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Theoretical investigation of electronic performance, half-metallicity, and magnetic properties of Cr-substituted BaTe

Khelifa Berriah¹ · Bendouma Doumi² · Allel Mokaddem^{3,4} · Mohammed Elkeurti^{1,2} · Adlane Sayede⁵ · Abdelkader Tadjer⁶ · João Pedro Araújo⁷

Abstract

We have investigated the structural, electronic, and ferromagnetic properties of chromium (Cr)-doped rocksalt BaTe (Ba_{1-x}Cr_xTe) compounds with compositions x = 0.25, 0.5, and 0.75, based on density functional theory with generalized gradient approximation of Wu–Cohen (GGA-WC) and Tran–Blaha-modified Becke–Johnson (TB-mBJ) potential using the WIEN2k package. We found that the electronic structure showed half-metallic ferromagnetic character with spin polarization of 100% around the Fermi level. In addition, the minority-spin bands depicted a half-metallic ferromagnetic (HMF) gap and half-metallic (HM) gap. The improved HMF and HM gaps found with the TB-mBJ potential are higher than with the GGA-WC approximation. These large HM gaps make Ba_{1-x}Cr_xTe compounds promising candidates for use in spintronics applications.

Keywords Cr-substituted BaTe · TB-mBJ approach · Half-metallic ferromagnetism · Electronic properties

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1 Introduction

In recent years, a new field of electronics called spintronics has been developing rapidly [1–3]. In this approach, the electron spin is used as another degree of freedom for information processing, offering a promising method to meet new challenges [2,4]. Spintronics devices offer several advantages compared with conventional electronics, including higher speed and lower power, because they are based on spin direction and spin coupling [5]. In particular, transitionmetal-doped III–V and II–VI semiconductors have emerged as a new type of compounds called dilute magnetic semiconductors (DMSs). These materials are main candidates for progress of spintronics applications, because they are ferromagnetically stable at higher temperatures than room temperature [6,7] and exhibit half-metallic ferromagnetic behavior [8–10].

The BaX (X = S, Se, Te) barium chalcogenides belong to the group of alkaline-earth chalcogenides, which are widegap insulators with closed-shell ionic systems [11]. BaX compounds have attracted increasing attention because of their potential applications in light-emitting and laser diodes [12–14] as well as luminescent and magnetooptical devices [15–18]. BaTe forms closed-shell ionic coordination and crystallizes as an NaCl-type (B1) phase at ambient conditions [14,19]; it has been widely researched owing to its interesting structural and electronic properties [14]. Syassen et al. [20] found that optical and absorption spectra of BaTe appear in the pressure range of 0–400 kPa at temperature of 300 K. Besides, Feng et al. [14] investigated the effect of spin–orbit coupling on the optical properties of BaTe, including the reflectivity and the real and imaginary parts of the dielectric function. BaTe has an indirect gap located between the Γ and X high-symmetry points, exhibiting a reduction of about 0.27 eV in its bandgap on application of spin–orbit coupling [21].

Recently, several experimental and theoretical works have been carried out on the half-metallic and magnetic properties of DMSs based on II–VI semiconductors, e.g., on the room-temperature ferromagnetism of half-metallic Ni-doped ZnS [22], half-metallic ferromagnetism of Fe-doped ZnS thin films formed by chemical bath deposition [23], halfmetallic ferromagnetism of Co- and Fe-doped CdSe [24], half-metallic ferromagnetism in V-doped ZnS [25], and halfmetallic ferromagnetic properties of V-doped BaS [26] and V-doped SrO [27].

To the best of the authors' knowledge, no experimental and theoretical studies have been performed on BaTe doped with chromium (Cr) atom. In the work presented herein, we investigated the half-metallic ferromagnetic behavior and improved electronic structure of Cr-doped BaTe to explore its ferromagnetic features for use in spintronic applications. For these calculations, we used first-principles calculations based on density functional theory (DFT) using the WIEN2k code [28] [29,30], treating the exchange and correlation potential using the Wu–Cohen generalized gradient approximation (GGA-WC) [31] and Tran–Blaha-modified Becke–Johnson (TB-mBJ) approach [32,33].

2 Calculation methods

We employed the WIEN2k package [28] based on the fullpotential linearized augmented plane-wave (FP-LAPW) method for first-principles DFT calculations [29,30] to study the structural, electronic, and magnetic properties of Ba_{1-x} Cr_x Te compounds with various compositions x =0.25, 0.5, and 0.75 of chromium (Cr) impurity. First, we calculated the structural parameters, electronic structure, and magnetic properties, using the Wu–Cohen generalized gradient approximation (GGA-WC) [31]. We then used the accurate Tran–Blaha-modified Becke–Johnson (TB-mBJ) exchange potential combined with the local density correlation potential [32,33] to improve the electronic properties and give the exact gaps of the band structure.

We chose the average of the muffin-tin radii of Ba, Te, and Cr atoms in such a way that the muffin-tin spheres do not overlap. Besides, the charge density was Fourier extended up to $G_{\text{max}} = 14$ (a.u.)⁻¹, where G_{max} is the largest vector of the Fourier expansion. We extended the basis functions and potential in combination with spherical harmonics around atomic sites with cutoff of $l_{\text{max}} = 10$. In the interstitial region, the Fourier series were extended in plane waves with cutoff of $R_{\rm MT}$ $K_{\rm max} = 9$ (where $R_{\rm MT}$ is the mean radius of the muffin-tin spheres). For Brillouin zone sampling, we utilized a Monkhorst-Pack mesh [34,35] of $(4 \times 4 \times 4)$ for BaTe, Ba_{0.75}Cr_{0.25}Te, and $Ba_{0.25}Cr_{0.75}Te$, and $(4 \times 4 \times 3)$ for $Ba_{0.5}Cr_{0.5}Te$, with self-consistent convergence of the total energy of 0.1 mRy. Note that the number of k-points was reduced in the tetragonal structure of Ba0.5Cr0.5Te in space group P4/mmm (no. 123) due to its lower number of symmetry operations compared with the cubic structure of Ba_{0.75}Cr_{0.25}Te and Ba_{0.25}Cr_{0.75}Te in space group $Pm\bar{3}m$ (no. 221).

3 Results and discussion

3.1 Structural parameters

Barium telluride (BaTe) crystallizes in a rocksalt NaCl (B1) phase; its conventional structure has two types of atoms, viz. Ba and Te, located at (0, 0, 0) and (0.5, 0.5, 0.5, 0.5)0.5) positions, respectively, with space group $Fm\bar{3}m$ (no. 225). In the Ba₄Te₄ structure with eight atoms, we substituted one, two, or three atoms of chromium (Cr) at Ba sites to create supercells of Ba_{0.75}Cr_{0.25}Te, Ba_{0.5}Cr_{0.5}Te, and Ba_{0.25}Cr_{0.75}Te with concentration x = 0.25, 0.5, and0.75, respectively. The $Ba_{0.75}Cr_{0.25}Te$ and $Ba_{0.25}Cr_{0.75}Te$ supercells have cubic structure in space group $Pm\bar{3}m$ (no. 221), while $Ba_0 {}_5Cr_0 {}_5Te$ has tetragonal structure in space group P4/mmm (no. 123). Note that the description of the Ba_{0.75}Cr_{0.25}Te, Ba_{0.5}Cr_{0.5}Te, and Ba_{0.25}Cr_{0.75}Te alloys using smaller supercells would necessarily introduce side effects on their ordered structure. Therefore, our calculations are valid only for ordered structures close to the stoichiometries of the ordered Ba_{0.75}Cr_{0.25}Te, Ba_{0.5}Cr_{0.5}Te, and $Ba_{0.25}$ Cr_{0.75}Te compounds.

Structural parameters of BaTe and $Ba_{1-x}Cr_xTe$ at various concentrations were determined by fitting the variation of the total energy as a function of volume using the Murnaghan equation [36]. The structural parameters such as the lattice constant (*a*), bulk modulus (*B*), and its pressure derivative (*B'*) calculated for BaTe and $Ba_{1-x}Cr_xTe$ at different concentrations using the GGA-WC approximation, together with other theoretical [37–40] and experimental data [41], are summarized in Table 1. The *a* and *B* parameters obtained for BaTe are in good agreement with experimental [41] and recent theoretical calculations [37,38] using the GGA-WC approximation [31]. In addition, the

Table 1 Calculated lattice constant (*a*), bulk modulus (*B*), and its pressure derivative (*B'*) for BaTe and Ba_{1-x}Cr_xTe with composition x = 0.25, 0.5, and 0.75 of Cr atoms

Compound	<i>a</i> (Å)	B (GPa)	B'	Method
This work				GGA-WC
ВаТе	6.944	31.89	4.56	
Ba _{0.75} Cr _{0.25} Te	6.735	33.51	5.02	
Ba _{0.5} Cr _{0.5} Te	6.457	38.27	5.47	
Ba _{0.25} Cr _{0.75} Te	6.073	50.66	6.42	
Other calculations				
ВаТе	6.934 [37]	28.81 [37]	5.191 [37]	GGA-WC
	6.955 [<mark>38</mark>]	30.91 [38]	4.012 [38]	GGA-WC
	7.075 [<mark>39</mark>]	27.04 [39]	4.63 [39]	GGA-PBE
	7.0667 [40]	28.5384 [40]	4.0526 [40]	GGA-PBE
	6.8690 [<mark>40</mark>]	35.3052 [40]	4.4247 [<mark>40</mark>]	LDA
	7.005 [41]	29.4 [41]	7.4 [41]	Experimental

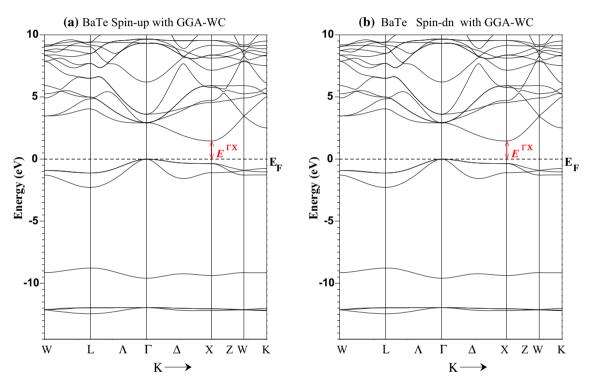


Fig. 1 Spin-polarized band structure obtained with GGA-WC for BaTe: a majority spin (up) and b minority spin (dn). The Fermi level is set to zero (horizontal dotted line)

results of these parameters are better than recent calculations [39,40] using the generalized gradient approximation of Perdew–Burke–Ernzerhof (GGA-PBE) [42] and local density approximation (LDA) with Teter–Pade parameterization [43]. This is due to the precise GGA-WC exchange potential for structural properties. For the doped $Ba_{1-x}Cr_xTe$ system, the lattice constant decreased with increasing concentration of chromium owing to the difference between the ionic radii of the Ba atom and the substituted Cr impurity. Consequently, $Ba_{1-x}Cr_xTe$ becomes harder with increasing Cr concentration. Note that there are no experimental and theoretical calculations to compare with our results for the structural parameters of $Ba_{0.75}Cr_{0.25}Te$, $Ba_{0.5}Cr_{0.5}Te$, and $Ba_{0.25}Cr_{0.75}Te$.

3.2 Electronic structure and half-metallic performance

We calculated the spin-polarized densities of states (DOS) and band structures of the BaTe and $Ba_{1-x}Cr_xTe$ materials using the optimized lattice constants. Firstly, we used the GGA-WC approximation to calculate the electronic structure

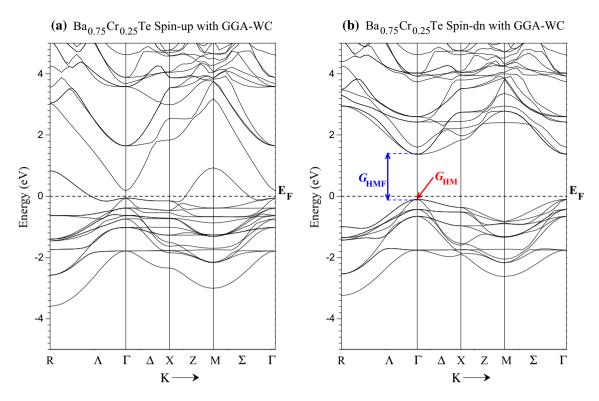


Fig. 2 Spin-polarized band structure obtained with GGA-WC for $Ba_{0.75}Cr_{0.25}Te$: **a** majority spin (up) and **b** minority spin (dn). The Fermi level is set to zero (horizontal dotted line)

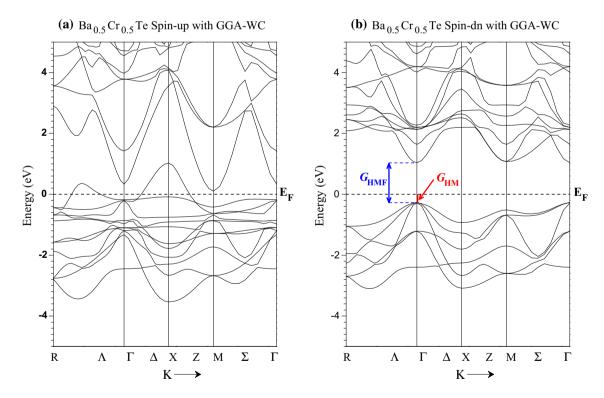


Fig. 3 Spin-polarized band structure obtained with GGA-WC for $Ba_{0.5}Cr_{0.5}Te$: **a** majority spin (up) and **b** minority spin (dn). The Fermi level is set to zero (horizontal dotted line)

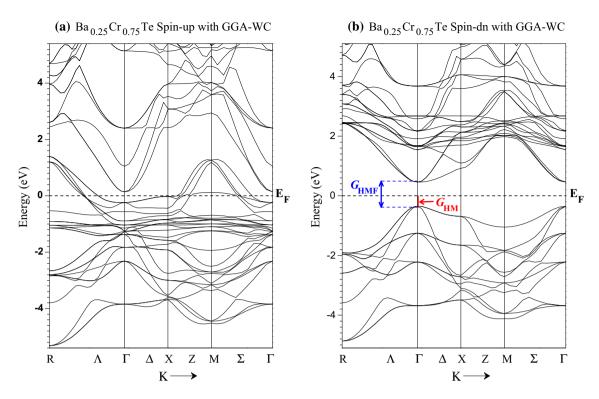


Fig. 4 Spin-polarized band structure obtained with GGA-WC for $Ba_{0.25}Cr_{0.75}Te$: **a** majority spin (up) and **b** minority spin (dn). The Fermi level is set to zero (horizontal dotted line)

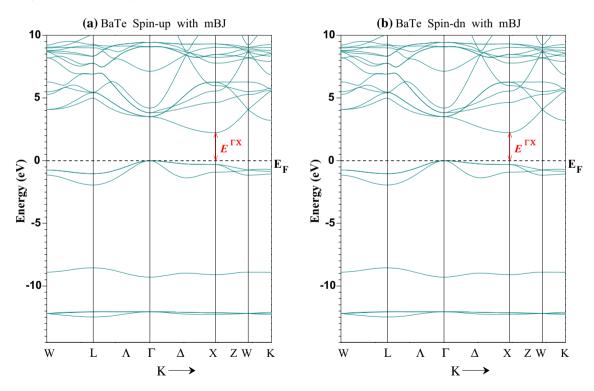


Fig. 5 Spin-polarized band structure obtained with TB-mBJ for BaTe: **a** majority spin (up) and **b** minority spin (dn). The Fermi level is set to zero (horizontal dotted line)

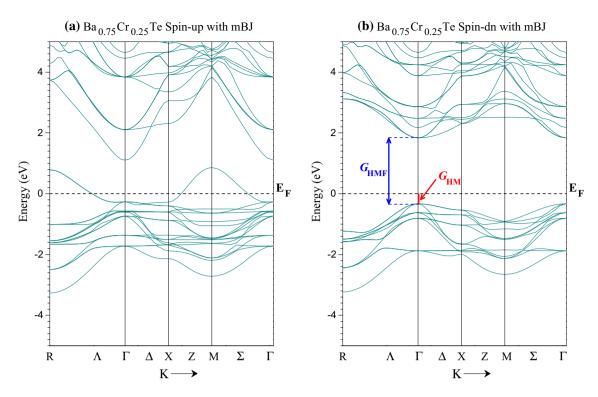


Fig. 6 Spin-polarized band structure obtained with TB-mBJ for $Ba_{0.75}Cr_{0.25}Te$: **a** majority spin (up) and **b** minority spin (dn). The Fermi level is set to zero (horizontal dotted line)

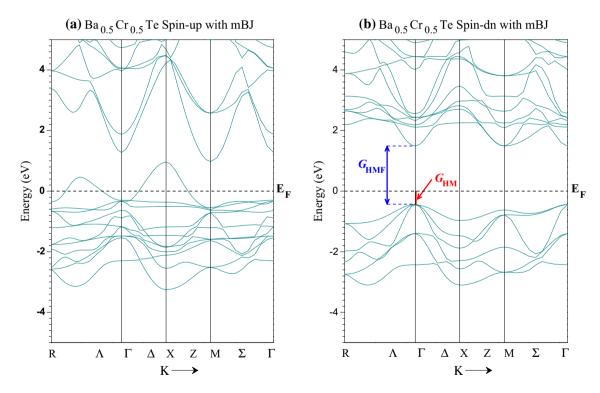


Fig.7 Spin-polarized band structure obtained with TB-mBJ for $Ba_{0.5}Cr_{0.5}Te$: **a** majority spin (up) and **b** minority spin (dn). The Fermi level is set to zero (horizontal dotted line)

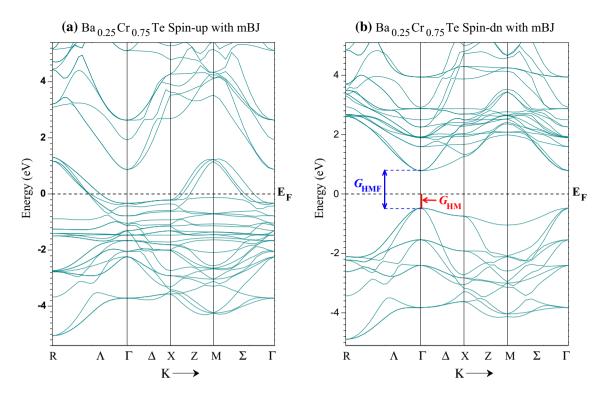


Fig. 8 Spin-polarized band structure obtained with TB-mBJ for $Ba_{0.25}Cr_{0.75}Te$: **a** majority spin (up) and **b** minority spin (dn). The Fermi level is set to zero (horizontal dotted line)

Compound	$G_{\rm HMF}({\rm eV})$	$G_{\rm HM}({\rm eV})$	$E^{\Gamma} X (eV)$	Method	Behavior
This work				GGA-WC	
ВаТе			1.455		
Ba _{0.75} Cr _{0.25} Te	1.491	0.136			HMF
Ba _{0.5} Cr _{0.5} Te	1.298	0.282			HMF
Ba _{0.25} Cr _{0.75} Te	0.816	0.360			HMF
This work				TB-mBJ	
ВаТе			2.256		
Ba _{0.75} Cr _{0.25} Te	2.167	0.329			HMF
Ba _{0.5} Cr _{0.5} Te	1.923	0.435			HMF
Ba _{0.25} Cr _{0.75} Te	1.250	0.469			HMF
Other calculations					
ВаТе			1.325 [37], 1.822 [38]	GGA-WC	
			1.656 [39], 1.6516 [40]	GGA-PBE	
			1.3649 [40]	LDA	
			2.453 [37], 2.254 [38]	TB-mBJ	
			2.365 [39]		
			3.08 [48], 3.1 [49]	Experimental	

of the compounds. We then improved the bandgap calculations of the electronic structure using the TB-mBJ approach. The spin-polarized band structure of the compounds is shown in Figs. 1, 2, 3, and 4 for the GGA-WC and Figs. 5, 6, 7, and 8 with the TB-mBJ potential. Figures 1 and 5 show that the two spin channels of BaTe have similar semiconducting band

Table 2 Calculated indirect bandgap $(E^{\Gamma}X)$ for BaTe, half-metallic ferromagnetic gap (G_{HMF}) , and half-metallic gap (G_{HM}) of minority-spin bands

concentrations x = 0.25, 0.5,and 0.75 of Cr atoms

for $Ba_{1-x}Cr_xTe$ at

structures with indirect bandgap $(E^{\Gamma}X)$ located between Γ and X high-symmetry points. The remaining figures for $Ba_{1-x}Cr_xTe$ with each concentration reveal a half-metallic feature resulting from the metallic nature of the majority spin and a bandgap for the minority spin. The minority-spin bands of $Ba_{1-x}Cr_xTe$ reveal two types of bandgap, viz.

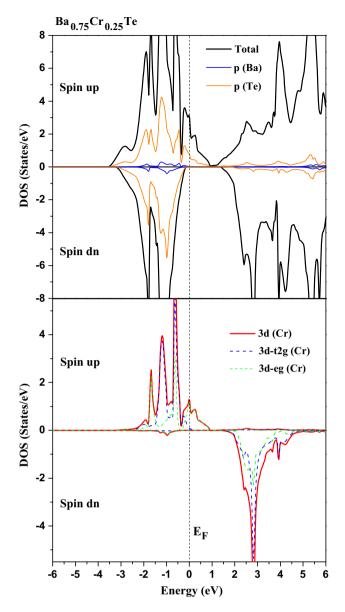


Fig. 9 Spin-polarized total and partial densities of states of Ba_{0.75}Cr_{0.25}Te. The Fermi level is set to zero (vertical dotted line)

a direct half-metallic ferromagnetic (HMF) gap and halfmetallic (HM) gap. The direct HMF gap is situated at Γ high-symmetry point between the valence-band maximum (VBM) and conduction-band minimum (CBM). The HM gap is defined as the minimum of the lowest energy of the majority (minority)-spin conduction bands with respect to the Fermi level and the absolute value of the highest energy of the majority (minority)-spin valence bands [44,45]. This factor describes the minimal energy bandgap for a spin-flip excitation needed to create a hole or electron with minority spin [9,46,47].

The calculated indirect gap $(E^{\Gamma X})$ of BaTe, HMF (G_{HMF}) and HM (G_{HM}) gaps of Ba_{1-x} Cr_x Te at all concentra-

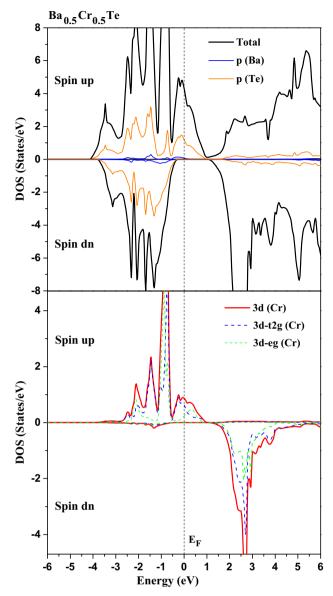


Fig. 10 Spin-polarized total and partial densities of states of $Ba_{0.5}Cr_{0.5}Te$. The Fermi level is set to zero (vertical dotted line)

tions, using the GGA-WC and TB-mBJ exchange potentials, together with other theoretical [37–40] and experimental [48,49] data, are presented in Table 2. The agreement is very good between our computed $E^{\Gamma X}$ for BaTe and calculations [37,38] using the same GGA-WC and TB-mBJ approaches. Also, our result for $E^{\Gamma X}$ with TB-mBJ is better than other calculations [39,40] using the GGA-PBE [42] and LDA [43] approximations. Note that the calculated $E^{\Gamma X}$, G_{HMF} , and G_{HM} gaps obtained with TB-mBJ are higher than obtained with the GGA-WC, because the TB-mBJ semilocal exchange correlation potential can provide perfect bandgaps compared with the LDA and different versions of the GGA approxima-

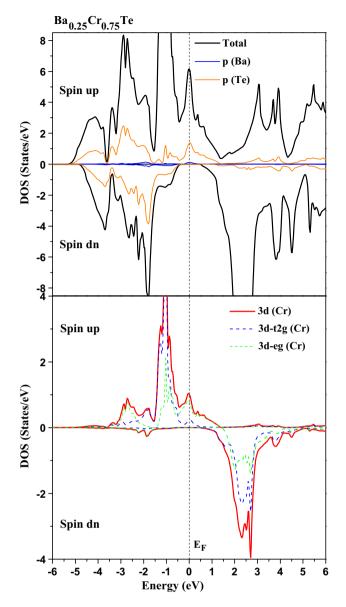


Fig. 11 Spin-polarized total and partial densities of states of Ba_{0.25}Cr_{0.75}Te. The Fermi level is set to zero (vertical dotted line)

tion for semiconductors and insulators [33,37,38]. Moreover, we understand that the half-metallic ferromagnetic gap of $Ba_{1-x}Cr_x$ Te decreases with increasing chromium concentration, due to broadening of 3*d* states of the Cr impurity in the gap. We found that $Ba_{0.75}Cr_{0.25}$ Te, $Ba_{0.5}Cr_{0.5}$ Te, and $Ba_{0.25}Cr_{0.75}$ Te compounds exhibited half-metallic character with HM gap of 0.329, 0.435, and 0.469 eV, making them promising materials for use in spintronics applications.

The contribution to the DOS around the Fermi level (E_F) determines the metallic or semiconductor nature of a material and its polarization. The spin polarization (P) of a material arises from the difference in the contribution of the densities of states of the majority spin $N^{\uparrow}(E_F)$ and minority spin

Table 3 Calculated total and local magnetic moments of relevant Cr, Ba, and Te atoms, and at interstitial sites (in Bohr magneton μ_B per Cr atom) for Ba_{1-x}Cr_xTe with compositions x = 0.25, 0.5, and 0.75 of Cr atoms

Compound	Total (μ_B)	$Cr\left(\mu_B ight)$	$Ba\left(\mu _{B}\right)$	$Te\left(\mu _{B}\right)$	Interstitial (μ_B)
This work					
Ba _{0.75} Cr _{0.25} Te	4	3.996	0.001	-0.261	0.265
Ba _{0.5} Cr _{0.5} Te	4	3.952	0.0005	-0.242	0.290
Ba _{0.25} Cr _{0.75} Te	4	3.788	-0.006	-0.189	0.408

 N^{\downarrow} (E_F) around E_F , defined by the expression [50]

$$P = \frac{N^{\uparrow} (E_{\rm F}) - N^{\downarrow} ({\rm E}_{\rm F})}{N^{\uparrow} (E_{\rm F}) + N^{\downarrow} ({\rm E}_{\rm F})}.$$
(1)

The spin-polarized total (T) and partial (P) densities of states (DOS) of Ba_{0.75}Cr_{0.25}Te, Ba_{0.5}Cr_{0.5}Te, and Ba_{0.25}Cr_{0.75} Te compounds are shown in Figs. 9, 10, and 11, respectively. These plots depict that the valence bands of the two spin directions are dominated by the major contribution of p (Te) and 3d (Cr) states and minor p (Ba) states in the energy ranges of -2.5 to -0.1 eV for Ba_{0.75}Cr_{0.25}Te, -2.7 to -0.6 eV for $Ba_{0.5}Cr_{0.5}Te$, and -3.5 to -0.3 eV for $Ba_{0.25}Cr_{0.75}Te$. The crystal field of the octahedral (Te) surrounding splits the 3d (Cr) levels into two types of states, viz. three low-lying t_{2g} $(d_{xy}, d_{xz}, \text{ and } d_{yz})$ levels and two high-lying e_g $(d_{z^2}$ and $d_{x^2-y^2}$) states [9,46], as shown in Fig. 9. For Ba_{1-x}Cr_xTe at all concentrations, the metallic nature of the majority spin results from p-d hybridization between p (Te) and 3d(Cr) states around $E_{\rm F}$, while the minority-spin states show a gap. Consequently, the $Ba_{1-x}Cr_xTe$ compounds exhibit half-metallic ferromagnetic behavior with spin polarization of 100%, making them potential candidates for use in spintronics applications.

3.3 Magnetic properties

The ferromagnetic state of $Ba_{1-x}Cr_xTe$ can be explained using the Zener carrier-mediated model [51], because the ferromagnetism is mediated by acceptor hole carriers generated by partially filled 3d (Cr) levels. Besides, the partially occupied 3d (Cr) states stabilize the ferromagnetic state configuration [52,53] associated with the double-exchange mechanism [54]. Therefore, the ferromagnetic state arrangement of $Ba_{1-x}Cr_xTe$ is stabilized by the contributions of p-dexchange and double-exchange mechanisms.

To describe the effect of the p-d exchange mechanism on the magnetic behavior, we calculated the total magnetic moment per Cr atom and local magnetic moments within the muffin-tin spheres of the relevant Cr, Ba, and Te atoms in $Ba_{1-x}Cr_xTe$ compounds. The total and local magnetic moments of Ba_{0.75}Cr_{0.25}Te, Ba_{0.5}Cr_{0.5}Te, and $Ba_{0.25}Cr_{0.75}Te$ are presented in Table 3, revealing that the total magnetic moment per Cr atom for each compound was equal to $4\mu_B$, principally due to the local magnetic moment of Cr atom. The total magnetic moment of $4\mu_B$ originates from the 3d (Cr) majority-spin states, which are partially filled with four electrons. On the other hand, the predicted magnetic moment of Cr atom is reduced to below $4\mu_{\rm B}$ and minor local magnetic moments are induced at Ba, Te, and interstitial sites due to the p-d exchange interaction between p (Te) and 3d (Cr) levels. For $Ba_{1-x}Cr_xTe$ at all concentrations, the magnetic spins of Cr and Te atoms have opposite sign, indicating antiferromagnetic interaction between Cr and Te. The interaction between Cr and Ba atoms is ferromagnetic for Ba_{0.75}Cr_{0.25}Te and Ba_{0.5}Cr_{0.5}Te, but becomes antiferromagnetic in the case of $Ba_{0.25}Cr_{0.75}Te$ compound.

4 Conclusions

We predicted the structural, electronic, and ferromagnetic properties of rocksalt BaTe doped with Cr atoms, i.e., $Ba_{1-x}Cr_xTe$ with compositions x = 0.25, 0.5, and 0.75, based on first-principles calculations in DFT with the GGA-WC and TB-mBJ exchange potentials using WIEN2k code. The electronic structure showed that the $Ba_{1-x}Cr_xTe$ compounds are half-metallic ferromagnetic with spin polarization of 100%. The total magnetic moment per Cr atom has integral value of $4\mu_B$, being principally formed by the main contribution of the local magnetic moment of Cr atom. The ferromagnetic state arrangement is stabilized by both p-dexchange and double-exchange mechanisms. The improved HM gaps of the $Ba_{1-x}Cr_xTe$ materials make them potential candidates for use in future spintronics applications.

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