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# Ultrafast dynamics and vibrational relaxation in six-coordinate heme proteins revealed by Femtosecond Stimulated Raman Spectroscopy.

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## Abstract

Identifying the structural rearrangement and the active sites in photo-induced reactions is a fundamental challenge to understand from a microscopic perspective the underlying dynamics ruling the functional mechanisms of heme proteins. Here, femtosecond stimulated Raman spectroscopy is used to follow the ultrafast evolution of two

six-coordinate heme proteins. By exploiting the sensitivity of Raman spectra to the structural configuration, we investigate the effects of photolysis and binding of amino acid residues in cytochrome *c* and neuroglobin. Comparing the system response for different time delays and Raman pump resonances, we show how details of atomic motions and energy redistribution can be unveiled.

## Introduction

Heme proteins constitute one of the most important families of macromolecular compounds, present in all living organisms and displaying a wide range of biological functions, including oxygen transport and intracellular trafficking. The heme protein's core is an iron atom, confined in a porphyrin ring via four bonds with nitrogen atoms, and able to bind two axial ligands. One of these ligands is usually an amino acid as histidine or methionine. The other ligation site can be occupied by a small gaseous molecule like O<sub>2</sub> or CO, with another amino acid, or be unoccupied. Changes in ligation state are often involved in the function of the heme protein. Therefore, the characterization of the heme-iron bonds and their connection with conformational rearrangements of the porphyrin are crucial for understanding the structural mechanisms ruling the biological role of such prototypical compounds. Notably, several concurring processes, such as structural reconfiguration, energy redistribution and relaxation of intermediate excited states, occur on picosecond and femtosecond timescales, and accompany bond breaking and recombination events. The advent of femtosecond laser sources and the development of optical nonlinear techniques have paved the way to a direct exploration of heme protein ultrafast dynamics.<sup>1-5</sup>

Within such studies, particular efforts have been devoted to the understanding of a rapidly expanding group of proteins, namely those able to form the iron atom sixth bond either with an internal residue or with an external ligand (like O<sub>2</sub>, NO or CO). Notably, within this class of proteins, the exchange of internal and external ligands is often thought to be functional. Upon (photo-)dissociation of a gaseous ligands, the kinetics of binding of the

internal residue is rate-limited by migration of the the gaseous ligand through the protein matrix and steric rearrangement,<sup>6-13</sup> whereas dissociation of the heme-residue bond in the absence of external ligands leads to at least 6 orders of magnitude faster rebinding.<sup>14-16</sup>

In particular, the photodissociation and fast rebinding dynamics of the heme-residue bond, occurring on the picosecond time scale (5-9 ps), have been assessed using transient absorption (TA).<sup>17,18</sup> Critically, TA spectroscopy is not able to directly monitor geometrical rearrangements of the heme, neither to distinguish between transient electronic intermediates or de-excitation to a vibrationally hot electronic ground state due to its lack of structural sensitivity.<sup>4,19-21</sup> In contrast, conventional spontaneous Time-Resolved Resonance Raman (TR<sup>3</sup>) spectroscopy can provide direct fingerprints of the transient structural configuration of a reacting heme. However, TR<sup>3</sup> temporal and energy resolutions are fundamentally constrained by the Fourier transform limit ( $\Delta E \Delta t \geq 15 \text{ cm}^{-1} \text{ ps}$ ), making it difficult to resolve spectral details arising from ultrafast dynamics on the  $\lesssim 1 \text{ ps}$  timescale.

Nevertheless, using an accurate balance of the two resolutions, Cianetti et al. provided useful insight on the photolysis observing the appearance, in ferrous six-coordinate (6-c) cytochrome *c* (cyt *c*), of a low frequency Raman band (centered at  $216 \text{ cm}^{-1}$  and assigned to the iron-histidine stretching mode) sensitive to the doming of the porphyrin ring at the picosecond time scale.<sup>22</sup>

In this work, in order to circumvent the TR<sup>3</sup> limitations, we take advantage of the recently introduced Femtosecond Stimulated Raman Spectroscopy (FSRS) non-linear technique,<sup>23-25</sup> combining both high spectral and temporal resolutions.<sup>26-29</sup> The experimental scheme requires three pulses, a femtosecond actinic pump (AP), that triggers the dynamic of interest, a Raman pulse (RP) and a broadband probe pulse (PP), to probe photoreactions with uncompromised temporal precision (down to 50 fs) and spectral resolution (a few wavenumbers) via stimulated Raman scattering.<sup>24</sup> Specifically, probing transient stimulated vibrational spectra and taking advantage of the wavelength tunability of a narrowband Raman pulse,<sup>30,31</sup> FSRS has proven able to assign contributions arising from different intermediate states and

to provide important insights on femtosecond structural changes upon photo-excitation.<sup>32-36</sup> By comparing the Raman band cross sections and positions, measured scanning the RP wavelength across the absorption bands of transient and relaxed species,<sup>37,38</sup> it is possible to extract detailed structural information on the system dynamics: for this reason we developed a FSRS setup with a tunable Raman pulse (Femtosecond Stimulated Resonance Raman Scattering, FSRRS) able to explore resonance effects across all the Soret absorption band (400-460 nm).<sup>39</sup> Here we study two heme proteins with the hemes in the ferrous 6-c form with two amino acid ligands: cyt *c* and neuroglobin (Ngb). Ngb is a relatively recently discovered protein,<sup>40</sup> which is present predominantly in brain and nerve tissues and whose biological function is still elusive. As the prototypical oxygen-storage protein myoglobin (Mb) it can easily bind small gaseous ligands to the *b*-type heme cofactor, but whereas the heme in unliganded deoxy-myoglobin is 5-c in the absence of external ligands in Ngb it is 6-c with two histidine residues as ligands. cyt *c* is an ubiquitous and well-studied soluble electron transfer heme protein. During its electron transfer function it switches between the ferric and ferrous state, always remaining 6-c with histidine and methionine as axial ligands. Other than globin heme proteins it does not readily bind external ligands, except when it is modified by interactions with lipids.<sup>41,42</sup> It is also relatively rigid,<sup>43,44</sup> a property thought favorable for its electron transfer function.<sup>45</sup> Another important difference with Ngb is that it carries a *c*-type heme, which implies that the heme binds to the protein not only via the iron atom but also via two covalent bonds of the peripheral vinyl groups with cysteine residues.<sup>46</sup> The external ligand<sup>47</sup> and one of the internal ligands in ferrous 6-c heme proteins not binding external ligands<sup>18</sup> can be photo-dissociated with high quantum yield. Our Raman results elucidate the detail of the dynamics upon the photo-excitation, allowing to discern between structural modification and energy transfer in such molecules. Specifically we find a new marker of internal ligand photolysis in the high frequency region ( $\nu_4$  at 1360  $\text{cm}^{-1}$ ) common for both the molecules and molecule-specific cooling processes that involve different Raman modes.

## Results

To visualize the electronical resonances of the system, TA spectra of photo-excited ferrous Mb (5-c), cyt *c* and Ngb are reported in Fig. 1, with a sketch of the heme pockets for the three different protein species reported in the right panel. Notably, in view of the absence of the axial distal ligand, the photo-dynamics of the 5-c heme in Mb is in principle distinguished from 6-c hemes due to the absence of photolysis and is characterized only by an energy transfer for times  $> 50$  fs.<sup>20,38</sup> Although the structural pathway is different, the TA spectra measured in the Soret band are qualitatively similar, characterized by a dispersive shape with a maximum absorption red-shifted with respect to the Soret band, testifying the poor structural sensitivity of TA.

More quantitatively, in Fig. 1a-b-c, after the instantaneous appearance of a dispersive feature in TA data, the recovery of the initial absorption profile occurs faster in the Mb case ( $\approx 3$  ps) than for the two 6-c hemes ( $\approx 8$  ps for Ngb and  $\approx 5$  ps for cyt *c*) (cf. refs.<sup>18,21</sup>).

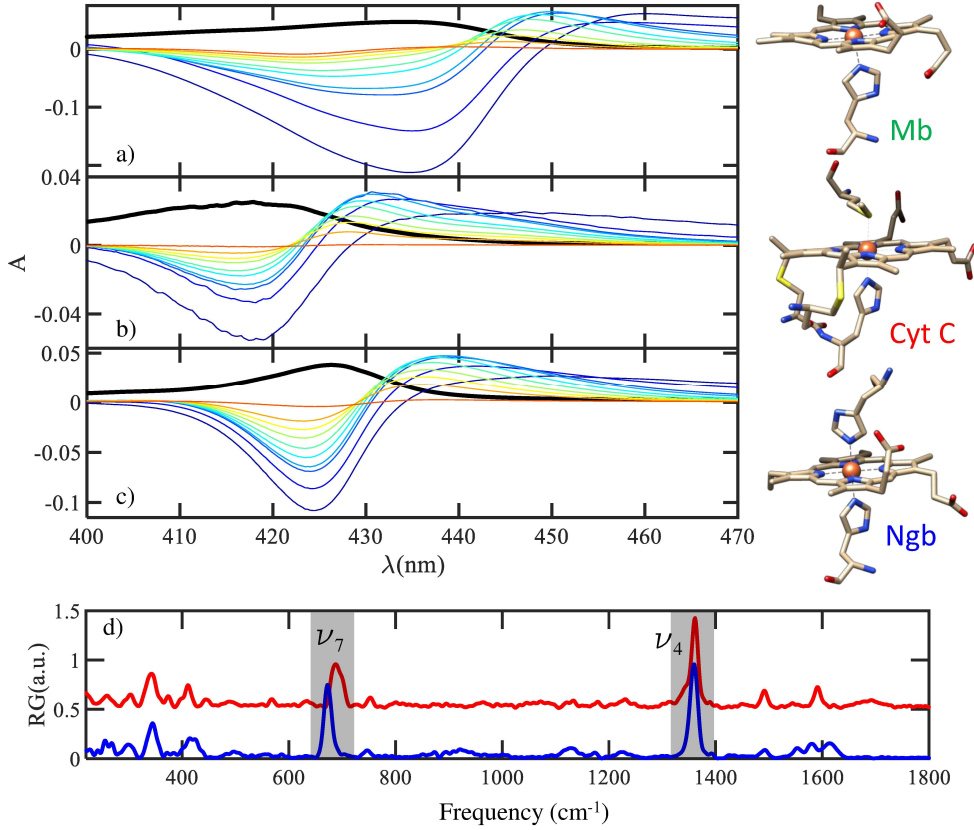


Figure 1: TA traces around the Soret band for different time delays in Mb (a), *cyt c* (b) and Ngb (c) are reported as colored lines in panels a-b-c. The time delays are 0.1, 0.3, 0.5, 1, 2, 3, 5, 7, 10, 30 ps from blue to red. The scaled static absorption profiles are reported as black lines. In d) we report the *cyt c* and Ngb static stimulated Raman spectra, measured with a RP tuned at 430 nm (red and blue lines, respectively). The two main peaks at  $\sim 700$  and  $\sim 1350$   $\text{cm}^{-1}$  are the  $\nu_7$  and  $\nu_4$  respectively.

A careful inspection of the TA spectra also reveals the presence of a sub-picosecond more red-shifted tail, possibly related to the heating effects. Such relaxation is present for all the three investigated species<sup>16,18,48</sup> and it occurs on a femtosecond timescale (from  $\approx 150$  fs in *cyt c* to  $\approx 400$  fs in Ngb and Mb).

Steady-state stimulated Raman spectra of *cyt c* and Ngb, measured tuning the RP in resonance with the Soret band absorption setup without AP, are also reported in fig. 1d and reveals two dominant modes ( $\nu_7$  and  $\nu_4$ ), corresponding to totally symmetric in-plane breathing modes of the porphyrin ring.

## Cytochrome c

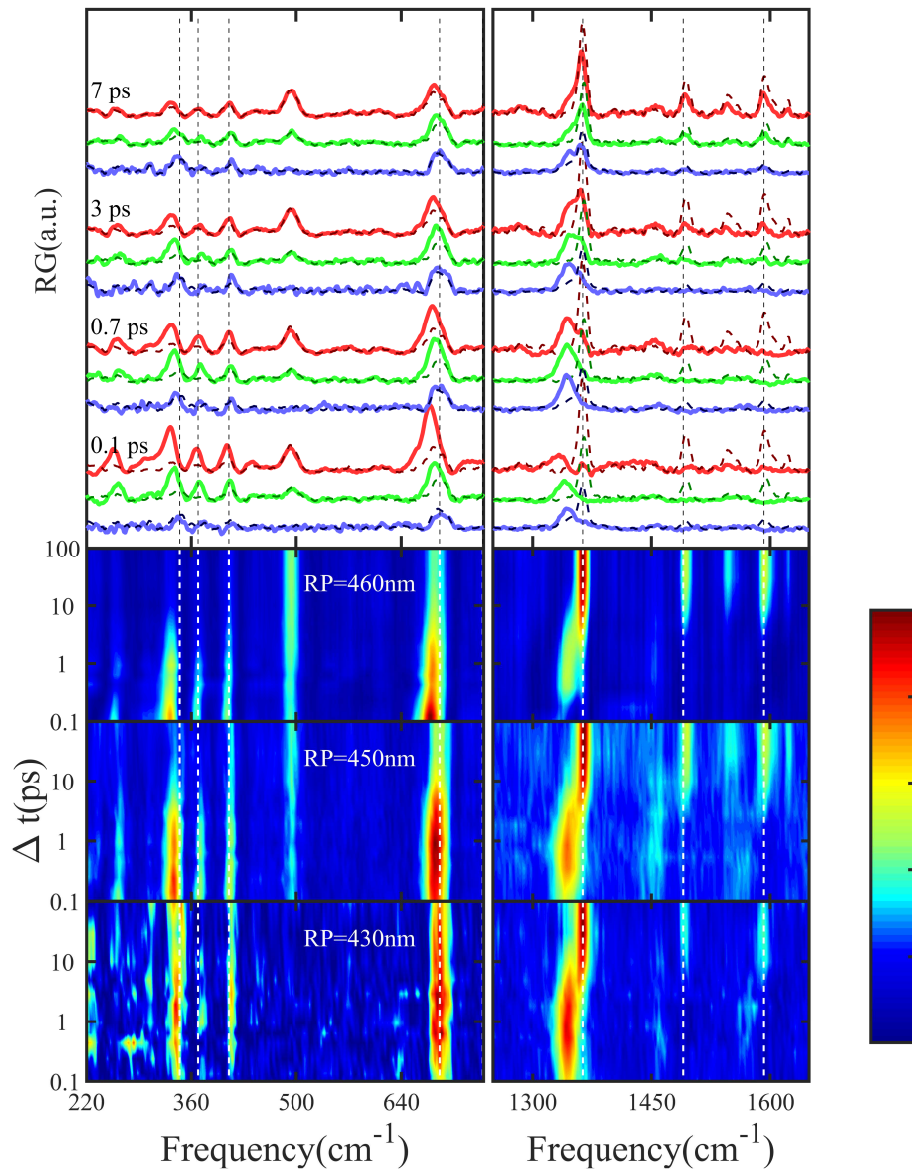


Figure 2: FSRRS spectra of ferrous *cyt c* (solid lines) are shown for selected time delays with the RP at 460 nm (red), 450 nm (green) and 430 (blue). Reference spectra in the absence of the AP are also reported as dashed lines. b, The corresponding FSRRS colour maps are reported for all the measured time delays from 0.1 to 100 ps. The vertical dashed lines are guides to the eye. The unphotoexcited contribution has been removed and probe chirp temporal shifts have been corrected.



The FSRRS measurements of *cyt c* offer a direct way to follow the molecular dynamics. The experimental data, reported in Fig. 2, show two main features: (i) a shift of  $\nu_4$  position from  $1360 \text{ cm}^{-1}$  to  $\sim 1342 \text{ cm}^{-1}$  on the time scale of a few picoseconds and (ii) a shift to lower energy of the vibrational mode up to  $700 \text{ cm}^{-1}$  more pronounced for higher RP wavelengths. The latter effect, more evident during the first 1 ps, can be attributed to an out-of-equilibrium heating of vibrational modes,<sup>17</sup> corresponding to an increase of the population of vibrationally excited states. This implies the appearance in the Raman spectrum of hot bands, i.e. Raman transition from vibrationally excited levels ( $n > 0$ ) to the higher ones ( $n + 1$ ). Hot bands are generally centered at lower wave-numbers with respect to the fundamental bands ( $0 \rightarrow 1$ ), due to anharmonic molecular potentials (a further discussion is reported in the SI), and are more visible for RP wavelengths which are red-shifted with respect to Soret band maximum.<sup>38</sup> On the contrary, the  $1342 \text{ cm}^{-1}$   $\nu_4$  band cannot be ascribed to a hot band of the "fundamental" contribution (at  $\sim 1360 \text{ cm}^{-1}$ ) because it does not undergo an amplitude enhancement for red-shifted RP wavelengths at any of the monitored time delays, and because the picosecond ( $\sim 1360 \text{ cm}^{-1}$ ) band is weaker than the shifted peak also in resonance with the vibrational ground state (at RP=430 nm), incompatible with a thermal distribution of vibrational levels, where the population of the  $n=1$  level can not be larger than that of the  $n=0$  level. Moreover, the shifted peak, reported as a red dashed line in Fig.3, shows a frequency dynamics compatible with the evolution in an out-of-equilibrium configuration (see below) and not with hot band population, which would show a time-independent energy difference between the vibrational levels.

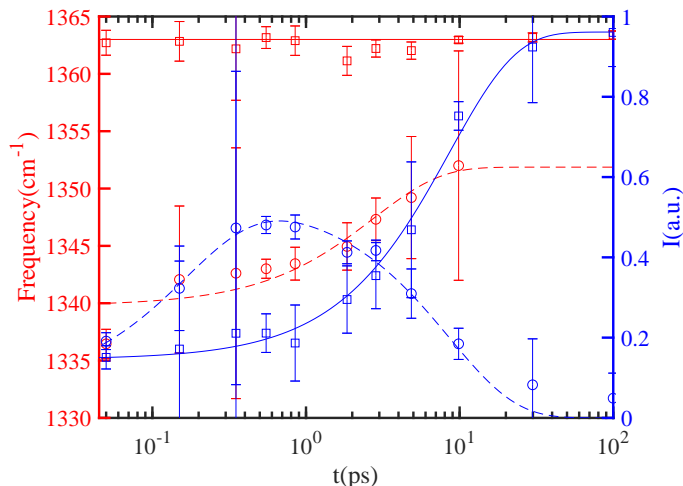


Figure 3: The intensity (blue axis) and frequency (red axis) of  $\nu_4^{gs}$  (squares) and  $\nu_4^{pd}$  (circles) are reported for different time delays, extracted from the measurements with the RP tuned at 450 nm. Time dependent amplitudes and positions are extracted fitting the experimental spectra with Gaussian profiles. The  $\nu_4^{pd}$  position is not reported for time delays with vanishing  $\nu_4^{pd}$  intensity. The recovery of the  $\nu_4^{gs}$  peak amplitude is simultaneous with the disappearance of  $\nu_4^{pd}$ , which shows a frequency shift (red dashed line) of  $\sim 10 \text{ cm}^{-1}$ . The kinetics of the relaxation process are obtained by mono and bi-exponential fits of peak amplitudes and positions (reported as lines). Specifically, the characteristic times are for  $\nu_4^{pd}$  intensity 0.18 and 8.6 ps, for  $\nu_4^{gs}$  intensity 8.6 ps and for  $\nu_4^{pd}$  peak position 2.8 ps.

Considering the recombination time scale of the residue binding,<sup>17,18,49</sup> the spectrally resolved transient 1340-1350  $\text{cm}^{-1}$   $\nu_4$  peak can be assigned to the methionine-photodissociated 5-c heme. Consequently, hereafter we refer to the two contributions as ground state and photodissociated state ( $\nu_4^{gs}$  and  $\nu_4^{pd}$ , respectively).

Interestingly, the same measurement was performed also with TR<sup>3</sup>,<sup>16</sup> where the upshift of the  $\nu_4$  band on the picosecond timescale was assigned to vibrational cooling related to anharmonic coupling of the  $\nu_4$  band to lower-frequency bands. This view is in agreement with the evolution of the  $\nu_7$  mode described here and discussed above. We note that the time constant of the shift was found to be somewhat shorter in the present FSRS work (2.8 ps) than in the TR3 experiments (6.8 ps). This difference is presumably related to the much lower TR3 spectral resolution, which leads to uncertainty in distinguishing the two  $\nu_4$  contributions.

## Neuroglobin

The FSRRS spectra of photoexcited ferrous Ngb, measured with a RP tuned at 460 nm, are reported in Fig. 4 and show at first sight similar trends as those recorded for the cyt *c* case. Evident are the downshift of the of vibrational modes up to  $700\text{ cm}^{-1}$  on the picosecond timescale due to heating and the intense shifted  $\nu_4$  component occurring within the first 10 ps. Interestingly, the transient  $\nu_4$  spectral component has a shape with a distinctly asymmetric red-tail (see the solid line at 0.7 ps of Fig. 4).

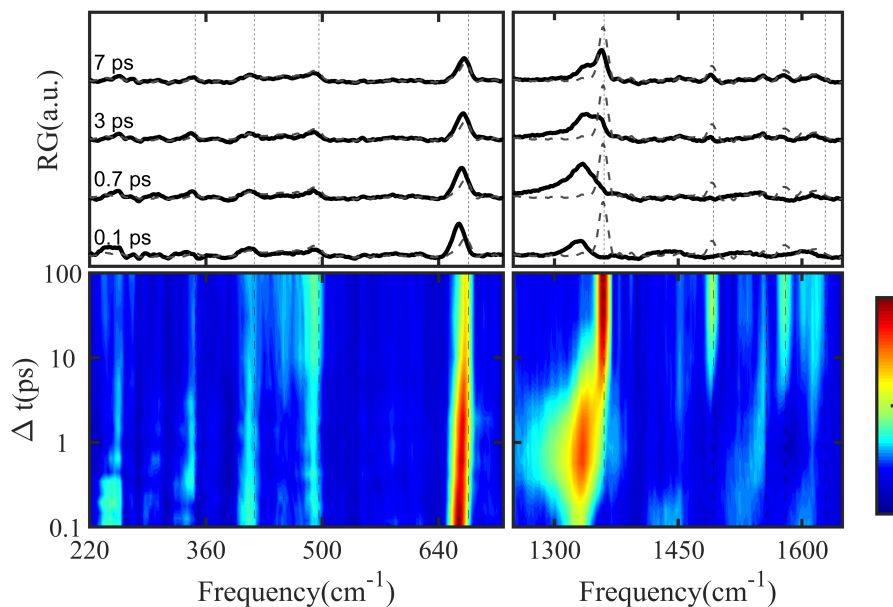


Figure 4: FSRRS spectra of ferrous Ngb (solid lines) are shown for selected time delays. Reference spectra in the absence of the actinic pulse are reported as dashed lines. b, The corresponding FSRRS colour maps are reported for all the measured time delays from 0.1 to 100 ps. The vertical dashed lines are guides to the eye. The RP wavelength is set at 460 nm, the unphotoexcited contribution has been removed and probe chirp temporal shifts have been corrected.

To understand the origin of this contribution we compare the same measurement at 1 ps of time-delay (in which the contribution of  $\nu_4^{gs}$  is negligible) for different RP wavelengths in the red side of Soret band, as shown in Fig.5. The results demonstrate the large dependence of the red tail on the RP wavelength, as expected for the scenario of hot bands.<sup>38</sup> Moreover,

the broad bandwidth of such contribution ( $\sim 50\text{cm}^{-1}$ , highlighted by the red area in Fig.5) indicates a lifetime of the first vibrationally excited state much shorter than that of the ground state.

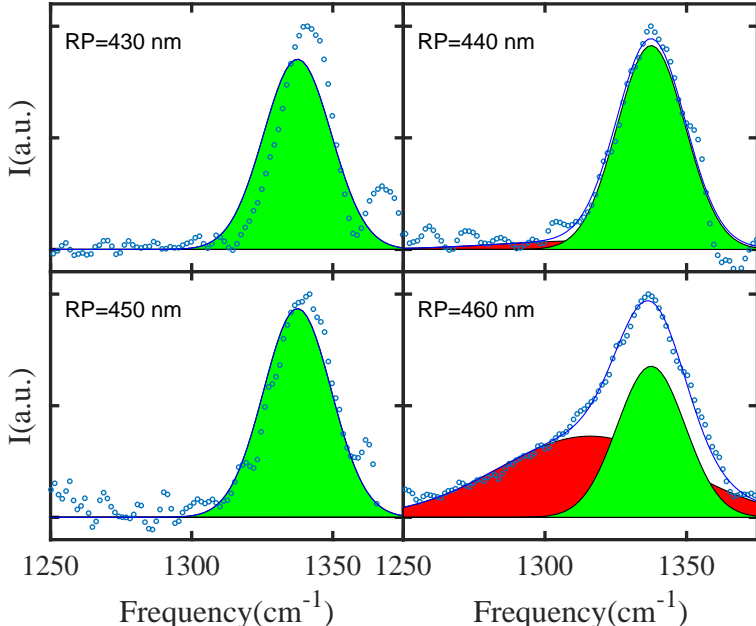


Figure 5: FSRRS  $\nu_4$  spectra of ferrous Ngb (blue circles), after subtraction of the non-excited molecular fraction, at 1 ps and different RP wavelength. For this time delay the experimental data can be fitted (blue line) with two gaussians, that represent  $\nu_4^{pd}$  (green area) and hot band of  $\nu_4^{pd}$  (red area). The contribution of  $\nu_4^{gs}$  is negligible for this time-delay. In agreement with Ref.,<sup>38</sup> the hot-band contribution is emphasized with RP red-shift.

Moreover, the peak positions of  $\nu_4^{pd}$  and hot band of  $\nu_4^{pd}$ , reported in fig. 6, have a similar trend and, consequently, a constant difference of  $\sim 25\text{ cm}^{-1}$ . This implies that during the structural evolution in the photo-dissociated state the anharmonicity factor is almost constant. Assuming a Morse potential, the anharmonicity factor  $\alpha$  is  $\sim 0.0088$  (see SI), 2 times greater than the value extracted for Myoglobin ( $\alpha = 0.004$ ).<sup>38</sup>

Summing up, Ngb shows also a Raman signature of the photodissociated state, in this case for the photolysis of histidine (ref.<sup>18</sup>), with a transient hot  $\nu_4$  vibrational state which decays on the picosecond time scale, analogously to the case of excited deoxy Mb<sup>38</sup> (see SI). The latter feature represents an important difference with respect to cyt *c*, that does not show this heating of the  $\nu_4$  vibrational mode.

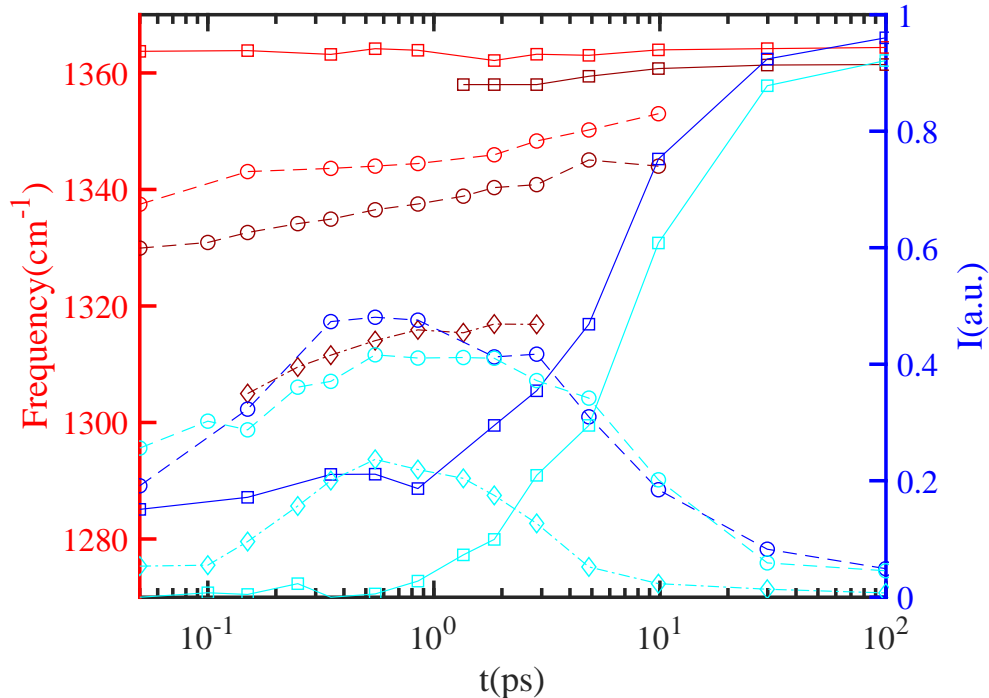


Figure 6: The intensity of cyt *c* (blue lines) and Ngb (cyan lines) and frequency of cyt *c* (red lines) and Ngb (brown lines) of  $\nu_4^{gs}$  (squares),  $\nu_4^{pd}$  (circles) and hot bands of  $\nu_4^{pd}$  (diamonds) are reported for different time delays. Time dependent amplitudes and positions are extracted fitting the experimental spectra with Gaussian profiles. The peak position is reported only for intensity  $I(t)$  larger than  $0.08I(-10\text{ps})$ . The recovery of  $\nu_4^{gs}$  intensity is simultaneous with the disappearance of  $\nu_4^{pd}$ , that has a frequency shift (red dashed line) of  $\sim 10\text{ cm}^{-1}$ .

## Discussion

In the literature photolysis of amino acid residues (i.e. histidine, methionine, etc..) from ferrous hemes has been mainly investigated by TA spectroscopy.<sup>18</sup> The only Raman evidence of bond breaking obtained by ps-time-resolution spontaneous TR<sup>3</sup> is the appearance in cyt *c* of the iron-histidine band,<sup>22</sup> following vibrational coherence spectroscopy assignments.<sup>17</sup> Vibrational cooling following photolysis on the picosecond time scale has been studied by TR<sup>3</sup>, but here the poor spectral and temporal resolution limits the obtainable information on the evolution of the dissociated configuration and in particular does not allow to resolve hot bands. FSRRS transient spectra allow to exploit the marker of photolysis in the region

of  $\nu_4$ . Moreover, FSRR's high spectral and temporal resolutions enabled us to disclose marked differences in the structural dynamics and energy flow in the investigated *b*- and *c*-type hemes. The results are summarized in fig. 7.

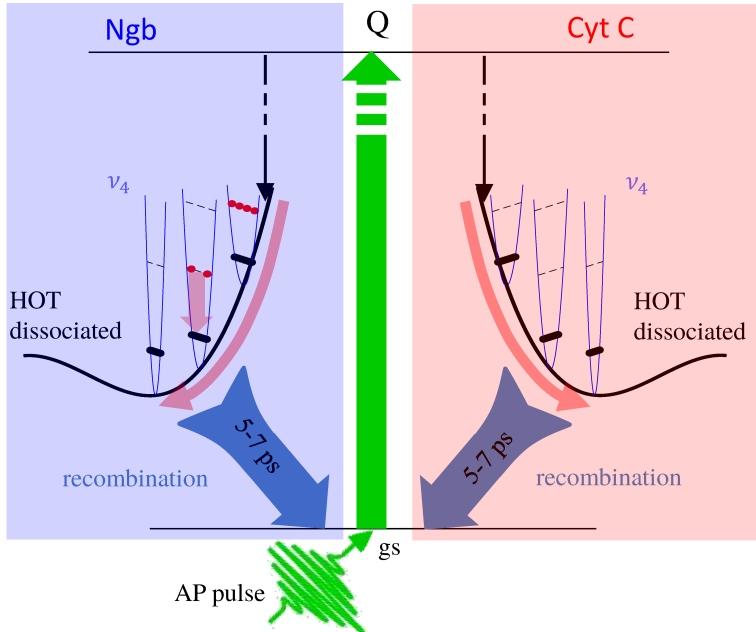


Figure 7: The interaction with the AP pulse promotes the system from the ground state to the Q band in both 6-c molecules. The system decays instantaneously ( $< 50$  fs) into the hot photodissociated state of Ngb (blue area) and cyt *c* (red area), in which the internal residue (histidine for Ngb and methionine for cyt *c*) is detached from the iron atom. In this condition a cooling process and a structural relaxation can be observed. The observation of hot band of  $\nu_4^{pd}$  testifies the population of vibrational excited in Ngb (red circles in  $\nu_4$  potential). The last step is the recovery of the ground state in 5-9 ps with the binding of the residue.

## Structural dynamics

In two 6-c molecules examined, the frequency position of  $\nu_4^{gs}$  is very similar and also the dynamics of the  $\nu_4^{pd}$  are practically equivalent, as quantified by red and brown dashed lines in fig. 6. Particularly, after an instantaneous softening of  $\nu_4^{pd}$  mode, a partial recovery toward the gs frequency is observed before the full relaxation to the 6-c ground state. This recovery, quantified in fig. 3, can be interpreted in two different ways: anharmonic mode coupling with a low frequency mode<sup>4</sup> or a picosecond geometrical reorganization able to accommodate

the recombination of the distal ligand after its instantaneous photolysis.<sup>36</sup> Notably, the observation of the  $\nu_4^{pd}$  hot band only for the Ngb case suggests a  $\nu_4$  coupling different for the two 6-c systems,<sup>38</sup> which would imply different trends (which was not observed) if the  $\nu_4^{pd}$  peak shifts were due to anharmonic coupling. Moreover, in striking contrast with the  $\nu_4^{pd}$  dynamics measured in Mb, where the anharmonic coupling induces an initial  $\approx 4 \text{ cm}^{-1}$  peak red-shift followed by a recovery,<sup>38</sup> the  $\nu_4^{pd}$  peak measured in both the 6-c molecules shows a strong ( $> 10 \text{ cm}^{-1}$ ) and monotonic (in time) blue-shift (see fig. S2 of the SI). Taken together, these results point to the presence of a picosecond geometrical reorganization. Therefore, we can deduce that the picosecond geometrical reorganization sensed by the  $\nu_4$  vibrational coordinate is a general feature in the process of bond breaking/reformation of ferrous heme with an internal residue.

## Energy flow

The hot bands dynamics, observed in the two 6-c systems, show an out-of-equilibrium energy distribution in the photo-dissociated intermediate state, with a key role of the vibrational modes up to  $700 \text{ cm}^{-1}$ , as in the case of 5-c Mb.<sup>38</sup> Specifically, these hot bands are the signature of a high temperature distribution of vibrational excited states and appear in less than 100 fs, testifying the crucial role of these modes as thermal receptors. Moreover the picosecond recovery of the cold state highlights their efficiency in the flowing of thermal energy out of the heme via coupling to low-frequency (delocalized) modes of the protein environment. The scenario is quite different for high frequency modes and, specifically, for the  $\nu_4$  band. In the case of 5-c Mb a picosecond trapping of thermal energy in the  $n=1$  level of the  $\nu_4$  mode is observed. Starting from 6-c heme systems a 5-c hot heme state is also populated, albeit with less vibrational energy due to the energy required for dissociating the Fe-residue bond (which is yet only a small fraction of the energy of the absorbed photon, at least for cyt *c*<sup>50</sup>) and the dynamics is complicated by the photolysis process. The photo-dissociated Ngb shows a very prominent  $\nu_4$  hot band effect very similar

to the Mb case. However, in cyt *c* no population of higher  $\nu_4$  vibrational levels is observed on the same timescale. This finding implies that either these levels don't get significantly populated from the initial broad manifold of vibrational states (ref.<sup>38</sup>) or that they very rapidly ( $< 100$ fs) decay to the  $n=0$  level. Notably, this interpretation implies a most efficient cooling process in cyt *c*, that can explain the shorter time scale of red-tail in TA (150 fs). We assign this striking difference between cyt *c* and Ngb to the fact that the *c*-type heme is not only bound axially to the protein via the heme iron but also via two covalent bonds with cysteine residues at the periphery of the heme plane. Therefore the effective coupling of the  $\nu_4$  in-plane mode to protein modes in the same frequency range will be much stronger and energy redistribution with high-frequency protein modes more efficient. Energy flow has been modelled in a number of heme proteins and several studies have been devoted to cyt *c*.<sup>43,51-54</sup> In comparison with similar studies on myoglobin<sup>52</sup> it was indeed observed that through-bond energy flow plays a much stronger role in cyt *c* in particular by flow through the cross-linked cysteine residues, including on the 100-fs timescale.<sup>55</sup> In a different study, the cysteines were found to respond with a rate  $> 500 \text{ ps}^{-1}$  even when the excess energy was deposited initially only in the close vicinity of the heme iron.<sup>54</sup> These studies thus appear in qualitative agreement with the present experimental work. Mode-resolved theoretical analyses may allow a more detailed comparison with the FSRS data.

## Experimental section

### Sample preparation

The murine deoxy Neuroglobin is prepared dissolving the freeze-dried ferric form<sup>56</sup> in a phosphate buffer, pH 7.4, and the obtained solution is purified into a biochemical basket centrifuge. Sodium dithionite ( $\text{Na}_2\text{S}_2\text{O}_4$ ) is used to reduce the ferric Neuroglobin to ferrous form. The reduction of the protein was done in a nitrogen-purged buffer to avoid exposure to atmospheric oxygen. The concentrations of the samples are between 100 and 200  $\mu\text{M}$ .



Horse-heart cyt *c* was purchased from Sigma and prepared as described above for Neuroglobin.

The sample is allowed to flow anaerobically through the transmission cell during the experiment due to a peristaltic pump, so as to guarantee a fresh sample at every laser shot (1 KHz). All Raman measurements were performed at room temperature. The transmission cell is mounted onto an upright motorized translator, to allow regular movements to minimize window damaging by the three beams.

## FSRS setup

A detailed description of our FSRRS setup has been given elsewhere.<sup>30,39,57</sup> The basic concept is to use a  $\sim 1$   $\mu$ J-energy, 50 fs-time-duration AP tuned in the heme Q-band (550 nm) to photo-excite the system, triggering the dynamics of interest. The system evolution is then monitored by a couple of a narrowband tunable Raman pulse (RP), which provides the spectral resolution, and a broadband femtosecond probe pulse (PP), which guarantees the time precision. The joint action of temporally and spatially overlapped RP and PP allows to record Stimulated Raman scattering (SRS) spectra, coherently generated on top of the highly directional PP and hence essentially free from fluorescence background. The TA measurements, recorded in absence of the RP, have been performed with the same pulse properties.

## Conclusions

In conclusion, the observation of Raman evolution in 6-c heme proteins with high spectral and temporal resolution allows to discriminate a new marker of photolysis related to a red shifted contribution of  $\nu_4$  Raman mode. In addition, the dynamics of hot band contribution unveils the pathway of thermal relaxation in such macromolecules. In particular, we detected an high out equilibrium condition in the photolysed state, consisting in a high temperature of

low vibrational modes (up to  $700\text{ cm}^{-1}$ ). Interestingly in Ngb a trapping of thermal energy in the  $n=1$  level is observed for the vibrational coordinate of  $\nu_4$ , but not in cyt *c*. This finding presumably finds its structural origin in the different heme-protein-bonding in these proteins and may be related to the difference in rigidity required for the respectively small molecule- and electron-transfer functions of these proteins. After the recombination of the internal ligand, the systems are fully relaxed in the gs, without any residual thermal heating effect.

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## Supporting Information Available

The following files are available free of charge.

- SupplementaryMaterial.pdf: supporting information with (i) anharmonicity factor extracted from the FSRRS spectra and (ii) comparison between five-coordinate (Mb) and six-coordinate (Ngb and cyt *c*)  $\nu_4$  peak dynamics.

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