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Radionuclide candidates for $\beta+\gamma$ coincidence PET: An overview

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Keywords

- PET
- Production of medical radionuclides
- Cyclotrons
- $\beta+\gamma$ coincidence

Highlights

This paper summarizes the properties, applications and feasible production of all known radioisotopes that can be used in the novel $\beta+\gamma$ coincidence PET technique. These radioisotopes include: ^{10}C , ^{14}O , ^{22}Na , $^{34\text{m}}\text{Cl}$, $^{44\text{g}}\text{Sc}$, ^{48}V , $^{52\text{m}}\text{Mn}$, ^{55}Co , ^{60}Cu , ^{66}Ga , ^{69}Ge , ^{72}As , ^{76}Br , $^{82\text{g}}\text{Rb}$, $^{86\text{g}}\text{Y}$, $^{94\text{m}}\text{Tc}$, $^{110\text{m}}\text{In}$ and ^{124}I .

Abstract

Following the advancement of the Positron Emission Tomography (PET), a novel technique emerged which takes advantage of the gamma quanta emitted, in some cases, after the $\beta+$ decay. While $^{44\text{g}}\text{Sc}$ is commonly agreed the best choice to validate this new modality, other radioisotopes must also be considered in the future. We present them in this paper, along with their possible applications, properties and the optimal production routes with the use of the accelerators.

Introduction

During the recent years, we have witnessed the birth and the steady development of a new approach in Positron Emission Tomography (PET) for the nuclear imaging, namely a $\beta+\gamma$ coincidence technique ([Martin et al., 1995](#); [Pentlow et al., 2000](#); [Herzog et al., 2002](#); [Lubberink et al., 2002a](#); [Beattie et al., 2003](#); [Buchholz et al., 2003](#); [Sandström et al., 2004](#); [Grignon et al., 2007](#); [Lang et al., 2013](#); [Thirolf et al., 2015](#)). Compared to the classical PET, it offers potentially better spatial resolution ([Sandström et al., 2004](#); [Grignon et al., 2007](#); [Duval et al., 2009](#)) or, alternatively, provides the same image quality with less radioactivity (hence reducing the dose to the patient). It also reduces the exposure time that may translate to the increased number of consultations in hospitals per day.

Today, the conventional PET is used worldwide and has been acknowledged as the gold standard in oncology imaging when used with FDG. Its principle and numerous applications were summarized recently by The International Atomic Energy Agency ([IAEA, 2008](#)) and The European Association of Nuclear Medicine ([EANM 2010, 2011](#)). In general, PET is a technique commonly used in medicine for diagnostic purposes as it allows quantitative in vivo measurements of the distribution of administered positron-emitting radioisotopes in the human body. In short, the radioactive tracer (which have been defined to accumulate in some specific cells) is injected in the patient, decays and produces the positron ($\beta+$) that travels few millimetres (depending on the energy) from the decay vertex before annihilating with an electron. Positron-electron annihilation converts them into two 511 keV energy photons (with over 99% probability), which are emitted in opposite directions (in positron-electron center of mass system) and can escape the human body. These photons are then detected in parallel rings of scintillation crystals surrounding the patient. Two detected photons of the right energy allow to assume that the positron emitter is located somewhere in between, at the Line of Response (LoR). In the first approximation, the intersection of multiple LoR provides the distribution of the tracer (although in practice many additional corrections are applied).

The $\beta+\gamma$ coincidence is one of the possible extensions of PET modality and its key element is the use of a $\beta+$ source that emits additional γ quanta (sometimes called “third γ ”). To prevent the overlap of the signals, the typical constraint in the acquisition systems is that the third γ should be emitted from the radioisotope few ps after the $\beta+$ decay allowing coincidence measurements ([Grignon et al., 2007](#)). Assuming that the $\beta+$ particle and the third γ are emitted at the same location, the position of the radioisotope is then obtained by the intersection of the arrival direction of the third γ with the conventional LoR (Fig. 1). This technique allows to localize the emission point on the event-by-event basis ([Martin et al., 1995](#); [Pentlow et al., 2000](#); [Grignon et al., 2007](#); [Duval et al., 2009](#)). The localization via direction cone implies the Compton scattering of the third γ and the detection of the scattered photon in the photoelectric interaction. This favours a third γ of high energy, around 1 MeV ([Grignon et al., 2007](#)). While it would give additional radiation dose to the patient, the precision of the $\beta+\gamma$ PET is supposed to allow reducing the administered activity, significantly reducing the overall dose ([Lang et al., 2013](#)).

In this article, we review the medical radioisotopes suitable for $\beta+\gamma$ coincidence PET. The most important criteria for the candidates are the $\beta+$ decay followed by a γ quanta of around 1 MeV emitted few ps later. However, there are many more requirements related to the practical aspects of medical radioisotopes. In particular, $\beta+$ branching ratio as well as the intensity of third γ should be as high as possible (preferably 100% for both) to assure the highest probability for the signal detection in the PET scanner. At the same time,

radioisotope of interest should emit a limited number of additional γ lines (which would only introduce the noise to the imaging and dose to the patient). Finally, the radioisotope should have feasible production route and convenient half-life (from few hours to few days), suitable for the clinical practice. Obviously, no radioisotope fulfils all these criteria. However, some are very close and could be potentially investigated in $\beta^+\gamma$ coincidence PET, depending on the parameters of available scanner and the aim of the research or imaging. The most promising radioisotopes are presented in this work, including their applications and production routes investigated so far. The physical properties of all candidates are listed in Table 1 and the natural abundance of the targets discussed for their production is shown in Table 2 (although in certain cases the discussed production route and its yield requires the use of an enriched target with modified composition discussed in the text).

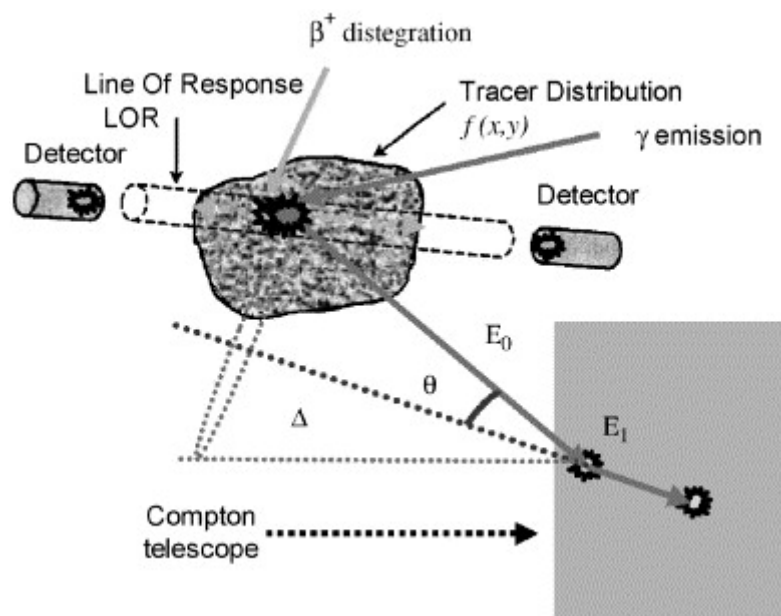


Fig. 1. Principle of the $\beta^+\gamma$ coincidence PET (Grignon et al., 2007).

Finally, it is worth mentioning that with the development of new PET scanners, further advancement in $\beta^+\gamma$ coincidence technique can be expected. Already mentioned XEMIS and XEMIS2 machines offer a sub-millimetre position resolution, a good energy resolution, and a possibility to measure third γ of up to 5 MeV (Grignon et al., 2007; Duval et al., 2009; Cussonneau et al., 2017; Gallego Manzano et al., 2018). At the same time, the first PET scanner build from plastic scintillators, J-PET, offers cost-effective whole-body scans with the feasibility of ortho-positronium imaging (Gajos et al., 2016; Kamińska et al., 2016; Moskal et al., 2019). Recently constructed EXPLORER might be also of interest as it has proved its high-sensitive total-body scan (Badawi et al., 2019).

As a side remark, it is possible to imagine an approach in which LoR is replaced with two (or more) γ -rays emitted from the nucleus, forcing the reconstruction based on three complete Compton events. Some radioisotopes satisfy this criterion, with three high-intensity γ lines and low β^+ branching ratio: ^{94g}Tc ($T_{1/2} = 4.9$ h), ^{96}Tc ($T_{1/2} = 4.3$ d), ^{108}In ($T_{1/2} = 58$ min), ^{110g}In ($T_{1/2} = 4.9$ h) and ^{206}Bi ($T_{1/2} = 6.2$ d). However, the imaging with this method would require more activity compared to the $\beta^+\gamma$ coincidence counterpart and is not discussed further in this paper.

There is also a similar PET technique which utilizes a rare three γ quanta emission from the ortho-positronium annihilation (Kacperski et al., 2004; Kacperski and Spyrou, 2005;

Abuelhia et al., 2007; Gajos et al., 2016; Kamińska et al., 2016; Moskal et al., 2019). This method can be applied with all β^+ radionuclides and is not discussed further here.

^{34m}Cl

Compared to other radioisotopes from this overview, ^{34m}Cl is less popular in the nuclear medicine field. However, it was recognized as potential PET radionuclide (Qaim and Stöcklin, 1983; Helus et al., 1985) and has already been used to label dopamine D1 agonists (DeJesus et al., 2007; Murali et al., 2011). The first factor limiting its popularity is the number of high-intensity γ lines increasing the dose without providing the relevant diagnostic data. The γ line of the highest intensity could be used in $\beta^+\gamma$ coincidence imaging although other radioisotopes discussed here offer better physical properties for this purpose.

The second limiting factor is the difficult production of ^{34m}Cl. The practical no-carrier-added production uses ^{nat}S(α ,x) reaction. According to the cross-section data (Hintz and Ramsey, 1952; Umbarger et al., 1970; Nagatsu et al., 2008), the required beam energy is around 65 MeV. The saturation of the thick target irradiation results in radio-contaminant-free 1500 MBq/ μ A ^{34m}Cl (Zatolokin et al., 1976; Takei et al., 2007; Nagatsu et al., 2008). A feasible production chain was also developed, consisting of the 80% effective separation using heated water and HPLC pump (Takei et al., 2007). Even though this production requires high energy α beam. Alternative methods face even more difficult problems.

^{44g}Sc

The ^{44g}Sc radioisotope is one of the most promising $\beta^+\gamma$ candidate (mentioned as such by Lang et al., 2013 and Thirolf et al., 2015) and the perfect one to set-up the proof of concept of this new imaging modality. It has a convenient half-life ($T_{1/2} = 3.9$ h) and emits only one, high-intensity γ -line with a desirable energy (1157 keV, 99%). It has been already used in the study of $\beta^+\gamma$ coincidence PET with the use of XEMIS and XEMIS2 detectors (Grignon et al., 2007; Duval et al., 2009; Cussonneau et al., 2017; Gallego Manzano et al., 2018). The general interest of ^{44g}Sc was also stressed by Huclier-Markai et al., 2018. It is not only related to its convenient physical properties, but also to the feasible chemistry (reported mainly for the labelling of DOTA-peptides) supported with the stability and biodistribution studies (Koumarianou et al., 2011; Cydzik et al., 2012; Krajewski et al., 2012, 2013; Pruszyński et al., 2012; Roesch, 2012; Severin et al., 2012; Müller et al., 2013; Huclier-Markai et al., 2014; Alliot et al., 2015a, 2015b; Valdovinos et al., 2015; van der Meulen et al., 2015; Domnanich et al., 2016; Kilian et al., 2018). Additionally, its metastable state, ^{44m}Sc ($T_{1/2} = 58.6$ h), can be used as long-lived in vivo generator (Huclier-Markai et al., 2014; Alliot et al., 2015a, 2015b; Duchemin et al., 2015) as it decays mainly by a low energy transition to the ground state. Finally, ^{44g}Sc can be used as counterpart of therapeutic partners. It forms a theranostic pair with ⁴⁷Sc ($T_{1/2} = 3.35$ d), which is a low-energy β^- emitter for targeted radiotherapy, allowing the application of the theranostic approach (Müller et al., 2014, 2018). It can also be used with the therapeutic ¹⁷⁷Lu as they share similar chemistry (Umbricht et al., 2017).

The most interesting production route is via ⁴⁴Ca(p,n) reaction with the CaCO₃ or CaO, (Khandaker et al., 2009; Krajewski et al., 2012, 2013; Severin et al., 2012; Müller et al., 2013, 2014; Hernandez et al., 2014; Hoehr et al., 2014; Valdovinos et al., 2015; van der Meulen et al., 2015; Singh et al., 2015; Domnanich et al., 2016; Carzaniga et al., 2017;

Sitarz et al., 2018) which can be later easily dissolved for the chemical separation (Cyzdik et al., 2012; Krajewski et al., 2012, 2013; Pruszyński et al., 2012; Severin et al., 2012; Müller et al., 2013; Huclier-Markai et al., 2014; Alliot et al., 2015b; Valdovinos et al., 2015; van der Meulen et al., 2015; Domnanich et al., 2016; Kilian et al., 2018). A standard 16 MeV proton beam of 1 μ A can produce up to 20 MBq (CaCO_3) or 35 MBq (CaO) after 1 h of irradiation of natural targets and 40-50 times more if the commercially available >90% enrichment is used. In case of natural target, ^{43}Sc ($T_{1/2} = 3.89$ h) is present at the level of 3% while the use of the enriched target eliminates the radioactive impurities almost completely (Sitarz et al., 2018). Alternatively, ^{44g}Sc can be obtained from $^{44}\text{Ti}/^{44g}\text{Sc}$ generator ($T_{1/2} = 59.1$ y) as pointed out by Filosofov et al., 2010; Pruszyński et al., 2010, 2012; Roesch, 2012 and summarized recently by Hassan et al., 2017. Currently, such generator is available in Mainz, with 185 MBq ^{44}Ti and the possibility to extract 97% ^{44g}Sc in 20 mL solution (Filosofov et al., 2010).

Still, the most accessible optimal production route is the irradiation of >90% enriched $^{44}\text{CaCO}_3$ or ^{44}CaO with proton beam, yielding up to 1000 MBq/ μAh or 1700 MBq/ μAh respectively and less than 1% of radioactive contaminants.

^{48}V

^{48}V radioisotope is well-known only as a monitor for the beam current measurements (IAEA, 2017). It emits two high-intensity γ lines and has a β^+ branching ratio of only 50%. However, it has the longest half-life (16 d) among all discussed $\beta^+\gamma$ candidates, suitable for the studies of slow metabolic processes and for the labelling of organic compounds (Qaim, 2011). The prospects of ^{48}V were recently reminded by Usman et al., 2017 as it is already finding applications as a tracer in biological actions in plants (Xuam Tham et al., 2001), in material science (Rorat et al., 2005) or in the renal artery brachytherapy (Arbabi et al., 2009). As suggested by Martin et al., 1995 it is also a promising candidate for the coincidence PET. However, it should be noted that the second high-intensity γ line from ^{48}V would introduce additional dose for the patient and noise in the imaging without providing the relevant diagnostic data.

As summarized in IAEA, 2017, the cross-sections for different production of ^{48}V are well measured and suggest the route via $^{nat}\text{Ti}(p,x)$ reaction (Smith et al., 2011). The major contribution comes from $^{48}\text{Ti}(p,n)$ reaction suitable for the energy of the commonly available cyclotrons and the target nuclide has high natural abundance. With 16 MeV beam and 1 μAh irradiation of 90% enriched ^{48}Ti target, over 20 MBq of ^{48}V can be produced, with less than 0.5% of radioactive impurity of ^{49}V ($T_{1/2} = 330$ d).

^{52m}Mn

The recent advancements in hybrid MRI induced the interest in ^{52g}Mn radioisotope. It was suggested as the tracer of Mn^{+2} ions (Lewis et al., 2015) that serve as the T_1 MRI relaxation agent in the Manganese-Enhanced Magnetic Resonance Imaging (MEMRI) (Koretsky and Silva, 2004; Silva et al., 2004; Wadghiri et al., 2004; Silva and Bock, 2008; Massaad and Pautler, 2011; Cacace et al., 2014). However, in the light of the reported risk regarding the use of the bulk manganese (Crossgrove and Zheng, 2004), the use of ^{52g}Mn in conventional PET/MRI was suggested (Graves et al., 2015) to obtain analogous data with lower biological toxicity. Meanwhile, its metastable level, ^{52m}Mn , has very high β^+ branching ratio and emits only one, high-energy and high-intensity γ line. Since the $\beta^+\gamma$ coincidence imaging requires less activity, the toxicity of manganese in PET/MRI could be further

reduced with the use of ^{52m}Mn (already suggested for $\beta+\gamma$ coincidence PET by [Martin et al., 1995](#)). However, it should be noted that significant modifications might be in order for PET hardware to permit the acquisition in the presence of a strong magnetic field and radiofrequency pulses ([Disselhorst et al., 2014](#)).

The best access to ^{52m}Mn is through a generator, $^{52g}\text{Fe}/^{52m}\text{Mn}$, since ^{52g}Fe decays in 100% to ^{52m}Mn . The co-produced ^{52m}Fe ($T_{1/2} = 46$ s) decays in 100% to ^{52g}Mn though so it should be taken into account by managing the post-irradiation separation, rapidly removing early ^{52g}Mn and later waiting for ^{52m}Mn to be formed from ^{52g}Fe . The possible production routes for $^{52g}\text{Fe}/^{52m}\text{Mn}$ generator have been summarized by [Atcher et al., 1980](#) and [Steyn et al., 1990](#). According to the literature, a reasonable option is to irradiate thick ^{nat}Mn target with protons of energy from 40-50 to 60-75 MeV. This procedure is routinely used in Brookhaven Linac Isotope Producer. Up to 22 MBq/ μAh of $^{52g}\text{Fe}/^{52m}\text{Mn}$ was reached, with no more than 1% of ^{55}Fe contaminant.

^{55}Co

^{55}Co has been acknowledged as “the emerging PET radionuclide” by [Amjed et al., 2016](#), as it features high $\beta+$ branching ratio, half-life favouring the studies of slow biological process and feasible labelling with different complexes as well as satisfactory biodistribution ([Srivastava, et al., 1994](#); [Thisgaard et al., 2011](#); [Mastren et al., 2015](#); [Dam et al., 2016](#); [Garousi et al., 2017](#)). The mentioned paper summarized numerous applications of ^{55}Co which include: the lung cancer detection ([Nieweg et al., 1982](#)), the renal imaging ([Goethals et al., 2000](#)) and the neuro-imaging ([Jansen et al. 1994, 1996, 1997](#); [Stevens et al., 1999](#); [Reuck et al., 2004](#)). These procedures can be also performed with $\beta+\gamma$ coincidence PET (suggested by [Martin et al., 1995](#)) since ^{55}Co emits one high-intensity high-energy γ line. Additionally, there is a theranostic matched pair with Auger-emitting ^{58m}Co . However, it is worth mentioning that two additional low-intensity γ lines from ^{55}Co would introduce additional dose to the patient and noise to the imaging.

The production routes of ^{55}Co have been thoroughly studied via different nuclear reactions, for which the cross-section and thick target yield data are very well reported. The most promising one is $^{54}\text{Fe}(d,n)$ reaction ([Sharma et al., 1986](#); [Zaman and Qaim, 1996](#); [Hermanne et al., 2000](#); [Zaman et al., 2003](#); [Nakao et al., 2006](#); [Király et al., 2009](#); [Thisgaard et al., 2011](#); [Závorka et al., 2011](#); [Khandaker et al., 2013](#); [Avriganu et al., 2014](#); [Valdovinos et al., 2017](#)) which requires enriched target to produce sufficient quantities as well as avoid the co-production of long-lived radioactive impurities of ^{56}Co and ^{57}Co as they emit high intensity γ lines which unnecessarily increase the dose. The commercially available enrichment exceeds 95% in the form of metal (^{54}Fe) or oxide ($^{54}\text{Fe}_2\text{O}_3$). For example, with the metal target and 1 μAh irradiation of 8 or 15 MeV deuteron beam, up to 20 MBq or 40 MBq of ^{55}Co respectively can be produced, with negligible amount of ^{56}Co and ^{57}Co ([Sharma et al., 1986](#)).

^{60}Cu

^{60}Cu is a short-lived $\beta+$ emitter with an additional γ line making it suitable for $\beta+\gamma$ coincidence PET ([Martin et al., 1995](#)). So far, it has only been used in standard PET technique along with the labelling studies of ^{60}Cu -ATSM for tumor hypoxia imaging ([Blower et al., 1996](#); [McCarthy et al., 1999](#); [Chao et al., 2001](#); [Dehdashti et al., 2003](#); [Dietz et al.,](#)

2008; Lewis et al., 2008). It also has a theranostic matched pair, ^{67}Cu , the β^- emitter for targeted radionuclide therapy.

The most commonly used production route of ^{60}Cu is via $^{60}\text{Ni}(p,n)$ or $^{\text{nat}}\text{Ni}(p,x)$ reactions (used in above-mentioned papers). The corresponding cross-sections have been already well reported (Blosser and Handley, 1955; Tanaka et al., 1972; Barrandon et al., 1975; Levkovskij, 1991; Singh et al., 2006; Al Saleh et al., 2007; Amjed et al., 2014). Irradiations of thick ^{60}Ni targets (with commercially available 99% enrichment) with proton beam of 16 MeV and 1 μA for 20 minutes, followed by 1 h post-irradiation processing time, are sufficient to achieve up to 400 MBq of ^{60}Cu .

^{66}Ga

The radioisotope of ^{66}Ga already has multiple applications and it has been widely used as PET radioisotope. Labelled with albumin colloids from commercially available kits designed for $^{99\text{m}}\text{Tc}$, ^{66}Ga was successfully used in the imaging of the lymphatic transport (Goethals et al., 1988). The feasible ^{66}Ga labelling and purification was also reported for DOTA-peptides ^{66}Ga -DOTA-Tyr3-octreotide and ^{66}Ga -DOTA-biotin (Lewis et al., 2002), the blood cells (Ellis and Sharma 1999; Jalilian et al. 2003) and ^{66}Ga -deferoxamine-folate for in vivo and in vitro imaging (Ke et al., 2003, 2004; Mathias et al., 2003). Additionally, the most abundant β^+ emitted by ^{66}Ga has a uniquely high energy, which allowed the use of ^{66}Ga -DOTATOC for both PET imaging and radiotherapy (Ugur et al., 2002). Finally, ^{66}Ga emits high-energy γ line making it appropriate for $\beta^+\gamma$ coincidence PET. However, it also emits many less intense but high-energy γ lines that will introduce noise in the imaging and dose to the patient.

The method of choice to produce ^{66}Ga is via $^{66}\text{Zn}(p,n)$ reaction. In this case, several cross-section data (Hille et al., 1972; Little and Lagunas-Solar, 1983; Kopecký, 1990; Tárkányi et al., 1990; Levkovskij, 1991; Hermanne et al., 1991; Nortier et al., 1991; Szelecsényi et al., 1998, 2005) and thick target yield data (Barrandon et al., 1975; Intrator et al., 1981; Dmitriev, 1986; Kopecký, 1990; Tárkányi et al., 1990; Nortier et al., 1991; Lewis et al., 2002; Rowshanfarzad et al., 2004) were investigated. This reaction was also summarized and re-evaluated with ALICE/ASH 0.1 and TALYS-1.2 by Sadeghi et al., 2010. The natural abundance of ^{66}Zn is quite low but a feasible method was reported for the preparation and recovery of the enriched target (Rowshanfarzad et al., 2004). Around 200 mg/cm^2 of ^{66}Zn material is enough for the optimal production of ^{66}Ga with the 15 MeV proton beam. The reported yield for 99% enriched target was around 500 $\text{MBq}/\mu\text{Ah}$ of radionuclidically pure ^{66}Ga . The optimal post-irradiation processing of Zn targets and ^{66}Ga separation was found to be a cation-exchange chromatography and/or liquid-liquid extraction method (Lewis et al., 2002; Rowshanfarzad et al., 2004).

^{69}Ge

The unravelled potential of ^{69}Ge (Mirzadeh and Lambrecht, 1996) is mainly related to the common interest shifted on a different germanium isotope (namely, the $^{68}\text{Ge}/^{68}\text{Ga}$ generator, summarized by Rösch, 2013 and Velikyan, 2015). However, its physical properties make it a possible $\beta^+\gamma$ coincidence PET agent, yet with an inconveniently low β^+ branching ratio, similarly low intensity of 1 MeV γ line, and the presence of additional γ lines. On the other hand, its chemistry is already well developed for the purpose of the mentioned generator. So far, it was only used to label nanoparticles for successful in vivo PET/MRI imaging (Chakravarty et al., 2014). Due to its long half-life, it can be also considered for

immunoPET studies and antibody labelling.

The production of the applicable amounts of ^{69}Ge was not yet investigated but the cross-section for $^{69,\text{nat}}\text{Ga}(p,n)$ reaction (Levkovskij, 1991; Porile et al., 1963; Johnson et al., 1964; Adam-Rebeles et al., 2013; Hermanne et al., 2015) seems the most cost-efficient and available for small cyclotrons. The 16 MeV beam interacting on a thick $^{\text{nat}}\text{Ga}$ target would produce around 110 MBq/ μAh . However, no cross-section data is available to estimate the co-production of radioactive impurity of ^{71}Ge . The measurement is challenging due to no γ emission but important as ^{71}Ge is the long-lived Auger-emitter which would contribute to the dose.

^{72}As

In the nuclear medicine, another popular PET radioisotope, ^{72}As , can be considered in the $\beta+\gamma$ coincidence PET thanks to its additional γ line. It has proven its favourable physical and chemical properties in labelling and preclinical studies (Hosain et al., 1982; Emran et al., 1984; Ballard et al., 2012; Ellison et al., 2016). Furthermore, it has a therapeutic matched pair in the form of the β^- emitting ^{77}As . Both radioisotopes have been studied in preclinical and clinical research (Nayak and Brechbiel, 2009; Ellison et al., 2016). Arsenic itself, in the form of the arsenic trioxide, is a popular anticancer drug (Ravandi, 2004), successfully used recently in the clinical treatment of the acute promyelocytic leukemia (Miller et al., 2002; Lu et al., 2007).

There are many methods for the direct production of ^{72}As , however much more attention is paid to the generator $^{72}\text{Se}/^{72}\text{As}$ due to its convenient half-life of 8.4 d. Many practical extraction methods for this generator have already been reported (Al-Kouraiishi and Boswell, 1978; Phillips et al., 1991; Jennewein et al., 2004, 2005; Ballard et al., 2012, 2012b; Chajduk et al., 2012; Wycoff et al., 2014; Feng et al., 2019). According to the literature, the best method for the production of $^{72}\text{Se}/^{72}\text{As}$ leads via $^{70}\text{Ge}(\square, 2n)$ reaction (Amiel, 1959; Al-Kouraiishi and Boswell, 1978; Calboreanu et al., 1987; Mushtaq and Qaim, 1990; Levkovskij, 1991; Jennewein et al., 2005; Szkliniarz et al., 2015; Takács et al., 2016; Feng et al., 2019). To reduce the formation of radioactive impurity ^{75}Se that decays to stable ^{75}As , the enriched ^{70}Ge target is recommended (commercially available enrichment is about 96%). The 6 h irradiation with 20 μA and the energy range of 47 \rightarrow 0 MeV produces around 250 MBq ^{72}Se (Feng et al., 2019) from which around 70 MBq ^{72}As can be extracted each day during the following week. The reported radioactive impurities of arsenic can be removed during the preparation of the generator.

^{76}Br

The radioisotope of ^{76}Br has a large number of accompanying γ rays, from which the most intensive makes ^{76}Br a possible $\beta+\gamma$ candidate (as suggested by Lubberink et al., 2002a; Sandström et al., 2004; Lang et al., 2013; Thirolf et al., 2015) whose coincidence PET imaging and γ cascade correction have already been investigated (Lubberink et al., 2002a). There are two factors limiting the possible interest in ^{76}Br : the γ line used for the coincidence has quite low energy and other γ lines contribute to the dose. However, the labelling chemistry of bromine is similar to that of iodine, which is relatively well investigated (Maziere and Loc'h, 1985) and might render ^{76}Br worth considering. So far, it has been successfully used to study the dopamine receptors associated with the diagnosis of schizophrenia (Martinot et al., 1991; 1994), as the amino acid tracer (Hanaoka et al., 2015),

as the monitor for corticotropin-releasing hormone (Jagoda et al., 2011), as the bromo analogue marker to diagnose heart disease (Loc'h et al., 1994), for the labelling of mouse epidermal growth factor (Scott-Robson et al., 1991) and to study the tumor angiogenesis by labelling a human antibody (Rossin et al., 2007). Additionally, ^{76}Br was used to verify the thymidine analogue, BUdR, as the tumor cell proliferation imaging agent (Gardelle et al., 2001).

As summarized by Hassan et al., 2004 and Sadeghi et al., 2010, the most feasible method for production of ^{76}Br is the direct route via $^{76}\text{Se}(p,n)$ reaction for which the cross-section data (Kovács et al., 1985, Levkovskij, 1991, Hassan et al., 2004; El-Azony et al., 2009) and experimental yields (Janssen et al., 1980; Tolmachev et al., 1998) are well measured. However, the favourable cross-section requires an enriched target due to the low abundance of ^{76}Se . The literature indicates that around 360 MBq/ μA of ^{76}Br can be produced by the irradiation of the commercially available 97% enriched metal target with 15–8 MeV proton beam. The radioactive impurity of ^{77}Br was observed at the level below 2% and originated from ^{77}Se impurity in the target.

^{82g}Rb

An interesting case, ^{82g}Rb ($T_{1/2} = 1.3$ min), emits β^+ radiation followed by only one low-intensity γ line. Still, it was introduced as the $\beta^+\gamma$ PET candidate (Lang et al., 2013; Thirolf et al., 2015).

The radioisotope of ^{82}Sr ($T_{1/2} = 25.4$ d) that decays to ^{82g}Rb is the only reasonable method to acquire ^{82g}Rb . The generator $^{82}\text{Sr}/^{82g}\text{Rb}$ has already gained incredible popularity and is widely used to diagnose the cardiovascular disease (a leading cause of death in modern industrialized countries) in myocardial perfusion imaging (Yano et al., 1977; Kensett et al., 1987; Go et al., 1990; Saha et al., 1990; Di Carli et al., 2007; Klein et al., 2007; Merhige et al., 2007; Yoshinaga et al., 2010; Dhar and Ananthasubramaniam, 2011; Scholtens and Barneveld, 2017). It provides significantly better precision compared to ^{201}Tl (Go et al., 1990) and presents less radiation exposure for patients compared to ^{99m}Tc scan (Yoshinaga et al., 2010). Many studies have also been performed on the elution system and the optimized chemical separation (Grant et al., 1975; Yano et al., 1977; Kensett et al., 1987; Mausner et al., 1987; Saha et al., 1990; Cackette et al., 1993; Bilewicz et al., 2005; Klein et al., 2007; Yoshinaga et al., 2010).

The method of choice for $^{82}\text{Sr}/^{82g}\text{Rb}$ generator production is the $(p,4n)$ reaction on ^{85}Rb which benefits from high natural abundance. This method was studied multiple times and is most often employed for the large-scale production (Mausner et al., 1987; Huszár et al., 1989; Deptula et al., 1990; Lagunas-Solar, 1992; Cackette et al., 1993; Gilabert et al., 1998; Ido et al., 2002; Buthelezi et al., 2006), although, as summarized by Takács et al., 2003, significant discrepancies still exist. For the proton energy of 70-60 MeV (with 40 MeV exiting from the thick Rb or RbCl target), the $^{82}\text{Sr}/^{82g}\text{Rb}$ production yields of 8-13 MBq/ μAh were reported. The observed long-lived radioactive impurity of ^{85}Sr was below 1%. As the typical generator activity used for clinical studies reaches 4 GBq (Saha et al., 1990), the typical irradiation lasts for few days and requires high beam current.

^{86g}Y

The potential of ^{86g}Y lies within its theranostic matched pair (Lopci et al., 2011; Rösch et al., 2017; Bandara et al., 2018), the β^- emitter ^{90}Y available from the long-lived $^{90}\text{Sr}/^{90}\text{Y}$ generator system, which is a versatile therapy agent (as reviewed by Goffredo et al., 2011). However, the emission of multiple intensive, dose-contributing γ rays from ^{86g}Y and the recent development of the ^{90}Y imaging with the bremsstrahlung photons (summarized by

Wright et al., 2015) might render the matched pair obsolete. Still, many radiochemicals and in vivo PET imaging studies were performed with ^{86g}Y (summarized by Nayak and Brechbiel, 2011) and its dominating γ line was recognized for the $\beta+\gamma$ coincidence PET (Pentlow et al., 2000; Beattie et al., 2003; Buchholz et al., 2003; Sandström et al., 2004; Lang et al., 2013; Thirolf et al., 2015). In fact, the quantitative coincidence imaging for this radioisotope have been already investigated (Pentlow et al., 2000; Beattie et al., 2003; Buchholz et al., 2003).

Several methods of ^{86g}Y production were investigated (in each, the co-produced ^{86m}Y decays with $T_{1/2} = 47.4$ min to ^{86g}Y). As reviewed by Schmitz, 2011, the most commonly used is the direct production via $^{86}\text{Sr}(p,n)$ reaction, for which the excitation function has been also re-evaluated with nuclear codes by Sadeghi et al., 2010. It requires about 14 MeV as higher energies increase the percentage of the contaminants. The reported irradiation of around 200 mg/cm² of the commercially available 95% enriched $^{86}\text{SrCO}_3$ target yields about 150 MBq/ μAh with less than 3% of radioactive impurities (Rösch et al., 1993a,1993b; Kettern et al., 2002; Yoo et al., 2005, Avila-Rodriguez et al., 2008, Lukić et al., 2009; Elbinawi et al., 2018). For comparison, around 10 MBq of ^{86g}Y is enough for in vivo mice studies (Lövgqvist et al., 2001; McQuade et al., 2005).

^{94m}Tc

Despite the clear dominance of ^{99m}Tc in nuclear medicine, other technetium radioisotope, ^{94m}Tc , is of a potential interest as a PET quantification of ^{99m}Tc -radiopharmaceuticals (Bigott et al., 2001; Qaim, 2012) due to the same chemistry of both isotopes. It decays completely to ^{94}Mo and emits one high-intensity high-energy γ line making it suitable for $\beta+\gamma$ coincidence imaging (Martin et al., 1995; Lang et al., 2013; Thirolf et al., 2015). The feasibility of standard in vivo PET studies has been already reported (Nickles et al., 1993; Stone et al., 1994; Luyt et al., 2003).

The method of choice for the ^{94m}Tc production is the bombardment of ^{94}Mo with medium energy proton beam (Rösch and Qaim, 1993c; Rösch et al., 1994; Bigott et al., 2001, 2006; Uddin et al., 2004; Kakavand et al., 2013). The commercially available enrichment of about 95% is available in MoO_3 powder form. The irradiation with the optimal energy range of 13→8 MeV produces about 2 GBq/ μAh ^{94m}Tc with about 8% of ^{94g}Tc impurity. A 1 h irradiation with 4 μA beam followed by 0.5 h thermochromatographic separation with 90% efficiency results in 1300 MBq $^{94m}\text{TcO}_4^-$, ready for medical application (Rösch et al., 1994). Other purification and target recovery methods were also investigated (summarized in Bigott et al., 2006).

^{110m}In

So far, the radioisotope of ^{110m}In has not drawn a lot of attention in the nuclear medicine field. However, as confirmed by in vivo clinical studies, it provides 3 times better resolution than the typical ^{111}In SPECT (Lubberink et al., 2002b) which opens the possibilities for the detection of small tumors with indium-labelled radiopharmaceuticals. This might be important in the light of an emerging radioisotope ^{114m}In (IAEA, 2007), an Auger emitter with almost instant β^- emissions (from its short-lived daughter), whose therapeutic properties are expected (Tolmachev et al., 2000). Additionally, ^{110m}In emits medium-energy but high-intensity γ line which could be potentially interesting for $\beta+\gamma$ coincidence PET.

Several cross-sections measurements are available, suggesting the potential ^{110m}In direct and indirect production options (summarized by Tárkányi et al., 2015). The direct production of ^{110m}In ($J^\pi = 2^+$) always leads to the co-formation of the radioactive impurity ^{110g}In ($J^\pi = 7^+$) but the higher isomeric ratio can be achieved with lower projectile energies. The recommended reaction, $^{110}\text{Cd}(p,n)$ (Otozai et al., 1966; Abramovich et al., 1975; Skakun et al., 1975; Kormali et al., 1976; Marten et al., 1985; Nortier et al., 1990; Tárkányi et al.,

2006, 2015; Al-Saleh, 2008; Khandaker et al., 2008; Büyüksulu et al., 2010), at 15 MeV energy (available in the commonly used machines) and with electroplated ^{nat}Cd target (Kakavand et al., 2015b) yields 160 MBq/ μAh of ^{110m}In with reported 3% of ^{110g}In and 6% of ^{111g}In radioactive impurities (Mukhammedov et al., 1984; Kakavand et al., 2015b).

^{124}I

A textbook medical radionuclide ^{124}I has relatively low β^+ branching ratio and the medium-intensive medium-energy γ line but was still suggested as $\beta^+\gamma$ PET candidate (Martin et al., 1995; Herzog et al., 2002; Sandström et al., 2004; Lang et al., 2013; Thirolf et al., 2015) and the quantitative coincidence imaging have also been investigated (Herzog et al., 2002). This is mainly because it is the only isotope of iodine suitable for PET that can be paired with the strategic therapeutic ^{131}I (Lopci et al., 2011) commonly used for the treatment of hyperthyroidism and thyroid cancer (overviewed by de Klerk, 2000 and Higashi et al., 2012). Furthermore, ^{124}I itself has already been used for the imaging of tissue proliferation (Blasberg et al., 1996; 2000; Roelcke et al., 2002) and for multiple in vivo cancer imaging studies (Langen et al., 1990; Snook et al., 1990; Wilson et al., 1991), including thyroid (Frey et al., 1986; Phan et al., 2008; Capocetti et al., 2009; Kharazi et al., 2011; Plyku et al., 2017). It is also considered as a potential Auger-emitter for the radiotherapy (Stepanek et al., 1996).

The methods for production of ^{124}I were thoroughly studied by many groups with the use of protons, deuterons and α -particles (summarized by Schmitz, 2011 and Azizakram et al., 2016). As one of the “emerging isotopes”, its production cross-sections are also collected in IAEA database (IAEA, 2007). The most typical ^{124}I production method is the $^{124}\text{Te}(p,n)$ reaction (Kondo et al., 1977; Scholten et al., 1995; Qaim et al., 2003; Sajjad et al., 2006; Nye et al., 2007; Aslam et al., 2010) suitable for popular small cyclotrons but requiring the enriched target material (which is crucial to increase the yield and to reduce the radioactive impurities). Recent thick target yield measurements (Sajjad et al., 2006) indicate that the irradiation of the commercially available $^{124}\text{TeO}_2$ targets (99.9% ^{124}Te) with the energy range of 14→7 MeV produces around 21 MBq/ μAh of ^{124}I with around 0.03% of radioactive impurities of ^{125}I and ^{126}I , followed by the dry distillation. Basically, a few hours run with around 20 μA is enough to produce several 50 MBq batches used for the imaging (Plyku et al., 2017).

Online monitors (^{10}C , ^{14}O)

There are two very short-lived $\beta^+\gamma$ PET radionuclide candidates mentioned in the literature (Martin et al., 1995; Lang et al., 2013; Thirolf et al., 2015), namely: ^{10}C ($T_{1/2} = 19.3$ s) and ^{14}O ($T_{1/2} = 70.6$ s). They are formed via the fragmentation of the high energy heavy ions or different nuclear reactions induced by the high-energy projectiles on ^{12}C , ^{14}N , ^{16}O nuclides which are immensely abundant in the organic compounds. The beams used in the ion therapy, apart from the delivery of the radiation dose, produce ^{10}C and ^{14}O allowing the visualization of the treatment with the online acquisition system.

Online PET scans with ^{10}C and ^{14}O have already been reported (Litzenberg et al., 1999; Enghardt et al., 2004; Cambraia Lopes et al., 2016; D’Ascenzo et al., 2018). Still, this technique poses a challenge from the point of the time resolution (Oelfke et al., 1996; Beebe-Wang et al., 2003; Cambraia Lopes et al., 2016). The method with $\beta^+\gamma$ coincidence PET can be potentially considered in cases where the lower statistic is expected.

Calibration source (^{22}Na)

^{22}Na is a common calibration source, used even in the recent PET research (D'Ascenzo et al., 2018). It has a very long half-life of 2.6 years and high β^+ branching ratio, making it a convenient β^+ emitter for the repetitive calibrations. ^{22}Na also emits 1274.5 keV gamma-line that was used as a trigger during the calibration of liquid argon detector (Amaudruz et al., 2016). Being already recognized as the $\beta^+\gamma$ PET candidate (Martin et al., 1995; Lang et al., 2013; Thirolf et al., 2015), ^{22}Na can be therefore considered as a calibration source for $\beta^+\gamma$ coincidence PET.

^{22}Na sources can be bought from the different suppliers or produced on-site with the proton beam of low and medium energy (as summarized by Takács et al., 1996). Today, the only large-scale production is performed with the use of 70 MeV proton cyclotron by iThemba LABS in South Africa.

Summary

In this paper, we summarized the properties and availability of radioisotopes suitable for the $\beta^+\gamma$ coincidence PET imaging. The $^{44\text{g}}\text{Sc}$ radioisotope is undoubtedly most useful for the proof of concept. However, there are many other attractive candidates, in particular those already popular in nuclear medicine: ^{66}Ga , ^{72}As , $^{82}\text{Sr}/^{82\text{g}}\text{Rb}$ and ^{124}I . The coincidence imaging can also be employed in PET/MRI technique with the use of $^{52\text{m}}\text{Mn}$ and ^{69}Ge or with many $^{99\text{m}}\text{Tc}$ -radiopharmaceuticals with the use of $^{94\text{m}}\text{Tc}$. The summary of $\beta^+\gamma$ PET radioisotopes is shown in Table 3. Despite discussed drawbacks, the presented radioisotopes can be useful in the $\beta^+\gamma$ coincidence PET depending on the available scanner and the nature of the study.

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Table 1. Physical properties of the possible $\beta^+\gamma$ PET radioisotopes (IAEA, 2019)

Isotope	$T_{1/2}$	Decay (%)	Main gamma lines [keV] (intensity; delay)
^{10}C	19.3 s	EC (0.1), β^+ (99.9)	718.4 (100%; 0.71 ns)
^{14}O	70.6 s	EC (0.1), β^+ (99.9)	2312.6 (99.4%; 68 fs)
^{22}Na	2.6 y	EC (10), β^+ (90)	1274.5 (99.9%; 3.6 ps)
$^{34\text{m}}\text{Cl}$	32 min	EC (1.1), β^+ (54.3), IT (44.6)	146.4 (40.5%; instant), 1176.6 (14.1%; 136 fs), 2127.5 (42.8%; 318 fs), 3304.0 (12.3%; 136 fs)
$^{44\text{g}}\text{Sc}$	3.9 h	EC (5.7), β^+ (94.3)	1157.0 (99.9%; 2.61 ps)
^{48}V	16.0 d	EC (50.1), β^+ (49.9)	983.5 (100%; 4.04 ps), 1312.1 (97.5%; 0.76 ps)
$^{52\text{m}}\text{Mn}$	21 min	EC (1), β^+ (97), IT (2)	1434.1 (98.3%; 0.783 ps)
^{55}Co	17.5 h	EC (24), β^+ (76)	477.2 (20.2%; 37.9 ps), 931.1 (75%; 8 ps), 1408.4 (16.9%; 37.9 ps)
^{60}Cu	24 min	EC (7), β^+ (93)	826.1 (22%; 0.59 ps), 1332.5 (88%; 0.735 ps)
^{66}Ga	9.49 h	EC (44), β^+ (56)	1039.2 (37%; 1.68 ps), 2752 (23%; not found)
^{69}Ge	39 h	EC (76), β^+ (24)	574.2 (13.3%; 1.7 ps), 872.1 (11.9%; 0.25 ps), 1107 (36%; 0.222 ps)
^{72}As	26 h	EC (14), β^+ (86)	834.0 (80%; 3.35 ps)
^{76}Br	16.2 h	EC (44.4), β^+ (55.6)	559.1 (74.0%; 12.3 ps), 657.0 (15.9%; 11 ps), 1853.7 (14.7%; not found)
$^{82\text{g}}\text{Rb}$	1.3 m	EC (5), β^+ (95)	776.5 (13%; 4.45 ps)
$^{86\text{g}}\text{Y}$	14.7 h	EC (68.1), β^+ (31.9)	443.1 (16.9%; 5 ns), 627.7 (32.6%; 0.9 ps), 703.3 (15.4%; 5 ns), 777.4 (22.4%; 0.386 ps), 1076.6 (82.5%; 1.46 ps), 1153.1 (30.5%; 1.73 ps), 1920.7 (20.8%; not found)
$^{94\text{m}}\text{Tc}$	52 min	EC (28), β^+ (72)	871.1 (94%; 2.77 ps)
$^{110\text{m}}\text{In}$	69.1 min	EC (38.8), β^+ (61.2)	657.8 (97.7%; 5.42 ps)
^{124}I	4.2 d	EC (77.3), β^+ (22.7)	602.7 (62.9%, 6.2 ps), 722.7 (10.4%; 1.04 ps), 1691.0 (11.2%; 0.17 ps)

Table 2. Natural abundance of the targets discussed in the text (IAEA, 2019).

Element	Natural isotopic composition
Al	$^{27}\text{Al} = 100\%$
C	$^{12}\text{C} = 98.93\%$, $^{13}\text{C} = 1.07\%$
Ca	$^{40}\text{Ca} = 96.94\%$, $^{42}\text{Ca} = 0.647\%$, $^{43}\text{Ca} = 0.135\%$, $^{44}\text{Ca} = 2.09\%$, $^{46}\text{Ca} = 0.04\%$, $^{48}\text{Ca} = 0.19\%$
Cd	$^{106}\text{Cd} = 1.25\%$, $^{108}\text{Cd} = 0.89\%$, $^{110}\text{Cd} = 12.49\%$, $^{111}\text{Cd} = 12.80\%$, $^{112}\text{Cd} = 24.13\%$, $^{113}\text{Cd} = 12.22\%$, $^{114}\text{Cd} = 28.73\%$, $^{116}\text{Cd} = 7.49\%$
Cr	$^{50}\text{Cr} = 4.35\%$, $^{52}\text{Cr} = 83.79\%$, $^{53}\text{Cr} = 9.5\%$, $^{54}\text{Cr} = 2.37\%$
Fe	$^{54}\text{Fe} = 5.85\%$, $^{56}\text{Fe} = 91.754\%$, $^{57}\text{Fe} = 2.12\%$, $^{58}\text{Fe} = 0.28\%$
Ga	$^{69}\text{Ga} = 60.11\%$, $^{71}\text{Ga} = 39.89\%$
Ge	$^{70}\text{Ge} = 20.57\%$, $^{72}\text{Ge} = 27.45\%$, $^{73}\text{Ge} = 7.75\%$, $^{74}\text{Ge} = 36.5\%$, $^{76}\text{Ge} = 7.73\%$
Mo	$^{92}\text{Mo} = 14.53\%$, $^{94}\text{Mo} = 9.15\%$, $^{95}\text{Mo} = 15.84\%$, $^{96}\text{Mo} = 16.67\%$, $^{97}\text{Mo} = 9.6\%$, $^{98}\text{Mo} = 24.39\%$, $^{100}\text{Mo} = 9.82\%$
N	$^{14}\text{N} = 99.64\%$, $^{15}\text{N} = 0.36\%$
Ni	$^{58}\text{Ni} = 68.08\%$, $^{60}\text{Ni} = 26.22\%$, $^{61}\text{Ni} = 1.14\%$, $^{62}\text{Ni} = 3.63\%$, $^{64}\text{Ni} = 0.93\%$
O	$^{16}\text{O} = 99.76\%$, $^{17}\text{O} = 0.04\%$, $^{18}\text{O} = 0.2\%$
Rb	$^{85}\text{Rb} = 72.17\%$, $^{87}\text{Rb} = 27.83\%$
S	$^{32}\text{S} = 94.99\%$, $^{33}\text{S} = 0.75\%$, $^{34}\text{S} = 4.25\%$, $^{36}\text{S} = 0.01\%$
Se	$^{74}\text{Se} = 0.89\%$, $^{76}\text{Se} = 9.37\%$, $^{77}\text{Se} = 7.63\%$, $^{78}\text{Se} = 28.77\%$, $^{80}\text{Se} = 49.61\%$, $^{82}\text{Se} = 8.73\%$
Sr	$^{84}\text{Sr} = 0.56\%$, $^{86}\text{Sr} = 9.86\%$, $^{87}\text{Sr} = 7.0\%$, $^{88}\text{Sr} = 82.58\%$
Ta	$^{180\text{m}}\text{Ta} = 0.012$, $^{181}\text{Ta} = 99.988\%$
Te	$^{120}\text{Te} = 0.09\%$, $^{122}\text{Te} = 2.55\%$, $^{123}\text{Te} = 0.89\%$, $^{124}\text{Te} = 4.74\%$, $^{125}\text{Te} = 7.07\%$, $^{126}\text{Te} = 18.84\%$, $^{128}\text{Te} = 31.74\%$, $^{130}\text{Te} = 34.08\%$
Ti	$^{46}\text{Ti} = 8.25\%$, $^{47}\text{Ti} = 7.44\%$, $^{48}\text{Ti} = 73.72\%$, $^{49}\text{Ti} = 5.41\%$, $^{50}\text{Ti} = 5.18\%$
Zn	$^{64}\text{Zn} = 49.17\%$, $^{63}\text{Zn} = 27.73\%$, $^{67}\text{Zn} = 4.04\%$, $^{68}\text{Zn} = 18.45\%$, $^{70}\text{Zn} = 0.61\%$

Table 3. The arbitrary summary of $\beta^+\gamma$ PET radioisotopes discussed in the text. “Plus” sign indicates favourable characteristic, “minus” stands for unfavourable, and “plus/minus” marks moderate. The order of the radioisotopes in the table corresponds to their overall score.

radio-isotope	T _{1/2}	high β^+ branching ratio	intensive third γ (~1 MeV)	no additional γ lines	feasible production	already used in practice	theranostic pair
^{44g} Sc	3.9 h	+	+	+	+	+/-	+
⁷² As	26 h	+	+	+	+/-	+	+
⁵⁵ Co	17.5 h	+	+	+/-	+/-	+	+
⁶⁰ Cu	24 min	+	+	+/-	+	-	+
^{52m} Mn	21 min	+	+	+	+/-	+/-	-
⁶⁶ Ga	9.49 h	+/-	+/-	+/-	+	+	+/-
^{110m} In	69.1 min	+/-	+/-	+	+	-	+
^{82g} Rb	1.3 m	+	-	+	+/-	+	-
^{94m} Tc	52 min	+/-	+	+	+	-	-
¹²⁴ I	4.2 d	-	+/-	+/-	+/-	+	+
⁴⁸ V	16.0 d	+/-	+	-	+	+/-	-
^{86g} Y	14.7 h	-	+	-	+	-	+
⁷⁶ Br	16.2 h	+/-	+/-	+/-	+	+/-	-
⁶⁹ Ge	39 h	-	+/-	+/-	+/-	-	-
^{34m} Cl	32 min	+/-	+/-	-	-	-	-

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