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▶ To cite this version:

Eduardo Paredes, Veronique Malard, Claude Vidaud, Emilie Avazeri, Richard Ortega, et al.. Isotopic variations of copper at the protein level in neuronal human cells exposed in vitro to uranium. Analyst, 2019, 144 (20), pp.5928-5933. 10.1039/C9AN01081E . hal-02447119

HAL Id: hal-02447119 https://hal.science/hal-02447119v1

Submitted on 21 Jan 2020 $\,$

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1	Isotopic variations of copper at the protein level in
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28 Abstract

29 The study of isotopic variations of endogenous and toxic metals in fluids and tissues is a recent 30 research topic with an outstanding potential in biomedical and toxicological investigations. Most of the analyses have been performed so far in bulk samples, which can make the 31 interpretation of results entangled, since different sources of stress or the alteration of different 32 metabolic processes can lead to similar variations in the isotopic compositions of the elements 33 34 in bulk samples. The downscaling of the isotopic analysis of elements at the sub-cellular level, is considered as a more promising alternative. Here we present for the first time the accurate 35 determination of Cu isotopic ratios in four main protein fractions from lysates of neuron-like 36 human cells exposed in vitro to 10µM of natural uranium for seven days. These protein 37 38 fractions were isolated by Size Exclusion Chromatography and analysed by Multi-Collector 39 Inductively Coupled Plasma Mass Spectrometry to determine the Cu isotopic variations in each protein fraction with regard to the original cell lysate. Values obtained, expressed as $\delta^{65}Cu$, 40 were -0.03 ± 0.14 ‰ (U_c, k=2), -0.55 ± 0.20 ‰ (U_c, k=2), -0.32 ± 0.21 ‰ (U_c, k=2) and $+0.84 \pm 0.01$ 41 42 0.21 % (U_c, k=2) for the four fractions, satisfying the mass balance. The results obtained in this 43 preliminary study pave the way for dedicated analytical developments to identify new specific 44 disease biomarkers, to get insight into the knowledge of stress-induced altered metabolic 45 processes, as well as to decipher metabolic pathways of toxic elements.

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56 Introduction

The study of the isotopic signatures of endogenous elements, such as Cu, Zn, and Fe, in body 57 58 fluids/ tissues and in vitro cultured human cell lines is a recent research topic which has shown an outstanding potential for biomedical investigations. (1,2,3) In particular, Cu and Zn isotopic 59 signatures seem to be promising tools for the diagnosis of cancer (4, 5, 6, 7) and 60 neurodegenerative diseases, (8,9,10,11) as well as for the follow-up of patients. (12) In vitro 61 cultured human cell lines are powerful tools to help identifying the altered metabolic processes 62 leading to disease-induced isotopic variations, (13) as well as performing other metabolic or 63 64 toxicological studies. (14,15,16,17) Isotopic fractionations may occur during the redistribution 65 of an element among different chemical species, as predicted by *ab initio* calculations. (18) The stress-induced alteration of the metabolic processes involving any of these chemical species 66 may lead to a modification of the element isotopic signatures. Since the altered metabolic 67 68 processes may depend on the source of stress (disease, toxic element, etc), the chemical species 69 undergoing isotopic signature variations may differ. Most of the studies performed so far have 70 been based on the isotopic analysis of the element in the bulk sample, but two articles recently 71 published (19,20) have demonstrated the potential of the determination of the isotopic 72 signatures at the sub-cellular level.

73 We have previously developed a procedure for the accurate isotope ratio determination of U, Zn 74 and Cu in bulk SH-SY5Y human neuroblastoma cell samples, after differentiation of the cells 75 into neuron-like cells and exposure to low concentrations of natural U for seven days. (21,22) In 76 the current study, isotopic analysis was downscaled at the protein level for the first time. To 77 meet this aim, additional analytical efforts were required to isolate the protein fractions 78 containing U, Zn and Cu, in combination with accurate isotope ratio measurements of the very 79 small amounts of elements contained in these fractions. In these sense, we developed a method 80 dedicated to the isotopic analysis of Cu, Zn and U in different protein fractions of cell lysates 81 obtained by Size Exclusion Chromatography (SEC), after exposure of SH-SY5Y neuron-like cells to 10 µM of natural U for seven days. Only Cu isotope ratio results are presented, since it 82

was not possible to accurately determine the isotope ratios of Zn and U in the protein fractions,as discussed in the following.

85

86 **Experimental**

87 **Reagents and solutions**

All the aqueous solutions were prepared using ultrapure water (resistivity > 18.2 M Ω cm at 25 88 °C) from a Milli-O[®] system (Millipore). Eagle's minimum essential medium (EMEM, ATCC, 89 30-2003), F12 medium (Life Technologies, 21765-029), fetal bovine serum (FBS, ATCC, 30-90 91 2020) and penicillin/ streptomycin (Gibco-Thermo Fisher Scientific, 15070-063) solutions were 92 used to prepare the culture medium for cell growing and exposure experiments. Retinoic Acid (RA, Sigma-Aldrich, R2625) and 12-O-tetradecanoylphorbol-13-acetate (TPA, Sigma Aldrich, 93 P8139) were used for cell differentiation. TrypLE Express 1X/EDTA (Gibco-Thermo Fisher 94 95 Scientific, 12605-010) was used for cell trypsinization. Phosphate buffer saline (PBS, pH 7.4) 96 free of CaCl₂ and MgCl₂ (Gibco 10010-015) was used to wash the cells after trypsinization. A 100mM ammonium acetate (Normapur grade, VWR Prolabo) solution in ultrapure water was 97 98 used as mobile phase for the SEC experiments.

99 Plasma Pure Plus 34-37% HCl and Plasma Pure Plus 67-70% HNO3 ultrapure reagents (SCP 100 Science) were used for the sample preparation steps and for the preparation of solutions analyzed by Multi-Collector Inductively Coupled Plasma Mass Spectrometry (MC-ICPMS). 101 IRMM-3702 $[n(^{66}Zn)/n(^{64}Zn) = 0.56397 \pm 0.00030 (U_c, k=2)]$ and ERM[®]-AE633 102 $[n(^{65}Cu)/n(^{63}Cu) = 0.44563 \pm 0.00042 (U_c, k=2)]$ isotopic certified reference materials (i-CRM) 103 104 traceable to SI units were purchased from the Institute for Reference Materials and Measurements (IRMM, Geel, Belgium). The ERM®-AE633 i-CRM was used to prepare a 105 106 bracket solution analyzed just before and after the samples for Cu isotope ratio measurements. 107 This bracket solution and the samples were spiked with the IRMM-3702 i-CRM, which was 108 used as an internal standard for the correction of mass discrimination.

In-house natural uranium powder (U_3O_8) was dissolved in 0.5 mol L⁻¹ HNO₃ to prepare a natural 109 uranium stock solution at 150 mM. An intermediate U solution (pH = 8-8.5) was then prepared 110 by 1:5 v/v dilution of this solution in a buffer solution containing 0.1 mol L^{-1} NaHCO₃ 111 (analytical reagent, Normapur), 0.1 mol L⁻¹ Na₂CO₃ (99.95% extra pure, Acros Organics), 0.15 112 mol L⁻¹ NaCl (puriss. p.a., Sigma-Aldrich), and 0.05 mol L⁻¹ TRIS (ultrapure grade \geq 99.9%, 113 114 Sigma-Aldrich) in ultrapure water. This dilution was performed by dropwise addition of the U 115 stock solution into the buffer solution to avoid U precipitation. Finally, the exposure solutions 116 (10 µM of U) were prepared by diluting the intermediate U solution into an appropriate volume 117 of the culture medium consisting in an equal mix of EMEM and F12 mediums supplemented with 10% FBS and 1% penicillin/ streptomycin. 118

119 Differentiation of cells, exposure to natural uranium and cell lysis.

120 Human SH-SY5Y (ATCC, CRL-2266, Batch 59740436) cells were first grown and differentiated into neuron-like cells. Cells were grown in 175 cm² flasks at 37°C in 5% CO₂ for 121 10 days. The culture medium was replaced with fresh medium every 3-4 days. Afterwards, the 122 123 cells were passaged by trypsinization using TrypLE Express 1X/EDTA and then seeded at 25000 cells cm⁻². The differentiation into neuron-like cells was performed according to the 124 125 method developed by Presgraves et al. (23) For this, the culture medium was replaced with 126 fresh medium containing 10 μ M RA, and cells were incubated for 3.5 days. This solution was 127 then replaced with fresh medium containing 80 nM TPA and left for 3.5 days again. The 128 effectiveness of the cell differentiation was assessed by phase contrast microscopy and western 129 blot showing neurite outgrowths and tyrosine hydroxylase expression in differentiated cells. 130 (24)

The differentiated cells were exposed to freshly-prepared 10 µM U exposure solutions for 7
days, with exposure solution renewal once after 3 days. After exposure, cells were trypsinized,
collected, washed twice with PBS, and finally counted using a Vi-CELL XR 2.04 cell viability
analyzer (Beckman Coulter).

135 Cell proteins were extracted using a mechanical method. For this, cells were suspended at 136 30×10^6 cells per mL in 0.02 M TRIS pH 7.5 containing anti-proteases (Roche cocktail-EDTA free) and 0.002 M spermine base (Sigma). The cells were disrupted by one shot at 1000 bars using a cell disruption system (Constant Systems) and then incubated for 30 min at 4 °C. The extracts were then ultra-centrifuged (1 h at 100,000 g) and the supernatants collected. Aliquots of the supernatants were stored at -20 °C until use.

141 Separation and detection of metal enriched protein fractions

142 A Smartline chromatographic system (KNAUER, Berlin, Germany) equipped with a dual Pump 143 1000 was used for the chromatographic separation of metal enriched protein fractions by SEC. 144 The column was a 4.6 x 150 mm BEH 200 (Waters, Milford, USA) with 1.7 µm particle size 145 and 200 Å pore size. The mobile phase consisted in a 100 mM ammonium acetate solution in ultrapure water, and the flow rate was 0.3 mL min⁻¹. Cell lysates (20 µL) were manually 146 147 injected in the column with a Rheodyne injector valve (model 9725). The column was coupled 148 to a UV detector (Smartline PDA detector 2800) connected in series to a quadrupole ICPMS (q-149 ICPMS, X7, Thermo) through a 1-meter length PEEK tubing (i.d: 125 µm). The monitoring of metal enriched protein fractions was performed by UV at $\lambda = 280$ nm, and by q-ICPMS by 150 following the signal of the isotopes ${}^{65}Cu^+$, ${}^{66}Zn^+$ and ${}^{238}U^+$. The integration time per 151 152 chromatographic point in q-ICPMS was 10 ms, and approximately 5800 points per isotope were 153 acquired for a run time of 10 min.

To estimate the molecular weight (MW) range of the protein fractions, the calibration of the column was performed by injection of 10 mg L⁻¹ standards of Aprotinin (6.5 kDa), Ovalbumin (43 kDa), Conalbumin (75 kDa), Monoclonal antibody (150 kDa), Ferritin (440 kDa) and Thyroglobulin (669 kDa) diluted in the mobile phase. The signal was monitored by UV at $\lambda =$ 280 nm and the retention time (average of 3 replicates) was determined.

For the collection of the main protein fractions, 100 injections of cell lysates (20 μ L) were performed, and the 100 eluate volumes corresponding to the same Cu protein fraction (F1, F2, F3 and F4) were pooled before Cu isotopic analysis. For comparison, 500 μ L of cell lysates were subjected to isotopic analysis without protein separation step. For matrix matching between the original lysate and the Cu protein fractions, the original lysate was diluted in the appropriate volume of the mobile phase before sample preparation for isotopic analysis. Furthermore, procedural blanks were prepared by subjecting the same volume of the mobilephase to the analytical procedure.

167 Sample preparation for isotopic measurements

168 Sample preparation and analysis were performed at the Laboratoire de développement 169 Analytique, Nucléaire, Isotopique et Elémentaire (LANIE, CEA). The labware used for sample 170 preparation previous to isotopic measurements was systematically pre-washed with acid 171 solutions in accordance with clean lab practices. A 120 x 52 x 70 cm glovebox (Ateliers de 172 Technochimie, Ivry sur Seine, France) was purposely designed to perform all sample 173 preparation steps in order to protect samples from atmospheric contamination. This glovebox 174 was made of polyvinyl chloride (PVC) and was equipped with High Efficiency Particulate Air 175 (HEPA) filters to avoid contamination by dust particles.

The sample preparation for the procedural blanks (mobile phase) and the samples (protein fractions and cell lysate diluted in the mobile phase) was similar to that published elsewhere. (*21*) First, the samples/ procedural blanks were evaporated at 80 °C on a heating block. The residues were then acid digested in 15 mL closed Savillex vessels at 85 °C for 2 h after the addition of 1 mL of 67-70% HNO₃. The acid was then evaporated at 85 °C until dryness and the residue was re-dissolved in 1 mL of 3 M HNO₃ for the purification of U, Cu and Zn according to a protocol described elsewhere. (*21*)

183 Isotope ratio measurements

All isotope ratio measurements were performed with a Neptune Plus MC-ICPMS (Thermo Fisher Scientific, Darmstadt, Germany) equipped with 9 Faraday detectors fitted to $10^{11} \Omega$ resistors. The sample introduction system consisted of an OpalMist nebulizer at around 10 µL min⁻¹ (Glass Expansion, Melbourne, Australia) coupled to an Apex HF desolvation system (Elemental Scientific, Omaha, USA). 'Jet' sampler and X-type skimmer cones adapted to dry plasma conditions (25) were employed. The cup configuration used can be found elsewhere. (21)

191 Preamp gain calibration was performed daily. The Zn hydride formation rates were measured by 192 monitoring the signal at m/z 69 (68 Zn¹H⁺) at the beginning of each measurement session. ZnH⁺ formation rates were 0.005-0.008% and the interference of ${}^{64}Zn^{1}H^{+}$ on ${}^{65}Cu^{+}$ was systematically corrected. A background correction was performed by running a 2% HNO₃ solution before the samples, procedural blanks and bracketing solutions. Three procedural blanks were run at the beginning of the session and the average signal measured for the procedural blanks was subtracted to the signal of the samples.

The so-called modified sample-standard bracketing (m-SSB) approach, (21, 26) which combines 198 199 the classical sample-standard bracketing and the inter-element correction approaches, was used to determine the δ^{65} Cu values. For this, the samples were bracketed by ERM[®]-AE633 solutions 200 201 and the Cu concentrations in the samples and the bracketing solutions were matched to less than 202 50 % difference. Both the samples and the bracketing solutions were spiked with IRMM-3702 at the same Zn (internal standard) concentration. For the determination of δ^{65} Cu in a sample, 203 $(\delta^{65}Cu)_{analyte}$ and $(\delta^{66}Zn)_{spike}$ were calculated using a classical sample-standard bracketing 204 205 as shown in equations 1 and 2:

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$$(\delta^{65}Cu)_{analyte} = \left(\frac{(r_{65/63})_{sample}}{(r_{65/63})_{bracketing}} - 1\right) x1000$$
(1)

207
$$(\delta^{66}Zn)_{spike} = \left(\frac{(r_{66/64})_{sample}}{(r_{66/64})_{bracketing}} - 1\right) x 1000$$
(2)

Where $(r_{65/63})_{sample}$ and $(r_{66/64})_{sample}$ are the measured ${}^{65}Cu^{+}/{}^{63}Cu^{+}$ and ${}^{66}Zn^{+}/{}^{64}Zn^{+}$ ratios 208 in the sample, whereas $(r_{65/63})_{bracketing}$ and $(r_{66/64})_{bracketing}$ are the average of the 209 measured ${}^{65}Cu^+/{}^{63}Cu^+$ and ${}^{66}Zn^+/{}^{64}Zn^+$ in the two bracketing solutions run just before and after 210 the sample. The final δ^{65} Cu in the sample (protein fractions or original lysate) with regard to the 211 ERM[®]-AE633 solution was obtained by subtracting $(\delta^{66}Zn)_{spike}$ to $(\delta^{65}Cu)_{analyte}$. With 212 these results, the $n({}^{65}Cu)/n({}^{63}Cu)$ isotope ratio in the samples was calculated. Finally, results 213 were expressed as $(\delta^{65}Cu)_{Fi}$, corresponding to the relative difference in parts per mil between 214 the $n({}^{65}Cu)/n({}^{63}Cu)$ isotope ratio in the protein fraction i and the $n({}^{65}Cu)/n({}^{63}Cu)$ isotope ratio in 215 216 the original lysate:

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$$(\delta^{65}Cu)_{Fi} = \left(\frac{(r_{65/63})_{fraction\,i}}{(r_{65/63})_{original\,lysate}} - 1\right) x1000$$
(3)

218 Where $(r_{65/63})_{fraction i}$ is the $n({}^{65}Cu)/n({}^{63}Cu)$ isotope ratio determined in the protein fraction i 219 and $(r_{65/63})_{original \ lysate}$ is the $n({}^{65}Cu)/n({}^{63}Cu)$ isotope ratio determined in the original lysate. 220 The methods used for U and Zn isotope ratio measurements can be found elsewhere (21) and are 221 not described in this article, since only Cu isotope ratios could be accurately determined in the 222 protein fractions.

223 Uncertainty estimation

The expanded uncertainty (U_c , k=2) of δ values was estimated by quadratic propagation of two sources of uncertainty (see ref. 21 for details): the within-day measurement reproducibility (SD_M) and the reproducibility associated to procedural blank correction (SD_{PB}), as shown in the following equation:

 $U_c = 2\sqrt{SD_M^2 + SD_{PB}^2} \tag{4}$

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230 **Results and discussion**

231 Under the experimental conditions used, the protein fractions from lysate injections eluted in 232 less than 8 min, as shown in the UV profile of Figure 1. The column was previously calibrated 233 with UV monitoring of standard proteins, which made it possible to assign the theoretical 234 masses of the Cu, Zn and U protein fractions from their retention times (peak maxima). The 235 different metal-containing protein fractions observed in the q-ICPMS chromatograms (Figure 1) 236 corresponded to theoretical masses of approximately > 600, 110, 32 and 6 kDa for Cu, > 600, 33237 and 6 kDa for Zn, and > 600, 70 and 3-6 kDa for U, respectively. We estimated the relative 238 proportions of metal in each one of the peaks by comparing the peak areas corresponding to the 239 main Cu, Zn and U peaks from 14 SEC-q-ICPMS chromatograms. These relative Cu 240 proportions for the peaks corresponding to > 600, 110, 32 and 6 kDa were 38 ± 5 % (SD, n=14), 241 20 ± 4 % (SD, n=14), 22 ± 3 % (SD, n=14) and 21 ± 4 % (SD, n=14), respectively. The relative 242 Zn proportions for the peaks corresponding to > 600, 33 and 6 kDa were 39 ± 5 % (SD, n=14), 42 ± 5 % (SD, n=14) and 18 ± 6 % (SD, n=14), respectively. A larger variability was found for 243 244 the U peaks, being the relative U proportions for the peaks corresponding to > 600, 70 and 3-6

245 kDa of 50 ± 13 % (SD, n=14), 36 ± 16 % (SD, n=14) and 13 ± 4 % (SD, n=14), respectively.

The proteomic analysis of these protein fractions was carried out with the aim of identifying theU-target proteins, and the results are reported elsewhere. (27)

248 Four protein fractions F1, F2, F3 and F4 (dotted rectangles, Figure 1) corresponding to the main 249 Cu protein fractions were collected. For Zn and U isotope ratio measurements, F2 and F3 were 250 mixed after the element purification, since they corresponded to the same Zn and U peaks. 251 However, the accurate determination of the U isotope ratios was not possible because the 252 recovered U amounts were too small. In the case of Zn, the amounts in the protein fractions 253 were high enough for precise isotope ratio measurements. However, the Zn isotope ratios could 254 not be accurately determined because of the high procedural blanks obtained (from 200 to 350 255 ng), corresponding to over 50% of the total Zn amounts in the samples. These high procedural 256 blanks cannot be attributed to the sample preparation procedure for isotope ratio measurements, 257 which were found to be 35 ± 19 ng, (21) and probably came from the mobile phase reagents 258 and/or the stationary phase. Indeed, high base lines were observed for Zn in the q-ICPMS 259 chromatograms (Figure 1).

260 On the contrary, accurate results could be obtained for the Cu isotope ratios. The Cu amounts 261 were high enough for the measurement of isotope ratios with external precisions better than 262 0.15‰, and the uncertainty associated to the procedural blank correction ranged from 0.08 to 263 0.17% (2SD, n = 6, see ref. 21 for details on the calculation method). Figure 2 shows the 264 $(\delta^{65}Cu)_{Fi}$ in the 4 protein fractions, reflecting the isotopic signature of Cu in each of the 265 fractions relative to the original lysate. Fraction 1 showed a similar isotopic signature as the original lysate with $(\delta^{65}Cu)_{F1} = -0.03 \pm 0.14 \%$ (U_c, k=2), whereas fractions 2 and 3 were 266 depleted in ${}^{65}Cu$ with $(\delta^{65}Cu)_{F2} = -0.55 \pm 0.20 \%$ (U_c, k=2) and $(\delta^{65}Cu)_{F3} = -0.32 \pm 0.21 \%$ 267 (U_c, k=2), respectively. Finally, fraction 4 was enriched in ${}^{65}Cu$ with $(\delta^{65}Cu)_{F4} = +0.84 \pm 0.21$ 268 $(U_c, k=2)$. The mass balance was verified from these results using equation 5: 269

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$$(\delta^{65}Cu)_{original\ lysate} = \frac{f_1(\delta^{65}Cu)_{F_1} + f_2(\delta^{65}Cu)_{F_2} + f_3(\delta^{65}Cu)_{F_3} + f_4(\delta^{65}Cu)_{F_4}}{100}$$
(5)

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Where $(\delta^{65}Cu)_{F1}$, $(\delta^{65}Cu)_{F2}$, $(\delta^{65}Cu)_{F3}$ and $(\delta^{65}Cu)_{F4}$ are the δ^{65} Cu in the protein fractions 1 to 4 with regard to the original lysate, and f_1 , f_2 , f_3 and f_4 are the percentages of Cu in the protein fractions 1 to 4 (see above). The $(\delta^{65}Cu)_{original lysate}$ calculated when applying Equation 5 was -0.02 ± 0.12 ‰, which is not significantly different from the expected value of 0 ‰.

278 The results obtained in this work clearly demonstrate a different isotopic distribution of Cu 279 among the four protein fractions. Recently, two papers have demonstrated that different Cu 280 pools of biological samples may show different isotopic signatures. In one of these articles, the 281 Cu isotope ratios were determined in the bulk serum of healthy individuals and alcoholic 282 cirrhosis patients, as well as in the exchangeable + ultrafiltrable (EXCH + UF) Cu fraction of 283 the serum, representing the labile Cu pool, and the non-exchangeable + non-ultrafiltrable 284 (NEXCH + NUF) Cu fraction containing the Cu bound to ceruloplasmin. (19) The results 285 showed a heavier Cu isotope ratio in the EXCH + UF fraction compared to the bulk serum and 286 the NEXCH + NUF fraction of healthy individuals, whereas this difference was not found in the 287 patients, potentially reflecting a labile Cu deregulation linked to the disease. (19) In another 288 article, the Cu isotope ratios were determined in sub-cellular fractions of the SH-SY5Y human neuroblastoma cell line, corresponding to the mitochondria and the rest of the cell lysate, 289 290 demonstrating different isotopic signatures. (20) The present work can be considered as a step 291 forward in the downscaling of Cu isotope ratio determinations at sub-cellular level, since we 292 performed accurate Cu isotope ratio determinations at the protein level. These three studies open 293 new perspectives on the identification of new disease-specific biomarkers. In previous 294 publications, the study of the isotopic signature of the element as prognostic biomarker was 295 performed on bulk samples. The main drawback of such approach is that similar isotopic variations in bulk samples can be induced by different types of diseases. For instance, similar 296 depletions in the heavier 65 Cu isotope have been found in the serum of cancer, (7) cirrhosis (28) 297 298 or Wilson disease (8) patients. Furthermore, in some cases, the isotopic variations observed between the healthy and cancerous tissues do not lead to changes in the blood isotopic
signatures, as observed for Zn isotope ratios in breast cancer patients. (5)

301

302 **Conclusions**

303 For the first time, the accurate determination of Cu isotope ratios was performed in intracellular 304 protein fractions isolated by SEC, which represents a significant progress in the downscaling of 305 isotope ratio determinations from the cellular level to the protein level. We evidenced 306 differences in the Cu isotopic signatures in the four main protein fractions from SH-SY5Y neuron-like cell lysates. Since the Cu isotopic signatures in the proteins might be modified by 307 308 stress sources altering the Cu metabolism, this work paves the way for the identification of new 309 disease biomarkers as well as for the development of new strategies to get insight into the 310 knowledge of stress-induced alteration of Cu metabolic processes, which can be critical in the 311 development of diagnostic tools. However in this study, Zn and U isotope ratios could not be 312 accurately determined in the protein fractions, and dedicated analytical developments are 313 required to extend this approach to other elements than Cu, as well as different types of 314 biological samples.

The upcoming major step is to perform isotopic analysis at the molecular level, through the study of isotopic fractionations of elements involved in different chemical species, in combination with *ab initio* theoretical calculations of these isotopic fractionations among these species. *(18)* This is key information for deeper investigation of stress-induced alteration of metabolic processes at the molecular level, as well as to aid deciphering the metabolic routes of toxic elements in a more specific manner.

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322 **Conflicts of interest**

323 There are no conflicts to declare.

324

325 Acknowledgements

The authors would like to acknowledge the Transversal Toxicology Program run by the CEA DRF (France) and the financial support from this program. E. Paredes would also like to thank the CEA - Enhanced Eurotalents program, co-funded by the European Commission through the Marie Sklodowska-Curie COFUND program under the 7th Framework Program for research and technological development (FP7). The authors also acknowledge the CNRS Interdisciplinary Mission through the PEPS (Projet Exploratoire Premier Soutien) Faidora program (Faibles Doses, Risques, Alertes).

333

334 Figure captions

Figure 1. SEC chromatograms of cell lysates of the SH-SY5Y human neuroblastoma cell line differentiated into neuron-like cells and exposed to 10 μ M of natural U for seven days. The elution profiles correspond to the UV signal monitored at 280 nm and the q-ICPMS signals of 65 Cu⁺, 66 Zn⁺ and 238 U⁺ isotopes. For each q-ICPMS chromatogram, the signal relative to the maximum signal measured for the isotope throughout the chromatogram is plotted. The dotted rectangles show the 4 protein fractions collected for Cu isotopic analysis.

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Figure 2. $(\delta^{65}Cu)_{Fi}$ values for the 4 Cu protein fractions with regard to the original lysate. The vertical bars correspond to the expanded uncertainty of the results (k=2). The plain and dotted red lines indicate the δ^{65} Cu value of the original lysate, $(\delta^{65}Cu)_{original lysate}$, defined as 0 ‰, and its expanded uncertainty.

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