

List of figures:

Fig. 1. XRD patterns of (a) Nb₂O₅, (b) Cu-Nb₂O₅, (c) Au-Nb₂O₅ AP, (d) AuCu-Nb₂O₅ DR and (e) Au-Nb₂O₅ DR.

Fig. 2. (A) Representative TEM images of the catalysts and (B) gold particle size distribution in (a) Au-Nb₂O₅ AP, (b) Au-Nb₂O₅ DR and (c) AuCu-Nb₂O₅ DR samples basing on the TEM images.

Fig. 3. (A) UV-vis spectra of (a) Nb₂O₅, (b) Au-Nb₂O₅ AP, (c) Au-Nb₂O₅ DR, (d) Cu-Nb₂O₅ and (e) AuCu-Nb₂O₅ DR. (B) Band gap estimation for parent Nb₂O₅ using Tauc's plot method.

Fig. 4. FTIR spectra of (a) Nb₂O₅, (b) Au-Nb₂O₅ AP, (c) Au-Nb₂O₅ DR, (d) Cu-Nb₂O₅, (e) AuCu-Nb₂O₅ DR after Py adsorption followed by evacuation at 150 °C. The spectra were subtracted from the spectra of the activated samples before Py adsorption and normalized to mass of wafer = 13.0 mg.

Fig. 5. Activity (A, C) and selectivity (B, D) of the catalysts at the steady state during the photocatalytic oxidation of methanol under UV irradiation (irradiance ≈ 205 mW/cm²) and visible light irradiation ($\lambda > 390$ nm; irradiance ≈ 190 mW/cm²).

Fig. 6. The influence of Lewis (A) and Brønsted (B) acidity of the catalysts on their activity in methanol photooxidation (UV light, light intensity ≈ 70 mW/cm²).

Fig. 7. IR spectra of the adsorbed surface species on Au-Nb₂O₅ DR and Au-Nb₂O₅ AP during the first minutes of irradiation with UV light (light intensity ≈ 205 mW/cm²). The spectra were subtracted from the IR spectra of the samples before irradiation.

Fig. 8. IR spectra of the adsorbed surface species on AuCu-Nb₂O₅ DR during the first minutes of irradiation with visible light ($\lambda > 390$ nm; light intensity ≈ 190 mW/cm²). The spectra were subtracted from the IR spectra of the sample before irradiation.

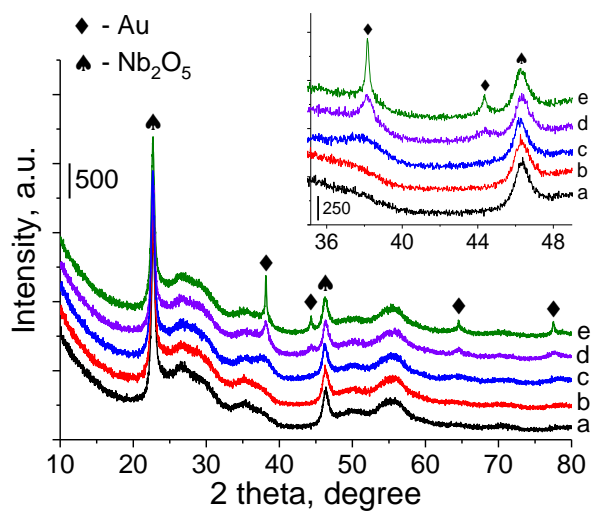


Fig. 1. XRD patterns of (a) Nb₂O₅, (b) Cu-Nb₂O₅, (c) Au-Nb₂O₅ AP, (d) AuCu-Nb₂O₅ DR and (e) Au-Nb₂O₅ DR.

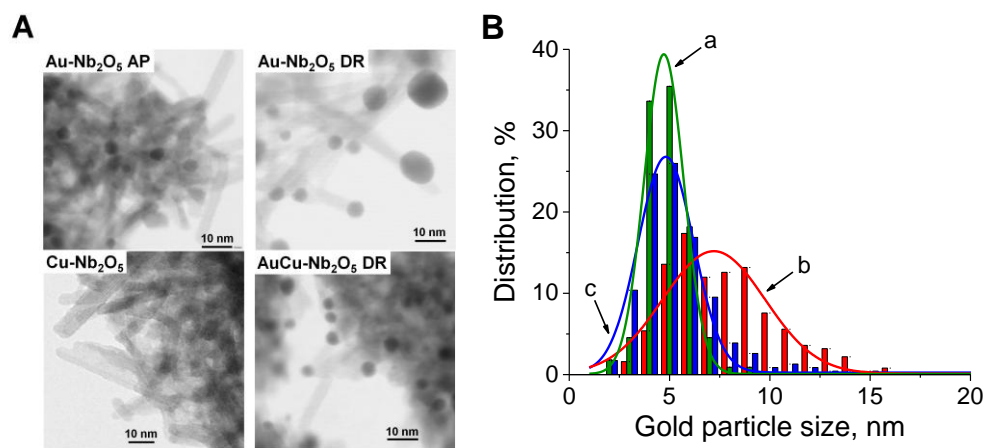


Fig. 2. (A) Representative TEM images of the catalysts and (B) gold particle size distribution in (a) Au-Nb₂O₅ AP, (b) Au-Nb₂O₅ DR and (c) AuCu-Nb₂O₅ DR samples basing on the TEM images.

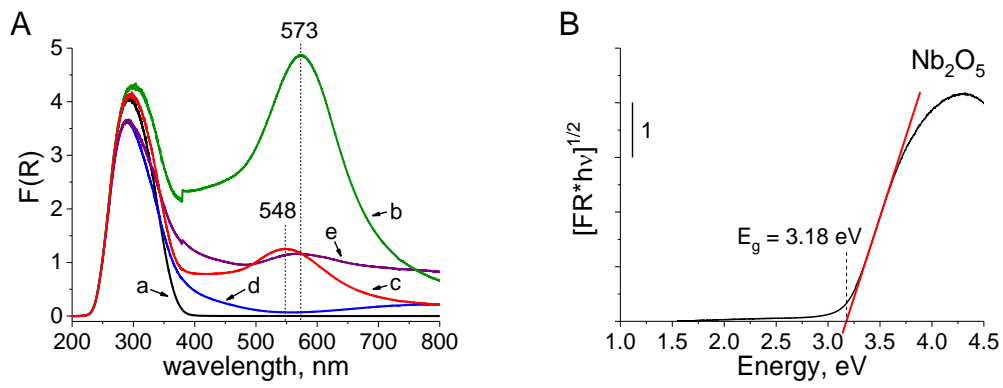


Fig. 3. (A) UV-vis spectra of (a) Nb₂O₅, (b) Au-Nb₂O₅ AP, (c) Au-Nb₂O₅ DR, (d) Cu-Nb₂O₅ and (e) AuCu-Nb₂O₅ DR. (B) Band gap estimation for parent Nb₂O₅ using Tauc's plot method.

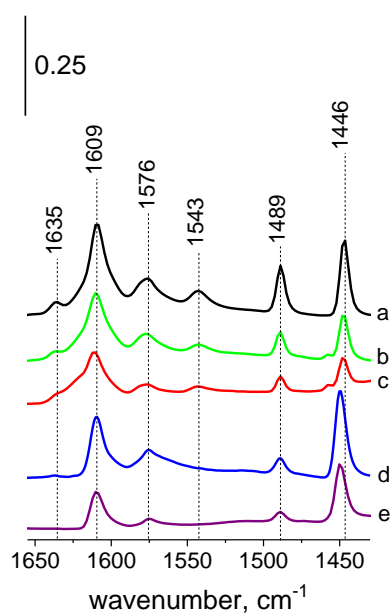


Fig. 4. FTIR spectra of (a) Nb₂O₅, (b) Au-Nb₂O₅ AP, (c) Au-Nb₂O₅ DR, (d) Cu-Nb₂O₅, (e) AuCu-Nb₂O₅ DR after Py adsorption followed by evacuation at 150 °C. The spectra were subtracted from the spectra of the activated samples before Py adsorption and normalized to mass of wafer = 13.0 mg.

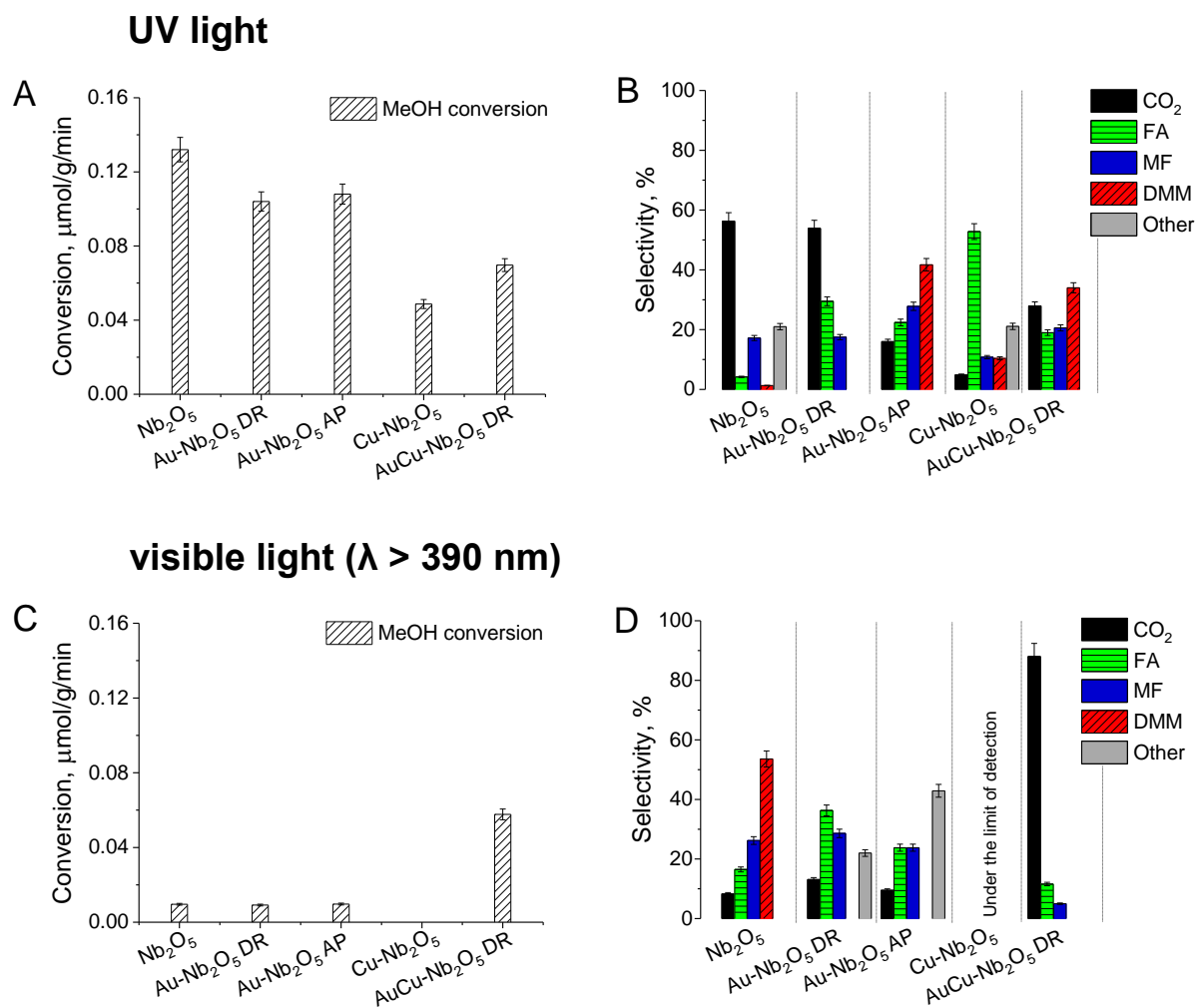


Fig. 5. Activity (**A**, **C**) and selectivity (**B**, **D**) of the catalysts at the steady state during the photocatalytic oxidation of methanol under UV irradiation (irradiance ≈ 205 mW/cm²) and visible light irradiation ($\lambda > 390$ nm; irradiance ≈ 190 mW/cm²).

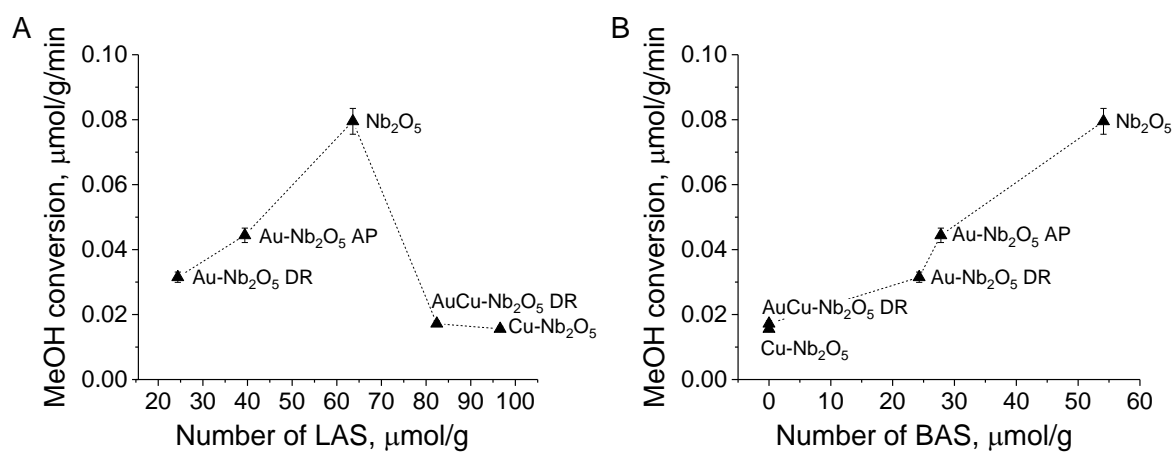


Fig. 6. The influence of Lewis (A) and Brønsted (B) acidity of the catalysts on their activity in methanol photooxidation (UV light, light intensity $\approx 70 \text{ mW/cm}^2$).

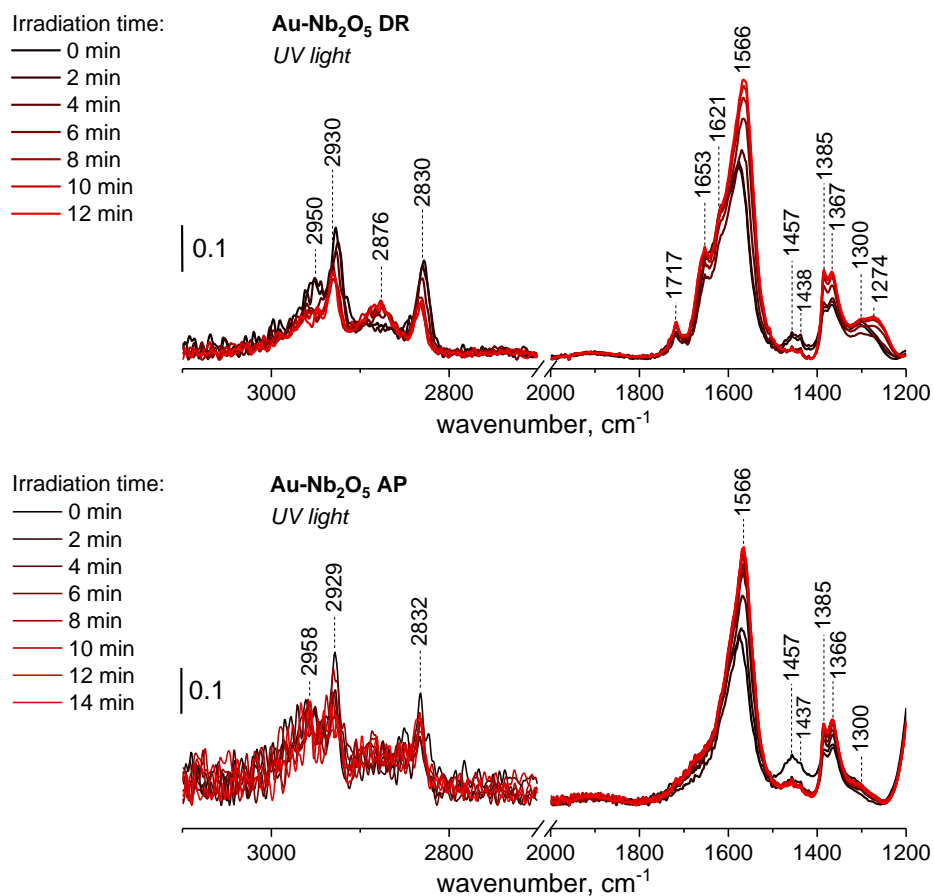


Fig. 7. IR spectra of the adsorbed surface species on Au-Nb₂O₅ DR and Au-Nb₂O₅ AP during the first minutes of irradiation with UV light (light intensity $\approx 205 \text{ mW/cm}^2$). The spectra were subtracted from the IR spectra of the samples before irradiation.

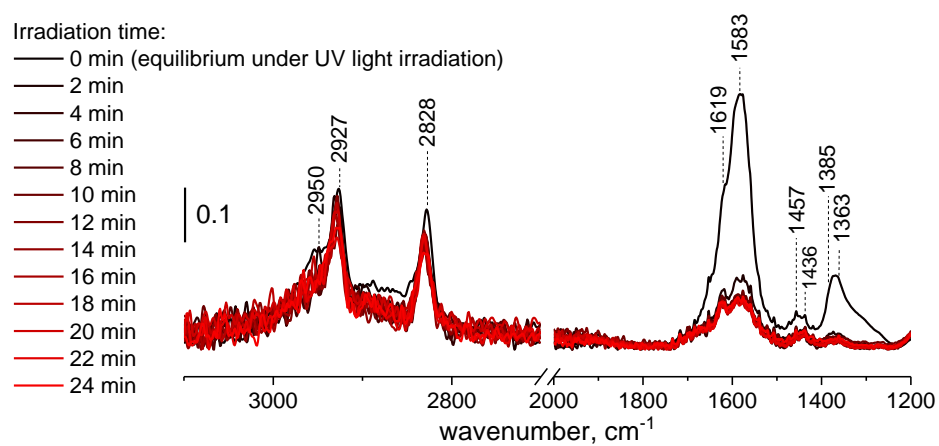


Fig. 8. IR spectra of the adsorbed surface species on AuCu-Nb₂O₅ DR during the first minutes of irradiation with visible light ($\lambda > 390$ nm; light intensity ≈ 190 mW/cm²). The spectra were subtracted from the IR spectra of the sample before irradiation.