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Influence of operating parameters on the single pass removal efficiency during the photocatalytic degradation of acrylonitrile.

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Highlights

- Single-pass removal efficiencies can be improved according operating parameters
- A complete mineralization can be reached irrespective of initial pollutant concentration
- Conversion vs. mineralization rates show the probable formation of intermediates

1. Introduction

The hospital is a place where the air quality is very important as it plays an important role in the health of both patients and hospital staff. Microbial contamination in relation to nosocomial infections has been at the fore front of hospital air quality whilst chemical contamination is rarely studied [1]. One of the most demanding areas in terms of the air quality is the operating room (OR). Surgeons and operating theatre staff are routinely exposed to pollution from the surgical smoke. The chemicals present in greatest quantity in surgical smoke are hydrocarbons and nitrile compounds among which is acrylonitrile. Barret and Garber [2] reported acrylonitrile concentrations to be between 1 and 1.6 ppm. In ORs air quality is assured mostly by HVAC. However, this is not always enough thus additional treatment devices can be used to purify air. Stand-alone devices that use photocatalytic oxidation (PCO) are becoming increasingly popular to treat indoor air. The aim of this work is to assess if PCO is efficient for removing acrylonitrile and to study the influence of three operational parameters (air velocity, light intensity and initial concentration) on the photocatalytic removal efficiency of acrylonitrile.

2. Methods

A 420 L multi-pass dynamic photocatalytic reactor was used to carry out the PCO experiments. It has been described and modelled in a previous articles [3],[4]. The media used was supplied by Saint-Gobain Quartz (QUARTZEL[®]) and consisted of quartz fibers coated with TiO₂ deposited through a sol-gel method. The media was irradiated with two 18-W UVC fluorescent tubes (Phillips PL-L series). Acrylonitrile was supplied from a certified commercial gas cylinder produced by Air products. The cylinder contained 150 ppm of acrylonitrile balanced in nitrogen at a pressure of 150 bars. The selected average light intensities (I) ranged from 1 to 4.5 mW.cm⁻². The air velocities corresponding to the selected flow rates (73 - 216 Nm³.h⁻¹) varied from 0.5 to 1.5 m.s⁻¹ and were determined as the ratio between the air flow rate and the cross-sectional area (0.04 m²) of the photocatalytic module. Initial concentrations (C₀) chosen for the experiments ranged from 0.5 to 10 ppm which were representative of concentrations reported in the literature. The relative humidity (RH) was maintained at 50% at 20°C to be representative of average humidity levels found in operating rooms in European countries.



3. Results and discussion

First results showed that acrylonitrile is degraded within 1h based on the experimental conditions of $C_0=2ppm$, $v=1 m.s^{-1}$ and $I = 4.5 mW.m^{-2}$ (Fig. 1). It was observed that the degradation followed a first order kinetic. The experimental points were then fitted to the model developed by Dumont and Héquet. The model describes the first order decay relationship of concentration vs. time and helps to determine the single pass removal efficiency. It is as follows:

$$C = C_0 \exp\left\{-\frac{t}{\tau_R} \left| [1 - \exp(-\alpha)] \right\}$$
(1)

In Eqn. 1 the term α is the single pass removal efficiency and represents the fraction of the total flow treated during the time τ_R (residence time in the reactor). C is the pollutant concentration at time t and C₀ is the initial pollutant concentration. Numerical resolutions are carried out using Excel[®] Solver which is based on the least- square method.



Figure 1. Experimental degradation curve of acrylonitrile fitted to model (C₀ = 2 ppm; v = 1 m.s⁻¹; I = 4.5 mW.cm⁻²).

4. Conclusions

Conclusively, PCO is able to degrade acrylonitrile. The experimental points are well fitted to the model and enable the calculation of the one-pass removal efficiency. Since the global objective is to study the effect that the operating parameters have on the single pass removal efficiency, experiments are conducted at the selected air velocities, light intensities and initial concentrations to determine their influence.

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