

Modeling and Computation of a liquid-vapor bubble formation

André Galligo, Frédéric Lesage, Sebastian Minjeaud

▶ To cite this version:

André Galligo, Frédéric Lesage, Sebastian Minjeaud. Modeling and Computation of a liquid-vapor bubble formation. Université Côte d'Azur Complex Days, M. Argentina; S. Barland; P. Reynaud-Bouret; F. Cauneau; K. Guillouzouic; U. Kuhl; T. Passot; F. Planchon, Jan 2018, Nice, France. hal-01984782

HAL Id: hal-01984782

https://hal.science/hal-01984782

Submitted on 17 Jan 2019

HAL is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers. L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.

Copyright

Modeling and Computation of a liquid-vapor bubble formation

Galligo André, Lesage Frédéric, Minjeaud Sebastian

Abstract The Capillary Equation correctly predicts the curvature evolution and the length of a quasi-static vapour formation. It describes a two-phase interface as a smooth curve resulting from a balance of curvatures that are influenced by surface tension and hydrostatic pressures. The present work provides insight into the application of the Capillary Equation to the prediction of single nucleate site phase change phenomena. In an effort to progress towards an application of the Capillary Equation to boiling events, a procedure to generating a numerical solution, in which the computational expense is reduced, is reported.

André Galligo

Université Côte d'Azur, CNRS, Inria, LJAD, France.

e-mail: galligo@unice.fr

Frederic Lesage

Université Côte d'Azur, Center of Modeling, Simulation and Interactions.

e-mail: Frederic.Lesage@unice.fr

Sebastian Minjeaud

Université Côte d'Azur, CNRS, Inria, LJAD, France.

e-mail: Sebastian Minjeaud@unice.fr

1 Introduction

Boiling is a complex phase transitioning process in which a new liquid-vapor interface (bubbles) is created. It is an effective mechanism widely used in energy conversion industrial facilities. However, a precise descriptive/predictive model of bubble formations is needed to better understand its heat transfer characteristics. As an important first step, the question can be addressed from a quasi-static viewpoint. Indeed, at the interface, a classical conservative law governs the surface curvature of the generated bubble. This non-linear law links the difference of pressure and the surface tension effect. After normalization, it can be expressed at each point of the surface by the following equation: $\kappa_1 + \kappa_2 - Bz$ =constant; where B is a Bond number, summarizing the context, z is the elevation and $\kappa_1 + \kappa_2$ is twice the Gaussian mean curvature. Therefore, the study of a growing bubble is amenable to solving geometric computations and non-linear differentiable equation. The main difficulty is to design criteria to detect when the growing bubble will detach from the substrate's nucleation site. Inspired by experimental observations, the second author was able to develop [1,2] a successful computational strategy. The computational model implemented for the axis-symmetric case (which relies on 2D geometric computations), provides useful information, fully confirmed by experimental observations. Moving forward, we were able to speed up the computations and introduce the use of splines for representing the plane section of the surfaces.

In future work, we will address with similar tools the case for which the nucleation site is on an inclined plane; this geometry better models the one effectively used in industrial plants. Our next target is to better formalize our detachment criteria, in collaboration with an experimental team in Dublin, Ireland; then develop and implement a 3D model for a more general situation. Future studies will also include external stress terms such as cross flows and electrical fields in an effort to move towards realistic and accurate boiling models.

2 Young-Laplace Equation and Capillary Equation

We consider an infinitesimal section of a vapour formation interface defined by two principal radii of curvature in perpendicular planes, as illustrated in Fig. 1.

We suppose that the work done on the surface is considered to be equal to the work attributed to the pressure difference, as expressed by the Young-Laplace Equation:

$$\Delta p = \sigma(\kappa_1 + \kappa_2) \tag{1}$$

in which κ_1 and κ_2 are the curvatures corresponding to the principal radii of curvature in orthogonal planes, and σ is defined as the work per unit area required to produce an area variation of the interface.

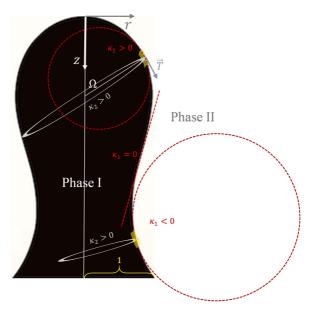


Fig. 1 Axisymmetric surface with smooth contour.

Applying the Young-Laplace pressure balance to the context of a single vapour formation event, we consider a surface that is axisymmetric. The surface can thus be described by a smooth curve $\mathcal C$ that defines the two-phase interface, as illustrated in Fig. 1. The horizontal coordinate is denoted by r and the vertical one, oriented downward from the apex, by z. We normalize

the problem by imposing a foot radius of unity, in other words the point $(\eta, 1)$ belongs to the curve \mathcal{C} .

The curvature κ_1 is in the z-r plane and the curvature κ_2 is in the plane orthogonal to the tangent unit vector **T**. Due to symmetry, the curvatures are equal at the apex origin, they are denoted by $\kappa_0 = \kappa_1 = \kappa_2$. Let g be the gravitational constant and $\Delta \rho$ the difference in density between the two phases, (1) implies the capillary equation:

$$2\sigma \kappa_0 - \Delta \rho gz = \sigma(\kappa_1 + \kappa_2), \quad \text{on } \mathcal{C}. \tag{2}$$

Introducing a dimensionless Bond number, $Bo = \Delta \rho r(\eta)^2 g/\sigma$, we get the adimensionalized form of the capillary equation:

$$2\kappa_0 = \kappa_1 + \kappa_2 + Bo \cdot z, \quad \text{on } \mathcal{C}. \tag{3}$$

2.1 Strategy for Solving the problem

By choosing a parametrization of the curve \mathcal{C} and explicit expressions of the curvatures, the equation (3) can be expressed as a second order Ordinary Differential Equation (ODE) with a free parameter κ_0 (the Bond number being determined by the physical properties of the two fluids). In order to simplify the bubble detachment problem, we decompose the difficulty by fixing a positive value for the parameter η and we consider the corresponding subproblem with the following boundary conditions:

$$(0,0) \in \mathcal{C}$$
 ; $(\eta,1) \in \mathcal{C}$ and $\frac{dz}{dr} = 0$ at the apex. (4)

Numerical procedures allow to compute a candidate curve solution for each value of the fixed parameter η . Roughly speaking, two boundary conditions are needed to determine the curve from the second order ODE (3) while the third one determines a value for the free parameter κ_0 . So, we obtain a familly of curves depending on the parameter η . Then we need a criterion to select the curve that best models the (physical) bubble detachment phenomena. We claim that the detachment profile is identified by noting that at a given height η , below a threshold, one smooth curve satisfies the boundary conditions (4) with at most one inflexion point. This is illustrated in Fig. 2.

The detachment height is then deduced through incremental increases in the parameter to maximize the height η till this property can be satisfied.

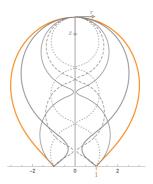


Fig. 2 Set of curves satisfying (3) with the boundary conditions (4) for fixed values of η

2.2 First numerical procedure

This first method is an established approach [1, 3]. The half (smooth) curve \mathbb{C} , is divided in two parts: first a graph of a function r(z) near the apex, then a graph of a function z(r) till the foot. Expressing the curvatures κ_1 and κ_2 in each case, relying on differential geometry formulas, gives rise to a pair of ODEs:

$$2\kappa_0 = \frac{z''(r)}{(z'(r)^2 + 1)^{3/2}} + \frac{z'(r)}{r\sqrt{1 + z'(r)^2}} + Bo \cdot z(r)$$
 (ODE1)

and

$$2\kappa_0 = \frac{-r''(z)}{(r'(z)^2 + 1)^{3/2}} + \frac{1}{r(z)\sqrt{1 + r'(z)^2}} + Bo \cdot z$$
 (ODE2)

For each η , we first choose a value of κ_0 . We integrate (ODE1) with the initial boundary conditions z(0) = 0 and z'(0) = 0. This is done till some value r_1 of r such that $z'(r_1) = 1$. We denote by $z_1 = z(r_1)$. Then we integrate (ODE2)

with the initial boundary conditions $r(z_1) = r_1$ and $r'(z_1) = 1$. This second integration is done until $z = \eta$. The $r(\eta)$ is computed, it will serve for a shooting method. Indeed, performing this process for different values of κ_0 , allows to find a minimum value of κ_0 for which $r(\eta) = 1$.

This numerical approach was successfully implemented using the software Mathematica.

2.3 Second numerical procedure

The half (smooth) curve \mathcal{C} is represented by a kind of arc-length parameterization $(0 \le t \le 1)$: r(t), z(t) with $|r'(t)|^2 + |z'(t)|^2 = \lambda^2$, depending on a new parameter λ . Again, we can express κ_1 and κ_2 , by similar differential geometry formulas, which provide a system of ODE with respect to the variable t:

$$\begin{cases} 2z'(t)r(t)z''(0) = -\lambda r(t)r''(t) + \lambda z'(t)^2 + Bo\lambda^2 z(t)z'(t)r(t), \\ r'(t)^2 + z'(t)^2 = \lambda^2. \end{cases}$$
 (5)

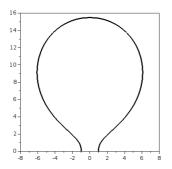
The boundary conditions can be expressed as

$$(r(0), z(0)) = (0, 0), \quad (r(1), z(1)) = (1, \eta), \quad z'(0) = 0.$$

Now, we can approximate the functions r(t) and z(t) by cubic spline functions on a subdivision of the interval (0,1) and express the Boundary Value Problem as a polynomial system of equations, which can be solved by standard techniques. The unknowns of the system are the coefficients of the two splines r and z and the parameter λ .

This second numerical approach was implemented using the software Scilab and is quite efficient since, for a given Bond number, it converges in few iterations of the nonlinear solver. Obviously, the curves obtained by the two numerical methods agree very well. Numerical results obtained with the second approach using a regular subdivision of 50 cells are presented in Fig. 3: at left, for a Bond number equal to 0.0052, the detachment height is estimated to 15.47 whereas at right, for a Bond number equal to 0.047, it is estimated to 7.90.

This second approach is still amenable when we replace the maximization parameter η by the volume of the bubble, an *a priori* more relevant physical quantity. The details will be given in a forthcomming article.



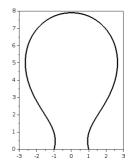


Fig. 3 Numerical solutions obtained for Bo = 0.0052 (left) and Bo = 0.047 (right)

3 Conclusion

Two numerical methods that solve the capillary equation are reported. The objective of the numerical treatments is to generate curves that model the interface of vapour formations growing from a nucleation site. The first method is an established approach and is used here to provide context and clarity to the solution procedure. The second method is a new refinement to the numerical solution of the Capillary equation. By implementing splines into the procedure, it reduces the computational expense and presents a stepping stone towards more accurate heat transfer predictions of single boiling events.

Acknowledgements

This work has been supported by the French government, through the UCA-JEDI Investments in the Future project managed by the National Research Agency (ANR) with the reference number ANR-15-IDEX-01.

References

 Frédéric J Lesage and Francis Marois. Experimental and numerical analysis of quasistatic bubble size and shape characteristics at detachment. *International Journal of Heat*

- and Mass Transfer, 64:53-69, 2013.
- 2. Frédéric J Lesage, James S Cotton, and Anthony J Robinson. Analysis of quasi-static vapour bubble shape during growth and departure. *Physics of Fluids*, 25(6):067103, 2013.
- 3. D. Gerlach, G. Biswas, F. Durst, and V. Kolobaric. Quasi-static bubble formation on submerged orifices. *International Journal of Heat and Mass Transfer*, 48(2):425 438, 2005.