the origin of the Ambawang river (BR, black river sites) to its downstream confluence with the Landak and Kapuas Kecil Kecil (WR, white river sites). White river samples collected upstream of the confluence with the white river (WRu). Drainage canals (DC) flowing into the black river were also sampled during the second sampling campaign. During the wet season campaign, a rain collector was installed on the roof of the Pontianak's meteorological station to collect rain samples for lead isotopic analysis. In situ parameters (pH, conductivity and dissolved oxygen) were measured using a multiparameter probe (WTW, Germany). Depth profiles of dissolved oxygen in the black river were also measured with an oxygen microelectrode (MI-730 dip-type micro-oxygen electrode and O2-ADPT adapter; Microelectrodes, Inc., Bedford, NH, USA). Frequent calibration was performed with a zero oxygen solution and distilled water equilibrated to ambient oxygen concentrations, where temperature was carefully monitored. To create the zero oxygen solution, 1 g of sodium sulfite ( $\text{Na}_2\text{SO}_3$ ) and a few crystals (~1 mg) of cobalt chloride ( $\text{CoCl}_2$ ) was dissolved in 1 L of distilled water. For measurements of additional parameters, a larger volume of water was collected for further analysis. Samples were filtered immediately following collection on the boat using a portable peristaltic pump (Geotech, USA) and prebaked and pre-weighed GF/F filters (0.7  $\mu\text{m}$ ) and stored in glass bottles for DOC,  $\delta^{13}\text{C}$ -DOC and optical properties of DOM analysis. Samples were filtered with cellulose acetate filters (0.22  $\mu\text{m}$ ) and stored in polypropylene vials for analysis of major nutrients and trace element. DOC analysis was repeated on the cellulose acetate samples and DOC concentrations did not differ significantly based on filtration at 0.2 or 0.7  $\mu\text{m}$ . During the wet season, at selected sites (8), samples were first filtered with GF/D filters (2.7 $\mu\text{m}$ ) to assess to the coarse colloidal fraction of trace metals and DOC.

## 3. Sample analysis

Non-purgeable organic carbon (NPOC, referred to hereafter as DOC) was analyzed on filtered (GF/F Whatman) samples after acidification to pH 2 (HCl) with a TOC-V CSH analyzer (Shimadzu, Japan), with a quantification limit of 1  $\text{mg L}^{-1}$ . Major cations and anions were analysed using an HPLC (Dionex, USA). The quantification limit was 0.5  $\text{mg L}^{-1}$  for chloride, nitrate and sulphates and 0.025  $\text{mg L}^{-1}$  for ammonium, potassium, magnesium and calcium. Certified material (ion 915 and ion 96.4 Environment and Climate Change Canada, Canada) was included in the analytical loop and recovery was >95% of the certified value. For trace element analysis, samples were acidified with ultrapure  $\text{HNO}_3$  prior to ICP-MS (7500 ce, Agilent Technologies) analysis.  $^{115}\text{In}$  was used as an internal standard, and SLRS-4 (River water certified for trace elements) was used as a reference material on every run and accuracy (recovery>95%) was checked. Determination limits were < 0.5  $\mu\text{g g}^{-1}$  for Fe and Al, <0.05  $\mu\text{g g}^{-1}$  for Ni, Cu and Zn and < 0.005  $\mu\text{g g}^{-1}$  for Cd and Pb. Pb isotope ratios ( $^{206}\text{Pb}/^{207}\text{Pb}$ ;  $^{208}\text{Pb}/^{206}\text{Pb}$ ) in water samples were analyzed using a

High Resolution ICP-MS (Thermo Element II XR; OMP service ICP-MS, Toulouse, France). Measurements were corrected for mass bias using individual sample bracketing with certified and adequately diluted NIST NBS-981 (100 ng L<sup>-1</sup> to 500 ng L<sup>-1</sup>) according to Krachler et al. (2004).

The UV absorption spectra of pore water were measured with a spectrophotometer (Secoman UVi-lightXT5) from 190 to 700 nm in a 1 cm quartz cell. The Specific UV Absorbance at 254 nm (SUVA, L mg<sup>-1</sup> m<sup>-1</sup>) was calculated as follows:  $SUVA = A_{254} / (b \cdot DOC)$  (Weishaar et al., 2003), where  $A_{254}$  is the sample absorbance at 254 nm (non-dimensional),  $b$  is the optical path length (m) and DOC is in mg L<sup>-1</sup>. The baseline was determined with ultra-pure water.

Emission Excitation Matrices (EEM) were acquired using a Hitachi F4500 fluorescence spectrometer. Emission spectra were acquired from 250 to 550 nm for excitation ranging from 250 to 550 nm. The slits were set to 5 nm for both the excitation and emission monochromators. The scan speed was 2400 nm min<sup>-1</sup> and the integration response was 0.1 s. Fluorescence intensity was corrected from the excitation beam to ensure stability. The inner filter effect was taken into account using a dilution approach as developed by Luciani et al. (2009). The fluorescence index was calculated as defined by McKnight et al. (2001) and Jaffé et al. (2008), by the ratio of the fluorescence intensity at 470 nm to the intensity at 520 nm for a 370 nm excitation. The PARAFAC analysis (PARAllel FACtor analysis (Bro, 1997)) was performed on all samples using the PROGMEEF program in Matlab (Luciani et al. 2008).

The isotopic composition ( $\delta^{13}C$ ) of DOC was determined at the UC Davis Stable Isotope Facility, following the described procedure (<http://stableisotopefacility.ucdavis.edu/doc.html>). Briefly, a TOC Analyzer (OI Analytical, College Station, TX) is interfaced to a PDZ Europa 20-20 isotope ratio mass spectrometer (Sercon Ltd., Cheshire, UK) utilizing a GD-100 Gas Trap Interface (Graden Instruments).

#### 4. Statistical analysis

Statistical analysis was performed using the Rstudio software (Version 1.2.1335), using ggplot, dplyr, ade4 and dunn.test packages. Significant differences ( $p < 0.05$ ) between groups were evaluated using Kruskal Wallis and Dunn's post hoc multiple test.

### 5. Results

#### 5.1. Trends in water chemistry from the source of the black river to the ocean

The observed water chemistry of the Ambawang river and drainage canals is typical of black rivers draining peatlands (Table 1, Figure 2), and does not show significant differences between the wet and dry seasons. It is acidic with a pH of  $3.2 \pm 0.6$  and  $3.5 \pm 0.3$  in the drainage canals (DC) and black river (BR) respectively, has a low conductivity (DC:  $89.8 \pm 21.4 \mu S cm^{-1}$ , BR:  $85.2 \pm 21.6 \mu S cm^{-1}$ ), is anoxic (DC:  $2.3 \pm 0.3$ , BR:  $1.9 \pm 0.7 mg L^{-1}$ ), and has extremely high DOC concentrations (DC:  $35.2 \pm 5.5$ , BR:  $35.8 \pm 3.4 mg L^{-1}$ ). The Cl concentrations are low and homogeneous (DC:  $2.6 \pm 0.7$ , BR:  $2.4 \pm 0.6 mg L^{-1}$ ). After the confluence with the White River, the chemistry of the river radically changes.

An abrupt increase in pH is observed (WR:  $5.3 \pm 0.7$ ). The dissolved oxygen concentration increases up to  $3.7 \pm 1.0$  mg L<sup>-1</sup> and DOC concentrations drop sharply to  $9.2 \pm 3.2$  mg L<sup>-1</sup>. Across all samples, the DOC concentrations show a significant negative correlation with DO concentrations ( $r^2=0.63$ ,  $n=40$ ,  $p<4.10^{-9}$ ). In contrast, no increase in Cl is observed until close to the ocean (3 samples corresponding to ocean water intrusion were excluded from Figure 2). After the confluence with the black river, The DOC/Cl ratio also drops sharply from 15 in the black river and drainage canals to 5 in the white river (Figure SI.1).

## **5.2. DOM optical characteristics and stable isotopic signature**

No systematic differences were observed for the DOM characteristics between the dry and the wet season sampling campaigns. The  $\delta^{13}\text{C}$  signature of DOC (Figure 3a) is very negative, reaching  $-30.3 \pm 0.4\text{‰}$  in the drainage canals. It increases along the continuum from upstream in the black river to the ocean. Significant differences are observed between the drainage canals and black river compared to the white river. The  $\delta^{13}\text{C}$  signature of DOC is significantly negatively correlated with DOC concentration ( $r^2=0.68$ ,  $p<10^{-4}$ ,  $n=40$ ), with the highest DOC values being associated with the lowest  $\delta^{13}\text{C}$ -DOC values.

The SUVA index (Figure 3b) shows high values in the black river ( $5.3 \pm 1.2$ ), the highest values measured upstream. A wide range of values is measured in the drainage canals ( $4.3 \pm 1.4$ ). The SUVA values of the Black River are significantly higher than those measured in the Kapuas Kecil ( $4.5 \pm 0.3$ ) and its tributaries ( $4.2 \pm 0.2$ ). The fluorescence index varies widely in the drainage canals ( $1.60 \pm 0.17$ ) and black river ( $1.55 \pm 0.06$ ), but is more uniform in the white river, both upstream ( $1.55 \pm 0.03$ ) and downstream ( $1.55 \pm 0.05$ ) of the confluence with the Black river. These two optical indices show coherent spatial patterns within the black river and drainage canals (Figure 3b&d), with lower SUVA values being associated with higher FI values, in 3 drainage canals and in the black river close to their connection, sampled during the wet season. Across all samples, a significant correlation is observed between FI and SUVA values ( $r^2=0.37$ ,  $p<10^{-4}$ ,  $n=41$ ).

The EEMS of all water samples have two main peaks (Figure SI.2). The main one ( $\lambda_{\text{ex}}=250$  nm,  $\lambda_{\text{em}}=460$  nm), is coupled with a less intense peak ( $\lambda_{\text{ex}}=350$  nm,  $\lambda_{\text{em}}=460$  nm). The peaks are typical of high molecular and aromatic molecules, which have been observed in wetlands (Fellman et al., 2010). The PARAFAC analysis reveals two fluorophores: C1 ( $\lambda_{\text{ex}}=255$  nm,  $\lambda_{\text{em}}=450$  nm) and C2 ( $\lambda_{\text{ex}}=285$  nm,  $\lambda_{\text{em}}=485$  nm, Figure SI. 2) The first component constitutes from 60 to 73 % of the total fluorescence of samples. The relative contribution of these two fluorophores evolves along the sampled continuum, with the lowest values measured upstream in the black river (Figure 3c). The spatial evolution of C1/C2 ratio and  $\delta^{13}\text{C}$ -DOC are consistent. A significant ( $r^2=0.43$ ,  $p<0.001$ ,  $n=41$ ) relationship is observed across all the samples. A stronger relationship ( $r^2=0.85$ ,  $p<0.001$ ,  $n=5$ ) is observed when the drainage canals samples alone are considered.

## **5.3. Trace element concentrations and physical fractionation**

Black rivers originating from drained peatlands have a unique composition of inorganic elements. The concentrations of trace metals (Pb, Ni, Zn, Cd) as well as Al and Fe and are significantly higher in the black river and drainage canals than the concentrations in the white river (Table 1, Figure 4). For Al, Fe and As, high concentrations were measured in the Black River during the dry season. In contrast to other TM, higher Cu concentrations are measured in the white river. A PCA analysis (Figure 5) of TM concentration and DOM properties reveals specific association between DOC, Fe and As and to a lesser extend Zn and Cd, while another group is formed by Al, Pb and Ni (Figure 5). Cu shows no association with DOM but does show increased concentrations with higher FI. The first axis of PCA (load of DOC, Fe, As) strongly discriminates the black river and drainage canals samples from the white river.

The distribution of DOC and TM is presented in Table 2. Dissolved organic matter is mostly (>98%) in the dissolved and fine colloids form (<0,22  $\mu\text{m}$ ) all along the studied continuum. Iron and As are mostly present in the form of dissolved and fine colloids in the black river and drainage canals (>96%). However, after transfer to the white river, half of Fe and a third of As is in the coarse colloidal form. Zinc and Cd do not show similar patterns. Aluminium is mostly present in the coarse colloidal phase (>60%) in the black river and drainage canal and this proportion further increases in the white river (>80%). Lead is mostly present in the dissolved and fine colloid phase (>75%) in the drainage canals and black river and shifts to coarse colloidal (>60%) forms after the confluence with the white river. Nickel and Cu is mostly present in the dissolved and fine colloidal phase in the DC and BR but almost entirely in the coarse colloidal fraction in the white river.

#### **5.4. Pb isotopic composition**

We observe distinct differences between the lead isotope ratios in the white river and those in the black river and drainage canals. A decrease in the  $^{206}\text{Pb}/^{207}\text{Pb}$  isotopic ratio is observed with increasing Pb concentrations in the black river but not the white river (Figure 6a). Furthermore, the biplot of the  $^{206}\text{Pb}/^{207}\text{Pb}$  and the  $^{208}\text{Pb}/^{206}\text{Pb}$  signatures illustrates significant differences between the white water and black river/drainage canal groups (Figure 6b).

### **6. Discussion**

#### **6.1. In-stream processing of DOM in black rivers**

We observe in-stream processing of DOM, but the total DOM exported from tropical peatlands exceeds the processing capacity of the rivers which drain them and a large proportion of DOM is transported to the ocean. We find persistently high DOC concentrations in both drainage canals and black rivers draining degraded peatlands consistent with the range of previously reported values in Borneo (Moore et al., 2013; Cook et al., 2018) and in the upper range of black rivers in Sumatra (Rixen et al., 2008; Baum et al., 2007). We also find indicators of in-stream processing of DOM. The transformation of DOM we observe along the continuum is likely due to dominant photo-oxidation and a lower contribution of microbial processing. We observe an increase in the  $\delta^{13}\text{C}$ -DOC values along the studied continuum (Figure 3a). This shift toward higher  $\delta^{13}\text{C}$ -DOC is correlated with an increase in the C1/C2 ratio of PARAFAC fluorophores (Figure 3c). The two fluorophores are typical of terrestrial input of DOM (Yamashita et al. 2008). An



increase in this C1/C2 ratio reflects a shift toward lower wavelengths and therefore toward lower aromaticity and lower molecular weight (Austnes et al. 2010). Moreover, a decreasing trend of SUVA values is observed along the continuum (Figure 3b). These observations indicate that at our site, aromatic features are preferentially processed in-stream, consistent with a dominant effect of photo-oxidation (Spencer et al., 2009; Amon and Benner, 1996; Sharpless et al., 2014). This has also been observed in the Congo River where photo-oxidation led to an increase in  $\delta^{13}\text{C}$ -DOC and a decrease in aromatic features (Spencer et al. 2009).

However, photo-oxidation is not the only process of DOM processing. The low oxygen levels in the black river and drainage canals and the significant relationship between DOC and DO concentrations suggest that nearly all oxygen entering the well-mixed water is quickly consumed by DOM oxidation (Figure 2a&b). The sharply decreasing oxygen profiles measured in the black river suggest that the transformation of DOM is restricted to the shallow surface layers of these waters (Figure SI.3). However, localized increase in FI index (reflecting a higher proportion of microbial derived DOM, Figure 3d) suggests that microbial processing occurs at a higher extend in drainage canals and the black river. Both photo-oxidation and microbial processing have been quantified in laboratory experiments for DOM originating from tropical peatlands. Martin et al. (2018) found that up to 25 % of riverine DOC from a black river in Sarawak, Malaysia, was lost within 5 days of exposure to natural sunlight. Microbial long-term incubation studies by Rixen et al. (2008), showed that 27% of DOC was degraded after two weeks, and that as a whole, 73 % of DOC was considered refractory to microbial degradation. Another hypotheses is that in-stream microbial processing of DOM is further limited by the low pH, and low nutrient levels (especially inorganic nitrogen), of the Black River (Wickland et al., 2018), rather than intrinsic refractory characteristics. Although the precise extent of in-stream processing cannot be quantified here, our results are consistent with in stream transformation of DOM by important photo-oxidation but also microbial degradation in the shallow surface layers. Additionally, quantitative assessment of outgassing in tropical peatland drainage canals would improve the evaluation of carbon release following peatland drainage. Overall, more work is needed to understand the extent of upstream processing of peatland DOM.

## **6.2. Role of DOM, Al and Fe in trace metal dynamics in peat draining waters**

This study provides the first record of trace metals in black rivers originating from degraded tropical peatlands. We observe strong enrichment of Al and Fe, as well as Pb, As, Ni and Cd in peat-draining waters. The measured concentrations are comparable to those measured by Kurasaki et al. (2000) in Borneo rivers for Pb, Zn, Cu and Cd, but significantly higher (5 to 10 times) for Fe. The concentration levels, however, remain low compared to highly impacted regions of Indonesia (Arifin et al., 2012). The elevated concentrations of Al and Fe in water draining tropical peatlands is consistent with existing observations of elevated Fe concentrations from black rivers in the tropics (Zhang et al., 2019) and northern peatlands. This enrichment is likely due to the weathering of mineral material under the peat during peat accumulation processes (Tipping et al., 2002; Pokrovsky et al., 2005). As a consequence, in water draining peatlands, strong organo-mineral associations between DOM and Fe (Krachler et al. 2010, 2012; Broder and Biester 2015), as well as DOM and Al (Helmer et al., 1990) have been observed. These colloidal associations between DOM and Al and Fe in the form of hydroxides strongly control TM transfer and speciation in peat draining waters (Tipping

et al., 2002). In the present study, specific associations of trace metals with Al and Fe are observed, including strong links between Al and Pb and Ni. However, the lack of a direct relationship between Pb and DOM contrasts with reported observations in the literature (Graham et al. 2006; Pokrovsky et al., 2016; Jeremiason et al., 2018). Despite this, we do observe strong links between Fe, As, Zn, Cd and DOM, which have been previously reported in water draining peatlands ( Broder and Biester, 2015; Neubauer et al., 2013; Pokrovsky et al., 2016). The coupled dynamic of Fe and As might be related to similar mobilization processes within the peat column, with the sorption of As to Fe(III)-(oxyhydr)oxides (ThomasArrigo et al., 2014) in anoxic peat water. Widespread drainage of tropical peatlands and the corresponding release of anoxic water to surface water networks could induce a coupled increase in DOM and Fe concentrations, similar to that which has occurred in Sweden (Kritzberg and Ekström, 2011).

### **6.3. Peatlands as secondary sources of atmospheric pollutants**

The isotopic composition of Pb in peat draining water strongly suggests its anthropogenic origin. The isotopic signatures measured in river samples are a combination of the signature of undisturbed soils of Borneo (Valentine et al., 2008), and a mix of both present and past anthropogenic inputs. Older anthropogenic inputs are reflected by the signature of atmospheric deposition from Java aerosols (Bollhöfer and Rosman, 2000), while the signature of recent regional anthropogenic inputs was characterized by rain samples collected in Pontianak as part of this study (Figure 6b). In the black river and drainage canals, the isotopic ratio is close to the aerosols and recently sampled rainwater and dominated by anthropogenic inputs, whereas the isotopic ratio in the white river is closer to the natural signal (Figure 6). This isotopic difference is consistent with the difference between the watersheds drained by these two rivers: tropical peatlands are ombrotrophic systems, and the trace metal content in peat soil is derived from the atmosphere (Weiss et al. 2002), whereas the Kapus Kecil is recharged from a larger watershed and reflects contribution of mineral soils. Tropical peatlands can serve as secondary sources of atmospheric pollutants to the environment. With peatland drainage, black rivers release the accumulated atmospheric deposition over hundreds of years on much shorter timescales. For example, the isotopic signature observed in the black river reflects anthropogenic sources deposited at different times, including older deposition such as the lead measured in the Java aerosols (Bollhöfer and Rosman, 2000), and more recent deposition following the widespread introduction of unleaded fuel (characterized by samples collected from rainwater during the wet 2014 sampling period in this study). This release of lead by degraded tropical peatlands has the potential to impact records from environmental archives, for example the orals of the Singapore Strait (Chen et al., 2015). Although this is the first measurement of the aquatic release of trace metals from tropical peatlands, the role of tropical peatlands as a secondary source of contaminants has also been highlighted by the trace metal content analysis of dust emitted to the atmosphere by peat fires (Betha et al., 2013).

### **6.4. From degraded tropical peatlands to the ocean**

Sharp changes in physico-chemical conditions are observed after the mixing of the black and the white river, including sharp increases in DO concentration and pH values. This strongly controls the transport of DOM and TM drained from degraded tropical peatlands. After the confluence with the white river, DOC concentrations decrease abruptly. At the same time, the DOC/Cl ratio decreases sharply from the black river and drainage canals to the white river. This suggests

that dilution alone is insufficient to explain the drop in DOC concentrations. Instead, in stream processing of DOM likely plays a role, with the sudden elevation of pH and DO creating favorable conditions for microbial processing of DOC, making the mixing zone a likely hotspot of GHG emissions (Palmer et al., 2016). This would also be consistent with the decrease in the SUVA index observed after the confluence. Despite processing of DOM in the mixing zone, a significant proportion of DOM originating from degraded peatlands actually reaches ocean. We observe high DOC concentrations at all sampling locations, with concentrations remaining high even close to the ocean (Figure 2a). Additionally, the results of our physical fractionation show that even close to the estuary, DOC remains in the dissolved and fine colloid form ( $<0.22\ \mu\text{m}$ ), and that flocculation processes might be limited. The decrease in trace metal concentrations after the confluence might be influenced by shifts in physical fractionation and increase proportion of colloidal form. This is especially true for Al and Pb. Some flocculation at the estuary might limit their transfer to the ocean. A higher proportion of Fe and As remains in the form of fine colloids after mixing with the whiter river, and is still associated with DOC. Similar conservative behavior of LMW organic molecules associated with Fe was observed at the outlet of northern peatlands (Krachler et al., 2012), and in Arctic rivers (Pokrovsky et al., 2014). This highlights that dissolved organic molecules derived from tropical peatlands can also act as carriers of trace metals to the ocean.

## 7. Conclusions

This study characterizes the composition and concentration of DOM and TM in the canals and rivers draining the degraded tropical peatlands of Indonesian Borneo. It highlights in-stream processing of DOM in drainage canals and rivers draining degraded peatlands. Both stable isotopic and optical properties of DOM are consistent with photo-oxidation along the continuum from the black river to the ocean. In the black river and drainage canals, rates of microbial processing are likely limited by the low dissolved oxygen concentrations, and concentrated at shallow depths. In contrast, after mixing with the white river, higher rates of microbial processing of DOM are expected. Along the continuum, DOM is found at relatively high concentrations in the dissolved and fine colloidal phases, suggesting a substantial fraction of DOM derived from degraded peatlands reaches the ocean. Additionally, we provide the first assessment of trace metal concentrations in rivers draining degraded tropical peatlands. Rivers draining these peatlands are enriched in some trace metals (Pb, Ni, Zn, Cd) as well as Al and Fe. Using the isotopic signature of Pb, we show that degraded tropical peatlands are secondary sources of atmospherically deposited contaminants to surface waters. Trace metal dynamics after transfer to the white river show clear trends: while Pb and Ni are associated with Al, As, Zn and Cd are associated with Fe and DOM. Lead and Al are present in coarse colloidal form and may be transferred to sediments after flocculation. In contrast, DOM, Fe and As are found predominantly in fine colloidal form even after the confluence with the white river, and as a result may be transferred to the ocean. The role of degraded tropical peatlands as a source of DOM, as well as Fe and As to the ocean requires further investigation.

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**Table 1.** Water chemistry of the white river, black river and drainage canals for the two sampling campaigns. The first line is the mean and the second line is the standard deviation. DO: Dissolved oxygen, FI: Fluorescence Index, SUVA: Specific UV Absorbance.

		pH	DO	Cond	DOC	$\delta^{13}\text{DOC}$	FI	SUVA	C1.C2	Cl	Al	Fe	Pb	As	Zn	Cu	Ni	Cd
	Unit	/	mg L <sup>-1</sup>	$\mu\text{S cm}^{-1}$	mg L <sup>-1</sup>	‰	-	L mg <sup>-1</sup> m <sup>-1</sup>	-	mg L <sup>-1</sup>	$\mu\text{g L}^{-1}$	$\mu\text{g L}^{-1}$	$\mu\text{g L}^{-1}$	$\mu\text{g L}^{-1}$	$\mu\text{g L}^{-1}$	$\mu\text{g L}^{-1}$	$\mu\text{g L}^{-1}$	$\mu\text{g L}^{-1}$
White River	dry	5.20	4.49	37.2	8.43	-29.46	1,51	4,28	2.26	4.83	312.160	444.893	0.262	0,28	18.572	1.140	0.528	0,005
		0.33	0.26	13.2	1.61	0.23	0,03	0,00	0.08	4.03	407.057	383.954	0.236	0,09	9.028	0.158	0.106	0,002
	wet	4.43	3.37	1220.6	11.25	-29.41	1,60	4,57	2.30	409.16	147.242	547.500	0.129	0,32	10.264	0.868	1.186	0,006
		0.86	0.96	981.8	4.29	0.41	0,04	0,33	0.18	342.02	124.616	497.278	0.102	0,09	6.136	0.089	1.169	0,003
White River Upstream	dry	5.45	4.91	24.0	6.89	-29.32	1,55	n.a	2.54	1.31	101.610	242.500	0.139	0,21	15.290	1.059	0.519	0,003
		5.71	0.17	3.8	1.28	0.06	0,08		0.22	0.64	27.490	44.500	0.027	0,02	3.060	0.007	0.077	0,000
	wet	5.37	4.15	268.7	8.69	-29.56	1,56	4,15	2.30	47.87	148.617	408.733	0.236	0,30	9.702	1.186	0.724	0,003
		0.06	0.39	92.0	0.78	0.13	0,01	0,15	0.14	65.16	71.999	170.153	0.167	0,04	6.233	0.244	0.441	0,000
Black River	dry	3.45	1.69	98.7	36.42	-30.29	1,50	5,97	1.77	2.78	592.804	2143.500	0.467	0,59	119.386	0.582	1.958	0,012
		0.06	0.39	18.2	2.54	0.38	0,06	0,46	0.07	0.50	43.008	187.609	0.054	0,04	86.473	0.194	1.672	0,006
	wet	2.97	1.98	77.3	35.37	-30.04	1,59	4,94	2.12	2.19	443.136	1440.991	0.316	0,40	10.948	0.720	1.294	0,007
		0.13	0.75	18.3	3.70	0.38	0,15	1,32	0.15	0.57	137.488	493.547	0.110	0,10	6.980	1.127	0.527	0,003
Drainage canal	wet	3.08	2.34	89.8	35.17	-30.27	1,60	4,33	2.05	2.57	489.183	1348.017	0.313	0,35	14.518	0.369	1.535	0,008
		0.43	0.30	19.6	5.47	0.40	0,17	1,27	0.18	0.64	194.889	494.078	0.048	0,08	11.620	0.070	0.801	0,006

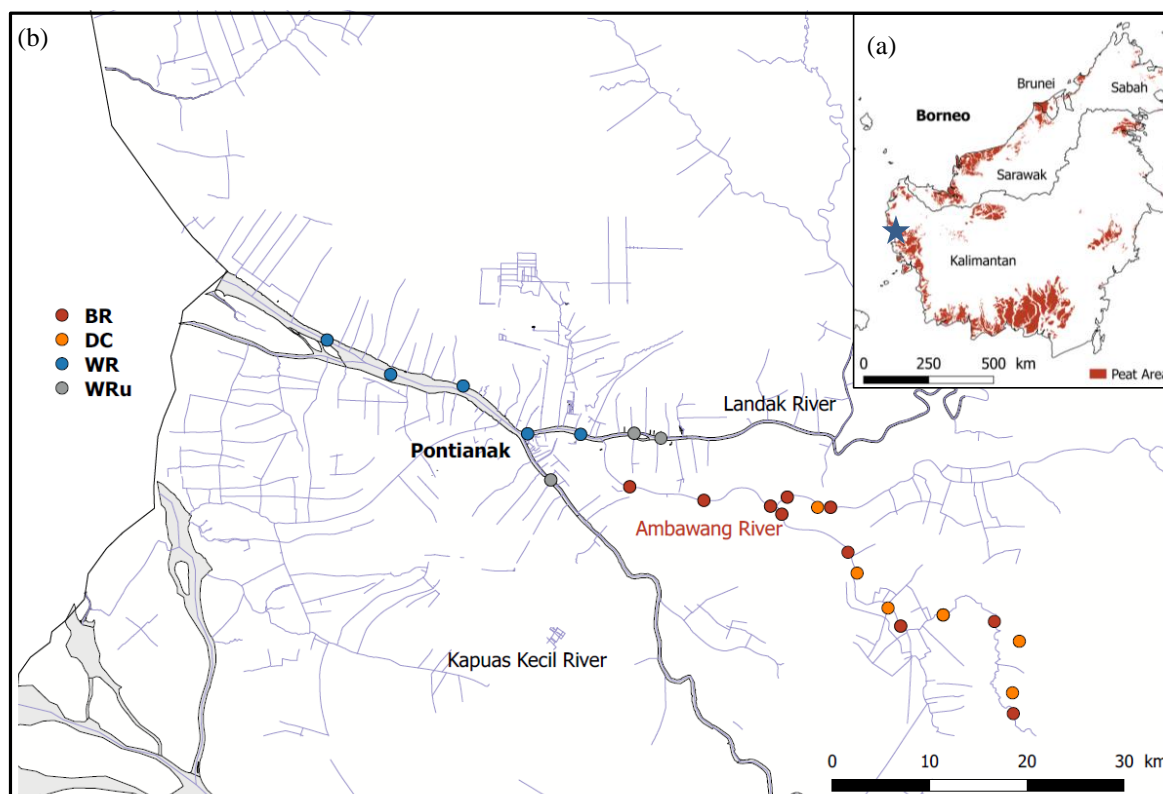
543

544 **Table 2.** Proportion of DOC and selected trace metals in the form of dissolved and fine colloids (< 0.22 µm) and coarse colloids (0.2-2.7 µm)

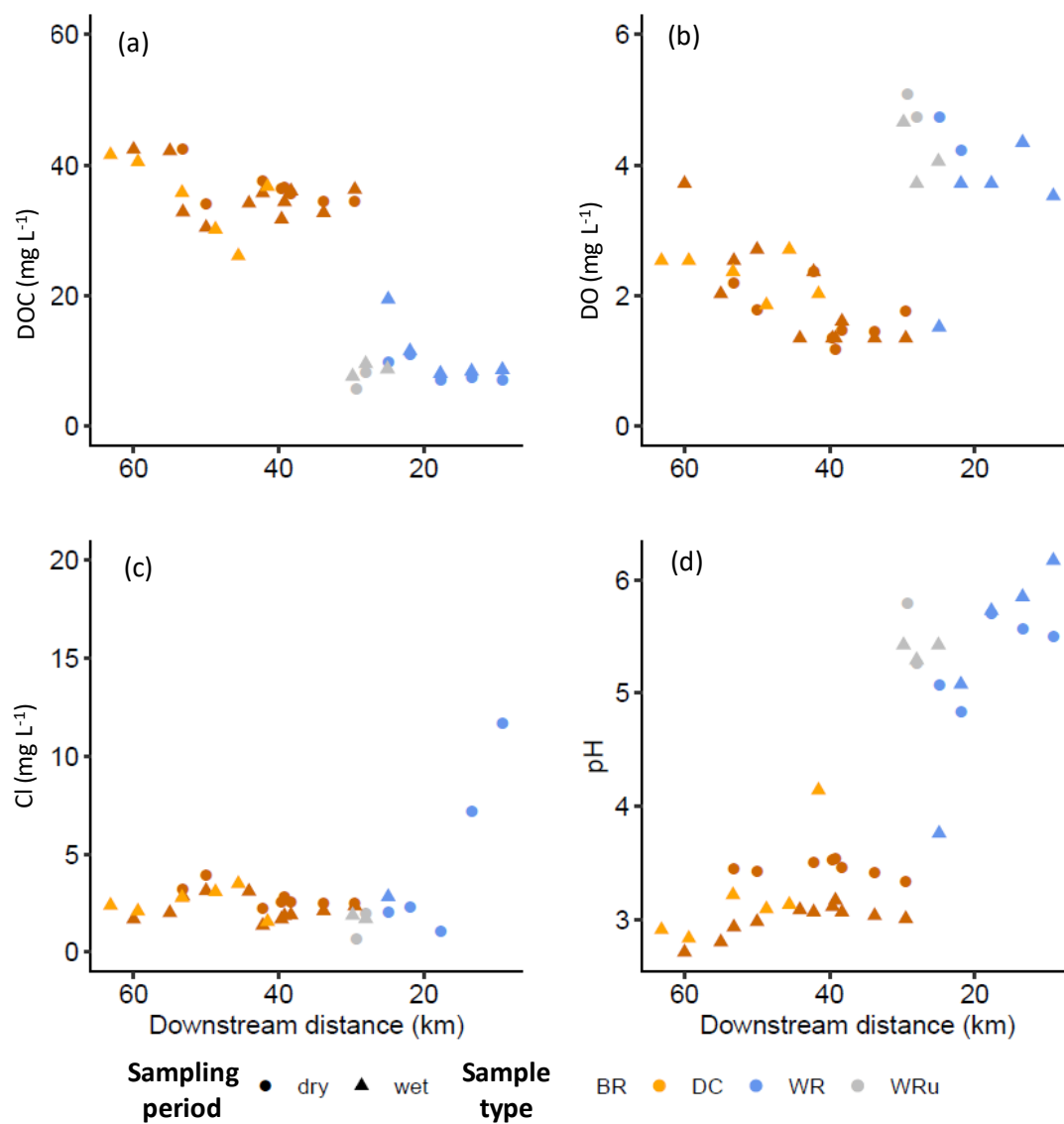
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	Drainage Canals		Black River		White River	
	<0.2 µm	0.2-2.7 µm	<0.2 µm	0.2-2.7 µm	<0.2 µm	0.2-2.7 µm
DOC	97	3	98	2	100	0
Al	39	61	36	64	18	82
Fe	100	0	99	1	45	55
Pb	75	25	78	22	34	66
As	98	2	96	4	67	33
Ni	72	28	50	50	1	99
Cu	68	32	48	52	1	99
Zn	13	87	12	88	26	74
Cd	66	34	100	0	83	17

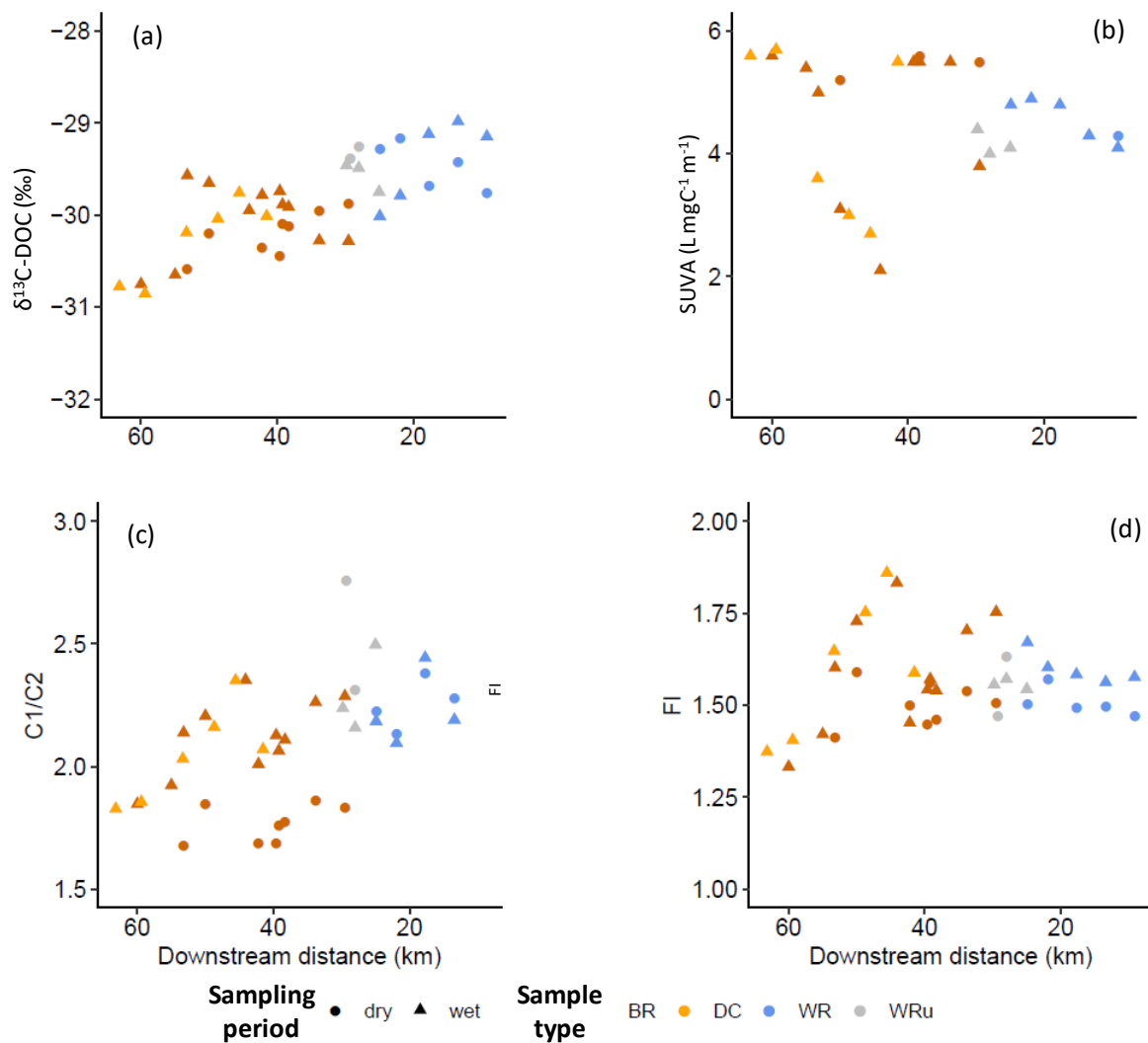
546



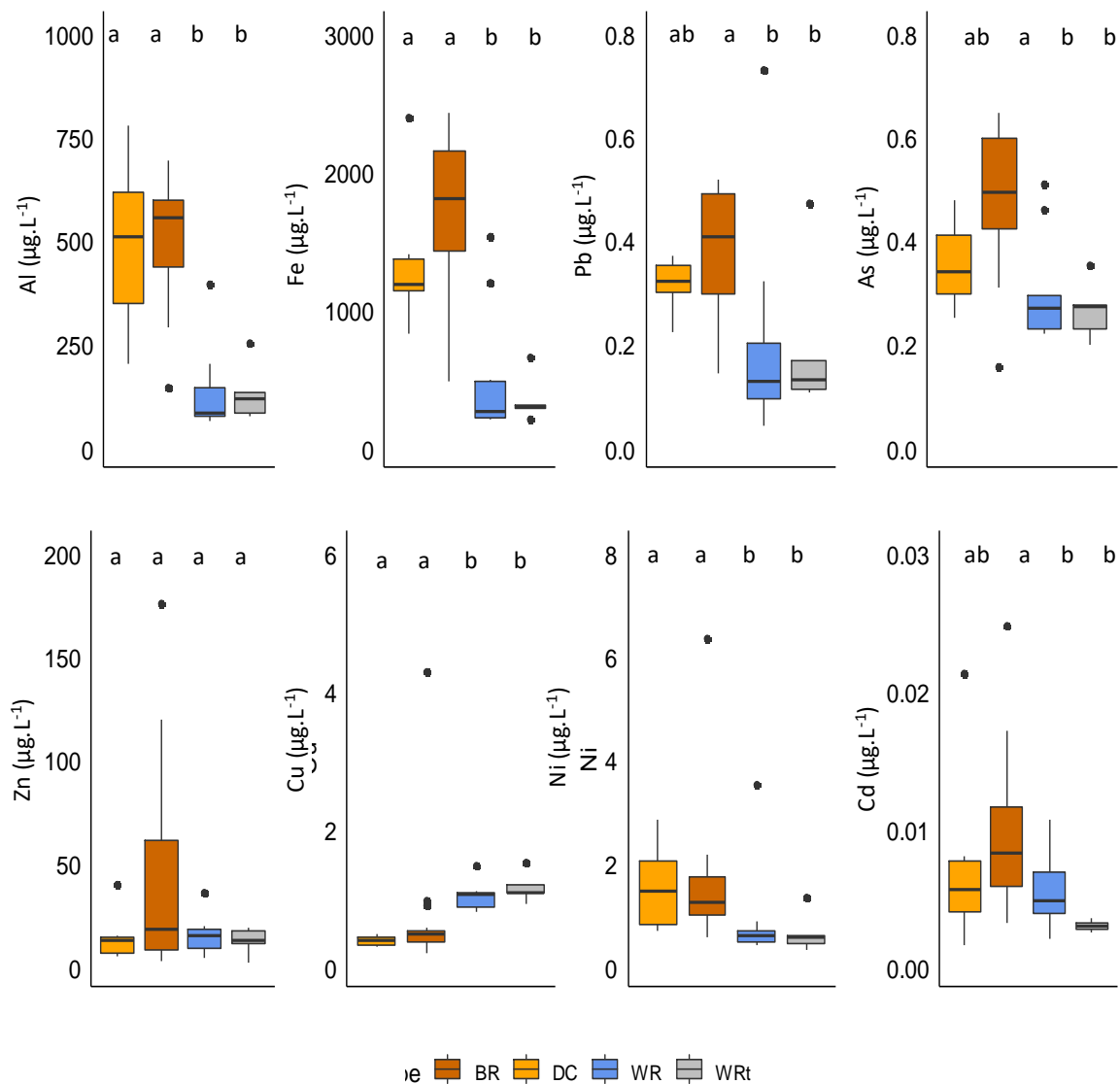
**Figure 1:** (a) Location of the study area on Borneo island. (b) Location of sampling sites and types of water: Black River (BR), Drainage Canals (DC), White River (WR), and white River upstream of the confluence with the black river (WRu).



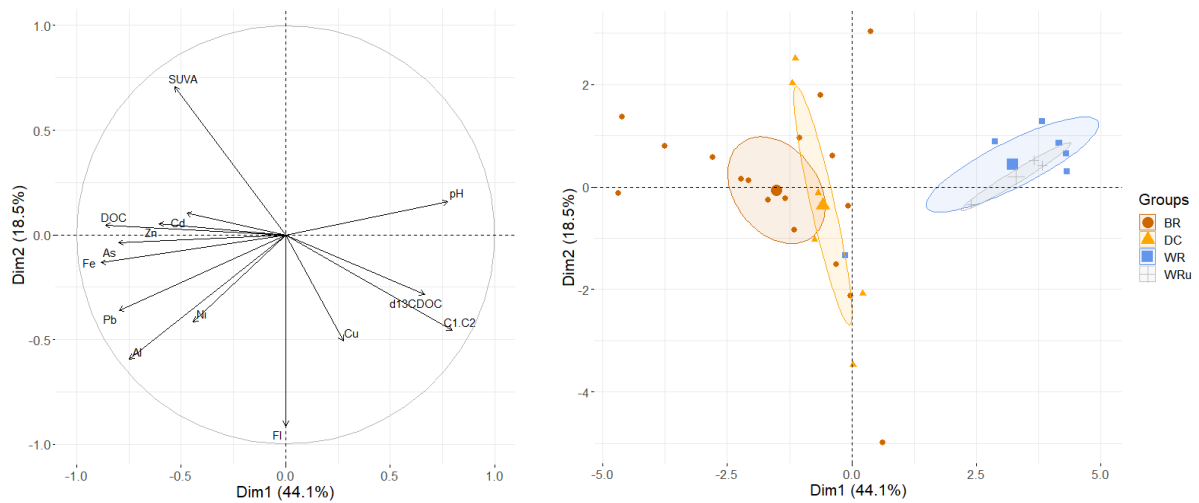
**Figure 2:** Evolution of (a) dissolved organic carbon concentration, (b) dissolved oxygen concentration, (c) chloride concentration and (d) pH along the continuum from the black river to the ocean.



**Figure 3:** Evolution of DOM along the black river to the ocean continuum. (a)  $\delta^{13}\text{C-DOC}$ , (b) SUVA (Specific UV Absorbance) index, (c) C1/C2, (d) FI (Fluorescence Index).

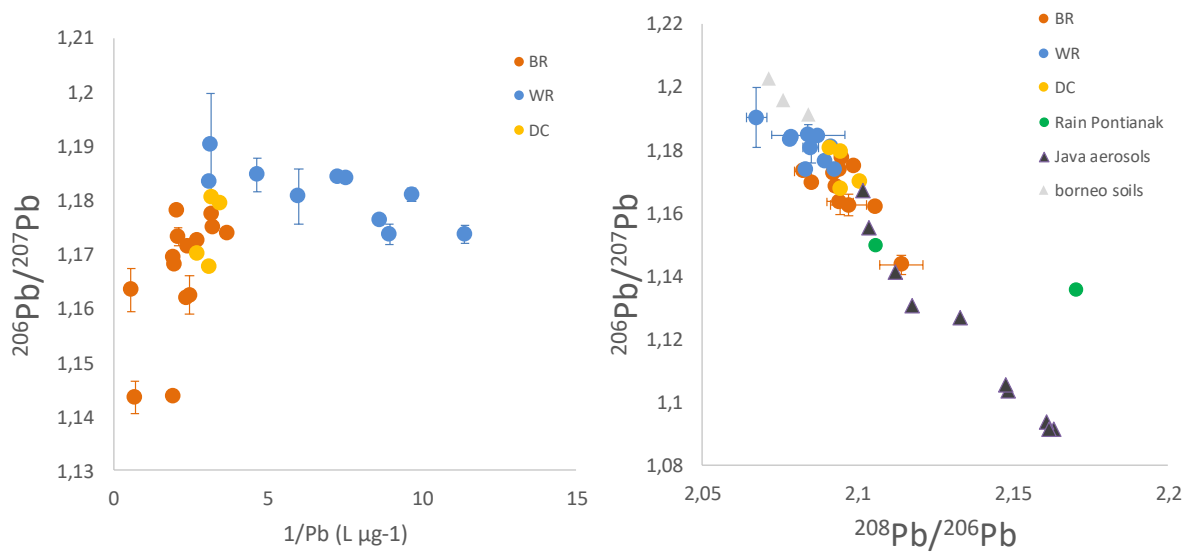


**Figure 4:** Ranges of selected TM concentration for different sampled water types. Letters represent significantly different groups (Kruskal Wallis and Dunn's post hoc multiple test ( $p < 0.05$ ))



**Figure 5:** The first two factors of the PCA (63,1 % of variance) by variables (a) and by observation (b) for the different sampled water types.





**Figure 6:** (a) Dependence of  $^{206}\text{Pb}/^{207}\text{Pb}$  ratio on Pb concentrations for the different water samples. (b) Relationship between  $^{206}\text{Pb}/^{207}\text{Pb}$  ratio and  $^{208}\text{Pb}/^{206}\text{Pb}$  ratio.