

# The interplay between total mercury, methylmercury and dissolved organic matter in fluvial systems: A latitudinal study across Europe

Andrea Garcia Bravo, Dolly Kothawala, Katrin Attermeyer, Emmanuel Tessier, Pascal Bodmer, José L.J. Ledesma, Joachim Audet, Joan Pere Casas-Ruiz, Nuria Catalan, Sophie Cauvy-Fraunié, et al.

#### ▶ To cite this version:

Andrea Garcia Bravo, Dolly Kothawala, Katrin Attermeyer, Emmanuel Tessier, Pascal Bodmer, et al.. The interplay between total mercury, methylmercury and dissolved organic matter in fluvial systems: A latitudinal study across Europe. Water Research, 2018, 144, pp.172-182. 10.1016/j.watres.2018.06.064. hal-01982331

#### HAL Id: hal-01982331

https://hal.science/hal-01982331

Submitted on 3 Nov 2020

**HAL** is a multi-disciplinary open access archive for the deposit and dissemination of scientific research documents, whether they are published or not. The documents may come from teaching and research institutions in France or abroad, or from public or private research centers.

L'archive ouverte pluridisciplinaire **HAL**, est destinée au dépôt et à la diffusion de documents scientifiques de niveau recherche, publiés ou non, émanant des établissements d'enseignement et de recherche français ou étrangers, des laboratoires publics ou privés.



Bravo, AG and Kothawala, DN and Attermeyer, K and Tessier, E and Bodmer, P and Ledesma, JLJ and Audet, J and Casas-Ruiz, JP and Catalán, N and Cauvy-Fraunié, S and Colls, M and Deininger, A and Evtimova, VV and Fonvielle, JA and Fuß, T and Gilbert, P and Herrero Ortega, S and Liu, L and Mendoza-Lera, C and Monteiro, J and Mor, JR and Nagler, M and Niedrist, GH and Nydahl, AC and Pastor, A and Pegg, J and Gutmann Roberts, C and Pilotto, F and Portela, AP and González-Quijano, CR and Romero, F and Rulík, M and Amouroux, D (2018)The interplay between total mercury, methylmercury and dissolved organic matter in fluvial systems: A latitudinal study across Europe. Water Research, 144. pp. 172-182. ISSN 0043-1354 (In Press)

Downloaded from: http://e-space.mmu.ac.uk/621418/

Version: Accepted Version

Publisher: Elsevier

**DOI:** https://doi.org/10.1016/j.watres.2018.06.064

Please cite the published version

53

Abstract

1 The interplay between total mercury, methylmercury and dissolved organic matter in fluvial systems: A latitudinal study across Europe 2 Andrea G. Bravo<sup>1,\*,\*</sup>, Dolly N. Kothawala<sup>2,\*</sup>, Katrin Attermeyer<sup>2</sup>, Emmanuel Tessier<sup>3</sup>, Pascal Bodmer<sup>4,5</sup>, José L. J. Ledesma<sup>6</sup>, Joachim Audet<sup>6</sup>, Joan Pere Casas-Ruiz<sup>7</sup>, Núria Catalán<sup>7</sup>, Sophie Cauvy-Fraunié<sup>8</sup>, Miriam Colls<sup>7</sup>, Anne Deininger<sup>9</sup>, Vesela V. Evtimova<sup>10</sup>, Jérémy A. Fonvielle<sup>11</sup>, Thomas Fuß<sup>12,13</sup>, Peter Gilbert<sup>14</sup>, Sonia Herrero Ortega<sup>11</sup>, Liu Liu<sup>5</sup>, Clara Mendoza-Lera<sup>8</sup>, Juliana Monteiro<sup>15</sup>, Jordi-René Mor<sup>7,16</sup>, Magdalena Nagler<sup>17</sup>, Georg H. Niedrist<sup>18</sup>, Anna C. Nydahl<sup>2</sup>, Ada Pastor<sup>7</sup>, Josephine Pegg<sup>19,20</sup>, Catherine Gutmann Roberts<sup>19</sup>, Francesca Pilotto<sup>9</sup>, Ana Paula Portela<sup>15</sup>, Clara Romero González-Quijano<sup>12</sup>, Ferran Romero<sup>7</sup>, Martin Rulík<sup>21</sup>, David Amouroux<sup>3,\*</sup> 3 4 5 6 7 8 9 10 11 12 <sup>1</sup> Department of Environmental Chemistry, Institute of Environmental Assessment and Water 13 Research (IDAEA), Spanish National Research Council (CSIC), Barcelona, Spain <sup>2</sup> Limnology/Department of Ecology and Genetics, Uppsala University, Uppsala, Sweden 14 <sup>3</sup> CNRS/ UNIV PAU & PAYS ADOUR, Institut des Sciences Analytiques et de Physico-Chimie pour 15 16 l'Environnement et les Materiaux, UMR5254, MIRA, Pau, France 17 Institute for Environmental Sciences, University of Koblenz-Landau, Landau, Germany <sup>5</sup> Chemical Analytics and Biogeochemistry, Leibniz-Institute of Freshwater Ecology and Inland 18 19 Fisheries, Berlin, Germany <sup>6</sup> Department of Aquatic Sciences and Assessment, Swedish University of Agricultural Sciences, 20 Uppsala, Sweden 21 22 Catalan Institute for Water Research (ICRA), Girona, Spain 8 IRSTEA, UR RIVERLY, 69616 Villeurbanne Cedex, France 23 24 Department of Ecology and Environmental Science, Umeå University, Umeå, Sweden <sup>10</sup> Department of Aquatic Ecosystems, Institute of Biodiversity and Ecosystem Research, Bulgarian 25 26 Academy of Sciences, Sofia, Bulgaria 27 Experimental Limnology, Leibniz-Institute of Freshwater Ecology and Inland Fisheries (IGB), 28 Stechlin, Germany 29 <sup>12</sup> Ecohydrology, Leibniz-Institute of Freshwater Ecology and Inland Fisheries (IGB), Berlin, Germany <sup>13</sup> WasserCluster Biological Station Lunz, Lunz am See, Austria 30 <sup>14</sup> The Environmental Research Institute, University of Highlands and Islands, Thurso, Scotland, UK, 31 <sup>15</sup> Research Centre in Biodiversity and Genetic Resources (CIBIO), University of Porto, Porto, Portugal 32 33 <sup>16</sup> Department of Evolutionary Biology, Ecology and Environmental Sciences, Faculty of Biology, University of Barcelona (UB), Barcelona, Spain 34 35 Microbial Resource Management, Institute of Microbiology, University of Innsbruck, Innsbruck, Austria
<sup>18</sup> River and Conservation Research, Institute of Ecology, University of Innsbruck, Innsbruck, Austria 36 37 <sup>19</sup> Department of Life and Environmental Sciences, Bournemouth University, UK 38 <sup>20</sup> University Centre Sparsholt, Winchester, UK 39 40 <sup>21</sup> Department of Ecology and Environmental Sciences, Palacky University in Olomouc, Olomouc, 41 Czech Republic 42 43 <sup>+</sup>These authors contributed equally to this work. Corresponding authors: Andrea G Bravo (jandriugarcia@gmail.com) & David Amouroux 44 45 (david.amouroux@univ-pau.fr) 46 Keywords: Mercury, methylmercury, streams, rivers, freshwaters, organic 47 48 matter, EuroRun, fluorescence 49 50 51 52

Large-scale studies are needed to identify the drivers of total mercury (THg) and monomethyl-mercury (MeHg) concentrations in aquatic ecosystems. Studies attempting to link dissolved organic matter (DOM) to levels of THg or MeHg are few and geographically constrained. Additionally, stream and river systems have been understudied as compared to lakes. Hence, the aim of the study was to examine the influence of DOM concentration and composition, morphological descriptors, land uses and water chemistry on THg and MeHg concentrations and the percentage of THg as MeHg (%MeHg) in 29 streams across Europe spanning from 41°N to 64 °N. THg concentrations (0.06 - 2.78 ng L<sup>-1</sup>) were highest in streams characterized by DOM with a high terrestrial soil signature and low nutrient content. MeHg concentrations (78 - 160 pg L<sup>-1</sup>) varied non-systematically across systems. Relationships between DOM bulk characteristics and THg and MeHg suggest that while soil derived DOM inputs control THg concentrations, autochthonous DOM (aquatically produced) and the availability of electron acceptors for Hg methylating microorganisms (e.g. sulfate) drive %MeHg and potentially MeHg concentration. Overall, these results highlight the large spatial variability in THg and MeHg concentrations at European scale, and underscore the importance of DOM composition on Hg cycling in fluvial systems.

#### 1. INTRODUCTION

Mercury (Hg) is a hazardous substance with the potential to produce significant adverse neurological and other health effects in severely exposed humans, with particular concerns on unborn children and infants (Mason et al., 2012). Anthropogenic emissions and long-range transport of atmospheric-Hg have led to Hg concentrations above background levels in surface reservoirs (atmosphere, oceans or terrestrial) on a global scale (Amos et al., 2013). Aquatic food webs are particularly vulnerable to increases in Hg, since its methylated form (monomethyl-mercury, MeHg) accumulates in organisms and biomagnifies in aquatic food webs (Morel et al., 1998), even at low concentrations. As a consequence, millions of people are readily exposed to harmful concentrations of MeHg via fish consumption (Mason et al., 2012).

In order to prevent human exposure to Hg, a global legally binding agreement, the Minamata Convention, was signed by 128 countries and entered into force in September 2017. Prior to the Minamata Convention, the European Water Framework Directive (WFD, 2008) classified Hg as one of 30 priority hazardous substances and set the maximum allowable concentrations of environmental quality standards to 70 ng L<sup>-1</sup>. However, in aquatic systems, without any point anthropogenic or geological sources, total-Hg (THg) concentration typically ranges from less than 0.05 to 12 ng L<sup>-1</sup> (Dennis et al., 2005; Driscoll et al., 2007; Wiener et al., 2003). Because the maximum allowable concentrations of environmental quality standards were much higher than naturally relevant concentration levels of THq, national monitoring programs did not have the incentive to detect THq at ambient concentrations. There is even less information about MeHg background levels, which generally are between one and two orders of magnitude lower as compared to THg levels (Driscoll et al., 2007). In this context, streams and rivers are of special concern because very few studies have reported THg and MeHg concentrations at a wide geographical scale (Balogh et al., 2002; Brigham et al., 2009; Eklöf et al., 2015; Lescord et al., 2018; Marvin-DiPasquale et al., 2009; Tsui et al., 2009). While rivers and streams have, however, been hypothesized to be important contributors to the global biogeochemical Hg budget (Amos et al., 2014; Kocman et al., 2017; Schartup et al., 2015a; Sunderland and Mason, 2007), quantitative information on Hg levels is largely lacking, especially in the temperate zone. Filling this gap in knowledge is an important prerequisite in the ongoing quest to reduce uncertainties associated with the global biogeochemical cycling of Hg and for the development of remediation and risk management strategies within the context of the Minamata Convention.

82

83

84

85

86

87

88

89

90

91

92

93

94

95

96

97

98

99

100

101

102

103

104

105

106

107

108

109

Freshwaters receive Hg from a variety of sources including atmospheric deposition (Fitzgerald et al., 1998), release from industrial discharge (Bravo et al., 2015, 2014), and lateral export from terrestrial sources (e.g. soils and wetlands) (Brigham et al., 2009; Eklöf et al., 2012). Diffuse inputs from land management activities and remobilization of Hg previously accumulated in terrestrial ecosystems can be an important source comparable in

magnitude to direct Hg anthropogenic inputs (e.g. industrial releases or mining activities) to surface waters (Kocman et al., 2017). Long-term accumulation of Hg in terrestrial systems has been explained by the strong binding between atmospherically deposited Hg and sulfur sites within organic matter (OM) (Skyllberg et al., 2006; Xia et al., 1999). In boreal ecosystems, characterized by OM-rich soils, the strong correlation between THg and dissolved organic carbon (DOC) found in several studies (Brigham et al., 2009; Driscoll et al., 1995; Eklöf et al., 2012; Tsui et al., 2009) denotes that the mobilization of OM is an important vector for Hg transport from catchment soils to surface waters (Eklöf et al., 2012; Grigal, 2002). Besides stream water DOC, the proportion of wetland area within the catchment and the discharge have been used to predict THg concentrations in surface waters (Brigham et al., 2009; Eklöf et al., 2015; Tsui et al., 2009). In contrast, the relationship between DOC and MeHg remains poorly understood (Brigham et al., 2009; Tsui et al., 2009). The variation of MeHg concentrations in aquatic ecosystems is indeed more complex, as MeHg is potentially derived from numerous sources, not only terrestrial sources. In boreal streams, the wetland proportion of the catchment has also been linked to MeHg freshwater concentrations (Brigham et al., 2009). Wetlands, characterized by oxygen-depletion and low to intermediate nutrient levels, are known to be important sites for MeHg production and a relevant source of MeHg to inland waters (Schaefer et al., 2014; Tjerngren et al., 2012). Besides the contribution of MeHg produced in wetlands to stream waters, in-stream MeHg formation might also be an additional key source of MeHg (Balogh et al., 2002; Eklöf et al., 2015; Tsui et al., 2009), especially in catchments lacking extensive wetland coverage. The methylation of inorganic divalent Hg (IHg) to MeHg is primarily mediated by anaerobic micro-organisms (Gilmour et al., 2013) and is controlled by temperature, availability of electron acceptors (e.g. SO<sub>4</sub><sup>2-</sup> and Fe<sup>3+</sup>) (Bravo et al., 2015) and the molecular composition of OM, which influences IHq-speciation (Graham et al., 2013; Jonsson et al., 2014), IHq-availability (Chiasson-Gould et al., 2014; Mazrui et al., 2016; Schaefer and Morel, 2009; Schartup et al., 2015b) and microbial activity (Bravo et al., 2017). Specifically, IHg methylation processes are enhanced by the presence of algal OM (Bravo et al., 2017; Schartup et al., 2013) and fresh humic OM

110

111

112

113

114

115

116

117

118

119

120

121

122

123

124

125

126

127

128

129

130

131

132

133

134

135

136

leached from organic rich soils (Herrero Ortega et al., 2018). Indeed, elevated MeHg concentrations have been linked to periods of high algal production (Gascon Diez et al., 2016; Soerensen et al., 2017) and leaf fall (Balogh et al., 2002). However most studies investigating the role of OM as a transport vector for Hg and/or a key factor controlling MeHg formation have been carried out in the boreal biome. Consequently, there is still a high amount of uncertainty in identifying factors governing THg and MeHg levels in stream waters, particularly at a broader geographical scale.

The molecular composition of dissolved organic matter (DOM) is variable and can be an important factor in predicting THg and potentially, MeHg (Bravo et al., 2017; Herrero Ortega et al., 2018; Lescord et al., 2018; Noh et al., 2017; Schartup et al., 2015b, 2013). A common fingerprinting approach used to characterize the bulk DOM composition in inland waters is fluorescence spectroscopy (Fellman et al., 2010). The increasingly widespread use of fluorescence can be attributed to its wide accessibility, ease of use, and low costs. Three commonly used indexes (fluorescence index (McKnight et al., 2001), freshness index (Parlanti et al., 2000) and humification index (Zsolnay et al., 1999)) are particularly useful for distinguishing between DOM derived from different sources (McKnight et al., 2001), and degrees of freshness and humification (Zsolnay et al., 1999). Likewise, commonly known reoccurring peaks distributed across the fluorescence excitation-emission matrix (EEM) can provide insights into sample specific DOM characteristics. Fluorescence has indeed demonstrated the potential to reveal key DOM molecular characteristics associated with THg and MeHg concentrations (Herrero Ortega et al., 2018; Lescord et al., 2018; Noh et al., 2017; Schartup et al., 2015b).

The aim of this study was to identify the relationships between THg and MeHg concentrations and DOM composition (sources and quality) across a diversity of European streams. Unraveling the factors determining THg and MeHg background levels in the aquatic environment at a broad geographical scale is crucial for a better understanding of Hg cycling. For this purpose, we measured THg and MeHg concentrations, DOM concentration and

composition, anions and cations in filtered and unfiltered stream water at 29 sites with contrasting catchment characteristic across a wide latitudinal gradient across Europe.

#### 2. MATERIAL AND METHODS

#### 2.1 Organizational Framework

Water sampling for THg and MeHg concentration measurements was organized within the framework of the project called EuroRun: Assessing CO<sub>2</sub> fluxes from European running waters (available from https://freshproject-eurorun.jimdo.com/). EuroRun represents the first Collaborative European Freshwater Science Project for Young Researchers aiming to create synergistic interactions among young and early career scientists across Europe. Within the context of the collaborative EuroRun project, the EuroRun\_Mercury initiative included 12 teams distributed in 8 countries, which sampled 29 stream sites (Fig. 1). With 56% of participants being female, EuroRun\_Mercury was gender balanced. The teams represented different regions across Europe. For a specific region, each team selected several sites with contrasting dominant land uses (i.e. forest or agriculture) and a wide range of hydrological and watershed characteristics (Table S1).

#### 2.2 Sampling

The sampling procedure was explained and demonstrated to one lead representative from each team, who participated in a workshop performed in Sweden in September 2016. All the material needed to collect water samples along with a detailed written protocol was provided to participants during the workshop (see SI). The sampling took place during 3<sup>rd</sup> – 14<sup>th</sup> October 2016 and teams from Austria (AUT), Bulgaria (BGR), the Czech Republic (CZE), Germany (DEU), Spain (ESP), France (FRA), Great Britain (GBR), Portugal (PRT) and Sweden (SWE) collected water samples for this study (Fig. 1). The 29 sampled sites were located across a wide latitudinal gradient (from 41°N to 64 °N) and altitude range (from 1 to 1560 m above sea level), and spanned a wide range of hydrological (discharge from 0.1 to 56 m³ s<sup>-1</sup>;) and catchment characteristics (Table S1). Each team provided hydrological

(discharge, stream wetted width and stream order) and catchment characteristics data (land use in % and catchment area).

A total of 4 water samples were collected at each site. Amber borosilicate bottles (VWR, 250 mL) were used to store water and analyze THg and MeHg species in filtered and unfiltered fractions. A previous study determined the cleaning procedures suitable for THg and MeHg analyses with these bottles (Cavalheiro et al., 2016). In the field, borosilicate bottles were rinsed three times with filtered (0.45 µm, Sterivex-HV) or unfiltered water. For dissolved (D) THgD and MeHgD analyses about 250 mL of water were collected directly into the amber borosilicate bottle, filtered (0.45 µm, Sterivex-HV) and acidified (1% v/v final concentration; HCl Ultrex II, J. T. Baker) on site. A subsample of filtered water was placed into 40 mL pre-muffled EPA borosilicate glass vials (VWR) with PTFE caps for analysis of DOC concentration, DOM optical properties and concentration of anions and cations. Unfiltered water samples were collected for determination of the total fraction (<sub>T</sub>) of THq<sub>T</sub>, MeHg<sub>T</sub> (acidified) and total organic carbon (TOC) concentration. Samples were kept at 4°C and shipped to the laboratory by courier within 48 h. A subsample (10 mL) of unfiltered water was filtered in the laboratory, and stored frozen for later determination of anions and cations. Because four of the vials broke during transport, these samples could not be determined (marked as not determined in Table 1 and S2).

Results for unfiltered water samples correspond to the total fraction ( $THg_T$ ,  $MeHg_T$  and TOC), whereas the results for filtered water samples represent the dissolved fraction ( $THg_D$ ,  $MeHg_D$ , DOC, DOM optical properties, anions and cations).

214

215

216

217

218

219

220

213

193

194

195

196

197

198

199

200

201

202

203

204

205

206

207

208

209

210

211

212

#### 2.3 Chemical Analyses

#### 2.3.1 Analysis of Inorganic-Hg and Methylmercury

Inorganic-Hg (IHg) and MeHg (monomethylmercury) concentrations (Table 1) were measured using species-specific isotope dilution and capillary gas chromatography (Trace GC Ultra, Thermo Fisher, Waltham, MA, USA equipped with a TriPlus RSH auto-sampler) hyphenated to inductively coupled plasma mass spectrometer (Thermo X Series 2)

(Cavalheiro et al., 2016; Rodriguez Martín-Doimeadios et al., 2003). The detection limits of this method are 0.02 ng L<sup>-1</sup> for IHg and 0.005 ng L<sup>-1</sup> for MeHg. The measurement error (calculated by analyzing each sample three times) was less than 15% and 10% for IHg and MeHg concentrations, respectively. Based on previous studies (Bravo et al., 2014; Monperrus et al., 2004; Rodriguez Martín-Doimeadios et al., 2003), THg was calculated as the sum of IHg and MeHg.

#### 2.3.2 TOC, DOC and DOM optical properties

221

222

223

224

225

226

227

228

229

230

231

232

233

234

235

236

237

238

239

240

241

242

243

244

245

246

247

248

TOC and DOC concentrations were determined on unfiltered and filtered samples, respectively. Prior to analysis, samples were acidified to a pH of 2 with 0.1 M HCl, and analyzed using a Shimadzu TOC-V<sub>CPH</sub> total carbon analyzer. Optical properties were determined on filtered water using an Aqualog spectrofluorometer (Horiba), which measures absorbance and fluorescence simultaneously. The fluorescence spectrophotometer has a routine protocol before running samples that involves i) instrument validation (ensuring the signal: noise ratio <20,000 counts ii) ensuring the excitation lamp is calibrated to a given wavelength (maximum intensity at 467 nm), iii) the emission energy is calibrated using pure water at an emission of 397 nm and an excitation of 350 nm. Fluorescence spectra were collected across excitation wavelengths (Ex<sub> $\lambda$ </sub>) 250 to 445 nm at 5 nm increments and across emission wavelengths (Em<sub>λ</sub>) ranging from 300–600 nm. Bandpass and resolution were 5 nm. All spectra were measured in a 1 cm quartz cuvette and automatically blank-subtracted, corrected for instrument biases and inner filter effects using the FluorEssence™ software (Horiba). Samples were measured for fluorescence at a 2 second integration time, with the exception of five more optical dense samples from Great Britain and Sweden, which were measured at 0.5 seconds. All intensities were normalized to 1 second integration time postanalysis and the excitation-emission matrixes (EEM) were expressed in fluorescence units (FU).

Peak picking was used to define the relative intensity of four common peaks across the EEM including the Peak A ( $\lambda_{ex}$ : 260 nm,  $\lambda_{em}$ : 400–460 nm), Peak C ( $\lambda_{ex}$ : 320–360 nm,  $\lambda_{em}$ : 420–460 nm), and Peak M ( $\lambda_{ex}$ : 290–310 nm,  $\lambda_{em}$ : 370–410 nm) associated with soil

terrestrial DOM, as well as a protein-like peak T ( $\lambda_{ex}$ : 275 nm,  $\lambda_{em}$ : 340-350 nm) that is considered to be of algal/bacterial origin (i.e. autochthonous to the aquatic environment) (Fellman et al., 2010; Huguet et al., 2009). The summed area of total fluorescence (TF) under the EEMs was used to normalize each of the four peaks, which were expressed as a percent of TF (i.e. %Peak A = Peak A/ TF x 100%). We also calculated three common fluorescence indexes: i) the fluorescence index (FI, calculated as the ratio between emission at wavelengths of 470 and 520 nm, at an excitation of 370 nm), an indicator of source being more microbial (≈1.8) or terrestrial (≈1.2) (Cory et al., 2007; McKnight et al., 2001), ii) the freshness index (calculated as the ratio of emission at 380 nm, divided by the emission maxima between 420 and 435 nm, at an excitation of 310 nm) which reflects the age of OM, with higher values representing more recently produced OM (Parlanti et al., 2000) and iii) the humification index (HIX, ratio of peak areas across an emission spectra [435-480 nm / 300-345 nm]), indicating the degree of humification (Zsolnay et al., 1999). The absorbance at 254 nm (A<sub>254</sub>) was used to calculate carbon specific absorbance (SUVA<sub>254</sub>), a commonly used metric of the degree of aromaticity, as A<sub>254</sub>/DOC concentration normalized to a 1 m path length (L mg C<sup>-1</sup> m<sup>-1</sup>) (Weishaar et al., 2003).

#### 2.3.3 Anions and Cations

249

250

251

252

253

254

255

256

257

258

259

260

261

262

263

264

265

266

267

268

269

270

271

272

273

274

275

276

Inorganic analyses were conducted on a Metrohm IC system (883 Basic IC Plus and 919 Autosampler Plus). Anions (F-, Cl-, NO<sub>3</sub>- and SO<sub>4</sub><sup>2-</sup>) were separated on a Metrosep A Supp 5 analytical column (150 x 4.0 mm) fitted with a Metrosep A Supp 4/5 guard column at 0.7 mL min<sup>-1</sup> using a carbonate eluent (3.2 mM Na<sub>2</sub>CO<sub>3</sub> + 1.0 mM NaHCO<sub>3</sub>). A Metrosep C4 column (250 x 2.0 mm) with a Metrosep C4 guard column was used for separation of the cations (Na<sup>+</sup>, NH<sub>4</sub><sup>+</sup>, K<sup>+</sup>, Ca<sup>2+</sup>, Mg<sup>2+</sup>) with an eluent of 1.7 mM nitric acid and 0.7 mM dipicolinic acid at a flow rate of 0.2 mL min<sup>-1</sup>.

#### 2.4 Statistical Analyses

Kolmogorov-Smirnov test and quantile-quantile plots were used to evaluate the normality of the data. The distribution of THg concentrations were slightly skewed due to the high concentrations measured at sites in Great Britain. The center and dispersion of THg

concentrations were summarized by the median and the inter-quartile range (IQR). DOM optical parameters, MeHg and %MeHg met normality after being log-transformed. Parametric tests were used to analyze linear correlations (Pearson) between DOM optical parameters, THg and MeHg concentrations, and %MeHg and to compare Hg concentrations in the total and dissolved fractions (t-test). All statistical tests were carried out in R 3.2.4 (<a href="http://www.R-project.org/">http://www.R-project.org/</a>) (R Core Team, 2016).

Principle component analysis (PCA) was used as an initial diagnostic tool to examine key relationships and identify strongly correlated variables including optical measurements (i.e. %Peak A, %Peak C, %Peak M and %Peak T, SUVA<sub>254</sub>, fluorescence index, freshness index, humification index), TOC, sulfate, nitrate and calcium concentrations, land use and stream order. This complete information was available for 21 of the 29 studied sites. Accordingly, the samples with incomplete information (DEU1\_1, DEU1\_2, GBR2\_2, FRA1\_2, PRT1\_1, CZE1\_1, GBR2\_1 and ESP2\_3) were excluded from the PCA analysis. All data were log-transformed, scaled and centered prior to PCA analysis. PCA analysis were carried out using the *prcomp* function of the package *stats* in R 3.2.4 (R Core Team, 2016).

Partial least squares (PLS) regression models were subsequently applied to predict THg concentration (Y response variable) from DOM optical properties, water chemistry and catchment characteristics (X predictor variables, Table S5 and Table S6). A second PLS, excluding the sites with high THg concentrations (Great Britain, GBR1\_1, GBR1\_2), was also tested (Supplementary information, Table S7). The PLS model performance was examined based on the cumulative goodness of fit (R²Y, explained variation), and the cumulative goodness of prediction (or Q², predicted variation). The variable influence on projections (VIP scores) provided a means of assigning the importance of X variables on the overall PLS model. X-predictor variables with an influence score ≥1 were considered highly influential, between 0.8 and 1.0 moderately influential, and <0.8 less influential predictors (Eriksson et al., 1999). The PLS models were performed using the *plsr* function of the *pls* package in R (Mevik and Wehrens, 2007). Graphics were built in R 3.2.4 (R Core Team, 2016) and modified to fit the journal requirements with Inkscape 0.92 (https://inkscape.org/es/).

THg and MeHg loads of both total and dissolved fractions (Table S3) were calculated with the stream discharge value provided by the participants (Table S1) and the concentrations measured in this study (Table 1).

#### 3. RESULTS

#### 3.1 Mercury Levels (THg, IHg, MeHg and %MeHg)

THg concentrations were consistently low among sites (Table 1). Median values for total (THg<sub>T</sub>) and dissolved fractions (THg<sub>D</sub>) were 0.46 ng L<sup>-1</sup> (interquartile range, IQR: 0.28 – 0.8) and 0.24 ng L<sup>-1</sup> (IQR: 0.28 – 0.49), respectively. Streams in Great Britain (GBR) and Austria (AUT) had the highest THg<sub>T</sub> and THg<sub>D</sub> concentrations (Fig. 2a, Table 1). THg<sub>T</sub> concentrations were generally higher than THg<sub>D</sub> (p = 0.004), although both were significantly correlated (R<sup>2</sup>=0.62, p = 0.04; Table S3). IHg<sub>T</sub> was also slightly higher than IHg<sub>D</sub> (Fig. 2b). MeHg<sub>T</sub> concentrations ranged between 7.8 and 159 pg L<sup>-1</sup> and were similar (p = 0.07) to MeHg<sub>D</sub> (6.6 – 134 pg L<sup>-1</sup>; Fig. 2c). The highest MeHg concentrations, for total and dissolved fractions, were found at sites from Sweden (SWE2\_1), Great Britain (GBR1\_1) and Spain (ESP1\_3). The %MeHg was very similar in both the total (%MeHg<sub>T</sub>) and dissolved (%MeHg<sub>D</sub>) fractions (Fig. 2d) but varied among sites, ranging from 1 to 43%. The highest %MeHg<sub>T</sub> was found at a Spanish site (ESP1\_3). The median of THg<sub>T</sub> and MeHg<sub>T</sub> loads were 3.9 kg year<sup>-1</sup> (IQR: 0.6 – 6.8) and 0.3 kg year<sup>-1</sup> (IQR: 0.1 – 0.5), respectively.

Significant correlations were found between the total and dissolved fraction of THg, IHg and MeHg (Table S3). As total and dissolved fractions were correlated, we used the total fraction (THg<sub>T</sub>, MeHg<sub>T</sub>, %MeHg<sub>T</sub>) for all the following statistical analyses (PCA and PLS).

#### 3.2 Water Chemistry

DOC and TOC concentrations ranged between 0.9 and 18.5 mg L<sup>-1</sup> and 0.9 and 22.2 mg L<sup>-1</sup>, respectively. DOC and TOC were significantly correlated (R<sup>2</sup> = 0.99, p < 0.01). The highest concentrations were found in Great Britain (GBR1\_1, GBR1\_2) and Sweden (SWE1\_1) and the lowest in Spain (ESP1\_2, ESP2\_1) and Austria (AUT2\_1, AUT2\_2, AUT2\_3; Table S2). The concentrations of sulfate, a well-known electron acceptor for Hg methylating microorganisms, varied largely and ranged from 1.3 (AUT1\_1) to 731 (ESP2\_3)

mg L<sup>-1</sup>. About 50% of the studied sites presented sulfate concentrations lower than 12.6 mg L<sup>-1</sup> (IQR: 6.8 - 66 mg L<sup>-1</sup>). Spanish sites presented the highest sulfate concentrations (Table S2). Nitrate concentration varied between 0.04 (GBR1\_1) and 23 (ESP2\_2) mg L<sup>-1</sup>. The median value for nitrate concentration was 1.9 mg L<sup>-1</sup> (IQR: 0.7 - 2.7).

#### 3.3 DOM Optical Characteristics

Very distinct fluorescence signatures were observed between study sites, with four illustrative EEMs representing contrasting signatures displayed in Figure 3. Sites from Great Britain showed a strong optical signature representing terrestrially derived DOM, as indicated by the high intensity of Peaks A and C, while in contrast, sites from Spain (ESP) and the Czech Republic (CZE), with high Peak T intensities, represented DOM optical signatures of waters typically enriched in autochthonous DOM (Fig. 3, Table S2). The wide range of bulk DOM optical characteristics captured in this dataset can be attributed to the broad geographical scale of this study.

Analyzed water samples also captured a wide range of DOM characteristics based on each of the measured optical indexes. The SUVA<sub>254</sub>, which is associated with the presence of aromatic (Weishaar et al., 2003) and high molecular mass compounds (Chowdhury, 2013) typically spans from <1 to a maximum of 6 L mg C<sup>-1</sup> m<sup>-1</sup> (Weishaar et al., 2003), and was found to range from 2.4 (CZE1\_2) to 5.2 (GBR1\_1, GBR1\_2) L mg C<sup>-1</sup> m<sup>-1</sup>. HIX values spanned from 2.2 to 25 (Table S2). The highest HIX values were measured in Austria (AUT1\_2) and Great Britain (GBR1\_1, GBR1\_2). Some of the lowest values were observed in the Czech Republic (CZE1\_2) and Austria (AUT2\_2). While high HIX values (> 10) have been linked to the presence of complex molecules and strongly humified organic material, mainly of terrestrial origin, low values (<4) have been associated with autochthonous OM (Huguet et al., 2009). The fluorescence index (FI) covered almost the full range of this index, from a value of 1.3, representing terrestrial sources (degraded plant and soil OM; lower values), in GBR1\_2, to a value of 1.8, indicating microbial sources (including extracellular release and leachate from bacteria and algae) in ESP2\_2, ESP3\_3 (Cory et al., 2007; McKnight et al., 2001). The freshness index, typically associated with the contribution of

recently produced microbial-derived DOM (Parlanti et al., 2000), also spanned a wide range from 0.34 to 0.91 and was positively correlated to the FI and negatively correlated to the HIX and SUVA<sub>254</sub> (Fig. 4).

361

362

363

364

365

366

367

368

369

370

371

372

373

374

375

376

377

378

379

380

381

382

383

384

385

386

387

388

The wide distribution of DOM composition between sites indicated differences in the source, processing pathways and biological reactivity of the DOM, as shown by the PCA analysis (Fig. 5). The first principal component (PC1) explained 41% of the total variance across 21 European sites (Fig. 5a). Positive loadings on PC1 were associated with high TOC, A<sub>254</sub> and HIX, indicating a terrestrial DOM signature. In contrast, negative loadings of PC1 were found for agricultural land use, NO<sub>3</sub>, SO<sub>4</sub><sup>2</sup>, Ca<sup>2+</sup>, fluorescence Peak T and Peak M, freshness index and FI, likely associated with fresh and recently produced algal and microbial DOM (Ohno, 2002). Running waters from Great Britain (GBR1\_1; GBR1\_2), Sweden (SWE1\_1, SWE2\_1) and Austria (AUT1\_1, AUT1\_2) were characterized by low anions, cations and high concentrations of terrestrial humidified DOM (positive scores in PC1 Fig. 5b). In contrast, negative scores (Fig. 5b) of running waters from Spain (ESP) and the Czech Republic indicated an enrichment in nutrients and fresh autochthonous OM. Also, negative loadings for the percentage of urban and agricultural areas in the studied catchments suggest that anthropogenic activities might enrich running waters in nutrients and enhance primary production (Fig. 5a). The second component (PC2) explained 14% of the variance. Positive loadings on PC2 were found for the fluorescence Peak A and the ratio of the Peaks A to C (A/C). Both Peaks A and C have been linked to the presence of humic substances (Fellman et al., 2010). However, an increase in the relative abundance of the peak A compared to Peak C (A/C peak ratio) has been related to DOM processing, as Peak C is lost preferentially to Peak A under both photo- and biodegradation processes (Kothawala et al., 2012). Therefore, positive loadings of PC2 indicated terrestrial DOM that had been processed (Fig. 5a). On the opposite end, negative loadings on PC2 were associated with aromatic compounds (SUVA<sub>254</sub>) (Weishaar et al., 2003) and terrestrial DOM (Peak C) yet not strongly degraded (Kothawala et al., 2012).

#### 3.4 Predictors of THg, MeHg and %MeHg Concentrations

A PLS model (PLS model I, Fig. 6) identified key factors driving THg<sub>T</sub> concentrations in running waters. From the 29 available sites, only 21 sites were included in the PLS model which contained a complete set of required variables. To interpret the PLS loading plots, the distance and positioning of Y variables (THg, MeHg or %MeHg) relative to X variables indicated how well, or poorly, they related to each other. The greater the distance a variable (X or Y) lies from the origin, the greater its overall influence. Variables situated close together on the PLS plot are positively correlated, while variables opposite to each other are negatively correlated. The proportion of variance in the X variables (i.e. DOM composition, anions, cations and environmental factors) explained by the model is indicated by the R<sup>2</sup>X. The R<sup>2</sup>Y value indicates the proportion of variance in THg explained by the model. With two predictive components (R<sup>2</sup>X, 50%), the variance of THg concentrations was well explained (R<sup>2</sup>Y, 83.2%) and relatively well predicted (Q<sup>2</sup>, 60%) by the PLS model I (Fig. 6, Table S5, Table S6). PLS model I indicates that while TOC and A254 presented the highest positive correlation with THq, freshness index, FI and agriculture were negatively correlated. When not including sites highly influenced by terrestrial DOM inputs, GBR1\_1 and GBR1\_2, a second PLS (PLS model II), with two predictive components (R<sup>2</sup>X, 40%), explained the variance of THg<sub>T</sub> concentrations relatively well (R<sup>2</sup>Y, 82%) but could not predict them (Q<sup>2</sup>, -2%) (Table S7).

It was not possible to predict a significant proportion of the variability in MeHg concentrations using a PLS model with the measured variables. Nevertheless, the concentration of MeHg was significantly correlated to %MeHg (Fig. 4). Additionally, the %MeHg was significantly and positively correlated to Peak M (%), FI and sulfate concentration and negatively related to HIX and  $THg_T$  (Fig. 4).

412

413

414

415

416

389

390

391

392

393

394

395

396

397

398

399

400

401

402

403

404

405

406

407

408

409

410

411

#### 4. Discussion

#### 4.1 Hg levels in European running waters

Our results provide new information on THg and MeHg concentrations in streams at a European scale. THg concentrations across the studied European running waters were in the same range as those measured in pristine streams of northern Sweden (Eklöf et al., 2012), USA (Oregon, Wisconsin and Florida, Brigham et al., (2009); California, Tsui et al., (2009) and Minnesota, Balogh et al., (2002)) and France (Cavalheiro et al., 2016) and were lower than those reported for rivers typically contaminated by point sources (Baptista-Salazar et al., 2017; Bravo et al., 2014). Therefore, none of the study sites appeared to be affected by point sources of Hg.

417

418

419

420

421

422

423

424

425

426

427

428

429

430

431

432

433

434

435

436

437

438

439

440

441

442

443

Fluvial systems are an important transport pathway and ultimately deliver Hg originating from catchment soils to the marine environment. At a broader scale, we attempted to provide a rough estimate discharge of THg<sub>T</sub> and MeHg<sub>T</sub> of European rivers considering median values obtained in this study and the total European river discharge (3100 km<sup>3</sup>) (Shiklomanov and Rodda, 2004). Without considering point source of contamination, such as chlor-alkali or mining activities, known to increase stream Hg concentrations, the estimated European river discharge of THg<sub>T</sub> and MeHg<sub>T</sub> would be 1426 kg year<sup>-1</sup> (IQR: 868 – 2494) and 97 kg year<sup>-1</sup> (IQR: 60 - 254), respectively. Particles are an important vector of Hg transport in streams (Schuster et al., 2011) and can influence hydrologically mediated processes such as sedimentation and/or remobilization (e.g heavy rain events) of Hg previously deposited within river systems, which can largely affect THg<sub>T</sub> and MeHg<sub>T</sub> loads (Baptista-Salazar et al., 2017; Moreno-Brush et al., 2016; Sunderland and Mason, 2007). Measured THg and MeHg concentrations were notably higher in the total fraction compared to the dissolved fraction, highlighting the role of particles in the transport and fate of Hg in the studied fluvial systems. As precise hydrological data were unavailable here, and our calculations are based on a single sampling campaign, and the discharge of THg and MeHg only provides an initial rough estimate to be considered with care. In order to improve upon current estimates of European Hg discharge into the oceans (Kocman et al., 2017), greater monitoring of THg and MeHg is required across fluvial systems, particularly beyond the boreal ecozone.

#### 4.2 Terrestrial DOM: An Important Source of THg in European Running Waters

Using a multivariate approach comparing 21 streams spanning a wide geographical area across Europe, we were able to discriminate between the relative influence of bulk DOM optical characteristics and environmental variables on THg and MeHg concentrations. PC1 of the PLS model showed positive correlations between the highly influential predictors TOC and A<sub>254</sub> with THg concentrations in the studied sites and suggests that terrestrial DOM is an important driver of THg in streams (Fig. 6). Sites characterized by humic-rich soils, such as those in Sweden and Great Britain (Jones et al., 2005), presented the highest THg<sub>T</sub> and THg<sub>D</sub> concentrations. The positive correlation between terrestrial DOM and THg concentration in boreal rivers and streams has been previously reported in the literature (Brigham et al., 2009; Eklöf et al., 2012). As atmospheric Hg deposition during the industrial era have led to the accumulation of Hg in organic rich soils (Johansson and Tyler, 2001), the remobilization of Hg-enriched soil OM is an important process controlling THg concentrations in streams (Baptista-Salazar et al., 2017; Brigham et al., 2009; Driscoll et al., 1995; Eklöf et al., 2012; Kirk and Louis, 2009; Tsui et al., 2009) and lakes (Bravo et al., 2017).

It is noteworthy to highlight the lack of a significant correlation between TOC and  $THg_T$  concentration (p = 0.6) when the sites with the highest TOC and  $THg_T$  concentrations (SWE2\_1, GBR1\_1 and GBR1\_2) were excluded from the statistical analyses. Also, a poor predictability was obtained in a PLS model (PLS model II) built for  $THg_T$  excluding the sites highly influenced by terrestrial DOM (GBR1\_1 and GBR1\_2, Table S7). Therefore, TOC, and DOM components associated with terrestrial DOM, failed to predict  $THg_T$  concentrations in southern parts of Europe (FRA, ESP, CZE, AUT, BGR and DEU). This suggests that other Hg sources are relevant for the studied southern streams. Direct inputs of Hg deposited by wet and dry atmospheric depositional processes (Brigham et al., 2009; Domagalski et al., 2016; Lee et al., 2001) could contribute to  $THg_T$  concentrations in streams with low inputs of terrestrial DOM. Besides atmospheric deposition, catchment characteristics and changes in runoff may be relevant in determining  $THg_T$  in streams and rivers (Baptista-Salazar et al., 2017; Domagalski et al., 2016). Therefore, our results confirm the role of terrestrial OM as an important source of Hg for streams but highlight the uncertainties on the drivers of THg

concentrations at broader global scale when considering systems with low loads of terrestrial OM.

#### 4.3 Autochthonous DOM and Nutrients: Important Drivers of MeHg Formation in

#### **European Running Waters**

In this study, MeHg concentrations measured in running waters were in the same range as those previously reported for Sweden (Skyllberg et al., 2003), USA (California, Tsui et al., (2009); Florida, Brigham et al., (2009)) and France (Cavalheiro et al., 2016). Similarly, the %MeHg measured in samples collected in Sweden and Great Britain were similar to those previously reported in the literature (Brigham et al., 2009; Skyllberg et al., 2003). In contrast to THg, MeHg concentrations were not correlated to TOC, A<sub>254</sub> or HIX (Fig. 4), suggesting that terrestrial sources are not the main vector for MeHg in the studied streams. MeHg concentrations have been related to the concentration of autochthonous OM in reservoirs, lakes and in the Baltic sea (Gascon Diez et al., 2016; Noh et al., 2017; Soerensen et al., 2017).

The %MeHg has been considered a proxy of net MeHg formation (Drott et al., 2008). In the studied European streams studied here, the positive correlation between %MeHg to Peak M and the FI indicated that autochthonous OM is important for MeHg formation. Peak M, as well as Peak A and Peak C, have been generally described as humic-like peaks (Fellman et al., 2010). Peaks A and C have been linked to DOM derived primarily from vascular plant sources, aromatic and with a higher-molecular-weight fraction of the DOM pool (Coble, 1996). In contrast, Peak M represents compounds less aromatic and of lower molecular weight as compared to peaks A and C (Fellman et al., 2010). In a previous study performed in freshwaters, strong correlations between Peak M and Peak T (protein-like) were reported, but weaker relationships were observed between Peak M and Peaks A and C (Kothawala et al., 2014). Indeed, Peak M was originally used as a marker for autochthonous DOM from planktonic production (Fellman et al., 2010). A positive correlation of the %MeHg with Peak M and the FI thus suggests that net MeHg production is linked to fresh autochthonous OM. In contrast, the negative correlation between %MeHg and HIX suggests

that high concentrations of highly humified terrestrial DOM hampered Hg methylation processes most likely by decreasing Hg bioavailability (Chiasson-Gould et al., 2014; Soerensen et al., 2017) and/or bacterial activity (Bravo et al., 2017). High Hg methylation rates have been associated with algal derived OM (Bravo et al., 2017; Gascon Diez et al., 2016; Herrero Ortega et al., 2018; Mazrui et al., 2016). Our study provides further evidence of the potential role of algal and/or microbial DOM in the *in situ* formation of MeHg in streams.

The methylation of IHg to MeHg in aquatic systems has been primarily attributed to the activity of sulfate-reducing bacteria, iron-reducing bacteria (FeRB), methanogens and Firmicutes (Gilmour et al., 2013) in oxygen limiting conditions of sediments (Bravo et al., 2017), hypolimnion (Eckley and Hintelmann, 2006) and settling particles (Gascon Diez et al., 2016). As sulfate can be used as an electron acceptor by Hg methylating microorganisms (Pak and Bartha, 1998), the positive significant correlation found between the %MeHg and the sulfate concentration, suggests an important role of SRB in MeHg formation in running waters and provides further evidence for the relevance of *in situ* MeHg formation.

Stream sulfate mainly originates from atmospheric deposition and/or geogenic sources (Ledesma et al., 2018). Because sulfate release is accompanied by cation leaching (Prechtel et al., 2001), high concentrations of both sulfate and cations (e.g. calcium, Table S2) in Spanish sites indicate an effect of geological substrate of the river basin on the stream water chemistry. Although it was not the focus of this work, these results underline the relevance of geology and catchment characteristics on stream water properties and ultimately also for Hg cycling.

#### Conclusion

Our findings indicated that streams enriched in terrestrial DOM, as typical of Northern study sites, revealed the highest THg concentrations, suggesting the source was from organic rich catchment soils. In contrast, Southern European streams were enriched in nutrients and autochthonous DOM and although inputs of THg and terrestrial DOM from the

catchment were lower, we propose that MeHg levels were regulated by. We propose that the elevated MeHg concentrations in some of these sites could be due to a higher *in situ* MeHg formation, likely related to enhanced microbial activity, which is boosted by the availability of autochthonous algal/microbial derived DOM and sulfate. Our results show the interplay between DOM components and THg and MeHg concentrations in streams and provide new insights into Hg cycling in aquatic systems at a broad geographical European scale.

534

535

554

555

528

529

530

531

532

533

#### References

536 Amos, H.M., Jacob, D.J., Kocman, D., Horowitz, H.M., Zhang, Y., Dutkiewicz, S., Horvat, M., Corbitt, 537 E.S., Krabbenhoft, D.P., Sunderland, E.M., 2014. Global biogeochemical implications of mercury 538 discharges from rivers and sediment burial. Environ. Sci. Technol. 48, 9514–9522. 539 doi:10.1021/es502134t 540 Amos, H.M., Jacob, D.J., Streets, D.G., Sunderland, E.M., 2013. Legacy impacts of all-time 541 anthropogenic emissions on the global mercury cycle. Global Biogeochem. Cycles 27, 410-421. 542 doi:10.1002/gbc.20040 543 Balogh, S.J., Huang, Y., Offerman, H.J., Meyer, M.L., Johnson, D.K., 2002. Episodes of elevated 544 methylmercury concentrations in prairie streams. Environ. Sci. Technol. 36, 1665-1670. 545 doi:10.1021/es011265w 546 Baptista-Salazar, C., Richard, J.H., Horf, M., Reic, M., Gosar, M., Biester, H., 2017. Grain-size 547 dependence of mercury speciation in river suspended matter, sediments and soils in a mercury 548 mining area at varying hydrological conditions. Appl. Geochemistry 81, 132–142. 549 doi:10.1016/j.apgeochem.2017.04.006 Bravo, A.G., Bouchet, S., Guédron, S., Amouroux, D., Dominik, J., Zopfi, J., 2015. High methylmercury 550 551 production under ferruginous conditions in sediments impacted by sewage treatment plant 552 discharges. Water Res. 80, 245-255. doi:10.1016/j.watres.2015.04.039 553 Bravo, A.G., Bouchet, S., Tolu, J., Björn, E., Mateos-Rivera, A., Bertilsson, S., 2017. Molecular

composition of organic matter controls methylmercury formation in boreal lakes. Nat. Commun.

8, 14255. doi:10.1038/ncomms14255

556 Bravo, A.G., Cosio, C., Amouroux, D., Zopfi, J., Chevalley, P.-A., Spangenberg, J.E., Ungureanu, V.-557 G., Dominik, J., 2014. Extremely elevated methyl mercury levels in water, sediment and 558 organisms in a Romanian reservoir affected by release of mercury from a chlor-alkali plant. 559 Water Res. 49, 391-405. 560 Brigham, M.E., Wentz, D. a., Aiken, G.R., Krabbenhoft, D.P., 2009. Mercury cycling in stream 561 ecosystems. 1. Water column chemistry and transport. Environ. Sci. Technol. 43, 2720–2725. 562 doi:10.1021/es802694n 563 Cavalheiro, J., Sola, C., Baldanza, J., Tessier, E., Lestremau, F., Botta, F., Preud'homme, H., 564 Monperrus, M., Amouroux, D., 2016. Assessment of background concentrations of 565 organometallic compounds (methylmercury, ethyllead and butyl- and phenyltin) in French aquatic 566 environments. Water Res. 94, 32-41. doi:10.1016/j.watres.2016.02.010 567 Chiasson-Gould, S.A., Blais, J.M., Poulain, A.J., 2014. Dissolved organic matter kinetically controls 568 mercury bioavailability to bacteria. Environ. Sci. Technol. 48, 3153-3161. 569 doi:10.1021/es4038484 570 Chowdhury, S., 2013. Trihalomethanes in drinking water: Effect of natural organic matter distribution. 571 Water 39, 1-8. 572 Coble, P.G., 1996. Characterization of marine and terrestrial DOM in seawater using excitation-573 emission matrix spectroscopy. Mar. Chem. 51, 325-346. doi:10.1016/0304-4203(95)00062-3 574 Cory, R.M., McKnight, D.M., Chin, Y.P., Miller, P., Jaros, C.L., 2007. Chemical characteristics of fulvic 575 acids from Arctic surface waters: Microbial contributions and photochemical transformations. J. 576 Geophys. Res. Biogeosciences 112, 1–14. doi:10.1029/2006JG000343 577 Dennis, I.F., Clair, T.A., Driscoll, C.T., Kamman, N., Chalmers, A., Shanley, J., Norton, S.A., Kahl, S., 578 2005. Distribution patterns of mercury in Lakes and Rivers of northeastern North America. 579 Ecotoxicology 14, 113-123. doi:10.1007/s10646-004-6263-0 580 Domagalski, J., Majewski, M.S., Alpers, C.N., Eckley, C.S., Eagles-Smith, C.A., Schenk, L., Wherry, 581 S., 2016. Comparison of mercury mass loading in streams to atmospheric deposition in 582 watersheds of Western North America: Evidence for non-atmospheric mercury sources. Sci.

Total Environ. 568, 638-650. doi:10.1016/j.scitotenv.2016.02.112

584 Driscoll, C.T., Blette, V., Yan, C., Schofield, C.L., Munson, R., Holsapple, J., 1995. The role of 585 dissolved organic carbon in the chemistry and bioavailability of mercury in remote Adirondack lakes. Water, Air, Soil Pollut. 80, 499-508. doi:10.1007/BF01189700 586 587 Driscoll, C.T., Han, Y.-J., Chen, C.Y., Evers, D.C., Lambert, K.F., Holsen, T.M., Kamman, N.C., 588 Munson, R.K., 2007. Mercury contamination in forest and freshwater ecosystems in the 589 northeastern United States. Bioscience 57, 17. doi:10.1641/B570106 590 Drott, A., Lambertsson, L., Björn, E., Skyllberg, U., 2008. Do potential methylation rates reflect 591 accumulated methyl mercury in contaminated sediments? Environ. Sci. Technol. 42, 153-158. 592 Eckley, C.S., Hintelmann, H., 2006. Determination of mercury methylation potentials in the water 593 column of lakes across Canada. Sci. Total Environ. 368, 111-125. 594 doi:10.1016/j.scitotenv.2005.09.042 595 Eklöf, K., Fölster, J., Sonesten, L., Bishop, K., 2012. Spatial and temporal variation of THq 596 concentrations in run-off water from 19 boreal catchments, 2000-2010. Environ. Pollut. 164, 102-597 109. doi:10.1016/j.envpol.2012.01.024 598 Eklöf, K., Kraus, A., Futter, M., Schelker, J., Meili, M., Boyer, E.W., Bishop, K., 2015. Parsimonious 599 model for simulating total mercury and methylmercury in boreal streams based on riparian flow 600 paths and seasonality. Environ. Sci. Technol. 49, 7851–7859. doi:10.1021/acs.est.5b00852 601 Eriksson, L., Johansson, E., Wold, S., 1999. Introduction to Multi- and Megavariate Data Analysis 602 Using Projection Methods (PCA & PLS). Umetrics. 603 Fellman, J.B., Hood, E., Spencer, R.G.M., 2010. Fluorescence spectroscopy opens new windows into 604 dissolved organic matter dynamics in freshwater ecosystems: A review. Limnol. Oceanogr. 55, 605 2452-2462. doi:10.4319/lo.2010.55.6.2452 606 Fitzgerald, W.F., Engstrom, D.R., Mason, R.P., Nater, E.A., 1998. The case for atmospheric mercury 607 contamination in remote Areas. Environ. Sci. Technol. 32, 1–7. 608 Gascon Diez, E., Loizeau, J.-L., Cosio, C., Bouchet, S., Adatte, T., Amouroux, D., Bravo, A.G., 2016. 609 Role of settling particles on mercury methylation in the oxic water column of freshwater systems. 610 Environ. Sci. Technol. 50, 11672–11679. doi:10.1021/acs.est.6b03260

Gilmour, C.C., Podar, M., Bullock, A.L., Graham, A.M., Brown, S.D., Somenahally, A.C., Johs, A.,

512	new environments. Environ. Sci. Technol. 47, 11810–11820. doi:10.1021/es403075t
514	Graham, A.M., Aiken, G.R., Gilmour, C.C., 2013. Effect of dissolved organic matter source and
515	character on microbial Hg methylation in Hg-S-DOM solutions. Environ. Sci. Technol. 47, 5746-
516	5754. doi:10.1021/es400414a
517	Grigal, D.F., 2002. Inputs and outputs of mercury from terrestrial watersheds: a review. Environ. Rev.
518	10, 1–39. doi:10.1139/a01-013
519	Herrero Ortega, S., Catalán, N., Björn, E., Gröntoft, H., Hilmarsson, T.G., Bertilsson, S., Wu, P.,
520	Bishop, K., Levanoni, O., Bravo, A.G., 2018. High methylmercury formation in ponds fueled by
521	fresh humic and algal derived organic matter. Limnol. Oceanogr. 63, S44–S53.
522	doi:10.1002/lno.10722
523	Huguet, A., Vacher, L., Relexans, S., Saubusse, S., Froidefond, J.M., Parlanti, E., 2009. Properties of
524	fluorescent dissolved organic matter in the Gironde Estuary. Org. Geochem. 40, 706–719.
525	doi:10.1016/j.orggeochem.2009.03.002
526	Johansson, K., Tyler, G., 2001. Impact of atmospheric long range transport of lead, mercury and
527	cadmium on the Swedish forest environment. Water, Air Soil Pollut. Focus 1, 279–297.
528	Jones, R.J.A., Hiederer, R., Rusco, E., Montanarella, L., 2005. Estimating organic carbon in the soils
529	of Europe for policy support. Eur. J. Soil Sci. 56, 655-671. doi:10.1111/j.1365-
530	2389.2005.00728.x
531	Jonsson, S., Skyllberg, U., Nilsson, M.B., Lundberg, E., Andersson, A., Björn, E., 2014. Differentiated
532	availability of geochemical mercury pools controls methylmercury levels in estuarine sediment
533	and biota. Nat. Commun. 5, 4624. doi:10.1038/ncomms5624
534	Kirk, J.L., Louis, V.L.S., 2009. Multiyear total and methyl mercury exports from two major sub-arctic
535	rivers draining into Hudson Bay, Canada. Environ. Sci. Technol. 43, 2254–2261.
536	doi:10.1021/es803138z
537	Kocman, D., Wilson, S.J., Amos, H.M., Telmer, K.H., Steenhuisen, F., Sunderland, E.M., Mason, R.P.
538	Outridge, P., Horvat, M., 2017. Toward an assessment of the global inventory of present-day
539	mercury releases to freshwater environments. Int. J. Environ. Res. Public Health 14.

640	doi:10.3390/ijerph14020138
641	Kothawala, D., Stedmon, C., Müler, R., Weyhenmeyer, G., Köhler, S., Tranvik, L.J., 2014. Controls of
642	dissolved organic matter quality: evidence from a large-scale boreal lake survey. Glob. Chang.
643	Biol. 20, 1101–1114. doi:10.1111/gcb.12488
644	Kothawala, D.N., von Wachenfeldt, E., Koehler, B., Tranvik, L.J., 2012. Selective loss and
645	preservation of lake water dissolved organic matter fluorescence during long-term dark
646	incubations. Sci. Total Environ. 433, 238–246. doi:10.1016/j.scitotenv.2012.06.029
647	Ledesma, J.L.J., Futter, M.N., Blackburn, M., Lidman, F., Grabs, T., Sponseller, R.A., Laudon, H.,
648	Bishop, K.H., Köhler, S.J., 2018. Towards an Improved Conceptualization of Riparian Zones in
649	Boreal Forest Headwaters. Ecosystems 21, 297–315. doi:10.1007/s10021-017-0149-5
650	Lee, D.S., Nemitz, E., Fowler, D., Kingdon, R.D., 2001. Modelling atmospheric mercury transport and
651	deposition across Europe and the UK 35, 5455–5466.
652	Lescord, G.L., Emilson, E.J.S., Johnston, T.A., Branfireun, B.A., Gunn, J.M., 2018. Optical Properties
653	of Dissolved Organic Matter and Their Relation to Mercury Concentrations in Water and Biota
654	Across a Remote Freshwater Drainage Basin. Environ. Sci. Technol. acs.est.7b05348.
655	doi:10.1021/acs.est.7b05348
656	Marvin-DiPasquale, M., Lutz, M.A., Brigham, M.E., Krabbenhoft, P., Aiken, G.R., Orem, W.H., Hall,
657	B.D., Lutz, M.A., Brigham, M.E., Krabbenhoft, D.P., 2009. Production and Bed Sediment. 2. Pore
658	Water Partitioning Mercury Cycling in Stream Methylmercury Production and Bed Sediment -
659	Pore Water Partitioning. Environ. Sci. Technol. 43, 2726–2732. doi:10.1021/es802698v
660	Mason, R.P., Choi, A.L., Fitzgerald, W.F., Hammerschmidt, C.R., Lamborg, C.H., Soerensen, A.L.,
661	Sunderland, E.M., 2012. Mercury biogeochemical cycling in the ocean and policy implications.
662	Environ. Res. 119, 101–117. doi:10.1016/j.envres.2012.03.013
663	Mazrui, N.M., Jonsson, S., Thota, S., Zhao, J., Mason, R.P., 2016. Enhanced availability of mercury
664	bound to dissolved organic matter for methylation in marine sediments. Geochim. Cosmochim.
665	Acta 194, 153–162. doi:10.1016/j.gca.2016.08.019
666	McKnight, D.M., Boyer, E.W., Westerhoff, P.K., Doran, P.T., Kulbe, T., Andersen, D.T., 2001.
667	Spectrofluorometric characterization of dissolved organic matter for indication of precursor

668	organic material and aromaticity. Limnol. Oceanogr. 46, 38-48. doi:10.4319/lo.2001.46.1.0038
669	Mevik, BH., Wehrens, R., 2007. The pls Package: Principal Component and Partial Least Squares
670	Regression in R. J. Stat. Software, Artic. 18.
671	Monperrus, M., Krupp, E., Amouroux, D., Donard, O.F., Rodríguez Martín-Doimeadios, R., 2004.
672	Potential and limits of speciated isotope-dilution analysis for metrology and assessing
673	environmental reactivity. TrAC Trends Anal. Chem. 23, 261–272. doi:10.1016/S0165-
674	9936(04)00313-9
675	Morel, F.M.M., Kraepiel, A.M.L., Amyot, M., 1998. The chemical cycle and bioaccumulation of
676	mercury. Annu. Rev. Ecol. Syst. 29, 543–566. doi:10.1146/annurev.ecolsys.29.1.543
677	Moreno-Brush, M., Rydberg, J., Gamboa, N., Storch, I., Biester, H., 2016. Is mercury from small-scale
678	gold mining prevalent in the southeastern Peruvian Amazon? Environ. Pollut. 218, 150–159.
679	doi:10.1016/j.envpol.2016.08.038
680	Noh, S., Kim, J., Hur, J., Hong, Y., Han, S., 2017. Potential contributions of dissolved organic matter to
681	monomethylmercury distributions in temperate reservoirs as revealed by fluorescence
682	spectroscopy. Environ. Sci. Pollut. Res. doi:10.1007/s11356-017-0913-2
683	Ohno, T., 2002. Fluorescence inner-filtering correction for determining the humification index of
684	dissolved organic matter. Environ. Sci. Technol. 36, 742–746. doi:10.1021/es0155276
685	Pak, K., Bartha, R., 1998. Mercury methylation and demethylation in anoxic lake sediments and by
686	strictly anaerobic bacteria. Appl. Environ. Microbiol. 64, 1013–1017.
687	Parlanti, E., Wörz, K., Geoffroy, L., Lamotte, M., 2000. Dissolved organic matter fluorescence
688	spectroscopy as a tool to estimate biological activity in a coastal zone submitted to
689	anthropogenic inputs. Org. Geochem. 31, 1765–1781. doi:10.1016/S0146-6380(00)00124-8
690	Prechtel, A., Alewell, C., Armbruster, M., Bittersohl, J., Cullen, J.M., Evans, C.D., Helliwell, R.,
691	Kopácek, J., Marchetto, A., Matzner, E., Meesenburg, H., Moldan, F., Moritz, K., Veselý, J.,
692	Wright, R.F., 2001. Response of sulphur dynamics in European catchments to decreasing
693	sulphate deposition. Hydrol. Earth Syst. Sci. 5, 311–326. doi:10.5194/hess-5-311-2001
694	R Core Team, 2016. R: A Language and Environment for Statistical Computing.

- Rodriguez Martín-Doimeadios, R.C., Monperrus, M., Krupp, E., Amouroux, D., Donard, O.F.X., 2003.
- 696 Using speciated isotope dilution with GC-inductively coupled plasma MS to determine and
- 697 unravel the artificial formation of monomethylmercury in certified reference sediments. Anal.
- 698 Chem. 75, 3202–3211. doi:10.1021/ac026411a
- 699 Schaefer, J.K., Kronberg, R.-M., Morel, F.M.M., Skyllberg, U., 2014. Detection of a key Hg methylation
- 700 gene, *hgcA*, in wetland soils. Environ. Microbiol. Rep. 6, 441–447. doi:10.1111/1758-2229.12136
- 701 Schaefer, J.K., Morel, F.M.M., 2009. High methylation rates of mercury bound to cysteine by
- Geobacter sulfurreducens. Nat. Geosci. 2, 123–126. doi:10.1038/ngeo412
- Schartup, A.T., Balcom, P.H., Soerensen, A.L., Gosnell, K.J., Calder, R.S.D., Mason, R.P.,
- Sunderland, E.M., 2015a. Freshwater discharges drive high levels of methylmercury in Arctic
- 705 marine biota. Proc. Natl. Acad. Sci. 112, 11789–11794. doi:10.1073/pnas.1505541112
- 706 Schartup, A.T., Mason, R.P., Balcom, P.H., Hollweg, T.A., Chen, C.Y., 2013. Methylmercury
- production in estuarine sediments: Role of organic matter. Environ. Sci. Technol. 47, 695–700.
- 708 Schartup, A.T., Ndu, U.C., Balcom, P.H., Mason, R.P., Sunderland, E.M., 2015b. Contrasting effects
- 709 of marine and terrestrially derived dissolved organic matter on mercury speciation and
- 5965–5972. bioavailability in seawater. Environ. Sci. Technol. 5965–5972.
- Schuster, P.F., Striegl, R.G., Aiken, G.R., Krabbenhoft, D.P., Dewild, J.F., Butler, K., Kamark, B.,
- Dornblaser, M., 2011. Mercury Export from the Yukon River Basin and Potential Response to a
- 713 Changing Climate 9262–9267.
- 714 Shiklomanov, I.A., Rodda, J.C., 2004. World Water Resources at the Beginning of the Twenty-First
- 715 Century, Cambridge University Press (UK).
- Skyllberg, U., Bloom, P.R., Qian, J., Lin, C.-M., Bleam, W.F., 2006. Complexation of mercury(II) in soil
- organic matter: EXAFS evidence for linear two-coordination with reduced sulfur groups. Environ.
- 718 Sci. Technol. 40, 4174–80.
- Skyllberg, U., Qian, J., Frech, W., Xia, K., Bleam, W.F., 2003. Distribution of mercury, methyl mercury
- 720 and organic sulphur species in soil, soil solution and stream of a boreal forest catchment.
- 721 Biogeochemistry 64, 53–76. doi:10.1023/A:1024904502633
- Soerensen, A.L., Schartup, A.T., Skrobonja, A., Björn, E., 2017. Organic matter drives high interannual

724	538. doi:10.1016/j.envpol.2017.06.008
725 726	Sunderland, E.M., Mason, R.P., 2007. Human impacts on open ocean mercury concentrations. Global Biogeochem. Cycles 21. doi:10.1029/2006GB002876
727	Tjerngren, I., Karlsson, T., Björn, E., Skyllberg, U., 2012. Potential Hg methylation and MeHg
728	demethylation rates related to the nutrient status of different boreal wetlands. Biogeochemistry
729	108, 335–350. doi:10.1007/s10533-011-9603-1
730	Tsui, M.T.K., Finlay, J.C., Nater, E.A., 2009. Mercury bioaccumulation in a stream network. Environ.
731	Sci. Technol. 43, 7016–7022. doi:10.1021/es901525w
732	Weishaar, J.L., Aiken, G.R., Bergamaschi, B. a, Fram, M.S., Fujii, R., Mopper, K., 2003. Evaluation of
733	specific ultraviolet absorbance as an indicator of the chemical composition and reactivity of
734	dissolved organic carbon. Environ. Sci. Technol. 37, 4702–4708.
735	WFD, 2008. Directive 2008/105/EC of 16 December 2008 on environmental quality standards in the
736	field of water policy, amending and subsequently repealing Council Directives 82/176/EEC,
737	83/513/EEC, 84/156/EEC, 84/491/ECC, 86/280/ECC and amending Directive 2000/60/EC, The
738	European Parliament and the Council of the European Union. Official Journal of the European
739	Union. doi:http://eur-lex.europa.eu/legal-content/EN/TXT/?uri=celex:32008L0105
740	Wiener, J.G., Krabbenhoft, D.P., Heinz, G.H., Scheuhammer, A.M., 2003. Ecotoxicology of mercury,
741	in: Hoffman, D.J., Rattner, B.A., Jr., G.A.B., Jr., J.C. (Eds.), Handbook of Ecotoxicology, Second
742	Edition. Lewis Publishers, Boca Raton, FL, pp. 409-463. doi:10.1201/9781420032505.ch16
743	Xia, K., Skyllberg, U., Bleam, W.F., Bloom, P.R., Nater, E.A., Helmke, P.A., 1999. X-ray absorption
744	spectroscopic evidence for the complexation of Hg(II) by reduced sulfur in soil humic substances
745	Env. Sci Technol 33, 257–261.
746	Zsolnay, A., Baigar, E., Jimenez, M., Steinweg, B., Saccomandi, F., 1999. Differentiating with
747	fluorescence spectroscopy the sources of dissolved organic matter in soils subjected to drying.
748	Chemosphere 38, 45–50. doi:10.1016/S0045-6535(98)00166-0

### Acknowledgements

This research was partially funded the Generalitat de Catalunya (Beatriu de Pinos BP-00385-2016) grants to AGB. DNK would like to acknowledge the Swedish National Science Foundation, Vetenskapsrådet (VR), for a starting grant. We thank the initiators of the first Collaborative European Freshwater Science Project for Young Researchers, the European Federation of Freshwater Sciences (EFFS) board, the European Fresh and Young Researchers (EFYR) and the representatives of the Fresh Blood for Fresh Water (FBFW) meetings. We also thank the seven national freshwater societies financing this project, namely the Deutsche Gesellschaft für Limnologie e.V. (DGL; Germany), Swiss Society for Hydrology and Limnology (Switzerland), Iberian Association for Limnology (AIL; Spain and Portugal), Italian Association of Oceanography and Limnology (Italy), Freshwater Biological Association (FBA; United Kingdom), French Limnological Association (AFL; France), Group of Austrian Members of SIL (SIL-Austria), as well as the Leibniz-Institute of Freshwater Ecology and Inland Fisheries for additional funds. CML was funded by the French Agency for Biodiversity (ONEMA-AFB, Action 13, "Colmatage, échange snappe-rivière et processus biogéochimiques). NC held a "Juan de la Cierva" postdoctoral grant (FJCI-2014-23064). Furthermore, we would like to thank Christian Noss, Elvira de Eyto, Brian C. Doyle, Adam Bednařík, Tea Basic, Georgina Busst, David Fletcher, Danny Sheath, Laura Barral-Fraga, Anna Freixa, Xisca Timoner, Marcus Klaus, Wiebke Schulz, Nina Pansch, Nikolay Simov and Lyubomir A. Kenderov for field assistance, Christoffer Bergvall for laboratory assistance and Viktor Rosenberg for the GIS map. David Kocman is acknowledged for the discussion on Hg loads in rivers.

771

772

773

774

775

776

777

750

751

752

753

754

755

756

757

758

759

760

761

762

763

764

765

766

767

768

769

770

#### **Authors Contribution:**

AGB conceived the study. KA and PB coordinated the sampling and the teams involved in the study. AGB and ET performed GC-ICP-MS analyses of Hg with guidance from DA. DNK conducted fluorescence, spectrometry and DOC and TOC analyses. KA measured cations and anions. AGB and DNK conducted all statistical analyses. AGB, DNK and KA built the figures. AGB wrote the manuscript with significant assistance and comments from DNK, DA.

KA and PB. The rest of the co-authors carried out the sampling and contributed to the writing

of the MS.

780

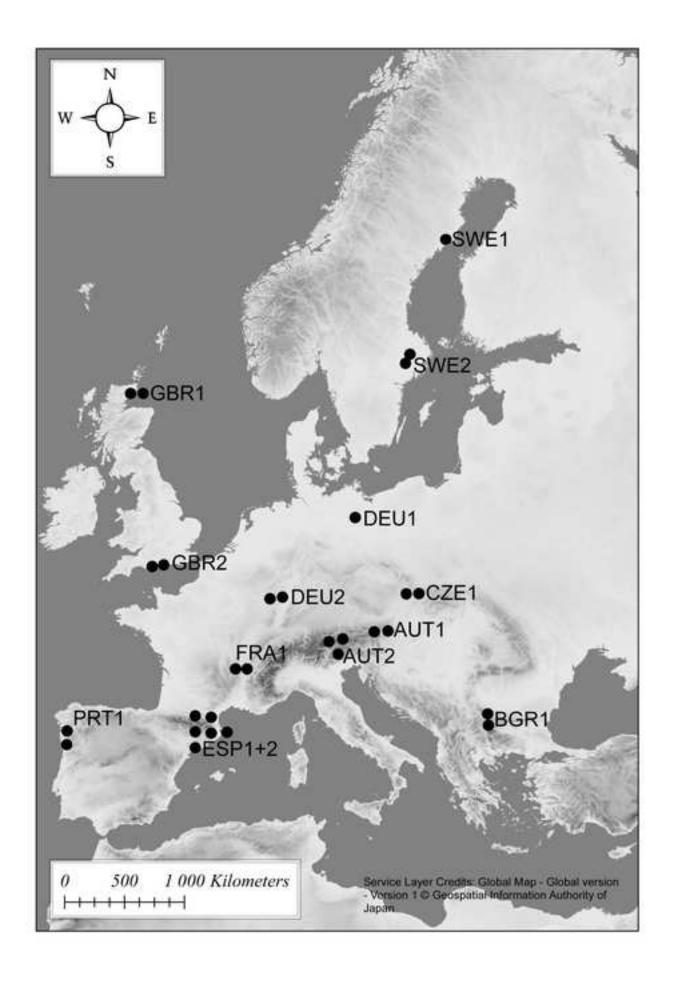
**Table 1.** Concentrations of inorganic-Hg (IHg), methylmercury (MeHg), total-Hg (THg = IHg+ MeHg) and the percentage of MeHg as THg (%MeHg) in the dissolved (filtered 0.45  $\mu$ m: <sub>D</sub>) and total (unfiltered: <sub>T</sub>) fraction of the studied streams.

Country	Stream ID	IHg <sub>D</sub>	MeHg <sub>D</sub>	THg <sub>D</sub>	MeHg <sub>D</sub>	IHg⊤	MeHg⊤	$THg_T$	MeHg <sub>⊤</sub>
_		ng L <sup>-1</sup>	pg L <sup>-1</sup>	ng L <sup>-1</sup>	%	ng L <sup>-1</sup>	pg L <sup>-1</sup>	ng L <sup>-1</sup>	%
Austria	AUT1 1	1.04	12	1.05	1.2	0.96	16	0.97	1.6
	AUT1_2	0.8	19	0.82	2.3	0.93	19	0.95	2.0
	AUT2_1	0.14	38	0.18	21.1	0.52	54	0.57	9.5
	AUT2_2	0.06	7	0.06	10.4	0.05	8	0.06	13.0
	AUT2_3	0.2	9	0.21	4.4	0.34	26	0.37	7.0
Bulgaria	BGR1_1	0.19	24	0.21	11.4	0.26	27	0.28	9.8
	BGR1_2	0.2	39	0.24	16.4	0.23	31	0.26	12.0
Czech Rep	CZE1_1	0.45	53	0.51	10.4	nd	nd	nd	nd
	CZE1_2	0.24	7	0.25	2.8	0.64	112	0.75	15.0
Germany	DEU1_1	0.17	67	0.23	28.7	0.3	12	0.31	3.9
·	DEU1_2	0.14	36	0.17	20.8	0.34	60	0.4	15.0
	DEU2_1	0.14	10	0.15	6.7	0.4	56	0.46	12.2
	DEU2_2	0.11	26	0.13	20.1	0.55	52	0.6	8.7
Spain	ESP1_1	0.3	67	0.37	18.2	0.37	89	0.46	19.3
•	ESP1_2	0.22	16	0.24	6.7	0.13	14	0.14	9.7
	ESP1_3	0.16	122	0.28	43.0	0.21	159	0.37	43.1
	ESP2_1	0.14	38	0.17	21.8	0.1	31	0.13	23.5
	ESP2_2	0.53	52	0.58	9.1	0.86	76	0.94	8.1
	ESP2_3	0.5	53	0.56	9.6	0.14	65	0.2	32.5
France	FRA1_1	0.47	15	0.48	3.1	0.49	15	0.51	2.9
	FRA1_2	0.39	17	0.41	4.2	0.86	20	0.88	2.2
Great Britain	GBR1_1	2.17	134	2.3	5.8	2.66	124	2.78	4.5
	GBR1_2	nd	nd	nd	nd	2.31	24	2.33	1.0
	GBR2_1	0.19	45	0.24	19.2	0.33	88	0.42	20.9
	GBR2_2	0.14	17	0.16	11.1	0.27	15	0.28	5.4
Portugal	PRT1_1	0.25	20	0.27	7.5	nd	nd	nd	nd
Sweden	SWE1_1	nd	nd	nd	nd	0.44	89	0.53	16.8
21100011	SWE2_1	0.48	135	0.61	21.9	0.7	153	0.86	17.8
	SWE2_2	0.13	26	0.16	16.2	0.15	30	0.18	16.4

- **Figure 1**. Distribution of the 29 sampling sites across nine European countries (see exact coordinates in Table 1).
- **Figure 2.** Box and whisker plots of the distribution of a) total Hg (THg) in the unfiltered (total) fraction  $THg_T$  and the filtered (dissolved) fraction  $THg_D$  (n = 27), b) inorganic Hg (IHg) in the total  $IHg_T$  and dissolved fractions  $IHg_D$  (n = 27), c) monomethyl-Hg (MeHg) concentrations (n = 27) in the total (MeHg<sub>T</sub>) and dissolved (MeHg<sub>D</sub>) fractions and d) the %MeHg (n = 27). Boxes represent  $25^{th}$  (Q1) and  $75^{th}$  (Q3) percentile. The median is indicated within the box. The lower whiskers represent the lowest value within 1.5 IQR of the Q1, and the highest within 1.5 IQR of Q3.
- **Figure 3**. Excitation-emission matrixes from sites presenting contrasted DOM characteristics a) and b) show samples with an eminently terrestrially derived DOM rich in humic peaks A and C. Plots c) and d) display samples with a marked microbial or algal signature as indicated by the high intensities in the peak T region. Intensities are shown in fluorescence units.
- **Figure 4.** Significant (p < 0.05, n=21) correlations between  $THg_T$ ,  $MeHg_T$ , the  $MeHg_T$ , DOM optical properties and stream characteristics. The ellipses have their eccentricity parametrically scaled to the Pearson correlation value (narrower ellipses represent higher correlation values). The orientation of the ellipses indicate negative (red) or positive (blue) correlations (scale on the right).
- **Figure 5.** Principal Component Analysis (PCA) of European streams (n = 21) showing the explained variability arising from principle components 1 (PC1) and 2 (PC2) based on bulk optical characteristics of dissolved organic matter (DOM) and environmental variables including stream order and catchment cover, as presented in (a) the loadings plot, and (b) the corresponding scores plot displaying the 21 study sites.
- **Figure 6.** Partial least squares loadings plot predicting the variability of total mercury (THg) concentration (Y variable, red) with several DOM optical properties and geographical and water chemistry characteristics as predictors (X variables). Fluorescence peaks are expressed as percentage of total fluorescence. Variables in blue were identified to be highly influential predictors of THg (with a variable influence on projection of  $\geq$  1.0)

**Graphical Abstract.** Conceptual framework of (a) streams characterized by high inputs of terrestrial DOM and high concentrations of THg and (b) streams enriched in microbial/algal DOM depicting high MeHg formation (%MeHg).

Figure 1 Click here to download high resolution image



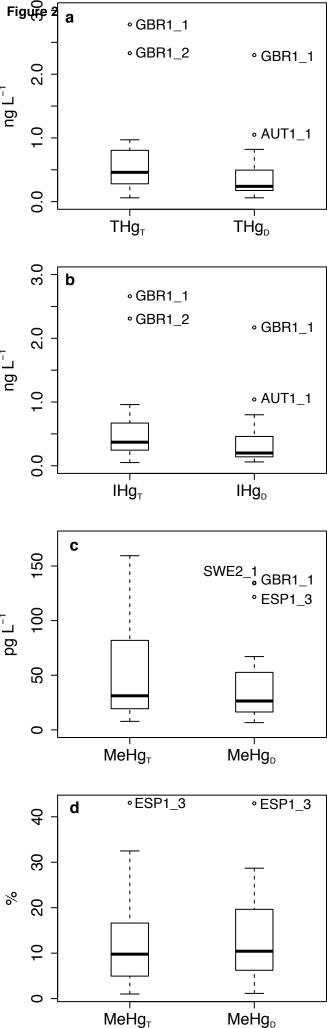
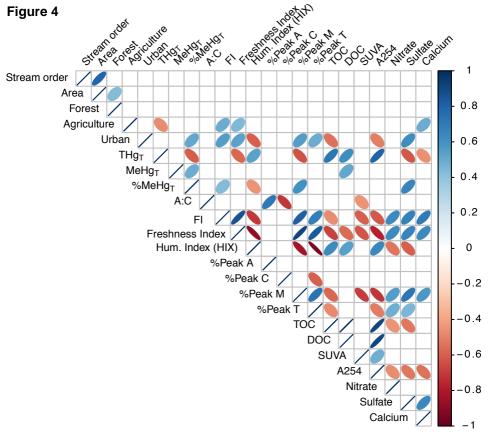


Figure 3 b) GBR1\_2 a) ESP1\_1 Fluorescence Units Peak A Peak C Peak M Emission Wavelengths (nm) Peak T -2000 250 270 290 310 330 350 370 390 410 430 450 c) ESP2\_2 d) CZE1\_2 Fluorescence Units 250 270 310 330 350 370 410 430 450

Excitation Wavelengths (nm)



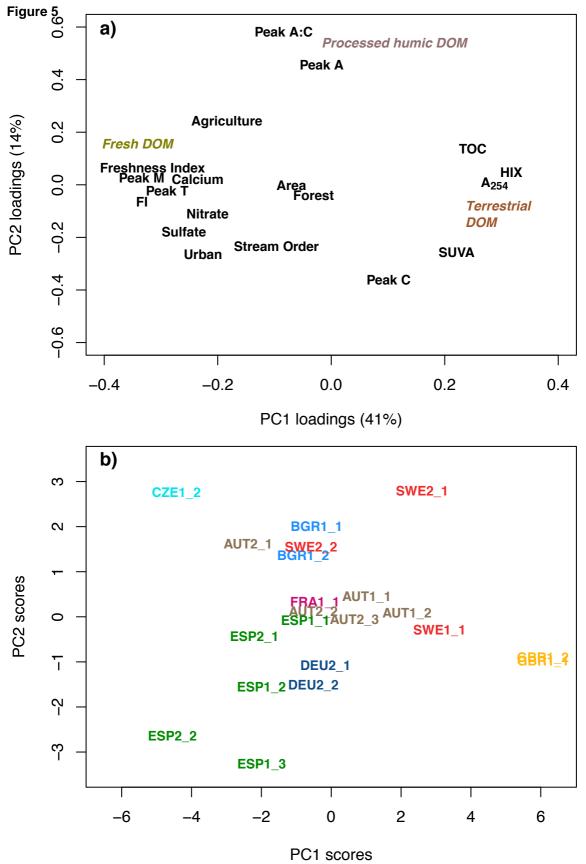
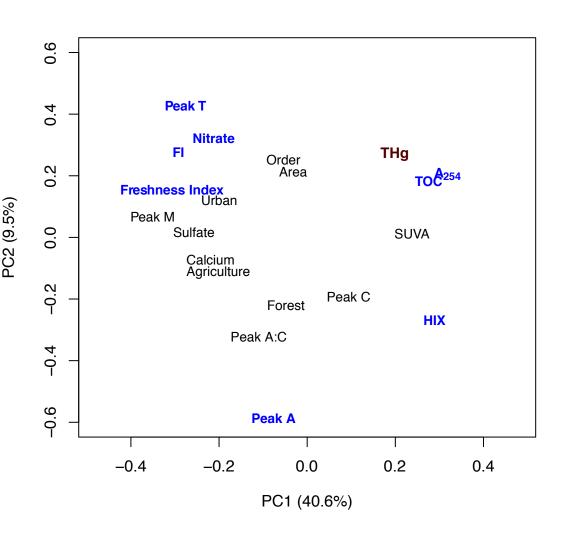


Figure 6



## Electronic Supplementary Material (for online publication only) Click here to download Electronic Supplementary Material (for online publication only): EuroRun\_Hg\_Sl\_v4.docx

Method Details (MethodsX)
Click here to download Method Details (MethodsX): MethodsX-Bravo et al\_v3.docx.zip