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**Key Points:**

- We observed high chlorophyll *a* and macronutrient concentrations co-occurring at the bottom of Antarctic land-fast sea ice in late spring
- Remineralization processes and adsorption within sea ice best explain the required sources of nitrate and nitrite, phosphate, and ammonium
- Biogeochemical modeling suggests that sea-ice diatoms have a lower affinity for silicic acid than pelagic diatoms

**Supporting Information:**

- Supporting Information S1

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## Field Observations and Physical-Biogeochemical Modeling Suggest Low Silicon Affinity for Antarctic Fast Ice Diatoms

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**Abstract** We use field observations from late spring and a one-dimensional sea-ice model to explore a high nutrient, high chlorophyll system in Antarctic land-fast ice. Lack of variability in chlorophyll *a* concentration and organic carbon content over the 17-day sampling period suggests a balance between macronutrient sources and biological uptake. Nitrate, nitrite, phosphate, and ammonium were measured at concentrations well above salinity-predicted levels, indicating nutrient accumulation fueled by remineralization processes. However, silicic acid (DSi) was depleted relative to seawater and was potentially limiting. One-dimensional physical-biogeochemical sea-ice model simulations at the observation site achieve extremely high algal growth and DSi uptake with a DSi half-saturation constant used for pelagic diatoms ( $K_{Si} = 3.9 \mu\text{M}$ ) and are not sufficiently improved by tuning the DSi:carbon ratio or DSi remineralization rate. In contrast, diatom biomass in the bottom ice, which makes up 70% of the observed chlorophyll, is simulated using  $K_{Si}$  an order of magnitude higher ( $50 \mu\text{M}$ ), a value similar to that measured in a few Antarctic diatom cultures. Some sea-ice diatoms may therefore experience limitation at relatively high ambient DSi concentrations compared to pelagic diatoms. Our study highlights the urgent need for observational data on sea-ice algal affinity for DSi to further support this hypothesis. A lower algal growth rate increases model predictions of DSi in the upper sea ice to more accurate concentrations. The model currently does not account for the non-diatom communities that dominate those layers, and thus, modeling diatom communities overpredicts DSi uptake in the upper ice.

**Plain Language Summary** Microscopic, single-celled algae growing inside Antarctic sea ice play a small, but important, role in the carbon cycle of polar oceans. These algae use photosynthesis to convert carbon dioxide to organic carbon and are an important food source for larger organisms. The growth of sea-ice algae is partly controlled by their uptake of essential nutrients under favorable light conditions. Therefore, a better understanding of nutrients in sea ice can improve ecosystem models. We studied two weeks of measurements of coastal sea-ice characteristics in late spring at a site in East Antarctica. There were high amounts of both algae and nutrients, indicating recycling processes that return nutrients to the sea ice, even as algae consume nutrients. Sticky biofilms may also prevent nutrients from being removed by seawater. We then tested a sea-ice model to see if it matched our field measurements. We had to adjust how readily algal cells in the model take up silicon—a nutrient used to build diatoms' (the dominant algal group) walls—from their environment to obtain a close match. Previously used values were taken from experiments with non-ice associated marine algae, so better estimates specific to sea-ice algae are needed to accurately include sea-ice processes in large-scale models.

### 1. Introduction

Covering approximately 7% of the planet at its maximum (Weeks & Hibler, 2010), sea ice is not only one of the world's largest biomes, supporting a variety of life, but also a fundamental feature of the Earth's climate system (Manabe & Stouffer, 1980). Sea ice insulates the polar oceans from the atmosphere, limiting heat, mass, and momentum exchanges and reflecting a large part of solar radiation back to the atmosphere.

The oceans have likely absorbed almost half of fossil-fuel emissions, with the Southern Ocean storing ~9% of that anthropogenic carbon (Sabine et al., 2004).

Sea ice provides a platform for many organisms, including primary producers, that live attached to the ice or in brine inclusions in the sea ice. With ice covering almost  $17.9 \times 10^6$  km<sup>2</sup> of the Southern Ocean at its maximum extent in 2018 (National Snow and Ice Data Center; Fetterer et al., 2017), sea-ice algae likely play a small but significant role in the Southern Ocean carbon cycle. Primary production in sea ice is estimated to contribute ~12% of total production in the Antarctic sea-ice zone and ~1% of total production in the Southern Ocean (Saenz & Arrigo, 2014). In particular, regions with land-fast ice, which is held motionless through attachment to the coast, shallow shoals, or grounded icebergs, have a disproportionately large fraction of primary productivity relative to their surface areas (Arrigo et al., 1997, 2003; Meiners et al., 2018). This is most likely due to greater nutrient availability in comparison to offshore, free-drifting pack ice (van Leeuwe et al., 2018). Fast ice's proximity to more dynamic coastal and shelf currents (Cota et al., 1991) brings seawater with nutrient concentrations that are often higher than those of open ocean waters (Gradinger, 2009). There is also evidence that fast ice has iron levels at least an order of magnitude higher than most Antarctic surface waters (Lannuzel et al., 2016), providing an important micronutrient in the high nutrient, low chlorophyll Southern Ocean (Boyd et al., 2000; Boyd et al., 2012; Martin, 1990). Thus, in order to improve estimates of fast ice productivity, it is necessary to understand nutrient availability and uptake dynamics. Overall, knowing what drives and quantifying patterns of sea-ice productivity will clarify its role in rapidly changing polar climate systems.

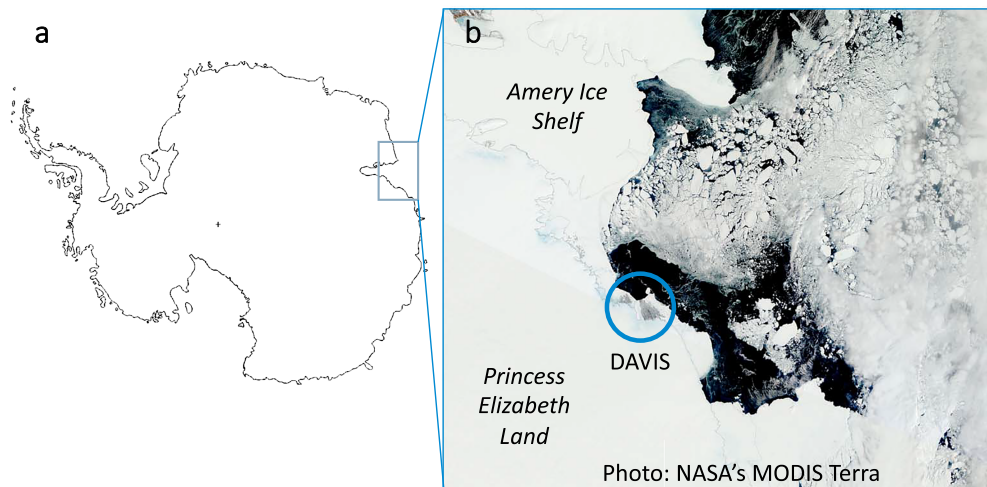
Currently, calculations of primary productivity in Antarctic sea ice have large uncertainties, with estimates ranging from 23 to 70 Tg C/year (Arrigo et al., 1997; Legendre et al., 1992; Saenz & Arrigo, 2014). Numerical models of sea-ice biogeochemistry improve our ability to study an environment challenging for field sampling and remote satellite measurements. One-dimensional modeling studies of biogeochemical sea-ice processes published over the last two decades (e.g., Arrigo et al., 1991; Duarte et al., 2017; Lavoie et al., 2005; Tedesco et al., 2010; Vancoppenolle & Tedesco, 2017) showed realistic simulations of the vertical structure and time variations of sea-ice chlorophyll, used as the main evaluation criterion when comparing observed and modeled quantities. However, Duarte et al. (2017) showed the difficulty of achieving realistic simulations of chlorophyll and nutrients in Arctic pack ice at the same time, partly due to limited knowledge of nutrient retention and motion within the brine matrix and at the ice-water interface. In addition, fundamental deficiencies in the predictive capability of such models remain. For instance, Lavoie et al. (2005) demonstrated that different parameters should be used for the same model to be successful at different sites. In this context, it is important to improve process understanding and observation-based model evaluations including all aspects of the system (algae, organic matter, nutrients, light, and brine dynamics). In particular, because macronutrient limitation often constrains primary productivity in Antarctic sea ice in the spring and summer, nutrient availability and supply processes are considered ideal subjects for joint field and model studies (Steiner et al., 2016).

The aim of this study is to characterize and quantify the biogeochemical processes that influenced primary production in Antarctic fast ice during a time series collected in spring. We combine sea-ice observations from East Antarctica and one-dimensional sea-ice modeling, focusing on both ice physics (e.g., temperature, salinity, and brine fraction) and biogeochemistry (e.g., chlorophyll *a* [Chl*a*], particulate organic carbon [POC], and macronutrient concentrations). We investigate the nutrient cycling processes that may underlie our measurements, and we examine possible limitations on sea-ice algal growth.

## 2. Methods

### 2.1. Data Collection

A time series of physical and biogeochemical data was collected from coastal land-fast sea ice and under-ice seawater at Davis Station, East Antarctica (68°34'36"S, 77°58'03"E; Figure 1) in late austral spring 2015 (Nov 16, Nov 20, Nov 23, Nov 26, Nov 29, and Dec 2). Sea-ice temperature, salinity, macronutrients (nitrate [NO<sub>3</sub><sup>-</sup>], nitrite [NO<sub>2</sub><sup>-</sup>], ammonium [NH<sub>4</sub><sup>+</sup>], phosphate [PO<sub>4</sub><sup>3-</sup>], and DSi), POC, and Chl*a* were measured. Ice core depths were measured from the top of the ice cover. Ice cores were collected following trace-metal clean methods described in van der Merwe et al. (2009) using a 0.14-m internal diameter corer (Lichtert Industrie, Belgium) driven by an electric drill. At each sampling time, three ice cores (for temperature and salinity, for



**Figure 1.** Location of Davis Station, Antarctica, outlined by blue circle on (a) general map of Antarctica and (b) image from NASA's Moderate Resolution Imaging Spectroradiometer (MODIS) Terra on 17 November 2015 (1 day after the start of the time series). Landmass of continent is outlined and on top is fast ice. The large glacier entering the sea is the Amery Ice Shelf.

Chla, and for POC and macronutrients) were collected about 0.1 m apart to minimize between-core variability. To minimize ice and brine loss during sampling, ice sections were cut directly after sampling (i.e., on the ice), starting from the bottom of the core, which is the most sensitive to light and brine loss. Under-ice seawater samples were collected following the methods from van der Merwe et al. (2011), with a custom-built polycarbonate Helmond-Byrne water sampler suspended from a Kevlar line and triggered using an all-Teflon® messenger (Sedwick et al., 1997). The sampler was lowered through the ice and triggered at 0-, 3-, and 10-m depths.

Ice core temperatures were measured from the first ice core approximately every 0.05 m, starting 0.025 m from the top, by inserting a Testotherm 720 temperature probe (precision of  $\pm 0.1$  °C) into a drilled hole. Salinity of 0.1-m melted ice core sections and seawater were measured using a Thermo Scientific Orion Model 125A Plus conductivity meter (precision of  $\pm 0.1$ ). To increase the vertical resolution where biomass was the highest, the basal 0.1-m ice core section was subsampled in two smaller sections of 0.06- and 0.04-m thickness.

Ice cores for biogeochemical parameters were cut in 0.1-m sections, again with 0.06 and 0.04 m subsampling in the lowest 0.1 m, using a medical grade stainless steel bone saw (Richards Medical). The collected ice sections were placed in individual acid-cleaned melting containers and allowed to melt in the dark at room temperature (Rintala et al., 2014). Filters and all glassware in contact with the seawater and melted sea-ice samples dedicated to POC analysis were combusted at 450 °C for 12 hr prior to sampling. Once fully melted, ice and seawater samples were gently stirred to homogenize before filtration (0.2–2.5 L) onto 25-mm-diameter GF/F filters (0.7  $\mu\text{m}$ , Whatman) for POC and Chla determination. Two 20-ml samples were filtered over 25-mm syringe filters (0.2  $\mu\text{m}$ , Acrodisc), collected in 30-ml plastic vials, and kept frozen at  $-20$  °C for macronutrient analysis. Filters for POC determination were directly dried overnight at 50 °C and stored in the dark at room temperature in pre-combusted vials. Filters for Chla were stored at  $-80$  °C until fluorometric analysis in the home laboratories.

Prior to analysis, inorganic carbon was removed from the POC filters by 24-hr exposure to concentrated hydrochloric acid fumes inside a closed glass container. Next, filters were packed in silver cups, and POC content was determined at the Vrije Universiteit Brussel (Belgium) using an elemental analyzer (EuroEA3000, Eurovector) coupled online via a Con-Flo III interface to a Thermo Delta V isotope ratio mass spectrometer (detection limit of 0.1  $\mu\text{g}$  and precision of 1%). Chla was extracted in 90% acetone according to standard techniques (Holm-Hansen et al., 1965; Lorenzen, 1966). Samples were stored in the dark at  $-20$  °C for 24 hr before analysis. A Turner Trilogy fluorometer, previously calibrated with Chla standards from Turner Designs (Sunnyvale, California), was then used to measure Chla (1.1% precision). Dissolved

inorganic nutrients ( $\text{NO}_3^- + \text{NO}_2^-$ ,  $\text{NH}_4^+$ ,  $\text{PO}_4^{3-}$ , and DSi) concentrations were determined using an automated QuAAtro39 analyzer (Seal Analytical Inc.) following standard colorimetric methods (Hansen & Koroleff, 1999). Nutrients were reacted and converted into a light-absorbing compounds whose absorbances were measured with a flow-through spectrophotometer. Concentrations were measured as bulk concentrations, representing the entire section of core (brine + ice). Precisions were  $\pm 0.6 \mu\text{M}$  for  $\text{NO}_3^- + \text{NO}_2^-$ ,  $\pm 0.1 \mu\text{M}$  for  $\text{NH}_4^+$ ,  $\pm 0.1 \mu\text{M}$  for  $\text{PO}_4^{3-}$ , and  $\pm 0.5 \mu\text{M}$  for DSi. Detection limits were  $1.52 \mu\text{M}$  for  $\text{NO}_3^- + \text{NO}_2^-$ ,  $0.37 \mu\text{M}$  for  $\text{NH}_4^+$ ,  $0.30 \mu\text{M}$  for  $\text{PO}_4^{3-}$ , and  $1.41 \mu\text{M}$  for DSi.

## 2.2. Analysis

Brine fraction, or the amount of a given section of sea ice that exists as liquid brine within the ice, was calculated as a function of temperature and salinity (Cox & Weeks, 1983). Because the depth intervals at which temperature and salinity were measured were not the same, temperatures matching the salinity measurement depths were interpolated using a parabolic best fit line ( $R^2 = 0.99$  Nov 16, 0.98 Nov 20, 0.99 Nov 23, 0.96 Nov 26, 0.96 Nov 29, and 0.89 Dec 30).

In abiotic ice systems, inorganic nutrients are generally considered to behave conservatively with salinity (Fripiat et al., 2017; Thomas et al., 2010). Theoretical dilution lines (TDLs), or the predicted concentration of nutrients for a given salinity, were established from nutrient concentrations in seawater 10 m below the ice, averaged for all six sampling days (standard deviations as a percent of the average: 2.8%  $\text{NO}_3^- + \text{NO}_2^-$ , 77.7%  $\text{NH}_4^+$ , 2.5%  $\text{PO}_4^{3-}$ , and 6.9% DSi). A depth of 10 m was chosen to minimize the influence of basal sea-ice algae communities and of fertilization as ice melts and releases nutrients at the ice-water interface.

## 2.3. Model

LIM1D (version 3, source 3.26) is a one-dimensional model that includes a representation of the major physical and biogeochemical sea-ice characteristics (<http://forge.ipsl.jussieu.fr/lim1d/wiki>; Vancoppenolle et al., 2010; Vancoppenolle & Tedesco, 2017). Sea ice is considered horizontally uniform with ten vertical layers of ice covered by a layer of snow; previous studies indicate that ten layers sufficiently resolve vertical profiles of biogeochemical processes (Moreau et al., 2014; Moreau et al., 2015; Vancoppenolle et al., 2010; Vancoppenolle & Tedesco, 2017). For physical processes, the model includes (a) ice thermodynamics (growth and melt of sea ice and diffusion of heat), (b) brine transport (natural brine convection and percolation), and (c) radiative transfer, which assumes that 0.03 m of snow and 0.1 m of ice scatter the majority of radiation. Photosynthetic algae are assumed to be responsible for most nutrient cycling and carbon uptake; heterotrophic grazers are not included. A pool of detrital organic matter, which contributes to nutrient recycling, is also included in the model. Bacteria are not explicitly resolved by the model, but bacterial respiration and remineralization rates are calculated based on their relationship to sea-ice temperature (Moreau et al., 2015). Photosynthesis is limited by temperature, light, brine salinity, and macronutrients. The C:Chl*a* ratio changes directly with photosynthetically active radiation (eq. 20.16 in Vancoppenolle & Tedesco, 2017; Aumont et al., 2015; Doney et al., 1996). Macronutrients follow calculations of carbon content at a constant C:Si:N:P = 106:12:16:1 molar ratio, according to the Redfield ratio with the addition of silicon for diatom uptake (Brzezinski, 1985; Redfield et al., 1963; Sarthou et al., 2005). Photosynthetic limitation by a specific factor is calculated and ranked on a scale from 0, *maximum limitation*, to 1, *minimum limitation* (Vancoppenolle & Tedesco, 2017). The model has a temporal resolution of 1 hr, for a total of 385 time steps over the 17-day period. For further details about LIM1D, please see Vancoppenolle et al. (2010) and Vancoppenolle and Tedesco (2017).

The model was initialized using observed thermodynamic and biogeochemical profiles from the first day of observations. We used atmospheric temperature, specific humidity, wind speed, and atmospheric pressure recorded over the whole sampling period (mid-November to early December) at Davis to force the model. Atmospheric data were retrieved from the Antarctic Climate Data Collected by Australian Agencies database (Barnes-Keoghan, 2000). Data for daily precipitation were not available, so a fixed rate of 0.01 m/day was used to match the observed snow cover.

The model was first run with default values for biological parameters (based on laboratory and field-based values from the literature; Table 1) to evaluate how closely the model outputs matched the biogeochemical observations.



**Table 1**  
Model Parameters

Parameter	Units	Control value	Sensitivity runs
Maximum specific growth rate <sup>a</sup>	s <sup>-1</sup>	$3.0 \times 10^{-5}$	$3.0 \times 10^{-6}$ , $3.0 \times 10^{-4}$
Algal decay rate <sup>a</sup>	s <sup>-1</sup>	$1.0 \times 10^{-6}$	
Remineralization rate <sup>b</sup>	s <sup>-1</sup>	$2.0 \times 10^{-7}$	$2.0 \times 10^{-8}$ , $2.0 \times 10^{-6}$
Si:C ratio <sup>c</sup>	—	0.11	0.04, 0.95
N:C ratio <sup>c</sup>	—	0.14	
P:C ratio <sup>c</sup>	—	0.0094	
Half-saturation constant for N <sup>c</sup>	μM	1.6	
Half-saturation constant for P <sup>c</sup>	μM	0.24	
Half-saturation constant for Si <sup>c</sup>	μM	3.9	50, 100

<sup>a</sup>Calibrated (Vancoppenolle & Tedesco, 2017). <sup>b</sup>Based on remineralization rate of carbon in sea ice (Moreau, Vancoppenolle, et al., 2015). <sup>c</sup>Based on observed pelagic diatom physiology (Redfield et al., 1963; Sarthou et al., 2005; Sommer, 1986).

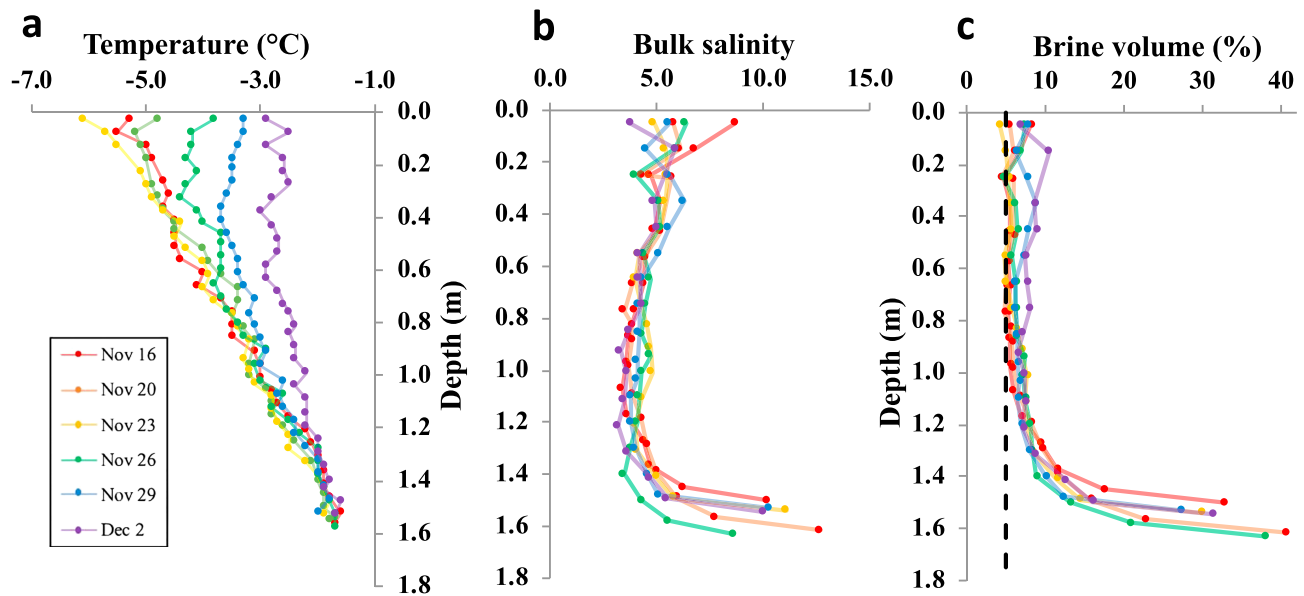
Several sensitivity assessments were then run to examine possible limitations on algal growth in the fast ice at Davis. In particular, parameters controlling DSi supply and uptake were changed according to ranges of values found in past laboratory and modeling assessments (Table 1). The rationale behind these sensitivity analyses is described in section 3.4, but we list here the technical details. The DSi:C incorporation ratio, set in the control run at 0.11, was also tested at 0.04 and 0.95 to simulate different extents of algal DSi uptake. For pelagic diatoms, Brzezinski's (1985) laboratory experiments found DSi:C = 0.12 and 0.13, while Sarthou et al.'s (2005) review found DSi:C = 0.11. Remineralization rates an order of magnitude above and below  $2 \times 10^{-7} \text{ s}^{-1}$  (Moreau, Kaartokallio, et al., 2015) were then tested to estimate the effect of diatom frustule dissolution. By testing  $3.0 \times 10^{-4} \text{ s}^{-1}$ ,  $3.0 \times 10^{-5} \text{ s}^{-1}$  (control), and  $3.0 \times 10^{-6} \text{ s}^{-1}$  for the maximum specific growth rate ( $\mu_{\text{max}}$ ), which controls biogenic silica (bSiO<sub>2</sub>) production, dissolution to production ratios were varied as 0.0007, 0.007, and 0.07, respectively. Though lower than Fripiat et al. (2014)'s bSiO<sub>2</sub> dissolution to production ratio ranging from 0.4 to 0.9, these were taken from a LIM1D calibration (Vancoppenolle & Tedesco, 2017). Finally, since no laboratory-based data on DSi half-saturation constants ( $K_{\text{Si}}$ ) in sea ice currently exist, different  $K_{\text{Si}}$  values were based on a range measured for pelagic diatoms:  $K_{\text{Si}} = 3.9 \text{ μM}$  (Sarthou et al., 2005; Vancoppenolle & Tedesco, 2017) as used in past LIM1D model simulations,  $K_{\text{Si}} = 100 \text{ μM}$  as an upper limit of the 88.7-μM maximum value measured for cultures of Southern Ocean diatoms (Sommer, 1986), and  $K_{\text{Si}} = 50 \text{ μM}$  as a midrange value.

### 3. Results

#### 3.1. Sea-Ice Physics

The site at Davis had a medium snow cover that increased from 0.18 to 0.36 m during the 17-day time series. Sea ice was on average  $1.66 \pm 0.04 \text{ m}$  (mean  $\pm$  standard deviation,  $n = 6$ ) thick. With the maximum ice observed on Nov 20 and Nov 29 (Stations 2 and 5, respectively), spatial heterogeneity may be partly responsible for the variability in ice thicknesses. Temperature in the fast ice generally increased with depth and over time (Figure 2a). The amplitude of change over time was greatest at the top of the ice, where temperature increased from  $-5.3 \text{ °C}$  on the first day to  $-2.9 \text{ °C}$  on the last. Temperature at the base of the ice remained consistently between  $-1.7 \text{ °C}$  and  $-2.0 \text{ °C}$ , close to seawater freezing temperature.

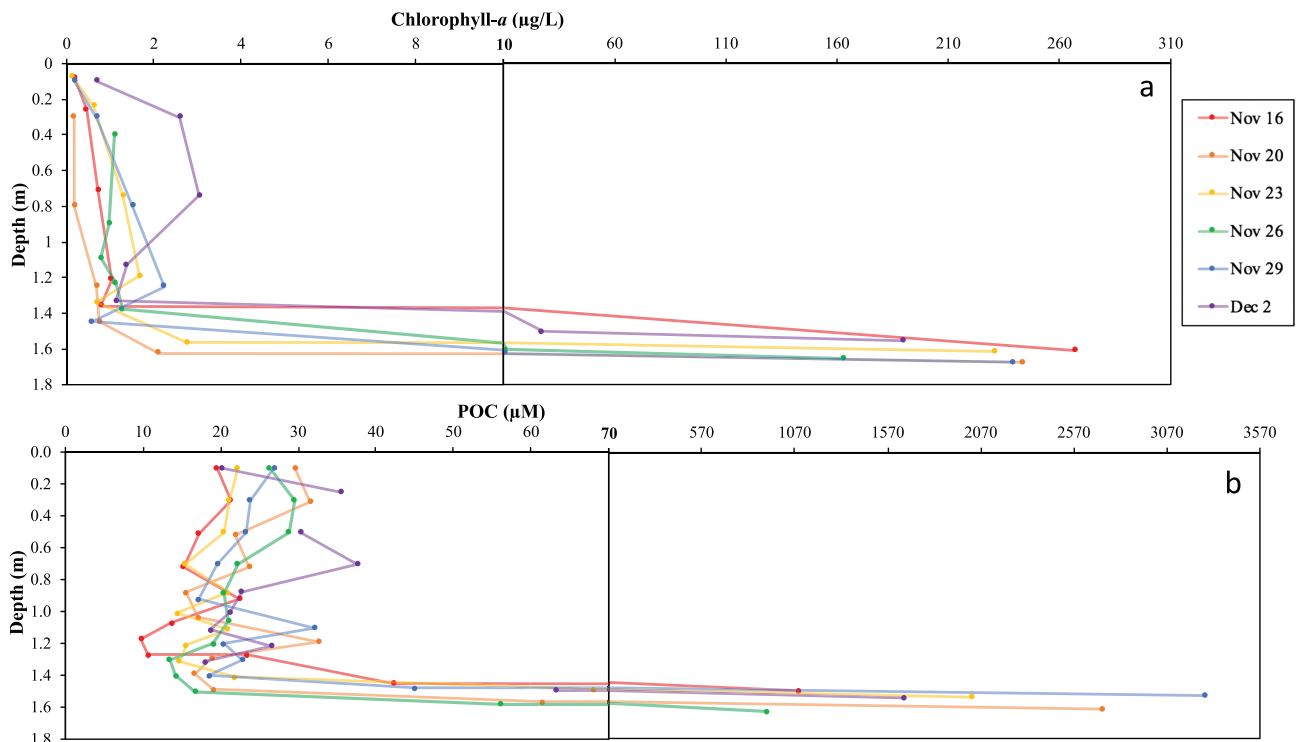
The sea ice had a “C-shaped” salinity profile, with bulk salinity averaging  $5.3 \pm 0.9$  ( $n = 35$ ) in the top of the ice (0–0.6 m),  $4.1 \pm 0.4$  ( $n = 44$ ) in the middle of the ice (0.6–1.3 m), and  $6.6 \pm 2.7$  ( $n = 22$ ) in the lower ice (1.30–1.65 m; Figure 2b). There was no strong temporal trend in sea-ice salinity. Brine volumes were largest at the base of the ice (five to eight times greater than at the middle or top; Figure 2c) where the ice is warmest and most saline. Sea ice with a columnar structure and a brine volume fraction over 5% is considered permeable, as previously isolated brine pockets connect via brine channels (Golden et al., 1998). The observed sea ice was permeable except some portions of the top 0.8 m in the earlier sampling days (mostly Nov 20 and 23, one point on Nov 26). Thus, brine and seawater could move and be exchanged throughout the fast ice for all days at the base and, during the second half of the time series, also at the top.



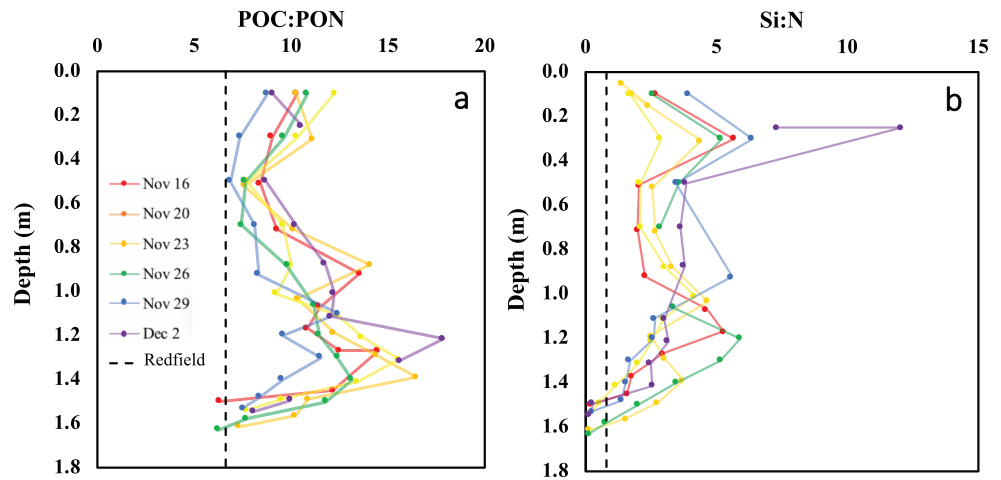
**Figure 2.** (a) Temperature, (b) bulk salinity, and (c) brine volume fraction vertical profiles for each sampling day. Dashed vertical line marks the 5% permeability threshold.

### 3.2. Sea-Ice Biogeochemistry

Chla was used as a proxy for biological activity by photosynthetic sea-ice algae. Although there was a small amount of Chla at depths of 0–1.3 m, over 70% of the total integrated Chla biomass was at the bottom of the ice, with concentrations between 163.5 and 267.4  $\mu\text{g/L}$  (Figure 3a). Microscopy qualitatively identified the pennate diatoms in the bottom ice as *Nitzschia stellata*, *Nitzschia lecoinei*, *Fragilariopsis cylindrus*, *Entomoneis kjellmanii*, and *Pleurosigma* sp., as well as some unidentified, chain-forming, centric diatoms



**Figure 3.** (a) Chla and (b) POC concentration vertical profiles for each sampling day.



**Figure 4.** (a) POC:PON and (b) DSi:N ( $\text{NO}_3^- + \text{NO}_2^-$ ) vertical profiles for each sampling day.

(Supporting Information, Figure S1, identified by Phillip Assmy; additional observations by Andreas Seger and Fraser Kennedy). There were no algal strands attached to the base of the ice. Chl $a$  concentrations in the seawater were up to three orders of magnitude smaller, averaging  $0.17 \pm 0.11 \mu\text{g/L}$  ( $n = 6$ ) right below the ice and  $0.12 \pm 0.07 \mu\text{g/L}$  ( $n = 6$ ) 10 m below. There was no clear temporal trend in the bottom sea-ice Chl $a$  concentrations. Evidence suggest a small increase in intermediate and upper ice Chl $a$  over time, from levels at or below  $1.03 \mu\text{g/L}$  on the first day to levels up to  $3.04 \mu\text{g/L}$  on the last day (Figure 3a). POC ranged between  $9.7 \mu\text{M}$  in internal sea ice and  $3,270 \mu\text{M}$  at the bottom of the ice (Figure 3b). Seawater POC was up to three orders of magnitude smaller, ranging from  $2.6$  to  $20.5 \mu\text{M}$ . There was no temporal trend in POC in the intermediate and upper ice nor the bottom ice (Figure 3b). The intermediate and upper ice had slightly higher POC:PON ratios ( $10.9 \pm 2.4$ ,  $n = 60$ ) than basal ice ( $7.1 \pm 0.8$ ,  $n = 6$ ; Figure 4a), which exhibited near-Redfield ratios ( $6.6$ , Redfield et al., 1963).

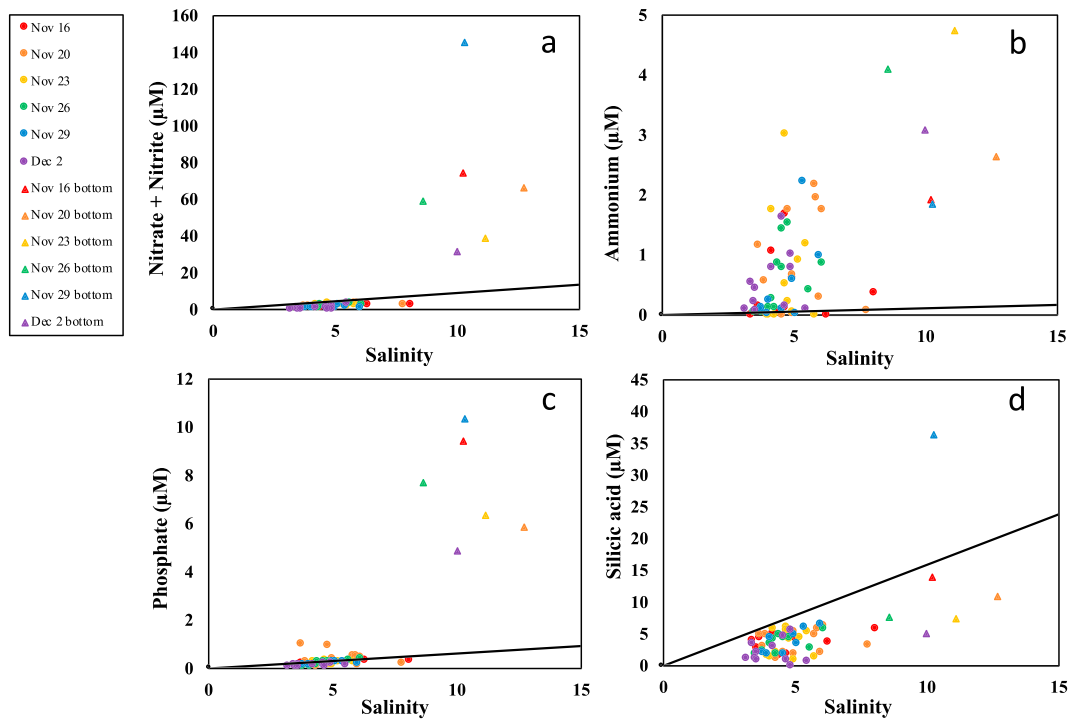
On every sampling day, macronutrients were much more concentrated in the bottom ice ( $\text{NO}_3^- + \text{NO}_2^- = 69.26 \pm 40.69$ ,  $\text{NH}_4^+ = 3.05 \pm 1.17$ ,  $\text{PO}_4^{3-} = 7.41 \pm 2.14$ ,  $\text{DSi} = 13.68 \pm 11.6 \mu\text{M}$ ,  $n = 6$ ) than in other ice layers ( $\text{NO}_3^- + \text{NO}_2^- = 1.38 \pm 0.86$ ,  $\text{NH}_4^+ = 0.76 \pm 0.75$ ,  $\text{PO}_4^{3-} = 0.23 \pm 0.19$ ,  $\text{DSi} = 4.11 \pm 1.58 \mu\text{M}$ ,  $n = 47$ ). For all macronutrients except DSi, the bottom ice was also enriched compared to seawater 0, 3, and 10 m under the ice ( $\text{NO}_3^- + \text{NO}_2^- = 29.28 \pm 0.69$ ,  $\text{NH}_4^+ = 0.29 \pm 0.23$ ,  $\text{PO}_4^{3-} = 2.10 \pm 0.22$ ,  $\text{DSi} = 52.40 \pm 2.74 \mu\text{M}$ ,  $n = 18$ ). Concentrations in the seawater, intermediate ice, and upper ice sometimes fell below the detection limits of the analytical instrument. Although there was some variability in measurements over the 17 days, overall, there was no discernable trend in macronutrient concentrations.

Bulk inorganic macronutrient concentrations are plotted as a function of bulk ice salinity (Figure 5). Nitrate and  $\text{NO}_2^-$  concentrations fell below the TDL, except for the bottom layer in which they were consistently well above the TDL (Figure 5a). Ammonium was often well above the TDL (Figure 5b), with sea-ice concentrations sometimes an order of magnitude larger than that of the seawater. In a pattern similar to  $\text{NO}_3^- + \text{NO}_2^-$ ,  $\text{PO}_4^{3-}$  followed or was slightly below the TDL, except for all the bottom sections and a few other depths from Nov 20 (Figure 5c). Other than one point (bottom layer on Nov 29), DSi was always below the TDL (Figure 5d). At times, the difference between DSi and the TDL was over  $5 \mu\text{M}$ , a fairly large amount given typical bulk sea-ice DSi concentrations of  $0$ – $10 \mu\text{M}$  (Fripiat et al., 2017). Compared to Brzezinski (1985) DSi:N ratio of  $0.95$ – $1.12$  for marine diatoms, there was relatively more DSi than N in the intermediate and upper ice ( $3.2 \pm 1.8$ ,  $n = 60$ ) and less in the basal ice ( $0.2 \pm 0.04$ ,  $n = 6$ ; Figure 4b).

### 3.3. Control Simulation Output

In the control simulation using previously calibrated, literature-based parameters (Table 1), the model agreed closely with the ice thickness and snow depths (blue plus signs, Figure 6a). There was  $0.022$  m of modeled ice loss over 17 days. Given the spatial heterogeneity between cores, this is a decent estimation of the observed ice loss of  $\sim 0.04$  m. The model calculated brine volume fraction, with permeability above





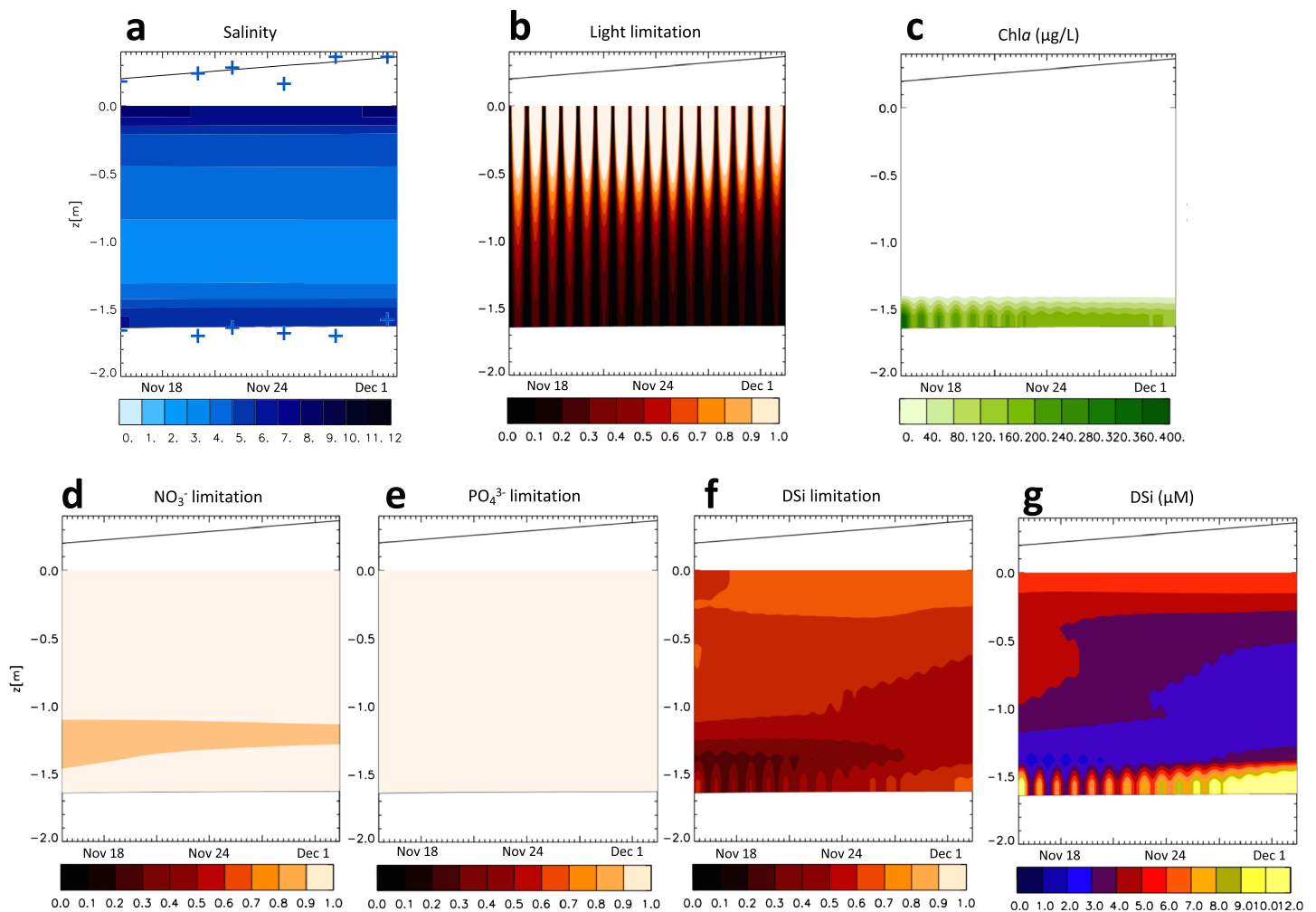
**Figure 5.** Macronutrient bulk concentrations (a)  $\text{NO}_3^- + \text{NO}_2^-$ , (b)  $\text{NH}_4^+$ , (c)  $\text{PO}_4^{3-}$ , and (d) DSi as a function of salinity for each sampling day. Theoretical dilution lines (solid black lines) were established from nutrient concentrations in seawater 10 m below the ice, averaged for all six time points.

the 5% theoretical threshold throughout the whole time series (Figure 7c). Light limited primary production on a daily cycle (Figure 6b), and brine salinity was limiting in the upper ice cover throughout the whole time series (not shown). Chla was enriched in the bottom sea ice, with values above  $100 \mu\text{g/L}$ , and showed some regulation by the light-dark cycle (Figure 6c). Nitrate,  $\text{NO}_2^-$ , and  $\text{PO}_4^{3-}$  never reached limiting concentrations (Figures 6d and 6e). Silicic acid was limiting (Figure 6f), with a decrease in concentration in the internal ice but some increase in the bottom ice over time (Figure 6g).

Because there was generally little temporal evolution in macronutrient concentrations, depth profiles, both for observations and the model outputs, are depicted as an average over the entire sampling period (Figure 7). Initialized with values from the first day of observations, the model simulated the sea-ice temperature (Figure 7a), salinity (Figure 7b), and brine volume (Figure 7c) at different depths relatively well, except for an underestimation of salinity in the bottom ice (Figure 7b). Chla was greatly overestimated in the bottom ice (Figure 7d). For biogeochemical parameters, the control simulation showed a slight overestimation of POC at 0.7–0.9 relative depth, but otherwise mostly matched observed profiles (Figure 7e). It reproduced  $\text{NO}_3^- + \text{NO}_2^-$  (Figure 7f) and  $\text{PO}_4^{3-}$  (Figure 7g) well. However, the model largely underestimated DSi, with a depth-integrated value of  $2.61 \pm 1.89 \text{ mmol/m}^2$  ( $n = 385$ ) compared to the observed  $5.88 \pm 1.24 \text{ mmol/m}^2$  ( $n = 6$ ; Figure 7h). Both in the observations and control simulation, DSi in the bottom ice was depleted relative to seawater (Figure 7h).

### 3.4. Sensitivity Analysis of the Model to DSi Limitation

Given the control simulation's DSi limitation and difficulty matching Chla and DSi observations, these characteristics were chosen as foci for model tuning and to elucidate the biogeochemical processes in the fast ice system. In all sensitivity runs, the model's simulations of physical characteristics,  $\text{NO}_3^- + \text{NO}_2^-$ , and  $\text{PO}_4^{3-}$  were not affected. Compared to the control run, decreasing the DSi:C ratio to 0.04 overestimated Chla (red line, Figure 8a) but caused a more accurate fit to DSi observations in the upper part of the ice (red line, Figure 8b). Increasing DSi:C to 0.95 led to a relatively accurate Chla simulation in the upper ice (blue line, Figure 8a) but a slight overestimation of DSi concentrations in the bottom ice (blue line, Figure 8b). The



**Figure 6.** Contoured (a) salinity, (b) light limitation, (c) Chla bulk concentration, (d) NO<sub>3</sub><sup>-</sup>/NO<sub>2</sub><sup>-</sup> limitation, (e) PO<sub>4</sub><sup>3-</sup> limitation, (f) DSi limitation, and (g) DSi bulk concentration for a control run. Limitations are without units on a scale from 0 (*maximum limitation*) to 1 (*no limitation*). Vertical axes are depth. Observations of snow depth and ice thickness are also plotted (plus signs) on (a). Black line on top of each panel represents modeled snow thickness.

intermediate DSi:C value of 0.11 produced the closest fit to both Chla and DSi observations (green line, Figure 8), although all three runs overestimated Chla in the bottom of the sea ice.

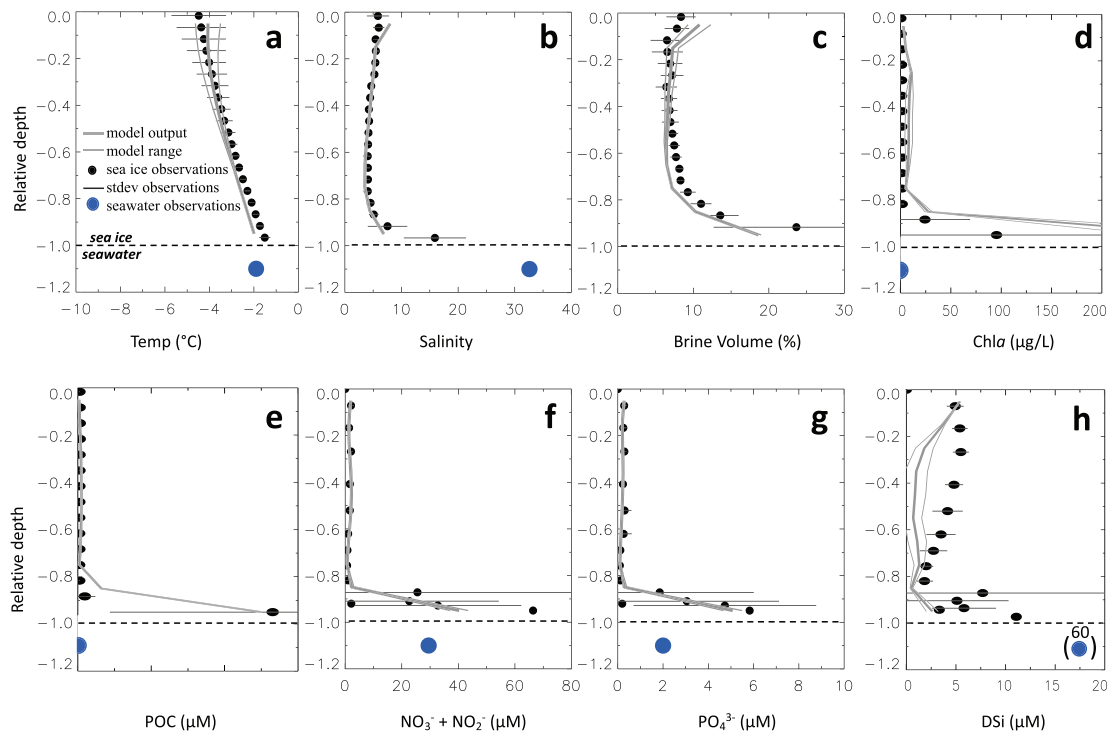
Decreasing the remineralization rate 10 times from the control run ( $2 \times 10^{-8} \text{ s}^{-1}$ ) led to a slight underestimation of DSi observations in the basal ice (red line, Figure 9b), while increasing the remineralization rate ( $2 \times 10^{-6} \text{ s}^{-1}$ ) led to a large overestimation of DSi and Chla concentrations at the bottom of the ice (blue line, Figure 9). In the upper ice, decreasing  $\mu_{\text{max}}$  by an order of magnitude ( $3 \times 10^{-6} \text{ s}^{-1}$ ) better matched DSi observations (red line, Figure 10b) compared to the control run, but otherwise changing  $\mu_{\text{max}}$  did not particularly improve model output of Chla and DSi (Figure 10).

When the lowest  $K_{\text{Si}}$  value was used, the model output overestimated Chla and underestimated DSi (green line, Figure 11). When the highest  $K_{\text{Si}}$  value was used, DSi output was too high in the lower ice, while Chla matched DSi observations fairly well (blue line, Figure 11). An intermediate  $K_{\text{Si}}$  value of 50 μM produced the closest combined fit to both Chla and DSi observations (red line, Figure 11).

## 4. Discussion

### 4.1. Status of the Biogeochemical Reservoirs

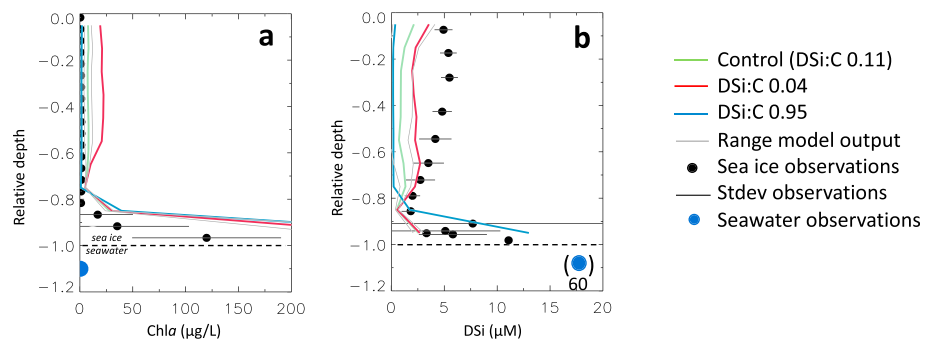
Sea ice at Davis displayed physical characteristics that are typical of Antarctic fast ice during spring, such as warming temperatures at the surface of the ice and a “C”-shaped salinity profile (Petrich & Eicken, 2010).



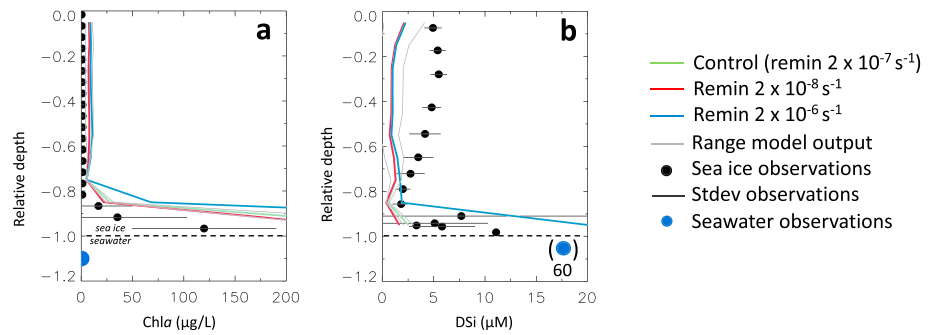
**Figure 7.** Mean LIM1D outputs for a control run. Observations and model outputs for (a) temperature, (b) salinity, (c) brine volume, (d) Chl<sub>a</sub>, (e) POC, (f) NO<sub>3</sub><sup>-</sup> + NO<sub>2</sub><sup>-</sup>, (g) PO<sub>4</sub><sup>3-</sup>, and (h) DSi. All concentrations are bulk ice concentrations. Vertical axes are normalized relative depths, with a maximum depth of one.

The model closely matched the observed ice physics, although it should be noted that the time series is relatively short and has little evolution. Because the model was initialized with the first day of observations, this likely contributes to the agreement between the observed and modeled quantities over the 17-day simulation period.

Since there are little to no temporal trends in the Chl<sub>a</sub>, POC, and macronutrients measurements throughout the time series, our sampling appears to mark a period without major biomass accumulation or decline. Previous work in the area suggests that sea-ice algal biomass near Davis generally peaks in October (Swadling et al., 2004). Compared to other Antarctic fast ice studies in November-December, basal Chl<sub>a</sub> concentrations in this study (163.5–267.4 μg/L) are one order of magnitude higher than maximum concentrations observed at Terra Nova Bay (34.2 μg/L, Grotti et al., 2005) and at Davis in 1993 (69 μg/L, Archer et al., 1996), but similar to those at Casey Station (235 μg/L, van der Merwe et al., 2011). As this study falls



**Figure 8.** LIM1D sensitivity tests for different DSi:C ratios, comparing mean observations to mean model outputs for (a) Chl<sub>a</sub> and (b) DSi concentrations. Vertical axes are relative depths, with a maximum depth of one.

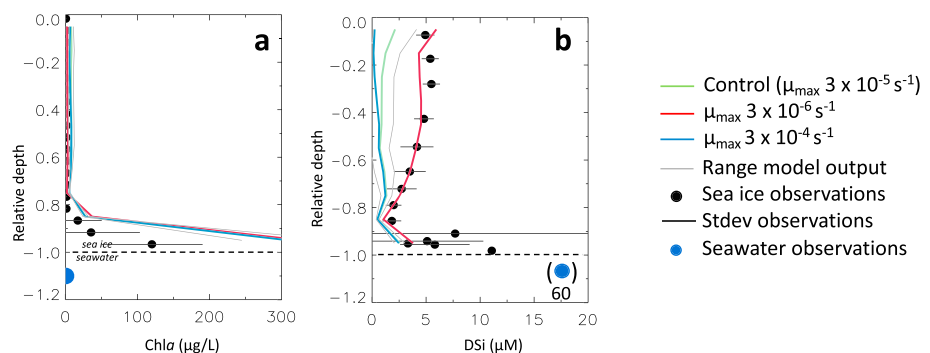


**Figure 9.** LIM1D sensitivity tests for different remineralization rates, comparing mean observations to mean model outputs for (a) Chl $a$  and (b) DSI concentrations. Vertical axes are relative depths, with a maximum depth of one.

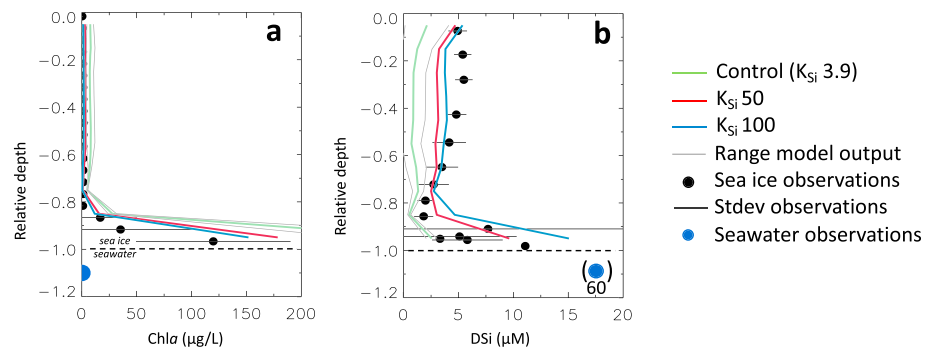
on the high end of observed values in Antarctic fast ice (excluding sites with platelet ice, e.g., Meiners et al., 2018), the algal population may be unlikely to accumulate further.

Two factors that challenge our ability to observe changes in the algal population, particularly in the bottom ice, are gradual ice melt as the spring warms into summer, and the possible loss of material via brine drainage from the bottom of the cores during sampling. Either of these could lead to an underestimation of Chl $a$ . Over the 17-day period, we estimate that ice thickness loss due to melting was 0.02–0.04 m (Figure 6a). Given the dense accumulation of algae in the bottom ice, it is possible that this contributes to the apparent stability of Chl $a$ , and slight ice loss should be retained as a limitation of this study. This is perhaps reflected in a minor decrease in Chl $a$  concentrations in the model simulation (Figure 6c). With regard to sampling, along with the precautions taken to retain as much of the cores as possible, negative air temperatures ( $-2.9 \pm 1.5$  °C) kept cores below their melting point. In situ sea-ice temperatures were relatively cold (Figure 2a), helping to keep the biomass attached to the ice matrix. What is lost is generally associated with the dissolved phase, while particulate matter (and therefore Chl $a$ ) tends to remain attached to the walls of the brine channels (Becquevort et al., 2009; Lannuzel et al., 2016). Overall, within these considerations, we maintain the integrity of the observations as generally representative of the fast ice system and dynamics.

Our observations of co-occurring high macronutrients ( $\text{NO}_3^- + \text{NO}_2^-$ ,  $\text{NH}_4^+$ ,  $\text{PO}_4^{3-}$ , not DSI; Figure 5) and chlorophyll reveal a relatively stable system. High nutrient concentrations are able to support a large sea-ice algae community, while other processes simultaneously replace these nutrients. Note that the nutrient measurements were for melted ice sections (i.e., bulk concentration), which include liquid brine and solid ice. As macronutrients are in reality dissolved within brine inclusions, the nutrient concentrations that algae experience in situ (i.e., brine-normalized) are therefore higher than the reported bulk concentrations. Unless sea-ice algae rely primarily on intracellular nutrient pools, there must exist processes that replenish these concentrated sea-ice nutrient pools. Physical supply processes, such as natural brine convection due to a cooler ice surface (Reeburgh, 1984) or forced brine convection by tides (Cota et al., 1987; Cota & Horne,



**Figure 10.** LIM1D sensitivity tests for different maximum growth rates, comparing mean observations to mean model outputs for (a) Chl $a$  and (b) DSI concentrations. Vertical axes are relative depths, with a maximum depth of one.



**Figure 11.** LIM1D sensitivity tests for different  $K_{Si}$  values, comparing mean observations to mean model outputs for (a) Chla and (b) DSi concentrations. Vertical axes are relative depths, with a maximum depth of one.

1989; Eicken, 1992), would contribute but not be able to supply nutrients above seawater levels ( $NO_3^- + NO_2^-$ ,  $NH_4^+$ , and  $PO_4^{3-}$ ; Figures 5 and 7). Similarly, surface flooding, which occurs when the snow-ice interface sinks below sea level, is virtually absent at Davis (snow depth is typically below 0.05 m) and fails to explain observed nutrient concentrations. This suggests that a biogeochemical supply process is active for the nutrients that are above the TDL. A limitation of this study is the lack of baseline seawater data for establishing the TDLs: Ideally, they would be calculated from seawater in winter, when light and temperature minimize biological activity. Since 10-m seawater concentrations of  $NO_3^- + NO_2^- = 29.49 \pm 0.81$ ,  $PO_4^{3-} = 2.03 \pm 0.05$ , and  $DSi = 51.86 \pm 3.60 \mu M$  ( $n = 6$ ) were close to median late winter seawater values in the East Antarctic sector of 29, 2, and  $60 \mu M$ , respectively (Becquevort et al., 2009), they were considered sufficient approximations for winter conditions.

In situ recycling processes within the sea-ice community could be partly responsible for replenishing nutrient supplies. Ammonium consistently above the TDL (Figure 5) suggests remineralization by bacteria (Arrigo et al., 1995; Grossmann et al., 1996) that is greater than consumption (assimilation + nitrification) at all depths. In Fripiat et al.'s (2017) compilation of Antarctic pack ice cores,  $NH_4^+$  was found to similarly accumulate—often at higher concentrations—at all depths during the spring. Nitrate,  $NO_2^-$ , and  $PO_4^{3-}$  above the TDL in the bottom layer similarly point to remineralization, perhaps with organic matter and brine channel walls as favorable microenvironments for bacteria (Meiners et al., 2004). A well-established bacterial community is able to break down detritus, making nutrients bioavailable for new growth. We suggest that the resulting  $NO_3^- + NO_2^-$  and  $PO_4^{3-}$  are additionally retained by adsorption in these microenvironments (Becquevort et al., 2009; Meiners & Michel, 2017), preventing removal by physical exchanges. Although there is little evidence of adsorption for nitrogen in the form of  $NO_3^- + NO_2^-$ ,  $NH_4^+$  is known to adsorb onto organic matter, especially the high pool of exopolymeric particles in sea ice (Becquevort et al., 2009). Oxidation then converts  $NH_4^+$  to  $NO_2^-$  and  $NO_3^-$ ; nitrification has been shown to be significant in sea ice, especially in microbe-rich biofilms (Baer et al., 2015; Firth et al., 2016; Fripiat et al., 2014; Fripiat et al., 2015; Priscu et al., 1990). Though commonly overlooked, nutrient adsorption after bacterial remineralization is a plausible explanation for  $NO_3^- + NO_2^-$  and  $PO_4^{3-}$  accumulation in the bottom of the sea ice.

Could brine-dwelling, heterotrophic protists and metazoans be driving the stable conditions? Their consumption of sea-ice algae could prevent increases in Chla, then playing a similar role to bacteria in returning inorganic nutrients to the sea ice. In a summer study at Davis, heterotrophic grazing was estimated to have consumed up to 25% of primary production (Archer et al., 1996). Although no quantitative community-composition measurements were made in our spring study, the ratio of POC:Chla was ~49–313 in the bottom ice and  $4.06 \pm 3.07$  ( $n = 6$ ) in the 10-m seawater sample. The ratio in sea ice is approximate, because POC and Chla were measured in separate cores at slightly different depths. For comparison, in late winter-early spring, East Antarctic pack ice had POC:Chla ~219–248 and a median heterotroph to autotroph ratio of 0.03 (range 0.01–1.15) with <1% protozoa; seawater with POC:Chla ~1,000 had a 8.5 heterotrophs to autotrophs ratio and 11% protozoa (Becquevort et al., 2009). Thus, while the POC:Chla in this study does imply a sizeable amount of non-chlorophyll containing matter, it does not necessarily indicate a large grazer population. Detritus and bacteria also contribute to POC and are necessary for the remineralization



processes proposed earlier. Overall, van Leeuwe et al.'s (2018) compilation of cores found that heterotrophic grazers likely have only a small presence in sea ice. On average, heterotrophs have a 10% abundance ( $SEM = 3$ , range = 0–91) in Antarctic fast ice and are most dominant in early spring, when light availability and phototrophs are low (van Leeuwe et al., 2018). The lack of observational data prevents us from including them in the model. While grazing is outside of the scope of this study, food web interactions should be recognized as a phenomenon poorly constrained in sea ice. Even so, with the given POC:Chl $a$  data and knowledge of other fast ice systems, nutrient availability remains the most plausible control of sea-ice algal biomass.

The model and observations both suggest that  $NO_3^- + NO_2^-$  and  $PO_4^{3-}$  concentrations are sufficient to sustain the high biomass observed in the bottom sea ice. While the model is limited in that its nutrient uptake follows the Redfield ratio, this assumption appears valid for nitrogen and phosphorus observed in the bottom ice (Figure 4). However, the DSi:N ratios in the bottom of the sea ice, where the majority of biomass was concentrated, were extremely low (Figure 4b). Observed DSi concentrations in the bottom ice were consistently below the TDL, and the model output also indicates DSi limitation (Figure 6f). Since nitrogen species and phosphate are above the TDL and seawater levels, we argue that DSi supply is the ultimate control of the system's evolution. This is consistent with a 3-D modeling study, in which Si limitation became more important relative to light and N limitation in the late Antarctic spring and summer (Saenz & Arrigo, 2014). Furthermore, we found no evidence in the literature that DSi adsorbs onto organic matter in the same manner as  $NH_4^+$  and  $PO_4^{3-}$ . Limited physical exchange with seawater constrains DSi in a way that remineralization cannot compensate for. Hence, DSi supply and demand are critical aspects to examine in order to understand why model simulations tend to overestimate Chl $a$  and underestimate DSi as compared with observations (Figure 7).

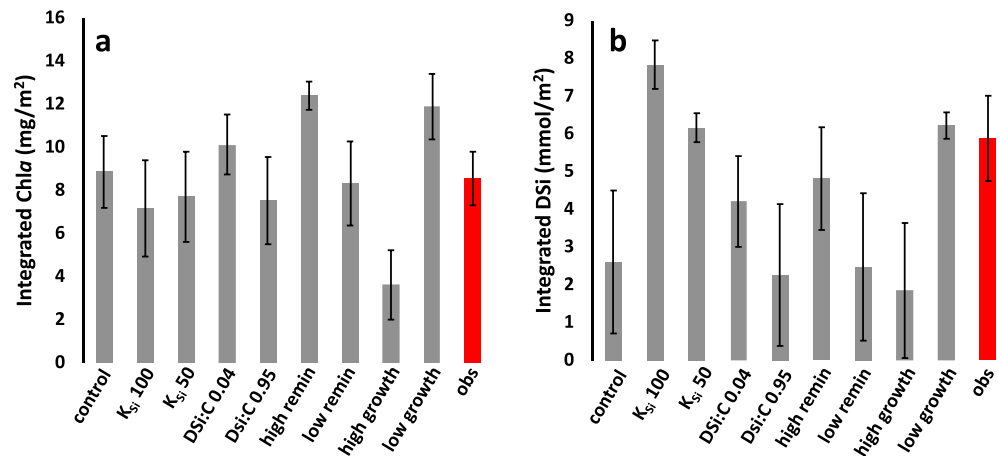
#### 4.2. DSi Supply and Affinity

Diatoms are the main users of DSi, converting it to  $bSiO_2$  to build their cell wall frustules. Our observations that diatoms are the major primary producers in the basal fast ice are consistent with past studies at Davis, which primarily found *Nitzschia frigida* (McConville & Wetherbee, 1983), *E. kjellmanii*, *N. stellata*, *Thalassiosira australis*, *Berkeleya adeliense*, *Navicula glacei* (Gibson et al., 1999), and other *Entomoneis* spp. (Archer, Leakey, Burkill, & Sleigh, 1996). The dominance of pennate diatoms in the bottom ice has also been characterized in a compilation of over 300 sea-ice cores from the Arctic and Antarctic (van Leeuwe et al., 2018).

We propose that DSi is the most limiting nutrient for algal growth in the bottom ice. It remains unresolved why other algal groups or smaller, less-silicified diatoms are unable to take advantage of this, but bottom ice communities—in this study and in others—are consistently dominated by large diatoms (van Leeuwe et al., 2018). We also hypothesize that sea-ice diatoms require more DSi than pelagic diatoms. To come to this, we acted on three specific model features that may explain the discrepancies in observations of high Chl $a$  and low DSi in the bottom sea ice: (a) the silicon quota in organic matter (DSi:C ratio), (b) the dissolution and production rates of  $bSiO_2$ , and (c) the affinity of sea-ice diatoms for DSi.

The DSi:C ratio = 0.11 used in the model simulations is based on marine diatoms (Sarhou et al., 2005) and therefore not necessarily suited to the sea-ice environment. Sea-ice diatoms are frequently larger than pelagic diatoms (Arrigo et al., 2010), altering the surface area to volume ratio, while lower temperatures tend to increase silicification (Martin-Jezequel et al., 2000). However, changing DSi:C does not alter Chl $a$  and POC biomass because it affects  $bSiO_2$  dissolution and DSi uptake in a similar way (Figures 8a and 8b). The increase in DSi reflects the higher amount of DSi implicitly stored in organic matter when using a higher DSi:C. It should be noted that on the last day, DSi:C = 0.95 overpredicts DSi, which may increase with time. In the end, there is no reason to change our DSi:C quota in sea ice as compared to pelagic diatoms.

In sea ice, the dissolution of  $bSiO_2$  is likely the major biological source of DSi (Fripiat, Tison, et al., 2014). An accumulated bacterial population (Deming, 2010) degrades the protective organic matrix around the frustules, exposing them to the surrounding undersaturated brine environment (Bidle & Azam, 1999). Even so,  $bSiO_2$  dissolution is a slow process, particularly under cold conditions, and is made possible by substantially higher residence times for detritus in the brine network compared to in the ocean (Fripiat et al., 2017). In Antarctic pack ice, estimates of the dissolution to production ratio range from 0.4 to 0.9 (Fripiat, Tison, et al., 2014). We tested ratios between 0.007 and 0.07 by changing  $\mu_{max}$  (production) relative to the



**Figure 12.** Depth-integrated observations (red,  $n = 6$ ) and model outputs (gray,  $n = 385$ ) averaged over the course of the time series for (a) DSi and (b) Chla concentrations. Error bars indicate standard deviation.

remineralization (dissolution) rate. With respect to the bottom ice, results show that neither the process of frustule dissolution nor production can fully explain the observed DSi and Chla profiles during this time series (Figures 9 and 10). However, in the upper sea ice, a lower  $\mu_{max}$  better simulated DSi concentration (Figure 10b), consistent with the fact that these communities are mainly flagellates and *Phaeocystis* spp. (van Leeuwe et al., 2018) that do not uptake DSi. At Davis in particular, the interior ice community tends to be dominated by *Gymnodinium* sp., a small dinoflagellate, and autotrophs  $<20 \mu\text{m}$  long (Archer, Leakey, Burkill, & Sleight, 1996). Since the model is based on pelagic diatoms, a slower growth rate in the upper ice would better reproduce the lack of DSi use, even though it does not correctly represent the growth of non-silicified algae. As flagellates make up  $\sim 76\%$ ,  $65\%$ , and  $31\%$  of surface, interior, and bottom ice communities, respectively, in Antarctic fast ice (van Leeuwe et al., 2018), future studies should account for different types of phytoplankton assemblages across the ice cover, for example, following the approaches of Tedesco et al. (2012) and Aumont et al. (2015).

Although there are no current studies that quantify sea-ice diatoms' DSi requirement for growth, the model predicts that sea-ice diatoms may require one order of magnitude more DSi than observed for cultured pelagic diatoms (i.e., average  $K_{Si} = 3.9 \mu\text{M}$ ; Sarthou et al., 2005). Comparing all of the model parameterizations,  $K_{Si} = 50 \mu\text{M}$  is best able to match depth-integrated observations of both DSi and Chla, while most sensitivity runs only reproduce the Chla stocks well (e.g., the control, low remineralization, and DSi:C = 0.95 runs, Figure 12). The kinetics of DSi uptake may be related to the type and expression of DSi transporter in the cells, which is not only species dependent but may also change during the course of cell wall synthesis (Martin-Jezequel et al., 2000). In general, the kinetics of diatoms typical at Davis, including the genera *Entomoneis* and *Nitzschia* (Archer, Leakey, Burkill, & Sleight, 1996; Gibson et al., 1999; McConville & Wetherbee, 1983), are poorly, if at all, characterized (Martin-Jezequel et al., 2000). Importantly, cultures of diatoms collected from Antarctic waters reported  $K_{Si}$  one to two orders of magnitude higher than diatoms from temperate or tropical oceans (Jacques, 1983; Sommer, 1986; Sommer, 1991), though field-based experiments question those high  $K_{Si}$  values (Nelson & Treguer, 1992). Some Antarctic sea ice models used  $K_{Si} = 60 \mu\text{M}$  (Arrigo & Sullivan, 1994; Saenz & Arrigo, 2012, 2014), based on unpublished sea-ice algae data from C. W. Sullivan (referenced in Arrigo et al., 1993). In the end, Arrigo et al. (1993) concluded that light availability was more important than DSi as a control on the fast ice system at McMurdo Sound in the Ross Sea. However, at sites like Davis or in 3-D models like in Saenz and Arrigo (2014), a much lower affinity for DSi would cause sea-ice diatoms to experience limitation at relatively high ambient DSi concentrations, including the brine-normalized concentrations observed in this study. Based on our model simulation with  $K_{Si} = 50 \mu\text{M}$ , this hypothesis may well be true in sea ice and calls for a focused investigation on DSi uptake kinetics for cultured and natural assemblages of sea-ice diatoms. Meanwhile, this information might be used in following attempts to model the large-scale impact of sea-ice primary productivity on the global carbon cycle, in the likes of Castellani et al. (2017).

## 5. Conclusion

This study presents and discusses a 17-day time series of biogeochemical sea-ice observations from Davis, East Antarctica, together with one-dimensional sea-ice modeling, to examine fast ice processes in late austral spring. Ice temperature, salinity, and brine volume showed depth-wise and temporal patterns characteristic of first-year sea ice. With initialization using the first day of observations and little temporal evolution, model outputs of ice physics matched observations fairly well. The co-occurrence of high nutrients and high chlorophyll indicates sufficient sources of most nutrients. Sea ice was replenished with  $\text{NO}_3^- + \text{NO}_2^-$ ,  $\text{PO}_4^{3-}$ , and  $\text{NH}_4^+$ , but our observations and model simulations suggest that DSI was limiting. Bacterial remineralization and adsorption within sea ice may constitute the main source of  $\text{NO}_3^- + \text{NO}_2^-$ ,  $\text{PO}_4^{3-}$ , and  $\text{NH}_4^+$ . Comparisons between model simulations and observations suggest that models likely need to account for the distinct compositions of algal communities at different depths in the ice. We further propose that some sea-ice diatoms may have a much lower affinity for DSI than pelagic diatoms. Future research should include quantification of DSI cellular requirements by sea-ice algae, grazing, the role of biofilms in retaining organic matter and nutrients, and modeling ocean-ice biogeochemical interactions.

## Conflict of Interest

The authors declare they have no competing interest of any kind with this research and the publication of this manuscript.

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