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Quantum and dielectric Confinement Effects in Lower-Dimensional Hybrid Perovskite Semiconductors

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ABSTRACT

Hybrid halide perovskites are now superstar materials leading the field of low-cost thin film photovoltaics technologies. Following the surge for more efficient and stable 3D bulk alloys, multilayered halide perovskites and colloidal perovskite nanostructures appeared in 2016 as viable alternative solutions to this challenge, largely exceeding the original proof of concept made in 2009 and 2014, respectively. This triggered renewed interest in lower-dimensional hybrid halide perovskites and at the same time increasingly more numerous and differentiated applications. The present paper is a review of the past and present literature on both colloidal nanostructures and multilayered compounds, emphasizing that availability of accurate structural information is of dramatic importance to reach fair understanding of quantum and dielectric confinement effects. Layered halide perovskites occupy a special place in the history of halide perovskites, with a large number of seminal papers in the 1980s and 1990s. In recent years, the rationalization of structure-properties relationship has greatly benefited from new theoretical approaches dedicated to their electronic structures and optoelectronic properties, as well as a growing number of contributions based on modern experimental techniques. This is a necessary step to provide in-depth tools to decipher their extensive chemical engineering possibilities which surpasses the ones of their 3D bulk counterparts. Comparisons to classical semiconductor nanostructures and 2D van der Waals heterostructures are also stressed. Since 2015, colloidal nanostructures undergo a quick development for applications based on light emission. Although intensively studied in the last two years by various spectroscopy techniques, the description of quantum and dielectric confinement effects on their optoelectronic properties is still in its infancy.

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1. INTRODUCTION

3D hybrid halide perovskites, or ABX_3 compounds where A is a small organic (*e.g.* methylammonium MA^+ , formamidinium FA^+) or an inorganic cation (*e.g.* Cs^+ , Rb^+), B a metal Pb^{2+} , Sn^{2+} , Ge^{2+} and X a halogen I^- , Br^- , Cl^- , are now superstar materials for photovoltaic applications and have attracted worldwide efforts from the scientific community since mid-2012. These materials have opened a route towards low-cost manufacture of solar cells while offering currently certified conversion efficiencies of more than 23% at the level of the best known thin film technologies like CIGS (a copper indium gallium selenide material alloy), CdTe or Si and not far from monocrystalline silicon (25%).¹⁻⁷ Halide perovskite solar cell technologies use various three-dimensional (3D) alloys in mesoporous and planar architectures either with n-i-p or p-i-n junctions.⁸⁻¹¹ Solar cells with multiple junctions and 2 or 4 terminals, use also a 3D perovskite alloy associated with another perovskite material, CIGS but mostly silicon as a low band gap absorber.¹²⁻¹⁷

Recently, layered halide perovskites¹⁸⁻²⁰ and colloidal perovskite quantum dots (QD)^{21,22} have also revealed great potential for solar cells, although further progresses must be achieved to reach efficiencies on par with their bulk 3D counterparts. Very recently, halide perovskites also demonstrated attractive photo and electroluminescence either as bulk materials or colloidal nanostructures, leading to optically pumped laser devices, with potential for display technology or exhibiting attractive photo-detection capabilities. However, before industrialization of devices based on halide perovskites, a number of serious technological locks have still to be solved, *e.g.* related to photo-stability, degradation under ambient conditions, the potential toxicity of lead and upscaling of device size.⁵

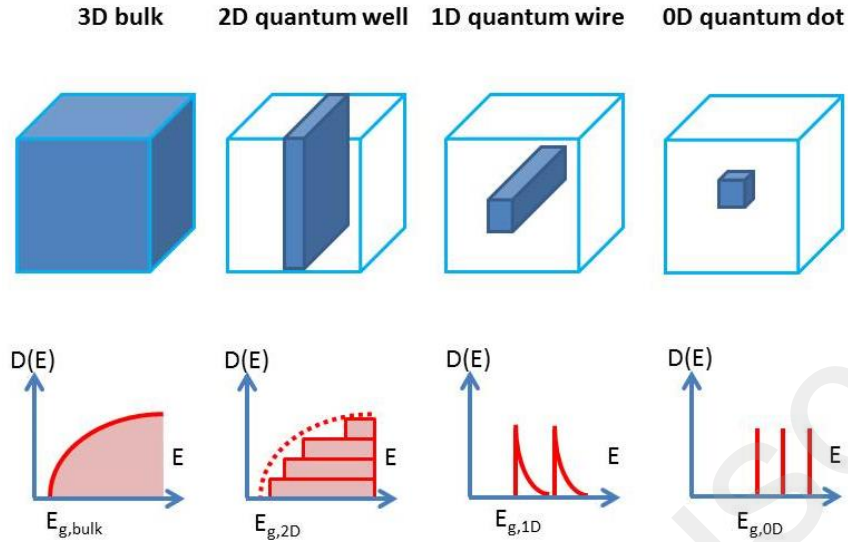


Figure 1. Schematic representation of a three-dimensional (3D) bulk semiconductor and the related two-dimensional (2D), one-dimensional (1D) and zero-dimensional (0D) quantum nanostructures. The joint density of states $D(E)$ between the conduction band (CB) and the valence band (VB) is represented as a function of the energy, assuming infinite confinement potentials and an effective mass model for both the CB and the VB. The quantum confinement also results in an increase of the electronic band gaps when compared to the bulk material. The 3D density of states (DOS) can be reconstructed by merging 2D DOS related to electronic subbands both in the CB and VB.

Prior to the perovskite fever, especially in the 1980s and 1990s, most experimental efforts on halide perovskites were focused on chemistry, structural and optical characterizations of layered halide perovskites. Chemical engineering of the organic part led to an extensive exploration of the possible structural organizations and optical properties of these hybrid compounds.^{23–25} Given the diversity of structural networks attainable with metal halides, we stress here that the minimum structural condition needed to be grouped under the terms “perovskite network or perovskite sub-network” is to maintain some component of

corner-sharing metal halide octahedra. Quantum and dielectric confinement effects were identified very early as a relevant factor underlying the unusual optoelectronic properties of monolayered and multilayered perovskites, among which the room temperature emission from excitonic resonances provides one notable example.^{26–29} The concept of quantum confinement is widely used for heterostructures and nanostructures made of classical semiconductors, especially quantum wells (QW). For epitaxially grown classical semiconductor nanostructures, the dimensionality and size of the nanostructure have direct consequences on the electronic band gap and joint Density Of States (DOS) close to the band gap (Figure 1).

A schematic QW picture was introduced for layered perovskites yielding a simple and efficient framework to perform analogies with classical semiconductor heterostructures.^{26,29,30} Progress on the understanding of their electronic structures^{31–33} was strongly connected to detailed experimental investigations of their excitonic properties. Moreover, D. Mitzi and coworkers obtained important conceptual advances, developing a technological approach for tin-based layered halide perovskites leading to the demonstration of transistor devices and to impressive anisotropic carrier transport properties.^{34–37} Researchers from Greece,^{29,38,39,33,23,40–45} and Japan,^{26,46,27,31,47–52,32,53–58} were the other most important contributors to the field at its early stage. The vast majority of the studies on low dimension halide perovskite structures were initially dedicated to monolayered systems such as A'_2BX_4 compounds (where A' is typically an extended non-conjugated primary ammonium cation, $R-NH_3^+$). These systems correspond to $n = 1$, where n is the number of vertically and periodically ordered perovskite octahedra sheets sandwiched between organic barriers. Monolayered compounds offer extended chemical engineering possibilities by

comparison with 3D halide perovskites. They also afford an ideal platform to investigate fundamental physical processes in a hybrid 2D system with self-organized vertical growth.^{59–}

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Until very recently, experimental studies on pure multilayered halide perovskites ($n > 1$) were scarce as a consequence of growth and purity issues and were basically limited to bilayered systems ($n = 2$).^{28,31,29,52,32,61,77,18,78,19,79–88} Thanks to new synthetic routes,⁸⁰ phase-pure higher n -members have recently been obtained leading to breakthroughs reported in Nature²⁰ and Science.⁸⁹ These achievements are related to three major experimental advances demonstrated for $A'_2A_{n-1}B_nX_{3n+1}$ multilayered perovskites known as Ruddlesden-Popper phases:

- Synthetic control to achieve phase purity and new methodologies for thin film material growth with well-defined crystallographic structures and orientations;
- Major progress in photoconversion efficiency of layered halide perovskites for photovoltaics from about 4% to 15.4%, along with increased stability against illumination and moisture;
- Detailed investigations of localized versus delocalized and light induced electronic states in thin films and ultrathin exfoliated crystals.

Similar demonstrations are now expected for many other multilayered halide perovskites.

All the recent progresses on multilayered halide perovskites reveal that a change of paradigm is occurring at the moment in a similar manner as the one observed for 3D halide perovskites in the past few years.⁹⁰ The understanding of the optoelectronic properties of multilayered halide perovskites may indeed greatly benefit from concepts developed in the

realm of conventional semiconductors, based on solid-state physics,^{64,65,67,90–94} and may open the route towards new applications. Such studies may provide guidance for new synthetic targets and open the route towards new applications as well as help answering a number of fundamental open questions. The possibilities offered by monolayered and multilayered perovskites for various applications are now intensively explored,^{86,88,95–102,102–140} as well as various fundamental properties beyond the initial studies essentially based on structural, excitonic and carrier transport properties.^{85,141–165}

Initial results on perovskite solar cells were obtained using hybrid halide perovskite QD^{1,2}, but these nanostructures were rapidly overshadowed by bulk thin films of 3D materials to enhance absorbance. Recently, colloidal perovskite QD clearly demonstrate potential for light emission and photovoltaic applications,^{21,166} which led to the resurgence of interest in these nanostructures. Colloidal halide perovskites nanostructures indeed show great promises for applications,^{167,168} and thus also deserve some attention as part of this review, even so quantum and dielectric confinements have been yet little explored by the scientific community in these new colloidal nano-objects. Beside colloidal QD, colloidal quantum wire (QWr), quantum rods (QR) and quantum nanoplatelets (QNPL) are also investigated. In this review, we choose the International System of Units (S.I.), unless explicitly stated otherwise, and use bold typeface for vector and tensor quantities in the present manuscript.

2. A CHEMICAL AND STRUCTURAL VIEWPOINT

Summary: Perovskites are defined by a framework of corner-sharing octahedra. The <100>, <110> and <111> layered perovskite series are obtained by cutting the 3D mother structure in planes perpendicular to the <100>, <110> and <111> directions. Most of the crystallographically characterized multilayered perovskites belong to the <100> series.

2.1. Perovskite networks

2.1.1. Introduction

The perovskite type structure ABX_3 , which refers to the mineral $CaTiO_3$, can be described as a three-dimensional (3D) anionic network of corner-sharing BX_6 octahedra counterbalanced by cations (Ca^{2+} in $CaTiO_3$) filling the space defined by eight adjacent corner-sharing BX_6 octahedra. In the case of halide-based perovskites, X^- is a halide, B^{2+} is a divalent metal and A^+ an inorganic cation such as Cs^+ , Rb^+ or an organic cation. In this last situation, the related compounds are called hybrid perovskites or organic-inorganic perovskites. While the archetype compound is $(CH_3NH_3)PbI_3$ ($CH_3NH_3^+$ = methylammonium, MA^+), a lot of other 3D hybrid perovskites ABX_3 are known where A^+ is MA^+ or the formamidinium (FA^+) organic cations, B^{2+} is Pb^{2+} , Sn^{2+} , Ge^{2+} , Cu^{2+} , Mn^{2+} , Cd^{2+} , Cr^{2+} , Zn^{2+} , etc., and X^- is I^- , Br^- or Cl^- .^{169,24} Moreover, a lot of derived compounds have been successfully prepared these recent years thanks to different substitutions on the A, B or X sites. Particularly, 3D perovskites based on mixed organic cation MA^+/FA^+ or even $MA^+/FA^+/Cs^+$ (triple cation approach), mixed metal cation (ex : Pb^{2+}/Sn^{2+}), and mixed halides (ex : I^-/Br^-) have been developed.⁸⁻¹⁰ Finally, it is worth noting that in the search of lead-free materials, the strategy consisting of replacing 2 Pb^{2+} cations by one M^{+} and one M'^{3+} cations has led to 3D double perovskite $A_2M'M''X_6$ of Elpasolite type.¹⁷⁰ Nevertheless, if the 3D structural dimensionality is retained in these mixed metal hybrid

perovskites, the electronic dimensionality is lowered.¹⁷¹ Most colloidal perovskite nanostructures share the same crystallographic structures as the corresponding 3D bulk perovskites. The cubic, tetragonal, and orthorhombic polymorphs are observed, but surface effects may adjust their relative stabilities.¹⁶⁷

The lower-dimensional perovskites are characterized by a non-3D anionic inorganic framework of corner-sharing octahedra, meaning that these networks can be described from the mother ABX_3 structure by the dimensional reduction concept: adding a certain amount of AX per ABX_3 unit, a decondensation of the 3D network will occur leading to a lower-dimensional perovskite network.¹⁷² For instance, the addition of one AX per ABX_3 unit leads to a A_2BX_4 compound based on a 2D corner-sharing octahedra network. Thus, a lot of different kinds of low-dimensional perovskite networks can be imagined, but with the exception of a few atypical cases, all known networks are mainly the $\langle 100 \rangle$ - and $\langle 110 \rangle$ -oriented layered perovskites, and to a less extent the $\langle 111 \rangle$ -oriented perovskites (Figure 2), and several 1-D networks which derive from the layered perovskites. The description of lower-dimensional perovskites in terms of $\langle 100 \rangle$, $\langle 110 \rangle$ and $\langle 111 \rangle$ families had been reported by D. Mitzi as soon as in 1999,^{24,173} and updated by the same author in 2016.²⁵ The different networks of each of these families can be obtained by cutting the ABX_3 mother structure in planes perpendicular to the $\langle 100 \rangle$, $\langle 110 \rangle$ and $\langle 111 \rangle$ directions, every n, m or q layers leading to B_nX_{3n+1} , B_mX_{3m+2} and B_qX_{3q+3} anionic frameworks for the $\langle 100 \rangle$ -, $\langle 110 \rangle$ - and $\langle 111 \rangle$ -oriented families, respectively.

2.1.2. The $\langle 100 \rangle$, $\langle 110 \rangle$ and $\langle 111 \rangle$ layered perovskite series

Another way to describe the $\langle 100 \rangle$ - and $\langle 110 \rangle$ -oriented low-dimensional perovskites (Figure 2) consists of removing neutral « $B^{(II)}X_2$ » units from the mother structure, or substituting $(B^{(II)}X)^+$ units by organic cations. This point of view can be very interesting to design and

prepare new hybrid perovskites. In the case of the $\langle 100 \rangle$ -oriented family, the removal of one BX_2 plane every $n+1$ layers leads to B_nX_{3n+1} ($B_{(n+1)}X_{3(n+1)} - BX_2$) networks. For instance, the removal of one BX_2 plane every three layers ($A_3B_3X_9$) leads to the $A_3B_2X_7$ bilayer perovskite. The general formula of such $\langle 100 \rangle$ -oriented hybrid perovskites are $(A')_2(A)_{n-1}B_nX_{3n+1}$ or $(A'')(A)_{n-1}B_nX_{3n+1}$. When the interlayer cations is a monocation (A'^+), typically a primary ammonium cation $R-NH_3^+$, a head-to-tail arrangement of organic cations in the interlayer space occurs, involving that two adjacent layers are staggered to each other, usually by $(\frac{1}{2} \ \frac{1}{2})$. When the interlayer is a dication (A''^{2+}), typically a primary diammonium cation $(NH_3-R-NH_3)^{2+}$, the staggered situation can also be encountered. However, dications favor the eclipsed situation of consecutive layers (Figure 2). Finally, when $n > 2$, the perovskite network can also be considered as resulting from the combination of $(n-2)$ BX_3 inner layers, (as in the mother 3D structure) and two BX_4 outer layers, here also leading to the general formula of B_nX_{3n+1} . In the case of the $\langle 110 \rangle$ -oriented family, the substitution of $(BX)^+$ cationic units in the (110) planes by organic cations (A'^+ or $\frac{1}{2} A''^{2+}$) every $m+1$ layers leads to B_mX_{3m+2} ($B_{(m+1)}X_{3(m+1)} - BX$) networks. For instance, the substitution of one $(BX)^+$ by one A^+ cation every three layers ($A_3B_3X_9$) leads to the $A_4B_2X_8$ layered perovskite. Compounds belonging to this class of hybrid perovskites are quite rare, and only one series with $m = 1$ to 4 has been described,³⁵ the general formula being $(A')_2(A)_mSn_mI_{3m+2}$ where A^+ and A''^{2+} are methylammonium and iodoformamidinium, respectively. The $m = 2$ member of this $\langle 110 \rangle$ family is the most common since a few have been described recently. The formula can be $(A')_2(A)_2B_2X_8$, as in the case of the iodostannate hybrid, $(A)_4B_2X_8$ (one example: $A^+ = \text{guanidinium}$; $X = I$; $M = \text{Pb}^{174}$ or Sn^{175}), or $(A'')_2B_2X_8$ perovskites (A'' : dications, Figure 2). The $m = 1$ member of the $\langle 110 \rangle$ series, BX_5 , is special since it is a 1-D network consisting of trans-connected BX_6 octahedra. This network has been

crystallized either with two kinds of cation, as in $(\text{NH}_2\text{C}(\text{I})=\text{NH}_2)(\text{NH}_2\text{CHNH}_2)\text{SnI}_5$,¹⁷⁶ or by one kind of cation, as in $(\text{CH}_3\text{SC}(\text{=NH}_2)\text{NH}_2)_3\text{PbI}_5$.⁴³ The case of the $\langle 111 \rangle$ -oriented perovskites is particular because there is no known 2D networks based on divalent B(II) metal ions, but only with trivalent M(III) ions ($\text{M} = \text{Bi}, \text{Sb}$). Thus, the related materials $(\text{CH}_3\text{NH}_3)_3\text{Bi}_2\text{Br}_9$ ¹⁷⁷ or $(\text{NH}_4)_3\text{Sb}_2\text{I}_9$ ¹⁷⁸ based on $\text{M}^{(\text{III})}_2\text{X}_9$ $q=2$ $\langle 111 \rangle$ networks can be conceptually obtained by the substitution of 3 B^{2+} cations of the mother structure ($\text{A}_3\text{B}_3\text{X}_9$) by two M^{3+} cations, the vacancies being ordered in (111) planes (Figure 2).

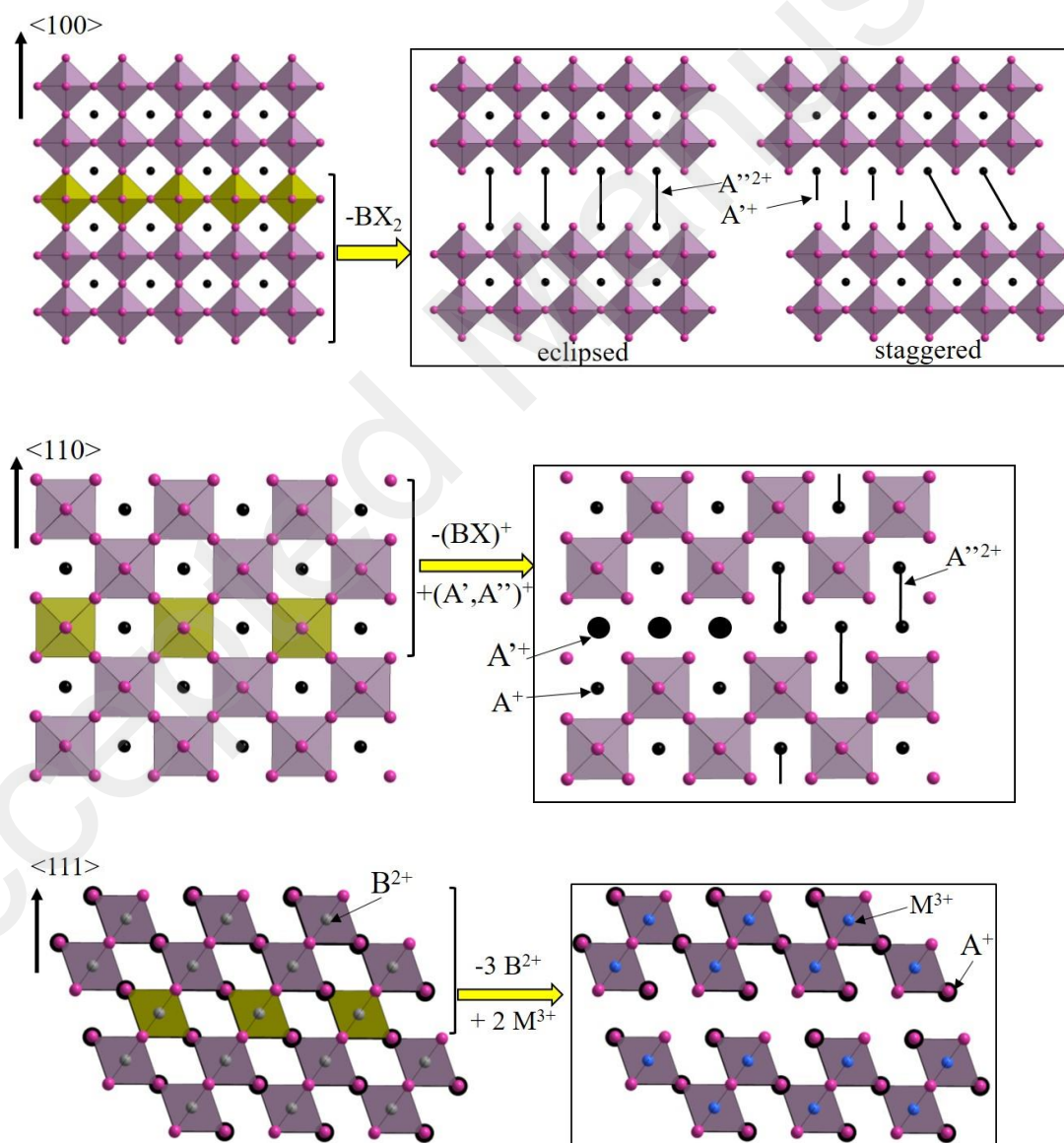


Figure 2. The $\langle 100 \rangle$ - (top), $\langle 110 \rangle$ - (middle) and $\langle 111 \rangle$ -oriented (bottom) low-dimensional perovskites conceptually defined from the mother ABX_3 perovskite structure (left) by elimination or substitution processes. A^{+} and A'^{2+} are monocations and dications, respectively. B^{2+} and M^{3+} are divalent and trivalent metal ions, respectively. (top) For the $\langle 100 \rangle$ -oriented family, the removal of one BX_2 plane every $n+1$ layers leads to B_nX_{3n+1} networks. (middle) For the $\langle 110 \rangle$ -oriented family, the substitution of $(BX)^+$ cationic units in the (110) planes by organic cations (A'^{+} or $\frac{1}{2} A'^{2+}$) every $m+1$ layers leads to B_mX_{3m+2} networks. (bottom) For the $\langle 111 \rangle$ -oriented perovskites, $M^{(III)}_2X_9$ $q=2$ $\langle 111 \rangle$ networks can be conceptually obtained by the substitution of 3 B^{2+} cations of the mother structure ($A_3B_3X_9$) by two M^{3+} cations, the vacancies being ordered in (111) planes.

If substitutions on the B and X sites of the 3-D BX_3 perovskite network have led to a lot of derived 3-D perovskites, such substitutions have also been carried out for the low-dimensional perovskites, particularly for the halide component. The preparation of a mixed halide solid solution is of high interest to tune electronic, optical or magnetic properties of corresponding hybrids. As a first example, we can mention the early work of Y. Susuki and H. Kubo (1983) who analyzed the Cl^- and Br^- distribution in the quasi-two-dimensional perovskite ferromagnets $(CH_3NH_3)CuCl_{1-x}Br_x$.¹⁷⁹ Later, in order to clarify the role of the halide in the superexchange mechanism of halochromate perovskite ferromagnets, C. Bellitto et al. reported on the mixed halide material $(C_6H_5CH_2NH_3)CrBr_{3.3}I_{0.7}$ exhibiting a 3-D ferromagnetic ordering at 51 K.¹⁸⁰ In the 1990s, G. C. Papavassiliou and N. Kitazawa brought an important contribution to the field by investigating the mixed crystals of $(A)_2PbBr_xI_{4-x}$ and $(A)_2PbCl_xBr_{4-x}$ ($A = C_6H_5CH_2NH_3^+$,^{181,182} $C_{10}H_{21}NH_3^+$)²⁹ showing the shift of the exciton band to the blue region by replacing I by Br, or

Br by Cl. More recently, M. G. Kanatzidis et al. prepared the first mixed halide multilayered 2D perovskite $(EA)_4Pb_3Br_{10-x}Cl_x$ ($\langle 100 \rangle/n = 3$; $EA^+ =$ ethylammonium). They show that the content of Cl⁻ (x) has a drastic influence on the distortions of the perovskite network and consequently on the luminescence properties (white emission) of related materials.⁸⁴ Finally, we can notice the first hybrid perovskite containing a pseudohalide anion, $(CH_3NH_3)_2PbI_2(SCN)_2$. First prepared as crystallized powders and thin films for solar cells application,^{183,184} its crystal structure has unambiguously revealed the layered nature of this compound, the terminal iodides of the ideal $n = 1$ PbI_4^{2-} layers being substituted by the SCN^- pseudohalide anions leading to $PbI_2(SCN)_2^{2-}$.¹⁸⁵ If a low band gap was first assigned to this material (1.55 eV), it has been further shown that this value was wrong,¹⁸⁵ and finally measured around 2 eV.¹⁸⁶

2.1.3. Atypical perovskite networks

Atypical hybrid perovskites are based on corner-sharing octahedra networks which cannot be deduced from the ABX_3 mother structure by simple cuts along (100), (110) or (111) planes. The first example is the series of “rooflike” networks. These networks are isomers (same formula as $BX_4 = BX_{4/2}X_2$ resulting from the corner-sharing of four halides, while two halides are terminal), including two types of octahedra, those with the two terminal halides in trans-position, as in the $n = 1$ members of the $\langle 100 \rangle$ series, and those with the two terminal halides in cis-position, as in the $m = 2$ member of the $\langle 110 \rangle$ series (Figure 3). They can be described as corrugated layers of « $n' \times n''$ rooflike » type where n' and n'' are the number of octahedra composing each half of the roof. Thus, a 4×4 rooflike network is encountered in $(NH_3-(CH_2)_5-NH_3)SnI_4$ ¹⁸⁷ (Figure 3, left), while a 3×3 rooflike network is encountered in $\alpha(H_2DMEN)PbBr_4$ ⁹⁵ (DMEN = 2-(dimethylamino)ethylamine) (Figure 3, middle). Recently, the first asymmetrical 3×2 rooflike

network has been crystallized with both formamidinium and guanidinium (Gua^+) cations, $(\text{FA})(\text{Gua})\text{PbI}_4$.¹⁸⁸ Another example is the Pb_4I_{14} network of Figure 3 (right) found in $(\text{MA},\text{HEA})_6\text{Pb}_4\text{I}_{14}$ (MA^+ , methylammonium, HEA^+ , ethanolammonium).¹⁸⁹ It can be easily described from the ABX_3 mother structure by the substitution of one $(\text{PbI})^+$ unit every five units by an organic cation A^+ leading to $\text{A}_6\text{Pb}_4\text{I}_{14}$ ($= \text{A}_5\text{Pb}_5\text{I}_{15} - (\text{PbI})^+ + \text{A}^+$). In the structure, the (PbI) vacancies are ordered, leading to a unit cell volume 5/4 bigger than the unit cell volume of $(\text{CH}_3\text{NH}_3)\text{PbI}_3$ (Figure 3, right). From a structural point of view, the perovskite network remains 3-D because the corner-sharing mode occurs in the three directions of space. However, as underlined by the red color of crystals, the electronic dimensionality is certainly lowered. This can be rationalized considering that chains of *trans*-connected octahedra are encountered only along the c direction, while along the $(2a + b)$ and $(a - 2b)$ directions, an alternation of four *trans*-connected octahedra and one Pb^{2+} vacancy occurs. Finally it must be noted that a solid solution can be obtained and it is shown that this material, preferentially named $(\text{MA},\text{HEA})_{1+x}\text{Pb}_{1-x}\text{I}_{3-x}$ to highlight the lead and iodide deficient PbI_3 mother network, can exist in the $x = 0$ to 0.2 range. $x = 0$ corresponds to the $(\text{CH}_3\text{NH}_3)\text{PbI}_3$ while $x = 0.2$ corresponds to the Pb_4I_{14} ($\text{Pb}_{0.8}\text{I}_{2.8}$) based compound where only organic cations fill cavities. For $0 < x < 0.2$, cavities contains organic cation as well as lead and iodine.¹⁸⁹

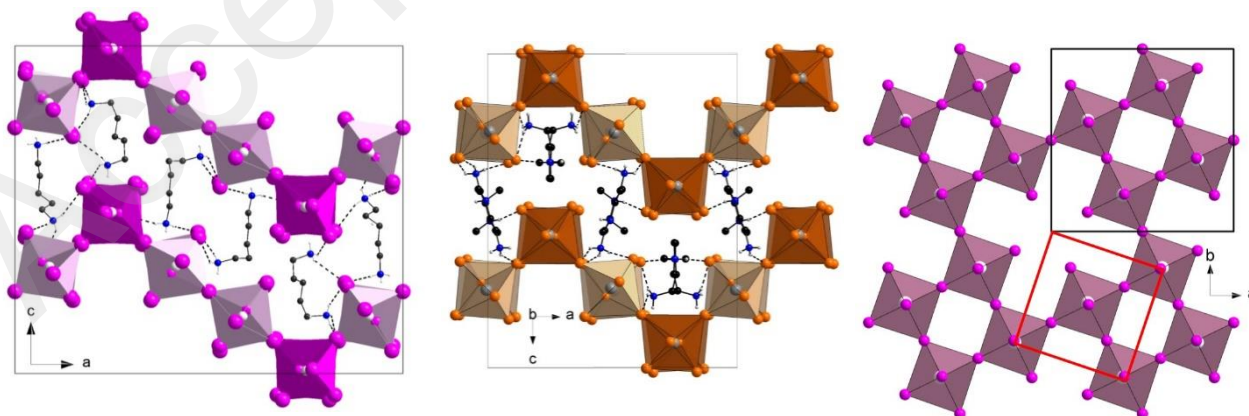


Figure 3. Atypical perovskite networks: (left) 4×4 rooflike network in $(\text{NH}_3\text{-(CH}_2)_5\text{-NH}_3)\text{SnI}_4$, (middle) 3×3 rooflike network in $\alpha(\text{H}_2\text{DMEN})\text{PbBr}_4$ (DMEN = 2-(dimethylamino)ethylamine), and (right) Pb_4I_{14} ($\text{Pb}_{0.8}\text{I}_{2.8}$) lead and iodide deficient BX_3 perovskite network in $(\text{HEA}, \text{MA})_{1.2}\text{Pb}_{0.8}\text{I}_{2.8}$ (HEA^+ : ethanolanmonium; MA^+ : methylammonium).

2.1.4. Halometallate structures with perovskite sub-networks

Perovskite networks are one sub-class of the more general family of B(II)/X(-I) halometallate networks which can be defined as anionic networks of corner, edge or face-sharing B_xX_y polyhedra. Among these networks, some are built in a large amount through the corner-sharing mode of octahedra, which is the signature of perovskite sub-networks. In Figure 4 three of them are reported. The first two structures are 2-D anionic networks of corner- and face-sharing octahedra. However, they are better described as 1-D $\text{B}_n\text{X}_{4n+2}$ perovskite ribbons^{190,191,192} linked together by B^{2+} cations. In $(\text{PPA})\text{Pb}_2\text{I}_7$ (PPA^+ = phenylpropylammonium), two octahedra thick ribbons of $(\text{Pb}_2\text{I}_{10})$ are alternatively linked to four octahedra thick ribbons of $(\text{Pb}_4\text{I}_{16})$ by Pb^{2+} leading to the Pb_2I_7 network (or $\text{Pb}_8\text{I}_{28} = \text{Pb}_2\text{I}_{10} + \text{Pb}_4\text{I}_{16} + 2 \text{Pb}^{2+}$). In $(\text{PBA})\text{Pb}_2\text{I}_7$ (PBA^+ = phenylbutylammonium), three octahedra thick ribbons of $(\text{Pb}_3\text{I}_{14})$ are linked together by Pb^{2+} leading to the Pb_2I_7 network (or $\text{Pb}_4\text{I}_{14} = \text{Pb}_3\text{I}_{14} + \text{Pb}^{2+}$) (Figure 4). If these networks are 2-D from a structural point of view, DFT calculations clearly show that there is a quantum confinement effect: the calculated band edges are flat along the [010] confinement direction, while the electronic wave functions corresponding to the states at the band edges are confined to the regions of ribbons of corner-sharing octahedra.¹⁹³ Thus, from

an electronic point of view, these materials can be considered as 1-D and comparable to 1-D perovskites B_nX_{4n+2} . The first member of this series characterized by B_nX_{4n+2} ribbons linked by B^{2+} cations is the B_3X_{10} network resulting from B_2X_{10} ($n' = 2$) ribbons fused by B^{2+} ions, as found for instance in $(\text{AESBT})\text{Pb}_3\text{I}_{10}$ ¹⁹⁴ (AESBT = 5-ammoniummethylsulfanyl-2,2'-bithiophene) or $((\text{CH}_3)_3\text{S})\text{Pb}_3\text{Br}_{10}$, this last material interestingly exhibiting a broadband emission with a massive Stokes shift.¹⁹⁵

Halometallate networks can also result from corner and edge-sharing octahedra. For instance in $(\text{H}_2\text{Aepz})_3\text{Pb}_4\text{I}_{14}$ (Aepz = 2-(2-aminoethyl)-pyrazole)¹⁹⁶ the Pb_4I_{14} bilayer network can be described as perovskite ribbons linked together through edge-sharing mode (Figure 4, right). In the same chemical system $\text{H}_2\text{Aepz-Pb-I}$, the $n = 1$ member of the $\langle 100 \rangle$ -oriented family has also been obtained, $(\text{H}_2\text{Aepz})\text{PbI}_4$. While it is well known in the $\langle 100 \rangle$ series that the band gap decreases from the monolayer BX_4 to the bilayer B_2X_7 compounds (see section 3.1), the band gap of the bilayered compound $(\text{H}_2\text{Aepz})_3\text{Pb}_4\text{I}_{14}$ is larger than the band gap of the monolayer compound $(\text{H}_2\text{Aepz})\text{PbI}_4$. This is most probably indicative of effect of quantum confinement in the regions of ribbons of corner-sharing octahedra, meaning that the electronic dimensionality doesn't correspond to the 2-D structural dimensionality.

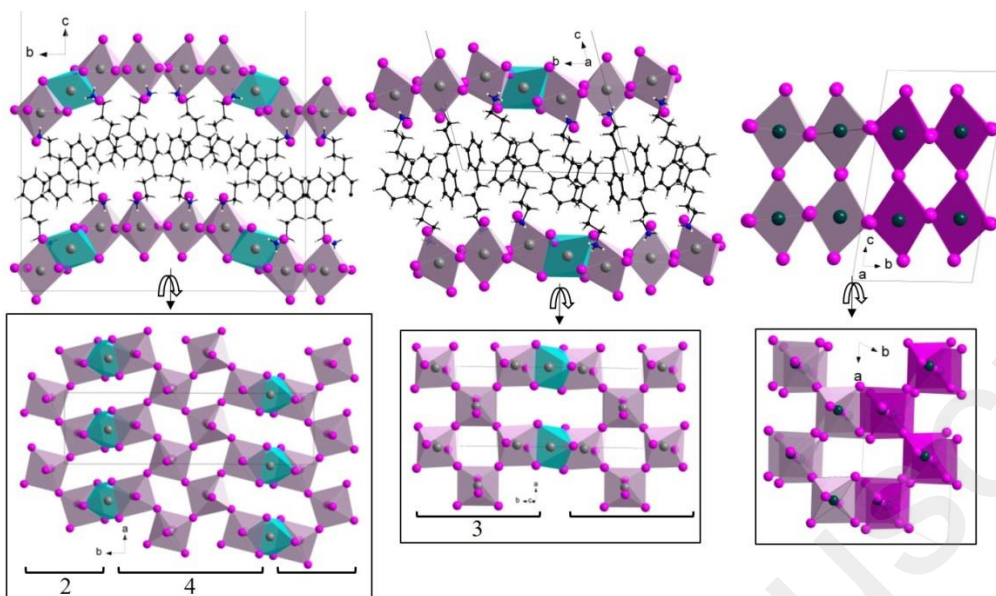


Figure 4. Layered halometallate networks built from ribbons of corner-sharing octahedra linked together either by Pb^{2+} ions (blue octahedra), in $(PPA)Pb_2I_7$ (PPA^+ = phenylpropylammonium) (left) and $(PBA)Pb_2I_7$ (PBA^+ = phenylbutylammonium) (middle), or through edge-sharing mode in $(H_2Aepz)_3Pb_4I_{14}$ ($Aepz = 2-(2-aminoethyl)-pyrazole$) (right). Adjacent ribbons appear as light pink octahedra and pink octahedra.

2.2. Formation of perovskite networks / choice of organic cations

2.2.1. Introduction

Due to the labile halide ligands, metal halide complexes are involved in fast dissociation and association in solution. The large number of different anionic halometallate anions, 0-D, 1-D, 2-D or 3-D that are found in the solid state, can be explained by a condensation of BX_n polyhedra during the crystallization process depending on several parameters. The stoichiometry of reagents is one of these parameters. Ideally, we could expect that the mixture of 1 BX_2 and 1

AX affords a ABX_3 compound, and that 1 BX_2 and 2 AX gives a A_2BX_4 compound, but it is not a rule. Many examples show that two or more different compounds can be obtained together, as in the system $SnI_2/CH_3-(CH_2)_{11}NH_3^+$ where both $(CH_3-(CH_2)_{11}NH_3)SnI_4$ and $(CH_3-(CH_2)_{11}NH_3)SnI_3$ are crystallized.¹⁹⁷ Moreover, as syntheses are often carried out with an excess of X^- anion in order to easily solubilize the BX_2 salts, the concentration of X^- is also a key parameter.¹⁹⁸ Finally, the importance of the stoichiometry is nicely illustrated by the preparation of different members of the layered $\langle 100 \rangle$ -oriented perovskites $(CH_3(CH_2)_3NH_3)_2(CH_3NH_3)_{n-1}Pb_nI_{3n+1}$ ($n = 1$ to 4). Thus, the $n = 1-4$ materials are prepared as pure phases in HI solution from the stoichiometric reaction of CH_3NH_3Cl and PbO , while half the expected stoichiometry is used for the *n*-butylamine (limiting reagent).⁸⁰ If in that case the formation of multilayered perovskites was expected because suitable cations were used, the nature of the BX_3^- or BX_4^{2-} networks of ABX_3 or A_2BX_4 materials prepared from any A^+ organic cation, is not predictable in many cases. The nature of the organic cation(s) is a crucial key point in the formation of the anionic halometallate network. For instance, in the case of a ABX_3 compound, BX_3^- can be either a 3-D perovskite network, a 1-D chain of face-sharing octahedra,¹⁹⁹ a 2-D sheet of MoO_3 type,¹⁹⁷ etc... all these networks being isomers. Besides the nature of A^+ , the pressure or temperature can influence the nature of the anionic network. For instance, the network found in $(FA)PbI_3$ is either a 1-D chain of face-sharing octahedra at room temperature, or a 3-D perovskite above 150 °C.¹⁹⁹ Different types of perovskite networks can also be isomers. Thus, the BX_4^{2-} anion can be the $n = 1$ member of the $\langle 100 \rangle$ -oriented family, the $m = 2$ member of the $\langle 110 \rangle$ -oriented family, or one of the exotic 4×4 , 3×3 or 3×2 rooflike networks (Figure 3). All these networks can be stabilized by non-bulky $R-NH_3^+$ or $^+H_3N-R-NH_3^+$ primary ammoniums, meaning that the nature of the R groups, particularly their ability to make intermolecular interactions, will template one

inorganic anion rather than one of its isomers. As another example of isomeric networks, we can cite the $B_2X_7^{3-}$ anion found as the $n = 2$ member of the $\langle 100 \rangle$ -oriented family (Figure 2), the atypical lead and iodide deficient $B_{1-x}X_{3-x}$ ($x = 0.2$) network (Figure 3), and the above described halometallate structures with perovskite sub-networks (Figure 4).

2.2.2. $\langle 100 \rangle$ -oriented layered perovskites

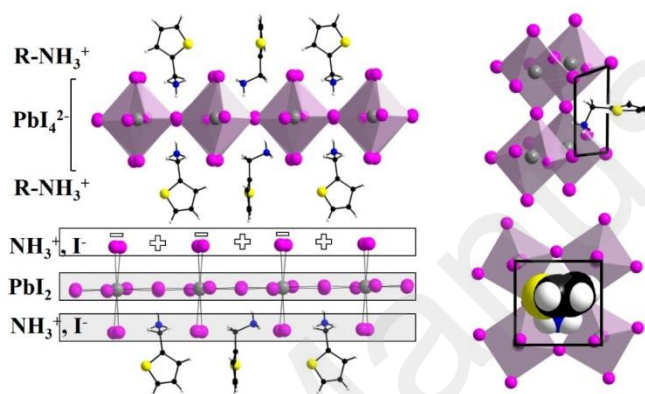


Figure 5. Organic-inorganic interface in a $n = 1$ $\langle 100 \rangle$ -oriented perovskite (example of $(2\text{-methylammonium-thiophene})_2\text{PbI}_4$) highlighting the cationic size (right) and charge (left) criteria necessary for organic cations to template the perovskite network.

Not only the $\langle 100 \rangle$ -oriented perovskites are by far the more common family of hybrid perovskites, but the prediction of achievement of such structures by selecting suitable cations is certainly much higher than for any other halometallate network. These B_nX_{3n+1} networks are characterized by terminal halides which define a square-like grid of anionic charges. On the other hand, suitable organic cations, typically non bulky primary (di)ammonium $R\text{-NH}_3^+$ or $^+\text{H}_3\text{N-R-NH}_3^+$, will form organic layers with ammonium heads defining a set of positive charges counter-balancing the anionic charges of the inorganic layers. Thus, if such hybrids are well defined as perovskite layers separated by organic cations, they can also be considered as neutral blocks ($n > 1$) or neutral PbI_2 layers ($n = 1$) ended by NH_3^+, X^- planes in order to highlight the

ionic interaction at the organic-inorganic interface that finally guides the formation of such structures. This is illustrated in Figure 5 for the ((2-methylammonium)thiophene)₂PbI₄ $n = 1$ compound.²⁰⁰ Besides the necessity to have a cationic head, the size of organic cations is also a key point. There is no restriction in length, meaning that the distance between adjacent perovskite layer can be tuned depending on the length of selected organic cations, as shown by a series of perovskites based on long alkylammonium cations.²⁰¹ However, their width, or cross-sectional area must fit in the surface defined by four adjacent terminal halides as illustrated in Figure 5 for the thiophene derivative cation. The great number of BX₄²⁻ ($n = 1$) based perovskites, especially for Pb²⁺ and Sn²⁺ ions is also explained by some degree of freedom of this soft network in order to accommodate different organic cations and hydrogen bonding schemes.

Three main structural features characterize the degree of distortion of the network. The first one is relative to the geometry of the BX₆ polyhedron. While an octahedral nature reveals the stereo-inactivity of the Pb²⁺ or Sn²⁺ ns² lone pair, a non-octahedral geometry usually reveals the lone pair stereo-activity, as found in (EDBE)PbCl₄ (EDBE : 2,2'-(ethylenedioxy)bis(ethylammonium)).²⁰² This structural feature is considered as responsible of self-trapped excitons in (EDBE)PbCl₄ and some other chloro- and bromo-plumbate perovskites, and consequently to a broadband white emission of these promising perovskites. The two other degrees of distortion are the in-plane and out-of-plane distortions.²⁰³ These distortions as well as the octahedral nature of the metal complexes, have direct impact on the electronic properties of related materials. The weaker they are the smaller is the band gap.^{76,203–206}

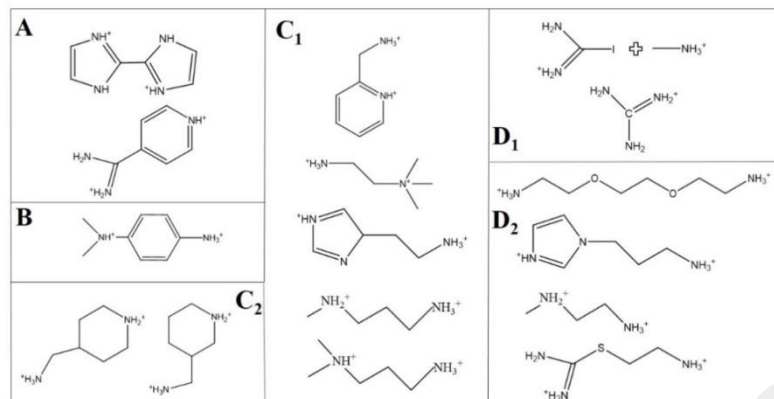


Figure 6. A, B, C₁ and C₂: Non all primary ammonium dications affording <100>-oriented perovskites (A, no primary ammonium; B, one primary ammonium; C₁ and C₂, one CH₂-NH₃⁺ fragment). D₁ and D₂: cations affording <110>-oriented perovskites.

If most of the <100>-oriented perovskites have been stabilized by primary ammonium mono- or dications R-NH₃⁺ or ⁺H₃N-R-NH₃⁺, other cations can template these perovskite networks. With the exception of the non-primary ammonium biimidazolium and 4-amidinopyridinium cations (Figure 6, A), leading to PbI₄²⁰⁷ and PbBr₄²⁰⁸ based layered perovskites, respectively, and the N,N-dimethylphenylene-p-diammonium (DPDA²⁺) dication in (DPDA)PbI₄ (Figure 6, B), all are dications that possess one -(CH₂)-NH₃⁺ fragment.²⁰⁹ The C₂ cations shown in Figure 6 have stabilized a series of multilayered (n>1) perovskites,²⁰⁴ and it is worth noting that they have all the same orientation in the interlayer space, meaning that all primary ammonium are in interaction with a perovskite sheet while all piperidinium fragments are in interaction with the adjacent perovskite sheet (Figure 7). In contrast, in the n = 1 perovskite, a head-to-tail situation of cations in the interlayer space is encountered, as well as between cations which are on both sides of the perovskite layer, as illustrated in Figure 7. This is

usually the best compromise for the inorganic layer to host the bulky non-primary ammonium head.²⁵

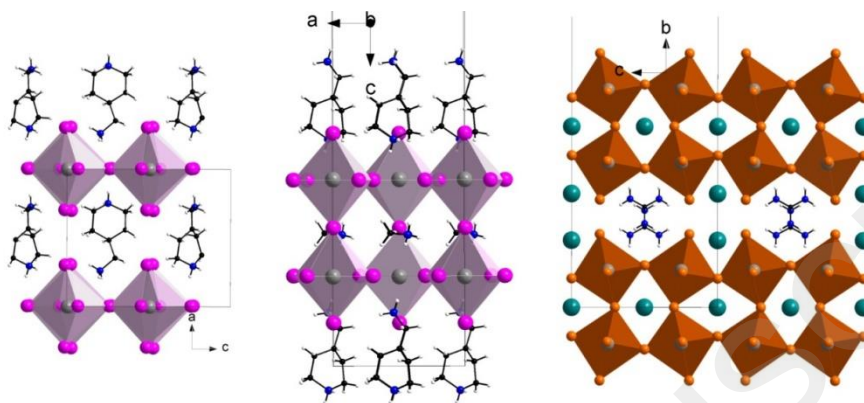


Figure 7. Structures of the $n = 1$ (4AMP) PbI_4 (left) and the $n = 2$ (4AMP)(MA) Pb_2I_7 (middle) perovskites (4AMP: 4-(methylammonium)piperidinium) showing the different molecular arrangement in the interlayer space. Structure of the $Cs_2(C(NH_2)_3)Pb_2Br_7$ perovskite with alternating ordering of cations in the interlayer space (right).

The first layered perovskite incorporating two kinds of well-ordered cations in the interlayer space is the $n = 1$ iodostannate hybrid (5FPEA,NEA) SnI_4 (5FPEA⁺ = 2,3,4,5,6-pentafluorophenethylammonium, NEA⁺ = 2-naphtylenethylammonium). More recently, other examples including multilayered ($n > 1$) hybrids have been reported: Cs(Gua) PbX_4 ($X = Br, I$) and $Cs_2(Gua)Pb_2Br_7$ (Gua = guanidinium, Figure 7)²¹⁰ as well as (Gua)(MA) Pb_nI_{3n+1} ($n = 1, 2, 3$).⁸³ Interestingly, these perovskites adopt a higher crystal symmetry than most 2D perovskites, which consequently leads to lower band gap with the same perovskite layer thickness n .⁸³

The polarizability is also a main characteristic of organic cations that can influence the properties of layered hybrid perovskites.²¹¹ More specifically, depending on the nature of the interlayer organic cation, the electronic and dielectrical confinements can be tuned (vide infra). Organic cations based on neutral electron rich conjugated cycles will be more polarizable, thus

decreasing the electronic and dielectric confinement in perovskite layers. Several hybrid perovskites based on acene alkylammonium or thiophene alkylammonium cations have been reported. Besides the well-known 2-phenethylammonium cation based on a single benzene ring,³⁸ and its functionalized derivatives,²¹² cations based on a naphthyl group (two fused benzene rings), have also led to $n = 1$ perovskites.²¹³ Thiophene alkylammonium type cations with one,²⁰⁰ two,¹⁹⁴ or four²¹⁴ thiophene units have also led to $n = 1$ hybrid perovskites. Another strategy to tune the electronic and dielectric confinement in these layered materials is through intercalation of highly polarizable molecules. The intercalation process of neutral molecules within layered perovskites have first been demonstrated by D. Mitzi for a series of iodostannate perovskites incorporating benzene or hexafluorobenzene molecules.²¹⁵ More recently, the reversible incorporation of diiodide molecules into a $n = 1$ layered perovskite through a post-synthetic process has been demonstrated: the exposure to vapor of diiodide of $(\text{IC}_6\text{H}_{12}\text{NH}_3)\text{PbI}_4$, where $\text{IC}_6\text{H}_{12}\text{NH}_3^+$ is the iodo-hexylammonium cation, led to $(\text{IC}_6\text{H}_{12}\text{NH}_3)\text{PbI}_4 \cdot 2\text{I}_2$ (Figure 8). Furthermore, the authors demonstrated that the dielectric characteristics of the intercalated compound are very different from those of the non-intercalated compound, involving a reduction of the electronic confinement in these 2D hybrid perovskites.⁶⁹

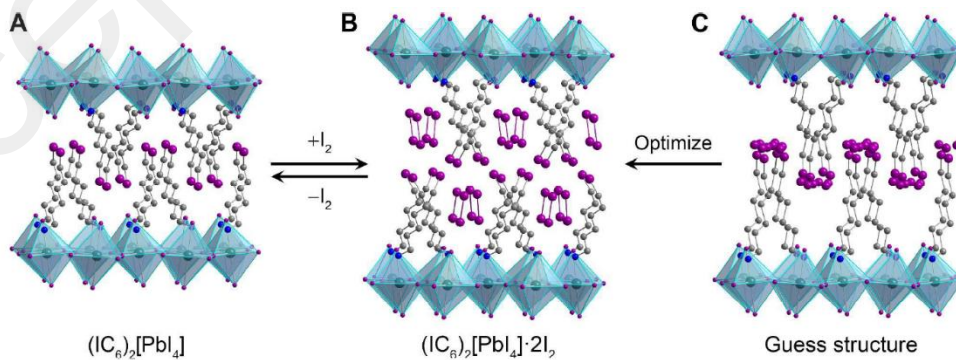


Figure 8. Intercalation of I_2 in the interlayer space of the hybrid perovskite $(IC6)_2PbI_4$ (A) ($IC6^+$: iodohexylammonium). C is the initial guess structure of the I_2 intercalation compound, and B is its DFT geometry-optimized structure. Reprinted with permission from ref⁶⁹
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If the criteria of charge and size that fulfill non bulky primary ammonium $R-NH_3^+$ or diammonium $^+H_3N-R-NH_3^+$ are necessary, they don't ensure systematic achievement of perovskite phases. This can be illustrated by a series of hybrids obtained with phenalkyl ammonium cations. If the phenethylammonium cation affords multilayered perovskites,^{18,28} the phenpropylammonium and the phenbutylammonium cations don't. Due to a longer alkyl chain (one or two supplementary CH_2 units), a head-to tail arrangement of molecules interacting together in the interlayer space occurs, which, combined to the quite bulky phenyl group, involves a decrease of the positive charge density at the organic-inorganic interface. The set of positive charges cannot fit with the set of negative charges brought by terminal halides of the $n = 1$ perovskite layer. As a consequence, a condensation of the BX_6 octahedra within the inorganic layers occurs during the crystallization process in order to equal the set of positive charges defined by the organic cations, finally leading to Pb_2I_7 halometallate structures with perovskite sub-networks. The resulting structures $(PPA)Pb_2I_7$ and $(PBA)Pb_2I_7$ are built from different inorganic networks, but both are based on corner- and face-sharing octahedra (Figure 4).¹⁹³ Another similar case involves two thiophene based cations. If the 2-methylammonium-thiophene afford the monolayer perovskite (Figure 5),²⁰⁰ the longer 5-ammoniummethylsulfanyl-2,2'-bithiophene cation led to a Pb_3I_{10} halometallate structure with a perovskite sub-network (corner- and face-sharing modes) as a result of a molecular head-to-tail arrangement.¹⁹⁴ Non-bulky

primary ammonium cations which possess heteroatoms along the molecular skeleton (O, S) or at the end (I, Br, CN...) can also preclude the formation of perovskite layers due to abilities of heteroatoms to make specific interactions through their free electron pairs. An interesting example is the series of hybrids obtained with the cystaminium $(\text{NH}_3(\text{CH}_2)_2\text{SS}(\text{CH}_2)_2\text{NH}_3)^{2+}$ cation. Depending on experimental conditions, the layered perovskite $(\text{NH}_3(\text{CH}_2)_2\text{SS}(\text{CH}_2)_2\text{NH}_3)\text{PbI}_4$ ²¹⁶ as well as a series of hybrids based on 1D $\text{Pb}_n\text{I}_{4n'+2}$ ($n' = 2, 3, 5$) perovskite networks have been obtained.^{192,190}

2.2.3. B_2X_8 $m = 2$ $\langle 110 \rangle$ -oriented layered perovskites

The layered compounds ($m > 1$) belonging to the $\langle 110 \rangle$ -oriented perovskites are rare and only $m = 2$ members have been structurally characterized. It is important to notice that the corresponding $\text{B}_2\text{X}_8^{4-}$ network and the BX_4^{2-} $n = 1 / \langle 100 \rangle$ are isomers meaning that for a given ratio « organic cation/ BX_4 », both perovskites are in competition. The comparative structural analysis of both networks as well as the structural analysis of the $m = 2$ $\text{B}_2\text{X}_8^{4-}$ perovskites can be of interest, first to understand why the $n = 1 \langle 100 \rangle$ network is preferentially formed, but also to design new $\langle 110 \rangle$ perovskites in the future. Organic cations which have stabilized $\langle 110 \rangle$ perovskites are depicted in Figure 6 (D1 and D2). As regards $\langle 100 \rangle$ and $\langle 110 \rangle$ networks, two main differences can be stressed. The first is the density of anionic charge at the organic-inorganic interface, which can be defined as the number of terminal halides by surface unit. The charge density is higher in the $\langle 110 \rangle$ network. In fact, while the distance between two adjacent terminal halides is roughly $d = (2 \times d(\text{B-X}))$ for the $\langle 100 \rangle$ network, the average distance is lower for the $\langle 110 \rangle$ network since there are two types of distances, $d = (2 \times d(\text{B-X}))$ along one direction (for instance a direction, Figure 9a) and $d = (2 \times d(\text{B-X}) \times \cos 45^\circ)$ along a

perpendicular direction (b direction in Figure 9a). The second difference concerns the interaction between organic cations and the inorganic anion. For the $\langle 100 \rangle$ BX_4^{2-} network, all cationic heads will be ideally localized within the inorganic network entering through square windows defined by four adjacent terminal halides (Figure 5). In contrast for the $\langle 110 \rangle$ network there will be two different situations. One cationic head will be localized within the inorganic layer entering through rectangular windows defined by four adjacent terminal halides whereas the other will take place in the interlayer space next to edges of metal halide octahedra (Figure 9). The structural analysis of the first reported $m = 2$ compound, $(NH_2C(I)=NH_2)_2(MA)_2Sn_2I_8$,³⁹ is of interest. Besides the fact that the smaller methylammonium cation is located in the inorganic layer while the bigger iodoformamidinium cation lies in the interlayer space, it is interesting to notice that the iodoformamidinium cation interacts through hydrogen bonding with terminal iodides of both adjacent layers (Figure 9a, dashed lines). This situation has been confirmed in the structure of the Pb analogue reported recently,²¹⁷ and a similar feature occurs in $(H_2D2d)_2Pb_2Br_8$ (D2d : 2-(aminoethyl)isothiourea)²¹⁸ and $(H_2D2b)_2Pb_2Br_8$ (D2b : N-(3-aminopropyl)imidazole)²¹⁹ (Figure 9b). In the structure of $(H_2D2c)_2Pb_2Br_8$ (H_2D2c : N-methylethane-1,2-diammonium; Figure 9d)²⁰² the primary ammonium head in the interlayer space is also interacting with terminal bromides of two adjacent layers. A special feature of this structure is the presence of the secondary ammonium within the perovskite layer while the primary ammonium is located in the interlayer space. Finally, the structure of $(EDBE)_2Pb_2Br_8$ (EDBE : 2,2'-(ethylenedioxy)bis(ethylammonium), Figure 9c)²⁰² is the only one based on a symmetric dication. This is quite surprising since, as expected for such a symmetric cation, the chloride compound $(EDBE)PbCl_4$ is a $\langle 100 \rangle$ perovskite. This may be related to the presence of

hydrogen bonds between one activated H atom, of a CH₂ unit next to the oxygen heteroatom, and terminal bromides, which are then stabilized as in all the other existing $m = 2$ structures.

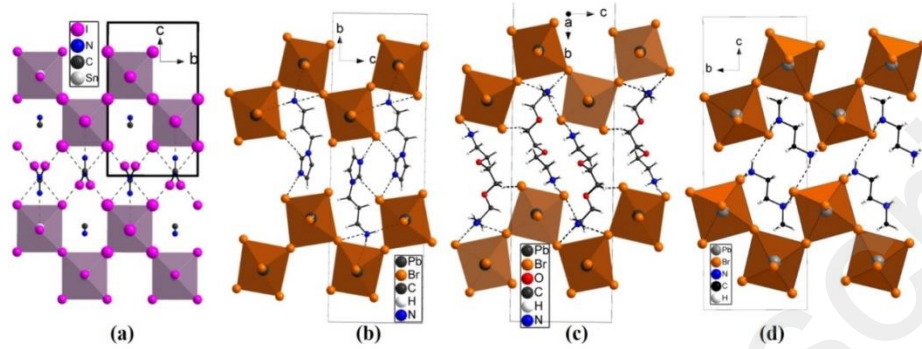


Figure 9. Structures of $m = 2$ $\langle 110 \rangle$ -oriented perovskites, $(\text{NH}_2\text{C}(\text{I})=\text{NH}_2)_2(\text{MA})_2\text{Sn}_2\text{I}_8$ with $\text{N}\dots\text{I}$ contacts ($d < 3.70 \text{ \AA}$) as dashed lines (H atoms not located) (a), $(\text{H}_2\text{D}2\text{b})\text{Pb}_2\text{Br}_8$ ($\text{D}2\text{b}$: N -(3-aminopropyl)imidazole) (b), $(\text{EDBE})_2\text{Pb}_2\text{Br}_8$ (EDBE : 2,2'-(ethylenedioxy)bis(ethylammonium)) (c) and $(\text{H}_2\text{D}2\text{c})_2\text{Pb}_2\text{Br}_8$ ($\text{H}_2\text{D}2\text{c}$: N -methylethane-1,2-diammonium) (d) showing the hydrogen bonding at the organic-inorganic interface.

2.3. Multilayered $\langle 100 \rangle$ -oriented perovskites

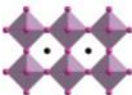


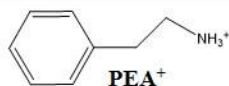
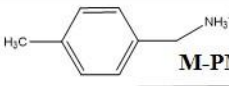
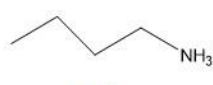
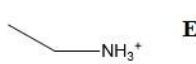
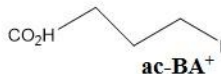
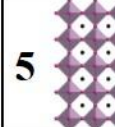
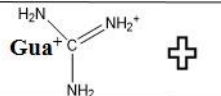
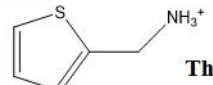
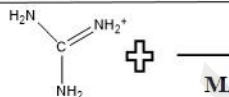
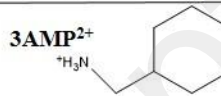
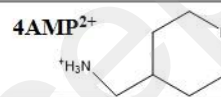
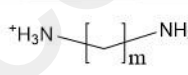
2.3.1. Introduction

The multilayered $\langle 100 \rangle$ -oriented perovskites ($n > 1$) have emerged as a promising class of light-absorbing materials for solid state photovoltaics due to a higher stability against humidity than their 3D analogues, while keeping a high power conversion efficiency up to 15.4% when a suitable preferential orientation of layers perpendicular to the substrate occurs.^{20,220–223}

Successful preparation of phase pure $n > 2$ multilayered perovskite⁸⁰ combined to thin films preparation using hot-casting methods¹¹ represents a breakthrough, pushing the 2D halide perovskites as serious candidates in the field of PSCs. The first multilayered hybrid perovskite, $(\text{PEA})_2(\text{MA})\text{Pb}_2\text{I}_7$ with $n = 2$, has been discovered by J. Calabrese, D. L. Thorn et al. in 1991.²⁸

They anticipated that this new material belongs to a more general family of multilayered halide perovskites: « ... and we began to suspect that the cubic perovskite $(\text{MeNH}_3)\text{PbI}_3$ and the layered perovskite $(\text{RNH}_3)\text{PbI}_4$ are actually two extremes of a hitherto unknown structural series $(\text{RNH}_3)_2(\text{MeNH}_3)_{n-1}\text{Pb}_n\text{I}_{3n+1}$. Here $n = \infty$ corresponds to the cubic perovskite compound $(\text{MeNH}_3)\text{PbI}_3$ and $n = 2, 3, \dots$ (« bilayer, « trilayer », ...) describe structures not previously known for any organic amine-metal halide system. ». In the next paragraph, we will highlight some structural features of $n > 1$ multilayered hybrid perovskites ($\text{B} = \text{Pb}, \text{Sn}, \text{X} = \text{I}, \text{Br}$) which have been crystallographically characterized so far.

Table 1 : <100>-oriented multilayered hybrid perovskites ($n > 1$): nature of the inorganic network $\text{B}_n\text{X}_{3n+1}$, nature of organic cations in the perovskite sheet (MA^+ or FA^+) and in the interlayer space (left column) and space group (SG) (MA^+ : methylammonium, FA^+ : formamidinium, PEA^+ : phenethylammonium, M-PMA^+ : 4-methyl-benzylammonium, BA^+ : butylammonium, EA^+ : ethylammonium, ac-BA^+ : ammonium 4-butyric acid, ThMA^+ : 2-methylammonium thiophene, Gua^+ : guanidinium, 3AMP^{2+} : 3-(methylammonium)piperidinium, 4AMP^{2+} : 4-(methylammonium)piperidinium).

Cations in the interlayer space	2 	3 	4 
 PEA ⁺	<u>Pb₂I₇/MA</u> ^a P-1	<u>Pb₃I₁₀/MA</u> ^l P1	<u>B_nX_{3n+1}/A</u> ^{ref} SG
 M-PMA ⁺	<u>Pb₂I₇/MA</u> ^b Pbcn		
 BA ⁺	<u>Pb₂Br₇/FA</u> ^c Cmc2 ₁	<u>Pb₃Br₁₀/MA</u> ^m Cmc2 ₁	
	<u>Sn₂I₇/MA</u> ^d Ama2	<u>Sn₃I₁₀/MA</u> ⁿ Cmca	
	<u>Pb₂I₇/MA</u> ^e Cc2m	<u>Pb₃I₁₀/MA</u> ^e C2cb	<u>Pb₄I₁₃/MA</u> ^e Cc2m
 EA ⁺		<u>Pb₃Br₁₀/EA</u> ^o C2cb	
 ac-BA ⁺	<u>Pb₂I₇/MA</u> ^f C2cb		 <u>Pb₅I₁₆/MA</u> ^q C2cb
 Gua ⁺ + Cs ⁺	<u>Pb₂Br₇/Cs</u> ^g Cmmm		
 ThMA ⁺	<u>Pb₂I₇/MA</u> ^h Aba2	<u>Pb₃I₁₀/MA</u> ^p Cmc2 ₁	
 MA ⁺	<u>Pb₂I₇/MA</u> ⁱ Bmm2	<u>Pb₃I₁₀/MA</u> ⁱ Imma	
 3AMP ²⁺	<u>Pb₂I₇/MA</u> ^j Ia	<u>Pb₃I₁₀/MA</u> ^j Pa	<u>Pb₄I₁₃/MA</u> ^j Ia
 4AMP ²⁺	<u>Pb₂I₇/MA</u> ^j Ia	<u>Pb₃I₁₀/MA</u> ^j Pc	<u>Pb₄I₁₃/MA</u> ^j Ia
 ⁺ H ₃ N-[CH ₂] _m -NH ₃ ⁺	<u>Pb₂I₇/MA</u> ^k m=8 Cc m=9 Cc	<u>Pb₃I₁₀/MA</u> ^k m=8 Pc m=9 Pc	<u>Pb₄I₁₃/MA</u> ^k m=8 Cc

^aref. 28, ^bref. 42, ^cref. 224, ^dref. 175, ^eref. 80, ^fref. 225, ^gref. 210, ^href. 200, ⁱref. 83, ^jref. 204, ^kref. 226, ^lref. 18,

^mref. 227, ⁿref. 35, ^oref. 95, ^pref. 220, ^qref. 82

2.3.2. Crystallographically characterized structures

The <100>-oriented multilayered perovskites $(A')_2(A)_{n-1}B_nX_{3n+1}$ or $(A'')(A)_{n-1}B_nX_{3n+1}$ that have been crystallographically characterized are mainly $n = 2$, $n = 3$ and $n = 4$ members of the series but also $n = 5$ for one structure. Several structural data of multilayered hybrid perovskites, where the inorganic network is based on Pb^{2+} or Sn^{2+} and I^- or Br^- , are reported in Table 1: nature of the inorganic network B_nX_{3n+1} , nature of organic cation(s) in the perovskite sheet and in the interlayer space along with the corresponding space group. We first notice that the organic cation in the inorganic sheet is in most cases the methylammonium cation. Two compounds are exceptions: in $(BA)_2(FA)Pb_2Br_7$ the formamidinium cation takes place in the $Pb_2Br_7^{3-}$ layer,²²⁴ while in $(EA)_4Pb_3Br_{10}$, the ethylammonium cation is present in the $Pb_3Br_{10}^{4-}$ sheet.⁸⁴ If this last multilayered perovskite is unique, since only one type of cation is present both in the interlayer space and in the perovskite sheet (in that case, $A' = A$), the presence of the quite big ethylammonium cation occupying the perovskite cages is surprising and unprecedented. In fact, in the 3D $APbX_3$ perovskites, the only known cations that can fit in the perovskites cages are MA^+ , FA^+ and Cs^+ . The incorporation of the larger size cation of EA^+ into the $n = 3$ perovskite sheet is explained by a highly distorted perovskite network, with quite strong distortion of PbI_6 octahedra (Pb-I bond distances in the 2.98-3.31 Å range), which finally allows to accommodate the extra $-CH_2-$ group of ethylammonium cations. For a long time, only primary ammonium monocations have afforded multilayered <100>-oriented perovskites, the head to tail arrangement of organic cations in the interlayers leading roughly to a $(\frac{1}{2} \ \frac{1}{2})$ shift of adjacent layers in the layer plane. Recently, a few series of Pb_nI_{3n+1} ($n = 2-4$) multilayered hybrids with dications in the interlayer space have been crystallized. The dications are either the long alkylidiammonium $(NH_3-(CH_2)_m-NH_3)^{2+}$ ($m = 8, 9$)²²⁶ or the shorter dications of 3AMP²⁺ (3-

(methylammonium)piperidinium) and $4AMP^{2+}$ (4-(methylammonium)piperidinium).²⁰⁴ In hybrids based on these last cations, due to their dicationic nature as well as their short length, no shift between adjacent perovskite layers is observed. A main characteristic of some of these perovskites is their enhanced electronic dimensionality as underlined by band structure calculations which reveals a band dispersion along the perovskite layer stacking direction.⁸³ The key factor is the short halogen-halogen contacts between apical halogenes of adjacent layers ($d = 4.00 \text{ \AA}$ in $(4-AMP)(MA)_3Pb_4I_{13}$). This feature is especially expected when short cations such as $3AMP^{2+}$ and $4AMP^{2+}$ or Gua^+ and MA^+ fill the interlayer space. A singular feature of most of these multilayered perovskite is their acentric and polar nature as they mainly crystallize in the monoclinic m and orthorhombic $mm2$ crystal classes (deduced from space groups given in Table 1). One exception is the first reported multilayered perovskite $(PEA)_2(MA)Pb_2I_7$ described in the P-1 space group.²⁸ However, the statistical disorder which affects some atoms may indicate that a lower symmetry could be refined. A few years later, the acentric nature of a $n = 2$ perovskite compound based on 2-methylammonium thiophene cation, by highlighting the acentric nature of the bilayer perovskite network, has been unambiguously demonstrated.²⁰⁰ Next, using the ammonium 4-butyric acid cations that forms supramolecular dimers in the interlayer space leading to an acentric organic layer, it was shown that the monolayer compound was centric due to a symmetry center located in the perovskite layer, while the $n = 2$ hybrid perovskite built from bilayer perovskite sheets along with the acentric organic layers was acentric and polar.²²⁵ Recently, the relative energies of some multilayered perovskites in both the centrosymmetric and non-centrosymmetric structures were computed. It has been found that the acentric structures were lower in energy, and that this trend was even more significant when the number of layers, n is increased.⁸⁰ This is in agreement with similar calculations on $(CH_3NH_3)PbI_3$ that predicted the

acentric structure ($I4cm$) to lay slightly lower in energy compared to the centric ($I4/mcm$) structure.²²⁸ Not only most of these compounds are acentric, but they are also polar, meaning that they may exhibit ferroelectric properties. If ferroelectricity in low-dimensional hybrid halobismuthate based on 1D BiI_3Cl_2 perovskite network has been evidenced a few years ago,²²⁹ some 2D hybrid perovskites ferroelectrics have been reported recently, either monolayered^{230,231} or multilayered materials, $(\text{BA})_2(\text{MA})_2\text{Pb}_3\text{Br}_{10}$ ²²⁷ and $(\text{BA})_2(\text{FA})\text{Pb}_2\text{Br}_7$.²²⁴ In $(\text{BA})_2(\text{MA})_2\text{Pb}_3\text{Br}_{10}$ an order-disorder transition occurs at 315 K, the BA^+ cations being disordered in the centro-symmetrical high temperature phase while they become ordered in the polar low temperature phase. At the same time, the inorganic network that adopts a highly symmetrical configuration above 315 K undergoes structural changes corresponding to a re-orientation and tilting of PbBr_6 octahedra.²²⁷

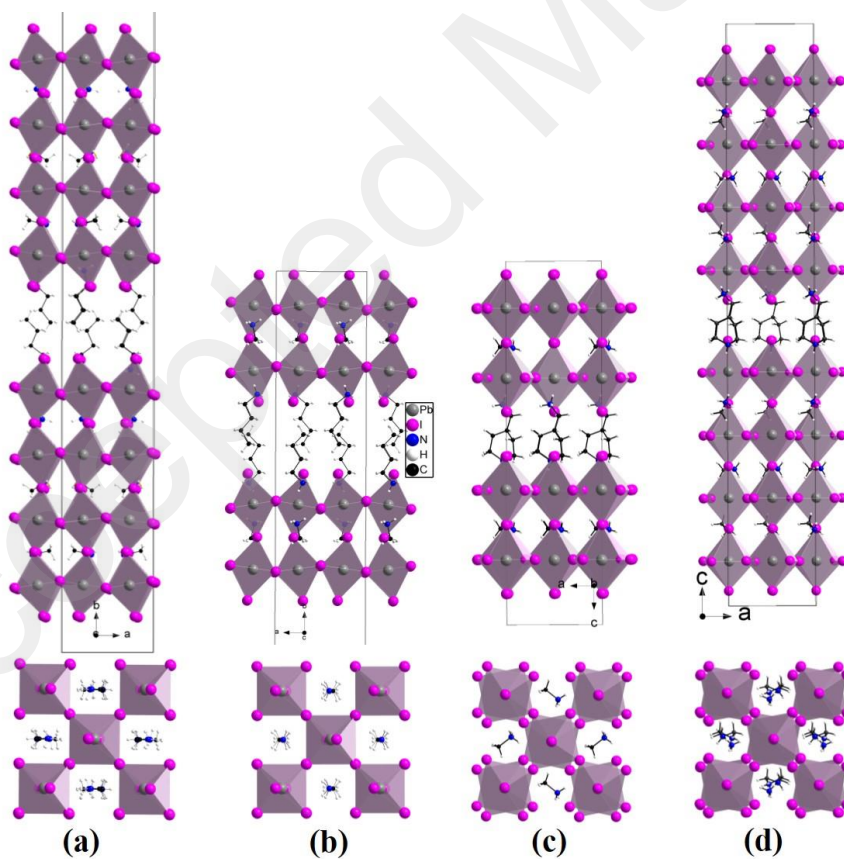


Figure 10. Structures of $n = 2$ and $n = 4$ $\langle 100 \rangle$ -oriented perovskites, $(BA)_2(MA)_3Pb_4I_{13}$ ⁸⁰ (BA^+ : butylammonium, MA^+ : methylammonium) (a), $(BA)_2(MA)Pb_2I_7$ ⁸⁰ (b), $(4AMP)_2(MA)Pb_2I_7$ ²⁰⁴ ($4AMP^{2+}$: 4-(methylammonium)piperidinium) (c) and $(4AMP)_2(MA)_3Pb_4I_{13}$ ²⁰⁴ (d) : general view (top) and view of one inorganic multilayer along the perpendicular direction (bottom).

If a large number of $n = 1$ $\langle 100 \rangle$ -monolayer perovskites have been stabilized by a variety of organic cations, the number of known $\langle 100 \rangle$ -multilayered perovskites are currently comparatively low. Then the question arises: could all organic cations that have afforded monolayered compounds stabilize multilayered materials? As already underlined, the $n = 1$ layer can be highly distorted (in-plane and out-of-plane distortion, octahedra distortion) in related materials depending on the nature of organic cations in the interlayer space. In contrast, and even if an out-of-phase tilting of octahedra is observed in the tetragonal room temperature structure of the 3D $(CH_3NH_3)PbI_3$, the distortions are limited with Pb-I-Pb bond angle greater than 164° . As a consequence, it can be expected that cations that induce only weak distortions in the monolayer perovskite network shall be well suited to afford multilayered materials. The $n = 2$ multilayered compound is unique since it is built from two outer layers, and, if distortions of in-plane type are the most commonly observed, large out-of-plane and in-plane distortions can also occur, as in $Cs_2(C(NH_2)_3)Pb_2Br_7$ (Figure 7, right).²¹⁰ The $n > 2$ multilayered compounds are different in account of the presence of two types of layers, outer layers and inner layers (one or more), the latter being comparable to layers of 3D bulk materials. Figure 10 shows an overall view, as well as that of one inorganic multilayer along the perpendicular direction, of $n = 2$ and $n = 4$ structures of BA/MA/Pb/I and 4AMP/MA/Pb/I compositions. It is first interesting to notice that the bilayer conformation is quite different in both $(BA)_2(MA)Pb_2I_7$ and $(4AMP)_2(MA)Pb_2I_7$. In

the BA⁺ based material, layers exhibit out-of-plane distortions ($\delta = 7.18^\circ$), the MA⁺ cation is roughly parallel to the stacking axis while only the ammonium part of BA⁺ cations lie in the inorganic sheet, the CH₂ group lying in the plane defined by apical iodides (Figure 10b). On the contrary in the structure of (4AMP)₂(MA)Pb₂I₇, layers exhibit in-plane distortions (Pb-I-Pb bond angle $\approx 156^\circ$) but no out-of-plane distortions (Figure 10c). The second interesting feature is the similar type of distortions occurring in the n = 2 and n = 4 structures for both systems. In (4AMP)₂(MA)₃Pb₄I₁₃, only in-plane distortions are observed, but with increased Pb-I-Pb bond angles compared to those observed in the bilayer compound, particularly for the inner layers (160° and 162°) (158° and 158.7° for the outer layers). As in the structure of (BA)₂(MA)Pb₂I₇, only out-of plane distortions are observed in the structure of (BA)₂(MA)₃Pb₄I₁₃, but here also the network is less distorted in the n = 4 structure ($\delta = 6.08^\circ$ -6.50°) than in the n = 2 structure (Figure 10a).

Given that different types of cations, namely ThMA⁺ or PEA⁺, including a thiophene or a phenyl group, BA⁺ and (NH₃-(CH₂)_m-NH₃)²⁺ (m = 8, 9) based on an aliphatic chain, 3AMP²⁺ and 4AMP²⁺ dications, including a non-primary ammonium head, and both Gua⁺ and MA⁺ in the interlayer space, have afforded n = 2-4 multilayered hybrid perovskites, shows that a large variety of cations can potentially stabilize such multilayered materials. However, other factors than steric considerations can preclude the formation of multilayered perovskites. In fact, during the synthetic process, by using a small (A = MA⁺ or FA⁺) cation and a longer one (for instance monocation A⁺), there is a competition between the formation of the expected multilayered perovskite and the formation of both the 3D ABX₃ compound and the monolayer (A')₂BX₄. The nature of the monolayered material (for instance based on highly distorted layers) and its high thermodynamic stability can favor the disproportionation into n = 1 and the 3D perovskite.

3. MULTILAYERED HALIDE PEROVSKITES: QUANTUM CONFINEMENT

Summary: The description of 3D halide perovskites basic electronic structures represents currently a major challenge for nowadays theoretical first-principles approaches due to the importance of relativistic effects, many-body effects, the dynamics of the organic cations and the coupling to low energy lattice excitations. For lower-dimensional perovskite networks, this is an even more challenging goal. Therefore, the understanding of quantum confinement effects in low dimensional structures and nanostructures is still at an early stage and is capitalizing on concepts and tools, including empirical methods, initially developed for conventional semiconductors.

3.1. Electronic band structure

Reviewing the main features of the electronic structure of hybrid halide perovskites is an essential prerequisite to understand the subtle interplay between their organic and perovskite components and to investigate effect of quantum and dielectric confinement. This section aims at providing basic considerations, such as those related to the atomic composition of the band edge electronic states, and related key findings that should also help to bridge the gap between solid state physics and molecular chemistry concepts.

3.1.1. Brief overview of theoretical approaches and relation to experiments

Band gap (alternatively bandgap) is among the most frequently used physical term to characterize the capabilities of a given material for a wide range of applications as well as to

conduct comparison between experimental and computed data, especially that derived from electronic structure calculations. To avoid any confusion, we introduce the following terminology:

- The HOMO-LUMO gap, which correspond to the energy difference between the highest occupied molecular orbital (HOMO) and the lowest unoccupied molecular orbital (LUMO) of an isolated (0D) species.
- The DFT gap ($E_{g,DFT}$), which corresponds to the energy difference between the highest occupied valence band (VB) state and the lowest empty conduction band state (CB) that is computed using Kohn-Sham (KS) Density Functional Theory (KS DFT)²³² and does not account for electron removal or addition neither for electron promotion from a VB state to a CB state.
- The transport or electrical or fundamental gap ($E_{g,trans}$), which corresponds to the energy difference between uncorrelated free electrons and holes; experimentally, this is to a good approximation accessible through ultraviolet photoemission spectroscopy (UPS) that provides information on the occupied VB states and inverse photoemission spectroscopy (IPES) for the empty CB states. It is related to the difference between ionization potential and electron affinity. As such, it can be computed to a good approximation using total energy (E_{tot}) differences: $E_{g,trans} = E_{tot}(N + 1) + E_{tot}(N - 1) - 2E_{tot}(N)$, with N the number of electrons in the system, or better based on Many-Body Perturbation Theory (MBPT)²³³ such as the quasiparticle self-consistent GW theory²³⁴ (sc-GW).
- The optical band gap energy ($E_{g,opt}$) is experimentally determined in many cases from photoluminescence (PL) experiments or the observation of a

resonance in optical absorption spectroscopy. It consists in the continuum band gap (E_g) corresponding to the lowest allowed interband absorption minus the exciton binding energy (E_b) corresponding to the energy required to form a bound exciton (electron-hole pair). In direct band gap semiconductors, the continuum band gap E_g usually matches the transport band gap $E_{g,trans}$. Noteworthy, the continuum gap may or not imply contributions from phonons either related to self-energy corrections or to interactions between the free electrons and holes with lattice phonons.

To some degree of accuracy, these quantities may be computed using various methods ranging from fully empirical ones²³⁵ to self-consistent first-principles calculations including relevant treatments to account for many-body interactions and electron-phonon coupling.^{93,236–240} The simplest approach is based on linear combinations of atomic orbitals (LCAO)²⁴¹ with mainly three variants with the Hückel and Extended Hückel theories (EHT)^{242–244} for quantum chemists and the tight-binding (TB) model²⁴⁵ for physicists. Currently, the most widely used and versatile technique is based on DFT within which the problem of j interacting electrons can be solved through a set j one-electron Schrödinger-like KS equations by introducing an effective potential.²³² Unfortunately, its exact form is unknown and a broad variety of approximated exchange-correlation density functionals (xc-kernel) have been developed over the last decades, the simplest being the local density approximation (LDA),^{246,247} and the step above includes gradient corrections (GGA).²⁴⁸ Noteworthy, KS DFT is a ground state theory and it has been shown that none of the band gaps experimentally measured can be derived from a single KS DFT band-structure calculation, even with the exact density functional. More recently, alternative realizations of DFT have been developed within generalized Kohn-Sham (GKS) schemes.^{249–251} It has been demonstrated that the GKS band gap provides a better approximate to $E_{g,trans}$ than

the KS band gap, and that it equals $E_{g,trans}$ for a given approximate functional. For instance, those containing a fraction of exact exchange (*e.g.* HSE),²⁵² the so-called hybrid functionals, are currently attractive computationally-efficient alternatives to many-body techniques.

DFT has proved efficient to afford useful structure-properties relationships for many classes of semiconductors as long as an appropriate xc-kernel is implemented. For instance, in such cases, computed VB and CB densities of states (DOS) can be compared to DOS derived from UPS and IPES, respectively. Unfortunately, when using standard xc-kernels such as PBE²⁴⁸ for the archetype $\text{CH}_3\text{NH}_3\text{PbI}_3$ compound, DFT computed band gaps show a fortuitous agreement with the experimental optical band gap.⁹¹ This is a direct consequence of two opposite but huge error compensations between many-body and relativistic effects.²⁵³ In fact, due to the large atomic number of lead (to a lesser extent for tin and halogens), spin-orbit coupling (SOC) has a dramatic effect on the electronic structure, inducing a large reduction of the band gap. The combination of no SOC with PBE leads to a fortuitous cancellation for the Pb bands but not for any other bands, thus the entire band structure will be incorrect in its details. To accurately estimate the band structure and continuum band gap, one needs to include the relevant many-body interactions that can be obtained using sc-GW, or alternatively to a cheaper computational cost by a one-shot GW correction (G_0W_0 approximation). Last, to account for excitonic effects, one needs to describe bound states that are well accounted for when solving the Bethe-Salpeter Equation (BSE) describing the bound states of a two-body relativistic system.²⁵⁴ This can be performed at various levels of theory (*vide infra*), *e.g.* within the framework of DFT or that of semi-empirical approaches such as the TB model.

An alternative approach to investigate electronic excitations (*e.g.* optical properties) relies on time-dependent DFT (TD-DFT).^{255,256} In the last two decades, TD-DFT schemes have been

extended to periodic solids²⁵⁷ but have yet rarely been applied to halide perovskites.²⁵³ In the context of photovoltaics, it has proved efficient for many dyes, for instance to compute optical absorption of TiO₂/Dyes interfaces.²⁵⁸ Among the few TD-DFT studies on halide perovskites that also include non-adiabatic molecular dynamics (NAMD)²⁵⁹⁻²⁶² except for the optical properties of the prototypical OD Cs₄PbBr₆ compound,²⁶⁰ effort has primarily concerned 3D networks to rationalize absorption spectra,²⁶³ the exciton^{259,262} and excited state geometries in the presence of defects.²⁶⁴ Noteworthy, none of these TD-DFT investigations includes SOC, which is believed to have no significant harmful effect on structural optimization,²⁶⁴ but sizeable impact is expected as far as optical properties are concerned.^{64,91,92,262}

3.1.2. Ways to handle the organic cations

For most of the organic cations (primary non-conjugated ammonium cations) used to stabilize 3D and lower dimension halide perovskites, their HOMO-LUMO gap is significantly larger than the transport gap of the inorganic framework and, therefore, the molecular states do not contribute to the top of VB (VBM for valence band maximum) and bottom of the CB (CBM for conduction band minimum). This has been noticed at the early stage, with for example the use of an isolated (PbX₆)⁴⁻ charged cluster²⁶⁵ or a single (SnI₄)²⁻ layer^{38,203} to compute the electronic structure of halide perovskites. This was assessed for lower dimension halide perovskites in the seminal work by Koutselas et al. who performed ETH calculations with and without the amines and showed that the band gap did not change.³³ Over the decades, alternative strategies have been explored with for example the use of a uniform background charge density in self-consistent LCAO modeling of 2D and 3D perovskites achieving fair comparison with VB DOS experimentally derived from UPS.²⁶⁵ Many computational tricks have also been explored in

the context of DFT computations, with compensation background charge densities or substitution of the organic cation by an alkali metal. In 2007, Sourisseau et al. reported that for 2D $(X-(CH_2)_2-NH_3)_2PbI_4$ ($X = Cl, I$), the best agreement with computed DOS and electronic band gaps is obtained when putting a sodium atom on the ammonium site.²⁰⁵

In the case of hybrid 3D halide perovskites, this leads in a situation conducive to computations as the organic cations are disordered near room temperature (both in cubic and tetragonal phases) as they do not fit the site symmetry.⁹¹⁻⁹³ This trick has also proved efficient for the modeling of lower dimension halide perovskites, both at the early stage and still recently, as it allows modeling the numerous disordered layered structures and at the same time reduces the computational cost. Replacement of the organic cation by a Cs^+ at the position of the ammonium group also helped developing model structures to introduce composite approaches designed to afford band alignments and dielectric profiles (section 3.2 and 4.4, respectively) or to rationalize effect of octahedral tilting on the band gap in layered perovskites (section 3.1.3.3).

However, such approximations will break down in several situations especially when dielectric confinement is at work. In fact, whereas $R-NH_3^+$ have low static dielectric constant of about 2 to 3, it may be significantly increased for bifunctional organoammoniums, non-primary ammoniums or as a result of intercalating species. Next, explicit account of the cations is a prerequisite for structural optimization. Noteworthy, most of the computed properties depend intimately on the structure of the perovskite framework (*vide infra*) and thus we stress that raw DFT structural optimization even with the most suited xc-kernel (*e.g.* corrected for dispersion forces) will afford the ground state static structure at zero kelvin (0 K). Moreover, explicit account of the A, A' and A'' cations is mandatory whenever information on lower-lying states is desired, as illustrated recently when comparing UPS VB DOS in BA-based 2D perovskites²⁶⁶ or

investigating VB DOS broadening in 3D perovskites.²⁶⁷ The same holds whenever detailed information is desired on higher-lying CB states.²⁶⁶ Besides, it has recently been demonstrated based on molecular dynamics simulations that the organic molecule can be highly dynamic in layered halide perovskites inducing dynamic potential fluctuation and subsequent wave function localization.²⁶⁸

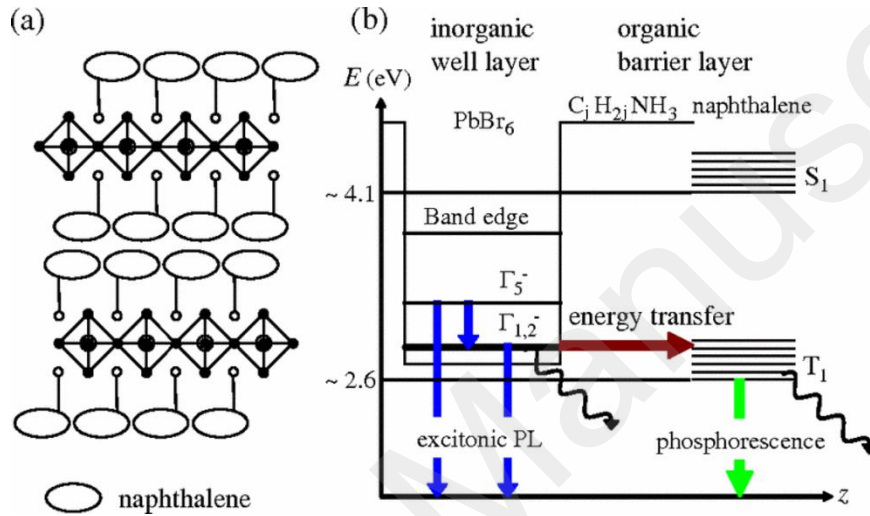


Figure 11. Schematics of the crystal structure of naphthalene-incorporated layered perovskite-type QW materials. It merges the Wannier and Frenkel pictures which are suited for the excitons in the inorganic and the organic parts, (a) Energy diagram and energy transfer dynamics of $(C_jH_{2j}NH_3PbBr_4)$ (b). Reprinted with permission from ref ²⁶⁹. Copyright 2008 American Physical Society.

Last but not least, quite a few lower dimension halide perovskites containing active organic cations have been reported (section 2.2 and ref. ^{25,270}). As early as in 1999, Mitzi et al. suggested possible energy transfer between the inorganic framework and the organic moiety for more conjugated cations, namely a quaterthiophene chromophore.^{214,271} This has recently been confirmed using GKS DFT calculations including SOC, revealing a type II band alignment.²⁷²

Among others, charge separation between the organic ammonium-propyl-imidazole (API) layer and the inorganic PbBr_4 one has been suggested to occur on the band edge states on the basis of early DFT computations lacking SOC.²¹⁹ Recent GKS DFT results with SOC included have belied this conclusion, showing that the optical response of single-layered Pb-based perovskites with single- π -conjugated layers such as ammonium-propyl-imidazole or phenylethylammonium is dominated by states localized within the perovskite layer.²⁴⁰

The description of energy transfer between the inorganic framework and the organic moiety is a complex question that has not yet received a clear theoretical framework. However, a qualitative picture has emerged from pioneering experimental works on this topic.^{273,274,214,269,25,270} An appropriate description entails the merging of the Wannier and Frenkel models which are suited for the excitons in the inorganic and the organic parts, respectively. Prospective schematic energy level diagrams mix energy levels related to singlet and triplet molecular states with mono-electronic states that are expected to describe the CB and VB edges of the inorganic QW.^{274,25} A few recent attempts have been conducted in this direction.^{240,272,274,} For instance, it has been shown from DFT simulations²⁰⁶ that chlorine-based systems shall present larger electronic band gaps and band alignments close to a type II QW. Even though it may favor mixing of inorganic and organic electronic wavefunctions, it shall only partially explain the results of a comparative study on the halide substitution using Cl/Br/I.²⁷³ The schematic descriptions of energy levels must be refined in order to introduce quantitatively the electron-hole pair states for both the inorganic and the organic parts (Figure 11).²⁶⁹ More, the singlet-triplet splitting of the Frenkel exciton as well as triplet-triplet coupling are expected to play a major role in the process of energy transfer. Finally, relevant wavefunction overlaps between the two excitonic states must be carefully introduced, in order to explain the resonant

optical nonlinearities. In fact, progresses have recently been made for the description of the Wannier exciton of the inorganic QW, including realistic dielectric profiles and the leakage of the spinor states into the organic barrier derived from DFT simulations (vide infra).²⁷⁵ A further step is necessary to introduce exchange energy into such simulations, using mono-electronic spinor states. Together with a correct description of the molecular excitonic states, it may lead to a more satisfying theoretical framework for situations where the organic cation plays an active role, *i.e.* largely beyond a mere cationic charge compensating molecule.^{25,276,277}

3.1.3. Electronic band edge states

3.1.3.1 Basics of the atomic composition of band edge states

As early as in 1978, Heidrich et al. developed an empirical LCAO model for CsPbI₃ by analogy with thallos halides and computed the first band diagram of halide perovskites.²⁷⁸ This early work, based on the ansatz that the average band energies can be approximated by energies of the free ions corrected by Madelung shifts, already stressed that the electronic band gap of the cubic *Pm3m* phase is direct but at the Brillouin Zone (BZ) boundary, namely at the R-point ($\mathbf{k}_R = (1/2, 1/2, 1/2)$), and that SOC and crystal field may split the possible transitions of optical spectra. At a first sight, the fact that the band gap shows up at the BZ boundary can be grasped considering a linear chain of atomic orbitals (Figure 12).²⁷⁹ Noteworthy, in this simple picture, only the main orbitals that contribute to VBM and CBM have been considered namely the p-orbitals of the halide atom with the s-orbitals of the metal for the VBM and the p-orbitals of the metal for the CBM (vide infra). Figure 12 introduces further the concept of band-folding illustrated in Figure 13 for the pseudo-cubic phase of a 3D halide perovskite. In the first BZ associated to the primitive unit cell of the cubic *Pm3m* space group, the DFT band structure

drawn along its principal directions indeed reveals a direct band gap at the R-point. With the very same crystallographic positions but representing the band structure along principal directions of the first BZ of the orthorhombic $Pnma$ unit cell evidences the band folding from BZ edges R, M ($\mathbf{k}_M = (1/2, 1/2, 0)$) and X ($\mathbf{k}_X = (0, 1/2, 0)$) to the BZ center Γ ($\mathbf{k}_\Gamma = (0, 0, 0)$). Further comparison to the true orthorhombic structure highlights the connections between the two band structures and additional effect related to the structural distortions (strain). Thus, depending on the unit cell used to compute band dispersion, the shape may appear different and the band gap may or not show up at the BZ center.

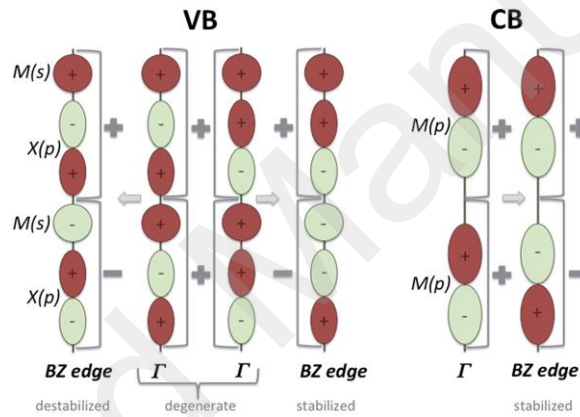


Figure 12. Atomic orbital hybridization and band folding in the Brillouin zone (BZ) exemplified on the simplest scenario, which consists of a linear chain. Left panel: hybridization between the p-orbitals of the halogen, $X(p)$, and the s-orbitals of the metal, $M(s)$. The valence band maximum (VBM) corresponds to the anti-bonding combination (left) that shows up at the BZ edge when the unit cell is built from one halogen and one metal (grey bracket). Upon unit cell doubling in real space (two grey brackets), the two combinations shown left and right fold back to the Γ point, centre of the new BZ. Right panel: hybridization between the p-orbitals of the metal, $M(p)$. The conduction band minimum (CBM) corresponds to the bonding combination (right) that shows up at the BZ edge when the unit cell is built from one metal (grey bracket).

Upon unit cell doubling, this combination folds back to the Γ point, centre of the new BZ.
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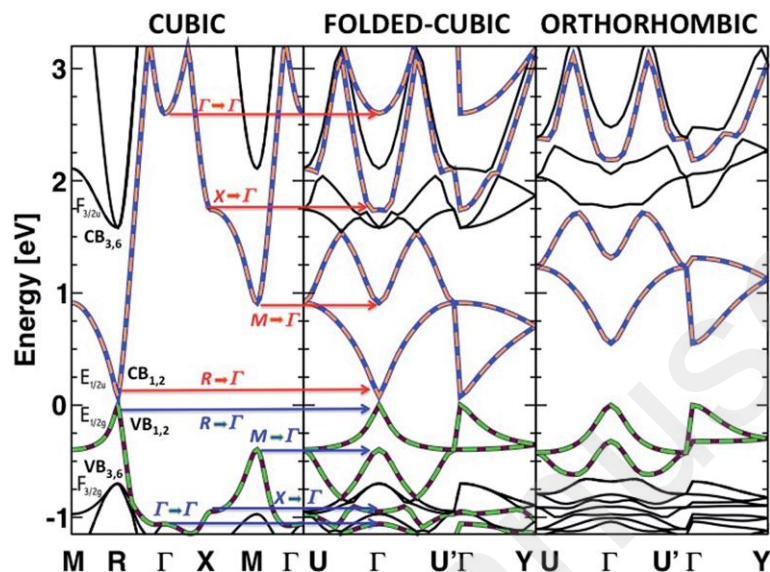


Figure 13. Band folding in the Brillouin zone (BZ) shown for $\text{CH}_3\text{NH}_3\text{PbI}_3$ with DFT calculations including spin-orbit coupling (SOC). Left panel: electronic band diagram calculated for the cubic $Pm3m$ (space group no. 221, $Z = 1$) crystal structure and represented along principal directions of the first BZ with irreducible representations of the double group. Middle panel: the very same band structure represented along principal directions of the first BZ of the orthorhombic $Pnma$ unit cell (space group no. 62, $Z = 4$) evidencing the band folding from R , M and X to Γ , which results in an apparent 4-fold increase of the number of bands. Right panel: electronic band diagram calculated for the real low-temperature orthorhombic $Pnma$ crystal structure shown along principal directions of its first BZ. It highlights effects of structural distortions on the band structure, namely the increased band-gap and the change in effective masses at the VBM and CBM. It is noteworthy that the band mixing remains moderate as a direct consequence of the large splitting of $\text{CB}_{1,2}$ and $\text{CB}_{3,6}$ (left panel) induced by SOC whereas such

mixing would have been largely overestimated without SOC. Reprinted with permission from ref 279– Published by the Royal Society of Chemistry.

The initial LCAO model was refined in 1981 with inclusion of spin-orbit coupling and the s-orbitals of Cs, leading to a Hamiltonian matrix of 28×28 .²⁸⁰ Despite irrelevant calibration of Cs-orbitals, it allowed to attribute the main feature of available experimental optical spectra and ultraviolet photoemission data, stressing the impact of a van Hove singularity at the M-point of the BZ. Since then, several other tight-binding models were devoted to 3D halide perovskites among which the work by Umebayashi et al.²⁶⁵ who provided the bonding diagram of a $[\text{PbI}_6]^{4-}$ cluster, the 3D crystal $\text{CH}_3\text{NH}_3\text{PbI}_3$ as well as for the 2D crystal $(\text{C}_4\text{H}_9\text{NH}_3)_2\text{PbI}_4$ (Figure 14). In the zero-dimensional system the higher occupied states were shown to be composed of σ -antibonding hybridization between metal s-orbitals and halogen p-orbitals. The lowest unoccupied states were stated to consist in metal p-orbitals hybridized either with halogen s-orbitals (σ -antibonding) or p-orbitals (π -antibonding). The later hybridization, even of moderate strength in metal halide crystals, was not taken into account in the early studies and their respective weight (also σ - versus π -antibonding) are seldom made clear.

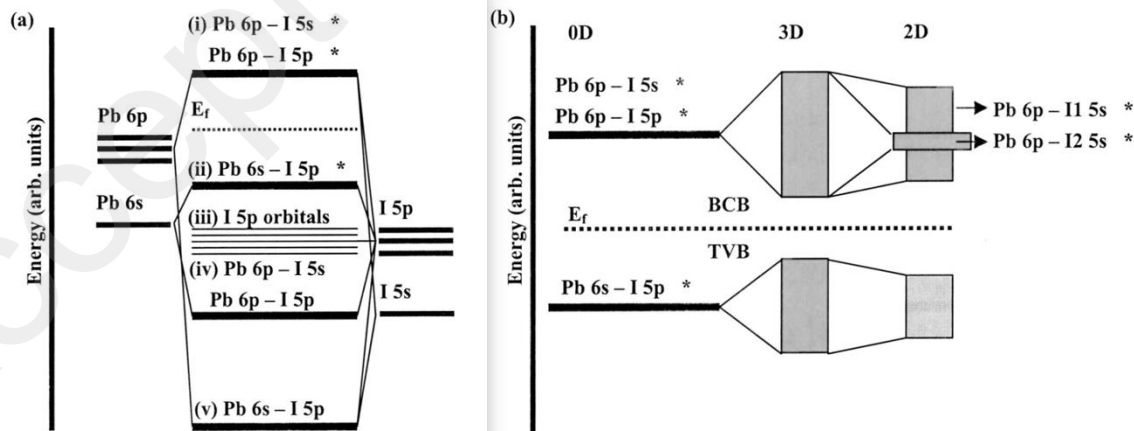


Figure 14. Bonding diagram of (a) $[PbI_6]^{4-}$ cluster (0D system), (b) 3D crystal $CH_3NH_3PbI_3$, and 2D crystal $(C_4H_9NH_3)_2PbI_4$ at the VBM (denoted TVB) and CBM (denoted BCB). Reprinted with permission from ref²⁶⁵ Copyright 2003 American Physical Society.

3.1.3.2 Electronic structure in 3D networks

Considering further TB modeling of cubic 3D halide perovskites²⁸¹ clearly reveals that the CBM at the high symmetry R-point of the BZ has a σ -antibonding character with

$$E_{CBM(R)} = \frac{E_{sX} + E_{pB}}{2} + \frac{\sqrt{(E_{pB} - E_{sX})^2 + 16V_{pBsX}^2}}{2}, \quad (1)$$

where E_{sX} and E_{pB} are the atomic energies and V_{pBsX} the overlap integral related to the p-orbitals of the metal (pB) and the s-orbitals of the halide (sX). Due to the stronger hybridization between metal s-orbitals (sB) and halogen p-orbitals (pX), the picture was clear from the early stage with

$$E_{VBM(R)} = \frac{E_{sB} + E_{pX}}{2} + \frac{\sqrt{(E_{pX} - E_{sB})^2 + 48V_{sBpX}^2}}{2}. \quad (2)$$

These expressions highlight several main features of the band diagram of 3D halide perovskites, for instance the reversed ordering of band edge states (VBM and CBM) as compared to tetrahedrally bonded semiconductors^{64,91}. They also provide a chemically intuitive tool to get a proper grasp of both halide (X) and metal (B) substitution based on atomic energies.²⁸² For example, when Pb is replaced by Sn, E_{sB} becomes less negative leading to a smaller band gap. Alternatively, substitution of I by Br (or Cl) leads to sizably more negative values of E_{pX} and the band gap significantly opens. Moreover, from expression (2) it is also easy to anticipate that for fixed lattice parameters (constant Pb-Pb distances) the band gap will

increase with increasing octahedral rotation or tilting (Pb-I-Pb angles deviating from 180°)^{228,283} as a result of a decrease of the overlap integral V_{sBpX} . Effect of lattice strain (Δa : small change of the lattice parameter a) can also be anticipated from such a simple TB model:²⁸¹

$$\delta E_{CBM(R)} \approx \frac{-16V_{pBsX}^2}{\sqrt{(E_{pB}-E_{sX})^2+16V_{pBsX}^2}} \frac{\Delta a}{a} \quad (3)$$

and

$$\delta E_{VBM(R)} \approx \frac{-48V_{sBpX}^2}{\sqrt{(E_{pX}-E_{sB})^2+48V_{sBpX}^2}} \frac{\Delta a}{a}, \quad (4)$$

showing that the largest variation is expected at the VBM, which is stabilized when the lattice parameter increases leading to a band gap increase.

Moreover, due to the presence of atomic species with large atomic numbers, relativistic effects and for instance spin-orbit coupling have a dramatic effect on the band diagrams. Combined to symmetry considerations, this complexifies further the band diagram as illustrated by the schematic energy diagram reported by Tanaka et al. (Figure 15).²⁸⁴ More recently, DFT calculations confirmed the major role of SOC evidencing the large splitting (Δ_{SOC}) of the first degenerated conduction levels (CB_{1,2} and CB_{3,6} in Figure 14), the CBM transforming to 2-fold

$E_{1/2u}$ degenerate spin-orbit split-off states (CB_{1,2}): $|1/2, 1/2\rangle = \frac{1}{\sqrt{3}} |(\mathbf{X} + i\mathbf{Y}) \uparrow\rangle + \frac{1}{\sqrt{3}} |Z \downarrow\rangle$ and

$|1/2, -1/2\rangle = \frac{1}{\sqrt{3}} |(\mathbf{X} - i\mathbf{Y}) \downarrow\rangle + \frac{1}{\sqrt{3}} |Z \uparrow\rangle$.⁹¹ Corresponding spinorial components will be

illustrated in the next section on a typical monolayered halide perovskite (Figure 16). This giant change of the whole electronic band structure has drastic effects on the electronic band gap, which changes according to

$$E_{CBM(R)} = \frac{E_{pB} + E_{sX} - 2\Delta_{SOC}/3}{2} + \frac{\sqrt{(E_{pB} - E_{sX} - 2\Delta_{SOC}/3)^2 + 16V_{pBsX}^2}}{2} \quad (5)$$

within the TB model,²⁸¹ but also on effective masses and on the optical absorption through a renormalization of the oscillator strengths.⁹¹ In fact, the CB_{1,2} lead to a strong and isotropic optical transition with the even $|S\rangle$ -like doubly E_{1/2g} degenerate VBM states that undergoes a threefold decrease due to SOC.⁹² Expression (5) and the one relevant to VBM combined with atomic energy levels tables also provide an intuitive way to grasp changes due to relativistic effects subsequent to metal and halide substitution.²⁷⁹

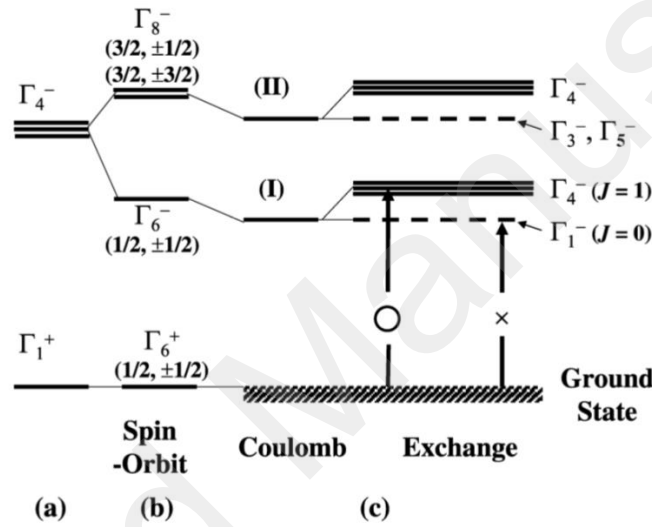


Figure 15. Schematic electronic energy diagram of 3D crystals for both mono-electronic and electron-hole pair (excitonic) states.²⁸⁴ Notice that a similar diagram including the correct mono-electronic irreducible representations at the R point instead of the irreducible representations at the Γ point quoted here is given in ref.^{92,285} (a) The CB and VB have Γ_4^- and Γ_1^+ symmetry, respectively. (b) With the spin-orbit interaction, the CB splits into fourfold Γ_8^- and twofold Γ_6^- states, while the VB transforms as twofold Γ_6^+ in double representation. (c) The lowest-energy exciton is split by the Coulomb and exchange interactions into sixfold Γ_4^- and twofold Γ_1^- states and transitions to these states are optically allowed and forbidden, respectively. Reprinted from ref²⁸⁴, Copyright 2003, with permission from Elsevier.

3.1.3.3 Electronic structure in 2D networks ($n = 1$, $\langle 100 \rangle$ networks)

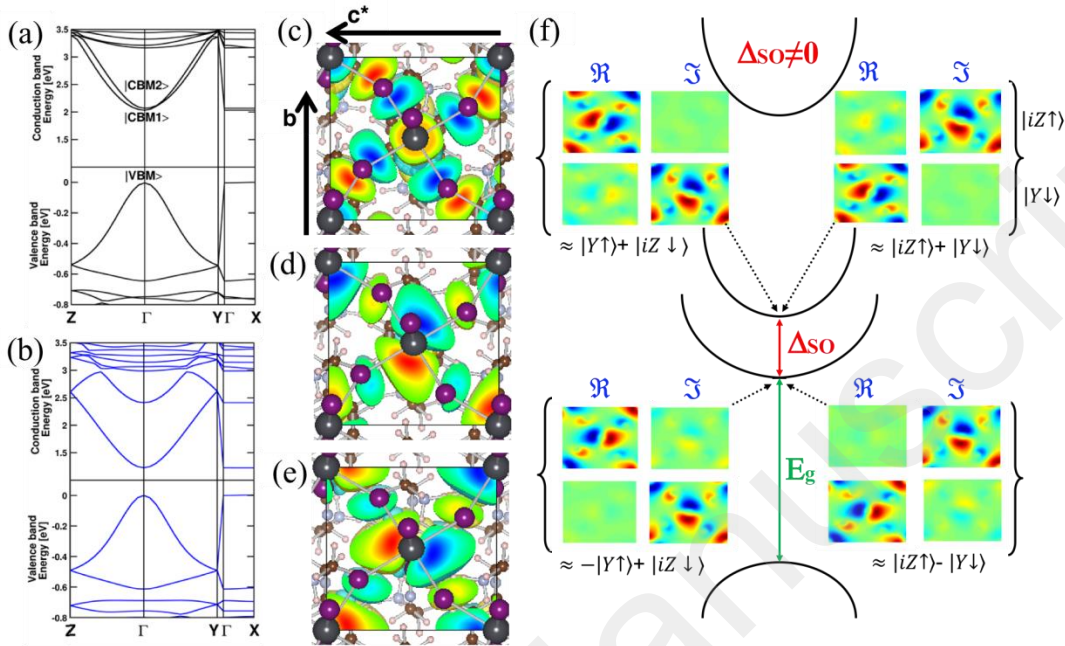


Figure 16. Electronic properties of $([pFC_6H_5C_2H_4NH_3]_2PbI_4)$ from DFT calculations. Band structure computed (a) without and (b) with SOC with energy levels referenced to the VBM. Both highlight the absence of dispersion and thus electronic coupling along Γ -X. Comparison of both band structures highlight the effect of SOC on the band gap, bandwidth and effective masses. Comparison to Figure 13 also illustrate the band gap increase related to quantum confinement. (c-e) Electronic wave functions (computed without SOC) of VBM, CBM1 and CBM2 as represented in the bc^* plane illustrating the σ -antibonding character of the in-plane hybridization between p -orbitals of halide and s -orbitals of metal in VBM (c) and p -orbitals of metal with s -orbitals of halide for CBMs (d,e). (f) Schematic representation of the electronic band diagram when considering effect of SOC (Δ_{SO}) with (\Re) real and (\Im) imaginary parts of the complex spinorial components of CBM1 and CBM2 (top panels: spin up components; bottom panels: spin down components). Reprinted with permission from ref ⁶⁴. Copyright 2012 American Physical Society.

As for 3D perovskite crystals, the band gap of layered halide perovskite may or not show up at the center of the BZ depending on the space group and the unit cell used to compute the band structure. The simplest approximation to investigate lower dimension halide perovskite consists in considering an isolated single charged inorganic moiety. This was implemented by G. Papavassiliou and coll. as early as in 1994 on a single $(\text{SnI}_4)^{2-}$ $\langle 100 \rangle$ layer based on EHT.³⁸ They underlined the main features of the band structure edge states, with VBM largely composed of in-plane σ -antibonding combination of halide p-orbitals and metal s-orbitals, whereas the CBM was found to be primarily composed of out-of-phase hybridization between the metal and halide p-orbitals. A decade later, DFT calculations revealed that CBM involves both halide p and s-orbitals and concluded that CBM remains composed primarily of σ -antibonding states.²⁶⁵ In 2005, Knutson and Mitzi highlighted the different symmetry allowed in-plane hybridizations at the high symmetry points of the BZ, for instance CBM consisting of hybridization between metal p-orbitals and halogen p or s-orbitals at the Γ - or M-point, respectively.²⁰³ All these early works did not account for SOC but already stressed effect of quantum confinement on the band gap, bandwidth, DOS and the flat dispersions revealing absence of electronic coupling along the stacking axis. The first quantitative inspection of SOC in monolayered $\langle 100 \rangle$ halide perovskites conducted in 2012 revealed its impact on electronic (summarized in Figure 16) and optical properties.⁶⁴ To conclude on the orbital nature of the band edge states in $\langle 100 \rangle$ monolayered perovskites, VBM can be clearly thought as a σ -antibonding combination between the p-like equatorial and axial halide orbitals and s-like metal ones whereas the situation is more involved for the CBM as a result of SOC. CBM retains features of bonding interaction between metal p-

like orbitals, antibonding interactions between the latter and s-like orbitals of equatorial halogens and p-like orbitals of axial ones.

All the early studies on monolayered $\langle 100 \rangle$ perovskites also underlined another important factor that directly influences the band gap value, band diagram and related optical properties, namely the structural distortions of the inorganic perovskite layer.^{38,265,203,205} Knutson et al. investigated the band structure of a series of idealized $(\text{SnI}_4)^{2-}$ perovskite $\langle 100 \rangle$ sheets using EHT to decipher the crystal orbital origin related to in- and out-of-plane angular distortion and demonstrated that the in-plane Sn-I-Sn bond angle has the greatest impact on the band gap.²⁰³ This effect is similar to the one already discussed for 3D frameworks based on expressions (1) and (2) or (5) when considering SOC. More recently, it has been further quantified based on DFT calculations using a $\langle 100 \rangle$ Cs_2BX_4 model structure having rigid octahedral with D_{4h} symmetry,⁷⁶ which is illustrated in Figure 17. This model provides a synthetic view for structure-band gap relationships not only for monolayered $\langle 100 \rangle$ perovskites,^{76,206} but it has also proved useful for qualitative assessments in multilayered ones.^{84,204} Figure 17 shows that the band gap significantly opens upon octahedral tilting, mainly as a result of a lesser antibonding character of the VBM that gets stabilized. In addition, octahedral tilting leads to reduced hybridization and in turn to significant reduction of VBM and CBM bandwidths.

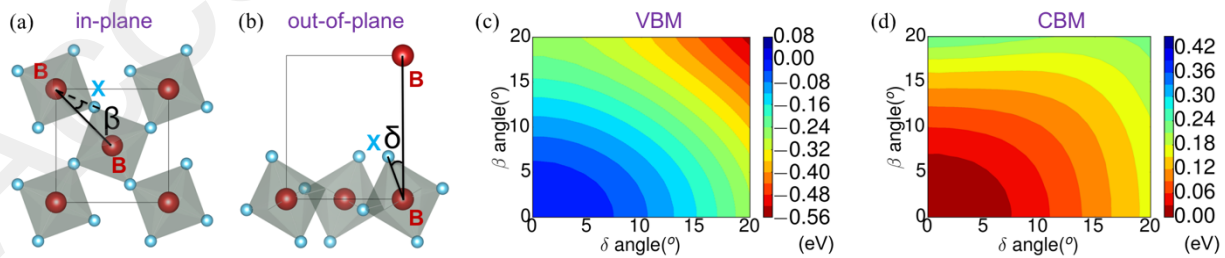


Figure 17. Cs_2BX_4 model to rationalize impact of octahedral tilting on band edge states. The model is composed of rigid $[\text{BX}_6]_2^-$ octahedra of D_{4h} symmetry.⁷⁶ Two angles characterize the

structural distortions: (a) β corresponds to the in-plane projection of the B-B-X angle and (b) δ corresponds to the out-of-plane B-B-X tilt angle. Color maps (c) and (d) report the variation of VBM and CBM state energies as a function of β and δ . They have been computed for $B = \text{Pb}$ and $X = \text{I}$ for a Pb-I distance of 3.18 Å using DFT and including SOC according to Ref.²⁰⁶. They reveal the larger impact of octahedral tilting on VBM (c) as compared to CBM (d) and the systematic band gap increase with structural distortion independently of any change in quantum or dielectric confinement.

The straightforward trends provided by such a model to experimentalists having on hand series of related crystal structures prompt its extension to $\langle 110 \rangle$ and $\langle 111 \rangle$ layered perovskites as well as implementation of more quantitative levels of theory (e.g. sc-GW, HSE or DFT-1/2). It may also be useful to further inspect a range of B-B in-plane distances as well as anisotropy in B-X bonds (apical versus equatorial), even so the effect of lattice strain has been demonstrated to have lesser impact than octahedral tilting (vide supra). Besides and to the best of our knowledge, a full TB model providing analytical expressions similar to (1)-(5) derived for 3D frameworks and tacking symmetry into account has still to be worked out for layered halide perovskites, whilst it is anticipated to afford similar sounded insight to the understanding of structure-properties relationships as well as an efficient tool to address large size structures, heterostructures and nanostructures.

3.1.3.4 Electronic structure and confinement effects

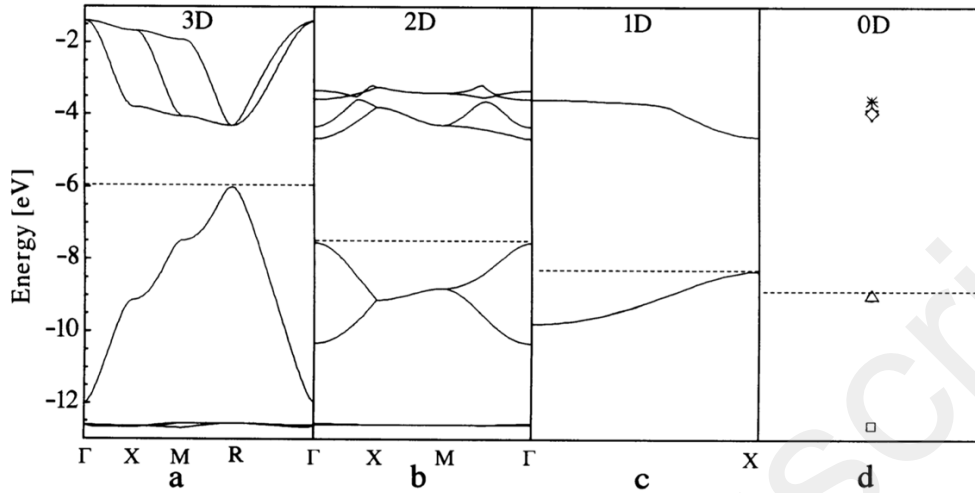


Figure 18. Electronic band structures of (a) $\text{CH}_3\text{NH}_3\text{PbI}_3$, (b) $(\text{C}_9\text{H}_{19}\text{NH}_3)_2\text{PbI}_4$, (c) $(\text{NH}_2\text{C}(\text{I})=\text{NH}_2)_3\text{PbI}_5$ and (d) $(\text{CH}_3\text{NH}_3)_4\text{PbI}_6 \cdot 2\text{H}_2\text{O}$. Dashed line indicates the position of E_{Fermi} . Reprinted with permission from ref³³ Copyright 1996 IOP Publishing.

In 1996, Koutselas et al. performed ETH calculations on a series of hybrid halide perovskites of various dimensions ranging from 3D down to 0D.³³ A modification of the Slater orbital exponent for the I 5p orbitals was used to obtain a close agreement between the calculated band gap and the experimental band gap in the case of the 3D network. This pioneering work demonstrated many of the main features of the band diagram of lower dimension halide perovskites. Figure 18 highlights: (i) the direct band gap, (ii) the increase in band gap, (iii) the decrease of bandwidth and (iv) the rise in DOS with decreasing dimensionality and thus effect of quantum confinement. As already mentioned in section 2.3, it is only recently that series of phase-pure multilayered halide perovskites have been achieved and crystal structures made available. These provide a fair playground to further investigate effect of quantum confinement based on electronic structure calculations. Figure 19a reports the band structures computed for a series of $(\text{CH}_3(\text{CH}_2)_3\text{NH}_3)(\text{CH}_3\text{NH}_3)_{n-1}\text{Pb}_n\text{I}_{3n+1}$ multilayered perovskites with $n = 1$ to 4.

Quantum confinement leads to a systematic decrease of the band gap with increasing thickness (n-value) of the perovskite layer sandwiched between the organic cations along with the increased number of subbands that equals the number n of octahedra in the perovskite layer. Similar systematic decrease has also been computed for model structures $A_{n+1}M_nX_{3n+1}$ ($A = \text{Cs, MA}$; $M = \text{Pb, Sn}$; $X = \text{I, Br, Cl}$).^{81,286} Corresponding local densities of states (computed within a small energy range close to the band edges) also reveal localization of electron and holes on specific sheets of the perovskite layer. The specific localization depends mainly on the amount of octahedral tilting within each layer, VBM being localized on the sheet with smallest octahedral tilting (Figure 17). Noteworthy, the density isosurfaces illustrate the knot spin texture^{287–289} of CBM subsequent to inclusion of SOC.

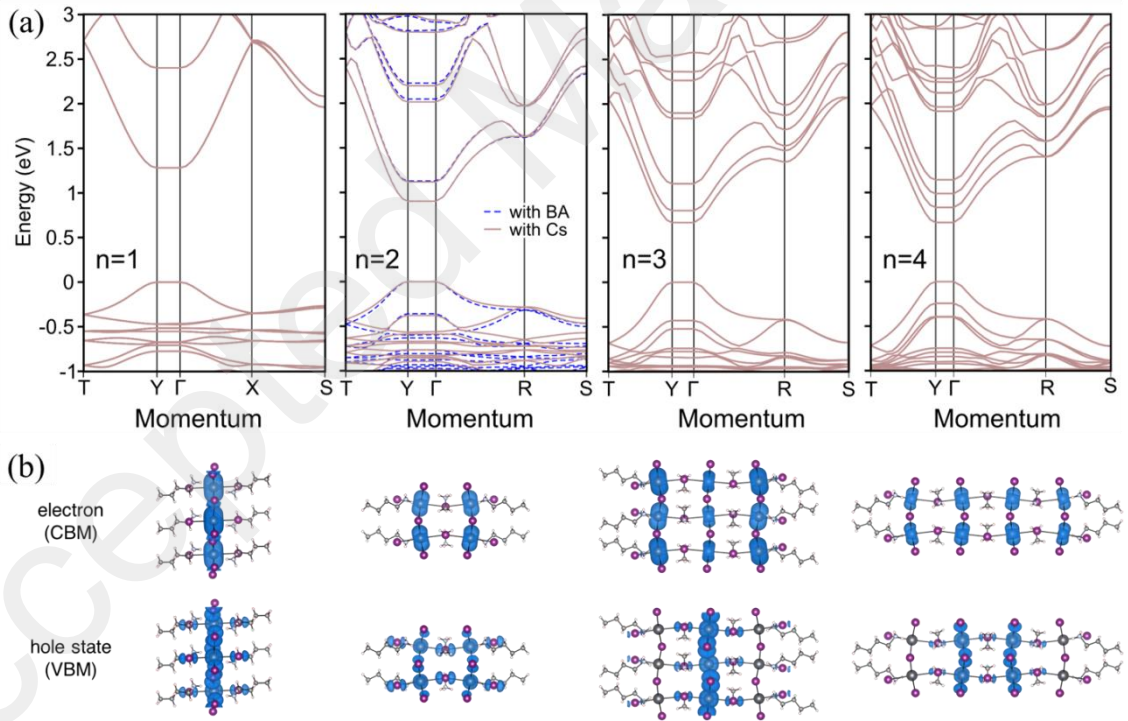


Figure 19. DFT computed electronic properties of $(\text{CH}_3(\text{CH}_2)_3\text{NH}_3)(\text{CH}_3\text{NH}_3)_{n-1}\text{Pb}_n\text{I}_{3n+1}$ with $n = 1$ to 4 including SOC. (a) Computed band structures for the experimental crystal structures obtained by replacing the organic cations $(\text{CH}_3(\text{CH}_2)_3\text{NH}_3)$ with Cs^+ (brown). Comparison to the

real structure is shown for $n = 2$ (dashed blue). (b) Corresponding local densities of states at CBM and VBM, evidencing the knot spin texture of CBM and localization on different perovskite sheets. Adapted with permission from ref²⁷⁵. Copyright 2018 NPG Publishing group.

We further stress that such localization on specific sheets is obtained based on static structures usually obtained from X-Ray diffraction. However, from molecular dynamics simulations²⁶⁸ and proton NMR²⁹⁰ there is clear evidence that both A and A' cations undergo a fast disordered dynamics (ps timescale), which has been suggested to lead to potential fluctuation sufficiently large to induce further localization of the band edge states.²⁶⁸ This also impels us to reconsider differently our view of hydrogen bonding in these intriguing hybrid halide perovskites. Meanwhile, when using static *average* structures, replacement of the molecules having a dipole with an alkali metal such as Cs⁺ may help to decipher whether the specific localization is mainly related to the specific dipolar orientations of the organic cations (*e.g.* CH₃NH₃⁺ Figure 19b) on hand or primarily to the detailed structure of the inorganic perovskite framework. Last but not least, we also suggest that caution be exercised when implementing DFT structural optimization of low dimensional hybrid halide perovskites. In fact, it may lead to significant structural distortions such as larger β and δ tilt angles (Figure 17a,b) that have been shown in section 3.1.3.3 to have a direct impact on their electronic structure with for instance increased band gaps (Figure 17c,d) that should not be traced back to quantum confinement. Changing the crystallographic space group for DFT structural optimization may induce additional structural distortions with direct impact on computed electronic properties.^{291,292} Another way to possibly tune confinement effects is to play with the thickness and composition of the organic layer that sandwiches the perovskite ones. It turns out that for most of the investigated layered structures, the first has minor effect on the computed electronic band

structure. For instance, in the series of alkylammonium lead iodide monolayers having from 4 up to 10 (and from 10 up to 18) carbon atoms in the alkyl chain, the increased size of the organic barrier does hardly alter band edge states and band gaps of the electronic band structure computed for the experimental room temperature structures.^{94,206,262} The band gap jump between the two sets of chain length can be traced back to a phase transition (section 3.2) that induces larger out-of-plane octahedral tilting (Figure 17).²⁰⁶ This is indicative of very small contribution from confinement effects and is in contradiction with results based on DFT optimized structures. For instance, predictions relevant to the room temperature optoelectronic properties of layered perovskites after systematic DFT structure optimization,²⁹³ must be considered with caution. Similar procedures are known to lead to sizeable variations of the band structure, band gap and effective masses in the case of 3D halide perovskites.^{294–296}

For small interlayer cations, especially for eclipsed configurations, electronic coupling between the perovskite layers may occur and lead to sizeable band dispersion in the direction of reciprocal space corresponding to the stacking axis,^{83,204} thus reducing effect of quantum confinement. The nature of the organic cation in the interlayer may also have significant influence on the band diagram due to varying dielectric constants of the organic layer. This is illustrated in Figure 20 using an extreme case where, the dielectric contrast is reversed after I₂ intercalation in (IC₆)₂[PbI₄] (Figure 20c-d).⁶⁹ The DFT band gap remains direct but is shown to decrease by 110 meV, (Figure 20a-b; 230 meV when including self-energy corrections, see section 4.3) and a set of narrow bands appears close to the Fermi level originating from the guest I₂ molecules. These bands show limited interaction with the levels of the inorganic layer and (IC₆)₂[PbI₄]·2I₂ can be considered as a weakly coupled composite. Halogen intercalation also reduces significantly the CBM bandwidth, which in turn increases its effective mass.

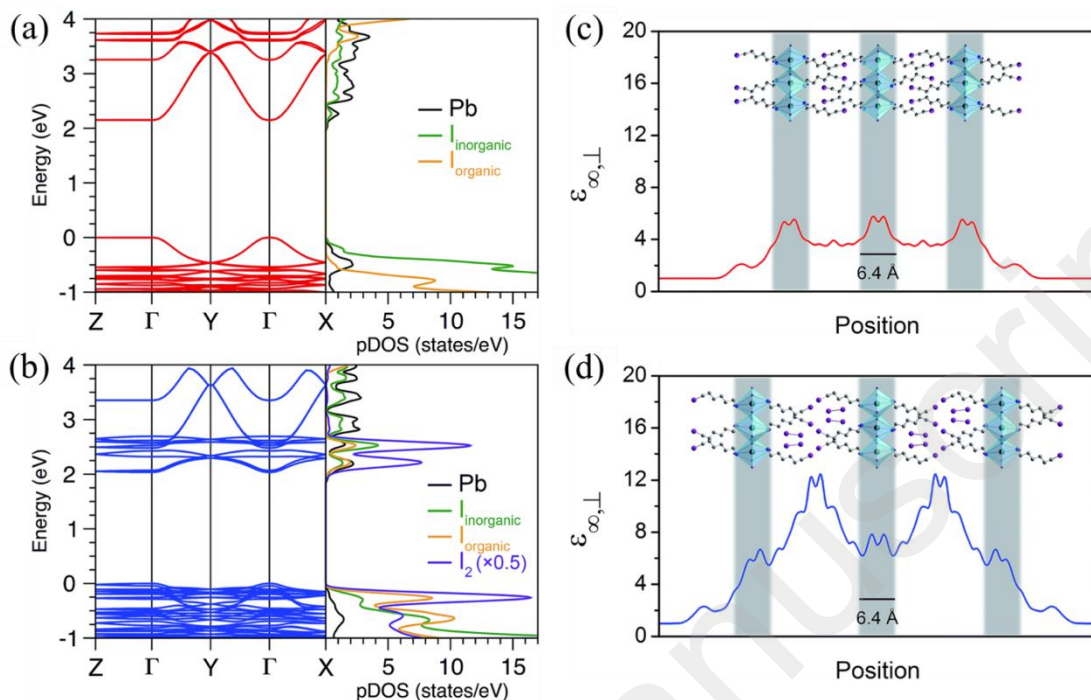


Figure 20. DFT computed electronic and dielectric properties for (a),(c) $(IC_6)_2[PbI_4]$ and (b), (d) after I_2 intercalation leading to $(IC_6)_2[PbI_4] \cdot 2I_2$. (a), (b) Band structure and projected density of states. (c), (d) corresponding calculated high-frequency dielectric profiles $\epsilon_{\infty,\perp}$ computed from slabs of (c) $(IC_6)_2[PbI_4]$ and (d) $(IC_6)_2[PbI_4] \cdot 2I_2$. Here, $\epsilon_{\infty,\perp}$ is the high-frequency dielectric constant perpendicular to the direction of layer propagation. Dark green, purple, blue, and grey spheres represent Pb, I, N, and C atoms, respectively. Hydrogen atoms are omitted for clarity. Reprinted with permission from ⁶⁹ – Published by the Royal Society of Chemistry.

3.1.3.5 Rashba effect in layered halide perovskites

A pending question is: are Rashba or Dresselhaus effects predicted so far by DFT for 3D halide perovskites also observable in layered halide perovskites? A spinor-splitting leading to such effects is generally expected when the spatial inversion symmetry is broken, while the time-reversal symmetry still holds for all the electronic states.⁹³ In layered heterostructures of classical

semiconductors, it yields spin-related splitting properties in all the directions perpendicular to the stacking axis because the stacking axis usually coincides with a symmetry axis. On the contrary, the stacking axis and the symmetry axis are predicted to be perpendicular in many practical cases in multilayered perovskites. A spin degeneracy is then expected to remain along one of the in-plane axis.^{66,297} Recent experimental results indicate that the observation of a Rashba or Dresselhaus effect might be possible in layered compounds (Figure 21).^{298,299}

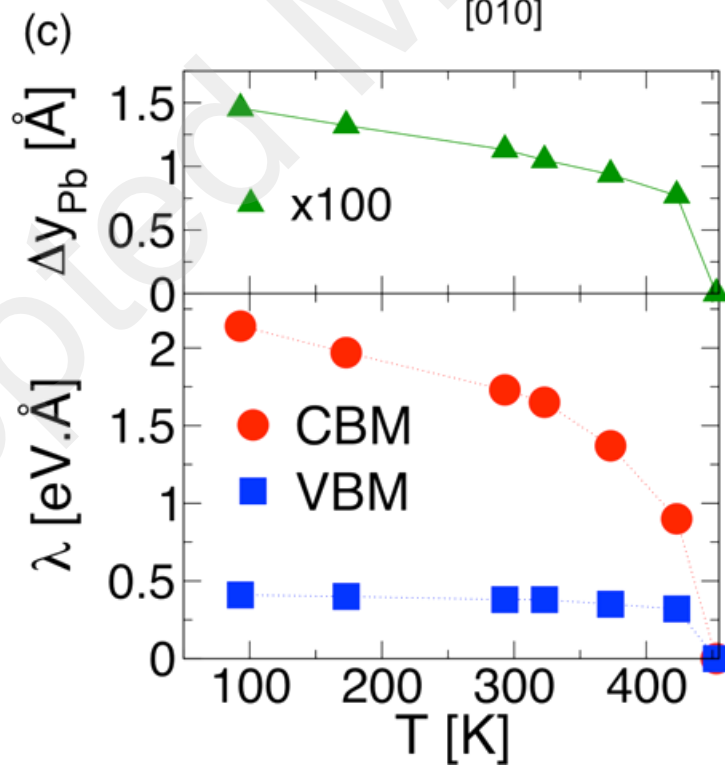
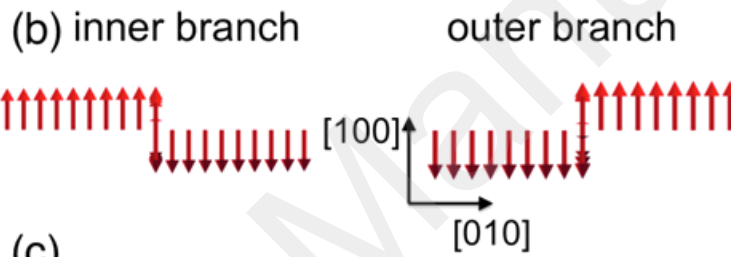
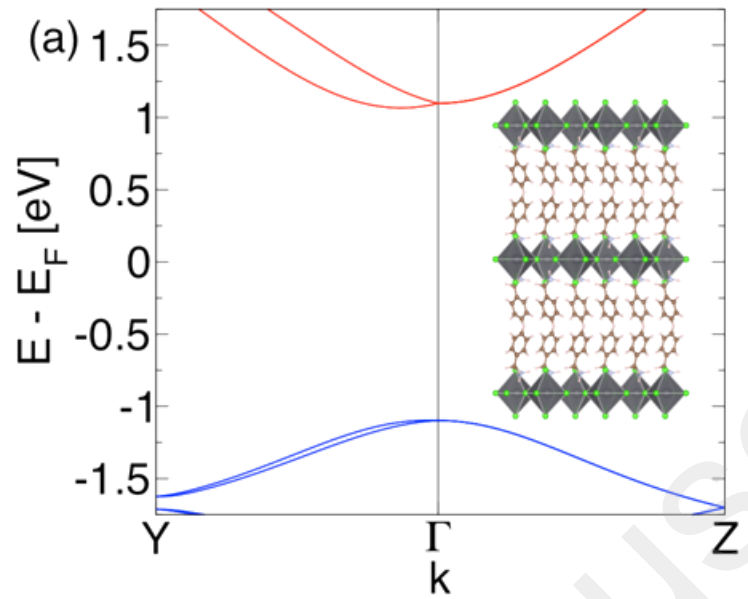


Figure 21. (a) Band structure of Bz_2PbCl_4 in the low temperature $Cmc21$ phase, computed with SOC. Blue and red bands correspond to occupied and unoccupied bands, respectively. (b) Spin textures for the inner and outer branches of the conduction band. (c) Temperature dependence of the Pb displacement along y and computed Rashba parameters for the conduction and valence bands. Reprinted with permission from ref⁶⁶. Copyright 2015 American Chemical Society.

3.1.4. Effective masses from band structures

Among the input parameters of models aiming at investigating effect of quantum and dielectric confinement as well as for device empirical simulation including transport simulations, the effective mass of charge carrier is an essential ingredient. The commonly used unit of measurement is the rest mass m_0 of an electron ($9.109 \cdot 10^{-31}$ kg or $5.486 \cdot 10^{-4}$ amu). It can be derived from electronic band structure calculations considering the band curvature near a high symmetry point of the BZ. As for other materials, deriving effective masses from DFT calculations raises technical issues. First and foremost, it relies on the assumption of parabolic band curvatures near the critical point and in the general case the effective mass has a tensor character as a result of anisotropy. In 3D halide perovskites, it has been demonstrated that the accuracy of effective masses much depends on the level of theory of the DFT approach,⁹³ and reliable estimates require self-consistent implementation of both relativistic (SOC) and many-body effects (sc-GW).³⁰⁰ This may lead to erroneous conclusions, as for instance those derived for multilayered perovskites of varying n-value.⁶¹ Unfortunately, implementation of SOC concomitantly with sc-GW is out of reach for most of the lower dimension halide perovskites as it is already a computationally challenging task for their 3D analogues.

Now, it has been shown that there is a direct relationship between the DFT computed band gaps and effective masses.³⁰¹ Using this relation of proportionality, the computed single particle DFT effective masses can be multiplied by the ratio between the experimental and DFT electronic band gaps. This procedure helped to reach good agreement between predicted and experimental effective masses for the $(\text{CH}_3(\text{CH}_2)_3\text{NH}_3)(\text{CH}_3\text{NH}_3)_{n-1}\text{Pb}_n\text{I}_{3n+1}$ series with $n = 1$ to 4 (Table 2). We also remind that this does not take the influence of thermal effects on the effective masses into account.

Table 2. DFT computed band gaps ($E_{g,DFT}$), experimental continuum band gaps ($E_{g,exp}$), experimental and DFT computed reduced masses $\mu = \left(\frac{1}{\mu} = \frac{1}{m_e} + \frac{1}{m_h}\right)$, as a function of n for $(\text{CH}_3(\text{CH}_2)_3\text{NH}_3)(\text{CH}_3\text{NH}_3)_{n-1}\text{Pb}_n\text{I}_{3n+1}$ multilayered perovskites.²⁷⁵ Computed masses are given for the [100] in-plane direction and [001] (except [010] for $n = 1$) out-of-plane direction, before ($\mu_{r,DFT}$) and after ($\mu_{r,scaled}$) band gap correction. The experimental exciton reduced mass, $\mu_{r,exp}$, is derived from fitting the diamagnetic shifts obtained from magneto-absorption spectroscopy to semi-empirical simulations of the Wannier-Mott exciton as described in Ref. [275].

n	infinity	1	2	3	4
$E_{g,DFT}$ (eV)	0,41	1,28	0,91	0,67	0,66
$E_{g,exp}$ (eV)	1,55	2,80	2,44	2,25	2,15
$\mu_{r,exp}$	0,104 ³⁰²	0,221	0,217	0,201	0,196
μ_{DFT} [100]	0,054 (0,049 ^a)	0,106	0,086	0,063	0,056
μ_{DFT} [001]	0,041 (0,037 ^a)	0,109 ^b	0,080	0,061	0,059
μ_{scaled} [100]	0,204 (0,186 ^a)	0,233	0,231	0,212	0,182

$\mu_{scaled} [001]$	0,155 (0,140 ^a)	0,238 ^b	0,216	0,206	0,191
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^a computed from DFPT

^b computed for the [010] direction

3.2. Concept of quantum confinement applied to multilayered halide perovskites

Before discussing the concept of quantum confinement applied to multilayered halide perovskites, it is useful to revisit some assumptions implicitly made by many authors performing theoretical analyses of the electronic structure of layer halide perovskites. Noteworthy as shown in the previous section, DFT simulations of the electronic properties of layered halide perovskite materials, usually assume the existence of a periodic lattice and a perfect lattice translational symmetry along the stacking axis for layered halide perovskites. 3D periodic Bloch functions are used to describe the electronic wavefunctions of the equivalent superlattice close to the electronic band gap: $\psi_{e,\mathbf{k}}(\mathbf{r}_e) = \frac{e^{i\mathbf{k}\cdot\mathbf{r}_e}}{\sqrt{V}} u_{e,\mathbf{k}}(\mathbf{r}_e)$ with a truly 3D periodic Bloch functions $u_{e,\mathbf{k}}(\mathbf{r}_e)$, where \mathbf{k} is the 3D wavevector inside the BZ of the periodic lattice, V a normalization volume and \mathbf{r}_e the position of the electron in real space. Such Bloch functions implicitly enforce delocalization of mono-electronic states along the stacking axis, although DFT studies on monolayered halide perovskite show that in most cases resulting superlattice effects are negligible.^{64,65,76,69} Without electronic coupling between the inorganic perovskite layers, flat dispersions of the electronic bands are computed from the BZ center Γ to a point located at the edge (\mathbf{Y} for example, Figure 19a) along the direction of the stacking axis. Mono-electronic states at the bottom of the CB or the top of the VB are thus degenerated along this line, and any superposition of the solutions corresponding to the \mathbf{k} wavevectors between Γ and \mathbf{Z} , shall then be considered as a possible solution as well. An opposite view consists in considering the perovskite

sheets as isolated entities from the electronic viewpoint leading to an alternative description of the electronic states close to the electronic band gap by:

$$\psi_{e,k_t}(\mathbf{r}_{te}, \mathbf{z}_e) = \frac{e^{i\mathbf{k}_t \cdot \mathbf{r}_{te}}}{\sqrt{A}} u_{e,k_t}(\mathbf{r}_{te}, \mathbf{z}_e) \quad (6)$$

with in-plane 2D periodic Bloch functions $u_{e,k_t}(\mathbf{r}_{te}, \mathbf{z}_e)$ and \mathbf{k}_t is the in-plane component of the wavevector, where A is a normalization surface, \mathbf{r}_{te} the in-plane position of the electron and \mathbf{z}_e the position along the stacking axis. In that case, the lack of out-of-plane periodicity or electronic coupling between the layers allows charge localization along the stacking axis. This description of mono-electronic states has been chosen for the simulation of the Wannier excitons.^{31–33,275} Carrier localization close to only one perovskite sheet for both electron and hole may be justified in that case by the attractive electron-hole interaction. In practical cases, even weak superlattice effects may indeed lead to charge localization along the stacking axis as a result of thermal or defect-induced fluctuations of the vertical ordering. Strong electron-phonon interaction leading to polaron formation,^{303,304} is another example where carrier localization along the stacking may occur. In order to analyze the localization of electronic states in the stack by DFT, very large supercell calculations including local lattice distortions and defects, as well as molecular dynamics²⁶⁸ will be certainly useful in the future and lead to more realistic simulations of mono-electronic states.

In classical multiple QW systems, a 3D DOS can be reconstructed by merging 2D-like DOS related to electronic subbands both in the CB and VB (Figure 1), as a result of i) an increase of the QW thickness (n in the case of multilayered halide perovskites, *e.g.* Figure 19), ii) a superlattice effect related to electronic coupling between QW (influence of the organic barrier

for multilayered perovskites), iii) both effects occurring simultaneously. In the future, understanding how the crossover occurs in terms of electronic structures between monolayered ($n = 1$) and 3D ($n = \infty$) halide perovskites, and how the 2D DOS merges into a 3D-like DOS in that case is of great interest. A complete symmetry analysis of the electronic structure of multilayered perovskites is also still lacking, although the spinors have been described for monolayered compounds (Figure 16).⁶⁴

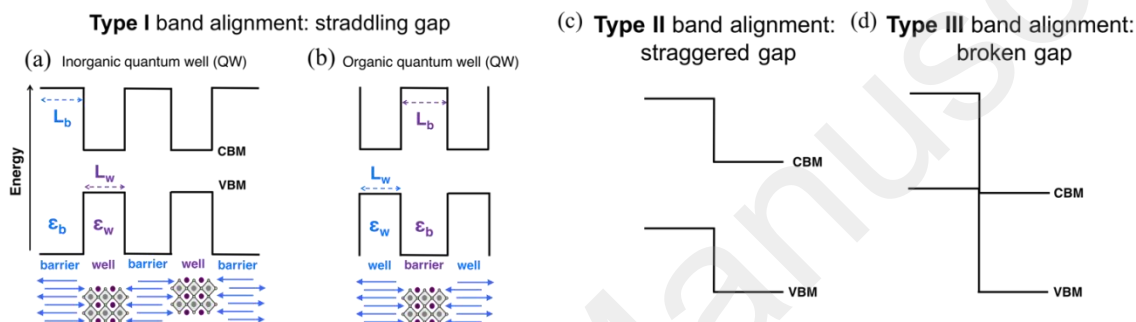


Figure 22. Schematic view of the three types of possible band alignments relative to VBM and CBM states of each material building the whole superlattice. (a) and (b) Type I band alignment in QW made of a well and a barrier of width and dielectric constant L_w/L_b and ϵ_w/ϵ_b , respectively: (a) Most common band alignment observed in hybrid halide perovskites illustrated for a layered structure where semiconducting perovskite sheets (mainly inorganic) alternate with organic layers having much wider band gaps. (b) Wider band gap perovskite and organic cation with smaller HOMO–LUMO gap result in the well/barrier roles of the organic and inorganic parts being switched. (c) and (d) Alternative band alignments obtained when shifting the electron affinity of the organic cation relative to the perovskite sheet, either affording a (c) type II alignment with a staggering of the energy levels or (d) a type III alignment having a broken gap.

For layered halide perovskites, the quantum confinement concept was discussed initially in terms of QW by several authors giving an appealing qualitative guidance for the researchers involved in the field of layered perovskites.^{26,38,30} A favorable situation for optoelectronic properties corresponds to the so-called type I QW case (Figure 22a), where both electrons and holes are confined within the well, leading to an enhancement of the radiative recombination and stronger binding energies for the exciton than in a 3D material. Changing the QW thickness gives also additional flexibility to tune the emission energy. The validity of a type-I QW to describe layered halide perovskites has been evaluated quantitatively using a specifically designed DFT-based approach.^{94,206} This approach shows that most known multilayered halide perovskite structures correspond to type I QW with very deep confinement potentials in a composite W/B structure (Figure 23c). Except a few cases related to short organic cations in the barrier (section 3.1.3.4), the electronic coupling between the perovskite sheets vanishes and the electronic band gap of the heterostructure does not depend strongly on the barrier thickness. In fact, figure 23 reports the computed band alignments for various $(C_jH_{2j+1}NH_3)_2PbI_4$ compounds with $4 \leq j \leq 18$. The electronic band gap computed for the RT crystallographic structures, undergoes a step-like increase for $j \geq 10$. This increase is related to a change of the molecular conformations leading to an increase of the in-plane and out of-plane tilt angles, being well reproduced within a Cs_2PbI_4 model structure (section 3.1.3.3 and Figure 17), *i.e.* without need of a quantum confinement effect.

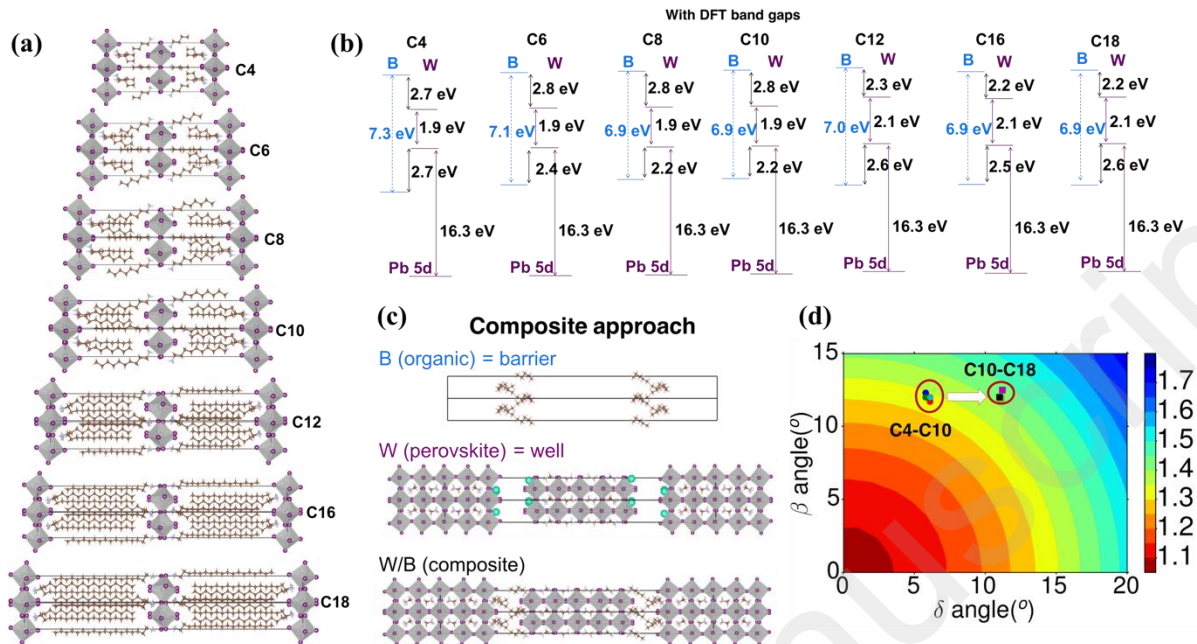


Figure 23. Band alignment for various barrier thicknesses. (a) Structures of $(C_jH_{2j+1}NH_3)_2PbI_4$. C_j next to the structures refers to the number (j) of carbon atoms in the alkyl chain. (b) Computed band alignments for the alkyl-based structures shown in panel a, taking crystallographic data recorded at room temperature. Since all the experimental band gaps are not available for the series, DFT band gaps for CB lineup were used. (c) Schematics of the composite model used to perform band alignment calculations defining the organic barrier (B), the perovskite well (W) of the whole composite structure (W/B). (d) Computed band gaps plotted on the 2D color map illustrating the variation of the band gap with octahedral in-plane (β) and out-of-plane (δ) tilt angles for a Cs_2PbI_4 model structure (Figure 17). Adapted with permission from ref²⁰⁶. Copyright 2018 American Chemical Society.

To compute the variation of the monoenergetic subband energies, the electronic band gap⁶¹ or the envelope functions of the edge states,³² it may appear appealing, at first sight, to use the type I QW picture (Figure 22a) together with an empirical transverse effective mass

Hamiltonian for the out-of plane carrier motion. The validity of such an approach can be inspected considering the series of BA-based multilayered perovskites. Figure 24 reports the band gap energy variation of $(\text{C}_6\text{H}_{13}\text{NH}_3)_2(\text{CH}_3\text{NH}_3)_n\text{Pb}_n\text{I}_{3n+1}$ with $n = 1-4$ as a function of the QW width using a transverse effective mass model.⁶¹ The confinement potentials used for the calculation are indicated as inset in the figure. A seemingly fair agreement is obtained with the reported experimental values, except for an overestimation of the band gap for the thinnest ($n = 1$) compound.

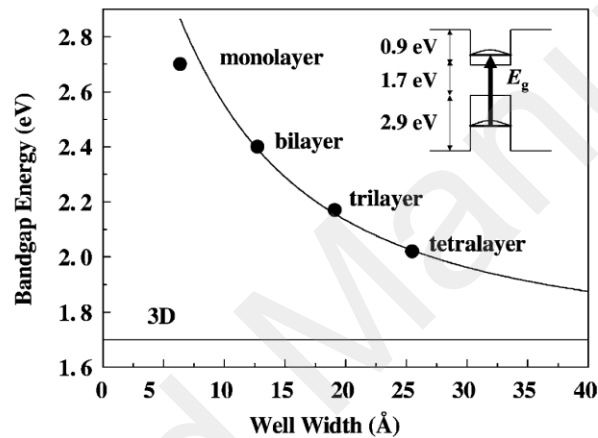


Figure 24. Band gap energy of $(\text{C}_6\text{H}_{13}\text{NH}_3)_2(\text{CH}_3\text{NH}_3)_n\text{Pb}_n\text{I}_{3n+1}$ as a function of the well width. The solid line represents the band gap energy calculated using a transverse effective-mass approximation. The inset provides the confinement energies and band gap used in the calculation. Reprinted with permission from ref⁶¹. Copyright © National Institute for Materials Science in partnership with Taylor & Francis, reprinted by permission of Taylor & Francis Ltd, <http://www.tandfonline.com> on behalf of © National Institute for Materials Science in partnership with Taylor & Francis.

The overestimation of the band gap for the monolayered compound is in fact not a detail. Indeed, any attempt to combine the type I QW picture and confinement potentials with an

effective mass Hamiltonian to better fit the whole curve leads to unrealistic parameters.^{94,206} Thus, the transverse effective mass model has no firm theoretical basis for layered halide perovskites, also predicting a strong electronic coupling between QW (perovskite sheets) in contradiction with DFT results (Figure 19).⁹⁴ The fundamental reason for this inconsistency is deeply rooted into the classical representation of the electronic states in QW superlattices:⁹⁴

$$\psi_e(\mathbf{r}_e) \approx \sum_i F_i(\mathbf{r}_e) U_{e,i}(\mathbf{r}_e) \quad (7)$$

where $F_i(\mathbf{r}_e)$ are slowly varying envelope functions along the stacking axis and $U_{e,i}(\mathbf{r}_e)$ periodic and rapidly oscillating ones. In order to understand the reasons for the failure of the type I QW picture associated to an effective mass model, it is important to realize first that for most multilayered halide perovskites, the actual QW are ultrathin. The equivalent QW thickness L_{QW} in multilayered perovskites of general formula $A'_2A_{n-1}B_nX_{3n+1}$ is typically on the order of $L_{\text{QW}}(\text{nm}) = n \times 0.6$ while in most cases $n < 3$. Ultrathin QW can also be encountered in conventional semiconductor heterostructures, but then great care should be taken to various effects such as the non-parabolicity of the electronic dispersions that requires to go beyond the effective mass approximation in order to compute the envelope functions $F_i(\mathbf{r}_e)$. The second fundamental issue is related to the implicit assumption that the periodic functions of the heterostructure entering formula (7) can be identified to bulk Bloch functions of both W and B partners of the composite structure (Figure 22a) $U_{e,i}(\mathbf{r}_e) \approx u_{e,W}(\mathbf{r}_e) \approx u_{e,B}(\mathbf{r}_e)$. This assumption holds true in many QW superlattices of classical semiconductors, because W and B partners are only slightly different by nature such as in GaAs/AlAs or InGaAs/InP based heterostructures. The truly hybrid character of multilayered halide perovskites, with W being

essentially inorganic and B organic (Figure 23c), obviously breaks this assumption. It is especially hard to make a direct connection between the properties of monolayered compounds and the properties of the reference 3D material (*e.g.* MAPbI₃). The chemical motif in monolayered compounds contains terminal halides along the stacking axis, by contrast to the corresponding 3D network. For those reasons any reliable theoretical description of hybrid layered halide perovskites should be based on DFT calculations although account of many-body corrections including spin-orbit coupling effect remain challenging.^{64,203,205,265,293,305}

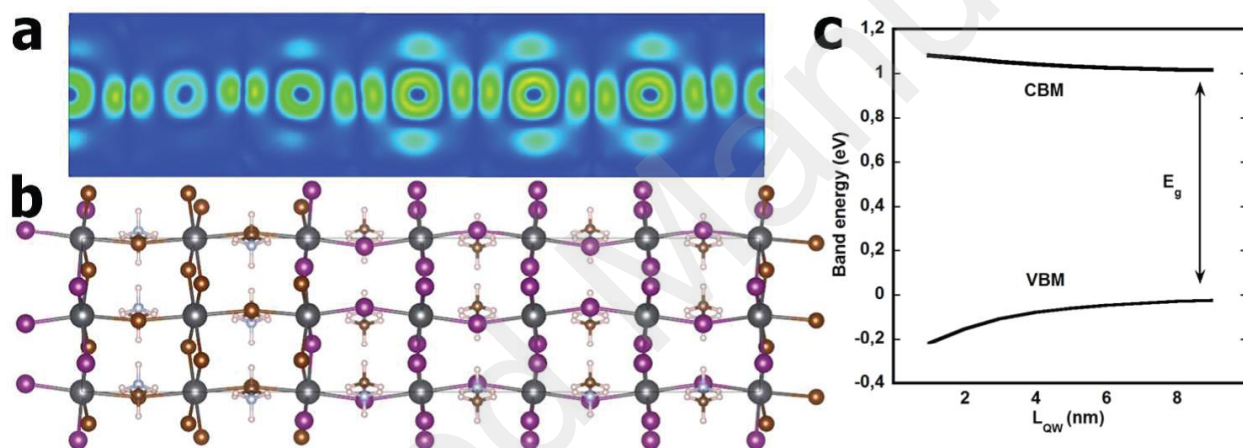


Figure 25. (a) Representation of the electronic density for the VBM of a short superlattice along the [010] direction of the low temperature bulk orthorhombic cell, containing two CH₃NH₃PbI₃ cells and one CH₃NH₃PbBr₃ cell. (b) Overview of the atomic structure. The diagrams are computed at the DFT level with a GGA exchange-correlation functional.²⁴⁸ SOC is not considered. (c) Electron (CB) and hole (VB) ground-state energies as a function of QW thickness (L_{QW}) computed with an empirical effective mass model. Reprinted with permission from ref⁶⁷. Copyright 2015 American Chemical Society.

Figure 25 shows a hypothetical short superlattice containing two $\text{CH}_3\text{NH}_3\text{PbI}_3$ cells and one $\text{CH}_3\text{NH}_3\text{PbBr}_3$ cell. In such a case, both compounds in the well and in the barrier share similar Bloch basis wavefunctions $U_{e,i}(r_e) \approx u_{e,W}(r_e) \approx u_{e,B}(r_e)$ and electronic density undergoes a slow variation along the stacking axis, corresponding to the envelope function picture $F_i(r_e)$. This shows that in such a case, the transverse effective mass approximation is valid, provided that non-parabolicity effects can be neglected.

4. MULTILAYERED HALIDE PEROVSKITES: DIELECTRIC CONFINEMENT

Summary: The localization of charge carriers in halide perovskite is strongly affected by the polarizability of the surrounding polar perovskite lattice. These self-energy contributions are enhanced in low dimensional perovskite structures and nanostructures by dielectric inhomogeneities. The latter also influence electron-hole pair interactions, giving rise to strong excitonic resonances. In this section, the theoretical schemes developed so far are reviewed. Current understanding relies on a combination of first-principles and empirical approaches, including semi-empirical resolution of the Bethe-Salpeter equation. Comparison to Van der Waals heterostructures³⁰⁶ are also emphasized to provide further insight into multilayered halide perovskites.

4.1. Excitons

The enhancement of the electron-hole interaction by dielectric confinement effects has been recognized, in the 1990s, by all the experimental groups working in the field and shown to lead to large binding energies for the exciton in monolayered perovskites. A few modelling attempts were based either on average dielectric constant or more elaborate dielectric well

models with abrupt interfaces.^{31–33} Since recently, realistic descriptions of the dielectric profiles are affordable thanks to specifically designed DFT-based approaches.^{307,81,95,83,69,275,206} It was also shown experimentally for crystals of $(C_6H_{13}NH_3)_2PbI_4$ (Figure 26,⁶² and more recently bromide compounds with $((C_4H_9NH_3)_2PbBr_4)$ and $(C_6H_5C_2H_4NH_3)_2PbBr_4$)³⁰⁸ as well as a $(C_4H_9NH_3)_2PbI_4$ nanosheet (Figure 27,³⁰⁹), that the excitonic bound state series (Ns, with N = 1, 2, 3...) exhibits strong deviations from the well-known 2D hydrogenic series, which yields the following expression for the binding energies $E_{b,N}$:

$$E_{b,N} = \frac{Ry}{\left(N - \frac{1}{2}\right)^2} \quad (8)$$

with Ry being the Rydberg energy of the equivalent 3D hydrogenic system. These deviations were attributed to dielectric confinement effects. The availability of precise crystallographic structures and the development of refined computational techniques for the electronic band structures, now allow addressing quantitatively the problem of the deviation of $E_{b,N}$ from the 2D hydrogenic series in multilayered perovskites (Figure 28).²⁷⁵

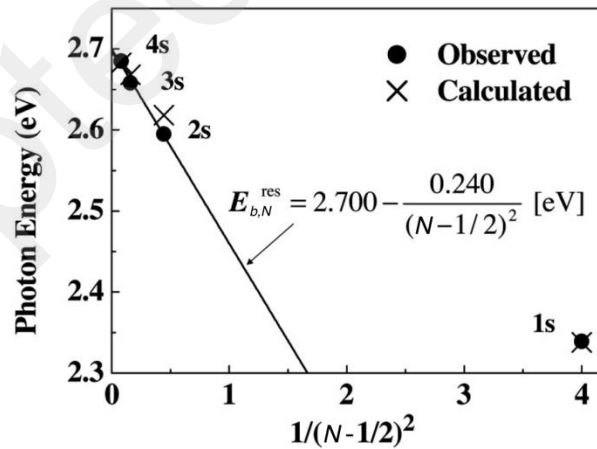


Figure 26. Resonance energies of Wannier-series excitons in $(C_6H_{13}NH_3)_2PbI_4$ as a function of $1/\left(N - \frac{1}{2}\right)^2$. Closed circles and crosses represent the observed and calculated

energies of the excitons, respectively. The solid line shows the fitting based on a simple two-dimensional Wannier exciton model. Reprinted with permission from ref⁶². Copyright 2005 American Physical Society.

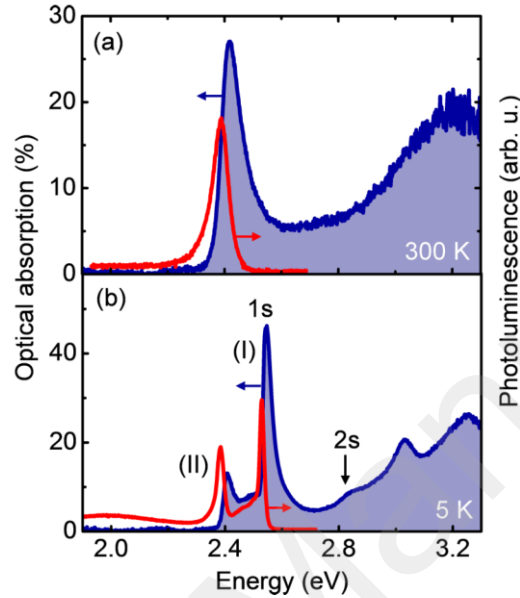


Figure 27. (a) Room-temperature optical absorption and PL spectra of an exfoliated $(C_4H_9NH_3)_2PbI_4$ nanosheet. (b) Optical absorption and PL from an organic-inorganic perovskite crystals nanosheet at 5 K. In about half of the nanosheets, an abrupt 100meV blueshift of the 1s excitonic state is observed at about 250K from position (II) to (I). It is attributed to a structural phase transition. Domains of both phases may coexist within a single flake as reported here. For each nanosheet, the N_s series is related to the dominant 1s excitonic peak at 5K. Reprinted with permission from ref³⁰⁹. Copyright 2015 American Physical Society.

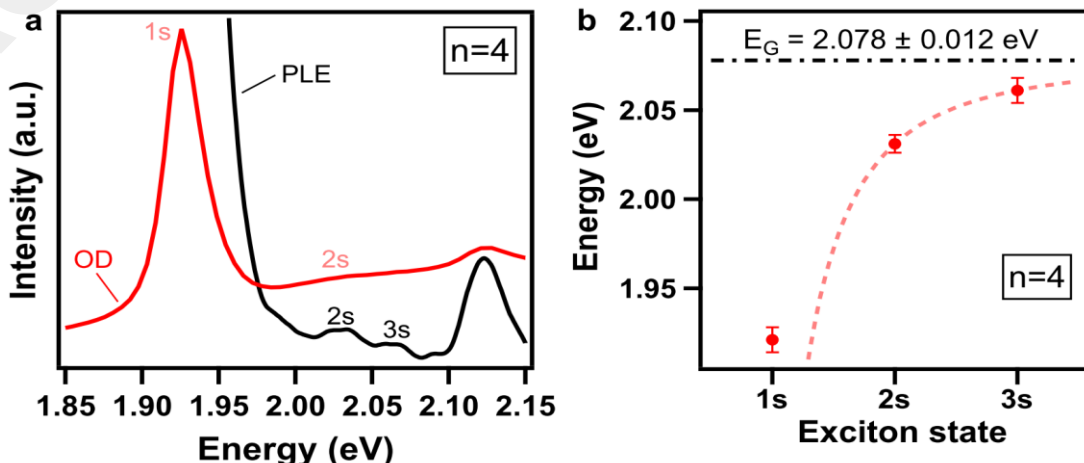


Figure 28. (a) Optical density (OD) and photoluminescence excitation (PLE) of the $(\text{CH}_3(\text{CH}_2)_3\text{NH}_3)(\text{CH}_3\text{NH}_3)_3\text{Pb}_4\text{I}_{13}$ multilayered perovskite ($n = 4$) clearly showing the exciton ground state $1s$, the excited exciton states $2s$ and $3s$. (b) Energy of the exciton Rydberg states for the multilayered perovskite $n = 4$. Dashed line is a fit to the $2s$ and $3s$ states with the 2D hydrogen model of exciton Rydberg series. Reprinted with permission from ref²⁷⁵. Copyright 2018 NPG Publishing group.

4.2. Single particle self-energies and electron-hole pair interactions – Shortcomings of simplified descriptions for layered perovskites

In order to properly understand the concept of dielectric confinement and its impact on the electronic and excitonic properties of composite compounds such as multilayered perovskites, it is important to review first a few basic physical concepts. The textbook of Jackson on classical electrodynamics³¹⁰ allows introducing in a simple manner the mutual pair interaction from opposite charges, while single particle self-energy contributions are considered as well. The mutual interaction energy of two point charges $\pm e$ (Figure 29) reads:

$$V(\mathbf{r}_h, \mathbf{r}_e) = \frac{-e^2}{4\pi \epsilon_M |\mathbf{r}_h - \mathbf{r}_e|}, \quad (9)$$

where ϵ_M is the macroscopic dielectric constant.

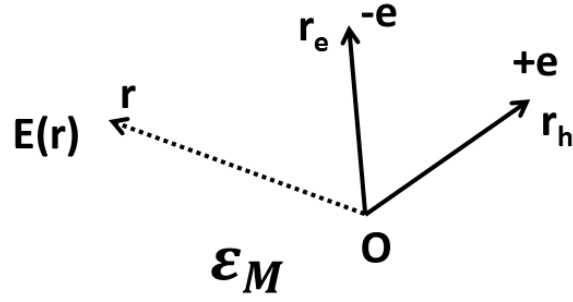


Figure 29. Classical representation of two opposite point charges $\pm e$ (electron (e) and hole (h)) located in a dielectric medium, whose polarizability is represented by a single effective parameter, the macroscopic dielectric constant ϵ_M . The analytic expression of the electrostatic field $\mathbf{E}(\mathbf{r})$ produced by the two charges at a general position \mathbf{r} is given by expression (13).

For a Wannier-Mott exciton, the mutual interaction between an electron and a hole is computed from a similar simple expression. It leads to the resolution of a Wannier-Mott equation for the relative motion of the electron-hole pair, which is directly connected in 3D to the well-known solution of the Schrödinger equation describing the electronic states in the hydrogen atom. This approach was generalized for any fractional space dimension α between 1 and 3, leading to the following Schrödinger equation for a Wannier-Mott exciton:³¹¹

$$\left[\frac{-\hbar^2}{2\mu\rho^{\alpha-1}} \frac{\partial}{\partial\rho} \left[\rho^{\alpha-1} \frac{\partial}{\partial\rho} \right] + \frac{l^2}{2\mu\rho^2} - \frac{e^2}{4\pi\epsilon_M\rho} \right] \psi(\rho, \theta) = (E - E_g) \psi(\rho, \theta) \quad (10)$$

where E_g is the electronic band gap of the semiconductor, μ is the reduced mass ($\frac{1}{\mu} = \frac{1}{m_e} + \frac{1}{m_h}$), l^2 corresponds to the angular-momentum operator, $\rho = |\mathbf{r}_h - \mathbf{r}_e|$ is the radial distance between the electron and hole ($0 \leq \rho \leq \infty$) and θ the related angle ($0 \leq \theta \leq \pi$). The discrete bound state energies are given by a generalized Rydberg series:

$$E_{b,N} = \frac{\text{Ry}}{\left(N + \frac{\alpha-3}{2}\right)^2} \quad (11)$$

where $N = 1, 2, 3, \dots$. This expression allows recovering the 2D and 3D hydrogenic series respectively for $\alpha = 2$ and $\alpha = 3$, and also conveniently provides a simple way to analyze anisotropic or confined systems considering α as an adjustable parameter, including a complete calculation of the absorption spectrum.³¹¹ This method was used by Tanaka and coworkers⁵⁹ to analyze spectroscopy results on the 1s and 2s exciton resonances in $(\text{C}_6\text{H}_{13}\text{NH}_3)_2\text{PbI}_4$ leading to a fractional dimension $\alpha = 2.3$. In classical semiconductor QW it is in principle further possible to account for both the effect of finite thickness and reduced mass mismatch between the well and barrier materials, by using expressions such as $\alpha = 3 - e^{-\frac{L}{2a_B}}$ where a_B is the Bohr radius and L an effective thickness.³¹² This approach was recently refined to study the 1s exciton bound state energy variation as a function of the number of inorganic layers (QW thickness) in halide perovskite multilayered perovskites, by tuning the value of α as a function of the QW thickness with (Figure 30):²⁷⁵

$$\alpha = 3 - \gamma e^{-\frac{L_W}{2a_B}} \quad (12)$$

The additional empirical factor γ was introduced to better account for the dielectric confinement effect and for the deviation from a pure quantum confinement enhancement of the exciton binding energy.

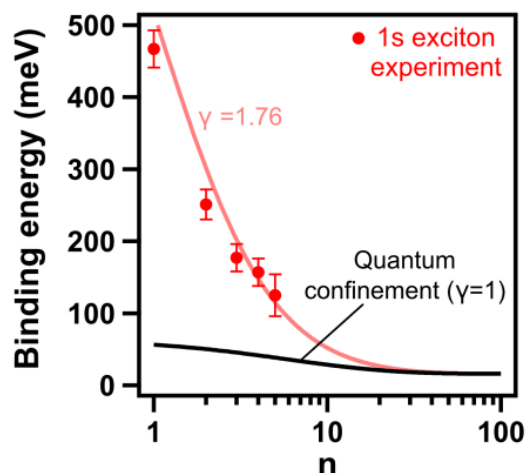


Figure 30. Results for the binding energy of the exciton ground state, showing the enhancement of the binding energy due to dielectric confinement. The black curve corresponds to the scaling law using expression (11) with $\gamma = 1$ in expression (12). The red curve gives the general scaling law of the exciton binding energy with the perovskite layer thickness using $\gamma = 1.76$. $R_y = 16 \text{ meV}$, $a_0 = 4.6 \text{ nm}$, and $L_w = 0.6292 \times n$ in nanometers. Reprinted with permission from ref²⁷⁵. Copyright 2018 NPG Publishing group.

It allows predicting that the exciton binding energy in multilayered perovskites is larger than room temperature thermal fluctuations up to $n \sim 20$ (equivalent QW thickness of $\sim 12.6 \text{ nm}$). A fine description of the complete series of the Ns resonances as a function of the QW thickness is not possible and the effect of dielectric confinement is not correctly captured by such simplified approaches.

More, the underlying representation of electron and hole as point charges, which corresponds to a convenient expression of the mutual interaction potential, simultaneously leads to a major physical inconsistency on the associated self-energies. As quoted by Jackson in his textbook,³¹⁰ a negative potential energy is in apparent contradiction with the positive expression of the

electrostatic energy density $w(\mathbf{r}) = \frac{\epsilon_M E(\mathbf{r})^2}{2}$. The total electric field $E(\mathbf{r})$ produced by the two opposite charges (Figure 29) is indeed:

$$\mathbf{E}(\mathbf{r}) = e \frac{\mathbf{r}-\mathbf{r}_h}{4\pi \epsilon_M |\mathbf{r}-\mathbf{r}_h|^3} - e \frac{\mathbf{r}-\mathbf{r}_e}{4\pi \epsilon_M |\mathbf{r}-\mathbf{r}_e|^3} \quad (13)$$

leading to three terms for the electrostatic energy density:

$$w(\mathbf{r}) = \frac{e^2}{32\pi^2 \epsilon_M |\mathbf{r}-\mathbf{r}_h|^4} + \frac{e^2}{32\pi^2 \epsilon_M |\mathbf{r}-\mathbf{r}_e|^4} - \frac{e^2 (\mathbf{r}-\mathbf{r}_h) \cdot (\mathbf{r}-\mathbf{r}_e)}{16\pi^2 \epsilon_M |\mathbf{r}-\mathbf{r}_h|^3 |\mathbf{r}-\mathbf{r}_e|^3} \quad (14)$$

The last term is negative and, after integration over space, allows recovering the mutual interaction energy between the two charges (expression (9)). The first two terms are positive making the sum of the three terms positive. They are thus necessary to solve the above-mentioned paradox. These first two terms are only related to one charge (either $+e$ or $-e$) and called for that reason self-energies. They correspond to the interaction of a single charge with the background medium, represented by the electron gas in the VB of a semiconductor when considering $\epsilon_M = \epsilon_\infty$. At this level of theory, a major problem arises, as the integration of the first and second terms over space diverges. This inconsistency is related to the representation of the interacting electron and hole by point charges, and quantum theory approaches are required for a proper description. It also indicates that the simple representation of the mutual interaction between the electron and hole in a Wannier-Mott equation for the exciton can only be considered as an approximation.

More, the representation of the medium polarizability by a single dielectric parameter ϵ_M is not suited for complex situations such as the one encountered in multilayered halide perovskites. Indeed, these materials are composite and inhomogeneous materials at the atomic scale (*e.g.* Figure 23a,c). In the framework of electrodynamics, the retarded response of the polarization induced in a medium by an electric field is expressed by:

$$\mathbf{P}(\mathbf{r}, t) = \varepsilon_0 \iint \chi(\mathbf{r}, \mathbf{r}', t, t') \mathbf{E}(\mathbf{r}', t') d^3 \mathbf{r}' dt' \quad (15)$$

where $\chi(\mathbf{r}, \mathbf{r}', t, t')$ is the two-point susceptibility. For a system at equilibrium, this quantity can be considered homogeneous in time, depending only on $t - t'$. For a continuous dielectric medium, spatial homogeneity leads in addition to a $\mathbf{r} - \mathbf{r}'$ dependence of the susceptibility $\chi(\mathbf{r} - \mathbf{r}', t - t')$. After double Fourier transform, a linear expression can be obtained:

$$\mathbf{P}(\mathbf{q}, \omega) = \varepsilon_0 \chi(\mathbf{q}, \omega) \mathbf{E}(\mathbf{q}, \omega) \quad (16)$$

It shows that a better description of the problem relies on a refined expression of the macroscopic dielectric tensor $\boldsymbol{\varepsilon}_M(\mathbf{q}, \omega)$, including wave vector \mathbf{q} and frequency ω dependencies. However, for planar systems at the nanometer scale, such as QWs, the spatial homogeneity assumption breaks down.³¹³ This assumption is related to neglecting wavelength components smaller than the lattice parameter in periodic solids, which can more or less be justified in bulk solids in some situations (vide infra), but not in dielectric heterostructures especially close to the interfaces. This is at the heart of the so-called ‘‘dielectric confinement’’ effect.

At the atomic scale in solids, the concept of homogeneity is related to the crystal lattice periodicity. In a bulk periodic solid, the local ‘microscopic’ electric field is given at the atomic scale by (time dependence is not written for sake of clarity):

$$\mathbf{E}(\mathbf{r} + \mathbf{R}) = \sum_{\mathbf{q}, \mathbf{G}} e^{i\mathbf{G} \cdot \mathbf{r}} \mathbf{E}_{\mathbf{G}}(\mathbf{q}) \quad (17)$$

where \mathbf{R} is a crystal lattice vector, \mathbf{G} a wavevector of the associated reciprocal lattice, \mathbf{r} a vector representing a position within the primitive cell and \mathbf{q} a wavevector within the BZ. The ‘macroscopic’ electric field $\mathbf{E}_M(\mathbf{R})$ is defined at the nanoscale by averaging the local field over the primitive cell: $\mathbf{E}_M(\mathbf{R}) = \frac{1}{V_{cell}} \int_{cell} \mathbf{E}(\mathbf{r} + \mathbf{R}) d^3 \mathbf{r}$. The Fourier transform of the ‘macroscopic’

electric field is then defined by retaining only the $\mathbf{G} = \mathbf{0}$ component of the ‘microscopic’ electric

field in momentum space $\mathbf{E}_M(\mathbf{q}) = \mathbf{E}_{\mathbf{G}=\mathbf{0}}(\mathbf{q})$. The \mathbf{q} dependence allows spatial variation of the ‘macroscopic’ electric field over distances greater than the primitive cell dimension. Similar considerations for the ‘microscopic’ two-point dielectric permittivity in a periodic solid $\epsilon(\mathbf{r} + \mathbf{R}, \mathbf{r}' + \mathbf{R}')$, where \mathbf{R}, \mathbf{R}' are lattice vectors, leads to the definition of a matrix $\epsilon_{\mathbf{G}, \mathbf{G}'}(\mathbf{q})$ after Fourier transform $\epsilon(\mathbf{q} + \mathbf{G}, \mathbf{q}' + \mathbf{G}') = \epsilon_{\mathbf{G}, \mathbf{G}'}(\mathbf{q})\delta_{\mathbf{q}, \mathbf{q}'}$. The ‘macroscopic’ dielectric constant is then defined by $\epsilon_M^{-1}(\mathbf{q}) = \epsilon^{-1}_{\mathbf{G}=\mathbf{0}, \mathbf{G}'=\mathbf{0}}(\mathbf{q})$. Notice that the value of $\epsilon_M(\mathbf{q})$ is related to a single component of the inverse of the dielectric matrix $\epsilon^{-1}_{\mathbf{G}=\mathbf{0}, \mathbf{G}'=\mathbf{0}}(\mathbf{q})$ and not of the matrix itself $\epsilon_M(\mathbf{q}) \neq \epsilon_{\mathbf{G}=\mathbf{0}, \mathbf{G}'=\mathbf{0}}(\mathbf{q})$. This stems from inclusion of local field effects.³¹⁴ It shows that considering ‘macroscopic’ dielectric constants in dielectric heterostructures instead of truly ‘microscopic’ quantities, may lead to severe approximations if the averaging of the properties over distances on the order of a unit cell or a bond length in a covalent solid is not justified. In such cases, it is necessary to consider advanced descriptions of the electronic structures in order to better account for the dielectric properties. This is precisely one of the issues, when considering composite hybrid materials such as layered halide perovskites, with strong inhomogeneities over short length scales.

4.3. Many-body approaches to the single particle self-energies and electron-hole interaction

In quantum many-body theory of bulk materials, single particle self-energies can be calculated in the GW approximation from :³¹⁵

$$\Sigma(\mathbf{r}, \mathbf{r}', \omega) = i \int \frac{dE'}{2\pi} e^{-i\delta E'} G(\mathbf{r}, \mathbf{r}', E - E') W(\mathbf{r}, \mathbf{r}', E') \quad (18)$$

where $\delta = 0^+$ (approaches zero from positive values), $G(\mathbf{r}, \mathbf{r}', E - E')$ is the single-particle Green's function and $W(\mathbf{r}, \mathbf{r}', E')$ the screened Coulomb interaction W . The screened Coulomb interaction $W(\mathbf{r}, \mathbf{r}', \omega)$ can be computed from the unscreened interaction $V(\mathbf{r}, \mathbf{r}', \omega)$ when the two-point 'microscopic' dielectric permittivity is known:

$$W(\mathbf{r}, \mathbf{r}', \omega) = \int \varepsilon^{-1}(\mathbf{r}, \mathbf{r}'', \omega) V(\mathbf{r}'', \mathbf{r}', \omega) d^3 \mathbf{r}'' \quad (19)$$

Nowadays, single particle self-energies are often used to successfully correct the well-known under-estimation of the electronic band gaps of bulk semiconductors computed at the DFT atomistic level.³⁰¹ However performing such calculations for very large heterostructures, like layered halide perovskites, has a high and unaffordable computational cost, including the computation of the 'microscopic' two-point dielectric permittivity $\varepsilon(\mathbf{r}, \mathbf{r}'', \omega)$. More, the physical meaning of the dielectric confinement effect appears more clearly in simpler semi-empirical approaches (vide infra). Although the proper evaluation of $\varepsilon(\mathbf{r}, \mathbf{r}'', \omega)$ is a major issue from the computational view point, it shall not be forgotten that any modification of the dielectric permittivity by carrier injection, dielectric confinement effect or coupling to optical phonons shall affect the exciton binding energy, but also the single particle self-energies and thus in turn the electronic band gap.

In quantum many-body approaches of the two particles problem for an electron and a hole, the BSE yields the exciton energies and wavefunctions.²⁵⁴ It can be solved in various ways, by iterative inversion of a Dyson-like equation or by constructing the exciton wavefunction as a sum of electronic transitions between VB and CB states obtained from a DFT+GW initial calculation.³¹⁶

$$\Psi(\mathbf{r}_e, \mathbf{r}_h) = \sum_{vck} A^{vck} \psi_{vk}^*(\mathbf{r}_h) \psi_{ck}(\mathbf{r}_e) \quad (20)$$

where A^{vck} is the exciton envelope function, $\psi_{ck}(\mathbf{r}_e)$ ($\psi_{vk}(\mathbf{r}_h)$) the CB (VB) mono-electronic state wavefunctions. An effective Hamiltonian is obtained for the envelope function containing a diagonal related to the single particle eigenstates and a nondiagonal part, including both a statically screened Coulomb interaction and an unscreened exchange interaction (this last contribution is not reported here):

$$\hat{H}_{cv'}^{cv}(\mathbf{k}, \mathbf{k}') = (E_c(\mathbf{k}) - E_v(\mathbf{k}))\delta_{c,c'}\delta_{v,v'}\delta_{\mathbf{k},\mathbf{k}'} + W_{cv'}^{cv}(\mathbf{k}, \mathbf{k}') \quad (21)$$

where the screening of the Coulomb interaction $W_{cv'}^{cv}(\mathbf{k}, \mathbf{k}')$ is again related to the inverse of the dielectric matrix $\epsilon^{-1}_{G,G'}(\mathbf{q})$ ($\mathbf{q} = \mathbf{k} - \mathbf{k}'$). This approach was used for the exciton in 3D perovskites in 2016,³¹⁷ but similar theoretical analyses of the exciton in layered perovskites are extremely rare due to the large size of the system. It was performed only for a few monolayered compounds (see (EDBE)PbX₄ (EDBE = 2,2-(ethylenedioxy)bis-(ethylammonium), X = Cl, Br) on Figure 31).^{240,304,318}

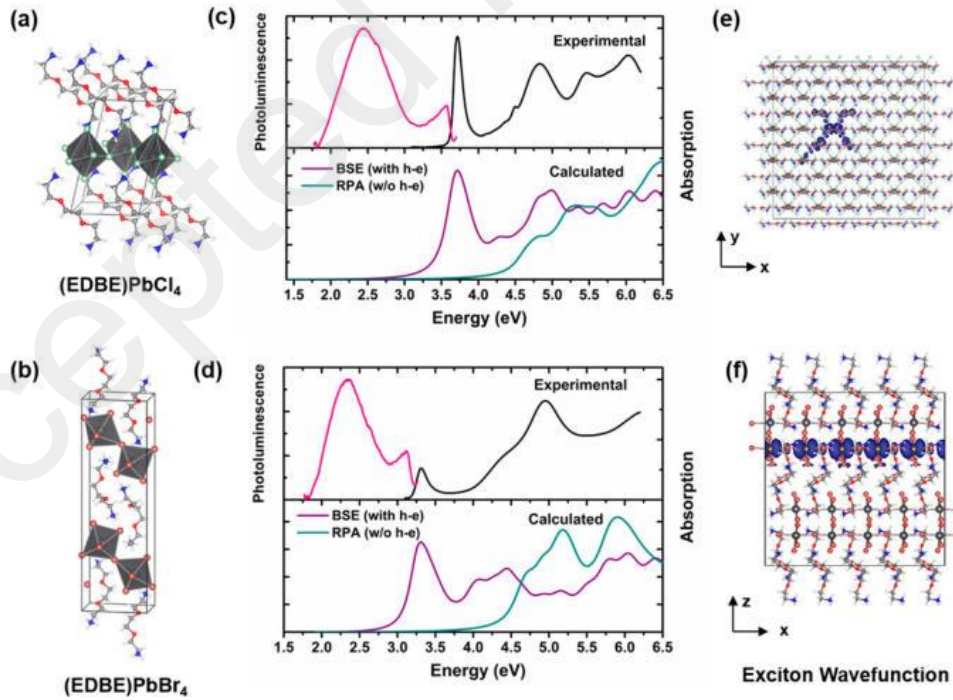


Figure 31. Crystallographic, optical, and excitonic characteristics of the two-dimensional (EDBE)PbCl₄ and (EDBE)PbCl₄ hybrid perovskites: (a,b) crystal structures; (c,d) experimental absorption and PL spectra at room temperature (upper panel) and calculated absorption spectra with and without consideration of hole–electron (h–e) interactions (bottom panel); and (e,f) illustration of the exciton wave functions corresponding to the excitonic peaks. Reprinted with permission from ref³⁰⁴. Copyright 2017 American Chemical Society.

Large exciton binding energies were obtained from the calculation for 1s resonance in reasonable agreement with the experimental results, but no results related to the higher orders (2s,3s, ...) of the exciton resonances were given and the broadening of the 1s exciton resonance was also overestimated (Figure 31). More, due to the large size of the systems, single particle energies $E_c(\mathbf{k}), E_v(\mathbf{k})$ and wavefunctions were only evaluated at the DFT level. Adhoc scissor corrections on the electronic band gaps were then applied to match the experimental absorption spectra. This interesting contribution clearly shows the possibilities and the limitations of such DFT-based atomistic approaches to scrutinize the exciton in layered halide perovskites.

A simplified Hamiltonian can be used to reduce the computational complexity:³¹⁶

$$\hat{H}_{c'v'}^{cv}(\mathbf{k}, \mathbf{k}') = \left(E_g + \frac{\hbar^2 k^2}{2\mu} \right) \delta_{c,c'} \delta_{v,v'} \delta_{\mathbf{k},\mathbf{k}'} - \frac{e^2}{\epsilon_\infty q^2} \quad (22)$$

After Fourier transform of the corresponding Schrödinger equation, the 3D Wannier-Mott equation in position space is recovered.³¹⁶ The electronic band gap E_g can be either evaluated from a DFT calculation including self-energy corrections or considered as an empirical input parameter. The reduced effective mass μ and the macroscopic dielectric constant ϵ_∞ may also be considered as empirical inputs.

In parallel with the development of many-body theories based on DFT, semiconductor Bloch equations were proposed to study the influence of many-body effects on the optoelectronic properties of semiconductors.³¹³ These equations describe the optical response of semiconductors under low or high levels of excitation and can be coupled to Maxwell's equations. The single particle energies are usually computed from the empirical effective mass approximation for single band Hamiltonians or from the $\mathbf{k}\cdot\mathbf{p}$ approximation for multiband empirical Hamiltonians. At low carrier injection and considering non-interacting electron-hole pairs, an effective Hamiltonian is obtained for the envelope function:

$$\hat{H}_{cv'}^{cv}(\mathbf{k}, \mathbf{k}') = \left(E_g + \frac{\hbar^2 k^2}{2\mu} \right) \delta_{c,c'} \delta_{v,v'} \delta_{\mathbf{k},\mathbf{k}'} + W_{cv'}^{cv}(\mathbf{k}, \mathbf{k}') \quad (23)$$

This effective Hamiltonian is similar to the previous one, but allows mixing an empirical description of the mono-electronic states with refined expressions of the electron-hole pair interaction in momentum space beyond the expression suited for 3D Wannier-Mott equation ($W_{cv'}^{cv}(\mathbf{k}, \mathbf{k}') = -\frac{e^2}{\epsilon_\infty |\mathbf{k}-\mathbf{k}'|^2}$). The excitons can be studied in 3D bulk semiconductors, but also in QWs provided that the proper expression of the screened interaction is used.³¹⁹ An equivalent approach at the same level of theory consists in computing the exciton Green function $G(\mathbf{k}, \hbar\omega + i\Gamma)$ from a Dyson like equation by iterative inversion to describe bulk 3D³²⁰ or QW³²¹ semiconductors:

$$G(\mathbf{k}, \hbar\omega + i\Gamma) = G_0(\mathbf{k}, \hbar\omega + i\Gamma) + \frac{1}{\mu(\mathbf{k})} \sum_{\mathbf{k}'} G_0(\mathbf{k}, \hbar\omega + i\Gamma) W(\mathbf{k}, \mathbf{k}') G(\mathbf{k}', \hbar\omega + i\Gamma) \quad (24)$$

where Γ is a broadening factor, $\mu(\mathbf{k})$ is the dipole matrix element characteristic of the oscillator strength of the optical transition between the CB and VB and

$G_0(\mathbf{k}, \hbar\omega + i\Gamma) = \frac{\mu(\mathbf{k})}{\left(E_g + \frac{\hbar^2 k^2}{2\mu} - \hbar\omega - i\Gamma \right)}$ is the Green function for interband free carrier transitions.

For 3D semiconductors the optical susceptibility is computed using the $G(\mathbf{k}, \hbar\omega + i\Gamma)$ solution from: $\varepsilon_0\chi(\hbar\omega) = \frac{2}{V} \sum_{\mathbf{k}} \mu^*(\mathbf{k})G(\mathbf{k}, \hbar\omega + i\Gamma)$ where V is a normalization volume. This method was used for 3D perovskites³²² (Figure 32) and for monolayered perovskites^{76,69,275} including the effect of dielectric confinement in $W(\mathbf{k}, \mathbf{k}')$ (Figure 33).

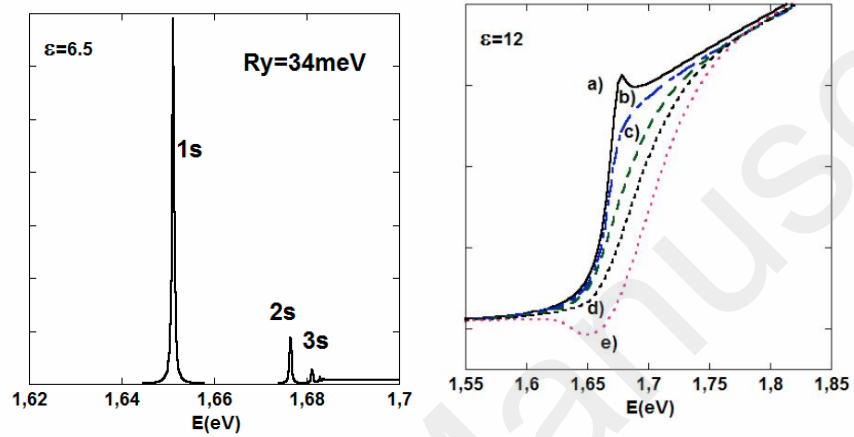


Figure 32. (left) The optical absorption of MAPbI₃ computed using the BSE and the exciton Green's function in the linear regime at low temperature. An empirical basis of electron and hole monoenergetic states obtained with the $\mathbf{k}\cdot\mathbf{p}$ method close to the R point is used to compute the exciton Green's function. The e-h Coulomb interaction is computed with an effective dielectric constant equal to 6.5. 1s-3s exciton resonances are shown (the electronic band gap is set at 1.685eV). (right) Optical absorption of MAPbI₃ computed using the BSE and the exciton Green's function in the non-linear regime at $T=160\text{K}$. The concentration of free carriers is equal to: a) 0, b) 10^{17}cm^{-3} , c) $5\cdot 10^{17}\text{cm}^{-3}$, d) 10^{18}cm^{-3} , e) $2\cdot 10^{18}\text{cm}^{-3}$. Reprinted from ref³²². Copyright 2016 Society of Photo Optical Instrumentation Engineers (SPIE).

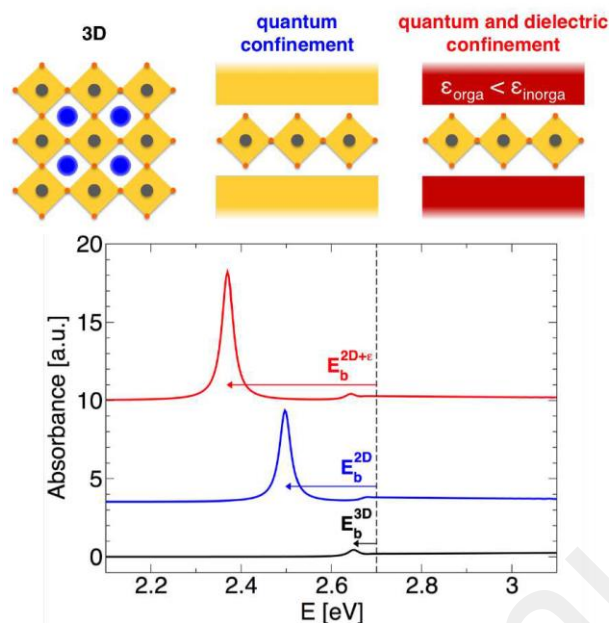


Figure 33. Optical absorption computed using the $\mathbf{k}\cdot\mathbf{p}$ /BSE approach for a model halide perovskite material considered as a 3D material with $\epsilon_\infty = 5.0$ (black line), a 2D material with only quantum confinement, and $\epsilon_\infty = 5.0$ (blue line), a 2D material with both quantum and dielectric confinement (red line). In the latter, the computed dielectric profile of (Decyl- NH_3) $_2\text{PbI}_4$ is used. In all cases, the band gap is set to 2.7 eV, being the value extrapolated at room temperature from available experimental data, and the reduced effective mass is set to 0.093. Reprinted with permission from ref⁷⁶. Copyright 2016 American Chemical Society.

More recently the complete series of Ns states was analyzed with the same method for the (C $_4$ H $_9$ NH $_3$) $_2$ (CH $_3$ NH $_3$) $_{n-1}$ Pb $_n$ I $_{3n+1}$ multilayered perovskites with $n = 1-4$.²⁷⁵ To achieve a quantitative agreement between theory and experiment, it was necessary to refine the method by computing for each n value, the electronic density and dielectric constant profiles from DFT calculations and extract the reduced effective mass from a simulation of the experimental diamagnetic shift (Table 2).

Finally, let's mention that more advanced problems of excitons in 3D can be easily treated using the exciton Green function $G(\mathbf{k}, \hbar\omega + i\Gamma)$ within the framework of the theory of the optical Bloch equations. It is for example possible to consider the screening of the electron-hole interaction by a gas of free carriers in the Thomas-Fermi approximation $W(\mathbf{k}, \mathbf{k}') = -\frac{e^2}{\epsilon_\infty(|\mathbf{k}-\mathbf{k}'|^2 + 1/\lambda^2)}$, where λ is the screening length.³¹³ It corresponds in the position space to the famous Yukawa's potential $V(\mathbf{r}_h, \mathbf{r}_e) = \frac{-e^2}{4\pi\epsilon_\infty|\mathbf{r}_h-\mathbf{r}_e|} e^{-|\mathbf{r}_h-\mathbf{r}_e|/\lambda}$. It is important to notice that either in the simple electrostatic potential or in Yukawa's potential, both potentials depend solely on the relative distance $\rho = |\mathbf{r}_h - \mathbf{r}_e|$. These potentials could alternatively be represented using a single variable $V(\rho)$ that is similar to a 'local' potential for the relative electron-hole motion. Conversely, the local potentials in momentum space are represented using only $\mathbf{q} = \mathbf{k} - \mathbf{k}'$ by $W(\mathbf{k}, \mathbf{k}') = W(\mathbf{q})$. The dielectric confinement effect in layered systems is related to the non-locality of the electron-hole interaction potential, but this effect can be conveniently treated using the framework of the optical Bloch equations (Figures 32,33).

4.4. Image charges and dielectric confinement

The image charge method is a classical method designed to handle the problem of planar interfaces between dielectrics, as well as other simple dielectric heterostructures.^{310,323} In order to compute in position space, the electrostatic potential due to a test charge e_0 located on one side of a planar interface separating two semi-infinite dielectric mediums, it is possible to include two fictitious 'image' charges e'_0 and e''_0 , located respectively at $z = -a$ and $z = a$ (Figure 34). e'_0 allows computing the additional contribution to the homogeneous potential on the positive side ($z > 0$) of the interface, whereas e''_0 allows computing the correct value of the potential on the

negative side ($z < 0$). In the case of an electron-hole pair, the test charge e_0 can be either the electron or the hole and the computed electrostatic potential is multiplied by the other charge to deduce the mutual interaction electrostatic energy.

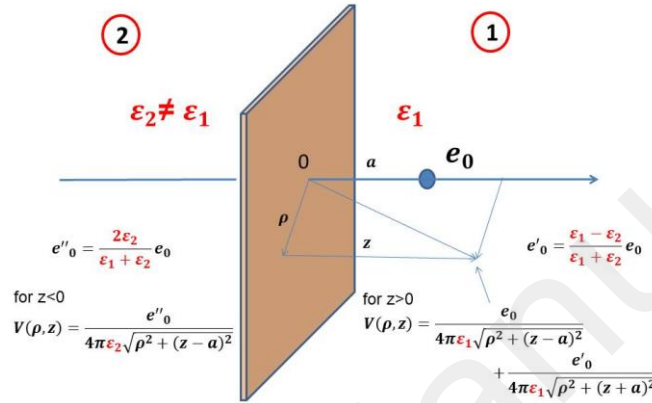


Figure 34. Electrostatic potential due to a test charge e_0 located on one side of a planar interface separating two semi-infinite dielectric mediums. e'_0 and e''_0 are two fictitious ‘image’ charges located respectively at $z = -a$ and $z = a$. e'_0 allows computing the additional contribution to the homogeneous potential on the positive side ($z > 0$) of the interface, whereas e''_0 allows computing the correct value of the potential on the negative side ($z < 0$).

In order to apply the same method for a dielectric well (DW) with abrupt interfaces (Figure 35), it is necessary to consider infinite series of image charges.³²⁴ For example, if the test charge e_0 is located inside a DW of thickness L_w ($-L_w/2 < z_0 < L_w/2$) surrounded by two semi-infinite dielectric mediums, the image charges are located at positions $z_n = nL_w + (-1)^n z_0$, $n = \pm 1, \pm 2, \pm 3 \dots$. The electrostatic potential is non-local because the expressions of the image charges are not equivalent if the potential is computed outside or inside the DW. In the

latter case and for a symmetric DW ($\varepsilon_2 = \varepsilon_3 = \varepsilon_b$ in the barrier and $\varepsilon_1 = \varepsilon_w$ in the well), the image charges located at positions z_n are given by $e_n = \left(\frac{\varepsilon_w - \varepsilon_b}{\varepsilon_w + \varepsilon_b}\right)^{|n|} e_0$. This approach was used initially by Hong and coworkers³¹ to explain the binding energies of the exciton for monolayered $(\text{C}_6\text{H}_5\text{NH}_3)_2\text{PbI}_4$ and $(\text{C}_{10}\text{H}_{21}\text{NH}_3)_2\text{PbI}_4$, and bilayered compounds $(\text{C}_6\text{H}_5\text{NH}_3)_2(\text{CH}_3\text{NH}_3)\text{Pb}_2\text{I}_7$ compounds. Unfortunately, a simple exciton wavefunction trial was used, relying on electron and hole mono-electronic wavefunctions computed for an infinite QW. The number of unknown parameters is important for this purely empirical method: ε_w , ε_b , L_w (Figure 22a). More, layered halide perovskites are rather dielectric superlattices (DSL) and an additional dielectric barrier thickness L_b must be introduced in that case (Figure 22a).^{32,33} This additional parameter may be deduced from the lattice periodicity $L_w + L_b$ along the stacking axis measured by X-ray diffraction. Using further the schematic picture of a composite DSL, a simple formula can be proposed for the average dielectric constant $\overline{\varepsilon_\infty} = \frac{\varepsilon_w L_w + \varepsilon_b L_b}{L_w + L_b}$.⁴⁶ It is not relevant to analyze the enhancement of the exciton binding energy by the dielectric confinement effect, but it shows that dielectric measurements can be used to reduce the number of unknown parameters to two.³¹ $\overline{\varepsilon_\infty} = 4.41$ and $\overline{\varepsilon_\infty} = 3.39$ were obtained experimentally for $(\text{C}_6\text{H}_5\text{NH}_3)_2\text{PbI}_4$ and $(\text{C}_{12}\text{H}_{25}\text{NH}_3)_2\text{PbI}_4$.³¹ Finally, the dielectric constant in the DW was considered to be the one of PbI_2 in some cases ($\varepsilon_w = 6.1$)^{31,46}, or the one of $\text{CH}_3\text{NH}_3\text{PbI}_3$ ($\varepsilon_w = 6.1$) in other cases,³³ but without clear justifications. It was in fact shown recently by using DFT approaches that the dielectric composite picture breaks down for low n values,^{81,206} and that smaller ε_w values than the above mentioned ones must be considered for monolayered compounds. In fact, in $\text{CH}_3\text{NH}_3\text{PbI}_3$ and PbI_2 covalent structures, all the iodine atoms are bonded to two Pb atoms. But, on one hand, in layered halide perovskites, the inorganic layers are disconnected along the stacking axis, with terminal iodides having a different electronic environment than the equatorial

ones. On the other hand, in 3D $\text{CH}_3\text{NH}_3\text{PbI}_3$ as well as in PbI_2 , all iodine atoms have similar chemical and electronic environments, being connected to two Pb atoms. Similar issues exist for the determination of the effective barrier dielectric constant in a DW or DSL model. The choice of the barrier reference material is somehow arbitrary, especially when the equivalent barrier thickness L_b is small as in the multilayered perovskites based on the BA cation, or even smaller for 3AMP, 4AMP and Gua/MA cations.^{83,204}

In most cases, the well thickness L_w was directly related to the lattice periodicity within the inorganic layer ($L_w = n \times 6.36 \text{ \AA}$) and the experimental exciton binding energies were fitted. From empirical resolution of the BSE, predicted reduced effective masses were nevertheless highly dispersed among the various authors, depending on the theoretical description ranging from $\mu = 0.09 \times m_0$,³¹ $\mu = 0.12 \times m_0$,³³ to $\mu = 0.17 \times m_0$.³² More recent approaches based on exciton diamagnetic shift measurements combined with DFT theoretical calculations for the dielectric profiles and electronic density profiles derived from the complex spinor Bloch functions densities,²⁷⁵ give $\mu = 0.221 \times m_0$ for $(\text{C}_4\text{H}_9\text{NH}_3)_2\text{PbI}_4$ (Table 2). The effective mass slightly decreases down to $\mu = 0.186 \times m_0$ when the number of layers in the QW increases up to $n = 5$.

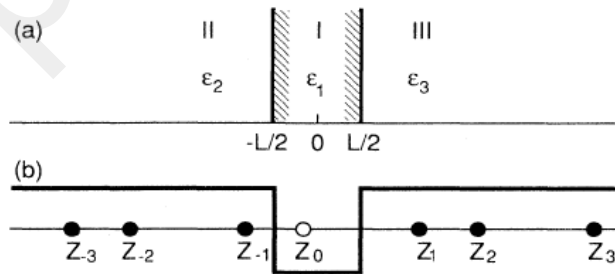


Figure 35. (a) Schematic structure of a general three-layer dielectric well (DW). (b) Typical configurations of the real (Z_0) and image charges are shown for the cases when a charged particle is in the well layer. The original charge is indicated by an open circle and the

induced image charges are given by solid circles. Reprinted with permission from Ref. ³²⁴
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Despite some issues related to unknown physical parameters, the DW model³¹ or DSL model^{32,33} with abrupt interfaces, afford convenient ways to go beyond the Wannier-Mott description and take into account advanced features such as the non-locality of the electron-hole interaction.

However a numerical intrinsic weakness of models based on abrupt dielectric interfaces is the divergence of the dielectric self-energy correction close to the interfaces ($z_0 = \pm L_w/2$ for a DW).^{307,81} This issue is related to the self-energy correction resulting from dielectric inhomogeneities:

$$\delta\Sigma = \lim_{r \rightarrow r_0} \frac{1}{2} e_0 (V_{DW \text{ or } DSL}(r, r_0) - V_{bulk}(r, r_0)) \quad (25)$$

which can be directly evaluated from image charge models.³²⁴ The bulk self-energy $V_{bulk}(r, r_0)$ is expected to be included in the well and barrier bulk single particle energies and electronic band gaps. The numerical divergence of the $\delta\Sigma$ correction at an abrupt dielectric interface leads to serious numerical issues when the mono-electronic wavefunctions extend over both the well and the barrier. A numerical trick consists in shifting the position of the interfaces^{324,32} or assuming that the electronic states are totally confined in the well.^{31,33} Another deficiency of the image charge method is the implicit assumption of isotropy for both materials forming the well and the barrier, which is obviously not fulfilled for the organic part, but also for the inorganic part due to the presence of terminal halides only along the stacking axis. Therefore, the relation for the above mentioned average dielectric constant only holds for the in-plane component $\overline{\varepsilon_{\infty, \parallel}} = \frac{\varepsilon_{w, \parallel} L_w + \varepsilon_{b, \parallel} L_b}{L_w + L_b}$. The average dielectric constant along the stacking axis, which is

also responsible for the enhancement of the exciton binding energy, is instead given

by $1/\overline{\varepsilon_{\infty,\perp}} = \frac{L_w/\varepsilon_{w,\perp} + L_b/\varepsilon_{b,\perp}}{L_w + L_b}$. Notice that $\overline{\varepsilon_{\infty,\perp}} = \overline{\varepsilon_{\infty,\parallel}}$ holds only in the $L_w/L_b \gg 1$ limit, and for a bulk inorganic part corresponding to 3D halide perovskite in the cubic phase.

In order to avoid the problems related to abrupt interfaces, a more general method may consist in using a partial Fourier transform of the inhomogeneous Poisson equation with a test charge e_0 at $(\mathbf{r}_{t0}, \mathbf{z}_0)$:

$$\frac{\partial}{\partial z} \left(\varepsilon_{\perp}(z) \frac{\partial}{\partial z} (V(q_t, z, z_0)) \right) - q_t^2 \varepsilon_{\parallel}(z) V(q_t, z, z_0) = -e_0 \delta(z - z_0) e^{-i\mathbf{q}_t \cdot \mathbf{r}_{t0}} \quad (26)$$

where $V(\mathbf{q}_t, z) = \iint d^2 \boldsymbol{\rho} e^{i\mathbf{q}_t \cdot \boldsymbol{\rho}} V(\boldsymbol{\rho}, z)$. Without loss of generality, it is possible to put the test charge on the reference axis ($e^{i\mathbf{q}_t \cdot \mathbf{r}_{t0}} = 1$). For a homogeneous bulk material ($\varepsilon_{\perp}(z) = \varepsilon_{\perp}$

and $\varepsilon_{\parallel}(z) = \varepsilon_{\parallel}$) the solution reads $V_{bulk}(\mathbf{q}_t, z, z_0) = \frac{e_0}{2\sqrt{\varepsilon_{\perp}\varepsilon_{\parallel}q_t}} e^{-i\mathbf{q}_t \cdot \mathbf{r}_{t0}} e^{-\sqrt{\frac{\varepsilon_{\parallel}}{\varepsilon_{\perp}}} q_t |z - z_0|}$, and

$V_{bulk}(\mathbf{q}_t, z, z_0) = e^{-i\mathbf{q}_t \cdot \mathbf{r}_{t0}} \frac{e_0}{2\varepsilon q_t} e^{-q_t |z - z_0|}$ in the isotropic case ($\varepsilon_{\perp} = \varepsilon_{\parallel} = \varepsilon$). The self-energy

correction can be analytically or numerically computed from:³⁰⁷

$$\delta\Sigma(z) = \lim_{z \rightarrow z_0} \frac{e_0}{4\pi} \int_0^{\infty} (V_{DworDSL}(q_t, z, z_0) - V_{bulk}(q_t, z, z_0)) q_t dq_t \quad (27)$$

considering $\varepsilon_{\perp}(z)$ and $\varepsilon_{\parallel}(z)$ in the expression of the $V_{bulk}(q_t, z, z_0)$ bulk solution.^{81,69,275}

More, recently a new numerical method was proposed to extract from DFT calculations and at low computational cost, $\varepsilon_{\perp}(z)$ profiles without any experimental input. It was applied to various multilayered perovskite phases (Figure 36).^{69,81,83,84,206,275}

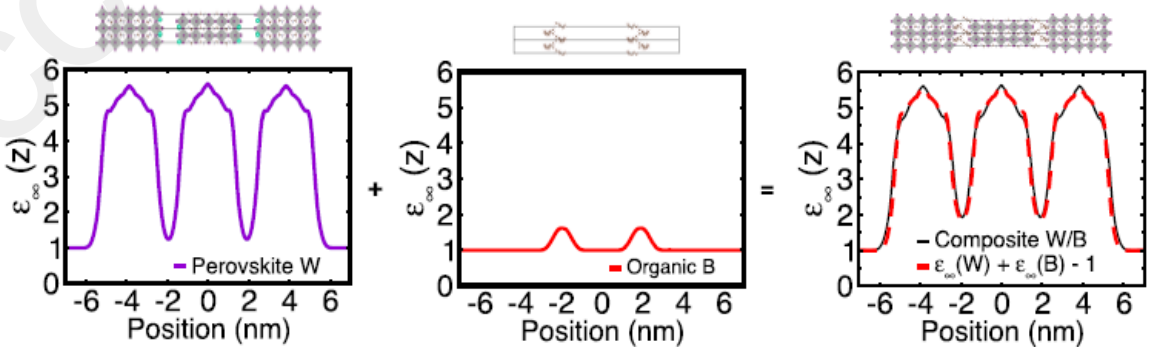


Figure 36. Dielectric profiles computed for each of the sections of a slab of the layered hybrid perovskite $(BA)_2(MA)_4Pb_5I_{16}$ (Figure 23c): (left) for the perovskite well (W) and (middle) for the organic barrier (B). (right) The dielectric profile of the real composite structure W/B computed for $(BA)_2(MA)_4Pb_5I_{16}$ is compared to the sum of the individual components W and B. It reveals the composite nature of the dielectric profiles for the five-octahedron thick layered perovskite. Reprinted with permission from Ref.²⁰⁶ Copyright 2018 American Chemical Society.

This method reveals the composite nature of the dielectric profiles for multilayered perovskites, but also shows that the 3D perovskite $CH_3NH_3PbI_3$ or PbI_2 cannot be considered as the reference material in the DW for small n values. The dielectric profiles do not exhibit abrupt interfaces (Figure 36), and complex situations can be analyzed theoretically such as the inversion of dielectric mismatch when I_2 molecules intercalated in the barrier (Figure 20).⁶⁹ However, the $\epsilon_{\parallel}(z)$ profile is usually not computed and the $\epsilon_{\parallel}(z) = \epsilon_{\perp}(z)$ condition was assumed for the calculations of the exciton state resonance in monolayered compounds⁶⁹ and multilayered ones.²⁷⁵ Noteworthy, the partial Fourier transform method was also used to analyze the exciton binding energy in DSL but including empirical bulk parameters.³²

A precise analysis of the exciton in multilayered halide perovskites also requires an accurate description of the single particle spinor states, as shown in the expression of the electron-hole interaction potential for a DW with abrupt interfaces:

$$\begin{aligned}
 V(q_t, z_e, z_h) = & \\
 & \frac{-e^2}{2\epsilon_w q_t} \iint_{\frac{-L_w}{2}, \frac{-L_w}{2}}^{\frac{L_w}{2}, \frac{L_w}{2}} \rho_e(z_e) \rho_h(z_h) \left[e^{-q_t |z_e - z_h|} + \Delta\chi \left(e^{-q_t |z_e + z_h - d|} + e^{-q_t |z_e + z_h + d|} \right) + \right. \\
 & \left. \Delta\chi^2 \left(e^{-q_t |z_e - z_h - 2d|} + e^{-q_t |z_h - z_e - 2d|} \right) \right] dz_e dz_h \quad (28)
 \end{aligned}$$

where $\rho_e(z_e)$ and $\rho_h(z_h)$ are the electron and hole density profiles, $\Delta = (1 - \chi^2 e^{-2q_t d})^{-1}$, and $\chi = \frac{\varepsilon_w - \varepsilon_b}{\varepsilon_w + \varepsilon_b}$. We may notice that this formula leads to:

$$V(q_t) = \frac{-e^2}{2\varepsilon q_t} \iint \rho_e(z_e) \rho_h(z_h) e^{-q_t |z_e - z_h|} dz_e dz_h \quad (29)$$

without dielectric confinement ($\varepsilon_w = \varepsilon_b = \varepsilon$).

Here the DW expression is strictly valid when the particles are completely confined in the well ($-L_w/2 < z_e < L_w/2$ and $-L_w/2 < z_h < L_w/2$), but refined expressions can be used when the electronic density profiles exhibit leakage in the barrier.^{325,275} Some of the oldest approaches assume simple expressions for the single particle wavefunctions derived from infinite confinement potentials,^{31,33} and still one relies on a more elaborate method including an Hamiltonian with effective masses along the stacking axis and finite confinement potentials corrected by self-energy expressions.³² The larger reduced effective mass found in this last case ($\mu = 0.17 \times m_0$) by comparison with the others^{31,33} is indeed in better agreement with a recent work including DFT results (Table 2).²⁷⁵ However the effective mass method for the carrier motion along the stacking axis is known to suffer from serious issues leading to the prediction of unphysical superlattice effects and dispersion non-parabolicity.⁹⁴ We stress that very large effective masses were considered for the barrier materials and the tunneling between neighboring layers was enforced.³² The most recent study derives the electronic density profiles from DFT calculation without relying on empirical effective masses approaches:²⁷⁵

$$\begin{aligned} \psi_{e,k_t}(\mathbf{r}_{te}, z_e) &= \frac{e^{-ik_t r_{te}}}{\sqrt{A}} u_{e,k_t=0}(\mathbf{r}_{te}, z_e) \\ \psi_{h,k_t}(\mathbf{r}_{th}, z_h) &= \frac{e^{-ik_t r_{th}}}{\sqrt{A}} u_{h,k_t=0}(\mathbf{r}_{th}, z_h), \end{aligned} \quad (30)$$

where $u_{e,k_t=0}(\mathbf{r}_{te}, z_e)$ ($u_{h,k_t=0}(\mathbf{r}_{th}, z_h)$) are the electron (hole) ground-state, complex spinor Bloch functions at the BZ center. The electron and hole probability density profiles along the

stacking axis are derived by in-plane averaging of the Bloch functions (see Figure 16f and 19b for corresponding LDOS):

$$\rho_e(z_e) = \iint |u_{e,k_t=0}|^2 d^2\mathbf{r}_{te} \text{ and } \rho_h(z_h) = \iint |u_{h,k_t=0}|^2 d^2\mathbf{r}_{th}. \quad (31)$$

4.5. Comparison to dielectric confinement effect in Van der Waals heterostructures

Atomically thin layer Van der Waals heterostructures, such as transition metal dichalcogenides, are known to exhibit very large exciton binding energies and self-energy effects as a result of dielectric confinement.³²⁶ It leads to deviations from the Ns series of resonance energies of 2D excitons similar to the ones reported in free standing monolayered iodide perovskites,³⁰⁹ monolayered iodide and bromide perovskite crystals^{62,308} and multilayered iodide perovskite crystals.²⁷⁵

The initial empirical analysis of the 2D dielectric confinement effect in ultrathin dielectric sheets was performed by Keldysh and coworkers,³²⁷ but recent developments for monolayered Van der Waals heterostructures aim at proposing connections between empirical and DFT approaches.³²⁸⁻³³² A theoretical framework mixing empirical and DFT approaches was also proposed recently for halide perovskites^{76,69,275} following the pioneer work based on purely empirical approaches.³¹⁻³³ However there are some differences between the two families of layered materials that deserve to be stressed.

Let's first consider the complete empirical relation proposed above for halide perovskites:²⁷⁵

$$V(q_t, z_e, z_h) = \frac{-e^2}{2\varepsilon_w q_t} \iint_{\frac{-L_w}{2}, \frac{-L_w}{2}}^{\frac{L_w}{2}, \frac{L_w}{2}} \rho_e(z_e) \rho_h(z_h) [e^{-q_t|z_e-z_h|} + \Delta\chi(e^{-q_t|z_e+z_h-d|} + e^{-q_t|z_e+z_h+d|}) + \Delta\chi^2(e^{-q_t|z_e-z_h-2d|} + e^{-q_t|z_h-z_e-2d|})] dz_e dz_h \quad (32)$$

This momentum space formula is similar to the one reported by Keldysh in his seminal paper:³²⁷

$$V(q_t, z_e, z_h) = \frac{-e^2}{2\varepsilon_w q_t} \iint_{\frac{-L_w}{2}, \frac{-L_w}{2}}^{\frac{L_w}{2}, \frac{L_w}{2}} [e^{-q_t|z_e-z_h|} + \Delta\chi(e^{-q_t|z_e+z_h-d|} + e^{-q_t|z_e+z_h+d|}) + \Delta\chi^2(e^{-q_t|z_e-z_h-2d|} + e^{-q_t|z_h-z_e-2d|})] dz_e dz_h, \quad (33)$$

assuming homogeneous electron and hole densities ($\rho_e(z_e) = \rho_h(z_h) = 1/L_w$) within the DW with no electronic leakage in the dielectric barrier. Thus, it is implicitly matching the DW and QW thicknesses. DFT calculations on layered halide perovskites have shown that the shape of the density distributions may deviate from a rectangular distribution, with especially out-of-plane contributions of the iodine atoms for the hole density.^{64,206,275} By comparison, Van der Waals heterostructures exhibit a higher carrier confinement and are closer to the simplifying hypothesis of Keldysh's theory on carrier densities (Figure 19b, bottom).³²⁶ Assuming instead a QW with vanishing thicknesses for the confinement of the band edge states ($\rho_e(z_e) = \delta_{z_e}$ and $\rho_h(z_h) = \delta_{z_h}$) while the DW has a non-zero thickness L_w :

$$V(q_t) = \frac{-e^2}{2\varepsilon_w q_t} \frac{1+\chi e^{-q_t L_w}}{1-\chi e^{-q_t L_w}} \quad (34)$$

which is equivalent to an effective 2D dielectric function of the form:

$$\varepsilon_{2D}(q_t) = \varepsilon_w \frac{1-\chi e^{-q_t L_w}}{1+\chi e^{-q_t L_w}} \quad (35)$$

Considering the approximation valid for 2D Wannier exciton $q_t L_w \ll 1$:

$$\varepsilon_{2D}(q_t) \approx \varepsilon_b + \frac{(\varepsilon_w^2 - \varepsilon_b^2)}{2\varepsilon_w} q_t L_w. \quad (36)$$

For a large dielectric mismatch $\varepsilon_w \gg \varepsilon_b$:²⁷⁵

$$\varepsilon_{2D}(q_t) \approx \varepsilon_b + \frac{\varepsilon_w q_t L_w}{2}. \quad (37)$$

The most popular formula of Keldysh's theory is usually given in position space.³²⁶ It is based on a number of additional approximations neglecting z_e and z_h dependences. Similar approximations for halide perovskites lead to:

$$V(q_t) = \frac{-e^2}{2\varepsilon_w q_t} \frac{1 + \chi e^{-q_t L_w}}{1 - \chi e^{-q_t L_w}}. \quad (38)$$

The second approximation of Keldysh's paper ($q_t L_w \ll 1$) corresponds to a long-range electron-hole interaction (small in-plane wavevectors), thus well suited for 2D Wannier-like excitons and materials such as Van der Waals heterostructures or layered halide perovskites. It leads to an expression in position space after Fourier transform:

$$V(r_t) = \frac{-e^2 \varepsilon_w}{2\pi(\varepsilon_w^2 - \varepsilon_b^2)} \int_0^\infty \frac{1}{t + \frac{2\varepsilon_w \varepsilon_b}{\varepsilon_w^2 - \varepsilon_b^2} \frac{r_t}{L_w}} J_0(t) dt =$$

$$\frac{-e^2 \varepsilon_w}{4(\varepsilon_w^2 - \varepsilon_b^2)} \left[H_0 \left(\frac{2\varepsilon_w \varepsilon_b}{\varepsilon_w^2 - \varepsilon_b^2} \frac{r_t}{L_w} \right) - N_0 \left(\frac{2\varepsilon_w \varepsilon_b}{\varepsilon_w^2 - \varepsilon_b^2} \frac{r_t}{L_w} \right) \right]. \quad (39)$$

Finally, assuming a large dielectric contrast $\varepsilon_w \gg \varepsilon_b$:

$$V(r_t) = \frac{-e^2}{4\varepsilon_w} \left[H_0 \left(\frac{2\varepsilon_b}{\varepsilon_w} \frac{r_t}{L_w} \right) - N_0 \left(\frac{2\varepsilon_b}{\varepsilon_w} \frac{r_t}{L_w} \right) \right], \quad (40)$$

which is the original approximate formula of Keldysh in position space.³²⁷ We may first notice that the approximation $\varepsilon_w \gg \varepsilon_b$ is less justified for monolayered halide perovskite crystals, than for free-standing mono-layered systems (the in-plane dielectric constant of bulk MoS₂ is on the order of 13-15).^{328,329}

Finally let's discuss a more fundamental difference between Van der Waals heterostructures and multilayered halide perovskites. In monolayered sheets related to Van der Waals heterostructures, a microscopic in-plane polarization \vec{P}_{DW} was introduced, instead of considering two different macroscopic dielectric constants. Indeed, a microscopically derived

polarizability makes more sense because the notion of the bulk dielectric constant has no clear meaning in atomically thin layers (Figure 37).³²⁸

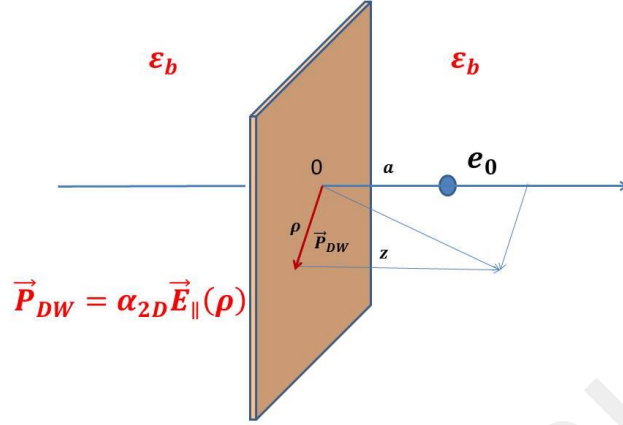


Figure 37. Schematic representation of a test charge e_0 located on a side of an ultrathin DW inducing an in-plane polarization related to a microscopic polarizability α_{2D} . ϵ_b is the barrier dielectric constant.

Starting from the partial Fourier transform of the inhomogeneous Poisson equation with a test charge e_0 at $(\mathbf{r}_{t0}, \mathbf{z}_0)$:

$$\epsilon_b \frac{\partial}{\partial z} \left(\frac{\partial}{\partial z} (V(q_t, z, z_0)) \right) - q_t^2 \epsilon_b V(q_t, z, z_0) = -e_0 \delta(z - z_0) e^{-i\mathbf{q}_t \cdot \mathbf{r}_{t0}} + \alpha_{2D} q_t^2 V(q_t, z, z_0) \delta(z), \quad (41)$$

where the last term is the 2D induced charge at the well position ($z = 0$) related to the 2D polarization $\mathbf{P}_{DW} = \alpha_{2D} \mathbf{E}_{\parallel}(\boldsymbol{\rho})$. The solution for a test charge on the reference axis ($e^{i\mathbf{q}_t \cdot \mathbf{r}_{t0}} = 1$) and in the DW ($z_0 = 0$) is:

$$V(\mathbf{q}_t, z, z_0) = \frac{e_0}{2\epsilon_b q_t \left(1 + \frac{q_t \alpha_{2D}}{2} \right)} \quad (42)$$

The effective 2D dielectric constant is thus

$$\epsilon_{2D}(q_t) \approx \epsilon_b + \frac{\epsilon_b q_t \alpha_{2D}}{2} \quad (43)$$

which can be compared to the above derived purely empirical relation:²⁷⁵

$$\varepsilon_{2D}(q_t) \approx \varepsilon_b + \frac{\varepsilon_w q_t L_w}{2} \text{ when } \varepsilon_w \gg \varepsilon_b \quad (44)$$

The in-plane polarization of the related 3D crystal can be estimated in Van der Waals heterostructures as an average over the number of monolayers,^{328,329} thus affording a convenient way to extract the value of the 2D in-plane polarizability α_{2D} from first-principles calculations of the 3D dielectric matrix. These methods were recently refined in the quasi-2D approach of Latini and coworkers.³³⁰ Actual density profiles were used to compute the screened electron-hole interaction in reciprocal space and the effect of the dielectric mismatch was introduced through a proper 2D transformation of the 3D bulk dielectric, accounting for local field effects. In the DFT approaches on layered perovskites,^{81,69,275} a direct computation of the dielectric profile was performed for each phase, due to the absence of a 3D reference system. In the 3D compound (MAPbI₃) that might be considered as a reference for many multilayered systems, covalent bonds are present along the stacking axis. On the other hand, only terminal iodides are present in the monolayered structures along the same axis. Thus, the MAPbI₃ crystal cannot be considered as a simple assembly of perovskite monolayers. This discrepancy is in fact related to the failure of the $\varepsilon_{\parallel}(z) = \varepsilon_{\perp}(z)$ condition. It shows that further progresses must be performed to better account for the anisotropy of the dielectric properties of the layered perovskite heterostructures.

4.6. Experimental results on the exciton fine structure and higher excitations in layered halide perovskites

As shown in the previous section, the increase of the 1s exciton binding energy and oscillator strength related to the dielectric confinement effect in layered halide perovskites is now a well-established general feature, leading for example to PL at room temperature and

distinct signatures in optical absorption. It was demonstrated for a large number of chemical compositions and extensive investigations on the optical characteristics of the 1s exciton in these systems were performed since the 1980s by many experimental groups. Progresses on the understanding of the quantum and dielectric confinements may benefit from recent results obtained for monolayered compounds such as, the exciton fine structure,^{49,333,334,308} multiexcitonic effects.^{47,54,60,335–337} Additional information may also come from various observations related to strong exciton-photon coupling in optical cavities,^{74,75,338,339} the influence of chirality,³⁴⁰ electrooptic effects,^{333,341} or spin-selective optical Stark effects.³⁴²

We will focus in this part on a brief overview of the experimental works dedicated to the exciton fine structure of monolayered perovskites. In fact, in-depths analyses have been proposed over the last two decades on the exciton fine structures of monolayered bromide and iodide compounds (Table 3).

Table 3: Table summarizing some important spectroscopy results on the exciton fine structure in monolayered halide perovskites.

$(C_jH_{2j+1}NH_3)_2PbBr_4$ (j = 4)	Biexciton binding energy $E_{bXX} = 60meV$	Kato et al, solid state comm 2003 ⁶⁰
$(C_jH_{2j+1}NH_3)_2PbBr_4$ (j = 4)	Exciton fine structure model	Tanaka et al, Jap. J. Appl. Phys. 2005 ³³³
$(C_jH_{2j+1}NH_3)_2PbBr_4$ (j = 4)	Exciton fine structure model	Ema et al, Phys. Rev. B 2006 ³³⁴
$(CH_3C_6H_4CH_2NH_3)_2PbBr_4$	Exciton and biexciton localization	Goto et al, Phys. Rev. B 2006 ³³⁶
$(C_jH_{2j+1}NH_3)_2PbBr_4$ (j = 8)	Temperature study	Kitazawa et al, Thin Solid Films 2006 ³⁴³
$(C_jH_{2j+1}NH_3)_2PbBr_4$ (j = 4, 5, 7, 12)	Influence of the organic barrier	Kitazawa et al, Thin Solid Films 2010 ³⁴⁴
$(C_6H_5C_jH_{2j}NH_3)_2PbBr_4$	Temperature study	Kitazawa et al, J. Phys.

(j = 1-4)		Chem. Solids 2010 ³⁴⁵
$(C_jH_{2j+1}NH_3)_2PbBr_4$ (j = 4)	Biexciton Exciton fine structure model	Yamamoto et al, Phys. Status solidi C 2012 ³⁴⁶
$(C_6H_5C_jH_{2j}NH_3)_2PbBr_4$ (j = 2) $(C_6H_5C_jH_{2j}NH_3)_2PbI_4$ (j = 2)	Comparison between Bromide and iodide compounds	Kitazawa et al, Mater. Chem. Phys. 2012 ³⁴⁷
$(C_jH_{2j+1}NH_3)_2PbBr_4$ (j = 4) $(C_6H_5C_jH_{2j}NH_3)_2PbBr_4$ (j = 2)	Influence of the organic barrier Exciton fine structure model	Takagi et al, Phys. Rev. B 2013 ³⁰⁸
$(C_jH_{2j+1}NH_3)_2PbI_4$ (j = 4, 8, 9, 10, 12)	Exciton fine structure (not analyzed)	Ishihara et al, Phys. Rev. B 1990 ⁴⁶
$(C_jH_{2j+1}NH_3)_2PbI_4$ (j = 10) $(C_6H_5C_jH_{2j}NH_3)_2PbI_4$ (j = 2)	Biexciton binding energy $E_{bXX} = 50meV$	Ishihara et al, Surf. Sci. 1992 ⁴⁷
$(C_jH_{2j+1}NH_3)_2PbI_4$ (j = 6)	Magneto-optical study Exciton fine structure model	Kataoka et al, Phys. Rev. B 1993 ⁴⁸
$(C_jH_{2j+1}NH_3)_2PbI_4$ (j = 6)	Biexciton lasing Biexciton binding energy $E_{bXX} = 40meV$	Kondo et al, Solid State. Comm. 1998 ⁵⁴
$(C_jH_{2j+1}NH_3)_2PbI_4$ (j = 6) $(C_6H_5C_jH_{2j}NH_3)_2PbI_4$ (j = 2)	Influence of the organic barrier	Shimizu et al, Phys. Rev. B 2005 ³³⁵
$(C_6H_5C_jH_{2j}NH_3)_2PbI_4$ (j = 2)	Triexciton	Shimizu et al, Phys. Rev. B 2006 ³³⁷
$(C_6H_5C_jH_{2j}NH_3)_2PbI_4$ (j = 2)	Exciton splitting at low temperature	Gauthron et al, Optics express 2010 ³⁴⁸
$(C_jH_{2j+1}NH_3)_2PbI_4$ (j = 4, 5, 7, 8, 9)	PL dynamics depending on the organic cation	Kitazawa et al, J. Phys. Chem. Solids 2011 ³⁴⁹

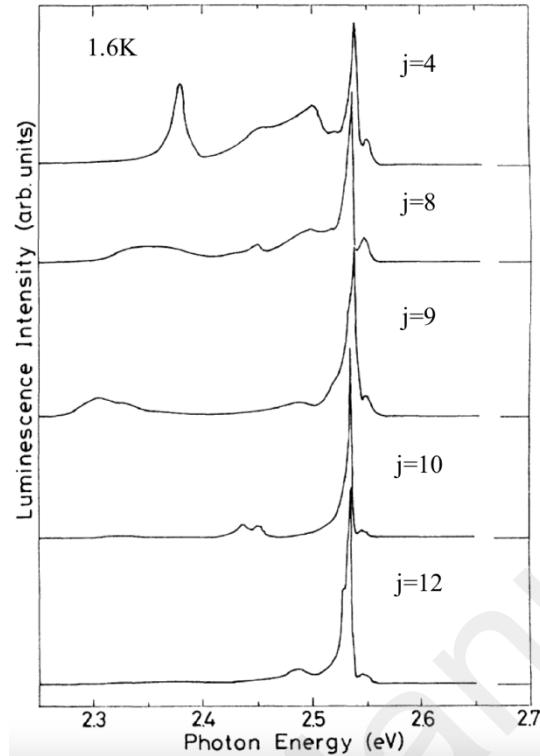


Figure 38. PL spectra in single crystals of $(C_jH_{2j+1}NH_3)_2PbI_4$ with $j = 4, 8, 9, 10,$ and 12 at $1.6 K$. Reprinted with permission from ⁴⁶ Copyright 1990 American Physical Society.

The first paper showing evidence of exciton fine structures at low temperature in monolayered halide perovskites was published by Ishihara and coworkers in 1990 (Figure 38),⁴⁶ although no analysis was provided. The paper was merely focused on the dielectric confinement enhancement of the exciton binding energy. The next important step was the observation of an additional broad contribution at high optical excitation in $(C_{10}H_{21}NH_3)_2PbI_4$.⁴⁷ This new band (M band on Figure 39) appears about 50 meV below the free exciton band (F on Figure 39). It grows as the square of the excitation intensity. As reported by the authors, the ratio of the binding energy of the biexciton to that of the exciton ($50/320 \sim 0.16$) is typical of QW-like structures. A lower biexciton binding energy of 40 meV was reported later on for $(C_6H_{13}NH_3)_2PbI_4$.⁵⁴ The large value of the biexciton binding energy is favorable for the observation of biexciton lasing,

which was observed up to 40K by Kondo and coworkers in 1998 in $(C_6H_{13}NH_3)_2PbI_4$ under optical pumping.⁵⁴ Moreover, as shown on Figure 39, other contributions are present in between the F and M peaks, but could not be resolved in the early study on $(C_{10}H_{21}NH_3)_2PbI_4$.⁴⁷ Due to smaller dielectric constants, bromide compounds exhibit somewhat larger exciton binding energy than iodide compounds. A larger biexciton binding energy was measured as well for $(C_4H_9NH_3)_2PbBr_4$ (Figure 39).⁶⁰ Noteworthy the intermediate contribution in between the EX and M peaks are better resolved and interpreted as exchange splitting states (referred to as Q on Figure 39), whereas the high energy peak is related to a singlet state. We may also notice that the Q peak contains in fact two contributions. The observation of triexciton formation with a dissociation energy of 14 meV was reported $(C_6H_5C_2H_4NH_3)_2PbI_4$ by Shimizu and coworkers.³³⁷ This energy was not directly determined from peak splitting, but estimated from the temperature dependence of the triexciton peak intensity.

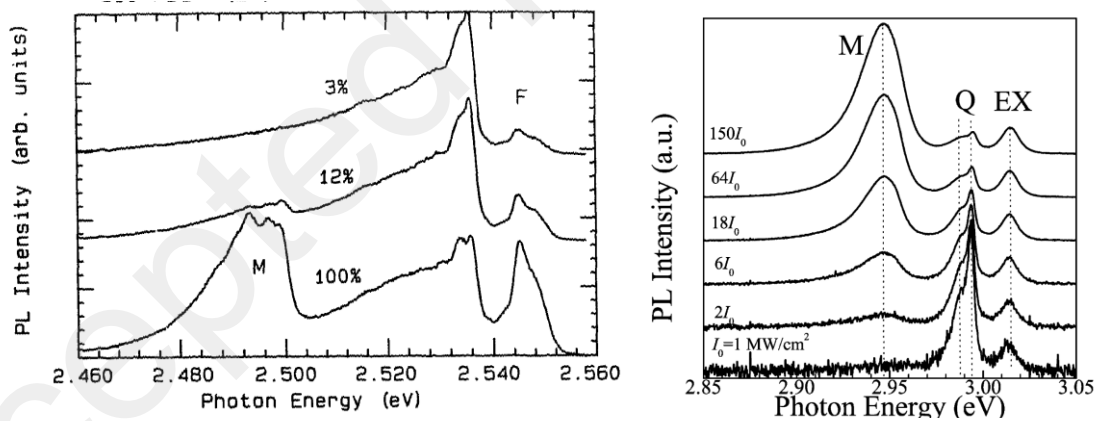


Figure 39. (left) PL spectra of $(C_{10}H_{21}NH_3)_2PbI_4$ at 2 K for different excitation levels. Highest intensity is estimated to be 100 kW/cm^2 . Ref⁴⁷ Copyright 1992, with permission from Elsevier. (right) PL spectra of a $(C_4H_9NH_3)_2PbBr_4$ single crystal at 10 K for various excitation intensities. EX and Q denote the exciton luminescence from the singlet state and the exchange splitting

states, respectively. M denotes the biexciton luminescence. All spectra are normalized with respect to the EX peak intensity. Ref ⁶⁰ Copyright 2003, with permission from Elsevier.

A magneto-optical study was performed on polycrystalline films of $(C_6H_{13}NH_3)_2PbI_4$ by Kataoka and coworkers combining Voigt and Faraday configurations.⁴⁹ A model was proposed for the first time for the exciton fine structure. This model was designed for a Frenkel exciton, but interestingly takes into account the strong SOC of the CB, as well as the double degeneracy of the in-plane p-orbitals of Pb atoms.⁶⁴ The additional CB states related to out of plane Pb p-orbitals which have a higher energy than in-planar Pb p-orbitals, are also taken into account through a crystal field term. This model was extensively used leading to precise peak assignments for bromide compounds.^{333,334,346,308} Figure 40a illustrates such analysis for $(C_4H_9NH_3)_2PbBr_4$ and $(C_6H_5C_2H_4NH_3)_2PbBr_4$ single crystals at $T=10K$. The computation of the exciton fine structure accounts for Coulomb interaction but also an exchange term (Figure 40b). Γ^-_5 is a doubly degenerate excitonic state (in the approximation of a tetragonal unit cell) with an optical activity corresponding to an in-plane electric field polarization (TE). Γ^-_1 is a dark state with no optical activity, and Γ^-_2 is a bright state corresponding to an out of-plane electric field polarization (TM), that may however appear dark in many experimental situations. The peak splitting appears much larger than in 3D halide perovskites or $CsPbBr_3$ QD,³⁵⁰ which is interpreted as a result of an exchange energy greatly enhanced (28 to 32 meV according to Ema et al³³⁴) by strong confinement and by image charge effects. Quantitative simulations of the splitting observed in the exciton fine structure of monolayered perovskites are still lacking.

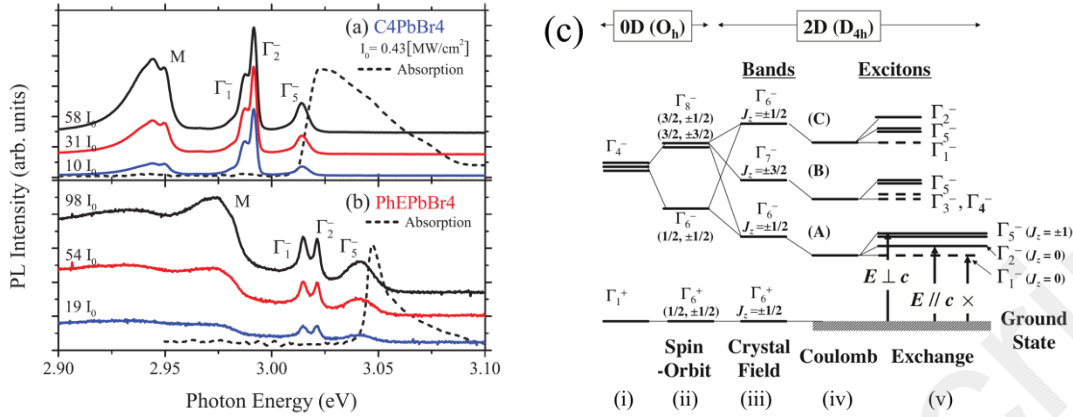


Figure 40. (a) and (b) PL spectra of (a) $(C_4H_9NH_3)_2PbBr_4$ (C_4PbBr_4) and (b) $(C_6H_5C_2H_4NH_3)_2PbBr_4$ ($PhEPbBr_4$) single crystals at 10 K for various excitation intensities. There are four main peaks in both samples. The assignments of these four peaks are described in the text. Dashed lines show absorption spectra obtained by the Kramers-Kronig relation of reflection spectra. Ref. ³⁰⁸ Copyright 2013 American physical society. (c) Schematic energy diagram of two-dimensional crystal $(C_4H_9NH_3)_2PbBr_4$ at Γ point of the BZ. (i) The CBM and VBM states of the $[PbBr_6]$ octahedron have Γ_4^- and Γ_1^+ symmetries, respectively. (ii) With spin-orbit interaction and (iii) tetragonal crystal field, the CBM state splits into the three bands A, B, and C, while the VBM state is not split. The excitons in each band are split by the (iv) Coulomb and (v) exchange interactions. Only the selection rules to the lowest-energy excitons (A band) are described. Ref. ³³³. Copyright 2005 The Japan Society of Applied Physics.

The situation is less clear for iodide compounds.⁴⁹ A possible explanation may rely on dark exciton and biexciton localization in bromide compounds.³³⁶ It is consistent with a comparative study of $(C_6H_5C_2H_4NH_3)_2PbBr_4$ and $(C_6H_5C_2H_4NH_3)_2PbI_4$ using temperature dependent time-resolved PL.³⁴⁷ It shows an apparent long lifetime for the dark exciton line in the bromide compound, not observed in the iodide one. The internal strain of the structure seems to

play a major role for the exciton localization. The size of the halogen atom as well as the internal dynamics of the organic cation may lead to large difference in the local structure as probed by NMR studies.^{290,351} Another study on $(C_jH_{2j+1}NH_3)_2PbI_4$ compounds with $j = 4, 5, 7, 8, 9$ shows that the PL dynamics is strongly affected by seemingly small changes of the organic cation in the barrier.³⁴⁹

More generally, the localization of charge carriers has been little explored from the fundamental viewpoint in multilayered halide perovskites, although it is obviously important to explain various experimental results and may lead to additional applications for white light emission.^{336,202,352,84,89,303,304,353–355}

5. CONFINEMENT EFFECTS IN COLLOIDAL NANOSTRUCTURES

Summary: Recent results obtained for colloidal perovskites nanostructures are reviewed. The analysis of confinement effects deserves additional research efforts to reach a suitable level of description on par with other colloidal systems or multilayered perovskite structures.

5.1. Introduction

Experimental studies on colloidal nanostructures of halide perovskite materials started very recently (2015), but they revealed almost immediately the good potential of these nanostructures for light emission. The development of perovskite colloidal QD solar cells made a significant step one year later, and these nanostructures are currently leading the sector of QD photovoltaics.^{166,21,167,356,168,22} New synthesis methods were proposed allowing virtually all perovskite chemical compositions to be explored with various nanostructure shapes, sizes and electronic dimensionalities ranging from QD, QWr, QR to QNPL.^{21,166–168,309,350,357–396} Inorganic

CsPbX₃ QD (X=I, Br, Cl or alloys) exhibit presently the most important potential for applications both for light emission and photovoltaics. Now, hybrid halide perovskites colloidal nanostructures deserve further investigations and present room for chemical engineering.

Most experimental studies on halide perovskite colloidal structures show that carrier confinement significantly improves the optoelectronic properties by comparison to the related bulk perovskite compounds, leading for example to enhanced and stable narrow-band emission, with suppressed blinking and small spectral diffusion at low temperature.³⁶⁶ However, the literature on quantum and dielectric confinement effects in halide perovskite nanostructures is still scarce. We present in the next section a short review of this topic that will certainly undergo a quick development in the near future.

5.2. Quantum and dielectric confinements

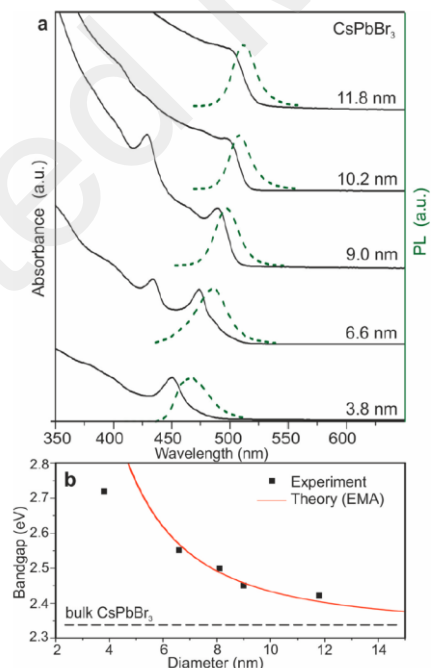


Figure 41. Quantum-size effects in the absorption and emission spectra of 5–12 nm CsPbBr₃ QD. (b) Experimental versus theoretical (effective mass approximation, EMA) size

dependence of the band gap energy. Reprinted with permission from ref ¹⁶⁶. Copyright 2015 American Chemical Society.

The experimental studies dedicated to these effects essentially report on the optical band gap variations as a function of the nanostructure size. Such variations were clearly observed in the first paper on CsPbX₃ QD (Figure 41).¹⁶⁶ An effective mass approximation was used to analyze the quantum confinement effect. The dielectric confinement effect was not considered (vide infra).

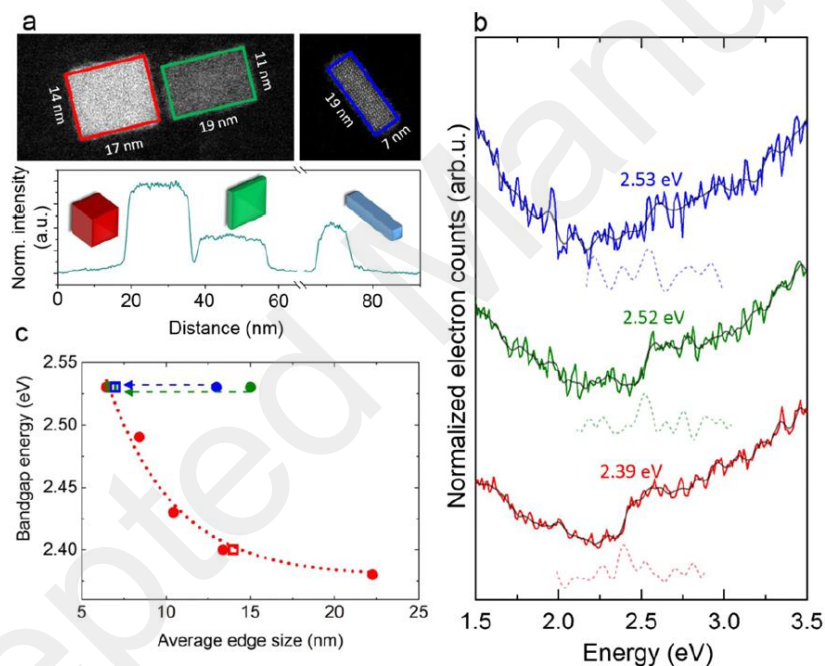


Figure 42. Shape-dependent energy band gap of isolated single nanostructures. (a) Z-contrast annular dark-field (ADF) images of three QD that have different shapes. Two edge lengths can be derived from the ADF image, while the thickness is estimated based on the normalized ADF image intensity as shown in the panel underneath. The thickness of each nanostructure is estimated qualitatively to be 13–14 nm (cube, red), 6–7 nm (QNPL, green), and 8–9 nm (QR, blue). (b) The low-loss EEL spectra of the corresponding nanostructure with

their band gap energies given. The semitransparent black solid lines are the smoothed spectra and the dashed lines their first derivatives. (c) A schematic comparison of the “average” edge size, as solely determined from the ADF image, vs band gap energy for these three shapes (solid dots) showing a clear discrepancy for the QNPL and QR. The average edge size of a (non cubic) nanostructure is determined as the square root of the surface. Taking now the thickness into account, the actual shortest edge of the nanostructure can be considered where subsequently the band gap values of the plate and rod-shaped nanostructures agree well with earlier observations (open squares). Reprinted with permission from ref³⁶⁵. Copyright 2016 American Chemical Society.

Other authors report on variations of the optical band gaps when the chemical synthesis is used to tune the sizes and shapes of perovskite nanostructures. However, quantum and dielectric confinement effects are not yet analyzed using geometrical and material parameters (Figures 42 and 43).^{365,380}

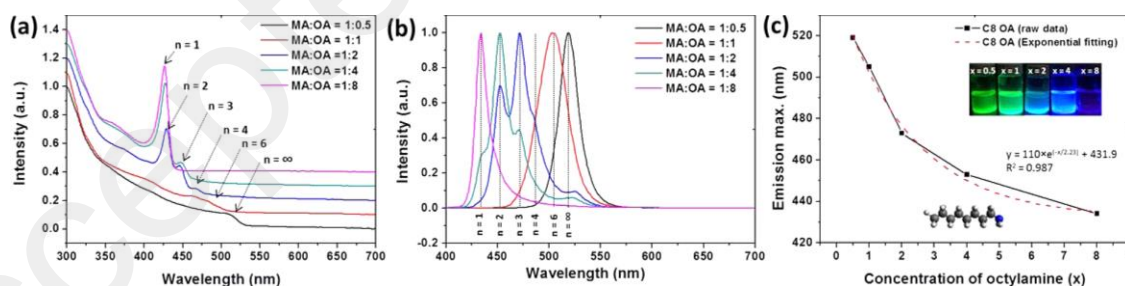


Figure 43. (a) UV–visible absorption spectra and (b) PLE spectra plotted for varying MA/OA ratios (OA is alkylamine) when octyl-alkylammonium ligand (C8 OA) is used as the capping ligand. The PL spectra have been normalized to a unitary intensity to illustrate variations of the peak maxima. (c) Plot of the peak position of the primary PLE band as a

function of the relative OA concentration. An exponential fit to the data yields a R^2 value of 0.987. The inset to panel c indicates a digital photograph of 2D QNPL capped with OA prepared at different concentrations of the capping ligand upon excitation with a 365 nm UV light source. Reprinted with permission from ref³⁸⁰. Copyright 2016 American Chemical Society.

In QNPL, the confinement effect is somehow simpler to analyze. The confinement effect due to the lateral dimensions can be neglected provided that the lateral size exceeds about 15nm (Figure 44).³⁸⁵ Confinement effects have been observed in colloidal QNPL made of various bulk compounds including CsPbX₃ and MAPbX₃, but also CsSnI₃ (Figure 45)³⁸⁶, or layered perovskites (C₄H₉NH₃)₂PbBr₄ (Figure 46)³⁹⁷ and (C₄H₉NH₃)₂PbI₄.³⁰⁹

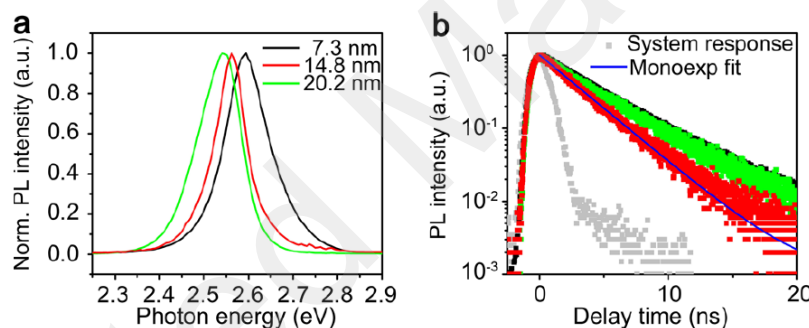


Figure 44. PL properties of the CsPbBr₃ QNPL. (a) PL spectra of the 7.3, 14.8, and 20.2 nm QNPLs. (b) PL decay traces of the 7.3, 14.8, and 20.2 nm QNPL. The system response is also plotted as the gray line. Reprinted with permission from ref³⁸⁵. Copyright 2017 American Chemical Society.

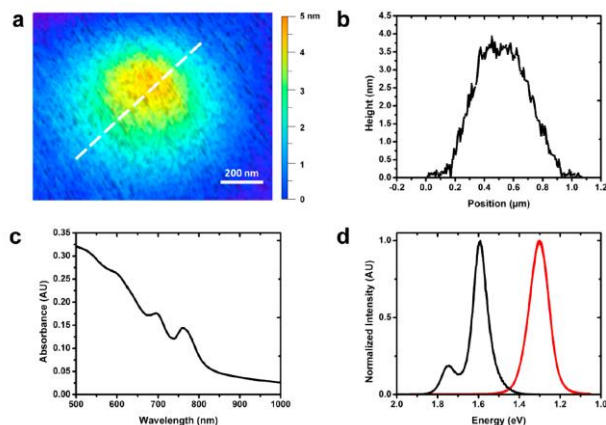


Figure 45. Confinement in CsSnI_3 studied by thickness measurements and optical characterization. (a) Atomic force microscopy (AFM) image of a single CsSnI_3 QNPL as measured in non-contact mode in an air-free cell. (b) AFM height profile across a single, isolated CsSnI_3 QNPL as shown in panel a by the dashed line. (c) Absorbance spectrum of CsSnI_3 QNPL. (d) PL spectra of quantum-confined CsSnI_3 QNPL (black) and bulk CsSnI_3 (red). Reprinted with permission from ref³⁸⁶. Copyright 2018 American Chemical Society.

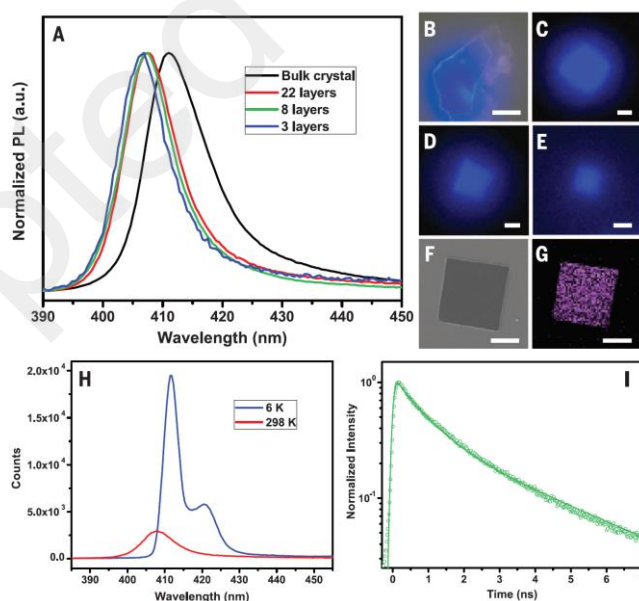


Figure 46. PL properties of the 2D $(\text{C}_4\text{H}_9\text{NH}_3)_2\text{PbBr}_4$ sheets. (A) Steady-state PL spectrum of a piece of bulk crystal and several 2D sheets. (B) The corresponding optical image of the bulk crystal under excitation. Scale bar, 20 mm. (C to E) Optical images of the 2D sheets with 22 layers, 8 layers, and 3 layers. Scale bars, 2 mm. (F) SEM image of a 2D sheet. Scale bar, 2 mm. (G) The corresponding cathodoluminescence image showing the emission (with a 40-nm bandpass filter centered at 415 nm). (H) PL spectra of a 2D sheet at 298 and 6 K. (I) Time-resolved PL measurements showing a bi-exponential decay. From ref³⁹⁷. Reprinted with permission from AAAS.

A purely DFT-based theoretical approach of the optoelectronic properties of halide perovskite nanostructures is not yet possible due to the very large number of atoms and electrons to be considered. It can be performed for very small nanostructures,³⁷⁶ but at a level of theory (without including many-body effects) where the description of bulk materials still deserves improvements. More importantly, the actual crystallographic structure inside a nanostructure may be different from the one of the bulk compound at the same temperature. The cubic phase is the crystallographic phase at high temperatures for all CsPbX_3 compounds, but they exhibit various structural phase transitions leading to tetragonal phases or orthorhombic perovskite as well as non perovskite phases. The case of CsPbBr_3 QD is the one with the most precise description and illustrates some of the possible issues. Some of the early works^{166,358} indeed report on a cubic phase at room temperature in contradiction with the known bulk orthorhombic phase from room down to very low temperatures³⁹⁵ and with other experimental results on QD.³⁶² Refined analysis of 12.5nm QD crystallographic structures at room temperature by X-ray diffraction indicates that the crystal structure is unequivocally an orthorhombic perovskite phase.³⁶⁴ However, the analysis of the exciton fine structure of a large number of CsPbBr_3 QD at

low temperature (Figure 47), shows that they may exhibit a tetragonal or orthorhombic phase.^{350,398} A further reduction of symmetry leading to a Rashba effect is also possible.³⁷⁹ More, the existence of coherent nanotwins was demonstrated in CsPbX₃ QD.³⁷⁴ These nanotwins together with dynamic disorder may results in an apparent higher-symmetry structure.

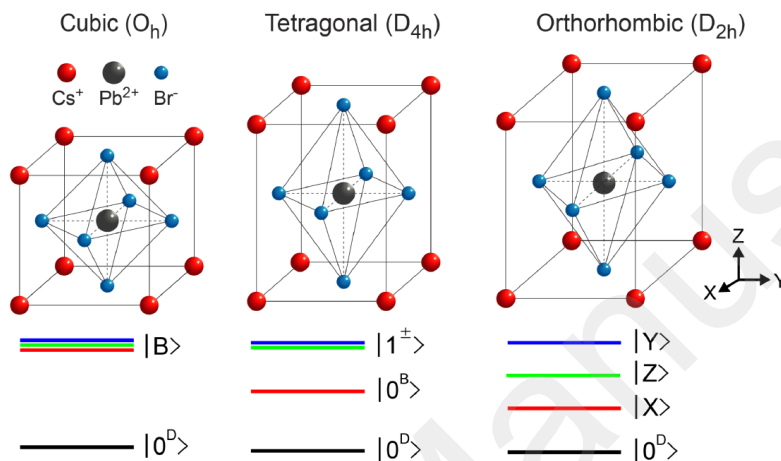


Figure 47. Diagrams of energy levels of the band-edge exciton fine structure of CsPbBr₃ perovskites for three different crystal structures. The lowest state is a dark state while the others are optically active. For the cubic phase, B represents a triply degenerate bright energy-level. Reprinted with permission from ref³⁵⁰. Copyright 2017 American Chemical Society.

A DFT-based theoretical analysis of quantum confinement effects is simpler in QNPL assuming lattice periodicity in the two directions perpendicular to the smallest dimension (Figure 48)?^{81,382} Sizeable increase of the electronic band gap by comparison to bulk materials is predicted for thicknesses of QNPL smaller than about 8-10 monolayers (about 6nm depending on the material). The variation of the band gap with the number of layers n , $E_g(n)$, can be expressed as: $E_g(n) = E_{g,bulk} + \delta E_g(n)$ where $E_{g,bulk}$ is the band gap of the bulk core material and $\delta E_g(n)$ stems from confinement effects. Although DFT-based many-body corrections are not

possible for such nanostructures, it is possible to propose an approximated decomposition of the various contributions: $E_g(n) = E_{g,DFT,bulk} + \delta E_{g,DFT}(n) + \Sigma_{bulk} + \delta \Sigma(n)$ where $E_{g,DFT,bulk}$ and Σ_{bulk} are respectively the bulk band gap evaluated at the DFT level and the self-energy corrections due to many-body effects. The band gap can also be extracted from experimental data.

The effect of quantum confinement can be estimated from $\delta E_{g,DFT}(n)$, provided that the effective masses of the bulk materials are not strongly affected by the underestimation of the band gap at the DFT level. SOC is in principle necessary for a precise determination of the electronic properties of halide perovskite (Table 4 and section 3.1.3).^{64,91} But a crude approximation of the electronic band gap variation may be obtained in many cases by neglecting the SOC, due to an error cancellation between SOC and many-body corrections (vide supra). However, one must keep in mind that the degeneracy of the CB is not correctly predicted.

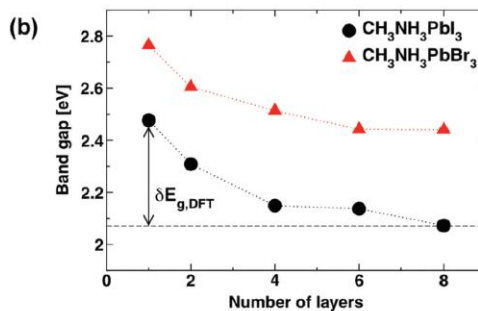
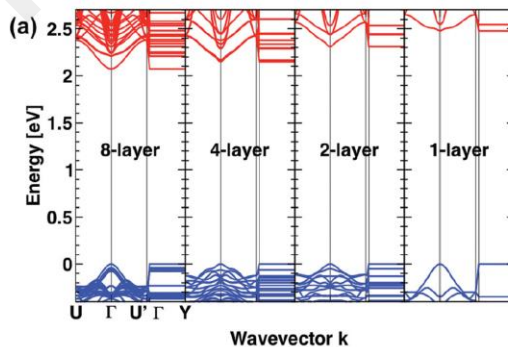
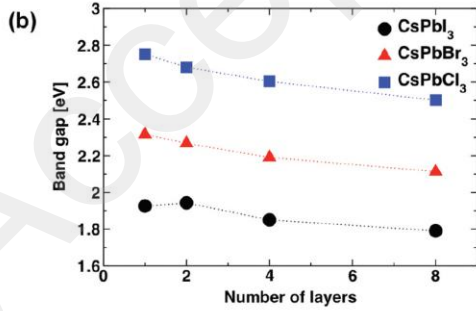
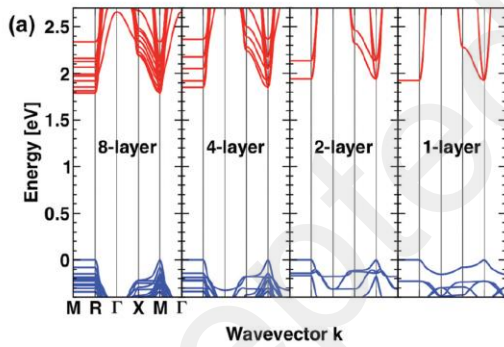


Figure 48. (left) (a) Band structures for slabs of CsPbI₃. (b) Band gaps with respect to the slab thickness of CsPbX₃ (X = I, Br, Cl). (right) (a) Band structures for slabs of CH₃NH₃PbI₃ with various thicknesses. (b) Band gaps with respect to the slab thickness of CH₃NH₃PbX₃ (X = I, Br). From ref⁸¹ – Published by the Royal Society of Chemistry.

Table 4: SOC-DFT calculated band gap $E_{g,DFT}$, electron/hole effective masses and reduced masses. Reprinted with permission from ref³⁸². Copyright 2016 American Chemical Society.

n	E_g^a (eV)	m_h/m_0	m_e/m_0	μ/m_0
1	2.84	0.246	0.242	0.122
3	2.67	0.224	0.226	0.112
5	2.59	0.211	0.215	0.107
11	2.51	0.198	0.202	0.100
15	2.47	0.191	0.195	0.096
∞	2.36	0.172	0.171	0.086

^aThe calculated band gap values were rescaled by 2.09 eV to match the measured PL band gap for the bulk.

The variation of the self-energy correction $\delta\Sigma(n)$ cannot be evaluated directly from GW computations. An alternative semi-empirical evaluation of this correction, similar to the one used for layered perovskites, is possible. A detailed description of the dielectric profile is needed for that purpose. DFT calculations on the in-plane and out-plane dielectric constants was performed considering the stacks of CH₃NH₃PbBr₃ perovskites with capping ligand octylammonium (CH₃(CH₂)₇NH₃⁺) layers as a single object (Figure 49.)³⁸⁴ It shows that a small dielectric anisotropy is present for thicknesses larger than 10 monolayers, whereas quantum confinement effects are expected to vanish beyond this limit. More importantly the average dielectric constant is smaller than the bulk one, thus leading to the prediction of an enhancement of the exciton binding energy.

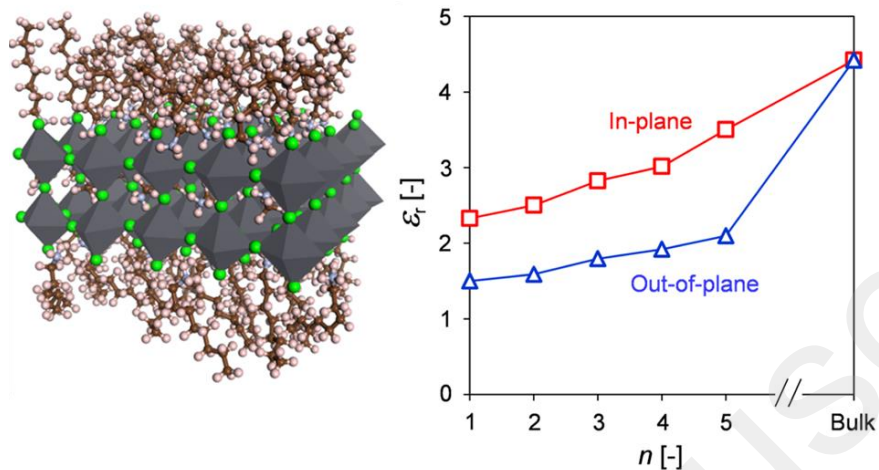


Figure 49. (left) Computer-generated molecular model (carbon, brown; hydrogen, pink; bromine, green; nitrogen, sky blue) of a 2D perovskite with $n = 2$ from $\text{CH}_3\text{NH}_3\text{PbBr}_3$. (right) DFT-calculated out-of-plane and in-plane relative permittivity ϵ_r as a function of n . Reprinted with permission from ref³⁸⁴. Copyright 2016 American Chemical Society.

A DFT computation of the dielectric profiles in QNPL was proposed for the first time by Sapori and coworkers⁸¹ and applied to many different halide perovskites (Figure 50.)²⁰⁶ This approach is able to predict a dielectric confinement in various situations including QNPL made of bulk 3D halide perovskites such as CsPbX_3 or MAPbX_3 compounds. It shows that the dielectric constant at the center of a QNPL is significantly smaller than the bulk dielectric constant for thickness smaller than about 8 monolayers (5nm). The binding energy of the 2D exciton is thus expected to be enhanced in thin QNPL both by a lower perovskite dielectric constant and by the dielectric mismatch with the external layer.

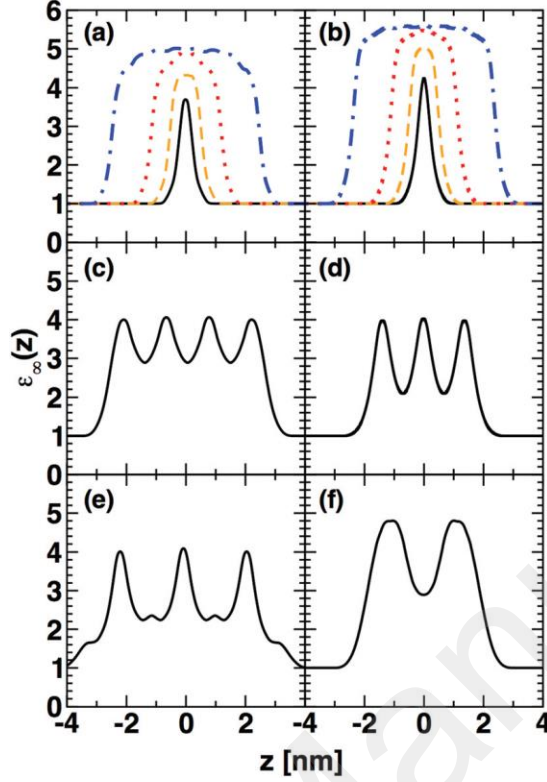


Figure 50. High frequency dielectric profiles $\epsilon_{\infty}(z)$ for slabs of (a) CsPbI_3 , (b) $\text{CH}_3\text{NH}_3\text{PbI}_3$ (in the $[010]$ direction). Black, orange, red and blue lines correspond to 1, 2, 4 and 8-layer slabs, respectively. $\epsilon_{\infty}(z)$ for (c) the 2D halide perovskite with an aromatic cation $(\text{C}_7\text{H}_{10}\text{N})_2\text{PbI}_4$, the 2D halide perovskite with aliphatic cations (d) $(\text{C}_4\text{H}_{12}\text{N})_2\text{PbI}_4$, and (e) $(\text{C}_{10}\text{H}_{24}\text{N})_2\text{PbI}_4$. (f) Same for the bilayered ($n = 2$) halide perovskite $(\text{C}_{10}\text{H}_{24}\text{N})_2\text{PbI}_3\text{CH}_3\text{NH}_3\text{PbI}_4$. From ref⁸¹ – Published by The Royal Society of Chemistry.

In order to evaluate the impact of dielectric self-energy correction on single particle states, the self-energy profiles may be deduced after a partial Fourier transform of the Poisson equation as previously described for layered perovskites:

$$\delta\Sigma(z) = \lim_{z \rightarrow z_0} \frac{e_0}{4\pi} \int_0^{\infty} \left(V_{QNPL}(q_t, z, z_0) - V_{bulk}(q_t, z, z_0) \right) q_t dq_t \quad (45)$$

In order to properly evaluate the $\delta\Sigma(n)$ correction, the self-energy profile $\delta\Sigma(z)$ shall be a weighted average over the electron or hole densities, but this has not yet been performed. A rough estimate of $\delta\Sigma(n)$ can be obtained by considering the profile value at the middle of the QNPL $\delta\Sigma(z = 0)$ (Figure 51). This approximation is strictly valid for ultrathin QNPL and considering single particle states highly confined inside the perovskite sheet ($\rho_e(z_e) = \delta_{z_e}$ and $\rho_h(z_h) = \delta_{z_h}$) It yields thus an upper limit for the self-energy correction in monolayered QNPL on the order of 0.3 eV for a single particle state and of 0.6 eV for the band gap. In principle this value should be added to the $\delta E_{g,DFT}(n)$ correction which is on the order of about 0.4 eV. The exciton binding energy which depends on the QNPL thickness as well ($E_{b,x}(n)$) must be subtracted from the electronic band gap formula:

$$E_{g,opt}(n) = E_{g,DFT,bulk} + \delta E_{g,DFT}(n) + \Sigma_{bulk} + \delta\Sigma(n) - E_{b,x}(n) \quad (46)$$

Experimental or theoretical studies on the enhancement of the exciton binding energies in QNPL $E_{b,x}(n)$ are scarce, by comparison to layered perovskites. Li and coworkers³⁸⁵ report a binding energy of about 120 meV for CsPbBr₃ QNPL with thicknesses on the order of 5-7 monolayers and larger lateral sizes, that is on the same order as the one predicted for layered perovskites with $n = 5-7$.²⁷⁵ More systematic studies will be certainly fruitful. Yaffe and coworkers³⁰⁹ indeed show that the binding energy of a free-standing monolayered perovskite sheet may be as high as 490 meV, which is slightly larger than the binding energy of the same compounds in bulk monolayered perovskites (467 meV)²⁷⁵.

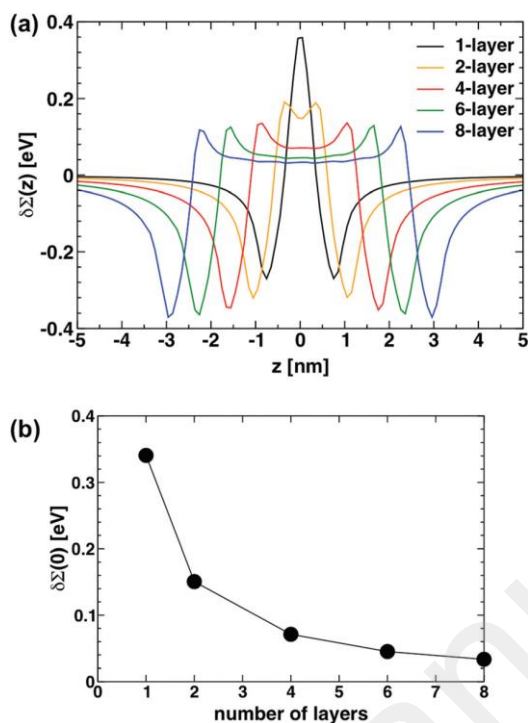


Figure 51. (a) Self-energy profile $\delta\Sigma(z)$ for slabs of $\text{CH}_3\text{NH}_3\text{PbI}_3$. (b) Self-energy taken at the slab centre $\delta\Sigma(0)$. From ref⁸¹ – Published by The Royal Society of Chemistry.

6. CONCLUSION

Despite decades of research on layered halide perovskites and the recent surge of interest on colloidal perovskite nanostructures, understanding on the seemingly well-defined quantum and dielectric confinement concepts in the context of halide perovskites is still scarce. Noteworthy, some basic optoelectronic properties of 3D bulk halide perovskites are still shrouded in mystery.^{399–404} Therefore, multilayered halide perovskites and colloidal perovskite nanostructures represent nano-objects of great complexity. Theoretical descriptions may hardly rely only on brute force atomistic methods based on DFT. Precise full DFT-based calculations based on the BSE equation will require significant progresses on the numerical side in order to address the

excitations in such big systems. From the review on the dielectric confinement effect in layered perovskites (section 4), we suggest some perspectives on the theoretical approaches of the dielectric properties of multilayered halide perovskites. The progresses on the understanding of these effects will probably rely first on the Poisson equation and more systematic investigations of the dielectric profiles using DFT including anisotropy effects ($\epsilon_{\parallel}(z) \neq \epsilon_{\perp}(z)$). DFT shall also be useful to obtain a proper description of single particle spinors. Mixing empirical approaches with modern DFT methods, as well as drawing analogies with classical semiconductor heterostructures or van der Waals heterostructures, is a more practical way to achieve quantitative and predictive descriptions. The success of nowadays studies will inevitably rely on precise structural characterizations and modern spectroscopy tools. Recent investigations of various families of natural multilayered halide perovskite (innate heterostructures) and exfoliated crystals allow making systematic variations of relevant important physical parameters, such as the QW thickness or the dielectric mismatch between the QW and the barrier. Furthermore, whereas most of the existing layered perovskites consist in type I inorganic quantum wells, engineering of the spacer cations that stabilize the perovskite slabs leaves room for significant diversification and improvement, raising issues of dimensionality (e.g. structural versus electronic) and/or energy or charge transfer phenomena. Similar questions are also to be anticipated in the realm of perovskite nanostructures, for instance with the ongoing development of core-shell nano-objects and the ever-growing list of materials and chemical moieties that are combined to perovskites to enrich their properties. We believe that this exciting class of materials provides an impressive playground both for new discoveries in fundamental science but also in the prospect to establish new stable and environmental friendly devices.

ACKNOWLEDGEMENTS

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ABBREVIATIONS

0D,1D, 2D, 3D	zero-, one-, two-, three-dimensional
3AMP	3-(methylammonium)piperidinium
4AMP	4-(methylammonium)piperidinium
5FPEA	2,3,4,5,6-pentafluorophenethylammonium
ac-BA	ammonium 4-butyric acid
ADF	annular dark field
AESBT	5-ammoniummethylsulfanyl-2,2'-bithiophene
AFM	atomic force microscopy
API	ammonium-propyl-imidazole
BA	butylammonium
BSE	Bethe–Salpeter equation
BZ	Brillouin zone
CB	conduction band
CBM	conduction band minimum
CIGS	solid solution of copper, indium, gallium and selenide used for thin-film solar cell technology
DFT	density functional theory

DOS	density of states
DPDA	N,N-dimethylphenylene-p-diammonium
DSL	dielectric superlattice
DW	dielectric well
EA	ethylammonium
EDBE	2,2'-(ethylenedioxy)bis(ethylammonium)
EEL	electron energy loss spectroscopy
ETH	extended Hückel tight-binding model
FA	formamidinium
GGA	generalized gradient approximation to the exchange–correlation energy functional in density functional theory including the density and its first derivative
GKS	generalized Kohn – Sham
GUA	guanidinium
GW	approximation of the self-energy of a many-body system of electrons as a product between the single particle Green's function G and the screened Coulomb interaction W
G_0W_0	first order perturbation of the GW approximation
H2D2c	N-methylethane-1,2-diammonium
HEA	ethanolammonium
HOMO	highest occupied molecular orbital

HSE	exchange–correlation energy functional incorporating a portion of exact exchange from Hartree–Fock theory proposed by Heyd, Scuseria and Ernzerhof in 2003
IC6	iodohexylammonium
IPES	inverse photoemission spectroscopy
KS	Kohn – Sham
LCAO	linear combination of atomic orbitals
LDA	local density approximation to the exchange–correlation energy functional in density functional theory
LDOS	local density of states
LUMO	lowest unoccupied molecular orbital
MA	methylammonium
NAMD	non-adiabatic molecular dynamics
NEA	2-naphtylenethylammonium
OA	octyl-alkylammonium
OD	optical density
PEA	phenethylammonium
PBE	generalized gradient approximation proposed by Perdew, Burke and Ernzerhof in 1996
PL	photoluminescence
PLE	photoluminescence excitation
QD	quantum dot
QNPL	quantum nanoplatelet

QR	quantum rod
QW	quantum well
QWr	quantum wire
RA	alkylamine
Sc-GW	self-consistent GW approximation
SEM	scanning electron microscopy
SOC	spin-orbit coupling
TB	tight binding
TD-DFT	time-dependent density functional theory
ThMA	2-methylammonium thiophene
UPS	ultraviolet photoelectron spectroscopy
UV	ultraviolet
VB	valence band
VBM	valence band maximum

SHORT BIOS

Claudine Katan

Dr. Claudine Katan, born Hoerner, is a CNRS Research Investigator since 1993 in France. She has a broad experience in interdisciplinary collaborative research thanks to her subsequent position at the Physics and Chemistry Departments of Rennes interspersed by a two year stay at FOTON Institute as well as fruitful international collaborations. Her theoretical research is usually conducted in close collaboration with experimentalists and covered structural and electronic properties of molecular charge-transfer crystals, linear/non-linear optical properties of molecular chromophores for various applications, including two-photon uncaging. Currently, understanding the physical and photo-physical properties of hybrid and all-inorganic metal-halide perovskites is her major theme. Those materials have recently demonstrated great potential for various sustainable technologies, including light emitting and photovoltaic devices, raising in turn many fundamental issues in Physics and Materials Science.

Nicolas Mercier

Nicolas Mercier is full Professor at the University of Angers (France). He obtained his PhD in 1994 from the University of Le Mans (France), under the supervision of Pr. M. Leblanc in Pr. G. Ferey's group. In 1996, he was an assistant professor in the chemistry department of Angers University. His major interests include the synthesis, crystallography, structure-property relationships of organic-inorganic hybrid materials. He started working in the field of hybrid perovskites (HPs) in 2002, showing the key role of organic cations to tune the band gap of 2D HPs, the potential of such hybrids in the field of SHG switchable materials and in the field of ferroelectrics. Recently, he has discovered the new family of lead and iodide deficient hybrid perovskites (d-HPs). His activity is also dedicated to viologen based photochromic materials, porous coordination polymers and mechanochromic luminescent hybrid materials.

Jacky Even

Jacky Even is full Professor at INSA Rennes engineering school since 1999 and senior member of Institut Universitaire de France (IUF). He received a PhD in Physics from the University of Paris VI in 1992. He was assistant professor at the Physics Department of Rennes University (1992–1999), where he gained broad experience in phase transitions and solid-state chemical reactions in molecular materials, using theoretical and experimental approaches: neutron and X-ray scattering, Raman/FTIR spectroscopy, calorimetry, among others. In 1999, he created FOTON's simulation team, to address fundamental questions on semiconductors at the atomistic level as well as to perform optoelectronic device simulation. Besides halide perovskites, his theoretical activity is dedicated to semiconductor buried nanostructures, colloidal nanostructures, photovoltaic and light-emitting (LED, laser, ...) devices for silicon photonics, and optical telecommunications.

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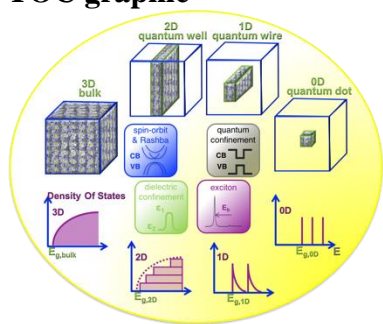
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