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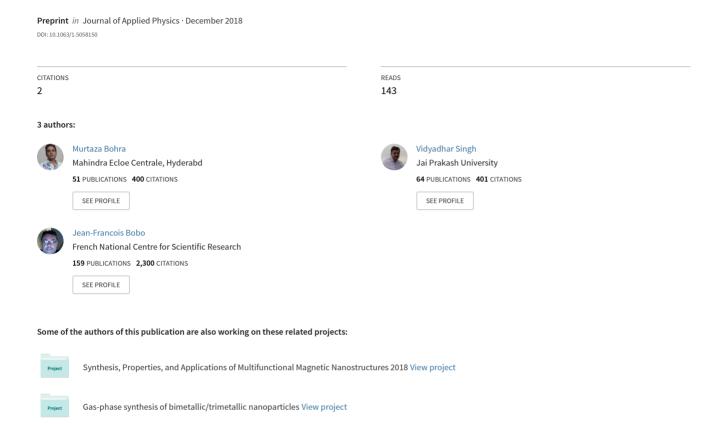
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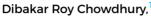
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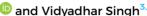




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ABSTRACT

Charge ordering (Fe⁺³/Fe⁺²) is a key concept in the Verwey transition of Fe₃O₄ because it frequently competes with functional properties (half-metallicity/ferromagnetism and structural transformation) and quantum confinement effect, especially at nanoscale dimensions. In this paper, we report the fabrication of nanocrystalline Fe₃O₄ thin films via two different reduction routes, namely, vacuum annealing and wet H2 annealing. While vacuum annealed films exhibit Verwey transition and resistivity values comparable to bulk Fe₃O₄, the same is not observed in electric transport properties of wet H₂ annealed films. However, this transition was visible in the magnetic characteristics exhibited by both the films though realized via different routes. This observation indicates the possibility of charge and spin ordering as two independent phenomena, and it is a coincidence that happens at the same Verwey transition region. It is seen that a crossover from thermally activated hopping (300-120 K) to Mott variable range hopping (VRH) (across Verwey transition) and then to Shklovskii-Efros VRH hopping (70-30 K) via the conduction mechanism takes place in vacuum annealed films in contrast to the typical semiconducting behavior (300-50 K) expected of wet H₂ annealed films. Different electric transport properties in both varieties of Fe_3O_4 films could be ascribed to the electronic disorder/defects affecting charge ordering Fe^{+3}/Fe^{+2} and trimerons $(Fe^{+3}-Fe^{+2}-Fe^{+3})$.

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INTRODUCTION

Verwey transition, particularly, in nano-sized strongly correlated magnetite [Fe₃O₄: (Fe⁺³)_A(Fe⁺², Fe⁺³)_BO₄] materials, continues to draw the attention of researchers involved with spintronics and the physics associated with it.1-5 Electron hopping between Fe⁺² and Fe⁺³ ions at octahedral B-sites facilitates room temperature electric conductivity σ = 200 $(\Omega\,cm)^{-1}$ in Fe₃O₄. Otherwise, most of the ferrites derived from it are insulators. Besides, Fe₃O₄ owns only minority t_{2g} spin band at the Fermi level, thereby demonstrating half-metallic character, which has triggered many applications.^{4,5} The transformation of this curious metallic state of Fe₃O₄ to an insulating state below 120 K (T_V) is essentially known as the Verwey transition.⁶ Consequently, the crystal structure of Fe₃O₄ also transforms from a cubic to a monoclinic structure due to the Jahn-Teller distortion of Fe⁺² ions. The monoclinic structure is connected to the complex electronic order of $\mathrm{Fe^{+2}/Fe^{+3}}$ charges, $\mathrm{Fe^{+2}}$ orbital states, and metal-metal weakly bonded Fe3 units known as trimerons, 4,7-9 which means that electrons are not fully localized as Fe⁺² states, but, instead, are spread over the three sites Fe⁺³-Fe⁺²-Fe⁺³. However, there still exist uncertainties as to the precise mechanism responsible for this electrical and structural transition. Further, the consequential influence of magnetic properties on the transition and its electronic characteristic is still unclear, 7-10 particularly for nanocrystalline Fe₃O₄ films.

The Verwey transition has found a lot of advanced technological applications, for example, in an enhancement of resistive switching, spin accumulation, spin Seebeck and

magneto-caloric effects, switchover from giant magnetoresistance (MR) to tunnel magnetoresistance in a single spin valve, 11-16 etc. It is, therefore, important to investigate the Verwey transition in nanocrystalline Fe₃O₄ films to get a better insight into the mechanism of spin-lattice-charge coupling. Most of these Fe₃O₄ films are post-annealed for the improvement of the electrical performance and stability of the spintronic devices. Typically in pulsed laser deposition and sputtering processes, ^{17,18} it is observed that Fe₃O₄ films grown at low temperatures are formed out of amorphous and mixed oxide phases and antiphase boundaries. Therefore, it is very important to optimize the different reduction processes to get bulk stoichiometric Fe₃O₄ properties.

In this study, we investigate the electric transport properties across the Verwey transition of nanocrystalline Fe₃O₄ films produced by two different reduction methods. Our findings suggest that electric transport is an independent process and it hardly has any correlation with the magnetization process and it depends on external and internal factors that arise from the involved complexities in the reduction methods.

EXPERIMENTAL

Thin films were first laser-ablated from an α-Fe₂O₃ ceramic target on to fused quartz substrates at three different growth temperatures (Tg): RT, 350, and 500 °C in a vacuum of 1×10^{-5} mbar. ^{17,23} These films undergo two different reduction processes, namely, vacuum annealing and wet H₂ annealing.¹ For vacuum annealing, after the deposition, the samples were in situ annealed at 500 °C for 1h in vacuum $\sim 4 \times 10^{-6}$ mbar. For wet H₂ annealing, as-grown films were ex situ annealed in a tubular furnace at 450 °C in wet H2 gas atmosphere for 15 min. X-ray diffraction (XRD) of the films was taken by a PANalytical X'Pert PRO X-ray diffractometer using CuK_α radiation. X-ray photoelectron spectra (XPS) were collected on a Microtech Unit ESCA 3000 Spectrometer with source MgKa (1256.6 eV). The transmission electron microscopy (TEM) micrographs were collected from a Philips CM 200 transmission electron microscope. Electrical transport measurements were performed using a four-probe method in between 30 and 300 K. The micro-Raman spectra were recorded in the range of 100-1000 cm⁻¹ with a 6 mW, 514.5 nm Ar⁺ laser source. Magnetization (M) vs temperature (T) curves were measured at various magnetic fields using a Quantum design PPMS.

RESULTS

Figure 1(a) represents the XRD patterns of vacuum and wet H₂ annealed films of different T_g values. Vacuum annealed films show sharp peaks corresponding to Fe₃O₄ phase at all Tg values. Wet H2 annealed films show comparatively broad Fe₃O₄ peaks along with the amorphous background of quartz substrates. This indicates a relatively better crystallinity in the vacuum deposited films. The average grain sizes estimated from the Scherrer formula using high-intensity Fe₃O₄ peak width were found to be in the range 50-83 nm (vacuum annealed) and 22-60 nm (wet H₂ annealed), respectively. Larger grain sizes in the former sets of films are further

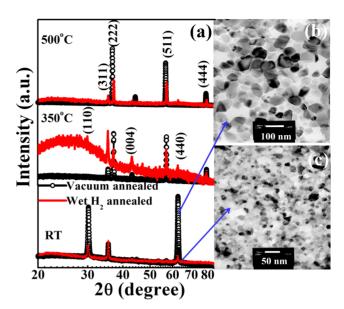


FIG. 1. (a) XRD patterns of vacuum and wet H_2 annealed films of different T_{q} values; 16 [(b) and (c)] TEM micrographs for both varieties of films for one of the representatives $T_{cr} = RT$.

confirmed from TEM images for one of the representative samples with $T_g = RT$ [Figs. 1(b) and 1(c)]. No other oxides $(\alpha - Fe_2O_3, FeO, and \gamma - Fe_2O_3)$ and elemental Fe impurity phases could be seen in either case within the accuracy limit set by the XRD and the XPS equipment (shown in Fig. S1 in the supplementary material), which is also supported by Raman studies (shown later). Irrespective of reduction methods, all films exhibit lattice constants (a = 8.37 - 8.39 Å) close to the known value for cubic bulk Fe₃O₄ (8.39 Å) structure.

To understand the effect of microstructural difference on electric transport behavior, the resistivity (ρ) vs temperature (T) curves for wet H₂ and vacuum annealed films grown at different Tg values are plotted in Figs. 2(a) and 2(b), respectively. The room temperature ρ values (as shown in Table I) of $4.4-40 \,\mathrm{m}\Omega\,\mathrm{cm}$ for vacuum annealed films are close to the known single crystal Fe₃O₄ value of $4 \text{ m}\Omega \text{ cm}^{24}$ but smaller than wet H_2 annealed films (90-250 m Ω cm). The ρ -T curves are also quite different in both sets of films. The resistivity of wet H₂ annealed films gradually increases upon lowering the temperatures (300-50 K), but in the case of vacuum annealing, the ρ -T curve initially varies very slowly, and after passing through a kink, it starts increasing rapidly. Such a kink (115–125 K) in the ρ –T curve is a clear signature of the Verwey transition. However, this transition is not as sharp as was reported in the case of single crystal Fe₃O₄ at 120 K.²

Interestingly, M-T curves [insets of Figs. 1(a) and 1(b)] show clear Verwey transition around 120 K irrespective of the methods used for reduction. Except for the Verwey transition,

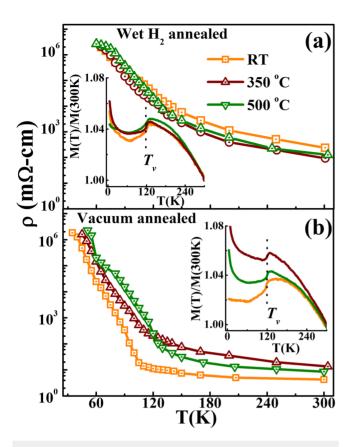


FIG. 2. (a) ρ -T curves of wet H₂ and vacuum annealed (b) films at different T_0 values. Corresponding M-T curves (measured at 2 T) are also given in their respective insets.

M-T curves do not show any other peaks, which exclude the possibility of spin blocking process because our grain sizes exceed the superparamagnetic size limit of Fe₃O₄ and show typical ferrimagnetic behavior. 13,17 Another apparent difference is that the ρ -T and M-T curves of vacuum annealed films are significantly influenced by in situ growth temperature T_a .

TABLE I. Room temperature resistivity (ρ) and magnetoresistance (MR) values, activation energy (E_g), T_M , and T_{SE} for vacuum and wet H_2 annealed Fe_3O_4 films at different T_{α} values.

| | Vacuum annealed films | | | | | Wet H ₂ annealed films | | |
|----------------------------|-----------------------|-------------------------|---------------------------------------|----------------------------|----------------------|-----------------------------------|-------------------------|-----------------------------|
| <i>T</i> _g (°C) | ρ (300 K) (mΩ cm) | E _g (meV) | Т _М (10 ⁸ K) | T _{SE} (K) | MR (300 K) (%) | ρ (300 K) (mΩ cm) | E _g (meV) | <i>MR</i> (300 K) (%) |
| RT 350 500 | 4.4 14 8.4 | 20 78 44 | 9.8 2.7 4.7 | 23 104 21 904 21 025 | -0.3 -0.5 -0.9 | 240 93 126 | 81 86 84 | -1.81 -2.05 -2.25 |

DISCUSSIONS

It is clear from the above results that the in situ growth temperature influences physical properties of vacuum annealed films to some extent, but the same is not evident in the cases of wet H₂ annealed films. This can be understood from the fact that vacuum annealing was carried out just after deposition while keeping the vacuum level almost intact and raising annealing temperature. This results in robust crystalline growth and reduction of any residual Fe₂O₃ phase into Fe₃O₄. We feel that during initial deposition, crystallites not detectable by XRD were formed, which act as nucleation centers for further crystallization during post-deposition annealing.²⁶ Thus, the initial deposition conditions leave their signature on the as-deposited films, which determine the texture and physical properties of the films when annealed. However, the wet H2 annealing process¹⁷ involves two steps: all the samples irrespective of their in situ growth temperatures were first ex situ annealed in air to convert them into complete Fe₂O₃ phase. In the second step, wet H₂ gas was passed over the samples to obtain the end product Fe₃O₄. In the wet H₂ reduction process, complete grain refinement takes place, which results in smaller grain sizes of Fe₃O₄ with presumably distorted surfaces compared to vacuum annealed films. Thereby, all wet H2 annealed films show almost identical behavior and the initial in situ growth temperature has hardly any effect.

A notable feature observed in the present study is that the vacuum annealed films show the Verwey transition in ρ -T curves, while the same is not seen in wet H2 annealed films. It is quite surprising that magnetization shows the Verwey transition in both sets of films, but their electronic transport properties behave differently. Systematic studies^{2,3} on stoichiometric Fe₃O₄ nanocrystals (5-100 nm), however, show the occurrence of concurrent Verwey transition in conductance, magnetization, nuclear magnetic resonance, cryo-XRD, and heat capacity measurements. The Verwey transition was weakly size-dependent below 40 nm and became suppressed in nanocrystals smaller than 20 nm before disappearing completely for less than 6 nm; even in that sense also, our samples overcome the grain size threshold. Kim et al.,²⁷ in their recent work, reported that the Verwey transition exhibits large size-dependent thermal hysteresis in the magnetization data and that the hysteresis width is correlated with the critical size of the single magnetic domain. They also anticipated stronger coupling between charge and spin degrees of freedom in Fe₃O₄. To understand spin-charge-lattice coupling in our nanocrystalline Fe₃O₄ films, we plotted normalized M (T)/M (300 K) vs T curves measured at different magnetic fields in Fig. 3 for one of the representative samples. We can see that the magnitude of the drop in magnetization below 120 K is strongly dependent on the applied magnetic field. This feature is an artifact arising out of insufficient magnetic field resulting in non-saturation [shown in Fig. S2(a) in the supplementary material]. This increased non-saturation effect is caused by increased anisotropy in the monoclinic phase of Fe₃O₄ below 120 K.²³ However, no such drop in magnetization is observed in M(T)/M(300 K) vs T curve when measured at a

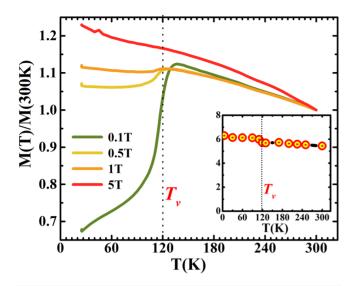


FIG. 3. Normalized M-T curves at various measuring fields for one of the representative samples (wet H₂ annealed films with T_g = RT). The inset shows magnetization at infinite field $4\pi M$ (∞) vs T curve.

fixed magnetic field of 5 T. This raises a question whether this drop in magnetization is due to a change of the magnetic moment of Fe₃O₄ or it is just a manifestation of the increased anisotropy. This controversy can be resolved by plotting saturation magnetization (i.e., magnetization at infinite field) $4\pi M$ (∞) vs T curve (inset of Fig. 3), estimated from the Chikazumi non-saturation formula, $4\pi M$ (H) = $4\pi M$ (∞)(1 – H/H*)^{1/2}.²³ This suggests that $4\pi M$ (∞) increases only below 120 K similar to M (T)/M(300 K) vs T curve measured at a sufficiently high field of 5 T (Fig. 3). Thus, this drop in magnetization occurs only because of the insufficient magnetic field applied which cannot saturate the monoclinic phase of Fe₃O₄; otherwise at high fields, sample results typically obey Bloch-T^{3/2} law of ferrites, ¹⁷ without inducing any change at Verwey transition. This also implies that irrespective of the crystal structure transition, this phenomenon can happen even if there is a change in anisotropy.²⁸ In the case of Fe₃O₄, it is a coincidence that such changes in anisotropy also occur around 120 K. Based on the above discussion, it can be argued that the Verwey transition may not be driven primarily by magnetic interactions and hence the metal-insulator transition has hardly any effect on magnetic properties. This argument is further supported if we consider the Verwey transition as a Mott-type metal-insulator transition;²⁹ then, in principle, it does not require a change in either crystal structure or magnetic symmetry, although in practice the coupling between charge, spin, and lattice means that other transitions tend to occur simultaneously.³⁰ Moreover, Wang et al.³¹ also support our decoupling argument by not showing the Verwey transition in the electric transport properties and temperature dependent structural evolution even in polycrystalline bulk

 $\rm Fe_3O_4$ samples. While their magnetization data still manifested a discontinuous decrease at the $\rm T_V$, they attributed these unusual results to the quenching related stresses and defects in the material.

After decoupling of magnetic phenomena, the controversy, often, is to find which comes first: does the structural transition drive the electronic one or is it the other way round. As the grain boundary volume is not only a factor in electric transport of nano-crystalline Fe₃O₄ thin films, the different post-annealing treatments are involved that could also affect the Fe⁺³/Fe⁺² stoichiometry, residual thermal stresses, and B-site cation and oxygen vacancies differently. 1-3,32 To elucidate this aspect, we have taken room temperature Raman spectra (within the spectral range 150-800 cm⁻¹) of two representative samples (Tg=RT) for each reduction process, as shown in Fig. 4. These spectra were fitted to a sum of Lorentzians which yields line shape parameters: peak position (ω) and full width at half maximum (γ). We observed three strongest modes (A_{1g} , T^3_{2g} , and T^2_{2g}) and one faint T^1_{2g} mode of Fe_3O_4 phase. The A_{1g} mode is responsible for the structural property (a symmetric stretch of oxygen atoms along Fe-O bonds of tetrahedral A site) and the T³_{2g} mode

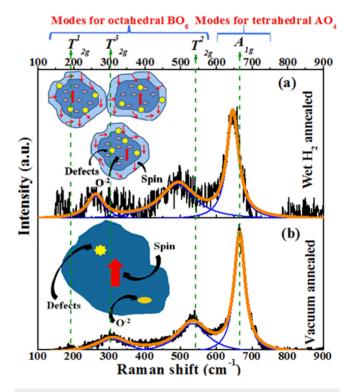


FIG. 4. Raman spectra recorded in the range $100-800 \, \mathrm{cm}^{-1}$ at room temperature for (a) wet H_2 and (b) vacuum annealed films of T_g = RT. Red arrow shows Raman mode position in bulk cubic $\mathrm{Fe_3O_4}$. The solid orange line shows the Lorentzian fitted curves of the peaks due to A_{1g} and T_{2g} modes. Insets show the schematic of the electronic and magnetic disordered grains in both sets of films.

more for the electronic property (symmetric and asymmetric bending of Fe-O bonds of the octahedral B site). In wet H₂ annealed films, the intensity of the modes is weaker in comparison to vacuum annealed films; their mode position is also significantly shifted toward lower wave numbers, A_{1g}: 646 cm⁻¹ (665 cm⁻¹) and T³_{2g}: 262 cm⁻¹ (308 cm⁻¹). It appears that most of these modes are closer to those of nonstoichiometric $Fe_{3(1-\delta)}O_4$.³³ When any structural defect/vacancies are created in the Fe₃O₄ thin films, the changes in the local electronic network signify a measure of the electron-phonon interaction, which would influence the hopping parameters. In metallic systems, a phonon can decay in an electron-hole pair excitation and extra broadening due to such a decay, as suggested by Allen [Eq. (1)],33 can be used to estimate electron-phonon coupling parameter (1) associated with a particular mode i as follows:

$$\lambda_{\rm i} = \frac{\gamma_{\rm i}}{\omega_{\rm i}^2} \left[\frac{g_{\rm i}}{2\pi \ {
m N(E_F)}} \right],$$
 (1)

where q_i is the mode degeneracy and $N(E_F)$ is the electronic density of states at the Fermi level. Though precise λ cannot be estimated, we attempted to estimate the relative variation in λ values. Using theoretically 18 calculated N(E_F) (3 states eV $^{-1} f.u.^{-1}$) value for RT, one estimates that λ for the T $_{2g}^3$ mode in wet H₂ annealed films (=2.01) is quite large in comparison to vacuum annealed films (=1.55) and bulk Fe_3O_4 (=0.51) values. For the A_{1g} mode, λ is fairly constant with 0.13 (0.11) in both the samples but higher than that of the bulk Fe₃O₄ value (0.045), indicating some sort of structural similarity. However, both the samples are different from the electronic structure point of view because in Fe₃O₄, near the Fermi level, the electronic states demonstrate T_{2g}^3 symmetry and hence couple strongly to T_{2g}^3 phonon as compared to $A_{\rm Ig}$ phonons. The enhanced value of λ for the T32g mode in wet H2 annealed films suggests more localization of electrons at the Fe sites, which could be due to defect/vacancy that affects hopping between Fe2+ and Fe3+ ions at octahedral sites. However, it has little impact on the magnetic interactions among the Fe ions at the tetrahedral A and octahedral B sites. 18 The higher value of λ for both T^{3}_{2g} and A_{1g} modes compared to bulk Fe₃O₄ indicates that wet H₂ annealed films of smaller grain sizes have more electronic and structural disorder. Thus, they behave more like loosely electron correlated systems which affect Fe⁺³/Fe⁺² charge ordering and disturb Fe⁺³-Fe⁺²-Fe⁺³ trimerons, eventually suppressing the Verwey transition, in electrical transport properties compared to the more ordered phase vacuum-annealed Fe₃O₄ films of bigger sizes. Sketches of such electronic and magnetic disorders are depicted in the insets of Fig. 4. It is important to note that though the structural transition is followed by charge ordering in both sets of reduced films, it is the presence of electronic disorder/defects that mask the Verwey transition in wet H₂ annealed films.

The transport properties support the Verwey transition and the impact of external and internal factors like annealing conditions and stoichiometry of the ingredients. Since these contain primary information on the mechanism of electronic transport in Fe₃O₄, we re-plot the same in terms of $ln(\rho)$ vs 1/T curves [as shown in Figs. 5(a) and 5(b)], and the results were fitted to the general expression $\rho = \rho_0 \exp(T_0/T)^{n18}$ in order to determine whether the conduction regime corresponds to a semiconductor or to a disordered metal. The value of the pre-exponential ρ_0 indicates whether the conduction occurred by the extended states or by the localized states. To is a degree of a disorder associated with an activation energy Eg (in the case of semiconductors) and electronic localization length $\xi_{\rm L}$ (in the case of disordered metals). We found that the best fit was obtained for n=1 in wet H_2 annealed films in the entire temperature range (300-50 K), indicating semiconducting type behavior with E_g = 93-240 meV, which is higher than the reported corresponding bulk Fe₃O₄ values $(10-60 \text{ meV})^{24}$ (shown in Table I). These values of E_g indicate thermally activated nearest-neighbor-hopping (NNH) mechanism between localized states. The higher values of ρ_0 (~10⁵-10⁷ Ω cm) further indicate the presence of a wide range of localization and conduction by hopping in the localized state. Likewise, bulk Fe₃O₄, vacuum annealed films exhibit semiconducting behavior with E_g = 20-78 meV above the Verwey transition (300-110 K), wherein thermally activated hopping is dominated at higher temperatures. Below the Verwey transition, these curves deviate from linearity and can be fitted with a value for n = 1/4 corresponding to Mott variable range hopping (VRH) in the interval (110-70 K) due to localization behavior of charge carriers. The in situ growth temperature not only influences ρ and T_V values but also controls the exact borders of the VRH mechanism in vacuum annealed films, which is also argued in the literature that it arises due to its dependence on purity and the method of preparation. 18,24 According to the VRH theory, localized carriers form an impurity band, and the low-temperature resistivity is governed by a phonon-assisted hopping of carriers

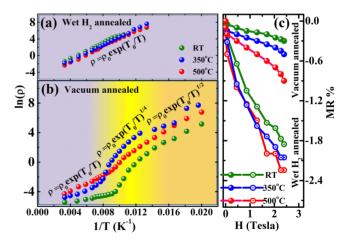


FIG. 5. Plot between (a) $\ln(\rho)$ vs 1/T for wet H₂ annealed; (b) vacuum annealed Fe₃O₄ films. (c) MR vs H curves for both sets of films at 300 K.

between localized states. The parameter T₀ (i.e., Mott temperature T_M in this case) is related to the density of localized states N(E_F) and localization length ξ_L by the Mott VRH²⁴ relation $\xi_L = [18/k_BN(E_F)]^{1/3}$. The observed T_M values $(2.4-10\times10^8)$ K) are slightly larger than the earlier reported values for Fe₃O₄ films $(1.68 \times 10^8 \text{ K})^{24}$ (shown in Table I). The localization length $\xi_{\rm L}$, which can be used to describe the spreading of the wave function, is estimated to be smaller than the distance of the nearest Fe ions, which may have occurred because we used N(E_F) value of high-temperature cubic Fe₃O₄ phase instead of low-temperature monoclinic phase of Fe₃O₄. However, at low temperatures (70-30 K), resistivity data are neither fit NNH nor Mott VRH expressions. Instead, they fit well to the Shklovskii and Efros (SE)-VRH model^{24,34} for n = 1/2. We may recall that the Coulomb interactions between charge carriers were ignored in the Mott VRH theory and that $N(E_{\rm F})$ was treated as a constant. If the long-range nature of the Coulomb interactions is taken into account, SE found that N(E_E) is no longer a constant, leading to a Coulomb gap $\Delta_{\rm CG} = k_{\rm B} (T_0 \times T_{\rm C})^{1/2}$, where $T_{\rm C}$ corresponds to the crossover temperature from Mott-VRH to SE-VRH^{21,29} and T₀ is basically SE temperature (shown in Table I) given by $T_{SE} = 2.8e^2/4\pi\xi_L k\varepsilon_0$ (where k is the dielectric constant). Now, ξ_L is approximated to more reasonable (1.9-2.1 nm) values which are nearly seven times the nearest neighbor hopping distance between Fe²⁺ and Fe³⁺ ions at octahedral B-sites. This indicates the presence of cationic/anionic defects in Fe₃O₄, which leads to the localization of charge carriers. When the temperature is high enough for a hopping electron to explore the energy range $\Delta_{\rm CG}$ (95–110 meV), the influence of the Coulomb gap can be ignored, and one recovers the Mott VRH conduction law. Below T_C, only electronic states inside the Coulomb gap are accessible, and the SE-VRH conduction law is expected to follow. Hence, the resistivity plots of our vacuum annealed films have three linear regions having different slopes, indicating three charge transport mechanisms.

The formation of a defect band and its spin-splitting can also be envisaged via magnetoresistance (MR) measurement, which is essential to study the effect of the density of defect states on the magnetic exchange. Figure 5(c) displays MR-H curves for all samples at room temperature. By comparing MR-H curves, we found that the MR in wet H₂ annealed films (1.8%-2.2%) is higher than that of vacuum annealed films (0.3%-0.9%) (cf. Table I), which further supports our argument about the presence of electronic disorder/defects. Interestingly, in situ growth temperature also affects the MR of vacuum annealed films like other electronic and magnetic properties. The higher value of MR in wet H₂ annealed films can be explained from the fact that alignment of the disordered spins of the smaller grain boundaries significantly affects the MR because the electrical transport mechanism of the films is due to tunneling.^{21,32} This suggests that the frozen interfacial/ surface moments caused by cationic/anionic defects affect the MR significantly even though they affect the magnetization of the films only slightly. Thus, we observed a slight change in the Fe^{3+/}Fe²⁺ ratio,³² and defects made the tunneling barrier higher and more stable, enhanced the room temperature MR,

and stabilized the Fe₃O₄ films chemically. Moreover, the electric transport measurement under various magnetic fields [shown in Fig. S2(b) in the supplementary material] shows no influence on the existence of the Verwey transition, except in the enhancement of MR values in the entire temperature range. This is indicative of no change of spin ordering at the Verwey transition and also the absence of its correlation with charge ordering.

These observations could be useful for the futuristic spintronic applications of nanocrystalline Fe₃O₄ thin film devices where annealing is invariably employed.

CONCLUSION

We have shown how the optimization of the postreduction treatment is crucial in achieving the best electrical transport and magnetic properties in nanocrystalline Fe₃O₄ thin films. The following conclusions can be made from the present study:

- 1. The Verwey transition is a structural transition, which drives charge ordering.
- 2. This structural transition causes non-saturation effects, and this can lead to a drop in magnetization values around the Verwey transition at low fields. However, application of large magnetic fields can saturate both cubic and monoclinic phases of Fe₃O₄, and there should not be any such drop. Saturation magnetization increases upon lowering the temperatures similarly to other ferrites without marking any changes at the Verwey transition. Thus, the Verwey transition has hardly any correlation with magnetic symmetry change, but at low fields owing to increased anisotropy, coercivity, and M_R/M_S, values normally may get changed.
- 3. Even though structural transition precedes charge ordering in both sets of reduction treated films, it is the electronic disorder/defects that mask the Verwey transition in wet H₂ films compared to the vacuum annealed films.
- 4. No correlation between spin and charge coupling has been found in our nanocrystalline Fe₃O₄ thin films.
- 5. The temperature during in situ growth influences physical properties of vacuum annealed Fe₃O₄ films.

SUPPLEMENTARY MATERIAL

See supplementary material for phase purity check of Fe₃O₄ thin films examined by room temperature X-ray spectroscopic data. The non-saturation behavior of M-H loops and the absence of magnetic field effect on electric transport properties across the Verwey transition have been demonstrated.

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REFERENCES

- ¹A. Mitra, J. Mohapatra, S. S. Meena, C. V. Tomy, and M. Aslam, J. Phys. Chem. C 118, 19356 (2014).
- ²S. Lim, B. Choi, S. Y. Lee, S. Lim, B. Choi, S. Y. Lee, S. Lee, H.-H. Nahm, Y.-H. Kim, T. Kim, J.-G. Park, J. Lee, J. Hong, S. G. Kwon, and T. Hyeon, Nano Lett. 18, 1745 (2018).
- ³J. Lee, S. G. Kwon, J.-G. Park, and T. Hyeon, Nano Lett. **15**, 4337 (2015).
- ⁴S. D. Jong, R. Kukreja, C. Trabant, N. Pontius, C. F. Chang, T. Kachel, M. Beye, F. Sorgenfrei, C. H. Back, B. Bräuer, W. F. Schlotter, J. J. Turner, O. Krupin, M. Doehler, D. Zhu, M. A. Hossain, A. O. Scherz, D. Fausti, F. Novelli, M. Esposito, W.-S. Lee, Y. D. Chuang, D. H. Lu, R. G. Moore, M. Yi, M. Trigo, P. Kirchmann, L. Pathey, M. S. Golden, M. Buchholz, P. Metcalf, F. Parmigiani, W. Wurth, A. Föhlisch, C. Schüßler-Langeheine, and H. A. Dürr, Nat. Mater. 12, 882 (2012).
- ⁵Z. Zhang and S. Satpathy, Phys. Rev. B **44**, 13319 (1991).
- ⁶E. J. W. Verwey and P. W. Haayman, Physica **8**, 979 (1941).
- ⁷H. Y. Huang, Z. Y. Chen, R.-P. Wang, F. M. F. de Groot, W. B. Wu, J. Okamoto, A. Chainani, A. Singh, Z.-Y. Li, J.-S. Zhou, H.-T. Jeng, G. Y. Guo, J.-G. Park, L. H. Tjeng, C. T. Chen, and D. J. Huang, Nat. Commun. 8, 1 (2017)
- (2017).

 ⁸J. P. Wright, J. P. Attfield, and P. G. Radaelli, Phys. Rev. Lett. **87**, 266401 (2001).
- ⁹X. Liu, W. Mi, Q. Zhang, and X. Zhang, Phys. Rev. B **96**, 214434 (2017).
- ¹⁰M. Hoesch, P. Piekarz, A. Bosak, M. L. Tacon, M. Krisch, A. Kozłowski, A. M. Oles, and K. Parlinski, Phys. Rev. Lett. 110, 207204 (2013).
- ¹¹M. Liu, J. Hoffman, J. Wang, J. Zhang, B. Nelson-Cheeseman, and A. Bhattacharya, Sci. Rep. 3, 1876 (2013).
- ¹²R. Ramos, T. Kikkawa, K. Uchida, H. Adachi, I. Lucas, M. H. Aguirre, P. Algarabel, L. Morellón, S. Maekawa, E. Saitoh, and M. R. Ibarra, Appl. Phys. Lett. **102**, 072413 (2013).
- ¹³M. Bohra and S. C. Sahoo, J. Alloys Compd. **699**, 1118 (2017).
- ¹⁴X. H. Liu, W. Liu, Z. D. Zhang, and C. F. Chang, J. Appl. Phys. **123**, 083903 (2018).
- ¹⁵S. G. Bhat and P. S. Anil Kumar, Mater. Res. Express **3**, 125905 (2016).

- ¹⁶P. Dey, R. Rawat, S. R. Potdar, R. J. Choudhary, and A. Banerjee, J. Appl. Phys. 115, 17C110 (2014).
- ¹⁷M. Bohra, K. E. Prasad, R. Bollina, S. C. Sahoo, and N. Kumar, J. Mag. Mag. Mater. 418, 137 (2016).
- ¹⁸A. Kumar, D. K. Pandya, and S. Chaudhary, J. Appl. Phys. **112**, 073909 (2012).
- ¹⁹X. Liu, W. Mi, Q. Zhang, and X. Zhang, Appl. Phys. Lett. **113**, 012401 (2018).
- ²⁰W. Mi, Z. Guo, Q. Wang, Y. Yang, and H. Bai, Scr. Mater. **68**, 972 (2013).
- ²¹W. B. Mi, J. J. Shen, E. Y. Jiang, and H. L. Bai, Acta Mater. **55**, 1919 (2007).
- ²²X. Liu, L. Yin, and W. Mi, Sci. Rep. 7, 43403 (2017).
- ²³M. Bohra, S. Prasad, N. Venketaramani, N. Kumar, S. C. Sahoo, and R. Krishnan, J. Mag. Mag. Mater. 321, 3738 (2009).
- ²⁴R. Prakash, R. J. Choudhary, L. S. Sharath Chandra, N. Lakshmi, and D. M. Phase, J. Phys. Condens. Matter 19, 486212 (2007).
- 25 X. Liu, C. F. Chang, A. D. Rata, A. C. Komarek, and L. H. Tjeng, npj Quant. Mater. 1, 16027 (2016).
- ²⁶B. Ramamurthy Acharya, S. Prasad, N. Venkataramani, S. N. Shringi, and R. Krishnan, J. Appl. Phys. **79**, 478 (1996).
- ²⁷T. Kim, S. Lim, J. Hong, S. Gu Kwon, J. Okamoto, Z. Y. Chen, J. Jeong, S. Kang, J. C. Leiner, J. T. Lim, C. S. Kim, D. J. Huang, T. Hyeon, S. Lee, and J.-G. Park, Sci. Rep. 8, 5092 (2018).
- ²⁸H. F. Tian, T. L. Qu, L. B. Luo, J. J. Yang, S. M. Guo, H. Y. Zhang, Y. G. Zhao, and J. Q. Li, Appl. Phys. Lett. **92**, 063507 (2008).
- ²⁹J. García and G. Subías, J. Phys. Condens. Matter **16**, R145 (2004).
- 30R. G. Moore, J. Zhang, V. B. Nascimento, R. Jin, J. Guo, G. T. Wang, Z. Fang, D. Mandrus, and E. W. Plummer, Science 318, 615 (2007).
- ³¹L. Wang, J. Li, W. Ding, T. Zhou, W. Zhong, J. Wu, and Y. Du, J. Mag. Mag. Mater. 207, 111 (1999).
- ³²W. B. Mi, E. Y. Jiang, and H. L. Bai, J. Phys. D Appl. Phys. **42**, 105007 (2009).
- (2009). ³³R. Gupta, A. K. Sood, P. Metcalf, and J. M. Honig, Phys. Rev. B **65**, 104430 (2002).
- ³⁴Y.-L. Huang, S.-P. Chiu, Z.-X. Zhu, Z.-Q. Li, and J.-J. Lin, J. Appl. Phys. **107**, 063715 (2010).