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1 Biodegradation of Emiliania huxleyi Aggregates by a Natural

2 Mediterranean Prokaryotic Community under Increasing

3 **Hydrostatic Pressure**

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18**Abstract**

19In the deep ocean, fluxes of particulate organic carbon (POC) and calcium carbonate are positively 20correlated, suggesting that CaCO₃ could increase sinking particle densities and/or protect the organic 21matter from degradation by prokaryotes, the so called "ballast effect". Here, we used the PArticle 22Sinking Simulator system to investigate the effect of increasing pressure on the biodegradation of 23calcifying Emiliania huxleyi aggregates. A ten days incubation was performed, simulating the changes 24in temperature and pressure experienced in situ in the NW Mediterranean Sea. Aggregates sinking 25from 200m to 1700m depth (assuming an average sinking velocity of 150 m d⁻¹) were exposed to a 26fresh natural mesopelagic prokaryotic community, with limited influence of culture-associated 27prokaryotes. E. huxleyi aggregates lipid composition revealed more degradation under increasing 28hydrostatic pressure (relative to constant atmospheric pressure), suggesting increased cell lysis. In 29parallel, changes in particulate inorganic carbon and total alkalinity indicated CaCO₃ dissolution, 30potentially accelerated under pressure. However, increased hydrostatic pressure also had a positive 31effect on particle aggregation, which may compensate the effect of increased cell lysis. Our results 32imply that in coccolithophorid-dominated sinking aggregates, the ballasting and protection effects of 33coccoliths may collapse throughout the water column. The increase in aggregation potential with 34pressure observed in controlled conditions might balance the loss of mineral ballast to a certain 35extent, although this needs to be confirmed in situ.

36Keyword: biological carbon pump; mesopelagic; mineral ballast; coccolithophorid; prokaryotes; 37biodegradation; hydrostatic pressure

38Introduction

39The ocean's biological carbon pump transports carbon into the deep ocean in the form of organic 40matter (Eppley and Peterson, 1979). Between 1-40% of the organic carbon newly formed from CO₂ 41fixation at the sea surface is transported to the deep sea (Ducklow et al., 2001), away from direct 42contact with the atmosphere. This process absorbs a large portion of the anthropogenic CO₂ 43emissions in the atmosphere, limiting greenhouse CO₂ gas build-up (Le Quéré et al., 2010). However, 44many climate-sensitive factors can influence organic carbon export and remineralization by 45mesopelagic heterotrophs, such as the mineral content of sinking particles (Armstrong et al., 2001; 46Kwon et al., 2009).

47Globally, approximately 5-20 Pg C yr⁻¹ leaves the euphotic zone (20-120m depth) in the form of 48sinking particulate organic carbon (POC) (Henson et al., 2011; Siegel et al., 2014). However, most of 49this POC is potentially remineralized to CO₂ in the upper mesopelagic zone (Buesseler and Boyd, 502009; Giering et al., 2014), where the activity of particle-attached prokaryotes and zooplankton play 51an important role in POC attenuation (Cho and Azam, 1988; Giering et al., 2014; Turley and Mackie, 521994, 1995). On one hand, sinking POC collected *in situ* can be rapidly remineralized by attached 53prokaryotes, as demonstrated by incubations done at atmospheric pressure (Goutx et al., 2007; 54Panagiotopoulos et al., 2002; Sempéré et al., 2000). On the other hand, the remineralisation of 55sinking particles (diatom aggregates or faecal pellets) and dissolution of biogenic minerals (opal) by 56fresh natural surface prokaryotic assemblages were observed to slow down under increasing pressure 57(Tamburini et al., 2006, 2009). Correlation between POC and mineral fluxes (calcium carbonate, opal, 58clays) in the deep ocean has lead to the interpretation that "ballast" minerals exert an important 59control over POC export (Armstrong et al., 2001; Francois et al., 2002; Klaas and Archer, 2002). 60Oceanic provinces in which calcifying plankton dominates display lower mesopelagic remineralisation 61fluxes than systems dominated by opal-producing diatoms (Cardinal et al., 2001, 2005; Jacquet et al.,

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622005). In sinking particles collected in situ, the presence of minerals has been suggested to preserve

63organic matter from degradation (Hedges et al., 2001; Ingalls et al., 2003). These observations suggest 64that mineral and POC dissolutions in the water column may be inter-related. To understand these 65interactions, a few biodegradation studies have been conducted using aggregates obtained from 66cultures of opal-protected diatoms, which form important seasonal blooms (Treguer and Pondaven, 672000). Biologically-mediated biogenic opal dissolution has been observed to increase with organic 68matter degradation in surface waters (Bidle and Azam, 1999, 2001; Goutx et al., 2007) and to 69diminish with temperature decrease (Bidle et al., 2002) or hydrostatic pressure increase (Tamburini et 70al., 2006). Adding another ballast mineral (CaCO₃) to diatom aggregates diminished diatom aggregate 71remineralisation by zooplankton (Le Moigne et al., 2013). Regarding CaCO₃, biologically-mediated 72dissolution has also been suggested (Antia et al., 2008; Harris, 1994; Jansen and Wolf-Gladrow, 2001; 73Milliman et al., 1999) to explain the fact that significant CaCO₃ dissolution is occurring in the upper 741000 m of the water column above the calcite or aragonite saturation horizon (Wollast and Chou, 751998, 2001). However, there is still no or contradictory evidence that this dissolution could be 76microbiologically-mediated (Bissett et al., 2011), and the link between CaCO₃ dissolution and POC 77degradation remains unclear.

78Ocean acidification experiments using diatom aggregates supplemented with CaCO₃ ballast indicate 79that CaCO₃ dissolution and POC degradation will potentially increase with increasing pCO₂, a 80phenomenon that is exacerbated under increasing pressure (de Jesus Mendes and Thomsen, 2012). 81However, little is known about the biodegradation of calcifiers aggregates. Among the calcifiers, 82coccolithophorids contribute to air-sea CO₂ exchange, through their combined photosynthesis (sink of 83CO₂) and calcification (source of CO₂) activities (Poulton et al., 2007). Coccolithophorid aggregates 84formation has been reported from many recent lab studies (Biermann and Engel, 2010; Chow et al., 852015; Engel et al., 2009a, 2009b; Iversen and Ploug, 2010), but they are considered as being more 86sensitive to fragmentation than diatom aggregates (De La Rocha and Passow, 2007).

87Laboratory biodegradation experiments using aggregates from the coccolithophore *Emiliania huxleyi* 88showed that heavily calcified strains have higher settling velocities and are better preserved over time

89than poorly calcified strains (Engel et al., 2009a, 2009b). Oxygen consumption within *E. huxleyi* 90aggregates can be 4 to 7 times larger than that of marine snow and diatom aggregates, indicating that 91they may be more prone to degradation (Ploug et al., 2008). Nevertheless, the carbon-specific 92respiration rate per meter settled was found to be lower in *E. huxleyi* aggregates than in diatom 93aggregates of similar size (Iversen and Ploug, 2010). This potentially confers coccolithophorid 94aggregates a larger potential for transporting organic carbon deeper in the ocean relative to diatom 95aggregates. It is important to note that the biodegradation experiments cited above were conducted 96in the presence of culture-associated prokaryotes or natural prokaryotic assemblages stored for a 97long time before use and the effect of fresh natural prokaryotic assemblages remains to be assessed. 98In addition, the increase in pressure experienced by aggregates sinking through the water column has 99yet to be included in such experiments.

100The objective of the present study was therefore to investigate the effect of hydrostatic pressure on 101the lipid composition (an indicator of cell degradation) and CaCO₃ dissolution within *E. huxleyi* 102aggregates in the presence of fresh free-living natural prokaryotic assemblages collected at 200 m 103depth in the NW Mediterranean Sea. Incubations were performed under controlled pressure 104conditions simulating a sinking rate of 150 m d⁻¹, in which we monitored the structure of the 105prokaryotic communities as well as the degradation of *E. huxleyi* lipids over 10 days following the 106experimental design presented in Tamburini et al. (2009). Total alkalinity, particulate inorganic and 107organic carbon (PIC and POC) and total organic carbon (TOC) concentrations were also monitored as 108indicators of CaCO₃ dissolution and organic carbon remineralization.

109Results

110Two sets of incubations (Exp A and Exp B) were performed at 13 °C, with two pressure conditions 111applied in parallel: constant atmospheric pressure (ATM) and increasing hydrostatic pressure 112corresponding to a sinking simulation from 200 to 1700 m depth (HP). The E. huxleyi aggregate 113dilutions in either sterile-filtered (Exp A) or fresh GF/F-filtered (Exp B) seawater were sampled at to

114and after 10 days in the incubation vessels (t10). Exp A was performed with sterile-filtered seawater 115to monitor the development of prokaryotes associated with the *E. huxleyi* culture, while Exp B was 116designed to observe the development on *E. huxleyi* aggregates of a natural freshly collected seawater 117free-living prokaryotic community, and its effect on the organic and inorganic carbon contents of the 118aggregates.

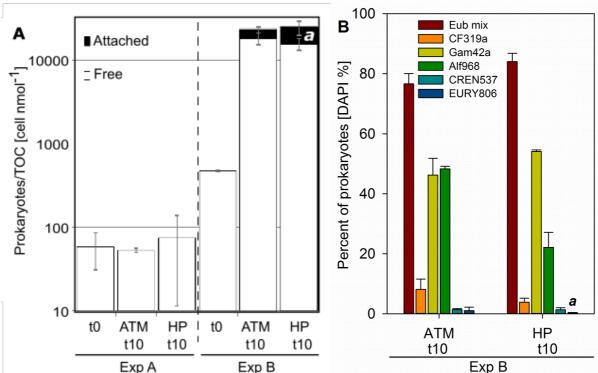
119**Prokaryotic development**

120The prokaryotic abundance associated with *E. huxleyi* culture aggregates (Exp A) did not change 121significantly between the start (t0: $7.7 \pm 1.8 \, 10^3$ cells mL⁻¹, average \pm SD) and end of the incubation 122(ATMt10: $6.5 \pm 0.2 \, 10^3$ cells mL⁻¹; HPt10: $9.4 \pm 3.92 \, 10^3$ cells mL⁻¹). Normalizing cell concentrations by 123the total organic carbon (TOC) concentration measured in each replicate to account for potential 124variations in the initial amount of aggregates added to each bottle, no significant difference was 125observed between both pressure conditions (Fig. 1A).

126In Exp B, the total prokaryotic abundance increased by two orders of magnitude after 10 days 127(ATMt10: $2.2 \pm 0.4 \pm 10^6$ cells mL⁻¹, HPt10: $2.7 \pm 0.0 \pm 10^6$ cells mL⁻¹), with no significant difference 128between pressure treatments, when normalised by TOC (Fig. 1A). The growth of prokaryotes attached 129to *E. huxleyi* aggregates was stimulated under increasing pressure incubation (HP), where large 130aggregates were observed at the end (Suppl. Fig. 1). In this condition, attached prokaryotes (> 2 μ m) 131were significantly more abundant (35.9 \pm 14.7 % of total DAPI positive cells) than under constant 132atmospheric pressure (20.9 \pm 4.7 %, Mann-Whitney-MW p < 0.05).

133Under both atmospheric and increasing pressure conditions, abundances of *Archaea* and *Bacteria* 134were two and three orders of magnitude higher than at t0, respectively. Although the proportion of 135*Euryarchaea* was 4.1 times lower under pressure (Fig. 1B, significant, MW p < 0.05), increasing 136pressure had no effect on the proportion and abundance of *Crenarchaea* (ATMt10: 2.3 ± 0.2 10⁴ cell 137mL⁻¹, HPt10: 2.6 ± 0.4 10⁴ cell mL⁻¹, MW p = 0.51) and *Bacteria* (ATMt10: 1.3 ± 0.2 10⁶ cell mL⁻¹, HPt10: 1381.4 ± 0.2 10⁶ cell mL⁻¹, MW p = 0.83). Among the *Bacteria*, Alpha- and Gamma-proteobacteria cell

139concentrations increased by three orders of magnitude during the incubations, and the fraction of 140Alphaproteobacteria was twice as low in conditions of increasing HP than at ATM (not significant, 141MW p = 0.13). The proportion of Bacteroidetes also tended to be on average 1.9 times lower in HP 142than in ATM conditions (not significant, MW p = 0.83), but their concentrations increased only by one 143to two orders of magnitude.

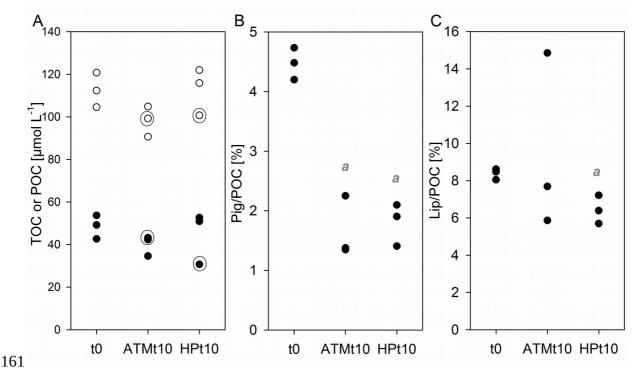


145Figure 1: (A) Total prokaryotic abundance (DAPI-stained cells) normalized by sample TOC concentration in 146Emiliania huxleyi aggregates biodegradation Exp A (no natural prokaryotes added) and Exp B (GF/F-filtered 147natural prokaryotic community added) initially (t0) and after 10 days under constant atmospheric pressure 148(ATMt10) or increasing hydrostatic pressure (HPt10). White bars: free-living prokaryotes; Black bars: >2 μ m 149particle-attached prokaryotes; (B) Percent of total prokaryotes (DAPI-stained cells) at the end of Exp B, 150detected by catalyzed reporter deposition coupled with fluorescence in situ hybridization (CARD-FISH). Eub 151mix (Eub338,Eub-II and Eub-III): Bacteria; CF319a: Bacteriodetes; Alf968, Gam42a: alpha- and gamma-152subclasses of Proteobacteria; Cren537: Crenarchaea; Eury806: Euryarchaea. Average±SD. Significant differences 153between ATM and HP are indicated as a (Mann-Whitney p < 0.05) for particle-attached (A) or total prokaryotes 154(B).

155E. huxleyi detritus organic carbon content and lipid degradation

156During the Exp B, TOC and particulate organic carbon (POC) concentrations tended to decrease over 157the 10 days incubation under atmospheric pressure only (Fig. 2A, ATMt10, not significant, p = 0.13). 158The POC pigment content decreased significantly after 10 days of exposure to natural prokaryotic

159assemblages, with no effect of pressure (Fig. 2B). In contrast, the lipid content (excluding pigments) 160decreased significantly in HPt10 but not in ATMt10 (Fig. 2C).

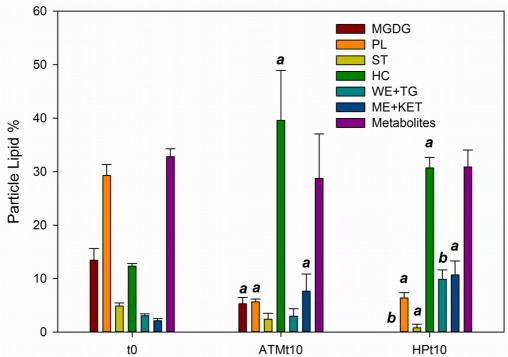


162 Figure 2: Variation in organic carbon composition in Exp B: (A) total or particulate organic carbon (TOC-empty 163 circles, or POC-filled circles, μ mol L⁻¹), (B) POC pigment fraction (%), (C) POC lipid fraction (pigments excluded, 164%). Significant differences (Mann-Whitney p < 0.05) are indicated as: a between t0 and t10. Circled data points 165 belong to the same ATM and HP replicates circled in Fig. 4.

166At the start of Exp B, the lipid composition in *E. huxleyi* aggregates was similar to the one described in 167an earlier study with the same strain of *E. huxleyi* in exponential growth (Bell & Pond, 1996; see 168suppl. Table 1). Overall, chloroplast lipids (including pigments and membrane monogalactosyl-169diacylglycerol-MGDG) and cell membrane lipids (phospholipids) dominated the lipid composition 170(61.9-63.1 %). Sterols, hydrocarbons, neutral lipids (wax esters, triglycerides, methyl esters, ketones, 171free fatty acids and alcohols) and metabolites (degradation lipids: mono- and di-acylglycerols) 172accounted for 35.1-37.2 % of the particle lipids.

173Within the non-pigment lipids, a significant decrease in the proportion of chloroplast membrane and 174whole cell membrane lipids (MGDG and PL, to 9.8-18.2 % and 25.4-33.5 %, respectively) was 175observed in both ATMt10 (2.5-6.8 % and 4.8-6.8 %) and HPt10 samples (0.0 % and 4.7-8.5 %). 176Meanwhile, the proportion of hydrocarbons, methyl esters and ketones increased significantly in both

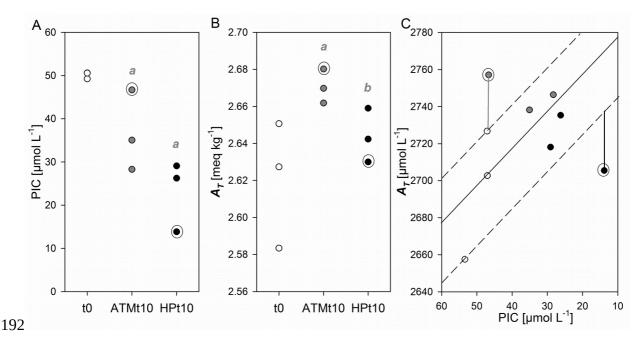
177conditions (Fig. 3). HPt10 samples experienced significant losses of membrane sterols and MGDG 178relative to ATMt10 samples. At the same time, the relative proportion of reserve wax esters and 179triglycerides were significantly higher in HP (HP: 5.9-12.5 %; ATM: 0.0-5.5 %), mainly due to the 180appearance of triacylglycerides.



181 t0 ATMt10 HPt10 182Figure 3: Variation in the lipid composition of *E. huxleyi* detritus in Exp B: Proportions of the different lipids 183(pigments are excluded, % L_{tot}). MGDG: monogalactosyldiacylglycerol, PL: phospholipids, ST: sterols, HC: 184Hydrocarbons, WE+TG: wax esters and triglycerides, ME+KET: methyl esters and ketones, ; Average±SD. 185Significant differences (Mann-Whitney p < 0.05) are indicated as: a between t0 and t10, b between ATM and HP.

186Inorganic carbon content variations

187A significant drop of particulate inorganic carbon (PIC, Fig. 4A) was observed in both ATM and HP 188conditions after 10 d (p < 0.05). It was larger under HP than at ATM (not significant, p = 0.13). Total 189alkalinity (A_T) increased significantly between t0 and ATMt10 (Fig. 4B, p < 0.05) but the increase was 190significantly lower under HP. The increase in A_T observed between t0 and HPt10 was indeed not 191statistically significant (p = 0.27).



193Figure 4: Variation in (A) particulate inorganic carbon (PIC, μ mol L⁻¹), (B) total alkalinity (A_T, meq kg⁻¹), and (C) 194A_T converted to μ mol L⁻¹ (using a seawater density of 1028.7 L kg⁻¹) vs PIC in Exp B. Full and dashed lines on 195panel C model the expected co-evolution of A_T and PIC during calcite dissolution. Significant differences (*Mann*-196Whitney p < 0.05) are marked with an "a" between t0 and t10, "b" between ATM and HP. Circled data points 197point at the ATM2 and HP3 replicate incubations displaying values out of the expected range for calcite 198dissolution from the measured t0 values shown in panel C.

199 Discussion

200Effect of hydrostatic pressure increase on the prokaryotic community

201During the present *E. huxleyi* aggregate biodegradation experiment, initial natural mesopelagic 202Mediterranean Sea prokaryotic concentrations increased by up to two orders after 10 days in 203controlled temperature and hydrostatic pressure (HP) conditions simulating particle sinking from 204200m to 1700m depth at a rate of 150 m d⁻¹. Such a high prokaryotic development has recently been 205observed with deep Mediterranean seawater natural prokaryotic communities amended with labile 206marine broth, both under atmospheric (ATM) and HP conditions (Wannicke et al., 2015). This might 207therefore indicate that *E. huxleyi* aggregates constitute a relatively labile substrate for mesopelagic 208Mediterranean prokaryotic communities to grow on.

209The natural prokaryotic diversity was dominated by *Bacteria*, with a lower proportion of *Archaea* at 210the end of the incubation, as previously observed in similar experiments (Tamburini et al., 2006,

2112009), despite a significant increase in absolute abundance. The initial fraction of *Bacteria*, 212Crenarchaea and Euryarchaea were lower in the present experiment (30%, 7% and 4%, respectively) 213than in previous experiments with 200m depth seawater sampled in different areas of the 214Mediterranean Sea (Tamburini et al., 2006, 2009). Different sea surface mixing conditions at the time 215of the natural prokaryotic community sampling for the different experiments might have resulted in 216different initial community compositions (Pinazo, pers. Com). When focusing on the classes of 217Bacteria generally found associated to marine aggregates (Acinas et al., 1999; Delong et al., 1993; 218Moeseneder et al., 2001; Ploug and Grossart, 1999; Simon et al., 2002; Thiele et al., 2015), only 219Bacteroidetes and Alphaproteobacteria tended to be sensitive to increasing hydrostatic pressure 220conditions (HP). Similar trends were observed during previous particle sinking simulation experiments 221 with natural faecal pellets (Tamburini et al., 2009). The latter authors proposed that the sensitivity of 222Bacteroidetes to pressure may have reduced the amount of particulate carbohydrate hydrolysed. This 223could result in a lower production of dissolved carbohydrates as substrate for Alphaproteobacteria 224(Tamburini et al., 2009). A similar mechanism might be at play in the present experiment with E. 225huxleyi detritus, which is supported by the observation of large sticky particles and aggregates under 226HP condition only, as in Tamburini et al. (2009)(Suppl. Fig. 1). These large sticky aggregates might 227 indeed be the result of less degradation of transparent exopolymeric particles (TEP) under HP relative 228to ATM (Tamburini et al., 2009), associated with the fact that more dissolved carbohydrates may have 229been released under HP by E. huxleyi, possibly resisting prokaryotic degradation (Nanninga and 230Tyrrell, 1996) and scavenged by TEP (Engel et al., 2004).

231Interestingly, we observed that prokaryotes grew significantly more on > 2 μ m aggregates under 232increasing pressure (HP) than at atmospheric pressure (ATM). This pattern was not observed in a 233previous similar controlled experiment studying the biodegradation of natural aggregates collected in 234the Mediterranean Sea (mainly faecal pellets) during 6.5 days, although the formation of visible sticky 235aggregates had also been observed under HP (Tamburini et al., 2009). The fact that the present 236simulation lasted longer and reached higher pressure levels could explain that the pattern observed

237here may not have been detected in the previous study. Finally, it also is also possible that 238coccolithophorid cells are more sensitive to pressure than faecal pellets or diatom aggregates 239(Tamburini et al., 2006, 2009). Hence, more labile intracellular organic matter would be released in 240the aggregates surroundings, favouring higher prokaryotic growth. To verify the latter hypothesis, we 241analysed the aggregate lipid contents and calculated a lipolysis index.

242**Degradation of E. huxleyi aggregates organic matter**

243Assuming a prokaryotic cellular carbon content of 12.4-53.0 fg C cell⁻¹ (Fukuda et al., 1998; Kogure 244and Koike, 1987), prokaryotes made up for 0.05-0.21% of the TOC at t0, which increased to 1.84-24511.44 % at ATMt10 and 2.32-11.94 % at HPt10. Heterotrophic growth could have been supported by 246the dissolved organic matter present at the start of the incubation, and/or by the degradation of E. 247huxleyi aggregates. A lipolysis index defined as the ratio of metabolites resulting from acyl-lipid 248degradation to the sum of acyl lipids can be used to characterize the degradation stage of labile 249 organic matter (Goutx et al., 2000, 2003). Although the lipid index in E. huxleyi aggregates increased 250significantly between the beginning (t0: 0.72±0.06) and the end of the experiment (ATMt10: 2512.07 \pm 1.22, HPt10: 1.93 \pm 0.32), there was no significant pressure effect (MW p = 0.51). However, the 252lipid fraction of E. huxleyi detritus organic carbon decreased significantly after ten days under HP, 253contrasting with previous observations using natural aggregates, mainly consisting in faecal pellets 254(Tamburini et al., 2009). In addition, compared to the lipid composition at ATM, membrane lipids 255(phospholipids, sterols, MGDG) disappeared from the particulate fraction significantly more under HP 256than at ATM, which could indicate higher E. huxleyi cell lysis under pressure (Baldi et al., 1997). E. 257huxleyi aggregates might therefore be more vulnerable to degradation under pressure than faecal 258 pellets.

259**Changes in mineral content**

260Particulate inorganic carbon (PIC) decreased more under HP than at ATM, suggesting that more 261calcite dissolution may have occurred under increased hydrostatic pressure (Fig. 4). Enhanced PIC 262dissolution under HP has been observed in a previous study with diatom-mineral (carbonate-263kaolinite-smectite) aggregates, together with a decreased POC decay rate (de Jesus Mendes and 264Thomsen, 2012). We here show that increased CaCO₃ dissolution can occur in calcite-sheathed cell 265 aggregates sinking through mesopelagic waters in the presence of natural prokaryotic communities. 266According to the stoichiometry of calcification, two moles of total alkalinity are produced per mole of 267CaCO₃ dissolved (approximated as moles of PIC in Fig. 4C). However, although there was more calcite 268dissolution under pressure, the increase in total alkalinity was larger at atmospheric pressure. 269Nonetheless, values were within the range expected for calcite dissolution, considering the starting 270total alkalinity and PIC conditions (Fig. 4C), except for the two outliers circled in Fig. 4 (namely ATM2 271and HP3 incubations, also circled in Fig. 2). Processes other than CaCO₃ dissolution may explain the 272outlying PIC-A_τ pairs measured in replicates ATM2 and HP3, which respectively also displayed the 273 lowest prokaryotic abundance (1.77 10⁶ cell mL⁻¹) and POC concentration (30.7 μmol L⁻¹). 274Total alkalinity may indeed be affected by microbial aerobic and anaerobic organic matter 275remineralization/transformation through the consumption or release of ammonium, nitrate, 276phosphate and sulphate (Gattuso et al., 1999). The processes of respiration, nitrification and sulphate 277 reduction have adverse effects on protons release/uptake. However, the difference of +31 μ mol L⁻¹ A_T 278between the value measured in ATM2 and the value expected from the triplicate t0 PIC 279concentrations cannot be solely explained by dissimilation of ammonium by organic matter 280 respiration. This process would result in a A_{τ} increase of 0.17 moles per mole of TOC respired (Gattuso 281et al., 1999), when the difference in TOC between t0 replicates and ATM2 was only 5.4 to 21.6 µmol L⁻ 282¹ (which was comparable to the other 2 ATM replicates). Since the t0 replicates consisted in 283independent splits of the initial aggregate-seawater mixture that was further split into ATM and HP 284replicate incubations, another explanation for this outlying pair could be that the ATM2 incubation 25 13 285may have started with a higher PIC concentration around $65~\mu M$ (higher concentration of detached 286coccoliths).

287Similarly, the -31 μ mol L⁻¹ difference in A_T between the value measured in HP3 and the expected value 288cannot be explained solely by the dissimilation of nitrate by organic matter respiration. Such a 289process would lead to a decrease in A_T of 0.13 moles per mol of CO_2 released (Gattuso et al., 1999), 290whereas the difference in TOC concentration between t0 replicates and HP3 comparable to that with 291ATM2 was observed (Fig. 2). Given the large A_T variability observed between the independent splits 292sampled at t0, it is more likely that the HP3 incubation started with a much lower A_T than the other 293HP replicates (around 2635 μ mol L⁻¹, or 2561 μ mol kg⁻¹ at 50 μ M PIC).

294Implications for the carbon pump and the ballast hypothesis

295In the Mediterranean Sea, CaCO₃-sheathed coccolithophores can dominate the phytoplankton 296community (Ignatiades et al., 2009). They are highly diverse, including large numbers of the 297ubiquitous species *E. huxleyi* (Cros and Fortuño, 2002; Knappertsbusch, 1993). Several studies have 298stressed the need for controlled experiments to validate the hypothesis that CaCO₃ may be a more 299efficient POC "carrier" than biogenic silica (De La Rocha and Passow, 2007; Passow and De La Rocha, 3002006; Passow, 2004). Our experiment contributes to a better understanding of this issue. In contrast 301with previous experiments using natural marine snow/faecal pellets (Tamburini et al., 2009; Turley, 3021993), organo-mineral aggregates (de Jesus Mendes et al., 2007) or diatom culture aggregates (de 303Jesus Mendes and Thomsen, 2012; Tamburini et al., 2006), CaCO₃-containing *E. huxleyi* culture 304aggregates appear more sensitive to the increase in hydrostatic pressure. Our results suggest that 305increased pressure results in higher cell lysis and CaCO₃ dissolution relative to ATM. The protecting 306effect of coccoliths may therefore not remain effective throughout the water column, since coccoliths 307might detach from cells and dissolve more readily under pressure. Calcium carbonate may therefore 308not be an efficient POC carrier to the deep ocean in the form of intact coccolithophorid cells.

309Coccolithophores are the main producers of the pelagic CaCO₃ biogenically produced in the global 310ocean (0.8–1.4 Gt year⁻¹) (Feely et al., 2004) and may contribute to about 50% of pelagic CaCO₃ 311deposition in sediments (Broecker and Clark, 2009; Poulton et al., 2007). Yet, the processes and rates 312of coccolith CaCO₃ dissolution in the upper layers of the ocean are poorly identified and quantified. 313Since calcification decreases the ocean CO₂ uptake capacity by reducing total alkalinity, the depth of 314calcite dissolution likely has important implications on the net gradient in pCO₂ between the ocean 315and atmosphere (Antia et al., 2001; Barrett et al., 2014).

316The dissolution of pelagic CaCO₃ particles has long been assumed to occur primarily at great depths 317below the calcite and aragonite saturation depths (Broecker, 1977). In fact, up to 80% of the CaCO₃ 318that is exported out of the surface ocean might actually dissolve in the upper 1000 m, well above the 319lysocline (Chung et al., 2003; Feely et al., 2004; Iglesias-Rodriguez et al., 2002; Jansen and Wolf-320Gladrow, 2001; Milliman et al., 1999; Wollast and Chou, 1998, 2001). Grazing by microzooplankton 321has been shown to play an important role in coccolithophorid calcite dissolution (Antia et al., 2008; 322Harris, 1994; Jansen and Wolf-Gladrow, 2001; Milliman et al., 1999). In contrast, carbonate 323dissolution on individual calcifying cells by microbes was deemed negligible (Bissett et al., 2011), but 324the presence of chemical gradients within sinking aggregates (Ploug and Bergkvist, 2015) may create 325micro-environments with lower pH and calcite saturation state that could facilitate (a)biotic 326dissolution. We show here that calcium carbonate dissolution in the presence of natural prokaryotic 327communities might be higher under increasing hydrostatic pressure. Pressure was recently suggested 328to affect dissolution kinetics over and above its influence on the saturation constants (Dong et al., 3292016). The combined effect of pressure and microbial activities therefore remains to be assessed in a 330more mechanistic manner.

3310n the other hand, we show that increasing hydrostatic pressure potentially triggers the 332development of Mediterranean mesopelagic prokaryotic communities on *E. huxleyi* aggregates (as 333opposed to diatom detritus or natural aggregates, Tamburini et al., 2006, 2009), as well as 334aggregation. The study by Tamburini et al. (2009) had already shown that particulate

335carbohydrates:POC and Transparent Exopolymer Particles (TEP):POC ratios could be significantly 336higher in conditions of increasing hydrostatic pressure. The prokaryotes themselves may promote 337aggregation, through the production of sticky extracellular muco-polysaccharides (Biddanda and 338Pomeroy, 1988; Biddanda, 1986; Turley, 1992). Moreover, coccoliths have been proposed as 339aggregation nuclei for colloidal DOM, therefore offering organic matter protection from 340solubilization/remineralization (Engel et al., 2009a). The increased aggregation observed under 341pressure in our reaction vessels could therefore compensate the loss of mineral ballast at depth, by 342increasing aggregate sizes and/or mineral scavenging potential. To what extent pressure-induced 343aggregation balances calcite dissolution *in situ* remains to be observed and assessed, by including 344model calculations, for instance.

345**Methods**

346Production of Emiliania huxleyi aggregates and experimental design

347The TW1 *E. huxleyi* strain (Algobank-Caen) was maintained in exponential growth phase in K/2 348medium (Keller et al., 2007) at 21°C, under a 12-12h light-dark cycle (intensity: 4300 Lux). The culture 349was supplemented with Provasoli's antibiotic concentrate (3% final concentration, Sigma) until the 350bacterial numbers reached a minimal proportion of one to two prokaryote per phytoplankton cell. 351The antibiotic solution was then removed for two culture passagings before *E. huxleyi* cells in 352exponential phase were concentrated by centrifugation (3000 rpm, 15 min) (Houdan et al., 2005; 353Sawada and Shiraiwa, 2004). Fresh detritus was obtained according to a procedure modified from 354(Tamburini et al., 2006) by exposing cell culture to 55 °C for 10 min only, which avoided cell disruption 355observed during freeze-thaw cycles. The detritus aliquots were immediately diluted in sterile-filtered 356or GF/F-filtered Mediterranean Seawater (for Exp A and Exp B, respectively, see details below) to 357reach a final concentration of about 600 μg POC L⁻¹ (cell density of 5000 cell mL⁻¹) close to natural 358coccolithophore bloom (Holligan et al., 1993). Aliquots were sampled at the starting point of the 359incubations (t0) for the various microbiological and chemical parameters described below.

360Incubation in the PArticle Sinking Simulator (PASS)

361The two experiments (Exp A and Exp B) were conducted in March 2007 using Mediterranean 362seawater (13°C) collected at 200 m depth, above the Marseilles Trench (43°01'48 N-05°11'158 E, 363600m bottom depth). In Exp A, seawater was left in the dark at 4 °C for 15 d and gently filtered 364through 0.2 μ m-porosity polycarbonate filters to subtract the prokaryotic community. In Exp B, fresh 365natural prokaryotic assemblages <0.7 μ m were recovered in the filtrate of a gentle filtration onto pre-366combusted GF/F filters of the freshly collected seawater sample.

367Exp A and Exp B were incubated for 10 d at *in situ* temperature (13°C). The PASS system was used 368following a procedure described in (Tamburini et al., 2009), except a peristaltic liquid dispenser 369(Jencons Scientific Ltd.) was used to obtain homogeneous replicate aliquots. Aliquots were incubated 370in 500 mL high-pressure bottles (HPBs) where pressure increased linearly using a piloted pressure 371generator in order to reproduce the increase in pressure that occurs when particles sink. 372Coccolithophores detritus was kept in suspension by regular half-revolutions of the HPBs in water 373baths kept at *in situ* temperature (Tamburini et al., 2006). For Exp A, two duplicate HPBs were kept at 374atmospheric pressure and two duplicate HPBs were exposed to increasing pressure (1.5 MPa d⁻¹ 375corresponding to a sinking rate of 150 m d⁻¹). This is applicable to mid-density particles, as discussed 376in (Tamburini et al., 2006) and corresponds to sinking rates observed in previous *E. huxleyi* aggregates 377studies (Engel et al., 2009b; Iversen and Ploug, 2010; Ploug et al., 2008). Exp B was performed with 378triplicate HPBs. After 10 days of incubation (HPt10, corresponding to a depth of 1700 m), the 379pressurised coccolithophores detritus dilutions were decompressed and sampled as described in 380(Tamburini et al., 2009) for the parameters described below.

381**Microbiological analyses**

382Samples for prokaryotic abundances (50-135 mL) were fixed using 0.2- μ m filtered formaldehyde (2% 383final concentration), kept at room temperature for 15 min, and subsequently stored at 4°C in the dark 384for around 12h. For total (>0.2 μ m) and free-living (0.2-2.0 μ m) prokaryotic abundances, samples 33

were filtered onto 0.2- μ m-pore-size polycarbonate filters supported by 0.45- μ m-pore-size cellulose 386nitrate filters directly, or after filtration by gravitation onto 2.0- μ m-pore-size polycarbonate filters, as 387in (Tamburini et al., 2009). Cells were stained with diamidinophenylindole (DAPI) according to (Porter 388and Feig, 1980). The fraction of "attached" or large prokaryotes >2 μ m was calculated by subtracting 389values of the 0.2-2.0 μ m fraction from the total fraction.

390Samples for catalyzed reporter deposition coupled with fluorescence *in situ* hybridization (CARD-FISH) 391were treated as in (Tamburini et al., 2009): briefly, samples were filtered onto 0.2-μm-pore-size 392polycarbonate filters as for the total prokaryotic abundances. Filters were washed twice with 0.2-μm 393filtered MilliQ water, air-dried, and stored in scintillation tubes at -20°C until analysis. Cells were 394embedded in low-gelling point 0.1% agarose (Sigma), dried at 37°C for 10min, and dehydrated with 39595% ethanol. Bacterial cell walls were permeabilized by subsequent treatments with lysozyme (10 mg 396ml⁻¹, Sigma)(Amann et al., 1995) and with achromopeptidase (60 U, Sigma)(Sekar et al., 2003). 397Archaeal cell walls were permeabilized with proteinase K (0.2 ml ml⁻¹, Fluka)(Teira et al., 2004). Acid 398treatment in 0.01M HCl was performed to denature endogenous peroxydase before hybridization 399with the horseradishperoxydase (HRP)-labeled oligonucleotide probes listed in Table1: for a detailed 400protocol see (Tamburini et al., 2009). The filter portions hydridized with each probe were mounted in 401a Citifluor:Vectashield:DAPI mix, and DAPI and CARD-FISH-stained cells were quantified under an 402Olympus BX61 microscope equipped with a 100-WHg-lamp and appropriate filter sets according to 403(Tamburini et al., 2009).

Table 1: 16S rRNA-targeted oligonucleotide probes used in this study.

	Sequence(5'-3')	Target organisms	%	References
Probe			Formamide	
Eub338/I	GCT GCCTCCCGTAGGAGT	Domain ofbacteria	55	(Amann et al., 1990)
EubII	GCA GCCACCCGTAGGTGT	Domain ofbacteria	55	(Daims et al., 1999)
EubIII	GCT GCCACCCGTAGGTGT	Domain ofbacteria	55	(Daims et al., 1999)
ALF968	GGTAAGGTTCTGCGCGTT	Most of Alfa of	55	(Manz et al., 1992)
		Proteobacteria		
GAM42a ^a	GCC TTCCCACATCGTTT	Gamma-Subclass of	55	(Manz et al., 1992)
		Proteobacteria		
CF319a	TGGTCCGTGTCTCAGTAC	Cytophaga-	55	(Manz et al., 1996)
		Flavobacter cluster		

Cren537	TGACCACTTGAGGTGCTG	Crenarchaea	20	(Teira et al., 2004)
Eury806	CACAGCGTTTACACCTAG	Euryarchaea	20	(Teira et al., 2004)
NegControl	TAGTGACGCGCTCGA	For non-specific	55	(Karner and Fuhrman, 1997)
		probe binding		

 $40\overline{5}^a$ Including an unlabeled competitor probe BET42a (5'-GCCTTCCCACTTCGTTT-3'), see (Manz et al., 1992) for 406details.

409All glassware was pre-combusted before use. Additionally, glassware was rinsed with 1 N HCl and

407

408Chemical analyses

410Milli-Q water after each sample. All plastic ware was rinsed with 1 N HCl and Milli-Q water and 411changed for each sample. Sample aliquots for particulate organic carbon (POC, 50-104 mL) and lipids 412(P-lips, 140-500mL) were filtered onto pre-combusted 0.7 µm GF/F filters (25 mm filter diameter) 413under a low vacuum (<50mm Hg) and preserved at -20°C or in liquid nitrogen, respectively. 414Filters for particulate total carbon (PtotC) and POC were divided in two and analyzed as in Raimbault 415et al. (2008). The filter portions used for PtotC were analyzed directly, while filter portions for POC 416 measurement were acidified with 100 μ L H_2SO_4 (0.5N) and subsequently analyzed using high 417combustion (1000°C) on a mass spectrometer (CN-Integra tracer-mass). The concentration of 418particulate inorganic carbon (PIC) was calculated from the difference between PtotC and POC. 419Lipids were extracted from filters using a Bligh and Dyer (1959) protocol, and lipid classes were 420separated on chromarods and quantified on a thin layer chromatography/flame ionization detection 421(TLC/FID) latroscan apparatus model MK-6s (latron,Tokyo) coupled to a PC equipped with a Chromstar 4226.1 integration system (Bionis, Paris), as in Goutx et al. (2007). The separation of the 16 following lipid 423 classes was obtained: neutral lipids including hydrocarbons (HC), sterol esters co-eluting with wax 424esters (WE), ketones (KET), triacylglycerols (TG), free fatty acids (FFA), alcohols (ALC), sterols and 425diglycerides (DG); chloroplast lipids containing pigments (Pig), glycolipids and 426monogalactosyldiacylglycerol (MGDG); monoacylglycerols (MG); and phospholipids (PL) including 427diphosphatidylglycerides co-eluting with phosphatidylglycerides (DPG+PG), phosphatidylethanol-428amine (PE) and phosphatidylcholine (PC). A lipolysis index was calculated as in Goutx et al. (2000, 4292003), defined as the ratio of the sum of the metabolites (DG, MG, FFA, ALC) to the sum of the intact 430acyl-lipids (TG, PE, PI, DPG+PG, PC, WE, MGDG).

431TOC concentrations were obtained by adding POC to DOC values. Samples for DOC were obtained and 432preserved as in Tamburini *et al.* (2009). The DOC measurements were carried out in the laboratory at 433the Pisa section of the Institute of Biophysics (IBF), CNR, with a Shimadzu TOC-VCSN. Samples were 434acidified with 2N HCl and sparged for 3 minutes with CO_2 -free ultra-high purity air in order to remove 435inorganic carbon. From 3 to 5 replicate injections were performed until the analytical precision was 436lower than 1%. The system blank was measured every day at the beginning and the end of analyses 437using low-carbon water (2-3 μ mol L⁻¹ C) produced by a Milli-Q system. Measurement reliability was 438assessed twice a day by comparison of data with DOC Consensus Reference Waters (CRM; Hansell, 4392005) (measured concentration = 42.7 μ mol L⁻¹, nominal concentration: 41-44 μ mol L⁻¹). For further 440analytical details refer to Santinelli et al. (2013)

4415amples for total alkalinity (A_T) were filtered on GF/F membranes (duplicate 50 mL filtrate aliquots 442from the PtotC and POC GF/F filtrations), immediately poisoned with 100 μ l of a 50% saturated HgCl₂ 443solution (37 g HgCl₂ in 1 L distilled water) and stored at 4°C in the dark pending analysis. A_T was 444determined on 20 mL subsamples using a home-made titration system composed of a 20 mL open 445temperature-controlled titration cell (kept at 25.0°C), an Orion 8103SC pH electrode calibrated on the 446National Bureau of Standards scale and a computer-driven Metrohm 665 Dosimat titrator, 447Switzerland. The seawater samples and the acid titrant (0.1 N HCl) were kept at a constant 448temperature of 25°C. Samples were weighted with a precision of 0.01 g before the titration to 449determine their exact volume from temperature and salinity. A_T was calculated from a Gran function 450applied to pH variations from 4.2 to 3.0 as a function of added volume of HCl. A_T measurements had a 451reproducibility of 3 μ mol kg⁻¹.

452**Statistical analyses**

453Differences in prokaryotic abundances (N = 3 replicate incubation bottles per condition) were 454analyzed by the non-parametric pairwise Mann-Whitney test on raw data (i.e. individual replicate 455values), due to the low number of replicates, using the Statistica 6 software. Unless otherwise stated, 456data are reported as averages and standard deviation (SD) and statistical difference was accepted at p 457 \leq 0.05.

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Supplementary data

Supplementary Table 1: Lipid proportions (% Total Particulate Lipids) in the *E. huxleyi* culture at day 4 737(Bell and Pond, 1996) and during Exp B. Numbers 1, 2 and 3 stand for replicate incubations. 738Chloroplastic: pigments + monogalactosyldiacylglycerol, Phospholipids: diphosphatidylglycerides + 739phosphatidylglycerides + phosphatidylethanolamine + phosphoinositol + phosphatidylcholine, ST: 740sterols, Neutral: wax esters + methyl esters + ketones + triglycerides + Free Fatty Acids + fatty 741alcohols, Metab: Monoglycerides + DiGlycerides, HC: Hydrocarbons.

	Chloroplastic	Phospholipids	Sterols	Neutral	Metab	HC
E. huxleyi 4 days (Bell and Pond, 1996)	41.3 ± 0.8	21.5 ± 0.4	3.6 ± 0.4	23.3 ± 1.3	ND	9.4 ± 0.3
t0-1	39.3	22.5	2.4	23.2	4.0	7.6
t0-2	44.8	18.2	3.3	17.9	5.5	8.4
t0-3	46.5	16.6	3.8	19.2	4.4	8.1
ATMt10-1	10.8	5.0	0.0	50.3	0.0	31.2
ATMt10-2	32.5	3.4	3.1	13.6	0.0	43.6
ATMt10-3	20.7	5.8	2.5	37.7	0.0	20.6
HPt10-1	22.5	3.6	1.9	37.2	5.2	25.6
HPt10-2	19.8	6.8	0.0	43.1	0.0	21.0
HPt10-3	23.0	4.6	0.0	35.3	0.0	25.3

Supplementary Figure 1: Photographs of the GF/F filters from Exp B at ATMt10 (A, B, C) and HPt10 (D, 743E, F). A-C, and B-F are the replicate incubations. Arrows point at some of the large visible aggregates 744in HPt10 incubations.

