

Soil Properties and Multi-Pollution Affect Taxonomic and Functional Bacterial Diversity in a Range of French Soils Displaying an Anthropisation Gradient

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1 Soil properties and multi-pollution affect taxonomic and functional bacterial

2 diversity in a range of French soils displaying an anthropisation gradient

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8

9 Abstract

10 The intensive industrial activities of the 20th century have left behind highly contaminated wasteland soils. It is 11 well known that soil parameters and the presence of pollutants shape microbial communities. But in such 12 industrial waste sites, the soil multi-contamination with organic (polycyclic aromatic hydrocarbons, PAH) and 13 metallic (Zn, Pb, Cd) pollutants and long-term exposure may induce a selection pressure on microbial 14 communities that may modify soil functioning. The aim of our study was to evaluate the impact of long-term 15 multi-contamination and soil characteristics on bacterial taxonomic and functional diversity as related to the 16 carbon cycle. 17 We worked on 10 soils from northeast of France distributed into 3 groups (low anthropised controls, slag

18 heaps, and settling ponds) based on their physico-chemical properties (texture, C, N) and pollution level. We 19 assessed bacterial taxonomic diversity by 16S rDNA Illumina sequencing, and functional diversity using Biolog[®]

and MicroResp[™] microtiter plate tools.

21 Although taxonomic diversity at the phylum level was not different among the soil groups, many OTUs were 22 influenced by metal or PAH pollution, and by soil texture and total nitrogen content. Functional diversity was 23 not influenced by PAH contamination while metal pollution selected microbial communities with reduced 24 metabolic functional diversity but more tolerant to zinc. Limited microbial utilisation of carbon substrates in 25 metal-polluted soils was mainly due to the nitrogen content. Based on these two observations, we 26 hypothesised that reduced microbial activity and lower carbon-cycle-related functional diversity may have 27 contributed to the accumulation of organic matter in the soils that exhibited the highest levels of metal 28 pollution.

30 Introduction

58

31 The decline of the steel industry in the north east of France at the end of the 20th century left behind more 32 than 6,000 ha of polluted wastelands [1]. On these sites, the soils are multi-contaminated with hydrocarbons

33 and metallic trace elements (MTEs) [2, 3]. On sites of former coking plants or slag heaps, polycyclic aromatic

34 hydrocarbon (PAH) pollution results from the use of coal tar during coke manufacturing. PAHs are ubiquitous 35

contaminants produced from the incomplete combustion of organic matter [4], especially during industrial

36 activities. In settling ponds resulting from the sewage sludge storage of blast furnace gases [5], high

37 concentrations of MTEs (Cd, Cu, Zn, Pb, Ni) are also often encountered.

- 38 In soils, the fate of PAHs and MTEs depends on various abiotic and biotic processes. Over time PAHs adsorb 39 onto organic matter and spread into the micropores of soil components. Thus is referred to as the pollutant 40 ageing, and leads to a decrease of pollutant availability [6]. MTE availability depends mostly on soil properties 41 (pH, organic matter, redox potential...) [7, 8]. PAH compounds can be degraded over time mainly through 42 microbial processes [9] that can be limited by low PAH availability [10, 11]. MTEs, cannot be biodegraded, but 43 microorganisms can modify their speciation through direct use or modifications of soil properties [12].
- 44 Numerous studies have focused on the short-term influence of PAHs or MTEs on microbial communities. The 45 selective pressure exerted by PAH toxicity [13, 14] could modify microbial community composition [14, 15], 46 reduce microbial taxonomic diversity [16], or inhibit microbial activity and the global activity of enzymes such 47 as arylsulphatase, phosphatase, urease, dehydrogenase [13]. Similarly, various studies have highlighted the 48 negative effects of MTE pollution. For example the effect of zinc addition has been studied on enzymatic 49 activity and microbial community composition [17], as well as on the taxonomic diversity of bacterial 50 communities [18].
- 51 In aged polluted soils, various adaptation processes enable the microbial community to cope with and even 52 benefit from pollution. The selection of PAH degraders [15, 19] and MTE-resistant microbes [17, 20] in aged-53 polluted soils has been shown using acquired tolerance tests [21]. However, the long-term impact of chronic 54 exposure to PAHs and MTEs on microbial community composition has been less addressed [22, 23]. The impact 55 of pollutants has been mainly studied separately, although multi-contamination can have a synergistic effect 56 [24, 25]. Studies considering aged pollution and the impact of multi-contamination on microbial taxonomic 57

diversity (richness and evenness) are scarce [26] and therefore needed.

- The carbon cycle is pivotal in soil functioning because it contributes to all biological processes, such as biomass 59 production or respiration by organisms. It also affects other nutrient cycles through organic matter recycling 60 and mineralisation. In a context of soil pollution, one can thus wonder how the different functions of the 61 carbon cycle are affected, and study the microbial functions involved in key C-cycle processes. Metabolic 62 functional diversity can be defined as the number (richness) and the evenness of the metabolic functions of the 63 microbial community. One approach to estimating functional diversity is by comparing the degradation 64 capacities of carbon substrates using Biolog[®] [27] and MicroResp[™] [28] microtiter plate methods in the 65 presence or absence of pollutants, as done with metals that decrease bacterial functional diversity [29]. 66 Apart from this impact of pollutants, it is well known that soil parameters, such as the pH, texture, the C
- 67 content or the C/N ratio, highly contribute to shape microbial community composition [30, 31] and functional

- diversity [32] in soils of various origins (forest, grassland, agricultural...). In polluted soils, such as urban,
 industrial, traffic, mining and military areas [3], physico-chemical parameters have also been found to largely
 influence the composition of the bacterial and fungal communities [22].
- 71 We therefore hypothesised that microbial diversity in multi-contaminated soils may be influenced by both soil 72 properties and pollutant concentrations and availability, but differently at the taxonomic and functional levels. 73 In this context, the present study aimed to: i) assess taxonomic diversity and metabolic functional diversity of 74 microbial communities in a context of aged multi-pollution, and ii) to determine which soil factors (physico-75 chemical parameters, pollutants...) affected these diversities. We studied microbial communities from a 76 collection of 10 soils presenting gradients of PAH and metal pollution, and under different land uses, ranging 77 from weakly to highly anthropised, i.e. from forest soil to industrial soil. We assessed bacterial taxonomic 78 diversity through 16S rRNA gene tag-amplicon sequencing (Illumina MiSeg). We estimated metabolic functional 79 diversity using the degradation functions of ecologically relevant carbon substrates through Biolog® and
- 80 MicroResp[™] microtiter plate methods.

81 Materials and methods

82

83 Study sites and soil sampling

84 Ten soils originating from industrial wastelands and waste sites, natural forests or ancient gravel pit, located in 85 the "Grand Est" region (north-east of France), were sampled in November 2015. This soil collection was chosen 86 to cover a wide range of anthropisation situations, i.e. from no or low contamination to high contamination, 87 and represented an anthropisation gradient. All sampling sites are within 100km. Three soils were considered 88 as the control soils (ctrl), i.e. two low-anthropised forest soils collected at Hémilly (He; 49°2'1"N/6°30'51"E; 89 Moselle; Stagnic Luvisol) and Montiers-sur-Saulx (Mo; 48°31'55"N/5°16'8"E; Meuse; Calcaric Cambisol), and 90 one anthropised but unpolluted former gravel pit soil collected at Dieulouard (Di; 48°49'44" N/6°5'2"E, 91 Meurthe-et-Moselle; Fluvisol). Seven anthropised soils (Technosol) known to be polluted by MTEs and/or PAHs, 92 where industrial activities stopped during 1980s, were collected from: i) former slag heaps (sh) at Homécourt 93 (Ho; 49°12'79"N/5°59'72"E, Meurthe-et-Moselle), Terville (Te; 49°20'25"N/6°08'33"E; Moselle), Uckange (Uc; 94 49°18'58"N/6°9'55"E; Moselle), and Neuves-Maisons (NM; 49°12'52"N/5°59'45"E; Meurthe-et-Moselle), and ii) 95 former settling ponds (sp), dried since at least 30 years, at Pompey (Po; 48°46'8"N/6°8'8"E; Meurthe-et-96 Moselle), Mont-St-Martin (MsM; 49°32'9"N/5°46'46"E; Meurthe-et-Moselle), and Russange-Micheville (RM; 97 49°28'58"N/5°55'51"E; Moselle).

98 All sites were colonized by trees (mainly birch and/or beech) and herbaceous plants, with a very heterogeneous 99 colonization density from one site to another, except for the NM soil which was colonized only by some herbs. 100 None of the soil samples were rhizosphere soils or directly under plant influence (sampling at least 2m from a 101 tree). As plant colonization was sparse by location, we collected samples on bare soils. At each of the ten sites, 102 samples were collected from three independent sub-sites 1 m apart. After removing the litter layer, if present, 103 one soil block per sub-site (block of 20 cm on the side and 30 cm deep) was removed using shovel. The three 104 sub-site soil blocks were mixed to get one composite sample per site. Back to the laboratory, the soil samples 105 were air-dried at room temperature for one week, and then sieved at 2 mm. The dried and sieved soils were 106 then stored at room temperature in the dark before chemical and microbiological analyses, and aliquots were 107 stored at -80 °C before freeze-drying and PAH analyses.

108

109 Soil physico-chemical characteristics

110 The soil physico-chemical characteristics were determined from the dried sieved soils. The pH was measured 111 (PHM210 Radiometer Analytical, equipped with a pH probe, Bioblock Scientific) in a soil suspension prepared in 112 distilled water (1:5 w:v). The CaCO₃ content was estimated from measurements of CO₂ concentrations released 113 after acid (4M HCl) decarbonation of 1 g of dw soil, using an infrared absorbance (Binos 1004 analyser), with A 2325.6 cm⁻¹. The water retention capacity was estimated from the difference in weight between dry and water-114 115 saturated soil, which corresponded to 100%. Measurements of the soil texture (clay fraction, 0-2 µm; silt 116 fraction, 2-63 µm; sand fraction 63-2,000 µm; ISO 11277), the cation exchange capacity (CEC; determined by 117 Metson method, NF X 31-130), organic carbon, organic matter, nitrogen concentrations (C, OM, and N,

- respectively; Dumas method, ISO 10694), the available phosphorus concentration (P; Olsen method, NF ISO 119 11263), total (fluoridric acid extraction) and available (calcium chloride extraction) concentrations of metals 120 (Cr, Cd, Cu, Ni, Pb, Zn for total and available concentrations, plus Tl, Mo, Al, Ca, Fe, K, Mg, Mn, and Na for total 121 concentrations) were performed at the laboratoire d'analyse des sols (INRA, Arras, France).
- We calculated a metal pollution index (Mi) in order to compare metallic pollution of soils using only one
 variable. Mi was the sum of the relative proportions (RP_{metal}) of nine metal concentrations (Zn, Pb, Cd, Fe, Cr,
 Cu, Ni, Co, Tl). RP_{metal} in soils were calculated using the following formula:

$$RP_{metal} = \frac{MC_{soil} \times 100}{MC_{max}}$$

- where MC_{soil} is the total metal concentration in the soil, and MC_{max} is the maximum total metal concentration found among all the soils of the collection. Total instead of available metal concentrations were used because some of the available metal values were lower than the detection limit. The nine metals were chosen because they commonly resulted from the steel industry, and when for at least one soil, the metal concentration was higher than the mud compost norm (NF 44 095) or the natural geochemical background of French soils [33].
- 130

131 PAH extraction and analysis

132 Soil samples stored at -80 °C were freeze-dried and ground to 500 µm (Mixed Mill MM 400, Retsch). Total PAHs 133 were extracted from 1 g dw soil in triplicate with dichloromethane, at 130 °C and 100 bars, using accelerated 134 solvent extraction (DIONEX® 200 ASE), as described in Cennerazzo et al. [34]. Available PAHs were extracted 135 from 1 g dw soil in triplicate with 20 ml of hydroxypropyl-β-cyclodextrin (50 mM) in Teflon™ FEP Oak Ridge 136 centrifuge tubes (Nalgene, USA) [35]. After mixing (16 h) at 24 °C and centrifugation, PAHs were extracted from 137 the aqueous solution through liquid:liquid extraction using dichloromethane. Both solvent extracts of total and 138 available PAHs were evaporated (nitrogen flow) and dissolved in acetonitrile for PAH analysis using a reverse-139 phase chromatography (UHPLC DIONEX® Ultimate 3000 system) equipped with a Diode Array Detector (UV 140 detection, 254 nm) and a Zorbax Eclipse PAH column (2.1 x 100 mm, 1.8 µm, Agilent).

141

142 Carbohydrate, organic acid, and dissolved organic carbon measurements

143 Carbohydrates, organic acids, and dissolved organic carbon (DOC) were measured on triplicate soil aqueous 144 extracts (1:5 w:v) using 6 g dw soil mixed with 30 ml of distilled water in Teflon™ FEP Oak Ridge centrifuge 145 tubes (Nalgene, USA) for 2h at 20 °C [36, 37]. After centrifugation and filtration, aqueous extracts were 146 recovered. Seven carbohydrates (inositol, trehalose, sucrose, glucose, xylose, mannose, and fructose) were 147 quantified by an ICS 3000 ion-exchange chromatograph equipped with a Dionex CarboPac SA10 column 148 (Thermo Scientific). Eleven organic acids (gluconic, lactic, acetic, propionic, formic, pyruvic, succinic, maleic, 149 oxalic, fumaric, and citric acids) were quantified using an ICS 2100 ion chromatographer equipped with a 150 Dionex IonPac AS11 HC column (Thermo Scientific). DOC was measured on a TOC-V analyser (Shimadzu).

151

152 Quantification of culturable bacteria

The most probable number (MPN) was measured on 1 g dw soil (in triplicate) moistened (to 60% of their water retention capacity) and pre-incubated for 2 days (24°C) for reactivation of microbial community. After soil aliquots were resuspended in 10 ml of NaCl (0.9%) and mixed for 1 h, ten-fold dilution series were prepared in NaCl (0.9%) and used to inoculate 96-well microtiter plates (16 wells *per* dilution) filled with nutrient Broth No 1 (Fluka analytical). After incubation for 3 days (24 °C), absorbance at 600 nm was measured using spectrophotometer (Safas, Monaco) and MPN counts were calculated using Mac Grady's tables, and expressed as MPN *per* gram of soil.

160

161 DNA extraction, real-time quantitative PCR, and sequencing

162 Genomic DNA (gDNA) was extracted in triplicate from ca. 0.5 g of moistened soil (as described above) stored at 163 -20 °C, using Fast DNA Spin Kit for Soil (MP Biomedicals, France), following the manufacturer's instructions. Concentration and purity (A_{260}/A_{280} ratio) were measured using a spectrophotometer (UV1800, Shimadzu) 164 equipped with a TrayCellTM adapter (Hellma[®]). gDNA was diluted to 5 ng μ l⁻¹ for real-time quantitative PCR 165 166 (qPCR) and sequencing library preparation. The abundance of fungi and bacteria was estimated by qPCR using 167 the primer sets Fung5F/FF390R [38], 968F/1401R [39], targeting 18S and 16S rRNA genes, respectively. The 168 qPCR assays were performed as previously described [40, 41]. Briefly, the reaction mixture (20 µl) was 169 composed of 10 µl iQ SYBR green SuperMix (Bio-Rad), 0.8 µl primers (10 µM), 0.4 µl bovine serum albumin 170 solution (3%), 0.2 µl dimethyl sulfoxide, 0.08 µl T4gp32 (MP Biomedicals, France), and 1 µl DNA (diluted gDNA or 10-fold dilution series from 10^8 to 10^1 copies μl^{-1} of the standard plasmids). Quantification was performed 171 172 using a CFX96 Real-Time PCR detection system (Bio-Rad), using 56 °C and 50 °C as annealing temperature for 173 16S and 18S rDNA quantification, respectively.

174 The V3/V4 region of bacterial 16S rRNA genes (ca. 550 bp) was amplified using primers 175 S-D-Bact-0341-a-S-17 and S-D-Bact-0787-b-A-20 [42, 43] and following a previously described dual-index 176 strategy [44] using PCR primers with Illumina adaptor, pad and index sequences [37]. PCR reactions were 177 performed on 2 µl of diluted gDNA using Phusion high-fidelity polymerase (Thermo Scientific). PCR reactions 178 consisted of 31 cycles with touchdown annealing temperature for 18 cycles (63 °C to 54°C with a decrease of 179 0.5 °C/cycle) and 13 cycles at 54 °C. Amplification products were checked on 1% agarose gel electrophoresis, 180 and purified using the UltraClean-htp 96 Well PCR Clean-Up kit (Qiagen) following the manufacturer's 181 instructions. After Quant-iT Picogreen ds-DNA assay Kit (Invitrogen) quantification, an amplicon library was 182 prepared (equimolar pool at 10 nM), purified on a QIAquick PCR purification kit column (Qiagen), and sent for 183 sequencing to Genewiz platform (South Plainfield, NJ, USA) using an Illumina MiSeq V2 Kit for 2 x 250 bp 184 paired-end sequencing. Illumina MiSeq paired-end reads have been deposited in the SRA database under 185 Bioproject accession number PRJNA450766. Sequence data were analysed following the MiSeq SOP procedure 186 available in March 2017 and described in Kozich et al. [44], using Mothur v.1.38.0 [45]. Paired-end reads were 187 trimmed using the following criteria: QS>20, 404 bp < length < 454 bp, and no ambiguous bases. Chimeras 188 detected using Uchime [46] and singletons (sequences appearing only once among all samples) were removed. 189 Alignment of unique sequences and taxonomy was assigned using the Silva bacteria database (cutoff = 80).

Sequences affiliated to archaea, eukaryota, unknown, mitochondria, and chloroplasts were removed. Sequences were clustered in Operational Taxonomic Units (OTUs) at 97% similarity. Finally, datasets were rarefied to the lowest number of sequences *per* sample (34 191 reads/sample). Alpha diversity was expressed by calculating Chao1, Pielou's evenness J', and Shannon H' indices [47] and Beta-diversity was calculated with Bray-Curtis index, using Mothur.

195

196 Carbon substrate utilisation using Biolog[®] plates and metal tolerance test

197 Two types of Biolog® microtiter plates were used to assess the metabolic functional diversity of bacterial 198 communities through utilisation patterns of 62 carbon substrates (Table S1): i) EcoPlates[™] containing 31 substrates of ecological relevance (5 guilds: 9 carboxylic acids, 6 amino acids, 8 carbohydrates, 4 polymers, and 199 200 4 miscellaneous), and ii) MT2 microplates[™] allowing to test 31 chosen substrates (4 PAHs, 6 aromatic acids, 8 201 carboxylic acids, 3 amino acids, 6 carbohydrates, 3 polymers, and 1 miscellaneous). MT2 microplates™ were 202 prepared following the manufacturer's instructions (0.3 mg of carbon in each well) except for organic acids, 203 cellulose, and lignin which were 10-fold diluted because the recommended concentrations were toxic or the 204 presence of solid particles flawed absorbance measurements (data not shown). PAHs and phenolic acids were 205 dissolved in n-hexane and added to empty plates, and the solvent was eliminated by drying for 12 h in a sterile 206 hood [37]. The water-soluble substrates (20 µl) were added just before inoculation and 20 µl of sterile water 207 were added in the wells containing PAHs and phenolic acids and in the EcoPlate[™] wells.

208 Microbial inocula were prepared by diluting soil aliquots (1 g of moistened soil prepared as described above) in 209 10 ml of NaCl (0.9%) and stirring with glass beads (1.5-mm diameter) for 1 h. Supernatant was decanted for 15 210 min and diluted in NaCl (0.9%) differently for each soil, based on MPN data to obtain a similar microbial 211 abundance (from 10^2 to 10^3 cultivable bacteria ml⁻¹) for all soil. Biolog[®] plates were inoculated with 100 µl of 212 the appropriate dilution, and incubated at 24 °C in a plastic bag containing wet cotton to avoid desiccation.

213 Zinc was used to test bacterial community tolerance to metals. $Biolog^{\circ}$ substrate utilisation was measured in 214 the presence of zinc, added as $ZnCl_2$ (10 mg l⁻¹ of Zn) directly in the inoculum. This zinc concentration was 215 chosen to obtain a zinc effect without total inhibition of bacteria (preliminary experiments not shown).

216 To summarise, our experimental design tested 62 carbon substrates on 10 soils in triplicate, with or without 217 zinc addition. The absorbance at 595 nm was monitored in an spectrophotometer (SAFAS, Monaco) over 7 days 218 at t = 0, 24, 48, 72, and 96 h. Absorbance values were corrected (Abs_{corr}) after subtraction of the absorbance of 219 mean blank wells at the corresponding times. Substrate utilisation was considered positive when the corrected 220 absorbance was > 0.2. For each soil, Abs_{corr} at 96 h was used to calculate functional alpha diversity estimators, 221 i.e. metabolic richness (number of carbon substrates significantly utilised), Pielou's evenness (I'; based on the 222 relative proportions of utilisation intensity for each substrate) and Shannon (H'; based on the two previous 223 indices) diversity indices using the vegan package in R [48]. For each soil, the average well color development 224 (AWCD) was calculated over time using the formula:

$$AWCD = \frac{\sum bs_{corr}}{62}$$

where *Abs_{corr}* was the corrected absorbance of the substrate. Zinc tolerance was assessed for each soil by calculating a percentage of inhibition based on a comparison of areas below the AWCD curves between conditions with or without zinc. Percentages of inhibition were compared to soil total and available zinc concentrations (transformed to 1 when below the detection limit, i.e. 10 µg kg⁻¹) through linear regressions.

230

231 Soil metabolic profiling using the MicroResp[™] method

232 The MicroResp[™] technique was used to measure basal respiration and substrate-induced respiration (SIR) [28]. 233 Soils were loaded (four replicates) into the 96-deep-well plates, and the mean mass of each soil was measured. 234 Soils were moistened to 45% water retention capacity using sterile distilled water, and plates were pre-235 incubated 3 days at 24 °C in a plastic bag containing wet cotton to avoid desiccation. After this step, six carbon 236 sources (pyruvate, succinate, citrate, L-asparagine, D-ribose, and D-mannitol), selected according to the 237 contrasting levels of substrate utilisation in Biolog® plates among soils, were added to reach 80% of soil water 238 retention capacities. Substrates were added at 20 mg of C per well except for L-asparagine (3 mg) and citrate (5 239 mg) having lower solubility. Water was added to plates for basal respiration. Finally, deep-well plates were 240 sealed to the CO₂-trap microtiter plates and incubated in the dark at 24 °C for 4 to 9 h depending on the carbon 241 source (Table S1) and for 8 h for basal respiration. Absorbance was measured at 570 nm using an 242 spectrophotometer (SAFAS, Monaco) just before sealing to measure blank values to subtract to the final 243 absorbance. Mean basal respiration (n = 8) was measured on two independent series and 6 SIR were measured 244 on 4 replicates for the 10 soils. Absorbance values for SIR were corrected by subtracting the absorbance mean 245 (n = 4) of the corresponding soil basal respiration series, and transformed into a CO₂ concentration using the 246 following formula:

- 247
- 248

$ppm of C02 = 138.72 \times exp^{(6.7974 \times ODC)}$

where *ODc* is the corrected absorbance. Then the CO_2 concentration was standardised by dividing it by the soil quantity and the incubation time. Based on MicroResp instruction manual, ppm values of CO_2 were then expressed as quantity of carbon released as CO_2 per gram of soil per hour. For each soil, these corrected and standardised absorbance values were used to calculate the same functional diversity estimators (richness, evenness, and Shannon diversity indices) as the ones described above for Biolog data.

254

255 Statistical analyses

All statistical analyses were performed using RStudio v1.0.136. Significant differences among soils or among soil groups were assessed using Kruskal-Wallis rank sum test followed by a multiple comparison test included in the Vegan R package [48]. For principal component analysis (PCA) nine soil variables among soil physico-chemical, pollution, and texture characteristics, were used, and soil groups were made with Monte Carlo test using 1,000 iterations. For redundancy analysis (RDA), explanatory variables were chosen by using the *ordistep* R function. Permutation tests on the RDA axes and variables were performed. Multivariate analyses were carried out on

- standardised data using the ade4 [49] and vegan R packages. Linear regression was assessed by fitting to a
- 263 linear model using "Im" function. The partial least square regression (PLSr) algorithm [50, 51] in regression
- 264 mode was used to identify possible relationships between the soil physico-chemical characteristics and 96
- bacterial OTUs (each representing at least 1% of the total). The two matrices were z-normalised. The PLSr was
- 266 implemented using the mixOmics package [52, 53]. Once calibrated, the PLSr model was validated using the
- 267 leave-one-out cross-validation method. The R² and the mean squared error of prediction were then used (data
- not shown) to select the appropriate number of principal components used to implement the model. The
- correlation between bacterial OTUs and soil physico-chemical parameters was displayed in a heatmap ("cim"
- 270 function) resulting from the similarity matrix obtained from the PLSr model.

271 **Results**

272

273 Grouping of soils according to their physico-chemical characteristics

274 The physico-chemical and microbial characteristics of the ten soils are summarised in Table 1 and 2. Many 275 characteristics differed among soils and soil groups (settling pond, slag heap, control). Although one of the 276 control forest soils (ctrl-He) had a pH of 5.4 while all other soils had a slightly basic pH (from 7.0 to 8.0), no 277 statistical pH difference was observed among the soil groups. The slag heap soils differed by a high available 278 PAH concentration and intermediate values of the metal pollution index, whereas the settling pond soils had a 279 high metal pollution index with especially high Zn, Pb, Cd, and Tl concentrations and intermediate values of 280 available PAH concentrations. These two groups also presented high total PAH concentrations and high Fe 281 concentrations as compared to the control soils. The control soils were mainly characterised by the highest 282 dissolved organic carbon concentration and the lowest total PAH concentration and metal pollution index. The 283 control soils also contained higher total potassium and aluminium concentrations than the other soils, probably 284 in part due to their proportion of clay. No difference among soil groups was observed for the C/N ratio or for 285 concentrations of total organic carbon, organic matter, nitrogen, organic acids, carbohydrates, and available 286 phosphorus. Our collection of soils also presented highly variable iron concentrations, ranging from 18 g kg⁻¹ in ctrl-Di to a particularly high concentration of 452 g kg⁻¹ in sh-Uc. Concerning microbial parameters such as MPN 287 288 counts and 16S rRNA gene abundance, soil groups were not statistically different. Interestingly, the ratios of 289 fungal to bacterial abundance (18S /16S rDNA ratios) were similar between the control and slag heap soils, and 290 the lowest for settling pond soils, probably because the three soil groups had similar 18S rRNA gene copy 291 numbers while the settling pond soils tended to have higher 16S rRNA gene copy numbers (p = 0.053).

292 We performed a PCA on the basis of soil physico-chemical, pollution, and texture characteristics (Fig. 1), which 293 confirmed the distribution of the 10 soils in 3 groups based on their use. The first three components of the PCA 294 explained 76.75% of total variance, with the first (PC1), the second (PC2) and the third (PC3) components of the 295 PCA accounting for 35.3%, 28.5%, and 12.8%, respectively. The main contributing variables were the metal 296 pollution index Mi (24.8% of total variance on the considered axis) and the total potassium concentration 297 (18.9%) on the first axis, soil texture (sand: 33.1%; clay: 24.9%) and the pH (13.8%) on the second axis, and PAH 298 (46.9%) and organic carbon (30.9%) concentrations on the third axis (not shown). On PC1, control soils were 299 mainly separated from slag heap and settling pond soils by the potassium and clay fraction parameters. 300 Interestingly, settling pond soils were separated from the control and slag heap soils by the metal pollution, 301 organic carbon, and nitrogen parameters. On PC2, slag heap soils were separated from the control and settling 302 pond soils by the sand fraction and pH parameters. PAH pollution was not a discriminative parameter (10.6% of 303 the weight on PC2) and did not separate the soil groups according to their use on the PC3 axis (46.86% of the 304 weight on PC3). To test if the soil groups were robust, we performed a Monte-Carlo test, and confirmed that our grouping was statistically significant (1,000 iterations, $p = 1.998 \times 10^{-3}$). We also performed a second PCA 305 306 without the pollution characteristics (data not shown) and confirmed the separation of the soils in three groups 307 corresponding to their use (Monte Carlo test, 1000 iterations, p = 0.016).

308

309 Bacterial taxonomic diversity

310 We determined bacterial taxonomic diversity by sequencing 16S rRNA genes. After read treatment and 311 rarefaction to 34,191 reads per sample, we classified OTUs with 97% similarity taxonomically. The taxonomic 312 alpha diversity estimators based on OTU data are shown in Table 2. Interestingly, slag heap soils harboured 313 more diversified bacterial communities with higher Shannon's diversity (mean value of 6.203) and Pielou's 314 evenness (mean value of 0.791) indices than settling pond soils. Compared to the other two soil groups, control 315 soils had similar bacterial taxonomic diversity indices due to the low and high bacterial taxonomic diversity 316 indices of ctrl-Mo and ctrl-Di, respectively. We investigated bacterial taxonomic diversity at the phylum level 317 (Alpha, Beta, Gamma, and Delta classes were found for Proteobacteria). Their relative proportions in each soils 318 are presented in Fig. 2. Based on the dissimilarity distance matrix, the soils were all highly different: the 319 minimum Bray-Curtis dissimilarity index between two different soils was 0.814, while it ranged from only 0.254 320 to 0.519 for replicates of a same soil (data not shown). Phylum proportions varied among soils, with a majority 321 of Acidobacteria, Actinobacteria, and Proteobacteria. Acidobacteria were dominant (10.8% to 31.4%), except in 322 ctrl-Mo (7.1%) and sh-NM (4.3%) communities. Within the soil collection, Actinobacteria represented 10.1% to 323 25.9% of bacterial communities. Proteobacteria represented 16.2% to 35.1% of soil bacterial communities and 324 were mainly represented by Alpha-Proteobacteria (up to 18.5% in sh-Te), except in ctrl-Mo where Beta-325 Proteobacteria were the dominant proteobacterial class and represented 13.2%. We can note that ctrl-He, sp-326 MsM, and sp-RM bacterial soil communities harboured the lowest proportions of Proteobacteria but high 327 proportions of Verrucomicrobia (22.1% in ctrl-He) and of unclassified OTUs (16.5% and 17.8% in sp-MsM and 328 sp-RM, respectively). Bacterial communities from soils sharing the same use had very different compositions. 329 For example, Firmicutes represented 13.8% of the sp-MsM soil community, while they represented a maximum 330 of 0.4% of the other settling pond soil bacterial communities. By comparing soil groups, we noted that 331 Verrucomicrobia were present in a significantly higher proportion in control soils than in polluted soils $(p = 9.415 \times 10^{-5})$. On the contrary, polluted soils harboured a significantly higher proportion of Candidate 332 333 division TM7 phyla than control soils (p = 0.001). We investigated the relationship between physico-chemical 334 soil parameters and bacterial phylum proportions by searching significant linear correlations. The proportion of Actinobacteria was positively correlated to the total nitrogen concentration (Pearson, $p = 6.74 \times 10^{-4}$, r = 0.88) 335 and the soil cation-exchange capacity (Pearson, $p = 2.04 \times 10^{-3}$, r = 0.85). The proportion of Proteobacteria was 336 positively correlated to total and available PAH concentrations (Pearson, $p = 1.59 \times 10^{-3}$, r = 0.86, and 337 $p = 9.99 \times 10^{-4}$, r = 0.87, respectively), mostly due to a correlation with gamma-Proteobacteria (Pearson, p =338 0.05, r = 0.63, and p = 0.02, r = 0.71, respectively). Positive correlations between the proportion of Chloroflexi 339 340 and total zinc and lead concentrations (Pearson, $p = 2.43 \times 10^{-3}$, r = 0.84, and p = 0.03, r = 0.70, respectively) were also found. The proportion of unclassified OTUs was correlated with the total zinc concentration 341 (Pearson, $p = 5.42 \times 10^{-3}$, r = 0.80) and the metal pollution index (Pearson, $p = 1.93 \times 10^{-2}$, r = 0.72). Similarly, 342 343 the proportion of Gemmatimonadetes was positively correlated with the metal pollution index (Pearson, p = 344 1.44 x 10^{-2} , r = 0.74) and with various metals such as total nickel and cobalt concentrations (Pearson, p = 1.66 x 345 10^{-2} , r = 0.73, and p = 1.52 x 10^{-2} , r = 0.74, respectively).

The analysis at the OTU level generated 32,920 OTUs, which ranged from 2,159 (sp-RM) to 4,434 (ctrl-Di) mean 346 347 OTUs per sample. The PLSr was computed to gain insights into the soil physico-chemical parameters that drove 348 bacterial taxonomic diversity. The heatmap in Fig. S1 shows that the distribution of correlations between 349 bacterial diversity and soil characteristics was rather scattered, reflecting a great heterogeneity of the 350 relationships linking bacterial diversity to soil properties. Each soil factor selectively correlated only to a small 351 number of OTUs (10-17). In particular, as identified by the high correlation coefficients (> 0.9), the strongest 352 correlations between soil properties and OTUs concerned the pH, the clay content, and the PAH content. 353 Whereas the pH and clay seemed to correlate mostly to OTUs affiliated to the Verrucomicrobia (negatively) and 354 Firmicutes (positively), respectively, the PAH content appeared to affect a more diverse range of OTUs. The soil 355 PAH content was indeed positively and strongly correlated to the abundance of 5 OTUs affiliated to 356 Bacteroidetes (Flexibacter and Flavobacterium members) but also to that of 3 OTUs affiliated to Chloroflexi, 357 and 4 affiliated to Alpha-Proteobacteria (2 OTUs belonging to Sphingomonadaceae), Nitrospirae, and 358 Actinobacteria (one OTU affiliated to Arthrobacter). Interestingly, these groups were mostly unaffected by all 359 other soil parameters and were the only taxa affected by PAHs. Moreover, no OTU was negatively affected by 360 PAHs. With respect to the other soil characteristics such as total N, available P, and the metal concentration 361 index, we observed weaker correlations with bacterial diversity. These factors were positively correlated to one 362 group of OTUs dominated by members of Actinobacteria. The total organic carbon and sand contents were 363 slightly correlated to 2 other groups of OTUs, but with opposite effects. Organic C was positively correlated to a 364 highly diversified taxonomic group, while the sand content was positively correlated with a group of 8 365 Acidobacteria.

366

367 Metabolic functional diversity

368 The soil basal respiration and the substrate-induced respiration (SIR) measured for six different carbon sources 369 (L-asparagine, citrate, succinate, pyruvate, D-mannitol, D-ribose) using MicroResp microtiter plates are shown 370 in Fig. 3. The functional alpha diversity estimators based on MicroResp data are shown in Table 2. No statistical 371 difference among soil groups emerged for substrate richness, probably due to the low number of carbon 372 substrates we tested. The control and slag heap soils had similar functional Shannon indices and higher ones 373 than the settling pond soils. Pielou's functional evenness index was statistically different among the 3 soil 374 groups, with the highest for slag heap soils and the lowest for settling pond soils. The sum of SIRs for the 6 375 different carbon sources was used as a proxy of the soil metabolic activity (Fig. 3). Most of the carbon sources 376 were utilised by every soil, i.e. the SIR level was higher than basal respiration, except with D-mannitol that was 377 not utilised by sh-Te, and citrate that was not utilised by ctrl-Mo, ctrl-Di, sh-Ho, and sp-Mi. The highest 378 measured SIR was with pyruvate and the lowest with citrate. sp-Po showed the highest metabolic activity, 379 while sh-Ho showed the lowest, mostly due to the great differences in SIR levels with the pyruvate substrate 380 (data not shown). Kruskal-Wallis tests showed that all SIR values except with D-ribose significantly differed

among the soils. Based on soil groups, Kruskal-Wallis tests revealed that the settling pond soils had a significantly higher soil metabolic activity than the slag heap and control soils ($p = 1.93 \times 10^{-3}$), mostly due to the high metabolic activity of sp-Po. Basal respiration and SIR with pyruvate were also the highest in the settling pond soils and the lowest in the slag heap soils ($p = 1.73 \times 10^{-7}$ and $p = 9.54 \times 10^{-5}$, respectively). Finally, a positive linear correlation (Pearson, $p = 6.99 \times 10^{-5}$, r = 0.94) was found between soil metabolic activity and soil basal respiration. Additionally, these last two parameters were positively correlated with the total nitrogen concentration (Pearson, $p = 8.42 \times 10^{-3}$; r = 0.77, and $p = 1.53 \times 10^{-2}$; r = 0.74, respectively).

388 We assessed carbon substrate utilisation using Biolog microtiter plates to calculate soil metabolic richness and 389 average well colour development (AWCD) (Fig. 4). The functional alpha diversity estimators based on Biolog 390 data are shown in Table 2. Kruskal-Wallis test showed that bacterial communities from the settling pond soils 391 had a significantly lower metabolic richness and Shannon's diversity indices than bacterial communities from 392 the slag heap and control soils, while no difference among the three soil groups was observed for Pielou's 393 evenness index. The metabolic richness of the ten soils ranged from 5 (sp-RM) to 31 (ctrl-Mo) utilised carbon 394 sources, out of a total of 62 carbon sources. None of the 10 soils significantly catabolised PAHs in the 395 conditions of the experiment, probably due to a too short incubation time. Thus, except PAHs for all soils, and 396 polymers and miscellaneous substrates for sp-RM, at least one substrate from each substrate guild was utilised 397 by each soil. Among the 62 carbon substrates, protocatechuic acid, L-asparagine, Tween 40, D-mannitol, 398 succinic acid, and putrescine were the best utilised carbon substrates in the aromatic acid, amino acid, 399 polymer, carbohydrate, carboxylic acid, and miscellaneous guilds, respectively. On the contrary, for these same 400 carbon guilds, the less utilised substrates were syringic acid, glycine, alpha-cyclodextrin, D-sucrose, propionic 401 acid, and pyrocatechol, respectively. Kruskal-Wallis tests showed significant differences in metabolic richness 402 among soils, especially between ctrl-Mo and sp-RM. ctrl-Mo indeed exhibited higher total metabolic richness 403 and a wider range of utilised carboxylic acids, carbohydrates, and polymers than sp-RM. No significant 404 difference was observed among soils for miscellaneous, amino acid, and aromatic acid metabolic richness. 405 Besides, as regards the soil groups, the settling pond soils showed significantly lower total metabolic richness 406 (p = 0.007), carboxylic acid (p = 0.017) and carbohydrate (p = 0.004) utilisation richness than the slag heap and 407 control soils. Regarding utilisation of substrates from the polymer guild, the settling pond soils showed the 408 lowest metabolic richness, and the control soils the highest (p = 0.03), respectively. Finally, a positive linear correlation (Pearson correlation, $p = 1.182 \times 10^{-5}$, r = 0.959) was found between total metabolic richness and 409 410 AWCD.

411

412 Zinc tolerance

Soil microbial zinc tolerance was assessed using Biolog microtiter plates by adding $ZnCl_2$ solution, and then percentages of inhibition due to zinc addition were calculated. Fig. 5 shows the linear regression between the soil available zinc concentration and the percentage of inhibition of carbon substrate utilisation. A similar linear regression was found with the total zinc concentration (p = 3.975×10^{-5} , R² = 0.45). The settling pond soils had the highest available zinc concentration, and their metabolic activity was the least inhibited by zinc addition.

- 418 The ctrl-He soil also had a high available zinc concentration (38.1 mg kg⁻¹), due to its low pH (5.4, Table 1)
- 419 leading to a higher MTE availability and a lower zinc inhibition of metabolic activity (46.2%). In comparison, ctrl-
- 420 Mo and ctrl-Di had neutral pH values (7.0 and 7.3, respectively), a low available zinc concentration (< 10 and 11
- 421 μg kg⁻¹, respectively) and a higher zinc inhibition of metabolic activity (49.3 and 54.7%, respectively).
- 422

423 Effects of the soil characteristics on taxonomic and metabolic functional diversity

424 We investigated the effects of soil environmental characteristics on bacterial taxonomic diversity (Fig. 6A), and 425 MicroResp (Fig. 6B) and Biolog (Fig. 6C) metabolic functional diversity through redundancy analyses. We 426 performed three RDAs using soil environmental characteristics as explanatory variables, and the taxonomic and 427 functional diversity indices of the soil microbial communities (Table 1 and 2) as response variables. The 428 explanatory variables that best explained the variation of diversity indices were selected by backward selection 429 to maximise the percentage of explained variability leading to at least 99.3% of total variation. The 430 supplemental Table S2 provides p values and coefficients for axes and soil properties significance. Only the first 431 axis (p < 0.005) in Fig. 6A and 6B was significant, and so were the first two axes (p < 0.005 and p < 0.02) in Fig. 432 6C. Among the explanatory variables (metal pollution index, PAH concentration, pH, and sand fraction), only 433 the sand fraction had a significant effect (first axis; Fig. 6A) on bacterial taxonomic diversity (p < 0.01, F =16.30). This relationship was confirmed by significant positive linear correlations between the proportions of 434 soil sand fractions and Chao1 ($p = 5.97 \times 10^{-3}$, r = 0.80), Shannon diversity ($p = 3.50 \times 10^{-3}$, r = 0.82) and Pielou's 435 436 evenness (p = 0.01, r = 0.76) indices. Total organic carbon and nitrogen concentrations were the two 437 significantly explanatory variables (p < 0.005, F = 8.71, and p < 0.01, F = 4.46, respectively) accounting for the 438 variation of MicroResp soil metabolic diversity (Fig. 6B). The total nitrogen concentration showed negative 439 relationships with Shannon diversity (H') and Pielou's evenness (J) indices, confirmed by significant Pearson 440 linear correlations (p = 1.50×10^{-4} , r = -0.92, and p = 3.08×10^{-3} , r = 0.83, respectively). The metal pollution 441 index and the total nitrogen concentration were the strongest determinants of both axes (Fig. 6C) and had a 442 significant effect on Biolog metabolic diversity indices (p < 0.01, F = 7.77, and p < 0.05, F = 4.71, respectively). 443 The metal pollution index showed a negative relationship with metabolic richness and Shannon diversity index 444 (H') confirmed by slightly significant negative linear correlations (p = 0.06, r = -0.61 for both of them). Total 445 nitrogen exhibited a negative relationship with Pielou's evenness index (J), confirmed by a significant linear 446 correlation (p = 0.02, r = -0.73).

447 **Discussion**

448

449 We studied ten different soils presenting metal and/or PAH pollution gradients and exhibiting a wide range of 450 anthropisation levels. Although the ten soils came from various sites with different types and levels of plant 451 colonization, they mostly clustered in three groups based on their physico-chemical properties and pollution 452 levels. These three groups reflected their use, whether industrial or low anthropised. Settling pond soils were 453 highly polluted with MTEs, with higher contents of total organic carbon and nitrogen and a siltier texture than 454 the soils from the other groups. Slag heap soils were the sandiest and the most PAH-polluted soils, and also 455 exhibited middle-high metal pollution: mostly MTEs for sh-Te, sh-NM, and sh-Ho, while sh-Uc contained a very 456 high iron concentration. The control soils were the least metal- and PAH-polluted ones; they were also the 457 most clayey soils, and had the highest dissolved organic carbon and potassium contents. Apart from the 458 identification of well known edaphic factors impacting the microbial diversity, we also aimed at identifying the 459 impact of pollutants on the microbial community composition. The taxonomic and metabolic functional 460 diversity of the microbial communities inhabiting the ten soils was characterized. We studied bacterial 461 taxonomic diversity based on 16S rRNA gene sequencing, and functional diversity based on measurements of 462 the utilisation of various carbon substrates involved in different functions to estimate the impact of land use on 463 microbial community diversity and functions. We focused on functions related to the carbon cycle, one of the 464 most important biogeochemical cycles for soil functioning [54, 55]. Although the taxonomic and functional 465 diversity of soil microorganisms is extensively studied to better understand ecosystem functioning, the real link 466 between taxon occurrence and functional richness remains poorly understood in most ecosystems [56], 467 particularly for microbial communities. Various authors suggest that the functions related to carbon 468 mineralisation are redundant [57, 58]. Besides, the loss of species or the modification of microbial community 469 composition does not necessarily induce a loss of functions. In order to better understand the effect of 470 pollution on microbial communities, it is thus important to study both taxonomic and functional diversity, using 471 a combination of tools.

One of the objectives of this study was to identify soil parameters mostly shaping microbial community diversity in the ten soils. Although, some of the measured diversity indicators seemed constant in all soils, our data revealed that both pollutant and physico-chemical parameters strongly affected part of the taxonomic and functional structures.

476

477 Non-impacted microbial parameters

We noted few variations in bacterial community composition among our soils, with all the major phyla present in each soil, and no phylum significantly dominant or under-represented in highly anthropised soils. The taxonomic diversity indices of the polluted soils were not significantly different from those of the control soils. Previous studies reported similar trends with no difference among the bacterial taxonomic diversity indices of historically mining-impacted *vs.* unpolluted sediments [59]. Indeed, long-term adaptation and resilience [60] of the bacterial communities, being as diverse as in control soil [22] or over a metal polluted gradient [61], was 484 previously observed. Additionally, except for sp-RM that exhibited a very strong decrease of metabolic 485 richness, at least one substrate from each carbon substrate guild from Biolog plates was utilised by each soil. In 486 some cases, metabolic potentials were not affected by metal pollution as shown in three forest soils gradually 487 polluted by Zn and Cd [62]. The differences of substrate-induced respiration observed among soils were less 488 pronounced using the MicroResp than the Biolog method, and were essentially linked to the soil basal 489 respiration rates as previously observed [63]. Contrarily to many studies [63, 64, 65], basal respiration rates and 490 substrate-induced respiration were mostly not affected by pollution (metals and PAH) level. Additionally to the 491 bacterial activity, MicroResp can also integrate fungal respiration. Although, fungi are major actors of the 492 carbon cycle [66], their relative abundances (ratio between 18S and 16S rDNA copy numbers) were not 493 correlated with the respiration rates because the three soil groups had similar fungal density (18S rDNA copies.g⁻¹ dw soil). Moreover, even if fungi harbor high metal tolerance [67] no link between fungal 494 495 abundances, respiration rates and soil pollution was highlighted.

496

497 Impact of metals on microbial diversity

The relative toxicity of a pollutant depends more on its available content than on its total content. The bioavailability of MTEs in these soils was relatively low since most of them had a pH value above 7. Only soil He with a pH of 5.4 had a relatively higher MTE availability than the other soils of the same group, although it was not contaminated. Except for this soil, using the relative metal pollution index or the available fraction of MTEs such as Zn showed the same tendency, that is why we mainly used the former to compare the soils.

503 The metal pollution index and/or the total zinc content were positively correlated with the relative proportion 504 of unclassified OTUs and Gemmatimonadetes and Chloroflexi phyla (correlation with available Zinc content 505 have p value of 0.077, 0.088 and 0.120, respectively). Unclassified OTUs may represent extensive unexplored 506 bacterial diversity that may have original properties allowing them to survive and grow in strongly metal-507 contaminated environments. Gemmatimonadetes are commonly found in heavy-metal contaminated soils [68]. 508 A positive relationship between the relative proportion of Gemmatimonadetes or Chloroflexi and metal 509 pollution was shown in lake sediment historically polluted with zinc and lead [69] and in the soil of an 510 abandoned Pb-Zn mine [70]. The Chloroflexi group is known to have a high stress tolerance ability [71], which 511 could explain its development in our metal-polluted soils. At the OTU level, a few OTUs were positively 512 correlated to the metal pollution index, and these OTUs were also positively correlated to the P and N 513 contents. These OTUs were affiliated to Actinobacteria members, closely related to the AKIW543 clone, the 514 Acidimicrobineae and Aeromicrobium genera, and to Firmicutes affiliated to the Bacillus genus. Very little 515 information is available about the two former but few strains belonging to Aeromicrobium and Bacillus genera 516 have already been isolated from former mining sites [72] and from heavy-metal contaminated soil [73].

517 The metal pollution index and/or the total and available zinc contents were negatively correlated to the 518 richness and diversity of the substrates utilised in Biolog plates, suggesting that functional diversity (number of 519 metabolic functions) was reduced in metal-polluted soils [74]. Similarly, when comparing soils sampled at 520 different distances from a site historically metal-polluted with As, Cu, and Pb, Boshoff et al. [29] found lower functional diversity in the soils closest to the contamination source. It may result from the replacement of sensitive species by more resistant ones having different and less versatile metabolic properties [75] maybe due to an energy cost balance between metal stress survival mechanisms and substrate utilisation efficiency [76]. Functional diversity may indicate the capacity of the community to adapt its metabolism, relative composition, and size to various changes in environmental conditions [77]. In our study, the low metabolic richness of the bacterial community in the metal-polluted soils may have disturbed carbon recycling and soil functioning.

Additionally, using Biolog for tolerance tests, lower zinc sensitivity (or higher tolerance) of the bacterial community inhabiting high metal-contaminated soils was shown as compared to low-polluted or unpolluted soils, suggesting an adaptation of communities to metal stress, as previously observed [78]. It is well accepted that microorganisms coming from highly polluted environments are more tolerant to pollution than those coming from less polluted environments [21].

533 Among our soil collection, sh-Uc had the highest iron concentration and relatively low concentrations of other 534 metals, resulting in a middle-high metal pollution index. This soil had similar taxonomic but slightly higher 535 functional diversity indices as compared to other soils presenting a middle-high metal pollution index. Unlike 536 other metals such as Cd or Pb, Fe is an essential element for almost all living organisms and is involved in 537 numerous metabolic processes. Moreover, due to its extremely high total concentration in sh-Uc, iron was 538 probably mostly present as insoluble oxides and unavailable forms. This finding suggests that although iron was 539 present in a very high total concentration in sh-Uc, it had a lower effect on microbial diversity than other 540 metals such as Zn, Cd, or Pb.

541

542 Impact of PAH on microbial diversity

543 We highlighted various correlations between PAH pollution and the proportions of some phyla or OTU 544 abundance, but found no relationship with functional metabolic diversity, suggesting a limited impact of PAH 545 on microbial community carbon-cycle related functions in our soils. Total and available PAH contents were both 546 positively correlated to the proportions of Proteobacteria (mostly Gamma-Proteobacteria) and Nitrospirae 547 phyla, even if in the literature the Nitrospirae phylum seemed to be negatively impacted by PAH content and 548 bioavailability [79]. Proteobacteria and Gamma-Proteobacteria were previously found in greater proportions in 549 pyrene-amended soil [80] and can be involved in PAH degradation [81]. The abundance of some OTUs was also 550 positively correlated to the total PAH pollution level, but was not affected by all other soil parameters, 551 suggesting an advantage conferred in the presence of PAHs. Most of them were found predominant in PAH-552 polluted environments, i.e. bacteria belonging to Flavobacterium [82], Flexibacter [83], Sphingomonadaceae 553 [84] and Chloroflexi [85] were detected or isolated from PAH-contaminated soils. Firstly, PAH pollution could 554 have contributed to shape bacterial communities by favouring species able to metabolise PAHs. Secondly, in 555 our soils, due to ageing, available PAH concentrations were low as compared to total concentrations and did 556 not negatively affect taxonomic or functional diversity at the community level, indicating no major toxicity 557 impact of PAHs on bacterial diversity.

558

559 Impact of physico-chemical parameters on microbial diversity

Apart from the soil pollutants, microbial communities can be affected by physico-chemical parameters. It is well known that the pH is one of the major parameters that shapes microbial community composition [30, 86]. Except one control soil (ctrl-He), our soils were chosen with a similar slightly alkaline pH to avoid a too high influence of this parameter that could have hidden the structuring impact of other parameters. Thus, the pH has little impact on taxonomic and functional diversity, although the proportions of few OTUs mostly affiliated to Verrucomicrobial DA101 members and commonly found as dominant bacteria in soils [87] increased when the pH decreased.

567 The soil texture, and especially the sand fraction proportion, was the strongest determinant of bacterial 568 taxonomic diversity in our soils, but no link with metabolic functional diversity was highlighted. The relative 569 proportion of the soil sandy fraction was indeed positively correlated with taxonomic richness, evenness, and 570 Shannon diversity indices, and the relative proportion of Alpha-Proteobacteria [88], whereas the relative 571 proportions of silt and clay negatively influenced the richness and evenness indices. Fine-textured soil can 572 protect microbes from predation by protozoans, reduce variations in water availability, and maintain higher 573 microbial abundance [89]. Conversely, coarse-textured soils allow for a better aeration and water circulation 574 providing various microhabitats and supporting growth of diverse microorganisms adapted to limited nutrient 575 conditions or able to use a wider range of substrates [88]. Similarly to our results, Chau et al. [90] found a 576 positive linear relationship between sand content and taxonomic richness, while Kandeler et al. [91] found that 577 the small-size fraction promoted bacterial diversity. These contradictory results could be related to soil 578 fractionation techniques [90], as we based textures on particle granulometry, not on mineralogy (i.e. the sandy 579 fraction could contain sand but also other particles). Finally, the texture may affect pollutant availability, which 580 could indirectly affect microbial diversity.

581 While the soil total organic carbon content was not the main parameter explaining taxonomic diversity, it was 582 negatively correlated to the functional diversity indices, calculated from MicroResp and Biolog data, to the 583 utilisation level of 6 substrates (phthalic, ferulic, succinic, and aconitic acids, Tween 40, and phenylethylamine), 584 and to the number of polymer, carboxylic acid and miscellaneous substrates utilised in Biolog microtiter plates. 585 We noticed that in the redundancy analysis, the total organic carbon content co-varied with the metal pollution 586 index, as previously shown by Valsecchi et al. [92] on a collection of 16 soils. As explained above, metal 587 pollution may have reduced the functional capacities of the microbial community. The resulting lower microbial 588 functional diversity and activity may have reduced organic carbon recycling in the soils, resulting in an 589 accumulation of total organic carbon and organic matter. Similarly, rates of organic matter (i.e. litter or plant 590 polymer such as cellulose) degradation were lower in metal-contaminated than unpolluted soils [93, 94]. This 591 finding could be explained by a decrease of enzymatic activities involved in the C cycle in metal-polluted soils 592 [91]. Moreover, Lucisine et al. [95] also explained that accumulation of organic matter at the surface of 593 brownfield soil could be partly due to the lack of endogenous metal-sensitive earthworm species limiting 594 organic matter incorporation in the deeper soil layers.

595 We observed that the total nitrogen content was positively correlated with soil basal respiration, with ribose-, 596 pyruvate-, and asparagine-induced respiration, and with 16S rRNA gene copy numbers. These relationships 597 suggest that in some of our soils, nitrogen present in very low concentrations was one of the limiting factors of 598 microbial growth and activity. N limitation can indeed occur in some cases depending on the quality of the 599 available carbon substrate [96, 97]. Additionally, the relative proportions of the Actinobacteria phylum and of a 600 few OTUs affiliated to Actinobacteria and Firmicutes were positively correlated to the total nitrogen content. 601 As discussed above, these OTUs were also positively correlated to the metal pollution index indicating a 602 simultaneous influence of various soil parameters on microbial community composition. As explained for total 603 organic carbon, metal pollution partly inhibits soil enzyme activities and organic matter decomposition [98], so 604 that the total nitrogen content potentially increases in the most polluted soils. Moreover, several strains of 605 Bacilli isolated from soils are known as nitrogen-fixing bacteria [99] and could contribute to increase the 606 nitrogen content. On the contrary, Bacteroidetes phyla and a few OTUs affiliated to Acidobacteria were 607 negatively correlated with the total nitrogen content. Acidobacteria are often classified as oligotrophs [100], 608 which are favoured in nutrient-poor environments. Moreover, a negative relationship between the total 609 nitrogen content and the functional evenness index was mostly explained by a higher mineralising activity in a 610 few substrates. Nitrogen may have impacted bacterial community composition, favouring the development of 611 certain species with higher metabolic activity for certain specific substrates.

612

613 **Conclusions**

614

615 We studied a collection of 10 soils presenting different anthropisation levels. Based on their physico-chemical 616 characteristics and on metal and PAH pollution levels, these soils were gathered in 3 groups corresponding to 617 their use, i.e. either industrial (slag heap and settling pond) or low anthropised (control soil). Although 618 taxonomic diversity largely varied among soils, the occurrence of a few bacterial phyla and OTUs was 619 influenced by metal and PAH pollution as well as by the pH, texture, and the total N content. Furthermore, 620 metal pollution, particularly zinc concentrations, seems to have selected more metal-tolerant communities but 621 with reduced metabolic functional diversity and activity. Apart from metal pollution, the N content appeared to 622 be one of the main factors limiting microbial activity in our soils. These reduced activity and metabolic 623 functional diversity may have contributed to an accumulation of organic matter in the most metal-polluted 624 soils. Considering all abiotic and biotic soil characteristics, especially metabolic functional diversity, we are now 625 able to rank our soils based on their impact on microbial communities (i.e. control soils have the lowest effect, 626 while settling pond soils have the highest). Although it would be very interesting to study a wider range of soil 627 to draw more generalizable conclusions, the study of 10 soils already allowed us to identify the drivers of 628 microbial diversity at both the taxonomic and functional level.

629

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638 Figure captions

639

640 Table 1 | Physico-chemical characteristics of the ten soils. Parameters were measured one time on one mean 641 sample. For soil groups, values are means (n=3 for ctrl and sp soils, and n=4 for sh soil) \pm standard error of the 642 mean. Significant results of Kruskal-Wallis tests (p < 0.05), testing differences among the 3 soil groups, are 643 indicated by different letters, and no letter appears when non-significant differences were found. For statistical 644 analyses, values lower than the detection limits of the method were replaced by 1, except for available 645 phosphorus for which the value was replaced by 0.05. The metals used for calculating the metal pollution index 646 (Mi) are indicated by a star. Abbreviations of soil names and groups: Di: Dieulouard, Ho: Homécourt, He: 647 Hémilly, MsM: Mont St Martin, Mo: Montiers, RM: Russange-Micheville, NM: Neuves Maisons, Po: Pompey, Te: 648 Terville, Uc: Uckange; ctrl: control, sh: slag heap, sp: settling pond. Abbreviations of soil characteristics: PAH: 649 polycyclic aromatic hydrocarbons, CEC: cation exchange capacity.

650

651**Table 2 | Microbial, taxonomic and functional characteristics of the ten soils.** Values are means $(n = 3 \text{ or } 4) \pm$ 652standard error of the mean. For soil groups, values are means $(n=3 \text{ for ctrl and sp soils, and } n=4 \text{ for sh soil}) \pm$ 653standard error of the mean. Significant results of Kruskal-Wallis tests (p < 0.05), testing differences among the 3654soil groups, are indicated by different letters, and no letter appears when non-significant differences were655found.

656

Figure 1 | Principal Component Analysis (PCA) and correlation circle based on the physico-chemical, pollution, and texture characteristics of the ten soils. On the PCA: Soils (Di: Dieulouard, Ho: Homécourt, He: Hémilly, MsM: Mont St Martin, Mo: Montiers, RM: Russange-Micheville, NM: Neuves Maisons, Po: Pompey, Te: Terville, Uc: Uckange) were grouped in three types according to their industrial history (sp: settling pond; sh: slag heap; ctrl: control). On the correlation circle: N: total nitrogen; C_org: total organic carbon; P_Olsen: available phosphorus; Mi: metal pollution index; PAH: sum of 16 regulatory PAHs; K_t: total potassium; sand and clay: proportions of the sand and clay fractions, respectively.

664

Figure 2 | Taxonomic composition of the bacterial communities of the ten soil samples. Values are means
(n = 3), and error-bars represent standard errors of the mean for each soil. The group named Others included
OTUs affiliated to the following phyla: BD1-5, Chlamydiae, Chlorobi, Cyanobacteria, Deinococcus-Thermus,
Fusobacteria, Spirochaetes, WCHB1-60, Synergistetes, and Candidate divisions BRC1, OD1, OP10, SR1, TG-1,
TM6, WS3, and WS6. Statistical differences among phylum proportions in the soil groups are indicated by
different letters (Kruskal-Wallis test, p < 0.05).

Figure 3 | Cumulated substrate-induced respiration (SIR) in the presence of six carbon sources, and basal soil respiration measured using the MicroResp method. Values are means (n = 4 for SIR and n = 8 for basal respiration), and error bars represent standard errors of the mean for each soil. Statistical differences among the soil groups are indicated by different letters (Kruskal-Wallis test, p < 0.05).

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Figure 4 | Catabolic richness and average well color development (AWCD) measured after 96h of incubation using Biolog microtiter plates. Values are means (*n* = 3), and error bars represent standard errors of the mean for each soil. Sixty-two carbon sources were tested: 4 PAHs, 6 aromatic acids, 9 amino acids, 7 polymers, 14 carbohydrates, 17 carboxylic acids, and 5 miscellaneous (for details see Supplemental Table S1). Statistical differences among the soil groups are indicated by different letters (Kruskal-Wallis test, p > 0.05). As none of the PAH compounds was significantly degraded, they were excluded from the statistical analysis.

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Figure 5 | Linear regression between the inhibition rate of carbon substrate utilisation measured using Biolog microtiter plates (n = 3) with and without zinc addition, and available zinc concentrations in the ten soils. The blue line represents the linear model of regression (y = -0.0006x + 60.98; where y is the inhibition rate and x is the available zinc content of the soil). The dark grey area represents the confidence interval, and the light grey area represents the prediction interval (95%). Settling pond, slag heap, and control soils are represented by red, green, and blue dots, respectively.

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Figure 6 | Redundancy analysis (RDA) of soil bacterial community diversity (A), MicroResp soil catabolic diversity (B), and Biolog catabolic diversity (C), each of them constrained by soil environmental characteristics.

694 For each RDA, the explanatory variables that best explained the variation of diversity indices (response 695 variables) were selected by backward selection. Explanatory and response variables are in red and black, 696 respectively. Abbreviations of explanatory variables: N: total nitrogen; C_org: total organic carbon; Mi: metal 697 pollution index; PAH: sum of 16 regulatory PAHs; K_t: total potassium; sand: proportion of the sandy fraction. 698 Abbreviations of response variables: S: richness; Chao1: extrapolated richness; H: Shannon's diversity index; J: 699 Pielou's evenness index; 16S: index for bacterial taxonomic diversity. MR: index for MicroResp soil catabolic 700 diversity. BLG: index for Biolog catabolic diversity. Abbreviations for soils: Di: Dieulouard, Ho: Homécourt, He: 701 Hémilly, MsM: Mont St Martin, Mo: Montiers, RM: Russange-Micheville, NM: Neuves Maisons, Po: Pompey, Te: 702 Terville, Uc: Uckange). Soils were grouped in three types according to their history (sp: settling pond; sh: slag 703 heap; ctrl: control). The p values and F coefficients, for axes and soil properties, are provided in the 704 supplemental Table S2.

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707 Supplemental material

- 708 Table S1 | Substrates used in the Biolog and MicroResp experiments.
- 709

Table S2 | The p values and F coefficients for axes and soil properties for the redundancy analyses presented
 on the Figure 6.

- 712
- Figure S1 | Heatmap based on the similarity matrix obtained from the PLSr analysis (implemented using 5
- 714 **components).** The heatmap depicts the positive (red) and negative (blue) correlations between OTU
- abundances (the predictor matrix) and the soil physico-chemical properties (the response variables) investigated
- in this study. Dendrograms indicate the level of similarity among the clusters present in the heatmap. The
- 717 *hierarchical clustering was based on the Euclidean distance and the complete linkage method.*

718	Refer	ences
719	_	
720	1.	(2016) BASOL. http://basol.developpement-durable.gouv.fr/. Accessed 11 Sep 2017
721 722 723	2.	Rachwał M, Magiera T, Wawer M (2015) Coke industry and steel metallurgy as the source of soil contamination by technogenic magnetic particles, heavy metals and polycyclic aromatic hydrocarbons. Chemosphere 138:863–873. doi: 10.1016/j.chemosphere.2014.11.077
724 725 726	3.	Joimel S, Cortet J, Jolivet CC, et al (2016) Physico-chemical characteristics of topsoil for contrasted forest, agricultural, urban and industrial land uses in France. Sci Total Environ 545:40–47. doi: 10.1016/j.scitotenv.2015.12.035
727 728 729	4.	Wilcke W (2000) SYNOPSIS Polycyclic Aromatic Hydrocarbons (PAHs) in Soil — a Review. J Plant Nutr Soil Sci 163:229–248. doi: 10.1002/1522-2624(200006)163:3<229::AID- JPLN229>3.0.CO;2-6
730 731	5.	Dhelft P (1994) Épuration du gaz de haut fourneau. Tech L'ingénieur Métaux Ferr Élabor Métal Prim TIB366DUO. (ref. article : m7422)
732 733	6.	Johnsen AR, Wick LY, Harms H (2005) Principles of microbial PAH-degradation in soil. Environ Pollut 133:71–84. doi: 10.1016/j.envpol.2004.04.015
734 735 736	7.	Janssen CR, Heijerick DG, De Schamphelaere KAC, Allen HE (2003) Environmental risk assessment of metals: tools for incorporating bioavailability. Environ Int 28:793–800. doi: 10.1016/S0160-4120(02)00126-5
737 738 739	8.	Kim R-Y, Yoon J-K, Kim T-S, et al (2015) Bioavailability of heavy metals in soils: definitions and practical implementationa critical review. Environ Geochem Health 37:1041–1061. doi: 10.1007/s10653-015-9695-y
740 741	9.	Cerniglia CE (1992) Biodegradation of polycyclic aromatic hydrocarbons. Biodegradation 3:351–368. doi: 10.1007/BF00129093
742 743	10.	Hatzinger PB, Alexander M (1995) Effect of aging of chemicals in soil on their biodegradability and extractability. Environ Sci Technol 29:537–545. doi: 10.1021/es00002a033
744 745 746 747	11.	Cébron A, Faure P, Lorgeoux C, et al (2013) Experimental increase in availability of a PAH complex organic contamination from an aged contaminated soil: consequences on biodegradation. Environ Pollut Barking Essex 1987 177:98–105. doi: 10.1016/j.envpol.2013.01.043
748 749	12.	Abdu N, Abdullahi AA, Abdulkadir A (2017) Heavy metals and soil microbes. Environ Chem Lett 15:65–84. doi: 10.1007/s10311-016-0587-x
750 751	13.	Andreoni V, Cavalca L, Rao MA, et al (2004) Bacterial communities and enzyme activities of PAHs polluted soils. Chemosphere 57:401–412. doi: 10.1016/j.chemosphere.2004.06.013
752 753 754	14.	Grant RJ, Muckian LM, Clipson NJW, Doyle EM (2007) Microbial community changes during the bioremediation of creosote-contaminated soil. Lett Appl Microbiol 44:293–300. doi: 10.1111/j.1472-765X.2006.02066.x

- Sawulski P, Clipson N, Doyle E (2014) Effects of polycyclic aromatic hydrocarbons on microbial
 community structure and PAH ring hydroxylating dioxygenase gene abundance in soil.
 Biodegradation 25:835–847. doi: 10.1007/s10532-014-9703-4
- Sutton NB, Maphosa F, Morillo JA, et al (2013) Impact of long-term diesel contamination on
 soil microbial community structure. Appl Environ Microbiol 79:619–630. doi:
 10.1128/AEM.02747-12
- 761 17. Kelly JJ, Häggblom M, Tate RL (1999) Changes in soil microbial communities over time
 762 resulting from one time application of zinc: a laboratory microcosm study. Soil Biol Biochem
 763 31:1455–1465. doi: 10.1016/S0038-0717(99)00059-0
- 18. Moffett BF, Nicholson FA, Uwakwe NC, et al (2003) Zinc contamination decreases the bacterial
 diversity of agricultural soil. FEMS Microbiol Ecol 43:13–19. doi: 10.1111/j.15746941.2003.tb01041.x
- 767 19. Cébron A, Norini M-P, Beguiristain T, Leyval C (2008) Real-Time PCR quantification of PAH-ring
 768 hydroxylating dioxygenase (PAH-RHDα) genes from Gram positive and Gram negative bacteria
 769 in soil and sediment samples. J Microbiol Methods 73:148–159. doi:
 770 10.1016/j.mimet.2008.01.009
- Diaz-Ravina M, Baath E (1996) Development of Metal Tolerance in Soil Bacterial Communities
 Exposed to Experimentally Increased Metal Levels. Appl Environ Microbiol 62:2970–2977.
- Blanck H (2002) A Critical Review of Procedures and Approaches Used for Assessing Pollution Induced Community Tolerance (PICT) in Biotic Communities. Hum Ecol Risk Assess Int J
 8:1003–1034 . doi: 10.1080/1080-700291905792
- Bourceret A, Cébron A, Tisserant E, et al (2016) The Bacterial and Fungal Diversity of an Aged
 PAH- and Heavy Metal-Contaminated Soil is Affected by Plant Cover and Edaphic Parameters.
 Microb Ecol 71:711–724. doi: 10.1007/s00248-015-0682-8
- 23. Lindgren JF, Hassellöv I-M, Nyholm JR, et al (2017) Induced tolerance in situ to chronically PAH
 exposed ammonium oxidizers. Mar Pollut Bull 120:333–339. doi:
 10.1016/j.marpolbul.2017.05.044
- 782 24. Thavamani P, Malik S, Beer M, et al (2012) Microbial activity and diversity in long-term mixed
 783 contaminated soils with respect to polyaromatic hydrocarbons and heavy metals. J Environ
 784 Manage 99:10–17. doi: 10.1016/j.jenvman.2011.12.030
- 25. Lu M, Xu K, Chen J (2013) Effect of pyrene and cadmium on microbial activity and community
 structure in soil. Chemosphere 91:491–497. doi: 10.1016/j.chemosphere.2012.12.009
- 787 26. Markowicz A, Cycon M, Piotrowska-Seget Z (2016) Microbial Community Structure and
 788 Diversity in Long-term Hydrocarbon and Heavy Metal Contaminated Soils. Int J Environ Res
 789 10:321–332. doi: 10.22059/ijer.2016.57792
- Zak JC, Willig MR, Moorhead DL, Wildman HG (1994) Functional diversity of microbial
 communities: A quantitative approach. Soil Biol Biochem 26:1101–1108. doi: 10.1016/00380717(94)90131-7
- 28. Campbell CD, Chapman SJ, Cameron CM, et al (2003) A rapid microtiter plate method to
 measure carbon dioxide evolved from carbon substrate amendments so as to determine the

- physiological profiles of soil microbial communities by using whole soil. Appl Environ Microbiol69:3593–3599
- 29. Boshoff M, De Jonge M, Dardenne F, et al (2014) The impact of metal pollution on soil faunal
 and microbial activity in two grassland ecosystems. Environ Res 134:169–180. doi:
 10.1016/j.envres.2014.06.024
- Fierer N, Jackson RB (2006) The diversity and biogeography of soil bacterial communities. Proc
 Natl Acad Sci 103:626–631. doi: 10.1073/pnas.0507535103
- 802 31. Nacke H, Thürmer A, Wollherr A, et al (2011) Pyrosequencing-based assessment of bacterial
 803 community structure along different management types in German forest and grassland soils.
 804 PloS One 6:e17000. doi: 10.1371/journal.pone.0017000
- 805 32. Bissett A, Richardson AE, Baker G, Thrall PH (2011) Long-term land use effects on soil
 806 microbial community structure and function. Appl Soil Ecol 51:66–78. doi:
 807 10.1016/j.apsoil.2011.08.010
- Baize D (2000) Teneurs totales en « métaux lourds » dans les sols français : résultats généraux
 du programme ASPITET. Courr Environ Inra 39.
- 810 34. Cennerazzo J, de Junet A, Audinot J-N, Leyval C (2017) Dynamics of PAHs and derived organic
 811 compounds in a soil-plant mesocosm spiked with 13C-phenanthrene. Chemosphere
 812 168:1619–1627. doi: 10.1016/j.chemosphere.2016.11.145
- 813 35. Reid BJ, Stokes JD, Jones KC, Semple KT (2000) Nonexhaustive Cyclodextrin-Based Extraction
 814 Technique for the Evaluation of PAH Bioavailability. Environ Sci Technol 34:3174–3179. doi:
 815 10.1021/es990946c
- 36. Jones DL, Willett VB (2006) Experimental evaluation of methods to quantify dissolved organic
 nitrogen (DON) and dissolved organic carbon (DOC) in soil. Soil Biol Biochem 38:991–999. doi:
 10.1016/j.soilbio.2005.08.012
- Thomas F, Cébron A (2016) Short-Term Rhizosphere Effect on Available Carbon Sources,
 Phenanthrene Degradation, and Active Microbiome in an Aged-Contaminated Industrial Soil.
 Syst Microbiol 92. doi: 10.3389/fmicb.2016.00092
- 82238.Lueders T, Wagner B, Claus P, Friedrich MW (2004) Stable isotope probing of rRNA and DNA823reveals a dynamic methylotroph community and trophic interactions with fungi and protozoa824in oxic rice field soil. Environ Microbiol 6:60–72. doi: 10.1046/j.1462-2920.2003.00535.x
- 825 39. Felske A, Akkermans ADL, Vos WMD (1998) Quantification of 16S rRNAs in Complex Bacterial
 826 Communities by Multiple Competitive Reverse Transcription-PCR in Temperature Gradient Gel
 827 Electrophoresis Fingerprints. Appl Environ Microbiol 64:4581–4587.
- 40. Cébron A, Beguiristain T, Bongoua-Devisme J, et al (2015) Impact of clay mineral, wood
 sawdust or root organic matter on the bacterial and fungal community structures in two aged
 PAH-contaminated soils. Environ Sci Pollut Res Int 22:13724–13738. doi: 10.1007/s11356-0154117-3
- 41. Thion C, Cebron A, Beguiristain T, Leyval C (2012) Long-term in situ dynamics of the fungal
 communities in a multi-contaminated soil are mainly driven by plants. Fems Microbiol Ecol
 834 82:169–181. doi: 10.1111/j.1574-6941.2012.01414.x

- 42. Muyzer G, de Waal EC, Uitterlinden AG (1993) Profiling of complex microbial populations by
 denaturing gradient gel electrophoresis analysis of polymerase chain reaction-amplified genes
 coding for 16S rRNA. Appl Environ Microbiol 59:695–700.
- 43. Caporaso JG, Lauber CL, Walters WA, et al (2011) Global patterns of 16S rRNA diversity at a
 depth of millions of sequences per sample. Proc Natl Acad Sci U S A 108 Suppl 1:4516–4522.
 doi: 10.1073/pnas.1000080107
- Kozich JJ, Westcott SL, Baxter NT, et al (2013) Development of a Dual-Index Sequencing
 Strategy and Curation Pipeline for Analyzing Amplicon Sequence Data on the MiSeq Illumina
 Sequencing Platform. Appl Environ Microbiol 79:5112–5120. doi: 10.1128/AEM.01043-13
- Schloss PD, Westcott SL, Ryabin T, et al (2009) Introducing mothur: Open-Source, PlatformIndependent, Community-Supported Software for Describing and Comparing Microbial
 Communities. Appl Environ Microbiol 75:7537–7541. doi: 10.1128/AEM.01541-09
- 84746.Edgar RC, Haas BJ, Clemente JC, et al (2011) UCHIME improves sensitivity and speed of848chimera detection. Bioinforma Oxf Engl 27:2194–2200. doi: 10.1093/bioinformatics/btr381
- 47. Hill TCJ, Walsh KA, Harris JA, Moffett BF (2003) Using ecological diversity measures with
 bacterial communities. FEMS Microbiol Ecol 43:1–11. doi: 10.1111/j.15746941.2003.tb01040.x
- 48. Oksanen J, Blanchet FG, Friendly M, et al (2017) vegan: Community Ecology Package. R
 package version 1.17-2. *R Development Core Team. R: A language and environment for*statistical computing. Vienna: *R Foundation for Statistical Computing*.
- 49. Dray S, Dufour A-B (2007) The ade4 Package: Implementing the Duality Diagram for Ecologists.
 J Stat Softw 22:1–20. doi: 10.18637/jss.v022.i04
- Wold S, Ruhe A, Wold H, Dunn I W (1984) The Collinearity Problem in Linear Regression. The
 Partial Least Squares (PLS) Approach to Generalized Inverses. SIAM J Sci Stat Comput 5:735–
 743. doi: 10.1137/0905052
- Wold S, Sjöström M, Eriksson L (2001) PLS-regression: a basic tool of chemometrics. Chemom
 Intell Lab Syst 58:109–130. doi: 10.1016/S0169-7439(01)00155-1
- Lê Cao KA, Rohart F, Gonzalez I, et al (2017) mixOmics: Omics Data Integration Project.
 Available from: <u>https://CRAN.R-project.org/package=mixOmics</u>.
- 864 53. Rohart F, Gautier B, Singh A, Cao K-AL (2017) mixOmics: An R package for 'omics feature
 865 selection and multiple data integration. PLOS Comput Biol 13:e1005752. doi:
 866 10.1371/journal.pcbi.1005752
- Nielsen UN, Ayres E, Wall DH, Bardgett RD (2011) Soil biodiversity and carbon cycling: a review
 and synthesis of studies examining diversity–function relationships. Eur J Soil Sci 62:105–116.
 doi: 10.1111/j.1365-2389.2010.01314.x
- Schimel JP, Schaeffer SM (2012) Microbial control over carbon cycling in soil. Front Microbiol
 3:348. doi: 10.3389/fmicb.2012.00348
- 872 56. Nannipieri P, Ascher J, Ceccherini MT, et al (2003) Microbial diversity and soil functions. Eur J
 873 Soil Sci 54:655–670. doi: 10.1046/j.1351-0754.2003.0556.x

- 57. Yin B, Crowley D, Sparovek G, et al (2000) Bacterial Functional Redundancy along a Soil
 875 Reclamation Gradient. Appl Environ Microbiol 66:4361–4365. doi: 10.1128/AEM.66.10.4361876 4365.2000
- 877 58. Rousk J, Brookes PC, Bååth E (2009) Contrasting Soil pH Effects on Fungal and Bacterial
 878 Growth Suggest Functional Redundancy in Carbon Mineralization. Appl Environ Microbiol
 879 75:1589–1596. doi: 10.1128/AEM.02775-08
- Reis MP, Barbosa FAR, Chartone-Souza E, Nascimento AMA (2013) The prokaryotic community
 of a historically mining-impacted tropical stream sediment is as diverse as that from a pristine
 stream sediment. Extremophiles 17:301–309 . doi: 10.1007/s00792-013-0517-9
- 60. Gillan DC, Danis B, Pernet P, et al (2005) Structure of sediment-associated microbial
 communities along a heavy-metal contamination gradient in the marine environment. Appl
 Environ Microbiol 71:679–690. doi: 10.1128/AEM.71.2.679-690.2005
- Azarbad H, Niklińska M, Laskowski R, et al (2015) Microbial community composition and
 functions are resilient to metal pollution along two forest soil gradients. FEMS Microbiol Ecol
 91:1–11. doi: 10.1093/femsec/fiu003
- 889 62. Niklińska M, Chodak M, Stefanowicz A (2004) Community level physiological profiles of
 890 microbial communities from forest humus polluted with different amounts of Zn, Pb, and Cd—
 891 Preliminary study with BIOLOG ecoplates. Soil Sci Plant Nutr 50:941–944. doi:
 892 10.1080/00380768.2004.10408558
- 893 63. Nordgren A, Bååth E, Söderström B (1988) Evaluation of soil respiration characteristics to
 894 assess heavy metal effects on soil microorganisms using glutamic acid as a substrate. Soil Biol
 895 Biochem 20:949–954. doi: 10.1016/0038-0717(88)90109-5
- 896 64. Bérard A, Capowiez L, Mombo S, et al (2016) Soil microbial respiration and PICT responses to
 897 an industrial and historic lead pollution: a field study. Environ Sci Pollut Res 23:4271–4281.
 898 doi: 10.1007/s11356-015-5089-z
- 899 65. Stazi SR, Moscatelli MC, Papp R, et al (2017) A Multi-biological Assay Approach to Assess
 900 Microbial Diversity in Arsenic (As) Contaminated Soils. Geomicrobiol J 34:183–192. doi:
 901 10.1080/01490451.2016.1189015
- 90266.Bardgett RD, Freeman C, Ostle NJ (2008) Microbial contributions to climate change through903carbon cycle feedbacks. The ISME journal, 2(8), 805.
- 804 67. Rajapaksha RMCP, Tobor-Kapłon MA, Bååth E (2004). Metal toxicity affects fungal and
 905 bacterial activities in soil differently. Applied and Environmental Microbiology, 70(5), 2966906 2973. doi: 10.1128/AEM.70.5.2966-2973.2004
- 90768.Sullivan TS, McBride MB, Thies JE (2013) Rhizosphere microbial community and Zn uptake by908willow (Salix purpurea L.) depend on soil sulfur concentrations in metalliferous peat soils. Appl909Soil Ecol 67:53-60. doi: 10.1016/j.apsoil.2013.02.003
- 91069.Ni C, Horton DJ, Rui J, et al (2016) High concentrations of bioavailable heavy metals impact911freshwater sediment microbial communities. Ann Microbiol 66:1003–1012. doi:91210.1007/s13213-015-1189-8

913 70. Epelde L, Lanzén A, Blanco F, et al (2015) Adaptation of soil microbial community structure 914 and function to chronic metal contamination at an abandoned Pb-Zn mine. FEMS Microbiol 915 Ecol 91:1-11. doi: 10.1093/femsec/fiu007 916 71. Fierer N (2017) Embracing the unknown: disentangling the complexities of the soil microbiome. Nat Rev Microbiol 15:579-590. doi: 10.1038/nrmicro.2017.87 917 918 72. Sprocati AR, Alisi C, Tasso F, et al (2014) Bioprospecting at former mining sites across Europe: 919 microbial and functional diversity in soils. Environ Sci Pollut Res Int 21:6824-6835. doi: 10.1007/s11356-013-1907-3 920 921 73. Ellis RJ, Morgan P, Weightman AJ, Fry JC (2003) Cultivation-Dependent and -Independent 922 Approaches for Determining Bacterial Diversity in Heavy-Metal-Contaminated Soil. Appl 923 Environ Microbiol 69:3223-3230. doi: 10.1128/AEM.69.6.3223-3230.2003 924 74. Giller KE, Witter E, Mcgrath SP (1998) Toxicity of heavy metals to microorganisms and 925 microbial processes in agricultural soils: a review. Soil Biol Biochem 30:1389-1414. doi: 926 10.1016/S0038-0717(97)00270-8 927 75. Hemme CL, Deng Y, Gentry TJ, et al (2010) Metagenomic insights into evolution of a heavy 928 metal-contaminated groundwater microbial community. ISME J 4:660-672. doi: 929 10.1038/ismej.2009.154 930 76. Chander K, Joergensen RG (2001) Decomposition of 14C glucose in two soils with different 931 amounts of heavy metal contamination. Soil Biol Biochem 33:1811-1816. doi: 10.1016/S0038-932 0717(01)00108-0 933 77. Preston-Mafham J, Boddy L, Randerson PF (2002) Analysis of microbial community functional 934 diversity using sole-carbon-source utilisation profiles – a critique. FEMS Microbiol Ecol 42:1– 935 14 . doi: 10.1016/S0168-6496(02)00324-0 936 78. Lock K, Janssen CR (2005) Influence of soil zinc concentrations on zinc sensitivity and 937 functional diversity of microbial communities. Environ Pollut Barking Essex 1987 136:275–281. 938 doi: 10.1016/j.envpol.2004.12.038 79. 939 Ni N, Wang F, Song Y, et al (2017) Effects of cationic surfactant on the bioaccumulation of 940 polycyclic aromatic hydrocarbons in rice and the soil microbial community structure. RSC Adv 941 7:41444-41451. doi: 10.1039/C7RA07124H 942 80. Ren G, Teng Y, Ren W, et al (2016) Pyrene dissipation potential varies with soil type and 943 associated bacterial community changes. Soil Biol Biochem 103:71-85. doi: 944 10.1016/j.soilbio.2016.08.007 945 81. Padmanabhan P, Padmanabhan S, DeRito C, et al (2003) Respiration of 13C-Labeled Substrates Added to Soil in the Field and Subsequent 16S rRNA Gene Analysis of 13C-Labeled Soil DNA. 946 947 Appl Environ Microbiol 69:1614–1622. doi: 10.1128/AEM.69.3.1614-1622.2003 948 82. Trzesicka-Mlynarz D, Ward OP (1995) Degradation of polycyclic aromatic hydrocarbons (PAHs) 949 by a mixed culture and its component pure cultures, obtained from PAH-contaminated soil. 950 Can J Microbiol 41:470-476. 951 83. Viñas M, Sabaté J, Espuny MJ, Solanas AM (2005) Bacterial Community Dynamics and 952 Polycyclic Aromatic Hydrocarbon Degradation during Bioremediation of Heavily Creosote955 84. Martirani-Von Abercron S, Marín P, Solsona-Ferraz M, et al (2017) Naphthalene 956 biodegradation under oxygen-limiting conditions: community dynamics and the relevance of biofilm-forming capacity. Microb Biotechnol 10:1781–1796. doi: 10.1111/1751-7915.12842 957 958 85. Cébron A, Beguiristain T, Faure P, et al (2009) Influence of Vegetation on the In Situ Bacterial 959 Community and Polycyclic Aromatic Hydrocarbon (PAH) Degraders in Aged PAH-Contaminated or Thermal-Desorption-Treated Soil. Appl Environ Microbiol 75:6322-6330. doi: 960 961 10.1128/AEM.02862-08 962 86. Rousk J, Bååth E, Brookes PC, et al (2010) Soil bacterial and fungal communities across a pH 963 gradient in an arable soil. ISME J 4:1340-1351. doi: 10.1038/ismej.2010.58 964 87. Brewer TE, Handley KM, Carini P, et al (2017) Genome reduction in an abundant and 965 ubiquitous soil bacterium 'Candidatus Udaeobacter copiosus.' Nat Microbiol 2:16198. doi: 10.1038/nmicrobiol.2016.198 966 967 88. Sessitsch A, Weilharter A, Gerzabek MH, et al (2001) Microbial population structures in soil 968 particle size fractions of a long-term fertilizer field experiment. Appl Environ Microbiol 67:4215-4224. doi: 10.1128/AEM.67.9.4215-4224.2001 969 970 89. Franzluebbers AJ, Haney RL, Hons FM, Zuberer DA (1996) Active fractions of organic matter in 971 soils with different texture. Soil Biol Biochem 28:1367-1372. doi: 10.1016/S0038-972 0717(96)00143-5 973 90. Chau JF, Bagtzoglou AC, Willig MR (2011) The Effect of Soil Texture on Richness and Diversity 974 of Bacterial Communities. Environ Forensics 12:333–341. doi: 975 10.1080/15275922.2011.622348 976 91. Kandeler F, Kampichler C, Horak O (1996) Influence of heavy metals on the functional diversity 977 of soil microbial communities. Biol Fertil Soils 23:299–306. doi: 10.1007/BF00335958 978 92. Valsecchi G, Gigliotti C, Farini A (1995) Microbial biomass, activity, and organic matter 979 accumulation in soils contaminated with heavy metals. Biol Fertil Soils 20:253–259. doi: 10.1007/BF00336086 980 981 93. Chew I, Obbard JP, Stanforth RR (2001) Microbial cellulose decomposition in soils from a rifle 982 range contaminated with heavy metals. Environ Pollut 111:367–375. doi: 10.1016/S0269-7491(00)00094-4 983 McEnroe NA, Helmisaari H-S (2001) Decomposition of coniferous forest litter along a heavy 984 94. 985 metal pollution gradient, south-west Finland. Environ Pollut 113:11-18. doi: 10.1016/S0269-7491(00)00163-9 986 987 95. Lucisine P, Lecerf A, Danger M, et al (2015) Litter chemistry prevails over litter consumers in 988 mediating effects of past steel industry activities on leaf litter decomposition. Sci Total Environ 989 537:213-224. doi: 10.1016/j.scitotenv.2015.07.112

Contaminated Soil. Appl Environ Microbiol 71:7008–7018. doi: 10.1128/AEM.71.11.7008-

953

954

7018.2005

990 96. Wardle DA (1992) A Comparative Assessment of Factors Which Influence Microbial Biomass
991 Carbon and Nitrogen Levels in Soil. Biol Rev 67:321–358. doi: 10.1111/j.1469992 185X.1992.tb00728.x

- 97. Johnson D, Leake JR, Lee JA, Campbell CD (1998) Changes in soil microbial biomass and
 994 microbial activities in response to 7 years simulated pollutant nitrogen deposition on a
 995 heathland and two grasslands. Environ Pollut 103:239–250. doi: 10.1016/S0269996 7491(98)00115-8
- 98. Kuperman RG, Carreiro MM (1997) Soil heavy metal concentrations, microbial biomass and
 998 enzyme activities in a contaminated grassland ecosystem. Soil Biol Biochem 29:179–190. doi:
 999 10.1016/S0038-0717(96)00297-0
- 1000 99. Moore A, Becking J (1963) Nitrogen Fixation by Bacillus Strains Isolated from Nigerian Soils.
 1001 Nature 198:915-916. doi: 10.1038/198915a0
- 1002100.Fierer N, Lauber CL, Ramirez KS, et al (2012) Comparative metagenomic, phylogenetic and1003physiological analyses of soil microbial communities across nitrogen gradients. ISME J 6:1007–10041017. doi: 10.1038/ismej.2011.159

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Table 1.

Solgroup ch ch sh <				Мо	Не	Di	Uc	Те	NM	Но	Ро	MsM	RM	Me Kruskall-Wa	eans (± SE) per g llis multiple com (95% confidence	r oup nparaison test e)
Total organic carbon (g kg ⁻¹) 81.2 72.1 24.3 29.9 68.2 72.2 1980	Se	oil group		ctrl	ctrl	ctrl	sh	sh	sh	sh	sp	sp	sp	ctrl	sh	sp
Disoleticitation (Large (Large Large	Total orga	nic carbon (g	kg ⁻¹)	81.5	72.1	24.3	29.9	88.2	75.2	159.0	109.0	119.0	149.0	59.3 ± 17.7	88.1 ± 26.7	125.7 ± 12.0
Organic matter (sk *) 19.1 125 42 52 153 130 276 180 280 130 132.7 130.	Dissolved or	ganic carbon ((g kg ⁻¹)	1.34	1.69	1.18	0.12	0.99	0.23	0.33	0.76	0.66	0.24	1.4 ± 0.2 ^a	0.4 ± 0.2 ^b	0.6 ± 0.2 ^b
Nitrogen (s is	Organic	matter (g kg	1)	141	125	42	52	153	130	276	189	206	257	102.7 ± 30.7	152.7 ± 46.5	217.3 ± 20.4
CN 17.9 25.3 16.0 55.6 9.03 20.0 56.0 15.2 8.8 19.1.3 40.1.8.4 27.5.5.6 Available phonus (set) 5.0 0.15 9.9 3.6 20.0 17.5 9.8 51.2 16.2 20.3 2	Nitr	ogen (g kg ⁻¹)		4.56	2.85	1.74	0.56	4.35	2.35	2.92	7.19	4.18	3.83	3.1 ± 0.8	2.5 ± 0.8	5.1 ± 1.1
Available prosphorus (k ⁺) 0.60 0.02 0.02 0.01 </td <td></td> <td>C/N</td> <td></td> <td>17.9</td> <td>25.3</td> <td>14.0</td> <td>53.6</td> <td>20.3</td> <td>32.0</td> <td>54.6</td> <td>15.2</td> <td>28.5</td> <td>38.8</td> <td>19.1 ± 3.3</td> <td>40.1 ± 8.4</td> <td>27.5 ± 6.8</td>		C/N		17.9	25.3	14.0	53.6	20.3	32.0	54.6	15.2	28.5	38.8	19.1 ± 3.3	40.1 ± 8.4	27.5 ± 6.8
CEC (mon kg *) 55.0 95.0 15.5 9.9 3.6 20.0 17.5 9.8 12.2 16.2 10.2 10.2 10.7	Available p	hosphorus (g	kg ⁻¹)	0.04	0.02	0.04	< 0.01	0.21	0.08	0.02	0.12	0.21	0.07	0.03 ± 0.01	0.08 ± 0.05	0.13 ± 0.04
Carbonates (gk ⁻¹) 55 0.8 1.43 8.4 1.81.8 1.42 1.14.0 866 1.20 2.19.1 6.91.3 7.61.02 7.61.02 pr(H,O) 70 5.4 7.3 7.8 7.2 7.6 6.61.06 7.61.02 7.61.02 Organic acids (mg (g ⁻¹) 0.28 0.20 0.00 0.00 0.00 0.05 0.08 0.05 0.07 0.22.1 0.06 0.08.1 0.04 0.07 to 7.5 Solf ecture (mg (g ⁻¹)) 0.28 0.20 0.10 0.13 7.0 7.6 0.51 0.51 0.56 0.00 0.02 0.05 0.08 0.07 0.22.1 0.06 0.08.1 0.01 0.07 to 7.5 Solf ecture (mg (g ⁻¹)) 0.28 0.55 7.4 7.50 0.53 3.58 8.10 0.20 1.56 0.51 0.51 0.51 0.51 0.51 0.51 0.51 0.51 0.51 0.51 0.51 0.51 0.51 0.51 0.51 0.51 0.51 0.51	CEC	(cmol kg ⁻¹)		35.0	15.5	9.9	3.6	20.0	17.5	9.8	51.2	16.2	10.8	20.1 ± 7.6	12.7 ± 3.7	26.1 ± 12.7
pH(H,0) 70 54 7.3 8.0 7.7 7.4 7.5 7.8 7.2 7.6 6.640.6 7.640.1 7.640.2 Organic adds (mg/a) 223 140.0 120 115 238.1 278.3 120.0 10.0 10.2 0.08 0.05 0.07 0.22.00 0.080.00 0.07 0.22.00 0.080.00 0.07 0.22.00 0.080.00 0.07 0.22.00 0.080.00 0.07 0.22.00 0.081.00 0.07 0.22.00 0.081.00 0.07 0.22.00 0.081.00 0.07 0.20.00 0.07.00 0.22.00 0.051.07 3.110.2 2.00.04 0.00 0.07 0.07 0.07.00	Carbo	onates (g kg ⁻¹)		5.5	0.8	14.3	8.4	181.8	14.2	114.0	89.6	152.0	219.3	6.9 ± 3.9	79.6 ± 41.8	153.7 ± 37.4
Organic acids (mg kg ¹) 222.9 1400 120.7 8.2 179.0 115. 28.1 279.3 154.0 154.2 154.0 152.21.4 180.25.10 180.25.10 180.25.10 130.25.00 0.05 0.05 0.05 0.07 0.22.20.05 0.082.00 0.03.21.00 131.12.2 Soit testure (%) Sit 4.60 0.00 1.15 5.5 1.00 1.30 7.00 1.02 1.03 0.02 ± 0.06 0.08 ± 0.01 0.02 ± 0.06 Soit testure (%) Sit 4.60 6.90 1.43 1.76 2.07 2.3 1.64 1.45 2.85±0.27 7.15±0 2.85±0.27 3.1±0.5 2.8±0.6 2.05 2.15 7.19 8.81 1.20 1.1±0.4 0.05±0.2 9.6±1.7 1.1±0.4 0.05±0.2 9.6±1.7 1.1±0.4 0.05±0.2 9.6±1.7 1.5±0 3.4 2.1±0.4 0.6±0.0 0.5±0.0 1.5±0.4 1.5±0.4 0.5±0.2 9.5±0.7 1.5±0.4 0.5±0.2 0.5±0.0 1.5±0.4 0.5±0.0 <td>I</td> <td>oH (H₂O)</td> <td></td> <td>7.0</td> <td>5.4</td> <td>7.3</td> <td>8.0</td> <td>7.7</td> <td>7.4</td> <td>7.5</td> <td>7.8</td> <td>7.2</td> <td>7.6</td> <td>6.6 ± 0.6</td> <td>7.6 ± 0.1</td> <td>7.6 ± 0.2</td>	I	oH (H ₂ O)		7.0	5.4	7.3	8.0	7.7	7.4	7.5	7.8	7.2	7.6	6.6 ± 0.6	7.6 ± 0.1	7.6 ± 0.2
Catholydrates (mp kg ³) 0.28 0.29 0.20 0.02 0.05 0.08 0.05 0.07 0.22 ± 0.05 0.08 ± 0.04 0.07 ± 0.01 Solt testure (h) Sint 4.80 6.25 11.5 5.5 100 13.3 7.0 17.5 13.0 22.0 8.10 31.12 22.0 8.10 31.12 22.0 8.10 31.12 22.0 8.10 31.12 22.0 8.10 31.12 22.0 8.10 31.12 22.0 8.11 31.12 22.0 8.10 31.12 22.0 11.1 4.0 31.12 22.0 11.1 4.0 22.0 11.1 4.0 22.0 9.0 11.1 4.0 11.0 4.0 10.1 4.0 0.0 31.0 11.0 4.0 0.0 0.0 11.0 4.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0	Organic	acids (mg kg	1)	222.9	140.0	120.7	43.2	179.0	11.5	238.1	279.3	154.0	114.2	161.2 ± 31.4	118.0 ± 54.1	182.5 ± 49.7
Solitexture (s) Gay 449 25 115 55 100 133 70 77 109 109 286+109 90:17 112:12 Solitexture (s) Sit 460 69.0 143 176 270 220 53.3 727 7.66 43:1159 196:83 669:68 Solitexture (s) Al 672 51.9 450 690 690 58.8 8.0 292 164 415 23:120 715:43 28:10.4 Permeting (g k) ⁻¹ Al 672 51.9 450 692 55.8 5.4 15.5 34 21:0.4 0.51:2.2 0.51:0.1 Vig k 58 4.8 32 5.9 25.5 7.4 9.9 113 4.3 39 0.52:01 12:0.4 0.65:00 0.05:00 10.0 10.0 10.0 140 0.56 0.40 0.00 0.56:0 12:0.0 15:0.4 11:0.2 10:0 10.0 10:0 10:0	Carbohy	drates (mg kg	-1)	0.28	0.29	0.10	0.04	0.20	0.02	0.05	0.08	0.05	0.07	0.22 ± 0.06	0.08 ± 0.04	0.07 ± 0.01
Solt exter (M) Sit 4.60 6.90 1.43 1.76 2.07 2.79 1.20 5.33 7.27 7.46 4.11 ±0 9.64 ±3 665 ±68 Sind 5.1 5.5 7.42 7.69 6.33 5.88 8.10 2.20 1.64 1.45 2.83 ±2.00 7.51 ±0 2.01 ±0 2.01 ±0 5.5 ±0.7 3.11 ±0.5 2.81 ±0 0.01 ±0.1 1.11 ±0.4 1.05 ±2 9.61 ±1 0.01 ±0.1 <			Clay	48.9	25.5	11.5	5.5	10.0	13.3	7.0	17.5	10.9	10.9	28.6 ± 10.9	9.0 ± 1.7	13.1 ± 2.2
Total concentration (marked) Sand S.1 S.5 7.42 7.69 6.93 5.88 8.10 2.92 1.64 1.45 2.83:2.30 7.15:4.9 2.00:4.64 Major Ga 1.62 3.81 1.16 4.67 1250 9.92 15.10 7.19 8.81 1200 1.11:0.4 1.05:2.2 9.61:1.7 Major Ga 1.62 3.88 1.66 6.6 5.4 9.22 5.8 5.4 1.5 3.44 2.10:4.4 0.55:0.12 0.36:0.1 1.2:0:4 0.5:0:0.2 0.3:0:0.1 0.2 0.5:0:0.1 0.2:0:0.5 0.0:0:0.0 0.0:0 0.0:0 0.4 0.40 0.40 0.40 0.40 0.5:0:0 0.5:0:0 0.5:0:0 0.5:0:0 0.5:0:0 0.5:0:0 0.5:0:0 0.5:0:0 0.5:0:0 0.5:0:0 0.5:0:0 0.5:0:0 0.5:0:0 0.5:0:0 0.5:0:0 0.5:0:0:0 0.5:0:0 0.5:0:0:0 0.5:0:0:0 0.5:0:0:0 0.5:0:0:0 0.5:0:0:0:0 0.5:0:0:0:0:0:0:0:0:0:0:0:0:0:0:0:0:0:0:	Soil textur	e (%)	Silt	46.0	69.0	14.3	17.6	20.7	27.9	12.0	53.3	72.7	74.6	43.1 ± 15.9	19.6 ± 3.3	66.9 ± 6.8
Name Name <th< td=""><td></td><td></td><td>Sand</td><td>5.1</td><td>5.5</td><td>74.2</td><td>76.9</td><td>69.3</td><td>58.8</td><td>81.0</td><td>29.2</td><td>16.4</td><td>14.5</td><td>28.3 ± 23.0</td><td>71.5 ± 4.9</td><td>20.0 ± 4.6</td></th<>			Sand	5.1	5.5	74.2	76.9	69.3	58.8	81.0	29.2	16.4	14.5	28.3 ± 23.0	71.5 ± 4.9	20.0 ± 4.6
Image of the state state of the state of the state of the state of the st		Major elements (g kg ⁻¹)	AI	67.2	51.9	45.0	19.9	25.5	39.9	40.0	39.6	21.0	22.0	5.5 ± 0.7 ^a	3.1 ± 0.5 ^b	2.8 ± 0.6 ^b
Image: relation (g g ^) for the constraint (g g g g g g g g g g g g g g g g g g g			Ca	16.2	3.8	11.6	46.7	125.0	99.2	151.0	71.9	88.1	129.0	1.1 ± 0.4	10.5 ± 2.2	9.6 ± 1.7
real Mg 5.8 4.8 3.2 5.9 2.35 7.4 9.9 11.3 4.3 3.9 0.5 ± 0.1 1.2 ± 0.4 0.6 ± 0.2 Total Na 2.52 5.87 8.87 0.40 1.61 2.73 1.49 0.76 0.40 0.28 0.58 ± 0.8 0.6 ± 0.0 0.05 ± 0.0 ref 4.41 2.73 1.80 452.0 108.0 193.0 65.8 93.0 304.0 53.6 3.0 ± 0.8* 2.0 ± 8.7* 15.0 ± 6.7* 15.0 ± 6.7* 15.0 ± 6.7* 15.0 ± 6.7* 12.0 ± 7.6* 64.0 0.7 0.3 0.1 4.3 3.3 1.1 15.20 17.6 2.20 0.4 ± 0.2* 2.2 ± 1.0* 63.9 ± 9.4.1 12.0 ± 7.0* 15.0 16.0 61.9 56.4 ± 0.1 44.11 ± 67.8 2.0 ± 2.4 ± 0.0* 63.9 ± 9.4.1 17.8 ± 0.6* 13.9 ± 9.0 12.0 ± 1.0* 16.0 61.9 56.4 ± 0.1 44.11 ± 67.8 12.0 ± 1.0* 12.0 ± 1.0 14.1 ± 67.8 ± 1.0* 12.0 ± 1.2 ± 1.0* 12.0 ± 1.2 ± 1.0*			к	14.6	18.6	28.6	0.6	5.4	9.2	5.8	5.4	1.5	3.4	2.1 ± 0.4 ^a	0.5 ± 0.2 ^b	0.3 ± 0.1 ^b
Total concentration (monomina circle) extraction) Na 2.52 5.87 8.87 0.40 1.61 2.73 1.49 0.76 0.40 0.28 0.58±0.18 0.16±0.05 0.05±0.01 Total concentration (monomina circle) (monomina circle) (monomina circle) Tat 144 248 605 74 1650 2540 314 29600 55400 119000 151±54 ³ 114±580 ⁵ 6800±2650 1260 14400 39500 56±19 ⁴ 363±134 ² 29533±769 1261 114 120 17.6 220 0.4±0.2 ⁵ 2.2±1.0 ⁸ 139±44.1 126 17.6 120 0.4±0.2 ⁹ 2.2±1.0 ⁸ 139±44.1 126 126.4±33.4 126.4±33.4 126.4±33.4 126.4±33.4 126.4±33.4 126.4±33.4 126.4±33.4 126.4±33.4 126.4±33.4 132.1±8 98.3±2.0 126.4±33.4 126.4±33.4 132.1±8.4 98.3±2.0 126.4±33.4 127.4±2.8 117.8±2.8 117.8±2.8 117.8±2.8 117.8±2.8 117.8±2.8 117.8±2.8 117.8±2.8 117.8±2.8 117.8±2.8			Mg	5.8	4.8	3.2	5.9	23.5	7.4	9.9	11.3	4.3	3.9	0.5 ± 0.1	1.2 ± 0.4	0.6 ± 0.2
Total concentration (hydrofluoric acid extraction) Fe* 44.1 27.3 18.0 452.0 108.0 193.0 65.8 93.0 93.0 53.6 30.0.8 20.5 k.7 150.7 k.7 total concentration (hydrofluoric acid (hydrofluoric acid extraction) Ta 144 248 605 74 1650 2540 314 29600 55400 119000 151.5 % 1144.5 % 80.00 2.5 k.7 % 30.3 1.7 % 30.3 34700 14400 39500 56.1 9 363.1 34 29.3 37.7 69 33.1 11 1520 17.6 0.20 0.4 4.0 2.9 22.1 10 63.9 4.4 11 20.0 17.0 1500 166.0 61.9 58.6 2.01 41.1 2.67.8 126.9 4.2 8 126.4 4.2 8			Na	2.52	5.87	8.87	0.40	1.61	2.73	1.49	0.76	0.40	0.28	0.58 ± 0.18	0.16 ± 0.05	0.05 ± 0.01
Total concentration (hydrofluoric acid extraction) Zn * * 144 248 60.5 74 1650 2540 314 2960 5540 1900 151 ±54 * 1144 ±50 * 6800 ±255 (hydrofluoric acid extraction) Pb * 44 93 31 23 475 653 303 3470 1440 39500 56±19 * 36±14 * 29533±7 69 (hydrofluoric acid extraction) frace 6d* 0.7 0.3 0.1 0.1 4.5 3.3 11 152.0 17.6 22.0 0.4±0.2 * 2.2±1.0 * 63.9±44.1 Ca * 164 12.9 10.2 9.4 146.0 144.0 93.6 148.0 16.0 61.9 58.4±0.1 44.1±267.8 126.9±3.0 125.9±3.1 125.9±3.1 126.9±3.0 127.9±3.1 127.9±3.1 126.9±3.0 127.9±3.1 127.9±3.1 126.9±3.0 127.9±3.1 127.9±3.1 127.9±3.1 127.9±3.1 127.9±3.1 127.9±3.1 127.9±3.1 127.9±3.1 127.9±3.1 127.9±			Fe *	44.1	27.3	18.0	452.0	108.0	193.0	65.8	93.0	304.0	53.6	3.0 ± 0.8 ^b	20.5 ± 8.7 a	15.0 ± 7.8 a
concentration (hydroluori, acid extraction) Pb* 44 93 31 23 475 653 303 34700 14400 39500 56±19 ^b 36±14 ^b 29533±769 extraction) Metallic trace elements (mg k1 ⁻¹) Cd* 0.7 0.3 0.1 0.1 455 3.3 1.1 1520 17.6 22.0 0.4±0.2 ^b 2.2±1.0 ^b 63.9±4.4.1 extraction) frace elements (mg k1 ⁻¹) 16.4 12.9 10.2 9.4 146.0 144.0 936 148.0 146.0 42.3 132±1.8 98.3±2.0 125.4±4.30 Cu* 16.4 12.9 10.2 7.3 55.3 1130 29.9 7.31 26.0 34.3 25.8±8.7 51.4±2.2 127.4±3.6 24.4±13.5 Cu* 14.0 14.0 94.0 15.0 0.9 0.5 0.6 4.6 12.3 10.0 7.9 21.0 1.5 1.0±0.3 6.9±2.6 101.15.7 Mo 1.5 0.9	Total		Zn *	144	248	60.5	74	1 650	2 540	314	29 600	55 400	119 000	151 ± 54 ^b	1 144 ± 580 ^b	68 000 ± 26 565 ª
Inverted extraction (vdfortuoric ecid extraction) Cd* 0.7 0.3 0.1 0.1 4.5 3.3 1.1 1520 17.6 22.0 0.4±0.2 ^b 2.2±10 ^b 63.9±44.1 Metallic trace elements (mg kg ⁻¹) Ct* 75.1 82.1 18.7 22.5 345.0 1220.0 177.0 150.0 166.0 61.9 58.6±20.1 441.1±267.8 126.0±32.4 Ni* 42.5 21.7 13.2 7.3 55.3 113.0 29.9 73.1 246.0 34.3 25.8±8.7 51.4±2.8 117.8±65.7 Co* 14.2 19.0 4.8 17.4 20.2 28.8 11.3 10.6 51.4 11.3 12.7±4.2 19.4±3.6 24.4±13.5 Co* 14.2 19.0 4.8 17.4 20.2 28.8 11.3 10.6 51.4 11.3 12.7±4.2 19.4±3.6 24.4±13.5 Mo 1.5 0.9 0.5 0.6 4.6 12.3 10.0 7.9 21.0 1.5 1.0±0.3 6.9±2.6 10.1±5.7 Mo 1.5	concentration		Pb *	44	93	31	23	475	653	303	34 700	14 400	39 500	56 ± 19 ^b	$363\pm134~^{\rm b}$	29 533 ± 7 692 ^a
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	(hydrofluoric acid extraction)		Cd *	0.7	0.3	0.1	0.1	4.5	3.3	1.1	152.0	17.6	22.0	0.4 ± 0.2 b	2.2 ± 1.0 $^{\rm b}$	63.9 ± 44.1 ^a
Available concentration (calcium chlorind extraction; (alcium chlorind		Metallic	Cr *	75.1	82.1	18.7	22.5	345.0	1 220.0	177.0	150.0	166.0	61.9	58.6 ± 20.1	441.1 ± 267.8	126.0 ± 32.4
elements (mg kg ⁻¹) Ni* 42.5 21.7 13.2 7.3 55.3 113.0 29.9 73.1 246.0 34.3 25.8±8.7 51.4±2.8 117.8±65.3 Co* 14.2 19.0 4.8 17.4 20.2 28.8 11.3 10.6 51.4 11.3 12.7±4.2 19.4±3.6 24.4±13.5 Ti* 0.77 0.59 1.13 0.09 0.52 0.83 0.26 88.30 4.79 8.63 0.83±0.16 ^b 0.43±0.16 ^b 39.1±27.27 Mo 1.5 0.9 0.5 0.6 4.6 12.3 10.0 7.9 21.0 1.5 1.0±0.3 6.9±2.6 10.1±5.7 Mn 932 1570 285 740 3480 9890 250 68100 3830 771 929±371 4158±1993 2423±219 Available concentration (calcium chloride extraction; (g kg ¹) Cd 2.3 82.8 1.3 <1.0		trace	Cu *	16.4	12.9	10.2	9.4	146.0	144.0	93.6	148.0	186.0	42.3	13.2 ± 1.8	98.3 ± 32.0	125.4 ± 43.0
Available concentration (calcium chlorind extraction); µkg ¹) Co * 14.2 19.0 4.8 17.4 20.2 28.8 11.3 10.6 51.4 11.3 12.7±4.2 19.4±3.6 24.4±13.5 T1 * 0.77 0.59 1.13 0.09 0.52 0.83 0.26 88.30 4.79 8.63 0.83±0.16 ^b 0.43±0.16 ^b 33.91±27.27 Mo 1.5 0.9 0.5 0.6 4.6 12.3 10.0 7.9 21.0 1.5 1.0±0.3 6.9±2.6 10.1±5.7 Mn 932 1570 285 740 3480 9890 2520 68100 3830 771 929±371 4158±1993 2423±219 Available concentration (calcium chlorid calcium chlorid extraction; µkg ¹) Cd 2.3 82.8 1.3 <1.0		elements (mg kg ⁻¹)	Ni *	42.5	21.7	13.2	7.3	55.3	113.0	29.9	73.1	246.0	34.3	25.8 ± 8.7	51.4 ± 22.8	117.8 ± 65.1
Available concentration (racium chlorin de stratcion (racium regulator) PAH Total 3.5 0.1 0.9 0.52 0.83 0.26 88.30 4.79 8.63 0.83 ± 0.16 ^b 0.43 ± 0.16 ^b 33.91 ± 27.27 Mo 1.5 0.9 0.5 0.6 4.6 12.3 10.0 7.9 21.0 1.5 1.0 ± 0.3 6.9 ± 2.6 10.1 ± 5.7 Mn 932 1570 285 740 3480 9890 2520 68100 3830 771 929 ± 371 4158 ± 1993 24 234 ± 21 9 Available concentration (racium chlorin (racium chlori (racium chlorin (racium chlorin (racium chlor		(1115 16)	Co *	14.2	19.0	4.8	17.4	20.2	28.8	11.3	10.6	51.4	11.3	12.7 ± 4.2	19.4 ± 3.6	24.4 ± 13.5
Mo 1.5 0.9 0.5 0.6 4.6 12.3 10.0 7.9 21.0 1.5 1.0±0.3 6.9±2.6 10.1±5.7 Mn 932 1570 285 740 3480 9890 2520 68100 3830 771 929±371 4158±1993 24234±199 24234±199 Available concentration (calcium chloride extraction; µkg ¹) <10			ті *	0.77	0.59	1.13	0.09	0.52	0.83	0.26	88.30	4.79	8.63	0.83 ± 0.16 ^b	$0.43\pm0.16~^{\rm b}$	33.91 ± 27.22 ^a
Mn 932 1570 285 740 3480 9890 2 520 68 100 3 830 771 929±371 4 158±193 24 234±219 Available concentration (calcium chloride extraction; µg kg ¹) 710 38 100 11 <10			Mo	1.5	0.9	0.5	0.6	4.6	12.3	10.0	7.9	21.0	1.5	1.0 ± 0.3	6.9 ± 2.6	10.1 ± 5.7
Available concentration (calcium chloride extraction; µg kg ⁻¹) Total 3.5 0.1 0 0.6 128 3.8 2.310 49.200 64.000 12.704±12.698 51±27 38.503±18.5 38.503±18.5 Available concentration (calcium chloride extraction; µg kg ⁻¹) Cd 2.3 82.8 1.3 <1.0			Mn	932	1 570	285	740	3 480	9 890	2 520	68 100	3 830	771	929 ± 371	4 158 ± 1 993	24 234 ± 21 951
Available concentration (calcium chloride extraction; µg kg ⁻¹) Pb <3 94 <3 <3 <3 <3 <3 32 397 1440 32±31 b 1±0 b 623±422 t Available concentration (calcium chloride extraction; µg kg ⁻¹) Cd 2.3 82.8 1.3 <1.0			Zn	< 10	38 100	11	< 10	36	128	38	2 310	49 200	64 000	12 704 ± 12 698	51 ± 27	38 503 ± 18 594
Available concentration (calcium chloride extraction; µg kg ⁻¹) Cd 2.3 82.8 1.3 < 1.0 1.6 2.4 1.7 48.7 111.0 25.3 28.8 ± 27.0 1.7 ± 0.3 61.7 ± 25.6 chloride extraction; µg kg ⁻¹) Cr < 10			Pb	< 3	94	< 3	< 3	< 3	< 3	< 3	32	397	1 440	32 ± 31 ^b	1 ± 0 ^b	623 ± 422 ^a
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	Available concentra	ation (calcium	Cd	2.3	82.8	1.3	< 1.0	1.6	2.4	1.7	48.7	111.0	25.3	28.8 ± 27.0	1.7 ± 0.3	61.7 ± 25.6
Cu 41 30 27 24 129 84 237 98 187 29 33 ± 4 118 ± 45 105 ± 46 Ni <15	chloride extracti	on; µg kg⁻¹)	Cr	< 10	< 10	< 10	< 10	< 10	20	< 10	29	< 10	< 10	1 ± 0	5.7 ± 4.7	10.5 ± 9.5
Ni < 15 390 < 15 < 15 90 < 15 < 15 126 59 131±130 23±22 62±36 Metal pollution index 71.2 66.8 27.2 144.0 198.6 329.0 115.6 475.5 480.9 299.8 55.1±14.0 ⁻⁶ 196.8±47.3 ^{-b} 418.7±59.5 Sum of 16 regulatory PAH Total 3.5 0.1 0.9 0.9 114.7 1095.9 937.7 21.2 49.6 6.6 1.5±0.6 ^{-b} 537.3±146.2 ^{+a} 25.8±6.4 ⁺ (n=3; mg kg ⁻¹) Available 0.0 0.0 6.3 83.0 54.1 0.2 46 0.0 200.005 37.5±140.2 ^{+a} 25.8±6.4 ⁺			Cu	41	30	27	24	129	84	237	98	187	29	33 ± 4	118 ± 45	105 ± 46
Metal pollution index 71.2 66.8 27.2 144.0 198.6 329.0 115.6 475.5 480.9 299.8 55.1 ± 14.0 ^c 196.8 ± 47.3 ^b 418.7 ± 59.5 Sum of 16 regulatory PAH Total 3.5 0.1 0.9 0.9 114.7 1095.9 937.7 21.2 49.6 6.6 1.5 ± 0.6 ^b 537.3 ± 146.2 ^a 25.8 ± 6.4 ^a (n = 3; mg kg ⁻¹) Available 0.0 0.0 6.3 83.0 54.1 0.2 46 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0		Ni		< 15	390	< 15	< 15	< 15	90	< 15	< 15	126	59	131 ± 130	23 ± 22	62 ± 36
Sum of 16 regulatory PAH Total 3.5 0.1 0.9 0.9 114.7 1095.9 937.7 21.2 49.6 6.6 1.5±0.6 ^b 537.3±146.2 ^a 25.8±6.4 ⁱ	Metal	pollution inde	x	71.2	66.8	27.2	144.0	198.6	329.0	115.6	475.5	480.9	299.8	55.1 ± 14.0 °	196.8 ± 47.3 ^b	418.7 ± 59.5 ^a
(n=3; mg kg ⁻¹) Available 0.0 0.0 0.0 63 83.0 54.1 0.2 4.6 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0.0 0	Sum of 16 regul	atory PAH	Total	3.5	0.1	0.9	0.9	114.7	1 095.9	937.7	21.2	49.6	6.6	1.5 ± 0.6 ^b	537.3 ± 146.2 ª	25.8 ± 6.4 ^a
······································	(n = 3; mg	kg⁻¹)	Available	0.0	0.0	0.0	0.0	6.3	83.0	54.1	0.2	4.6	0.0	0.0 ± 0.0 ^c	35.8 ± 10.7 $^{\rm a}$	1.6 ± 0.8 ^b

1011 Table 2.

		Мо	Не	Di	Uc	Те	NM	Но	Ро	MsM	RM	Me Kruskall-Wal	ans (± SE) per gro lis multiple comp (95% confidence)	oup baraison test
	Soil group	ctrl	ctrl	ctrl	sh	sh	sh	sh	sp	sp	sp	ctrl	sh	sp
	MPN (x10 ³ g ⁻¹)	1 434.5 ± 327.7	40.2 ± 8.1	488.6 ± 126.1	161.2 ± 53.9	586.1 ± 170.0	496.9 ± 214.8	2 071.4 ± 952.1	80.2 ± 19.1	38.5 ± 4.1	0.1 ± 0.1	654.5 ± 410.9	828.9 ± 424.1	39.6 ± 23.1
Microbial characteristics (n = 3)	16S rRNA gene copy number (x10 ¹⁰ g ⁻¹)	17.8 ± 2.4	25.0 ± 2.2	30.1 ± 1.1	6.9 ± 0.5	41.5 ± 1.2	4.4 ± 0.3	14.8 ± 0.7	41.5 ± 0.4	26.4 ± 1.2	17.8 ± 0.4	24.3 ± 2.1	16.9 ± 4.5	28.6 ± 3.5
	18S/16S rRNA gene copy ratio	0.03 ± 0.00	0.18 ± 0.01	0.07 ± 0.00	0.08 ± 0.01	0.08 ± 0.01	0.07 ± 0.01	0.08 ± 0.00	0.04 ± 0.00	0.04 ± 0.00	0.05 ± 0.00	0.09 ^a ± 0.02	0.08 ^a ± 0.00	0.04 ^b ± 0.00
Pactorial	Chao1 index	2 621 ± 205	3,161 ± 254	4 966 ± 32	4 293 ± 291	4 380 ± 38	2 541 ± 100	4 966 ± 100	3 422 ± 223	3 151 ± 2	2 478 ± 15	3,583 ± 367	4 045 ± 282	3 017 ± 154
Bacterial taxonomic diversity (n = 3)	Shannon's index (H')	5.40 ± 0.03	6.38 ± 0.05	7.12 ± 0.02	6.68 ± 0.03	6.97 ± 0.01	6.37 ± 0.01	7.06 ± 0.04	6.65 ± 0.02	6.06 ± 0.02	5.99 ± 0.02	6.30 ^{ab} ± 0.25	6.77 ^a ± 0.08	6.23 ^b ± 0.11
	Pielou's evenness index (J')	0.70 ± 0.01	0.80 ± 0.01	0.85 ± 0.01	0.81 ± 0.00	0.84 ± 0.00	0.82 ± 0.00	0.84 ± 0.00	0.83 ± 0.00	0.77 ± 0.00	0.78 ± 0.00	0.78 ^{ab} ± 0.02	0.83 ^a ± 0.00	0.79 ^b ± 0.01
MicroResp	Catabolic richness (n = 6)	5.0 ± 0.0	5.8 ± 0.3	5.0 ± 0.0	5.8 ± 0.3	4.5 ± 0.3	5.0 ± 0.4	4.5 ± 0.3	5.8 ± 0.3	5.0 ± 0.4	4.5 ± 0.3	5.3 ± 0.2	4.9 ± 0.2	5.1 ± 0.3
catabolic diversity	Shannon's index (H')	1.53 ± 0.03	1.62 ± 0.06	1.57 ± 0.01	1.72 ± 0.05	1.41 ± 0.07	1.57 ± 0.08	1.49 ± 0.07	1.13 ± 0.02	1.41 ± 0.09	1.39 ± 0.07	1.57 ^a ± 0.03	1.55 ^a ± 0.05	1.31 ^b ± 0.06
(n = 4)	Pielou's evenness index (J')	0.95 ± 0.02	0.93 ± 0.02	0.97 ± 0.01	0.98 ± 0.00	0.94 ± 0.02	0.98 ± 0.01	0.99 ± 0.00	0.65 ± 0.02	0.88 ± 0.03	0.93 ± 0.01	0.95 ^b ± 0.01	0.97 ^a ± 0.01	0.82 ^c ± 0.04
Biolog catabolic diversity (n = 3)	Catabolic richness (n = 62)	30.7 ± 1.3	17.7 ± 2.2	21.3 ± 1.45	25.7 ± 2.6	22.0 ± 9.5	12.7 ± 3.5	15.7 ± 5.8	15.0 ± 1.5	12.0 ± 4.4	5.0 ± 1.15	23.2 ª ± 2.1	19.0 ^a ± 3.0	10.7 ^b ± 2.0
	Shannon index (H')	3.27 ± 0.04	2.77 ± 0.13	2.97 ± 0.07	3.15 ± 0.10	2.78 ± 0.38	2.35 ± 0.36	2.43 ± 0.41	2.47 ± 0.09	2.22 ± 0.39	1.52 ± 0.25	3.00 ^a ± 0.08	2.68 ^a ± 0.17	2.07 ^b ± 0.20
	Pielou's evenness index (J')	0.96 ± 0.00	0.97 ± 0.01	0.97 ± 0.00	0.97 ± 0.01	0.95 ± 0.01	0.96 ± 0.02	0.94 ± 0.01	0.92 ± 0.02	0.95 ± 0.01	0.98 ± 0.01	0.97 ± 0.00	0.95 ± 0.01	0.95 ± 0.01



1017 Figure 2.



	Control	Slag heap	Settling pond	p-value
Unclassified OTUs	С	В	А	2.100 x 10 ⁻⁵
Others		n.s.		0.661
Verrucomicrobia	А	В	В	9.415 x 10 ⁻⁵
Gamma-proteobacteria	С	В	А	2.793 x 10 ⁻³
Delta-proteobacteria		n.s.		0.404
Beta-proteobacteria		n.s.		0.325
Alpha-proteobacteria	В	А	В	2.840 x 10 ⁻⁵
Candidate div. TM7	В	А	А	1.103 x 10 ⁻³
Nitrospirae	В	В	А	9.559 x 10 ⁻³
Gemmatimonadetes	с	В	А	5.512 x 10 ⁻⁵
Firmicutes		n.s.		0.084
Chloroflexi	С	В	А	1.989 x 10 ⁻⁴
Bacteroidetes	А	А	В	2.967 x 10 ⁻³
Actinobacteria		n.s.		0.396
Acidobacteria		n.s.		0.086

1020 Figure 3.







1026 Figure 5.







1033 Supplementary material

1035 Figure S1.



Table S1.

			Bi	olog	Micro	oResp
Chemical guild	Substrate	Chemical formula	Eco	MT2	Substrate	Time (h)
PAHs	Phenanthrene	C ₁₄ H ₁₀		~		
	Pyrene	C ₁₆ H ₁₀		~		
	Fluoranthene	C ₁₆ H ₁₀		√		
	Fluorene	C ₁₃ H ₁₀		~		
Aromatic acids	Benzoïc acid	C ₇ H ₆ O ₂		~		
	Protocatechuic acid	C ₇ H ₆ O ₄		✓		
	Phtalic acid	C ₈ H ₆ O ₄		~		
	P-Coumaric acid	C ₉ H ₈ O ₃		v		
	Ferulic acid	$C_{10}H_{10}O_4$		v		
<u> </u>	Syringic acid	C ₉ H ₁₀ O ₅				
Carboxylic acids	Pyruvic acid	$C_3H_4O_3$		v	~	4.67
	Propionic acid	$C_3H_6O_2$		v		
	D-Gluconic acid	$C_6H_{12}O_7$		v		
	Oxalic acid	C ₂ H ₂ O ₄		v		
	Succinic acid	$C_4H_6O_4$		~	~	6.67
	Citric acid	C ₆ H ₈ O ₇		~	\checkmark	6.17
	Trans-Aconitic acid	C ₆ H ₆ O ₆		✓		
	L-Tartaric acid	$C_4 H_6 O_6$		\checkmark		
	D-Galactonic acid y-lactone	$C_6H_{10}O_6$	~			
	D-Galacturonic acid	C ₆ H ₁₀ O ₇	\checkmark			
	γ-Hydroxybutyric acid	C ₄ H ₈ O ₃	\checkmark			
	D-Glucosaminic acid	C ₆ H ₁₃ NO ₆	\checkmark			
	Itaconic acid	$C_5H_6O_4$	\checkmark			
	α-Ketobutyric acid	C ₄ H ₆ O ₃	\checkmark			
	D-Malic acid	C ₄ H ₆ O ₅	\checkmark			
	2-Hydroxy benzoic acid	C ₇ H ₆ O ₃	\checkmark			
	4-Hydroxy benzoic acid	C ₇ H ₆ O ₃	\checkmark			
Amino acids	L-Proline	C ₅ H ₉ NO ₂		√		
	Glycine	$C_2H_5NO_2$		\checkmark		
	L-Alanine	C ₃ H ₇ NO ₂		\checkmark		
	L-Arginine	C ₆ H ₁₄ N ₄ O ₂	\checkmark			
	L-Asparagine	$C_4H_8N_2O_3$	\checkmark		\checkmark	5.33
	L-Phenylalanine	C ₉ H ₁₁ NO ₂	\checkmark			
	L-Serine	C ₃ H ₇ NO ₃	\checkmark			
	L-Threonine	$C_4H_9NO_3$	\checkmark			
	Glycyl-L-Glutamic Acid	C ₇ H ₁₂ N ₂ O ₅	~			
Carbohydrates	D-Ribose	C ₅ H ₁₀ O ₅		√	✓	8.00
	D-Glucose	$C_{6}H_{12}O_{6}$		✓		
	D-Fructose	$C_{6}H_{12}O_{6}$		✓		
	D-Sucrose	C ₁₂ H ₂₂ O ₁₁		\checkmark		
	D-Trehalose	C ₁₂ H ₂₂ O ₁₁		\checkmark		
	D-Mannose	$C_6H_{12}O_6$		✓		
	β-Methyl-D-glucoside	$C_7H_{14}O_6$	~			
	D-Xylose	C ₅ H ₁₀ O ₅	\checkmark			
	<i>i</i> -Erythritol	$C_4H_{10}O_4$	\checkmark			
	D-Mannitol	C ₆ H ₁₄ O ₆	~		✓	8.17
	N-Acetyl-D-glucosamine	C ₀ H ₁₅ NO ₆	✓			
	D-Cellobiose	C12H22O11	✓			
	α-D-Lactose	C12H22C11	~			
	Pyruvic acid methyl ester	C.H.O.	~			
Polymers	Lignin (alkali)	(C44H202S)		~		
	Cellulose (from Spruce)	$(C_{14}, 200, 30)$		✓		
	Xvlan	(C5H8O5)		1		
	Glycogen	(CcHaOc)	~			
	a-Cyclodextrine	(C ₆ , 1005)n	✓			
	Twoon 40	C361 60 C 30	•			
			•			
Miscellaneous	Glucose-1-phosphato	СНОР	* 			
wiscenarieous	DL a Cheard sharshata		•			
	Phonylothyl amino		•			
	Putroscipo		•			
	Catachal	C ₄ Π ₁₂ N ₂	v	./		
	Calechol	C61160 2		v		

classement d'après Analysis of microbial community functional diversity using sole-carbon-source utilisation profiles - a critique (Preston-Mafham et al. 2002) avec changement de groupes (*) ou ajout (#)

1043 Table S2	52.
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		F-value	P-value
	Axis 1	26.338	0.001 ***
	Axis 2	1.683	0.216
RDA-A	Ti_metals	2.072	0.198
	PAH	1.729	0.238
	sand	16.304	0.002 **
	pH	3.251	0.108
RDA-B	Axis 1	10.353	0.002 **
	Axis 2	2.815	0.111
	C_org	4.458	0.009 **
	N	8.709	0.002 **
RDA-C	Axis 1	19.903	0.002 **
	Axis 2	8.222	0.012 *
	Ti_metals	7.767	0.010 **
	PAH	1.047	0.412
	C_org	2.963	0.110
	N	4.708	0.032 *
	K_t	2.396	0.139

p < 0.05 * p < 0.01 ** p < 0.001 ***

1044 1045