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1	Higher plasma oxidative damage and lower plasma antioxidant defences in an
2	Arctic seabird exposed to perfluoroalkyl carboxylic acids
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26 ABSTRACT

27 Perfluoroalkyl and polyfluoroalkyl substances (PFASs) may cause detrimental effects on 28 physiological function and reproduction of Arctic animals. However, there is a paucity of 29 information on the link between PFASs and oxidative stress, which can have potential detrimental effects on key fitness traits, such as cellular homeostasis or reproduction. We 30 have examined the correlations between multiple blood-based markers of oxidative status 31 32 and several perfluoroalkyl carboxylic acids (i.e., with 8 or more carbons) in male Arctic black-legged kittiwakes (Rissa tridactyla) during the pre-laying period. Higher protein 33 34 oxidative damage was found in those birds having higher concentrations of acid (PFDoA), perfluorotridecanoic 35 perfluorododecanoic acid (PFTriA) and perfluorotetradecanoic acid (PFTeA). Lower plasmatic non-enzymatic micro-molecular 36 37 antioxidants was found in those birds having higher concentrations of perfluoroundecanoic acid (PFUnA), PFDoA and PFTeA. Effect size estimates showed 38 that the significant correlations between PFASs and oxidative status markers were 39 intermediate to strong. The non-enzymatic antioxidant capacity (including antioxidants 40 of protein origin) was significantly lower in those birds having higher plasma 41 concentration of linear perfluorooctanesulfonic acid (PFOSlin) only. In contrast, the 42 activity of the antioxidant enzyme glutathione peroxidase in erythrocytes was not 43 associated with any PFAS compounds. Our results suggest that increased oxidative stress 44 45 might be one consequence of long-chain PFAS contamination. Given the correlative nature of our study, experimental work will be needed to demonstrate whether PFASs 46 cause toxic effects on free-living vertebrates through increased oxidative stress. 47

49 **INTRODUCTION**

Ecotoxicological studies have so far extensively directed their attention toward legacy 50 persistent organic pollutants (POPs) like organochlorine pesticides (OCPs) and 51 polychlorobiphenyls (PCBs).¹ In contrast, less attention has been given to the 52 environmental toxicity of other organic contaminants in the Arctic, such as chlorinated 53 paraffins, phthalates, siloxanes or the perfluoroalkyl and polyfluoroalkyl substances 54 (PFASs).² Among these, PFASs remain comparatively much less investigated.³ PFASs 55 are synthetically manufactured chemicals, produced since the 1950s, that are widely used 56 57 for numerous industrial and commercial purposes as water repellents and surfactants (e.g., impregnation agents for carpets, papers and textiles, fire-fighting foam, non-stick coating 58 and waterproof clothing).⁴⁻⁵ Chemically and thermally stable, PFASs are highly persistent 59 in the environment and have been detected globally in both wildlife and humans.⁶⁻⁸ 60 Because of oceanic currents and atmospheric long-range transport, PFASs and their 61 precursors and breakdown products can reach high latitudes, such as the Arctic Ocean.⁹⁻ 62 ¹¹ Once deposited in the Arctic marine ecosystem, PFASs bio-accumulate in living 63 organisms and bio-magnify along the food webs.¹²⁻¹⁶ Importantly, (i) PFASs have long 64 half-lives, which facilitates their bio-accumulation through the food webs depending on 65 the species and PFAS congener⁸; (ii) the long and odd carbon-chain-length PFASs appear 66 to be more bio-accumulative and toxic than the short and even-chain length PFASs in 67 wildlife.17-20 68

It is worthwhile to note that while PFASs have been produced for over 50 years, it is only since late 1990s that their occurrence in the environment has come under scientific scrutiny. PFASs have raised recent concerns about their potential physiological disrupting properties and negative impacts on reproductive fitness in wildlife (multiple species in ²¹; lesser black-backed gull in ²²; northern fulmar in ²³; zebrafish in ²⁴; tree
swallow in ²⁵; black-legged kittiwake and northern fulmar in ²⁶; black-legged kittiwake
in ²⁷⁻²⁸; black-legged kittiwake in ²⁹⁻³⁰; glaucous gull in ³¹; common eider, black guillemot,
black-legged kittiwake, glaucous gull, arctic skua and great skua in ³²).

Increased molecular oxidative damage and disruption of antioxidant defences are 77 suspected as important mechanisms through which PFASs could be detrimental for cell 78 function and, possibly, for organism health. Experimental evidence on laboratory models 79 80 found that PFASs may increase production of reactive oxygen species (ROS), increase molecular oxidative damage and up- or down-regulate antioxidant defences.^{24,33-34} 81 Further work found that the perfluoroundecanoic and perfluorododecanoic acids (PFUnA 82 83 and PFDoA, respectively) are equally potent inducers of stress response genes relative to perfluorooctane sulfonic acid (PFOS) and perfluorononanoic acid (PFNA) and that the 84 effect of carbon-chain-length was more important than the functional group in 85 determining oxidative stress.³⁵ There is thus good reason to expect that long-chain PFASs 86 might cause dysregulation of the oxidative homeostasis, leading to accumulation of 87 88 oxidative damage to key biomolecules like proteins or nucleic acids. However, the effect of PFAS exposure on oxidative stress is almost unknown for wildlife. To the best of our 89 knowledge, only two studies have addressed this topic in free-living animals. ³⁶ did not 90 91 find any significant relationship between plasma PFAS concentrations and the activity of the antioxidant enzyme superoxide dismutase in plasma of white-tailed eagle nestlings, 92 while ³² did not find any significant relationship between PFASs and amount of DNA 93 94 damage in lymphocytes.

Long-lived species, like many polar seabirds that occupy high trophic levels, are
exposed to a greater risk of accumulation and sensitivity to high concentrations of

97 contaminants. In Svalbard (European Arctic), a number of studies showed that black-98 legged kittiwakes (*Rissa tridactyla*, hereafter "kittiwake") are chronically exposed to a 99 complex cocktail of organic contaminants and trace elements, which are known to 100 correlate with physiological metrics, impaired individual fitness and population 101 dynamics.^{27-30,37-39} It is, however, unknown whether exposure of kittiwakes to PFASs is 102 associated with markers of oxidative damage and antioxidant protection.

103 In this study, we have examined the correlations between blood-based markers of 104 oxidative status and several PFAS compounds in male kittiwakes during the pre-laying 105 period, while controlling for a number of potential confounding factors that might affect the oxidative status independently from PFASs (i.e., body condition, body size, both time 106 and day of blood sampling, hormonal status; reviewed in ⁴⁰). As with the hormonal status, 107 we measured plasma levels of testosterone, baseline corticosterone and luteinizing 108 hormone because prior work found large individual variation among kittiwakes^{27-28,38} and 109 significant effects on organism's oxidative status⁴⁰, which could affect the relationships 110 111 between PFAS and oxidative status markers. We focused on males because this 112 investigation on oxidative stress is part of a larger project aiming at assessing the overall 113 consequences (ornament coloration, fecundity, oxidative stress, sexual hormones) of PFASs exposure in males during the pre-laying stage (nest site defence, pair-bonding, 114 115 copulation, nest building) a period during which males appear to be sensitive to pollutants.³⁸ 116

We have also examined whether the effect size of the association between each
oxidative status marker and each PFAS compound varies according to their carbon-chainlength (C₈₋₁₄) because the toxicity of PFASs may increase with carbon-chain-length.

120

121 MATERIALS AND METHODS

122 Sampling

Fieldwork was conducted in 2016 on a colony of Arctic kittiwakes at Kongsfjord (78° 54' 123 N; 12° 13' E), Svalbard. Blood samples were collected from 50 adult males during the 124 pre-laying period (courtship and mating period), from 25th May to 6th June. Birds were 125 caught on their nest with a loop at the end of a long pole. Within 3 minutes since capture 126 127 0.5 ml of blood were taken from the brachial vein using a heparinized syringe and a 25gauge needle. This blood sample was used to measure oxidative status markers and 128 129 hormones. Straightaway, a second sample of venous blood (ca. 2 ml) was collected using another syringe and this blood sample was used to assess the PFASs burden. Then, tarsus, 130 131 skull (head + bill) and wing length were measured using a calliper (nearest 0.1 mm) and body mass was taken using a Pesola spring balance (nearest 5 grams). Blood samples 132 133 were stored on ice in the field. On average, blood samples were stored on ice for 3h40min 38s (min: 1h30; max 9h55) before being centrifuged and stored at -80°C. Plasma and red 134 blood cells, obtained after centrifugation, were kept frozen separately, either at -80°C for 135 136 subsequent oxidative status markers or at -20°C for PFAS analyses. All samples were analysed within 4 months since collection. 137

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139 Hormone assays

Plasma levels of testosterone, baseline corticosterone and luteinizing hormone (LH) were measured by radioimmunoassay at the Centre d'Etudes Biologiques de Chizé following protocols previously validated. Briefly, testosterone and corticosterone were extracted using diethyl ether and ethyl ether, respectively. Plasma concentrations of all three hormones were measured by radioimmunoassay. As with corticosterone, a commercial antiserum against corticosterone-3-(O-carboxy- methyl) oxime bovine serum albumin
conjugate (Biogenesis, UK) was used. The lowest detectable quantities significantly
different from zero at a 90% confidence level were 0.05 ng/ml for testosterone, 0.4 ng/ml
for corticosterone and 1.7 ng/ml for LH. All samples were analysed in duplicate.
Corticosterone and LH were analysed in a single run and the mean coefficient of variation
was 4.9 and 12.0%, respectively. Testosterone was analysed in two runs and the
coefficients of variation were 11.2 and 11.5%.

152

153 Oxidative status markers

154 One marker of plasma oxidative damage, one marker of plasma non-enzymatic 155 antioxidant capacity and one red blood cell antioxidant enzyme were measured at the Centre d'Etudes Biologiques de Chizé using standard methods. Protein carbonyls (marker 156 157 of oxidative protein damage) were measured using the Protein Carbonyl Colorimetric assay (Cayman Chemical Company, Ann Arbor, USA). This assay is based on the 158 colorimetric method proposed by ⁴¹. A same volume of plasma was used for all samples 159 160 and the amount of carbonyls was standardised by the plasma protein concentration 161 according to manufacturer's instructions. Protein carbonyls were derivatized to 2,4-162 dinitrophenylhydrazone by reaction with 2,4-dinitrophenylhydrazine (DNPH). The 163 absorbance was read at 370 nm. The extinction coefficient for DNPH (0.022/µM/cm) was 164 used to calculate the concentration of protein carbonyls, which was expressed as nmol/mg 165 protein (amount of carbonyls generated per unit of protein) or as total nmol obtained by 166 multiplying the concentration of carbonyls by the concentration of plasma proteins (i.e., total amount of carbonyls in the sample, which is also dependent on the amount of 167 substrates available, i.e., proteins). The mean coefficient of variation of duplicates was 168

169 11.5%. The metric expressed as nmol/mg indicates the amount of carbonyls that occurs 170 in a same amount of protein, thus this is standardised by the amount of substrates (i.e., proteins) that can be carbonylated. The second metric expressed as total amount of 171 carbonyls indicates the total amount of carbonyls that occurs in the tissue, which is 172 173 influenced by the amount of proteins available. This second metric is also important because accumulation of carbonyls is detrimental for the cells⁵⁰. The OXY-Adsorbent test 174 (Diacron International, Italy) was used to quantify the non-enzymatic antioxidant 175 176 capacity of plasma against HOCl. Values were expressed as either mM of HOCl neutralised or as mM of HOCl neutralised/mg protein to estimate the antioxidant potential 177 178 of micromolecular antioxidants (e.g., vitamins, carotenoids, glutathione) without the 179 contribution of proteins (i.e., non-enzymatic micro-molecular antioxidant capacity). The correlation between OXY values and protein concentration was actually high and 180 significant (r = 0.76, p < 0.001), which is to be expected because plasma proteins, such 181 as albumin, are prone to react with HOCl. Although free-radical trapping properties vary 182 among proteins, standardising OXY values by the concentration of total plasma proteins 183 184 gave us some kind of control about the contribution of proteins to OXY. The mean 185 coefficient of variation of duplicates was 9.4%. The Ransel assay (RANDOX Laboratories, UK) was used to measure the activity of the antioxidant enzyme glutathione 186 peroxidase (GPX) in haemolysates (red blood cells diluted with distilled water). Values 187 188 were expressed as Units of GPX/mg of protein of haemolysate. The mean coefficient of variation of duplicates was 8.1%. The Bradford protein assay (Bio-Rad Laboratories, 189 190 Hercules, USA) with bovine albumin as a reference standard was used to measure the concentration of proteins in both plasma samples and haemolysates. 191

193 **PFAS analyses**

Perfluoroalkyl carboxylic acids (i.e., PFASs with 8 or more carbons) were analysed in 194 plasma at the Norwegian Institute for Air Research (NILU) in Tromsø, Norway. The 195 following compounds were analysed in each plasma sample: perfluorooctanesulfonamide 196 197 (PFOSA), perfluorobutanesulfonic acid (PFBS), perfluoropropanesulfonic acid (PFPS), perfluorohexanesulfonic acid (PFHxS), perfluoroheptanesulfonic acid (PFHpS), 198 perfluorooctane sulfonic acid (PFOSlin), branched perfluorooctane sulfonic acid 199 200 (PFOSbr), perfluorononane sulfonic acid (PFNS), perfluorodecane sulfonic acid 201 (PFDcS), perfluorohexanoic acid (PFHxA), perfluoroheptanoic acid (PFHpA), perfluorooctanoic acid (PFOA), perfluorononanoic acid (PFNA), perfluorodecanoate 202 203 (PFDcA), perfluoroundecanoic acid (PFUnA), perfluorododecanoic acid (PFDoA), perfluorotridecanoic acid (PFTriA), perfluorotetradecanoic acid (PFTeA) and two 204 precursor compounds, the fluorotelomer sulfonates (6:2 FTS and 8:2 FTS). PFASs with 205 206 concentrations below the limit of quantification (LOQ) were replaced with a value equal 207 to (LOQ \times detection frequency) when the detection frequency (percentage of detection) 208 was > 50%. A 0.2 ml aliquot of plasma spiked with internal standards (carbon labeled PFAS)⁴² was extracted in methanol (1 ml) by repeated sonication and vortexing. The 209 supernatant was cleaned-up using ENVICarb graphitized carbon absorbent and glacial 210 211 acetic acid. Extracts were analysed by UPLC/MS/MS. Recovery of the internal standards 212 ranged between 86.3% and 120%. Results were validated with blanks (clean and empty glass tubes treated like a sample) and standard reference material (1957 human serum 213 214 from NIST) run with every 10 samples. The deviation of the target concentrations in the 215 SRMs were within the laboratory's accepted range (69 - 119%). Blanks varied between

concentrations below the instrument detection limits and 30 pg/g and were applied as the 216 LOD in the form of 3 times the average concentration. 217

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- 219

Body size and body condition

220 A body size index was estimated using the PC1 from a PCA of tarsus, skull and wing length (loadings, 0.73, 0.83 and 0.45, respectively). As far as the body condition, we did 221 222 not use the ratio of body mass onto body size nor the residuals of a linear regression of body mass onto body size because both indices have been criticized.⁴³ Rather we included 223 224 both body mass and body size index as factors in the models. In doing so, the coefficient 225 estimate of body mass is calculated considering the effect of body size, thus the outcome 226 reflects the effect of body condition on the given marker of oxidative status.⁴³

227

228 **Statistical analyses**

Generalized linear models were performed using the software STATISTICA 10 229 (StatSoft. Inc., Tulsa, OK, USA) to assess relationships between each oxidative status 230 231 marker (protein carbonylation, non-enzymatic antioxidant capacity, non-enzymatic 232 micro-molecular antioxidant capacity and glutathione peroxidase) and the following predictor variables: PFASs congener (PFOSlin, PFNA, PFDcA, PFUnA, PFDoA, PFTriA 233 234 or PFTeA), body size index, body mass, hormonal status, day of blood sampling and time 235 of blood sampling. All predictor variables were scaled to mean of 0 and standard deviation 236 of 1. These predictor variables were included in all models because prior work showed 237 that each of them can be significantly associated with oxidative status markers (reviewed in ⁴⁰). The two metrics of protein oxidative damage (protein carbonyls per mg of protein 238 and total protein carbonyls) were combined using the PC1 from a Principal Components 239

Analysis (PCA) because they were highly correlated (r = 0.79, p < 0.001). The hormonal 240 status was estimated using the PC1 from a PCA of corticosterone, testosterone and 241 242 luteinizing hormone (loadings, 0.39, 0.61 and 0.84, respectively). A normal error function and an identity-link function were applied to models of non-enzymatic antioxidant 243 244 capacity and GPX. A gamma error function and an identity-link function were applied to models of protein carbonyls and non-enzymatic micro-molecular antioxidant capacity. 245 246 These functions were selected because the model had the best fitting to the dataset 247 according to the Akaike Information Criterion. Preliminary analyses showed that the time 248 elapsed from the collection of blood to its storage was not significantly correlated (pvalue > 0.45) with any marker of oxidative status nor hormones, thus it was not further 249 250 considered in the analyses. Visual inspection of residuals, Q-Q plots and Cook's distance did not highlight any violation of normality and homogeneity of variance nor the presence 251 of outliers (all samples were below a 0.5 Cook's distance). The variance inflation factor 252 was always below 2, indicating that multicollinearity was low. The multicollinearity is 253 254 thought to be high and problematic when the variance inflation factor is higher than 5.

The "compute.es package"⁴⁴ in R⁴⁵ was used to calculate the standardized effect 255 256 size Hedges' g from test statistics of oxidative status markers that had significant associations with PFASs (i.e., protein oxidative damage and non-enzymatic micro-257 258 molecular antioxidant capacity). The "forestplots function" of the "metafor package" in 259 R was used to visualise boxplots of effect size and 95% confidence interval. Effect sizes were considered to be small (Hedges' g = 0.2, explaining 1% of the variance), 260 intermediate (g = 0.5, explaining 9% of the variance) or large (g = 0.8, explaining 25% 261 of the variance) according to 46 . 262

264 **RESULTS**

265 Concentrations of detectable PFASs are reported in Table 1 together with other variables 266 measured in kittiwakes. Six out of 20 PFASs (i.e. PFOSlin, PFNA, PFDcA, PFUnA, PFDoA, PFTriA) were detectable in all individuals, while one PFAS (i.e. PFTeA) was 267 268 detectable in 33 out of 50 individuals. PFOSlin, PFUnA and PFTriA concentrations were the highest of all PFASs measured in the investigated samples, with a percentage 269 270 contribution for each kittiwake ranging from 23.4 to 54.5%, from 20.3 to 34.8% and from 271 11.8 to 35.2%, respectively. The percentage contribution of all other detected PFASs ranged between 0.1 and 10.6%. PFOSA, PFBS, PFPS, PFHxS, PFHpS, PFOSbr, PFNS, 272 PFDcS, PFHxA, PFHpA, PFOA, and the two precursor fluorotelomer sulfonates (6:2 FTS 273 274 and 8:2 FTS) were below the detection limit in all the investigated samples.

Protein oxidative damage was significantly higher in those birds having higher 275 plasma concentration of PFDoA, PFTriA or PFTeA (Table 2). Effect size estimates 276 increased with chain length of PFASs (indicating an increase of protein damage with 277 278 chain length) and were significantly different from zero) for PFDoA (95% confidence 279 interval: 0.06 to 1.25), PFTriA (95% confidence interval: 0.01 to 1.20) and PFTeA (95% 280 confidence interval: 0.29 to 1.54; Fig. 1). The non-enzymatic micro-molecular antioxidant capacity was significantly lower in those birds having higher plasma concentration of 281 282 PFUnA, PFDoA or PFTeA (Table 2). There was also a near-significance tendency of the 283 non-enzymatic micro-molecular antioxidant capacity to be lower in birds with higher plasma PFTriA (Table 2). Effect size estimates were larger for longer PFASs (indicating 284 285 a decrease of micro-molecular antioxidants with chain length) and were significantly different from zero for PFUnA (95% confidence interval: -1.20 to -0.02), PFDoA (95% 286 confidence interval: -1.35 to -0.14) and PFTeA (95% confidence interval: -1.23 to -0.04; 287

Fig. 2). The non-enzymatic antioxidant capacity including the contribution of antioxidant of protein origin was significantly lower in those birds having higher plasma concentration of PFOSlin, but it was not associated with any other PFAS congener (Table 2). The activity of GPX was not associated with any PFAS compounds (Table 2).

Finally, our models showed that (i) kittiwakes in poorer body condition had more plasma protein carbonyls, (ii) the non-enzymatic antioxidant capacity was higher in kittiwakes sampled later in the day, and (iii) the non-enzymatic micro-molecular antioxidant capacity was higher in kittiwakes having lower concentrations of hormones.

296

297 **DISCUSSION**

298 Our results provide the first evidence in wild vertebrates that the correlation between oxidative status markers and PFASs is stronger for long-chain congeners. We found that 299 male kittiwakes having higher plasma concentrations of long-chain PFASs had more 300 301 protein oxidative damage and less plasma antioxidants after controlling statistically for 302 potentially confounding variables. The non-enzymatic antioxidant capacity (including antioxidants of protein origin) was significantly lower in kittiwakes having higher plasma 303 304 concentration of PFOSlin. On the other hand, the activity of glutathione peroxidase in 305 erythrocytes was not related to any PFAS congener. Effect size estimates were 306 intermediate to large, indicating that PFASs explained from 9 to more than 25% of the 307 variance in protein oxidative damage and non-enzymatic micro-molecular antioxidant capacity of plasma.⁴⁶ Intermediate effect sizes are suggested to be biologically 308 309 meaningful because average proportions of variance explained in ecological, evolutionary and physiological studies is usually below 7%.⁴⁷ Our effect size estimates were also larger 310

than those found in the comparison of oxidative status markers between animals living in
polluted (e.g., air pollution, heavy metals) and unpolluted sites.⁴⁸

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PFOSlin, PFUnA and PFTrA concentrations were the highest of all PFASs 313 measured in the investigated samples, with a percentage contribution for each kittiwake 314 315 ranging from 11.8 to 54.5%. The concentration of PFOSlin was higher than that previously recorded in males from the same kittiwake population in 2012 (13.4 vs. 10.6 316 pg/g ww in²⁹). In contrast, the average concentrations of PFUnA (10.3 vs. 12.1 pg/g ww 317 in²⁹) and of PFTrA (8.6 vs. 11.6 pg/g ww in²⁹) were both lower in our study than in prior 318 work.²⁹ One reason for such differences might be because ²⁹ measured PFASs of male 319 kittiwakes caught during the chick rearing phase. Work on male glaucous gulls during the 320 321 incubation period in Svalbard found levels of PFOSlin similar to ours, while those of PFUnA (4.4 pg/g ww) and of PFTrA (3.9 pg/g ww) were much lower than those we 322 recorded.³¹ 323

The strength of the correlation between oxidative status markers and PFASs 324 325 increased with the chain length of the congener. While persistent in the environment, 326 PFASs with fewer than eight carbons, such as PFHxA, and PFSAs with fewer than six 327 carbons, such as PFBS, are generally less toxic and bioaccumulative in wildlife and humans, while exposure of laboratory animals to long-chain congeners produces 328 detrimental reproductive, developmental, and systemic effects.^{4,6,18,49} For example, the 329 toxic effects of PFASs on rat brain cells decreased with increasing carbon chain length.¹⁷ 330 Interestingly, the toxic effects of PFASs were attenuated by the antioxidant vitamin E, 331 332 indicating a possible involvement of oxidative stress in the reduction of cell viability.⁴⁹ Further work showed that, compared to short-chain PFASs, long-chain PFASs are 333 stronger inducers of the response of genes regulating the cell oxidative status.³⁵ 334

Effects of PFAS exposure on oxidative status have been moderately investigated 335 336 in laboratory models and almost unexplored in wild animals. Thus, limited information is available for a comparison with our results and interpretation. Protein carbonylation 337 arises from overproduction of ROS by metabolic reactions that use oxygen and shift the 338 balance between oxidant/antioxidant statuses in favour of the oxidants.⁵⁰ Protein 339 carbonylation also occurs when carbonyls are introduced into proteins through the 340 reactions with lipid oxidative damage products (malondialdehyde and hydroxynonenal).⁵⁰ 341 342 Carbonylation is mostly irreversible and results in alteration of protein structure and 343 function. Only a small fraction of carbonylated proteins can be removed through proteasome-dependent proteolysis. Work carried out on laboratory models found 344 345 evidence that exposure to PFASs may change expression of genes regulating proteasome activation and proteolysis.^{6,51-52} It is, however, unclear, whether such changes in gene 346 expression make proteins one important target of the pro-oxidant effects of PFASs. This 347 is important because when protein carbonyls accumulate, they tend to aggregate leading 348 349 to cell death, tissue injury and development of disorders. Several studies found higher amounts of plasma protein carbonyls in individuals affected by a given disease,⁵³⁻⁵⁴ 350 351 suggesting a potential role of protein carbonylation in disease progression.

Depletion of circulating non-enzymatic antioxidants might reflect increased oxidation due to reaction with ROS, reduced intake from diet or mobilisation of antioxidants from blood to other target tissues. Irrespective of the reason, prior work on other bird species found evidence that circulating antioxidants may be linked to important individual or population fitness-related traits. For example, ⁵⁵ found that barn swallows (*Hirundo rustica*) with lower plasma non-enzymatic antioxidants had reduced probability of survival. ⁵⁶ found that Gentoo (*Pygoscelis papua*) and Adélie (*Pygoscelis adeliae*) penguins from increasing populations had higher plasma non-enzymatic antioxidantcapacity than those from decreasing populations.

Metabolic activity is one important source of ROS production. Thus the 361 association we found between PFASs and oxidative status markers might mirror an effect 362 363 of PFASs on metabolism. Prior work on the same kittiwake population found a positive association between the long-chain PFTriA and resting metabolic rate in females but not 364 in males.³⁰ Thus, the association between PFASs and oxidative status markers does not 365 366 appear, at least for males, to be due to a dysregulation of general body metabolism. The lack of an effect on metabolic rate does not reject the hypothesis that ROS production 367 368 might have been higher in the more contaminated birds. For example, PFASs might have 369 localised effects on important ROS generators (e.g., mitochondria of red blood cells or of other target tissues) without compromising the metabolism of the whole organism. In 370 vitro studies found that PFAS exposure may impair mitochondrial activity and lead to 371 increased rates of ROS production.⁵⁷ 372

373 It is unclear why the activity of glutathione peroxidase was not associated with 374 any PFAS compound. It might be that up-regulation of this enzyme might have been too 375 costly for the birds given the imminent start of breeding activity or that any effects of PFASs on oxidative status did not come through the pathways involving glutathione 376 377 peroxidase. The biochemical function of glutathione peroxidase is to reduce hydrogen peroxide to water and organic hydroperoxides to their corresponding alcohols.⁵⁰ Thus, we 378 379 cannot exclude that results would have been different if another antioxidant enzyme with 380 a different biochemical function would have been measured. There are, however, many discrepancies in the literature about the response of antioxidant enzymes to PFAS 381 exposure. For example, prior work did not find any association between PFASs exposure 382

and whole-body catalase activity in the planktonic crustacean *Daphnia magna*⁵⁸ or plasma superoxide dismutase activity in white-tailed eagle (*Haliaeetus albicilla*) nestlings.³⁶ In contrast, exposure to PFASs affected catalase activity in hepatocytes of freshwater tilapia *Oreochromis niloticus*²⁴ and expression of antioxidant genes Sod1, Sod2, Gpx2 and Nqo1 in mouse pancreas.⁵⁹ Irrespective of the mechanisms involved, the activity of GPX in erythrocytes does not appear to be an informative marker about the impact on oxidative status of the PFASs we have measured in this work.

390 In conclusion, our work shows that higher protein oxidative damage was found in 391 those birds having higher concentrations of PFDoA, PFTriA and PFTeA. Lower plasmatic non-enzymatic micro-molecular antioxidants was found in those birds having 392 393 higher concentrations of PFUnA, PFDoA and PFTeA. Experimental work will be needed to ascertain whether the correlation between individual PFAS burden and oxidative status 394 markers reflects a direct toxic effect of PFASs on oxidative homeostasis. It will also be 395 important to determine whether increased oxidative damage or decreased antioxidant 396 397 defences turn into a reduction in survival probability or lifetime reproductive success. Resistance against oxidative stress may also decrease with chronological age.⁶⁰ Thus, we 398 399 highlight the importance of assessing in future studies whether the effects of PFASs on 400 oxidative status markers are stronger in older individuals.

401

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418 **COMPETING INTERESTS**

419 The authors declare no competing financial interests.

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421 **REFERENCES**

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Table 1. Descriptive statistics of all poly- and perfluorinated alkyl substances congeners,

603 oxidative status markers (i.e., responde variables) and several individual metrics

604 (potential confounding factors) measured in 50 male black-legged kittiwakes from

605 Svalbard.

Variable	Mean ±	
	Standard	
	deviation	
CONTAMINANT		
PFOSlin - Linear perfluorooctane sulfonic acid (ng/g ww)	13.4 ± 6.2	
PFNA - Perfluorononanoic acid (ng/g ww)	2.0 ± 0.9	
PFDcA - Perfluorodecanoic acid (ng/g ww)	2.9 ± 1.2	
PFUnA - Perfluoroundecanoic acid (ng/g ww)	10.3 ± 3.7	
PFDoA - Perfluorododecanoic acid (ng/g ww)	1.7 ± 0.8	
PFTriA - Perfluorotridecanoic acid (ng/g ww)	8.6 ± 3.1	
PFTeA - Perfluorotetradecanoic acid (ng/g ww)	1.0 ± 0.8	
OXIDATIVE STATUS MARKERS – RESPONSE VARIABLES		
Protein carbonyls (nmol/mg protein)	1.8 ± 0.6	
Total protein carbonyls (nmol)	22.5 ± 8.2	
Glutathione peroxidase (Units/mg protein)	0.4 ± 0.2	
Non-enzymatic antioxidant capacity (mM HOCl neutralised)	169.9 ± 37.1	
Non-enzymatic micro-molecular antioxidant capacity (mM HOCl neutralised/mg protein)	13.0 ± 2.0	
INDIVIDUAL TRAITS		
Body mass (g)	416.2 ± 23.1	
Skull length (mm)	93.9 ± 1.6	
Tarsus length (mm)	34.9 ± 1.0	
Wing length (mm)	317.1 ± 6.3	
Baseline corticosterone (ng/ml)	9.6 ± 7.4	
Testosterone (ng/ml)	1.7 ± 1.4	
Luteinizing hormone (ng/ml)	6.2 ± 2.4	

607 Table 2. The table shows the outcomes of generalized linear models performed to test the effect of each PFASs congener on oxidative status markers. Each model also included a 608 609 number of potential confounding factors that may affect oxidative status markers 610 independently from PFASs. Significant effects are shown in bold type. Note that the coefficient estimate for the relationship between body mass and the dependent variable 611 (i.e., oxidative status marker) is calculated considering the effect of body size, thus it 612 actually indicates the covariation between a given marker and the individual body 613 condition.⁴³ GPX = glutathione peroxidase; N-E Antioxs = non-enzymatic antioxidant 614 capacity of plasma; N-E Micromol Antioxs = non-enzymatic micro-molecular 615 616 antioxidant capacity of plasma; $ce \pm se = coefficient estimate \pm standard error$.

	Protein		GPX		N-E		N-E	T
	Oxidative		_		Antioxs		Micromol	
	Damage		-				Antioxs	+
Main effect	C0 + 50	P	C0 + 50	P	C0 + 50	P	C0 + S0	n
included in the	00 1 30		00 1 30	'	00 1 30		00 1 30	٢
model								
PFOSlin	0.03±0.12	0.817	0.02±0.02	0.462	-13.7±5.0	0.006	-0.20±0.28	0.484
sampling date	0.11±0.13	0.406	0.03±0.03	0.309	-16.6±5.0	0.001	0.18±0.31	0.546
sampling time	-0.20±0.11	0.074	0.03±0.02	0.196	14.3±4.6	0.002	0.22±0.28	0.425
body mass	-0.29±0.12	0.013	-0.02±0.03	0.407	4.0±5.0	0.424	0.00±0.29	1.000
body size	-0.02±0.11	0.840	0.02±0.02	0.403	-5.5±4.7	0.236	0.10±0.27	0.704
hormonal status	-0.10±0.11	0.391	0.02±0.02	0.348	4.0±4.7	0.391	-0.94±0.28	0.001
				0.70/				
PENA	-0.03±0.13	0.852	0.01±0.03	0.764	-1.0±6.0	0.872	-0.09±0.34	0.804
sampling date	0.08±0.15	0.580	0.02±0.03	0.433	-10.7±6.0	0.077	0.22±0.35	0.534
sampling time	-0.20±0.11	0.074	0.03±0.02	0.208	14.2±5.0	0.004	0.22±0.28	0.423
body mass	-0.30±0.12	0.011	-0.02±0.03	0.395	5.2±5.4	0.330	0.01±0.29	0.977
bouy size	-0.01 ± 0.11	0.690	0.02±0.02	0.353	-7.4±0.0	0.137	0.00±0.27	0.700
nonnonai siaius	-0.00±0.11	0.400	0.03±0.02	0.202	0.20±4.9	0.955	-0.90±0.27	<0.001
PEDcA	0.22+0.13	0.106	0.00+0.03	0 904	-0.65+5.8	0.910	-0.49+0.31	0 116
sampling date	0.22+0.14	0.100	0.02+0.03	0.493	-10 5+5 0	0.075	-0.02+0.33	0.950
sampling time	-0.18+0.11	0.088	0.03+0.02	0.100	14 1+4 9	0.004	0.21+0.27	0.000
body mass	-0.22+0.12	0.056	-0.02+0.03	0.407	5 1+5 6	0.357	-0.07+0.29	0.809
body size	-0.03±0.11	0.795	0.02±0.02	0.352	-7.4±5.0	0.140	0.09±0.27	0.723
hormonal status	-0.14±0.11	0.204	0.03±0.02	0.249	0.24±4.9	0.960	-0.92±0.27	0.001
PFUnA	0.14±0.13	0.268	0.00±0.03	0.970	-3.3±5.6	0.562	-0.62±0.29	0.035
sampling date	0.19±0.14	0.181	0.02±0.03	0.517	-12.0±5.8	0.038	-0.07±0.32	0.822
sampling time	-0.18±0.11	0.098	0.03±0.02	0.196	13.9±4.9	0.005	0.19±0.27	0.483
body mass	-0.24±0.12	0.045	-0.02±0.03	0.403	4.2±5.7	0.453	-0.13±0.29	0.659
body size	-0.03±0.11	0.807	0.02±0.02	0.348	-7.1±5.0	0.154	0.12±0.26	0.644
hormonal status	-0.13±0.11	0.237	0.03±0.02	0.247	0.86±4.9	0.861	-0.87±0.26	0.001
PFDoA	0.29±0.13	0.024	-0.01±0.03	0.761	-1.5±5.4	0.779	-0.73±0.29	0.012
sampling date	0.26±0.14	0.066	0.01±0.03	0.625	-10.9±5.6	0.052	-0.10±0.30	0.736
sampling time	-0.18±0.10	0.088	0.03±0.02	0.199	14.1±4.9	0.004	0.21±0.26	0.421
body mass	-0.20±0.11	0.076	-0.02±0.03	0.343	4.9±5.6	0.378	-0.16±0.28	0.563
body size	-0.02±0.11	0.833	0.02±0.02	0.325	-7.2±5.0	0.148	0.14±0.25	0.574
normonal status	-0.14±0.10	0.176	0.03±0.02	0.220	0.21±4.8	0.965	-0.95±0.25	<0.001
PFTri∆	0 30+0 14	0.036	0.00+0.03	0.923	2 0+5 9	0.735	-0.58+0.32	0.064
sampling date	0.26±0.14	0.075	0.00±0.00	0.520	-8.8+6.1	0.153	-0.12+0.34	0.725
sampling time	-0.18+0.11	0.094	0.02±0.00	0.007	14 5+5 1	0.100	0.10+0.27	0.725
body mass	-0.26+0.11	0.026	-0.02+0.02	0.368	5 7+5 5	0.299	-0.08+0.28	0.723
body size	-0.06+0.12	0.632	0.02+0.02	0.348	-7 9+5 2	0.124	0.20+0.27	0.454
hormonal status	-0.14±0.11	0.181	0.03±0.02	0.226	-0.13±4.9	0.979	-0.92±0.26	<0.001
PFTeA	0.33±0.11	0.003	-0.02±0.03	0.482	7.4±5.4	0.171	-0.65±0.29	0.028
sampling date	0.22±0.12	0.070	0.01±0.02	0.664	-7.4±5.1	0.145	0.05±0.29	0.852
sampling time	-0.27±0.11	0.013	0.03±0.02	0.149	12.3±5.0	0.015	0.40±0.27	0.144
body mass	-0.25±0.11	0.018	-0.03±0.03	0.317	6.4±5.3	0.227	-0.05±0.28	0.858
body size	-0.05±0.11	0.668	0.03±0.02	0.275	-9.0±5.0	0.072	0.19±0.26	0.461
hormonal status	-0.19±0.10	0.062	0.03±0.02	0.171	-1.9±4.9	0.704	-0.84±0.27	0.001

618 **Figure captions**

Figure 1. Mean estimates of effect size and 95% confidence interval are shown. These were calculated from the statistical outcomes showing the effect of each PFAS congener on protein oxidative damage. Estimates are positive when values of damage are higher in birds having higher plasma concentration of a given PFAS congener. Note that effect size estimates are significant when the 95% confidence interval does not include zero. Numbers in bracket indicate carbon-chain-length of each PFAS congener.

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Figure 2. Mean estimates of effect size and 95% confidence interval were calculated from the statistical outcomes showing the effect of each PFAS congener on plasma nonenzymatic micromolecular antioxidant capacity. Estimates are negative when values of antioxidants are lower in birds having higher plasma concentration of a given PFAS congener. Note that effect size estimates are significant when the 95% confidence interval does not include zero. Numbers in bracket indicate carbon-chain-length of each PFAS congener.



636 Figure 1



