



**HAL**  
open science

## $\alpha,\beta$ -Unsaturated 2-Acyl-Imidazoles in Asymmetric Biohybrid Catalysis

Justine Mansot, Jean-Jacques Vasseur, Stellios Arseniyadis, Michaël Smietana

► **To cite this version:**

Justine Mansot, Jean-Jacques Vasseur, Stellios Arseniyadis, Michaël Smietana.  $\alpha,\beta$ -Unsaturated 2-Acyl-Imidazoles in Asymmetric Biohybrid Catalysis. *ChemCatChem*, 2019, 11 (23), pp.5686-5704. 10.1002/cctc.201900743 . hal-02401171

**HAL Id: hal-02401171**

**<https://hal.science/hal-02401171v1>**

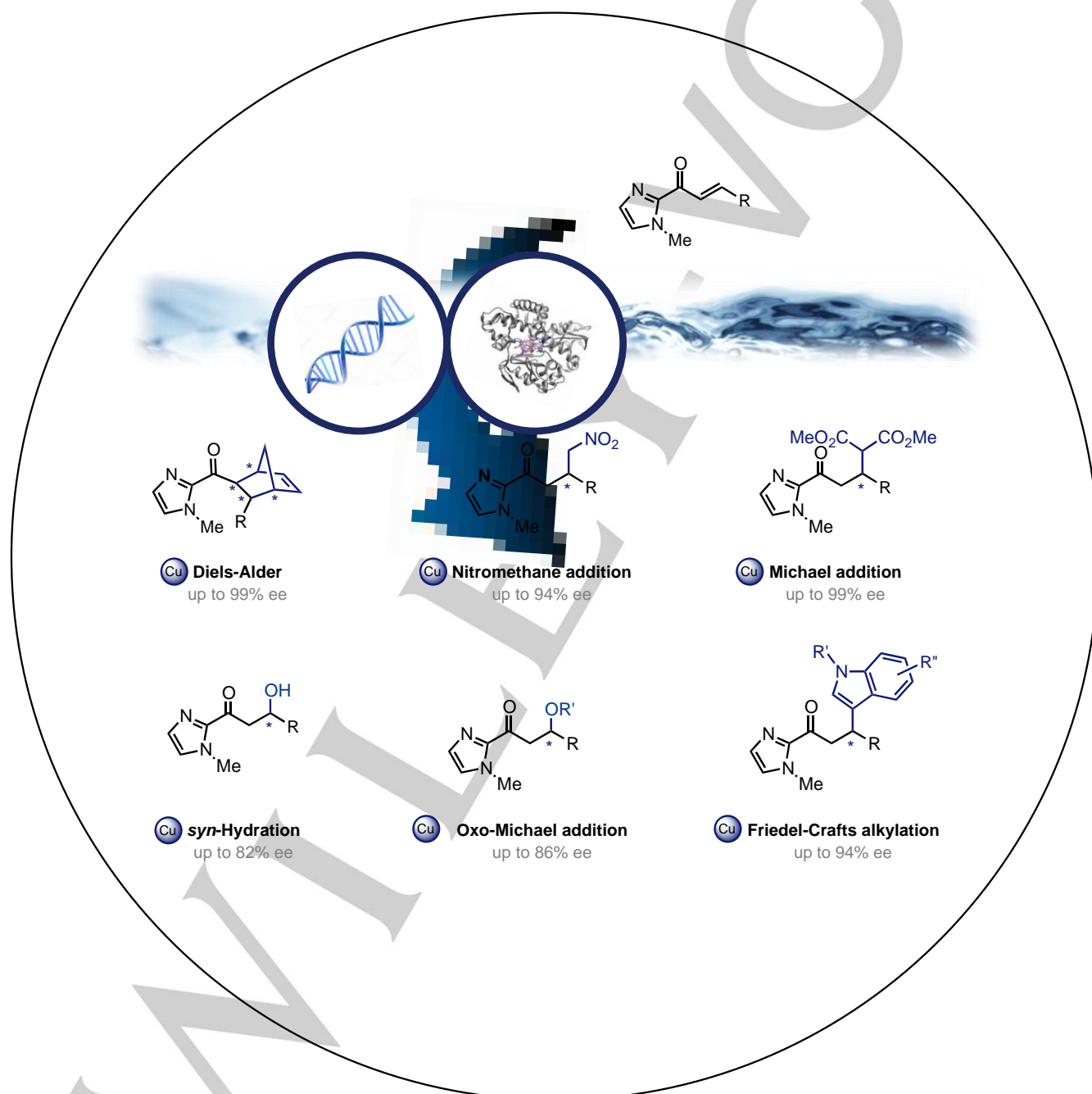
Submitted on 6 Nov 2020

**HAL** is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

# $\alpha,\beta$ -Unsaturated 2-Acyl-Imidazoles in Asymmetric Biohybrid Catalysis

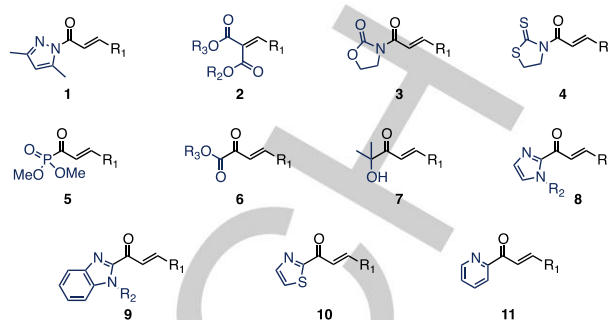
Justine Mansot,<sup>[a]</sup> Jean-Jacques Vasseur,<sup>[a]</sup> Stellios Arseniyadis,<sup>[b]\*</sup> and Michael Smietana,<sup>[a]\*</sup>



$\alpha,\beta$ -Unsaturated acylimidazoles have been used in a plethora of enantioselective transformations over the years and have unsurprisingly become privileged building blocks for asymmetric catalysis. Interestingly however, their use in asymmetric biohybrid catalysis as bidentate substrates able to interact with artificial metalloenzymes has only recently emerged, expanding considerably in the last few years. Easy to prepare and particularly versatile,  $\alpha,\beta$ -unsaturated acylimidazoles appear as leading synthons for the asymmetric construction of C–C and C–O bonds. This Minireview highlights the current and increasing interest of these key building blocks in the context of asymmetric biohybrid catalysis with the aim to stimulate further research into their still unexploited potential.

## 1. Introduction

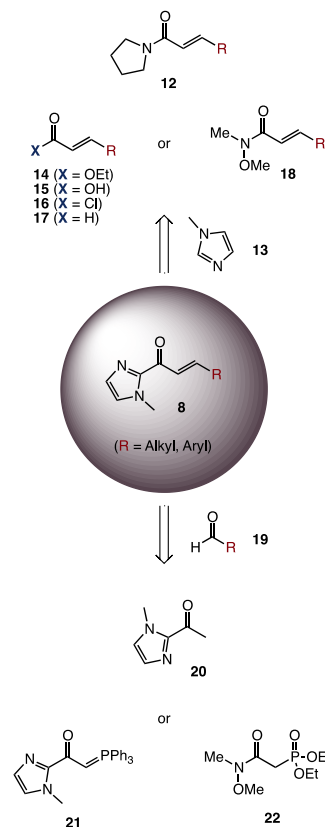
The conjugate addition of nucleophiles to electron-deficient alkenes is a common strategy for the formation of C–C and C–heteroatom bonds which dates back to the pioneering work of Komnenos in 1883<sup>[1]</sup> and Michael in 1887.<sup>[2]</sup> After more than a century of improvements devoted to the optimization of all the reaction parameters, catalytic asymmetric conjugate additions have become highly selective and, most importantly, highly reliable transformations leading to a broad range of optically active compounds with use in fields such as natural product synthesis and medicinal chemistry.<sup>[3]</sup> Over the years, it has been demonstrated that both organometallic and non-organometallic nucleophiles could be involved.<sup>[4]</sup> This later class emerged in the early 80's with the description of a diamine/Co(II) complex<sup>[5]</sup> and later by using chiral bisoxazoline type ligands<sup>[6]</sup> able to coordinate a variety of metals.<sup>[7]</sup> It soon emerged that bidentate substrates were ideally fitted to provide the necessary synergistic effect to reach high catalytic performance through an efficient coordination with the chiral metallic complex. Over the years, a number of  $\alpha,\beta$ -unsaturated acyl derivatives (compounds 1–7, Figure 1) were evaluated in a variety of synthetic transformations including Michael and Mukaiyama-Michael type additions, Diels-Alder and hetero-Diels-Alder cycloadditions as well as Friedel-Crafts alkylations.<sup>[8–16]</sup> However, all the reactions involving the aforementioned substrates required cryogenic conditions to reach high levels of enantioselectivity ultimately limiting the scope. To overcome this drawback, Evans and co-workers introduced the use of  $\alpha,\beta$ -unsaturated 2-acyl-imidazole 8 as a new bidentate substrate.<sup>[17]</sup> The latter showed an increased reactivity attributed to a lower steric demand and a higher electrophilicity of the enone when compared to the  $\alpha,\beta$ -unsaturated-acyl-pyrazoles (1), oxazolidinones (3), thiazolidine-2-thiones (4), phosphonates (5), benzimidazoles (9), thiazoles (10) and pyridines (11).<sup>[18]</sup> In addition, the chiral Lewis



**Figure 1.** Enone commonly used in asymmetric catalysis ( $R_1$  = Alkyl, Aryl,  $R_2$  = Alkyl,  $R_3$  = Alkyl).

acid-catalysed conjugate addition of various indoles in the presence of (PyBox)-Sc(III) led to the corresponding Friedel-Crafts alkylation products with excellent levels of enantioselectivity.

The synthesis of  $\alpha,\beta$ -unsaturated 2-acyl-imidazoles is relatively trivial and usually relies on either the introduction of the imidazole ring on an  $\alpha,\beta$ -unsaturated acyl precursor or on the formation of the double bond *via* some sort of olefination reaction (Figure 2). Okamoto *et al.* were, to the best of our knowledge, the first to describe the synthesis of  $\alpha,\beta$ -unsaturated acyl imidazoles such as 8 through the addition of the deprotonated *N*-methylimidazole 13 onto  $\alpha,\beta$ -unsaturated-*N*-acyl-pyrrolidine 12 (Figure 2).<sup>[19]</sup> Other syntheses involve the direct addition of the deprotonated form of 13 on an  $\alpha,\beta$ -unsaturated ester (14), a carboxylic acid (15), an acyl chloride (16) or on a Weinreb amid (18). Similarly, the addition of 13 on an enal (17) followed by an oxidation of the resulting alcohol intermediate can also be envisioned.<sup>[21]</sup> Ultimately, a tandem aldol/crotonization between aldehyde 19 and *N*-methyl-2-acylimidazole 20<sup>[22,23]</sup> or a Wittig reaction with the appropriate phosphorous ylide 21<sup>[17]</sup> also leads to the title compound. Finally, the phosphorous ylide can be replaced by a phosphonate derivative such as 22, which can readily undergo a Horner-Wadsworth-Emmons type olefination.<sup>[24]</sup>



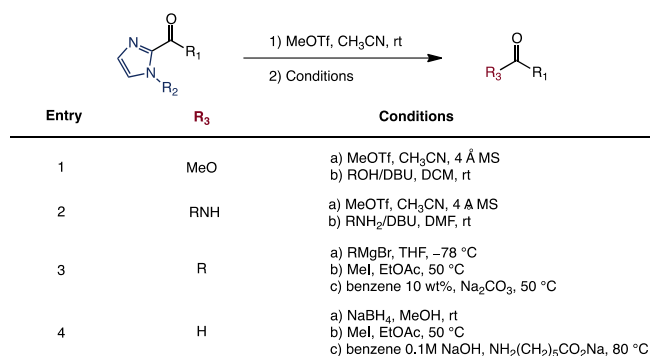
**Figure 2.** Synthesis of 8

In addition to improving catalytic performances, the imidazole moiety presents the advantage of being very versatile.

[a] J. Mansot, Dr. J.-J. Vasseur, Prof. Dr. M. Smetana  
Institut des Biomolécules Max Mousseron, UMR 5247 CNRS  
Université de Montpellier, ENSCM Place Eugène Bataillon, 34095  
Montpellier (France)

E-mail: michael.smetana@umontpellier.fr

[b] Dr. S. Arseniyadis  
Queen Mary University of London  
School of Biological and Chemical Sciences.  
Mile End Road, E1 4NS, London (UK)  
E-mail: s.arseniyadis@qmul.ac.uk



**Figure 3.** Post-transformation of the imidazole moiety.

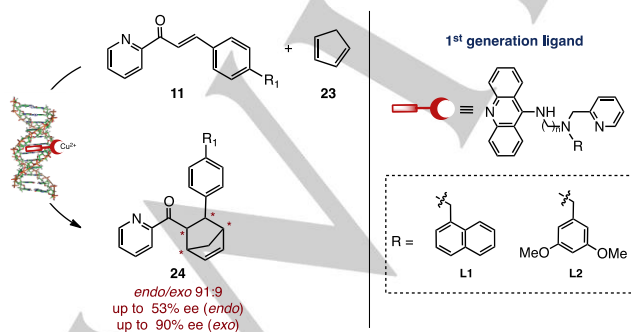
Indeed, it can be readily converted to a wide variety of useful chemical functionalities such as ketones, aldehydes,<sup>[25]</sup> amides, esters,<sup>[26]</sup> and carboxylic acids through simple synthetic transformations (Figure 3).<sup>[25,27]</sup>

## 2. DNA-based asymmetric catalysis

### 2.1. Natural DNA-helix

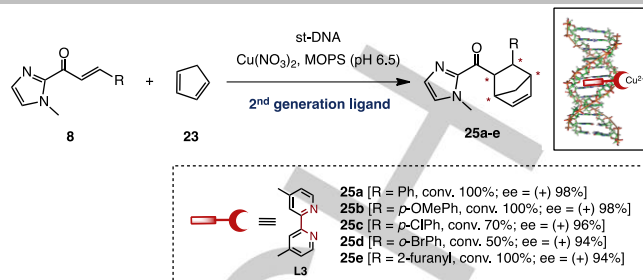
The concept of DNA-based asymmetric catalysis, first introduced by Roelfes and Feringa in 2005,<sup>[31]</sup> relies on a transfer of chirality from the DNA double helix to a pro-chiral substrate using an achiral transition metal catalyst imbedded within the DNA scaffold (Figure 4). The resulting supramolecular architecture provides the necessary chiral microenvironment to induce sufficient enantiodiscrimination.<sup>[32]</sup>

This concept was initially tested on a Diels-Alder reaction between azachalcone **11** and cyclopentadiene **23** in the presence of commercially available DNA extracted from salmon testes (st-DNA). The first generation of ligands that were used were derived from the 9-aminoacridine intercalant, which was covalently bound to a Cu(II) binding diamine. High *endo/exo* selectivities were obtained (up to 98/2 in favour of the *endo* product) albeit moderate enantioselectivities (up to 53% ee). Nonetheless, these results demonstrated that the DNA double helix could be used as a powerful source of chirality in the context of asymmetric catalysis.<sup>[33]</sup>



**Figure 4.** DNA-based asymmetric Diels-Alder.

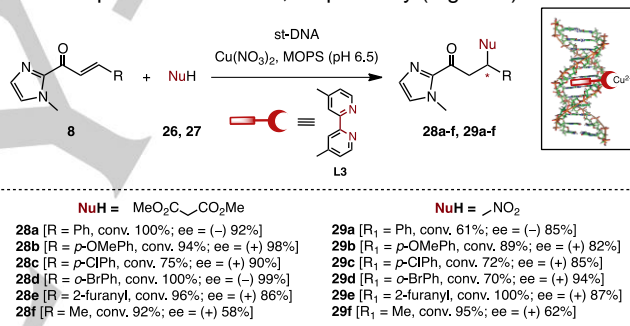
Roelfes and co-workers were later able to improve the selectivities by simply replacing the pyridine moiety by an imidazole.<sup>[34-35]</sup> By doing so,  $\beta$ -aryl-substituted  $\alpha,\beta$ -unsaturated acyl imidazoles could be converted to the corresponding *endo*



**Figure 5.** DNA-based asymmetric Diels-Alder.

bicyclic product in up to 98% ee (Figure 5). The best results were obtained when using commercially available 4,4'-dimethyl-2,2'-bipyridine **L3** as ligand, also known to be a DNA groove binder. This study also revealed that the catalyst loading could be lowered to 5 mol% without any loss of either enantioselectivity or diastereoselectivity.

These results prompted Roelfes and co-workers to evaluate other C–C bond forming reactions in presence of DNA.<sup>[36]</sup> Hence, the addition of nitromethane or dimethyl malonate to **8** allowed the formation of the corresponding Michael addition products with ees up to 94% and 99%, respectively (Figure 6).



**Figure 6.** DNA-based asymmetric Michael.

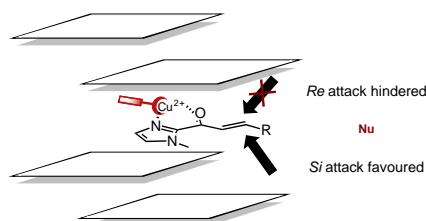
Dr. Stelios Arseniyadis received his PhD from the University of Strasbourg (with Dr. C. Mioskowski). After post-doctoral studies at Rhodia Chirex Inc. (with Prof. Dr. S. L. Buchwald), Imperial College London (with Prof. Dr. A. C. Spivey) and at The Scripps Research Institute (with Prof. Dr. K. C. Nicolaou), he joined the CNRS as a permanent researcher in 2005 and was promoted to the rank of Director in 2015. The same year, he joined Queen Mary University of London to start his own group which is focused on the development of new synthetic tools in the field of asymmetric catalysis.



Prof. Dr. Michael Smietana was educated at the Ecole Normale Supérieure in Paris and received his PhD from the University of Strasbourg (with Dr. Charles Mioskowski). After post-doctoral studies at Stanford (with Prof. Eric T. Kool) he joined Montpellier University in 2004 and was promoted to full Professor in 2012. In 2018 he received the Jean-Marie Lehn award from the Organic Chemistry Division of the French Chemical Society. His current research interests focus on molecular recognition and DNA-based asymmetric catalysis.

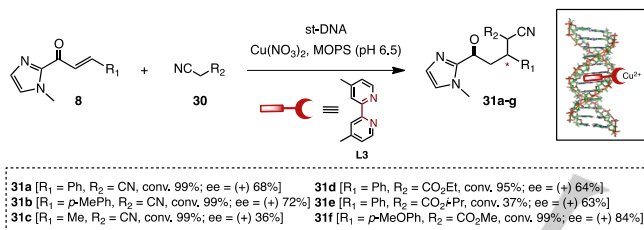


Mechanistically, the formation of the major enantiomer was proposed to result from the *Si* face attack of the complexed acylimidazole substrate as shown in Figure 7.



**Figure 7.** Schematic representation of the complexed acylimidazole substrate in DNA.

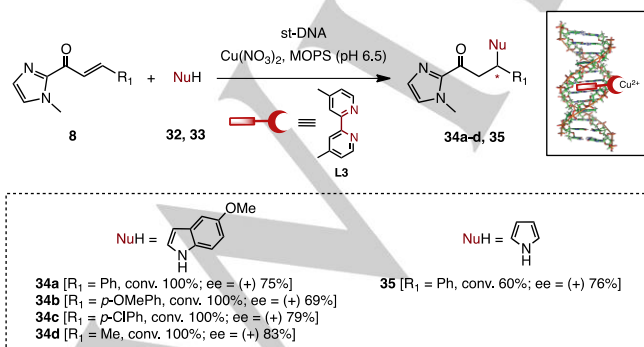
The nature of the Michael donor was later extended to malonitrile by Li *et al.*<sup>[37]</sup> Hence, after a thorough screening of the reaction conditions, the use of **L3**/Cu(II) in conjunction with st-DNA was shown to afford excellent conversions and high levels of selectivity (up to 72% ee, Figure 8). Interestingly, the addition of dissymmetric cyanoacetate derivatives generated the two diastereoisomers in roughly the same ratio in up to 84% ee.



**31a** [ $R_1 = \text{Ph}$ ,  $R_2 = \text{CN}$ , conv. 99%; ee = (+) 68%]    **31d** [ $R_1 = \text{Ph}$ ,  $R_2 = \text{CO}_2\text{Et}$ , conv. 95%; ee = (+) 64%]  
**31b** [ $R_1 = p\text{-MePh}$ ,  $R_2 = \text{CN}$ , conv. 99%; ee = (+) 72%]    **31e** [ $R_1 = \text{Ph}$ ,  $R_2 = \text{CO}_2i\text{Pr}$ , conv. 37%; ee = (+) 63%]  
**31c** [ $R_1 = \text{Me}$ ,  $R_2 = \text{CN}$ , conv. 99%; ee = (+) 36%]    **31f** [ $R_1 = p\text{-MeOPh}$ ,  $R_2 = \text{CO}_2\text{Me}$ , conv. 99%; ee = (+) 84%]

**Figure 8.** DNA-based conjugate addition of malonitrile derivatives.

Another benchmark in the field concerned the optimization of the enantioselective Friedel-Crafts alkylation with indoles.<sup>[38]</sup> Hence, through binding affinity studies, Roelfes and co-workers were able to show that only 16% of the catalyst was bound to DNA, while at the same time a 30-fold rate enhancement could be observed when the reaction was run in the presence of DNA, thus suggesting that the background reaction was negligible. As a general trend, the reactions all led to full conversions and ees reaching up to 83% (Figure 9).



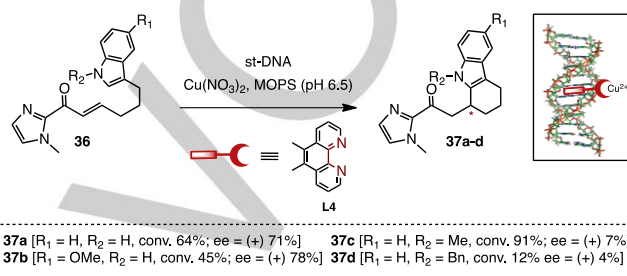
**34a** [ $R_1 = \text{Ph}$ , conv. 100%; ee = (+) 75%]    **35** [ $R_1 = \text{Ph}$ , conv. 60%; ee = (+) 76%]  
**34b** [ $R_1 = p\text{-OMePh}$ , conv. 100%; ee = (+) 69%]  
**34c** [ $R_1 = p\text{-ClPh}$ , conv. 100%; ee = (+) 79%]  
**34d** [ $R_1 = \text{Me}$ , conv. 100%; ee = (+) 83%]

**Figure 9.** DNA-based asymmetric Friedel-Crafts alkylation.

As a general trend, the best selectivities were obtained with substrates bearing an alkyl substituent at the  $\beta$  position of the enone. Most importantly, the authors were able to show that reaction was scalable and that the catalysts could be recycled

without any noticeable loss of reactivity or selectivity. Finally, pyrroles were also shown to be compatible with ees reaching up to 76%.

A few years later, Sugiyama *et al.* reported an intramolecular version of the Friedel-Crafts alkylation using the bifunctional acylimidazoles **36** (Figure 10).<sup>[39]</sup> The reactions catalysed by a st-DNA/5,6-dimethyl-1,10-phenanthroline **L4**/Cu(II) complex proved highly selective as up to 78% ee were obtained with a 5-methoxy group on the indole ring. In sharp contrast, substrates substituted on the nitrogen afforded much lower selectivities.

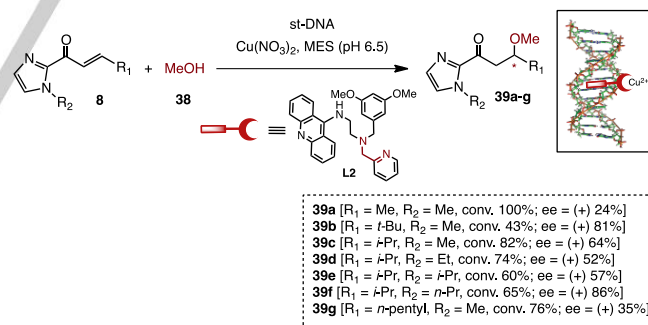


**37a** [ $R_1 = \text{H}$ ,  $R_2 = \text{H}$ , conv. 64%; ee = (+) 71%]    **37c** [ $R_1 = \text{H}$ ,  $R_2 = \text{Me}$ , conv. 91%; ee = (+) 7%]  
**37b** [ $R_1 = \text{OMe}$ ,  $R_2 = \text{H}$ , conv. 45%; ee = (+) 78%]    **37d** [ $R_1 = \text{H}$ ,  $R_2 = \text{Bn}$ , conv. 12% ee = (+) 4%]

**Figure 10.** DNA-based intramolecular Friedel-Crafts alkylation.

The influence of the co-solvent in the Friedel-Crafts alkylation and the Michael addition was also evaluated.<sup>[40]</sup> It was notably shown that up to 10% v/v of acetonitrile in water did not affect the binding constant of the catalyst to DNA while allowing an efficient scale-up of the reactions. Furthermore, an acceleration rate of the reaction was noted and attributed to a fast dissociation constant of the product.

The catalytic enantioselective oxa-Michael addition was also reported by Roelfes and co-workers with up to 86% ee (Figure 11).<sup>[41]</sup>

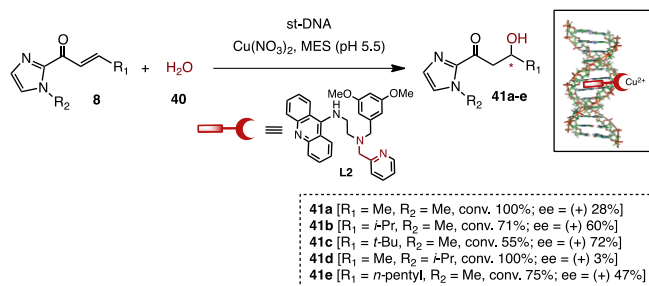


**39a** [ $R_1 = \text{Me}$ ,  $R_2 = \text{Me}$ , conv. 100%; ee = (+) 24%]  
**39b** [ $R_1 = t\text{-Bu}$ ,  $R_2 = \text{Me}$ , conv. 43%; ee = (+) 81%]  
**39c** [ $R_1 = i\text{-Pr}$ ,  $R_2 = \text{Me}$ , conv. 82%; ee = (+) 64%]  
**39d** [ $R_1 = i\text{-Pr}$ ,  $R_2 = \text{Et}$ , conv. 74%; ee = (+) 52%]  
**39e** [ $R_1 = i\text{-Pr}$ ,  $R_2 = i\text{-Pr}$ , conv. 60%; ee = (+) 57%]  
**39f** [ $R_1 = i\text{-Pr}$ ,  $R_2 = n\text{-Pr}$ , conv. 65%; ee = (+) 86%]  
**39g** [ $R_1 = n\text{-pentyl}$ ,  $R_2 = \text{Me}$ , conv. 76%; ee = (+) 35%]

**Figure 11.** DNA-based asymmetric oxa-Michael addition.

In these experiments, methanol but also ethanol and isopropanol were used in 40 v/v%, however the conversions dropped when increasing the size of the alkyl group at the  $\beta$ -position of the enone, while no reactivity was observed with  $\beta$ -aromatic substituents. Also worth noting is the formation of the hydration product in all cases, however this could be minimized by lowering the temperature to  $-18^\circ\text{C}$ .

The first example of a non-enzymatic *syn*-hydration was reported by Roelfes and co-workers in 2010 using st-DNA in association with the same **L2**/Cu(II) metallic co-factor used for the oxa-Michael addition (Figure 12).<sup>[42]</sup> Under these conditions,  $\alpha,\beta$ -unsaturated 2-acyl imidazoles could be readily converted to the corresponding  $\beta$ -hydroxy ketones in up to 72% ee. In all these experiments, only the  $\beta$ -methyl substituted acyl imidazole led to full conversions albeit with low ees, while substrates



**Figure 12.** DNA-based asymmetric *syn*-hydration.

bearing a  $\beta$ -*t*-butyl moiety led to higher *ees* but lower conversions. Surprisingly, replacing the methyl by an *i*-propyl group on the imidazole ring led to a racemic mixture of products, thus highlighting the impact of the steric hindrance induced by the substrate as well as the coordination sphere environment on the stereoselectivity of the reaction.

## 2.2 Synthetic DNA-helix

### 2.2.1 Supramolecular approach

The results obtained with natural DNA clearly highlight the importance of the binding between the metallic co-factor and the double helix. This prompted the community to next evaluate the influence of specific sequences on the reaction rate and ultimately on the enantioselectivity. A first set of experiments was run on the Diels-Alder cycloaddition between acylpyridine **11** (Figure 1) and cyclopentadiene using various **L3**/Cu(II)/DNA complexes.<sup>[43]</sup> The results showed an interesting correlation between the high affinity of the **L3**/Cu(II) metallic-co-factor for the GC-rich 12-mer sequence d(TCAGGGCCCTGA)<sub>2</sub>, **ODN1**, which is higher than that for natural st-DNA, and the increase of both the reaction rate (100-fold) and the selectivity.

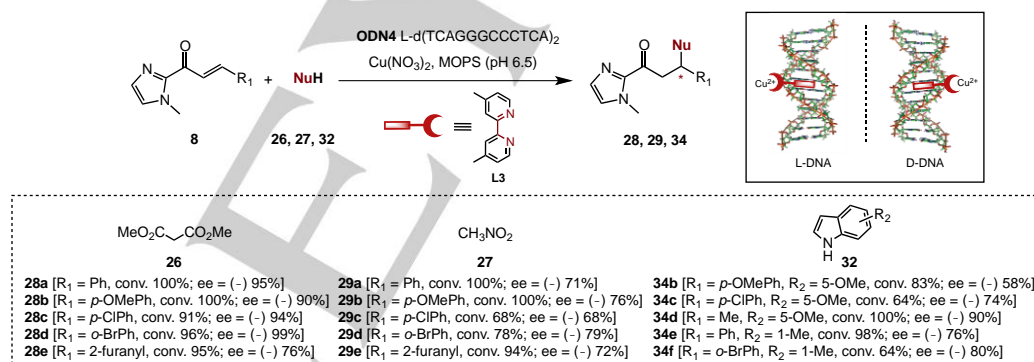
A similar study was performed on the Friedel-Crafts alkylation using this time  $\alpha,\beta$ -unsaturated 2-acyl imidazole **8** as a model substrate.<sup>[38]</sup> Similarly to the Diels-Alder cycloaddition, **ODN1** led to usually higher *ees* and conversions than st-DNA, while AT-rich oligonucleotides were found to induce higher *ees*.

Interestingly, in the case of the intramolecular Friedel-Crafts alkylation with 2-acyl imidazole **36**, Sugiyama and co-workers showed that AT-rich duplexes gave rise to higher enantioselectivities.<sup>[39]</sup> To understand these results a theoretical investigation was undertaken by modelling 28 conformations of **ODN2** d(CAAAAATTTTG)<sub>2</sub> with **L4**/Cu(II). The simulations revealed a higher binding energy of **ODN2** toward the pro-(*S*) conformation of 2-acylimidazole **36**. Moreover, a shorter distance between the two carbons involved in the newly formed C–C bond was observed in the case of the pro-*S* complex more inline with the expected transitional state.<sup>[44]</sup>

This sequence-selectivity was also evaluated in the case of the *syn*-hydration, where higher selectivities were obtained with sequences bearing an AT-rich central segment, however the structure of the ligand was also shown to be crucial.<sup>[42b]</sup>

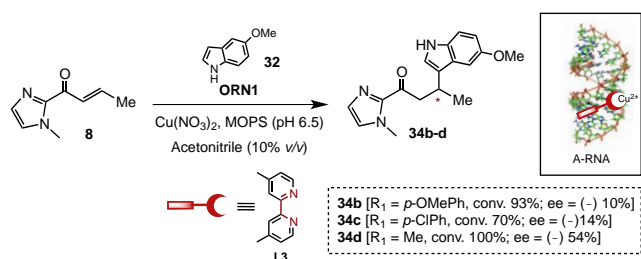
Finally, in the case of the oxa-Michael addition, AT-rich **ODN2** gave rise to higher *ees* [conv. 82%; 43% ee] than the GC-rich sequence **ODN1** [conv. 68%; 19% ee] (Figure 11), albeit still lower than st-DNA.<sup>[41]</sup>

These results prompted our group to devote our first efforts in the field of biohybrid catalysis toward the control of the enantioselective outcome. While many studies evaluated various parameters including the influence of the metallic co-factor to selectively access one specific enantiomer, the inversion of selectivity were generally partial, often very ligand-specific and thus unpredictable.<sup>[31,39,45,46]</sup> Taking into account the best selectivities obtained so far in the Friedel-Crafts alkylation of  $\alpha,\beta$ -unsaturated 2-acyl imidazole **8**, we synthesized the mirror image of **ODN1**, **ODN4**, using L-nucleotide phosphoramidites. As expected, the use of the left-handed L-d(TCAGGGCCCTGA)<sub>2</sub> **ODN4** sequence in the presence of the **L3**/Cu(II) complex afforded the opposite enantiomer with the exact the same enantioselectivity than the one obtained with **ODN1** (Figure 13).<sup>[22]</sup> The method was also validated on the conjugate addition of dimethyl malonate and nitromethane, thus showcasing the generality of the method.



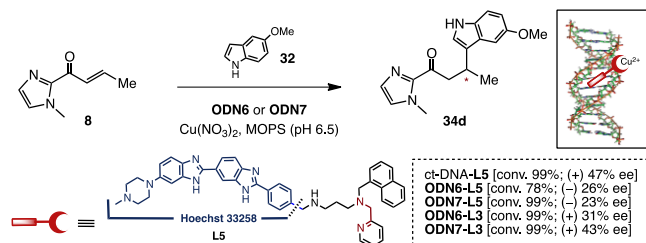
**Figure 13.** Mirror image DNA-based asymmetric catalysis

We later applied this concept to short double-stranded RNA sequences,<sup>[47]</sup> which are known to adopt a different helical shape (A-form helix) with a large and shallow minor groove and a deeper and tighter major groove. Although the enantioselectivities achieved were lower than the ones obtained with DNA, we found that the GC rich 16-mer **ORN1**, L-(CAGUCAGUACUGACUG)<sub>2</sub>, provided the best selectivities with up to 54% ee and an expected inversion of selectivity depending on the D or L nature of the sequence (Figure 14).



**Figure 14.** RNA-based asymmetric Friedel-Crafts alkylation. [**ORN1** L-(CAGUCAGUACUGACUG)<sub>2</sub>].

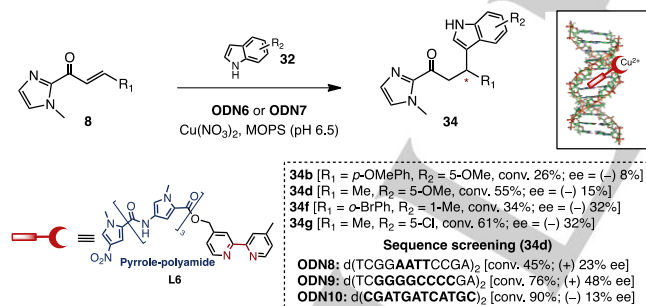
Achieving sequence-specificity was another challenge we decided to tackle. Hence, instead of evaluating the sequence dependency of a given reaction, we envisioned to design sequence-specific catalysts. The well-known minor groove binder Hoechst-33258, recognized for its strong affinity for AT-rich regions of DNA, was selected and modified to incorporate a Cu(II) binding site (Figure 15).<sup>[45]</sup>



**Figure 15.** DNA-based asymmetric Friedel-Crafts alkylation with minor groove-binding ligands. [ODN6: 5'-d(CGAAATTCGTTTCGAAATTCG)-3', ODN7: 5'-d(CGTATACGTTTTCGTATACG)-3'].

Among all the ligands that were synthesized and evaluated in the Cu(II)-catalysed Friedel-Crafts alkylation of  $\alpha,\beta$ -unsaturated 2-acyl imidazole **8** with 5-methoxyindole **32**, ligand **L5** provided the best selectivities (up of 47% ee with ct-DNA). Although moderate, this result reinforced the indisputable correlation between affinity and selectivity as the highest selectivities were obtained with the ligand having the highest binding affinity with DNA.

Sequence specific targeting of DNA was also at the centre of the recent work published by Qiao and co-workers.<sup>[46]</sup> A four unit pyrrole-polyamide groove binder known for its high affinity toward DNA was selected and covalently modified with a 4,4'-dimethyl-2,2'-bipyridine (**L6**, Figure 16).



**Figure 16.** DNA-based asymmetric Friedel-Crafts alkylation using pyrrole-polyamide **L6**.

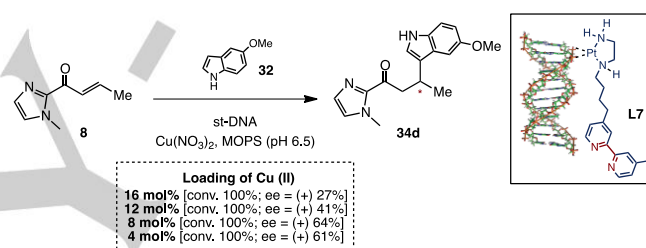
The Cu(II)-catalysed Friedel-Crafts alkylation of  $\alpha,\beta$ -unsaturated 2-acyl imidazole **8** with 5-methoxyindole was again selected as a model reaction. A screening of both self-complementary AT and GC-rich oligonucleotides identified **ODN9**, d(TCGGGGCCCCGA)<sub>2</sub>, as the lead sequence (up to 48% ee). Circular dichroism spectra and singular value decomposition analyses suggested that different binding modes could co-exist depending on the nature of the sequence therefore affecting the enantioselectivity.

### 2.2.2 Covalent approach

The anchoring strategy used to link the transition metal complex to DNA is a critical parameter that influences the nature of the

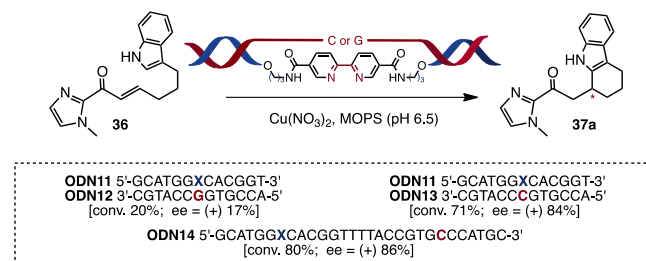
second coordination sphere provided by the bio-scaffold. The supramolecular anchoring approach of a metallic-cofactor to DNA allows a modular assembly of the biohybrid catalyst and permits a rapid optimization of the reaction conditions. In contrast, the covalent anchoring approach allows to specifically position the metallic co-factor independently of its binding affinity to DNA but requires multiple synthetic steps to assemble each biohybrid catalyst.

Taking advantage of the reactivity of bifunctional platinum complexes towards DNA, Gjonaj and Roelfes synthesized a bipyridine-platinum complex and covalently attached it to st-DNA. After addition of Cu(NO<sub>3</sub>)<sub>2</sub>, the new DNA-based catalyst **L7** was evaluated in the benchmark Friedel-Crafts alkylation of  $\alpha,\beta$ -unsaturated 2-acyl imidazole **8** with 5-methoxyindole **32** (Figure 17).<sup>[48]</sup> Although the ees were slightly lower than the ones obtained with the typical supramolecular approach, this new biohybrid catalyst proved particularly robust as showcased by a series of 10 recycling experiments that led to no noticeable erosion of either the reactivity or the selectivity.



**Figure 17.** DNA-based asymmetric Friedel-Crafts alkylation using a cisplatin ligand.

Park and Sugiyama later developed various synthetic DNA-based hybrid catalysts containing an intrastrand bipyridine ligand. The latter were prepared by incorporating a 2,2'-bipyridine-5,5'-dicarboxylic acid moiety within a specific 13-mer (**ODN11**) or an auto-complementary single-strand 29-mer (**ODN14**) sequence (Figure 18).<sup>[49]</sup> The modified 13-mer oligonucleotide was then associated with a complementary strand containing either a G or a C nucleobase against the ligand and the resulting biohybrid catalysts were evaluated in the intramolecular Friedel-Crafts alkylation of acyl imidazole **36**. As a general trend, the catalysts containing a cytosine facing the bipyridine led to higher selectivities (84% ee with **ODN11/ODN13** and 86% ee with the hairpin **ODN16** vs 17% ee with **ODN11/ODN12**).



**Figure 18.** DNA-based asymmetric intramolecular Friedel-Crafts alkylation using an intrastrand bipyridine ligand.

These results prompted the authors to investigate further the role of the cytosine within the catalytic pocket. Hence, various

counter-stands incorporating a triethylene glycol (**ODN15a**) or an alkyl linker (**ODN15b** and **ODN15c**) opposite to cytosine were prepared and evaluated in the Diels-Alder reaction between both  $\alpha,\beta$ -unsaturated 2-acyl imidazole **8** and  $\alpha,\beta$ -unsaturated 2-acyl pyridine **11** and cyclopentadiene (**23**). The best results were obtained with the triethylene glycol linker which afforded ees ranging between 88 and 92% (Figure 19).<sup>[50]</sup>

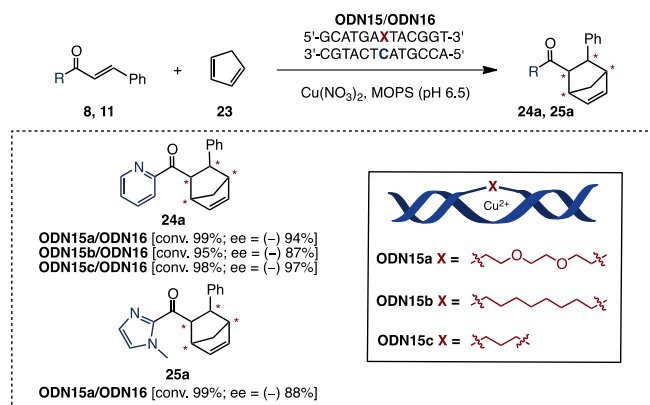


Figure 19. DNA metalloenzyme-based asymmetric Diels-Alder.

### 2.2.3 Solid-Supported DNA helix

The use of solid-supported DNA for asymmetric catalysis was first introduced by Park and Sugiyama based on a non-covalent immobilization of DNA on ammonium-functionalized silica beads.<sup>[51]</sup> The method proved highly effective with selectivities reaching up to 94% ee in the Diels-Alder cycloaddition between  $\alpha,\beta$ -unsaturated 2-acyl pyridine **11** and cyclopentadiene. Following these results, the authors also developed DNA-silica minerals made from cationic Mg(II) ions, *N*-trimethoxysilylpropyl-*N,N,N*-trimethylammonium chloride (TMAPS), tetraethoxysilane (TEOS) and a Cu/ligand complex.<sup>[52]</sup>

Considering the crucial importance of heterogeneous catalysis for industrial applications, our group has also embarked on the development of an eco-friendly and recyclable biohybrid catalyst. Our approach, which relied on the use of commercially available cellulose-supported (CS) DNA in conjunction with a Cu(II)/L3 complex, led to high levels of selectivity in both the Friedel-Crafts alkylation and the Michael addition on  $\alpha,\beta$ -unsaturated 2-acyl imidazole **8** (Figure 20).<sup>[53]</sup> Moreover, we demonstrated that the CS-ct-DNA-Cu(II)/L3 biohybrid catalyst could be recycled up to 10 times without any loss of reactivity or selectivity and even adapted to a single-pass continuous-flow process.

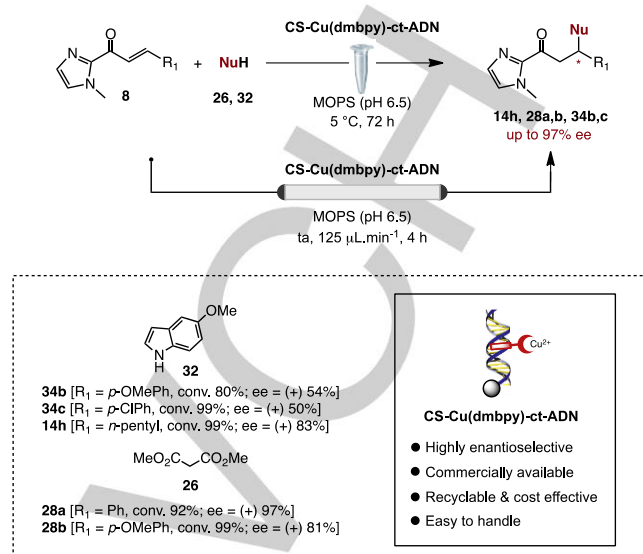


Figure 20. CS-ct-DNA-based asymmetric Friedel-Crafts alkylation and Michael addition.

## 2.4 G-quadruplex

### 2.4.1 Supramolecular approach

G-Quadruplex (G4DNA) are secondary structures composed of four-stranded G-rich DNA sequences. The core consists of four guanines bases associated through Hoogsteen hydrogen bonds forming planar square arrangements. These structures are stabilized by monovalent cations (preferably K<sup>+</sup>) that sit in the centre and interact with eight guanines. Depending on their unimolecular or bimolecular nature, G4DNA are distinguished by their topological variety which include mainly parallel, anti-parallel or '3+1' hybrids (Figure 21).<sup>[54-57]</sup>

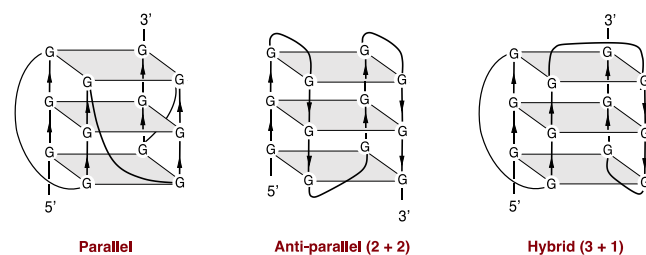


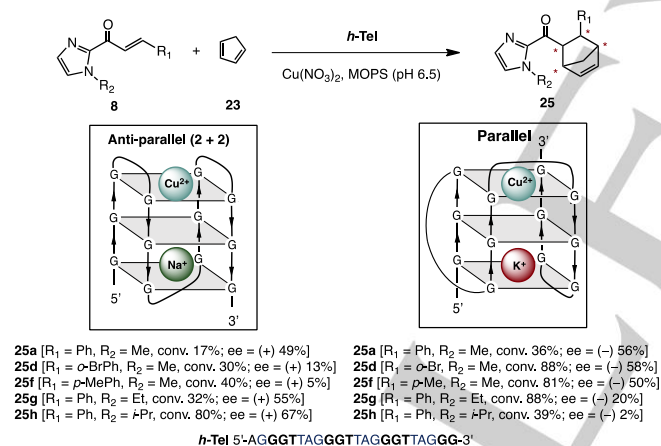
Figure 21. G4DNA conformations.

Considering these different arrangements and their respective ability to bind to a large variety of ligands, G4DNA also appeared as privileged bioscaffolds for asymmetric catalysis.<sup>[58]</sup> In the case of G4DNA-based biohybrid catalytic systems, both the supramolecular and covalent anchoring strategies were evaluated on the benchmark Diels-Alder cycloaddition, the Friedel-Crafts alkylation and the Michael addition.



The first example of a G4-based asymmetric catalysis was reported by Moses and co-workers in 2010 on the standard Diels-Alder cycloaddition between azachalcone **11** and cyclopentadiene.<sup>[59]</sup> G4DNA forming sequences such as *h-Tel* d(AGGGTTAGGGTTAGGGTTAGGG) and *c-Kit* d(AGGGAGGGCGCTGGGAGGAGG) were selected and evaluated in the presence of Cu(II) and ligand **L3**. Despite an excellent *endo/exo* diastereoselectivity and high conversions (up to 85%), the ees were rather moderate (up to 34%).

Nonetheless, these results demonstrated the ability of G4DNA to act as good chiral scaffolds. Interestingly, Li *et al.* later demonstrated that G4DNA/Cu(II) could catalyse the reactions in the absence of ligand in up to 74% ee.<sup>[60]</sup> Moreover, the importance of the conformation was emphasized by the observation of an inversion of selectivity between parallel and anti-parallel G-quadruplexes. This reversal of selectivity was also observed with *h-tel*-Cu(II) in the presence of different alkali metal ions (Figure 22).<sup>[61]</sup> Hence, the enantioselectivity could be switched from (–) 58% to (+) 67% by simply changing the nature of the alkali metal ion from Na<sup>+</sup> to K<sup>+</sup>. This inversion was attributed to a structural transformation from an antiparallel to a hybrid-type G-quadruplex structure.<sup>[62]</sup>

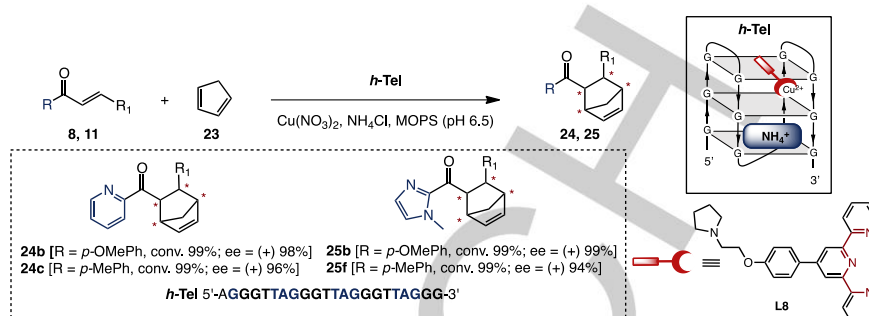


**Figure 22.** *h-Tel* G4DNA-catalysed asymmetric Diels-Alder.

The same group also prepared a terpyridine **L8**/Cu(II) complex having a high affinity to *h-tel* (Scheme 23). The resulting biohybrid catalyst induced a 73-fold rate acceleration compared to **L8** and ees up to 99%. Interestingly, a comparison between substrates **8** and **11** confirmed the prevalence of the acylimidazole scaffold over the acylpyridyl.<sup>[63]</sup>

The importance of the conformation of the G4DNA was also confirmed in both the Friedel-Crafts alkylation<sup>[64]</sup> and the Michael addition.<sup>[65]</sup> Hence, the selectivity outcome of a given reaction can easily be set by fine-tuning the conditions (choice of the alkali metal ions, the ligands, the co-solvents...). On the down side, the ees obtained with G4DNA catalysts never exceeded 75%, far below the selectivities generally obtained in the field.

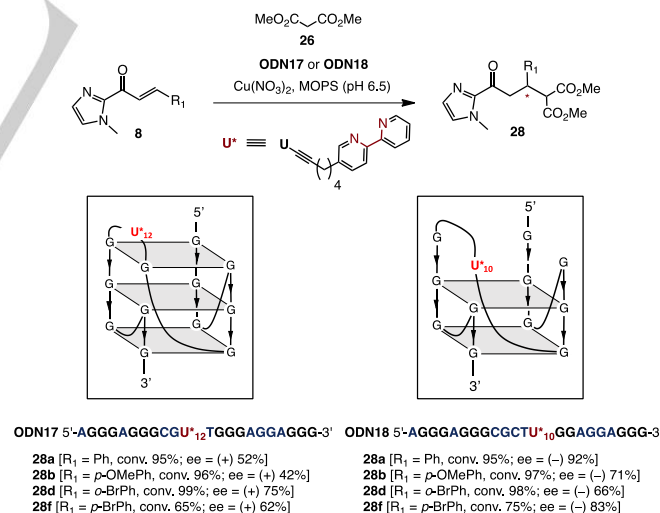
In order to improve the performances of the G4DNA, a covalent anchoring strategy was also evaluated, thus circumventing the issues related with the binding affinity of the ligand.



**Figure 24.** G4DNA-based asymmetric Diels-Alder.

## 2.4.2 Covalent approach

The covalent positioning of a bipyridine ligand into the G4DNA was first reported by Jäschke and co-workers.<sup>[66]</sup> New G4DNA biohybrid catalysts were obtained by covalently attaching a **L3**/Cu(II) complex *via* different linkers of varying length. The evaluation of these new G4DNA catalysts on the asymmetric Michael addition of dimethyl malonate on  $\alpha,\beta$ -unsaturated 2-acyl imidazole **8** revealed the importance of the positioning of the ligand, the length of the linker and the topology of the G4DNA. Interestingly, a stunning inversion of the selectivity was observed when using **ODN17**, which carries the modification at position 12 compared to **ODN18** which bears the exact same ligand at position 10 (Figure 24). Also worth pointing out is the catalyst recyclability, which was later demonstrated by reusing the catalyst over 10 times without any erosion of the selectivity.<sup>[67]</sup>

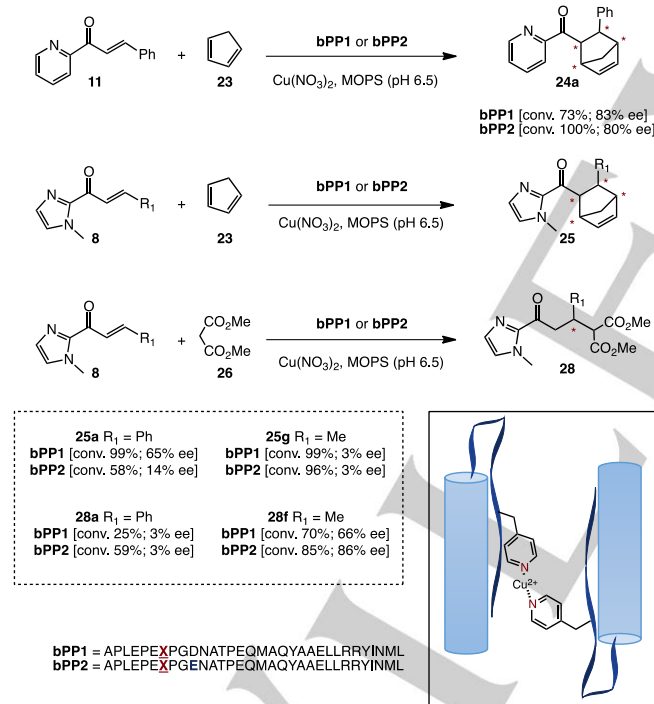


**Figure 24.** G4DNA-based asymmetric Michael addition through a covalent modification.

### 3. Peptide and Protein-based asymmetric catalysis

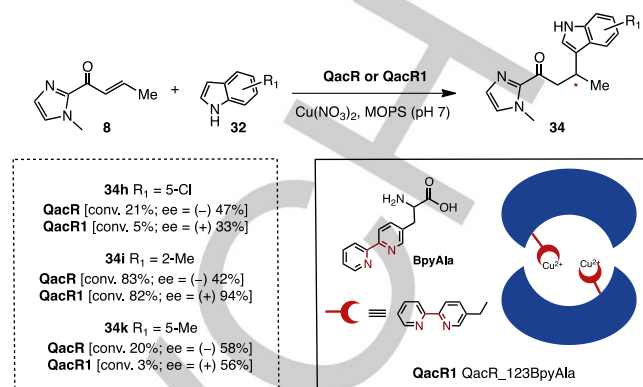
#### 3.1 Artificial metalloenzymes

The design of artificial metalloenzymes incorporating a synthetic transition-metal catalyst into a protein-binding site also represents an emerging approach to engineer biohybrid catalysts. Similar to DNA-based biohybrid catalysts, covalent, dative, or non-covalent anchoring strategies can be applied to polypeptidic scaffolds. In this respect, Roelfes *et al.* selected the bovine Pancreatic Polypeptide **bPP**,<sup>[65,66]</sup> a 36 residue polypeptide hormone as a new biohybrid scaffold.<sup>[67]</sup> The natural peptidic hormone was truncated to 31 residues and grafted at the Tyr7 position with an active site by introducing a non-proteinogenic amino acid capable of binding a Cu(II) ion. The **bPP** variants were evaluated in the Cu(II)-catalysed Diels-Alder and Michael addition with  $\alpha,\beta$ -unsaturated substrates **8** and **11** (Figure 25). The results showed that peptides containing L-3-pyridylalanine led to good enantioselectivities on both the Diels-Alder (up to 83% ee) and the Michael addition (up to 86% ee).



**Figure 25.** Bovine Pancreatic Polypeptide **bPP** as scaffold for asymmetric transformations.

While this latter biohybrid catalyst was obtained by solid phase synthesis, the same group reported a few years later the *in vivo* incorporation of a metal-binding non-proteinogenic amino acid into the transcription factor Lactococcal multidrug resistance Regulator (**LmrR**).<sup>[68]</sup> The *in-vivo* incorporation enabled the generation of **LmrR** mutant variants that permit a rapid optimization and a precise positioning of the bipyridine ligand (BpyAla) inside the **LmrR**. Application of these artificial metalloenzymes in the catalytic enantioselective Friedel-Crafts alkylation of  $\alpha,\beta$ -unsaturated acyl imidazole **8** with various

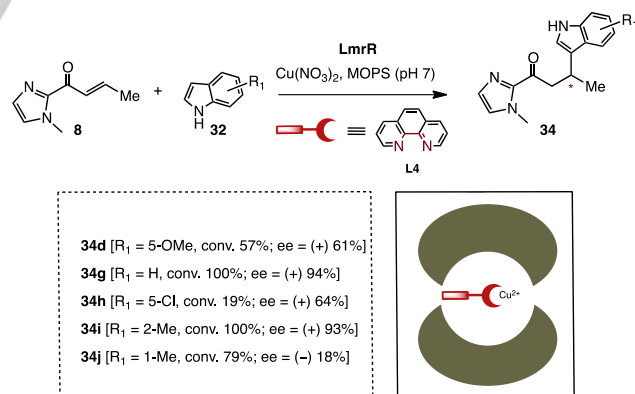


**Figure 26.** **QacR**-based asymmetric Friedel-Crafts alkylation.

indoles led to ees up to 83% and paved the way for directed evolution of artificial metalloenzymes (Figure 26).

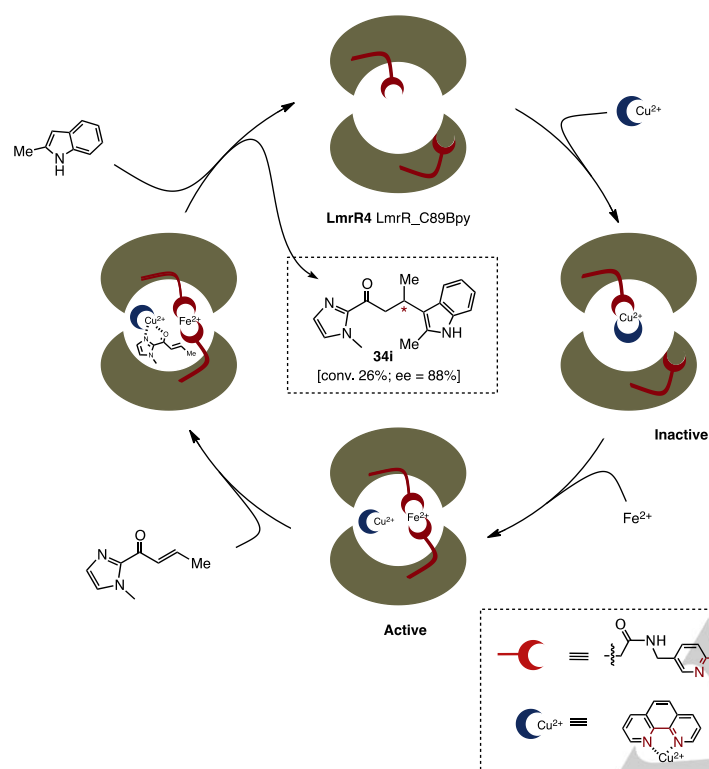
The concept was later extended to three other proteins (**QacR**, **CgmR** and **RamR**) belonging to the **TetR** family of Multidrug Resistance Regulators.<sup>[69]</sup> Among them, one **QacR** mutant induced up to 94% ee in the Cu(II)-catalysed enantioselective Friedel-Crafts alkylation with a reverse of selectivity compared to other mutants including **LmrR**-based artificial metalloenzymes.

The supramolecular assembly of an **LmrR**-based artificial metalloenzyme was also envisioned by combining a **LmrR** variant with a Cu(II) complex of phenanthroline ligand **L4** (Figure 27).<sup>[70]</sup> The rationale behind the study was based on the ability of planar aromatic ligands to bind to the **LmrR** hydrophobic pocket. The supramolecular approach proved to be superior to the covalent approach with ees up to 94% in the Cu(II)-catalysed enantioselective Friedel-Crafts alkylation.



**Figure 27.** Supramolecular approach for the **LmrR**-based asymmetric Friedel-Crafts alkylation.

These results prompted the authors to design a metal regulated **LmrR**-based artificial metalloenzyme combining both supramolecular and covalent anchoring strategies. The catalytic site was composed of the Cu(II)/**L4** ligand described above, while the regulatory site was introduced covalently by standard site directed mutagenesis technique<sup>[71]</sup> resulting in the formation of a dimeric protein containing two metal binding moieties. (Figure 28).<sup>[72]</sup>



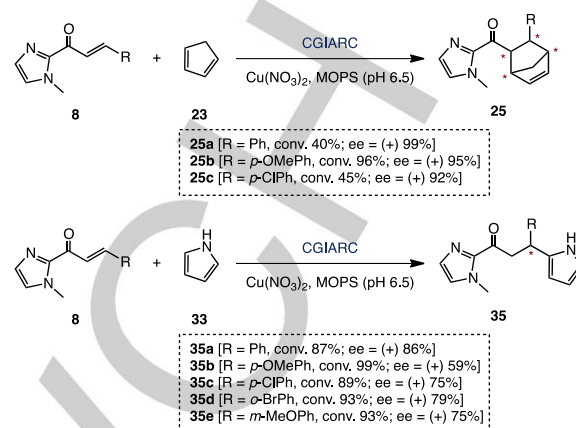
**Figure 29.** Metal ion regulated **LmrR**-based asymmetric Friedel-Crafts alkylation

In the absence of  $\text{Fe}^{2+}$  ions, the 2-2'-bipyridine ligand covalently linked to **LmrR** binds to the catalytically active  $\text{Cu(II)/L4}$  complex and deactivates the artificial metalloenzyme. The addition of  $\text{Fe}^{2+}$  ions traps the 2-2'-bipyridine ligands and frees the active  $\text{Cu}^{2+}$  complex, which subsequently reactivates the catalyst. Hence, when applied to the Friedel-Crafts alkylation between  $\alpha,\beta$ -unsaturated acyl imidazole **8** and 2-methyl indole, selectivities up to 88% ee were obtained thus demonstrating the synthetic potential of programmable artificial biohybrid catalysts.

### 3.2 Peptide

Recently, Hermann and co-workers identified cyclic peptides as a novel metalloprotein design containing only natural amino acids. The peptide scaffold was assembled by solid-phase peptide synthesis and cyclized by an intramolecular disulfide linkage between cysteines at the *N*- and *C*-termini. To determine the importance of a particular residue, a small library of 15 cyclic nonapeptides was synthesized and optimized using the alanine scanning technique.<sup>[73]</sup> The ability of the histidine to complex  $\text{Cu}^{2+}$  ions allowed to avoid the use of additional binding ligands.

These peptides were first evaluated on the  $\text{Cu(II)}$ -catalysed Diels-Alder cycloaddition, which led to the selection of shorter peptides that were ultimately used in the asymmetric  $\text{Cu(II)}$ -catalysed Friedel-Crafts alkylation. Interestingly, the truncated hexapeptide CGIARC afforded the highest selectivities with ees up to 99% in the Diels-Alder cycloaddition and 86% in the Friedel-Crafts alkylation (Figure 29).



**Figure 29.** Cyclic peptide-based asymmetric Diels-Alder and Friedel-Crafts alkylation.

## 4. Conclusion

While the use of acyl-imidazole in classical organic media are described by Campagne and co-workers in the back-to back accompanying mini-review, biohybrid catalysis has emerged recently as a particularly powerful tool for the enantioselective synthesis of chiral building blocks and  $\alpha,\beta$ -unsaturated acyl imidazoles have undoubtedly been a key to this success. Indeed, these substrates have not only been found to be highly versatile, they were also shown to exhibit interesting reactivity patterns and a high compatibility with a large variety of bioscaffolds. While various highly enantioselective C–C and C–O bond forming reactions have been unveiled, the use of  $\alpha,\beta$ -unsaturated acyl imidazoles in the context of asymmetric biohybrid catalysis is clearly still in its infancy and a plethora of new reactions are expected in the years to come.

## Acknowledgements

The Agence Nationale de la Recherche is gratefully acknowledged for funding project D-CYSIV (ANR-2015-CE29-0021-01).

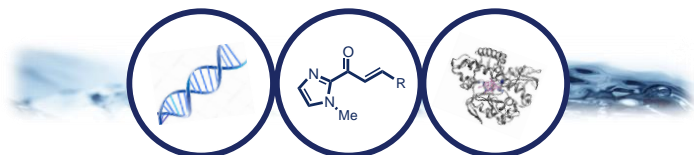
**Keywords:** Biohybrid catalysis • DNA • Acyl imidazole • Artificial metalloenzyme

- [1] T. Komnenos, *Justus Liebigs Ann. Chem.* **1883**, 218, 145-167.
- [2] A. Michael, *J. Für Prakt. Chem.* **1887**, 35, 349-356.
- [3] C. Hui, F. Pu, J. Xu, *Chem Eur J* **2016**, 23, 4023-4036.
- [4] K. Zheng, X. Liu, X. Feng, *Chem. Rev.* **2018**, 118, 7586-7656.
- [5] H. Brunner, B. Hammer, *Angew. Chem. Int. Ed. Engl.* **1984**, 23, 312-313.
- [6] A. Bernardi, G. Colombo, C. Scolastico, *Tetrahedron Lett.* **1996**, 37, 8921-8924.
- [7] G. Desimoni, G. Faita, K. A. Jørgensen, *Chem. Rev.* **2006**, 106, 3561-3651.

- [8] K. Itoh, **2002**, 124, 13394-13395.
- [9] D. A. Evans, T. Rovis, M. C. Kozlowski, C. W. Downey, J. S. Tedrow, **2000**, 122, 9134-9142.
- [10] D. A. Evans, S. J. Miller, T. Lectka, P. von Matt, *J. Am. Chem. Soc.* **1999**, 121, 7559-7573.
- [11] D. A. Evans, J. S. Johnson, E. J. Olhava, *J. Am. Chem. Soc.* **2000**, 122, 1635-1649.
- [12] J. Zhou, Y. Tang, *Chem Commun* **2004**, 4, 432-433.
- [13] K. B. Jensen, J. Thorhauge, R. G. Hazell, K. A. Jørgensen, *Angew. Chem. Int. Ed.* **2001**, 40, 160-163.
- [14] J. Zhou, M.-C. Ye, Z.-Z. Huang, Y. Tang, *J. Org. Chem.* **2004**, 69, 1309-1320.
- [15] C. Palomo, M. Oiarbide, F. Dias, A. Ortiz, A. Linden, *J. Am. Chem. Soc.* **2001**, 123, 5602-5603.
- [16] D. A. Evans, K. A. Scheidt, J. N. Johnston, M. C. Willis, *J. Am. Chem. Soc.* **2001**, 123, 4480-4491.
- [17] D. A. Evans, K. R. Fandrick, H.-J. Song, *J. Am. Chem. Soc.* **2005**, 127, 8942-8943.
- [18] D. A. Evans, K. R. Fandrick, H.-J. Song, K. A. Scheidt, R. Xu, *J. Am. Chem. Soc.* **2007**, 129, 10029-10041.
- [19] M. Okamoto, S. Ohta, S. Hayakawa, H. Moriwaki, S. Tsuboi, *Heterocycles* **1985**, 23, 1759-1764.
- [20] N. Barthes, C. Grison, *Bioorganic Chem.* **2012**, 40, 48-56.
- [21] L. Coudray, R. M. de Figueiredo, S. Duez, S. Cortial, J. Dubois, *J. Enzyme Inhib. Med. Chem.* **2009**, 24, 972-985.
- [22] J. Wang, E. Benedetti, L. Bethge, S. Vonnhoff, S. Klussmann, J.-J. Vasseur, J. Cossy, M. Smietana, S. Arseniyadis, *Angew. Chem. Int. Ed.* **2013**, 52, 11546-11549.
- [23] M. C. Myers, A. R. Bharadwaj, B. C. Milgram, K. A. Scheidt, *J. Am. Chem. Soc.* **2005**, 127, 14675-14680.
- [24] S. Drissi-Amraoui, T. E. Schmid, J. Lauberteaux, C. Crévisy, O. Baslé, R. M. de Figueiredo, S. Halbert, H. Gérard, M. Mauduit, J.-M. Campagne, *Adv. Synth. Catal.* **2016**, 358, 2519-2540.
- [25] S. Ohta, S. Hayakawa, K. Nishimura, M. Okamoto, *Chem. Pharm. Bull. (Tokyo)* **1987**, 35, 1058-1069.
- [26] A. Miyashita, Y. Suzuki, I. Nagasaki, C. Ishiguro, K. Iwamoto, T. Higashino, *Chem. Pharm. Bull. (Tokyo)* **1997**, 45, 1254-1258.
- [27] D. A. Evans, K. R. Fandrick, *Org. Lett.* **2006**, 8, 2249-2252.
- [28] K. Yamamura, E. T. Kaiser, *J. Chem. Soc. Chem. Commun.* **1976**, 830-831.
- [29] M. E. Wilson, G. M. Whitesides, *J. Am. Chem. Soc.* **1978**, 100, 306-307.
- [30] F. Schwizer, Y. Okamoto, T. Heinisch, Y. Gu, M. M. Pellizzoni, V. Lebrun, R. Reuter, V. Köhler, J. C. Lewis, T. R. Ward, *Chem. Rev.* **2018**, 118, 142-231.
- [31] G. Roelfes, B. L. Feringa, *Angew. Chem. Int. Ed.* **2005**, 44, 3230-3232.
- [32] M. Gao, R. Ye, W. Shen, B. Xu, *Org. Biomol. Chem.* **2018**, 16, 2602-2618.
- [33] N. Duchemin, I. Heath-Apostolopoulos, M. Smietana, S. Arseniyadis, *Org. Biomol. Chem.* **2017**, 15, 7072-7087.
- [34] A. J. Boersma, B. L. Feringa, G. Roelfes, *Org. Lett.* **2007**, 9, 3647-3650.
- [35] D. A. Evans, H.-J. Song, K. R. Fandrick, *Org. Lett.* **2006**, 8, 3351-3354.
- [36] D. Coquière, B. L. Feringa, G. Roelfes, *Angew. Chem. Int. Ed.* **2007**, 46, 9308-9311.
- [37] Y. Li, C. Wang, G. Jia, S. Lu, C. Li, *Tetrahedron* **2013**, 69, 6585-6590.
- [38] A. J. Boersma, B. L. Feringa, G. Roelfes, *Angew. Chem. Int. Ed.* **2009**, 48, 3346-3348.
- [39] S. Park, K. Ikehata, R. Watabe, Y. Hidaka, A. Rajendran, H. Sugiyama, *Chem. Commun.* **2012**, 48, 10398-10400.
- [40] R. P. Megens, G. Roelfes, *Org. Biomol. Chem.* **2010**, 8, 1387-1393.
- [41] R. P. Megens, G. Roelfes, *Chem. Commun.* **2012**, 48, 6366-6368.
- [42] (a) A. J. Boersma, D. Coquière, D. Geerdink, F. Rosati, B. L. Feringa, G. Roelfes, *Nat. Chem.* **2010**, 2, 991-995. (b) F. Rosati, G. Roelfes, *ChemCatChem* **2011**, 3, 973-977.
- [43] A. J. Boersma, J. E. Klijjn, B. L. Feringa, G. Roelfes, *J. Am. Chem. Soc.* **2008**, 130, 11783-11790.
- [44] G. P. Petrova, Z. Ke, S. Park, H. Sugiyama, K. Morokuma, *Chem. Phys. Lett.* **2014**, 600, 87-95.
- [45] K. Amirbekyan, N. Duchemin, E. Benedetti, R. Joseph, A. Colon, S. A. Markarian, L. Bethge, S. Vonnhoff, S. Klussmann, J. Cossy, J.-J. Vasseur, S. Arseniyadis, M. Smietana, *ACS Catal.* **2016**, 6, 3096-3105.
- [46] H. Zhou, D. Chen, J. K. Bai, X. L. Sun, C. Li, R. Z. Qiao, *Org. Biomol. Chem.* **2017**, 15, 6738-6745.
- [47] N. Duchemin, E. Benedetti, L. Bethge, S. Vonnhoff, S. Klussmann, J.-J. Vasseur, J. Cossy, M. Smietana, S. Arseniyadis, *Chem Commun* **2016**, 52, 8604-8607.
- [48] L. Gjonaj, G. Roelfes, *ChemCatChem* **2013**, 5, 1718-1721.
- [49] S. Park, L. Zheng, S. Kumakiri, S. Sakashita, H. Otomo, K. Ikehata, H. Sugiyama, *ACS Catal.* **2014**, 4, 4070-4073.
- [50] S. Park, I. Okamura, S. Sakashita, J. H. Yum, C. Acharya, L. Gao, H. Sugiyama, *ACS Catal.* **2015**, 5, 4708-4712.
- [51] S. Park, K. Ikehata, H. Sugiyama, *Biomater. Sci.* **2013**, 1, 1034-1036.
- [52] S. Sakashita, S. Park, H. Sugiyama, *Chem. Lett.* **2017**, 46, 1165-1168.
- [53] E. Benedetti, N. Duchemin, L. Bethge, S. Vonnhoff, S. Klussmann, J.-J. Vasseur, J. Cossy, M. Smietana, S. Arseniyadis, *Chem. Commun.* **2015**, 51, 6076-6079.
- [54] J. L. Huppert, *FEBS J.* **2010**, 277, 3452-3458.
- [55] G. N. Parkinson, M. P. H. Lee, S. Neidle, *Nature* **2002**, 417, 876-880.
- [56] A. Ambrus, D. Chen, J. Dai, T. Bialis, R. A. Jones, D. Yang, *Nucleic Acids Res.* **2006**, 34, 2723-2735.
- [57] K. N. Luu, A. T. Phan, V. Kuryavyi, L. Lacroix, D. J. Patel, *J. Am. Chem. Soc.* **2006**, 128, 9963-9970.
- [58] L.-X. Wang, J.-F. Xiang, Y.-L. Tang, *Adv. Synth. Catal.* **2015**, 357, 13-20.
- [59] S. Roe, D. J. Ritson, T. Garner, M. Searle, J. E. Moses, *Chem. Commun.* **2010**, 46, 4309-4311.
- [60] C. Wang, G. Jia, J. Zhou, Y. Li, Y. Liu, S. Lu, C. Li, *Angew. Chem. Int. Ed.* **2012**, 51, 9352-9355.
- [61] C. Wang, G. Jia, Y. Li, S. Zhang, C. Li, *Chem. Commun.* **2013**, 49, 11161-11163.
- [62] J. Dai, M. Carver, C. PUNCHIHEWA, R. A. Jones, D. Yang, *Nucleic Acids Res.* **2007**, 35, 4927-4940.

- [63] Y. Li, M. Cheng, J. Hao, C. Wang, G. Jia, C. Li, *Chem. Sci.* **2015**, *6*, 5578-5585.
- [64] C. Wang, Y. Li, G. Jia, Y. Liu, S. Lu, C. Li, *Chem. Commun.* **2012**, *48*, 6232-6234.
- [65] H. Zhao, K. Shen, *Biotechnol. Prog.* **2016**, *32*, 891-898.
- [66] S. Dey, A. Jäschke, *Angew. Chem. Int. Ed.* **2015**, *54*, 11279-11282.
- [67] S. Dey, C. L. Rühl, A. Jäschke, *Chem. – Eur. J.* **2017**, *23*, 12162-12170.
- [68] X. Li, M. J. Sutcliffe, T. W. Schwartz, C. M. Dobson, *Biochemistry* **1992**, *31*, 1245-1253.
- [69] M. Lerch, V. Gafner, R. Bader, B. Christen, G. Folkers, O. Zerbe, *J. Mol. Biol.* **2002**, *322*, 1117-1133.
- [70] D. Coquière, J. Bos, J. Beld, G. Roelfes, *Angew. Chem. Int. Ed.* **2009**, *48*, 5159-5162.
- [71] J. Bos, A. García-Herraiz, G. Roelfes, *Chem. Sci.* **2013**, *4*, 3578-3582.
- [72] I. Drienovská, A. Rioz-Martínez, A. Draksharapu, G. Roelfes, *Chem. Sci.* **2015**, *6*, 770-776.
- [73] J. Bos, W. R. Browne, A. J. M. Driessen, G. Roelfes, *J. Am. Chem. Soc.* **2015**, *137*, 9796-9799.
- [74] M. Bersellini, G. Roelfes, *Dalton Trans.* **2017**, *46*, 4325-4330.
- [75] M. Bersellini, G. Roelfes, *Org. Biomol. Chem.* **2017**, *15*, 3069-3073.
- [76] L. Zheng, A. Marcozzi, J. Y. Gerasimov, A. Herrmann, *Angew. Chem. Int. Ed.* **2014**, *53*, 7599-7603.

## REVIEW



$\alpha,\beta$ -Unsaturated acylimidazoles have become particularly useful building blocks in asymmetric catalysis. Surprisingly however, their use in asymmetric biohybrid catalysis as bidentate substrates able to interact with artificial metalloenzymes has only recently emerged. This review highlights the current efforts made in the field.

*Justine Mansot, Jean-Jacques Vasseur, Stellos Arseniyadis,\* and Michael Smietana\**

**Page No. – Page No.**

**$\alpha,\beta$ -Unsaturated 2-Acyl-Imidazoles in Asymmetric Biohybrid Catalysis**