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**Influence of surface charge and polymer coating on internalization and biodistribution of PEG-modified iron oxide nanoparticles**

Lionel Maurizi<sup>1†</sup>, Anne-Laure Papa<sup>1††</sup>, Laure Dumont<sup>2</sup>, Frédéric Bouyer<sup>1</sup>, Paul Walker<sup>3</sup>,  
David Vandroux<sup>2</sup>, and Nadine Millot<sup>1\*</sup>

<sup>1</sup> Laboratoire Interdisciplinaire Carnot de Bourgogne, UMR 6303 CNRS/Université de Bourgogne, 9 avenue Alain Savary, BP 47870, 21078 Dijon cedex, France

<sup>2</sup> NVH Medicinal, 57 rue Vannerie, 21000 Dijon, France

<sup>3</sup> Département de Spectroscopie par Résonance Magnétique, Centre Hospitalier Universitaire de Dijon, 2 Boulevard Maréchal de Lattre de Tassigny, 21033 Dijon, France

\* *Corresponding author: nmillot@u-bourgogne.fr (+333 803 959 37; Fax: +333 803 961 67)*

† Current address: Powder Technology Laboratory, Ecole Polytechnique Fédérale de Lausanne (EPFL), CH-1015 Lausanne, Switzerland

†† Current address: Wyss Institute for Biologically Inspired Engineering, Harvard University, Boston, MA 02115, USA

## **Abstract**

The aim of this study was to investigate the influence of the surface charge and coating of Superparamagnetic Iron Oxide Nanoparticles (SPIONs) on their *in vitro* and *in vivo* behaviors. Neutral and negatively-charged PEG-based SPIONs were synthesized and compared to Resovist®, a carboxydextran-based SPION currently used in clinics. Their cytotoxicity, cell internalization, and potential as contrast agents for magnetic resonance imaging were assessed. Neutral pegylated SPIONs were internalized less readily by the reticuloendothelial system and showed a lower uptake by the liver, compared to negatively-charged SPIONs (with carboxydextran and PEG). These results suggested that the charge of functionalized SPIONs was more relevant for their biological interactions than the nature of their coating.

**Keywords:** *In vivo* assays, Macrophages, MRI (magnetic resonance imaging), Nanoparticles, Surface grafting.

## 1 Introduction

Superparamagnetic iron oxide nanoparticles (SPIONs) are currently being developed for several biological applications such as hyperthermia, drug delivery <sup>1-3</sup>, or Magnetic Resonance Imaging <sup>4-6</sup> (MRI). For the latter application, SPIONs are usually used as Water proton transverse relaxation time ( $T_2$ ) contrast agents <sup>7,8</sup>, whereas conventional gadolinium chelates ( $Gd^{3+}$ ) are used as  $T_1$  contrast agents <sup>9</sup>. However, iron oxide nanoparticles must fulfill certain requirements in order to be used for MRI applications: (i) they should crystallize in a spinel structure ( $AB_2O_4$ ) such as  $Fe_3O_4$  (magnetite) or  $\gamma-Fe_2O_3$  (maghemite) <sup>10,11</sup> and (ii) their crystallite size should be smaller than 20 nm <sup>12,13</sup>. The crystallite size of SPIONs also influences their magnetic response in MRI <sup>14,15</sup>.

Moreover, the intrinsic instability of SPIONs in physiological conditions does not allow their use without surface modification. Two types of stabilizing agents are commonly used on the surface of nanoparticles. Electrostatic agents are able to modify the charge of SPIONs in the positive or negative direction. For example, amine <sup>16</sup>, carboxylic acid <sup>17</sup>, or phosphate groups <sup>18</sup> can be grafted during or after SPION synthesis. They are directly linked to the SPION surface due to the chelating interaction of OH groups with carboxylic or phosphate groups <sup>19,20</sup>, or by condensation of silane precursors <sup>21</sup>. Alternatively, steric agents stabilize the nanoparticles by physical hindrance. Polymers <sup>22</sup> with long hydrophilic chains, such as Polyethylene Glycol (PEG) and Polyvinyl alcohol (PVA), are commonly used <sup>23-25</sup> to avoid aggregation at physiological conditions <sup>26</sup>. Coating with polysaccharides such as dextran provides another approach to stabilizing these particles and is commonly used in commercial iron oxide nanoparticle MRI contrast agents <sup>27</sup>.

These surface modifications usually influence SPION size and biodistribution. Because they are bigger than gadolinium molecules ( $Gd^{3+}$ ), iron oxide nanoparticles usually have either a rapid hepatic uptake (when they are larger than 50 nm) <sup>4,27</sup>, or a delayed hepatic or

ReticuloEndothelial System (RES) uptake when they are below 40-50 nm<sup>28</sup> and could even reach sites of inflammation by lymphocyte transport<sup>29</sup>. In contrast, Gd<sup>3+</sup> complexes have a much faster renal clearance<sup>30</sup> than SPIONs<sup>31</sup>. Surface functionalization seems to influence the biological behavior of iron oxide nanoparticles. When they are in contact with nanoparticles, plasma proteins usually adhere to their surface, leading to a size increase<sup>32</sup>. This phenomenon, called opsonization, influences the half-life and biodistribution of nanoparticles *in vivo*; the faster the proteins are adsorbed onto the surface, the faster the nanoparticles will be taken up by the RES<sup>33</sup>. Finally, the initial surface charge could also influence nanoparticle uptake, as it is commonly observed that negatively-charged nanoparticles have faster interactions with cells or organs<sup>34,35</sup>. Faure *et al.* also demonstrated that the clearance and biodistribution of pegylated gadolinium oxide nanoparticles were governed by both the polymer length and terminal group<sup>35</sup>. Ge *et al.* recently reported that meso-2,3-dimercaptosuccinic acid (DMSA)-Fe<sub>2</sub>O<sub>3</sub> particles showed dose-dependent cytotoxicity to human aortic endothelial cells<sup>36</sup> and highlighted the need for careful evaluation of these particles before their medical application.

Surface modification is a key parameter affecting the biodistribution of nanoparticles. The present study investigated the influence of surface coating and charge on the initial *in vitro* and *in vivo* behaviors of SPIONs. SPIONs were synthesized with similar crystallite and hydrodynamic sizes as the commercially-available carboxydextran-coated iron oxide MRI contrast agent (Resovist®). They were functionalized with silanated polymer, methoxy-PEG<sub>2000</sub><sup>37,38</sup> (PEG<sub>2k</sub>-Si), or with a combination of PEG<sub>2k</sub>-Si and carboxylic acid (DMSA)<sup>39</sup>. These two types of nanoparticles were coated with the same stabilizing agent (PEG polymers), but had different charges, respectively neutral and negative. This new dual combination permitted us to obtain sterically stabilized pegylated SPIONs with thiol groups, providing negative charges and anchoring groups for further modification<sup>40</sup>. Three different

particles were compared; one with dextran and an initial negative charge, one with PEG and an initial neutral charge, and one with PEG and an initial negative charge. The cytotoxicity and interactions of these SPIONs were studied in mouse macrophages (RAW 264.7) and human hepatic cells (HepG2). To further investigate the RES behavior, Transmission Electron Microscopy (TEM) was performed on RAW 264.7 cells to observe SPION internalization. Finally, these nanoparticles were studied by MRI to estimate their potential as contrast agents, their biodistribution, and their clearance. Histology staining was also carried out to confirm the *in vivo* behaviors of these SPIONs.

## **2 Materials and methods**

### *2.1 Materials*

#### *2.1.1 Reagents*

Iron (II) chloride tetrahydrate ( $\text{FeCl}_2 \cdot 4\text{H}_2\text{O}$ ), iron (III) chloride hexahydrate ( $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$ ), sodium hydroxide (NaOH), hydrochloric acid (HCl 37%), nitric acid ( $\text{HNO}_3$  69%), meso-2,3-dimercaptosuccinic acid (DMSA), PEG<sub>2k</sub> (molecular weight = 2000 g·mol<sup>-1</sup>), 3-isocyanatopropyltriethoxysilane (ICPTS), dibutyltin dilaurate (DBTL), *n*-hexane, and tetrahydrofuran (THF) were purchased from Sigma Aldrich and used without any further purification. THF was distilled over sodium metal and benzophenone just before use. 3-(4,5-Dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) reagent and Prussian blue (Accustain® iron stain) were purchased from Sigma.

#### *2.1.2 Control iron oxide suspension used as an MRI contrast agent*

Resovist® from Schering is a negatively-charged carboxydextran SPION suspension used in the clinical context at the University Hospital of Dijon. This commercially available MRI T<sub>2</sub> contrast agent was used as a reference for this study <sup>8</sup>.

### 2.1.3 SPION preparation

Naked SPIONs were synthesized using a classical co-precipitation protocol<sup>41,42</sup>. Briefly, a stoichiometric mixture of 20 mL of 2 M iron II chloride solution and 80 mL of 1 M iron III chloride solution (molar ratio of Fe<sup>II</sup>:Fe<sup>III</sup> = 1:2) was added dropwise to 800 mL of 0.75 M NaOH solution (25°C). The naked SPION suspension formed was then washed and peptized several times with 1 M HNO<sub>3</sub> solution and water. The suspension was then dialyzed at pH 3 with a HNO<sub>3</sub> solution for 2 days, and was subsequently stored at this pH. The particle concentration was 23 mg·mL<sup>-1</sup>. X-Ray Diffraction (XRD) patterns of the nanoparticles were characteristic of the magnetite spinel structure (Fe<sub>3</sub>O<sub>4</sub>) (results not shown).

PEG<sub>2k</sub>-Si was obtained by reacting ICPTS with PEG<sub>2k</sub><sup>43</sup> in anhydrous THF at 60°C under nitrogen flow for 48 h, with DBTL as a catalyst. The PEG<sub>2k</sub>-Si was then precipitated in hexane. Pegylated SPIONs were obtained as follows: 0.13 mmol of PEG<sub>2k</sub>-Si was dissolved in 40 mL ethanol/water (50/50: v/v), pH 4, at 25°C<sup>43-45</sup>. One hundred milligrams of SPIONs were added and the reaction proceeded for 48 h under N<sub>2</sub> flow. The suspension was then dialyzed with water for one week. The resultant nanoparticles were called PEG<sub>2k</sub>-SPIONs.

PEG<sub>2k</sub>-SPIONs were also grafted with DMSA<sup>40</sup>. Pegylated particles were obtained as described above, except for the addition of 0.0439 mmol of DMSA 2 h prior to the end of the reaction. The modified SPIONs were then sedimented 3 times at pH 3 in the presence of a magnetic field, peptized at pH 10, and adjusted to pH 7. These nanoparticles were called DMSA + PEG<sub>2k</sub>-SPIONs.

Naked SPIONs, PEG<sub>2k</sub>-SPIONs and DMSA + PEG<sub>2k</sub>-SPIONs are schematically represented in Figure 1-a.

#### 2.1.4 Cell culture

Macrophage cell types from mice with leukemia (RAW 264.7) and HepG2 hepatocytes from a human liver carcinoma were cultured in 10 cm diameter petri dishes, in DMEM medium supplemented with 10% fetal bovine serum (FBS).

#### 2.1.5 Animal study

3 month old Swiss Albino mice (30 g) were purchased from the animal care unit of the University Hospital of Dijon.

#### 2.2 Physicochemical characterization

The zeta potential ( $\xi$ ) and the hydrodynamic size ( $d_H$ ) of the nanoparticles were measured with a Malvern Zetasizer, using DTS Nano 4.20 software. The nanoparticle suspensions were diluted to approximately  $200 \mu\text{g}\cdot\text{mL}^{-1}$  in  $10^{-2}$  M NaCl solution. The NaCl stock solution had previously been filtered through a  $0.8 \mu\text{m}$  filter. The  $\xi$  potentials of functionalized and naked SPION suspensions were measured from pH 3 to pH 12; pH was adjusted with 0.1 M HCl and 0.1 M NaOH, respectively.

TEM observations were performed with a JEOL JEM-2100 instrument operating at 200 kV. The copper grids were dipped in a dilute suspension of samples and left to dry completely before analysis. The mean size of the crystallites,  $d_{\text{TEM}}$ , was obtained from at least one hundred particles.

Specific surface area measurement of naked SPIONs was performed using a 5 isotherm point BET (Brunauer-Emmett-Teller) method with a BELSORP-mini apparatus with  $\text{N}_2$  gas adsorption.

Thermogravimetric analyses (TGA) were performed with a symmetrical thermobalance (SETARAM TAG24). The heating rate was  $2^\circ\text{C}\cdot\text{min}^{-1}$  up to  $800^\circ\text{C}$  under  $\text{N}_2/\text{O}_2$  (0.12/0.04  $\text{L}\cdot\text{min}^{-1}$ ). Sample weights were around 5 to 10 mg. Any weight loss was assumed to be due to



polymer degradation. The amount of DMSA on the particles was measured with Ellman's reagent assay titration<sup>40,46</sup>.

The particle relaxivities ( $r_2$  in  $\text{mM}_{\text{Fe}}^{-1} \cdot \text{s}^{-1}$ ) were calculated using the following method<sup>47</sup>. Water proton transverse relaxation time ( $T_2$ ) measurements at 3.0 T were carried out at  $298 \pm 1$  K with a Siemens Magnetom Trio TIM, using a commercially-available birdcage head coil. Relaxation time measurements were performed on test tubes containing SPIONs, with or without surface modification, in acrylamide gel at a range of concentrations (0, 0.013, 0.038, 0.063, 0.088, 0.125, and 0.25  $\text{mM}_{\text{Fe}}$ ). For  $T_2$  determination, a multi-echo spin-echo (SE) pulse sequence was used, with  $T_R$  (repletion time) = 5000 ms, matrix =  $256 \times 204$ , Field of View (FOV) =  $100 \times 80$  mm and slice thickness = 5 mm. Images were acquired at 32 echoes, from  $T_E = 8.2$  ms to 275.2 ms, with an 8.2 ms interval. Image analysis was performed using ImageJ (Image Analysis Software developed by the NIH, USA). The signal decay curve was fitted to the equation  $S(T_E) = A \exp(-T_E/T_2)$  using a non-linear function. The relaxivity,  $r_2$  (in  $\text{mM}_{\text{Fe}}^{-1} \cdot \text{s}^{-1}$ ), was then determined by fitting the relaxation rate ( $1/T_2$ ) as a function of iron concentration in  $\text{mM}_{\text{Fe}}$ .

## 2.3 Biological observations

### 2.3.1 Cell viability

The MTT assay was used to measure the cytotoxicity of SPIONs in two different cell lines (RAW 264.7 and HepG2). This measured mitochondrial metabolic activity, which converted the MTT reagent from a yellow tetrazolium salt to the dark blue formazan by reductive cleavage of the tetrazolium ring. Formazan crystals were then dissolved in an organic solvent and quantified by measurement of the absorbance of the solution at 570 nm, with the resultant value being proportional to the number of living cells. To determine cell viability, the cells were seeded in a 96-well plate and incubated at 37°C in a 5%  $\text{CO}_2$  atmosphere for 24 h until they reached 50% confluency. Cells were then incubated in the

presence of the different types of SPIONs or Resovist® (at two concentrations: 25 and 50  $\mu\text{g}_{\text{Fe}}\cdot\text{cm}^{-2}$  well area), and physiological saline (NaCl at 0.15 M). These two concentrations were chosen on the basis of the dose injected in the *in vivo* study. Briefly, 5.6  $\mu\text{g}_{\text{Fe}}\cdot\text{g}^{-1}$  mouse were injected (30 g mice, 2 mL of circulating blood<sup>48</sup>), i.e. 168  $\mu\text{g}_{\text{Fe}}/\text{mouse}$  and 84  $\mu\text{g}\cdot\text{mL}^{-1}$  blood. In 96-well plates, the working volume was 100  $\mu\text{L}$  so we selected 8.4  $\mu\text{g}_{\text{Fe}}/\text{well}$  (about 25  $\mu\text{g}_{\text{Fe}}\cdot\text{cm}^{-2}$  since each well had a surface area of 0.32  $\text{cm}^2$ ). This concentration was then doubled for high-dose cytotoxicity assessment. Previous studies have employed these doses to investigate both MRI biodistribution and cytotoxicity (MTT assays) of SPIONs<sup>49,50</sup>. After 1, 3, and 24 h of incubation, the wells were washed 2 times with PBS at 37°C. Then 100  $\mu\text{L}$  MTT (2  $\text{mg}\cdot\text{mL}^{-1}$ ) was added in Puck G+ medium (126.6 mM NaCl, 3.8 mM KCl, 0.11 mM  $\text{CaCl}_2\cdot 2\text{H}_2\text{O}$ , 0.62 mM  $\text{MgSO}_4\cdot 7\text{H}_2\text{O}$ , 0.86 mM  $\text{Na}_2\text{HPO}_4\cdot 2\text{H}_2\text{O}$ , 0.61 mM  $\text{KH}_2\text{PO}_4$ , 14.3 mM  $\text{NaHCO}_3$ , and 1.1  $\text{g}\cdot\text{L}^{-1}$  glucose, pH 7.4) for 1 h at 37°C. Puck G+ corresponded to basic osmotic medium and was used to avoid interference during the MTT reaction. Excess MTT was then removed and 200  $\mu\text{L}$  of isopropanol with HCl (0.1 M) was incubated with the cells for 1 h at room temperature. The absorbance in each well was then read in a spectrophotometer (Wallac Victor3, Perkin Elmer) at 570 nm. Average absorbance (from n = 4) was normalized to the control. The viability (in %) was then calculated by  $\text{absorbance}_{\text{test}}/\text{absorbance}_{\text{control}} \times 100$ .

### 2.3.2 SPION interactions with macrophages and hepatocytes

Optical microscopy: RAW 264.7 and HepG2 cells seeded on 1 cm diameter glass coverslips in 24-well plates were incubated for 3 h in the presence of Resovist®, PEG<sub>2k</sub>-SPIONs, or DMSA + PEG<sub>2k</sub>-SPIONs at a concentration of 25  $\mu\text{g}_{\text{Fe}}\cdot\text{cm}^{-2}$ . The cells were then washed 2 times with PBS and fixed with 4% paraformaldehyde, 0.5% glutaraldehyde, and 0.1% triton for 15 min. Then the cells were fixed in sodium borohydride (25  $\text{mg}\cdot\text{mL}^{-1}$  in PBS) for 10 min, washed once with PBS, incubated with 2% potassium ferrocyanide and 0.6

mmol/L HCl for 10 min, and washed with water. Cells were then stained with 0.2% pararaxilin for 1 min, before washing consecutively with water, 95% and 100% ethanol, and 100% xylene for 10 min. The dissolved iron from the SPIONs reacted with potassium ferrocyanide to give a blue product. The cells were then mounted on glass slides using Eukitt® (Sigma) and observed by optical microscopy. RAW 264.7 and HepG2 cells incubated with complete media were observed as control.

Transmission Electron Microscopy: RAW 264.7 cells were incubated with Resovist®, PEG<sub>2k</sub>-SPIONs, or DMSA + PEG<sub>2k</sub>-SPIONs at a concentration of 25  $\mu\text{g}_{\text{Fe}}\cdot\text{cm}^{-2}$ . Cells were then washed 3 times with PBS and fixed at room temperature with 4% paraformaldehyde and 1.5% glutaraldehyde in a phosphate buffer (0.1 M, pH 7.4) for 30 min. The cells were then washed 4 times in phosphate buffer and then one last time overnight at 4°C. Cells were then post-fixed using 1% osmium tetroxide solution in phosphate buffer, washed and dehydrated by 15 min incubations in successive ethanol baths (from 30% to 100%). Cells were then embedded in epoxy resin (EMBed-812) overnight at 37°C, followed by 48 h at 60°C. Resins were then cut with an ultramicrotome (slices from 80 to 90 nm). After cutting, the slices were stained with uranyl acetate and lead citrate solution for contrast enhancement and observed by TEM on a HITACHI H-7500 operating at 80 kV.

### 2.3.3 Mice MRI analyses

Mice were injected with 100  $\mu\text{L}$  of physiological saline (NaCl at 0.15 M) (n = 6), Resovist® (n = 6), PEG<sub>2k</sub>-SPIONs (n = 6), or DMSA + PEG<sub>2k</sub>-SPIONs (n = 4). SPION suspensions were diluted in physiological saline to a concentration of 1.68  $\text{mg}_{\text{Fe}}\cdot\text{mL}^{-1}$ . To reach a concentration of 5.6  $\text{mg}_{\text{Fe}}\cdot\text{kg}^{-1}$  mouse, the injected dose was around 168  $\mu\text{g}_{\text{Fe}}$  per mouse. This experiment was conducted in accordance with French ethical guidelines for *in vivo* studies (authorization number B0410).

Injected mice were then analyzed by 3.0 T MRI at 1, 3, and 24 h after injection. Gradient echo  $T_2^*$  weighted was used for acquisition with echo-time  $TE = 2.9$  ms, repetition time  $T_R w = 300$  ms, slice thickness = 2 mm and an in-plane resolution of  $0.35 \mu\text{m}^2$ . Images were acquired in sagittal, longitudinal and transverse slice orientations. On coronal images, the average contrast of the liver was calculated using Image J and reported in grey level (where 0 was full black and 2048 was full white). Half of each mouse population was sacrificed 3 h after injection for organ analysis and the other half was sacrificed after 24 h.

#### 2.3.4 *SPION distribution in organs*

The livers and hearts of the sacrificed mice were harvested and sliced. These organs were then treated with Prussian blue to reveal the presence of iron from dissolved SPIONs. Optical microscopy was then performed on the slices and the images obtained from test and control animals were compared.

#### 2.3.5 *Statistics*

All statistics were performed using the GraphPad Prism software and an unpaired t-test.

### **3 Results and discussion**

#### 3.1 *Physicochemical characterization of SPIONs*

Some physicochemical characteristics of the synthesized and functionalized SPIONs are summarized in **Table 1** and compared to those of the commercial Resovist® SPION suspension used as the reference nanomaterial in this study. No significant differences were observed in the crystallite sizes of Resovist® and the synthesized SPIONs. The hydrodynamic diameter of naked SPIONs increased to around 40 nm after PEG<sub>2k</sub> functionalization. This size was close to that of Resovist®. The major difference between PEG<sub>2k</sub>-SPIONs and Resovist® was their charge. At pH 7, Resovist® was negatively charged, while PEG<sub>2k</sub>-SPIONs were only slightly negatively charged (almost neutral). The addition of 0.3 molecule of DMSA per nm<sup>2</sup> of PEG<sub>2k</sub>-SPIONs did not change the size of the nanoparticles, but induced a negative

charge ( $\xi = -30\text{mV}$ ) at pH 7. According to the study performed by Fauconnier *et al.* <sup>51</sup>, the maximum concentration of DMSA/nm<sup>2</sup> should be around 2. However, Ge *et al.* demonstrated that this concentration could induce some cytotoxicity <sup>36</sup>. The lower DMSA concentration and the presence of PEG grafted on the surface of DMSA + PEG<sub>2k</sub>-SPIONs should decrease the potential for cytotoxicity. The presence of DMSA on these SPIONs also decreased the PEG density on the surface from 3.3 to 0.6 molecule/nm<sup>2</sup>, probably by removing excess of polymer from the surface of the SPION. The relaxivities ( $r_2$ ) were all in the same range, regardless of the nature of the surface coating. These measurements confirmed that all of the SPIONs had roughly the same size because the relaxation time ( $T_2$ ) linked to  $r_2$  was size-dependent <sup>15,52,53</sup>. It has also been demonstrated that greater aggregation of magnetic particles could increase  $r_2$  <sup>54</sup> by increasing the presence of magnetic mono-domains in the aggregates <sup>55</sup>. In addition, LaConte *et al.* recently showed that the charges of these nanoparticles did not influence  $r_2$  <sup>56</sup>. Finally, the concentration (molecule/nm<sup>2</sup>) at which the SPIONs were coated with either carboxydextran or PEG<sub>2k</sub>, with or without DMSA, did not influence their magnetic properties. These similarly-sized nanoparticles were studied *in vitro* and *in vivo* to determine the potential influence of their initial surface charge and coating on their biological behaviors. Figure 1-b shows the  $\zeta$  potentials of naked SPIONs, PEG<sub>2k</sub>-SPIONs, and DMSA + PEG<sub>2k</sub>-SPIONs at different pH conditions (from 3 to 12). The isoelectric point of naked and PEG<sub>2k</sub> SPIONs was around pH 7, showing the absence of charge on these two types of nanoparticles. However, the absolute  $\zeta$  values of PEG<sub>2k</sub>-SPIONs (+10 mV) were lower than those of naked SPIONs (+40 mV) at acidic pH; this shielding effect proved that PEG<sub>2k</sub> had been efficiently grafted onto the SPIONs. For the DMSA + PEG<sub>2k</sub>-SPIONs, the isoelectric point was below pH 3, proving that these particles were negatively charged at the pH range used in subsequent analyses.

The cytotoxicity of naked and functionalized-SPIONs was evaluated in macrophages (RAW 264.7) and hepatic cells (HepG2), as shown in

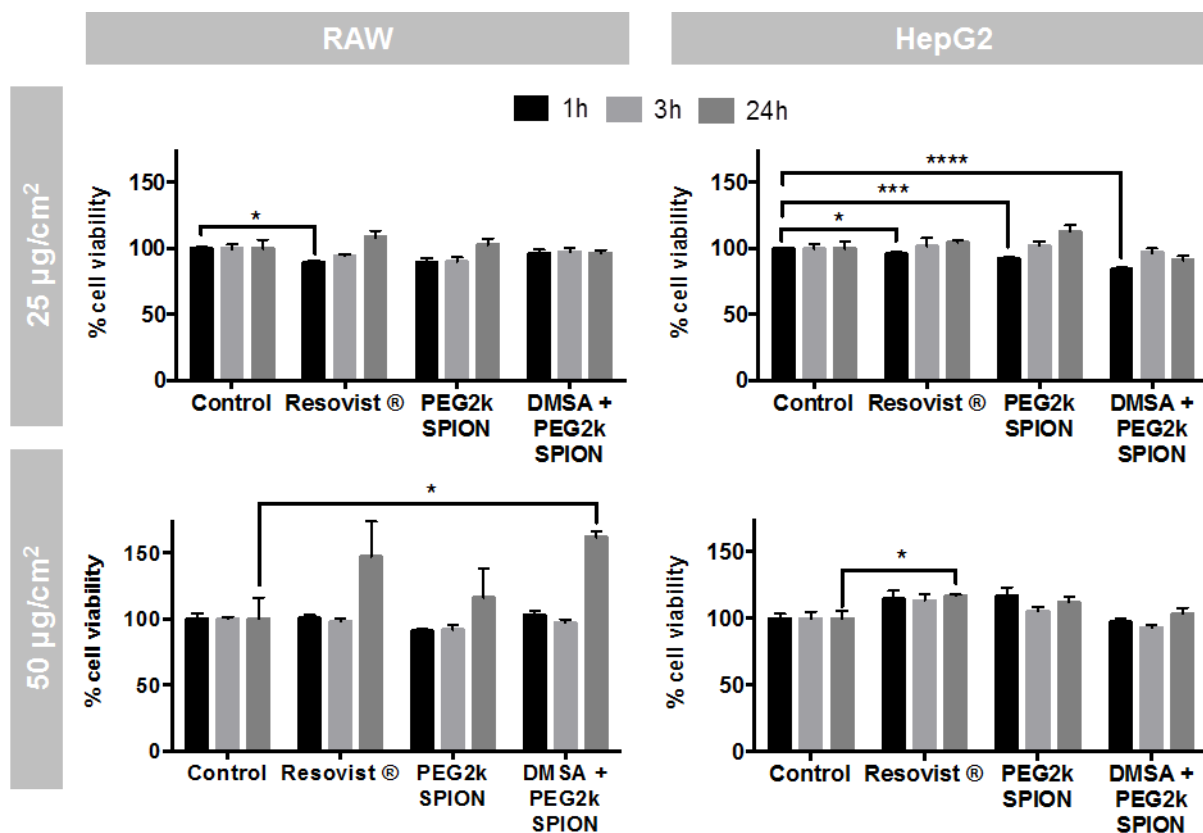


Figure 2. These two cell lines were chosen to model the usual *in vivo* targets of SPIONs: the RES and the liver. Compared with control cells, 24-h incubations with the two test concentrations (25 and 50 µg<sub>Fe</sub>.cm<sup>-2</sup>) of functionalized nanoparticles did not produce significant cytotoxicity, with cell viability remaining above 80% (Figure 2). The unexpectedly high assay values observed (only at the high concentration, after a 24-h incubation with one cell line) may have been due to interference between iron oxide and the assay.

Prussian blue staining experiments were carried out on RAW 264.7 and HepG2 cells incubated for 3 h in the presence of the three different functionalized SPIONs (at 25 µg<sub>Fe</sub>.cm<sup>-2</sup> of well) and compared with control cells, incubated with medium only (Figure 3). Blue colored spots observed on the images were caused by Prussian blue reacting with iron (dissolved from SPIONs) and potassium ferrocyanide. Thus, SPION-containing areas were labeled with the blue pigment. In RAW 264.7 cells incubated with Resovist®, blue spots co-

localized with the cells. No blue spots were observed at this magnification in RAW 264.7 cells incubated with PEG<sub>2k</sub>-SPIONs, while many were observed in and around these cells following incubation with DMSA + PEG<sub>2k</sub>-SPIONs. The distribution of the blue stain in the latter case may have indicated some stronger interactions between these nanoparticles and RAW 264.7 cells. This could reflect particle aggregation under these experimental conditions. Resovist® was also found to co-localize with HepG2 cells. Although very few PEG<sub>2k</sub>-SPIONs seemed to have interacted with HepG2 cells, DMSA + PEG<sub>2k</sub>-SPIONs interacted more with these cells. Some aggregations of these latter particles were observed on these hepatic cells, but not to the extent found on RAW 264.7 macrophage cells. These observations suggested that these particles had aggregated in some places before their dissolution and reaction with potassium ferrocyanide. As the Prussian blue was observed with an optical microscope, it was not possible to distinguish between co-localization and real internalization of the particles. However, these results showed that DMSA + PEG<sub>2k</sub>-SPIONs interacted more with the cells than PEG<sub>2k</sub>-SPIONs, which may be due to their surface charge and some aggregation of the negatively-charged particles. Moreover, Resovist® particles did not show any obvious aggregation (compared to DMSA + PEG<sub>2k</sub>) but showed more interaction with cells than PEG<sub>2k</sub>-SPIONs. Thus, macrophage cells should interact more with SPIONs than hepatic cells because of their phagocytosis role *in vivo*. Regarding cellular uptake, there was no evidence of PEG<sub>2k</sub>-SPION internalization by RAW 264.7 cells.

To obtain more insight into these interactions, TEM observations of RAW 264.7 cells incubated for 3 h with the three different nanoparticles (at 25 µg<sub>Fe</sub>.cm<sup>-2</sup> of well) were carried out (Figure 4). Resovist® (top images) and DMSA + PEG<sub>2k</sub>-SPIONs (bottom images) were easy to detect inside vesicles that were probably phagosomes. As observed in the optical Prussian blue staining, the SPIONs were aggregated, with sizes greater than 100 to 200 nm for Resovist® and around 500 nm for DMSA + PEG<sub>2k</sub>-SPIONs. These particles were entrapped

and therefore concentrated in vesicles. It was obvious from these observations that their sizes had increased from 40-50 nm (Table 1, PCS measurements) to a few hundred nm in contact with cells and their medium. This phenomenon has been described previously under biological conditions. For instance, Schlorf *et al.* reported the aggregation effect of fibrinogen and plasma after a 1-h incubation with dextran-coated SPIONs<sup>57</sup>. In addition, Khandhar *et al.* showed that the Resovist® response to MPS (Magnetic Particle Spectrometry) characterization was modified after incubation with media<sup>58</sup> while their neutral Poly(maleic anhydride-alt-1-octadecene) (PMAO)-PEG-coated iron oxide nanoparticles remained unaffected<sup>59</sup>. On the other hand, it was very difficult to detect PEG<sub>2k</sub>-SPIONs inside RAW 264.7 cells, even at a high magnification. The small black dots around 50 nm circled in yellow in Figure 4 could potentially be pegylated SPIONs, because the electron contrast was slightly higher than the background. Consequently, some PEG<sub>2k</sub>-SPIONs could be present inside vesicles, but in very small quantities (compared with other conditions) and with small sizes, as they were not detected with Prussian blue staining. No aggregates of PEG<sub>2k</sub>-SPIONs were detected inside RAW 264.7 cells following a meticulous observation by TEM. These results suggested that PEG<sub>2k</sub>-SPIONs were more stable under biological conditions than the other nanoparticles tested.

Optical and electron microscopy observations demonstrated that the negatively charged particles interacted more with hepatic (HepG2) and macrophage (RAW 264.7) cell lines than did pegylated neutral nanoparticles. Resovist® and DMSA + PEG<sub>2k</sub>-SPIONs were internalized in vesicles of around 500 nm and were easily detectable by Prussian blue staining or in TEM. In contrast, PEG<sub>2k</sub>-SPIONs were more difficult to observe in the cell cultures and this could be due to a reduced internalization of these nanoparticles by RAW 264.7 and HepG2 cells. If these PEG<sub>2k</sub>-SPIONs avoid being taken up by RES cells, they could remain in the blood circulation for longer than the negatively-charged particles tested.



Swiss albino mice were injected with physiological saline or 168  $\mu\text{g}_{\text{Fe}}$  of Resovist®, PEG<sub>2k</sub>-SPIONs, or DMSA + PEG<sub>2k</sub>-SPIONs. Mice were then analyzed using MRI at 1, 3, and 24 h after injection. On the coronal images obtained 3 h after injection (Figure 5a, livers circled in red), liver and heart contrasts were analyzed and reported in arbitrary units (Figure 5b and 5c). On the MRI images of Figure 5a, the liver of the control mouse appeared less dark than the livers of SPION-injected mice. Moreover, the liver of mice injected with PEG<sub>2k</sub>-SPIONs looked less dark than the other two. This result was confirmed by quantification, showing a significant difference (Figure 5b). No other difference in contrast was observed in any other organ during the whole experiment. As shown in Figure 5b, the average contrast of control mice livers did not change significantly during the experiment (around 700 to 800 a.u.), which suggested that there was no physiological influence on contrast during the experiment. For mice injected with SPIONs, the average contrast of the livers decreased significantly after 1 h, illustrating the negative contrast effect of the SPIONs on this organ. For the negatively-charged particles, the liver contrast reached final values of 50-100 a.u. 1 h after injection. However for PEG<sub>2k</sub>-SPIONs, even though the liver contrast had decreased to similar values of around 100 a.u. 24 h after injection, the contrast was relatively high, around 500 and 400 a.u. at 1 and 3 h post-injection, respectively. As the final contrast value of PEG<sub>2k</sub>-SPIONs after 24 h was similar to that of the other SPIONs, and all the nanoparticles had approximately the same  $r_2$  relaxivities, this indicated that the PEG<sub>2k</sub>-SPIONs accumulated less in the liver 1 and 3 h after injection and would therefore probably have remained in the circulation during this period. For other organ, no significant differences were observed between the contrast of control and injected mice, as shown for the heart in Figure 5c. The functionalized SPIONs used during this study therefore only showed a negative contrast in the liver. Moreover, neutral pegylated SPIONs seemed to be taken-up more slowly than their negatively-charged counterparts.

Prussian blue staining of livers and hearts was conducted to determine whether less PEG<sub>2k</sub>-SPIONs were taken up by the liver than Resovist® and DMSA + PEG<sub>2k</sub>-SPIONs (Figure 6). As the slices observed were taken in the middle of the organs, any nanoparticles detected were assumed to have been taken up by the organ. Firstly, no SPIONs were observed in heart slices, consistent with the MRI findings. However, livers from mice injected with SPIONs had blue spots, revealing the presence of iron oxide nanoparticles, even 3 h after injection. It is also important to note that fewer blue spots were observed in PEG<sub>2k</sub>-SPION-injected mice than in mice administered with the other two types of nanoparticle. By 24 h after injection, the quantity of SPIONs appeared to be very similar. These Prussian blue observations, coupled with the MRI images and contrast analyses, demonstrated that less PEG<sub>2k</sub>-SPIONs were taken up by the liver 3 h after injection, whereas approximately all the Resovist® and DMSA + PEG<sub>2k</sub>-SPIONs were within the liver, even 1 h after injection.

One of the reasons for these observations of lower liver uptake of PEG<sub>2k</sub>-SPIONs, compared with the negatively-charged SPIONs, could be explained by the *in vitro* results. Internalization experiments showed less internalization of PEG<sub>2k</sub>-SPIONs by hepatic and macrophage cells after 3-h incubations. This lower uptake of PEG<sub>2k</sub>-SPIONs by macrophages could help explain their longer circulation time. With PEG, the opsonization of particles by blood proteins and their subsequent uptake by macrophages is usually slower<sup>32</sup>, resulting in an extended blood circulation of PEG-coated nanoparticles which could limit the negative contrast of the liver. These protein adsorption differences have also been observed *in vitro*. The agglomerated SPIONs seen on Prussian blue microscopy imaging could be due to interactions between SPIONs and plasma proteins (usually albumin)<sup>60</sup>. The absence of PEG<sub>2k</sub>-SPION detection correlated with the lower opsonization rate of these particles, compared with those that were negatively-charged. However this phenomenon could not fully explain the longer blood circulation observed with PEG<sub>2k</sub>-SPIONs, because DMSA + PEG<sub>2k</sub>-

SPIONs also had PEG on their surface. The negative charge of these latter SPIONs could provide an explanation for their increased interactions with hepatic cells<sup>35,61</sup>. It was observed that DMSA + PEG<sub>2k</sub>-SPIONs tended to be taken up more by hepatocytes than PEG<sub>2k</sub>-SPIONs, leading to a longer latency in liver than the negatively-charged particles. Moreover, when the SPIONs were charged at physiological pH, they aggregated in the presence of cells and their medium. A final surface charge-related explanation for the slow liver uptake of only PEG<sub>2k</sub>-SPIONs is the aggregation of negatively-charged SPIONs. It has already been reported that particles larger than 40 nm are easily taken up by the liver<sup>7,62-64</sup>. In the present study, the microscopy observations were consistent with this theory.

#### **4 Conclusion**

This study prepared and investigated two different types of SPIONs as potential MRI contrast agents and compared them to commercial SPIONs (Resovist®) that were coated with carboxydextran and negatively-charged.

In conclusion, it was observed that negatively-charged and pegylated nanoparticles had the same biological behavior (cytotoxicity, cellular internalization, and MRI biodistribution) as nanoparticles coated with negatively-charged carboxydextran, even though the surfaces of these particles were very different. The results indicated that the biological behavior of nanoparticles with similar sizes and magnetic properties was mainly governed by the charge, rather than the coating. However, the nature of the coating may also play a role as the density of PEG<sub>2k</sub> was much lower with DMSA than without it.

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## References

1. Q. A. Pankhurst, N. T. K. Thanh, S. K. Jones, J. Dobson, Progress in Applications of Magnetic Nanoparticles in Biomedicine, *J. Phys. -Appl. Phys.* 42 (2009).
2. I. Hilger, W. A. Kaiser, Iron Oxide-Based Nanostructures for MRI and Magnetic Hyperthermia, *Nanomed.* 7, 1443–1459 (2012).
3. A. K. Gupta, R. R. Naregalkar, V. D. Vaidya, M. Gupta, Recent Advances on Surface Engineering of Magnetic Iron Oxide Nanoparticles and Their Biomedical Applications, *Nanomed.* 2, 23–39 (2007).
4. Z. L. Zhang, R. Dharmakumar, N. Mascheri, Z. Y. Fan, S. Y. Wu, D. B. Li, Comparison of Superparamagnetic and Ultrasmall Superparamagnetic Iron Oxide Cell Labeling for Tracking Green Fluorescent Protein Gene Marker with Negative and Positive Contrast Magnetic Resonance Imaging, *Mol. Imaging* 8, 148–155 (2009).
5. F. Liu, S. Laurent, H. Fattahi, L. V. Elst, R. N. Muller, Superparamagnetic Nanosystems Based on Iron Oxide Nanoparticles for Biomedical Imaging, *Nanomed.* 6, 519–528 (2011).
6. R. Singh, H. S. Nalwa, Medical Applications of Nanoparticles in Biological Imaging, Cell Labeling, Antimicrobial Agents, and Anticancer Nanodrugs, *J. Biomed. Nanotechnol.* 7, 489–503 (2011).
7. Y. X. J. Wang, S. M. Hussain, G. P. Krestin, Superparamagnetic Iron Oxide Contrast Agents: Physicochemical Characteristics and Applications in MR Imaging, *Eur. Radiol.* 11, 2319–2331 (2001).
8. P. Reimer, T. Balzer, Ferucarbotran (Resovist): A New Clinically Approved RES-Specific Contrast Agent for Contrast-Enhanced MRI of the Liver: Properties, Clinical Development, and Applications, *Eur. Radiol.* 13, 1266–1276 (2003).
9. K.-J. Chen, S. M. Wolahan, H. Wang, C.-H. Hsu, H.-W. Chang, A. Durazo, L.-P. Hwang, M. A. Garcia, Z. K. Jiang, L. Wu, Y.-Y. Lin, H.-R. Tseng, A Small MRI Contrast Agent Library of gadolinium(III)-Encapsulated Supramolecular Nanoparticles for Improved Relaxivity and Sensitivity, *Biomaterials* 32, 2160–2165 (2011).
10. P. Poix, Sur Une Methode de Determination Des Distances Cation-Oxygene Dans Les Oxydes Mixtes a Structure Spinelle - Application Des Valeurs a Quelques Cas Particuliers, *Bull. Soc. Chim. Fr.*, 1085–1087 (1965).
11. T. Belin, N. Millot, N. Bovet, M. Gailhanou, In Situ and Time Resolved Study of the Gamma/alpha-Fe<sub>2</sub>O<sub>3</sub> Transition in Nanometric Particles, *J. Solid State Chem.* 180, 2377–2385 (2007).
12. J. L. Dormann, D. Fiorani, E. Tronc, Magnetic Relaxation in Fine-Particle Systems, *Adv. Chem. Phys.* 98, 283–494 (1997).
13. S. Laurent, D. Forge, M. Port, A. Roch, C. Robic, L. V. Elst, R. N. Muller, Magnetic Iron Oxide Nanoparticles: Synthesis, Stabilization, Vectorization, Physicochemical Characterizations, and Biological Applications, *Chem. Rev.* 108, 2064–2110 (2008).
14. H. Duan, M. Kuang, X. Wang, Y. A. Wang, H. Mao, S. Nie, Reexamining the Effects of Particle Size and Surface Chemistry on the Magnetic Properties of Iron Oxide Nanocrystals: New Insights into Spin Disorder and Proton Relaxivity, *J. Phys. Chem. C* 112, 8127–8131 (2008).
15. U. I. Tromsdorf, N. C. Bigall, M. G. Kaul, O. T. Bruns, M. S. Nikolic, B. Mollwitz, R. A. Sperling, R. Reimer, H. Hohenberg, W. J. Parak, S. Forster, U. Beisiegel, G. Adam, H. Weller, Size and Surface Effects on the MRI Relaxivity of Manganese Ferrite Nanoparticle Contrast Agents, *Nano Lett.* 7, 2422–2427 (2007).
16. S. Campelj, D. Makovec, M. Drogenik, Functionalization of Magnetic Nanoparticles with 3-Aminopropyl Silane, *J. Magn. Mater.* 321, 1346–1350 (2009).

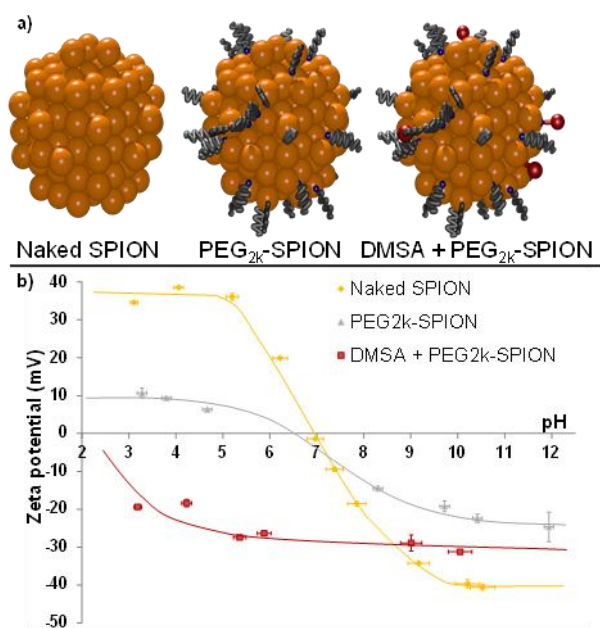
17. C. M. Cheng, G. Kou, X. L. Wang, S. H. Wang, H. C. Gu, Y. J. Guo, Synthesis of Carboxyl Superparamagnetic Ultrasmall Iron Oxide (USPIO) Nanoparticles by a Novel Flocculation-Redispersion Process, *J. Magn. Magn. Mater.* 321, 2663–2669 (2009).
18. T. J. Daou, S. Begin-Colin, J. M. Greneche, F. Thomas, A. Derory, P. Bernhardt, P. Legare, G. Pourroy, Phosphate Adsorption Properties of Magnetite-Based Nanoparticles, *Chem. Mater.* 19, 4494–4505 (2007).
19. L. Maurizi, F. Bouyer, J. Paris, F. Demoisson, L. Saviot, N. Millot, One Step Continuous Hydrothermal Synthesis of Very Fine Stabilized Superparamagnetic Nanoparticles of Magnetite, *Chem. Commun.* 47, 11706–11708 (2011).
20. X. F. Xue, K. Hanna, C. Despas, F. Wu, N. S. Deng, Effect of Chelating Agent on the Oxidation Rate of PCP in the magnetite/H<sub>2</sub>O<sub>2</sub> System at Neutral pH, *J. Mol. Catal. Chem.* 311, 29–35 (2009).
21. L. Maurizi, U. Sakulkhu, L. A. Crowe, V. M. Dao, N. Leclaire, J.-P. Vallée, H. Hofmann, Syntheses of Cross-Linked Polymeric Superparamagnetic Beads with Tunable Properties, *RSC Adv.* 4, 11142–11146 (2014).
22. M. Chanana, Z. Mao, D. Wang, Using Polymers to Make Up Magnetic Nanoparticles for Biomedicine, *J. Biomed. Nanotechnol.* 5, 652–668 (2009).
23. M. D. Butterworth, L. Illum, S. S. Davis, Preparation of Ultrafine Silica- and PEG-Coated Magnetite Particles, *Colloids Surf. -Physicochem. Eng. Asp.* 179, 93–102 (2001).
24. A. Petri-Fink, M. Chastellain, L. Juillerat-Jeanneret, A. Ferrari, H. Hofmann, Development of Functionalized Superparamagnetic Iron Oxide Nanoparticles for Interaction with Human Cancer Cells, *Biomaterials* 26, 2685–2694 (2005).
25. S. Nayar, A. Guha, A Novel Biomimetic Material—Glycine-PVA Ferrofluid That Crosses the Blood-Brain Barrier, *J. Biomed. Nanotechnol.* 7, 78–78 (2011).
26. L. Kaufner, R. Cartier, R. Wuestneck, I. Fichtner, S. Pietschmann, H. Bruhn, D. Schuett, A. F. Thuenemann, U. Pison, Poly(ethylene Oxide)-Block-Poly(glutamic Acid) Coated Magnetite Nanoparticles: In Vitro Characterization and in Vivo Behaviour, *Nanotechnology* 18 (2007).
27. P. Reimer, B. Tombach, Hepatic MRI with SPIO: Detection and Characterization of Focal Liver Lesions, *Eur. Radiol.* 8, 1198–1204 (1998).
28. V. Dousset, L. Ballarino, C. Delalande, M. Coussemaq, P. Canioni, K. G. Petry, J. M. Caille, Comparison of Ultrasmall Particles of Iron Oxide (USPIO)-Enhanced T<sub>2</sub>-Weighted, Conventional T<sub>2</sub>-Weighted, and Gadolinium-Enhanced T<sub>1</sub>-Weighted MR Images in Rats with Experimental Autoimmune Encephalomyelitis, *Am. J. Neuroradiol.* 20, 223–227 (1999).
29. A. Bjornerud, L. Johansson, The Utility of Superparamagnetic Contrast Agents in MRI: Theoretical Consideration and Applications in the Cardiovascular System, *Nmr Biomed.* 17, 465–477 (2004).
30. A. J. L. Villaraza, A. Bumb, M. W. Brechbiel, Macromolecules, Dendrimers, and Nanomaterials in Magnetic Resonance Imaging: The Interplay between Size, Function, and Pharmacokinetics, *Chem. Rev.* 110, 2921–2959 (2010).
31. C. Sun, J. S. H. Lee, M. Q. Zhang, Magnetic Nanoparticles in MR Imaging and Drug Delivery, *Adv. Drug Deliv. Rev.* 60, 1252–1265 (2008).
32. E. Duguet, S. Vasseur, S. Mornet, J. M. Devoisselle, Magnetic Nanoparticles and Their Applications in Medicine, *Nanomed.* 1, 157–168 (2006).
33. S. M. Moghimi, A. C. Hunter, J. C. Murray, Long-Circulating and Target-Specific Nanoparticles: Theory to Practice, *Pharmacol. Rev.* 53, 283–318 (2001).
34. C. Corot, P. Robert, J. M. Idee, M. Port, Recent Advances in Iron Oxide Nanocrystal Technology for Medical Imaging, *Adv. Drug Deliv. Rev.* 58, 1471–1504 (2006).

35. A. C. Faure, S. Dufort, V. Jossierand, P. Perriat, J. L. Coll, S. Roux, O. Tillement, Control of the in Vivo Biodistribution of Hybrid Nanoparticles with Different Poly(ethylene Glycol) Coatings, *Small* 5, 2565–2575 (2009).
36. G. Ge, H. Wu, F. Xiong, Y. Zhang, Z. Guo, Z. Bian, J. Xu, C. Gu, N. Gu, X. Chen, D. Yang, The Cytotoxicity Evaluation of Magnetic Iron Oxide Nanoparticles on Human Aortic Endothelial Cells, *Nanoscale Res. Lett.* 8, 215 (2013).
37. C. Barrera, A. P. Herrera, C. Rinaldi, Colloidal Dispersions of Monodisperse Magnetite Nanoparticles Modified with Poly(ethylene Glycol), *J. Colloid Interface Sci.* 329, 107–113 (2009).
38. C. Oh, Y. G. Lee, T. S. Choi, C. U. Jon, S. G. Oh, Facile Synthesis of PEG-Silica Hybrid Particles Using One-Step Sol-Gel Reaction in Aqueous Solution, *Colloids Surf. - Physicochem. Eng. Asp.* 349, 145–150 (2009).
39. F. Bertorelle, C. Wilhelm, J. Roger, F. Gazeau, C. Menager, V. Cabuil, Fluorescence-Modified Superparamagnetic Nanoparticles: Intracellular Uptake and Use in Cellular Imaging, *Langmuir* 22, 5385–5391 (2006).
40. L. Maurizi, H. Bisht, F. Bouyer, N. Millot, Easy Route to Functionalize Iron Oxide Nanoparticles via Long-Term Stable Thiol Groups, *Langmuir* 25, 8857–8859 (2009).
41. R. Massart, E. Dubois, V. Cabuil, E. Hasmonay, Preparation and Properties of Monodisperse Magnetic Fluids, *J. Magn. Magn. Mater.* 149, 1–5 (1995).
42. J. Boudon, J. Paris, Y. Bernhard, E. Popova, R. A. Decreau, N. Millot, Magneto-Optical Nanomaterial: A SPIO-Phthalocyanine Scaffold Built Step-by-Step towards Bimodal Imaging, *Chem. Commun.* 49, 7394–7396 (2013).
43. S. Jo, K. Park, Surface Modification Using Silanated Poly(ethylene Glycol)s, *Biomaterials* 21, 605–616 (2000).
44. C. Flesch, M. Joubert, E. Bourgeat-Lami, S. Mornet, E. Duguet, C. Delaite, P. Dumas, Organosilane-Modified Maghemite Nanoparticles and Their Use as Co-Initiator in the Ring-Opening Polymerization of Epsilon-Caprolactone, *Colloids Surf. -Physicochem. Eng. Asp.* 262, 150–157 (2005).
45. R. Shen, P. H. C. Camargo, Y. Xia, H. Yang, Silane-Based Poly(ethylene Glycol) as a Primer for Surface Modification of Nonhydrolytically Synthesized Nanoparticles Using the Stober Method, *Langmuir* 24, 11189–11195 (2008).
46. P. Eyer, F. Worek, D. Kiderlen, G. Sinko, A. Stuglin, V. Simeon-Rudolf, E. Reiner, Molar Absorption Coefficients for the Reduced Ellman Reagent: Reassessment, *Anal. Biochem.* 312, 224–227 (2003).
47. A.-L. Papa, L. Maurizi, D. Vandroux, P. Walker, N. Millot, Synthesis of Titanate Nanotubes Directly Coated with USPIO in Hydrothermal Conditions: A New Detectable Nanocarrier, *J. Phys. Chem. C* 115, 19012–19017 (2011).
48. K. H. Diehl, R. Hull, D. Morton, R. Pfister, Y. Rabemampianina, D. Smith, J. M. Vidal, C. van de Vorstenbosch, European Federation of Pharmaceutical Industries Association and European Centre for the Validation of Alternative Methods, A Good Practice Guide to the Administration of Substances and Removal of Blood, Including Routes and Volumes, *J. Appl. Toxicol. JAT* 21, 15–23 (2001).
49. D. W. Yang, Q. Ye, M. Williams, Y. Sun, T. C. C. Hu, D. S. Williams, J. M. F. Moura, C. Ho, USPIO-Enhanced Dynamic MRI: Evaluation of Normal and Transplanted Rat Kidneys, *Magn. Reson. Med.* 46, 1152–1163 (2001).
50. D. E. Sosnovik, M. Nahrendorf, R. Weissleder, Magnetic Nanoparticles for MR Imaging: Agents, Techniques and Cardiovascular Applications, *Basic Res. Cardiol.* 103, 122–130 (2008).
51. N. Fauconnier, A. Bee, J. Roger, J. N. Pons, Synthesis of Aqueous Magnetic Liquids by Surface Complexation of Maghemite Nanoparticles, *J. Mol. Liq.* 83, 233–242 (1999).

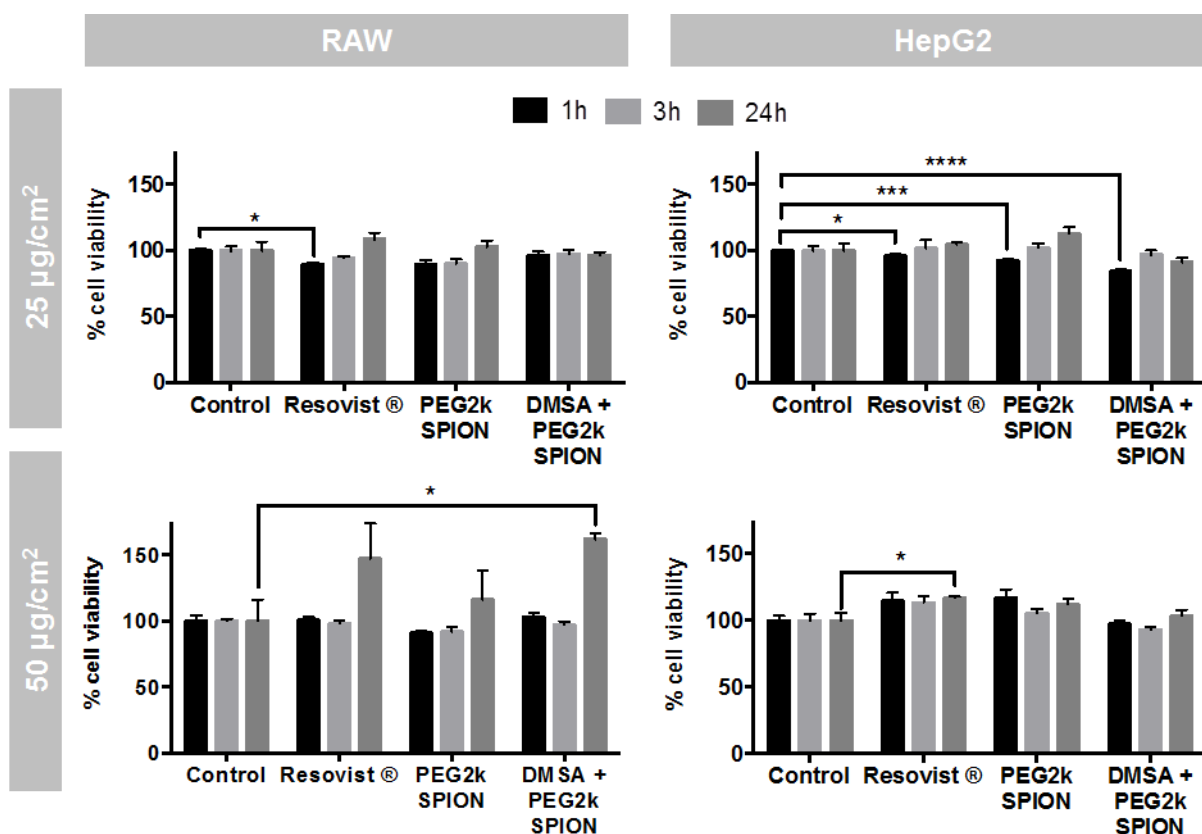
52. Y. W. Jun, Y. M. Huh, J. S. Choi, J. H. Lee, H. T. Song, S. Kim, S. Yoon, K. S. Kim, J. S. Shin, J. S. Suh, J. Cheon, Nanoscale Size Effect of Magnetic Nanocrystals and Their Utilization for Cancer Diagnosis via Magnetic Resonance Imaging, *J. Am. Chem. Soc.* 127, 5732–5733 (2005).
53. Y. W. Jun, J. W. Seo, A. Cheon, Nanoscaling Laws of Magnetic Nanoparticles and Their Applicabilities in Biomedical Sciences, *Acc. Chem. Res.* 41, 179–189 (2008).
54. B. A. Larsen, M. A. Haag, N. J. Serkova, K. R. Shroyer, C. R. Stoldt, Controlled Aggregation of Superparamagnetic Iron Oxide Nanoparticles for the Development of Molecular Magnetic Resonance Imaging Probes, *Nanotechnology* 19 (2008).
55. A. Roch, Y. Gossuin, R. N. Muller, P. Gillis, Superparamagnetic Colloid Suspensions: Water Magnetic Relaxation and Clustering, *J. Magn. Magn. Mater.* 293, 532–539 (2005).
56. L. E. W. LaConte, N. Nitin, O. Zurkiya, D. Caruntu, C. J. O'Connor, X. P. Hu, G. Bao, Coating Thickness of Magnetic Iron Oxide Nanoparticles Affects R2 Relaxivity, *J. Magn. Reson. Imaging* 26, 1634–1641 (2007).
57. T. Schlorf, M. Meincke, E. Kossel, C.-C. Glüer, O. Jansen, R. Mentlein, Biological Properties of Iron Oxide Nanoparticles for Cellular and Molecular Magnetic Resonance Imaging, *Int. J. Mol. Sci.* 12, 12–23 (2010).
58. A. P. Khandhar, R. M. Ferguson, J. A. Simon, K. M. Krishnan, Tailored Magnetic Nanoparticles for Optimizing Magnetic Fluid Hyperthermia, *J. Biomed. Mater. Res. A* 100, 728–737 (2012).
59. A. P. Khandhar, R. M. Ferguson, H. Arami, K. M. Krishnan, Monodisperse Magnetite Nanoparticle Tracers for in Vivo Magnetic Particle Imaging, *Biomaterials* 34, 3837–3845 (2013).
60. M. F. Calmon, A. T. de Souza, N. M. Candido, M. I. B. Raposo, S. Taboga, P. Rahal, J. G. Nery, A Systematic Study of Transfection Efficiency and Cytotoxicity in HeLa Cells Using Iron Oxide Nanoparticles Prepared with Organic and Inorganic Bases, *Colloids Surf. B Biointerfaces* 100, 177–184 (2012).
61. M. Goutayer, S. Dufort, V. Jossierand, A. Royere, E. Heinrich, F. Vinet, J. Bibette, J. L. Coll, I. Texier, Tumor Targeting of Functionalized Lipid Nanoparticles: Assessment by in Vivo Fluorescence Imaging, *Eur. J. Pharm. Biopharm.* 75, 137–147 (2010).
62. Y. Okuhata, Delivery of Diagnostic Agents for Magnetic Resonance Imaging, *Adv. Drug Deliv. Rev.* 37, 121–137 (1999).
63. S. Mornet, S. Vasseur, F. Grasset, P. Veverka, G. Goglio, A. Demourgues, J. Portier, E. Pollert, E. Duguet, Magnetic Nanoparticle Design for Medical Applications, *Prog. Solid State Chem.* 34, 237–247 (2006).
64. L. Maurizi, U. Sakulkhu, A. Gramoun, J.-P. Vallee, H. Hofmann, A Fast and Reproducible Method to Quantify Magnetic Nanoparticle Biodistribution, *Analyst* 139, 1184–1191 (2014).



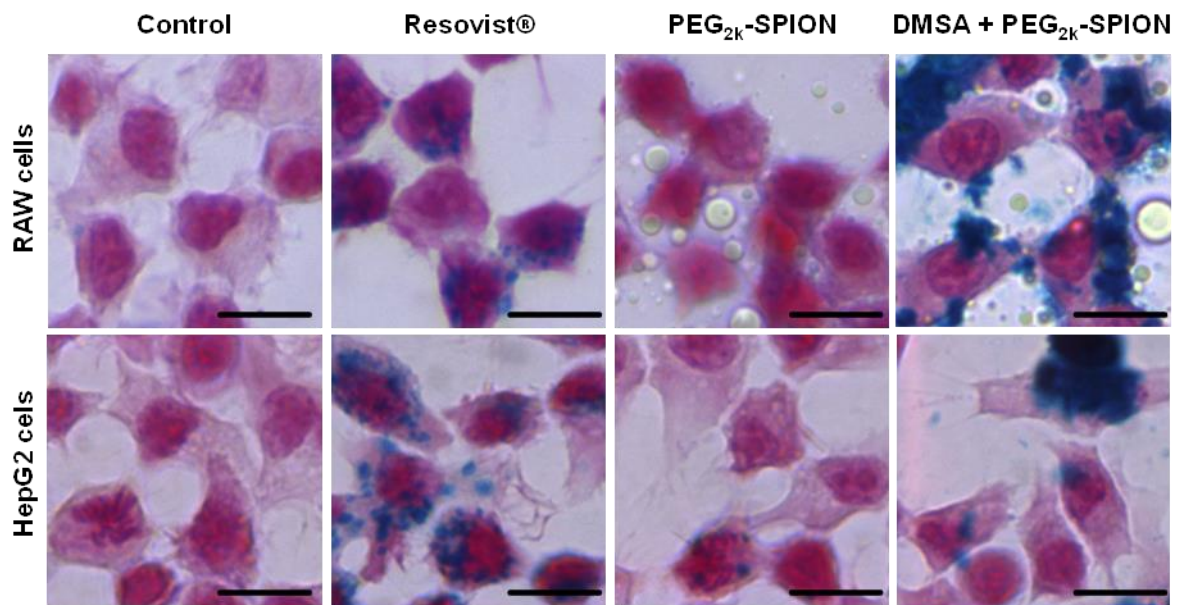
## Figures



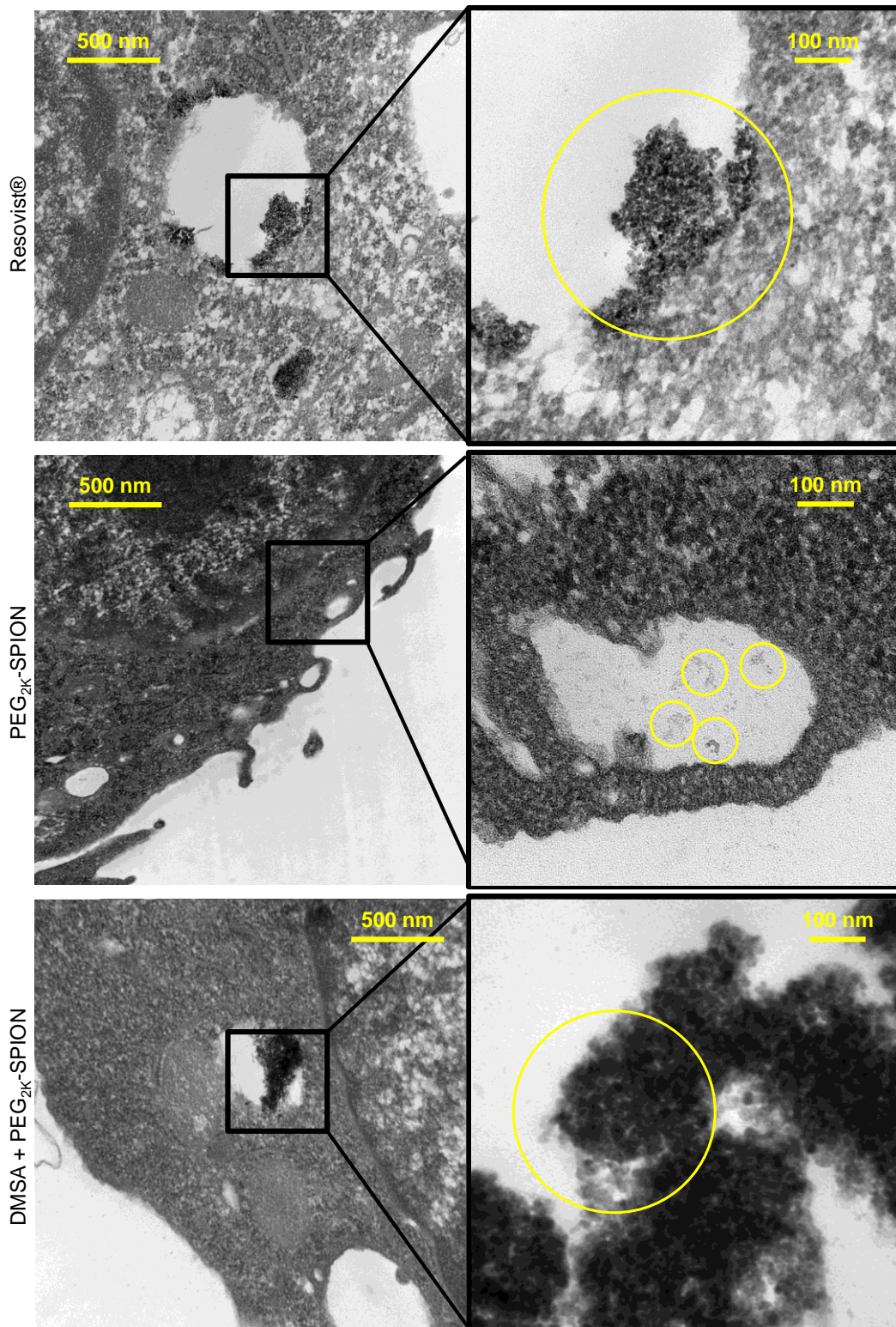
**Figure 1:** a) Representation of the three types of synthesized SPION (shown in orange). From left to right: naked SPION, SPION coated with PEG<sub>2k</sub> (shown in grey), and SPION coated with PEG<sub>2k</sub> (shown in grey) + DMSA (shown in red); b)  $\zeta$ -potential at the indicated pH for the three types of SPION, measured in  $10^{-2}$  M NaCl solution.



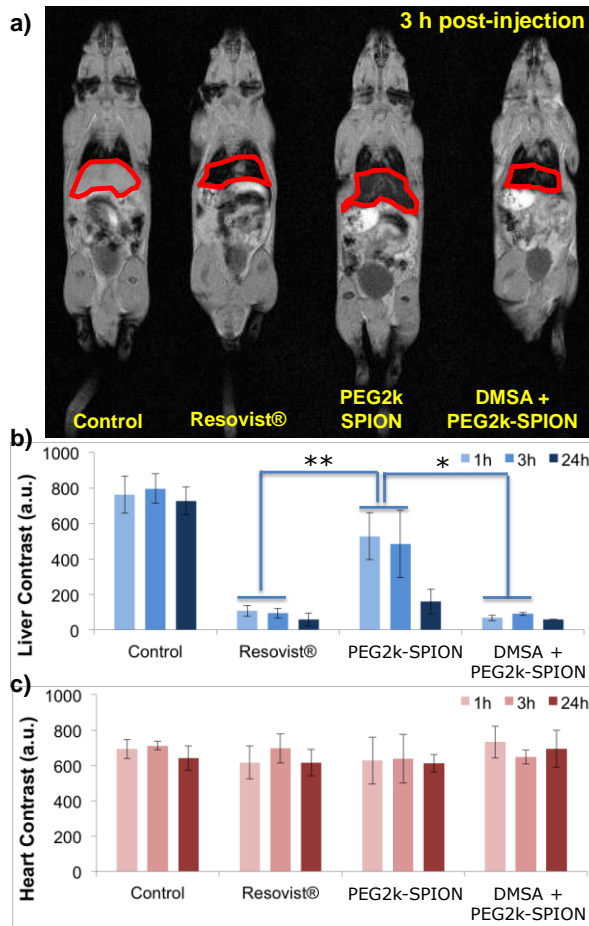
**Figure 2:** Cytotoxicity results from MTT assays performed on RAW 264.7 and HepG2 cells with Resovist®, PEG<sub>2k</sub>-SPION, and PEG<sub>2k</sub>-SPION + DMSA after 1-, 3-, and 24-h incubations. The results are given as a percentage of the control cells, incubated with medium only. Concentration used: 25 and 50 µg<sub>Fe</sub>.cm<sup>-2</sup>. \*p ≤ 0.05, \*\*p ≤ 0.01, \*\*\*p ≤ 0.001, \*\*\*\*p ≤ 0.001, for the comparison with control cells.



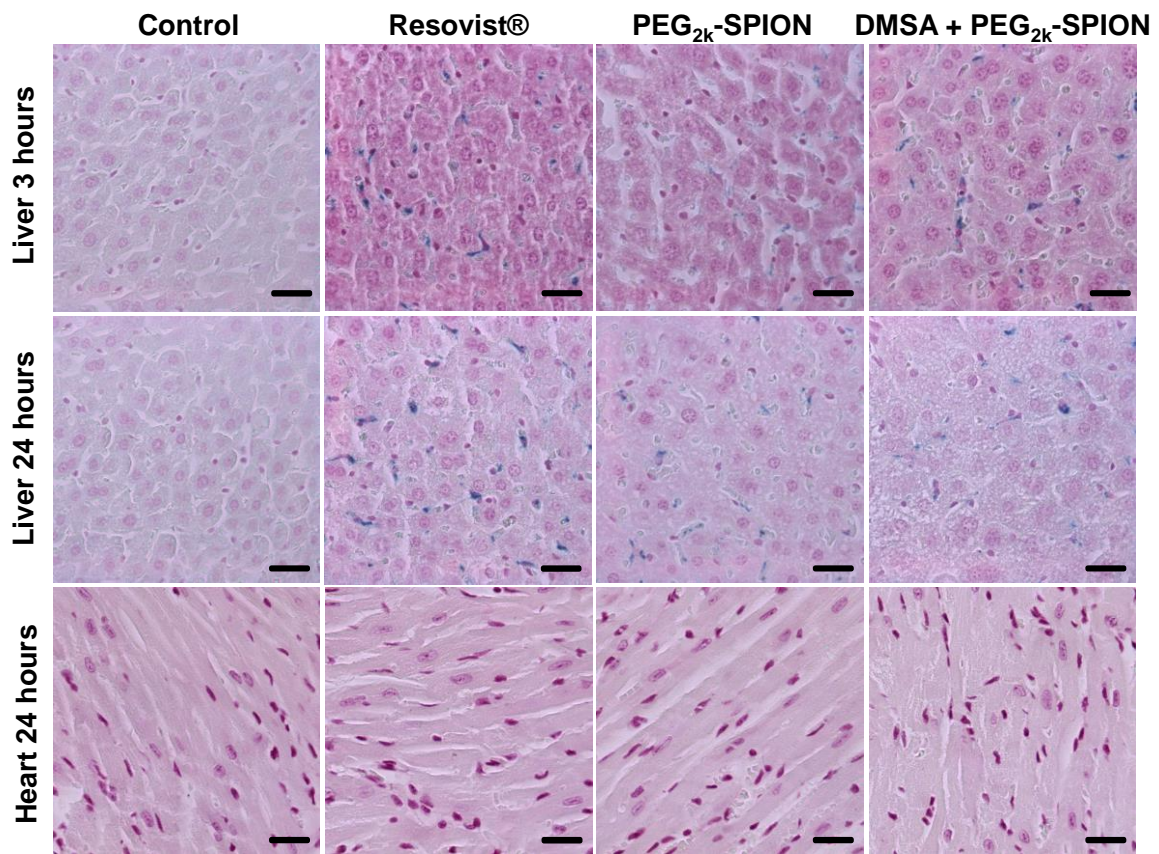
**Figure 3:** Representative optical microscope images of RAW 264.7 and HepG2 cells stained with Prussian blue after incubation with: complete medium (Control), Resovist®, PEG<sub>2k</sub>-SPION, or DMSA + PEG<sub>2k</sub>-SPION. Blue color represents the presence of iron from dissolved SPIONs. Concentration used: 25  $\mu\text{g}_{\text{Fe}}\cdot\text{cm}^{-2}$ . Scale bar: 10  $\mu\text{m}$ .



**Figure 4:** Representative Transmission Electron Microscope images of RAW 264.7 cells incubated with: Resovist®, PEG<sub>2k</sub>-SPION, or DMSA + PEG<sub>2k</sub>-SPION, showing 2 magnifications per condition. The yellow-circled darkest points indicate the presence of SPIONs. Concentration used: 25  $\mu\text{g}_{\text{Fe}}/\text{cm}^2$ .



**Figure 5:** a) Coronal magnetic resonance images (MRI) of mice 3 h after injection of (from left to right): physiological saline (Control), Resovist®, PEG<sub>2k</sub>-SPION, or DMSA + PEG<sub>2k</sub>SPION; b) Relative MRI contrasts measured in livers from mice injected with physiological saline (Control), Resovist®, PEG<sub>2k</sub>-SPION, or DMSA + PEG<sub>2k</sub>-SPION at 1, 3, and 24 h after injection; c) Relative MRI contrasts measured in hearts from mice injected with physiological saline (Control), Resovist®, PEG<sub>2k</sub>-SPION, or DMSA + PEG<sub>2k</sub>-SPION at 1, 3, and 24 h after injection. The red line in part a) surrounds the liver. Concentration used: 5.6 mg<sub>Fe</sub>.kg<sup>-1</sup>. MRI pictures at T<sub>E</sub>: 2.9 ms, T<sub>R</sub>: 300 ms, slice thickness: 2 mm, resolution: 0.35 μm<sup>2</sup>. Significances were reported as: \*p ≤ 0.05, \*\*p ≤ 0.01. Administration of all SPIONs significantly altered the liver contrast, compared with the control, at all time-points.



**Figure 6:** Representative optical microscope images of liver and heart slices stained with Prussian blue. Livers were obtained from mice injected with: physiological saline (Control), Resovist®, PEG<sub>2k</sub>-SPION, or DMSA + PEG<sub>2k</sub>-SPION and sacrificed 3 and 24 h after injection. Hearts were obtained from mice injected with: physiological saline (Control), Resovist®, PEG<sub>2k</sub>-SPION, or DMSA + PEG<sub>2k</sub>-SPION and sacrificed 24 h after injection. Blue spots indicate the presence of iron from dissolved SPIONs. Concentration used: 5.6 mg<sub>Fe</sub>.kg<sup>-1</sup> mouse. Scale bar: 10 μm.

**Table 1:** Physicochemical properties of Resovist®, naked SPIONs, SPIONs functionalized with PEG<sub>2k</sub>, and PEG<sub>2k</sub> + DMSA.

Nanoparticles	d <sub>TEM</sub> (nm)	d <sub>H</sub> (nm)	ξ potential at pH 7 (mV)	S <sub>BET</sub> m <sup>2</sup> .g <sup>-1</sup>	Molec.nm <sup>-2</sup>	r <sub>2</sub> relaxivity (mM <sup>-1</sup> <sub>Fe</sub> .s <sup>-1</sup> )
Resovist®	8 ± 2	22 ± 4	-22 ± 2	n.d.	n.d.	260 ± 14
Naked SPION	9 ± 3	23 ± 1	0 ± 3	110	0	285 ± 12
PEG <sub>2k</sub> -SPION	9 ± 3	38 ± 5	-3 ± 3	110	3.3 PEG	260 ± 9
DMSA + PEG <sub>2k</sub> -SPION	9 ± 3	36 ± 4	-30 ± 3	110	0.6 PEG 0.3 DMSA	297 ± 7

Molec: molecule ; n.d.: not determined