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Clarissa Detomi de Albuquerque, Sébastien Curet, Lionel Boillereaux. A 3D-CFD-heat-transferbased model for the microbial inactivation of pasteurized food products. Innovative Food Science & Emerging Technologies / Innovative Food Science and Emerging Technologies , 2019, 10.1016/j.ifset.2019.04.007 . hal-02332197

HAL Id: hal-02332197 https://hal.science/hal-02332197

Submitted on 22 Oct 2021

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Version of Record: https://www.sciencedirect.com/science/article/pii/S1466856418303837 Manuscript_746e2383ff96e06b7a8ec03f5317ad7c

A 3D-CFD-heat-transfer-based model for the microbial inactivation of pasteurized food products

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9 ABSTRACT

This study proposes to couple a 3D-CFD and heat transfer finite elements model with the 10 11 microbial inactivation approach proposed by Geeraerd et al. (2000). The CFD-heat transfer 12 model was developed using thermophysical properties for both heating fluid (water) and the processed sample (ground beef). The kinetic microbial parameters were estimated using 13 experimental data from the inactivation of Escherichia coli K12 in a packaged sample. The 14 proposed inactivation model was tested under more severe dynamic conditions than usual 15 (heating rates from 1 to 13 °C/min). The inactivation kinetic parameters were found 16 independent of the heating rate applied. In addition, the results reveal that the Geeraerd et al. 17 (2000) model without shoulder is sufficient to fit the experimental data. Such a model could 18 be beneficial in simulating microbial inactivation for food products, thus ensuring food safety 19 by limiting, as far as possible, overtreatment. 20

21 Keywords: Microbial inactivation; Modeling; 3D-CFD; Heat transfer; Pasteurization.

Nomenclature

a, b	dimensions of the elliptical sample (mm)
C _c	critical component related to the physiological state of the cells (-)
CP	apparent specific heat (kJ kg ⁻¹ K ⁻¹)
$D_{ m ref}$	decimal reduction time (s)
g	gravitational constant (m/s ²)
h _{air}	heat transfer coefficient of air (W m ⁻² K ⁻¹)
k _{max}	specific inactivation rate (s ⁻¹)

L	sample length (mm)
LL	maximum likelihood
n	number of experimental data points
Ν	microbial population (CFU /g)
N_0	initial microbial population (CFU /g)
\mathbf{N}_{simu}	simulated microbial population (CFU/g)
Nexp	experimental microbial population (CFU /g)
p	number of independently adjusted parameters within the model
Р	absolute pressure (Pa)
Q	volumetric heat generation term (W/m ³)
R	radius (mm)
Т	temperature (°C)
T ₀	initial temperature of product (°C)
T_{∞}	ambient temperature (°C)
t _i	simulated time (s)
T _{ref}	microbial inactivation reference temperature (°C)
T _{cylinder cell}	temperature of lateral surfaces of the cylinder (°C)
\vec{u}	velocity field (m/s)
u, v, w	spatial components of velocity field
V	volume of the meat sample (m ³)
x, y, z	spatial coordinates in the three dimensions (m)
Z	thermal resistance constant (°C)
Cuesly lattons	

Greek letters

ρ	density (kg/m ³)
μ	dynamic viscosity (Pa s)
σ	standard deviation of experimental data
λ	thermal conductivity (W m ⁻¹ K ⁻¹)

22 **1. Introduction**

Prediction of microbial inactivation during thermal treatment is crucial not only to ensure the safety of food products but also to avoid overtreatment (Boillereaux, Curet, Hamoud-Agha, & Simonin, 2013). Although conventional processing technologies produce safe products, they can also lead to significant changes to the sensory and nutritional attributes of foods. In today's quality conscious world, much attention is given to producing foods that retain superior sensorial quality but must also remain safe (Bott, 2014; Stratakos & Koidis, 2015). Consequently, the determination of the microbial destruction level that a thermal treatment can deliver to a product requires both understanding of: (*i*) the heat transfer within the food product and (*ii*) the destruction kinetics of the microorganism of interest (Valdramidis et al., 2005). The necessity of such an approach is reinforced regarding the thermal heterogeneities that occur during a pasteurization process.

A reliable inactivation parameter estimation is essential for building predictive models (Chen, 34 2013) and is necessary to establish safe minimal cooking-cooling conditions (Marcotte, Chen, 35 Grabowski, Ramaswamy, & Piette, 2008). The determination of these parameters has been the 36 subject of numerous studies, either under static (Ahmed, Conner, & Huffman, 1995; Smith, 37 Maurer, Orta-Ramirez, Ryser, & Smith, 2001) or dynamic conditions (Garre et al., 2018; 38 Hassani, Cebrián, Mañas, Condón, & Pagán, 2006; Juneja & Marks, 2003; Valdramidis, 39 Geeraerd, Bernaerts, & Van Impe, 2006; Valdramidis, Geeraerd, & Impe, 2007). Different 40 models have also been proposed to describe the microbial inactivation kinetics. 41

42 Numerous studies focusing on the estimation of microbial kinetic parameters are performed in isothermal and dynamic conditions, but under ideal laboratory conditions (capillary tubes, 43 44 liquid inoculum and small sample mass). In real time industrial processing, food products are often submitted to time-varying temperatures. Moreover, due to several factors affecting heat 45 46 transfer, i.e. geometry, thermophysical properties or external convection, significant thermal heterogeneities can occur. In the literature, the effect of capillary tube diameters on the 47 48 inactivation parameters of E. Coli, in solid food, was investigated from a two-dimensional 49 heat transfer approach (Chung, Wang, & Tang, 2007).

A recent study by Garre et al., (2018) presented a mathematical model to describe the nonisothermal microbial inactivation process. The model was dedicated to *Escherichia coli* inactivation under linear heating rates of 1, 5, 15, 35 and 40 °C/min, from 35 to 70 °C. The authors developed a mathematical model taking into account the thermotolerance of the microbial cells. The model was able to describe the experimental data using a unique set of model parameters.

Another investigation, Hamoud-Agha, Curet, Simonin, & Boillereaux, (2013) proposed to estimate D_{ref} and z-value of Bigelow's (1921) equation. These parameters were estimated for *E. coli* K12 inoculated into a model food under dynamic conditions (homogeneous temperature distribution due to a very small solid sample). These parameters were included in the Geeraerd et al., (2000) inactivation model coupled with a 3D-heat transfer model under microwaves to illustrate the resulting inactivation heterogeneity.

The literature is rich concerning CFD and heat transfer modelling dedicated to pasteurization 62 of food products (Bhuvaneswari & Anandharamakrishnan, 2014; Cordioli, Rinaldi, Copelli, 63 Casoli, & Barbanti, 2014; Denys, Pieters, & Dewettinck, 2003; Dimou & Yanniotis, 2011). 64 However, except in Hamoud-Agha et al., (2013), it is rare to find these models coupled with 65 microbial inactivation equations and even less with real products under external dynamic 66 conditions. These elements reflect the originality of this work proposal. In this study, a 3D-67 CFD and heat transfer approach is coupled with the Geeraerd et al. (2000) inactivation model. 68 External temperature conditions (heating rates from 1 to 13 °C/min) are applied to 69 Escherichia coli K12 inactivation in pre-packed ground beef. Model parameters are estimated 70 from experimental data using the Levenberg-Marquardt algorithm (Marquardt, 1963). 71

72 **2.** Materials and methods

73 2.1 Experimental section

74 2.1.1 Sample preparation

Raw ground beef (Beef Steak 5% fat, Charal[®], Cholet, France) was obtained from a local retail store. The proportion of lean meat and fat was determined on the basis of information on the package. Ground beef was provided in vacuum-packed portions of 100 g. The samples were quickly frozen in a -20 °C freezer (Servathin, Carrières-sous-Poissy, France) and stored in a freezing chamber at -20 ± 1 °C. Before each experiment, a bag of the sample was thawed for 18 h in a cooling chamber (4 °C).

81 2.1.2 Preparation of bacteria inoculum

The strain of *Escherichia coli* K12 (CIP 54.117, N° 11.612) used in this study was provided by the Pasteur Institute, France, in the form of a freeze-dried sample. The strain was rehydrated in 0.2 mL and inoculated in 5 mL of Brain Heart Infusion (BHI) (Panreac Applichem – Darmstadt, Germany) at 37 °C for 18 h at 150 rpm (Incubator Shaker Series – Model Excella E24, New Jersey, USA). A pre-culture was prepared with 0.1 mL of the preceding suspension and inoculated in blood agar (Columbia + 5% Sheep blood - Bio-Rad - Marnes-la-Coquette, France) at 37 °C, for 18 h. A clone of the pre-culture was then dissolved in 2 mL of sterile water (0.85% NaCl - Medium, BioMerieux). 0.1 mL of this solution was isolated in several Petri dishes of the blood agar with a sterile Pasteur pipette and incubated under the same conditions. The colonies of this second isolation were removed with a sterile swab and dissolved in 1 mL of the beef serum (Eurobio Serum Bovin, France) supplemented with 20% (v/v) glycerol and stored frozen at -80 °C (Thermo Scientific TSE - USA).

To reactivate the strain, 0.1 mL of the stocked culture was inoculated into the blood agar (Columbia agar + 5% sheep blood – 43041, BioMérieux, France) and incubated at 37 °C for 18 h. Fresh colonies were carefully removed with a sterile swab and dissolved in 5 mL of medium suspension (Api 20150, BioMérieux, France) until a cellular concentration of 0.5 McFarland was reached (Densitometer Densimat, BioMérieux, Italy). The optical density of 0.5 McFarland standard is comparable to that of a bacterial suspension concentration of approximately 1.5 x 10^8 CFU/mL (CLSI, 2012).

101 2.1.3 Inoculation of ground beef

Raw ground beef samples (36 g) were removed aseptically from their refrigerated packaging (7 °C) and transferred into sterile polypropylene sampling pouches with 4 g of bacterial suspension. Ground beef was mixed manually to evenly distribute the inoculum throughout the sample, producing an initial cell concentration of approximately 10^7 CFU/g. For heat treatment, 3 g of inoculated ground beef samples were aseptically placed in packaging of 11.6 mm wide x 60 mm long.

108 2.1.4 Heat treatment for microbial inactivation

As illustrated in Figure 1, each sample was inserted into the cylindrical vessel and submitted 109 110 to a linear heating ramp in a programmable Peltier-based effect water bath (Hart Scientific AOIP, FC 9105, USA). After the heating treatment, the sample was quickly cooled to 20 °C 111 by ice immersion. The temperature was measured with a thermocouple (type-K) positioned 112 near the bottom of the sample. The data were collected by a data logger (AOIP Datalog, 113 91133 Ris Orangis, France) with output recordings every 1 second. To obtain the inactivation 114 kinetics curves, eight final temperatures (50, 52, 54, 56, 58, 60, 62 and 64 °C) were 115 116 considered. Six different heating rates (1, 3, 5, 7, 10 and 13 °C/min) were applied. Triplicates were carried out for each experiment. 117

118 2.1.5 Microbial enumeration and expression of results

The method of analysis was according to ISO 16649-1 (2001). This standard procedure specifies a horizontal method for the enumeration of β -glucuronidase-positive *Escherichia coli* in products intended for human consumption. It is based on a colony-count technique at 44 °C in a solid medium containing a chromogenic ingredient for detection of the β glucuronidase enzyme (ISO 16649-1, 2001).

Untreated inoculated ground beef and treated ground beef were aseptically removed from the 124 packaging and transferred to a sterile filtered stomacher bag (400 mL, BagPage, St. Nom, 125 126 France). The samples were diluted with 1:9 (w/w) of buffered peptone water (BK131HA, Biokar Diagnostics, France) and stomached for 2 min at 230 rpm (Stomacher Lab Blender 127 InterscienceBagMixer® - 400, Grosseron, St. Herblain, France). Decimal serial dilutions were 128 performed in buffered peptone water. 1 mL of the initial suspension or decimal dilution was 129 inoculated onto duplicate plates of tryptone-bile-glucuronic (TBX) medium (BK146HA, 130 Biokar Diagnostics, France) and incubated between 18 h to 24 h at 44 °C. The presence of 131 blue colonies is considered to be β -glucuronidase-positive *Escherichia coli*. 132

Each dish containing less than 150 typical CFU was numbered and the result was calculated as the arithmetic mean from two parallel plates. The results expressed to the base 10 logarithm of the microbial population for each temperature measurement were collected. The detection limit was fixed to 1 CFU/g of the ground beef.

137 2.2 Model design section

138

2.2.1 Heat transfer and computational fluid dynamics (CFD)

The experimental methodology described in section 2.1.4 was considered in a CFD-heattransfer model, with the following assumptions:

- *Assumption 1:* Water in the cylindrical vessel and the packaged sample were at the
 same initial homogeneous temperature (7 °C).
- *Assumption 2*: The ground beef sample was considered as being homogeneous and isotropic.
- Assumption 3: Specific heat, density and thermal conductivity of the ground beef were
 considered as being constant within the temperature range (Pan & Paul Singh, 2001;
 Tsai, Unklesbay, Unklesbay, & Clarke, 1998).

- Assumption 5: Axial symmetry was supposed for both thermal problems and fluid
 mechanics, limiting the size to one quarter of the experimental apparatus.
- Assumption 6: The mass transfer and the shrinkage of the sample were assumed to be
 negligible (in-package thermal treatment).
- Assumption 7: The package thickness was sufficiently thin to neglect its impact on the
 heat transfer (low thermal resistance).

The model is aimed at predicting the temperature profile when a heating rate is applied to the wall of the water bath heating cell. The model consists of two main computational domains: the first relates to liquid water which partially fills the cylindrical cell, the second is the ground beef sample located at the centre of the cell. As illustrated in Figure 2, the sample was immersed in water and its upper surface was exposed to air surrounding the medium (natural convection). The lateral surfaces of the cylinder are modelled with a first type boundary condition (T = T_{cylinder cell}).

163 2.2.2 Heat transfer modelling

Heat transfer is based on the general heat equation, which depends on thermophysicalproperties. For the ground beef sample, the heat equation is reduced to:

166
$$\rho_{(sample)} C_{p(sample)} \frac{\partial T}{\partial t} = div \left(\lambda_{(sample)} \nabla T\right)$$
(1)

167 For liquid water, a convective term is added to take into account the variation of the 168 thermophysical properties of water as a function of temperature:

169
$$\rho_{(water)} C_{p(water)} \frac{\partial T}{\partial t} + \rho_{(water)} C_{p(water)} \vec{u} \nabla T = div \left(\lambda_{(water)} \nabla T\right)$$
(2)

170 Where \vec{u} is the velocity field at any point of the 3D sample domain (*u*, *v*, *w* components). 171 The initial and boundary conditions are defined with the following mathematical expressions:

172
$$T = T_0, \quad t = 0, \quad \forall (x, y, z), \quad T_0 = 7^{\circ}C$$
 (3)

173
$$\lambda_{(sample / water)} \frac{\partial T}{\partial z}\Big|_{z=L} = h_{air}(T - T_{\infty}), \quad \forall (x, y,), \forall t > 0, \quad T_{\infty} = 20^{\circ}C$$
(4)

The boundary condition expressed by equation 4 is applied to the upper surface of water andsample.

176
$$T_{cylindrical \ cell} = T_0 + \frac{dT}{dt} t \begin{cases} for \ (x, y) \in [0; R] \ and \ Z = 0, \ \forall t > 0 \\ for \ x, y = R, \ \forall t > 0, \ for \ Z \in [0, L] \end{cases}$$
(5)

177 Figure 2 displays the coordinate system used to describe these mathematical expressions.

The convective heat transfer coefficient is due to natural convection between the surrounding air medium and the upper surface of the product. Free convection occurs only at the upper surface of both the water and the sample (Figure 2). An empirical correlation was conducted to compute the heat transfer coefficient on the upper cooled side of the cylinder. The air heat transfer coefficient (h_{air}) at the surface of the vertical cylinder with its axis running vertically is the same as that from a vertical plate, so long as the thermal boundary layer is thin (Churchill & Chu, 1975).

185 2.2.3 Fluid flow modelling

The fluid flow is due to natural convection occurring within the liquid water. This convective flow is modelled from the Navier Stokes equations (Newtonian fluid with incompressible flow). Both the continuity and momentum equations are described as follows:

189
$$\begin{cases} \frac{\partial u}{\partial x} + \frac{\partial v}{\partial y} + \frac{\partial w}{\partial z} = 0 & (continuity) \\ \rho_{water} \ \frac{d\vec{u}}{dt} = \rho_{water}g - \nabla P + \mu_{water}\Delta\vec{u} & (momentum) \end{cases}$$
(6)

190 with

191
$$\frac{d\vec{u}}{dt} = \frac{\partial\vec{u}}{\partial t} + \left(\vec{u}\,\nabla\right)\vec{u}$$
(7)

192 The associated initial and boundary conditions are given as:

193
$$P = P_{atm} + \rho_{water} g z \quad at \quad t = 0, \forall x, y, \quad for \ z \in [0; L]$$
(8)

194
$$\vec{u} = 0 \quad at \quad R = x, y, \quad \forall t > 0, \quad for \ z \in [0; L]$$
 (9)

For validation purposes, coupled heat transfer and fluid flow were solved for a heating rate of $10 \,^{\circ}$ C /min to compare the simulation to experimental results in the same conditions.

197 2.2.4 *Thermophysical properties*

Thermophysical properties of the sample used for modelling purposes are listed in Table 1. The apparent specific heat (C_p) of ground beef (5% fat) was determined using a differential scanning calorimetry (µDSC VII Evo - Setaram Instrumentations, France) at a constant pressure. Samples (~ 500 mg) were weighed in aluminium cells and sealed. A hermetically sealed empty cell was used as a reference. The samples were equilibrated at 20 °C and then heated to 65 °C, at a heating rate of 1 °C/min. The measurements were made in triplicate and the average values were determined.

The density and the thermal conductivity of sample were taken from Pan & Paul Singh (2001). The authors reported densities of 1006-1033 kg/m³ and thermal conductivities of 0.35-0.41 W m⁻¹ °C⁻¹ of ground beef (4.8% fat). These values do not change significantly as a function of temperature range (between 5 and 75 °C).

Thermophysical properties of water were considered as temperature dependent (Green & Perry, 2007). These properties were directly implemented in the model using polynomial interpolation functions (Table 2).

212 2.2.5 Computational details for CFD simulations

Based on the finite element method, the model solved the partial differential equations numerically, with the computational code COMSOL[®] Multiphysics 5.3a. In this study, the geometry consisted of irregular shapes (ellipsoid-cylinder with axial symmetry) justifying the use of the finite element method to simulate the coupled CFD-heat transfer phenomenon.

As shown in Figure 3, the generated mesh was designed for a 3D configuration, to ensure good accuracy for the numerical resolution (522 830 tetrahedral elements with appropriate boundary layers at the near-cell wall zone). Additional mesh sensitivity studies were performed to obtain accurate number of mesh elements in relation to the computational cost. The mesh independency to the numerical results was also verified. First, the fluid flow was solved as a stationary study in order to initialize the velocity field for a better consistent initial value.

Due to the temperature-dependent thermophysical properties of water, the coupled equations 224 (Eq. 6 and 7) within fluid and heat transfer within the meat sample (Eq. 1) and water (Eq. 2) 225 were solved as a time-dependent study with a direct solver (strong coupling). To ensure a 226 good consistent initial value, the initial time step for the resolution was fixed at 10^{-7} s. Based 227 on these computational details, the computational time did not exceed 4 hours 9 minutes and 228 12 seconds on a Dell[®] Precision TM Workstation computer, equipped with 2×Intel[®] Xeon 229 processors (8 cores), at 2.5 GHz, with 256 GB of RAM, running on Windows®8 Professional, 230 64 bits. 231

232 Several simulations were performed depending on the heating rate being tested (from 1 to 13 °C/min).

234 2.2.6 Microbial inactivation model

The dynamic non log-linear model developed by Geeraerd et al. (2000) was selected in this study. The reduced model, without the tailing effect, consists of two coupled ordinary differential equations as follows:

$$\frac{dN}{dt} = -k_{\max} \left(\frac{1}{1+C_c}\right) N \tag{10}$$

$$\frac{dC_c}{dt} = -k_{\max}C_c \tag{11}$$

where *N* represents the microbial cell density (CFU/g). In equation 10, $\frac{dN}{dt}$ is forced to zero once the limit of detection (1 CFU/g) is reached. k_{max} denotes the specific inactivation rate (s⁻

¹) and C_c is related to the physiological state of cells (-). A single value for $C_c(0)$ can be used for all experiments, since the methodology of inoculation is standardized. Consequently, a similar initial physiological state is expected for cells

The temperature effect on kinetic parameters (k_{max}) was expressed using Bigelow's (1921) equation:

247
$$k_{\max}\left(T\right) = \frac{\ln 10}{D_{ref}} \exp\left(\frac{\ln 10}{z} \left(T - T_{ref}\right)\right)$$
(12)

where D_{ref} denotes the decimal reduction time (s) at the reference temperature T_{ref} and *z*-value the thermal resistance constant (°C).

250 2.2.7 Parameter estimation procedure

The approach consists in estimating the model parameters of Equation 12 by minimizing the residual sum of squares (RSS):

253
$$RSS = \sum_{i=1}^{n} \left(log \frac{\frac{1}{v} \int N_{simu}(t_i) dv}{N_0} - log \frac{\overline{N}_{exp}(t_i)}{N_0} \right)^2$$
(13)

where t_i are the times at which the thermocouple located in the sample reached the expected the temperatures (respectively 50, 52, 54, 56, 58, 60, 62 and 64 °C), *n* is the number of experimental data points, N_{simu} is the volume-averaged value of the simulated cell density at t_i . N_{exp} is the volume-averaged value of the triplicate experimental cell density at t_i . *V* is the volume of the sample and N_0 is the initial cell density.

To minimize the RSS, the Levenberg-Marquardt algorithm (*lsqnonlin* available in Matlab[®]7.10) was used. This algorithm is based on the least-squares minimization technique and is an improvement of the Gauss-Newton algorithm. In addition, confidence intervals at 95% were calculated for the estimated parameters using *nlparci* function available in Matlab[®]7.10.

264 2.2.8 Methodology used for model comparison and selection

Considering the model proposed by Geeraerd et al. (2000), it appears that parameters $C_c(0)$, D_{ref} and z-value can be estimated to fit the experimental data. Nevertheless, it is necessary to evaluate the relevance of the number of adjusted parameters. In this contribution, a non-linear least squares approach based on minimization of the RSS (residual sum of squares) is used. It evaluates the well-known Akaike Information Criterion (AIC) (Akaike, 1974), which provides a means for model selection. AIC is established as follows, on the assumption that the residues distribution is Gaussian:

272
$$AIC = 2 p - 2 \ln (LL)$$
 (14)

273 Where p is number of independently adjusted parameters within the model and LL is the 274 maximum likelihood. If the AIC criterion is used only with the aim of comparing models, the 275 following simplified expression is used:

$$AIC^* = 2 p + \frac{RSS}{\sigma^2}$$
(15)

277 When the residual sum of squares are almost equal for two different models, the minimum 278 AIC estimation selects the model with the lowest number of parameters according to the 279 "principle of parsimony" (Yamaoka, Nakagawa, & Uno, 1978). In addition, a variance σ^2 was 280 calculated for each heating rate from the experimental results obtained in triplicate.

3. Results and discussion

282 **3.1 Specific heat of ground beef**

The apparent specific heat (C_p) of ground beef at 5% fat was determined from 20 to 65 °C. As can be observed in Figure 4, the C_p values were found to be quite constant on the temperature range. The average measured value $(C_p = 3.69 \pm 0.01 \text{ kJ kg}^{-1} \text{ K}^{-1})$ is close to the reported one in the literature for meat product (Zhang, Lyng, Brunton, Morgan, & McKenna, 2004). The C_p average value was integrated directly into the heat transfer model.

288 **3.2** Heat transfer and CFD model validation

To validate the modelling approach, the experimental temperatures measured as a function of 289 time in both the water and the sample were compared to the one simulated from the CFD-heat 290 transfer model. Figure 5 illustrates the case for a heat treatment at 10 °C/min. The standard 291 deviation of the experimental data was about ± 0.3 °C, corresponding to the measuring 292 accuracy of the thermocouple used. Figure 5 depicts the linear setting temperature. The 293 agreement between experimental and model temperatures of the water and the sample is 294 noticeable. Non-negligible thermal delays (about 26 s and 57 s) can be observed respectively 295 between the water and the sample and the set temperature and the sample, justifying the 296 interest of such an approach when heating rates are applied. 297

The CFD modelling enabled prediction of the velocity fields due to the natural convection problem within a static enclosure. The Navier-Stokes equation was solved by considering a laminar flow. The norm of the velocity field, for example at 10 °C/min, is depicted in Figure 6 for different processing times. Fluid motion occurs (illustrated by black arrows in Figure 6), because of the temperature dependent thermophysical properties of water. Recirculation of the heating fluid leads to velocity gradients, with a maximum value of 2.5 mm/s around the sample (norm of the velocity field). The Reynold number was calculated considering this maximum fluid velocity of 10 °C/min with a value of 9.4. This result clearly demonstrates a natural convection of water within a confined space, in which the flow regime is laminar.

Once the CFD approach is validated, the model illustrates the temperature gradients when heating rates are applied. Figure 7 illustrates the temperatures at the final time of simulation for the six different heating rates considered. Simulation was stopped when the simulated temperature of the corresponding point to the thermocouple position reached 64 °C.

For all treatments the hottest zone was located at the surface of the sample, except for the upper central part, due to contact with ambient air (cooling). The coldest zone was along the central axis of the product. The temperature gap between the hottest and coldest zones increased with the applied heating rates, so 4.3, 6.2, 8.1, 9.8, 12.4, and 15.8 °C for 1, 3, 5, 7, 10 and 13 °C/min, respectively.

To confirm the effect of temperature heterogeneities on the volumetric microbial inactivation, equations 10, 11 and 12 were used. The simulation of the 3D local temperature distribution in the meat sample (CFD and heat transfer model solved by COMSOL[®] and transposed to MATLAB[®]) was performed to quantify the 3D local microbial inactivation. For example, under a thermal treatment at 10 °C/min, the numerical result illustrated a thermal cartography with hot and cold points. These temperature heterogeneities lead to heterogeneous local microbial inactivation represented by the hatched area in Figure 8.

In Figure 8, a total inactivation can be seen for the hottest point after 5.8 minutes. For the coldest point, the maximum inactivation achieved was of 2 Log₁₀, at the end of treatment. However, it is not possible to experimentally validate these local microbial inactivation data, at every geometrical point of the domain. Consequently, the use of a mathematical model to describe the local microbial inactivation during a pasteurization process remains a relevant tool. More, it confirms the importance of considering the volumetric distribution of temperature in the meat product to accurately predict microbial inactivation.

330 **3.3 Estimation of kinetic parameters of inactivation**

Various results can be found in the literature concerning evaluation of kinetic parameters 331 (Dref, Tref and z-value) under isothermal conditions. In a study of heat resistance of 332 Escherichia coli 0157:H7 inoculated into 7% fat ground beef, Ahmed et al. (1995) obtained a 333 $D_{\rm ref}$, by linear regression, of 684 s (at 55 °C) and 27 s (at 60 °C), and a z-value equal to 334 4.78 °C. In the same strain, but in ground beef with 4.8% fat, Smith et al. (2001) calculated by 335 first-order kinetics a *D*_{ref} equal to 73 s (at 58 °C) and 19 s (at 61 °C), for a *z*-value of 3.79 °C. 336 These variations might be attributed to differences in the strains of E. coli used, the fat content 337 of meat, along with differences in recovery methods and sample size (Smith et al., 2001; 338 Stringer, George, & Peck, 2000). In both studies, the values of D_{ref} differ greatly whereas z-339 value remains in the same order of magnitude. 340

In this study, ground beef with 5% fat was used and the heating rates described above were 341 applied. Despite these dynamical conditions, in a first approach, the parameters $D_{ref} = 73$ s, z-342 value = 3.79 °C and T_{ref} = 58 °C obtained by Smith et al. (2001) in isothermal conditions and 343 $C_c(0) = 0.23$, as reported in Hamoud-Agha et al. (2013), were used in the Geeraerd et al. 344 345 (2000) model. Results are illustrated in Figure 9. Black circles illustrate the experimental data in triplicate, whereas the predictions obtained using the aforementioned parameters are 346 represented by broken blue lines. Note that the model underestimates the experimental 347 inactivations, but this underestimation has the same order of magnitude whatever the heating 348 rate. 349

A second approach consisted in estimating the model parameters from the dynamic 350 experiments. First, the triplet parameters $C_c(0)$, D_{ref} and z-value were estimated. Poschet et al., 351 (2005) showed that the uncertainties on D_{ref} and z-value are minimal when T_{ref} is equal to the 352 middle of the lethal experimental temperature range. As D_{ref} is closely linked to T_{ref} , this value 353 was fixed at 58 °C, as defined in Smith et al. (2001). The C_c(0) value was estimated with a 354 lower bound equal to zero. In table 3, $C_c(0)$ is not significantly different from zero for all 355 heating rates. Consequently, equation (11) can be removed and the Geeraerd et al. (2000) 356 model becomes a classical first order inactivation model. 357

With the Bigelow model, the D_{ref} and z-values were estimated (Table 4). It should be noted that these AIC^{*} values are lower than the AIC^{*} values obtained in previous estimate procedure with three parameters (Table 3). The confidence intervals for D_{ref} increase for extreme heating rates (10 and 13 °C/min), certainly due to the large thermal heterogeneities. Another explanation is based on Van Derlinden, Balsa-Canto, & Van Impe, (2012), where the authors discussed the link between parameter identifiability and the shape of the temperature profile. The results of *z*-value indicated in Table 4 are consistent with those presented by Smith et al. (2001), for a similar product at 58 °C (*z*-value = 3.79 °C). Indeed, this value (3.79 °C) is almost included in the confidence intervals and will thus be considered hereafter.

With the estimation limited to D_{ref} , the confidence intervals are widely reduced, even if they remain significant for heating rates of 10 and 13 °C /min (12.52 s and 8.75 s). These results are summarized in Table 5. Compared to Table 4, quite similar or lower values of the AIC^{*} were obtained. It appears that adjusting D_{ref} is sufficient to fit the experiments without degrading the AIC^{*} values. Moreover, it also reduces the parameter uncertainties as verified with 95% confidence intervals.

373 Thereby, for one estimated parameter, the D_{ref} values remain in the same order of magnitude.

Consequently, the average of the previously estimated D_{ref} values (Table 5) of 47 s was used. The predicted inactivation curves obtained in this case are presented in black continuous line in Figure 9. They are very close to inactivation results obtained with each estimated D_{ref} (red dotted line is superimposed on black continuous line in figure 9). Hence, for D_{ref} values ranging from 40 to 50 s, considering a unique value of 47 s does not significantly affect the inactivation predictions.

Finally, results reveal that the k_{max} expression with temperature dependence proposed by Bigelow is sufficient to fit experimental data obtained in dynamic conditions with large temperature heterogeneities (>15 °C for 13 °C /min), provided that these heterogeneities are considered in the prediction model. Constant values for T_{ref} , z and D_{ref} can be considered for all heating rates in the range 1 to 13 °C /min. Additionally, T_{ref} and z-values of literature (Smith et al., 2001) were used with only a fitting of D_{ref} .

4. Conclusion

The contribution of this work lies in the integration of experimental and numerical approaches dedicated to the real-food pasteurization process including both thermal and inactivation kinetics with high heating rates.

The originality of this paper lies (i) in the integration of three-dimensional heat transfer conditions in parameter estimations applied to the inactivation model and (ii) in the application of extreme heating rates to challenge the validity of the Geeraerd modelling approach developed for isothermal conditions. The heat transfer model was validated during blank tests, and the volume-average of logarithmic inactivation was considered to be comparable with experimental data. In addition, the experiments were carried out on real ground beef, and not in capillary tubes, as usually proposed. The inactivation model was coupled with heat transfer to design a 3D simulator to predict the spatial distribution of logarithmic inactivation of *E. coli* in a meat sample.

Such a 3D simulator illustrates how the temperature gradients can provoke large inactivation heterogeneities, and thus the necessity to consider them in realistic pasteurization conditions. The satisfactory agreement between model and experiments allows coupling of the modelling approach with process control procedures, allowing energy supply to reach the expected microbial inactivation while limiting temperature heterogeneities. Such an objective could be reached for example by using combined energy sources with volumetric heating to avoid overtreatment at the surface of the food product (microwaves, ohmic heating).

406 Acknowledgments

The authors are grateful to the National Council for Scientific and Technological
Development (CNPq, Brazil) for C. D. Albuquerque's PhD scholarship and financial support
(process number 232767/2014-9).

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- 520













mm/ s



1 °C/min



5 °C/min





3 °C/min



7 °C/min

















Table 1. Thermophysical properties of ground beef.

Properties	Value	Unit	Source
Apparent specific heat (C _p)	3.69 ± 0.01	kJ kg ⁻¹ K ⁻¹	Experimental
Thermal conductivity (λ)	0.35	W m ⁻¹ $^{\circ}$ C ⁻¹	Pan and Singh (2001).
Density (ρ)	1006	kg/m ³	Pan and Singh (2001).

Table 2. Thermophysical properties of water as a function of temperature (Kelvin).

Functions of the properties*	Unit
$C_p = 12010.15 - 80.41 \times T + 0.31 \times T^2 - 5.38 \times 10^{-4} \times T^3 + 3.62 \times 10^{-7} \times T^4$	J kg ⁻¹ K ⁻¹
$\lambda = -0.87 + 8.95 \times 10^{-3} \times T - 1.58 \times 10^{-5} \times T^{2} + 7.97 \times 10^{-9} \times T^{3}$	W m ⁻¹ $^{\circ}$ K ⁻¹
$\rho = 838.47 + 1.40 \times T - 3.01 \times 10^{-3} \times T^2 + 3.72 \times 10^{-7} \times T^3$	kg /m ³
*Green & Perry, 2007	

Table 3. Estimated values of D_{ref} , *z*-value and $C_c(0)$, for $T_{\text{ref}} = 58 \,^{\circ}C$. Standard deviation of experimental data (σ^2), 95% confidence intervals (CI) and simplified Akaike Information Criterion (AIC^{*}).

Heating rate	σ^2	$D_{\rm ref}(s)$	CI D _{ref}	z-value	CI <i>z</i> -value	$C_{c}(0)$	CI	AIC*
(°C /min)				(°C)			$C_c(0)$	
1	0.24	33.1	3.8	1.96	1.41	0	0.30	11.8
3	0.20	48.9	6.8	3.81	0.63	0	0.21	12.2
5	0.23	41.7	5.3	7.32	1.53	0.11	0.15	10.2
7	0.34	45.7	8.1	3.59	0.61	0.21	0.18	9.3
10	0.50	56.2	22.4	6.19	2.3	0.08	0.17	11
13	0.63	74.7	33.1	2.99	0.69	0.04	0.15	9.2

Heating rate (°C /min)	σ^2	$D_{\mathrm{ref}}\left(\mathrm{s}\right)$	CI D _{ref}	<i>z</i> -value (°C)	CI z-value	AIC*
1	0.24	33.1	3.8	1.96	1.41	8.8
3	0.20	48.9	6.8	3.81	0.63	9.2
5	0.23	40.8	5.1	6.70	1.41	7.2
7	0.34	31.4	12.4	4.80	0.81	7.2
10	0.50	62.1	23.2	5.98	2.65	8.2
13	0.63	73.1	32.6	3.02	0.71	6.6

Table 4. Estimated values of D_{ref} and *z*-value ($T_{ref} = 58 \ ^{\circ}C$). Standard deviation of experimental data (σ^2), 95% confidence intervals (CI) and simplified Akaike Information Criterion (AIC^{*}).

Table 5. Estimated values of D_{ref} (z = 3.79 and $T_{ref} = 58$ °C). Standard deviation of experimental data (σ^2), 95% confidence intervals (CI) and simplified Akaike Information Criterion (AIC^{*}).

Heating rate (°C/min)	σ^2	$D_{\mathrm{ref}}(\mathrm{s})$	CI D _{ref}	AIC*
1	0.24	40.1	0.03	9
3	0.20	47.5	0.38	6.2
5	0.23	50.0	0.59	8.2
7	0.34	50.0	0.48	3.6
10	0.50	50.1	12.52	6.8
13	0.63	46.7	8.75	3.6