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A. Bourles, H. Amir, S. Gensous, F. Cussonneau, V. Medevielle, et al.. Investigating some mechanisms underlying stress metal adaptations of two Burkholderia sensu lato species isolated from New Caledonian ultramafic soils. European Journal of Soil Biology, 2020, 97, pp.103166. 10.1016/j.ejsobi.2020.103166 hal-02509749

# HAL Id: hal-02509749 https://hal.science/hal-02509749

Submitted on 21 Jul2022

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Version of Record: https://www.sciencedirect.com/science/article/pii/S1164556319304698 Manuscript\_758fad7cef5da2f36ef881afa3bbb9f4

#### 1 Investigating some mechanisms underlying stress metal adaptations of two *Burkholderia* sensu lato

#### 2 species isolated from New Caledonian ultramafic soils

- 3 Alexandre BOURLES<sup>a</sup>, Hamid AMIR<sup>a</sup>, Simon GENSOUS<sup>a</sup>, Flavie CUSSONNEAU<sup>a</sup>, Valérie MEDEVIELLE<sup>a</sup>,
- 4 Farid JUILLOT<sup>c</sup>, Alexis BAZIRE<sup>d</sup>, Michaël MEYER<sup>a</sup>, Valérie BURTET-SARRAMEGNA<sup>a</sup>, Yvon CAVALOC<sup>a</sup>,
- 5 Philippe JOURAND<sup>e</sup>, Linda GUENTAS<sup>a,b,\*</sup>
- <sup>a</sup>Institut des Sciences Exactes et Appliquées, Université de la Nouvelle-Calédonie, BR R4 98851 Nouméa, NouvelleCalédonie
- 8 <sup>b</sup>Université de Toulon, Laboratoire MAPIEM EA 4323, SeaTech-Ecole d'ingénieurs BP 20132 83957 La Garde
- 9 Cedex, France
- 10 °UMR IRD 206 Institut de Recherche pour le Développement, Centre IRD Nouméa, 101 Promenade Roger
- 11 Laroque, BP A5 98848 Nouméa Cedex Nouvelle-Calédonie
- 12 d'Laboratoire de Biotechnologie et Chimie Marines, EA 3884, LBCM, IUEM Université de Bretagne-Sud, 56100
- 13 Lorient, France
- 14 eIRD, Laboratoire des Symbioses Tropicales et Méditerranéennes (LSTM UMR040), Campus International de
- 15 Baillarguet, TA A-82/J, Montpellier F-34398 Cedex 05, France
- 16 \*Corresponding author: linda.guentas@univ-nc.nc (L. Guentas), Université de la Nouvelle-Calédonie, BR R4
- 17 98851 Nouméa, Nouvelle-Calédonie. Phone Number: +687290262
- 18

#### 19 Abstract

20 In New Caledonia, one third of the main island is covered by ultramafic soils characterized by high levels of metals 21 and low levels of plant nutrients. In these soils, metal tolerant bacteria may play a role in plant adaptation to the 22 edaphic constraints. Recently, two new bacteria species belonging to the genus Burkholderia sensu lato were isolated 23 from rhizospheric roots of pioneer plant growing in New Caledonian ultramafic soils and were able to tolerate high 24 metal concentrations. The purpose of this study was to determine the major mechanisms by which these two bacteria 25 were able to tolerate metals and if they were able to reduce metal absorption by plants from which they were 26 isolated. Bioaccumulation, adsorption on cell surface, ability to form biofilm and production of exopolysaccharides 27 were assessed. A focus was carried out on Ni, generally the most toxic metal in New Caledonian ultramafic soils. All 28 experiments were performed in comparison with a reference strain isolated from a non-ultramafic habitat, 29 Paraburkholderia caribensis MWAP64<sup>T</sup>. The two ultramafic bacterial species showed a high tolerance to Ni with a Ni-IC<sub>50</sub> of 5 mM for Caballeronia novacaledonica STM10272<sup>T</sup> and 30 mM for Paraburkholderia ultramafica 30 31 STM10279<sup>T</sup>, values significantly higher than those of *P. caribensis* MWAP64<sup>T</sup> (2 mM). The major mechanism 32 involved in the bacterial Ni tolerance was related to biofilm formation for C. novacaledonica STM10272<sup>T</sup> and to 33 acidic exopolysaccharide production for P. ultramafica STM10279<sup>T</sup>. Both species alleviated metal contents in roots 34 of Tetraria comosa, an endemic pioneer herbaceous species growing on New Caledonian ultramafic soils.

35 Key words: Metal tolerance, Biofilm; Exopolysaccharides; *Paraburkholderia*; *Caballeronia*; Plant metal alleviation

36

#### 38 1 Introduction

39 Ultramafic soils (also known as serpentinic soils) are resulting from weathering and pedogenesis of ultramafic 40 bedrocks [1]. These soils are characterized by (i) the dominance of iron oxides with high levels of potentially toxic 41 metals including Ni, Co, Cr and Mn, (ii) low levels of major plant nutrients such as N, P, and K, and (iii) a strong 42 unbalanced calcium/magnesium (Ca/Mg) ratio (<1) [1]. This a priori disadvantageous substrate constitutes a specific 43 habitat, which frequently hosts unique plants communities and "clusters" of biodiversity consisting of restricted and 44 endemic taxa [2]. In these extreme edaphic conditions, plants have developed several adaptive mechanisms such as 45 (i) slower growth (ii) ability to limit Mg absorption, and (iii) strategies to tolerate high metal concentrations [2]. 46 Among these strategies, the association with specific bacteria play a key role in plant adaptation to harsh edaphic 47 conditions [3,4].

The study of microbial communities from ultramafic soils led to the identification of new metal tolerant strains and species [5,6]. Metal tolerant-bacteria show a wide panel of responses towards metals such as biosorption, bioaccumulation, precipitations, chelation, biofilm production, exopolysaccharides (EPS) production or metal transformation, thereby reducing metal toxicity towards themselves [7]. Moreover, metal tolerant bacteria have been shown to reduce the bioavailability of metals and thus to alleviate their toxicity to plants exposed to metal stress [8].

53 In New Caledonia, a tropical archipelago in the South Pacific Ocean, one third of the main island is covered by 54 ultramafic soils. As a result, about 85% of the vascular flora growing on these soils are endemic [9]. However, 55 ultramafic settings are exploited for economically valuable ores such as Ni, Co and Cr, disturbing ultramafic 56 landscapes and classifying New Caledonia among the biodiversity hotspots [10]. New Caledonian ultramafic soils 57 are characterized by high richness and phylogenetic bacterial diversity compared with other environment 58 [11], however the interest for bacteria of these soils is recent [12]. In 2016, two new Burkholderia sensu lato (s.l.) 59 species (Burkholderia novacaledonica and Burkholderia ultramafica), isolated by Gonin et al. (2013) [13] from roots 60 of pioneers plants belonging to the Cyperaceae family growing on ultramafic soils in New Caledonia, have been 61 characterized [6]. Recently, Dobritsa and Samadpour (2019) [14] have proposed the transfer of B. novacaledonica STM10272<sup>T</sup> and B. ultramafica STM10279<sup>T</sup> to the genus Caballeronia and Paraburkholderia respectively. 62

These two species showed tolerance to high levels of the metals found in New Caledonian ultramafic soils [13] suggesting the existence of mechanisms involved in bacterial adaptation to the ultramafic constraint. In this study, we investigated the mechanisms occurred in *Paraburholderia ultramafica* STM10279<sup>T</sup> and *Caballeronia novacaledonica* STM10272<sup>T</sup> in comparison with *Paraburholderia caribensis* MWAP64<sup>T</sup> isolated from a nonultramafic soil [15]. We also investigated the implication of *P. ultramafica* STM10279<sup>T</sup> and *C. novacaledonica* STM10272<sup>T</sup> on the adaptation to ultramafic constraint of *Tetraria comosa*, an endemic pioneer Cyperaceae used in restoration programs [16].

#### 70 2 Material and Methods

#### 71 2.1 Bacterial species

For ultramafic species, plant sampling and bacterial strains isolation were performed by Gonin et al. (2013) [13]. Full description of *C. novacaledonica* STM10272<sup>T</sup> and *P. ultramafica* STM10279<sup>T</sup> including morphological, biochemical and molecular characterization was carried out by Guentas et al. (2016) [6] and are reported in Supplementary Table 1. According to Gonin et al. (2013) [13], the two species isolated from ultramafic soils, tolerate edaphic constraints such as low pH, unbalanced Ca/Mg ratio and high metals concentrations (Supplementary Table 2). Concerning *P. caribensis* MWAP64<sup>T</sup>, bacterial strain was isolated from Vertisol microaggregates in Martinique by Achouak et al. (1999) [15].

#### 79 2.2 Test for metal tolerance (50% inhibition concentration; IC<sub>50</sub>)

80 The metal tolerance was performed in aerobic conditions on microplates (sterile transparent 96 wells PS flat bottom, Greiner) in Lysogenic Broth Lennox medium (LB, tryptone: 10 g L<sup>-1</sup>, yeast extract 5 g L<sup>-1</sup>, NaCl: 5g L<sup>-1</sup>; 81 82 CONDA, Spain) supplemented with different metals concentrations (3 replicates by plate) as suggested by Chien et 83 al. (2013) [17]. For these experiments, 1 M metal stock solutions were prepared from solid salts (Ni as NiSO<sub>4</sub>·6H<sub>2</sub>O, 84 Co as CoCl<sub>2</sub>·6H<sub>2</sub>O, Cr as K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub>, and Mn as MnSO<sub>4</sub>·H<sub>2</sub>O). To avoid metal precipitation in the assays, like 85 suggested by Hartley et al. (1997) [18], these stock solutions were sterilized separately. Each well was filled with 20  $\mu$ L of an overnight bacterial culture adjusted to an OD<sub>600</sub> of 1, (which is approximately equal to 10<sup>8</sup> CFU mL<sup>-1</sup>) and 86 87 180 µL of broth tested. The plate was incubated at 28°C under 125 r min<sup>-1</sup> orbital shaking in a microplate incubator reader (EPOCH 2 microplate reader; BioTek Instruments, Winooski, Vermont, USA). Cell growth was monitored by measuring the OD<sub>600</sub> every hour for 48 h. The experimentations were done at least twice independently. According to the preliminary result of Gonin et al. (2013) [13], metals concentration tested ranged from 0.5 to 100 mM for Ni and Mn and from 0.01 to 50 mM for Co and Cr. The strain growth rates ( $\mu$ ) were determined for each concentration. A sigmoid dose–response curve was obtained by plotting growth rate against metal concentrations, and IC<sub>50</sub> values were determined using GraphPad Prism software (version 6, GraphPad Software, La Jolla California USA).

#### 94 2.3 Ni biosorption and bioaccumulation

95 The metal adsorbed onto the surface of the bacterial cell and the metal accumulated inside the cell were 96 determined according to Giovanella et al. (2017) [19], with modifications as follows. A primary culture was done by 97 inoculating 10 mL of LB Lennox medium with one colony of the chosen bacteria and then incubating for 24 h at 98 28°C and 125 r min<sup>-1</sup>. A culture (25 mL) inoculated at an OD<sub>600</sub> of 0.1 by a primary culture was incubated for 24 h at 99 28°C under rotary shaking at 125 r min<sup>-1</sup>; then the culture was artificially contaminated by aseptically adding Ni 100 such that Ni reached a final concentration of 1 mM, 5 mM and 50 mM (based on Ni-IC<sub>50</sub> values). After 24 h, 1 mL 101 samples were removed from the culture and centrifuged at 13,000 r min<sup>-1</sup> for 10 min (Sigma, 2-16K). The pellet was 102 washed thrice in sterile deionized water to remove free metal ions and finally suspended in 1 mL of sterile deionized 103 water. The OD<sub>600</sub> was read and the number of colony-forming units (CFU) per milliliter of the inoculum was carried 104 out. To determine Ni accumulation in the cell walls and intracellular spaces of these bacteria, another aliquot (1 mL) 105 was centrifuged at 13,000 r min<sup>-1</sup> for 10 min; the pellet was washed with ethylenediaminetetraacetic acid (EDTA, 10 106 mM, pH: 6.5) for Ni desorption of from the cell surface, centrifuged, and re-suspended in 1 mL of sterile deionized 107 water. Finally, the total amount of Ni accumulated intracellularly (bioaccumulation) and sorbed onto the cells 108 (biosorption) was determined by digesting the 1 mL aliquots in a nitric acid (69%, 1 mL) and hydrogen peroxide 109 (30%, 0.5 mL) solution for 1 night at room temperature. The digested samples were then adjusted to 10 mL with 110 deionized water, and the Ni content in the final solution was quantified by Inductively Coupled Plasma Optical 111 Emission Spectrometry (ICP OES, Varian, Varian 730-ES, Palo Alto, USA) at Laboratoire des Moyens Analytiques 112 (LAMA-US IMAGO-IRD, New Caledonia). The amount of Ni accumulated intracellularly (bioaccumulation) was 113 determined by substracting the Ni content adsorbed onto the cell walls (biosorption) from the total Ni amount.

#### 115 2.4 Biofilm formation assay in the presence of Ni

116 The biofilm formation assay was performed on a microtiter plate [20]. Aliquots of 90  $\mu$ L of LB culture medium 117 supplemented with different concentrations of Ni [(as NiSO<sub>4</sub>  $\square$  6H<sub>2</sub>O]) to reach 0.5, 1, 2.5, 5, 8, and 10 mM Ni 118 concentration (based on Ni- IC<sub>50</sub> values) were loaded into 96-well microtiter plates (sterile transparent 96 wells PS 119 flat bottom, Greiner). Wells were inoculated with 10 µL of a 48 h primary culture adjusted to an OD<sub>600</sub> of 1. For 120 these experiments, a non-inoculated LB medium was used as the negative control, whereas an inoculated LB culture 121 medium without Ni was used as the positive control. After, 24 h, 48 h, and 72 h of incubation at 28°C without 122 agitation, the medium and the non-adhering bacteria biofilm were eliminated by simple inversion and each well was 123 washed three times with 100  $\mu$ L of sterile deionized water. The microplates were then dried at 50°C for 30 min. The 124 cells from the adhering biofilm were dyed with 100 µL of 0.02% crystal violet (CV) solution (30 min at room 125 temperature). The excess of dye was removed by simple inversion and the biofilm was washed three times with 100 126 µL of sterile deionized water at each wash. The microplates were dried at room temperature for 10 min before 127 resuspending CV by adding 200 µL of 95% ethanol solution. After 10 min orbital shaking at 125 r min<sup>-1</sup>, the content 128 of each well was transferred into a new microplate and the OD<sub>590</sub> was read with a microplate reader (EPOCH 2 129 microplate reader; BioTek Instruments, Winooski, Vermont, USA). The results obtained were then transformed into 130 a quantification of biofilm formation by calculating the average absorbance of six replicates for each species and for 131 each concentration and dividing by the average absorbance of the blank (without bacteria). The results expressed as 132 this ratio can be considered as non-dependent from the non-specific coloring of the surface of the wells of the 133 microplates by the CV. Following this procedure, biofilm formation was considered to have occurred when the ratio 134 is greater than 2 [21].

#### 135 2.5 Quantification of biofilm formation

Since the concentration of bioavailable Ni frequently reaches 1 mM in ultramafic soils, especially in the rhizosphere from New Caledonia [22], quantification of biofilm production without Ni and at 1 mM of Ni was carried out. The biofilm formation was visualized by confocal laser scanning microscopy (CLSM) after 24 h. For these experiments, a 24 h primary culture was first adjusted at a 0.1 OD<sub>600</sub> in LB medium and used to inoculate a 24wells in glass bottom opaque black screening plate (SensoPlate, Greiner Bio-One) (1 mL per well). The microtiter
dish was then incubated for 24 h and the biofilms were stained for 10 min with 1% of Syto 61 red fluorescent nucleic
acid strain (5 mM, Molecular Probes, USA). CLSM observations were performed with a TCS-SP2 microscope
(Leica Microsystems, Heidelberg, Germany), using a 63x immersion objective. The excitation and Emission
wavelengths for Syto 61 Red were set at 633 nm and 645 nm, respectively. The images collected by confocal stacks
were processed using Leica Confocal Software and the biovolume, thickness and substratum coverage of biofilms
were quantified with COMSTAT [23].

#### 147 2.6 Production and gross chemical composition of EPS

148 Regarding its ability to produce an exopolysaccharide [15], P. caribensis MWAP64<sup>T</sup> was used as a positive 149 control in order to approve the experimental design. EPS production was quantified after 96 h of culture at 28°C under 125 r min<sup>-1</sup> orbital shaking in minimum broth medium (MB; KH<sub>2</sub>PO<sub>4</sub> · 3H<sub>2</sub>O, 0.66 g L<sup>-1</sup>; NaCl, 50 mg L<sup>-1</sup>; 150 MgSO<sub>4</sub> · 7 H<sub>2</sub>O, 10 mg L<sup>-1</sup>; FeCl<sub>3</sub> 4 mg L<sup>-1</sup>; CaCl<sub>2</sub> 40 mg L<sup>-1</sup>) supplemented with glucose (30 g L<sup>-1</sup>). In order to 151 152 produce biomass, 8 L of LB were inoculated using a 48 h grown culture (800 mL; 28°C, 125 r min<sup>-1</sup> orbital shaking). After 72 h of incubation (28°C, 125 r min<sup>-1</sup> orbital shaking), the biomass was recovered by centrifugation at 9,000 r 153 min<sup>-1</sup> (13,700 g) for 25 min (Sorvall RC5C Centrifuge, DuPont) and washed once with sterile deionized water. Cells 154 155 were transferred at 28°C to 10 L reactors containing 8 L of MB medium supplemented with glucose. The bacteria were removed from the medium by centrifugation at 9,000 r min<sup>-1</sup> (18,894 g) for 25 min. The supernatants were 156 157 collected and the EPS were purified by ultrafiltration on a 100,000 molecular weight cut-off membrane from 158 Sartorius and immediately freeze-dried. Protein content was determined by the Lowry method [24], modified by 159 Smith et al. (1985) [25] and Brown et al. (1989) [26]. Sugar contents, was quantified by the method described by 160 DuBois et al. (1956) [27] and completed by Chaplin (1986) [28], and uronic acid contents by the Blumenkrantz and 161 Asboe-Hansen method [29], modified by Filisetti-Cozzi and Carpita (1991) [30]. Infrared spectroscopy was used to 162 complete the biochemical characterization of EPS. The infrared spectrum of the freeze-dried EPS was recorded using 163 a Fourier Transform Infrared (FT-IR) spectrometer Frontier TM (Perkin Elmer) at the spectroscopy lab of the IRD 164 center in Noumea (New Caledonia). This spectrum corresponds to an average of four successive spectra that were

165 collected in Attenuated Total Reflectance mode (diamond crystal) over the 500 cm<sup>-1</sup> – 3500 cm<sup>-1</sup> range with a 4 cm<sup>-1</sup> 166 <sup>1</sup> resolution.

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#### 169 2.7 Effect of microbial inoculation on Tetraria comosa plants under ultramafic conditions

The inocula for pot trial were prepared as follows: A pre-inoculum was performed as described previously in the section 2.3. After 2 days, the aerobic cultures were used to inoculate 100 mL of LB media at an  $OD_{600 \text{ nm}}$  of 0.1. After 3 days at 28°C under orbital agitation (125 r min<sup>-1</sup>), the cells were pelleted by centrifugation (SIGMA®, 2K15, 5,000 g, 20 min), washed twice with sterile deionized water and, then re-suspended in sterile deionized water at a bacterial concentration between 10<sup>7</sup> and 10<sup>8</sup> CFU mL<sup>-1</sup> (corresponding to an  $OD_{600}$  of 0.5), as suggested by Larcher et al. (2003) [31]. The purity was checked and the concentration of the inoculum was carried out after a bacterial count on LB agar plate.

177 SIRAS Pacifique (Noumea, New Caledonia) provided seeds of Tetraria comosa. Germination occurred on a 178 compost amended ultramafic soil composed of (i) 2 mm sieved colluvial lateritic soil (ferralsol) sampled in Plum 179 area in New Caledonia (22°16'59''S, 166°39'12''E) and (ii) commercial compost (4:1 v:v). Characteristics of the ferralsol are: coarse sand, 39.4%; fine sand, 22.1%; silt-clay, 37.2%; pH H<sub>2</sub>0, 5.9; pH KCl, 5.6; total C, 42.1 g kg<sup>-1</sup>; 180 total N, 2.2 g kg<sup>-1</sup>; total P, 147 mg kg<sup>-1</sup>; available P (Mehlich), 3 mg kg<sup>-1</sup>; total Ca, 1.06 g kg<sup>-1</sup>; total Mg, 5.08 g kg<sup>-1</sup> 181 182 and Ca/Mg, 0.207. The total metals contents of the ferralsol measured after an alkaline fusion are as follows: Co, 0.87 g kg<sup>-1</sup>; Cr, 22 g kg<sup>-1</sup>; Fe, 348 g kg<sup>-1</sup>; Mn, 10 g kg<sup>-1</sup>; Ni, 5.6 g kg<sup>-1</sup>. The available metals extracted by 183 dimethylene triaminopentaacetic acid (DTPA) are as follows: Co<sub>DTPA</sub>, 68 mg kg<sup>-1</sup>; Cr<sub>DTPA</sub>, 0.17 mg kg<sup>-1</sup>; Fe<sub>DTPA</sub>, 71 184 mg kg<sup>-1</sup>; Mn<sub>DTPA</sub>, 1.1 g kg<sup>-1</sup>; Ni<sub>DTPA</sub>, 130 mg kg<sup>-1</sup>. The composition of commercial compost is as follows: N, 1.7 185 mg g<sup>-1</sup>; P<sub>total</sub>, 150 mg kg<sup>-1</sup>; P<sub>Olsen</sub>, 7 mg kg<sup>-1</sup>; K,139 mg kg<sup>-1</sup> (Terreau universel, Agrofino, France). The compost 186 187 amended-ultramafic soil was autoclaved 3 times at 120°C for 1 h, with an interval of 24 h before being used in the 188 greenhouse experiment. Although, T. comosa was able to develop on ultramafic soil in natural condition, the addition 189 of this commercial compost to the ultramafic soil was necessary because the growth of the plantlets of T. comosa 190 could be inhibited under greenhouse condition in a pure sterile lateritic topsoil [32]. After two months, roots 191 seedlings were washed with sterile water to remove any contaminants and were transplanted into one-liter plastic containers filled with the compost amended-ultramafic soil described above. A 5 mL bacterial inoculum was applied 192 193 on roots of each plantlet during transplantation in 1 L pots. The same volume of sterilized water was applied to a 194 control set of plants. A set of 15 plants was inoculated per treatment. The plants were grown under greenhouse 195 conditions as reported by Lagrange et al. (2011) [33]. After 15 months of growth, the plants were harvested. Shoot 196 and root biomasses were determined by measuring the dry mass tissue (3 days at 60°C). The 15 samples of plant 197 tissues were randomly pooled by five and ground to powder. The contents of P, K, Ca, Mg, Co, Cr, Mn, and Ni in 198 tissues were quantified by ICP OES with a Varian 730-ES spectrometer after digestion in a mixture (4:1, v/v) of 199 HNO<sub>3</sub> (69%) and H<sub>2</sub>SO<sub>4</sub> (37%) at the Laboratoire des Moyens Analytiques (LAMA-US IMAGO-IRD, New 200 Caledonia).

201 2.8 Statistical analysis

All statistical analysis were performed with the R software version 3.3.1 [34]. All data were analyzed with parametric tests (one-way analysis of variance (ANOVA), followed by Tukey's honestly significant difference (HSD) test; P < 0.05) or with non-parametric tests (Kruskal-Wallis test; followed by the Fisher's least significant difference (LSD) test; P < 0.05) when data did not follow the assumptions of a parametric test.

206 3 Results

207 3.1 Test for metal tolerance

The metals  $IC_{50}$  determined for the three species are reported in Table 1. *P. ultramafica* STM10279<sup>T</sup> and *C. novacaledonica* STM10272<sup>T</sup> showed a significant higher tolerance to Co, Mn and Ni compared to *P. caribensis* MWAP64<sup>T</sup>. For instance, *P. ultramafica* STM10279<sup>T</sup> Ni-IC<sub>50</sub> value was 16 times significantly higher and *C. novacaledonica* STM10272<sup>T</sup> Ni-IC<sub>50</sub> value was 2.5 times significantly higher than Ni-IC<sub>50</sub> value of *P. caribensis* MWAP64<sup>T</sup>. The two ultramafic species showed significantly different metal tolerance profiles for Cr and Ni. Indeed, the Ni IC<sub>50</sub> of *P. ultramafica* STM10279<sup>T</sup> was 6.4 times higher compared to Ni-IC<sub>50</sub> value of *C. novacaledonica* STM10272<sup>T</sup>. Conversely, the Cr IC<sub>50</sub> of *P. ultramafica* STM10279<sup>T</sup> was significantly lower than Cr-IC<sub>50</sub> value of *C.*  *novacaledonica* STM10272<sup>T</sup>. No significant differences between the two ultramafic species were observed for the
 tolerance to Co and Mn.

#### 217 3.2 Ni biosorption and bioaccumulation

218 Biosorption of Ni onto the cell surface and intracellular compartment were evaluated. Results expressed in ng 219  $(10^7 \text{CFU})^{-1}$  are reported in Fig. 1. The fraction of Ni inside the cells and onto cell wall surface were significantly lower for P. ultramafica STM10279<sup>T</sup> and C. novacaledonica STM10272<sup>T</sup> than for P. caribensis MWAP64<sup>T</sup>. 220 Moreover, significant differences are observed between P. ultramafica STM10279<sup>T</sup> and C. novacaledonica 221 222 STM10272<sup>T</sup>. Major differences appeared at 5 mM of initial Ni concentration, the mean intracellular Ni content in P. 223 ultramafica STM10279<sup>T</sup> (0.76  $\pm$  0.02 ng (10<sup>7</sup>CFU)<sup>-1</sup>) was 430 times lower than the value of *C. novacaledonica* 224 STM10272<sup>T</sup> and was 11,000 times lower compared to the value of *P. caribensis* MWAP64<sup>T</sup>. Ni content on cell surface of *P. ultramafica* STM10279<sup>T</sup> (3.45 ± 0.34 ng (10<sup>7</sup>CFU)<sup>-1</sup>) was 86 times lower than the value of *C.* 225 novacaledonica STM10272<sup>T</sup> and was significantly 675 times lower compared to the value of *P. caribensis* 226 MWAP64<sup>T</sup>. 227

#### 228 3.3 Biofilm formation assay in the presence of Ni

229 Biofilm formation by the three species studied in the presence of different Ni concentrations and their variation 230 with time are presented in Fig. 2. At 24 h increase of Ni concentration decreased biofilm formation for all the strains then, biofilm formation increased with time particularly for C. novacaledonica STM10272<sup>T</sup>. At 48 h and 72h 231 232 biofilm formation occurred until 2.5 mM of Ni initial concentration for P. ultramafica STM10279<sup>T</sup> and until 1 mM of Ni initial concentration for C. novacaledonica STM10272<sup>T</sup> and P. caribensis MWAP64<sup>T</sup>. The amount of biofilm 233 234 formed by C. novacaledonica STM10272<sup>T</sup> was significantly higher than those formed by P. ultramafica STM10279<sup>T</sup> 235 and P. caribensis MWAP64<sup>T</sup>. The differences found ranged from 1.8 times, 2.6 times and 3 times higher at 236 respectively 0, 0.5 and 1 mM initial Ni concentration at 48 h and 72 h, in comparison to the control. At 48 h and 72 h 237 initial Ni addition did not influence biofilm formation by C. novacaledonica STM10272<sup>T</sup> until 2.5 mM. At the opposite, biofilm formation by P. ultramafica STM10279<sup>T</sup> and P. caribensis MWAP64<sup>T</sup> decreased with the increase 238 of Ni concentration until 2.5 mM and 1 mM of Ni for P. ultramafica STM10279<sup>T</sup> and P. caribensis MWAP64<sup>T</sup> 239 240 respectively.

#### 241 3.4 Quantification of biofilm production

The biovolume, thickness and substratum coverage of the biofilms produced by the three species studied are presented in Fig. 3. The biovolume, the thickness average and the substratum coverage of the biofilm produced by *P*. *ultramafica* STM10279<sup>T</sup> and *C. novacaledonica* STM10272<sup>T</sup> were significantly higher compared to the biofilm of *P*. *caribensis* MWAP64<sup>T</sup>. Moreover, the addition of 1 mM of Ni to the medium has significantly reduced the substratum coverage by *P. caribensis* MWAP64<sup>T</sup> biofilm from 60% to 39%. For *C. novacaledonica* STM10272<sup>T</sup>, the addition of Ni has significantly enhanced the thickness average of the biofilm from 21 μm to 26 μm.

#### 248 3.5 EPS production and its gross chemical composition

249 Under the experimental conditions used in this study, P caribensis MWAP64<sup>T</sup> was able to produce EPS at a production yield around 1.2 g L<sup>-1</sup>. Regarding ultramafic species, only *P. ultramafica* STM10279<sup>T</sup> was able to 250 produce EPS at a production yield around 140 mg L<sup>-1</sup>. FT-IR spectroscopy was employed to detect the main 251 252 functional groups in the EPS produced by P. ultramafica STM10279<sup>T</sup>. The FT-IR spectrum obtained (Fig. 4) 253 showed a broad and large absorbance band centered at 3276 cm<sup>-1</sup> that was assigned to hydroxyl groups or hydrogen 254 bonds. It also showed two peaks at 2925 and 2868 cm<sup>-1</sup> that were respectively assigned to CH-stretching of 255 methylene (-CH<sub>2</sub>) and methyl (-CH<sub>3</sub>) groups. Several absorbance peaks between 1500 cm<sup>-1</sup> and 1760 cm<sup>-1</sup> that were 256 assigned to the stretching modes of C=O bonds were also visible. The peak at 1719  $cm^{-1}$  indicated the presence of 257 carboxyl of uronic acid groups, the peak at 1248 cm<sup>-1</sup> is assigned to the deformation vibration of C=O bonds and the 258 peak at 1032 cm<sup>-1</sup> is assigned to the stretching modes of C-O from intra-molecular C-O-C ring. This latter peak thus 259 strongly suggested that the monosaccharides in the EPS have pyranose rings. In addition, the two weak peaks near 260 841cm<sup>-1</sup> and 896 cm<sup>-1</sup> could be respectively assigned to the two  $\alpha$ - and  $\beta$ -ends of carbon-glucoside bonds. Finally, 261 two additional peaks could be seen at 783 and 701 cm<sup>-1</sup> but they could not be unambiguously assigned to specific 262 molecular groups. The colorimetric test carried out (Supplemental Table 3) has shown that the ratio of carbohydrates 263 to proteins (C/P) was almost 15, with values of 61% carbohydrates to 4% of proteins residues and the presence of 264 uronic acid (17%).

#### 265 3.6 Effect of microbial inoculation in Tetraria comosa under ultramafic conditions

266 Most significant data variations were observed for the contents of metal in plant tissue (Table 2). In shoot, no 267 significant differences were measured between the inoculated plants and the control, but in roots, the three strains 268 reduced metal contents even if no significant differences were observed with the inoculation of C. novacaledonica STM10272<sup>T</sup> for Co and Mn contents compared with control. The values of the 4 elements were about two times 269 lower in presence of *P. ultramafica* STM10279<sup>T</sup> and *P. caribensis* MWAP64<sup>T</sup> in comparison with control plants. 270 271 The effect of C. novacaledonica STM10272<sup>T</sup> on root metal contents was less pronounced but the differences with P. 272 ultramafica STM10279<sup>T</sup> were not significant. The inoculation of *T. comosa* plants with *P. ultramafica* STM10279<sup>T</sup> and P. caribensis MWAP64<sup>T</sup> did not induce significant changes in plant biomass (Supplemental Table 4) and main 273 274 nutritional element contents in plant organs (Supplemental Table 5). However, the inoculation with C. 275 novacaledonica STM10272<sup>T</sup> significantly decreased the shoot and root biomasses compared to the other treatments 276 (50% lower in average).

#### 277 4 Discussion

In our experiments, C. novacaledonica  $STM10272^{T}$  and P. ultramafica  $STM10279^{T}$  showed a higher metal 278 279 tolerance than *P. caribensis* MWAP64<sup>T</sup> a non-ultramafic species isolated from Vertisol in Martinique [15]. These 280 results are in agreement with previous studies, which have shown that ultramafic soils are known to hold bacterial 281 communities more adapted to metal stress [4]. For instance, a set of Burkholderia s.l. strains isolated from Ni-rich 282 ultramafic soil, tolerated up to 15 mM Ni, 10 mM Co and 5 mM Cr [35]. In New Caledonia, Stoppel and Schlegel (1995) [36] reported the occurrence of Burkholderia s.l. strains that could tolerate up to 30 mM Ni which is 283 comparable to P. ultramafica STM10279<sup>T</sup> Ni-IC<sub>50</sub>.. Differences in profiles of metal tolerance were observed between 284 285 the two ultramafic species and more particularly for Ni suggesting that different mechanisms are involved depending 286 of the species.

Intracellular sequestration has already been reported as a mechanism of Ni tolerance for *Burkholderia* s.l. species [37]. However, our data indicated a preferential biosorption of Ni onto the cell surface of the three species studied. This result suggests that the tolerance mechanism used by these three species was to limit the movement of Ni across the cell envelope. *Burkholderia* s.l. species are gram-negative bacteria that exhibit a complex cell-membrane structure composed of numerous phosphorylated biomolecules (i.e. phospholipids, lipoprotein, lipopolysaccharide)
located on the external membrane and involved in metal chelation [38]. In fact, the various components (proteins,
exopolysaccharides layers, extracellular capsules) of the surface of bacteria cells are known to provide non-specific
sorption capacity toward metals [39]. This non-specific binding capacity of the cell wall has been already
demonstrated to prevent the entry of nickel into the bacteria cell and thus to hinder its interference with essential
cellular components [7].

297 Biofilm formation is known to be involved in bacterial tolerance to Ni [17,40]. In our study, biofilm production 298 occurred until 1 mM for C. novacaledonica STM10272<sup>T</sup> and P. caribensis MWAP64<sup>T</sup> and until 2 mM for P. 299 ultramafica STM10279<sup>T</sup>. It is not surprising that biofilm production was reduced at concentrations below the IC<sub>50</sub>. 300 Indeed, Vega et al. (2014) [41] have shown that Ni inhibited the Quorum Sensing and biofilm formation without 301 affecting viability in Burkholderia multivorans. Confocal analysis clearly demonstrated the negative effect of nickel 302 on the formation of biofilm in *P. caribensis* MWAP64<sup>T</sup> suggesting that another mechanism must be involved in the 303 metal tolerance. Until 1 mM of Ni, biofilm formation appeared to be an important mechanism for C. novacaledonica 304  $STM10272^{T}$  in view of the large quantities of biofilm found and the enhancement of thickness average with the 305 addition of Ni. Since the concentration of bioavailable Ni frequently reaches 1 mM in New Caledonian ultramafic 306 soils, especially in the rhizosphere [22], we can conclude that Ni-concentration in these soils could induce biofilm 307 formation. These results corroborate those obtained by Perrin et al. (2009) [42] who showed in vitro biofilm 308 stimulation by nickel for Escherichia coli.

309 In order to go further into the mechanisms developed by the three Burkholderia s.l. species, we investigated their 310 ability to excrete EPS. Indeed, the high molecular sugar-based polymers that constitute these bacterial extracellular 311 substances have already been described to play a role in Burkholderia s.l. adaptation to different stress conditions 312 [43]. Bacterial EPS are known to play a significant role in the complexation of metals and subsequent decrease in their mobility in soils [39]. Under our experimental conditions, P. ultramafica STM10279<sup>T</sup> and P. caribensis 313 314 MWAP64<sup>T</sup> were able to produce EPS. P. caribensis MWAP64<sup>T</sup> is a well-known EPS-producing bacterium [15]. 315 Structural analysis of the polysaccharide excreted by P. caribensis strain MWAP71, a strain belonging to the P. 316 caribensis cluster [15] has shown the presence of 3-deoxy-D-manno-oct-2-ulosonic acid (Kdo) [44]. Anionic groups 317 such as carbonyl, carboxyl and hydroxyl groups have been reported to scavenge several metals and to be involved in

metal biosorption [39,45]. Colorimetric assays showing that carbohydrates were the main component in the crude 318 319 EPS produced by P. ultramafica STM10279<sup>T</sup> and the FT-IR peaks found suggest that the EPS of P. ultramafica 320 STM10279<sup>T</sup> is a polysaccharide containing pyranose and uronic acid groups [46–48]. All the features on the FT-IR 321 spectrum of the EPS of *P. ultramafica* STM10279<sup>T</sup> indicated that (i) carbohydrates were the main component in the 322 crude EPS produced by *P. ultramafica* STM10279<sup>T</sup> and (ii) the EPS produced by *P. ultramafica* STM10279<sup>T</sup> was a 323 polymer of a repeating unit presenting uronic acid residues. The pyranose and uronic acid groups that are expected to 324 constitute the polysaccharides are considered involved in the metal binding process. Such a behavior has already 325 been shown for Rhizobium etli, which was able to chelate Mn in these EPS [49].

326 The two ultramafic species clearly showed different metal tolerance patterns and mechanisms. C. novacaledonica 327 STM10272<sup>T</sup> was found to develop a biofilm to protect from metals, whereas *P. ultramafica* STM10279<sup>T</sup> excreted 328 EPS to sequester metals, although it was able to develop a biofilm. Finally, our results are in agreement with the 329 literature, as biofilm formation has already been reported for Paraburkholderia [50] and Caballeronia species [51], 330 while EPS production was only observed in the Paraburkholderia genus [15,52,53]. However, other mechanisms 331 may occur in metal tolerance of *P. ultramafica* STM10279<sup>T</sup> and *C. novacaledonica* STM10272<sup>T</sup>. Gonin et al. (2013) 332 [13] have shown that these bacteria possess cnr (Co-Ni resistance) and nre (Ni resistance) genes involved in metal 333 exclusion and active removal. Such mechanisms may operate in C. novacaledonica  $STM10272^{T}$  and in P. 334 ultramafica STM10279<sup>T</sup>.

335 All these mechanisms put in place by bacteria to protect themselves can be also involved in the variations of 336 metal absorption by inoculated plants [8]. Alleviation of metal contents in plants by metal tolerant Burkholderia s.l. 337 have been reported in the literature. Madhaiyan et al. (2007) [54] showed a reduction of Ni uptake and translocation 338 in tomato after inoculation with a Burkholderia s.l. strain. These authors suggested that this reduction might be 339 attributed to the biosorption of Ni by the bacterial cells, which would have prevented the metal from being 340 transported into the plant. Kuffner et al. (2010) [55] reported that a Burkholderia s.l. strain had the capacity of 341 reducing Zn and Cd accumulation in the roots of Salix caprea by favoring the immobilization of these metals in the soil. Our results on the effect of P. ultramafica STM10279<sup>T</sup> and C. novacaledonica STM10272<sup>T</sup> are in agreement 342 343 with these studies. However, the alleviation of metals observed in T. comosa were more pronounced with P.

*ultramafica* STM10279<sup>T</sup> than *C. novacaledonica* STM10272<sup>T</sup>, in possible relation to the differences of their metal
profile tolerance.

346 Although it is important to underline that an increase of the plant biomass is not the unique aspect of plant health 347 and adaptation [56], the lack of positive effect of bacterial inoculation on the biomass of T. comosa plants suggests 348 that metal alleviation had no influence on T. comosa plant growth. Other adaptation mechanisms inherent in the plant 349 may explain this result. Indeed, Gonin et al. (2013) [13] using sorghum, a plant non-native of ultramafic maquis, 350 reported a significant enhancement of plant biomass when inoculated by P. ultramafica STM10279<sup>T</sup>. It must be 351 outlined that Tetraria species are considered as oligotrophic (with low nutrition requirements) and their low growth 352 rate is an adaptive trait to ultramafic constraints [57], which can also explain the absence of positive effect of 353 bacterial inoculation on their biomass production. Concerning the detrimental effect of C. novacaledonica 354  $STM10272^{T}$  on plant biomass, it can be possibly explained by their ability to produce high level of biofilm. Although 355 biofilm plays a major role in the attachment of bacteria to plant roots, known biofilm-forming rhizosphere 356 Burkholderia s.l. are reported to inhibit root development [58].

#### 357 5 Conclusions

358 The results of the present study clearly demonstrated that the two ultramafic bacterial species used different 359 mechanisms to cope with metal toxicity such as cell wall absorption, biofilm formation and EPS production. These 360 mechanisms can explain the decrease of metal contents in the plant and underlined the importance to consider the plant/rhizobacteria interaction in ultramafic soils in order to understand the mechanisms occurred in plant adaptation 361 to the extreme edaphic constraints. Although no improvement in plant growth could be noticed, the ability of P. 362 ultramafica STM10279<sup>T</sup> to produce an EPS can create a more hospitable environment for other organisms that will 363 364 promote plant growth. It will then be interesting to experiment the co-inoculation of the studied bacteria with 365 arbuscular mycorrhizal fungi, to combine effects on metal alleviation in plant organs and improvement of plant 366 mineral nutrition.

- 367 Disclosure statement
- 368 Declaration of interest: none.

369 Acknowledgments

370	Tł	he authors thank J.L. Duprey and L. Jamet from the LAMA laboratory (LAMA-US IMAGO-IRD, New
371	Caled	onia) for chemical analyses and E. Chalkiadakis from BIOTECAL Company for the technical support. This
372	work	was supported by the Grand Observatoire du Pacifique Sud (ARBRE project); Province Sud of New Caledonia
373	(Prix	d' Encouragement à la Recherche: Thesis Grant and Vivaldi program: Master internship allowance).
374	Арре	ndix A. Supplementary data
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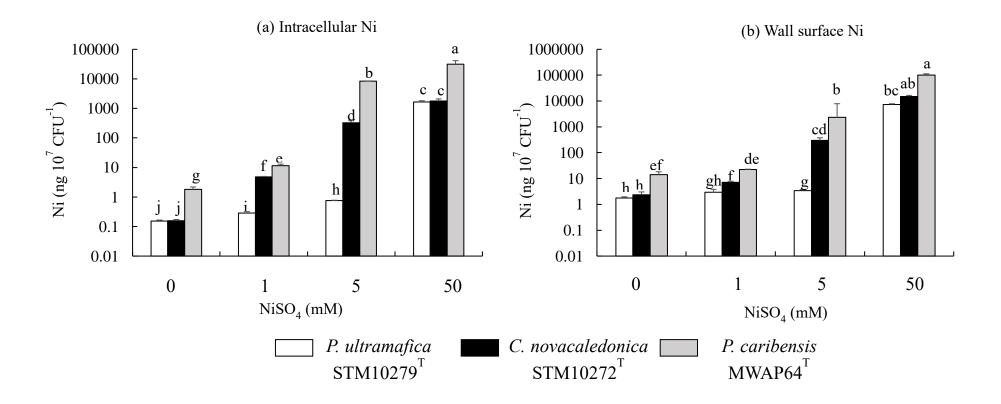
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Fig. 1.



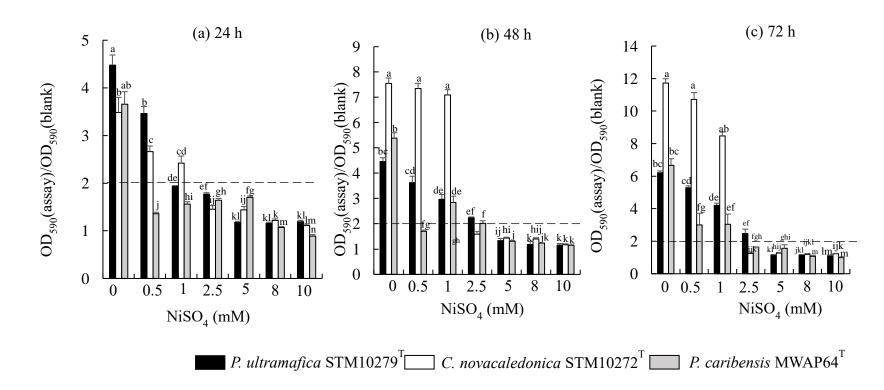


Fig. 2.

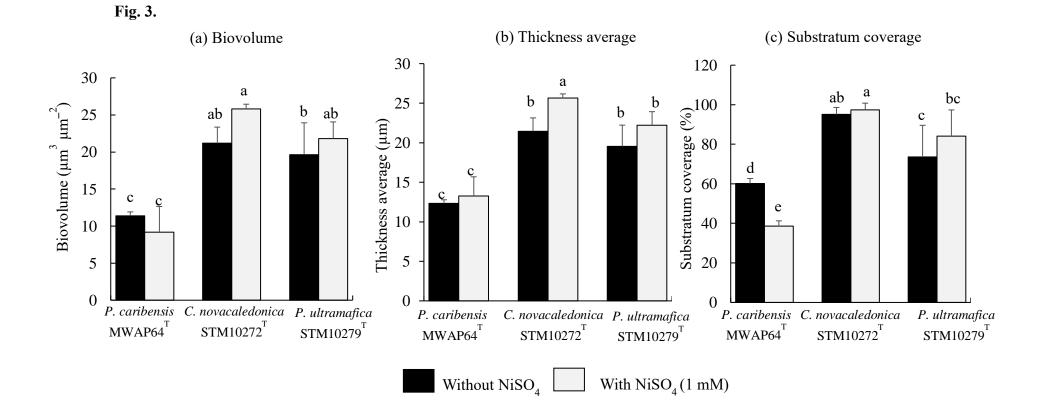
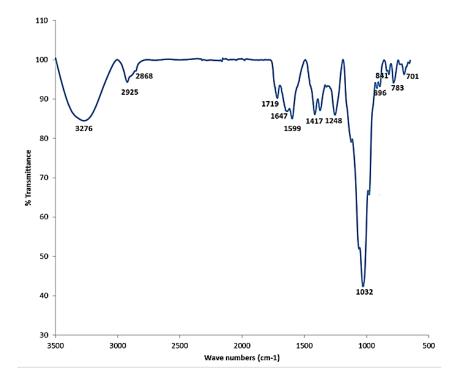


Fig. 4.



## **Figure Captions**

**Fig. 1.** Intracellular Ni (a) and wall surface Ni (b) in *Paraburkholderia ultramafica* STM10279<sup>T</sup>, *Caballeronia novacaledonica* STM10272<sup>T</sup> and *Paraburkholderia caribensis* MWAP64<sup>T</sup> as a function of the Ni concentration in the growing medium. Bars represent means, and error bars represent standard errors (3 replicates). Value with the same lower case letters are not significantly different at P < 0.05 using Fisher's LSD test.

**Fig. 2.** Formation of biofilm by *Paraburkholderia ultramafica* STM10279<sup>T</sup>, *Caballeronia novacaledonica* STM10272<sup>T</sup> and *Paraburkholderia caribensis* MWAP64<sup>T</sup> in response to nickel stress and as a function of the exposure time 24 h (a), 48 h (b) and 72 h (c). Bars represent means, and error bars represent standard errors (6 replicates). Biofilm production occurs when the ratio is greater than two (dotted line on the graphic). Value with the same lower case letters are not significantly different at P < 0.05 using Fisher's LSD test.

**Fig. 3.** Biovolume (a), Thickness average (b) and substratum coverage (c) for *Paraburkholderia* caribensis MWAP64<sup>T</sup>, Caballeronia novacaledonica STM10272<sup>T</sup> and Paraburkholderia ultramafica STM10279<sup>T</sup> with and without nickel in the growing medium. Bars represent means, and error bars represent standard errors (6 replicates). Value with the same lower case letters are not significantly different at P < 0.05 using Tukey's HSD test or Fisher's LSD test when data did not follow the assumptions of a parametric test.

**Fig. 4.** Fourier Transform Infrared (FT-IR) spectrum of the *Paraburkholderia ultramafica* STM10279<sup>T</sup> EPS in the 500-3500 cm<sup>-1</sup> range showing absorbance bands or peaks characteristic of polysaccharides.

### **Table 1** Metals IC<sub>50</sub> (mM) for Paraburkholderia ultramafica STM10279<sup>T</sup>, Caballeronia novacaledonica STM10272<sup>T</sup> and Paraburkholderia caribensis MWAP64<sup>T</sup>

### 2 (Mean $\pm$ Standard Error, n = 6).

IC <sub>50</sub> (mM)												
Species	Ni			Co			Cr			Mn		
<i>P. caribensis</i> MWAP64 <sup>T</sup>	1.9	±	0.9c	1.1	±	0.1b	0.2	±	0.1b	11.1	±	0.9b
P. ultramafica STM10279 <sup>T</sup>	30.7	±	2.8a	3.9	±	0.9a	0.5	±	0.1b	32.4	±	1.6a
C. novacaledonica $STM10272^{T}$	4.8	±	0.5b	5.8	±	0.9a	1.5	±	0.3a	29.1	±	5.7a

3 Value with the same lower case letters within a column are not significantly different at P < 0.05 using Fisher's LSD test.

4 **Table 2** Metals contents in the shoots and roots of *Tetraria comosa* as a function of various treatments (mean ± SE n = 3).

Dlant nort	The stars and		Ni (mg Kg <sup>-1</sup> )			Co (mg Kg <sup>-1</sup> )			Cr (mg Kg <sup>-1</sup> )			Mn (mg Kg <sup>-1</sup> )		
Plant part	Treatment	(m												
	Control	132	±	18a	13.5	±	1.8a	191	±	29ab	415	±	29a	
<u>C1</u>	<i>P. caribensis</i> MWAP64 <sup>T</sup>	159	±	18a	11.2	±	1.0a	242	±	12a	380	±	9a	
Shoot	P. ultramafica STM10279 <sup>T</sup>	155	±	12a	9.4	±	1.0a	191	±	13b	388	±	19a	
	C. novacaledonica $STM10272^{T}$	154	±	10a	15.2	±	0.5a	227	±	16ab	468	±	7a	
	Control	1377	±	111a	284	±	26a	2168	±	29a	3351	±	232a	
	<i>P. caribensis</i> MWAP64 <sup>T</sup>	582	±	54c	126	±	13b	811	±	12b	1806	±	150c	
Root	<i>P. ultramafica</i> $STM10279^{T}$	766	±	68bc	155	±	13b	1158	±	13b	2097	±	159b	
	C. novacaledonica STM10272 <sup>T</sup>	966	±	81b	206	±	19ab	1414	±	16b	2698	±	165ab	

5 Value with the same lower case letters within a column for each plant part are not significantly different at P < 0.05 using Tukey's HSD test or Fisher's LSD test

6 when data did not follow the assumptions of a parametric test.