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# **Tyrosine metabolism: identification of a key residue in the acquisition of prephenate**  aminotransferase activity by  $1\beta$  aspartate aminotransferase.

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Running title: *Signature of prephenate competent 1 AAT*

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**Abreviation**: AAT Aspartate aminotransferase, PAT Prephenate aminotransferase

#### **ABSTRACT**

Alternative routes for the post-chorismate branch of the biosynthetic pathway leading to tyrosine exist, the 4-hydroxyphenylpyruvate or the arogenate route. The arogenate route involves the transamination of prephenate into arogenate. In a previous study, we found that, depending on the microorganisms possessing the arogenate route, three different aminotransferases evolved to perform prephenate transamination *i.e.* 18 aspartate aminotransferase (18 AAT), *N*-succinyl-L,L-diaminopimelate aminotransferase, and branched-chain aminotransferase. The present work aimed at identifying molecular determinant(s) of 1 $\beta$  AAT prephenate aminotransferase (PAT) activity. To that purpose, we conducted Xray crystal structure analysis of two PAT competent 1 AAT from *Arabidopsis thaliana* and *Rhizobium meliloti*, and one PAT incompetent 1<sub>8</sub> AAT from *Rhizobium meliloti*. This structural analysis supported by site directed mutagenesis, modelling and molecular dynamics simulations allowed us to identify a molecular determinant of PAT activity in the flexible N-terminal loop of  $1\beta$  AAT. Our data reveal that a Lys/Arg/Gln residue in position 12 in the sequence (numbering according to *Thermus thermophilus* 1 AAT), present only in PAT competent enzymes, could interact with the 4-hydroxyl group of the prephenate, substrate and thus may have a central role in the acquisition of PAT activity by  $1\beta$  AAT.

The aromatic amino acid biosynthesis pathway occurs in microorganisms, fungi and plants, and is not found in animals. This pathway links the metabolism of carbohydrates to the biosynthesis of aromatic amino acids involved in protein synthesis, and leads to the synthesis of a large diversity of aromatic secondary metabolites [1-3]. The aromatic amino acid biosynthesis pathway plays a central role for the growth of these organisms and their interaction with environment. Phenylalanine and tyrosine also represent an important source of high-value aromatic natural compounds for industry, and human health.

Two alternative routes leading from chorismate to tyrosine and phenylalanine exist: the 4 hydroxyphenylpyruvate route, and the arogenate route (Figure 1). All photosynthetic organisms and many microorganisms synthesize L-tyrosine *via* the arogenate route [4-7]. In addition, most viridiplantae and some microorganism also synthesize L-phenylalanine *via* the arogenate route [8]. In this case, a prephenate aminotransferase transforms prephenate into arogenate, and an arogenate dehydrogenase (EC 1.3.1.43) or an arogenate dehydratase (EC 4.2.1.91) decarboxylates arogenate into L-tyrosine or L-phenylalanine, respectively (Figure 1) [7,8]. Alternatively, some microorganisms and fungi such as *Escherichia coli*, or yeast, first transform prephenate into 4-hydroxyphenylpyruvate by a prephenate dehydrogenase (EC 1.3.1.13) or into penylpuruvate by a prephenate dehydratase (EC 4.2.1.51) [9], and then both are transaminate into L-tyrosine or L-phenylalanine, respectively.

Prephenate aminotransferase (PAT) activity was first identified in plants [10,11]. In *Arabidopsis thaliana,* a specific chloroplastic aspartate aminotransferase is in charge of PAT activity (At2g22250 gene, EC 2.6.1.1, EC 2.6.1.78, and EC 2.6.1.79) [10,11]. This prephenate aminotransferase belongs to aspartate aminotransferase (AAT) subgroup 1 $\beta$  [12]. In this subgroup, the dicarboxylic substrate interacts with a lysine and an arginine (K101 and R361, numbering according to *Thermus thermophilus* 1 $\beta$  AAT enzyme: Q56232 abbreviated as Tth-PAT) instead of two arginines in the  $1\alpha$  AAT subgroup. In a previous study, we showed that the PAT competent 1 $\beta$  AAT from *A. thaliana* (Ath-PAT) catalyzes the transamination of oxaloacetate into aspartate as efficiently as prephenate into arogenate [10].

Historically in microorganisms, arogenate route has been detected in cyanobacteria, actinobacteria, some  $\alpha$ - and  $\beta$ -proteobacteria by the group of Jensen [4-6,13]. In a previous study [14] we reported that, in these arogenate competent microorganisms, PAT activity is carried out by three different aminotransferases, which present two different fold-types. Furthermore, we showed that the capacity to transaminate prephenate is not a general catalytic property of the three classes of aminotransferases, since other members were devoid of detectable PAT activity [14]. These results revealed that the arogenate route emerged independently several times during evolution. To go further, structure-function studies are needed to identify molecular signature(s) of the competent prephenate/arogenate aminotransferases, in order to investigate their presence in genomic databases.

In the present study, by combining X-ray crystal structure analysis of two PAT competent  $1\beta$  AAT from *Arabidopsis thaliana* and *Rhizobium meliloti*, and one PAT incompetent 1 AAT from *Rhizobium meliloti*, site directed mutagenesis, modelling and molecular dynamics simulations, we show that K/R/Q12 residue (numbering according to *T. thermophilus*  $1\beta$  AAT) is a signature of the PAT function of  $1\beta$  AAT. It is present in the N-terminal flexible loop only in PAT competent 1 $\beta$ -AAT and our data strongly support its possible role in stabilizing prephenate by interacting with its 4-hydroxy group.

#### **Results and discussion**

#### **Resolution of 3D structures of PAT competent and PAT incompetent 1 AAT.**

In our previous study [14], of the two 1 $\beta$  AATs from *Rizobium meliloti* tested, one (Q02635) showed a good prephenate aminotransferase activity while the other (P58350) had no detectable prephenate aminotransferase activity under our test conditions.

We thus decided to identify molecular determinant(s) involved in the acquisition of prephenate aminotransferase function by the  $18$  AAT through a comparative X ray structural study of PAT competent and PAT incompetent 1B AAT. To do this, we solved the three-dimensional structures of PAT competent from *Arabidopsis thaliana* (Ath\_Q9SIE1, Ath-PAT) and from *Rizobium meliloti* (Rme\_Q02635, Rme-PAT), and the PAT incompetent 1 $\beta$  AAT from *Rizobium meliloti* (Rme\_P58350, Rme-AAT).

#### *Overall structures*

The 3D structures of PAT competent 1 $\beta$  AAT Ath-PAT, and Rme-PAT and PAT incompetent 1 $\beta$  AAT Rme-AAT show homodimeric proteins, covalently bound to pyridoxal phosphate (PLP) and sharing a common fold (Figure 2). The monomers are related by a non-crystallographic 2-fold symmetry axis. Residues in or preceding the first  $\alpha$ -helix are observed in different positions when monomers of Ath-PAT are compared, suggesting flexibility. A citrate molecule is bound to the catalytic site of 3 monomers of Ath-PAT (two dimers are in the asymmetric unit). It is stabilized by interacting with Lys122, Trp146, Asn196, Arg398 side chains, the Gly60 NH group and water molecules. In Rme-PAT, pyridoxal phosphate cofactor is not covalently bound to the protein probably as a consequence of radiation damages.

#### *Structure comparisons*

The C $\alpha$  coordinates rms deviations calculated between all the monomers are low indicating that the structures are globally very similar. The largest difference is observed when Rme-AAT is compared with Ath-PAT (1.56 Å) or Tth-PAT (PDB code 1BKG) (1.63 Å). Dimer comparison after superimposing monomers shows moderate differences in the positioning of monomers. The catalytic site is large and solvent accessible and residues from the two monomers participate to the catalytic pockets. Most of the residues observed in the Ath-PAT catalytic site are conserved in Rme-PAT, Rme-AAT and Tth-PAT. From structure comparison with Tth-PAT, residues from the first  $\alpha$ -helix (Lys33(A), Pro34(A), Ser35(A), Lys36(A), Thr37(A), Met38(A) in Ath-PAT [15,16]) are also involved in the closing of the catalytic site upon ligand binding. Lys259 of Ath-PAT is covalently bound to PLP through an aldimine link. The interactions observed between maleate and residues of the catalytic site in Tth-PAT could also take place in Ath-PAT. Maleate observed in the Tth-PAT catalytic site is superimposable to citrate observed in Ath-PAT, but the position of the first  $\alpha$ -helix is different since the Ath-PAT catalytic site is in open conformation.

3D X ray structures of PAT competent 1 $\beta$  AAT Ath-PAT and Rme-PAT, and PAT incompetent 1 $\beta$ AAT Rme-AAT, present a very similar organization. Comparison of these three 3D structures did not reveal a clear molecular determinant of PAT activity, most certainly because they are in an open conformation. Indeed, during the catalytic reaction, the keto acid substrate (oxaloacetate or prephenate) binds to a PMPenzyme complex and a short N-terminal flexible loop move to close the active site [15,16]. Only a closed conformation of prephenate-PMP-enzyme complex is relevant for structural comparison of PAT competent and incompetent  $1\beta$  AAT. Unfortunately, all our attempts to obtain either a PMP-prephenate- or PMPmaleate-enzyme complex of Rme-PAT or Ath-PAT failed.

However, the observation that the  $1\beta$  AAT from *Thermus thermophilus* is PAT competent ([17], this study) allowed us to use its 3D structure in a closed conformation, in complex with PMP and maleate (PDB code 1BKG), resolved by Nakai *et al*. [15], to model the binding of prephenate in the active site.

#### **Identification of a potential molecular determinant of PAT capacity of 1 AAT**

Taking advantage of molecular elucidation of enzyme/ligand interactions in PAT competent  $1\beta$  AAT from *T. thermophilus* by Nakai et al [15], we manually replaced the substrate analogue maleate with prephenate in the active site of the 3D structure 1BKG. Prephenate differs from oxaloacetate and ketoglutarate, the two ketoacid substrates of aspartate aminotransferase, by the presence of its 2,5 cyclohexadiene-4-hydroxy group (Figure 3A). The idea is therefore to identify potential residue(s) present only in the PAT competent 1 $\beta$  AAT that could interact with the 2.5-cyclohexadiene-4-hydroxy group of prephenate. We placed the two prephenate carboxylate groups in interaction with Lys101 and Arg361 while its keto group points to the PMP NH2 group and the catalytic lysine, Lys234 (Figure 3B). In this position, the 2,5- cyclohexadiene-4-hydroxy group of the prephenate substrate points toward the entrance of the active site, its 4-hydroxyl facing the flexible N-terminal loop 10AMKPSA15, which is known to move and close the active site while the substrate binds to the active site [15,16]. In that context, it is interesting to note that the alignment of the N-terminal sequences of already characterized PAT competent 1 $\beta$  AAT (*Arabidopsis thaliana* [10], *Sorghum bicolor* [18], *R. meliloti*, *Rhodobacter sphaeroides* [14], *T. thermophilus* [17] (this study) and *Chlorobium tepidum* [17]) with those of the already characterized PAT incompetent 1 $\beta$  AAT (*R. meliloti, Streptomyces avermitilis* and *Synechocystis* [14]) (Figure 4), reveals that a lysine residue belonging to this flexible loop, Lys12 (numbering according to *T. thermophilus*  $1\beta$  AAT), is present only in the PAT competent  $1\beta$  AAT or is replaced by an arginine in all Poaceae like *S. bicolor* or *Zea mays* or a glutamine in some Proteobacteria like *Rhodobacter sphaeroides* and in Chlorobi like *Chlorobium tepidum*. Each of these three amino acids has the ability to stabilize prephenate in the active site by interacting with its 4-hydroxyl group. In contrast, in PAT incompetent  $1\beta$  AAT, Lys12 is replaced by a glycine, a valine or a serine. None of the later three can play the same role.

However, in the ternary maleate-PMP-enzyme complex (1BKG), the Lys12 side chain points outside the active site [15] (Figure 3). We hypothesize that in the presence of prephenate, the Lys12 side chain rotates to interact with the prephenate cyclohexadiene-4-hydroxy group during closure of the active site, thereby stabilizing the prephenate within the active site.

## **Lys12 plays a specific role in prephenate binding by PAT competent 1 AAT from** *R. meliloti* **and** *T. thermophilus*

The potential interaction of Lys12 with the prephenate 4-hydroxyl group was first addressed by using a site directed mutagenesis approach performed on PAT competent 1 $\beta$  AAT from *R. meliloti* and *T. thermophilus*. We replaced the lysine of the <sub>10</sub>AMKPSA<sub>15</sub> motif by a glycine. Steady-state kinetic properties of wild‐type and K12G mutants were determined with a fixed concentration (25mM) of Glutamate as amino donor and variable concentrations of either prephenate or oxaloacetate. Modification of the apparent Michaelis-Menten constant  $(K_M)$  will reflect a modification in the affinity of the 1 $\beta$  AAT K12G mutants for the substrate whereas modification of the apparent catalytic constant  $(k_{cat})$  will reflect an impact of the mutation on the reaction rate. If our assumption was correct we anticipate that the mutation will have a significant impact on the affinity for prephenate, much less on the affinity for oxaloacetate, and will have minor impact on the reaction rate of both oxaloacetate and prephenate aminotransferase activity. Apparent  $K_M$  for prephenate was found 7 time higher in the K12G mutant of PAT competent 1 $\beta$  AAT from *R. meliloti* (Q02635) (K12G:  $692\pm143$  µM versus WT:  $106\pm12$ µM). In the other hand, apparent  $K_M$  for oxaloacetate was only weakly affected in K12G mutant (Table 3). Furthermore, the apparent catalytic constants of the two aminotransferase reactions  $(k_{cat})$  were not affected.

The effect of the K12G mutation was also tested on the PAT activity of the *T. thermophilus* 1 $\beta$  AAT (Q56232). The K12G mutation gave similar results with a 10 time increase of the apparent  $K_M$  for prephenate (K12G: 2332 $\pm$ 204  $\mu$ M versus WT: 150 $\pm$  8.7  $\mu$ M) and no effect on the apparent  $K_M$  for oxaloacetate (WT:  $25.3\pm1.8 \mu$ M versus K12G:  $26.4\pm2.3 \mu$ M). It should be noted that in our conditions, the *T. thermophilus* 1 $\beta$  AAT apparent catalytic activity is slightly affected by the K12G mutation ( $k_{\text{cat}}$ ) oxaloacetate WT:  $32.2 \pm 4.6 \text{ s}^{-1}$  versus K12G:  $12.3 \pm 3.7 \text{ s}^{-1}$ ). Prephenate aminotransferase apparent catalytic activity seems less affected by the mutation ( $k_{cat}$  for prephenate WT: 7.7 $\pm$ 1.6 s<sup>-1</sup> versus K12G: 4.8  $\pm$  2.1 s<sup>-</sup> <sup>1</sup>), although the strong degradation of affinity for prephenate of the K12G mutant makes the determination of its kinetic parameters less accurate. Steady state kinetics analysis confirms thus that in both enzymes K12G mutation induces a specific decrease in affinity for prephenate with limited effect on their prephenate aminotransferase reaction rate.

In our previous study [14], the two incompetent PATs of *R. meliloti* (P58350), or *Synechocystis* (Q55128) which respectively have a G or a V instead of K12, were found to have no PAT activity at prephenate concentrations up to 2.5 mM, by our coupled test. However, in the present study, the two K12G mutants maintain prephenate aminotransferase activity at high prephenate concentrations since the K12G mutation does not affect their reaction rate.

In their study, Dornfield et al  $[17]$  propose that two highly conserved residues among 1 $\beta$  AAT, T84 and K169, (numbering according to A. thaliana 1 $\beta$  AAT, T16 and L101 according to *T. thermophilus* 1 $\beta$  AAT) (Figure 3 and 4), are important residues for substrate recognition in PAT competent 1 $\beta$  AAT. Furthermore, during the final stage of writing our manuscript, a new structural study on Arabidopsis PAT competent 1 $\beta$  AAT (At2g22250) from the same group was published in Plant Journal [16]. In this study, the authors proposed that, in addition to T84 and K169, E108 (numbering according to A. thaliana 1 $\beta$  AAT, E40 according to *T. thermophilus*  $1\beta$  AAT) (Figure 3 and 4) plays an important role in the acquisition of PAT competencies by 1 $\beta$  AAT. However, unlike K12, these residues are conserved among all (K169, and E108) or almost all (T84) 1β AAT, regardless their ability to catalyze the transamination of prephenate, or not [14,17]. Therefore, although these residues may have contributed to the evolution of prephenate aminotransferase activity within class 1 $\beta$  AAT and not class 1 $\alpha$  AAT, they can hardly be considered to have a specific role in the prephenate aminotransferase activity of  $1\beta$  AAT. On the contrary, the presence of K/R/Q12 only in PAT competent 1 $\beta$  AAT, as well as the strong and selective impact of the K12G mutation on the binding affinity for prephenate, strongly support the specific role played by Lys12 in the binding of prephenate, most likely by its stabilization at the active site through interaction with its 4 hydroxyl group.

A clear support to the specific role played by Lys12 in the binding affinity for prephenate was afforded by Nature through the case of *Synechocystis* Q55128 and *Synechococcus* Q3AK33 1 $\beta$  AAT. As already mentioned *Synechocystis* O55128 16 AAT have a V in place of a K, and was found inactive in our coupled assay [14]. However, in their study Dornfield et al [17] were able to detect traces of arogenate by HPLC in their end point test [17]. They reported a weak activity with an apparent  $K_M$  for prephenate in the range of 2600  $\mu$ M (in the range of the apparent  $K_M$  for prephenate of the *T. thermophilus* 1 $\beta$  AAT K12G mutant). Blast search revealed that the Cyanobacteria *Synechococcus* Q3AK33 1 $\beta$  AAT, which is a close relative to *Synechocystis* Q55128 1 $\beta$  AAT, have a K in place of a V (Figure 5A). In order to compare the catalytic properties of these two  $1\beta$  AAT, we have cloned, expressed in *E. coli* and partially purified the recombinant Q3AK33 1 AAT (Figure 5B). In our conditions, in contrast with *Synechocystis* Q55128 [14], a weak prephenate activity could be detected with the recombinant *Synechococcus* O3AK33 1B AAT by our coupled assay. Kinetic characterization of prephenate aminotransferase activity of recombinant Q3AK33 (Figure 5C) reveals that its apparent  $K_M$  for prephenate was found in the range of  $207\pm23$  µM µM versus 2600 μM for *Synechocystis* Q55128 1β AAT [17], nicely confirming that the presence of K12 strongly

improves binding affinity for prephenate. Our data also indicate that other residues that remains to identify are responsible for the high reaction rate exhibited by all tested  $1\beta$  AAT-PAT from higher plants [10 11] and by some 1 $\beta$  AAT from bacteria like *Rhizobium meliloti* (Q02635) [14] or *Chlorobium tepidum* (Q8KDS8) [17].

#### **Structural evidence for Lys12 involvement in prephenate stabilization within the active site**

We wanted to confirm, at the structural level, the specific role of Lys12 pointed out by the structure comparisons of PAT competent and incompetent 1B AAT, and by kinetics analysis of K12G mutants. Unfortunately, as already mentioned, all our attempts to obtain either a PMP-prephenate-enzyme complex of PAT competent 1 $\beta$  AAT from *A. thaliana, R. meliloti* or *T. thermophilus* failed, and at that time the only available 3D structure of a PAT competent 1B AAT in a closed conformation was the PMP-maleate-enzyme complex of 1AAT from *T. thermophilus*. (PDB code 1BKG), previously solved by Nakai *et al*. [15]. We thus used this 3D structure as a starting model to perform structural analyses *via* modeling and molecular dynamics simulations of the PMP-prephenate-enzyme complex to support the interaction between Lys12 and prephenate suggested by our kinetic experiments.

After model building, introduction of sodium counter-ions and solvation in a periodic box of TIP3P water molecules (structure *1BKG\_Preph\_Mod*, see methods), a short equilibration MD simulation of the PAT dimer in the presence of PMP and prephenate was run at 300 K for 20 ns leading to structure *1BKG\_Preph\_MD* (See methods). Starting from a structure with low energy and low Lys12 NZ / prephenate O1 oxygen distance (dK-Prf) of 3.2Å from this equilibration, we have run a QM/MM optimization followed by a calculation of the charges adequate to fit the calculated electrostatic potential around the ligands (PMP and prephenate). In addition to prephenate, PMP and Lys12, the QM part consisted of side chains of Lys101, Trp125, Ser127, Lys234, Arg242, Tyr322, Arg321 from chain A and Tyr64 from chain B. 14 water molecules were also included in this quantum part for a total of 191 QM atoms. All MM atoms belonging to residues within a distance of  $3 \text{ Å}$  of the QM part were allowed to relax during geometry optimizations, and the remaining structure was kept frozen.

The optimized structure of the catalytic site (see materials and methods for details) is shown in Figure 6 and 7. Stabilizing interactions through hydrogen bonds exist between Lys12, Lys101, Trp125, Arg361, Tyr322 and prephenate, between prephenate and PMP and between PMP and Tyr64, Arg242, Tyr322 and Lys234.

The atomic partial charges calculated using the Merz-Kollman routine [20] in Gaussian were directly transferred to the protein structure file describing the MM system for prephenate, PMP and surrounding residues.

#### **Molecular dynamics simulations**

Using this modified structure file, 24 ns constant pressure and temperature CPT (Constant Pressure and Temperature) dynamics were run restarting from structure *1BKG\_Preph\_MD*. The nuclear Overhauser enhancement (NOE) type restraints used to obtain *1BKG\_Preph\_MD* (see materials and methods for details) were kept for 4 ns and then removed for the remaining 20 ns. This distance dK-Prf oscillates between 2.5 and 4.5 Å during the first 20 ns and increases to up to 14 Å when the restraints are released (See Figure 8). Interestingly, this distance drops back down to around 2.8 Å after 42 ns, meaning that without any restraint on the protein, the distance between Lys12 and prephenate is occasionally compatible with the reaction proposed in the experimental section.

#### **Improving conformational sampling**

The previous MD simulation confirms that the N-terminal part of the protein could undergo conformational transitions during substrate binding as shown by the structural study of Nakai *et al* [15], enabling Lys12 to be stabilized in the vicinity of the prephenate. Still, a method to explore more conformational space is to use Replica eXchanging Self-Guided Langevin Dynamics [21]. 16 replicas were created starting from structure *1BKG\_Preph\_Mod* and vacuum RXGLD simulations were run for 10 ns in the presence of restraints (see materials and methods). Coordinates from Met1 to Gln29 (chain A), PMP and prephenate were extracted from the MD trajectory every ps (20000 structures) and grouped into 16 clusters corresponding to coordinate root mean square deviation (rmsd) larger than 3.5Å. From the evolution of cluster population versus time (shown in Figure 9), 6 most populated clusters, visited many times all along the simulation, seem particularly interesting and have been studied further.

## **Final production dynamics from selected clusters**

The coordinates of the lowest energy structure in the 6 selected clusters were computed and one molecular dynamics simulation was started from each resulting minimized structure (see materials and methods). The evolution of dK-Prf is plotted in Figure 10 for the six selected clusters. One particular cluster (number 6) simulated during 85 ns, shows very large fluctuations of dK-Prf from 2.6 to more than 21.6  $\AA$ , demonstrating the high mobility of the N-terminal part of the protein. The corresponding two structures are shown in Figure 11 after superimposition of backbone atoms of residues 33 to 382. Backbone atom coordinate rmsd is 4.18 Å for residues 13 to 32 and 1.09 Å for residues 33 to 382. Between the two structures, conformational changes around Lys12 appear with transitions of the backbone dihedral angle Φ=(Met11)C-N-Cα-Cβ from 180° to 120° and the side chain dihedral angle χ1=N-Cα-Cβ-Cγ from 60° to - 60°, for "far" and "close" conformation, respectively.

## **The presence of Lys/Arg/Gln12 could be considered as a molecular signature of the PAT competent 1 AAT**

Our experimental kinetics data, revealed that Lys12 plays a specific role in the acquisition of PAT activity by *R. meliloti* and *T. thermophilus* 1<sup>8</sup> AAT most certainly by stabilizing prephenate though an interaction with its 4-hydroxyl group. Accordingly, we were able to optimize at the QM level a structure of the active site respecting this restraint. Releasing this restraint during MD simulations results in a rapid drift of dK-Prf followed by frequent returns to dK-Prf lower than  $3 \text{ Å}$ . This means that, although no crystal structure of PAT competent 16 AAT from *T. thermophilus*, or homologues, has been obtained showing proximity between Lys12 and the catalytic site, interactions are possible and even frequent between Lys12 and prephenate, on the short time scale accessible to MD simulations.

These simulations strongly support, at the structural level, our kinetics data. All together our data point lysine12 and the corresponding arginine or glutamine, present in other PAT competent  $1\beta$  AAT, as key residues in the acquisition of a physiologically relevant prephenate aminotransferase activity by 1 $\beta$  AAT during evolution. As already mentioned, our data also revealed that other residues that remain to be identified are responsible for the good reaction rate exhibited by all tested plant and some bacteria 1  $\beta$  AAT PAT.

Phylogenetic distribution of  $1\beta$  AAT among different classes of organisms (Figure 12) confirms the presence of a 1 $\beta$  AAT with a Lys/Arg/Gln in position 12 in all photosynthetic organisms at the exception of most Cyanobacteria, which are known to synthesize arogenate via a BCAT-PAT [14]. The relative distribution of K/R/Q12 reveals the presence of a Lysine residue (corresponding to K12) in the majority of land plants except in the Poaceae (like *Zea maïs* or *Sorghum bicolor*) which possess an arginine. Most green algae also have a Lysine residue, but in the 1 $\beta$  AAT from *Ostreococcus lucimarinus* (A4S857) the lysine is replaced by an arginine, like in Poaceae. It is difficult to have a clear idea of the situation for the brown and the red algae due to the small number of sequenced genomes. Some brown algae like *Sargassum thunbergii* have a 1<sub>8</sub> AAT with a serine residue (AOA097IUX3) and presumably synthesize arogenate with a BCAT-PAT like Cyanobacteria or a N-succinyl-l,l-diaminopimelate aminotransferase like the two tested actinobacteria, *Streptomyces avermitilis* and *Mycobacterium tuberculosis* [14], or do not synthesize tyrosine and phenylalanine via the arogenate route. In the other hand, some brown algae like *Punctaria latifolia* possess a 1 $\beta$  AAT (AOA097IUR7) with an arginine like Poaceae. The few sequenced genomes of red algae seem indicate that they are devoid of 1 $\beta$  AAT, the first hit concern aspartate aminotransferases with a highly divergent N-terminal sequence but with a conserved K101 and a more conserved C-terminal sequence. In contrast, Diatoms which result from secondary endosymbiosis events between an algae and a eukaryote, have either a Lysine like green algae, or a Q like some photosynthetic bacteria, confirming the possible dual origin (green or red algae) of their chloroplasts [19]. Intriguingly, *Nanochloropsis* have a 1 AAT with a Glu in the corresponding 12 position. Chlorobi and Bacteroidetes have a Q, and interestingly the 1 $\beta$  AAT of the symbiotic cyanobacteria *Synechococcus spongiarum* also have à Q and branch with Chlorobi and Bacteroidetes rather than with other Cyanobacteria. Lys12 residues are found in 1 $\beta$  AAT from many non-photosythetic bacteria (most  $\alpha$ -proteobacteria and hadobacteria and some  $\gamma$ -proteobacteria) known to be arogenate competent organisms. In the other hand, most Actinobacteria like *Streptomyces* *avermitilis* have 1 $\beta$  AAT with a Serine12, which is in agreement with their capacity to synthesize arogenate through an S-DAP-PAT [14].

#### **Experimental procedures**

**Bacterials strains-** Two strains of *E. coli* were used in this work: the strain  $DH5\alpha$  (Invitrogen strain; F- $\Phi$ 80lacZ $\Delta M$ 15  $\Delta$  (lacZYA-argF) U169 deoR recA1 endA1 hsdR17 (rk-, mk +) phoA supE44 thi-1 gyrA96 relA1  $\lambda$ -) was used for plasmid DNA amplification and subcloning. The strain BL21 Rosetta 2 (DE3) (Merck; F-ompThsdSB (rB-mB-) gal dcm (DE3) pRARE2 (CamR)) was used for overproduction of recombinant proteins. These two strains were grown at 37°C in Luria-Bertani medium supplemented with 1% agar (w/v) for the agar media, and supplemented with Carbenicillin  $1^{\degree}$  (DH5 $\alpha$ ) or Kanamycine 1<sup>%</sup> (BL21). The 2011 RCR strain of *Rhizobium meliloti* was provided by G. Alloing (UMR6192, Sophia Antipolis). *Thermus thermophilus* (strain HB8) Genomic DNA was provided by A. Atteia (UMR 7281, Marseille). *Synechococcus* CC9605 genomic DNA was provided by F. Partensky (UMR7144, Roscoff)

**Chemicals-** Prephenate, oxaloacetate, and glutamate were purchased from Sigma.

**Primers**- Primers used for amplification of the 1<sub>B</sub> aspartate aminotransferase genes from the corresponding genomic DNA are presented in Table 4. Restriction sites introduced by PCR are underlined. These cDNAs were cloned into  $pET30 b(+)$  vector, and the recombinant proteins were produced without any tag.

**Overproduction and partial purification of recombinant proteins***-* Fresh colonies of transformed BL21 (DE3) Rosetta2 bacteria (Novagen, Darmstadt, Germany) containing pET30 b(+) vector expressing the desired recombinant protein were transferred into 15 ml of LB medium supplemented with Kanamycine  $1^{\degree}\prime\degree$ , and grown at 37°C. Saturated culture was transferred into 800 ml LB medium supplemented with Kanamycine  $1^{\degree}/\degree$ , and growth was continued until DO<sub>600 nm</sub>=0.6. IPTG (0.4 mM) was added to the medium, and the bacteria were grown at 20°C for 16 h. Cells were harvested by centrifugation (4000 g, 45 min), and pellets were re-suspended in 30 ml of 25 mM HEPES pH 8.0, 1 mM EDTA, 1 mM DTT, 10% glycerol  $(v/v)$ , 50 µM PLP, 5 mM  $\varepsilon$ -aminocaproic acid, and 1 mM benzamidine (Buffer A), and sonicated for 10 min at  $4^{\circ}$ C on a Branson sonicator. Streptomycin sulphate (0.1% (w/v)) was added to precipitate DNA and the extract was centrifuged for 45 min at 40 000 g (Sorvall SS-34) at 4°C. The resulting supernatant of soluble proteins were stored at -80°C until used for purification.

Partial purification of recombinant proteins from *R. meliloti* Q02635 WT and K12G mutant were performed as follows: crude extract of soluble proteins (20-50 mg) were loaded on a 30 ml of Q-sepharose resin (GE Healthcare) in an XK-16 column (Amersham Pharmacia) equilibrated with Buffer A, proteins were eluted by a linear gradient (0-0.5 M) of NaCl in buffer A. Active fractions were pooled and loaded on HiPrep 16/60 Sephacryl S-200 column (Amersham Pharmacia) equilibrated with Buffer A supplemented with 100 mM NaCl (Figure 13). The most active fractions were pooled, desalted in buffer A supplemented with 10% glycerol, frozen in liquid nitrogen and stored at  $80^{\circ}$ C.

Partial purification of recombinant proteins from *Thermus thermophilus* Q56232 WT and K12G mutant were performed as follows: crude extract of soluble proteins (20-50 mg) were loaded on 100 ml of EMD DEAE 650(M) resin Merck) in an XK 26 column (Amersham Pharmacia) equilibrated with Buffer A, proteins were eluted by a linear gradient (0-0.5 M) of NaCl in buffer A (Figure 14). The most active fractions were pooled, desalted in buffer A supplemented with 10% glycerol, frozen in liquid nitrogen and stored at -80°C.

Partial purification of recombinant proteins from *Synechoccocus* Q3AK33 were performed as follows: crude extract of soluble proteins (20-50 mg) were loaded on a 30 ml of Q-sepharose resin (GE Healthcare) in an XK-16 column (Amersham Pharmacia) equilibrated with Buffer A, proteins were eluted by a linear gradient (0-0.5 M) of NaCl in buffer A (Figure 5). The most active fractions were pooled, desalted in buffer A supplemented with 10% glycerol, frozen in liquid nitrogen and stored at -80°C

**Determination of enzyme activities***-* Glutamate-oxaloacetate aminotransferase activities were assayed as described in [8].

Prephenate aminotransferase activities were assayed by coupling the reaction with purified Tyrinsensitive arogenate-specific dehydrogenase from *Synechocystis* [22] and following the reduction of NADP at 340 nm. The reaction was carried out at 30 °C in 50 mM HEPES buffer (pH 8.0) in the presence of 40 nM coupling enzyme, 100 µM NADP and variable amounts of prephenate, and fixed concentration of glutamate (25mM). Activities were calculated using an epsilon for NADPH of 6250  $M^{-1}$ .cm<sup>-1</sup> at 340 nm. Arogenate formation was confirmed by HPLC analyses as described in [10].

**Kinetic data analysis***-* Initial velocity data were fitted to the Michaelis–Menten equation using the Kaleidagraph program (Synergy Software, Reading, PA, USA).

**Crystallization***-* Extensive searches for crystallization conditions were performed in 96 well plates at 293 K using the high throughput crystallization facilities of EMBL (Grenoble). Crystallization hits were refined in 24 well Linbro plates. All the crystals were grown at 293 K. Ath-PAT crystals grew in 0.1 M Na citrate pH 4.0, 11 % PEG 4K. Rme-AAT/PAT crystals were obtained in 0.2 M ammonium formate, 19 % PEG 3350 and Rme-AAT grew in 0.1 M Na acetate pH 4.5, 5% PEG 4K. The protein concentration was 10 mg/ml.

**Data collection***-* Diffraction data (Table 1) were collected at 100 K at the European Synchrotron Facilities either on FIP-BM30A [23] (Rme-AAT) or ID23-1 (Ath-PAT and Rme-AAT/PAT) using an ADSC 315r detector. Prior freezing, the crystals were soaked in mother liquor with glycerol in a concentration range from 20 % (Ath-PAT and Rme-AAT/PAT) to 25 % (Rme-AAT). All the data were processed and scaled using XDS [24].

**Phasing and Refinement***-* All the phase calculations were performed by molecular replacement after modifying the models using CHAINSAW [25] from CCP4 [26]. Phases for Ath-PAT were calculated using Molrep [27] and the atomic coordinates of aspartate aminotransferase from *Phormidium lapideum* (PDB code 1J32) as a model. Phaser [28] was used to calculate phases for Rme-AAT/PAT and Rme-AAT using the atomic coordinates of Ath-PAT and those of the aspartate aminotransferase from *Phormidium lapideum* (PDB code 1J32), respectively. All the refinements and rebuilding were carried out using Phenix [29] and COOT [30]. Non-crystallographic symmetry was applied in all cases during refinement. Rmsd calculations were done using PDBefold [31].

Atomic coordinates and X-ray data were deposited in the PDB with the accession numbers: At-PAT (6F5V), Rme-PAT (6F77) and Rme-AspT (6F35).

#### **Structural modeling**

#### *I- Modelling of Lys12 conformational transitions using MD simulations*

*Initial model-* It was built from the structure of aspartate aminotransferase from *Thermus thermophilus* (Tth-PAT) complexed with maleate (PDB id 1BKG). 1BKG gives the structure from Met1 to Leu382 in both A and B chains. It was obtained in the presence of 4'-deoxy-4'-aminopyridoxal-5' phosphate (PMP) and maleate.

Experimentalists have modified the structure "manually" changing the position of Lys12 "by eye" to fit experimental observations and introducing global rotations and translations with respect to 1BKG to obtain a structure called Sexp. They also manually inserted an ideal structure of prephenate in place of maleate. The global atomic coordinate rms deviation between Sexp and 1BKG is 0.85 Å (considering the 5888 atoms of the dimer without water and ligands). To start from the experimental structure 1BKG without losing the information coming from experiments, we started by calculating the reverse transformation to be applied to Sexp to bring it back close to 1BKG while maintaining the rearrangement of Lys12. Then, this transformation was applied to prephenate and crystal waters, as well.

*Structure preparation-* Finally the structure of the dimer of Tth-PAT from residue 1 to 382, PMP, prephenate and 141 crystal waters was completed and refined with the molecular dynamics program CHARMM [32]. The all-atom force field *all27* for proteins [33,34] was used. An initial structure of prephenate was obtained from the Zinc database (zinc1532584) while the initial structure of pyridoxamine phosphate (PMP) was loaded into chimera with its PubChem ID (cid1053). Both structures were saved to mol2 format and submitted to the SwissParam server which provided initial topology and parameter files for CHARMM. All histidines were protonated on their N*δ* atom, whereas aspartate and glutamate residues were kept unprotonated (negative charge). All arginine and lysine residues were positively charged. Missing hydrogen atom coordinates were built with CHARMM and the structure was energy minimized down to a gradient of 0.1 kcal/mol/Å with the ABNR algorithm and subject to harmonic restraints on heavy atoms with force constant of 5 kcal/mol/ $\AA^2$ . This led to an initial refined structure with a dK-Prf of 2.9 Å. To complete the initial unrestrained model, a 1 ns Langevin Molecular Dynamics simulation at a temperature of 300 K was run on this system containing the 1BKG dimer, the X-ray waters, prephenate and PMP. (The frame giving the highest CHARMM interaction energy between prephenate or PMP and the protein is shown in Figure 3).

**Solvation and restrained optimization***-* Twelve sodium ions were placed around the protein at initial locations minimizing their interaction energy with the protein to neutralize the system. Then the system was solvated inside a  $109*66*66 \text{ Å}^3$  water box initially containing 15886 waters molecules. Periodic boundary conditions were applied in the orthorhombic symmetry. The protein atoms were initially fixed, the crystal water oxygen atoms were harmonically restrained with a force constant of 2.0 kcal/mol/ $\AA^2$  and the solvent around the protein was equilibrated with 5000 steps (10 ps) Langevin Molecular Dynamics simulation at a temperature of 300 K yielding structure *1BKG\_Preph\_Mod* (after introduction of prephenate into 1BKG, solvation and energy minimization).

Then all protein atoms were set free to move, and harmonic restraints were introduced on protein backbone atoms, crystal water oxygens and heavy atoms of PMP with a force constant of 1.0 kcal/mol/ $\AA^2$ . Additionally, Nuclear Overhauser Enhancement (NOE)-type distance restraints using a bi-harmonic potential with a force constant of 40 kcal/mol/ $\AA$ <sup>2</sup> were applied between:

- Lys12 NZ atom and prephenate O1 oxygen, dK-Prf (above 2.0 and below 3.5 Å)
- PMP H1 or H2 atoms and prephenate O2 oxygen (above 1.0 and below 3.5 Å)
- Lys101 NZ atom and prephenate O6 oxygen (above 2.0 and below 3.5 Å)

- Arg361 HH11 and HH21 atoms and prephenate O3 or O5 oxygen (above 1.0 and below 3.0 Å) The system was energy minimized and subject to 10 more ps Langevin MD.

*Equilibration-* Starting with the equilibration phase, electrostatics was calculated using the Particle Mesh Ewald summation and the protein center of mass was restrained to the center of the coordinate system. The NOE-type distance restraints previously described were applied together with harmonic restraints on protein backbone atoms and heavy atoms of PMP with a force constant of 1.0 kcal/mol/ $\AA^2$ . Two rounds of steepest descent followed by ABNR minimization were run, the latter acting on the box dimensions as well as the atom coordinates. The system was then equilibrated for 100 ps using Langevin dynamics at 300 K. Harmonic restraints on the protein and PMP were removed leaving NOE-restraints only. Then, the system was subject to a constant pressure of 1 atm and temperature of 300 K and a constant pressure and temperature CPT (Constant Pressure and Temperature) dynamics based on Berendsen's weak coupling method [35] was run for 20 ns leading to Structure *1BKG\_Preph\_MD* (after introduction of prephenate into 1BKG, solvation, equilibration and 20 ns production Molecular Dynamics simulation).

#### **II-Parameter optimization**

*Partial charge optimization through QM/MM study-* A refined parameterization of the two substrates: prephenate and PMP and surrounding residues was done. Indeed, the atomic charges given by SwissParam are probably deduced from small compounds and certainly do not take into account the specific environment of the molecules. Starting from a structure with low energy and low dK-Prf (3.2Å) from previous equilibration, we have run a QM/MM optimization followed by a calculation of the charges adequate to fit the calculated electrostatic potential around the ligands.

We have used the Gaussian 03 package [36] for the quantal parts (QM parts hereafter) and Tinker 4.2 [37] for the part described by molecular mechanics (MM part). In addition to prephenate PMP and residue Lys12, the QM part consisted of side chains of Lys101, Trp125, Ser127, Lys234, Arg242, Tyr322, Arg361 from chain A and Tyr64 from chain B. 14 water molecules were also included in this quantum part for a total of 191 QM atoms. All MM atoms belonging to residues within a distance of 3 Å of the QM part were allowed to relax during geometry optimizations, and the remaining structure was kept frozen. Dangling bonds between the QM and MM subsystems were treated using the Link-Atom scheme [38] as implemented in Ref. [39].

For the QM part, the B3LYP density functional method [40,41,42] with a 6-31G\* basis set for all QM atoms was used, which represents 1484 basis functions. The MM part was treated with the ChARMM27 force field. Missing residues and hydrogens were added to complete the structure. From this energy-minimized structure, charges on atoms in the active region were calculated using the Merz-Kollman routine [20] in Gaussian to fit the *ab initio* electrostatic potential. After checking that the calculated charges for water atoms were very close to those used in the CHARMM MM force field, these charges were directly transferred to the protein structure file (psf) describing the MM system for prephenate, PMP and surrounding residues.

*New molecular dynamics simulations-* Using this modified structure file, 24 ns Constant Pressure and Temperature dynamics were run restarting from structure *1BKG\_Preph\_MD*. The NOE (Nuclear Overhauser Enhancement**)** type restraints used to obtain *1BKG\_Preph\_MD* were kept for four ns and then removed for the remaining 20 ns.

*Improving conformational sampling-* A method to explore more conformational space is to use Replica eXchanging Self-Guided Langevin Dynamics as implemented in CHARMM in the RXSGLD module [21]. SGLD simulations enhance conformational search efficiency through acceleration of low frequency motions in a molecular system. The low frequency motion represents the slow conformational change that contributes the most in conformational search. A guiding force resonant with the low frequency motion, which is calculated based on a local average of momentums, is introduced into the equation of motion to enhance the low frequency motion. Here, temperatures of replicas and their selfguiding temperatures were exponentially spaced from 300 to 400 K and from 300 to 600 K, respectively. The guiding factor increment was set to 0.2 and exchange was attempted every 50 steps [43]. 16 replicas were created starting from structure *1BKG\_Preph\_Mod* and vacuum RXGLD simulations were run for 10 ns with a time step of 1 fs on 16 processors. To explore conformational transitions compatible with a proximity between Lys12 and prephenate, NOE-type distance restraints on Lys12, prephenate and PMP were turned on and protein backbone atom coordinates were harmonically restrained with force constant of 2.0 kcal/mol/ $A^2$  except for the 29 N-Terminal amino acids of monomer A which were left completely free to move. Coordinates of Met1 to Gln29, chain A and PMP + prephenate were extracted from the MD trajectory every ps (20000 structures) and grouped into 16 clusters corresponding to coordinate rmsd larger than 3.5 Å.

*Final production dynamics from selected clusters* - The coordinates of the lowest energy structure in the 6 selected clusters were computed and one molecular dynamics simulation was started from each resulting minimized structure. Each system was neutralized, solvated and equilibrated for 100 ps with 4 NOE-type distance restraints as explained above. Harmonic restraints on the protein and PMP were removed leaving NOE-restraints only. Then, for the production phase, the system was subject to a constant pressure of 1 atm and temperature of 300 K and a CPT dynamics based on Berendsen's weak coupling method [35] was run for 5 ns. Then the NOE-restraint on Lys12 was removed and the dynamics was continued.

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# **Footnote**

\*During the final stage of writing our Manuscript, a new structural study on *Arabidopsis* prephenate amino transferase appeared online in plant journal (14). This study confirms that Ath-PAT behaves very similarly to Tth-PAT during substrate binding.

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# **Author contribution**

All the authors have contributed to the writing of the manuscript. CG and MG performed biochemical and crystallization experiments, DC and AR solved the 3D structures, SG performed molecular modeling, DC, SC, GC and MM designed the research

# **Table 1: data collection statistics**





Rsym= $\frac{\sum \sum [I_i - I_m]}{\sum \sum I_i}$ , where I<sub>i</sub> is the intensity of the measured reflection and I<sub>m</sub> is the mean intensity of this reflection. Values indicated in parentheses correspond to the statistics in the highest resolution shell.

# **Table 2: Refinement statistics**



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# **Table 3: kinetics analyses of 1b AAT Wild Type and K12G mutants from** *R. meliloti* **(Q02635) and**  *T. thermophilus* **(Q56232).**





Values given are the average of at least three independent determinations from a single enzyme preparation  $\pm$  SD. \*Note that K12G mutant kinetic apparent parameters  $K_M$  and  $k_{cat}$  for prephenate were difficult to determine accurately due to the important deterioration of the apparent  $K_M$ .

## **Table 4: primers used for amplification of** *Rhizobium meliloti* **(Q02635),** *Thermus thermophilus* **(Q56232)** *and Synechochoccus* **Q3AK33 1AAT***,* **and primers used for directed mutagenesis of K12G mutants**.



## **Figures Legends**

**Figure 1. Tyrosine biosynthesis pathway**. The alternative routes for the synthesis of tyrosine are represented. Synthesis of phenylalanine from arogenate occurring in plants is also represented. Multiple enzymatic steps involved are shown by dotted lines. ADH: arogenate dehydrogenase; ADT: arogenate dehydratase; E4P: erythrose 4-phosphate; PAT: prephenate aminotransferase, PDH: prephenate dehydrogenase; PEP: phosphoenolpyruvate; TYR-AT: tyrosine aminotransferase

## **Figure 2: Three dimensional structure comparison of PAT competent and incompetent 1AAT.**

A-3D structure of *A. thaliana* PAT competent 1βAAT dimer with PLP in yellow sticks. B- Superimposition of 3D structures of a monomer of *A. thaliana* PAT competent 1BAAT (green), and *R. meliloti* PAT competent 1 $\beta$ AAT (blue) and *R. meliloti* PAT incompetent 1 $\beta$ AAT (red).

# **Figure 3: Docking of prephenate in the active site of** *Thermus thermophilus* **PAT competent 1AAT (PDB id 1BKG)**

A Structure of the three ketoacid substrates of PAT competent 1 $\beta$  AAT, oxaloacetate (PDB ID 3PDB) (blue),  $\alpha$  ketoglutarate (build from prephenate) (cyan), and prephenate (green).

B Initial docking of prephenate in the active site of Tth-PAT (PDB id 1BKG). Prephenate was inserted manually in the active site of 1BKG in place of maleate and the resulting structure energy minimized. Then a short 1 ns unrestrained MD simulation was run. The structure giving the highest interaction energy between prephenate or PMP and the protein is displayed. The two carboxylate groups of prephenate interact with Lys101 and Arg361, and its keto group points in direction of the NH2 of PMP and the catalytic Lys234. In this position, the hydroxyl group of prephenate faces a  $_{10}$ AMKPSA<sub>15</sub> amino-acid motif of the N-terminal flexible loop. In this motif Lys12 is only present in PAT competent enzymes. In the maleate-PMP-Enzyme ternary complex (1BKG), the side chain of Lys12 points outside the active site (10.5 Å from the prephenate OH group). In this simulation E40 and T16 were close to the active site but not in interaction with either PMP or prephenate.

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**Figure 5: Analysis of recombinant** *Synechochoccus* **Q3AK33 1AAT**. A: N-terminal sequence alignment between *Synechochoccus* Q3AK33 and *Synechocystis* Q55128 1AAT. B: SDS-PAGE of recombinante Sco\_Q3AK33 Q sepharose purification: 1 total protein extract 20 µg; 2: Soluble protein extract 20 µg; 3 most active fraction eluted from a Qsepharose column 10 µg. C: Kinetic characterization of *Synechochoccus* Q3AK33 1AAT. Glutamate 25mM was used as amino donor**.**  Values given are the average of at least three independent determinations from a single enzyme preparation ± SD.

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The evolutionary history was inferred by using the Maximum Likelihood method based on the JTT matrix-based model. The tree with the highest log likelihood (-16448,6813) is shown. Initial tree(s) for the heuristic search were obtained by applying the Neighbor-Joining method to a matrix of pairwise distances estimated using a JTT model. The tree is drawn to scale, with branch lengths measured in the number of substitutions per site. The analysis involved 36 amino acid sequences. All positions containing gaps and missing data were eliminated. There were a total of 361 positions in the final dataset. Evolutionary analyses were conducted in MEGA6. Nature of the amino acid residue in position 12 (according to *Thermus thermophilus* 1AAT) is represented.

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# **Figures**



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							K12 116						E40		
A. thal PAT			55 MAKPNDAETLSSSVDMSLSPRVOSLKPSKTMVITDLAATLVOSGVPVIRLAAGEPDFDTP												
S. Bico PAT			47 -----MVTVRAEAVDTTISPRVSALRPSKTMAITDQATALRQAGVPVIGLAAGEPDFDTP												
	T. ther PAT		----------------- MRGLSRRVOAMKPSATVAVNAKALELRROGVDLVALTAGEPDFDTP												
	C. tepi PAT		--------MSVESFERFLSRRVLSMOESOTMKITGLAKKMOAEGKDVVSLSAGEPDFPTP												
	R. spha PAT														
R. meli PAT.			--------------- MAFLADALSRVKPSATIAVSOKARELKAKGRDVIGLGAGEPDFDTP												
R. meli AAT			1 ----MTINATVKEAGFQPASRISSIGVSEILKIGARAAAMKREGKPVIILGAGEPDFDTP												
	S. aver AAT		1 ------MSAATPPTERRVSARVGAISESATLAVDAKAKALKAAGRPVIGFGAGEPDFPTP												
Synech AAT			---------------- MRLT QRVS QVVPS ITLEITAK AKAMRTEGIDVLSFTAGE PDFTTP												
										. * * . *				***** **	
A. thal PAT			115 KVVAEAGINAIR-EGFTRYTLNAGITELREAICRKLKEENGLSYAPDQILVSNGAKQSLL												
S. Bico PAT			107 AVIAEAGMNAIR-DGATRYTPNAGTLELRKAICKKLEEENGLSYTPDQVLVSNGAKQCIT												
T. ther PAT		47	EHVKEAARRALA-QGKTKYAPPAGIPELREALAEKFRRENGLSVTPEETIVTVGGKQALF												
C. tepi PAT			53 ENVCEAGIEAIR-KGFTRYTANSGIPELKKAIIRKLORDNGLEYAEDEIIVSNGGKOALA												
R. spha PAT		49	VHIRQAAAAAIE-AGQTRYTDVDGTPELKAAIVEKFRRENGLSYETNQISVGTGGKQVLY												
R. meli PAT		47	DNIKKAAIDAID-RGETKYTPVSGIPELREAIAKKFKRENNLDYTAAQTIVGTGGKQILF												
R. meli AAT		57	EHVKOAASDAIH-RGETKYTALDGTPELKKAIREKFORENGLAYELDEITVATGAKOILF												
S. aver AAT		55	DYIVQAAIEACSNPKYHRYTPAGGLPELKAAIAAKTLRDSGYEVDASQVLVTNGGKQAIY												
Synech AAT			46 PHIVEAAKLALD-EGKTRYGPAAGEPALROAIAKKLREKNNLPYEAANILVTNGGKHSLF												
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														K101	

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$A$ Sco Q3AK33	Syn 055128 Sco Q3AK33 Syn Q55128	MPRPPELSDRAVALKPSLTLEISAKAKALRDSGKDICSLSAGEPDFATPPFIVEAAQHALGSGFTRY ----MRLTQRVSQVVPSITLEITAKAKAMRTEGIDVLSFTAGEPDFTTPPHIVEAAKLALDEGKTRY GPAAGDPDLRAALAHKLSVENGIPTQPEQVLVTNGGKQAIYNLFQVLLNPGDEVLLPAPYWLSYPEM GPAAGEPALRQAIAKKLREKNNLPYEAANILVTNGGKHSLFNLMLAMIEQGDEVIIPAPYWLSYPEM								
B	MW $\top$	S Qseph.								
	250 150 100					Sco Q3AK33				
	$75 -$			Substrate varied/fixed	$K_{\rm M}$ (µM)	$K_{cat}(S^{-1})$				
	$50 -$ $37 -$			Oxaloacetate/Glu	$12+2$	16±1,8				
				Prephenate/Glu	$207 + 21$	$0.25 \pm 0.05$				
	$25 -$									
	$20 -$									

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