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► To cite this version:

Camille Banc, Denise Blanc, Mathieu Gautier, Maria Lupsea-Toader, Rémy Gourdon. Modelling of trace metals leaching behaviour in sludge deposits with high organic contents.. European Meeting on Environmental Chemistry, Dec 2018, Clermont, France. hal-01973320

HAL Id: hal-01973320 https://hal.science/hal-01973320

Submitted on 8 Jan 2019

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Modelling of trace metals leaching behaviour in sludge deposits with high organic contents



OM -

62%

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sod

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he matrix considered in this study comes from a small wastewater treatement plant called Vertical Flow Constructed Wetland. These are increasingly used in France with some 5000 plants already installed in 2018. Sludge deposits accumulated at the top of the filter are composed by organic matter but also by inorganic pollutants such as trace metals. Improving $\overline{\mathbf{\sigma}}$ the understanding of the interactions between minerals and organic matter is $\mathbf{\breve{A}}$ necessary to predict the pollutant retention / release dynamics.

Experimental

<u>Results of elemental analysis</u>

Others analysis : . Acid and Base neutralisation (CENTS/TS 14429)

Concentrations

(mg/kgDM)

1219

623

64.2

57.4

30.5

1.7

Definition of a solid organo-mineral as- \rightarrow semblage representative of the matrix;

Evaluate the impact of different scenarios on the release and speciation of pollutants into the matrix;



Better predict the behavior of trace \rightarrow metals.

Ca²⁺

Fe²⁺



Na⁺





lonic

Strength

Phenolic site

Humic substance

Electrostatic part

Depends on coulombic interactions that results from the **negative charge density** present on the molecule. Model VII considers that the charges of humic substances are located only on the surface.

'थ⁄ु Sequential extraction (method from Tessier et *al.*, 1979) . X-Ray diffraction

Trace

elements

Zn

Cu

Pb

Cr

Ni

Cd

Pure mineral phases

NTOT

By their dissolution / precipitation properties, these phases will influence the concentration available in major elements but also in pollutant. The following example shows the CaCO, dissolution reaction :

 $CaCO_{3} + H^{+} = Ca^{2+} + HCO^{3-}$ \rightarrow Log_k 1,85



Mineral assemblage

Zn²

PhreeQc software (Parkhurst and Appelo, 1999) allows to simulate the dissolution of the different mineral phases implemented. For this, the software uses the thermodynamic data available in the database that has been specified. In our case, pure and mineral-bearing

Mineral-bearing phases

These matrices are also composed of mineral surfaces that can adsorb / release some pollutants through their binding sites. Example of Iron oxyhydroxides :

> . Hfo_sOH + Cd^{2+} = Hfo_sOCd⁺ + H⁺ $\rightarrow \log_k 0.47$

> . Hfo_wOH + Cd^{2+} = Hfo_sOCd⁺ + H⁺ $\rightarrow \log_k -2.91$

Evolution of Zn-leaching as a function of pH

pН



phases were implemented.

Conc

ptu

mode

Thresolds limits

(mg/kgDM)

3000

1000

800

1000

200

20

Organic assemblage

Resul

12

Model VII, developed by Tipping and its teams in 2011, is a database for simulating the adsorption capacity of humic substances (HS) via carboxylic and phenolic sites. The HS affinity for cations has two components: the intrinsic part and the electrostatic part.

Evolution dissolved organic carbon leaching <u>concentration as a function of pH</u>

Intrinsic part

It takes into account the specific binding properties of humic substances. 50 different sites are specified including mono, bi and tridentate site. Example of binding of Lead with a Fulvic acid site :

> $Fa_a + Pb + 2 = Fa_aPb +$ $\rightarrow \log_k 2.15$

Conclusions

Combining PhreeQc and Model VII allows to \rightarrow reconstruct a complex organo mineral assemblage;

The implementation of humic substances to the model provides good predictions of trace metals behavior;

Test on other type of matrices such as sedi- \rightarrow ment from stormwater management pond.

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