

Extreme geochemical variability through the dunitic transition zone of the Oman ophiolite: Implications for melt/fluid-rock reactions at Moho level beneath oceanic spreading centers

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1	Extreme geochemical variability through the dunitic transition zone of the
2	Oman ophiolite: implications for melt/fluid-rock reactions at Moho level
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4	
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14	
15	Abstract
16	The Maqsad area in the Oman ophiolite exposes a >300 m thick dunitic mantle-crust
17	transition zone (DTZ) that developed above a mantle diapir. The Maqsad DTZ is primarily
18	made of "pure" dunites (olivine with scattered chromite and chromite seams) and
19	"impregnated" dunites, which exhibit a significant lithological variability, including various
20	kinds of clinopyroxene-, plagioclase-, orthopyroxene-, amphibole (hornblende/pargasite)-
21	bearing dunites. These minerals are interstitial between olivine grains and their variable
22	abundance and distribution suggest that they crystallized from a percolating melt. Generally
23	studied through <i>in-situ</i> mineral characterization, the whole rock composition of dunites is
24	poorly documented. This study reports on whole rock and minerals major and trace element

25 contents on 79 pure to variably impregnated dunites collected systematically along cross sections from the base to the top of the DTZ. In spite of its high degree of depletion, the 26 27 olivine matrix is selectively enriched in the most incompatible trace elements such as LREE, HFSE, Th, U, Rb and Ba. These data support the view that this enrichment has been acquired 28 29 early in the magmatic evolution of the DTZ, during the dunitization process itself. The 30 dissolution of orthopyroxene from mantle harzburgites enhanced by the involvement of 31 hydrothermal fluids produced low amounts of melts enriched in silica and in some trace 32 elements that re-equilibrated with the olivine matrix. This pristine signature of the DTZ dunite 33 was eventually variably altered by percolation of melts with a Mid-Ocean Ridge Basalt (MORB) affinity but displaying a wide spectrum of composition attributable to evolution by 34 35 fractional crystallization and hybridization with the silica enriched, hydrated melts. The 36 olivine matrix has been partially or fully re-equilibrated with these melts, smoothing the early 37 strong concave-upward REE pattern in dunite. The chemical variability in the interstitial minerals bears witness of the percolation of MORB, issued from the mantle decompression 38 39 melting, variably hybridized with melt batches produced within the DTZ by melt-rock 40 reaction and poorly homogenized before reaching the lower crust. Our results lead to the 41 conclusion that pure and impregnated dunites are end-members that recorded different stages 42 of the same initial igneous processes: pure dunites are residues left after extraction of a 43 percolating melt while impregnated dunites correspond to a stage frozen before complete melt 44 extraction. Therefore dunites trace elements contents allow deciphering the multi-stage 45 processes that led to their formation at the mantle-crust transition zone.

46

47 Keywords

48 Oman ophiolite; dunitic mantle-crust transition zone; trace elements; melt-rock reactions;
49 melt percolation; refertilization

50

51 **1. Introduction**

52 The boundary between the Earth mantle and the oceanic crust is underlined by a dunitic horizon in most ophiolites worldwide, regardless the tectonic setting in which they evolved 53 54 (mid-ocean ridges vs. supra-subduction zone) (e.g. Moores and Vine, 1971; Prinzhofer et al., 55 1980; Boudier and Coleman, 1981; Quick, 1981; Jan and Howie, 1981; Kelemen et al., 56 1997a; Bédard and Hébert, 1998; Ceuleneer and le Sueur, 2008; Bouilhol et al., 2009). Since dunites are observed in present-day oceans in different tectonic settings, associated to mantle 57 58 harzburgites and to troctolitic-gabbroic veins and sills like in the dunitic mantle-crust 59 transition zone (DTZ) (e.g. Arai and Matsukage, 1996; Dick and Natland, 1996; Parkinson 60 and Pearce, 1998; Savov et al., 2005; Godard et al., 2008), it is reasonable to suppose that 61 ophiolitic DTZ formed prior to the intra-oceanic thrusting leading to obduction.

62

A residual origin for dunites after very high (>40%) degrees of partial melting of the mantle 63 64 (Green and Ringwood, 1967) requires unrealistic thermal regimes, at least in the modern 65 Earth and in oceanic spreading contexts (Herzberg et al., 1983). Alternatively, dunite may 66 result from the accumulation of olivine grains issued from fractional crystallization from a 67 melt sufficiently rich in Mg to remain during a significant time in the fractionation field of 68 olivine alone (+/- spinel) (Bowen, 1915; O'Hara, 1965; Elthon, 1979). Finally, dunite may be replacive after interaction ("reactional melting") between melts and peridotite in dry (Bowen, 69 70 1927; Berger and Vannier, 1984) or hydrated systems (Bowen and Tuttle, 1949). The 71 interaction between harzburgite and a melt undersatured in silica accounts for the formation 72 of dunite through the dissolution of pyroxenes, leading to the enrichment in silica of the melt 73 with concomitant crystallization of olivine (Dick, 1977; Quick, 1981; Kelemen, 1990; Kelemen et al., 1992). Petrological evolutions across a densely sampled cross section along a 74

75 330 m thick DTZ of the Oman ophiolite revealed that the DTZ does essentially but not 76 exclusively derive from melt-peridotite interaction, its shallowest part (topmost 50 m) 77 showing fractional crystallization trends (Abily and Ceuleneer, 2013).

78

79 In the melt-rock reaction model, it was proposed that reactional melting and orthopyroxene 80 consumption leads to high intergranular permeability enhancing the efficiency of melt 81 extraction from the mantle (Ortoleva et al., 1987; Daines and Kohlstedt, 1994; Kelemen et al., 82 1995a; Kelemen et al., 1997b). In this way, the DTZ appears to be a reactive interface where 83 melts are focused, transformed and potentially accumulated, then distributed beneath the midocean ridge. Crystallization associated to melt migration through the mantle-crust transition 84 85 zone accounts for the wide diversity of lithological facies observed in the DTZ. Dunites evolve from "pure dunites", i.e. compacted at very high temperature when complete melt 86 87 extraction occurred, to "impregnated dunites", highly refertilized before incipient cooling and 88 that potentially leads in extreme cases to the formation of hybrid olivine-rich troctolites or 89 plagioclase-bearing wehrlites (e.g. Arai and Matsukage, 1996; Abily and Ceuleneer, 2013; 90 Sanfilippo and Tribuzio, 2013a). Although interstitial minerals in ophiolites or abyssal DTZ 91 are regularly found in equilibrium with melts that fed the overlying crust (e.g. Drouin et al., 92 2009; Ghosh et al., 2014; Nicolle et al., 2016), the Oman ophiolite DTZ was also percolated 93 by melts with more exotic compositions (Koga et al., 2001) that, for a part of them, may be 94 attributable to hybridization with a hydrothermal fluid component (Rospabé et al., 2017). 95 Especially the presence of orthopyroxene, amphibole, garnet and diopsides both interstitially 96 between olivine grains and in inclusion in chromite calls for early, high temperature, 97 hydration of the DTZ. In this frame, to define the origin of the DTZ has direct implications on 98 melt generation, transfer and evolution from the mantle to the crust, as well as on the deep 99 hydrothermal circulations that may influence both the chemical exchanges and the thermal

100 structure of the oceanic lithosphere.

101

102 Most geochemists use the trace element contents of interstitial clinopyroxene and plagioclase 103 (if any) scattered in dunites to build petrological scenario (e.g. Kelemen et al., 1995a; Nicolle 104 et al., 2016; Akizawa et al., 2016a). However, it concerns actually the genesis of 105 crystallization products from melt fractions that travelled through the dunites, which might be 106 risky to extrapolate to the genesis of the dunitic matrix itself. Very few whole rock analysis or 107 olivine trace element contents are available due to their very low content in olivine. The 108 sporadic studies providing trace elements data on dunite show that re-equilibration processes 109 occurred between the olivine matrix, selectively enriched in some trace elements, and the 110 melts percolating interstitially between olivine grains (e.g. Godard et al., 2000; Gerbert-111 Gaillard, 2002; Sanfilippo et al., 2014, 2017).

112

To better constrain the processes leading to the formation of both pure and impregnated dunitic products constituting the DTZ as well as to their chemical signature, we combined major and trace elements in whole rock and in mineral phases for 79 samples of dunite from the Oman ophiolite mantle-crust transition zone. We show that the important chemical variability recorded by the dunites at Moho level is issued from a combination of harzburgite dunitization, re-equilibration associated to melt percolation, and refertilization from unmixed and contrasted MORB-derived melt batches variably hybridized with hydrous fluids.

120

121 **2. Geological background**

122 2.1. Geology of the Oman ophiolite

The Oman ophiolite exposes a 30.000 km² remnant of the Tethyan Ocean. Its creation
occurred ~95-97 million years ago (Tilton et al., 1981; Tippit et al., 1981; Rioux et al., 2012)

125 and was directly followed by an oceanic detachment close to the ridge axis that eventually led 126 to its obduction on the Arabian margin (Lanphere, 1981; Boudier et al., 1985; Boudier et al., 127 1988; Montigny et al., 1988). The synoptic mapping of mafic-ultramafic dykes cropping out in the mantle section (Python and Ceuleneer, 2003; Python et al., 2008) revealed that two 128 129 main magmatic suites contributed to the igneous evolution of the ophiolite: (1) a MORB suite 130 mostly present in the southeastern massifs of the ophiolite (Nakhl, Sumail and Wadi Tayin) 131 and in some restricted areas in the north, and (2) a more widespread depleted-andesitic type 132 attributed either to subduction or to hydrated re-melting of the shallow lithosphere during 133 spreading (Pearce et al., 1981; Benoit et al., 1999; Python and Ceuleneer, 2003; Yamasaki et 134 al., 2006; MacLeod et al., 2013) (Fig. 1a).

135

136 Along the Oman ophiolite, spatially constrained vertical flow structures within the mantle 137 section were interpreted as former asthenospheric diapirs distributed along the oceanic ridge 138 (Ceuleneer et al., 1988; Nicolas et al., 1988; Jousselin et al., 1998; Nicolas et al., 2000). The 139 Sumail massif, focus of the present study, exposes in the Maqsad area a well preserved paleo 140 mantle diapir and its associated N130 paleo-spreading center (Ceuleneer, 1986; Ceuleneer et 141 al., 1988; Ceuleneer, 1991; Jousselin et al., 1998). This structure is located in the central part of the largest (80×25 km) MORB district of Oman (Python and Ceuleneer, 2003) (Fig. 1). 142 143 The MORB affinity has been evidenced by the crystallization sequence and confirmed by 144 trace elements and isotopic data on both mantle and crustal rocks (Benoit et al., 1996; 145 Ceuleneer et al., 1996; Godard et al., 2000; Koga et al., 2001; Godard et al., 2003; Clénet et al., 2010). 146

147

148 2.2. The dunitic transition zone around the Maqsad diapir

149 The thickness of the DTZ reaches 300 to 350 m above the central part of the Maqsad mantle

150 diapir (Rabinowicz et al., 1987; Boudier and Nicolas, 1995; Jousselin and Nicolas, 2000; 151 Abily and Ceuleneer, 2013) and thins progressively off-axis due to tectonic transposition by 152 mantle plastic flow as it diverges from the diapir (Ceuleneer, 1991). In this area the DTZ was 153 mainly described as the uppermost part of the mantle, modified after complete orthopyroxene 154 consumption resulting from MORB-peridotite interaction (Godard et al., 2000; Gerbert-155 Gaillard, 2002), while its upper part (top most ~50 m) displays all the characteristic of 156 cumulates (Abily and Ceuleneer, 2013). The interstitial plagioclase and clinopyroxene located 157 between olivine grains exhibit chemical signature in good agreement with a fractionation 158 from a percolating MORB (Koga et al., 2001; Abily and Ceuleneer, 2013; Nicolle et al., 159 2016). It was also recently evidenced that some horizons of the DTZ contain interstitial 160 orthopyroxene and pargasite that support the percolation of a hybrid liquid, i.e. a blend of 161 variably evolved MORB and of hydrous silica-rich liquids (Rospabé et al., 2017). The 162 ubiquitous occurrence of hydrothermal diopside and grossular, both in interstitial position 163 between olivine grains and enclosed in disseminated chromite together with orthopyroxene, 164 pargasite, biotite and aspidolite, suggests them as primary high temperature features and 165 further challenges the pure igneous origin of the DTZ. According to our observations, they are 166 not systematically associated to cracks affecting chromite grain, as would be expected if they 167 were low temperature alteration products resulting from serpentinization and Ca-rich 168 metasomatism having affected mafic and ultramafic rocks (e.g. Palandri and Reed, 2004; 169 Python et al., 2011; Akizawa et al., 2016b) (see also Tamura et al., 2014). The possible 170 involvement of high temperature hydrothermal fluids during the magmatic stage that edified 171 the DTZ allows reconsidering the former hypothesis of a hydrothermal origin for dunites, 172 originally proposed by Bowen and Tuttle (1949).

173

174 **3. Methods**

175 *3.1. Sampling strategy*

176 The Sumail massif was slightly tilted during the Miocene uplift of the Djebel Akhdar. Its regional dip does not exceed 10° to the SE. For this reason, to perform sampling along 177 178 distributed cross sections is a straightforward way to reconstruct the three-dimensional 179 structuration of the DTZ around the Maqsad diapir and its associated paleo-ridge axis. In this 180 study we report on whole rock and mineral major elements and whole rock trace elements for 181 79 samples, and on interstitial clinopyroxenes trace elements for 11 impregnated dunites. 182 Samples were collected systematically every 10 to 20 m vertically along four cross sections 183 (Fig. 1b). Their location and mineral contents are presented in Table 1.

184

185 *3.2. Analytical methods*

186 In situ mineral major element concentrations were acquired by electron microprobe using a 187 Cameca SX 100 (Microsonde Ouest, Brest, France) and a Cameca SXFive (Centre de 188 MicroCaractérisation Raimond Castaing, Toulouse, France) with a 20 kV accelerating 189 voltage, a beam current of 20 nA, an electron beam diameter of 1 µm for all analyses and a 190 counting time of 10 s on peak for each element and 5 s on backgrounds on both sides of the 191 peak. In situ clinopyroxene trace element concentrations were obtained by laser ablation-192 inductively coupled plasma-mass spectrometry (LA-ICP-MS) at the Institute of Earth 193 Sciences, University of Lausanne (Switzerland). We used a NewWave UP-193 ArF 194 Excimer based laser ablation system coupled with a Thermo ScientificTM ELEMENT XRTM 195 high resolution ICP-MS. Spot size ranged between 75 and 100 µm and operating conditions were a 20 Hz pulse rate and an energy density of 6.0 J/cm². The SRM612 NIST was used as 196 external standard following the values proposed by Jochum et al. (2011) and ²⁹Si was used as 197 198 internal standard.

199

200 Whole rock major element concentrations were obtained by X-ray Fluorescence Spectrometry 201 at the Central Analytical Facilities, Stellenbosch University (South Africa). Whole rock trace 202 element concentrations were acquired on a Thermo Scientific[™] ELEMENT XR[™] HR-ICP-203 MS at the Géosciences Environnement Toulouse laboratory (Observatoire Midi-Pyrénées), 204 Université of Toulouse III. The sample preparation consisted in a HF-HClO₄ digestion 205 procedure (Yokovama et al., 1999) combined with a thulium addition (Barrat et al., 1996). 206 The concentrations in Li and Large Ion Lithophile Elements (LILE; Rb, Sr, Cs and Ba) and of 207 most of transition elements (Cr, Mn, Co, Ni, Cu, Zn) and Ga were acquired after a direct 208 digestion/dilution method. Sc, V, Rare Earth Elements (REE) and Y, High Field Strength 209 Elements (HFSE; Zr, Hf, Nb, Ta, Ti) and U, Th and Pb contents were determined after the 210 double coprecipitation procedure detailed in Rospabé et al. (2018). BHVO-2 was used as 211 calibrator following the values proposed by Barrat et al. (2012). The yield was monitored and 212 concentrations calculated using the Tm anomaly appearing in the samples REE pattern (Barrat 213 et al., 1996). The international rock standards UB-N (serpentinite) and DTS-2B (dunite) were 214 analysed as unknown during the course of this study. They are given in Appendix A together 215 with the samples major and trace element compositions.

216

217 4. Lithological diversity in the Maqsad DTZ

Olivines exhibit a recrystallized equigranular mosaic texture with frequent triple junctions at 120° (Fig. 2a). Its average grain size is about 1 mm. All samples are affected to some degree by serpentinization, which ranges from about 30 to 60%. Submillimeter round-shaped chromite grains are interstitial between olivine grains. They are generally disseminated but chromite schlierens occur in some samples. Small grains of clinopyroxene are also frequently observed at olivine triple junctions or along the olivine grain rims, with a grain size no larger than few tens of microns. Samples with less than 0.5% of clinopyroxene are considered as 226

227 The remaining 41 samples contain well-developed oikocrystic of clinopyroxene and/or other 228 interstitial minerals. These interstitial phases are classically interpreted as crystallization 229 products fractionated from a melt that percolated through the dunitic matrix and are referred 230 to as "impregnated dunites" (Benn et al., 1988; Boudier and Nicolas, 1995; Koga et al., 2001; 231 Abily and Ceuleneer, 2013). Two contrasted clinopyroxene compositions are observed; one 232 consistent with an igneous origin and the other one calling for crystallization from 233 hydrothermal fluids or water-rich melts (Python et al., 2007; Rospabé et al., 2017). 234 Accordingly, we use "clinopyroxene" and "diopside" to distinguish clinopyroxene with 235 igneous and hydrothermal origin respectively. Clinopyroxene is the most common 236 impregnating phase (14 samples are cpx-bearing dunites). Its texture changes with its modal 237 abundance, from small interstitial at olivine grains boundaries (up to hundreds microns width 238 in size) to oikocrystic crystals (millimeter to up to centimetre in size) (Fig. 2b). Smaller 239 interstitial diopsides are also sometimes observed both in pure and impregnated dunites, 240 showing different texture and chemical content than the clinopyroxene oikocrysts (Fig. 2c). 241 Plagioclase appears as a well-developed interstitial intercumulus phase, generally as 242 oikocrysts (Fig. 2d), and is regularly associated to clinopyroxene, characterizing the pl/cpx-243 bearing dunites (14 samples). Plagioclase-bearing cpx-free dunites are rarely observed (one 244 sample only). Plagioclase/olivine contacts are also regularly underlined by a clinopyroxene 245 corona. In addition to clinopyroxene and plagioclase, the dunites collected in the DTZ from 246 the Maqsad area show orthopyroxene oikocrysts in pl/cpx-bearing dunites (7 samples) (Fig. 247 2e), and amphibole, observed as subhedral crystals in few samples with no preference to the 248 other minerals present (7 samples) (Fig. 2f). Interstitial subhedral grossular garnet was 249 observed in 12 impregnated dunites (Rospabé et al., 2017). All these interstitial minerals are

also present as round-shaped primary inclusions in chromite grains together with micas (phlogopite and aspidolite), the latter being never observed interstitially within the dunitic matrix.

253

254 **5. Results**

255 5.1. Whole rock major and minor elements

The dunites from the DTZ plot above the mantle fractionation array in the MgO/SiO₂ vs. 256 257 Al₂O₃/SiO₂ diagram (Fig. 3a). Pure dunites and slightly impregnated ones (<5% of cpx or cpx 258 + pl) exhibit a more refractory characteristic than the most depleted harzburgites from the 259 mantle section with higher MgO/SiO₂ (>1.1) and lower Al₂O₃/SiO₂ (<0.02) ratios. 260 Impregnated samples are generally characterized by a progressive decrease of MgO/SiO₂ and an increase of Al₂O₃/SiO₂ that reflects an increasing amount of clinopyroxene and 261 262 plagioclase. The loss on ignition (LOI) is comprised between 6.9 and 10.9 wt.% for pure 263 dunites and 4.7 and 10.8 wt.% for impregnated dunites, and reflects primarily the degree of 264 serpentinization that is similar in both dunite types (Appendix A). The variation of FeO and 265 MgO in pure and slightly impregnated dunites perfectly mimics the stoichiometric variation 266 of the olivine Fe-Mg composition (FeO + MgO = 66.67 in mol%) while other higher 267 impregnated samples contain a lower MgO content similar to mantle harzburgites (mainly 268 ~43-47 wt.%) (Fig. 3b).

269

Low CaO and Al_2O_3 also characterize the pure and slightly impregnated dunites with concentrations lower than 1 wt.% (Fig. 3c). The higher CaO and Al_2O_3 contents are strongly correlated to the abundance of impregnating minerals. The modal proportion of clinopyroxene controls the CaO content that increases fast with increasing amount of cpx ($Al_2O_3/CaO < 1$ in cpx-bearing dunites) while plagioclase and amphibole exert the main control on the Al_2O_3 275 content in other impregnated dunites ($Al_2O_3/CaO>1$).

276

277 In pure dunites the Mg# (100 \times molar Mg/(Mg + Fe_{total})) is more variable (from 86 to 92.3, except a pure dunite containing schlierens of chromites with a whole rock Mg# = 83.4) than 278 279 in impregnated dunites (from 87.2 to 90.9) and peridotites from the mantle section (from 89.8 280 to 91.6). The dunites from the DTZ also display a much more variable Ni (1207-2870 μ g/g) 281 and a higher Co (110-155 μ g/g) contents than mantle peridotites (Ni = 1747-2671 μ g/g; Co = 282 95-122 µg/g) (Fig. 3d and 3e). These compositions are not clearly related to the degree of 283 impregnation minerals with high Ni and Co concentrations also observed in the most 284 impregnated samples.

285

286 5.2 Mineral major elements

287 The Fo in olivine $(100 \times \text{molar Mg}/(\text{Mg} + \text{Fe}_{\text{total}}))$ in dunites from the DTZ generally decreases from 92.5 to 87.8 concomitantly with the NiO content (0.45 to 0.19 wt.%) (Fig. 4a). 288 289 This compositional range is broader than in the peridotites from the mantle section 290 (89.5<Fo<92.1; 0.25<NiO<0.53 wt.%). The CaO content in olivine, mainly <0.05 wt.% in 291 mantle harzburgites, reaches 0.32 wt.% in the DTZ. Higher CaO values are generally 292 observed in olivine grains from pure dunites, also displaying the lower Fo and NiO contents 293 (Fig. 4b), and may consequently not be attributed to re-equilibration with other CaO-rich minerals. Fo in olivine and Mg# in whole rocks are highly correlated (Fig. 4c). 294

295

The Cr# (100 × molar Cr/(Cr + Al)) variation range in chromite is more restricted in dunites from the DTZ (43.3-63.4) than in harzburgites from the mantle section (20.6-70). Cr# gently decreases with the Fo and progressively shifts from the olivine-spinel mantle array (OSMA) (Arai, 1987, 1994) (Fig. 4d). Mg# (100 × molar Mg/(Mg + Fe²⁺)) in chromite is quite similar between the different types of dunite (43.3-63.4) while the TiO₂ content is on the contrary much more variable depending on the host lithological facies (Fig. 4e), being lower (0.20-0.60 wt.%) in pure dunites than in impregnated ones (mainly comprised between 0.15 and 0.9 wt.% and occasionally reaching 1.3 wt.%). On the contrary YFe^{3+} (100 × molar $Fe^{3+}/(Cr + Al$ + Fe^{3+})) is slightly higher in chromite from the pure dunites (Appendix B).

305

306 Minute clinopyroxenes were observed at the triple junction of olivine grains in what we 307 defined as pure dunites. They exhibit a variable Mg# (86.9-94.2) and a CaO content 308 exceeding 24.5 wt.% that contrast with the clinopyroxene oikocrysts from the impregnated dunites (Mg# = 89.6-92.9; CaO = 21.1-24.8 wt.%) (Appendix B). All the clinopyroxenes 309 310 display a wide variation range in TiO₂ (0.16-0.71 wt.%) (Fig. 4f), Cr_2O_3 (0.79-1.32 wt.%) and 311 Al₂O₃ (2.6-4.9 wt.%) (Appendix B). Zoning was observed in few clinopyroxenes only, with 312 the concomitant increase of Mg#, SiO₂ and TiO₂ together with the decrease of Al₂O₃ and 313 Cr₂O₃ from the core to the rim. Small interstitial diopside composition ranges between 314 igneous and hydrothermal end-members, with higher Mg# and lower TiO₂ (Fig. 4f), Cr₂O₃ 315 and Al₂O₃ contents (see also Rospabé et al., 2017).

316

Oikocrystic orthopyroxene grains observed in interstitial position in dunites present a high TiO₂ content (0.15-0.23 wt.%), specific to the dunites from the Maqsad area (Rospabé et al., 2017) (Appendix B). They display Mg# and Al₂O₃ contents varying from 88.8 to 90.8 and 1.15 to 2.22 wt.% respectively. As clinopyroxene, orthopyroxene rarely exhibits zoned compositions with the increase of SiO₂ concomitantly to the decrease of Al₂O₃, and Cr₂O₃ from the core to the rim. Mg# is quite variable as MgO and FeO both increase with SiO₂.

323

324 Plagioclase was not systematically analysed in each pl-bearing dunites due to its higher

325 degree of alteration than other phases. It presents rather primitive compositions in 326 (opx/)pl/cpx-bearing dunites (An (100 × molar Ca/(Ca + Na + K)) = 86.3-90.1) while being 327 more evolved in one amphibole/pl/cpx-bearing dunite (An = 74.6) (Appendix B).

328

Interstitial amphiboles in the studied samples are pargasites, pargasitic- or edenitichornblende. They display Mg# ranging from 85.7 to 91.4 and extremely variable Al_2O_3 (11.0-16.3 wt.%), Na₂O (1.98-3.75 wt.%), TiO₂ (0.02-1.24 wt.%) and Cr₂O₃ (0.01 to 1.81 wt.%) contents (Appendix B).

333

334 5.3. Whole rock trace elements

335 The dunites from the DTZ exhibit variable trace element compositions (Appendix A). Pure 336 dunites are highly depleted in lithophile trace elements with concentrations below both the 337 chondritic and the Primitive Mantle (PM) values (mostly comprised between 0.001 and 0.1 times the PM concentrations). Pure dunites are characterized by U- or V-shape REE 338 339 chondrite-normalized patterns (Fig. 5a), reflecting the major MREE (Sm, Eu and Gd) 340 depletion relative to LREE (La_{CN}/Sm_{CN} <3.76; 1.9 in average) and HREE (0.04< Gd_{CN}/Yb_{CN} 341 <0.28). We define the U- or V-shape appellation depending on the presence or absence of 342 negative Eu anomaly. The (Eu/Eu*)_{CN} ratio ranges between 0.22 and 1 in 29 samples, while 9 343 samples exhibit a positive Eu anomaly with $(Eu/Eu^*)_{CN} < 2.4$ $((Eu/Eu^*)_N = Eu_{CN}/\sqrt{(Sm_{CN} \times 10^{-5})})_N$ Gd_{CN})). Pure dunites display relative homogeneous HREE concentrations (Yb_{CN} = 0.10-0.24) 344 345 in comparison to more variable MREE ($0.005 < Gd_{CN} < 0.059$; $0.004 < Sm_{CN} < 0.065$) and especially LREE (0.004< La_{CN} <0.17). The depletion from HREE to MREE is not linear with 346 347 several samples exhibiting instead concave-upward patterns with a progressive Gd_{CN}/Dy_{CN} 348 increase for a constant Er_{CN}/Yb_{CN}. Pure dunites U-shaped PM-normalized multi-elements patterns exhibit significant enrichments in LILE, Th, U, Nb and Ta relative to LREE 349

350 (Rb_{PMN}/La_{PMN} = 1.5-85; Th_{PMN}/La_{PMN} = 0.7-5.3) and strong Pb and Sr positive anomalies 351 similar to mantle peridotites (2.19< Pb_{PMN}/Ce_{PMN} <1294) (Fig. 5b). Zr and Hf are generally 352 enriched relative to MREE (Zr_{PMN}/Sm_{PMN} = 0.6-27; 2.9 in average) as well as Ti that displays 353 a strong positive anomaly (1.3< Ti_{PMN}/Gd_{PMN} <76).

354

355 Cpx-bearing dunites exhibit variable REE and multi-element patterns partly attributable to 356 variations in the interstitial clinopyroxene content. Slightly impregnated ones (<5% cpx) 357 exhibit linear LREE-depleted or slightly concave-upward REE patterns ($La_{CN}/Yb_{CN} = 0.06$ -358 0.97) with similar HREE concentrations (Yb_{CN} = 0.12-0.29) and more restricted LREE and MREE variation ranges (La_{CN} = 0.01-0.23; Gd_{CN} = 0.02-0.12) than pure dunites (Fig. 5c). 359 360 Dunites containing a higher amount of clinopyroxene have higher HREE contents ($Yb_{CN} =$ 0.56-0.83). They exhibit clear convex-upward patterns with a nearly flat slope of the HREE 361 362 segment ($Gd_{CN}/Yb_{CN} = 0.74-0.85$) followed by a progressive depletion from MREE to LREE $(La_{CN}/Sm_{CN} = 0.06-0.09)$. It is worth noting that the cpx-bearing dunites display La and Ce 363 364 contents within the variation range of pure dunite ($La_{CN} = 0.01-0.23$). This observation may 365 be extended to other incompatible trace elements like Th, U, HFSE that are not more 366 concentrated in cpx-bearing dunites than in pure dunites, excepting one sample that displays spectacular HFSE positive anomalies (Fig. 5d). Cpx-bearing dunites exhibit similar 367 368 enrichment in LILE relative to LREE and the same strong positive Pb and Sr anomalies than 369 in pure dunites. The positive Ti anomaly observed for the slightly impregnated samples 370 disappears for the more impregnated ones in which the Ti content is buffered around 0.25 371 times the PM value.

372

373 The pl-bearing dunite displays a W-shaped REE pattern with strong depletion in MREE 374 ($Sm_{CN} = 0.005$; $Gd_{CN} = 0.003$) relative to LREE ($La_{CN}/Sm_{CN} = 4.3$) and HREE ($Gd_{CN}/Yb_{CN} =$ 375 0.002) bordering a strong positive Eu anomaly ($(Eu/Eu^*)_{CN} = 54$) (Fig. 5e). Other pl/cpx-376 bearing chondrite-normalized REE patterns evolve from concave-upward to convex upward 377 with the increasing amount of clinopyroxene. This evolution is accounted by a progressive 378 decrease of the HREE segment slope (Gd_{CN}/Yb_{CN} from 0.17 to 0.63). They are also 379 characterized, in regard to pure and cpx-bearing dunites, by higher Li and Rb contents, a 380 higher range of Cs, a well depletion in their Th and U contents, and a slighter positive Pb 381 anomaly (Fig. 5f).

382

383 Opx/pl/cpx-bearing dunites and amphibole-bearing dunites (\pm opx, pl and cpx) have REE 384 contents similar to pl/cpx-bearing dunites with patterns intermediate between highly 385 impregnated pl-bearing and cpx-bearing dunites (Fig. 5g). They also display similar PM-386 normalized multi-elements patterns than pl/cpx-bearing dunites, with Th, U, and with few 387 exceptions Nb and Ta contents similar to LREE (averaged Th_{PMN}/La_{PMN} = 0.9) (Fig. 5h).

388

389 5.4. Clinopyroxene minor and trace elements

390 Trace elements concentrations were obtained for unzoned crystals of clinopyroxene in 11 391 impregnated dunites (2 cpx-bearing, 6 pl/cpx-bearing, 3 opx/pl/cpx-bearing, and 1 392 amph/opx/pl/cpx-bearing). Their composition is homogeneous within a given sample. In 393 clinopyroxenes, lower trace elements contents are observed in the amph/opx/pl/cpx-bearing dunite while higher contents much characterize cpx-bearing dunites; clinopyroxenes in 394 395 (opx/)pl/cpx-bearing dunites display both low and high trace elements concentrations. 396 Regarding the chondrite-normalized REE patterns, all the clinopyroxenes are depleted in 397 LREE relative to MREE ($La_{CN}/Sm_{CN} = 0.03-0.32$) and the slope in HREE is highly variable, 398 from strongly positive to moderately negative ($Gd_{CN}/Yb_{CN} = 0.61-2.28$) (Fig. 6a). The most 399 LREE-depleted patterns are rather observed in (amph/)opx/pl/cpx-bearing dunites whatever

the REE concentrations, while the less depleted ones are observed in cpx-bearing dunites. Negative anomalies in Nb and Ta (Nb_{PMN}/La_{PMN} = 0.08-1) relative to LREE and in Zr and Hf relative to MREE ($Zr_{PMN}/Sm_{PMN} = 0.14-0.43$) are systematically observed whatever the host lithology (Fig. 6b). With few exceptions, negative anomalies in Pb and Sr relative to LREE-MREE are also observed as well as in Ti for the most enriched clinopyroxenes.

405

406 **6. Discussion**

407 The dunitic transition zone from the Magsad area exposes various types of dunites, 408 characterized from their mineralogical content (Boudier and Nicolas, 1995; Koga et al., 2001; 409 Abily and Ceuleneer, 2013; Rospabé et al., 2017) and from their geochemical diversity within 410 each lithological facies. Previous studies proposed different processes to explain the 411 formation of pure and impregnated dunites and/or their geochemical signature: (1) 412 serpentinization or weathering, (2) fractional crystallization and accumulation of olivine and 413 other minerals, (3) melt-rock reactions comprising harzburgite reactional melting and re-414 equilibration related to melt percolation and, (4) refertilization associated to melt transport. In 415 the next sections we will discuss the relative influence of each process in regard to the 416 geochemical variations revealed by our study, then we will propose a more general model for 417 the Maqsad DTZ genesis.

418

419 6.1. Serpentinization or secondary weathering effects

The dunites are partly serpentinized with a serpentine abundance of about 40 to 50% in average, according to thin sections observation and loss of ignition (LOI) values. Serpentinization is known to be nearly isochemical regarding the bulk rock major element concentrations (e.g. Coleman and Keith, 1971; Deschamps et al., 2013). In our samples, the LOI is not or poorly correlated with any major element contents attesting that the 425 serpentinization did not extensively affect bulk rock major element concentrations (see also 426 Rospabé, 2018). More specifically, the samples plot above the terrestrial array in the 427 MgO/SiO₂ vs. Al₂O₃/SiO₂ diagram (Fig. 3a) and do not show any evidence for MgO loss or SiO₂ enrichment as reported in pervasively serpentinized abyssal peridotites or talc-bearing 428 429 serpentinites (Snow and Dick, 1995; Bach et al., 2004; Paulick et al., 2006; Boschi et al., 2008; Malvoisin, 2015). The perfect consistency between the Fo in olivine and the Mg# in 430 431 whole rock (Fig. 4c) further demonstrates the absence of influence of serpentinization on the 432 bulk dunite major element compositions.

433

434 Regarding trace elements, with few exceptions, the La content is strongly correlated to the Th 435 content, in both pure and impregnated dunites (Fig. 7a). The Th, as well as HFSE, are 436 immobile during post-magmatic aqueous alteration (You et al., 1996; Kogiso et al., 1997; 437 Niu, 2004; Paulick et al., 2006). Thus, correlated enrichments of La with Th testify that LREE 438 enrichments derived from magmatic processes, as seen in mantle harzburgites from the 439 Sumail massif (Godard et al., 2000; Gerbert-Gaillard, 2002). A poorer but significant 440 correlation exists between HFSE (Zr, Hf, Nb, Ta) and U contents on one hand and Th and 441 LREE contents on the other (Fig. 7b, c and d). Rb and Ba seem to be vaguely correlated to 442 HFSE in pure dunites (especially to Zr and Hf) as well as to Th (Fig. 7e and f) and thus were 443 probably enriched during an initial igneous stage before being potentially affected again 444 during secondary weathering processes. Other LILE, such as Sr, and Pb contents are not 445 correlated to the LOI but the absence of correlation with Th suggests that their enrichment 446 could be related to a secondary alteration/serpentinization processes post-dating the igneous 447 history. This is confirmed by the fact that all the dunitic bulk rock analyses from the DTZ 448 exhibit strong positive Sr and Pb anomalies, while interstitial igneous clinopyroxenes exhibit 449 the opposite negative Sr and Pb anomalies. These trace element evolutions in regard to

immobile elements have been already described in Oman ophiolite mantle harzburgites
(Gerbert-Gaillard, 2002). The co-variations of Th, U, LREE and HFSE at least may thus be
interpreted in the framework of igneous processes.

453

454 *6.2. Cumulative origin*

455 A cumulative origin for dunites (Bowen, 1915; Bowen and Tuttle, 1949; O'Hara, 1965; 456 Elthon, 1979) implies to preserve a high-Mg parental melt in the crystallization field of the 457 single olivine (+/- spinel) during differentiation. This is quite problematic to account for thick 458 layers (several hundred meters or large bodies of dunites), and the production of such high 459 temperature melts is inconsistent with the inferred thermal conditions in the present day Earth 460 interior (Verhoogen, 1954; McKenzie and Bickle, 1988; Tilley et al., 2007). However, it was 461 proposed that the shallowest part of the Magsad DTZ, where both Fo and NiO in olivine 462 decrease together upsection, was generated by olivine crystallization and accumulation while 463 the main part of the DTZ was a residue of reaction melting (Abily and Ceuleneer, 2013).

464

465 Dunites from the Maqsad DTZ exhibit Ni and Co contents that are much closer to the mantle 466 peridotites compositions (Fig. 3d and 3e) than to the most primitive troctolite cumulates 467 cropping out within the mantle harzburgitic section in the Maqsad area (Ni < 650 μ g/g; Co < 468 40 µg/g; Benoit et al., 1996). This observation, together with the absence of clear correlation between Mg# and Ni in whole rock (Fig. 3d), the negative correlation between Mg# and Co in 469 470 whole rock (Fig. 3e), and with the consistent compositional variations between the different 471 types of dunites, suggests that both pure and impregnated dunites cannot be related by 472 fractional crystallization from a common parent melt. Regarding the pure dunites, a pure 473 cumulative origin does not account for the normalised U-/V-shaped REE patterns (Fig. 5a and 5b). The enrichment in LREE relative to MREE is not predicted by REE partition coefficients 474

475 between olivine and melt (Frey et al., 1978; McKenzie and O'Nions, 1991; Kelemen et al., 476 1993; Hauri and Hart, 1995; Lee et al., 2007; Sun and Liang, 2014) and may better be explain 477 by melt-rock reactions leading to peridotite metasomatism (e.g. Navon and Stolper, 1987; 478 Godard et al., 1995; Vernières et al., 1997). Models presented in Appendix C evidence that 479 the olivine matrix hosting impregnations exhibit similar REE patterns than for whole rock 480 pure dunites, with the same range of variability. In this way, impregnated dunites may be 481 definitely interpreted as hybrid lithologies, made of a mantle-derived olivine matrix 482 associated to other minerals crystallized during melt percolation (e.g. Rabinowicz et al., 1987; 483 Benn et al., 1988; Boudier and Nicolas, 1995; Koga et al., 2001; Abily and Ceuleneer, 2013; 484 Rospabé et al., 2017). The modelling allows us to conclude that the whole rock trace element 485 signatures of impregnated dunites result from a blend of three main variables: the amount of 486 impregnating mineral phases and their chemical signatures as expected, but also the 487 composition of the dunitic matrix itself (Appendix C).

488

489 6.3. Melt-rock reaction product: dunitization and melt transport

490 6.3.1. State of melt-rock reactions leading to incompatible trace element enrichments in
491 mantle peridotites

492 Incompatible trace element enrichments in peridotites, as U-shaped REE patterns, are 493 documented in numerous peridotites from various tectonic settings, from ophiolitic and 494 orogenic massifs (e.g. Prinzhofer and Allègre, 1985; Bodinier et al., 1990; Gruau et al., 1998; 495 Bouilhol et al., 2009) to abyssal peridotites at mid-ocean ridges (Niu, 2004; Godard et al., 496 2008) or in forearc context (Parkinson and Pearce, 1998). It has been proposed that melt-rock 497 reactions may lead to some chromatographic effects such as melt-peridotite re-equilibration or 498 to trapped melt between olivine and pyroxene grains. This can be accounted for by (1) a two-499 stages process involving mantle partial melting later percolated by a LREE-enriched melt (i.e. 500 cryptic metasomatism in the peridotitic matrix without mineralogical reaction) (Navon and 501 Stolper, 1987) or, (2) a single-stage process by combining partial melting (or melt-rock 502 reaction at increasing melt mass; Vernières et al., 1997) and melt transport concomitantly, 503 leading to a MREE depletion relative to HREE whereas LREE remain constant (Godard et al., 504 1995). In the Oman ophiolite, U-shaped REE patterns were described in mantle harzburgites 505 (Godard et al., 2000; Gerbert-Gaillard, 2002; Girardeau et al., 2002; Le Mée et al., 2004; 506 Monnier et al., 2006; Hanghøj et al., 2010), especially in the Magsad diapir area where it is 507 attributed to an association between partial melting, deformation related to mantle flow and 508 MORB percolation (Godard et al., 2000; Gerbert-Gaillard, 2002).

509

510 6.3.2. Extreme trace elements enrichments in DTZ dunites: an early dunitization-related
511 signature

512 The melt-rock reaction leading to dunite formation from a harzburgite through orthopyroxene 513 dissolution is expected to result in a high intergranular permeability of the dunite, that 514 becomes a porous media for efficient melt transfer and extraction (Toramaru and Fujii, 1986; 515 Kelemen et al., 1995b). Dunite channels and horizons, filled or not with troctolitic cumulates, 516 have been mapped within the Magsad mantle section and interpreted in terms of frozen 517 compaction waves (Ceuleneer et al, 1996; Rabinowicz and Ceuleneer, 2005). They are 518 common features in the Magsad diapir but not significant in terms of volume. Hence, it is 519 unlikely that the DTZ formed by the transposition in the solid state of such dunitic bodies 520 generated at greater depth, especially since the DTZ dunites do not show any sign of deformation. They better represent the final *in situ* reacting residue of complete dunitization 521 522 of the uppermost few hundred meters of the mantle induced by melt migration and extraction. 523 Regarding REE, pure dunites from the DTZ display lower HREE, similar and more variable 524 MREE, and higher and more variable LREE contents than harzburgites from the Maqsad

525 mantle section (Fig. 5a) (Godard et al. 2000; Gerbert-Gaillard, 2002). The lower HREE 526 content is in good agreement with a more refractory signature, i.e. depleted, following the 527 complete resorption of orthopyroxene. However, their LREE, Zr, Hf and Th contents are systematically higher than in mantle harzburgites, as Nb, Ta and U in many cases (Fig. 5a and 528 529 5b). Considering a reacting melt with MORB-like affinities (e.g. Ceuleneer et al., 1996; 530 Benoit et al., 1996; Kelemen et al., 1997a; Abily and Ceuleneer, 2013), these signatures are 531 inconsistent with a genesis at the expense of the surrounding harzburgites alone following the 532 single-stage process described above (Godard et al., 1995; Vernières et al., 1997). Moreover, 533 the occurrence of interstitial minerals implies that the percolating melt suffered crystallization 534 after the harzburgite dunitization, associated to the last stage of percolation during the cooling 535 of the system. Then, one can envision the overprinting of two- or several contemporaneous or 536 successive stages processes in order to explain the dunite trace element signatures. It is worth 537 mentioning that the single- and two-stages processes were proposed in order to mimic harzburgite signatures (Navon and Stolper, 1987; Godard et al., 1995; Vernières et al., 1997), 538 539 but it may be different for dunites.

540

541 As the trace element enrichment in pure dunites relative to harzburgites cannot be accounted 542 for by V1-type (e.g. Godard et al., 2003) MORB percolation alone, it therefore calls for the 543 additional involvement of a more enriched melt. In the same spirit Godard et al. (2000) 544 proposed the pervasive infiltration of volatile- and LILE-enriched small melt fractions to 545 account for the enrichment in Th, Nb and Ta relative to LREE in mantle harzburgites. The interaction between an incompatible trace elements-enriched melt or fluid and the 546 547 surrounding dunites does not contradict the percolation of MORB through the DTZ, and the 548 migration of different kinds of melt is further supported by the variable composition of clinopyroxenes located in interstitial position between olivine grains (Fig. 6) (see also Koga et 549

550 al., 2001) and by the presence of amphibole and orthopyroxene among other which are not 551 expected in the fractionation path of primitive MORB (Rospabé et al., 2017). Furthermore, 552 two strong arguments in favour to MORB percolation through the DTZ are the particularly high CaO content in olivine especially observed in pure dunites (Fig. 4b), consistent with 553 554 MORB-dunite equilibration before cooling (Abily and Ceuleneer, 2013), and the high TiO₂ 555 content in disseminated chromites and impregnant pyroxenes (Fig. 4e and f) (see also 556 Rospabé et al., 2017). In a profile view along the base (pure dunites) of the Buri cross section 557 (Fig. 1b), chemical evolutions from about 640 m to 700 m high show that high CaO values in 558 olivine are strongly correlated to other chemical species (Fig. 8), independently from the very 559 small amount of interstitial clinopyroxene (Table 1). This includes high Gd_{CN}/Yb_{CN} ratio in whole rock (Gd increases from 2.5 to 9.5 ng/g respectively) and a higher concavity of the 560 561 REE patterns (i.e. well-defined U-/V-shaped REE patterns), but also higher Co content in 562 whole rock and TiO₂ content both in chromite and scarce minute clinopyroxenes. On the 563 contrary, the 700 m level is characterized by lower Fo in olivine (and Ni content both in 564 olivine and whole rock) and La_{CN}/Sm_{CN} ratio in whole rock (faster increase of Sm (from 4.8 to 7.8 ng/g) relative to La (from 1.1 to 5.0 ng/g)). Assuming that high CaO in olivine recorded 565 566 the last re-equilibration with a percolating MORB before cooling (Abily and Ceuleneer, 567 2013), these co-evolutions highlight that dunites had overprinted a prior enrichment in LREE 568 and a strong concave-upward REE pattern signature before the accumulation and the re-569 equilibration with MORB at 700 m followed by the compaction of the dunitic matrix.

570

571 These observations have several major implications: (1) considering that such enrichments in 572 incompatible elements may be attributable to a low melt-rock ratio (i.e. low porosity) and/or 573 to chromatographic effects induced by the migration of a highly enriched melt (Navon and 574 Stolper, 1987; Godard et al., 1995; Vernières et al., 1997), the low CaO in olivine for the most 575 LREE-enriched samples better supports the re-equilibration with an incompatible elementsrich melt or fluid early in the magmatic history of the DTZ; (2) dunites displaying the highest 576 577 La_{CN}/Sm_{CN} ratios do not represent an end-member that was percolated by a higher amount of melt (Navon and Stolper, 1987) but result from the more or less significant interaction with 578 579 this early enriched melt or fluid; (3) the percolation of MORB-type melts did not occur all 580 over the DTZ and seems to have been focused at some particular horizons at an advanced 581 stage of the DTZ structuration (i.e. here the 700 m high level); (4) the increase of the La, Sm 582 and Gd contents until 700 m implies that the re-equilibration with MORB indeed enhanced 583 the enrichment in the most incompatible LREE, but mostly affected and re-enriched the 584 dunitic matrix in the previously most depleted MREE. Especially, from the base to the 585 elevation of 700 m along the Buri cross section, the faster increase of Gd_{CN}/Dy_{CN} (0.33 to 0.53) relative to Er_{CN}/Yb_{CN} (0.47 to 0.53) follows the smoothing of MREE-HREE segment 586 587 pattern. In this way the linear REE patterns in dunites and harzburgites have been interpreted 588 as resulting from re-equilibration induced by MORB-type melt transport (Godard et al., 589 2000), while concave-upward patterns observed through the DTZ and in mantle harzburgites 590 from the Magsad area should be better related to the dunitization process. Therefore, MORB 591 percolation led to a decrease of the LREE/MREE and an increase and MREE/HREE ratios, 592 and to smoother REE patterns (i.e. exhibiting a lower concavity). Accordingly, dunites from 593 the DTZ appear to be generated after a multi-stages process: (1) a complete harzburgite 594 reactional melting during which enriched melt batches are generated probably in low volume 595 (i.e. melt generated by the incongruent melting of orthopyroxene associated with the 596 contribution of fluids and minor amount of MORB) followed by (2) a chemical re-597 equilibration with larger amount of percolating MORB. In opposition with the model of 598 Navon and Stolper (1987), we propose that the incompatible trace elements-enriched 599 signature is a feature acquired early during the dunitization rather than later by the percolation

600 of enriched melts.

601

602 6.3.3. About the habitus of incompatible trace elements in the dunitic matrix

As previously discussed, the trace element enrichment in dunite, especially in LREE, may 603 604 better reflect magmatic processes rather than secondary alteration events (paragraph 6.1). It 605 supports the view that trace elements are hosted by the olivine itself rather than by interstitial 606 serpentine surrounding preserved grains and is in accordance with recent trace element data in 607 olivine obtained by laser ablation methods (e.g. Drouin et al., 2009; Rampone et al., 2016; 608 Stead et al., 2017). Sub-solidus re-equilibration between olivine and other mineral phases may eventually occur (Sun and Liang, 2014) and lead to trace element enrichments at olivine 609 610 grains boundary (Stead et al., 2017), while other studies propose that these particular 611 geochemical signatures reflect analytical artefacts or the entrapment of small LREE-rich melt 612 inclusions (e.g. Lee et al., 2007). The analytical method we developed (Rospabé et al., 2018) 613 allowed us to establish with confidence that the extensive enrichment in LREE relative to 614 MREE in pure dunites (Fig. 5) is a reliable result. These enrichments thus cannot be 615 satisfactorily explained both by analytical effects or re-equilibration with other minerals. 616 Furthermore, we did not detect melt inclusions through microscopic observations and during 617 the electron microprobe sessions, although the existence of inclusions smaller than about 10 618 um cannot be excluded. Finally, the absence of zoning in olivine grains, at least in terms of 619 major elements, does not militate for a possible concentration of incompatible elements at 620 grain boundaries. All these observations support the hosting of the incompatible trace 621 elements by crystal lattices in olivine consecutively to re-equilibration of the olivine matrix 622 with interstitial melts during their migration through the DTZ. Sub-solidus re-equilibration 623 between different olivine grains may lead to the trace element content smoothing between 624 rims and cores, but stays a hypothetical track at this stage and should be explored in future

625 works.

626

627 6.4. Refertilization induced by melt transport

Interstitial minerals in impregnated dunites may be considered as crystallization products 628 629 from a melt percolating through the residual and porous dunitic media (e.g. Benn et al., 1988; 630 Rabinowicz et al., 1987; Kelemen et al., 1995a; Boudier and Nicolas, 1995; Koga et al., 631 2001), rather than as trapped melts not perfectly extracted (e.g. Faul, 1997; Sundberg et al., 632 2010). This reasonable assumption is supported by the systematic high Mg# in pyroxenes and 633 generally high An in plagioclases, as well as by the low REE contents both in clinopyroxenes 634 and whole rocks, that are all together not expected in the case of MORB-like melt entrapment between olivine grains (see also Koga et al., 2001). Basically, the whole impregnated dunite 635 636 composition is strongly impacted by the modal abundance of these interstitial minerals, 637 especially by the amount of clinopyroxene (Fig. 5 and Appendix C).

638

639 6.4.1. Diversity of the percolating melts within the Maqsad DTZ

640 Calculated melts in equilibrium with clinopyroxenes impregnating the DTZ appear much 641 more variable than the melts that migrated through the mantle section (Benoit et al., 1996; 642 Benoit, 1997) and the V1 lavas constituting the upper crust in the Sumail massif (Godard et 643 al., 2003) (Fig. 9). Three main types of patterns are observed: clinopyroxenes in equilibrium 644 with (1) a N-MORB type melt (1 sample of cpx-bearing dunite) (Fig. 9a), (2) LREE-enriched 645 melts (mainly in (pl/)cpx-bearing dunites) (Fig. 9b), (3) strongly LREE-depleted melts 646 (mainly in (amph/)opx/pl/cpx-bearing dunites) (Fig. 9c). HREE melt concentrations are 647 basically in the variation range of the MORB-type (or V1, Godard et al., 2003) lavas but are 648 much more variable in LREE. The presence of minerals supposed to be in equilibrium with 649 both LREE-enriched and LREE-depleted melts through the DTZ may reflect either the mixing 650 between various melt batches, including a high proportion of deep melts issued from low 651 mantle partial melting degrees, or the reaction at shallow depth between a primitive MORB 652 and residual depleted peridotites (Sobolev, 1996; Kelemen et al., 1997b; Koga et al., 2001). 653 The first hypothesis would seem unlikely here since the absence of such derived enriched-like 654 cumulates in the mantle section. Indeed, MORB-parent melts that crystallized primitive 655 troctolite sills cropping out in the Magsad harzburgites show quite geochemical variations 656 attributable to different degrees of partial melting within the diapir (Benoit, 1997), but that are 657 much more restricted than the melt diversity described here.

658

659 6.4.2. Early generation of enriched melt batches during dunitization

660 The Maqsad area is characterized by the absence of cumulates crystallized from an enriched 661 parent melt within both the mantle section and the DTZ (Benoit et al., 1996; Kelemen et al., 662 1997a; Python and Ceuleneer, 2013). According to this and to the trace element enrichments in pure dunites (Fig. 5), we propose the early production of LREE-enriched melts at Moho 663 664 level (Fig. 9b) further related to melt-rock reactions rather than originated from depth. These 665 melts can be related to the SiO₂-enriched melts produced during incongruent melting of the 666 orthopyroxene and expected following the study of other ophiolitic massifs (Dick, 1977; 667 Quick, 1981; Kelemen, 1990; Kelemen et al., 1992). The early involvement of an enriched 668 melt responsible of the incompatible trace element enrichment in olivine, deduced from the 669 whole pure dunite composition, is in accordance with this interpretation.

670

The genesis of highly enriched melts during dunitization, and the incipient crystallization, may be related to reactive crystallization (Collier and Kelemen, 2010). This process was identified as the main cause of clinopyroxene chemical zoning in olivine-rich troctolites from slow spreading environments: cores were generated by fractional crystallization from a 675 percolating melt while enriched rims result from the fractionation after the contamination of 676 the melt by assimilation of the surrounding peridotite (e.g. Suhr et al., 2008; Drouin et al., 677 2009; Sanfilippo et al., 2013, 2015; Rampone et al., 2016). Especially, the resorption of orthopyroxene seems to be a credible source for this particular signature, allowing to 678 679 important Zr_{CN}/Nd_{CN} fractionation (until 1.5) with the increase of REE contents (Sanfilippo et al., 2015). In this way, the dunitization, when directly followed by interstitial mineral 680 681 fractionation, can be presented as a kind of reactive crystallization process considering the 682 enrichment in silica and incompatible trace elements of the residual melt. However, in this 683 study, clinopyroxene grains analyzed for their trace elements are not zoned and present low Zr_{CN}/Nd_{CN} ratios (0.34-0.74; 0.47 in average) with no observed correlation with Yb, similar to 684 685 clinopyroxene cores in olivine-rich troctolites. This highlights that the chemical evolution of 686 enriched melt batches is probably, in the frame of the Maqsad DTZ, firstly dominated by the 687 residual orthopyroxene assimilation process by MORB (i.e. reactive crystallization in the case 688 of clinopyroxene fractionation at this stage). Following the complete orthopyroxene 689 resorption, fractional crystallization occurs while MORB issued from partial melting within 690 the Maqsad diapir are introduced and transferred through the porous olivine matrix without 691 any further reaction.

692

693 6.4.3. Differentiation and hybridization of the percolating melts within the Maqsad DTZ

Oppositely, further hypotheses may be proposed concerning the LREE-depleted melts (Fig. 9c), which may reflect (1) the assimilation of the surrounding dunite at an advanced refertilization stage (i.e. reactive crystallization products), (2) the later sub-solidus reequilibration between clinopyroxene crystals and neighbouring mineral phases (e.g. Sun and Liang, 2014), (3) the result of hybridization between a latter percolating MORB and hydrous fluids (Rospabé et al. 2017). The two first hypotheses difficultly account for the large 700 chemical variations in clinopyroxene, that is the most common and abundant impregnant 701 phase in DTZ dunites, especially concerning the variable slope of the HREE segment (Figs. 6 702 and 9). However, the third hypothesis is supported by the fact that the LREE-depleted 703 signature is observed mainly in samples containing interstitial orthopyroxene and/or amphibole, minerals previously interpreted as fractionated from a hybrid melt between 704 705 MORB and hydrous fluids (Rospabé et al., 2017). In this scenario this hybrid component may 706 contribute to the exotic melt involved in the formation of disseminated chromites and 707 chromitite ore bodies as suggested by the numerous inclusions of hydrous minerals in 708 chromite (Lorand et Ceuleneer, 1989; Leblanc et Ceuleneer, 1991; Borisova et al., 2012; 709 Rospabé et al., 2017; Johan et al., 2017).

710

711 To conclude, our study reveals that the variability in the composition of the melts responsible 712 of the dunite formation and of their petrological and geochemical evolution is extremely 713 pronounced (see also Sanfilippo et al., 2014, 2017). A working hypothesis is to consider that 714 such wide melt variability is a specificity of the DTZ and reflects local melt/fluid-rock 715 reactions rather than a complex mixing of deep-seated melt batches. This assumption is 716 supported by the fact that some clinopyroxene grains exhibiting different patterns were 717 analysed in samples collected in nearly same locations (Table 1 and Appendix B). According 718 to the observed clinopyroxene chemical heterogeneity, it appears that the DTZ evolved as an 719 open reactive system in which melts fractions issued from strong melt-rock interactions 720 evolved by fractional crystallization and hybridization and were far from being fully 721 homogenised before being supplied to the lower crust.

722

723 6.5. Synthetic model for the formation of the Maqsad DTZ

As pointed out here above, none of the four main processes that are discussed - (1)

serpentinization or secondary weathering, (2) fractionation and accumulation of olivine, (3) harzburgite reactional melting and re-equilibration related to melt percolation into dunite porosity, and (4) refertilization through minerals crystallization from percolating melts interstitially between olivine grains - may account alone for the wide spectrum of geochemical signatures observed in whole rock and mineral compositions of the DTZ, although each can contribute to account for part of this variability, depending on the elements considered. This indicates that the DTZ records an overprinting of several processes.

732

733 In attempt to propose a general scenario describing the development of the Maqsad DTZ, 734 Figure 10 summarizes the different steps deduced from the whole rock and mineral major and 735 trace element signatures. The starting lithology, i.e. the harzburgite from the specific Maqsad 736 area, present a slight but significant LREE enrichment relative to the harzburgites from the 737 main mantle section located far away from the diapir (Fig. 10a) (Godard et al., 2000). Acting 738 as a filter for ascending melts, the top of the diapir is progressively transformed into dunites 739 by interaction with interstitial melts, probably of MORB affinity (Fig. 10b). This process 740 leads to the depletion in HREE in the residual dunite due to the resorption of the last 741 pyroxenes and to the enrichment in LREE and other incompatible trace elements (HFSE, Th, 742 U and possibly some LILE as Rb and Ba). This signature is better explained by the early 743 generation of small batches of enriched melts in response to orthopyroxene incongruent 744 melting (i.e. enriched residual melt). At start, the reaction is driven by MORB/harzburgite 745 interaction but the olivine matrix records the solely incompatible trace elements enrichment, 746 calling for a very low porosity needed to preserve this signature. It has been proposed that 747 high temperature hydrothermal fluids were involved in dunite formation or as early as the 748 mantle dunitization process (Bowen and Tuttle, 1949; Rospabé et al., 2017; Johan et al., 2017); the enriched signature described in this paper may be another evidence for the early 749

750 hydration of the DTZ. Following this, the dunitization will increase the permeability, as 751 MORB-like melts delivered by the mantle continue to invade the matrix. Those melts no 752 longer contribute to mineralogical reaction following the complete orthopyroxene resorption 753 within the host rock, and then may accumulate at Moho level or feed the crust (Fig. 10c and 754 10d). Chemical re-equilibration occurs during MORB percolation, leading to a slight LREE 755 and mainly MREE enrichment in the host dunite, i.e. the re-equilibration process seems to 756 mainly affect the most depleted trace elements in the dunitic matrix independently from the 757 partition coefficient between those elements and olivine. The chemical overprinting of this 758 stage occurs until the compaction of the matrix and the resulting melt extraction. It is worth 759 mentioning that all the intermediate cases may exist between a pure dunite having preserved 760 its prior enriched signature (Fig. 10e) and another one having been fully re-equilibrated with 761 MORB (Fig. 10f). Therefore, pure dunites may be considered as a witness of a magmatic 762 system closed at very high temperature (i.e. above the olivine-plagioclase-clinopyroxene 763 ternary cotectic). Depending on the thermal state of the DTZ, and certainly on the primitive or 764 evolved nature of the percolating melt, crystallization will or not occur interstitially. As 765 supposed following the various REE patterns calculated for the olivine matrix hosting some 766 impregnating clinopyroxene (Appendix C), mineral fractionation occurs concomitantly with 767 the melt percolation and re-equilibration processes (Fig. 10g). Among other, the circulation of 768 hydrous fluids at Moho level during the structuration of the DTZ explains: (1) the higher 769 amount of impregnations approaching the base of the crust witnessing the former thermal 770 gradient across DTZ (Rospabé et al., 2017; see also Dygert et al., 2017), (2) the harzburgite 771 dunitization in the specific context of the DTZ, and not to deeper dunite channels within the 772 mantle section, and (3) the interstitial mineralogy indicative of the percolation of exotic 773 hybrid melts certainly involved in the transport of high amount of Cr - potentially released from orthopyroxene during the dunitization (Johan et al., 2017) - and its concentration in 774

775 disseminated chromites and chromitite pods (Rospabé et al., 2017). In the case of a much 776 colder environment as in slower-spreading environment (Collier and Kelemen, 2010), 777 extreme refertilization potentially occur through a more efficient assimilation of the olivine 778 matrix coupled to fractional crystallization (reactive crystallization), leading to the formation 779 of olivine-rich troctolites as observed in samples collected along present-day mid-ocean 780 ridges or within ophiolites DTZ (e.g. Suhr et al., 2008; Drouin et al., 2009; Dick et al., 2010; 781 Sanfilippo and Tribuzio, 2013b; Sanfilippo et al., 2013, 2014, 2015; Rampone et al., 2016). 782 At this stage, it cannot be excluded that the fractional crystallization dominated the 783 refertilization process within the Maqsad DTZ, relative to reactive crystallization, due to the 784 fast spreading environment in which the Oman ophiolite likely evolved. Another possibility 785 consists in the fast cooling of this oceanic lithosphere fragment, related to the oceanic 786 thrusting that directly followed the lithosphere accretion and that fossilized the system in a 787 high temperature configuration (Boudier et al., 1988). Finally, in regard to the whole dunites 788 and mineral major and trace elements, it appears that both pure and variably impregnated 789 dunites represent end-members that can be generated independently.

790

791 As shown, the re-equilibration with percolating MORB did not occur equally all along the 792 DTZ but was more efficient at some specific levels (Fig. 8) (see also Abily and Ceuleneer, 793 2013), suggesting that the dunitic matrix was either unequally compacted when MORB 794 migrated through it or that the MORB percolation was locally focused. This feature may be 795 partially explained by a model of progressive downward dunitization of the mantle 796 harzburgites at the top of the Maqsad diapir (Abily and Ceuleneer, 2013). During this process, 797 the compaction may progress downward, possibly in response from cooling from above, 798 through successive steps and then may spatially constrain the porous fronts in which MORB 799 may have accumulated. Clearly, some horizons of the DTZ remained isolated early after

dunitization and other horizons were intensely re-equilibrated with MORB more or lesshybridized with hydrous, likely hydrothermal fluids.

802

803 **7. Conclusion**

804 Major and trace elements of 79 pure to highly impregnated dunites from the Oman ophiolite 805 and the composition of the impregnation minerals revealed a large petrological and chemical 806 variability. The dunitic matrix - measured pure dunites and modelled olivine matrix in 807 impregnated dunites - is characterized by a more or less pronounced concave-upward REE 808 pattern and a variable enrichment in other incompatible HFSE, Th, U, Rb and Ba. Their REE 809 and multi-elements patterns are highly scattered implying that no dominant single process 810 controlled the dunite formation. Chemical evolutions suggest that the enriched signature was 811 recorded early during the dunitization by interaction between the surrounding olivine matrix 812 and incompatible trace elements-rich small melt batches at low melt/rock ratios. The latest 813 MORB percolation leads to a cryptic metasomatism expressed as smoothest REE patterns, 814 among others, inducing flattest slopes in HREE and LREE patterns. This scenario contradicts 815 the later percolation of a deep-seated enriched melt and better supports that hybrid enriched 816 melts are generated by melt/rock reactions at an early magmatic stage within the DTZ beneath 817 mid-ocean spreading centers.

818

Our survey shows that these different signatures are variably distributed along the DTZ. This implies that some horizons have been compacted early immediately following the mantle harzburgites dunitization, while others maintained a high enough permeability to drain MORB and make possible re-equilibration. The physical processes that condition the heterogeneous chemical distribution along the DTZ remains quite enigmatic at this stage. Pure and impregnated dunites may be considered as two lithological end-members starting from a common dunitized peridotitic matrix: the first have been compacted at very high temperature after various chemical re-equilibration degrees and the complete extraction of interstitial percolating melts, while the others have been modally refertilized before the extraction of the residual melt. In this frame, the combination of whole dunites and *in situ* minerals major and trace elements compositions reveals undoubtedly the multi-stages processes involved in the DTZ generation.

831

832 Finally, based on the chemical variability within the different minerals observed in dunites 833 from the DTZ, we conclude that if MORB magmas generated by mantle partial melting may have been partly homogenized within the uppermost mantle before being supplied to the 834 835 mantle-crust transition zone, melt-rock reactions and hybridization processes occurring within 836 the DTZ may generate new, exotic and/or hybrid, melt batches that will participate to the 837 oceanic crust genesis and potentially to the MORB heterogeneity. The precise relationship 838 between the evolution of the Maqsad diapir, the N130 paleo-ridge axis, and the vertical 839 petrological and geochemical signatures across the DTZ is a complex three-dimensional 840 problem and could be deciphered by a more comprehensive structural, petrological and 841 geochemical survey of the area.

842

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857

Figure 1. Geological map of the Oman ophiolite showing (1) the spatial distribution of the mantle and crustal sections and (2) the two main magmatic suites displayed along the ophiolite regarding the nature of the dikes cross-cutting the mantle section, characterized as the depleted-andesitic series and the and N-MORB tholeiitic one (a). The Sumail massif exposes a dunitic transition zone of about 300 m thick above the Maqsad mantle diapir (b). Four cross sections were sampled in this area.

864

Figure 2. Photomicrographs illustrating the mineralogical diversity that allowed to define the different lithological facies observed within the Oman ophiolite DTZ, with the olivines constituting the matrix of a pure dunite (a), the comparison between a well developed impregnant clinopyroxene oikocryst in a cpx-bearing dunite (b) with a small interstitial diopside in a pure dunite (c), a plagioclase oïkocryst in a pl/cpx-bearing dunite (d), an orthopyroxene oikocryst in a opx/pl/cpx-bearing dunite (e), and an amphibole in a amphibolebearing dunite.

872

Figure 3. Whole rock major and minor element compositions of the studied dunites in the Maqsad dunites (coloured circles) compared to other DTZ dunites (open circles), mantle

875 harzburgites (open squares) and cpx-harzburgites/lherzolites (open diamonds) from the Oman 876 ophiolite (Godard et al., 2000; Gerbert-Gaillard, 2002; Takazawa et al., 2003; Monnier et al., 877 2006; Hanghøj et al., 2010; Khedr et al., 2014; Nicolle et al., 2016). (a) MgO/SiO₂ versus 878 Al₂O₃/SiO₂, (b) total iron as FeO versus MgO, (c) Al₂O₃ versus CaO, and (d) Ni and (e) Co as 879 a function of the Mg# ratio ($100 \times \text{molar Mg}/(\text{Mg} + \text{Fe}_{\text{total}})$). Compositions are recalculated on 880 a volatile-free basis. Grey bar in panel (a) represents the silicate Earth differentiation trend (or 881 "terrestrial array") (Jagoutz et al., 1979). Dashed black line in panel (b) represents the 882 variation of olivine composition constrained by FeO + MgO = 66.67 mol% following Godard 883 et al. (2000). Dashed grey lines in panels (b) and (c) represent constant Mg# and Al₂O₃/CaO 884 ratios respectively.

885

886 Figure 4. Mineral major element compositions of the studied dunites from the Magsad area illustrated in (a) NiO in olivine, (b) CaO in olivine and (c) Mg# in whole rock as a function of 887 888 Fo in olivine, in (d) $Cr\# (100 \times molar Cr/(Cr + Al))$ in chromite versus Fo in olivine with the 889 olivine-spinel mantle array (OSMA) from (Arai, 1987; Arai, 1994) (grey field), (e) TiO₂ versus Mg# ratio (100 × molar Mg/(Mg + Fe²⁺) in chromite and (f) TiO₂ as a function of Mg# 890 ratio (100 \times molar Mg/(Mg + Fe²⁺) in clinopyroxene. Dark blue dots in clinopyroxene plots 891 892 are the small diopsides observed at olivine grains rims in both pure and impregnated dunites. 893 They are plotted whatever their hosting lithological facies, displaying a composition range 894 spanning between igneous clinopyroxenes and pure hydrothermal diopsides (Rospabé et al., 895 2017).

896

Figure 5. Whole rock chondrite-normalized REE (a, c, e and g) and Primitive Mantlenormalized multi-elements (b, b, f and h) patterns of the studied dunites from the Maqsad dunitic transition zone. Colour labels following the figure 2 for pure dunites (a and b), cpx-

900 bearing dunites (c and d), pl/cpx-bearing dunites (e and f), and opx/pl/cpx-bearing dunites and 901 amphibole-bearing dunites (±opx, pl and cpx) (g and h). In the panels (a) and (b) pure dunites 902 patterns (blue lines) are compared to the harzburgites from the whole Oman ophiolite mantle 903 section (white field framed with a dashed black line) (Gerbert-Gaillard, 2002; Takazawa et 904 al., 2003; Monnier et al., 2006; Hanghøj et al., 2010; Khedr et al., 2014; Nicolle et al., 2016) 905 and to the main mantle harzburgites from the Sumail massif (light grey field), the harzburgites 906 from the Magsad diapir area (medium grey field) and few other dunites from the Magsad 907 DTZ (dark grey field) (Godard et al., 2000; Gerbert-Gaillard, 2002). In other panels each 908 lithological facies is compared to the field of pure dunites from this study (blue fields). 909 Normalizing chondrite and Primitive Mantle values are from Barrat et al. (2012) and Sun and 910 McDonough (1989) respectively.

911

Figure 6. Clinopyroxene chondrite-normalized REE (a) and Primitive Mantle-normalized
multi-elements patterns (b). Normalizing chondrite and Primitive Mantle values are from
Barrat et al. (2012) and Sun and McDonough (1989) respectively.

915

Figure 7. Trace element co-evolutions in pure dunites represented as (a) La, (b) U, (c) Zr, (d)
Ta, (e) Rb and (f) Ba as a function of Th.

918

Figure 8. Chemical evolutions along the base of the Buri cross section, made of pure dunites, illustrated in Fo and NiO in olivine, and whole rock La_{CN}/Sm_{CN} and Gd_{CN}/Yb_{CN} ratios as well as the degree of concavity of their REE patterns, all as a function of the absolute elevation of the sampling site. The concavity of the REE patterns was calculated for each sample as $\sqrt[[]{(Sm_{CN} \times Gd_{CN})}/\sqrt{(Ce_{CN} \times Yb_{CN})}$ -1]. All these co-evolutions are a further strong evidence for the hosting of incompatible trace elements by olivine itself. 925

926 Figure 9. Calculated equilibrium melts with the interstitial clinopyroxenes in impregnated 927 dunites from the Magsad DTZ. Clinopyroxene compositions were averaged for each sample 928 to perform the calculation. Compositions are presented following the shape of the REE 929 patterns with (a) MORB-type melts, (b) LREE-enriched melts and (c) LREE-depleted melts. 930 Calculated melts are compared to the Oman ophiolite V1 (red field) and V2 (green fields) 931 lavas (Godard et al., 2003; Godard et al., 2006) and boninites (dark green field) (Ishikawa et 932 al., 2002; Ishikawa et al., 2005). N-MORB values are from Sun and McDonough (1989). 933 Partitioning coefficients used in the calculation are from Bédard (1994) and normalizing 934 chondrite values are from Barrat et al. (2012). It is worth mentioning that clinopyroxene is the 935 most abundant impregnating phase, and is supposed to display the higher trace element 936 contents in dunites, together with amphibole. In regard of the large HREE content variations 937 in clinopyroxene, it is difficult to account for this variability following sub-solidus re-938 equilibration with other minerals present (e.g. Sun and Liang, 2014), especially in the 939 restricted temperature range expected with the cotectique ol-pl-cpx. Thus it is assumed that re-equilibration between clinopyroxene and neighbouring olivine, orthopyroxene, plagioclase 940 941 and/or amphibole is negligible as far as the REE are concerned (i.e. the assumption of a low 942 REE diffusivity at subsolidus temperature following Van Orman et al., 2001). Then, the 943 calculation of the inferred melt in equilibrium from which the clinopyroxene crystallized 944 allows investigating the nature of the melt that percolated and fractionated interstitial minerals 945 through dunitic rocks.

946

947 Figure 10. Synthetic model for the genesis of the dunites from the Maqsad DTZ,
948 summarizing the melt-rock reactions leading to their high petrological and geochemical
949 variability: (a) incongruent melting of orthopyroxene in mantle harzburgites just below Moho

950 level, (b) neo-formed dunite comprising interstitial high-silica and trace elements-enriched 951 small melt batches, (c) MORB infiltration within the porosity, leading to a hybridization 952 process between the former enriched melt and the incoming MORB, and (d) intense migration 953 of variably evolved MORB batches, leading to variably extensive chemical re-equilibration 954 with the surrounding dunitic matrix. Pure dunites may be produced at different stages over the 955 successive melt-rock reactions, directly following the dunites genesis when they are enriched 956 in the most incompatible elements as LREE (U-shaped REE pattern as the blue line) (e), or 957 after strong re-equilibration with the percolating MORB that allowed to the enrichment in 958 MREE in the olivine matrix (i.e. smoothing of the REE pattern) (f). The chemical signature of 959 pure dunites will first depend on the amount of melt that percolated and re-equilibrated with 960 the host dunites and second on the chemistry of this melt. In some cases, mineral crystallization occurs through the DTZ leading to the formation of impregnated dunites after 961 962 extraction of the residual melt and compaction of the matrix (g). Refertilization may occur at 963 any moment of the evolution of the system, either within a neo-formed dunite or within a 964 fully re-equilibrated one after intense MORB percolation. In that case the whole rock chemical signature will depend on (1) the composition of the olivine matrix, (2) the amount of 965 966 fractionated minerals interstitially and (3) the composition of these minerals.

967

Table 1. Location and modal composition of the samples analysed in this study. The modal
abundance of each mineral was calculated using both whole rock and mineral major elements.
Mode calculations were systematically compared to and are in good agreement with thin
section observations.

972

973 Appendix A. Whole rock major and trace element compositions.

974

975 Appendix B. Mineral major element compositions and clinopyroxene trace element976 compositions.

977

978 Appendix C. Modelling of the interstitial minerals modal abundance influence on the whole979 rock chemical content

980

981 **References**

- Abily B. and Ceuleneer G. (2013) The dunitic mantle-crust transition zone in the Oman
 ophiolite: Residue of melt-rock interaction, cumulates from high-MgO melts, or both? *Geology* 41, 67-70.
- Akizawa N., Ozawa K., Tamura A., Michibayashi K. and Arai S. (2016a) Three-dimensional
 evolution of melting, heat and melt transfer in ascending mantle beneath a fast-spreading
 ridge segment constrained by trace elements in clinopyroxene from concordant dunites
 and host harzburgites of the Oman ophiolite. *J. Petrol.* 57, 777-814.
- 989 Akizawa N., Tamura A., Fukushi K., Yamamoto J., Mizukami T., Python M. and Arai S.
- 990 (2016b) High-temperature hydrothermal activities around suboceanic Moho: An example
 991 from diopsidite and anorthosite in Wadi Fizh, Oman ophiolite. *Lithos* 263, 66-87.
- Arai S. (1987) An estimation of the least depleted spinel peridotite on the basis of olivinespinel mantle array. *Neues Jahrb. für Mineral. Monatshefte* 8, 347-354.
- Arai S. (1994) Characterization of spinel peridotites by olivine-spinel compositional
 relationships: Review and interpretation. *Chem. Geol.* 113, 191-204.
- 996 Arai S. and Matsukage K. (1996) Petrology of gabbro-troctolite-peridotite complex from Hess
- Deep, equatorial Pacific: implications for mantle-melt interaction within the oceanic
 lithosphere. *Proc. Ocean Drill. Program, Sci. Results* 147, 135-155.
- 999 Bach W., Garrido C. J., Paulick H., Harvey J. and Rosner M. (2004) Seawater-peridotite

1000 interactions: First insights from ODP Leg 209, MAR 15°N. Geochemistry, Geophys.
1001 Geosystems 5.

Barrat J.-A., Keller F., Amossé J., Taylor R. n., Nesbitt R. w. and Hirata T. (1996) Determination of Rare Earth Elements in Sixteen Silicate Reference Samples by Icp-Ms After Tm Addition and Ion Exchange Separation. *Geostand. Newsl.* 20, 133-139.

- 1005 Barrat J.-A., Zanda B., Moynier F., Bollinger C., Liorzou C. and Bayon G. (2012)
- 1006 Geochemistry of CI chondrites: Major and trace elements, and Cu and Zn Isotopes.
 1007 *Geochim. Cosmochim. Acta* 83, 79-92.
- Bédard J. H. (1994) A procedure for calculating the equilibrium distribution of trace elements
 among the minerals of cumulate rocks, and the concentration of trace elements in the
 coexisting liquids. *Chem. Geol.* 118, 143-153.
- Bédard J. H. and Hébert R. (1998) Formation of chromitites by assimilation of crustal
 pyroxenites and gabbros into peridotitic intrusions: North Arm Mountain massif, Bay of
- 1013 Islands ophiolite, Newfoundland, Canada. J. Geophys. Res. Earth 103, 5165-5184.
- Benn K., Nicolas A. and Reuber I. (1988) Mantle-crust transition zone and origin of wehrlitic
 magmas: Evidence from the Oman ophiolite. *Tectonophysics* 151, 75-85.
- Benoit M. (1997) Caractérisation géochimique (traces, isotopes) d'un système de drainage
 magmatique fossile dans l'ophiolite d'Oman. *PhD thesis* Université Toulouse III Paul
 Sabatier, 257 pp.
- Benoit M., Ceuleneer G. and Polvé M. (1999) The remelting of hydrothermally altered
 peridotite at mid-ocean ridges by intruding mantle diapirs. *Nature* 402, 514-518.
- Benoit M., Polvé M. and Ceuleneer G. (1996) Trace element and isotopic characterization of
 mafic cumulates in a fossil mantle diapir (Oman ophiolite). *Chem. Geol.* 134, 199-214.
- 1023 Berger E. T. and Vannier M. (1984) Les dunites en enclaves dans les basaltes alcalins des iles
- 1024 oceaniques: approche petrologique. *Bull. Mineral.* **107**, 649-663.

- Bodinier J. L., Vasseur G., Vernières J., Dupuy C. and Fabries J. (1990) Mechanisms of
 Mantle Metasomatism: Geochemical Evidence from the Lherz Orogenic Peridotite. J. *Petrol.* 31, 597-628.
- 1028 Borisova A. Y., Ceuleneer G., Kamenetsky V. S., Arai S., Béjina F., Abily B., Bindeman I.
- 1029 N., Polvé M., De parseval P., Aigouy T. and Pokrovski G. S. (2012) A new view on the
- 1030 petrogenesis of the Oman ophiolite chromitites from microanalyses of chromite-hosted
- 1031 inclusions. J. Petrol. **53**, 2411-2440.
- Boschi C., Dini A., Früh-Green G. L. and Kelley D. S. (2008) Isotopic and element exchange
 during serpentinization and metasomatism at the Atlantis Massif (MAR 30°N): Insights
- 1034 from B and Sr isotope data. *Geochim. Cosmochim. Acta* **72**, 1801-1823.
- 1035 Boudier F., Bouchez J. L., Nicolas A., Cannat M., Ceuleneer G., Misseri M. and Montigny R.
- 1036 (1985) Kinematics of oceanic thrusting in the Oman ophiolite: model of plate 1037 convergence. *Earth Planet. Sci. Lett.* **75**, 215-222.
- Boudier F., Ceuleneer G. and Nicolas A. (1988) Shear zones, thrusts and related magmatism
 in the Oman ophiolite: Initiation of thrusting on an oceanic ridge. *Tectonophysics* 151,
 275-296.
- Boudier F. and Coleman R. G. (1981) Cross section through the peridotite in the Samail
 Ophiolite, southeastern Oman Mountains. *J. Geophys. Res. Solid Earth* 86, 2573-2592.
- Boudier F. and Nicolas A. (1995) Nature of the moho transition zone in the Oman ophiolite. *J. Petrol.* 36, 777-796.
- 1045 Bouilhol P., Burg J. P., Bodinier J. L., Schmidt M. W., Dawood H. and Hussain S. (2009)
- Magma and fluid percolation in arc to forearc mantle: Evidence from Sapat (Kohistan,
 Northern Pakistan). *Lithos* 107, 17-37.
- Bowen N. L. (1915) The later stages of the evolution of the igneous rocks. J. Geol. 23, 1–91.
- Bowen N. L. (1927) The Origin of ultra-basic and related rocks. Am. J. Sci. 14, 89-108.

- Bowen N. L. and Tuttle O. F. (1949) The System MgO-SiO₂-H₂O. *Geol. Soc. Am. Bull.* 60,
 439.
- 1052 Ceuleneer G. (1986) Structure des ophiolites d'Oman : flux mantellaire sous un centre
 1053 d'expansion océanique et charriage à la dorsale. *PhD thesis* Université de Nantes, 349
 1054 pp.
- Ceuleneer G. (1991) Evidence for a Paleo-Spreading Center in the Oman Ophiolite: Mantle
 Structures in the Maqsad Area. In Springer Netherlands. pp. 147-173.
- 1057 Ceuleneer G. and le Sueur E. (2008) The Trinity ophiolite (California): The strange
 1058 association of fertile. *Bull. la Soc. Geol. Fr.* **179**, 503-518.
- 1059 Ceuleneer G., Monnereau M. and Amri I. (1996) Thermal structure of a fossil mantle diapir
 1060 inferred from the distribution of mafic cumulates. *Nature* **379**, 149-153.
- Ceuleneer G., Nicolas A. and Boudier F. (1988) Mantle flow pattern at an oceanic spreading
 centre: the Oman peridotite record. *Tectonophysics* 151, 1-26.
- 1063 Clénet H., Ceuleneer G., Pinet P., Abily B., Daydou Y., Harris E., Amri I. and Dantas C.
- 1064 (2010) Thick sections of layered ultramafic cumulates in the Oman ophiolite revealed by
- an airborne hyperspectral survey: Petrogenesis and relationship to mantle diapirism. *Lithos* 114, 265-281.
- Coleman R. G. and Keith T. E. (1971) A chemical study of serpentinization-burro mountain,
 California. *J. Petrol.* 12, 311-328.
- Collier M. L. and Kelemen P. B. (2010) The case for reactive crystallization at mid-ocean
 ridges. J. Petrol. 51, 1913-1940.
- Daines M. J. and Kohlstedt D. L. (1994) The transition from porous to channelized flow due
 to melt/rock reaction during melt migration. *Geophys. Res. Lett.* 21, 145-148.
- 1073 Deschamps F., Godard M., Guillot S. and Hattori K. (2013) Geochemistry of subduction zone
- serpentinites: A review. *Lithos* **178**, 96-127.

- 1075 Dick H. J. B. (1977) Evidence of partial melting in the Josephine peridotite. *Magma Genes*.
 1076 96, 59-62.
- 1077 Dick H. J. B., Johan Lissenberg C. and Warren J. M. (2010) Mantle melting, melt transport,
- and delivery beneath a slow-spreading ridge: The paleo-MAR from 23°15'N to 23°45'N.
- 1079 *J. Petrol.* **51**, 425-467.
- Dick H. J. B. and Natland J. H. (1996) Late-stage melt evolution and transport in the shallow
 mantle beneath the East Pacific Rise. *Proc. ODP Sci. Res.* 147, 103-134.
- 1082 Drouin M., Godard M., Ildefonse B., Bruguier O. and Garrido C. J. (2009) Geochemical and
- 1083 petrographic evidence for magmatic impregnation in the oceanic lithosphere at Atlantis
- 1084 Massif, Mid-Atlantic Ridge (IODP Hole U1309D, 30°N). *Chem. Geol.* **264**, 71-88.
- Dygert N., Kelemen P. B. and Liang, Y. (2017) Spatial variations in cooling rate in the mantle
 section of the Samail ophiolite in Oman: Implications for formation of lithosphere at
 mid-ocean ridges. *Earth Planet. Sci. Lett.* 465, 134-144.
- 1088 Elthon D. (1979) High magnesia liquids as the parental magma for ocean floor basalts. *Nature*1089 **278**, 514-518.
- Faul U. H. (1997) Permeability of partially molten upper mantle rocks from experiments and
 percolation theory. J. Geophys. Res. 102, 10299.
- Frey F. A., Green D. H. and Roy S. D. (1978) Integrated models of basalt petrogenesis: A
 study of quartz tholeiites to olivine melilitites from South Eastern Australia utilizing
 geochemical and experimental petrological data. *J. Petrol.* 19, 463-513.
- 1095 Gerbert-Gaillard L. (2002) Caractérisation Géochimique des Péridotites de l'ophiolite
 1096 d'Oman : Processus Magmatiques aux Limites Lithosphère/Asthénosphère. *PhD Thesis*1097 Université Montpellier II Sciences et Techniques du Languedoc, 238 pp.
- 1098 Ghosh B., Morishita T., Gupta B. Sen, Tamura A., Arai S. and Bandyopadhyay D. (2014)
- 1099 Moho transition zone in the Cretaceous Andaman ophiolite, India: A passage from the

- 1100 mantle to the crust. *Lithos* **198–199**, 117-128.
- Girardeau J., Monnier C., Le Mée L. and Quatrevaux F. (2002) The Wuqbah peridotite,
 central Oman Ophiolite: Petrological characteristics of the mantle in a fossil overlapping
 ridge setting. *Mar. Geophys. Res.* 23, 43-56.
- 1104 Godard M., Bodinier J. and Vasseur G. (1995) Effects of mineralogical reactions on trace
- element redistributions in mantle rocks during percolation processes : A chromatographic
 approach. *Earth Planet. Sci. Lett.* 133, 449-461.
- Godard M., Bosch D. and Einaudi F. (2006) A MORB source for low-Ti magmatism in theSemail ophiolite. *Chem. Geol.* 234, 58-78.
- Godard M., Dautria J. M. and Perrin M. (2003) Geochemical variability of the Oman
 ophiolite lavas: Relationship with spatial distribution and paleomagnetic directions. *Geochemistry, Geophys. Geosystems* 4.
- Godard M., Jousselin D. and Bodinier J. L. (2000) Relationships between geochemistry and
 structure beneath a palaeo-spreading centre: A study of the mantle section in the Oman
 ophiolite. *Earth Planet. Sci. Lett.* 180, 133-148.
- 1115 Godard M., Lagabrielle Y., Alard O. and Harvey J. (2008) Geochemistry of the highly
- depleted peridotites drilled at ODP Sites 1272 and 1274 (Fifteen-Twenty Fracture Zone,
- 1117 Mid-Atlantic Ridge): Implications for mantle dynamics beneath a slow spreading ridge.
- 1118 *Earth Planet. Sci. Lett.* **267**, 410-425.
- Green D. H. and Ringwood A. E. (1967) The genesis of basaltic magmas. *Contrib. to Mineral. Petrol.* 15, 103-190.
- 1121 Gruau G., Bernard-Griffiths J. and Lécuyer C. (1998) The origin of U-shaped rare earth
- patterns in ophiolite peridotites: assessing the role of secondary alteration and melt/rock
- reaction. *Geochim. Cosmochim. Acta* **62**, 3545-3560.
- 1124 Hanghøj K., Kelemen P. B., Hassler D. and Godard M. (2010) Composition and genesis of

depleted mantle peridotites from the Wadi Tayin massif, Oman ophiolite; Major and
trace element geochemistry, and Os isotope and PGE systematics. *J. Petrol.* 51, 201-227.

1127 Hauri E. H. and Hart S. R. (1995) Correction to "Constraints on melt migration from mantle

- 1128 plumes: A trace element study of peridotite xenoliths from Savai'i, Western Samoa'' by
- 1129 Erik H. Hauri and Stanley R. Hart. J. Geophys. Res. Solid Earth 100, 2003-2003.
- Herzberg C. T., Fyfe W. S. and Carr M. J. (1983) Density constraints on the formation of the
 continental Moho and crust. *Contrib. to Mineral. Petrol.* 84, 1-5.
- Ishikawa T., Fujisawa S., Nagaishi K. and Masuda T. (2005) Trace element characteristics of
 the fluid liberated from amphibolite-facies slab: Inference from the metamorphic sole
 beneath the Oman ophiolite and implication for boninite genesis. *Earth Planet. Sci. Lett.*240, 355-377.
- Ishikawa T., Nagaishi K. and Umino S. (2002) Boninitic volcanism in the Oman ophiolite:
 Implications for thermal condition during transition from spreading ridge to arc. *Geology* **30**, 899-902.
- Jagoutz E., Palme H., Baddenhausen H., Blum K., Cendales M., Dreibus G., Spettel B.,
 Lorenz V. and Wanke H. (1979) The abundances of major, minor and trace elements in
 the earth's mantle as derived from primitive ultramafic nodules. *Proc. Lunar Planet. Sci. Conf. 10th*, 2031-2050.
- Jan M. Q. and Howie R. A. (1981) The Mineralogy and Geochemistry of the Metamorphosed
 Basic and Ultrabasic Rocks of the Jijal Complex, Kohistan, NW Pakistan. *J. Petrol.* 22,
 85-126.
- Jochum K. P., Weis U., Stoll B., Kuzmin D., Yang Q., Raczek I., Jacob D. E., Stracke A.,
 Birbaum K., Frick D. A., Günther D. and Enzweiler J. (2011) Determination of reference
 values for NIST SRM 610-617 glasses following ISO guidelines. *Geostand*. *Geoanalytical Res.* 35, 397-429.

- 1150 Johan Z., Martin R. F. and Ettler V. (2017) Fluids are bound to be involved in the formation 1151 of ophiolitic chromite deposits. Eur. J. Mineral. 29, 543-555.
- 1152 Jousselin D. and Nicolas A. (2000) The Moho transition zone in the Oman ophiolite-relation with wehrlites in the crust and dunites in the mantle. Mar. Geophys. Res. 21, 229-241. 1153
- 1154 Jousselin D., Nicolas A. and Boudier F. (1998) Detailed mapping of a mantle diapir below a
- paleo-spreading center in the Oman ophiolite. J. Geophys. Res. Earth 103, 18153-18170.

Kelemen P. B. (1990) Reaction between ultramafic rock and fractionating basaltic magma I.

1155

1156

- 1157 phase relations, the origin of calc-alkaline magma series, and the formation of discordant 1158 dunite. J. Petrol. 31, 51-98.
- Kelemen P. B., Dick H. J. B. and Quick J. E. (1992) Formation of harzburgite by pervasive 1159 1160 melt/rock reaction in the upper mantle. Nature 358, 635-641.
- Kelemen P. B., Hirth G., Shimizu N., Spiegelman M. and Dick H. J. (1997b) A review of 1161 1162 melt migration processes in the adiabatically upwelling mantle beneath oceanic spreading ridges. Philos. Trans. R. Soc. A Math. Phys. Eng. Sci. 355, 283-318. 1163
- 1164 Kelemen P. B., Koga K. and Shimizu N. (1997a) Geochemistry of gabbro sills in the crust-1165 mantle transition zone of the Oman ophiolite: implications for the origin of the oceanic 1166 lower crust. Earth Planet. Sci. Lett. 146, 475-488.
- 1167 Kelemen P. B., Shimizu N. and Dunn T. (1993) Relative depletion of niobium in some arc 1168 magmas and the continental crust: partitioning of K. Nb. La and Ce during melt/rock 1169 reaction in the upper mantle. Earth Planet. Sci. Lett. 120, 111-134.
- 1170 Kelemen P. B., Shimizu N. and Salters V. J. M. (1995a) Extraction of mid-ocean-ridge basalt
- from the upwelling mantle by focused flow of melt in dunite channels. Nature 375, 747-1171 1172 753.
- 1173 Kelemen P. B., Whitehead J. A., Aharonov E. and Jordahl K. A. (1995b) Experiments on flow
- focusing in soluble porous media, with applications to melt extraction from the mantle. J. 1174

- 1175 *Geophys. Res. Solid Earth* **100**, 475-496.
- Khedr M. Z., Arai S., Python M. and Tamura A. (2014) Chemical variations of abyssal
 peridotites in the central Oman ophiolite: Evidence of oceanic mantle heterogeneity. *Gondwana Res.* 25, 1242-1262.
- 1179 Koga K. T., Kelemen P. B. and Shimizu N. (2001) Petrogenesis of the crust-mantle transition
- zone and the origin of lower crustal wehrlite in the Oman ophiolite. *Geochemistry*, *Geophys. Geosystems* 2.
- Kogiso T., Tatsumi Y. and Nakano, S. (1997) Trace element transport during dehydration
 processes in the subducted oceanic crust: 1. Experiments and implications for the origin
 of ocean island basalts. *Earth Planet. Sci. Lett.* 148, 193-205.
- Lanphere M. A. (1981) K-Ar ages of metamorphic rocks at the base of the Samail Ophiolite,
 Oman. J. Geophys. Res. Solid Earth 86, 2777-2782.
- Leblanc M. and Ceuleneer G. (1991) Chromite crystallization in a multicellular magma flow:
 Evidence from a chromitite dike in the Oman ophiolite. *Lithos* 27, 231-257.
- Lee C. T. A., Harbert A. and Leeman W. P. (2007) Extension of lattice strain theory to mineral/mineral rare-earth element partitioning: An approach for assessing disequilibrium and developing internally consistent partition coefficients between olivine, orthopyroxene, clinopyroxene and basaltic melt. *Geochim. Cosmochim. Acta* **71**, 481-496.
- Lorand J. P. and Ceuleneer G. (1989) Silicate and base-metal sulfide inclusions in chromites
 from the Maqsad area (Oman ophiolite, Gulf of Oman): A model for entrapment. *Lithos*22, 173-190.
- MacLeod C. J., Johan Lissenberg C. and Bibby L. E. (2013) "Moist MORB" axial
 magmatism in the Oman ophiolite: The evidence against a mid-ocean ridge origin. *Geology* 41, 459-462.

- Malvoisin B. (2015) Mass transfer in the oceanic lithosphere: Serpentinization is not
 isochemical. *Earth Planet. Sci. Lett.* 430, 75-85.
- 1202 McKenzie D. and Bickle M. J. (1988) The volume and composition of melt generated by 1203 extension of the lithosphere. *J. Petrol.* **29**, 625-679.
- McKenzie D. and O'Nions R. K. (1991) Partial melt distribution from inversion of rare earth
 element concentrations. *J. Petrol.* 32, 1021-1091.
- Le Mée L., Girardeau J. and Monnier C. (2004) Mantle segmentation along the Oman
 ophiolite fossil mid-ocean ridge. *Nature* 432, 167-172.
- Monnier C., Girardeau J., Le Mée L. and Polvé M. (2006) Along-ridge petrological
 segmentation of the mantle in the Oman ophiolite. *Geochemistry, Geophys. Geosystems*7.
- Montigny R., Le Mer O., Thuizat R. and Whitechurch H. (1988) K-Ar and Ar study of
 metamorphic rocks associated with the Oman ophiolite: Tectonic implications. *Tectonophysics* 151, 345-362.
- Moores E. M. and Vine F. J. (1971) The Troodos Massif, Cyprus and other Ophiolites as
 Oceanic Crust: Evaluation and Implications. *Philos. Trans. R. Soc. A Math. Phys. Eng. Sci.* 268, 443-467.
- 1217 Navon O. and Stolper E. (1987) Geochemical consequences of melt percolation: the upper
 1218 mantle as a chromatographic column. *J. Geol.* 95, 285-307.
- Nicolas A., Bouldier F., Ildefonse B. and Ball E. (2000) Accretion of Oman and United Arab
 Emirates ophiolite Discussion of a new structural map. *Mar. Geophys. Res.* 21, 147179.
- Nicolas A., Ceuleneer G., Boudier F. and Misseri M. (1988) Structural mapping in the Oman
 ophiolites: Mantle diapirism along an oceanic ridge. *Tectonophysics* 151, 27-56.
- 1224 Nicolle M., Jousselin D., Reisberg L., Bosch D. and Stephant A. (2016) Major and trace

- element and Sr and Nd isotopic results from mantle diapirs in the Oman ophiolite:
 Implications for off-axis magmatic processes. *Earth Planet. Sci. Lett.* 437, 138-149.
- Niu Y. (2004) Bulk-rock major and trace element compositions of abyssal peridotites:
 Implications for mantle melting, melt extraction and post-melting processes beneath
 Mid-Ocean ridges. J. Petrol. 45, 2423-2458.
- 1230 O'Hara M. J. (1965) Primary magmas and the origin of basalts. *Scottish J. Geol.* 1, 19–40.
- Ortoleva P., Chadam J., Merino E. and Sen A. (1987) Geochemical self-organization II: the
 reactive-infiltration instability. *Am. J. Sci.* 287, 1008-1040.
- 1233 Palandri J. L. and Reed M. H. (2004) Geochemical models of metasomatism in ultramafic
- 1234 systems: Serpentinization, rodingitization, and sea floor carbonate chimney precipitation.
- 1235 *Geochim. Cosmochim. Acta* **68**, 1115-1133.
- 1236 Parkinson I. and Pearce J. (1998) Peridotites from the Izu-Bonin-Mariana Forearc (ODP Leg
- 1237 125): Evidence for mantle melting and melt-mantle interaction in a supra-subduction
 1238 zone setting. *J. Petrol.* **39**, 1577-1618.
- 1239 Paulick H., Bach W., Godard M., De Hoog J. C. M., Suhr G. and Harvey J. (2006)
- 1240 Geochemistry of abyssal peridotites (Mid-Atlantic Ridge, 15°20'N, ODP Leg 209):
- 1241 Implications for fluid/rock interaction in slow spreading environments. *Chem. Geol.* 234,
 1242 179-210.
- Pearce J. A., Alabaster T., Shelton A. W. and Searle M. P. (1981) The Oman ophiolite as a
 Cretaceous arc-basin complex: evidence and implications. *Philos. Trans. R. Soc. London*300, 299-317.
- Prinzhofer A. and Allègre C. J. (1985) Residual peridotites and the mechanims of partial
 melting. *Earth Planet. Sci. Lett.* 74, 251-265.
- 1248 Prinzhofer A., Nicolas A., Cassard D., Moutte J., Leblanc M., Paris J. P. and Rabinovitch M.
- 1249 (1980) Structures in the new caledonia peridotites-gabbros: Implications for oceanic

- 1250 mantle and crust. *Tectonophysics* **69**, 85-112.
- Python M. and Ceuleneer G. (2003) Nature and distribution of dykes and related melt
 migration structures in the mantle section of the Oman ophiolite. *Geochemistry*, *Geophys. Geosystems* 4.
- Python M., Ceuleneer G. and Arai S. (2008) Chromian spinels in mafic-ultramafic mantle
 dykes: Evidence for a two-stage melt production during the evolution of the Oman
 ophiolite. *Lithos* 106, 137-154.
- Python M., Ceuleneer G., Ishida Y., Barrat J.-A. and Arai S. (2007) Oman diopsidites: a new
 lithology diagnostic of very high temperature hydrothermal circulation in mantle
 peridotite below oceanic spreading centres. *Earth Planet. Sci. Lett.* 255, 289-305.
- Python M., Yoshikawa M., Shibata T. and Arai S. (2011) Diopsidites and Rodingites:
 Serpentinisation and Ca-Metasomatism in the Oman Ophiolite Mantle. In *Dyke Swarms: Keys for Geodynamic Interpretations* Springer Berlin Heidelberg, Berlin, Heidelberg. pp.
 401-435.
- Quick J. E. (1981) Petrology and petrogenesis of the Trinity Peridotite, an upper mantle diapir
 in the Eastern Klamath Mountains, Northern California. J. Geophys. Res. 86, 1183711863.
- Rabinowicz M. and Ceuleneer G. (2005) The effect of sloped isotherms on melt migration in
 the shallow mantle: a physical and numerical model based on observations in the Oman
 ophiolite. *Earth Planet. Sci. Lett.* 229, 231-246.
- Rabinowicz M., Ceuleneer G. and Nicolas A. (1987) Melt segregation and flow in mantle
 diapirs below spreading centres: evidence from the Oman ophiolite. *J. Geophys. Res.* 92,
 3475-3486.
- 1273 Rampone E., Borghini G., Godard M., Ildefonse B., Crispini L. and Fumagalli P. (2016)
 1274 Melt/rock reaction at oceanic peridotite/gabbro transition as revealed by trace element

1275 chemistry of olivine. *Geochim. Cosmochim. Acta* **190**, 309-331.

- Rioux M., Bowring S., Kelemen P., Gordon S., Dudás F. and Miller R. (2012) Rapid crustal
 accretion and magma assimilation in the Oman-U.A.E. ophiolite: High precision U-Pb
 zircon geochronology of the gabbroic crust. *J. Geophys. Res. Solid Earth* 117.
- 1279 Rospabé M. (2018) Etude pétrologique, géochimique et structurale de la zone de transition
 1280 dunitique dans l'ophiolite d'Oman: Identification des processus pétrogénétiques à
 1281 l'interface manteau/croûte. *PhD thesis* Université Toulouse III Paul Sabatier.
- Rospabé M., Benoit M. and Candaudap F. (2018) Determination of Trace Element Mass
 Fractions in Ultramafic Rocks by HR-ICP-MS: A Combined Approach Using a Direct
 Digestion/Dilution Method and Pre-concentration by Co-precipitation. *Geostand*. *Geoanalytical Res.* 42, 115-129.
- Rospabé M., Ceuleneer G., Benoit M., Abily B. and Pinet P. (2017) Origin of the dunitic
 mantle-crust transition zone in the Oman ophiolite: The interplay between percolating
 magmas and high-temperature hydrous fluids. *Geology* 45, 471-474.
- Sanfilippo A., Dick H. J. B. and Ohara Y. (2013) Melt-rock reaction in the mantle: Mantle
 troctolites from the parecevela ancient back-arc spreading center. *J. Petrol.* 54, 861-885.
- 1291 Sanfilippo A., Morishita T., Kumagai H., Nakamura K., Okino K., Hara K., Tamura A. and
- Arai S. (2015) Hybrid troctolites from mid-ocean ridges: Inherited mantle in the lower
 crust. *Lithos* 232, 124-130.
- Sanfilippo A. and Tribuzio R. (2013a) Building of the deepest crust at a fossil slow-spreading
 centre (Pineto gabbroic sequence, Alpine Jurassic ophiolites). *Contrib. to Mineral. Petrol.* 165, 705-721.
- Sanfilippo A. and Tribuzio R. (2013b) Origin of olivine-rich troctolites from the oceanic
 lithosphere: A comparison between the Alpine Jurassic ophiolites and modern slow
 spreading ridges. *Ofioliti* 38, 89-99.

1300	Sanfilippo A., Tribuzio R., Ottolini L. and Hamada M. (2017) Water, lithium and trace
1301	element compositions of olivine from Lanzo South replacive mantle dunites (Western
1302	Alps): New constraints into melt migration processes at cold thermal regimes. Geochim.
1303	<i>Cosmochim. Acta</i> 214 , 51-72.
1304	Sanfilippo A., Tribuzio R. and Tiepolo M. (2014) Mantle-crust interactions in the oceanic

- lithosphere: Constraints from minor and trace elements in olivine. *Geochim. Cosmochim. Acta* 141, 423-439.
- Savov I. P., Ryan J. G., D'Antonio M., Kelley K. and Mattie P. (2005) Geochemistry of
 serpentinized peridotites from the Mariana Forearc Conical Seamount, ODP Leg 125:
 Implications for the elemental recycling at subduction zones. *Geochemistry, Geophys. Geosystems* 6.
- Schiano P., Clocchiatti R., Lorand J.-P., Massare D., Deloule E. and Chaussidon M. (1997)
 Primitive basaltic melts included in podiform chromites from the Oman Ophiolite. *Earth Planet. Sci. Lett.* 146, 489-497.
- Snow J. E. and Dick H. J. B. (1995) Pervasive magnesium loss by marine weathering of
 peridotite. *Geochim. Cosmochim. Acta* 59, 4219-4235.
- 1316 Sobolev A. (1996) Melt Incusions in Minerals as a source of princple petrological
 1317 information. *Petrology* 4, 209-220.
- Stead C. V., Tomlinson E. L., McKenna C. A. and Kamber B. S. (2017) Rare earth element
 partitioning and subsolidus exchange behaviour in olivine. *Chem. Geol.* 475, 1-13.
- 1320 Suhr G., Kelemen P. and Paulick H. (2008) Microstructures in Hole 1274A peridotites, ODP
- Leg 209, Mid-Atlantic Ridge: Tracking the fate of melts percolating in peridotite as the
 lithosphere is intercepted. *Geochemistry*, *Geophys. Geosystems* 9.
- 1323 Sun C. and Liang Y. (2014) An assessment of subsolidus re-equilibration on REE distribution
- among mantle minerals olivine, orthopyroxene, clinopyroxene, and garnet in peridotites.

- 1325 *Chem. Geol.* **372**, 80-91.
- Sun S.-S. and McDonough W. F. (1989) Chemical and isotopic systematics of oceanic
 basalts: implications for mantle composition and processes. *Geol. Soc. London, Spec. Publ.* 42, 313-345.
- 1329 Sundberg M., Hirth G. and Kelemen P. B. (2010) Trapped melt in the Josephine peridotite:
- 1330 Implications for permeability and melt extraction in the upper mantle. *J. Petrol.* 51, 185-1331 200.
- Takazawa E., Okayasu T. and Satoh K. (2003) Geochemistry and origin of the basal
 lherzolites from the northern Oman ophiolite (northern Fizh block). *Geochemistry*, *Geophys. Geosystems* 4.
- Tamura A., Morishita T., Ishimaru S. and Arai S. (2014) Geochemistry of spinel-hosted
 amphibole inclusions in abyssal peridotite: Insight into secondary melt formation in
 melt-peridotite reaction. *Contrib. to Mineral. Petrol.* 167, 1-16.
- Tilley C. E., Thompson R. N. and Lovenbury P. A. (2007) Melting relations of some oceanic
 basalts. *Geol. J.* 8, 59-64.
- 1340 Tilton G. R., Hopson C. A. and Wright J. E. (1981) Uranium-lead isotopic ages of the Samail
- Ophiolite, Oman, with applications to Tethyan ocean ridge tectonics. *J. Geophys. Res. Solid Earth* 86, 2763-2775.
- Tippit P. R., Pessagno J. E. A. and Smewing J. D. (1981) The biostratigraphy of sediments in
 the volcanic unit of the Samail ophiolite. *J. Geophys. Researc* 86, 2756-2762.
- 1345 Toramaru A. and Fujii N. (1986) Connectivity of melt phase in a partially molten peridotite.
 1346 *J. Geophys. Res. Solid Earth* 91, 9239-9252.
- Van Orman J. A., Grove T. L. and Shimizu N. (2001) Rare earth element diffusion in
 diopside: influence of temperature, pressure, and ionic radius, and an elastic model for
 diffusion in silicates. *Contrib. to Mineral. Petrol.* 141, 687-703.

- 1350 Verhoogen J. (1954) Petrological evidence on temperature distribution in the mantle of the
 1351 earth. *Eos, Trans. Am. Geophys. Union.*
- 1352 Vernières J., Godard M. and Bodinier J.-L. (1997) A plate model for the simulation of trace
- element fractionation during partial melting and magma transport in the Earth's upper
- 1354 mantle. J. Geophys. Res. 102, 24771.
- Yamasaki T., Maeda J. and Mizuta T. (2006) Geochemical evidence in clinopyroxenes from
 gabbroic sequence for two distinct magmatisms in the Oman ophiolite. *Earth Planet. Sci. Lett.* 251, 52-65.
- Yokoyama T., Makishima A. and Nakamura E. (1999) Evaluation of the coprecipitation of
 incompatible trace elements with fluoride during silicate rock dissolution by acid
 digestion. *Chem. Geol.* **157**, 175-187.
- You C. F., Castillo P. R., Gieskes J. M., Chan L. H. and Spivack, A. J. (1996) Trace element
 behavior in hydrothermal experiments: Implications for fluid processes at shallow depths
 in subduction zones. *Earth Planet. Sci. Lett.* 140, 41-52.





















Figure 10

