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Regioselective 1,6-conjugate addition of organocuprate reagents to γ -alkylidenebutenolides

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ABSTRACT

This work describes the study of behavior of the δ -substituted and δ -non-substituted *exo*-cyclic double bond of γ -alkylidenebutenolides with dialkylcuprates. The addition reaction was found to be regioselective to afford the 1,6-conjugate adduct up to 87% yield.

1.4- and 1.6-conjugate addition reactions of carbon nucleophiles to electron-deficient alkenes represents a powerful tool for the formation of carbon-carbon bonds and more interestingly for the creation of tertiary or quaternary carbon stereocenters [1]. Due to the poor propagation of the electronic effect through the π -conjugated system (lower reactivity of the δ -position as compared to the β -position), the regioselectivity of the 1,6-conjugate addition reactions is a real challenge to overcome. In addition, γ -alkylidenebutenolides **1** are an important class of naturally occurring γ -lactones endowing with diverse biological properties including herbicidal, antimicrobial, antitumor, antidiabetic and much more [2]. Possessing an $\alpha, \beta, \gamma, \delta$ unsaturated moiety, as well as a cyclic enol acetate, those γ -lactones are interesting building blocks useful in various synthetic applications [3] and more specifically for the study of the regioselectivity of the 1,6-conjugate addition reactions of organocuprate reagents. To the best of our knowledge, only one example was reported by J. Font and coll. [4]. Unfortunately, the regioselective addition of dimethyl- and dibutylcuprates onto protoanemonine, the simplest y-alkylidenebutenolide **1** ($R^1 = R^2 = R^3 = H$), proceeded with moderate yields, 27 and 17% respectively. Herein, we wish to report our study on the behavior of the $\delta\text{-substituted}$ and $\delta\text{-non-substituted}$ exo-cyclic double bond of γ -alkylidenebutenolides 1 with dialkylcuprates. (Scheme 1).

Our initial plan was to use γ -alkylidenebutenolide **1a** (R¹ = H, R² = Me, R³ = Bu) as a model substrate. **1a**, as well as the requisite

 γ -lactones **1**, were prepared according to our reported procedure [5]. The reaction was initially performed with 1.5 equivalent of Me₂CuLi-LiI at 0 °C followed by a saturated aqueous NH₄Cl hydrolysis (Table 1, entry 1). In these conditions, we were delighted to observe that the reaction proceeded in a complete regioselective way. The 1,6-conjugate adduct **2a** was obtained in good yield (87%) after the selective α -protonation of the corresponding lithium enolate intermediate.

Quenching the reaction with a buffer solution (NH_4Cl/NH_4OH) at 0 °C or at -78 °C (Table 1, entries 2–3) didn't improve the isolated yield of **2a**.

In addition, no better results (Table 1, entries 4–5) were observed when the reaction was carried out neither with organocopper reagent MeCu-LiI nor with higher-order cyanocuprate reagent Me₂CuLi-LiCN.

Under the optimal reaction conditions, the scope of 1,6-conjugate addition was then assessed through the variation of the dialkylcuprate reagents (Me₂CuLi·LiI, Et₂CuMg·MgI and Bu₂-CuLi·LiI) and the nature of the substituents R¹ (Me, H), R² (Me, H) and R³ (Bu, H, Ph, CH₂OTBS and CH₂OMe) onto the γ -alkylidenebutenolide **1** (Table 2). The methodology was found to be



Scheme 1. Regioselective 1,6-conjugate addition of organocuprate reagents to γ -alkylidenebutenolides **1**.

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Table 1Optimization of reaction conditions.

	Bu 1a	$\begin{array}{c} 1) [Me_nCuM] \\ \hline Et_2O, 0 \ ^{\circ}C \\ 2) \text{ hydrolysis} \end{array} \xrightarrow[Bu]{} 0 \ ^{\circ}C \\ Bu \\ 2a \\ 3a \end{array} \xrightarrow[Bu]{} 3a \\ Et_2O, 0 \ ^{\circ}C \\ Bu \\ Bu \\ Ca \\ Bu \\ Ca \\ Bu \\ Ca \\ C$	4a	
Entry	Me _n CuM·LiX	hydrolysis	2a/3a/4a	Yield 2a
1 2 3 4 5	Me2CuLi-Lil Me2CuLi-Lil Me2CuLi-Lil MeCu-Lil Me2CuLi-LiCN	NH₄CI, 0 °C NH₄CI/NH₄OH (9/1) 0 °C NH₄CI/NH₄OH (9/1) –78 °C NH₄CI, 0 °C NH₄CI, 0 °C	1/0/0 1/0/0 1/0/0 No reaction 1/0/0	87% 66% 86% 52%

Table 2

Scope and Limit of the regioselective 1,6-conjugate addition of organocuprate reagents to γ -alkylidenebutenolides 1.

		$ \begin{array}{c} R^2 \\ R^3 \end{array} \begin{array}{c} R^1 \\ R^3 \end{array} \begin{array}{c} 1) \\ 1 \\ 2 \end{array} $	[R ⁴ ₂ CuM], Et ₂ O, 0 °C NH ₄ Cl, 0 °C R ⁴ R ³	R^1 R^2 R^4 R^4 R^3	R^1 R^2 R^1 R^4 R^4 R^1 R^4 R^1 R^2 R^1 R^3		
Entry		1	2 R ₂ ⁴ CuM·MX	3 2/3/4ª	4 Major Proc	luct	Yield
1	200	1a	Me ₂ CuLi·LiI	1/0/0	, , , , , , , , , , , , , , , , , , , ,	2a-Me	87%
2	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	1a	Bu₂CuLi·LiI	1/0/0	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	2a-Bu	81%
3	20	1a	$Et_2CuMg\cdot MgI$	1/0/0	~~~~~	2a-Et	66%
4	200	1b	Me ₂ CuLi·LiI	1/0/0	100	2b-Me	62%
5	TBSO	1b	Bu ₂ CuLi·LiI	1/0/0	TBSO-	2b-Bu	47%
6	TBSO	1b	$Et_2CuMg\cdot MgI$	1/0/0	TBSO	2b-Et	39%
7	TBSO	1c	Me ₂ CuLi·LiI	0/1/0	TBSO	3c-Me	60% d.r. = 2:1
8	MeO	1d (Z/E:9/1)	Me ₂ CuLi·LiI	1/0/0	MeO	2d-Me	78%
9	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	1e (<i>Z</i> / <i>E</i> :86/14)	Me ₂ CuLi·LiI	1/0/0		2e-Me	46%
10	TBSO-	1f	Me ₂ CuLi·LiI	1/0/0	TBSO	2f-Me	54%
11	Pn	1g	Me ₂ CuLi·LiI	0/1/0		3g-Me	52%
12		1g	Bu ₂ CuLi·LiI	0/1/0		3g-Bu	41%
13		1a	Me ₂ CuLi-Lil ^b	7/93/0		За-Ме	56% d.r. = 7:3

^a Ratio calculated from crude ¹H NMR.

^b 2 equiv. of TMSCl were used as additive.

general with an excellent control of the regioselectivity of the conjugate addition affording the desired 1,6-adducts. When $R^2 = Me$ and $R^3 = Bu$ or CH₂OTBS (Table 2, entries 1–6), the deconjugated γ -lactones **2a** were isolated in acceptable to good yields whatever the nature of the dialkylcuprates. The moderate yield (Table 2, entry 6) of the isolated **2b-Et** could be explained by its degradation during the purification step.

Indeed, unidentified compounds, which were non-present in the crude of the reaction, were isolated after the purification step. Interestingly, when $R^3 = CH_2OTBS$ is replaced by $R^3 = CH_2OMe$ (entry 7), the regioselective 1,6-conjugate addition still occurred

but with the formation of the conjugated γ -lactone **3c-Me** (60%), as a 2/1 mixture of two diastereomers, after the selective γ -protonation of the corresponding lithium enolate intermediate.

The regioselectivity was then assessed by using non- β -substituted γ -alkylidenebutenolides **1d-1f** (Table 2, entries 8–10). When **1d** and **1e** were subjected to 1,5 eq. of Me₂CuLi-LiI, we were delighted to observe a total regioselective 1,6-conjugate addition in satisfactory yields. This selectivity was also observed starting from the all conjugated γ -alkylidenebutenolide **1f** bearing a phenyl group (R² = H, R³ = Ph) at the γ -position and promoting the 1,4-addition. Pleasantly, only the 1,6-adduct **2f-Me** was formed

(entry 10) without observing the presence of the other regioisomer arising from the 1,4-conjugate addition. In addition, starting from non- δ -substituted γ -alkylidenebutenolide **1h** (Table 2, entries 11–12), the regioselective 1,6-conjugate addition of Me₂CuLi-LiI and Bu₂CuLi-LiI occurred to give after purification, the conjugated γ -lactones **3g-Me** and **3g-Bu** in 52 and 41% yield respectively.

Finally, the 1,6-conjugate addition of Me₂CuLi-LiI to γ -alkylidenebutenolide **1a** was then performed by using 2 eq. of chlorotrimethylsilane as additive [6]. Unfortunately, these conditions did not increase the yield of the reaction but, interestingly, allow us to mainly obtain, in moderate yield, the conjugated γ -lactone **3a-Me**, as a 7:3 mixture of 2 diastereomers, instead the deconjugated γ -lactones 2 (Table 2, entries 13). With these conditions in hand, we have a tunable 1,6-conjugate addition towards either deconjugated γ -lactones **2** or conjugated γ -lactones **3** starting from β -substituted γ -alkylidenebutenolides.

In conclusion, the conjugate addition of dialkylcuprate onto δ substituted γ -alkylidenebutenolides is described. This addition was found to be highly regioselective to afford the corresponding 1,6-conjugate adduct.

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References

- (a) A.G. Csakÿ, G. de la Herrán, M.C. Murcia, Chem. Soc. Rev. 39 (2010) 4080;
 (b) E.M.P. Silva, A.M.S. Silva, Synthesis 44 (2012) 3109;
 - (c) P. Chauhan, U. Kaya, D. Enders, Adv. Synth. Catal. 359 (2017) 888.
- [2] (a) T. Kajikawa, K. Aoki, T. Iwashita, D.M. Niedzwiedzki, H.A. Frank, S. Katsumura, Org. Biomol. Chem. 8 (2010) 2513;
 - (b) M.J. Ortega, E. Zubia, J.M. Ocana, S. Naranjo, J. Salva, Tetrahedron 56 (2000) 3963;
 - (c) S. Miao, R.J. Andersen, J. Org. Chem. 56 (1991) 6275;
 - (d) A.R. Carroll, P.C. Healy, R.J. Quinn, C.J. Tranter, J. Org. Chem. 64 (1999) 2680; (e) C.J. Smith, R.L. Hettich, J. Jompa, A. Tahir, M.V. Buchanan, C.M. Ireland, J. Org.
- Chem. 63 (1998) 4147. [3] (a) L.C.A. Barbosa, R.R. Teixeira, G.W. Amarante, Curr Org. Synt. 12 (2015) 746,
- and references cited therein;

(b) A. Bartoli, F. Rodier, L. Commeiras, J.-L. Parrain, G. Chouraqui, Nat. Prod. Rep. 28 (2011) 763, and references cited therein;

(c) G.-Z. Liu, H.-W. Xu, G.-W. Chen, P. Wang, Y.-N. Wang, Bioorg. Med. Chem. 18 (2010) 1626;

(d) B. Vaz, M. Domínguez, R. Álvarez, A.R. de Lera, J. Org. Chem. 71 (2006) 5914; (e) F. Rodier, J.-L. Parrain, G. Chouraqui, L. Commeiras, Org. Biomol. Chem. 11 (2013) 4178;

(f) F. Rodier, M. Rajzmann, J.-L. Parrain, G. Chouraqui, L. Commeiras, Chem. Eur. J. 19 (2013) 2467;

(g) S. Dubois, F. Rodier, R. Blanc, R. Rahmani, V. Héran, J. Thibonnet, L. Commeiras, J.-L. Parrain, Org. Biomol. Chem. 10 (2012) 4712;

(h) R. Blanc, V. Héran, R. Rahmani, L. Commeiras, J.-L. Parrain, Org. Biomol. Chem. 8 (2010) 5490.

- [4] J. Bigora, J. Font, C. Jaime, R.M. Ortuño, F. Sanchez-Ferrando, Tetrahedron 41 (1985) 5577.
- [5] (a) S. Inack-Ngi, R. Rahmani, L. Commeiras, G. Chouraqui, J. Thibonnet, A. Duchêne, M. Abarbri, J.-L. Parrain, Adv. Synth. Catal. 351 (2009) 779;
 (b) S. Inack Ngi, K. Cherry, Héran, L. Commeiras, J.-L. Parrain, A. Duchêne, M.
- Abarbri, J. Thibonnet, Chem. Eur. J. 17 (2011) 13692. [6] (a) E.J. Corey, N.W. Boaz, Tetrahedron Lett. 26 (1985) 6015;
- (b) E.J. Corey, N.W. Boaz, Tetrahedron Lett. 26 (1985) 6019.