



**HAL**  
open science

## Bio-catalytic hydrolysis of paper pulp using in- and ex-situ multi-physical approaches: Focus on semidilute conditions to progress towards concentrated suspensions

Tien Cuong Nguyen, Dominique Anne-Archard, Xavier Cameleyre, Eric Lombard, Kim Anh To, Luc Fillaudeau

### ► To cite this version:

Tien Cuong Nguyen, Dominique Anne-Archard, Xavier Cameleyre, Eric Lombard, Kim Anh To, et al.. Bio-catalytic hydrolysis of paper pulp using in- and ex-situ multi-physical approaches: Focus on semidilute conditions to progress towards concentrated suspensions. *Biomass and Bioenergy*, 2019, 122, pp.28-36. 10.1016/j.biombioe.2019.01.006 . hal-02177352

**HAL Id: hal-02177352**

**<https://hal.science/hal-02177352v1>**

Submitted on 21 Oct 2021

**HAL** is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.



Distributed under a Creative Commons Attribution - NonCommercial 4.0 International License



19 **Abstract**

20 In order to make 2<sup>nd</sup>-generation biofuels more competitive, high solid-matter content  
21 has to be reached. To progress towards this target, the mechanism for destructuring  
22 lignocellulose fibres in semidilute conditions has to be well understood, as this  
23 configuration shows the basic mechanism which limits transfers and efficiency. This  
24 study aims to delve deeply into the biophysical and transfer limitations occurring during  
25 enzymatic hydrolysis. A specific experimental set-up associating in-situ and ex-situ  
26 physical (rheometry, chord length analysis) and biochemical analysis was used to  
27 expand the knowledge of hydrolysis of extruded softwood paper pulp over 24 h under  
28 different substrate concentrations (1% to 3%) and enzyme doses (Accellerase 1500, 5  
29 and 25 FPU/g cellulose). Non-Newtonian behaviour associated with pronounced yield  
30 stress stand as the major factors limiting process efficiency. A critical time was deduced  
31 from viscosity evolution, and the existence of a unique, dimensionless viscosity-time  
32 curve was established, suggesting similar mechanisms for fibre degradation. In addition,  
33 chord length distribution allowed for the description of population evolution and was  
34 discussed in the light of in-situ viscosity and hydrolysis yield. Physical (viscosity,  
35 particle size) and biochemical (substrate) kinetics were modelled (second-order) and  
36 coefficients identified. A chronology of the encountered phenomenological limitations  
37 demonstrates the necessity of optimising bioprocesses by considering physical  
38 parameters. A reference feed rate is proposed in order to reach high solid loading under  
39 fed-batch strategy.

40

41 **Keywords:** biorefinery; enzymatic hydrolysis; kinetics modelling; paper pulp;  
42 rheometry; yield stress.

43

#### 44 Nomenclature

|                 |                                   |   |
|-----------------|-----------------------------------|---|
| C               | Torque (mixing system)            | (N.m)                                     |
| C <sub>m</sub>  | Mass concentration                | (gdm.L <sup>-1</sup> )                    |
| Cellu           | Cellulose content                 | (%)                                       |
| DM              | Dry matter content                | (% g.L <sup>-1</sup> )                    |
| DP              | Degree of polymerisation          | (/)                                       |
| E/S             | Enzyme/substrate ratio            | (mL.g <sup>-1</sup> )                     |
| [Glc]           | Glucose concentration             | (g.L <sup>-1</sup> )                      |
| G'              | Elastic modulus                   | (Pa)                                      |
| G''             | Viscous modulus                   | (Pa)                                      |
| K <sub>p</sub>  | Power constant                    | (/)                                       |
| K <sub>s</sub>  | Metzner-Otto constant             | (/)                                       |
| k <sub>μ</sub>  | Rheological kinetic coefficient   | (Pa <sup>-1</sup> .s <sup>-2</sup> )      |
| k <sub>lc</sub> | Granulometric kinetic coefficient | (μm <sup>-1</sup> .s <sup>-1</sup> )      |
| k <sub>s</sub>  | Biochemical kinetic coefficient   | ((gdm/L) <sup>-1</sup> .s <sup>-1</sup> ) |
| lc              | Chord length                      | μm  |
| lc <sub>m</sub> | Mean chord length                 | μm  |
| m <sub>s</sub>  | Quantity of substrate             | (g humid matter)                          |
| N               | Rotation speed                    | (revolutions per second)                  |
| N <sub>p</sub>  | Power number                      | (/)                                       |
| P               | Power consumption                 | (W)                                       |
| Re              | Reynolds number                   | (/)                                       |
| R <sup>2</sup>  | Correlation coefficient           | (/)                                       |
| V <sub>w</sub>  | Water volume                      | (L)                                       |
| γ̇              | Shear rate                        | (s <sup>-1</sup> )                        |
| ρ               | Density                           | (kg.m <sup>-3</sup> )                     |

|              |                         |                        |
|--------------|-------------------------|------------------------|
| $\rho_s$     | Substrate density       | ( $\text{kg.m}^{-3}$ ) |
| $\mu$        | Viscosity               | (Pa.s)                 |
| $\mu_0$      | Initial viscosity       | (Pa.s)                 |
| $\mu_\infty$ | Final viscosity         | (Pa.s)                 |
| $\mu^*$      | Dimensionless viscosity | (/)                    |
| $\tau$       | Shear stress            | (Pa)                   |
| $\tau_0$     | Yield stress            | (Pa)                   |

45

## 46 **1 Introduction**

47 The latest global agreement, the Paris agreement, was established in 2015 to reduce the  
48 scale of the fossil-fuel-based economy. Greater use of renewable resources with low  
49 carbon footprints is being advocated to substitute the usage of fossil fuels and to achieve  
50 the goal of a decarbonised economy. Biofuel stands as one potential alternative, and is  
51 being scrutinised in many countries [1]. This development is not only justified by  
52 economic reasons, but also by societal demand and environmental constraints that  
53 necessitate the move towards renewable energies. Cellulosic biomass provides a low-  
54 cost, renewable, and abundant resource that has the potential to support large-scale  
55 production of fuels and chemicals via biotechnological routes [2]. Among the major  
56 users of lignocellulose resources generated through forestry and agricultural practice,  
57 the pulp and paper industry holds a strategic position. Currently, the promotion of  
58 biorefineries producing multiple products, including higher-value chemicals as well as  
59 fuels and power, is a major objective of numerous consolidated programs in the world.  
60 In order to achieve economic viability, the biorefinement of lignocellulosic resources  
61 must be operated at very high feedstock dry matter content. This strict prerequisite  
62 imposes a considerable constraint, particularly on the physicochemical and biocatalytic  
63 steps, which, overall, aim to produce high-quality, fermentable sugar syrups. Moreover,  
64 industrial criteria regarding maximum reactor volumes, energy and water consumption,  
65 and wastewater management must also be respected. The pulp and paper industry is able  
66 to provide a tried and tested industrial model for processing lignocellulose biomass into  
67 pre-treated cellulose pulps. The pulp product of this industry is appropriate for modern  
68 biorefining because it displays low lignin content, is free of inhibitory compounds that  
69 can perturb fermentations, and is devoid of microbial contaminants [3]. The process of  
70 biofuel production can thus be coupled with the pulp and paper industry. Two main  
71 advantages are highlighted here. The first is the perfect control of paper pulp quality due

72 to the efficiency of woody substrate pre-treatment and to the elimination of the lignin  
73 fraction in the initial biomass. The second is that the pulp and paper industry produces  
74 energy by valorising residues and co-products, thus reducing final biofuel costs [4].

75 In order to produce 2<sup>nd</sup>-generation biofuels, pre-treated paper pulp would be hydrolysed  
76 and then fermented to convert the simple sugars (hexoses and also pentoses) into  
77 molecules of interest. Conventionally, this process involves two separate steps:  
78 hydrolysis and fermentation (SHF). In addition, it can also use simultaneous  
79 saccharification and fermentation processes or, more recently, consolidated  
80 bioprocessing, which integrates enzyme production, saccharification, and fermentation  
81 into a single process. In the SHF process, cellulose is enzymatically hydrolysed by  
82 cellulases during the first step to form simple sugars (e.g. glucose), which are consumed  
83 in the second step by *Saccharomyces*, *Zymomonas*, or other microorganisms to obtain  
84 the desired product [5-7]. The main advantage of SHF is the ability to carry out each  
85 step in its optimum conditions: temperature, pH, etc.

86 The key steps are always the pre-treatment techniques and the conversion into  
87 fermentable sugars [8]. A better scientific understanding, and ultimately exact technical  
88 control, of these critical biocatalytic reactions, which involve complex matrices at high  
89 solid content, currently pose a major challenge that must be overcome to facilitate the  
90 intensification of biorefining operations. Among the main parameters to be studied, the  
91 rheological behaviour of the hydrolysis suspension and the fibre particle size stand out  
92 as major determinants of process efficiency, and are responsible for the choice of  
93 equipment to be used and the strategies to be applied [9]. Nevertheless, only a limited  
94 number of studies cover the rheological behaviour of lignocellulose matrices in highly  
95 concentrated suspensions during biocatalytic degradation [10].

96 The purpose of this study was to investigate the destructuring of fibre during enzyme  
97 attack under a multi-scale approach using different analytical techniques. Semidilute  
98 conditions were chosen, as they introduce the complexity of particle-particle

99 interactions, which are strongly involved in the transfer limitations observed in high dry  
100 matter content, without inhibiting the bioreactions. **Figure 1** illustrates the three blocks  
101 corresponding to the three levels of observation: macro-scale with viscometry and  
102 rheometry, micro-scale with chord length distribution (CLD) and molecular scale with  
103 biochemistry (chemical analyses of soluble fraction and solid fraction). This three-  
104 membered framework permits the analysis and comparison of in- and ex-situ methods  
105 (excluding biochemical analysis). Experiments were conducted using three substrate  
106 concentrations in a semidilute regime and two enzyme concentrations to explore the  
107 impact of substrate properties and enzyme ratio on fibre destructuring kinetics  
108 (bioconversion rate). Then, phenomenological models can be established and  
109 considered as a whole to provide a full overview of the mechanism. In this regard, the  
110 phenomenological models should fulfill certain criteria, such as reliability, simplicity,  
111 and congruity with the experimental information. It is hoped that a global result from  
112 these three sub-blocks will provide a "knowledge block" to explain certain scientific  
113 limitations and lead to the implementation and intensification of considered  
114 bioprocesses.

## 115 **2 Materials and Methods**

### 116 **2.1 Experimental set-up**

117 An experimental set-up was specifically developed which consisted of a bioreactor and  
118 a home-designed impeller system associated with several in-situ sensors (temperature,  
119 pH, rotation speed, torque, FBRM sensor). The bioreactor was a homemade glass tank  
120 (diameter: 130 mm,  $H_{\max}$ : 244 mm, V: 2.0 L) with a water jacket for thermal regulation.  
121 A specific agitator included a double impeller to minimise the difficulty in substrate  
122 mixing and ensure suspension homogeneity (**Figure 2**). The first impeller consisted of  
123 three inclined blades (diameter: 73.5 mm, angle: 45°, h = 38 mm) located 75 mm above  
124 the bottom to ensure mixing. The second mixer, which had 2 large blades (diameter:



125 120 mm, h = 22 mm), was set close to the bottom to avoid substrate decantation. The  
126 impeller shaft was connected to a viscometer working at a set speed (Viscotester Haake  
127 VT550, Thermo Fisher Scientific). This allowed for on-line torque measurements. The  
128 rotational speed ranged between 0.5 and 800 rpm, and torque ranged between 0.1 and  
129 30 mN.m (< 400 rpm) and 20 mN.m (> 400 rpm) (accuracy  $\pm 0.5\%$ ). Temperature was  
130 controlled by circulation (Haake DC30 cryostat, -50 to 200 °C  $\pm 0.01$ , Thermo  
131 Scientific) through the water jacket. Suspension pH was monitored by a pH meter  
132 (Mettler Toledo Seveneasy S20, 0-14 $\pm$ 0.01, -5 to 105 °C) and the pH adjusted with 0.5  
133 N NaOH or 0.5 N H<sub>2</sub>SO<sub>4</sub>. The viscometer and the cryostat were controlled by  
134 HaakeRheoWin Job Manager software (Thermo Fisher Scientific) which also ensured  
135 data recording (temperature, torque, mixing rate). A focused beam reflectance sensor  
136 (FBRM-G400-Mettler Toledo, range of 0.1 to 1000  $\mu\text{m}$ ) was located in the reactor in  
137 order to measure the distribution of particle chords.

## 138 **2.2 Substrates and enzyme**

139 Paper pulp from coniferous wood (Softwood, obtained via the Kraft process, with pulp  
140 extracted before bleaching, Tembec Co., Saint-Gaudens, France, type FPP27) after  
141 extrusion (Extruder Eurolab 16, 400 mm failure, extrusion line: 25 L/D 18/25  
142 conveying, 7/25 shear stress) was selected for study. The humidity of this substrate was  
143 72%. Per unit dry weight it contained 82% cellulose, 8% hemicellulose, 2% lignin, the  
144 remainder being ashes (3%) and extractive fractions (2-4%) whose composition has not  
145 been determined. The substrate density is 1034 $\pm$ 9 (kg.m<sup>-3</sup>) and the mean volume  
146 diameter of particles is 497 $\pm$ 77 ( $\mu\text{m}$ ). This pulp is favourable for enzymatic hydrolysis  
147 because of its low lignin content (2%). It also contains a low proportion of  
148 hemicellulose (xylan, mannan, etc.).

149 An enzyme cocktail (ACCELLERASE<sup>®</sup> 1500 Genencor, USA, ref. 3015155108)  
150 containing exoglucanases, endoglucanases (2200 to 2800 CMC U/g), hemicellulases

151 and  $\beta$ -glucosidases (525 to 775 pNPG U/g) was used. Its optimal temperature and pH  
152 were 50 °C (range 50 to 65 °C) and 4.8 (range 4 to 5), respectively. An ACCELLERASE®  
153 1500 dosage rate of 0.1 to 0.5 mL per gram of cellulose or roughly 0.05 to 0.25 mL per  
154 gram of biomass (depending on the biomass composition) was recommended by the  
155 suppliers. Inactivation may occur at temperatures higher than 70 °C and for pH < 4 or  
156 pH > 7. Enzyme activities were characterised in the range of 50 to 60 FPU/mL as  
157 reported [11, 12].

## 158 **2.3 Physical analysis**

159 Lignocellulose suspensions demonstrate complex rheological behaviour, and there is no  
160 standard method for studying their flow behaviour. To characterise their rheological  
161 properties as finely as possible, two measurement strategies were combined: (i) ex-situ  
162 rheometry (oscillation mode), which provided yield stress and elasticity information,  
163 and (ii) in-situ viscometry, which followed suspension viscosity in real time during  
164 enzyme attack. Particle size was analysed by in-situ chord length measurement  
165 (FBRM).

### 166 **2.3.1 Ex-situ rheometry**

167 Classic rheometry based on continuous permanent shear rate appears irrelevant for  
168 lignocellulosic substrates because of the rapid aggregation of fibres [10]. Thus,  
169 oscillatory measurements (including strain sweeps and frequency sweeps) were  
170 performed. These measurements offer several advantages: firstly, they prevent fibre  
171 aggregation caused by constant unidirectional shear flow; secondly, they provide  
172 additional information on the rheological behaviour of the suspensions (yield stress in  
173 this case). The storage modulus  $G'$  and the loss modulus  $G''$  were measured using a  
174 Mars III rheometer (Thermo Scientific). Dynamic measurements were performed with  
175 serrated plates (60 mm, roughness 400  $\mu\text{m}$ , gap size: 1.5 mm) on samples taken during  
176 hydrolysis. First, an oscillatory shear flow was set up with increasing shear stress

177 amplitude from 0.1 to 20 Pa and at a fixed frequency (1 Hz). This first measurement  
 178 was used to determine the linear domain. Then a scan was carried out in the linear  
 179 domain for frequencies from 0.5 Hz to 20 Hz and fixed shear stress amplitude. The  
 180 analysis was performed at 20 °C. Several methods can be used for yield stress  
 181 determination. It can be interpreted as the stress amplitude at which the elastic modulus  
 182  $G'$  becomes smaller than the shear modulus  $G''$ , or as the stress amplitude at which the  
 183 loss modulus  $G''$  reaches a maximum. It can also be identified as the maximum elastic  
 184 stress verifying a linear relation  $\tau = G' \cdot \gamma$  where  $\gamma$  is the strain amplitude [13, 14].  
 185 Especially for lignocellulose substrates, which cannot stand long measurement times, it  
 186 was defined as the first departure from the linear viscoelastic region [9, 15]. In the  
 187 present study, a 20% reduction in  $G'$  was chosen.

### 188 **2.3.2 In-situ rheometry**

189 Ex-situ measurement was limited by the number of samples and the substrate properties,  
 190 predominately decantation and flocculation of material. To overcome these difficulties,  
 191 in-situ viscometry was conducted throughout hydrolysis. It was based on the  
 192 determination of power consumption (or power number  
 193  $Np = \frac{P}{d^5 \cdot \rho \cdot N^3}$ ;  $P = 2\pi \cdot N \cdot C$ ) versus the Reynolds number ( $Re = \frac{\rho \cdot N \cdot d^2}{\mu}$ ) during  
 194 suspension mixing (see [16] for details). The viscosity was calculated from the power  
 195 consumption curve of the mixing system under consideration using a semi-empirical  
 196 model including laminar and transition regions for the reference curve with a one-to-one  
 197 relationship between  $Np$  and  $Re$ :

$$198 \quad Np = \left( \left( \frac{Kp}{Re} \right)^q + N_{p0}^q \right)^{1/q} \text{ with : } N_{p0} = 0.128; q = 0.782 \text{ (Eq. 1)}$$

199 The  $Kp$  constant for the mixing system was 97.9, while the Metzner-Otto constant used  
 200 to estimate an equivalent shear rate was  $Ks = 32$ . All the geometric constants were

201 determined from measurements using Newtonian (water, Marcol oil, and glycerol) and  
202 non-Newtonian fluids (xanthan-sucrose solutions). Once the experimental set-up was  
203 characterised by its power consumption curve  $N_p(\text{Re})$  and its  $K_s$  value, in-situ  
204 viscometry of the suspension was performed before the addition of enzymes and then  
205 throughout the biocatalytic reaction.

### 206 **2.3.3 In-situ particle size analysis**

207 Focus beam reflectance measurement (FBRM) enabled in-situ quantification and  
208 characterisation of chord length distribution (CLD). The FBRM sensor (FBRM G400,  
209 Mettler-Toledo, range: 0.1 to 1000  $\mu\text{m}$ ) was set up in the bioreactor to detect and  
210 monitor the changes of particle dimensions during enzymatic hydrolysis in real time.  
211 FBRM measurement is a laser-based technique. A solid-state laser light source ( $\lambda = 795$   
212 nm) provides a continuous beam of monochromatic light that is sent down the FBRM  
213 probe. A precision motor—pneumatic or electric—is used to rotate the precision optics  
214 at a constant speed. The scan speed is fixed at  $2 \text{ m}\cdot\text{s}^{-1}$ . As the scanning-focused beam  
215 sweeps across the face of the probe window, individual particles or particle structures  
216 (agglomerated or floc) backscatter the laser light towards the probe. Particles and  
217 droplets closest to the probe window are located in the scanning focused spot and  
218 backscatter distinct pulses of reflected light, which are detected by the probe and  
219 translated into chord lengths based on the simple calculation of the scan speed (velocity)  
220 multiplied by the pulse width (time). A chord length is simply defined as the straight-  
221 line distance from one edge of a particle or particle structure to another edge. Typically,  
222 thousands of individual chord lengths are measured each second to produce the chord  
223 length distribution, which is the fundamental measurement provided by FBRM.

## 224 **2.4 Biochemical analysis**

### 225 **2.4.1 Dry matter content**

226 The water content of substrates and hydrolysed suspensions was determined by drying  
227 at low temperature and pressure. Volumes of samples ( $\approx 1$  mL) were put in Eppendorf  
228 tubes (known mass,  $m_{\text{epp}}$ ). These Eppendorfs ( $m_{\text{ini}}$ ) were then placed in an oven at 60  
229 °C, 200 mbar for 5 days, and afterwards, weighed ( $m_{\text{fin}}$ ). Water content (W) and dry  
230 matter content (DM) were calculated using **Eq. 2** (accuracy  $\pm 0.5\%$ ):

$$231 \quad W(\%) = \frac{m_{\text{ini}} - m_{\text{fin}}}{m_{\text{ini}} - m_{\text{epp}}} \cdot 100; DM(\%) = 100 - W \text{ (Eq. 2)}$$

### 232 **2.4.2 Monomers and oligomers**

233 Samples were centrifuged at 13000 rpm for 5 min. Glucose and soluble cello-  
234 oligosaccharides, (i.e. with a degree of polymerization lower than 6) in the supernatants  
235 were quantified by an AMINEX HPX-87P carbohydrate analysis column (Bio-Rad  
236 Laboratories, Richmond, CA) using a high-performance liquid chromatography system  
237 (separations module: Waters Alliance 2690; refractometer detector: Waters 2414,  
238 Milford, MA). Conditions were optimised to detect and quantify the different cello-  
239 saccharides. Analysis was performed at 60 °C with deionised water and 0.1M  $\text{Pb}(\text{NO}_2)_2$   
240 as mobile-phase (ratio 80/20 v/v) at a flow rate of 0.5 mL/min for 30 min. Glucose and  
241 cello-oligosaccharide standards with a DP ranging from 2 to 5 were used (Sigma  
242 Chemical Co., St. Louis, MO).

## 243 **2.5 Experimental strategy**

244 Enzyme hydrolysis was carried out at 40 °C due to enzyme heat stability (activity  
245 reduction at high temperature), energy conservation, and taking into account the optimal  
246 conditions for the cell cultivation step. The pH of the medium was adjusted to 4.8. To  
247 prevent microbial contamination, 0.2 mL of a solution of chloramphenicol ( $5 \text{ g.L}^{-1}$ ) was

248 added. Enzymes were added when the suspension reached homogeneity (visual  
 249 monitoring and stabilised torque, reached within 30 min). Hydrolysis was carried out  
 250 over 24 h with a mixing rate of 100 rpm (corresponding approximately to a shear rate of  
 251  $50 \text{ s}^{-1}$ ), reactor volume 1300 mL and using three substrate concentrations, 1% (10.1 and  
 252 9.7 gdm/L formally), 2% (only for in-situ viscometry analysis, 19.6 and 20.0 gdm/L  
 253 formally) and 3% w/v (29.2 and 28.7 gdm/L formally), two enzyme/substrate ratios, 0.1  
 254 and 0.5 mL enzyme/g cellulose respectively (corresponding to 5 and 25 FPU/g  
 255 cellulose). Suspensions were sampled with a 7 mm diameter flexible tube connected to  
 256 a 50 mL syringe. Each sample was about 15 mL at 0 h, 15min, 1h, 2h, 3h, 5 h, 7h, 10 h  
 257 and 24 h of hydrolysis time. The enzyme reaction was stopped by adding 0.1 mL 10 N  
 258 NaOH. The total volume of samples removed was approximately 130 mL (< 10% of  
 259 initial volume). Samples were analysed in terms of rheological, granulo-metric and  
 260 biochemical properties during enzyme degradation.

### 261 **3 Results and Discussions**

#### 262 **3.1 Bioconversion rate**

263 The production of monomers (C6, C5) and water-soluble oligosaccharides is crucial for  
 264 the development of new intensified bioprocesses. First of all, the bioconversion rate was  
 265 calculated from the glucose produced using **Eq. 3**.

$$266 \quad \text{Bioconversion} (\%) = \frac{[Glc]_{\text{measured}}}{[Glc]_{\text{theory}}} \quad (\text{Eq. 3})$$

$$[Glc]_{\text{theory}} = \left( m_s \cdot \frac{DM}{100} \cdot \frac{Cellu}{100 \cdot 0.9} \right) / \left( V_w + \frac{m_s}{\rho_s} \right)$$

267 The theoretical conversion coefficient from cellulose to glucose is 1/0.9.

268 Under enzyme activity, the polymeric cellulose chain is broken down to produce  
 269 monomers. Oligomers (DP > 2) were not detected for any of the experiments. Xylose  
 270 was quantifiable only in the case of 3% w/v - 0.5mL enzyme/g cellulose.

271 As expected, the concentration of monomers (glucose, xylose) increases with hydrolysis  
272 time. For example, in the 3% w/v cases, the monomer concentration increased regularly  
273 during hydrolysis, so that enzyme loading had an impact on the quantity of the glucose  
274 released: 5.1 to 19.1 g.L<sup>-1</sup>, which corresponds to 19 and 73% bioconversion for 0.1 and  
275 0.5 mL/g cellulose respectively (**Table 1**). Few articles have examined the evolution of  
276 water-soluble cello-oligosaccharides. Sun and Cheng [17] analysed the hydrolysis of  
277 microcrystalline cellulose (10 g.L<sup>-1</sup>) by cellulase produced by *Cellulomonas fimi*. They  
278 did not detect soluble oligomers with DP<sub>≥</sub>4, but the cellotriose concentration varied  
279 between 0.2 and 0.6 g.L<sup>-1</sup> depending on the enzyme used—endo-glucanase or  
280 cellobiohydrolase. Solubilisation reached 61% and 50% respectively for each enzyme  
281 separately. In contrast, the intermediates of cellulose hydrolysis were not found because  
282 the Accellerase 1500 cocktail contains all the types of activity required to rapidly  
283 degrade these intermediates during hydrolysis.

284 The bioconversion yields are comparable to data reported in the literature. For a 2%  
285 (w/w) suspension of oven-dried corncob, after 24 h hydrolysis the glucose conversion  
286 varied between 30 and 82% for 6 and 30 FPU/g [18]. With the hydrolysis of a pre-  
287 treated wheat straw suspension at 1% (w/v), a doubled bioconversion (from 45.9 to  
288 87%) after 18 h with enzyme loading between 9.6 and 57.6 FPU/g was reported [19].

289 Changes in dry matter content were monitored by determining the water content in the  
290 sample at different hydrolysis times. This technique involves the error inherent to  
291 sampling heterogeneous suspensions. Beside this technique, the dry matter content was  
292 also determined from dissolved and undissolved substrate measurements. According to  
293 the hypothesis of the conservation of substrate mass before and after hydrolysis, the dry  
294 matter in suspension can be deduced from the initial quantity and the hydrolysed  
295 quantity (which released soluble components—monomers and cello-oligosaccharides).

296 As previously reported, the water-soluble cello-oligosaccharides with DP > 2 were not

297 quantifiable, so only glucose, xylose, and cellobiose were used to calculate the quantity  
298 of hydrolysed substrate. Mass balance was conserved around 95% (**Table 1**).

## 299 **3.2 Rheological behaviour during hydrolysis**

### 300 **3.2.1 Viscosity and yield stress**

301 In-situ viscometry showed that substrate suspensions were non-Newtonian shear-  
302 thinning fluids. The viscosity of suspensions at 100 rpm (corresponding to an equivalent  
303 shear rate of  $50 \text{ s}^{-1}$ ) as a function of hydrolysis progress is illustrated in **Figure 3-A**. As  
304 expected, slurry viscosity decreased during hydrolysis. Excluding the very beginning of  
305 the reaction ( $t < 0.2 \text{ h}$ ), viscosity changed drastically at the beginning of hydrolysis.  
306 This sharp reduction in viscosity was observed whatever the concentration or  
307 enzyme/substrate ratio. For example, at 3% dm (w/v), 0.5 mL E/g cellulose, the  
308 viscosity decreased from 68 to 3 mPa.s. This change in the physical appearance of the  
309 slurry is associated with the biochemical changes and particle size changes occurring in  
310 the fibres [20, 21]. The drop in viscosity is a combination of the decrease of solid  
311 concentration (solubilisation) and of the fragmentation of cellulose fibres [22, 23].  
312 Under the effects of enzymes, the cellulose chains are broken up to produce smaller  
313 particles and non-dissolved cellulose is converted into soluble compounds such as  
314 monomers and water-soluble oligomers. In addition, the viscosity drop is suggested to  
315 be strongly connected with the degradation and decrease in water binding capacity of  
316 the lignocellulose matrices during enzyme-based hydrolysis [24].  
317 The initial viscosities of the 3% (w/v) suspensions were 14-fold higher than those of the  
318 1% (w/v) suspension ( $\sim 70$  and  $\sim 5$  mPa.s, respectively). The nonlinear dependence of  
319 viscosity on the concentration is expected and can be explained by an increase in  
320 particle interactions, less free water, and hydrogen bonding between cellulose chains.  
321 As the solid concentration increases, the average distance between particles in the slurry  
322 decreases, leading to enhanced contact between particles, especially if there is an



323 entanglement of amorphous fibres between particles [25]. The high increase in initial  
324 viscosity of between 1% (w/v) and 3% (w/v) signifies that the increase of substrate  
325 concentration becomes the major stress for enzymatic hydrolysis at high (and very high)  
326 dry matter content levels. That is why a special strategy to reach the condition of high  
327 substrate concentration is absolutely necessary. During hydrolysis, a significant drop in  
328 slurry viscosity is observed within the first 10 h of the hydrolysis reaction. These results  
329 are supported by the literature for a wide range of matrices, particle sizes, and  
330 enzyme/cellulose ratios [26-28], although a large heterogeneity of viscosity data is  
331 reported. This can be explained by different substrates, pre-treatments, morpho-  
332 granulometry and enzymes, and operating conditions.

333 The effect of enzyme concentration on viscosity is clearly demonstrated even for the  
334 lowest concentration in **Figure 3-A**. For the same substrate concentration, the higher the  
335 enzyme/g cellulose ratio (E/C), the faster the viscosities decrease. Within 5 h, the  
336 viscosity of a 3% (w/v) suspension was reduced by 85% for 0.5 mL E/g cellulose;  
337 whereas its reduction was limited to 30% for 0.1 mL E/g cellulose. This observation is  
338 in agreement with other authors [22, 26, 29]. Furthermore, two tendencies in viscosity  
339 evolution were found for suspensions containing 0.1 mL/g cellulose. Firstly, the  
340 suspension viscosity increased during the first hour consecutive to swelling and  
341 unbinding effects (clearly observed for 2% w/v and 3% w/v, not significant for 1%  
342 w/v); after that, the viscosity decreased because of depolymerisation by the enzymes.

343 Shear stress sweep and frequency sweep experiments were performed in oscillation  
344 mode (see §2.3.1). The frequency sweep revealed that the elastic modulus  $G'$  and the  
345 viscous modulus  $G''$  do not depend on frequency (**Figure 3-B**). In addition,  $G'$  is  
346 always greater than  $G''$ , with a ratio  $G'/G''$  which can be identified in Figure 3-B. This  
347 is characteristic of viscoplastic behaviour, and shear stress sweeps were then used to  
348 evaluate yield stress. All suspensions showed viscoplastic shear-thinning behaviour  
349 both initially and throughout hydrolysis. Shear-thinning has already been mentioned by

350 other authors [9, 27, 29]. Before the introduction of enzyme, the rheological behaviour  
351 for small deformations is mainly elastic, with a high value of  $G'$  and a ratio  $G'/G''$  close  
352 to 5. For a 3% (w/v) suspension at  $t = 0$  h and in the linear domain,  $G'$  and  $G''$  reached  
353 about 1000 and 200 Pa. This is supported by the literature for different matrices. For  
354 corn stover, at 12% dm,  $G'$  and  $G''$  were reported as 2000 and 300 Pa [15]; for acid pre-  
355 treated softwood, these values were 100 and 50 Pa respectively [9]. When increasing the  
356 shear stress, all the samples exhibited two zones: a first where  $G'$  and  $G''$  did not depend  
357 on shear stress (linear domain) and a second where both moduli decreased. In this  
358 second zone,  $G'$  deviated sharply from a stable curve and crossed (or tended to cross)  
359  $G''$ . During hydrolysis, a regular decrease in both moduli was observed. For example, in  
360 the 3%-0.5 case, a 1000-fold reduction was observed for both  $G'$  and  $G''$  after 24 h of  
361 enzyme attack. Moreover, the elastic character is noticeably preserved. One also notes  
362 that the higher the enzyme activity, the faster the decrease in elasticity (in agreement  
363 with the viscosity results above). This discussion highlights, yet again, the strong impact  
364 of enzyme concentration on how rheological behaviour varies during hydrolysis.

365 As viscometric yield stress measurements were not possible for these suspensions, this  
366 yield stress  $\tau_0$  was deduced from the elastic modulus ( $G'$ ) (see **Figure 3-B**). Yield stress  
367 can be regarded as the stress required to initiate flow. Yield stress values between 1 and  
368 20 Pa were determined at 1% and 3% dm (w/v) before hydrolysis. With an increase in  
369 substrate concentration from 1% to 3% dm (w/v), yield stress increased 20-fold (**Figure**  
370 **4**). This confirmed the critical substrate concentration ( $\approx 3\%$  dm (w/v) determined by in-  
371 situ measurements) beyond which the viscosity increased exponentially [16]. The  
372 results presented were slightly higher than those obtained from pre-treated softwood: 0-  
373 28 Pa for 4-12% substrate concentration [9] and from pre-treated corn stover: 0.26-22.9  
374 Pa for 5-17% dm [30]. This difference could be due to the nature and the physical  
375 characteristics of the matrices and to the method used to determine yield stress.

376 During hydrolysis, yield stress decreased sharply as enzyme attack progressed (**Figure.**  
377 **4**). This can be explained by the drop in solid content in suspension and by a reduction  
378 in fibre-fibre interactions due to hydrolysis progress. The decrease of yield stress was  
379 greater for experiments with higher enzyme concentrations. The final yield stress (at 24  
380 h hydrolysis) of 3%-0.1 was 10 times higher than that of 3%-0.5. This observation is  
381 directly correlated to the impact of enzyme activities on fibre degradation. In addition,  
382 the yield stress of the hydrolysed samples (3%-0.5 at 24 h, DM = 1.14%) exhibited  
383 lower values for similar DM content compared to the original material (yield stress at 0  
384 h of 1% - 0.5, DM = 0.97%). This can be explained by the modification of fibre  
385 structure, diameter, and shape. Decreasing yield stress during enzyme hydrolysis was  
386 previously reported for corn stover [21] and pre-treated softwood [9]. These two studies  
387 also reported a lower yield stress value for hydrolysed slurries when compared to non-  
388 hydrolysed slurries at the same solid content.

389 **Figure 4** shows the dependence of yield stress during hydrolysis presented in terms of  
390 dry matter content. Taking the criterion that the fluid behaves as a pourable liquid at a  
391 yield stress below 1 Pa (the values of 1% are negligible), interestingly, results for 3%  
392 show that the two yield stress curves collapse onto a single curve when plotted against  
393 dry matter content. It decreased exponentially with the decrease in dry matter content  
394 during hydrolysis. Yield stress becomes negligible when solid matter < 2%. This has  
395 real significance for the choice of substrate flow rate for the cumulative feed strategy or  
396 if the slurry has to be pumped into another bioreactor.

### 397 **3.2.2 Uniqueness of viscosity-time curve**

398 In order to explore the change in viscosity during enzyme hydrolysis and to compare its  
399 kinetics between different experimental conditions, a critical time is estimated through a  
400 normalised viscosity defined as follows:

401 
$$\mu^* = \frac{\mu_t - \mu_{fin}}{\mu_0 - \mu_{fin}} \text{ (Eq. 4)}$$

402  $\mu^*$  is a dimensionless viscosity; and  $\mu_0$ ,  $\mu_t$ ,  $\mu_{fin}$  are the viscosities at  $t = 0$  h,  $t_i$  and the  
 403 final viscosity (the viscosity of a suspension containing the whole dry matter as soluble  
 404 fractions – hypothesizing total conversion).

405  $\mu^*$  describes the reduction of viscosity during enzyme hydrolysis. It starts from 1 ( $t = 0$   
 406 h) and tends to 0 ( $t = 24$  h). From this quantity, a critical time noted  $t(\mu^* = x)$  can be  
 407 defined as the time at which the viscosity was  $x\%$  of the initial suspension viscosity.  
 408 The relationship between these critical times  $t(\mu^* = x)$  and dimensionless viscosity,  $\mu^*$   
 409 only depends on the enzyme concentration used regardless of the substrate  
 410 concentration. These critical times decreased exponentially with the dimensionless  
 411 viscosity. The ratio of these critical times between both conditions was around 4-fold.  
 412 Thus the hydrolysis time,  $t$ , can be normalised using a selected critical time  $t(\mu^* = x_0)$   
 413 **(Eq. 5):**

414 
$$t^* = \frac{t}{t(\mu^* = x_0)} \text{ (Eq. 5)}$$

415 The dimensionless time-viscosity curves are plotted in **Figure 5** ( $\mu^* = 0.25$ ).  
 416 Interestingly, a single curve is exhibited for  $t^* > 0.05$ , i.e. when the possible swelling  
 417 and unbinding effects can be neglected. This uniqueness of dimensionless time-  
 418 viscosity curves was observed whatever the hydrolysis conditions in the studied cases  
 419 (substrate concentrations, enzyme ratio). These results suggest that during the period  
 420 and for a 75% reduction of initial viscosity, a similar degradation mechanism could be  
 421 assumed in semidilute regime ( $\leq 3\%$  dm w/v), although rheological behaviour strongly  
 422 differs between 1% and 3% w/v suspensions.  
 423 The uniqueness of viscosity-time curves at macroscopic observation suggests a similar  
 424 mechanism of fibre degradation. Considering these in-situ and ex-situ rheological

425 results, macro scales will be explored with chord length distribution analysis, and then  
426 kinetics will be modelled.

### 427 **3.3 Particle size evolution**

428 The time evolution of the mean chord length and chord number for the four different  
429 conditions is shown in **Figure 6**. Considering chord length distribution (CLD) for the  
430 initial suspension, we define four classes (I:  $< 60 \mu\text{m}$ , II: 60 to 90, III: 90 to 160  $\mu\text{m}$ , IV:  
431 160 to 600  $\mu\text{m}$ ) which correspond to 25% of this population. **Figure 7** illustrates the  
432 evolution of each class during hydrolysis for the case 1% w/w - 0.5 mL enzymes/g  
433 cellulose.

434 A sharp decrease in mean chord length from 120 to 60  $\mu\text{m}$  (**Figure 6**) is observed:  
435 within two hours for  $E/S = 0.5 \text{ mL/g}$  and six hours for 0.1 mL/g. Considering chord  
436 number, initial values are almost proportional to concentration and during hydrolysis,  
437 two successive trends are observed. During the first step, the total chord number  
438 increased for all experiments, although the durations of this step differed strongly:  
439 around 8 to 10 hours for  $E/S = 0.1 \text{ mL/g}$ , and 1 to 3 hours for  $E/S = 0.5 \text{ mL/g}$ .  
440 Meantime, the relative increase of chord number is lower for the largest  $E/S$  ratio.  
441 Considering now **Figure 7**, coarse population (class IV) decreased regularly while the  
442 finest population (Class I) increased.

443 The initial augmentation of chord number can be explained by the fragmentation of  
444 cellulose fibres. Coarse particles are attacked and divided into several fine particles.  
445 Beyond that, enzyme activity occurs and the finest particles are converted into dissolved  
446 compounds, generating a reduction in chord number. As demonstrated in **Figure 6**, the  
447 fragmentation mechanisms appear to be the dominant effect in the strong viscosity  
448 reduction observed during the first 5 h, while solubilisation increased to 31%. In  
449 contrast, after 5 h, fragmentation is negligible (class IV almost constant) while

450 solubilisation increased to 76% (at 24 h). Correlatively, the viscosity evolution was  
451 limited and the finest populations still increased from 38% to 50%.

### 452 **3.4 Modelling kinetics at macro-, micro-, and biochemical scales**

453 Hydrolysis induces a reduction in particle size, dry matter content, and viscosity, which  
454 promotes mixing and fibre accessibility. The kinetics of viscosity (macro), chord length  
455 (micro) and substrate and product concentrations (biochemical) stand as key indicators  
456 for understanding and controlling bioprocess performance. The kinetics of biochemical  
457 enzymatic reactions have been extensively reported in the literature [31], contrary to  
458 physical parameters such as viscosity and granulometry. A lot of enzyme reactions (e.g.  
459 hydrolysis, oxidation, and reduction or cofactor mechanisms) are second- or higher  
460 order reversible reactions. Many are limited by diffusion and physical accessibility.  
461 Adsorption (reaction) in high molecular weight structures (for example, proteins,  
462 polynucleotides, polysaccharides, or heterogeneous protein-phospholipid, protein-  
463 nucleotide, and protein-polysaccharide structures) is more complex. Among the  
464 numerous models reported in the literature, one of the best known is the Michealis-  
465 Menten equation [32] but a first-order model was also reported by Chrastil [33].

466 The time evolution of viscosity  $\mu$ , mean chord length  $lc_m$ , and residual substrate  
467 concentration in dry matter  $S$  (%) are modelled by the general equation **Eq. 6**:

$$468 \quad -\frac{dX}{dt} = k \cdot X^\alpha \text{ (Eq. 6)}$$

469 where  $X$  is the variable being modelled ( $\mu$ ,  $lc_m$  and  $S$ );  $k$  is the kinetic constant, and  $\alpha$   
470 the model/reaction order ( $l$ ). As boundary conditions,  $X$  varies from  $X_0$  to  $X_\infty$ ,  
471 corresponding to initial and final values respectively.

472 A second-order model accurately describes the time dependence of the physico-  
473 chemical parameters—viscosity, chord length, and dry matter content. For the lowest  
474 enzyme ratio, an increase of the mean chord length was observed during the first hours.

475 In these cases, the proposed model was restricted to the time interval corresponding to  
 476 the maximal value of  $lc_m$  until the end of hydrolysis (experiments with 0.1 mL  
 477 enzyme/g cellulose). This  $lc_m$  increase step could be explained by fibre swelling and  
 478 unwinding due to limited enzyme activity. To sum up, the most suitable kinetic models  
 479 are second order and written as **Eq. 7 to 9**:

$$480 \quad \mu = \frac{\mu_0 - \mu_\infty}{(\mu_0 - \mu_\infty) \cdot k_\mu \cdot t + 1} + \mu_\infty \text{ (Eq. 7)}$$

$$481 \quad lc_m = \frac{lc_{m0} - lc_{m\infty}}{(lc_{m0} - lc_{m\infty}) \cdot k_{lc} \cdot t + 1} + lc_{m\infty} \text{ (Eq. 8)}$$

$$482 \quad S = \frac{S_0 - S_\infty}{(S_0 - S_\infty) \cdot k_S \cdot t + 1} + S_\infty \text{ (Eq. 9)}$$

483 The model was adjusted via the least squares method for each of the physico-  
 484 biochemical parameters. The coefficients and the correlation coefficients are presented  
 485 in **Table 2**. The correlation coefficients indicated the fine agreement between the  
 486 models and experimental data. The enzyme cocktail (Accellerase 1500) contained only  
 487 cellulases and hemicellulases, so the lignin fraction can be considered as contributing to  
 488 the non-hydrolysable fraction  $S_\infty$ . For our substrate, lignin represents less than 2%, and  
 489 thus can be neglected. The final viscosity  $\mu_\infty$  corresponds to the viscosity of a  
 490 suspension containing all soluble fractions (total conversion). In the present case, this  
 491 value would be close to the supernatant viscosity (0.7 mPa.s) and  $\mu_\infty$  is negligible  
 492 compared to the initial suspension viscosity. In ideal conditions, the solid fractions  
 493 should be converted into soluble fractions and the ultimate chord length  $lc_{m\infty}$  would then  
 494 be null. However, experimentally, there is always some part that remains non-  
 495 hydrolysable substrate, so  $lc_{m\infty}$  cannot be neglected compared to  $lc_{m0}$ , and is assumed  
 496 equal to  $lc_{m24h}$ .

497 Effects of enzyme and substrate concentrations are clearly observed for all constants.  
 498 With  $k_S$ , the increase of the E/S ratio from 0.1 to 0.5 mL/g cellulose led to a 7- to 9-fold

499 increase. When the initial substrate concentration increased from 1 to 3% w/v, a  
500 reduction of 2-3 times was observed. The same tendency was found with  $k_{\mu}$ , which  
501 demonstrated the strong impact of enzyme and substrate concentrations on the variation  
502 of suspension viscosity during hydrolysis. Similar impacts were observed for enzyme  
503 concentration and substrate concentration: 8-fold and 4-fold respectively. The final  
504 value of the mean chord length,  $lc_{m\infty}$ , seemed identical in magnitude; however, the  
505 impact of enzyme ratio can be clearly distinguished by the absolute values of  $k_{lc}$ . An  
506 increase of 3.5-5.5 -fold of  $k_{lc}$  was observed when the enzyme concentration passed  
507 from 0.1 to 0.5 mL/g cellulose.

508 Using equations 7 to 9, **Figure 8** illustrates the successive phenomenological limitations  
509 observed during hydrolysis. Rheological behaviour appears to be the primary limiting  
510 factor. Transfer limitation due to high viscosity and yield stress is, however, temporary  
511 (during the first 5 h) but constitutes an important phenomenon. Viscosity collapse and  
512 particle fragmentation are concomitant. Beyond this, the limited reduction in size  
513 indicates a threshold for biocatalytic reaction. This could be due to recalcitrant fractions  
514 or inefficient enzyme activities. It is therefore necessary to study the physical and  
515 biochemical structures of this recalcitrant fraction in order to identify how it could be  
516 degraded.

517 The major challenge in 2<sup>nd</sup>-generation biofuels is to reduce costs so as to compete with  
518 1<sup>st</sup>-generation ones. Therefore, processing at high solid content is mandatory. However,  
519 the rheological behaviour of the hydrolysis suspension stands out as the first and major  
520 determinant of process efficiency, and has to be considered a key criterion in proposing  
521 a rational strategy for reaching high dry matter content.

522 To increase transfers and bioreaction efficiency, a strategy can be built from the  
523 identified parameters in a semidilute regime: the critical substrate concentration  $C^*$   
524 (meaning a drastic increase of initial suspension viscosity beyond  $C^*$ ) and a targeted



525 hydrolysis time,  $t(\mu^* = x_0)$  (corresponding to a chosen relative reduction of initial  
526 viscosity), can be used to define a reference feed rate  $Q_c$  (**Eq. 10**):

527 
$$Q_c = \frac{c^* \cdot V}{t(\mu^* = x_0)} \text{ (g} \cdot \text{h}^{-1}\text{) (Eq. 10)}$$

528 A cumulative feeding strategy can be defined, based on  $Q_c$ , which would allow working  
529 in a favourable regime so as to reach high hydrolysis yield by avoiding instantaneous  
530 high substrate concentrations.

531

#### 532 **4 Conclusion**

533 This study aimed to delve deeply into the biophysical and transfer limitations occurring  
534 during enzymatic hydrolysis in the cellulosic biofuels context. Considering paper pulp  
535 substrates, semidilute conditions were chosen, as they introduce the complexity of  
536 particle-particle interactions, which are strongly involved in transfer limitations, without  
537 inhibiting the bioreaction. Biochemical and physical phenomena were explored through  
538 a specific experimental set-up associating in-situ and ex-situ analyses.

539 Non-Newtonian behaviour associated with non-negligible yield stress stand as the major  
540 factors limiting process efficiency and progress towards high solid loading. The  
541 uniqueness of dimensionless time-viscosity curves was observed whatever the operating  
542 conditions, and suggests a similar mechanism for fibre degradation. The evolution of  
543 biophysical parameters during hydrolysis was observed as the results of a combination  
544 of fibre fragmentation, which dominated during the first step, and solid solubilisation.  
545 Kinetics modelling enabled demonstration of the successive phenomenological  
546 limitations and their magnitude. All these elements lead to the proposal of a reference  
547 feed rate which would be used in a cumulative feeding strategy for reaching high solid  
548 loading and balancing energy consumption and process efficiency.

#### 549 **5 Acknowledgements**

550 This work was realised within the ProBio3 project ANR-11-BTBR-0003, selected in the  
551 Investissements d'Avenir Programme, with financial support from the French  
552 government, managed by the National Research Agency. The authors are grateful to the  
553 "Programme de Bourses d'Excellence 2011" of the French Embassy in Vietnam.

554

- 556 [1] K.-H. Chang, K.-R. Lou, C.-H. Ko, Potential of bioenergy production from biomass  
557 wastes of rice paddies and forest sectors in Taiwan, *Journal of Cleaner Production* 206  
558 (2019) 460-476.
- 559 [2] W.R. Gibbons, S.R. Hughes, Integrated biorefineries with engineered microbes and  
560 high-value co-products for profitable biofuels production, *In Vitro Cellular &*  
561 *Developmental Biology - Plant* 45(3) (2009) 218-228.
- 562 [3] P. Vallette, C. De Choudens, *Le bois, la p tes, le papier*, Centre Technique de  
563 l'Industrie des Papiers, Cartons et Celluloses, 1987.
- 564 [4] M. Moshkelani, M. Marinova, M. Perrier, J. Paris, The forest biorefinery and its  
565 implementation in the pulp and paper industry: Energy overview, *Applied Thermal*  
566 *Engineering* 50(2) (2013) 1427-1436.
- 567 [5] Z.L. Fan, C. South, K. Lyford, J. Munsie, P. van Walsum, L.R. Lynd, Conversion of  
568 paper sludge to ethanol in a semicontinuous solids-fed reactor, *Bioprocess. Biosyst.*  
569 *Eng.* 26(2) (2003) 93-101.
- 570 [6] A. Herrera, S.J. T llez-Luis, J.A. Ram rez, M. V zquez, Production of Xylose from  
571 Sorghum Straw Using Hydrochloric Acid, *Journal of Cereal Science* 37(3) (2003) 267-  
572 274.
- 573 [7] X. Zhao, K. Cheng, D. Liu, Organosolv pretreatment of lignocellulosic biomass for  
574 enzymatic hydrolysis, *Appl Microbiol Biotechnol* 82(5) (2009) 815-827.
- 575 [8] T. Eggeman, R.T. Elander, Process and economic analysis of pretreatment  
576 technologies, *Bioresource Technology* 96(18) (2005) 2019-2025.
- 577 [9] M. Wiman, B. Palmqvist, E. Tornberg, G. Liden, Rheological characterization of  
578 dilute acid pretreated softwood, *Biotechnol Bioeng* 108 (2011) 1031 - 1041.
- 579 [10] T.C. Nguyen, D. Anne-Archard, L. Fillaudeau, Rheology of Lignocellulose  
580 Suspensions and Impact of Hydrolysis: A Review, in: R. Krull, T. Bley (Eds.),  
581 *Filaments in Bioprocesses*, Springer International Publishing 2015, pp. 325-357.
- 582 [11] P. Alvira, M.J. Negro, M. Ballesteros, Effect of endoxylanase and alpha-L-  
583 arabinofuranosidase supplementation on the enzymatic hydrolysis of steam exploded  
584 wheat straw, *Bioresource Technology* 102(6) (2011) 4552-4558.
- 585 [12] S.P. Govumoni, S. Koti, S.Y. Kothagouni, V. S, V.R. Linga, Evaluation of  
586 pretreatment methods for enzymatic saccharification of wheat straw for bioethanol  
587 production, *Carbohydrate Polymers* 91(2) (2013) 646-650.
- 588 [13] R. Damani, R.L. Powell, N. Hagen, Viscoelastic characterization of medium  
589 consistency pulp suspensions, *Canadian Journal of Chemical Engineering* 71(5) (1993)  
590 676-684.
- 591 [14] H.J. Walls, S.B. Caines, A.M. Sanchez, S.A. Khan, Yield stress and wall slip  
592 phenomena in colloidal silica gels, *Journal of Rheology* (1978-present) 47(4) (2003)  
593 847-868.
- 594 [15] J.J. Stickel, J.S. Knutsen, M.W. Liberatore, W. Luu, D.W. Bousfield, D.J.  
595 Klingenberg, C.T. Scott, T.W. Root, M.R. Ehrhardt, T.O. Monz, Rheology  
596 measurements of a biomass slurry: an inter-laboratory study, *Rheologica Acta* 48(9)  
597 (2009) 1005-1015.
- 598 [16] T.C. Nguyen, D. Anne-Archard, V. Coma, X. Cameleyre, E. Lombard, C. Binet, A.  
599 Nouhen, K.A. To, L. Fillaudeau, In situ rheometry of concentrated cellulose fibre  
600 suspensions and relationships with enzymatic hydrolysis, *Bioresource Technology* 133  
601 (2013) 563-572.
- 602 [17] Y. Sun, J. Cheng, Hydrolysis of lignocellulosic materials for ethanol production: a  
603 review, *Bioresource Technology* 83(1) (2002) 1-11.
- 604 [18] F. Carvalheiro, L.C. Duarte, F.M. G rio, Hemicellulose biorefineries: a review on  
605 biomass pretreatments, *Journal of Scientific & Industrial Research* 67 (2008) 849-864.

606 [19] G. Pierre, Z. Maache-Rezzoug, F. Sannier, S.A. Rezzoug, T. Maugard, High-  
607 performance hydrolysis of wheat straw using cellulase and thermomechanical  
608 pretreatment, *Process Biochemistry* 46(11) (2011) 2194-2200.

609 [20] W.E. Kaar, M.T. Holtzapple, Benefits from Tween during enzymic hydrolysis of  
610 corn stover, *Biotechnology and Bioengineering* 59(4) (1998) 419-427.

611 [21] C. Roche, C. Dibble, J. Knutsen, J. Stickel, M. Liberatore, Particle concentration  
612 and yield stress of biomass slurries during enzymatic hydrolysis at high-solids loadings,  
613 *Biotechnol Bioeng* 104 (2009) 290 - 300.

614 [22] L. Rosgaard, P. Andric, K. Dam-Johansen, S. Pedersen, A.S. Meyer, Effects of  
615 substrate loading on enzymatic hydrolysis and viscosity of pretreated barley straw,  
616 *Applied Biochemistry and Biotechnology* 143(1) (2007) 27-40.

617 [23] M. Vázquez, M. Oliva, S.J. Téllez-Luis, J.A. Ramírez, Hydrolysis of sorghum  
618 straw using phosphoric acid: Evaluation of furfural production, *Bioresource Technology*  
619 98(16) (2007) 3053-3060.

620 [24] M. Chang, T.C. Chou, G. Tsao, Structure, pretreatment and hydrolysis of cellulose,  
621 *Bioenergy*, Springer Berlin Heidelberg 1981, pp. 15-42.

622 [25] R. Dasari, K. Dunaway, R. Berson, A scraped surface bioreactor for enzymatic  
623 saccharification of pretreated corn stover slurries, *Energy Fuel* 23 (2009) 492 - 497.

624 [26] C.C. Geddes, J.J. Peterson, M.T. Mullinnix, S.A. Svoronos, K.T. Shanmugam,  
625 L.O. Ingram, Optimizing cellulase usage for improved mixing and rheological  
626 properties of acid-pretreated sugarcane bagasse, *Bioresource Technology* 101(23)  
627 (2010) 9128-9136.

628 [27] L.T.C. Pereira, L.T.C. Pereira, R.S.S. Teixeira, E.P.D. Bon, S.P. Freitas, Sugarcane  
629 bagasse enzymatic hydrolysis: rheological data as criteria for impeller selection, *Journal*  
630 *Of Industrial Microbiology & Biotechnology* 38(8) (2011) 901-907.

631 [28] B.-H. Um, Optimization of Ethanol Production from Concentrated Substrate,  
632 Auburn University, 2007, p. 268.

633 [29] J. Du, F.Z. Zhang, Y.Y. Li, H.M. Zhang, J.R. Liang, H.B. Zheng, H. Huang,  
634 Enzymatic liquefaction and saccharification of pretreated corn stover at high-solids  
635 concentrations in a horizontal rotating bioreactor, *Bioprocess. Biosyst. Eng.* 37(2)  
636 (2014) 173-181.

637 [30] N.V. Pimenova, A.R. Hanley, Effect of corn stover concentration on rheological  
638 characteristics, *Applied Biochemistry and Biotechnology* 113 (2004) 347-360.

639 [31] B. Frémaux, *Eléments de cinétique et de catalyse*, Lavoisier, Paris, 1989.

640 [32] L. Michaelis, M.L. Menten, Die Kinetik der Invertinwirkung, *Biochem* 49 (1913)  
641 333-369.

642 [33] J. Chrastil, Enzymatic product formation curves with the normal or diffusion  
643 limited reaction mechanism and in the presence of substrate receptors, *Int. J. Biochem.*  
644 20(7) (1988) 683-693.

645

646

*Figure 1: Experimental methodology and strategy*

*Figure 2: Configuration of bioreactor and impellers (dimensions in mm)*

*Figure 3: Overview of in-situ and ex-situ rheometry. A: In-situ viscometry as a function of hydrolysis time. B: Viscous,  $G''$  (open symbols), and elastic,  $G'$  (filled symbols), as a function of shear stress (3%-0.5 signifies 3% dm and 0.5 mL enzyme/g cellulose).*

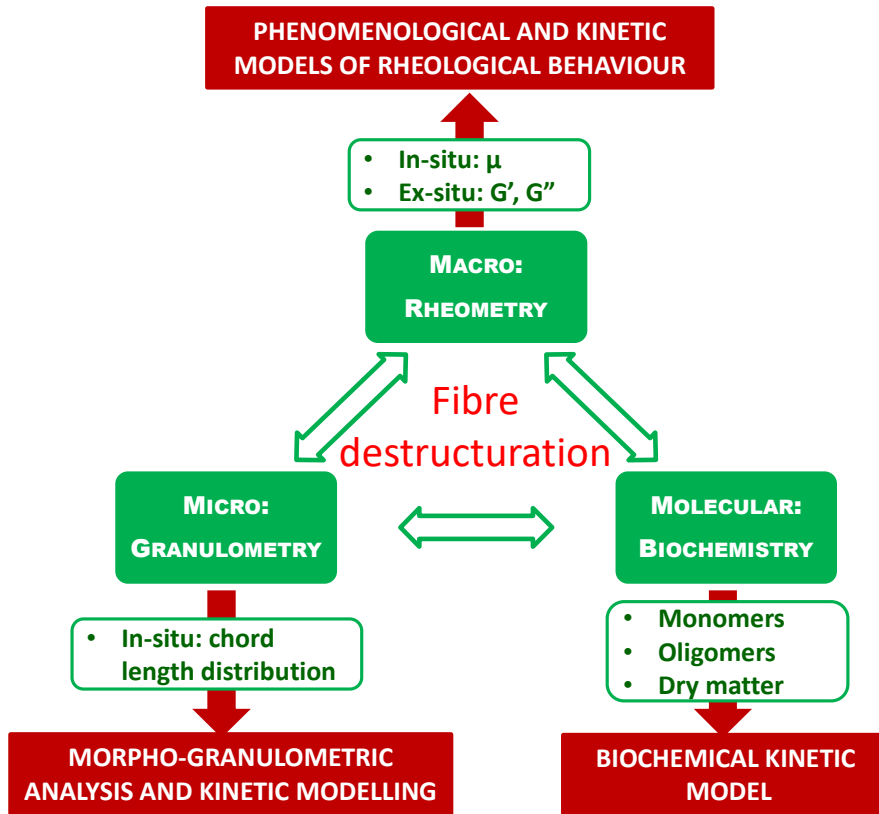
*Figure 4: Yield stress during hydrolysis versus dry matter content (1%-0.5 stands for: 1% dm and 0.5 mL enzyme/g cellulose)*

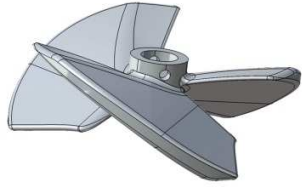
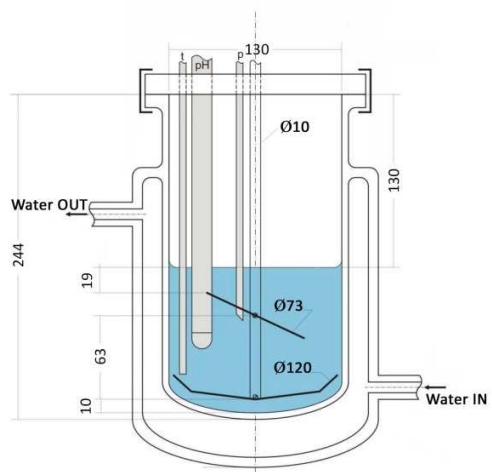
*Figure 5: Uniqueness of dimensionless viscosity-time curves (1%-0.5 signifies 1% dm and 0.5 mL enzyme/g cellulose).*

*Figure 6: Overview of in-situ (FBRM) particle size analysis: A: Chord number and B: Mean chord.*

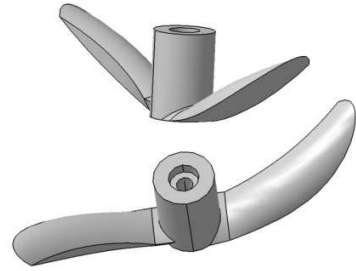
*Figure 7: Population balance (classes defined from initial chord length distribution), in-situ viscosity, and hydrolysis yield during hydrolysis (1%-0.5 mL enzyme/g cellulose case).*

*Figure 8: Reduction of physico-biochemical parameters during hydrolysis using kinetic models (for 3% w/v, 0.5 mL enzyme/g cellulose)*

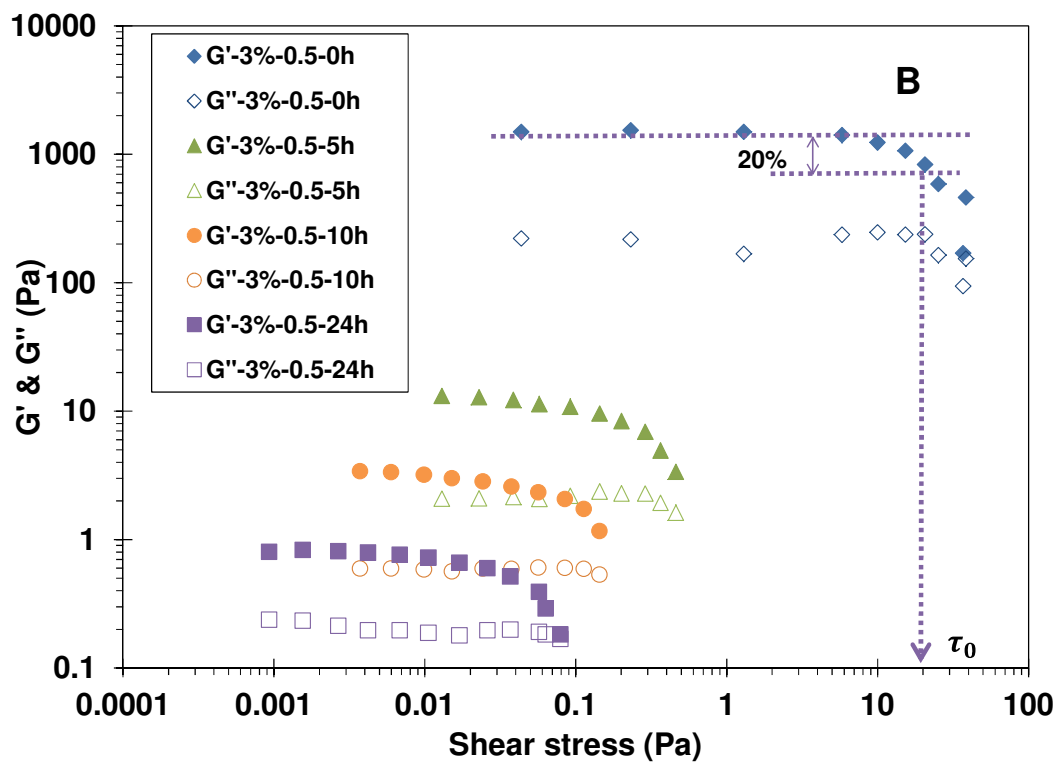
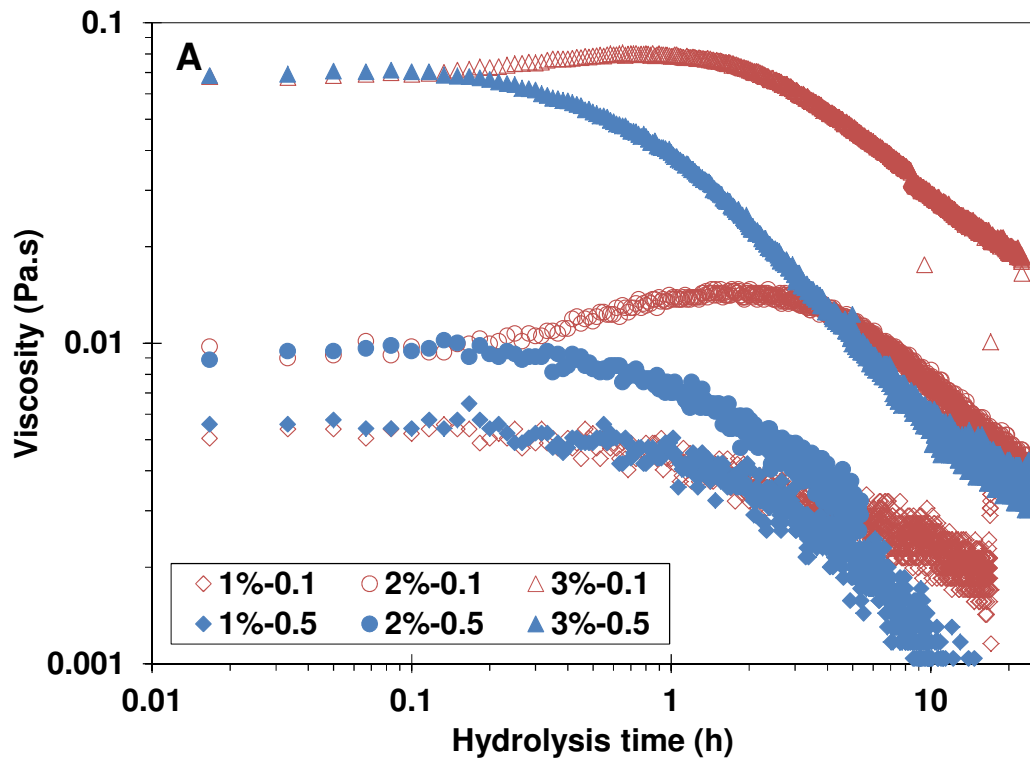




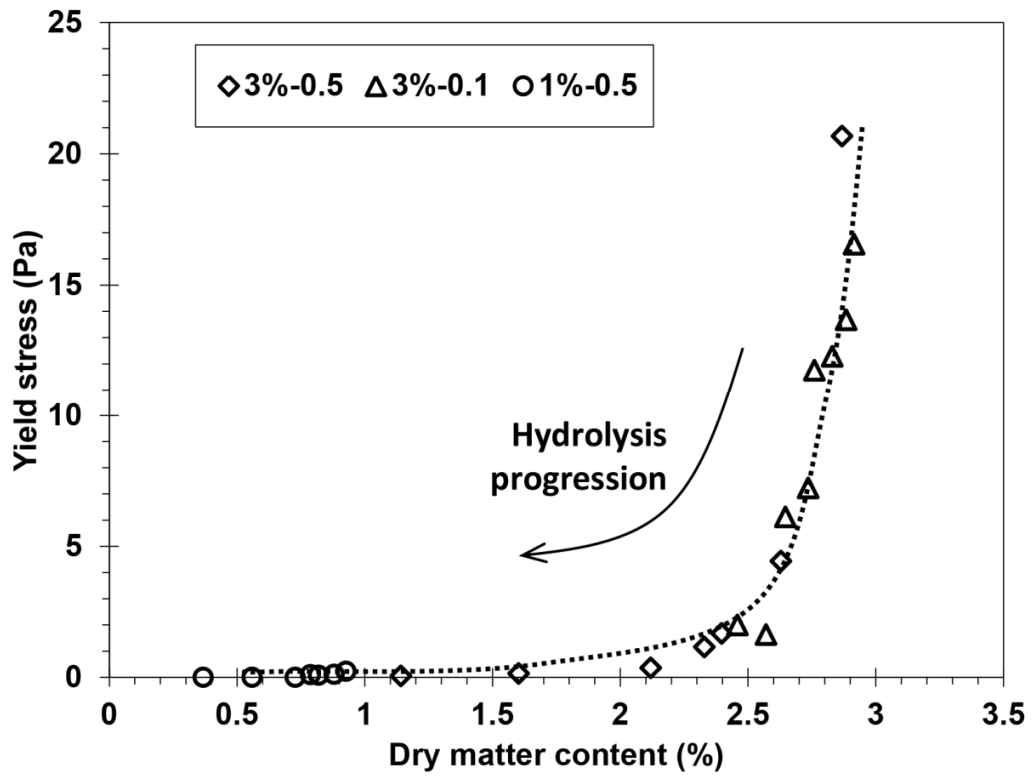
Agitator  $\phi 73$   
Height: 38

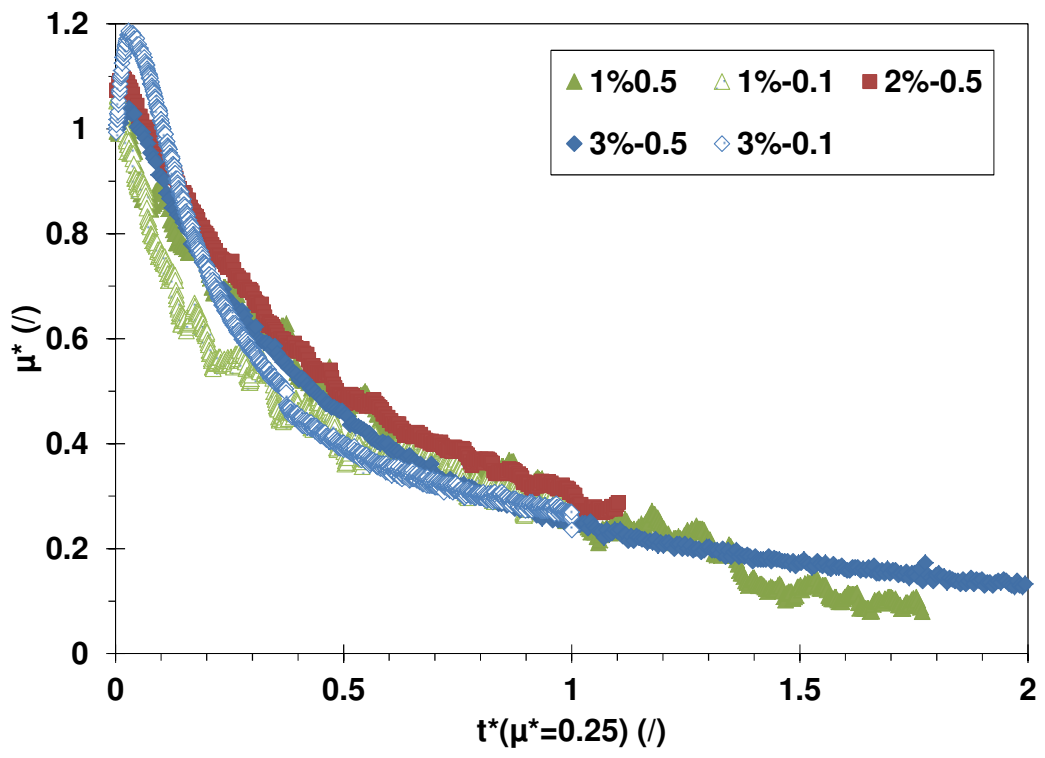


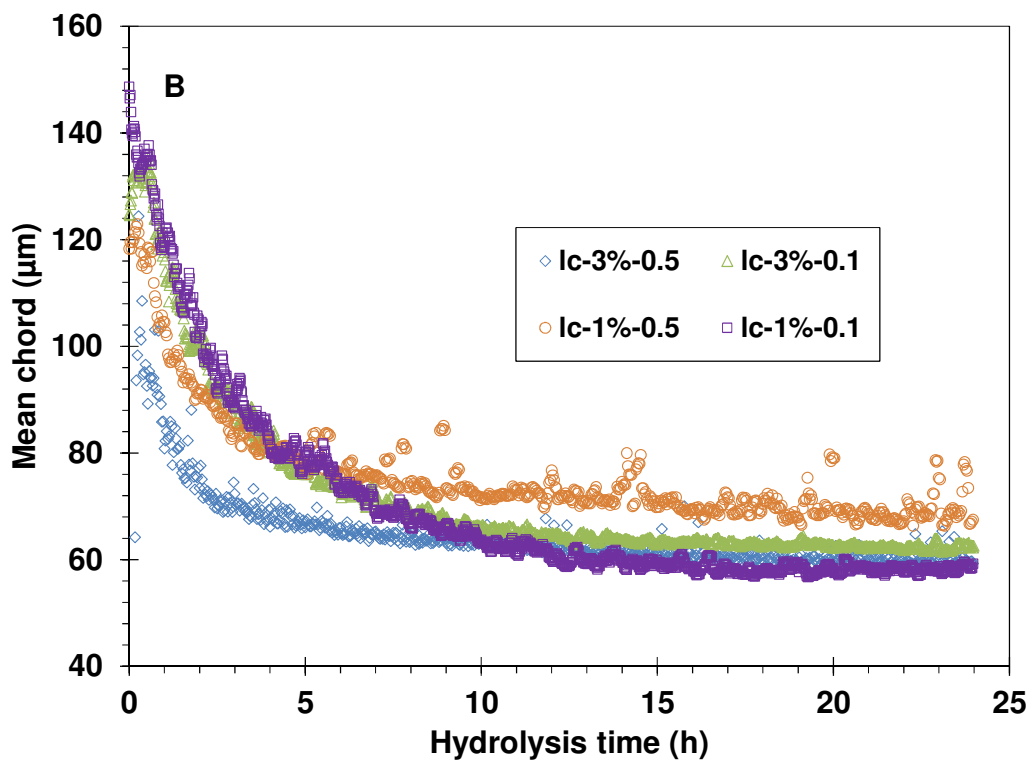
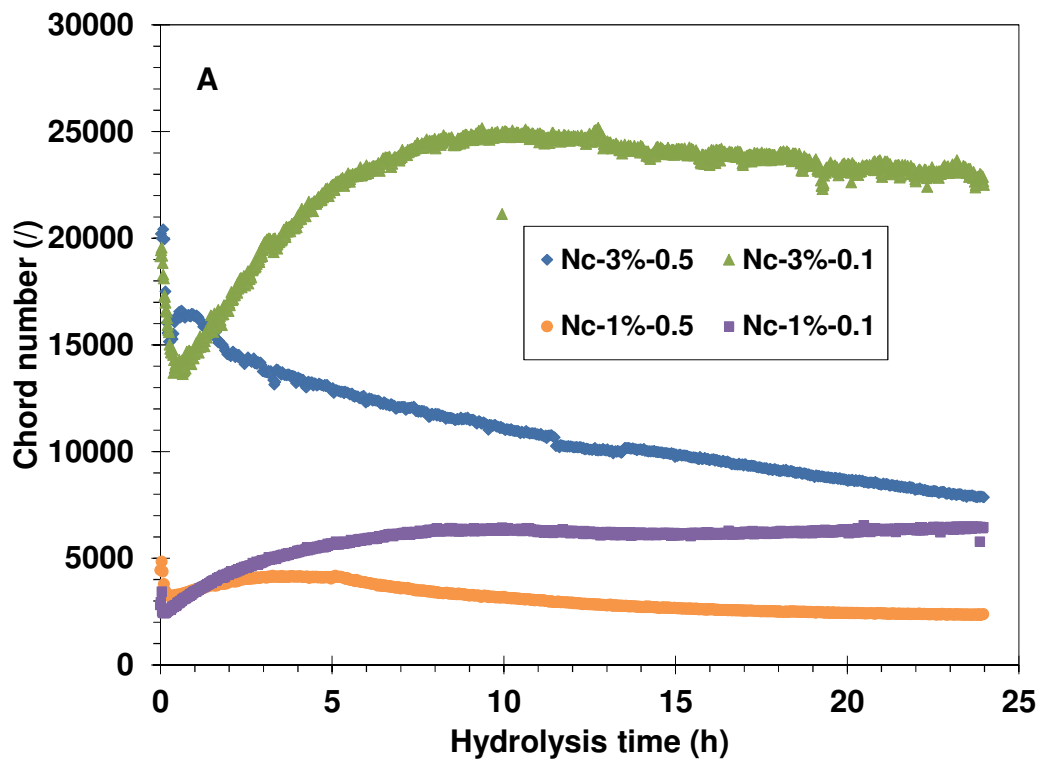
Agitator  $\phi 120$   
Height: 22

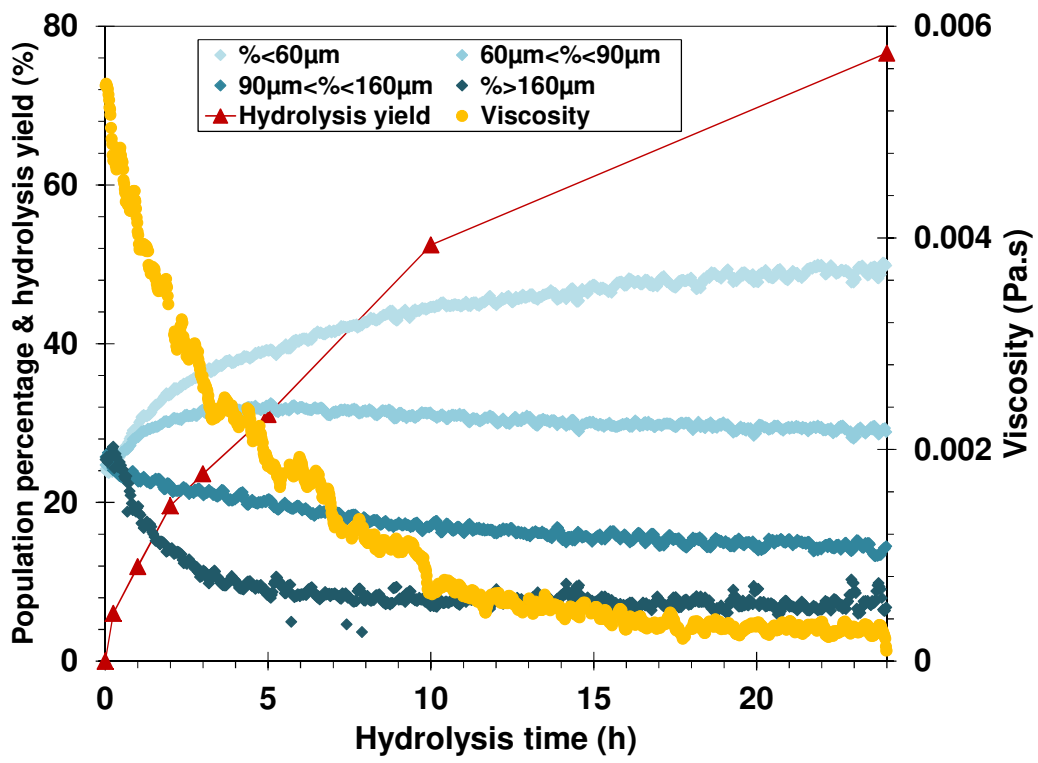


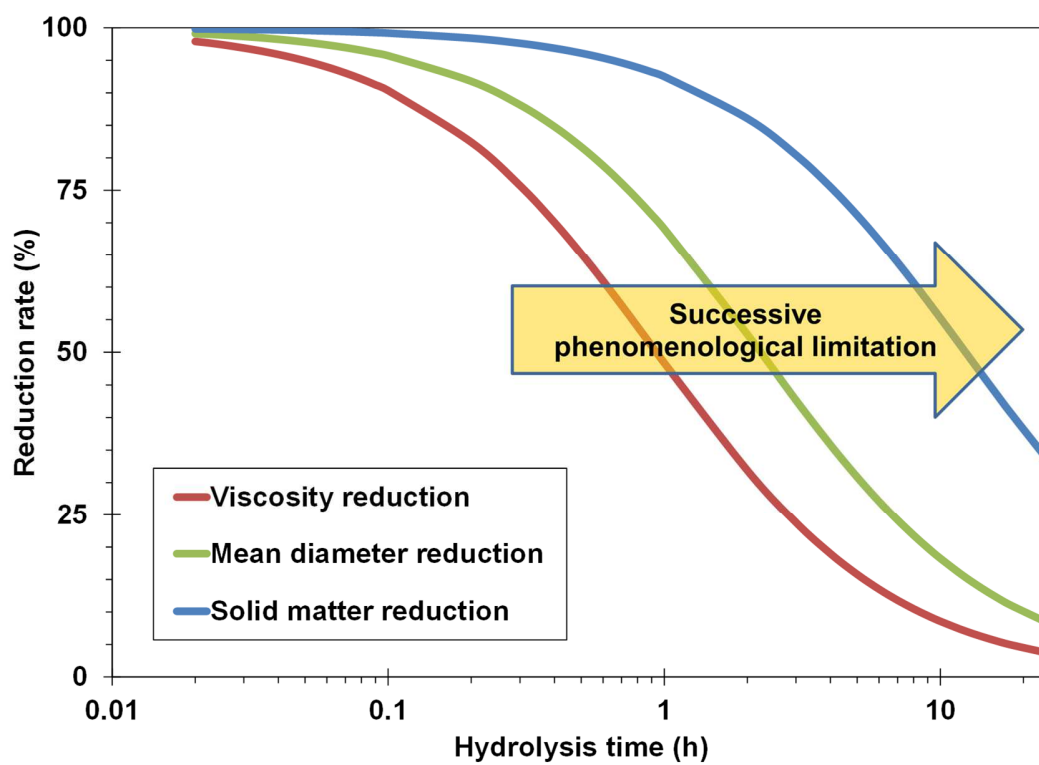












*Table 1: Bioconversion rate (%) during enzyme hydrolysis*

| Substrate concentration<br>(%gdm.L <sup>-1</sup> ) | Enzyme/Substrate<br>(mL/g cellulose) |    |       |      |      |      |      | Mass balance |
|--|--------------------------------------|----|-------|------|------|------|------|--------------|
|  |                                      | 0h | 0.25h | 2h   | 5h   | 10h  | 24h  | (24h,%)      |
| 1  | 0.1                                  | 0  | 1.0   | 4.7  | 9.6  | 11.6 | 13.2 | 93.7         |
| 1  | 0.5                                  | 0  | 6.0   | 19.6 | 31.1 | 52.4 | 76.6 | 94.1         |
| 3  | 0.1                                  | 0  | 1.3   | 6.4  | 11.3 | 14.5 | 19.1 | 94.6         |
| 3  | 0.5                                  | 0  | 6.0   | 20.6 | 32.7 | 54.1 | 73.5 | 95.2         |

Table 2: Kinetic coefficients of rheological, granulometric ( $l_{c_{\infty}}$  in  $\mu\text{m}$ ) and biochemical parameters, and associated correlation coefficients  $R^2$ .

| Coefficient   | 1%w/v                                       |  | 3%w/v                                       |  |
|---|---|--|---|--|
|   | 0.1mL/g<br>cellulose                        | 0.5mL/g<br>cellulose                         | 0.1mL/g<br>cellulose                        | 0.5mL/g<br>cellulose                         |
| $k_{\mu}$ ( $\text{Pa}^{-1} \cdot \text{s}^{-2}$ )  | -   | 61.5   | 2.1   | 15.3   |
| $R^2$   | -   | 0.936  | 0.979                                       | 0.988  |
| $k_{lc}$ ( $\mu\text{m}^{-1} \cdot \text{s}^{-1}$ ) | $8.1 \times 10^{-3}$<br>(65 $\mu\text{m}$ ) | $44.7 \times 10^{-3}$<br>(66 $\mu\text{m}$ ) | $9.0 \times 10^{-3}$<br>(60 $\mu\text{m}$ ) | $32.1 \times 10^{-3}$<br>(59 $\mu\text{m}$ ) |
| $R^2$   | 0.875                                       | 0.876  | 0.950                                       | 0.938  |
| $k_S$ ( $(\text{gdm/L})^{-1} \cdot \text{s}^{-1}$ ) | $7.6 \times 10^{-3}$                        | $71.7 \times 10^{-3}$                        | $3.7 \times 10^{-3}$                        | $26.9 \times 10^{-3}$                        |
| $R^2$   | 0.822                                       | 0.987  | 0.885                                       | 0.995  |