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## How reversible are the effects of silver nanoparticles on macrophages? A proteomicinstructed view†

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Silver nanoparticles are known to strongly affect biological systems, and numerous toxicological studies have investigated their effects. Most of these studies examine the effects immediately following acute exposure. In this work, we have conducted further investigation by studying not only the acute, postexposure response, but also the cellular response after a 72 hour-recovery-phase post exposure. As a biological model we have used macrophages, which are very important cells with respect to their role in the immune response to particulate materials. To investigate the response of macrophages to nanoparticles and their recovery post exposure, we have used a combination of proteomics and targeted experiments. These experiments provided evidence that the cellular reaction to nanoparticles, including the reaction during the recovery phase, is a very active process involving massive energy consumption. Pathways such as the oxidative stress response, central and lipid metabolism, protein production and quality control are strongly modulated during the cellular response to nanoparticles, and restoration of basic cellular homeostasis occurs during the recovery period. However, some specialized macrophage functions, such as lipopolysaccharide-induced cytokine and nitric oxide production, did not return to their basal levels even 72 hours post exposure, showing that some effects of silver nanoparticles persist even after exposure has ceased

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## **Environmental significance**

Silver nanoparticles are known to have profound effects on living cells. Because of their widespread use, contamination is almost unavoidable. In this context, it is important not only to assess the immediate effects of silver nanoparticles on cells, but also how they recover after exposure. Using macrophages as the target cell type and a combination of proteomic and targeted experiments, we show here that the recovery phase is not just a "return to normal" condition. For example, some of the specialized macrophage functions are still not restored after this time, showing that subtle but sustained effects can occur after a single exposure. Although cell survival is not affected, such effects may impact the health status of living beings.

## 1. Introduction

Silver nanoparticles (AgNPs) are used as a biocide in a variety of consumer and medical products, because of the toxic effects that they have on microorganisms. However, as an unwanted side effect, AgNPs also show toxicity to mammalian cells. Thus, numerous toxicological studies have been conducted both in vitro in bacterial and eukaryotic model cell systems and in vivo in animal models, including at the microbiome level. 1-3 Although some studies, especially those using

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animal models, have examined the effects of NPs after repeated, low-dose exposures, 4-8 most of the cellular, toxicological studies have been conducted in an acute exposure mode, meaning that high doses have been used and exposure times have been short, spanning less than or equal to 24 hours. In these types of studies, toxicological endpoints are investigated at the end of the exposure period. 9-16 In terms of exposure, the acute schemes relate most closely to accidental, acute exposures, while chronic schemes correspond highly to occupational exposures. It is of interest to study how biological systems recover after a single, high, non-lethal exposure. This is not a trivial question, as some mineral toxicants such as beryllium or crystalline silica show very prolonged effects, even after the exposure has ceased. 17,18 This poor recovery has also been observed for silver nanomaterials such as silver nanowires19 and silver nanoparticles.20

Many toxicological studies have reported inflammatory responses<sup>5,11,12,21</sup> and/or immunological effects<sup>15,22</sup> pointing to macrophages as a cell type of major interest in toxicological studies of AgNPs. This is in accordance with the important scavenging function of macrophages in many different tissues.

This work investigates the recovery of macrophages after a unique, acute, but subtoxic dose of AgNPs. Targeted experiments carried out primarily on macrophages showed that several functional effects (e.g. LPS-induced cytokine and NO production, mitochondrial transmembrane potential and phagocytosis) were altered immediately after exposure to AgNPs. However, these functions tended to return to normal after a 72 h recovery period.<sup>23</sup> We therefore decided to get a broader view of cellular recovery using a proteomic approach.

Proteomic approaches have been used to study cellular responses to AgNPs in several models such as intestinal cells<sup>9,24,25</sup> or liver cells.<sup>10</sup> This allowed several important pathways modulated in response to acute exposures to AgNPs to be highlighted. These included mitochondrial proteins, pointing at potential mitochondrial dysfunction, 9,10,24 and proteins involved in intermediate metabolism<sup>9,10,24</sup> or in inflammatory responses.<sup>10</sup> The latter findings reinforced the interest in studying macrophages, which are major players in inflammation. We therefore decided to use proteomics to study the cellular responses using a macrophage J774 cell line, directly after acute exposure and after a recovery period had lapsed. This model cell line was selected because, unlike human monocyte cell lines, 26,27 mouse macrophage cell lines such as RAW264.7 and J774 do not need to be chemically differentiated into macrophages, a process that deeply alters cell physiology and causes some cell death. Furthermore, as opposed to primary macrophages derived from human blood, these mouse cell lines do not show extensive variability from one experiment to another. This explains why these cell lines are extensively used as models for testing of a wide variety of nanomaterials, 28-31 including large scale projects.32-34

## 2. Experimental

Most experiments have been performed essentially as described in previous publications.35-37 Details are given here for the sake of the consistency of the paper. All biological experiments were carried out at least on three independent biological replicates.

#### 2.1 Nanoparticles

PVP-coated silver nanoparticles were purchased from Sigma, directly as a concentrated suspension (catalog number #758329). Their characteristics have been published previously.<sup>23,38</sup> These experiments confirmed the manufacturer's data and showed that the nanoparticles are spherical, with a diameter in the 50-110 nm range, and did not aggregate upon dilution in water or culture medium.

#### 2.2 Cell culture

The mouse macrophage cell line J774 was obtained from the European Cell Culture Collection (Salisbury, UK). The cells were cultured in DMEM + 10% fetal bovine serum (FBS). For routine culture, cells were seeded in non-adherent flasks (e.g. suspension culture flasks from Greiner) at 200 000 cells per ml and harvested 48 hours later, at 1000000 cells per ml. Cell viability was measured by a dye exclusion assay, either with eosin (1 mg ml<sup>-1</sup>)<sup>39</sup> under a microscope or with propidium iodide (1 µg ml<sup>-1</sup>)<sup>40</sup> in a flow cytometry mode.

For determination of the useful dose, cells were seeded at 500 000 cells per ml. They were treated with nanoparticles on the following day and harvested after a further 24 hours in culture.

For treatment with nanoparticles, cells were seeded in classical cell culture flasks and left for 24 hours at 37 °C for cell adhesion and confluence. For the recovery condition, cells were treated with nanoparticles on the following day. After 24 hours of exposure, the cell culture medium was removed and replaced by fresh medium. For consistency reasons, this operation was also carried out on cells used for control and for acute exposure. Another medium culture change was carried out 36 hours after the initial medium change, i.e. mid-term of the 72 hour recovery period. Finally, acute exposure was carried out for the final 24 hours and the cells were used immediately afterwards.

#### 2.3 Phagocytosis and particle internalization assay

The phagocytic activity was measured using fluorescent latex beads (1 µm diameter, green labelled, catalog number L4655 from Sigma). The beads were pre-incubated at a final concentration of 55 µg mL<sup>-1</sup> for 30 minutes at 37 °C in PBS/FBS (v/ v). Then, they were incubated with cells (5 μg mL<sup>-1</sup>) for 2 h 30 at 37 °C. Cells were harvested and washed with PBS. Cells were resuspended by vortexing with addition of 3/4 water volume and then 1/4 NaCl (35 mg mL<sup>-1</sup>) volume was added under vortexing in order to clean the cell surface of adsorbed particles. Cells were harvested in PBS with propidium iodide

(1 µg mL<sup>-1</sup>). Viability and phagocytic activity were measured simultaneously by flow cytometry on a FacsCalibur instrument (Beckton Dickinson). The dead cells (propidium positive) were excluded from the analysis.

#### 2.4 Mitochondrial transmembrane potential measurement

The mitochondrial transmembrane potential was assessed by rhodamine 123 uptake. Cells were incubated with rhodamine 123 (80 nM) for 30 minutes at 37 °C, 5% CO2 then rinsed twice in cold glucose (1 mg mL<sup>-1</sup>)-PBS (PBSG) and harvested in cold PBSG supplemented with propidium iodide (1 µg mL<sup>-1</sup>). The mitochondrial potential of the cells was analysed by flow cytometry on a FacsCalibur instrument (Beckton Dickinson). The dead cells (propidium positive) were excluded from the analysis. The low rhodamine concentration was used to avoid intramitochondrial fluorescence quenching that would result in a poor estimation of the mitochondrial potential.41

#### 2.5 Enzyme assays

The enzymes were assayed according to published procedures (see below).

The cell extracts for enzyme assays were prepared by lysing the cells for 20 minutes at 0 °C in Hepes (20 mM, pH 7.5), MgCl<sub>2</sub> (2 mM), KCl (50 mM), EGTA (1 mM), and tetradecyldimethylammonio propane sulfonate (SB 3-14) (0.15% (w/v)), followed by centrifugation at 15 000g for 15 minutes to clear the extract. The protein concentration was determined by a dye-binding assay. 42 The dehydrogenase dehydrogenase-coupled activities were assayed at 500 nm using the phenazine methosulfate/iodonitrotetrazolium coupled assay. 43 The enzyme assay buffer contained 25 mM Hepes, NaOH (pH 7.5), 5 mM magnesium acetate, 100 mM potassium nitrate and 1% Triton X-100. It also contained 30 μM phenazine methosulfate, 200 μM iodonitrotetrazolium chloride, 250 µM of the adequate cofactor (NAD or NADP) and 1-5 mM of the organic substrate, which was used to start the reaction. For phosphate-dependent enzymes such as glyceraldehyde dehydrogenase (GAPDH) and purine phosphorylase (PNPH), 50 mM potassium phosphate (pH 7.5) was added to the enzyme assay buffer. Triose phosphate isomerase was assayed with dihydroxyacetone phosphate and a glyceraldehyde dehydrogenase-coupled assay. 44 Purine phosphorylase (PNPH) was assayed by a xanthine oxidase-coupled assay.45 Hexokinase was assayed by a glucose phosphate dehydrogenase (G6PDH)-coupled assay. 46 Biliverdine reductase was assayed at 450 nm as described. 47 Pyridoxal kinase was assayed directly at 388 nm. 48 Enolase was assayed at 340 nm by a pyruvate kinase-lactate dehydrogenase-coupled assay. 49

## 2.6 NADP/NADPH and glucose assays

The glucose concentration in conditioned media was determined by a hexokinase-G6PDH assay. 50 Briefly, culture media collected at the end of the exposure period were centrifuged for 15 min at 15 000g to pelletize the particulate material.

The supernatant was collected, and diluted 100-fold in the dehydrogenase assay buffer described above, containing also 1 mM EGTA, 30 µM phenazine methosulfate, 200 µM iodonitrotetrazolium chloride, 10 IU ml<sup>-1</sup> G6PDH and 7.5 IU ml<sup>-1</sup> hexokinase. The reaction was started by the addition of ATP (1 mM final concentration) and the increase in absorbance at 500 nm read for 1 minute. The linear part of the absorbance curve was used to determine the reaction speed, which was proportional to the glucose concentration. Fresh culture media supplemented with 10% fetal calf serum, i.e. high glucose DMEM (4.5 g L<sup>-1</sup> glucose), RPMI1640 (2 g L<sup>-1</sup> glucose) and 199 medium (1 g L<sup>-1</sup> glucose) were used as standards.

The NADP-NADPH concentration was determined using an adapted alkaline extraction buffer. 51 Briefly, at the end of the exposure period, cells were collected by scraping, rinsed twice in PBS and pelleted. The packed cell pellet (PCP) volume was estimated, and the cells were lysed in 10 PCP volumes of 10 mM CAPS, 1 mM EGTA and 2 mM MgCl2 for 10 minutes on ice with occasional vortexing. The suspension was centrifuged (10000g, 5 minutes, 4 °C), the viscous cell pellet discarded and the supernatant collected and split into two aliquots. The first aliquot was neutralized on ice by adding 0.1 volume of 1 M tricine. This aliquot contained both the oxidized and reduced forms of the pyridine nucleotides, and was also used to determine the protein concentration by a dye-binding assay.<sup>42</sup>

The other aliquot was heated at 60 °C for 30 minutes in a thermostated water bath to destroy the oxidized forms of the pyridine nucleotides.<sup>51</sup> It was then cooled on ice, neutralized by adding 0.1 volume of 1 M tricine, and centrifuged for 10 minutes at 10 000g at 4 °C to eliminate any particulate material.

The NADP-NADPH concentration was then determined by using an enzyme cycling assay and standard NADP solutions.52

#### 2.7 NO production and cytokine production

The cells were grown to confluence in a 6-well plate and pretreated with nanoparticles as described above. For the final 18 hours of culture, half of the wells were treated with 100 ng ml<sup>-1</sup> LPS (from salmonella, purchased from Sigma), and arginine monohydrochloride was added to all the wells (5 mM final concentration) to give a high concentration of the substrate for nitric oxide synthase. After 18 hours of incubation, the cell culture medium was recovered and centrifuged at 10 000g for 10 minutes to remove cells and debris, and the nitrite concentration in the supernatants was read at 540 nm after addition of an equal volume of Griess reagent and incubation at room temperature for 30 minutes.

For cytokine production, a commercial kit (BD Cytometric Bead Array, catalog number 552364 from BD Biosciences) was used.

#### 2.8 F-actin staining

The experiments were performed essentially as previously described.53 The cells were recovered at the mid-term medium

change (see above) and cultured on coverslips placed in 6-well plates. For the acute exposure condition, the cells were exposed to silver nanoparticles for 24 h at 37 °C. At the end of the exposure time, the cells were washed twice for 5 min at 4 °C in PBS and fixed in 4% paraformaldehyde for 30 min at room temperature. After two washes (5 min/4 °C in PBS), they were permeabilized in 0.1% Triton X100 for 5 min at room temperature. After two more washes in PBS, 500 nM Phalloidin-Atto 550 (Sigma) was added to the cells and left for 20 min at room temperature in the dark. Coverslipattached cells were washed, placed on microscope slides (Thermo Scientific) using a Vectashield mounting medium containing DAPI (Eurobio) and imaged using a Zeiss LSM 800 confocal microscope. The images were processed using ImageJ software.

#### 2.9 Silver assay by ICP-MS

For measuring the Ag-NP uptake and release in cells, 2 mL cell cultures in 6-well plates were used, with the exposure scheme described above in section 2.2. At the end of the exposure period, the culture medium was removed and saved and the cell layer was gently rinsed twice with culture medium without serum.

The cells were then lysed by scraping in 2 ml of lysis buffer (50 mM Hepes (pH 7.5), 4 mM magnesium acetate, 200 mM sorbitol, and 0.1% (w/v) tetradecyldimethylammonio propane sulfonate (SB 3-14)). The lysate was pipetted into a microtube and incubated for 20 minutes on ice to complete cell lysis. For determination of soluble silver, aliquots of the culture media and of the cell lysates were centrifuged for 45 minutes at 16 000g, and the upper half of the supernatant was collected.

The samples were mineralized by the addition of one volume of suprapure 65% HNO3 and incubation on a rotating wheel at room temperature for 18 h.

Mineralized samples were diluted in 0.5% (v/v) HNO3 and analysed using an iCAP RQ quadrupole mass instrument (Thermo Fisher Scientific GmbH, Germany) equipped with an ASX-560 auto-sampler (Teledyne CETAC Technologies, Omaha, USA). The instrument was used with a MicroMist U-Series glass concentric nebulizer, a quartz spray chamber cooled at 3 °C, a Qnova quartz torch, a nickel sample cone, and a nickel skimmer cone equipped with a high-sensitivity insert. <sup>24</sup>Mg, <sup>25</sup>Mg, <sup>107</sup>Ag and <sup>109</sup>Ag concentrations were determined using standard curves and corrected using an internal standard solution of 103Rh added online. Data integration was done using the Qtegra software (version 2.8.2944.115). The results were normalized using the Mg concentration (4 mM in the cellular extracts and 0.82 mM in culture medium). To take into account the cellular concentration effects, the protein concentration of the extracts was determined by a dye-binding assay.42

#### 2.10 Proteomics

The 2D gel-based proteomic experiments were essentially carried out as previously described,35 at least on independent biological triplicates. However, detailed materials and methods are provided for the sake of paper consistency.

2.10.1 Sample preparation. The cells were collected by scraping and then washed three times in PBS. The cells were then washed once in TSE buffer (10 mM Tris-HCl (pH 7.5), 0.25 M sucrose, and 1 mM EDTA), and the volume of the cell pellet was estimated. The pellet was resuspended in its own volume of TSE buffer. Then 4 volumes (respective to the cell suspension just prepared) of concentrated lysis buffer (8.75 M urea, 2.5 M thiourea, 5% w/v CHAPS, 6.25 mM TCEP-HCl, 20 mM spermine base, and 20 mM spermine tetrahydrochloride) were added and the solution was left to extract at room temperature for 1 hour. The nucleic acids were then pelleted by centrifugation (270 000g at room temperature for 1 h), and the protein concentration in the supernatant was determined by a dye-binding assay.42 Carrier ampholytes (Pharmalytes, pH 3-10) were added to a final concentration of 0.4% (w/v), and the samples were kept frozen at -20 °C until use.

2.10.2 Isoelectric focusing. Home-made 160 mm long 4-8 linear pH gradient gels<sup>54</sup> were cast according to published procedures.<sup>55</sup> Four mm-wide strips were cut and rehydrated overnight with the sample, diluted to a final volume of 0.6 ml of the rehydration solution (7 M urea, 2 M thiourea, 4% CHAPS, 0.4% carrier ampholytes (Pharmalytes 3-10) and 100 mM dithiodiethanol).56

The strips were then placed in a Multiphor plate (GE Healthcare), and IEF was carried out with the following electrical parameters: 100 V for 1 hour, then 300 V for 3 hours, then 1000 V for 1 hour, then 3400 V up to 60-70 kVh. After IEF, the gels were equilibrated for 20 minutes in 125 mM Tris, 100 mM HCl, 2.5% SDS, 30% glycerol and 6 M urea.<sup>57</sup> They were then transferred on top of the SDS gels and sealed in place with 1% agarose dissolved in 125 mM Tris, 100 mM HCl, 0.4% SDS and 0.005% (w/v) bromophenol blue.

2.10.3 SDS electrophoresis and protein detection. Ten percent gels (160 × 200 × 1.5 mm) were used for protein separation. The Tris taurine buffer system<sup>58</sup> was used and operated at an ionic strength of 0.1 and a pH of 7.9. The final gel composition is thus 180 mM Tris, 100 mM HCl, 10% (w/v) acrylamide, and 0.27% bisacrylamide. The upper electrode buffer is 50 mM Tris, 200 mM taurine, and 0.1% SDS. The lower electrode buffer is 50 mM Tris, 200 mM glycine, and 0.1% SDS. The gels were run at 25 V for 1 hour, then 12.5 W was applied per gel until the dye front reached the bottom of the gel. Detection was carried out by tetrathionate silver staining<sup>59</sup>.

2.10.4 Image analysis and global analysis of the spot abundance data. The gels were scanned after silver staining on a flatbed scanner (Epson perfection V750), using 16 bit grayscale image acquisition. The gel images were then analyzed using the Delta 2D software (v 3.6). The spots that were never expressed above 100 ppm of the total spots were first filtered out. Then, significantly-varying spots were selected on the basis of their Student's T-test p-value between the treated and the control groups. Spots showing a p-value lower than 0.05

were selected. This strategy is used to avoid the use of arbitrary thresholds, which can result in discarding statisticallyvalid relevant changes and including non-valid changes.<sup>60</sup> The false positive concern was taken into account using several approaches such as the Benjamini Hochberg approach, 61 the sequential goodness of fit approach<sup>62</sup> and the sequential Fisher test approach.<sup>63</sup> Furthermore, we checked that all the spots that we found through the *T*-test also had a p < 0.05 in a non-parametric Mann-Whitney U-test.

For the global analysis of the spot abundance data, we used directly the spot abundance data as provided by the gel analysis software. The software directly normalizes each spot abundance by the sum of all spot abundances detected on the gel. These relative abundance data were used directly for global analysis using the PAST software suite<sup>64</sup> without any transformation. No limitation in the number of principal components was implemented in the principal component analysis.

2.10.5 Mass spectrometry. The spots selected for identification were excised from silver-stained gels and destained with ferricyanide/thiosulfate on the same day as silver staining in order to improve the efficiency of the identification process. 65,66 In-gel digestion was performed with an automated protein digestion system, MassPrep Station (Waters, Milford, USA). The gel plugs were washed twice with 50 µL of 25 mM ammonium hydrogen carbonate (NH4HCO3) and 50 μL of acetonitrile. The cysteine residues were reduced by 50 μL of 10 mM dithiothreitol at 57 °C and alkylated by 50 μL of 55 mM iodoacetamide. After dehydration with acetonitrile, the proteins were cleaved in the gel with 10 µL of 12.5 ng μL<sup>-1</sup> modified porcine trypsin (Promega, Madison, WI, USA) in 25 mM NH<sub>4</sub>HCO<sub>3</sub>. The digestion was performed overnight at room temperature. The generated peptides were extracted with 30 µL of 60% acetonitrile in 0.1% formic acid. Acetonitrile was evaporated under vacuum before nanoLC-MS/MS analysis.

NanoLC-MS/MS analysis was performed using a nano-ACQUITY Ultra-Performance-LC (Waters Corporation, Milford, USA) coupled to a Synapt™ High Definition Mass Spectrometer™ (Waters Corporation, Milford, USA), or to a TripleTOF 5600 (Sciex, Ontario, Canada).

The nanoLC system was composed of an ACQUITY UPLC® CSH130 C18 column (250 mm × 75 μm with a 1.7 μm particle size, Waters Corporation, Milford, USA) and a symmetry C18 precolumn (20 mm × 180 µm with a 5 µm particle size, Waters Corporation, Milford, USA). The solvent system consisted of 0.1% formic acid in water (solvent A) and 0.1% formic acid in acetonitrile (solvent B). 4 µL of the sample were loaded into the enrichment column over 3 min at 5 µL min<sup>-1</sup> with 99% of solvent A and 1% of solvent B. Elution of the peptides was performed at a flow rate of 300 nL min<sup>-1</sup> with a 8-35% linear gradient of solvent B in 9 minutes.

The Synapt™ High Definition Mass Spectrometer™ (Waters Corporation, Milford, USA) was equipped with a Z-spray ion source and a lock mass system. The system was fully controlled by MassLynx 4.1 SCN639 (Waters Corporation, Milford, USA). The capillary voltage was set at 2.8 kV and the cone voltage at 35 V. Mass calibration of the TOF was achieved using fragment ions from Glu-fibrino-peptide B in the [50;2000] m/z range. Online correction of this calibration was performed with Glu-fibrino-peptide B as the lock-mass. The ion  $(M + 2H)^{2+}$  at m/z 785.8426 was used to calibrate MS data and the fragment ion  $(M + H)^+$  at m/z 684.3469 was used to calibrate MS/MS data during the analysis.

For tandem MS experiments, the system was operated with automatic switching between the MS (0.5 s per scan in the m/z range [150:1700]) and MS/MS mode (0.5 s per scan in the m/z range [50;2000]). The two most abundant peptides (intensity threshold 20 counts per s), preferably doubly and triply charged ions, were selected in each MS spectrum for further isolation and CID fragmentation using a collision energy profile. Fragmentation was performed using argon as the collision gas.

Mass data collected during analysis were processed and converted into .pkl files using ProteinLynx Global Server 2.3 (Waters Corporation, Milford, USA). Normal background subtraction was used for both MS and MS/MS with a 5% threshold and polynomial correction of order 5. Smoothing was performed on MS/MS spectra (Savitzky-Golay, 2 iterations, window of 3 channels). Deisotoping was applied for MS (medium deisotoping) and for MS/MS (fast deisotoping).

The TripleTOF 5600 (Sciex, Ontario, Canada) was operated in positive mode, with the following settings: ion spray voltage floating (ISVF) 2300 V, curtain gas (CUR) 10, interface heater temperature (IHT) 150, ion source gas 1 (GS1) 2, declustering potential (DP) 80 V. The informationdependent acquisition (IDA) mode was used with Top 10 MS/MS scans. The MS scan had an accumulation time of 250 ms in the m/z [400;1250] range and the MS/MS scans 100 ms in the m/z [150;1800] range in high sensitivity mode. Switching criteria were set to ions with a charge state of 2-4 and an abundance threshold of more than 500 counts; the exclusion time was set at 4 s. The IDA rolling collision energy script was used for automatically adapting the CE. Mass calibration of the analyser was achieved using peptides from digested BSA. The complete system was fully controlled by AnalystTF 1.7 (Sciex). Raw data collected were processed and converted with MSDataConverter in the .mgf peak list format.

For protein identification, the MS/MS data were interpreted using a local Mascot server with the MASCOT 2.4.1 algorithm (Matrix Science, London, UK) against UniProtKB/SwissProt (version 2016\_01, 550299 sequences). The research was carried out in all species. Spectra were searched with a mass tolerance of 15 ppm for MS and 0.05 Da for MS/MS data, allowing a maximum of one trypsin missed cleavage. Carbamidomethylation of cysteine residues and oxidation of methionine residues were specified as variable modifications. Protein identifications were validated with at least two peptides with a Mascot ion score above 30.

#### 2.11 Electron microscopy

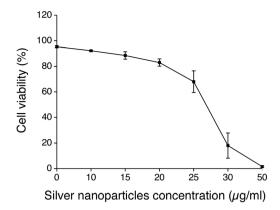
J774 macrophage cells were grown on coverslips and treated with AgNPs as described above. Cells were allowed to recover for 72 hours before being fixed with 2.5% glutaraldehyde in 0.1 M cacodylate buffer (pH 7.4) for 2 hours at room temperature. Cells were then washed with buffer, post fixed with 1% osmium tetroxide in the same buffer for 1 hour at 4°, and washed with water. Cells were then dehydrated through a series of graded alcohol (30-60-90-100-100-100%) and infiltrated with a mix of 1/1 epon/alcohol (100%) for 1 hour before several baths of fresh epon (Fluka) for 3 hours. Finally, a capsule full of epon was placed on the surface of the cells and the resin was allowed to polymerise for 72 h at 60 °C. The polymerised bloc was then detached from the coverslip with HF (48%) over 1 hour. Ultrathin sections of the cell monolayer were cut with an ultramicrotome (Leica). For transmission electron microscopy (TEM) observation, the sections were post-stained with 5% uranyl acetate and 0.4% lead citrate and observed with a transmission electron microscope at 80 kV (JEOL 1200EX). Images were acquired with a digital camera at magnification ranging from 15k to 50k (Veleta, Olympus). For scanning transmission electron microscopy observation (STEM) and energy dispersive spectroscopic analysis (EDS), the sections were not post-stained. They were coated with a thin layer of carbon and observed and analyzed with an OSIRIS microscope (TECNAI) operating at 200 kV.

## 3. Results

# 3.1 Silver accumulation upon treatment with silver nanoparticles and after recovery

First, the effect of silver nanoparticles and soluble silver on cell viability was assayed, and the results are shown in Fig. 1. For all the subsequent experiments, a dose of 20  $\mu g$  ml $^{-1}$  was selected, corresponding to the lethal dose 20% (LD20). This dose ensures strong effects on cells while keeping cellular mortality at an acceptable level for subsequent experiments. We also checked the presence of nanoparticles in cells by electron microscopy. Control cells did not show any electron-dense materials (Fig. 2A and B), while cells exposed to Ag nanoparticles acutely showed some electron-dense particles located in large membrane compartments called macropinosomes (Fig. 2C and D). After the recovery period, these electron dense particles were still observed in the macropinosomes (Fig. 2E and F). Energy dispersive spectroscopic analysis of these electron-dense features confirmed that they contained Ag (Fig. 2G–I).

Silver accumulation in cells was measured by ICP-MS. The results are summarized in Table 1. They showed a strong accumulation of silver in cells at the end of the 24 hour exposure followed by a moderate loss of total silver. The cells were able to cycle once during the 72 h recovery period, as shown by the increase in the protein amount in the extracts, which resulted in a further reduction of the cellular silver content. A moderate but detectable silver excretion was also evidenced. It should be also noted that the protein concentration is used in toxicology as an indicator for cell number, <sup>67,68</sup>



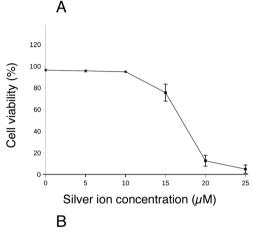


Fig. 1 Dose response curve to silver nanoparticles and ionic silver. Cell viability was measured by dye exclusion. The cells were exposed to silver nanoparticles (panel A) or silver lactate (panel B) for 24 hours before measurement of cell viability (n = 3).

#### 3.2 Global analysis of the proteomic results

From the total protein expression data obtained from the gel analysis software, a subset of variable proteins (p < 0.25 in either the acute vs. control or the recovery vs. control comparisons) was extracted. This allowed the noise brought by proteins that do not show a consistent variation in the biological phenomena under investigation to be decreased. This subset was then tested using the PAST statistical suite.<sup>64</sup> As a first test, a hierarchical clustering was performed, and the results are shown in Fig. 3A. Clustering clearly separates the control group from the two silver-treated ones, but gives no information regarding which silver treatment condition is closer to the control. Principal component analysis gave a more valuable indication, as shown in Fig. 3B. The recovery group appeared to be in an intermediate position between the control and acute groups along component one, which carries most of the variance. This is an expected position in a "return to normal" model during the recovery phase. However, the recovery group separated from both the control and the acute groups in component 2. This suggested that the three conditions were fairly different from each other. This was further confirmed by analysis of similarities (ANOSIM), which gave the following results for the *p*-values of the pairwise comparisons:

p = 0.031 for the acute vs. control comparison

p = 0.032 for the recovery vs. control comparison

p = 0.027 for the acute vs. recovery comparison

These figures suggest that the recovery phase was a very active state, which did not represent just an intermediate state between the control and the acute stage, i.e. immediately after exposure to silver nanoparticles. To further investigate this aspect, we extracted the intensity values for the spots that were significantly altered (p < 0.05) between the control group and at least one of the silver-treated groups. In order

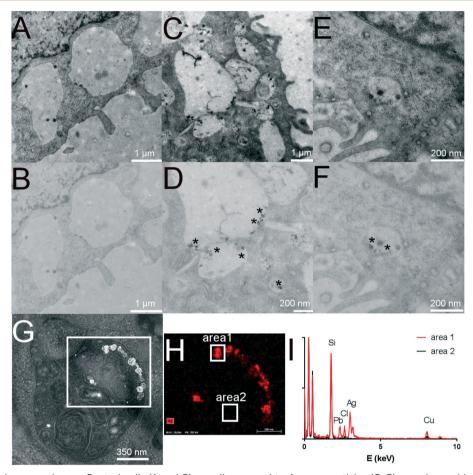


Fig. 2 Ag distribution in macrophages. Control cells (A and B) or cells exposed to Ag nanoparticles (C-G) were imaged by transmission electron microscopy (A-F) or scanning transmission electron microscopy (G). Panels A and B: control cells; panels C and D: cells acutely exposed to Ag nanoparticles; panels E and F: recovering cells. A, C and E: High contrast images; B, D and F: low contrast images highlighting the presence of electron-dense materials, which are probably Ag nanoparticles or debris of Ag nanoparticles (\*). Panel G is a HAADF image of a multilamellar body containing electron-dense materials. H and I: Chemical element distribution of the area of G delimited by the white square, analyzed by energy dispersive spectroscopy. Panel H shows the Ag distribution and panel I shows the EDS spectra of area1 (an intracellular region containing some Ag) and area2 (an intracellular region that does not contain any Ag) of H.

Table 1 Measurements of silver concentrations

	Unexposed	Exposed to 20 μg mI <sup>-1</sup> AgNPs	72 h recovery after exposure to AgNPs
Protein concentration in the cell extract (mg ml <sup>-1</sup> )	$0.352 \pm 0.02$	$0.322 \pm 0.02$	$0.657 \pm 0.04$
Total silver concentration in the cell extract ( $\mu g l^{-1}$ )	0	9794 ± 1159	$8258 \pm 1536$
Soluble silver concentration in the cell extract ( $\mu g l^{-1}$ )	0	$39.86 \pm 11.83$	47.57 ± 3.11
Soluble silver concentration in culture medium ( $\mu g l^{-1}$ )		$457 \pm 64^a$	$142 \pm 4$
Normalized silver content (ng Ag mg <sup>-1</sup> protein)	0	$30608\pm5055$	$12530\pm1984$
Soluble silver fraction in cells	0	$0.0041 \pm 0.0014$	$0.0059 \pm 0.001$
Fraction of silver excreted <sup>b</sup>	0	ND	$0.017 \pm 0.003$

<sup>&</sup>lt;sup>a</sup> This value contains both the soluble silver that is actively excreted by cells during the exposure and the soluble silver that arises from nanoparticle dissolution directly in the culture medium. <sup>b</sup> Defined as soluble silver in medium/total silver (i.e. silver in medium + silver in cells).

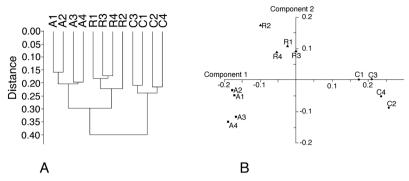


Fig. 3 Global analysis of proteomic data. Global data analysis was performed on the proteomic data to derive the relationships between the various samples used. Panel A: Hierarchical clustering. The Euclidean distance similarity index and Ward's algorithm were used for this analysis. C1–C4: Control (non-exposed) cells. A1–A4: Acutely exposed cells (24 hours). R1–R4: Recovering cells (72 hours post-exposure). Panel B: Principal component analysis. A representation displaying the first two principal components is shown. The PCA yielded 11 components, of which the first two, represented in the figure, explained 70% of the total variance (axis 1 – 54%, axis 2 – 16%). Component 3 explained 7.5% of the variance, component 4 explained 6%, and the remaining components explained less than 4% of the variance each. Samples labelled as in panel A.

to cope with the multiple testing issue, we used several approaches such as the Benjamini Hochberg approach,<sup>61</sup> the sequential goodness of fit approach<sup>62</sup> and the sequential Fisher test approach.<sup>63</sup> All the selected spots showed a very low *q*-value for the condition for which they were variable compared to the control (Table S1†).

We then divided this dataset of 239 spots into lists (Table S2†). The first list, labelled as "returning spots", contained the spots for which the amplitude of the change in abundance between the recovery and control stages is lower than the amplitude of the change in abundance between the acute exposure and control stages. 135 spots belonged to this list. It was then divided into subsets. The first subset contained "fast returning spots". It was defined as spots for which the quantitative change in the acute vs. control comparison shows an amplitude at least twice that in the recover vs. control comparison (i.e. more than 50% convergence). 44 spots belonged to this subset. The other subset of this list contained the 91 "slow returning spots", for which the return-to-normal trend was lower than 50% in the 72 hour recovery period.

The other list, labelled as "diverging spots", contained the 104 spots for which the amplitude of the change in abundance between the recovery and control stages is higher than the amplitude of the change in abundance between the acute exposure and control stages. This list was also divided into subsets. The "fast diverging spots" subset was defined as the spots for which the quantitative change in the recover vs. control comparison was at least 1.5 higher than the quantitative change in the acute vs. control comparison (i.e. more than 50% divergence). 53 spots belonged to this subset. The other subset of this list contained the 51 "slow diverging spots", for which the diverging trend was lower than 50% in the 72 hour recovery period.

#### 3.3 Detailed analysis of the proteomic results

In order to gain more precise insights, we investigated which proteins were significantly altered (p < 0.05) between the

control group and at least one of the silver-treated groups. The list of modulated proteins is given in Table 2 and the corresponding 2D gel images are shown in Fig. 4. As 2D gels separate protein forms, a variation in a spot does not necessarily mean that the whole amount of the protein is modulated. In fact, a single protein can be represented by various 2D gel spots, which do not necessarily all change upon the treatment with nanoparticles. To alleviate this problem, we systematically identified by mass spectrometry not only the modulated spots, but also the neighboring spots to check for such isoform issues. The quantitative results have been integrated in Table 2 and the mass spectrometry identification parameters are detailed in Table S3.† The results for some classes (apoptosis, cell fate and energy metabolism) are also shown in gel images in Fig. 4, and the results for the other classes are shown in Fig. S1-S5.† For global analysis of the modulated proteins, we used the DAVID annotation tool, 69,70 available at https://david.ncifcrf.gov/, and the results are shown in Tables S4 and S5.† The results showed that some functional classes such as energy metabolism, cytoskeleton especially actin cytoskeleton, protein folding and quality control (especially the ubiquitin/proteasome pathway), nucleotide and nucleic acid metabolism, cell signaling, oxidative stress response and detoxification proteins are strongly represented among the modulated proteins. This prompted us to carry out validation experiments on some of these proteins or pathways.

### 3.4 Enzymatic activities

For some of the modulated proteins, we first checked whether their enzyme activities (Table 3), *i.e.* the relevant parameter in terms of cell physiology, followed or not the trends observed *via* proteomics. There is no obvious inference between the two parameters, as the activity may be modulated by PTM and thus correlate with one or several spots, or with the sum of the spots representing the same protein. One good example is represented by malate dehydrogenase,

Table 2 Differentially-expressed proteins identified in the proteomic screen

Spot number	Protein name	Accession number	Ratio acute/ctrl	$T$ test acute $\nu s$ . ctrl	Ratio recov/ctrl	$T$ test recov $\nu s$ . ctr
A1	Casp3	P70677	0.59	0.04	1.00	0.99
A2	Efhd2	Q9D8Y	1.24	0.02	1.18	0.23
C1	Actin	P60710	1.51	< 0.01	1.67	0.01
C2a	Actn4/1	P57780	1.45	0.04	1.18	0.16
C2b	Actn4/2	P57780	1.43	0.04	1.38	0.01
32c 32d	Actn4/3 Actn4/4	P57780 P57780	1.25 1.44	0.33 0.20	1.17 1.28	0.03 0.05
C2S	∑ Actn4	P57780	1.36	0.16	1.23	0.03
C3a	Arp2/1	P61161	0.90	0.06	0.76	< 0.02
C3b	Arp2/2	P61161	1.10	0.41	0.94	0.63
C3c	Arp2/3	P61161	1.19	0.04	1.32	< 0.01
C3S	$\sum$ Arp2	P61161	1.15	0.07	1.14	0.06
C4a	Arpc2/1	Q9CVB6	0.57	0.03	0.52	0.02
C4b	Arpc2/2	Q9CVB6	0.76	0.04	0.78	0.06
C4c	Arpc2/3	Q9CVB6	0.91	0.51	1.12	0.41
C4S	∑Arpc2	Q9CVB6	0.80	0.11	0.90	0.39
C5a	Arpc5/1	Q9CPW4	0.59	0.03	0.25	0.01
C5b	Arpc5/2	Q9CPW4	1.06	0.51	0.99	0.90
C5S	∑ Arpc5	Q9CPW4	0.88	0.30	0.71	0.04
C6a	Capg/1	P24452	0.81	0.01	1.00	0.97
C6b	Capg/2	P24452	1.16	0.15	1.04	0.58
C6S C7	∑ capg	P24452	0.35	0.39	0.35	0.39
C8a	Capza2 Capzb/1	P47754 P47757	1.30 1.07	0.01 0.37	1.28 1.20	0.01 0.06
C8b	Capzb/1 Capzb/2	P47757	0.80	0.02	1.01	0.95
C8S	$\sum$ capzb	P47757	0.98	0.60	1.13	0.14
C9a	Cof/1	P18760	0.67	0.05	0.60	0.03
C9b	Cof/2	P18760	1.49	0.01	1.26	0.10
C9c	Cof/3	P18760	0.74	< 0.01	0.73	< 0.01
C9S	$\sum$ Cof	P18760	0.86	0.02	0.79	< 0.01
C10a	Dyncli2/1	O88487	1.21	0.49	2.20	< 0.01
C10b	Dyncli2/2	O88487	1.43	0.26	1.69	0.04
C10S	$\sum$ dyncli2	O88487	1.36	0.30	1.84	0.01
C11a	Gelsolin/1	P13020	1.16	0.44	0.94	0.71
C11b	Gelsolin/2	P13020	1.16	0.36	1.00	0.97
C11c	Gelsolin/3	P13020	1.39	0.16	1.27	0.26
C11d	Gelsolin/4	P13020	1.57	0.05	1.32	0.13
C11S	∑ gelsolin	P13020	1.33	0.15	1.15	0.39
C12 C13	Gmfg Lasp1	Q9ERL7 Q61792	$0.54 \\ 0.41$	<0.01 <0.01	0.66 0.38	<0.01 <0.01
C13	Ml12b	Q3THE2	0.92	0.51	0.81	0.01
C15a	Moesin/1	P26041	0.94	0.75	0.86	0.41
C15b	Moesin/2	P26041	0.90	0.53	0.78	0.08
C15c	Moesin/3	P26041	1.22	0.17	0.89	0.13
C15d	Moesin/4	P26041	1.18	0.07	1.09	0.10
C15e	Moesin/5	P26041	1.52	0.03	1.24	0.04
C15S	$\sum$ Moesin	P26041	1.20	0.08	1.01	0.91
C16	Mtap	Q9CQ65	1.04	0.83	1.30	0.02
C17	RhoA	Q9QUI0	0.59	< 0.01	0.76	0.02
C18a	Rhogdi1/1	Q99PT1	0.71	< 0.01	0.67	< 0.01
C18b	Rhogdi1/2	Q99PT1	1.16	0.13	1.11	0.08
C18S	∑ RhoGdi1	Q99PT1	0.99	0.85	0.95	0.28
C19a	Rhogdi2/1	Q61599	0.64	<0.01	0.54	< 0.01
C19b	Rhogdi2/2	Q61599	1.00	0.99	0.95	0.17
C19S	$\sum$ rhogdi2 Stmn/1	Q61599	0.87 0.84	0.02 0.19	0.80 0.70	$< 0.01 \\ 0.01$
C20a C20b	Stmn/2	P54227 P54227	0.60	0.19	0.53	0.01
C20b C20c	Stmn/3	P54227 P54227	0.88	0.02	0.76	0.01
C20S	$\sum$ Stmn	P54227	0.83	0.05	0.70	0.01
C21a	Sw70/1	Q6A028	1.63	0.02	1.50	0.20
C21b	Sw70/2	Q6A028	2.02	0.02	1.76	0.08
C21S	$\sum \text{Sw70}$	Q6A028	1.77	0.01	1.60	0.14
C22	Tbcb	Q9D1E6	1.31	0.03	1.11	0.45
C23a	Tetp/1	P63028	0.60	< 0.01	0.54	< 0.01
C23b	Tctp/2	P63028	0.84	0.03	0.85	0.05
C23S	$\sum$ Tetp	P63028	0.84	0.03	0.72	< 0.01
C24a	Twf1/1	Q91YR1	0.58	< 0.01	0.93	0.39

Table 2 (continued)

Spot number	Protein name	Accession number	Ratio acute/ctrl	T test acute vs. ctrl	Ratio recov/ctrl	$T$ test recov $\nu s$ . ctr
C24b	Twf1/2	Q91YR1	0.25	0.01	0.86	0.52
C24S	$\sum$ Twf1	Q91YR1	0.40	< 0.01	0.89	0.47
C25	Twf2	Q9Z0P5	0.70	0.01	0.60	0.01
C26	Vime	P20152	1.54	0.05	1.35	0.04
C27a	Vinculin/1	Q64727	1.28	0.25	1.24	0.11
C27b	Vinculin/2	Q64727	1.32	0.07	1.16	0.24
C27c	Vinculin/3	Q64727	1.63	0.01	1.12	0.28
C27S	∑ Vinculin	Q64727	1.39	0.07	1.18	0.16
D1	Cbx1	P83917	0.91	0.27	0.72	< 0.01
D2a	Ddb1/1	Q3U1J4	1.74	0.04	1.40	0.16
D2b	$\frac{\mathrm{Ddb1}/2}{\sum \mathrm{Ddb1}}$	Q3U1J4	0.93	0.65	0.78	0.21
D2S D3	Nt5c	Q3U1J4 Q9JM14	1.31 1.18	0.14 0.08	1.07 1.39	0.73 0.03
04a	Pena/1	P17918	0.57	<0.01	0.57	<0.03
D4a D4b	Pena/2	P17918	0.82	0.02	0.80	0.03
D4c	Pena/3	P17918	1.00	0.92	0.98	0.73
D4S	∑ Pena	P17918	0.84	<0.01	0.82	0.02
D5	Ruvb2	Q9WTM5	1.46	0.06	1.42	0.02
E1a	6pgd/1	Q9DCD0	0.88	0.05	0.95	0.42
E1b	6pgd/2	Q9DCD0	1.05	0.32	1.01	0.90
E1S	$\sum 6$ pgd	Q9DCD0	0.98	0.45	0.99	0.82
E2a	Aacs/1	Q9D2R0	1.79	0.11	1.39	0.52
E2b	Aacs/2	Q9D2R0	3.12	0.01	2.56	< 0.01
E2S	$\sum$ Aacs	Q9D2R0	2.17	0.01	1.73	0.16
E3	Acadl	P51174	1.13	0.14	1.27	0.01
E4a	Eno1a/1	P17182	0.99	0.96	0.89	0.28
E4b	Eno1a/2	P17182	1.10	0.50	0.89	0.42
E4c	Eno1a/3	P17182	1.18	0.10	1.21	0.10
E4d	Eno1a/4	P17182	1.50	< 0.01	1.42	< 0.01
E4S	∑ Eno1A	P17182	1.26	0.02	1.19	0.06
E5	GalK	Q9R0N0	1.45	<0.01	1.50	0.03
E6	Gapdh	P16858	0.56	<0.01	0.60	< 0.01
E7	Gpd1L	Q3ULJ0	2.27	<0.01	1.42	0.14
E8a	Hxk3/1	Q3TRM8	1.91	0.02	1.80	0.03
E8b	Hxk3/2	Q3TRM8	1.23	0.04	1.22	0.09
E8c	Hxk3/3	Q3TRM8	1.52	0.02	1.52	0.02
E8S	∑ Hxk3	Q3TRM8	1.51	0.01	1.49	0.01
E9	Idhe	O88844	1.24	0.02	1.19	< 0.01
E10a	Ldha/1	P06151	0.79	0.02	0.86	0.06
E10b	Ldha/2	P06151	0.97	0.81	0.89	0.40
E10S	$\sum$ Ldha	P06151	0.89	0.26	0.88	0.21
E11a	Mdhc/1	P14152	0.47	0.02	0.72	0.11
E11b	Mdhc/2	P14152	0.83	0.01	0.84	0.01
E11c	Mdhc/3	P14152	1.38	0.02	1.40	< 0.01
E11S	$\sum$ Mdhc	P14152	1.01	0.91	1.05	0.20
E12	Pckgm	Q8BH04	1.30	0.20	0.61	0.01
E13	Pfkap	Q9WUA3	0.84	0.01	0.86	0.01
E14	Pfkl	P12382	1.83	0.01	1.62	0.03
E15a	Pgls/1	Q9CQ60	0.78	0.02	0.97	0.75
E15b	Pgls/2	Q9CQ60	1.11	0.16	1.22	0.01
E16	Pgp	Q8CHP8	1.37	0.02	1.60	0.05
E17a	Taldo1	Q93092	0.80	0.02	0.66	< 0.01
E17b	Taldo/2	Q93092	1.17	0.12	1.18	0.08
E17c	Taldo/3	Q93092	1.24	0.02	1.23	< 0.01
E17d	Taldo/4	Q93092	0.81	0.06	0.79	0.02
E17e	Taldo/5	Q93092	1.22	0.06	1.36	0.02
E17S	∑ Taldo	Q93092	1.08	0.16	1.09	0.06
E18a	Tpis/1	P17751	0.74	0.01	0.82	0.04
E18b	Tpis/2	P17751	1.04	0.61	1.20	0.09
E18c	Tpis/3	P17751	1.45	0.04	1.43	0.05
E18S	$\sum$ Tpis	P17751	1.09	0.29	1.16	0.17
F1	Cdk4	P30285	0.69	< 0.01	0.71	< 0.01
	Cdk6	Q64261	0.71	0.04	0.77	0.02
F2						
	Ndrg1/1	Q62433	1.32	0.09	1.52	0.04
F3a	Ndrg1/1 Ndrg1/2	Q62433 Q62433	1.32 1.38	0.09 0.01	1.52 1.33	$0.04 \\ 0.02$
F2 F3a F3b F3S		•				

Table 2 (continued)

Spot number	Protein name	Accession number	Ratio acute/ctrl	T test acute $vs.$ ctrl	Ratio recov/ctrl	$T$ test recov $\nu s$ . ctrl
F5a	Vma5a/1	Q99KC8	1.60	0.06	0.99	0.97
F5b	Vma5a/2	Q99KC8	1.31	0.11	1.19	0.10
F5c	Vma5a/3	Q99KC8	1.84	<0.01	1.59	0.02
F5S	∑ Vma5a	Q99KC8	1.54	0.02	1.25	0.12
G1	GalE	Q8R059	0.50	0.03	1.09	0.50
G2	Gnpda1	O88958	0.47	< 0.01	0.63	0.01
G3	Gt25c	Q8K297	1.53	0.01	1.48	0.01
G4	Mlec	Q6ZQI3	1.49	0.02	1.79	< 0.01
G5a	Naga/1	Q8JZV7	1.19	0.07	1.19	0.06
G5b	Naga/2	Q8JZV7	1.05	0.56	1.18	0.08
G5S	∑Naga	Q8JZV7	1.12	0.16	1.19	0.06
H1	Adh5	P28474	0.72	0.09	0.74	0.11
H2a	Aldr/1	P45376	0.57	0.01	0.66	0.01
H2b	Aldr/2	P45376	0.84	0.09	0.73	0.04
H2c	Aldr/3	P45376	1.18	0.04	1.25	0.01
H2S	∑Aldr	P45376	0.85	0.02	0.89	0.10
H3	Aldr2	P47738	1.16	0.08	1.23	< 0.01
H4a	Bvra/1	Q9CY64	0.54	0.02	0.62	0.03
H4b	Bvra/2	Q9CY64	1.23	0.08	1.30	0.06
H4S	∑ Bvra	Q9CY64	0.98	0.81	1.06	0.54
H5	Bvrb	Q923D2	1.45	0.01	0.96	0.77
H6	Ca13	Q9D6N1	0.62	0.01	0.83	0.11
H7a	Esd/1	Q9R0P3	0.56	< 0.01	0.64	0.01
H7b	Esd/2	Q9R0P3	0.78	< 0.01	0.92	0.15
H7S	∑Esd	Q9R0P3	0.73	< 0.01	0.86	0.02
H8	Frih	P09528	0.61	0.01	0.67	0.02
H9	Gclm	O09172	1.32	0.01	0.98	0.86
H10	Hmox2	O70252	1.53	< 0.01	1.03	0.69
H11a	Lgul/1	Q9CPU0	0.80	0.04	0.77	0.05
H11b	Lgul/2	Q9CPU0	1.10	0.24	1.10	0.16
H11S	$\sum$ Lgul	Q9CPU0	1.00	0.99	1.00	0.96
L1	Anxa1	P10107	0.78	0.02	1.11	0.06
L2a	Anxa2/1	P07356	0.79	0.01	0.60	< 0.01
L2b	Anxa2/2	P07356	0.44	< 0.01	0.56	< 0.01
L2S	∑ Anxa2	P07356	0.67	< 0.01	0.59	< 0.01
L3	Anxa3	O35639	1.34	0.01	1.30	0.02
L4	Anxa4	P97429	1.30	0.01	1.33	< 0.01
L5a	Anxa5/1	P48036	0.88	0.02	0.81	< 0.01
L5b	Anxa5/2	P48036	1.09	0.34	1.07	0.37
L5S	∑ Anxa5	P48036	0.95	0.35	0.90	0.08
L6	Anxa6	P14824	1.54	0.02	1.80	< 0.01
L7	Anxa7	Q07076	1.14	0.44	1.57	0.02
L8	Idi1	P58044	1.40	0.02	1.34	0.01
L9	Lypla2	Q9WTL7	1.07	0.28	1.22	0.02
L10	Mbd3	Q9Z2D8	1.52	0.02	1.77	< 0.01
L11	Pipna	P53810	1.22	0.11	1.32	0.01
L12	Ppt1	O88531	1.24	0.14	1.34	0.03
M1a	Clic4/1	Q9QYB1	0.90	0.12	0.95	0.58
M1b	Clic4/2	Q9QYB1	1.27	0.02	1.27	0.04
M1S	∑ Clic4	Q9QYB1	1.12	0.04	1.14	0.01
M2	Clybl	Q8R4N0	1.80	0.03	1.41	0.19
M3	Gatm	Q9D964	0.68	0.01	0.54	0.01
M4	Hmgcl	P38060	0.61	< 0.01	0.76	0.07
M5	Nduv2	Q9D6J6	1.24	0.01	1.49	0.05
M6	Oat	P29758	1.30	0.05	1.50	0.01
M7	Odba	P50136	1.54	0.02	1.28	0.12
M8	Phb	P67778	1.12	0.16	1.18	0.05
M9	Tmem11	Q8BK08	1.78	0.04	1.46	0.41
M10	Vdac2	Q60930	0.63	0.02	0.72	0.05
M11a	Atpb ac	P56480	1.23	0.16	1.41	0.03
M11b	Atpb bas	P56480	1.60	0.04	1.58	< 0.01
M11S	∑Atpb	P56480	1.41	0.06	1.49	< 0.01
M12	Clpp	O88696	1.28	0.33	1.84	0.02
M13a	Trap1/1	Q9CQN1	1.74	0.02	1.32	0.13
M13b	Trap1/2	Q9CQN1	1.21	0.25	0.92	0.64
M13S	∑Trap1	Q9CQN1	1.47	0.03	1.11	0.51

Table 2 (continued)

Spot number	Protein name	Accession number	Ratio acute/ctrl	T test acute $vs.$ ctrl	Ratio recov/ctrl	T test recov vs. ctrl
N1b	Aprt/2	P08030	1.06	0.43	1.08	0.39
N1S	$\sum$ Aprt	P08030	0.92	0.19	0.99	0.85
N2	Bpnt1	Q9Z0S1	1.77	0.02	1.87	0.01
N3	Guaa	Q3THK7	1.42	0.02	1.12	0.25
N4	Hint1	P70349	1.46	0.30	0.64	0.02
N5	Ndka	P15532	0.42	0.03	0.75	0.22
N6	Paps1	Q60967	5.54	0.04	2.44	0.01
N7a	Pnph/1	P23492	0.83	0.12	0.72	0.05
N7b	Pnph/2	P23492	1.25	0.02	1.22	0.08
N7S	$\sum$ Pnph	P23492	1.08	0.17	1.02	0.76
N8a	Prps1/1	Q9D7G0	0.69	0.01	0.92	0.48
N8b	Prps1/2	Q9D7G0	0.57	<0.01	0.77	0.16
N8c	Prps1/3	Q9D7G0	0.52	0.01	0.83	0.21
N8S	$\sum$ Prps1	Q9D7G0	0.60	< 0.01	0.83	0.14
N9a	Pur4/1	Q5SUR0	1.71	0.14	1.27	0.07
N9b	Pur4/2	Q5SUR0	2.53	0.01	2.01	0.01
N9c	Pur4/3	Q5SUR0	1.15	0.53	1.25	0.22
N9S	∑ Pur4	Q5SUR0	1.65	0.07	1.43	0.01
N10	Pur9	Q9CWJ9	1.16	0.27	1.23	0.01
N11	Nt5c3b	Q3UFY7	1.55	< 0.01	1.64	0.01
O1	Prx1ox	P35700	0.58	0.02	0.70	0.04
O2a	Prx3/1	P20108	0.93	0.69	1.13	0.08
O2b	Prx3/2	P20108	1.23	0.06	1.32	0.01
O2S	∑ Prx3	P20108	1.16	0.22	1.27	0.01
O3	Prx4	O08807	0.75	0.06	0.56	0.01
O4a	Prx6/1	O08709	0.56	0.02	0.45	0.01
O4b	Prx6/2	O08709	1.05	0.64	0.97	0.76
O4S	$\sum Prx6$	O08709	0.96	0.63	0.87	0.23
P1	Aars	Q8BGQ7	1.65	0.01	1.18	0.20
P2a	Eef2/1	P58252	0.77	0.02	0.63	< 0.01
P2b	Eef2/2	P58252	1.06	0.41	0.96	0.39
P2c	Eef2/2 Eef2/3	P58252	0.84	0.07	0.90	0.20
P2d	Eef2/4	P58252	0.99	0.88	0.98	0.61
P2e	Eef2/5	P58252	1.39	0.01	1.29	0.01
P2S	$\sum \text{Eef2}$	P58252	1.02	0.72	0.97	0.43
P3	If3g	Q9Z1D1	0.77	0.72	0.56	0.43
P4a	If5a/1	P63242	0.54	<0.01	0.62	< 0.03
	If5a/2	P63242 P63242				
P4d			1.07	0.26	1.03	0.58
P4S	$\sum_{\mathbf{f} \in A} \mathbf{f} \mathbf{f} 5 \mathbf{a}$	P63242	0.86	0.04	0.87	0.04
P5	If4a1	P60843	1.55	0.01	1.51	0.01
P6	If4a3	Q91VC3	1.27	0.05	1.09	0.28
P7	Rla0	P14869	1.28	0.07	1.33	0.04
P8	Rs4y1	P62702	0.50	0.04	0.62	0.07
P9a	Sars/1	P26638	1.42	0.03	1.05	0.59
P9b	Sars/2	P26638	1.52	0.04	1.10	0.42
P9S	$\sum$ Sars	P26638	1.48	0.02	1.08	0.46
Q1a	Dpp3/1	Q99KK7	1.28	0.22	1.75	<0.01
Q1b	Dpp3/2	Q99KK7	1.55	0.04	1.84	< 0.01
Q1S	∑ Dpp3	Q99KK7	1.40	0.09	1.79	< 0.01
Q2a	Hsp74/1	Q61316	0.77	0.22	0.77	0.23
Q2b	Hsp74/2	Q61316	1.33	0.08	1.18	0.23
Q2c	Hsp74/3	Q61316	1.31	0.09	1.14	0.29
Q2d	Hsp74/4	Q61316	1.26	0.15	1.31	< 0.01
Q2S	$\sum$ Hsp74	Q61316	1.07	0.58	1.03	0.80
Q3a	Hyou1/1	Q9JKR6	1.18	0.25	0.83	0.20
Q3b	Hyou1/2	Q9JKR6	1.38	0.01	1.18	0.03
Q3c	Hyou1/3	Q9JKR6	1.35	0.01	1.21	0.05
Q3d	Hyou1/4	Q9JKR6	0.80	0.23	0.79	0.11
Q3S	∑ Hyou1	Q9JKR6	1.04	0.67	0.93	0.27
Q4	Lxn	P70202	0.98	0.80	1.26	< 0.01
Q5	Pfd2	O70591	0.72	0.01	0.70	0.01
Q6a	Ppce/1	Q9QUR6	1.40	0.15	1.78	0.01
Q6b	Ppce/2	Q9QUR6	1.58	0.04	1.66	0.01
Q6S	$\sum$ Ppce	Q9QUR6	1.50	0.07	1.71	0.01
Q7a	Ppia/1	P17742	0.68	0.11	0.57	0.05
Q7b	Ppia/2	P17742	0.80	0.21	0.79	0.03
4. n	1 P14/2	- 1// 14	3.00	0.21	3.7 3	0.21

Table 2 (continued)

Spot number	Protein name	Accession number	Ratio acute/ctrl	T test acute vs. ctrl	Ratio recov/ctrl	T test recov vs. ctrl
Q7S	∑ Ppia	P17742	0.78	0.09	0.83	0.21
Q8a	Spb6/1	Q60854	1.66	0.02	1.49	0.01
Q8b	Spb6/2	Q60854	1.43	0.01	1.42	0.01
Q8S	∑ Spb6	Q60854	1.57	0.01	1.46	<0.01
Q9	Stip1	Q60864	1.72	<0.01	0.94	0.67
Q10	Tcpa	P11983	1.55	0.02	1.09	0.55
R1a	Btf3/1	Q64152	0.56	0.01	0.50	0.01
R1b	Btf3/2	Q64152	0.76	0.05	0.76	0.05
R1S R2	∑ Btf3	Q64152 Q9DAR7	0.66 1.30	0.02 < 0.01	0.64 $1.20$	0.01 0.02
R3	Dcps Ddx39a	Q8VDW0	1.64	0.03	1.31	0.02
R4	Ddx39b	Q9Z1N5	1.59	0.03	1.35	0.02
R5	Mgn	P61327	0.96	0.60	0.56	0.01
R6a	Sf3b2/1	Q3UJB0	1.03	0.91	0.87	0.49
R6b	Sf3b2/1 Sf3b2/2	Q3UJB0	1.16	0.44	1.32	< 0.01
R6c	Sf3b2/2 Sf3b2/3	Q3UJB0	1.26	0.38	1.80	<0.01
R6S	$\sum \text{Sf3b2}$	Q3UJB0	1.13	0.54	1.27	0.03
R7	ExoS4	Q921I9	0.88	0.42	0.51	0.03
R8	ExoS6	Q8BTW3	1.50	0.02	1.53	0.02
S1a	14-3-3 gam/1	P61982	0.85	0.28	0.67	0.02
S1b	14-3-3 gam/1 14-3-3 gam/2	P61982	0.83	0.28	0.07	0.24
S1S	$\sum 14-3-3 \text{ gam/}2$	P61982	0.84	0.10	0.82	0.07
S2	∠ 14-3-3 gain 14-3-3 th	P68254	1.31	0.10	1.19	0.04
S3	Cab39	Q06138	0.70	0.03	0.62	< 0.01
S4	Fam49b	Q921M7	1.47	0.03	1.59	0.05
S5	Gbb1	P62874	1.17	0.02	1.18	0.03
S6a	Gnai2/1	P08752	1.33	0.06	1.42	0.05
S6b	Gnai2/2	P08752	1.51	<0.01	1.67	0.03
S6S	∑ Gnai2	P08752	1.45	0.01	1.59	0.02
S7	Gnb2L1	P68040	0.89	0.02	0.81	<0.01
S8	Grb2	Q60631	1.06	0.83	1.46	0.02
S9	Igbp1	Q61249	1.36	0.06	1.36	0.04
S10	In35	Q9D8C4	1.25	0.04	1.21	0.14
S11	Inpp	P49442	2.46	0.01	1.76	0.17
S12	Ppp1ca	P62137	1.17	0.07	1.27	0.02
S13a	Ppp1r7/1	Q3UM45	0.30	0.02	0.38	0.04
S13b	Ppp1r7/2	Q3UM45	0.67	0.02	0.73	0.08
S13c	Ppp1r7/3	Q3UM45	0.90	0.51	0.94	0.72
S13S	$\sum Ppp1r7$	Q3UM45	0.61	0.03	0.68	0.06
S14	Ppp6	O00743	1.48	0.03	1.04	0.82
S15a	Snd1/1	Q78PY7	0.65	0.03	0.66	0.04
S15b	Snd1/2	Q78PY7	0.59	0.02	0.50	0.01
S15c	Snd1/3	Q78PY7	0.89	0.34	0.78	0.18
S15d	Snd1/4	Q78PY7	1.11	0.34	0.91	0.47
S15S	$\sum$ Snd1	Q78PY7	0.79	< 0.01	0.70	0.01
U1	Chip	Q9WUD1	0.59	< 0.01	0.73	0.11
U2	Csn4	O88544	1.34	0.03	1.37	0.03
U3a	Prs8/1	P62196	0.60	< 0.01	0.65	< 0.01
U3b	Prs8/2	P62196	0.85	0.04	0.85	0.22
U3S	∑ Prs8	P62196	0.31	0.33	0.32	0.34
U4a	Psa2/1	P49722	0.58	< 0.01	0.58	< 0.01
U4b	Psa2/2	P49722	1.15	0.12	1.28	0.07
U4S	∑ Psa2	P49722	0.86	0.04	0.92	0.36
U5a	Psa5/1	Q9Z2U1	0.80	0.01	0.84	0.11
U5b	Psa5/2	Q9Z2U1	0.92	0.33	0.99	0.83
U5S	∑ Psa5	Q9Z2U1	0.86	0.07	0.92	0.22
U6	Psb10	O35955	0.73	0.12	0.43	0.02
U7a	Psb2/1	Q9R1P3	0.67	0.02	0.76	0.14
U7b	Psb2/2	Q9R1P3	0.90	0.36	1.01	0.91
U7S	∑ Psb2	Q9R1P3	0.82	0.09	0.92	0.50
U8	Psb3 mod	Q9R1P1	0.47	0.01	0.65	<0.01
U9a	Psb4/1	P99026	0.74	0.05	0.69	0.03
U9b	Psb4/2	P99026	0.98	0.71	0.94	0.25
U9S	$\sum Psb4$	P99026	0.90	0.05	0.86	0.01
U10	Psmd14	O35593	1.03	0.61	1.32	<0.01
U11a	Psmd2/1	Q8VDM4	0.82	0.22	1.12	0.24
	/ -					

Table 2 (continued)

Spot number	Protein name	Accession number	Ratio acute/ctrl	T test acute $vs.$ ctrl	Ratio recov/ctrl	T test recov vs. ctr
U11c	Psmd2/3	Q8VDM4	1.39	0.02	1.13	0.17
U11S	$\sum Psmd2$	Q8VDM4	1.03	0.70	1.12	0.12
U12	Psmd7	P26516	0.68	0.03	0.68	0.02
U13	Psme1	P97371	1.40	< 0.01	1.40	< 0.01
U14a	Psme2/1	P97372	1.00	0.98	0.71	< 0.01
U14b	Psme2/2	P97372	1.22	< 0.01	1.07	0.34
U14S	∑ Psme2	P97372	1.15	< 0.01	0.96	0.50
U15	Ubl7	Q91W67	0.41	0.02	0.88	0.55
V1	Asna1	O54984	1.29	0.09	1.46	< 0.01
V2	Chm2a	Q9DB34	0.83	< 0.01	0.92	0.09
V3	CopE	O89079	1.35	0.01	1.41	0.07
V4	Emc2	Q9CRD2	0.56	0.01	0.84	0.07
V5	Erp29	P57759	0.78	0.12	0.68	0.05
V6	Mss4	Q91X96	0.52	0.04	0.64	0.10
V7	Nsf1c	Q9CZ44	1.55	0.01	1.51	0.01
V8	Pef1	Q8BFY6	0.64	0.05	0.50	0.02
V9	Snaa	Q9DB05	1.53	< 0.01	1.99	0.05
V10	Snap23	O09044	0.53	0.02	0.34	0.01
V11	Stxb2	Q64324	2.07	< 0.01	1.30	0.09
V12a	Tera/1	Q01853	1.31	0.13	1.18	0.16
V12b	Tera/2	Q01853	1.49	0.02	1.06	0.38
V12c	Tera/3	Q01853	1.39	0.02	1.29	0.04
V12d	Tera/4	Q01853	1.48	0.10	1.35	< 0.01
V12S	∑ Tera	Q01853	1.43	0.02	1.25	0.01
V13	Tpd52	Q62393	1.42	0.04	1.31	0.10
V14	Vat1	Q62465	1.48	0.01	1.22	0.15
V15	Vps29	Q9QZ88	1.30	0.05	1.29	0.08
V16a	Vps35/1	Q9EQH3	2.44	0.16	2.33	< 0.01
V16b	Vps35/2	Q9EQH3	1.80	0.17	2.03	0.01
V16c	Vps35/3	Q9EQH3	1.53	0.33	2.25	< 0.01
V16S	$\sum$ Vps35	Q9EQH3	1.79	0.18	2.20	< 0.01
V17	Vta1	Q9CR26	2.84	0.01	2.97	0.02
X1	Pdxk	Q8K183	1.29	< 0.01	1.45	0.02
X2	Spee	Q64674	1.55	0.02	1.43	0.07
X3	Sps1	Q8BH69	1.49	0.02	1.95	0.01
X4a	Ran/1	P62827	0.83	0.04	0.87	0.12
X4b	Ran/2	P62827	1.18	0.08	1.11	0.17
X4S	∑ Ran	P62827	1.03	0.62	1.00	0.98
Y1	CatD	P18242	1.38	0.01	1.48	< 0.01
Y2	CatZ	Q9WUU7	1.79	0.04	1.08	0.85
Y3	Ncf4	P97369	1.15	0.02	1.19	0.01

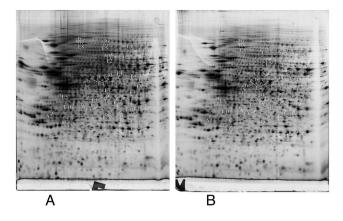
∑: sum of the different spots identified for the same protein.

which is known to be up-regulated by acetylation.71 In our case, the activity correlated with the amount of the acidic, modified spots and not with the total amount of the protein. For other activities, e.g. lactate dehydrogenase, triose phosphate isomerase and purine phosphorylase, the activity correlated with the total protein amount and not with a single spot. For several other enzymes such as enolase, hexokinase, pyridoxal kinase and isocitrate dehydrogenase, we detected an increase of the protein amount by proteomics and a stable or decreased enzyme activity.

## 3.5 Mitochondrial potential and glucose consumption

Several mitochondrial proteins appeared modulated upon exposure of macrophages to silver nanoparticles, among which are a few subunits of the respiratory complexes (NDUV2 and ATPB), some components of the mitochondrial protein quality control (TRAP1 and CLPP) and an antioxidant protein (PRX3). As this may suggest perturbations in the mitochondrial functions, we investigated the mitochondrial transmembrane potential. The results, shown in Fig. 5, indicated that all viable cells accumulated rhodamine 123 and thus had a strong mitochondrial transmembrane potential. However, the amount of accumulated rhodamine 123 and thus the level of the mitochondrial transmembrane potential were lower for the acute condition and went back to normal after 72 hours of recovery. As a lower transmembrane potential is indicative of a less efficient oxidative phosphorylation chain, we investigated the global metabolism under the three experimental conditions.

We thus measured the glucose consumption during the last 36 hours of the experiments by measuring the remaining glucose in the culture medium at the end of the experiment. The initial medium contained 4.1 g of glucose per liter due



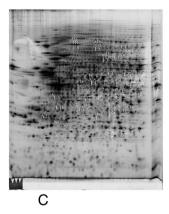


Fig. 4 Proteomic analysis of total cell extracts by 2D electrophoresis. Total cell extracts of RAW274.7 cells were separated by two-dimensional gel electrophoresis. The first dimension covered a 4-8 pH range and the second dimension a 15-200 kDa range. Total cellular proteins (150  $\mu$ g) were loaded on the first dimension gel. A: Gel obtained from control cells. B: Gel obtained from cells treated for 24 hours with 20 µg ml<sup>-1</sup> silver nanoparticles. C: Gel obtained from cells treated for 24 hours with 20 µg ml<sup>-1</sup> silver nanoparticles and left to recover for 72 hours. Only the proteins involved in apoptosis (A), energy metabolism (E) and control of cell fate and proliferation (F) are shown in this figure.

to the dilution brought by the addition of fetal serum to obtain the complete medium. In the untreated, control cells,  $1.8 \pm 0.2$  g of glucose per liter remained in the culture medium after 36 hours of culture. In contrast,  $1 \pm 0.05$  g of glucose and  $0.9 \pm 0.04$  g of glucose remained in the medium for the acute and recovery conditions, respectively. These differences were statistically significant (Mann Whitney U test, p < 0.05).

The increased glucose consumption in the acute exposure condition could be correlated with the less efficient mitochondria via a Warburg effect. However, this correlation did not hold for the recovery condition, in which the mitochondria appear as efficient as those of the control cells. This may suggest that the silver expulsion process that takes place during the recovery period consumes a lot of energy.

#### 3.6 Glutathione levels

One of the proteins that is induced after acute exposure to silver nanoparticles is GCLM, i.e. the regulatory subunit of the enzyme involved in the first step of glutathione biosynthesis, which is the limiting step of the pathway.<sup>72</sup> We thus investigated the levels of free glutathione, as studies on primary macrophages had shown that treatment with silver nanoparticles decreased the levels of free glutathione.<sup>23</sup> This result may be due to the formation of silver-glutathione complexes.<sup>73</sup> We thus measured by flow cytometry the proportion of cells with high levels of reduced glutathione, and the level of glutathione in this population. The results, shown in Fig. 6, indicated that the glutathione level is only 70% of the normal one just after the exposure to silver nanoparticles, and is 85% of the normal after 3 days of recovery. Thus, here again, the increase in GCLM amount can be interpreted as a cellular mechanism to compensate for the decrease of free glutathione caused by silver.

## 3.7 Actin cytoskeleton and phagocytosis

Cytoskeletal proteins, especially proteins implicated in the actin cytoskeleton dynamics, were among the most important classes emerging from the proteomic screen. Of note, many of these proteins have activities that are modulated by phosphorylation, such as cofilin<sup>74,75</sup> or Rho-GDIs.<sup>76,77</sup> As the regulation patterns

Table 3 Enzyme activities

Enzyme	Control	Acute	Recov	Ctrl + silver ion
LDH	$74.4 \pm 2.75$	68.9 ± 1.14*	73.8 ± 4.13	ND
MDH	$37.4 \pm 0.95$	$31.3 \pm 2.85*$	$31.7 \pm 2.36*$	ND
TPIS	$70.7 \pm 8.75$	$76.6 \pm 5.58$	$76.3 \pm 7.80$	ND
BVR	$0.67 \pm 0.04$	$0.53 \pm 0.1^*$	$0.79 \pm 0.06*$	ND
PDXK	$7.4 \pm 0.95$	$5.9 \pm 0.6*$	$7.5 \pm 0.68*$	ND
GAPDH	$157.12 \pm 14.5$	$144.25 \pm 9.77$	$157.44 \pm 5.91$	ND
6PGDH	$30.69 \pm 3.17$	39.37 ± 4.92*	$38.75 \pm 1.24*$	ND
PNPH	$8.97 \pm 1.17$	$9.12 \pm 1.6$	$9.06 \pm 0.77$	ND
HXK	$23.4 \pm 3.04$	$16.3 \pm 4*$	$23.3 \pm 2.45$	$(1 \mu M) 14.2 \pm 2.93$
ENO	$214.5 \pm 13.6$	$166.4 \pm 28.1^*$	$184 \pm 41$	$(10 \mu M) 167 \pm 12.4$
IDHC	$16.37 \pm 1.48$	$17.62 \pm 2.79$	23.75 ± 2.97**	$(5 \mu M) 8.06 \pm 1.09$

All the activities are expressed in nmol substrate converted per min per mg total protein. Statistical significance of the results in the Student T test: \*p < 0.05. ND: not determined. Abbreviations: LDH: lactate dehydrogenase; MDH: malate dehydrogenase; TPIS: triose phosphate isomerase; BVR: biliverdine reductase; PDXK: pyridoxal kinase; GAPDH: glyceraldehyde phosphate dehydrogenase; 6PGDH: 6-phosphogluconate dehydrogenase; PNPH: purine nucleoside phosphorylase; HXK: hexokinase; ENO: enolase; IDHC: NADPH-dependent isocitrate dehydrogenase.

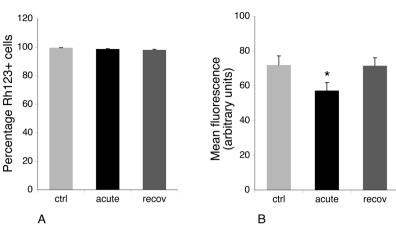


Fig. 5 Mitochondrial transmembrane potential analysis. The rhodamine 123 accumulation method was used. Panel A: Proportion of rhodamine123-positive cells in the viable cell population. Panel B: Mean rhodamine 123 fluorescence (in the viable cell population only). Symbols indicate the statistical significance (Student's T-test): \*: p < 0.05. ctrl: Non exposed cells. acute: Cells exposed for 24 hours to silver nanoparticles (20  $\mu$ g ml $^{-1}$ ). recov: Recovering cells (72 hours post exposure to silver nanoparticles). Symbols indicate the statistical significance (Student's T-test): \*: p < 0.05.

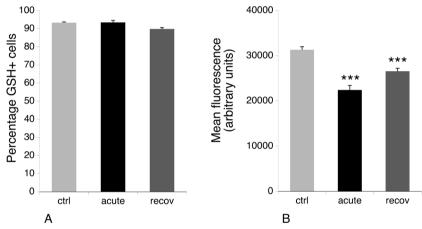


Fig. 6 Free glutathione levels. The chlorobimane (mCB) conjugation method was used. Panel A: Proportion of GSH-positive cells in the viable cell population. Panel B: Mean GSH-mCB conjugate fluorescence (in the viable cell population only). Symbols indicate the statistical significance (Student's T-test): \*\*\*: p < 0.001. ctrl: Non exposed cells. acute: Cells exposed for 24 hours to silver nanoparticles (20  $\mu$ g ml $^{-1}$ ). recov: Recovering cells (72 hours post exposure to silver nanoparticles).

that emerged from the proteomic screen for these proteins were very complex, with some spots modulated while others were constant, we directly examined the consequences on the actin cytoskeleton. The results, shown in Fig. 7, indicated a strong vesicularization of the cells during the recovery period.

Phagocytosis is one of the specialized functions of the macrophages that is highly dependent on the actin cytoskeleton. We thus tested the phagocytic activity of the cells. The results, shown in Fig. 8, indicated that the proportion of phagocytic cells does not change upon treatment with silver nanoparticles. However, the number of internalized particles per cell, as described by the mean fluorescence, slightly decreased immediately after exposure to nanoparticles and went back to quasi-normal values after 72 hours of recovery.

#### 3.8 NO and cytokine production, redox balance

Production of nitric oxide and pro-inflammatory cytokines such as IL-6 and TNF upon stimulation is another specialized

function of macrophages. We thus investigated their production after treatment of macrophages with silver nanoparticles under two schemes: either after treatment with nanoparticles only, or after treatment with nanoparticles and lipopolysaccharide (LPS) for the last 24 hours of culture. The first scheme investigated the intrinsic pro-inflammatory action of the nanoparticles, while the second one investigated the interference of nanoparticles with a standard inflammatory response induced by a bacterial stimulus. The results, shown in Fig. 9, indicated that silver nanoparticles have a weak but significant intrinsic pro-inflammatory effect, as detected from NO and TNF production, which returned to normal values after the recovery period. The situation was however very different for the combined nanoparticle-LPS treatment. In this case, the acute exposure condition did not differ significantly from the control cells (exposed to LPS only), while the cytokine and NO production were significantly lower after the recovery period. As the NO production

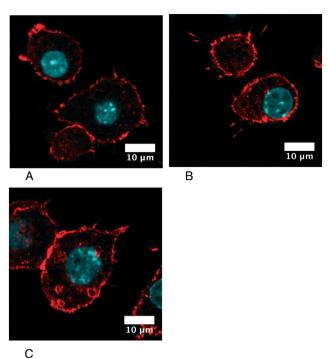


Fig. 7 Confocal imaging of the F-actin cytoskeleton. Only one confocal plane inside the cells is shown (going through the cell nucleus). A: Unexposed cells. B: Acutely-exposed cells. C: Recovering cells. Silvertreated cells show an important vesicularization, compatible with the presence of nanoparticles within the cell phagosomes.

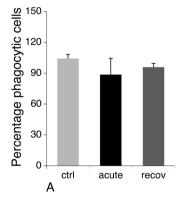
is NADPH-dependent, we investigated the NADP-NADPH levels after treatment with nanoparticles. The results, shown in Table 4, indicated that both the total NADP + NADPH and NADPH levels were similar for the three conditions tested.

## 4. Discussion

Silver nanoparticles are usually produced through a wet route and need to be stabilized as soon as prepared to avoid irreversible aggregation. Stabilization of AGNPs is usually achieved through interaction with complex ions such as citrate, or by polymers such as polyethylene glycol or polyvinylpyrrolidone (PVP). As such, PVP-coated silver nanoparticles are industrial products used in several applications including as antibacterial compounds, 78 conductive inks, 79-81 and sensors, 82 and in electromechanics. 83 In the course of their use, nanoparticles can be aerosolized. In this case they can contaminate living organisms without being modified. In the specific case of inhalation contamination, the nanoparticles reach the alveolae where there will be internalized by the lung macrophages, which is the reason why we made the choice to use this cell type in this study. Regarding other environmental dispersion routes, PVP-coated silver nanoparticles have been shown to be the most resistant in freshwater environments,84 and thus recommended for ecotoxicological testing.84 The PVP-coated silver nanoparticles, therefore, appear to be relevant for assessing the toxicity of silver nanoparticles.

Most of the toxicological studies on nanoparticles use an acute exposure scheme; they expose cells to a high, but nonlethal dose. Such a scheme does not represent chronic daily exposure under, for example, occupational conditions, but may represent the dose following an accidental exposure. To fully evaluate the toxicity of AgNPs, it is therefore important to know not only the acute effects post exposure, but also how cells recover from such exposures to high dosages of AgNPs. To accomplish this, a ternary comparison must be made between cells before and immediately after treatment and also between cells at the end of a recovery period. In this experimental frame, the acute exposure point is not the primary focus of attention. We therefore decided not to investigate in detail the role of silver ions in the acute response, based on the fact that the rather large silver nanoparticles used in this study dissolve to a very low extent, 23 producing free silver ion concentrations that are far below the LD20 observed for silver ions. Furthermore, this aspect has been investigated in numerous studies (e.g. in ref. 8, 15, 59 and 85) and the general outcome of these studies is that the mechanisms observed for silver nanoparticles cannot be explained by silver ions, at least for large nanoparticles.<sup>15</sup>

In this frame, the proteomic approach is of interest because it can be used to explore not only a few parameters, as



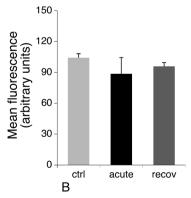


Fig. 8 Phagocytosis. The phagocytic capacity was assessed by fluorescent latex bead internalization. Panel A: Proportion of positive cells in the viable cell population. Panel B: Mean cellular fluorescence (in the viable cell population only). ctrl: Non exposed cells. acute: Cells exposed for 24 hours to silver nanoparticles (20  $\mu$ g ml<sup>-1</sup>). recov: Recovering cells (72 hours post exposure to silver nanoparticles).

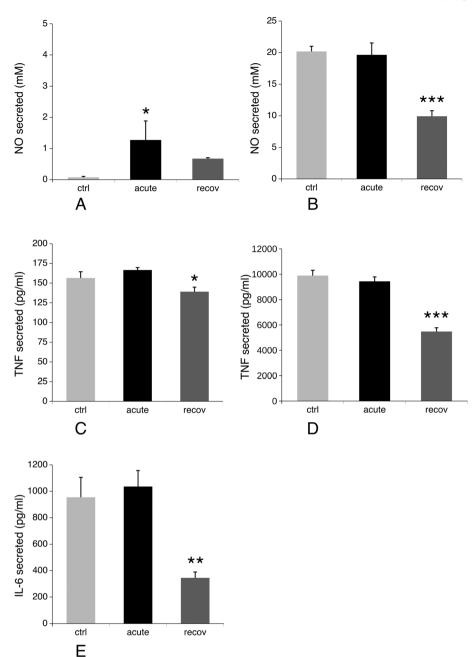


Fig. 9 NO and cytokine production. NO, TNF-alpha and IL-6 production were measured under the different cell culture conditions, with and without further exposure to LPS. ctrl: Non exposed cells. acute: cells exposed for 24 hours to silver nanoparticles (20  $\mu$ g ml $^{-1}$ ). recov: Recovering cells (72 hours post exposure to silver nanoparticles). Symbols indicate the statistical significance (Student's *T*-test): \*: p < 0.05; \*\*: p < 0.01; \*\*\*: p < 0.001. A and B: NO production, without (panel A) and with (panel B) exposure to LPS. C and D: TNF-alpha production, without (panel C) and with (panel D) exposure to LPS. E: IL-6 production with exposure to LPS. IL-6 is undetectable under the three culture conditions tested when the cells are not exposed to LPS.

targeted approaches do, but also can be used to investigate a few hundred parameters at the same time, *i.e.* the abundances of the various proteins analyzed in a proteomic screen. In this respect, the use of a 2D gel-based proteomic approach may appear as a medium scale approach, compared to deep shotgun proteomic approaches. However, compared to shotgun proteomics, 2D gels have the unique ability to separate protein forms, without the requirement that they appear as a single product. With growing recognition of the im-

portance of post-translational modifications, the ability to do this is biologically relevant. Indeed, we have shown both in this study and in previous ones<sup>53,86</sup> that some enzyme activities correlate with a single protein form and not with the sum of all the protein forms, showing the relevance of this parameter.

First, we performed a global analysis of the proteomic results. Both the principal component analysis and the analysis of similarity (ANOSIM) indicated that the recovery state was

Table 4 NADP(H) concentrations in cells

	Ctrl	Acute	Recov
NADPH	$1.87 \pm 0.12$	$1.88 \pm 0.03$	$1.85 \pm 0.01$
NADP + NADPH	$2.5 \pm 0.09$	$2.33 \pm 0.04$	$2.62 \pm 0.06$
NADPH/(NADP + NADPH)	$0.75 \pm 0.07$	$0.81 \pm 0.02$	$0.71 \pm 0.02$

All the concentrations are expressed in nmol nucleotide per mg total protein.

not just an intermediate state between the unexposed cells and the acutely-exposed cells. If this were the case, the recovery state should not have been separated from the other two states as it is in the principal component analysis. This is further confirmed by ANOSIM. The fact that all the p-values in the binary comparison are low and rather similar shows that the three states are significantly different from each other. If the recovery stage was just between the other two stages, at least one of the p-values between the recovery stage and one of the other two stages should be much higher than the p-value between the control and the acute stage. As this is not the case, we can reject the hypothesis that the recovery phase is only an intermediate between the control and acutely-treated stages.

We then performed a more detailed analysis of the proteomic results. The first important challenge was to determine to what extent the changes observed through proteomics are specific to silver nanoparticles and which are due to the simple presence of a PVP-coated nanoparticle. We have addressed this question in previously published studies.35,53 In these studies, we showed that PVP-coated zirconium oxide nanoparticles induced minimal changes at the proteomic level. This ruled out the possibility that the changes that we observe in the present study could be due either to the internalization process per se or are due to the effect of the addition of PVP.

As a further step, we compared our results obtained immediately after exposure to those published in the literature. 9,10,24 It should be noted that this comparison is challenging because these other studies did not use the same cell type as was used here (although two of them use human intestinal cell lines, however different ones<sup>9,24</sup>) or the same silver nanoparticles. Citrate-coated silver nanoparticles were used in one study,<sup>24</sup> while the other two used surfactantcoated nanoparticles, 9,10 and in this study PVP-coated nanoparticles were used. These differences are very likely to explain the wide differences observed between the studies in which proteins were differentially modulated by treatment, with a minimal overlap between the results. Despite this general trend, several convergences could be observed between our study and those previously published. For example, increases in the abundances of Ran, Moesin, RuvB2, EfhD2, Pfkl, Anxa3, Tcpa and Tbcb were observed both in our study and the one published by Verano-Braga et al.24

However, proteomics is prone to multiple testing issues, so that the results obtained by proteomics must be verified by independent, targeted experiments. To accomplish this, we analyzed enzyme activities. Our enzyme activity results were consistent with observations made about changes at the proteomic level. However, a few discrepancies occurred, which were always of the same type: increases in the protein level by proteomics, corresponding to stable or decreased activities. We could, however, show that these enzyme activities are very sensitive to the silver ion, which was released within cells during their exposure to silver nanoparticles. Thus, the increase in protein levels can be seen in such cases as a cellular mechanism to compensate for the decrease of activity brought by silver.

We then carried out indirect validation studies, for example on the free glutathione levels, which were decreased immediately after exposure, a phenomenon that was previously observed for copper nanoparticles.35 We also tested phagocytosis, which is an important function of macrophages and is important for clearing bacteria. For this assay we used micron-sized fluorescent beads, which functioned as bacterial mimics. This allowed us to investigate whether nanoparticletreated cells were still able to clear bacteria. The proteomic screen suggested that phagocytosis might be altered after treatment with silver nanoparticles, which was confirmed in our assay testing the ability of the cell to clear bacteria. These results were previously observed for primary macrophages.<sup>23</sup>

The key question asked in this study, however, does not revolve around the acute response to silver nanoparticles, but instead asks to what extent and by which means do cells recover after such an exposure. In a simple model for recovery, protein changes should exhibit trends that return to normal, i.e. the amplitude of the change in abundance between the recovery and control stages should be lower than the amplitude of the change in abundance between the acute exposure and control stages. Out of the 239 spots that are highlighted in the proteomic screen, 135 (56%) show such a trend. This means that 104 (44%) show a stronger response at the recovery phase than immediately after exposure. With such a split trend, it is interesting to evaluate the results of targeted experiments during the recovery phase.

Interestingly, this split trend was also apparent in the results of targeted experiments. General metabolic processes seem to return to normal (mitochondrial potential, glutathione levels, and most enzymatic activities), as well as phagocytosis and basal NO production (without LPS stimulation). Oppositely, the LPS-induced activities (e.g. NO, IL-6 and TNF production) worsen during the recovery phase, as well as the isocitrate dehydrogenase activity, and the glucose consumption, which stays much higher at both the recovery and acute exposure phases than in unexposed cells.

It was the necessary to check in detail how proteomic changes correlate with observed functional changes. For example, for mitochondrial proteins, the proteomic screen detects an increase in the beta subunit of ATP synthase (ATPB) and an increase in one subunit of respiratory complex I (NDUV2) upon cell exposure to silver nanoparticles. At the same time the mitochondrial transmembrane potential

decreases slightly, showing that this increase in mitochondrial proteins could be an attempt to restore normal energy production in cells. Upon recovery, one of the ATPB spots slightly decreases compared to the value immediately after exposure, while the other spot further increases. The NDUV2 spot is also enhanced. However, even the ATPB spot that decreased is still significantly more abundant at this stage of recovery than in control cells. This shows that the cellular situation is still not back to normal, as can be expected from the still high intracellular silver content, and that restoring the metabolic activity of a cell is an active process requiring an upregulation of several proteins.

Opposite to the delayed return to control values, GCLM, for example, shows an increase just after exposure. When the glutathione demand is high and the free glutathione levels are low, there is a return to control values at the end of the recovery period, at which point free glutathione levels are close to normal again.

In contrast to this simple case, the regulation of the actin cytoskeleton is much more complex. Many of the proteins interacting with actin and regulating its dynamics are regulated by phosphorylation. This holds true for cofilin,<sup>74</sup> actinin 4,87 vinculin,88 swap70,89 arpc5,90 gelsolin91 and Arp2. 92 While this further demonstrates the interest of taking into account modified protein forms, there are no obvious rules to predict which modified spot(s) is (are) effector(s) on the actin cytoskeleton. For example, the effector spot for cofilin is the median spot.<sup>93</sup> Consistent with what has been described in the literature, we observe mostly modulations on acidic, modified forms of the above-cited proteins.

In line with the sustained changes in the actin cytoskeleton observed even at the recovery stage, we observe sustained changes in these modified spots, as exemplified by spots 2 of actinin 4 and cofilin, and spots 1 for Arp2 and Arpc5. It must be recalled, however, that such changes are not induced by the particle internalization process per se, as they were not observed for cells treated with the non-toxic zirconium oxide nanoparticles.53 They are also not induced by every toxic nanoparticle, as they were observed neither with zinc oxide<sup>53</sup> nor with copper oxide, except for arpc5 which also decreased in response to treatment with copper oxide.37

Overall, the proteomic results suggest that the recovery process is a slow process. To evaluate this aspect, we determined the number of spots that showed a fast recovery. This category was defined as spots for which the quantitative change in the acute vs. control comparison showed an amplitude at least twice that in the recovery vs. control comparison. Only 44 spots were identified that fell in this category (including GCLM).

A striking result of our study was the observation that some changes in protein levels that were not immediately significant after exposure became significant at the end of the 72 h recovery phase. Such changes were observed both in targeted experiments (e.g. for the LPS-dependent responses) and in the proteomic screen. 51 spots showed such an expression profile, i.e. 21% of the total variable spots.

Functionally speaking, the results obtained when evaluating LPS-induced activities contrast with those previously described on primary macrophages,<sup>23</sup> in which the cytokine and TNF productions also tended to return to normal values during the recovery phase. This discrepancy may have several origins. One such cause for the discrepancy could be differences in the persistence of silver between differing experimental conditions. In the experiments on primary macrophages, measurements of silver by PIXE showed a strong decrease (50%) in cellular silver content upon recovery, without any cellular multiplication (primary macrophages are post-mitotic in vitro). In the current experiments using the J774 cell line, the total silver content decreased by only 15%, and the decrease in cellular silver content is due mostly to cell division. The simple fact that the cells are able to divide shows that the silver concentration present in the cells is not toxic and explains also why, generally, the cellular parameter tends to return to normal levels. The remaining silver concentration may, however, may be high enough to inhibit specialized macrophage functions that depend on cell signaling, such as LPS-dependent activities.

In this context, it is tempting to dismiss the results obtained on the cell lines and favor those obtained using primary cells. The situation may, however, not be so simple. First, primary macrophages in vitro survive for only a few days once differentiated, while resident macrophages, such as those used in experiments using cell lines, have a very long lifespan, 94,95 Furthermore, the condition in which the results between cell lines and primary macrophages diverge corresponds to an acute exposure to silver nanoparticles followed by a massive exposure to bacteria (mimicked by exposure to LPS), i.e. conditions that have not been tested in vivo. Therefore, we are currently unable to determine which system best represents the in vivo situation.

## Conclusions

Overall, our results obtained both by targeted measurements and by proteomic experiments involving hundreds of proteins show that recovery after an acute exposure to silver nanoparticles is a very active process that involves a massive energy consumption, 50% higher than normal. This process leads to the quasi restoration of cellular homeostasis within 72 hours post exposure, as confirmed by targeted experiments testing the activity of enzymes, on mitochondrial function and on phagocytosis. Proteomic experiments also show, however, that this restoration of cellular homeostasis involves changes in the levels of many proteins, of which many are sustained, or even amplified, at the end of the 72 hour recovery period studied. The persistence of proteomic changes might be linked with the persistence of intracellular silver during the recovery period, which may require a long-lasting cellular adaptation process. Some other specialized macrophage functions, such as LPS-induced cytokine or nitric oxide production, did not return to normal within the 72 hour recovery period.

## Author contributions

AT performed the phagocytosis, rhodamine uptake and glutathione assay experiments. BD performed the cytokine assays and confocal microscopy experiments. HD, SC, VSF and TR performed the proteomic experiments. TR performed in addition the enzymatic, NADP(H) and glucose assays. SR and JB performed the ICP-MS experiments. MC, KPG and PHJ performed the electron microscopy experiments.

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## Conflicts of interest

There are no conflicts of interest to declare.

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