DE LA RECHERCHE À L'INDUSTRIE





CHARACTERIZATIONS AND TREATMENT TESTS OF GRAPHITE WASTE FROM FRENCH GAS-COOLED REACTORS

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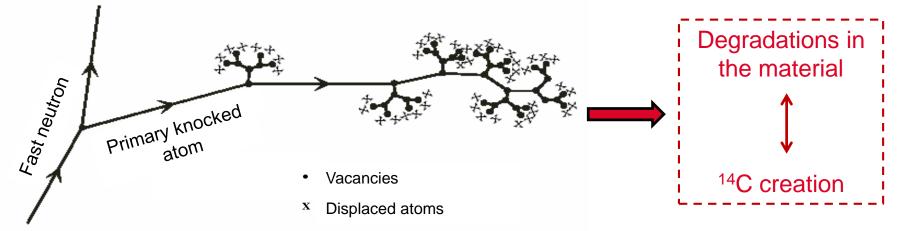
- I Nuclear graphite waste presentation
- II An original treatment proposition for ¹⁴C extraction
- III CO₂ gasification treatment tests

I – GRAPHITE WASTE CONTEXT INEUTRON IRRADIATION EFFECT

Graphite in French gas-cooled reactors

Neutron moderator in first generation of nuclear reactors

Neutron irradiation



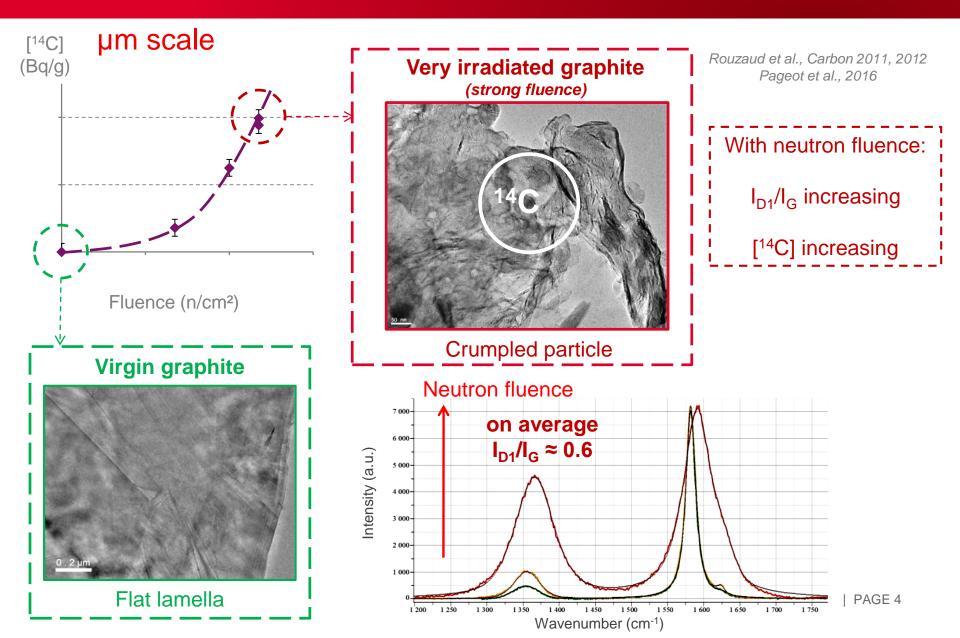
Bonal, Gosmain, 2007

Consequences of dismantling

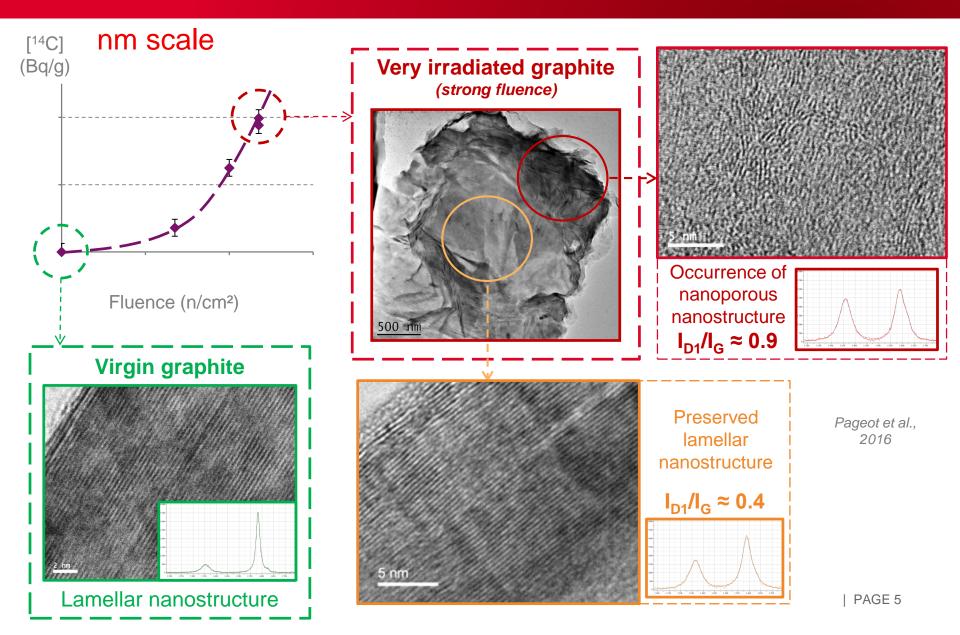
23,000 tons of graphite waste will be generated in France (250,000 tons at the world scale)

- ¹⁴C and ³⁶Cl are still active after a thousand years
 - ¹⁴C will be the main contributor to the activity
- Our aim: to reduce ¹⁴C concentration of graphite waste

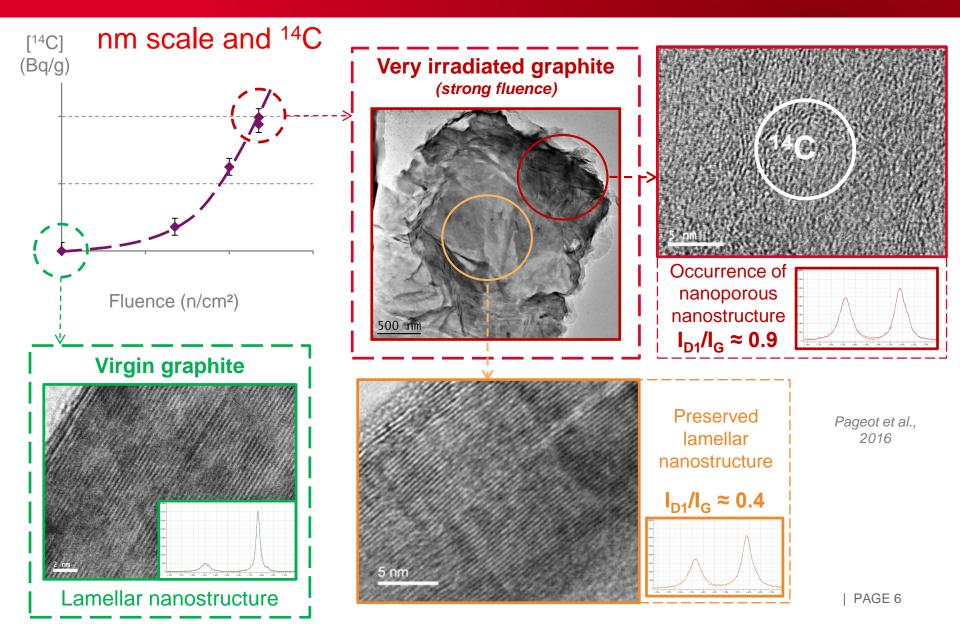
I – GRAPHITE WASTE CONTEXT INEUTRON IRRADIATION EFFECT



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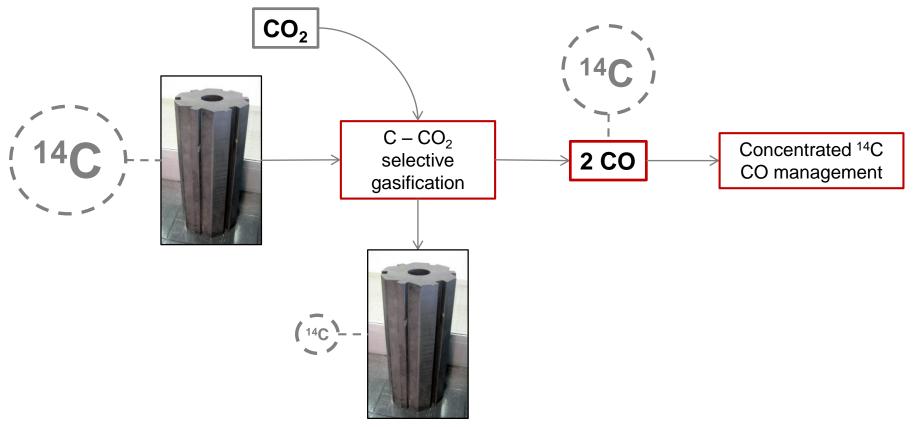
I – GRAPHITE WASTE



II – GRAPHITE WASTE TREATMENT PROPOSITION ■ PRINCIPLE □ C – CO₂ SELECTIVE REACTIVITY

Graphite waste CO₂ gasification treatment proposition

Treatment principle and global vision of the ¹⁴C management

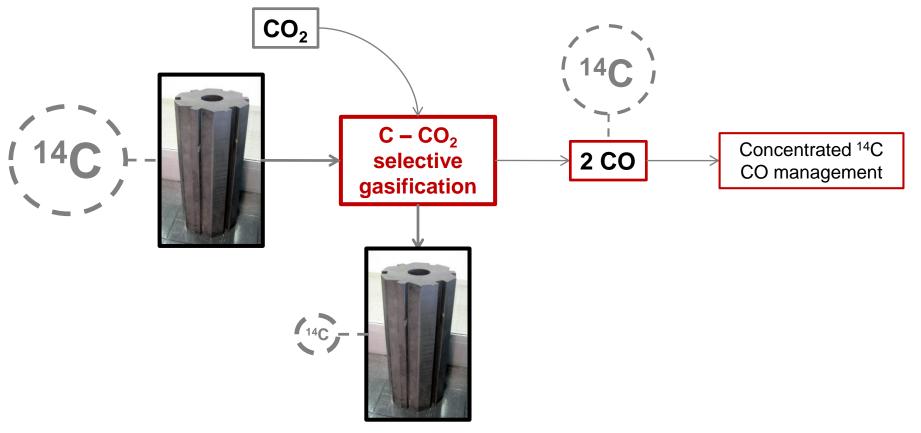


To selectively extract ¹⁴C from graphite stacks with the lowest weight loss as possible To concentrate ¹⁴C into an insoluble carbon

II - GRAPHITE WASTE TREATMENT PROPOSITION ■ PRINCIPLE □ C - CO₂ SELECTIVE REACTIVITY

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 To concentrate ¹⁴C into an insoluble carbon

$\begin{array}{c} \textbf{II} - \textbf{GRAPHITE WASTE TREATMENT PROPOSITION} \\ & \square PRINCIPLE \\ \end{array} \begin{array}{c} \textbf{C} - \textbf{CO}_2 \end{array} \\ \begin{array}{c} \textbf{SELECTIVE REACTIVITY} \end{array}$

Nanostructure and C-CO₂ reactivity

Hypothesis on ¹⁴C localization and nanostructure



After irradiation

 Lamellar nanostructure
 Active sites = edges of sheets
 14C
 Nanoporous nanostructure

 Image: Active site density
 Image: Active site density
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Lamellar and nanoporous areas are deeply associated at the nanostructure scale.

Hypothesis: ¹⁴C should be localized into nanoporous areas.

Rouzaud et al., Carbon 2011

Moneger et al., Eurocarbon 1998 Rodriguez-Reinoso et al., 1974

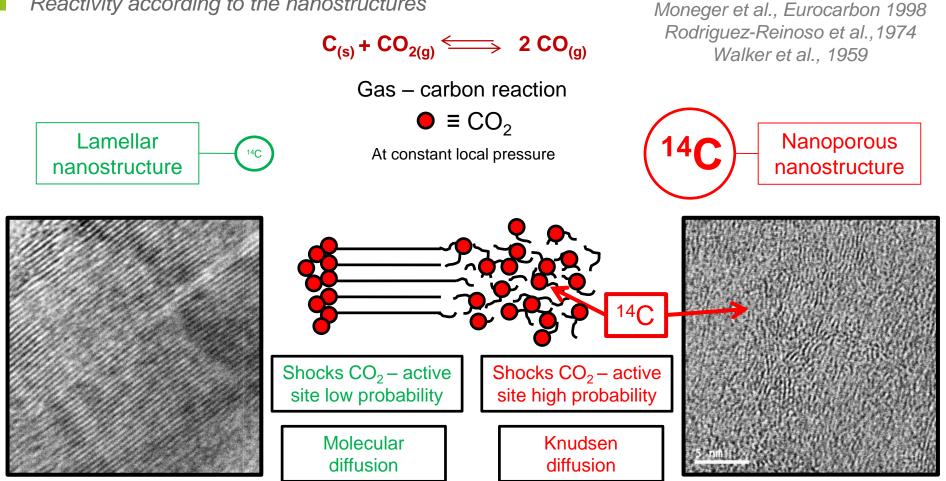
Walker et al., 1959

II – GRAPHITE WASTE TREATMENT PROPOSITION □ PRINCIPLE □ C – CO₂ SELECTIVE REACTIVITY

Rouzaud et al., Carbon 2011

Nanostructure and C-CO₂ reactivity

Reactivity according to the nanostructures

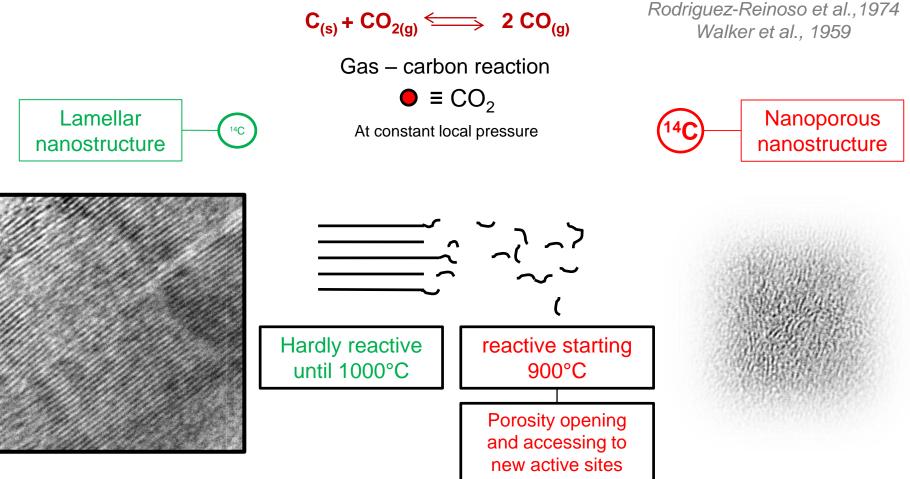


C-CO₂ reaction takes place on edges of graphene sheets and depends on active sites density. PAGE 10 $C-CO_2$ reactivity also depends on the probability of shocks CO_2 – active site.

$\begin{array}{c} II - GRAPHITE WASTE TREATMENT PROPOSITION \\ & \square PRINCIPLE \\ \end{array} C - CO_2 SELECTIVE REACTIVITY \end{array}$

Nanostructure and C-CO₂ reactivity

Selective reaction on nanoporous and ¹⁴C rich areas



Neutron irradiated graphite partial gasification ¹⁴C global concentration decrease

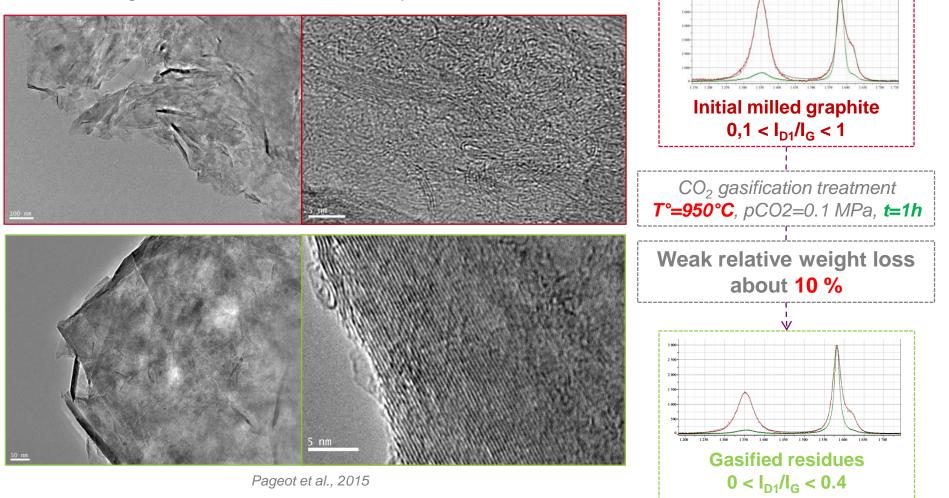
Rouzaud et al., Carbon 2011

Moneger et al., Eurocarbon 1998

III - CO₂ GASIFICATION TREATMENT TESTS ANALOGUES NUCLEAR GRAPHITE WASTE

Irradiated graphite analogue: milled graphite

Similar structure (Raman, TEM) nanostructure (TEM) and heterogeneities
 Advantages: non radioactive and important available volume

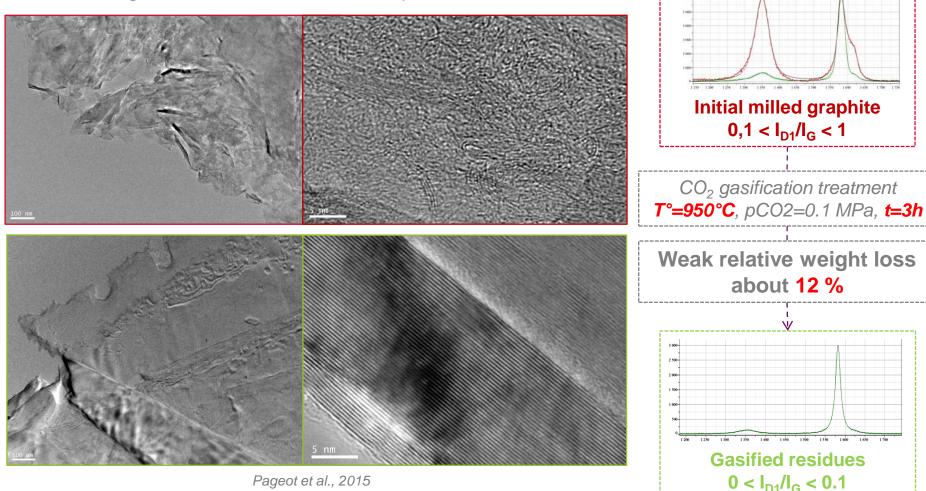


Reactivity experiments carried out at the IS2M laboratory

III - CO₂ GASIFICATION TREATMENT TESTS ANALOGUES NUCLEAR GRAPHITE WASTE

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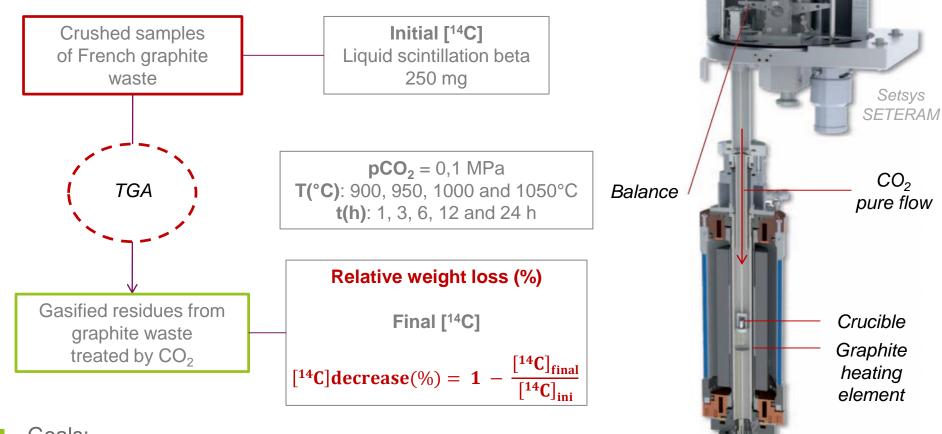


Reactivity experiments carried out at the IS2M laboratory

III - CO2 GASIFICATION TREATMENT TESTS ANALOGUES NUCLEAR GRAPHITE WASTE

Experiments on nuclear graphite waste

A. Duhart-Barone, S. Lutzler and J. Comte (CEA Cadarache) <u>Method:</u>



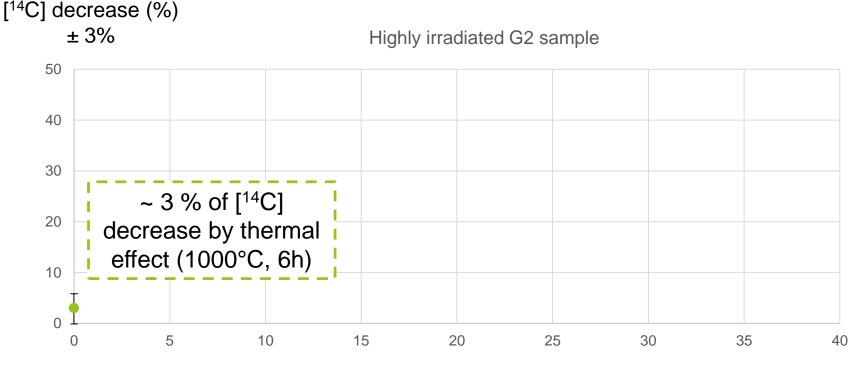
Goals:

Determine the optimal temperature in our conditions

Gasification duration effect

III - CO₂ GASIFICATION TREATMENT TESTS □ ANALOGUES ■ NUCLEAR GRAPHITE WASTE

[¹⁴C] decrease versus relative weight loss



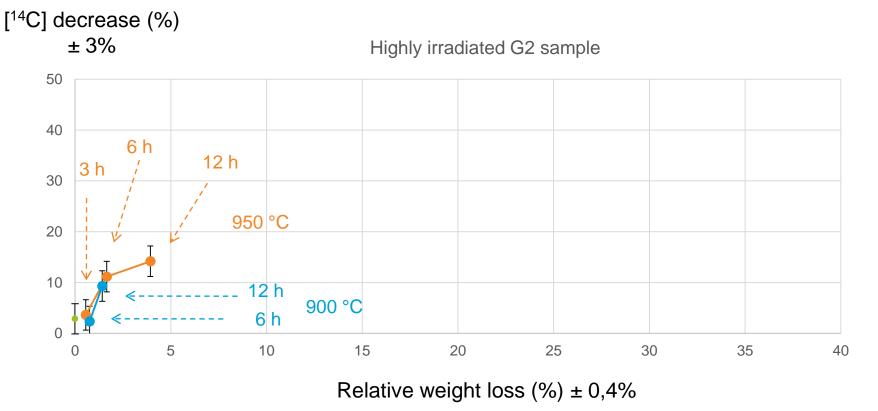
Relative weight loss (%) ± 0,4%

Thermal effect at 1000 °C (He)

- Very weak ¹⁴C extraction but measurable
- Interpretations : some of the ¹⁴C is weakly linked to the graphite
- In an other sample, no thermal effect is measurable

III - CO₂ GASIFICATION TREATMENT TESTS □ ANALOGUES ■ NUCLEAR GRAPHITE WASTE

[¹⁴C] decrease versus relative weight loss

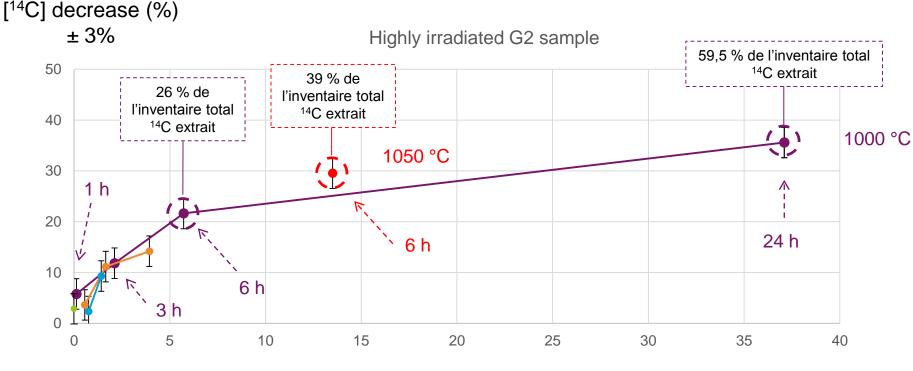


T ≤ 950 °C

- Weak and slow reactivity.
- Weak ¹⁴C extraction.
- Bad accessibility to active sites in closed and nanometer sized porosity ?

III – CO₂ GASIFICATION TREATMENT TESTS ANALOGUES INUCLEAR GRAPHITE WASTE

[¹⁴C] decrease versus relative weight loss



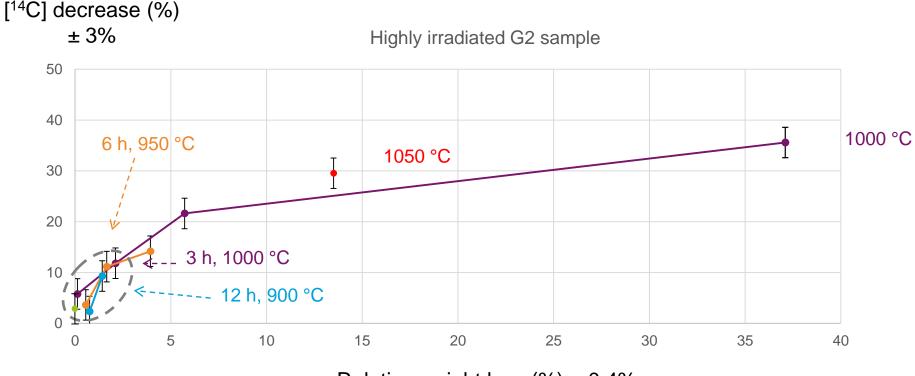
Relative weight loss (%) $\pm 0.4\%$

T ≥ 1000°C

- High temperatures and long durations required
- Significant ¹⁴C extraction
- Interpretations : porosity opening
- With gasification duration, selectivity decrease

III - CO₂ GASIFICATION TREATMENT TESTS □ ANALOGUES ■ NUCLEAR GRAPHITE WASTE

[¹⁴C] decrease versus relative weight loss

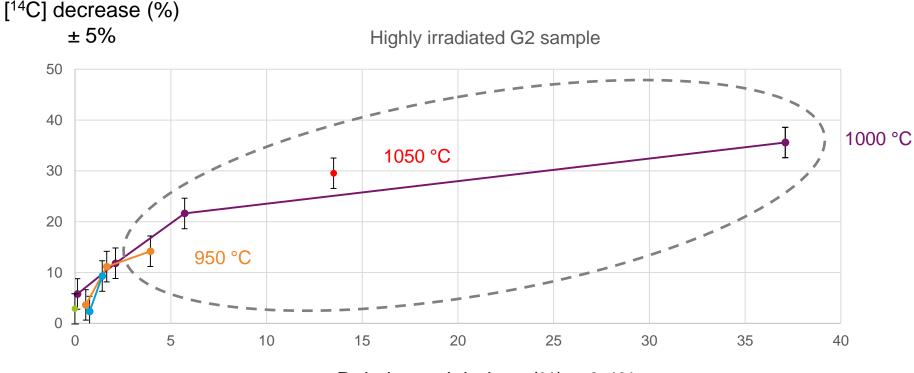


Relative weight loss (%) \pm 0,4%

- ¹⁴C extraction interpretation: 1st step (between 5 and 10% of [¹⁴C]_{ini})
 - For the same relative weight loss, whatever the T°, ¹⁴C extraction is quite the same.
 Easy ¹⁴C extraction.
 - By thermal effect or selectivity on the most degraded and accessible areas.

III – CO₂ GASIFICATION TREATMENT TESTS ANALOGUES INUCLEAR GRAPHITE WASTE

Diminution en [¹⁴C] en fonction de la perte de masse relative



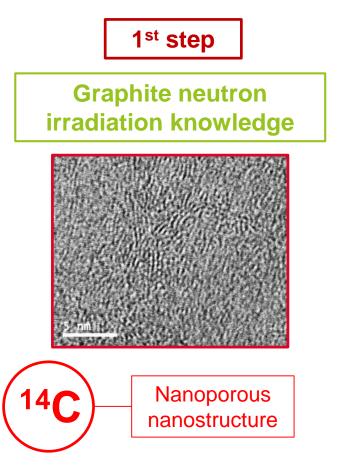
Relative weight loss (%) $\pm 0.4\%$

¹⁴C extraction interpretation: 2nd step (remaining ¹⁴C after the 1st step, 90–95% of [¹⁴C]_{ini})

- **—** Dependence according to the gasification temperature.
- Extraction of ¹⁴C strongly linked to graphite waste structure and hardly accessible.
- ¹⁴C maybe trapped in closed porosity.
- Significant material gasification should be required.



The main goal was to extract ¹⁴C from graphite waste



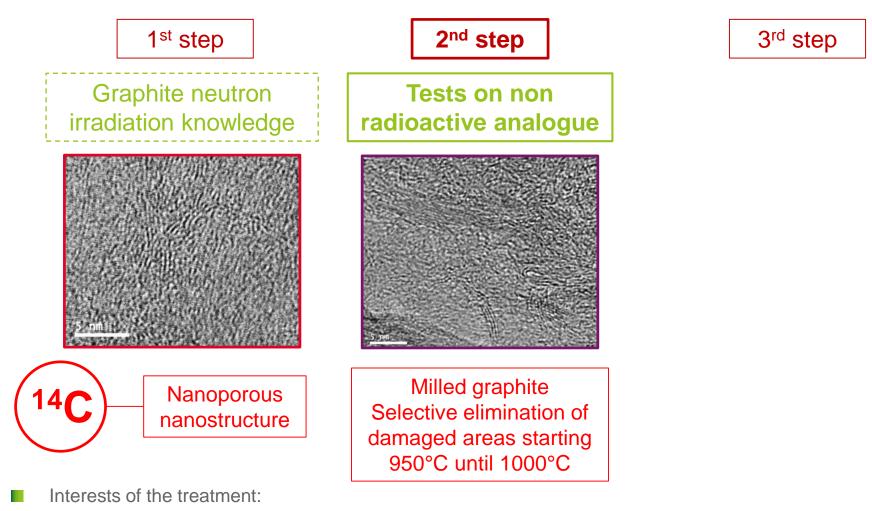
Interests of the treatment:





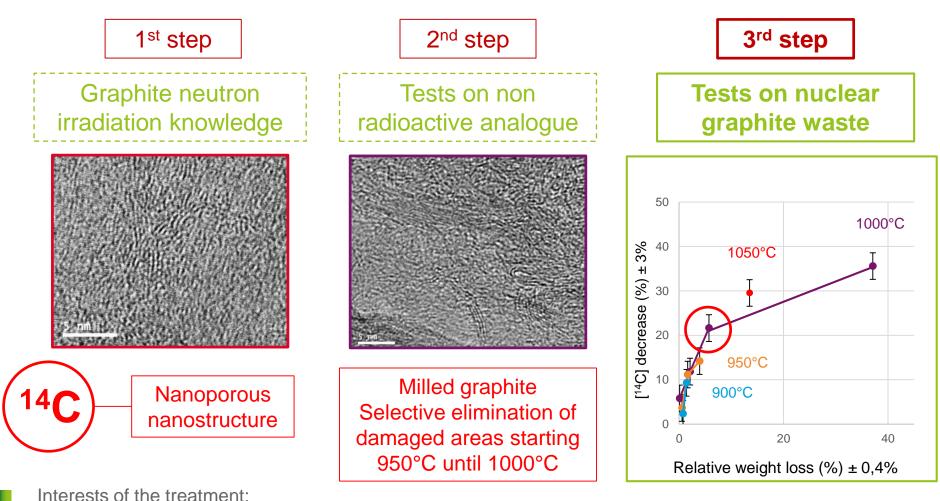
CONCLUSIONS

The main goal was to extract ¹⁴C from graphite waste



CONCLUSIONS

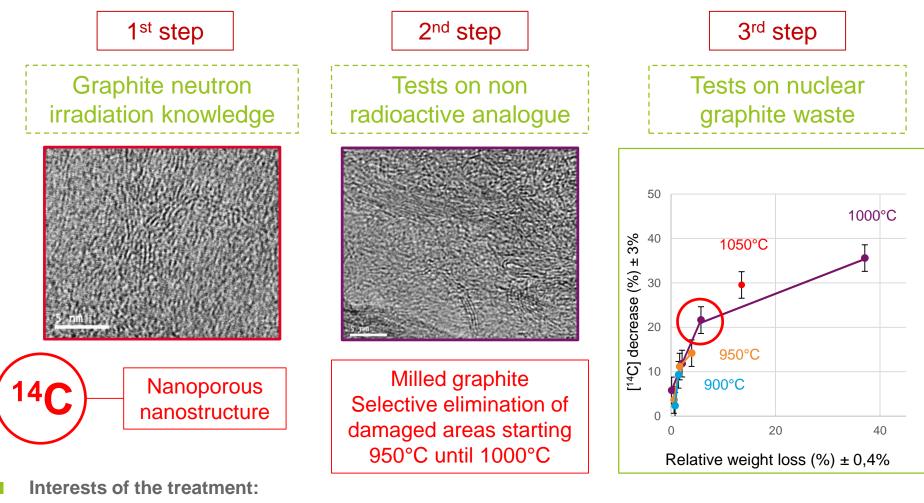
The main goal was to extract ¹⁴C from graphite waste



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CONCLUSIONS

The main goal was to extract ¹⁴C from graphite waste



- To decrease the activity
- To extract the most « mobile » ¹⁴C
- The treatment process is easily optimizable

Acknowledgments









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- Université Paris 11/ED MIPEGE

Thank you for your attention Questions ?



